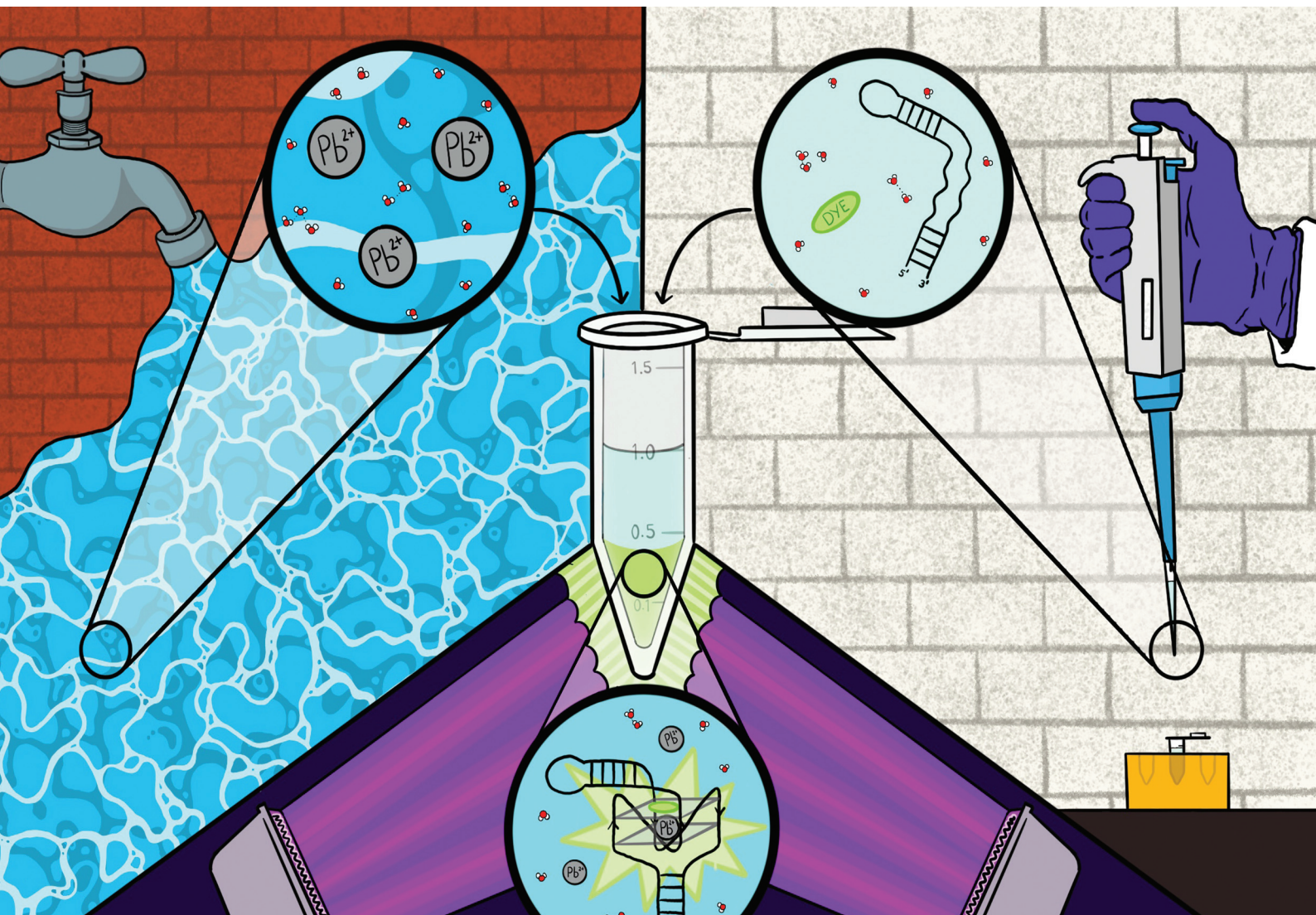


Organic & Biomolecular Chemistry

Volume 23
Number 43
21 November 2025
Pages 9725-9982

rsc.li/obc



Themed issue: New Talent 2025

ISSN 1477-0520



RNA Mango-based sensors for lead†

Cite this: *Org. Biomol. Chem.*, 2025, **23**, 9871
Accepted 26th June 2025

Annyesha Biswas ^a and Saurja DasGupta ^{*,a,b,c}

Received 9th May 2025,
Accepted 26th June 2025

DOI: 10.1039/d5ob00765h

rsc.li/obc

Lead (Pb²⁺) toxicity poses a serious threat to human health and remains a global concern; therefore, there is a critical need for the development of easy-to-use and cost-effective tools for the rapid monitoring of Pb²⁺. In this study, we demonstrate the potential of the RNA Mango aptamer as a sensitive and selective sensor for Pb²⁺. Our findings reveal that trace amounts of Pb²⁺ induce the formation of a G-quadruplex motif in RNA Mango, which facilitates dye binding and activates fluorescence. A detailed investigation of the fluorescence properties of RNA Mango with three different dyes, TO1-biotin, TO3-biotin, and thioflavin-T, in the presence of Pb²⁺ shows that RNA Mango has the highest binding affinity for Pb²⁺ in combination with TO1-biotin, with a *K_D* value as low as ~100 nM. In the presence of Pb²⁺, RNA Mango has sub-micromolar affinity for all three dyes, showing the tightest binding to TO1-biotin (*K_D* ~ 40 nM). Mango lead sensors detect low nanomolar concentrations of Pb²⁺ with limits of detection of 2–16 nM, which are significantly lower than its allowable limit in drinking water. RNA Mango exhibits remarkable selectivity toward Pb²⁺ and can detect Pb²⁺ in tap water samples. This work reports a new class of simple and inexpensive fluorescence-based sensors for lead and expands the repertoire of RNA-based lead sensors.

Introduction

Heavy metal contamination is a persistent global threat to environmental and human health. Among these metals, lead (Pb²⁺) is of particular concern due to its high toxicity even at low exposure levels. Industrial processes such as the mining and smelting of lead ores,¹ the production of leaded gasoline,² and the manufacturing of lead-acid batteries³ and lead-containing pigments⁴ release large amounts of lead into the environment. Importantly, chronic exposure to low amounts of lead, primarily through contaminated drinking water,⁵ industrial waste,⁶ and deteriorating infrastructure such as lead-based pipes,⁷ has been linked to cognitive deficits,⁸ developmental disorders,⁹ cardiovascular diseases,¹⁰ and kidney dysfunction.¹¹ Pb²⁺ toxicity poses a heightened risk to children, where exposure is associated with neurodevelopmental impairments, behavioral disorders, and reduced attention spans.^{12,13} Given these health risks, regulatory agencies such as the U.S. Environmental Protection Agency (EPA) and the World Health Organization (WHO) have set stringent limits for lead in drink-

ing water at 15 parts per billion (72 nM) and 10 parts per billion (46 nM), respectively.^{14,15} Ensuring compliance with these regulations requires sensitive and reliable detection methods.

Traditional analytical techniques for Pb²⁺ detection include atomic absorption spectrometry, inductively coupled plasma mass spectrometry, anodic stripping voltammetry, and X-ray fluorescence spectroscopy.¹⁶ These methods allow highly sensitive Pb²⁺ detection; however, the requirements for expensive and elaborate instrumentation, trained personnel, and complex sample preparation make them less suited for rapid, on-site monitoring. Organic scaffolds, like small molecule chelators, macrocycles, or even metal-organic frameworks (MOFs), have been used for fluorescence-based and colorimetric detection of Pb²⁺.^{17–19} Although these chemical sensors are capable of detecting Pb²⁺ without sophisticated instrumentation, they are not sufficiently selective and often fail to detect trace amounts of Pb²⁺.

To address these limitations, there is increasing interest in developing biosensors that offer rapid, cost-effective, and portable lead detection with minimal technical expertise. Biosensors generally outperform chemical sensors in selectivity and sensitivity metrics, presumably due to more sophisticated molecular recognition by biomolecules.^{20,21} Nucleic acid-based biosensors detect Pb²⁺ by leveraging structural changes induced by Pb²⁺ binding.^{22,23} For example, Pb²⁺ binding to RNA-cleaving DNazymes induce the formation of their catalytic conformation, which results in the cleavage of an RNA strand.²⁴ Pb²⁺-induced cleavage of a fluorescently-

^aDepartment of Chemistry and Biochemistry, University of Notre Dame, Notre Dame, IN 46556, USA. E-mail: sdasgupta@nd.edu

^bDepartment of Biological Sciences, University of Notre Dame, Notre Dame, IN 46556, USA

^cBerthiaume Institute for Precision Health, University of Notre Dame, Notre Dame, IN 46556, USA

†Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d5ob00765h>



labeled RNA substrate spatially separates the fluorophore on the RNA from the quencher on the DNAzyme, generating a fluorescence 'On' signal. A similar strategy can be used to create colorimetric DNAzyme-based sensors by conjugating them to gold nanoparticles.²⁵ G-quadruplex-based biosensors exploit the Pb^{2+} -induced folding of guanine-rich sequences into a G-quadruplex. G-quadruplexes are typically formed in the presence of monovalent cations like Na^+ and K^+ .²² Pb^{2+} , due to its similar ionic radius to K^+ and twice the charge, facilitates the formation of a stable G-quadruplex structure at lower concentrations. G-quadruplex-based sensors exploit the structural change from single-stranded to a G-quadruplex in different ways.²⁶ Pb^{2+} -induced quadruplex formation may bring the two DNA termini, modified with a fluorophore and quencher, into close proximity, resulting in a fluorescence 'Off' signal.²⁷ Certain G-quadruplex sequences exhibit peroxidase activity upon binding to hemin and appropriate reactants like ABTS, generating colored, chemiluminescent, or fluorescent products.²⁸ Therefore, the catalytic activity of these quadruplex-hemin DNAzymes has also been used as a Pb^{2+} sensing platform. DNA-based electrochemical sensors convert G-quadruplex formation into an electrochemical signal, which is detected by techniques such as cyclic voltammetry.²⁹ However, the requirement for nucleic acid modification with fluorophores/quenchers or nanoparticles, pH-sensitive chemical reactions, or expensive electrochemical assemblies poses significant challenges to the practical application of these biosensors.

