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Penrose triangles of the fossil-to-bio-based transition

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Received 16th July 2017, Accepted 25th July 2017

DOI: 10.1039/c7fd00178a

Transition within the chemical industry from fossil to green feedstocks is a complex process characterized by the generation of commercially viable feedstock–process–product triangles. The research in this area encompasses a great diversity of relevant topics. A number of those topics have been addressed within this volume of *Faraday Discussions* and are summarized in this paper. They are categorized and discussed along with seven general questions arising from the feedstock–process–product triangles. Opportunities are identified that should make more of these triangles technically and economically feasible. The future role of renewable electricity as the primary energy source for the bio-based industry is emphasized.

Introduction

The transition from a fossil-based to a bio-based chemical industry is a global process occurring in the context of sustainable development of our planet. Although the process itself is inevitable and irreversible, the vast majority of researchers working in the field agree that the pace at which it is proceeding is still far from satisfactory. The essence of the fossil-to-bio transition lies in fact in the successful generation of commercially viable feedstock–process–product triangles that offer clear benefits to the stakeholders concerned (farmers, industry, governments, end-users, consumers, *etc.*). Unfortunately, for the time being, many of these triangles still appear to be a kind of “Penrose Triangle”† (Fig. 1), described by Lionel and Roger Penrose¹ as “impossibility in its purest form”. What challenges need to be met and what questions need to be answered in order to make more of these triangles viable? The Faraday Discussion “Bio-Resources: Feeding a Sustainable Chemical Industry” has allowed debate of the

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† Triangular object made of three straight bars that have a square cross-section, which meet at right angles at the vertices of the triangle they form. Also called an “impossible tribar”, since such a combination of properties cannot be realized by any three-dimensional object in ordinary Euclidean space. First created by the Swedish artist Oscar Reutersvärd and described by the British psychiatrist Lionel Penrose and his son, mathematician Sir Roger Penrose.¹





Fig. 1 Penrose triangle for a fossil-to-bio transition.

latest developments in science and technology that address those questions and challenges and create new opportunities.

From fossil-based to bio-based: important differences

Before more deeply discussing the challenges occurring in the transition from a fossil-based to bio-based industry, it is good to realize some important characteristic features and differences between the two types of industry. These features and differences are briefly described below. Some of them may present additional challenges when it comes to development of a viable bio-based process.

Feedstock diversity and variability

Generally speaking, the variability of bio-feedstocks in terms of chemical composition and physical properties is much greater than in fossil fuels. Although both natural gas and crude oil stocks occur in different grades and compositions, the feedstocks for the bioprocesses are more diverse ranging from crops, through to wood, algae, industrial wastewaters, municipal organic wastes, animal wastes, food wastes, *etc.*² Furthermore, within biologically the same feedstock differences in the composition and properties are often seen, depending on the climate and water/soil conditions in which the feedstock has been grown. The diversity of the bio-feedstocks is usually considered a drawback because of the increased flexibility of the processing plant that is required to deal with varying feedstocks. However, it can also be seen as an opportunity to bring a more diverse range of products to the market.

Diversity of processes and operations

Usually, a biorefinery includes more diverse types of processes and operations than an oil refinery. Next to “conventional” chemical and catalytic processes that occur in a liquid or gas phase, biorefineries often include mechanical/chemical pre-processing of solid feedstocks, aerobic and anaerobic fermentations, enzymatic reactions, plus a range of microbial processes.³ This brings additional challenges related to process plant and site integration.

Transporting feedstocks

Contrary to the transport of gas and oil, which largely occurs *via* pipeline infrastructures, the vast majority of bio-feedstocks need to be transported on land or



water from the place of harvest to the processing location. Two alternative transport scenarios (centralized plant *versus* decentralized plant) have been considered in the literature,⁴ but for both cases transport remains an important cost.

