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# Metallohelices that kill Gram-negative pathogens using intracellular antimicrobial peptide pathways†

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A range of new water-compatible optically pure metallohelices – made by self-assembly of simple non-peptidic organic components around Fe ions – exhibit similar architecture to some natural cationic antimicrobial peptides (CAMPs) and are found to have high, structure-dependent activity against bacteria, including clinically problematic Gram-negative pathogens. A key compound is shown to freely enter rapidly dividing *E. coli* cells without significant membrane disruption, and localise in distinct foci near the poles. Several related observations of CAMP-like mechanisms are made *via* biophysical measurements, whole genome sequencing of tolerance mutants and transcriptomic analysis. These include: high selectivity for binding of G-quadruplex DNA over double stranded DNA; inhibition of both DNA gyrase and topoisomerase I *in vitro*; curing of a plasmid that contributes to the very high virulence of the *E. coli* strain used; activation of various two-component sensor/regulator and acid response pathways; and subsequent attempts by the cell to lower the net negative charge of the surface. This impact of the compound on multiple structures and pathways corresponds with our inability to isolate fully resistant mutant strains, and supports the idea that CAMP-inspired chemical scaffolds are a realistic approach for antimicrobial drug discovery, without the practical barriers to development that are associated with natural CAMPs.

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## Introduction

Bacteria naturally evolve resistance to most antimicrobial compounds, and the widespread careless use of antibiotics is exacerbating the situation. When coupled with a dearth of discovery of new antimicrobial drugs, antimicrobial resistance has emerged as a grave threat to current and future medical practice and socioeconomic activity, as recently highlighted by numerous organizations and governments.<sup>1,2</sup>

Antimicrobial peptides are part of the innate immune systems of plants and animals that evolved to defend against a range of microbes.<sup>3,4</sup> They are typically small molecules, comprising 10–50 amino acids. The majority – the so-called cationic antimicrobial peptides (CAMPs) – have an excess of both cationic and hydrophobic subunits arranged into secondary structures, such that the folded assembly exhibits an amphipathic or patchy charge distribution. While the surfaces of eukaryotic cell membranes are typically neutral, those of bacteria are intrinsically anionic due to the incorporation of negatively charged phospholipids, teichoic acid or lipopolysaccharides (LPS).<sup>5,6</sup> Therefore, CAMPs have strong electrostatic interactions with bacterial cell membranes but not those of the host cells, providing generalised specificity. In some cases, the site of action is the membrane itself, with the CAMP inducing fatal changes in membrane structure and fluidity, or the creation of ungated pores allowing uncontrolled solute exchange with the environment.<sup>7,8</sup> Alternatively, CAMPs may cross the membrane barrier and interact with intracellular targets with lethal consequences<sup>9–11</sup> or indeed have combinations of modes of action.<sup>4,12</sup> CAMPs and derivatives have been investigated as clinical antibiotics,<sup>13–15</sup> but unfortunately their pharmacokinetic profiles tend to be unfavourable and large scale manufacture presents a challenge.<sup>16</sup> Nevertheless it is worth noting

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that a number of CAMPs are currently under commercial development, with attempts being made to enhance the properties of natural systems using strategies such as peptide stapling<sup>17</sup> to improve structural integrity and enzymatic stability.

Our recently-developed ranges of cationic metallohelices, based on helical arrays of fully-encapsulated Fe ions (*e.g.* Scheme 1), have similar diameter to CAMPs, are optically pure, water-soluble and are available *via* efficient self-assembly on a multi-gram scale. Recently we have shown that one class<sup>18,19</sup> has a facially amphipathic architecture akin to that of natural antifreeze peptides<sup>20</sup> and a similar structure binds to the central hydrophobic  $\alpha$ -helical region of an amyloid  $\beta$  protein and attenuates toxicity.<sup>21</sup> Members of another class<sup>22</sup> have excellent and highly selective activity against a variety of cancer cell lines. In stark contrast, none of the above ranges have significant antimicrobial activity. Intriguingly however, a lone prototype compound of another class<sup>23</sup> **5a** (Scheme 1), was shown to have good activity against a limited panel of microbes. A new synthetic protocol has opened up this area of chemical space

and we report here compounds with structure-dependent activity against Gram-positive and -negative bacteria. One potent compound is shown to freely enter *E. coli* cells and to exhibit antimicrobial peptide-like mechanisms of action.