A new strategy that leveraged Pb^{2+} -induced G-quadruplex formation in the fluorogenic RNA aptamer, Spinach, generated the first RNA-based biosensor for Pb^{2+} .³⁰ Stabilization of the Spinach G-quadruplex by trace amounts of Pb^{2+} allows its cognate dye, DFHBI, to bind to this platform, activating strong and stable green fluorescence. Spinach demonstrated moderate binding affinity for Pb^{2+} ($K_D \sim 1.3 \mu\text{M}$) but high sensitivity, with a limit of detection (LOD) of 6 nM. This system consists of just the RNA aptamer and its cognate dye, where the fluorescence signal results from non-covalent dye binding. Therefore, fluorogenic RNA aptamers offer a simple and efficient platform for Pb^{2+} detection, addressing key limitations related to covalent modifications of nucleic acids with dyes or nanoparticles, optimal reaction conditions for colorimetric or chemiluminescence detection, and elaborate electrochemical assemblies. Despite the creation of numerous G-quadruplex-containing fluorogenic RNA aptamers since its development, the Spinach sensor remained the only RNA-based biosensor for Pb^{2+} . In this study, we expand the repertoire of RNA-based Pb^{2+} sensors by evaluating alternative fluorogenic RNA aptamers for enhanced binding affinity to Pb^{2+} , increased sensitivity, and improved selectivity. We identify the RNA Mango aptamer as an excellent candidate for creating Pb^{2+} sensors and describe its Pb^{2+} -sensing properties in combination with three dyes: TO1-biotin and TO3-biotin, the cognate dyes for this aptamer, and thioflavin-T (ThT), a sequence agnostic but G-quadruplex-specific dye.^{31–34} A detailed characterization of the fluorescence properties of the Mango aptamer with these dyes in the presence of Pb^{2+}

allowed us to develop the most sensitive RNA-based Pb^{2+} sensors to date.

Results and discussion

RNA Mango exhibits strong fluorescence in the presence of Pb^{2+}

Since the development of the Spinach sensor, other G-quadruplex-containing fluorogenic RNA aptamers such as Broccoli, Mango, Peach, and Beetroot have been artificially evolved (Fig. S1†).^{35–38} The Broccoli (Fig. S1A†) and Beetroot (Fig. S1B†) aptamers were isolated from random RNA libraries by virtue of their ability to bind to DFHBI-1T and DFHO and exhibit green and yellow fluorescence, respectively (Table S1†).^{35,36} The Mango aptamer (Fig. S1C†) was similarly obtained for binding to thiazole orange derivatives, namely TO1-biotin and TO3-biotin.³⁹ Various RNA Mango aptamers have been reported in the literature – all of which bind to their cognate dyes with nanomolar affinities. Mango III exhibits the brightest fluorescence among the existing Mango aptamers (Mango aptamers I–IV),⁴⁰ and the A10U mutant of Mango III (MangoIII_A10U) was reported to show a higher quantum yield and 18% higher fluorescence enhancement over wild-type Mango III.³⁸ Given our goal of developing a fluorescence sensor for Pb^{2+} , we selected the Mango III_A10U aptamer due to its superior fluorescence properties. The availability of a high-resolution crystal structure of Mango III_A10U during our screening process afforded opportunities for engineering a truncated version of this aptamer (Fig. S1C†).³⁸ Peach, an aptamer related to Mango, was selected for binding to TO3-biotin.³⁷ The G-quadruplex motif in each aptamer, stabilized by 100 mM K^+ in the folding buffer, serves as the dye binding platform and is consequently key to the aptamer's fluorescence properties. To assess whether these G-quadruplex-containing RNA aptamers can be used to detect Pb^{2+} , we examined the fluorescence properties of Broccoli, Mango III_A10U (henceforth as Mango), Peach, and Beetroot in the presence of their respective dyes, both in the absence and presence of $1 \mu\text{M}$ Pb^{2+} (Fig. 1 and S2†).

Although all four aptamers use G-quadruplexes for binding to their cognate dyes, only RNA Mango in combination with

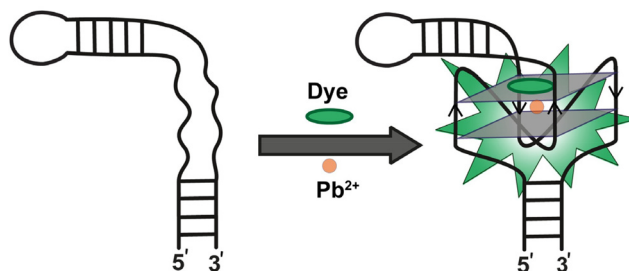


Fig. 1 Schematic illustration of a fluorogenic RNA-based lead sensor. The unstructured region (shown by curved lines) in the RNA aptamer folds into a G-quadruplex structure in the presence of Pb^{2+} and binds to the fluorophore, triggering a fluorescence signal.



TO1-biotin exhibited a significant fluorescence enhancement (~ 8 -fold) in the presence of $1\ \mu\text{M}\ \text{Pb}^{2+}$ (Fig. S2A†). The Peach/TO3-biotin combination also showed fluorescence enhancement upon the addition of Pb^{2+} ; however, this enhancement was less pronounced (3-fold) with a less intense signal than Mango/TO1-biotin (Fig. S2B†). No noticeable enhancement in fluorescence was observed for the Broccoli/DFHBI-1T and Beetroot/DFHO aptamer-dye pairs (Fig. S2C and S2D†). This could be due to the relatively lower affinity of Broccoli and Beetroot for their cognate dyes DFHBI-1T ($K_D \sim 0.36\ \mu\text{M}$) and DFHO ($K_D \sim 2.6\ \mu\text{M}$) compared to the low nanomolar affinities of Mango and Peach to their dyes in $\sim 100\ \text{mM}\ \text{K}^+$.^{35–37} Despite the lower affinity for their dyes, the poor fluorescence enhancement observed in Broccoli and Beetroot is unexpected and may reflect the differences in their dye binding affinities in the presence of $1\ \mu\text{M}\ \text{Pb}^{2+}$. Fluorescence enhancement exhibited by Mango/TO1-biotin in the presence of $1\ \mu\text{M}\ \text{Pb}^{2+}$ was 33-fold, 308-fold, and 1449-fold higher relative to Peach/TO3-

biotin, Broccoli/DFHBI-1T, and Beetroot/DFHO, respectively (Fig. 2A and B). These results suggested that, among the aptamers tested, Mango was the most promising candidate for a Pb^{2+} sensor.

As Mango showed the highest increase in fluorescence in the presence of Pb^{2+} , we studied its fluorescence properties with TO3-biotin and ThT, dyes that have been previously reported to bind Mango in addition to TO1-biotin.^{38,40} While TO3-biotin is a cognate dye of the Mango and Peach aptamers, ThT exhibits green fluorescence upon specifically binding to G-quadruplex structures in a context-independent manner (Table S1†).^{31–34} Mango showed fluorescence enhancements of ~ 2 -fold and ~ 5 -fold in the presence of TO3-biotin and ThT, respectively, upon Pb^{2+} addition (Fig. 2C and D). This ability of Mango to exhibit fluorescence enhancements with all three dyes in the presence of just $1\ \mu\text{M}\ \text{Pb}^{2+}$ highlights its robustness and versatility as a potential Pb^{2+} sensor.