Feedstock *versus* market – regionalization

Since the long-distance transport of biomass is troublesome and expensive, regionalization of bio-based production will occur. Contrary to the current situation, where many oil refineries are built and operated in regions and countries that do not have oil deposits themselves, with a bio-based economy, countries lean in biomass, such as the Gulf states, will have to import bio-based platform chemicals from elsewhere. Furthermore, since the product portfolio of a bio-refinery depends strongly on the type of feedstock it converts,^{4,5} and the feedstocks often have local character, the manufacturing of certain types of bio-based chemical products will occur locally.

Susceptibility to the force of nature

Lastly, but definitely not least, natural disasters such as floods, droughts, wildfires or agriculture and forest pests can affect the availability of the biomass in a given region and consequently the prices of the respective bio-based products, both regionally and globally. In the long term, climate change will also influence the dynamics of the bio-based economy.

Questions arising from feedstock–process–product triangles

Numerous questions arise within each feedstock–process–product triangle and these questions need to be adequately addressed. The papers presented in this volume of Faraday Discussions can be categorized according to seven general questions dealing with various elements of the triangle, located either at its vertices or between them.

Question 1 (feedstock): how to improve the analytical tools and develop standardized methods for the evaluation of feedstocks?

The need for improvement of the current analytical tools has been addressed by Galkin *et al.* (DOI: 10.1039/C7FD00046D), who have pointed out the shortcomings of the current analytical techniques used for biomass evaluation in the pulping industry. The authors postulated development of analytical tools for targeting all the wood components, including the generated fractions, as well as standardized methods for evaluating and reporting yields. In another paper by Hayes *et al.* (DOI: 10.1039/C7FD00081B), near-infrared spectroscopy was proposed as a suitable method for rapid, low-cost analysis of the major lignocellulosic components of waste paper/cardboard.



Question 2 (feedstock ↔ product): how to select a suitable feedstock for the desired products?

The importance of this issue was demonstrated in the paper by Wood *et al.* (DOI: 10.1039/C7FD00044H) where eight different feedstocks (hardwood, softwood, cereal straws and dicotyledonous crops) were exposed to microwave-assisted liquid hot water pre-treatment. The paper shows that fundamental differences in the cell wall composition resulted in considerable differences in feedstock suitability with regard to the quantity of released products.

Question 3 (feedstock ↔ process): how to pretreat and process feedstocks?

Selection of a pre-treatment technology is of fundamental importance to the entire process economics. Currently, steam-explosion is the most commonly used method for the pre-treatment of the lignocellulosic biomass. Seidel *et al.* (DOI: 10.1039/C7FD00066A) have pointed out that the explosive decompression at the end of this step could enhance the enzymatic cellulose digestibility of hardwood and herbaceous plants. Weigand *et al.* (DOI: 10.1039/C7FD00059F) proposed the use of protic low-cost ionic liquids in order to increase enzymatic glucose yields from willow biomass. In addition, Ferrini *et al.* (DOI: 10.1039/C7FD00069C) have shown the importance of solvent effects during functionalization of the propyl side-chain in lignin oil obtained from the deconstruction of lignocellulosic materials *via* catalytic upstream biorefining. Microwave-assisted acidolysis of the lignocellulosic biomass presents an effective approach to produce high purity lignin and fermentable chemicals from softwood, as demonstrated by Zhou *et al.* (DOI: 10.1039/C7FD00102A). Importantly, the lignin was isolated largely intact and retained the original structure of the native lignin in the feedstock. Interesting findings with regard to microalgae biomass were presented by Zhou *et al.* (DOI: 10.1039/C7FD00065K). These authors pyrolyzed raw biomass samples along with samples pre-treated through extraction of the lipids or saccharides. It appeared that fractional pyrolysis of the pretreated microalgae not only increased the bio-oil yield but also improved its quality.

Question 4 (feedstock ↔ process ↔ product): how to develop a new process?

