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Antimicrobial textiles based on photocrosslinked poly(ethylene-*co*-acrylic acid)†

Yimin Zeng [ID] and Michael O. Wolf [ID] *

Photocrosslinking of a series of amine-containing poly (ethylene-*co*-acrylic acids) (PEAAs) on textile surfaces by reaction of photogenerated singlet oxygen at the amine groups to form imines is reported. The materials are hydrophilic and show antimicrobial activity. Modified PEAAs with 3 different lengths of side chains are synthesized to study the effect of varying side chain length on polymer photooxidative crosslinking and antimicrobial activity. Materials with longer side chains show elimination of *E. coli* and MRSA by contact lysis which is enhanced under irradiation with green light.

Introduction

Viral and bacterial infections can be transmitted by surface contamination.[1] Bacterial infections are the second leading cause of death worldwide, affecting individuals in all age groups. In 2019, an estimated 7.7 million deaths were attributed to 33 bacterial pathogens, constituting 13.6% of all worldwide fatalities.[2] Generally, disinfectants are used to sanitize surfaces to prevent infection by bacteria, viruses, and other pathogens. This labour-intensive process involves repeatedly wiping or spraying disinfectants to ensure continuous sanitization. In the process, volatile and toxic disinfectants can be released into the environment possibly resulting in ecological damage. To address this issue, modified surfaces that kill or inhibit the growth of microbes to reduce or eliminate disease transmission have been developed.[3] Modification of textiles with coatings that have inherent antimicrobial properties that can prevent the spread of diseases is one approach that is being explored. By applying an antimicrobial coating to textiles, the spread of pathogens can be mitigated in health care and other high-risk settings, for example on hospital gowns, bedding, and clothing.[4,5]

Antimicrobial surfaces can be prepared by functionalization with primary amine or phosphonium containing polymers, quaternary ammonium salts (QASs), or metal nanoparticles.[6–10] Surfaces modified with these functionalities can act to kill bacteria through passive pathways, such as contact lysis or ion release, where the bacterial cell membranes are disrupted. Hydrophobic polymer coatings can also prevent

moisture accumulation and reduce microbial adhesion on surfaces.[11] Additionally, photosensitizers can be used to generate singlet oxygen ($^1O_2$, the first excited state of oxygen) or free radical ions that destroy microbial membranes, a process called antimicrobial photodynamic inactivation (aPDI).[12–14] Coatings based on QASs and metal nanoparticles are passive while aPDI gives the possibility of active photosensitization utilizing light energy to destroy pathogens *via* singlet oxygen generation. Typically, the photosensitizer is excited to a singlet state with visible light and then undergoes intersystem crossing to a triplet state, followed by triplet–triplet energy transfer to ground state oxygen, forming $^1O_2$.[15,16]

Our group recently developed dual-functional antimicrobial polymers, which employ primary amine-containing aminopropylmethylsiloxane (6–7 wt%)-dimethylsiloxane copolymers (PDMS-NH$_2$) with a covalently attached photosensitizer, rose bengal (RB), that generate $^1O_2$ on irradiation with green light.[17] Others have trapped RB in the pores of organic polymers on cotton surfaces and demonstrated biocidal activity.[18] Our PDMS-NH$_2$ polymer is photocrosslinked by reaction with $^1O_2$ at the amine groups to form imines. By forming covalent bonds between polymer chains a robust coating on textile surfaces results which prevents polymer leaching into the aqueous environment in real-world applications.[11,19] Textiles can be modified by soaking with a solution of the polymer followed by photocrosslinking resulting in the formation of a durable antimicrobial coating.[17,20]

Photocrosslinking PDMS-NH$_2$ results in a hydrophilic coating on textile substrates attributed to the presence of the alkyl side chains on the polysiloxane backbone. Although a hydrophobic polymer coating on fabric can reduce surface bacterial growth, there are disadvantages associated with the use of hydrophobic materials, such as poor durability in clothing applications.[21] Moreover, silicones are not biodegradable or recyclable raising sustainability concerns. Here we report