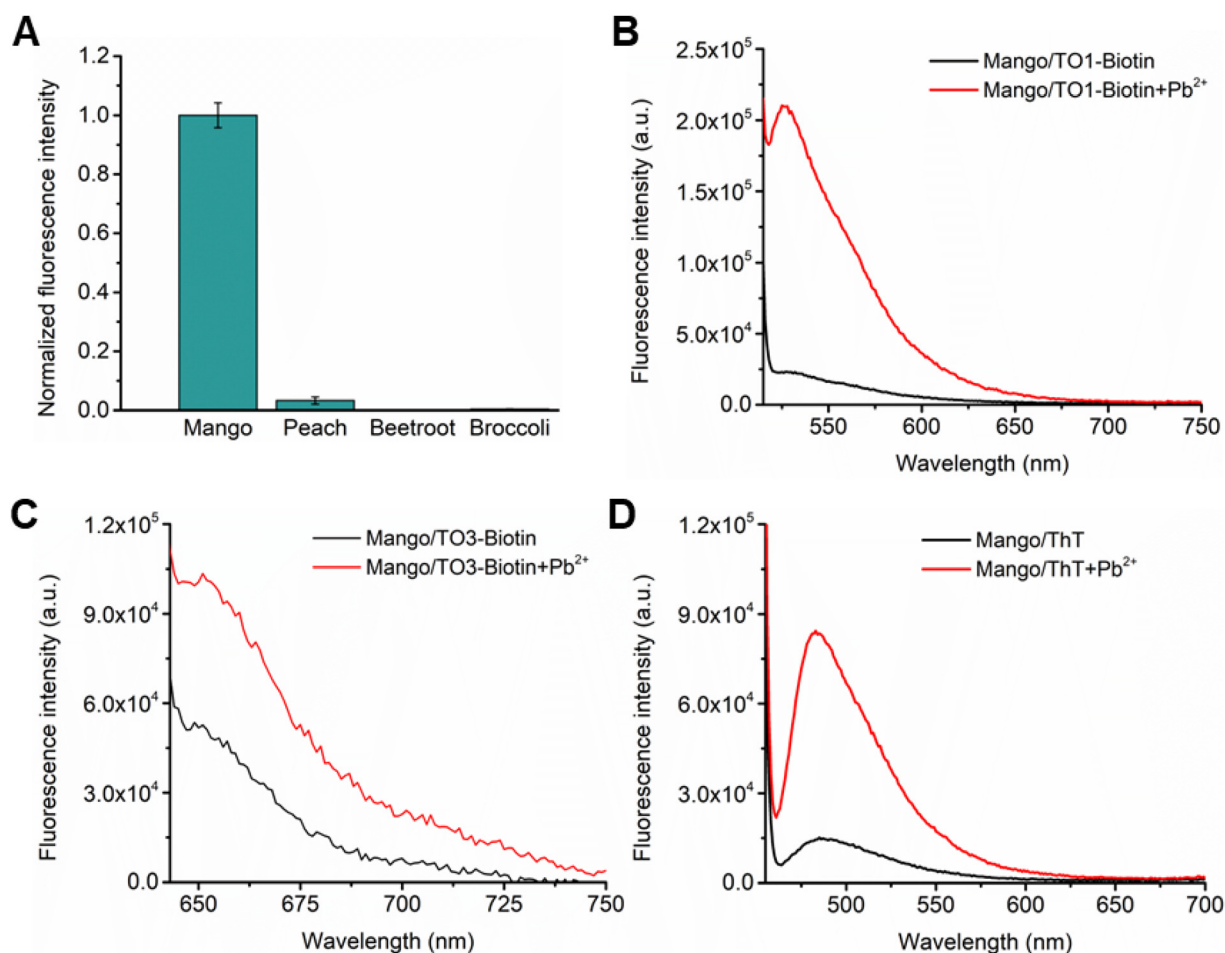


Fig. 2 RNA Mango exhibits fluorescence enhancement in the presence of Pb^{2+} . (A) Fluorescence enhancement in different RNA aptamers/dye combinations upon the addition of $1\ \mu\text{M}\ \text{Pb}^{2+}$. (B)–(D) Fluorescence spectra of (B) Mango-TO1-biotin, (C) Mango-TO3-biotin, and (D) Mango-ThT in the absence (black) and presence of $1\ \mu\text{M}\ \text{Pb}^{2+}$ (red). Mango-TO1-biotin and Mango-ThT showed significant fluorescence enhancement with Pb^{2+} . Concentrations of the RNA and dye were $300\ \text{nM}$ and $3\ \mu\text{M}$, respectively. Experiments were performed at pH 8 and $5\ \text{mM}\ \text{Mg}^{2+}$. Error bars in (A) correspond to the standard deviation of three independent experiments.



Pb²⁺-induced fluorescence activation of RNA Mango is dependent on G-quadruplex stabilization

Stabilization of the G-quadruplex motif in Mango is central to its fluorescence activation by Pb²⁺ as this motif serves as the binding platform for the dye. RNA G-quadruplexes usually exist in the parallel topology and are characterized by a prominent positive peak at ~260 nm and a prominent negative peak at ~240 nm.⁴¹ We observed an enhancement of the positive peak at 263 nm upon the addition of 1 μ M Pb²⁺ to Mango in the absence of any dye, which further increased in intensity in the presence of 10 μ M Pb²⁺ (Fig. 3A). This Pb²⁺-induced increase in CD intensity at 263 nm was accompanied by a decrease in intensity at ~240 nm. These results provide direct evidence of the Pb²⁺-induced stabilization of the G-quadruplex motif in Mango that is independent of dye binding.

To further investigate the effect of Pb²⁺-induced G-quadruplex formation on the fluorescence properties of Mango, we mutated guanine residues critical for quadruplex formation to adenines (Fig. 3B and Table S2†). Mutating these key guanine residues in the G-quadruplex resulted in a near-complete loss of fluorescence for TO1-biotin and ThT (Fig. 3C and E). Although significantly reduced, some fluorescence was detected in the quadruplex mutants of Mango when bound to TO3-biotin (Fig. 3D), likely due to the higher intrinsic fluorescence of TO3-biotin in buffer. Collectively, these results confirm that Pb²⁺-induced fluorescence activation of Mango is dependent on G-quadruplex formation.

RNA Mango is a sensitive Pb²⁺ sensor

An ideal Pb²⁺ sensor should exhibit high sensitivity with a limit of detection (LOD) in the low nanomolar range. To evaluate its sensitivity, we investigated Mango fluorescence with TO1-biotin, TO3-biotin, or ThT in response to increasing concentrations of Pb²⁺. The binding affinity of Mango to Pb²⁺ in the presence of each dye was determined by fitting the data to the Hill equation (Fig. 4A–C). The apparent K_D values for Pb²⁺ binding with Mango in the presence of TO1-biotin, TO3-biotin, and ThT dyes were calculated as 113 $\pm$ 31 nM, 295 $\pm$ 52 nM, and 157 $\pm$ 32 nM, respectively (Fig. 4D). These results revealed that sensors created with Mango/TO1-biotin, Mango/TO3-biotin, and Mango/ThT bind Pb²⁺ with 12-fold, 4-fold, and 8-fold greater affinities than the Spinach sensor (K_D ~ 1.3 μ M).³⁰ LOD of the Mango lead sensor with TO1-biotin, TO3-biotin, and ThT were calculated as 2 nM, 16 nM, and 4 nM (Fig. 4E and S3†), respectively (based on the 3 σ /slope method, where σ is the standard deviation of the blank). This is consistent with the trend in their binding affinities for Pb²⁺. Sensors prepared with Mango/TO1-biotin and Mango/ThT are the most sensitive RNA-based Pb²⁺ sensors reported to date. Importantly, the LOD values for all three Mango-based lead sensors are significantly lower than the acceptable limits for Pb²⁺ in drinking water.

The increase in the fluorescence signal in response to increasing concentrations of Pb²⁺ was directly imaged for all three Mango-based sensors (Fig. S4†). While the Mango/TO1-

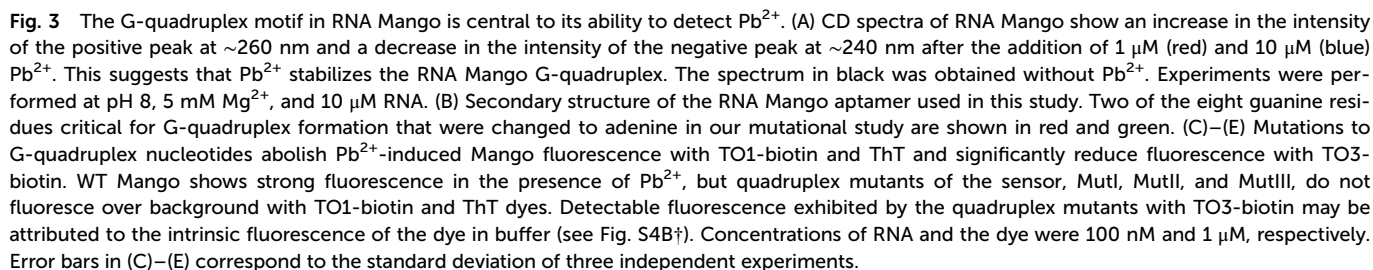
biotin and Mango/ThT sensors exhibited a clear, concentration-dependent enhancement of green fluorescence upon Pb²⁺ addition with a negligible signal without Pb²⁺, the red fluorescence enhancement in the Mango/TO3-biotin sensor was less discernible due to the high intrinsic fluorescence of the TO3-biotin dye. Although Pb²⁺-induced fluorescence enhancement in the Mango/TO3-biotin sensor can be clearly measured by fluorimetry, its inability to provide a distinct visual readout of this enhancement prevents its use as a point-of-care lead sensor in its current form. In contrast, the ease of Pb²⁺ detection using the Mango/TO1-biotin and Mango/ThT combinations makes them attractive candidates for cheap, easy-to-use, point-of-care sensors for lead.