This short question actually hides a multitude of situations, strategies and approaches. In some cases, plant flexibility is the central issue when feedstocks (and sometimes also products) vary. In other cases, the optimum process configuration needs to be found for a narrowly defined feedstock and product. Lapkin *et al.* (DOI: 10.1039/C7FD00073A) have proposed an interesting generic method for automation of route identification and optimisation based on data-mining and network analysis. The method was applied to generate multiple possible reaction routes for converting limonene into paracetamol. In the future, this approach should enable rapid concurrent optimization of the reaction network and the corresponding processes. The paper by Coma *et al.* (DOI: 10.1039/C7FD00070G) focusses on flexibility issues during development of a process for converting a highly variable feedstock (organic waste) into platform chemicals. The key to success is having flexible anaerobic fermentation and hydrothermal processes that can treat complex biomass as a whole to obtain a range of products within an integrated biorefinery concept. Cárdenas-



Fernández *et al.* (DOI: 10.1039/C7FD00094D) have presented a concept for developing an integrated biorefinery to convert sugar beet pulp into chemicals and pharmaceutical intermediates. The process is based on steam explosion (thermal hydrolysis) of wet sugar beet pulp, followed by bioethanol fermentation, enzyme-membrane fractionation of the sugar beet pectin, and bioconversion of D-galacturonic acid (transaminase) and L-arabinose (transketolase). Another integrated process, to synthesize 1,5-pentanediol (PDO) and 1,6-hexanediol (HDO) from lignocellulosic biomass, has been proposed by He *et al.* (DOI: 10.1039/C7FD00036G). The route goes *via* furfural and tetrahydrofuran-dimethanol, respectively, and techno-economic analysis demonstrated that this approach could produce HDO and PDO at a minimum selling price of \$4090 per ton. In addition, Bajracharya *et al.* (DOI: 10.1039/C7FD00050B) have studied bio-electrochemical conversion of carbon dioxide, as a sustainable feedstock, into chemicals using microorganisms as the catalyst. This microbial electrosynthesis-based process produces acetate as the primary product.

Question 5 (process): how to improve the effectiveness of the reactions?

The most common answer to this question is through development of new, better catalysts. Liu *et al.* (DOI: 10.1039/C7FD00041C) have disclosed Pt nanoparticles supported on bamboo shoot-derived porous heteroatom doped carbon materials as highly active catalysts for controlled hydrogenation of furfural in aqueous media. They have shown that the product selectivity could be easily modulated by controlling the carbonization temperature of the porous heteroatom doped carbon support and the reaction conditions (temperature and H₂ pressure). Albert (DOI: 10.1039/C7FD00047B) has investigated optimization of polyoxometalate catalysts for a fractionated oxidation of lignocellulosic biomass to produce formic acid and high-grade cellulose. One of those catalysts, the Lindqvist-type POM K₅V₃W₃O₁₉, has been shown to catalyse selective oxidation of only the hemi-cellulose and lignin to formic acid, while the cellulose fraction remained untapped. Finally, Huang *et al.* (DOI: 10.1039/C7FD00039A) have investigated the role of acid co-catalysts during selective production of mono-aromatics from lignocellulose over a Pd/C catalyst and they found that HCl and H₂SO₄ showed superior catalytic performances over H₃PO₄ and CH₃COOH.

Question 6 (process): how to improve the product separation?

New and interesting methods for improved product separation have been disclosed within this current volume of Faraday Discussions. Xia and Matharu (DOI: 10.1039/C7FD00035A) have reported, for the first time in the literature, an acid-free subcritical water extraction of pectin from mango peel. Yields of up to 18.34%, with the degree of esterification exceeding 70%, were reported. Lorenz *et al.* (DOI: 10.1039/C7FD00053G) have suggested hydrolyzing cellulose with HCl vapour, in order to facilitate the isolation of cellulose nano-crystals.

Question 7 (product): how to increase the market share of bio-based products?