## Results and discussion

We reported that optically pure phenylglycinol **1** (Scheme 1) reacts in the presence of NaH with benzylic bromide **2a** to give diamine **3a**.<sup>24</sup> However, all other examples of such etherifications we have attempted led to inseparable mixtures containing the *N*-alkylated isomers. After extensive experimentation with solvents, bases and *N*-protection strategies we found that performing the reaction in the presence of [15]-crown-[5]<sup>25</sup> gave clean products. The optical purity of the subsequent range of new diamines **3b–h** was established *via* synthesis of the (*R*)-(+)-Mosher's acid derivatives (ESI<sup>†</sup>). This achieved, the self-assembly of the various metallohelices incorporating 2-pyridinecarboxaldehyde **4** and M(II) salts proceeded smoothly to give a range of structures.

With iron(II) chloride as the metal ion source, both enantiomers of each of the water-soluble systems **5a–h** were produced while zinc(II) perchlorate gave the isostructural analogues **6a–h**. All compounds were characterised by multi-nuclear NMR techniques, mass spectrometry, microanalysis, and where possible single crystal X-ray diffraction. Circular dichroism (CD) spectroscopy, thermo-gravimetric analysis (TGA), IR and UV-visible spectroscopies were additionally applied to the Fe compounds.

The complexes contained a single set of ligand peaks in the <sup>1</sup>H and <sup>13</sup>C NMR spectra (ESI) consistent with the presence of single diastereomers (Fig. 1), and since the ligands are optically pure, the complexes are single enantiomers. As we have described in detail,<sup>18,23,25,26</sup> the mechanisms for the extremely reliable diastereoselection using our  $\alpha$ -methylbenzyliminopyridine unit used in these compounds also contribute substantially to the unusual stability to hydrolysis (*vide infra*).

Slow vapour diffusion of ethyl acetate into a concentrated acetonitrile solution of **6b** afforded single crystals suitable for XRD. The cationic unit of the compound (Fig. 2) shows the approximate octahedral coordination environments at the Zn centres, and the now familiar inter-ligand  $\pi$ - $\pi$  stacking interactions, responsible in large part for the excellent diastereoselection.<sup>18,23,26</sup>

The overall size and shape of the **6b** cation is more markedly folded or “concertinaed” than **6a**.<sup>26</sup> The inter-metallic distance is substantially shortened (12.4 Å *vs.* 14.0 Å) while the circumference of the molecule is increased, as evidenced by the average distance between the centroids of each xylenyl aromatic ring: 7.6 Å, *vs.* 6.3 Å for **6a**. Correspondingly the *m*-xylenyl bridge also folds with a more pronounced helical turn along the metal-metal axis: torsion angles between Zn–N(imine) bonds from the same ligand, along the Zn–Zn axis [*e.g.* N(6)–Zn(1)–Zn(2)–N(7)] range from 100.9° to 101.0°. It should be noted however that as with **6a** the *P*-helicity generated by each  $\Lambda$ -Zn centre is inverted at the C-stereogenic centre such that the bridge units fold with *M*-helicity. In **6b** all the *m*-xylenyl groups are oriented such that



**Scheme 1** Synthesis and characterisation of new metallohelices. Self-assembly of series 5 and 6 incorporating a range of linker units *via* optically pure diamines **3**.





Fig. 1  $^1\text{H-NMR}$  spectra of metallohelices  $\Delta\text{-6b}$  and  $\Delta\text{-6h}$  (500 MHz in  $d^3\text{-acetonitrile}$  at 298 K). Note the presence of single diastereomers and the unusual chemical shifts of the nuclei at position b in both cations.



Fig. 2 Structure of the cationic unit of  $\Delta_{\text{zn}}\text{-6b}$ . Thermal ellipsoid plot (50% probability) with H atoms, solvent molecules and counterions removed for clarity.



Fig. 3 Structure of the cationic unit of **6e**. Thermal ellipsoid plot (50% probability) with H atoms, solvent molecules and counterions removed for clarity.

the C-H groups at C(21), C(57), and C(93) point towards the centre of the complex. Since the plane of each xylenyl group intersects this cavity, the centre of the complex is expected to be a deshielding environment as a result of the coinciding ring currents. This explains the unusually downfield-shifted  $^1\text{H-NMR}$  signal for these nuclei (8.30 ppm). The system **6h** (Fig. 1) has a similar feature.