In addition to the sensors' high affinity for Pb²⁺, the strong fluorescence signal in the presence of a trace amount of Pb²⁺ could be due to tighter binding of the dye. To assess the affinity of RNA Mango to the three dyes in the presence of 1 μ M Pb²⁺, we performed fluorescence titrations in the presence of increasing concentrations of TO1-biotin, TO3-biotin, and ThT (Fig. S5A–C†). The apparent K_D values for dye binding were determined as 39 $\pm$ 8 nM for TO1-biotin (Fig. S5A and D†), 792 $\pm$ 72 nM for TO3-biotin (Fig. S5B and D†), and 193 $\pm$ 27 nM for ThT (Fig. S5C and D†). The highest affinity of Mango to TO1 is consistent with the highest sensitivity and Pb²⁺ affinity exhibited by the Mango/TO1-biotin sensor. In comparison, the apparent K_D value for DFHBI binding for the Spinach sensor in the presence of 10 μ M Pb²⁺ was determined to be ~1 μ M.³⁰ Therefore, Mango-based lead sensors bind to their dyes with at least 1.4–28.5-fold greater affinity than the Spinach sensor. Consequently, Mango-based lead sensors require lower dye concentrations for Pb²⁺ detection.

RNA Mango is a selective Pb²⁺ sensor

To investigate the selectivity of Mango sensors for Pb²⁺ detection, we measured their fluorescence in the presence of thirteen other metal ions that may be encountered in environmentally relevant samples (Fig. 5). Pb²⁺-specific fluorescence signals could be detected visually for the Mango/TO1-biotin and Mango/ThT sensors; however, this was not possible for the Mango/TO3-biotin sensor, which showed a high background signal due to the intrinsic fluorescence of the TO3-biotin dye (Fig. S6†). In fluorimetric experiments, we observed strong fluorescence enhancement only upon Pb²⁺ addition with all three dyes; however, weak signals were detected in the presence of Ca²⁺ and K⁺ with TO1-biotin (Fig. 5A and B) and TO3-biotin (Fig. 5C and D). In contrast, the Mango/ThT sensor showed weak but detectable fluorescence only when Ca²⁺ was present (Fig. 5E and F). These results are consistent with reports of G-quadruplex stabilization by Ca²⁺ and K⁺, with the lower signals reflective of much weaker binding than Pb²⁺.³⁰ The Mango/TO1-biotin sensor exhibited ~20-fold and ~25-fold higher signals for Pb²⁺ over K⁺ and Ca²⁺, respectively (Fig. 5B), whereas the Mango/TO3-biotin sensor showed an ~10-fold higher signal for Pb²⁺ relative to both Ca²⁺ and K⁺ (Fig. 5D). In contrast, the Mango/ThT sensor exhibited a 53-fold and





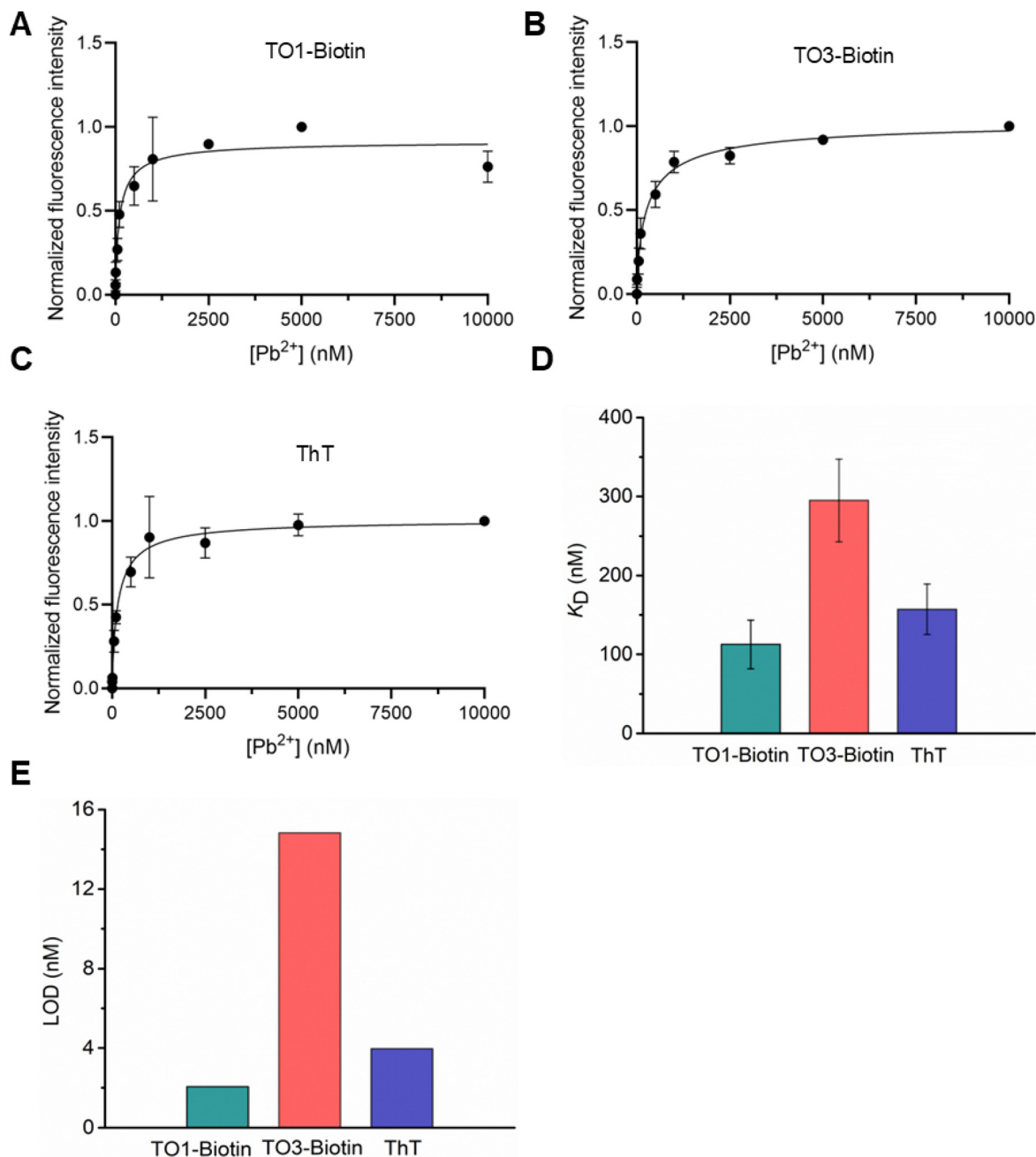


Fig. 4 RNA Mango exhibits high sensitivity for Pb²⁺ detection. (A–C) Fluorescence signals exhibited by RNA Mango in response to increasing concentrations of Pb²⁺ with (A) TO1-biotin, (B) TO3-biotin, and (C) ThT. (D) RNA Mango binds to Pb²⁺ with high affinity in the presence of all three dyes, exhibiting K_D values of 100–300 nM. (E) RNA Mango is highly sensitive to low nanomolar concentrations of Pb²⁺ in the presence of all three dyes, with LOD of 2–16 nM. Experiments were performed at pH 8 and 5 mM Mg²⁺. Concentrations of RNA and the dye were 100 nM and 1 μ M, respectively, with Pb²⁺ concentrations in the range of 5 nM–10 μ M. Error bars correspond to the standard deviation of three independent experiments.

142-fold higher signal for Pb²⁺ compared to Ca²⁺ and K⁺, respectively (Fig. 5F). The superior selectivity for Pb²⁺, combined with a low limit of detection, makes Mango/ThT the best Mango-based lead sensor. The combined attributes of high binding affinity, excellent sensitivity, and strong selectivity for Pb²⁺ detection underscore the potential of Mango-based sensors as robust tools for the quantitative detection of Pb²⁺ at low concentrations.

RNA Mango detects Pb²⁺ in tap water

A reliable sensor must demonstrate functionality in practical, real-world applications. We tested the applicability of the Mango lead sensors for detecting Pb²⁺ in tap water. RNA Mango generated fluorescence within minutes of exposure to Pb²⁺ and remained stable in tap water containing 1 μ M Pb²⁺ for at least 24 hours – the longest duration tested (Fig. S7†).