The enormous variety of bio-based products results in a multitude of answers to the above question and related strategies. According to Bomtempo *et al.* (DOI: 10.1039/C7FD00052A), new platform chemicals derived from biomass should fulfil several criteria, in that they should: “be an intermediate molecule, have



a flexible structure to make a wide range of derivatives possible, be cost competitive at the level of the platform molecule and at the level of the derivatives, be capable of generating scale and scope economies in the value chain, be organized within an innovation ecosystem and have associated well-developed mechanisms of governance". Jin *et al.* (DOI: 10.1039/C7FD00049A) have proposed a 10-step procedure that should be applied when developing new bio-based solvents. The procedure includes various functional, technical and economic criteria that a bio-based solvent needs to fulfil, in order to become a marketable product. Bio-based solvents are an example of bio-based products that have been developed for use in chemical manufacturing. Other examples of such products include catalysts and sorbents. In this context, Golikova *et al.* (DOI: 10.1039/C7FD00042A) have studied a new biocatalyst based on glucose oxidase, while Zuin *et al.* (DOI: 10.1039/C7FD00056A) have presented polysaccharide-derived mesoporous materials (Starbon® materials) as sustainable sorbents for solid-phase extraction of naturally-occurring bioactive phenolic compounds. In the case of bio-based polymeric materials, which are closer to the consumer market, new functionalities or improved properties are usually the deciding factor with respect to their commercial success. An example is the polymeric materials composed only of methylated softwood lignin derivatives reported by Wang *et al.* (DOI: 10.1039/C7FD00083A) that can exhibit better tensile behaviour than polystyrene. As pointed out by these authors, mistaken assumptions from the past about the lignin configuration have hindered the development of these materials for more than 50 years. In another paper, Pérocheau Arnaud *et al.* (DOI: 10.1039/C7FD00057J) have presented novel polyesters with higher glass transition temperatures that are based on branched diols from biomass, while Alberts and Rothenberg (DOI: 10.1039/C7FD00054E) have described the development and commercialization of Plantics-GX: a plant-based biodegradable thermoset plastic.

Michael Faraday and the bio-world

The ground-breaking fundamental works of Michael Faraday on electricity and electromagnetism, originally published as a series of articles in the Philosophical Transactions of the Royal Society of London,^{6,7} have largely contributed to the utilization of electricity in technology. The link between electricity and the bio-world is clear, at least hypothetically. The works of Miller and Urey^{8,9} as well as some later studies (*e.g.* by the group of Bada^{10,11}) point at electricity as the "Mother of All Bio" as an electric spark discharged circa 3.5 billion years ago converted the primordial soup into the first amino acids. In other words: "no electricity = no bio". Back to the 21st century, electricity-based technologies attract more and more interest from researchers working in the area of bio-based processing. Electric fields have been investigated in the context of electro-fermentation (*e.g.* Schievano *et al.*¹² and Chandrasekhar *et al.*¹³). Pulsed electric fields have various bio-related applications including within algae treatment, large-scale biomass (*e.g.* sugar beets) processing or the recovery of valuable products from plants and microorganisms (Golberg *et al.*,¹⁴ Frey *et al.*¹⁵). Algae are harvested using electromagnets and submicron-sized magnetic particles (*e.g.* Xu *et al.*¹⁶ and Cerff *et al.*¹⁷). Induction heating has been proposed for fast pyrolysis of different types of biomass, including rice straw, sugarcane bagasse, coconut shells, Napier grass, pinewood sawdust and sewage sludge (*e.g.* Tsai^{18,19} and Muley²⁰). The heating



rates in those reactors are as high as $500\text{ }^{\circ}\text{C min}^{-1}$, which leads to higher quality bio-oils being obtained and lower carbon deposition. Biomass pyrolysis can also be advantageously performed using microwaves^{21,22} (a conventional microwave oven is a Faraday cage!). Microwave heating is also applicable to other bio-based processes, for instance esterifications²³ or plant extraction,²⁴ while a microwave-induced plasma can be used for gasification of waste biomass to synthesis gas.²⁵ Finally, another form of electromagnetic radiation – light – is widely used in the cultivation of microalgae.^{26,27}