Department of Chemistry, University of British Columbia, 2036 Main Mall, Vancouver, British Columbia, V6T 1Z1, Canada. E-mail: mwolf@chem.ubc.ca

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crosslinking of a hydrophilic carbon-based polymer, poly(ethylene-*co*-acrylic acid) (PEAA) that contains 15 wt% of carboxylic acid groups, and exploration of the antimicrobial properties of textile coatings based on this polymer. PEAA is widely applied in biosensor fabrication and in the food packaging industry due to its adhesive properties towards polar substrates.[22,23] In 2017, Noh and co-workers reported an antimicrobial material consisting of QAS grafted onto PEAA.[24] In the present work, both the benefits of PEAA as a carbon-based polymer and the dual-functional antimicrobial approach pioneered by our group are combined. Modified PEAAs with 3 different lengths of side chains are synthesized to study the effect of varying side chain length on polymer photooxidative crosslinking and antimicrobial activity. The modified PEAA-NH$_2$ polymers, **PEAA-3C-NH$_2$**, **PEAA-6C-NH$_2$**, and **PEAA-8C-NH$_2$** have amine groups that have 3 carbon, 6 carbon, and 8 carbon spacers between the polymer backbone and amine group, respectively.

Results and discussion

PEAA-NH$_2$ was synthesized by reacting PEAA with *N,N*-dicyclohexylcarbodiimide (DCC) and diamino alkanes with three different alkyl chain lengths (Scheme 1). Chain lengths were selected based on readily available diaminoalkanes, where the longer chain alkanes are typically more costly. 1,3-Diaminopropane, 1,6-diaminohexane and 1,8-diaminooctane were used to leave a terminal amino group available for photocrosslinking after formation of an amide link with one of the amino groups. **PEAA-3C-NH$_2$**, **PEAA-6C-NH$_2$**, and **PEAA-8C-NH$_2$** were prepared, in each case the product after coupling was isolated as a fine white powder, which precipitated with methanol and was washed by centrifugation. Cotton pieces are soaked in
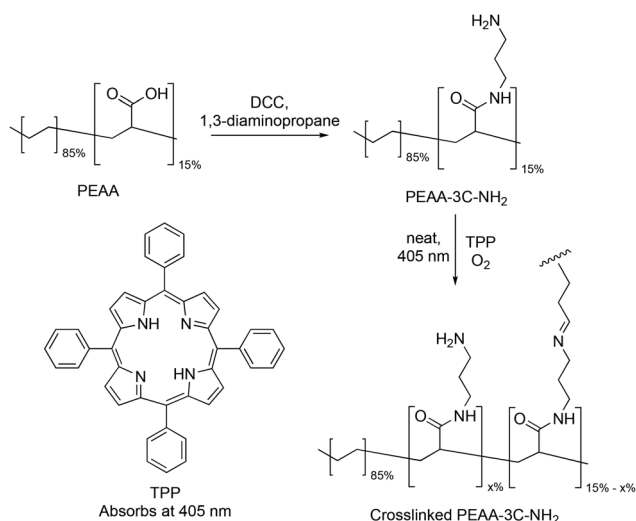
a hot DMSO solution of the polymer to coat the textile surface followed by crosslinking with light to create a robust coating.

The PEAA-NH$_2$ solution-treated cotton has both passive and active photosensitized functionality to kill microbes, through contact lysis and singlet oxygen formation. Crosslinking of PEAA-NH$_2$ was carried out by $^1O_2$ produced by tetraphenylporphyrin (TPP) co-adsorbed with the polymer on irradiation with a 36 W 405 nm LED bulb for 30 minutes. In a second step, polymer-coated cotton was washed with THF to remove remaining TPP and then soaked in a 0.1 wt% ethanol solution of RB to introduce the photosensitizer (Fig. 1). TPP is used to crosslink the polymer on the textile surface while RB is present to generate $^1O_2$ to kill microbes under irradiation. TPP photosensitization is not initiated by the 530 nm green light used for the antimicrobial tests.