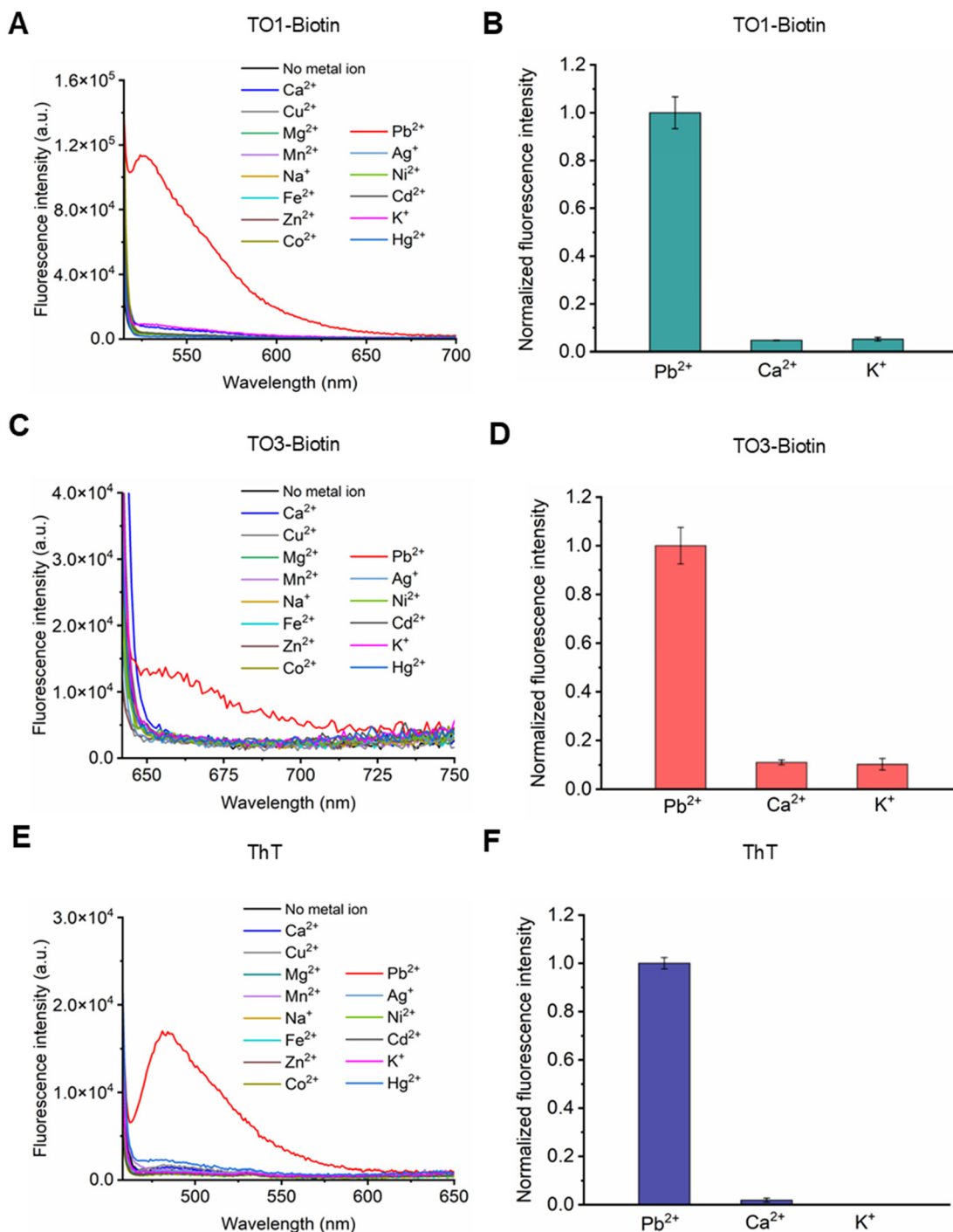


Fig. 5 RNA Mango is selective for Pb²⁺. Fluorescence spectra of the Mango sensor with (A) TO1-biotin, (C) TO3-biotin, and (E) ThT in the presence of various metal ions. Relative fluorescence intensities of the Mango sensor in the presence of Pb²⁺, Ca²⁺, and K⁺ with (B) TO1-biotin, (D) TO3-biotin, and (F) ThT. Experiments were performed with 100 nM RNA, 250 nM dye, and 1 μ M metal ions at pH 8 and 5 mM Mg²⁺. Error bars in (B), (D), and (F) correspond to the standard deviation of three independent experiments.

Encouraged by this observation, we assessed the performance of the Mango/TO1-biotin and Mango/ThT sensors by spiking tap water samples with different concentrations of Pb²⁺. We observed an increase in the fluorescence signal with Pb²⁺ concentrations ranging from 5 nM to 5 μ M for TO1-biotin (Fig. 6A)

and ThT (Fig. 6B), indicating that the Mango sensor can detect Pb²⁺ in tap water. The increase in the fluorescence signal observed within minutes after the addition of Pb²⁺-spiked tap water and the stability of the Mango sensor in tap water demonstrate the potential applicability of our sensor in real-



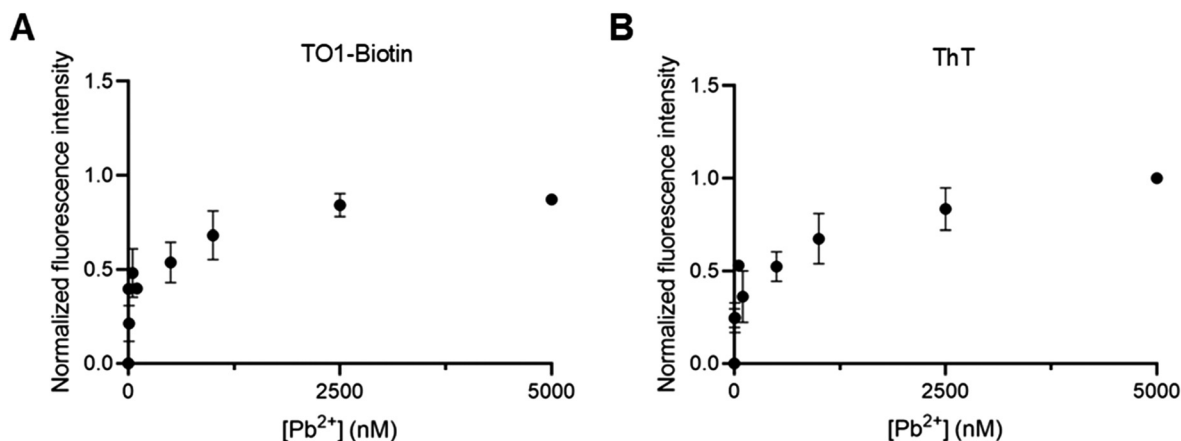


Fig. 6 Mango lead sensors are functional in tap water. Fluorescence signals exhibited by RNA Mango in response to increasing concentrations of Pb²⁺ in tap water for (A) TO1-biotin and (B) ThT. Experiments were performed at pH 8 and 5 mM Mg²⁺. Concentrations of RNA and the dye were 100 nM and 1 μ M, respectively, with Pb²⁺ concentrations in the range of 5 nM–5 μ M. Error bars correspond to the standard deviation of three independent experiments.

world samples. However, both Mango sensors showed modest background fluorescence in tap water samples, even in the absence of Pb²⁺. High concentrations of Ca²⁺ in the hard water samples may interfere with Pb²⁺ detection by Mango sensors; however, this can be remedied by several approaches, as discussed in the next section.

Conclusions

In this work, we created a new class of fluorescence biosensors for detecting Pb²⁺ based on the RNA Mango aptamer, thus expanding the analytical applications of RNA in heavy metal sensing. Notably, our sensors demonstrate excellent sensitivity with limits of detection (2–16 nM) significantly below the permissible level of Pb²⁺ in drinking water (72 nM). The Mango sensors are able to discriminate between the Pb²⁺ and other metal ions, showing only a weak signal with K⁺ and Ca²⁺. The Mango/ThT sensor emerged as the most promising candidate exhibiting high affinity for its analyte, Pb²⁺ and its dye, ThT in the presence of Pb²⁺, and showing the highest selectivity among the Mango sensors. The versatility of Mango-based lead sensors is highlighted by their compatibility with dyes spanning different spectral regions, including TO1-biotin and ThT in the green and TO3-biotin in the red regions of the spectra, which allows users to select the dye best suited to their needs. Additionally, to detect lead, one simply needs to add the sample to the RNA/dye solution, which eliminates the need for complex chemical reactions or sophisticated sensor assemblies. Fluorescence signal is observed in the presence of Pb²⁺ within minutes by simply shining light from a handheld UV lamp. These features make Mango lead sensors attractive for on-site detection of Pb²⁺.