Naturally, the longer bridge of thioether **6e** results in a longer complex than **6a** or **b** with an inter-metal distance of 17.4 Å (Fig. 3). Interestingly, the 'hinge' created by the sulfur atoms

appears to aid the folding process by allowing the bridge to arch outwards, with the two flanking *p*-tolyl units able to orient separately, and furthermore allowing the bridge a remarkable twist along the metal-metal axis.

The torsion angles between Zn-N(imine) bonds from the same ligand range from  $167.3^\circ$  to  $179.1^\circ$ . As a consequence of this arching of the bridge the sulfur atoms protrude from the helix and the system appears to have a larger circumference than earlier examples (S-S distances range 8.0 Å to 9.2 Å).

The structure does not appear to be distorted significantly from threefold symmetry; the C-S-C bond angles [e.g. C(19)-S(1)-C(22)] fall within a narrow and conventional range ( $103\text{--}104^\circ$ ). The  $\text{C}^{\text{Ph}}\text{-CH}_2\text{-O-CH}_2$  torsion angles [e.g. C(25)-C(28)-





O(2)–C(29)] range from 162° to 171°; slightly lower than the essentially antiperiplanar arrangement in *meta*-bridged **6b**. An angle between imine N-atom planes [N(2)–N(6)–N(10) and N(3)–N(7)–N(11)] of 9.9° (comparable to **6a** and **6b**) is also supportive of there being little torsional strain in this complex.

The  $\Delta_{\text{Zn}}$  cationic unit of dibenzofuran-bridged **6h** is shown in Fig. 4. The intermetallic distance of 14.4 Å is rather smaller than that observed in the related structure **6e** corresponding to a more compact concertina fold. As expected, the circumference is slightly larger, with distances between 'apex' O atoms [O(11), O(14) and O(17)] falling into a narrow range (9.9–10.1 Å). These three atoms form an approximate equilateral triangle, indicating high symmetry of the system. There is also a lack of distortion arising from ring strain: each C<sup>Ph</sup>–CH<sub>2</sub>–O–CH<sub>2</sub> torsion angle [e.g. C(142)–C(141)–O(10)–C(140)] falls into the range 173–180° (effectively antiperiplanar). As a consequence of this efficient helication the CH groups at C(227) and C(230) are directed inwards, similar to the observation for **6b**, leading to a strong downfield shift in the <sup>1</sup>H-NMR spectrum (9.0 ppm, Fig. 1). In addition, the angle between imine N atom planes [N(2)–N(6)–N(10) and N(3)–N(7)–N(11)] is exceptionally shallow at 3.3°. Torsion angles between Zn–N(imine) bonds from the same ligand, along the Zn–Zn axis range 159.8° to 162.5°, greater than **6b** and less than in **6e**; this is in line with **6h** having the intermediate inter-metal distance of the three.

In summary, while the helical configurations of all these metallohelicities are controlled by  $\pi$ -stacking and steric effects<sup>23,27</sup> so as to be exclusively right or left handed ( $\Delta$  or  $\Lambda$ ) depending on choice of enantiomer **1**, the various diamines **3** cause different folding behaviours in the assemblies **5** and **6**. Overall the range of architectures is such that intermetallic distances – the relative positions of the seats of encapsulated positive charge – vary from 12.4 Å (**5/6b**) to over 17.4 Å (**5/6e**), and the van der Waals diameters of the structures are in a range from *ca.* 11.5 Å (**5/6a**) to 15.0 Å (**5/6h**).

Water compatible systems suitable for antimicrobial screening were prepared by similar self-assembly reactions (Scheme 1) using iron(II) chloride in methanol. <sup>1</sup>H and <sup>13</sup>C NMR spectra were very similar to the Zn counterparts, suggesting strong structural similarity. Typically, <sup>1</sup>H-NMR spectra were slightly broader, and all contained a large peak at *ca.* 4.9 ppm due to the presence of water of crystallization. Quantification of

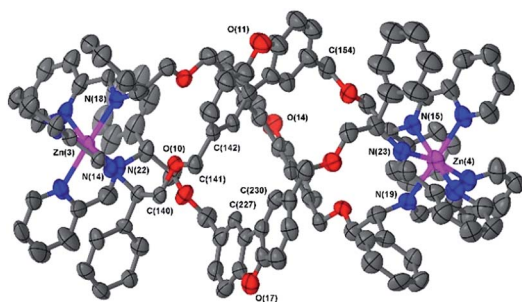


Fig. 4 Structure of the cationic unit of **6h**. Thermal ellipsoid plot (50% probability) with H atoms, solvent molecules and counterions removed for clarity.