On July 7, 1855, Michael Faraday took a short boat trip on the River Thames. That trip resulted in a famous letter, written on the same day, addressed to the Editor of The Times. In this letter, Faraday expressed his deep concerns about pollution of the river, calling it “a real sewer”. Two weeks later Punch magazine illustrated this with a cartoon of the famous scholar giving his card to the “Dirty Fellow” – Father Thames. If Michael Faraday were with us today, he would be happy to see that electricity-based processing methods can deliver promising results in the context of wastewater management and sewage treatment. Some of those methods go beyond only treatment and convert sewage sludge into useful products. Examples include production of pyrolytic liquids from industrial sewage sludge using induction heating,²⁸ production of bio-fuels *via* microwave-assisted pyrolysis of sewage sludge²⁹ or plasma gasification of sewage sludge.^{30,31}

Concluding remarks

The contributions present in this volume of Faraday Discussions not only address some of the questions and challenges of the fossil-to-bio-based transition but also demonstrate new opportunities which, if properly addressed and developed further, should lead to novel, commercially attractive concepts. Electricity-based



Fig. 2 Future bioprocessing plant: modular and green electricity-driven (plant model designed by LEGO® Ideas member Ymarilego). Image used by permission, ©2017 The LEGO Group; source for the portrait of Michael Faraday: “Michael Faraday” by John Hall Gladstone (3rd edn, Macmillan and Co., London, 1874).



technologies that build on Michael Faraday's scientific legacy are examples of such opportunities. In the long term, electricity is destined to play the key role in process industries, especially if obtained from a fully renewable source, as it is the most widely available and most flexible form of energy. This also holds true for the fossil-to-bio-based transition, where energy consumption is presently one of the most important challenges. Bioprocesses are often highly energy-demanding and the energy cost significantly affects the process economy. A bio-based industry that utilizes fossil fuel energy will never be really "green" and a gradual shift to cheap, renewable electricity as the primary energy source is needed. In the long term, such a shift, along with the introduction of modularity during plant design, should result in a dramatic change in the bio-process economy. Future bio-processing plants will be modular and green electricity-driven (Fig. 2). Such an evolution will further help in converting the "Penrose" triangles of today that are associated with the fossil-to-bio-based transition into viable feedstock–process–product solutions.

References

- 1 L. S. Penrose and R. Penrose, *Br. J. Psychol.*, 1958, **49**, 31–33.
- 2 J. S. Golden and R. B. Handfield, *Why Biobased? Opportunities in the Emerging Bioeconomy*, 2014, <https://www.biopreferred.gov/files/WhyBiobased.pdf>.
- 3 S. K. Maity, *Renewable Sustainable Energy Rev.*, 2015, **43**, 1427–1445.
- 4 T. T. H. Nguyen, Y. Kikuchi, M. Noda and M. Hiraoa, *Environ. Prog. Sustainable Energy*, 2016, **35**, 174–182.
- 5 V. Chambost and P. R. Stuart, *Ind. Biotechnol.*, 2007, **3**, 112–119.
- 6 M. Faraday, *Experimental Researches in Electricity*, Richard and John Edward Taylor, London, 1839, vol. I.
- 7 M. Faraday, *Experimental Researches in Electricity*, Richard and John Edward Taylor, London, 1844, vol. II.
- 8 S. L. Miller, *Science*, 1953, **117**, 528–529.
- 9 S. L. Miller and H. C. Urey, *Science*, 1959, **130**, 245–251.
- 10 A. P. Johnson, H. J. Cleaves, J. P. Dworkin, D. P. Glavin, A. Lazcano and J. L. Bada, *Science*, 2008, **322**, 404.
- 11 E. T. Parker, H. J. Cleaves, J. P. Dworkin, D. P. Glavin, M. Callahan, A. Aubrey, A. Lazcano and J. L. Bada, *Proc. Natl. Acad. Sci. U. S. A.*, 2011, **108**, 5526–5531.
- 12 A. Schievano, T. Pepé Sciarria, K. Vanbroekhoven, H. De Wever, S. Puig, S. J. Andersen, K. Rabaey and D. Pant, *Trends Biotechnol.*, 2016, **34**, 866–878.
- 13 K. Chandrasekhar, K. Amulya and S. Venkata Mohan, *Waste Manag.*, 2015, **45**, 57–65.
- 14 A. Golberg, M. Sack, J. Teissie, G. Pataro, U. Pliquet, G. Saulis, T. Stefan, D. Miklavcic, E. Vorobiev and W. Frey, *Biotechnol. Biofuels*, 2016, **9**, 94.
- 15 W. Frey, C. Gusbeth, T. Sakugawa, M. Sack, G. Mueller, J. Sigler, E. Vorobiev, N. Lebovka, I. Álvarez, J. Raso, L. C. Heller, M. A. Malik, C. Eing and J. Teissie, in *Bioelectrics*, ed. H. Akiyama and R. Heller, Springer Japan, Tokyo, 2017, ch. 6, pp. 389–486.
- 16 L. Xu, C. Guo, F. Wanga, S. Zheng and C.-Z. Liu, *Bioresour. Technol.*, 2011, **102**, 10047–10051.
- 17 M. Cerff, M. Morweiser, R. Dillschneider, A. Michel, K. Menzel and C. Posten, *Bioresour. Technol.*, 2012, **118**, 289–295.