However, to enable broad applicability in point-of-care devices, we are currently optimizing the concentrations of the Mango RNA aptamer and its fluorogenic dyes to reduce

responsiveness to Ca²⁺, which, if present in real-world samples, may trigger weak false positives. Alternatively, water samples may be pre-treated with NH₄OH to precipitate Ca²⁺ as Ca(OH)₂, which can then be removed *via* filtration.⁴² A subsequent pH adjustment to ~pH 8 with HCl will allow Mango sensors to function as intended. As Ca²⁺ is primarily present as Ca(HCO₃)₂ in hard water, boiling water samples will remove dissolved Ca²⁺ as CaCO₃, allowing Pb²⁺ detection by Mango sensors.⁴³ Another consideration for on-site detection is the potential to incorporate Mango sensors into inexpensive and portable point-of-care analytical devices. Toward this goal, we are developing paper-based analytical devices (PADs) and microfluidic chips for Pb²⁺ detection that use the Mango sensors reported in this work.

Experimental

Materials

Tris-HCl (pH 8) buffer was purchased from Invitrogen. Metal salts Pb(NO₃)₂·3H₂O, AgNO₃, CdCl₂·H₂O, CuSO₄·5H₂O, FeSO₄·7H₂O, Ni(NO₃)₂·6H₂O, ZnCl₂, Hg(NO₃)₂·H₂O, and Mn(CH₃COO)₂·4H₂O were purchased from Sigma-Aldrich (ACS grade). MgCl₂, NaCl, KCl, and CaCl₂ were purchased as filtered solutions from Invitrogen. Dyes DFHBI, DFHBI-1T, and DFHO were purchased from Lucerna Technologies. TO1-biotin and TO3-biotin were purchased from Applied Biological Materials Inc. (abm) and thioflavin-T (ThT) was purchased from Millipore Sigma. The details of all the dyes used in this work are listed in ESI Table S1.† The RNA oligonucleotides used in this work are listed in ESI Table S2† and were purchased from Integrated DNA Technologies (IDT).

Methods

Instrumentation for fluorescence assays. Fluorescence emission was measured using a Fluorolog-3 spectrofluorometer



equipped with a thermocontroller (Horiba Inc.) and a slit width of 2 nm for all the measurements. Excitation wavelengths of 510 nm, 637 nm, and 450 nm and emission ranges of 515–700 nm, 642–750 nm, and 455–600 nm were used for the fluorescence measurements for TO1-biotin, TO3-biotin, and ThT, respectively (see ESI Table S1†). All measurements were performed in triplicate at 25 °C. Data were plotted using Origin 2025 and GraphPad Prism 8.4.3.

Identification of optimal aptamer-dye pairs for Pb²⁺ detection. To identify the optimal RNA-dye pairs for Pb²⁺ detection, we screened four fluorogenic RNA aptamers that use G-quadruplex motifs as their dye-binding platform. In these experiments, 300 nM RNA was heated in the presence of 10 mM Tris-HCl (pH 8) at 90 °C for two minutes, followed by incubation with 5 mM MgCl₂ (for Mango) or 1 mM MgCl₂ (for the Peach, Broccoli, and Beetroot aptamers) at 50 °C for 15 minutes. 3 μM dye (TO1-biotin for Mango, DFHBI-1T for Broccoli, TO3-biotin for Peach, or DFHO for Beetroot) was added to this solution and incubated at 37 °C for 10 minutes. Finally, 1 μM Pb(NO₃)₂ was added and incubated at 37 °C for 10 minutes. Fluorescence intensities for samples containing RNA and the dye ('blank') were subtracted from the fluorescence intensities for samples containing RNA, dye, and Pb²⁺.

Sensitivity and selectivity assays. Samples for sensitivity assays contained 1 μM dye, 100 nM RNA, and 5 nM–10 μM Pb²⁺. Binding affinities of RNA Mango for Pb²⁺ in the presence of the three dyes were determined by plotting normalized fluorescence (normalization was performed with respect to the highest signal) for each concentration of Pb²⁺ and fitting the data to the Hill equation: $y = B_{\max} \times x^h / (K_D^h + x^h)$, where $B_{\max}$ = highest fluorescence signal, K_D = dissociation constant, and h = hill slope. The limit of detection (LOD) was measured from sensitivity assays using the $3\sigma/\text{slope}$ method (where σ represents the standard deviation of the 10 blanks). The fluorescence signal corresponding to the linear region of the binding plot was fitted into a simple linear regression model to obtain the slope. Samples for selectivity assays contained 100 nM RNA, 250 nM dye, and 1 μM metal ions (Pb²⁺ or non-Pb²⁺).

Sensitivity and selectivity assays were performed with Mango and each of the three dyes (TO1-biotin, TO3-biotin, or ThT). In addition to measuring fluorescence intensities using a fluorimeter, we imaged samples containing different Pb²⁺ concentrations (0.01, 0.1, 0.5, 1, 5, and 10 μM) or different metal ions at 1 μM using an Azure 600 imaging system with filters closest to the excitation and emission wavelengths of each dye. Samples containing TO1-biotin, TO3-biotin, and ThT were imaged under AzureRed ($\lambda_{\text{ex}} = 524 \text{ nm}/\lambda_{\text{em}} = 572 \text{ nm}$), AzureSpectra 650 ($\lambda_{\text{ex}} = 628 \text{ nm}/\lambda_{\text{em}} = 684 \text{ nm}$), or AzureSpectra 490 ($\lambda_{\text{ex}} = 472 \text{ nm}/\lambda_{\text{em}} = 513 \text{ nm}$), respectively.

Dye-binding assays. 5 nM–10 μM of each of the three dyes – TO1-biotin, TO3-biotin, or ThT – were added to 100 nM RNA in the presence of 1 μM Pb²⁺ and incubated at 37 °C for 10 minutes. Fluorescence intensities at each dye concentration in buffer containing Pb²⁺, but in the absence of RNA, were sub-

tracted from those observed for samples containing both the dye and RNA. Binding affinities of the Mango lead sensor for each dye in the presence of Pb²⁺ were determined by plotting normalized fluorescence at each dye concentration and fitting the data to the Hill equation.

Mutational studies on the RNA Mango G-quadruplex. We determined fluorescence intensities for three RNA Mango constructs (MutI, MutII, and MutIII; see Table S2†) with mutations to their G-quadruplex regions. The fluorescence was measured for each mutant construct in the presence of TO1-biotin, TO3-biotin, or ThT. For each construct, the fluorescence signal without Pb²⁺ was subtracted from that observed with Pb²⁺.

Stability of the RNA Mango aptamer in tap water. The stability of RNA Mango was evaluated under conditions that mimic potential real-world applications. For tap water assays, the sensor was incubated with 1 μM Pb²⁺ at 25 °C in tap water for 0, 30, 60, 120, 180 minutes and 24 h. Aliquots were quenched with 5 volumes of quench buffer (8 M urea, 100 mM Tris-HCl, 100 mM boric acid, 100 mM EDTA) and loaded on an analytical 10% denaturing PAGE. The gel was stained with SYBR Gold for visualization and scanned using an Amersham Typhoon RGB Instrument (Cytiva).

Detection of Pb²⁺ in tap water. To demonstrate the Pb²⁺ detection ability of the Mango sensors in a real-world sample, we added tap water samples spiked with 5 nM–5 μM Pb²⁺ to the Mango sensors (100 nM RNA + 1 μM TO1-biotin or ThT incubated at 37 °C for 10 minutes). The samples were further incubated at 37 °C for 10 minutes, and fluorescence intensities were measured as described above. The average fluorescence intensity of 10 samples containing 100 nM RNA, 1 μM TO1-biotin or ThT, and tap water was subtracted from each of the above Pb²⁺-spiked samples to calculate the fluorescence intensity solely due to Pb²⁺ in the spiked tap water samples.

Circular dichroism (CD) spectroscopy. CD samples were prepared in the same way as in the case of fluorescence experiments, except that they contained 10 μM RNA Mango and no dye. CD measurements were carried out in a Jasco J-1500 CD spectrometer in the wavelength range from 180 to 320 nm, using a path length of 1 mm. Each CD spectrum represents the average of three scans measured at a rate of 50 nm min⁻¹ with a 1 nm interval. All measurements were conducted at 25 °C. Data were smoothed and plotted using Origin 2025.

Author contributions

A. B. and S. D. designed research; A. B. performed research; A. B. and S. D. analyzed the data; A. B. and S. D. wrote the paper.

Conflicts of interest

The authors declare that a provisional patent application has been filed related to the work described in this manuscript.



Data availability

Additional supporting data for this article have been included in the ESI.† This includes (1) chemical structures of the dyes and the sequences of the oligonucleotides used in this work, (2) secondary structures of the fluorogenic RNA aptamers tested and their fluorescence spectra with their corresponding dyes in the absence and presence of Pb^{2+} , (3) sensitivity plots of the Mango sensor with TO1-biotin, TO3-biotin and ThT for Pb^{2+} , (4) images of the Mango lead sensors with increasing concentrations of Pb^{2+} and different metal ions, (5) binding curves for measuring the affinity of TO1-biotin, TO3-biotin, and ThT to RNA Mango in the presence of $1\ \mu\text{M}\ \text{Pb}^{2+}$, and (6) a denaturing PAGE gel to demonstrate the stability of the Mango lead sensor in tap water in the presence of $1\ \mu\text{M}\ \text{Pb}^{2+}$.