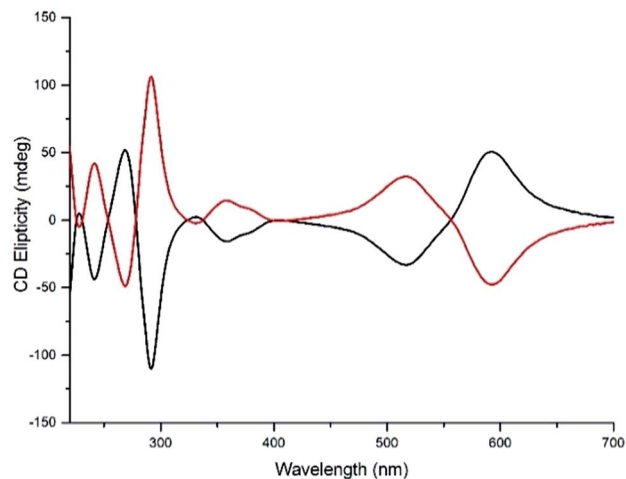


Fig. 5 Circular dichroism spectra of enantiomers  $\Delta$ -**5b** (black) and  $\Lambda$ -**5b** (0.03 mM in water).

the latter was achieved through a combination of microanalysis and thermogravimetric analysis (TGA) – see ESI† The Fe compounds **5** gave excellent electrospray-ionisation mass spectra (ESI-MS), whereby the  $[\text{Fe}_2\text{L}_3]^{4+}$  ion was readily observed at high resolution. Circular dichroism (CD) spectra of pairs of enantiomers **5** were mirror images within error (Fig. 5 and ESI†).

The diverse but efficient folding behaviours described above contribute to exceptional stability to hydrolysis in this class of metallohelicity and *e.g.* **5b** has a  $t_{1/2}$  of >10 d even at pH 1.5 (Fig. S8†).

### *In vitro* antibacterial activity

The compounds were screened using standard microbial assays to provide minimum inhibitory concentrations (MIC). While most of the compounds showed relatively high activity [(MIC) 1–8  $\mu\text{g ml}^{-1}$ ] against the Gram-positive microbe *B. subtilis* (168, ATCC 6051 – see Table S3†), perhaps due to the broad antimicrobial sensitivity of the species in liquid,<sup>28</sup> a range of potencies was observed against the more challenging *E. coli* (TOP10 strain, ATCC PTA-10989) and methicillin resistant *S. aureus* (USA300, ATCC BAA-1717); see Fig. 6. Generally, the enantiomers with  $\Lambda\text{Fe}$  helicity were more potent than their  $\Delta\text{Fe}$  counterparts, particularly against *E. coli*, though the differences were subtle. Compound  $\Lambda$ -**5b** gave a MIC of 2  $\mu\text{g ml}^{-1}$  against *E. coli*; at least as good as kanamycin given that the molecular weight of the metallohelicity is *ca.* four times higher. Additionally, the compounds gave minimum bactericidal concentration (MBC)/MIC ratios mostly in the range 1–2 against *E. coli* and *S. aureus*, making them by definition bactericidal [MBC/MIC  $\leq$  4].<sup>29</sup> Indeed, the lethal effect of  $\Lambda$ -**5b** upon *E. coli* TOP10 could be observed within 20–40 min. Interestingly, stationary phase *E. coli* cells were much less susceptible to these compounds (see Fig. S12†), suggesting this compound is most toxic to actively growing or dividing cells (*vide infra*). Similar effects have been seen with well-established antibiotics.<sup>30,31</sup> In addition, application of the LIVE/DEAD BacLight™ assay indicated that  $\Lambda$ -**5b** did