All three PEAA-NH$_2$ derivatives are insoluble or very poorly soluble in most organic solvents and water, and only partially dissolve in dimethyl sulfoxide (DMSO) at 120 °C. Synthesis of PEAA-NH$_2$ is confirmed by a shift in the α-H peak in the ^{1}H NMR spectrum at 120 °C from 2.22 ppm in PEAA to 2.51 ppm in PEAA-NH$_2$ due to the formation of the amide bond (Fig. S2–S7†). The NMR spectrum of **PEAA-6C-NH$_2$** could not be obtained due to the poor solubility of this polymer. In the FTIR spectrum, PEAA shows a carbonyl stretch due to the carboxylic acid groups at 1701 cm^{-1}. After coupling, the characteristic carboxylic acid peak of PEAA is no longer present, and an amide C=O stretch at 1639 cm^{-1} and an amine N–H bend at 1539 cm^{-1} appear, confirming the conversion from PEAA to PEAA-NH$_2$. An imine C=N stretch at 1690 cm^{-1} is observed in the crosslinked PEAA-NH$_2$ along with the presence of amine and amide peaks (Fig. 2). The FTIR spectrum shows that not all amine groups are involved in crosslinking PEAA-NH$_2$, thus the remaining free amine groups in the polymer can remain active for contact lysis. Crosslinking does not occur if PEAA-NH$_2$ is irradiated without TPP, indicating the necessity of the photosensitizer to be present for crosslinking. Ninhydrin tests were performed to determine the percentage of amine groups remaining after crosslinking (Table S3†). The percentage of amine remaining after crosslinking is 62 ± 5% for **PEAA-3C-NH$_2$**, 24 ± 2% for **PEAA-6C-NH$_2$**, and 23 ± 4% for **PEAA-8C-NH$_2$**. Based on this data, **PEAA-3C-NH$_2$** showed less crosslinking compare with the other two polymers.

The average water contact angle of cotton treated with 1 wt% PDMS-NH$_2$ is 134 ± 9°, while for all PEAA-NH$_2$ samples water is absorbed, and a contact angle cannot be measured on the cotton surface (Fig. S10†). This indicates that while the
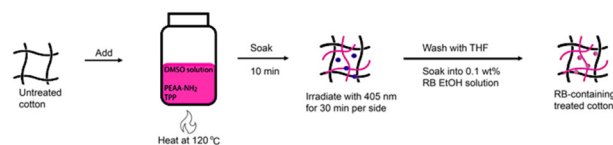

Scheme 1 Synthesis of **PEAA-3C-NH$_2$**. **PEAA-6C-NH$_2$** and **PEAA-8C-NH$_2$** are prepared the same way using 1,6-diaminohexane and 1,8-diaminooctane respectively.

Fig. 1 Scheme of textile treatment with solution soak and light irradiation resulting in RB-containing PEAA-NH$_2$ solution-treated cotton.
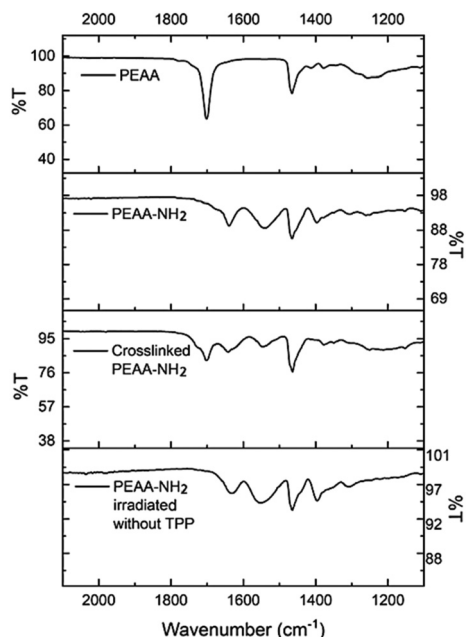
Fig. 2 A comparison of FTIR spectra of PEAA, **PEAA-8C-NH$_2$**, cross-linked **PEAA-8C-NH$_2$**, and **PEAA-8C-NH$_2$** irradiated without photosensitizer. Similar spectra were observed for **PEAA-3C-NH$_2$**, **PEAA-6C-NH$_2$**, and **PEAA-8C-NH$_2$** between 1100 and 2100 cm^{-1}.