- 18 W. T. Tsai, M. K. Lee and Y. M. Chang, *J. Anal. Appl. Pyrolysis*, 2006, **76**, 230–337.
- 19 M. K. Lee, W.-T. Tsai, Y.-L. Tsai and S. H. Lin, *J. Anal. Appl. Pyrolysis*, 2010, **88**, 110–116.
- 20 P. D. Muley, C. Henkel, K. K. Abdollahi and D. Boldor, *Energy Fuels*, 2015, **29**, 7375–7385.
- 21 Y.-F. Huang, P.-T. Chiueh and S.-L. Lo, *Sustainable Environ. Res.*, 2016, **26**, 103–109.
- 22 H. Marion Morgan Jr, Q. Bu, J. Liang, Y. Liu, H. Mao, A. Shi, H. Lei and R. Ruan, *Bioresour. Technol.*, 2017, **230**, 112–121.
- 23 N. Azcan and A. Danisman, *Fuel*, 2008, **87**, 1781–1788.
- 24 R. B. Mato Chain, J. Monzó-Cabrera and K. Solyom, in *Alternative Energy Sources for Green Chemistry*, ed. G. Stefanidis and A. Stankiewicz, RSC, Cambridge, 2016, ch. 2, pp. 34–63.
- 25 G. S. J. Sturm, A. Navarrete Muñoz, P. V. Aravind and G. D. Stefanidis, *IEEE Trans. Plasma Sci.*, 2016, **44**, 670–678.
- 26 M. Glemser, M. Heining, J. Schmidt, A. Becker, D. Garbe, R. Buchholz and T. Brück, *Appl. Microbiol. Biotechnol.*, 2016, **100**, 1077–1088.
- 27 C.-Y. Chen, K.-L. Yeh, R. Aisyah, D.-J. Lee and J.-S. Chang, *Bioresour. Technol.*, 2011, **102**, 71–81.
- 28 W.-T. Tsai, J.-H. Chang, K.-J. Hsien and Y.-M. Chang, *Bioresour. Technol.*, 2009, **100**, 406–412.
- 29 W. Zuo, Y. Tian and N. Ren, *Waste Manag.*, 2011, **31**, 1321–1326.
- 30 J. Balgaranova, *Waste Manage. Res.*, 2003, **21**, 38–41.
- 31 A. Mountouris, E. Voutsas and D. Tassios, *Energy Convers. Manage.*, 2008, **49**, 2264–2271.