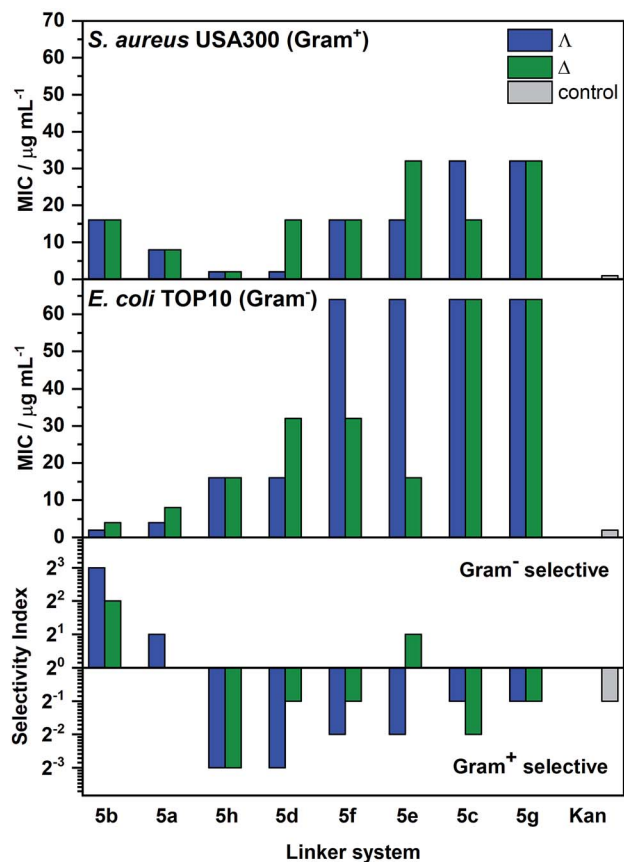


Fig. 6 Activity of the metallohelices against *S. aureus* and *E. coli*, and corresponding selectivity. Pairs of enantiomers (blue and green bars) are ordered along the x-axis by intermetallic distance (increasing left to right). The selectivity index is defined as  $MIC_{S. aureus}/MIC_{E. coli}$ . Hence, while the mid-length 5h and 5d provide the most active and selective compounds against *S. aureus*, for *E. coli*.  $\Delta$ -5b is most promising. Kanamycin (Kan) is included for reference.

not cause significant membrane damage to *E. coli* TOP10 at four times the MIC (Table S4<sup>†</sup>). This demonstrates that  $\Delta$ -5b is not acting as a surfactant.

From Fig. 6 we see that the compound  $\Delta$ -5b, with the shortest intermetallic distance (12.4 Å) has the highest selectivity for *E. coli*, as well as the highest activity. The nearest competitor is the enantiomer  $\Delta$ -5b. In contrast, compounds with mid-range intermetallic distance especially those containing an ether O atom in the bridge, are more selective for Gram-positive *S. aureus*; the enantiomers 5h and  $\Delta$ -5d which have the highest selectivity are also the most active.

Given the ever more acute clinical need for novel antimicrobial systems active against Gram-negative pathogens, we focused on this area, and the most selective enantiomers 5a and 5b were screened against a wider panel of clinically relevant microbes (Table 1). The panel included the notable *Escherichia coli* pathogens: enterohemorrhagic *E. coli* (EHEC) strains of the serotype O157:H7, Sakai (ATCC BAA-460)  $\Delta$ *stx1-2* and EDL933 (ATCC 700927)  $\Delta$ *stx1-2*; uropathogenic *E. coli* (UPEC) O6:H1:K2 strain CFT073 (ATCC 700928); an archetypal K12 *E. coli* control strain, MG1655 (ATCC 700926). Added to this panel were

examples of Gram-negative *ESKAPE* pathogens isolated from the clinic: *Klebsiella pneumoniae* strains K6 (ATCC 700603) exhibiting extended spectrum  $\beta$ -lactam resistance,<sup>32</sup> and KP02 (NCTC 13442) exhibiting carbapenem resistance via the OXA-48  $\beta$ -lactamase;<sup>33</sup> an *Acinetobacter baumannii* reference strain (NCTC 13420) from an MDR UK outbreak;<sup>34,35</sup> *Enterobacter cloacae* strain 684 (NCTC 13405) exhibiting extended spectrum  $\beta$ -lactam resistance via the AmpC enzyme.<sup>36</sup> The compound  $\Delta$ -5b demonstrated particularly good activity against the Enterobacteriaceae tested, especially strains of *E. coli*. Interestingly a striking difference in susceptibility is observed between the two *K. pneumoniae* isolates tested. We also note that none of the cationic metallohelices 5a–h showed significant activity against *Pseudomonas aeruginosa*, which can deploy a highly negatively charged alginate-based capsule, and correspondingly none of the cationic metallohelices 5a–h showed significant activity against this pathogen (e.g. strains PAO1, ATCC 15692). Overall, on the basis of the above performance,  $\Delta$ -5b was selected for mechanistic study.