1 wt% PDMS-NH$_2$ solution-treated cotton is hydrophobic, the PEAA-NH$_2$ samples are very hydrophilic. Tensile tests (fabric stretching) were performed to probe if the physical properties of the modified fabrics are changed from prior to treatment. The load average of untreated plain cotton is 16 ± 2 N and 17 ± 2 N for 1 wt% **PEAA-8C-NH$_2$** solution-treated cotton (Table S2†). Thus, polymer treatment results in a minimal tensile strain change.

Antimicrobial tests were carried out with *Escherichia coli* (*E. coli*) and methicillin resistant *Staphylococcus aureus* (MRSA) with a concentration of 10^5 bacteria per mL. Identical sized pieces of coated cotton were used to compare the bacterial killing efficiency between PDMS-NH$_2$ and the three PEAA-NH$_2$ derivatives. To allow generation of $^1O_2$ under irradiation, all coated fabrics were soaked in a 0.1 wt% RB solution for 3 min. The average loadings of RB on fabrics treated with **PEAA-3C-NH$_2$**, **PEAA-6C-NH$_2$** and **PEAA-8C-NH$_2$** were determined by measuring the absorption of extracted RB at 560 nm (Table S4†). Unmodified cotton was also treated with RB under the same conditions for comparison. The average weight percentages (wt%) of RB on solution treated fabric were determined to be 3.7 ± 0.1% for plain cotton, 3.3 ± 0.2% for **PEAA-3C-NH$_2$** solution-treated cotton, 3.0 ± 0.2% for **PEAA-6C-NH$_2$** solution-treated cotton, and 3.3 ± 0.4% for **PEAA-8C-NH$_2$** solution-treated cotton. The PEAA coatings result in only a minor change in the RB loading relative to plain cotton.

Contact lysis experiments were carried out by soaking the polymer solution-treated cotton sample in a bacterial culture solution in the dark. Photosensitization was conducted by irra-

diating a bacterial culture containing coated fabrics with a 15 W 530 nm green light. In both experiments, the bacterial cultures were assessed over 150 min to monitor bacterial growth under treatment. No reduction in the number of bacteria was observed when bacterial solutions or untreated textiles were irradiated with green light (Fig. S11†). Both 0.5 wt% **PEAA-8C-NH$_2$** and 0.5 wt% PDMS-NH$_2$ solution-treated cotton behaved the same by gradually reducing the number of *E. coli* colony-forming units (CFUs) over 150 min (Fig. S12†). In this case, the 0.5 wt% **PEAA-8C-NH$_2$** solution-treated cotton showed an average of 85% of *E. coli* elimination and 30% killing of MRSA through contact lysis (Fig. 3 and Fig. S13†). All bacteria were killed under 10 min irradiation with green light. This indicates that changing the hydrophobicity of the polymer coating significantly does not change the dual-functional antimicrobial capability. The bacterial killing behaviour under irradiation is dominated by $^1O_2$ produced by excitation of the RB. Identical treatment and antimicrobial assays were carried out with **PEAA-3C-NH$_2$** and **PEAA-6C-NH$_2$** in comparison to **PEAA-8C-NH$_2$** to study the impact of amine chain
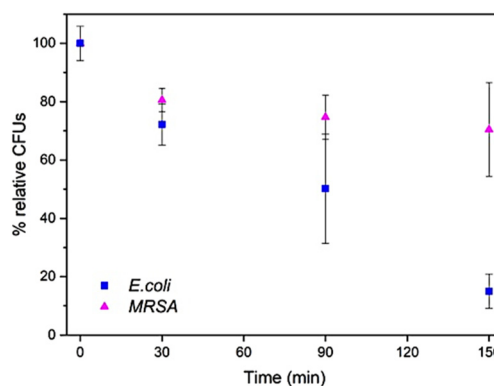