Acknowledgements

We thank Professor Bradley Smith and Hailey Salaberry for providing access to their lab's fluorimeter. We thank the members of the DasGupta Lab for their valuable feedback on the manuscript. We thank Professor Marya Lieberman and Vikrant Jandev for sharing the salts used in this study. We thank high school summer students Areej Arif, Jennifer Yang, and Ishita Awasthi for their help during the initial phase of the project. We thank the Biophysics Core Facility for providing the CD spectrometer and Azure 600 imager. CD analyses were performed at the Notre Dame Biophysics Instrumentation Core Facility with support from NIH grant S10OD036273. This work was supported by the University of Notre Dame Start Up funds to S. D.

References

- 1 J. Yang, X. Li, Z. Xiong, M. Wang and Q. Liu, Environmental pollution effect analysis of lead compounds in China based on life cycle, *Int. J. Environ. Res. Public Health*, 2020, **17**(7), 2184, DOI: [10.3390/ijerph17072184](https://doi.org/10.3390/ijerph17072184).
- 2 S. M. C. L. Gioia, M. Babinski, D. J. Weiss, B. Spiro, A. A. F. S. Kerr, T. G. Verissimo, I. Ruiz and J. C. M. Prates, An isotopic study of atmospheric lead in a megacity after phasing out of leaded gasoline, *Atmos. Environ.*, 2017, **149**, 70–83, DOI: [10.1016/j.atmosenv.2016.10.049](https://doi.org/10.1016/j.atmosenv.2016.10.049).
- 3 Z. Sun, H. Cao, X. Zhang, X. Lin, W. Zheng, G. Cao, Y. Sun and Y. Zhang, Spent lead-acid battery recycling in China – A review and sustainable analyses on mass flow of lead, *Waste Manag.*, 2017, **64**, 190–201, DOI: [10.1016/j.wasman.2017.03.007](https://doi.org/10.1016/j.wasman.2017.03.007).
- 4 D. E. Jacobs, R. P. Clickner, J. Y. Zhou, S. M. Viet, D. A. Marker, J. W. Rogers, D. C. Zeldin, P. Broene and W. Friedman, The Prevalence of Lead-Based Paint Hazards in U.S. Housing, *Environ. Health Perspect.*, 2002, **110**(10), A599–A606, DOI: [10.1289/ehp.021100599](https://doi.org/10.1289/ehp.021100599).
- 5 Y. T. Endale, A. Ambelu, G. Sahilu, G. B. Mees and G. Du Laing, Exposure and health risk assessment from consumption of Pb contaminated water in Addis Ababa, Ethiopia, *Heliyon*, 2021, **7**, e07946, DOI: [10.1016/j.heliyon.2021.e07946](https://doi.org/10.1016/j.heliyon.2021.e07946).
- 6 K. N. Wahyusi, L. I. Utami, S. Aprilio and N. Fergina, Reduction of Pb and Cr levels in paper industrial Liquid waste with ion exchange method, *J. Phys.: Conf. Ser.*, 2020, **1569**, 042054, DOI: [10.1088/1742-6596/1569/4/042054](https://doi.org/10.1088/1742-6596/1569/4/042054).
- 7 S. Zahran, D. Mushinski, S. P. McElmurry and C. Keyes, Water lead exposure risk in Flint, Michigan after switch-back in water source: Implications for lead service line replacement policy, *Environ. Res.*, 2020, **181**, 108928, DOI: [10.1016/j.envres.2019.108928](https://doi.org/10.1016/j.envres.2019.108928).
- 8 D. Ramírez Ortega, D. F. González Esquivel, T. Blanco Ayala, B. Pineda, S. Gómez Manzo, J. Marcial Quino, P. Carrillo Mora and V. Pérez de la Cruz, Cognitive impairment induced by lead exposure during lifespan: Mechanisms of lead neurotoxicity, *Toxics*, 2021, **9**(2), 23, DOI: [10.3390/toxics9020023](https://doi.org/10.3390/toxics9020023).
- 9 C. Gundacker, M. Forsthuber, T. Szigeti, R. Kakucs, V. Mustieles, M. F. Fernandez, E. Bengtsen, U. Vogel, K. S. Hougaard and A. T. Saber, Lead (Pb) and neurodevelopment: A review on exposure and biomarkers of effect (BDNF, HDL) and susceptibility, *Int. J. Hyg. Environ. Health*, 2021, **238**, 113855, DOI: [10.1016/j.ijheh.2021.113855](https://doi.org/10.1016/j.ijheh.2021.113855).
- 10 L. He, Z. Chen, B. Dai, G. Li and G. Zhu, Low-level lead exposure and cardiovascular disease: the roles of telomere shortening and lipid disturbance, *J. Toxicol. Sci.*, 2018, **43**(11), 623–630, DOI: [10.2131/jts.43.623](https://doi.org/10.2131/jts.43.623).
- 11 M. L. Adham, Renal Effects of Environmental and Occupational Lead Exposure, *Environ. Health Perspect.*, 1997, **105**, 928–939 <https://ehis.niehs.nih.gov>.
- 12 A. Reuben, M. L. Elliott, W. C. Abraham, J. Broadbent, R. M. Houts, D. Ireland, A. R. Knodt, R. Poulton, S. Ramrakha, A. R. Hariri, A. Caspi and T. E. Moffitt, Association of childhood lead exposure with MRI measurements of structural brain integrity in midlife, *JAMA, J. Am. Med. Assoc.*, 2020, **324**, 1970–1979, DOI: [10.1001/jama.2020.19998](https://doi.org/10.1001/jama.2020.19998).
- 13 B. P. Lanphear, R. Hornung, J. Khoury, K. Yolton, P. Baghurst, D. C. Bellinger, R. L. Canfield, K. N. Dietrich, R. Bornschein, T. Greene, S. J. Rothenberg, H. L. Needleman, L. Schnaas, G. Wasserman, J. Graziano and R. Roberts, Low-level environmental lead exposure and children's intellectual function: An international pooled analysis, *Environ. Health Perspect.*, 2005, **113**, 894–899, DOI: [10.1289/ehp.7688](https://doi.org/10.1289/ehp.7688).
- 14 P. Levallois, P. Barn, M. Valcke, D. Gauvin and T. Kosatsky, Public Health Consequences of Lead in Drinking Water, *Curr. Environ. Health Rep.*, 2018, **5**(2), 255–262, DOI: [10.1007/s40572-018-0193-0](https://doi.org/10.1007/s40572-018-0193-0).
- 15 S. C. Izah, N. Chakrabarty and A. L. Srivastav, A review on heavy metal concentration in potable water sources in Nigeria: Human health effects and mitigating measures, *Toxics*, 2016, **5**(1), 1, DOI: [10.3390/toxics5010001](https://doi.org/10.3390/toxics5010001).