### EHEC Sakai mutants exhibiting increased tolerance

Resistant strains may emerge when mutants in the bacterial population survive because they are less affected by the drug molecule. The *E. coli* strain EHEC O157:H7 Sakai, a notorious Gram-negative food pathogen, was exposed to inhibitory levels of  $\Delta$ -5b in so-called selection experiments to determine if it was possible to isolate mutants, which had become spontaneously resistant to the antimicrobial compound.

While 17 independent bacterial isolates were recovered, they were found to have achieved only a slight increase in tolerance rather than genuine target site resistance (Table S5<sup>†</sup>); MICs were only 2–4 times that of the parent strain. Whole genome DNA sequencing (Illumina) was performed on the isolates, and comparison of all the mutant strain genome sequences with that of the original wild-type parent strain allowed the identification of the specific nucleotide changes that gave rise to their increases in MIC. This was achieved using a combination of genome assembly and sequence comparison methods, specifically Bowtie2 and VarScan2 and the web-based Enterobase service,<sup>37</sup> allowing any single nucleotide polymorphisms (SNPs) and/or small sequence insertions or deletions (Table S5<sup>†</sup>) to be accurately identified. The mutants could be classed into four sub-types in respect of their differences to the wild-type organism, showing defects in genes involved in maintenance or modification of the outer membrane of the cell.

Two of the sub-types had acquired enzyme-related mutations which would alter the biophysical properties of the LPS outer membrane and thus create tolerance or resistance to the antimicrobial action of the metallohelix: (i) the WaaG enzyme, which has previously been identified as a potential antibiotic target,<sup>38,39</sup> is located at the cytosolic side of the inner membrane and catalyzes the transfer of the first outer-core glucose to the inner core during LPS synthesis for the outer leaflet;<sup>38,40</sup> (ii) *galU* – UTP-glucose-1-phosphate uridylyltransferase is also involved in LPS synthesis.<sup>41</sup> The LPS O-antigen of EHEC O157:H7 comprises *N*-acetyl-D-perosamine, L-fucose, D-glucose, and *N*-







up regulation, we note that a second porin, OmpW showed a significant down regulation. Under microaerobic conditions OmpW functions in the transport of hydrophobic molecules across the outer membrane.<sup>55</sup> It also represents a receptor for the antimicrobial protein colicin S4.<sup>56</sup> As for the B12 gene mentioned above, this porin could be responsible for influx of the compound.

Increased transcription of the gene for the YbjG undecaprenyl pyrophosphate phosphatase is also relevant as YbjG family proteins can confer bacitracin resistance.<sup>57</sup> Accordingly, disruption of ybjG causes increased bacitracin sensitivity, while over-expression causes increased resistance to bacitracin. Interestingly we also see an increase in transcription of the gene for FabR which is known to be involved in regulating genes responsible for unsaturated fatty acid (UFA) biosynthesis.<sup>50</sup> By controlling UFA production, FabR may indirectly influence the physical properties of the membrane bilayer. Two further, tightly linked genes, also relevant to CAMP resistance that show an increase in transcription, are those for the EptA phosphoethanolamine transferase and its regulator sensor protein, PmrB. Like PhoPQ, the PmrAB 2C system is also known to regulate resistance to CAMPs.<sup>58</sup> The PmrAB system upregulates EptA, which in turn catalyzes the addition of a phosphoethanolamine moiety to the KDO component of lipid A,<sup>59</sup> facilitating resistance to polymyxin B. Interestingly the sensor domain of PmrB is known to respond to several stimuli including acidic pH, which provides a link to another group of up regulated genes, involved in coping with acid stress. These observations highlight the complex interrelationships and overlaps between different forms of stress-response often seen in bacteria.