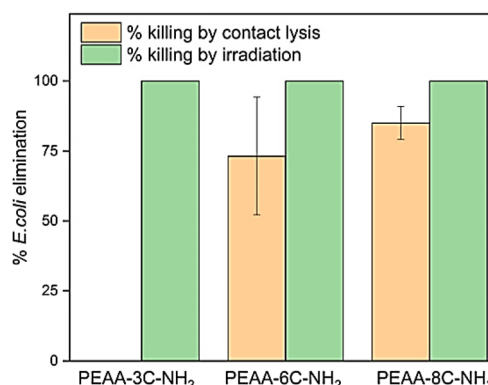**Fig. 3** A comparison of contact lysis for *E. coli* and MRSA on 0.5 wt% **PEAA-8C-NH$_2$** solution-treated cotton over time.

Fig. 4 A comparison of percent *E. coli* elimination by singlet oxygen and contact lysis on 0.5 wt% **PEAA-3C-NH$_2$**, **PEAA-6C-NH$_2$**, and **PEAA-8C-NH$_2$** solution-treated cotton over time. *E. coli* killing by singlet oxygen was performed under 10 min irradiation with green light, where contact lysis was performed in the dark over 150 min.

length on bacterial killing. All bacteria were killed with 10 min irradiation by **PEAA-3C-NH$_2$**, **PEAA-6C-NH$_2$**, and **PEAA-8C-NH$_2$**. **PEAA-6C-NH$_2$** behaved the same as both **PEAA-8C-NH$_2$** and **PDMS-NH$_2$**, while **PEAA-3C-NH$_2$** had no effect on bacteria present by contact lysis, even with a higher amount of amine groups remained after crosslinking (Fig. 4 and Table S1†). This can be rationalized by the short alkyl chain in **PEAA-3C-NH$_2$** which is less flexible for insertion into the bacteria cell membrane compared with the longer analogs. Although PDMS-NH$_2$ also has amine groups that are 3 carbons away, the siloxane backbone in PDMS-NH$_2$ may be more flexible than the carbon backbone in **PEAA-3C-NH$_2$**.

Conclusions

In summary, we report a series of new dual-functional organic antimicrobial polymer analogs to PDMS-NH$_2$ to explore the effect of modifying the hydrophilicity of polymer coatings on fabrics while retaining its bacterial killing property. **PEAA-3C-NH$_2$**, **PEAA-6C-NH$_2$**, and **PEAA-8C-NH$_2$** can all be crosslinked under 405 nm light with an average of 36% amine remained after crosslinking. **PEAA-6C-NH$_2$** and **PEAA-8C-NH$_2$** behave the same as PDMS-NH$_2$ in antimicrobial tests, where *E. coli* and MRSA were eliminated over 150 min by contact lysis and 10 min under irradiation of green light. The PEAA-NH$_2$ treatment results in a hydrophilic coating on fabrics while not changing the tensile properties of the textile.

Experimental section

General

PEAA and RB lactone (95%) were obtained from Sigma-Aldrich. DCC, 1,3-diaminopropane, 1,6-diaminohexane, and 1,8-diaminooctane were purchased from Thermo Fisher Scientific. TPP was purchased from STREM. Synthesis of PEAA-NH$_2$ was carried out under nitrogen with properly dried glassware and solvents. ^{1}H, HSQC and HMBC NMR experiments were carried out to obtain NMR peak assignments in the NMR lab at UBC. FTIR experiments were carried out in the Shared Instrument Facility (SIF) at UBC. Scheme 1 was followed to synthesize the desired products.