- 16 E. Bulska and A. Ruszczyńska, Analytical techniques for trace element determination, *Phys. Sci. Rev.*, 2017, **2**(5), 20178002, DOI: [10.1515/psr-2017-8002](https://doi.org/10.1515/psr-2017-8002).
- 17 Z. ul Nisa, N. A. Ashashi, R. Singhaal, M. Ahmad, R. M. Gomila, A. Frontera and H. N. Sheikh, Selective and efficient detection of Pb²⁺ in aqueous solution by lanthanoid-organic frameworks bearing pyridine-3,4-dicarboxylic acid and glutaric acid, *CrystEngComm*, 2023, **25**, 2418–2440, DOI: [10.1039/D3CE00099K](https://doi.org/10.1039/D3CE00099K).
- 18 M. Sadia, J. Khan, R. Khan, S. W. Ali Shah, A. Zada, M. Zahoor, R. Ullah and E. A. Ali, Trace level detection of Pb²⁺ ion using organic ligand as fluorescent-on probes in aqueous media, *Heliyon*, 2024, **11**(1), e41125, DOI: [10.1016/j.heliyon.2024.e41125](https://doi.org/10.1016/j.heliyon.2024.e41125).
- 19 S. Suguna, D. Parimala devi, A. Abiram, P. Mukhil sukitha, V. Rajesh kannan, R. Suresh Kumar, A. I. Almansour, K. Perumal, J. Prabhu and R. Nandhakumar, Symmetric and disulfide linked reversible fluorescent organic material: A chemosensor for Pb²⁺ ion and its applications in real world sample analysis, *J. Photochem. Photobiol., A*, 2023, **442**, 114777, DOI: [10.1016/j.jphotochem.2023.114777](https://doi.org/10.1016/j.jphotochem.2023.114777).
- 20 C. I. L. Justino, A. C. Freitas, R. Pereira, A. C. Duarte and T. A. P. Rocha Santos, Recent developments in recognition elements for chemical sensors and biosensors, Elsevier B. V, *TrAC, Trends Anal. Chem.*, 2015, **68**, 2–17, DOI: [10.1016/j.trac.2015.03.006](https://doi.org/10.1016/j.trac.2015.03.006).
- 21 X. Wang, X. Lu and J. Chen, Development of biosensor technologies for analysis of environmental Contaminants, *Trends Environ. Anal. Chem.*, 2014, **2**, 25–32, DOI: [10.1016/j.teac.2014.04.001](https://doi.org/10.1016/j.teac.2014.04.001).
- 22 B. Yue Zhang, L. Shi, X. Ying Ma, L. Liu, Y. Fu and X. Feng Zhang, Advances in the Functional Nucleic Acid Biosensors for Detection of Lead Ions, *Crit. Rev. Anal. Chem.*, 2023, **53**(2), 309–325, DOI: [10.1080/10408347.2021.1951648](https://doi.org/10.1080/10408347.2021.1951648).
- 23 S. Zhan, Y. Wu, L. Wang, X. Zhan and P. Zhou, A mini-review on functional nucleic acids-based heavy metal ion detection, *Biosens. Bioelectron.*, 2016, **86**, 353–368, DOI: [10.1016/j.bios.2016.06.075](https://doi.org/10.1016/j.bios.2016.06.075).
- 24 J. Li and Y. Lu, A highly sensitive and selective catalytic DNA biosensor for lead ions, *J. Am. Chem. Soc.*, 2000, **122**(42), 10466–10467, DOI: [10.1021/ja0021316](https://doi.org/10.1021/ja0021316).
- 25 J. Liu and Y. Lu, A colorimetric lead biosensor using DNzyme-directed assembly of gold nanoparticles, *J. Am. Chem. Soc.*, 2003, **125**, 6642–6643, DOI: [10.1021/ja034775u](https://doi.org/10.1021/ja034775u).
- 26 H. Sun, L. Yu, H. Chen, J. Xiang, X. Zhang, Y. Shia, Q. Yang, A. Guan, Q. Lia and V. Tang, A colorimetric lead(II) ions sensor based on selective recognition of G-quadruplexes by a clip-like cyanine dye, *Talanta*, 2015, **136**, 210–214, DOI: [10.1016/j.talanta.2015.01.027](https://doi.org/10.1016/j.talanta.2015.01.027).
- 27 S. Zhan, Y. Wu, L. Liu, H. Xing, L. He, X. Zhan, Y. Luobe and P. Zhou, A simple fluorescent assay for lead(II) detection based on lead(II)-stabilized G-quadruplex formation, *RSC Adv.*, 2013, **3**, 16962, DOI: [10.1039/c3ra42621a](https://doi.org/10.1039/c3ra42621a).
- 28 T. Li, E. Wang and S. Dong, Lead(II)-induced allosteric G-quadruplex DNzyme as a Colorimetric and chemiluminescence sensor for highly sensitive and selective Pb²⁺ detection, *Anal. Chem.*, 2010, **82**, 1515–1520, DOI: [10.1021/ac902638v](https://doi.org/10.1021/ac902638v).
- 29 F. Li, L. Yang, M. Chen, Y. Qian and B. Tang, A novel and versatile sensing platform based on HRP-mimicking DNzyme-catalyzed template-guided deposition of polyaniline, *Biosens. Bioelectron.*, 2013, **41**, 903–906, DOI: [10.1016/j.bios.2012.09.048](https://doi.org/10.1016/j.bios.2012.09.048).
- 30 S. Dasgupta, S. A. Shelke, N. S. Li and J. A. Piccirilli, Spinach RNA aptamer detects lead(II) with high selectivity, *Chem. Commun.*, 2015, **51**, 9034–9037, DOI: [10.1039/C5CC01526J](https://doi.org/10.1039/C5CC01526J).
- 31 J. Mohanty, N. Barooah, V. Dhamodharan, S. Harikrishna, P. I. Pradeepkumar and A. C. Bhasikuttan, Thioflavin T as an efficient inducer and selective fluorescent sensor for the human telomeric G-quadruplex DNA, *J. Am. Chem. Soc.*, 2013, **135**, 367–376, DOI: [10.1021/ja309588h](https://doi.org/10.1021/ja309588h).
- 32 T. Bradford, P. A. Summers, A. Majid, P. S. Sherin, J. Y. L. Lam, S. Aggarwal, J. B. Vannier, R. Vilar and M. K. Kuimova, Imaging G-Quadruplex Nucleic Acids in Live Cells Using Thioflavin T and Fluorescence Lifetime Imaging Microscopy, *Anal. Chem.*, 2024, **96**(51), 20223–20229, DOI: [10.1021/acs.analchem.4c04207](https://doi.org/10.1021/acs.analchem.4c04207).
- 33 L. Sjekloča and A. R. Ferré-D'Amaré, Binding between G Quadruplexes at the Homodimer Interface of the Corn RNA Aptamer Strongly Activates Thioflavin T Fluorescence, *Cell Chem. Biol.*, 2019, **26**, 1159–1168, DOI: [10.1016/j.chembiol.2019.04.012](https://doi.org/10.1016/j.chembiol.2019.04.012).
- 34 A. R. De La Faverie, A. Guédin, A. Bedrat, L. A. Yatsunyk and J. L. Mergny, Thioflavin T as a fluorescence light-up probe for G4 formation, *Nucleic Acids Res.*, 2014, **42**(8), e65, DOI: [10.1093/nar/gku111](https://doi.org/10.1093/nar/gku111).
- 35 G. S. Filonov, J. D. Moon, N. Svensen and S. R. Jaffrey, Broccoli: Rapid selection of an RNA mimic of green fluorescent protein by fluorescence-based selection and directed evolution, *J. Am. Chem. Soc.*, 2014, **136**, 16299–16308, DOI: [10.1021/ja508478x](https://doi.org/10.1021/ja508478x).
- 36 J. Wu, N. Svensen, W. Song, H. Kim, S. Zhang, X. Li and S. R. Jaffrey, Self-Assembly of Intracellular Multivalent RNA Complexes Using Dimeric Corn and Beetroot Aptamers, *J. Am. Chem. Soc.*, 2022, **144**, 5471–5477, DOI: [10.1021/jacs.1c13583](https://doi.org/10.1021/jacs.1c13583).
- 37 K. Y. S. Kong, S. C. Y. Jeng, B. Rayyan and P. J. Unrau, RNA Peach and Mango: orthogonal two-color fluorogenic aptamers distinguish nearly identical ligands, *RNA*, 2021, **27**(5), 604–615, DOI: [10.1261/rna.078493.120](https://doi.org/10.1261/rna.078493.120).
- 38 R. J. Trachman, A. Autour, S. C. Y. Jeng, A. Abdolazadeh, A. Andreoni, R. Cojocar, R. Garipov, E. V. Dolgosheina, J. R. Knutson, M. Ryckelynck, P. J. Unrau and A. R. Ferré-D'Amaré, Structure and functional reselection of the Mango-III fluorogenic RNA aptamer, *Nat. Chem. Biol.*, 2019, **15**, 472–479, DOI: [10.1038/s41589-019-0267-9](https://doi.org/10.1038/s41589-019-0267-9).
- 39 E. V. Dolgosheina, S. C. Y. Jeng, S. S. S. Panchapakesan, R. Cojocar, P. S. K. Chen, P. D. Wilson, N. Hawkins, P. A. Wiggins and P. J. Unrau, RNA Mango aptamer-fluorophore: A bright, high-affinity complex for RNA labeling and tracking, *ACS Chem. Biol.*, 2014, **9**, 2412–2420, DOI: [10.1021/cb500499x](https://doi.org/10.1021/cb500499x).



- 40 A. Autour, S. C. Y. Jeng, A. D. Cawte, A. Abdolazadeh, A. Galli, S. S. S. Panchapakesan, D. Rueda, M. Ryckelynck and P. J. Unrau, Fluorogenic RNA Mango aptamers for imaging small non-coding RNAs in mammalian cells, *Nat. Commun.*, 2018, **9**, 656, DOI: [10.1038/s41467-018-02993-8](https://doi.org/10.1038/s41467-018-02993-8).
- 41 R. del Villar-Guerra, J. O. Trent and J. B. Chaires, G-Quadruplex Secondary Structure Obtained from Circular Dichroism Spectroscopy, *Angew. Chem., Int. Ed.*, 2018, **57**, 7171–7175, DOI: [10.1002/anie.201709184](https://doi.org/10.1002/anie.201709184).
- 42 N. V. Malanova, V. V. Korobochkin and V. I. Kosintsev, The Application of Ammonium Hydroxide and Sodium Hydroxide for Reagent Softening of Water, *Procedia Chem.*, 2014, **10**, 162–167, DOI: [10.1016/j.proche.2014.10.028](https://doi.org/10.1016/j.proche.2014.10.028).
- 43 A. Al-Gailani, T. V. J. Charpentier, O. Sanni, R. Crisp, J. H. Bruins and A. Neville, Inorganic mineral precipitation from potable water on heat transfer surfaces, *J. Cryst. Growth*, 2020, **537**, 125621, DOI: [10.1016/j.jcrysgro.2020.125621](https://doi.org/10.1016/j.jcrysgro.2020.125621).