### Acid stress response

It is interesting that a link between transcriptional acid stress response and resistance to various membrane active environmental stresses, including polymyxin B have been observed in other bacteria suggesting a mechanistic relationship.<sup>60,61</sup> Indeed, an upregulated gene we observed, involved in both acid response and LPS synthesis is lpxT. LpxT is an enzyme, which is induced under acidic-aerobic conditions, that catalyses the phosphorylation of lipid A.<sup>62</sup> Furthermore, also of relevance here is the upregulation of the gene *asr*, which is dependent upon PhoB and RstA (see above), and facilitates an acid-induced protective response in cells exposed to very low pH.<sup>63,64</sup> Other upregulated genes observed that are known to be involved with acid stress response are those encoding for EfeB (a periplasmic deferrochelataase which extracts iron from exogenous heme), YdeP (an oxidoreductase) and RibB (involved in riboflavin biosynthesis in low pH). Note, we also observed the upregulation of response genes for other forms of stress including the heat shock Hsp90 and tellurite resistance TerW proteins. Upregulated oxidative stress response genes observed include; *sodA* superoxide dismutase, *yearR* which is induced by nitric oxide, *napF* a ferredoxin-type protein predicted to play a role in the oxidative stress response and *tatD* (*mttC*) a DNase potentially involved with the repair of H<sub>2</sub>O<sub>2</sub> induced DNA damage.

### Bacterial sub-cellular compound localization

We sought to investigate whether these large metallohelices could freely enter bacterial cells using fluorescence microscopy. A simple alkyne derivative of compound **Λ-5b** *i.e.* **Λ-5b'** (see ESI methods†) was synthesised for use in fluorescent “click” labeling chemistry using an Alexa Fluor® 488 azide (AF-488). Antimicrobial assays confirmed that **Λ-5b'** remained active against *E. coli*.

When cultured bacteria cease replication due to exhausting all the nutrients from the medium, they enter a quiescent state referred to as the stationary phase. We observed very little fluorescence staining of **Λ-5b'** in stationary phase *E. coli*, with the exception of a few cells with apparently compromised membranes. In contrast the compound was observed to enter the cytoplasm of exponentially growing cells and concentrate in punctate regions near the poles (Fig. 7A). Occasionally, the puncta were also positioned laterally. Importantly, we did not observe **Λ-5b'** localising preferentially in the membrane and the levels of staining for **Λ-5b'** were not homogeneous among cells suggesting differential uptake of the compound.

Several structures and phenomena are known to be associated with the poles of *E. coli* cells; these include anionic phospholipids such as cardiolipin that enrich in microdomains at regions of high negative membrane curvature<sup>65,66</sup> and chromosome partitioning machinery.<sup>67–69</sup> It is interesting to note that Keene and Collins<sup>70</sup> have observed the accumulation of antimicrobial dinuclear ruthenium tetracationic polypyridyl complexes at the cell poles by luminescence microscopy of *E. coli*. They ascribe this to localisation in the RNA of ribosomes, following imaging studies in the presence of antibiotics that are known to disrupt RNA/ribosomal distributions. However, we observed no signs of protein synthesis inhibition in our transcriptomic data.

A control imaging study with cells treated without compound **Λ-5b'** showed no staining with AF-488 (Fig. 7B). Interestingly, we noticed that cells which stained brightly for the test compound had reduced fluorescence on staining with DAPI – a DNA binder. This was confirmed by image analysis, which demonstrated bimodal frequency distribution of pixel mean intensity in the DAPI and AF-488 channels (Fig. S13†), suggesting competitive DNA binding.

### DNA interactions and enzyme inhibition *in vitro*

Following our observations above on DAPI stain exclusion, we decided to look directly at the interaction of **Λ-5b** with DNA. In addition, we note previous studies have demonstrated interactions of related metallohelices with naked DNA.<sup>23,71,72</sup> Ethidium bromide displacement studies on calf thymus DNA (CT-DNA) by **Λ-5b**, gave an apparent binding constant  $K_{app}$  of  $1.2 \times 10^8 \text{ M}^{-1}$  ( $\pm 1.2 \times 10^7 \text{ M}^{-1}$ ). Concentration-dependent increases in CT-DNA melting temperature [ $\Delta T_m$  of  $+4.3(0.3)^\circ\text{C}$  and  $+6.5(1.8)^\circ\text{C}$ , at DNA base to metallohelix ratio 20 : 1 and 10 : 1 respectively] were observed.