PEAA-8C-NH$_2$

0.5 g of PEAA was mixed with 0.1 g (0.48 mmol) of DCC and 0.8 g (5.5 mmol) of 1,8-diaminooctane. The neat reaction mixture was heated at 120 °C under N$_2$ for 24 h. The desired amine containing PEAA-NH$_2$ was precipitated with methanol and washed by centrifugation. The product was dried under vacuum. ^{1}H NMR (120 °C, 400 MHz, DMSO): δ 2.51 (m, α-H, overlap with DMSO), δ 2.58 (t, amide-H$_2$).

PEAA-3C-NH$_2$ and **PEAA-6C-NH$_2$** were prepared using the same procedure with excess 1,3-diaminopropane or 1,6-diaminohexane instead of 1,8-diaminooctane.

Polymer crosslinking

In 100 μL of DMSO solution, 50 mg of PEAA-NH$_2$ was mixed with 0.1 mg (0.16 μmol) of TPP. The reaction mixture was irradiated with a 36 W 405 nm LED bulb for 30 min at 120 °C to crosslink the PEAA-NH$_2$.

Textile treatment

Cotton pieces were soaked in polymer solutions in DMSO at 120 °C with TPP for 10 min in the dark. DMSO was removed by soaking fabrics in stirring water. The fabrics were air dried and irradiated with a 36 W 405 nm LED bulb for 30 minutes per side. The light bulb was placed 9 cm above the samples. Once the polymer was coated onto cotton, residual TPP was removed by a THF wash. The treated cottons were soaked into a 0.1 wt% RB EtOH solution for 3 min and dried under vacuum.

Loading of RB on treated fabrics

After treating coated fabrics with 0.1 wt% RB EtOH solution, RB was redissolved into 8 mL of EtOH by sonication. The absorption of redissolved RB was measured at 560 nm using a Cary 5000 UV/Vis spectrometer (Agilent Technologies Inc., Santa Clara).

Antimicrobial tests

E. coli and MRSA (level 2, from the Biological Services Laboratory, UBC Chemistry) were used. Samples from the same batch of tensile tests were employed to ensure identical surface area. The cultures were prepared with a concentration of 10^5 bacteria per mL and 10 mL was transferred into Falcon tubes contain polymer-coated cotton. Samples were treated with or without irradiation for comparison. A 15 W 530 nm LED bulb was set 9 cm above the samples. The optical power at the center of the lamp was 13 mW cm^{-2} measured using a thermal power meter (Ophir 3A-P-SH). 20 μL of sample solution was taken at each time point and incubated on agar gel. Results were collected in triplicate.

Hydrophobicity test

A drop of deionized water (10 μL) was placed on each fabric sample and measured from the horizon. Photos were taken with an iPhone XR and analysed by ImageJ.

Tensile tests

Polymer coated cottons were cut into identical dog bone shape (10 mm in the course direction and 37 mm in the wale direction) by laser cutting. Tensile measurements on polymer-treated cottons were measured until the break point using an Instron 5980 instrument under a 2 kN load at 20 cm min^{-1} with a clamp distance of 30 mm. Results were averaged from ten samples for each comparison group.

Ninhydrin test

50 mg of PEAA-NH$_2$ was mixed with 0.1 mg (0.16 μmol) of TPP in 100 μL of DMSO. The reaction mixture was irradiated under

identical conditions to those used to crosslink PEAA-NH$_2$. For comparison, samples were also prepared the same way without TPP. A ninhydrin solution in ethanol (10 μL, 3.5 mg mL^{-1}) was added to samples while heating. The absorption of ninhydrin after reaction was obtained using an Infinite M1000 Pro plate reader (Tecan Ltd, Morrisville) on a non-binding 96-well plate. Measurements were carried out in triplicate.

Author contributions

The manuscript was written through contributions of all authors.

Data availability

The data supporting this article have been included as part of the ESI.†

Conflicts of interest

The materials disclosed in this paper are included in a provisional patent that has been filed.

Acknowledgements

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