Linear dichroism (LD) studies show that only **5a** and **5b** (Fig. S9†) bind in an aligned *i.e.* not simply electrostatic manner with DNA – probably in the major groove.<sup>73</sup> Correspondingly,



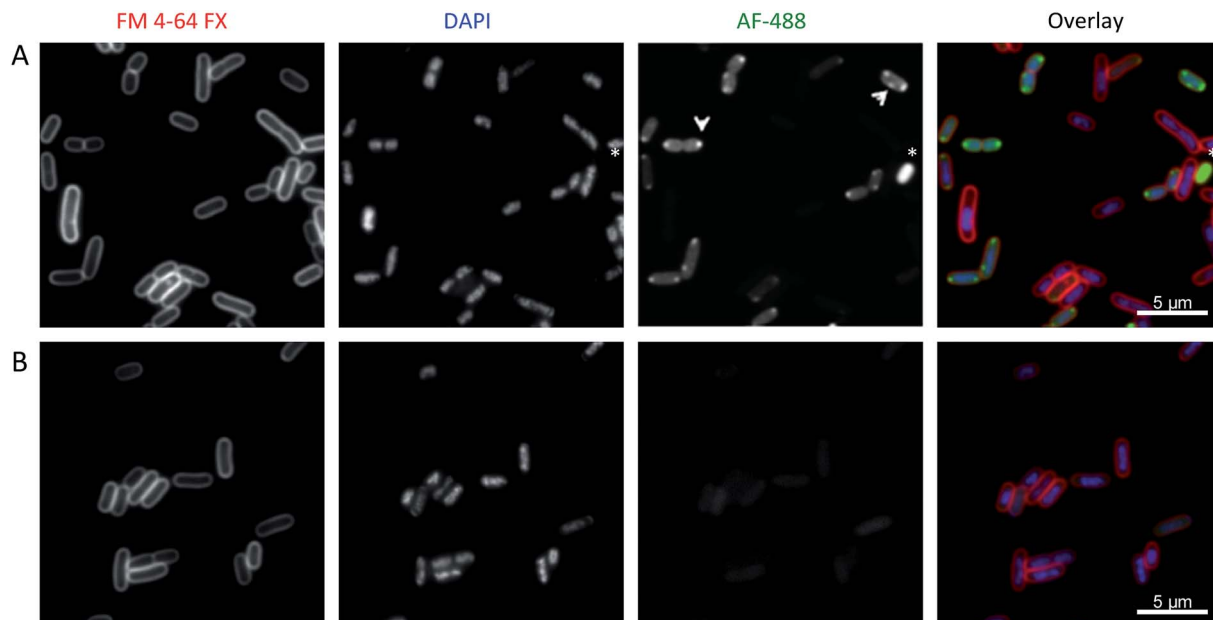


Fig. 7 Sub-cellular localization of  $\Delta$ -5b' in EHEC Sakai. Exponentially growing EHEC Sakai cells were treated with either (A)  $8 \mu\text{g ml}^{-1}$   $\Delta$ -5b' or (B) methanol and stained for membrane (FM 4-64 FX), nucleic acid (DAPI) and the  $\Delta$ -5b' (via click reaction with AF-488 azide). The arrows indicate the punctate localisation pattern of  $\Delta$ -5b'. The asterisk indicates an example of a bacterial cell with prominent  $\Delta$ -5b' but lack of DAPI staining.

these LD studies also indicate concentration-dependent structural deformation of DNA, and this was confirmed by atomic force microscopy on linearised plasmid pSP73 DNA (Fig. S10†).

Since G-quadruplex motifs – four-stranded structures formed by guanine-rich DNA – have been shown to be important in the regulation of the transcription of certain genes and potentially other processes in bacteria including *E. coli*,<sup>74–76</sup> we investigated their interactions with enantiomers **5b**. A FRET melting assay<sup>77</sup> was used to assess the change in stability of G-

quadruplex structures on addition of a metallohelix (Fig. 8). Three representative fluorophore labelled oligonucleotides were used, plus a negative control hairpin duplex which does not form a G-quadruplex motif. Further, by performing these studies in the presence and absence of an excess of calf thymus DNA (ctDNA) we were able to assess the selectivity of the compounds for binding the G-quadruplex motif over major groove binding (*vide supra*). In Fig. 8A–C we see that G-quadruplex structures were significantly stabilised by both

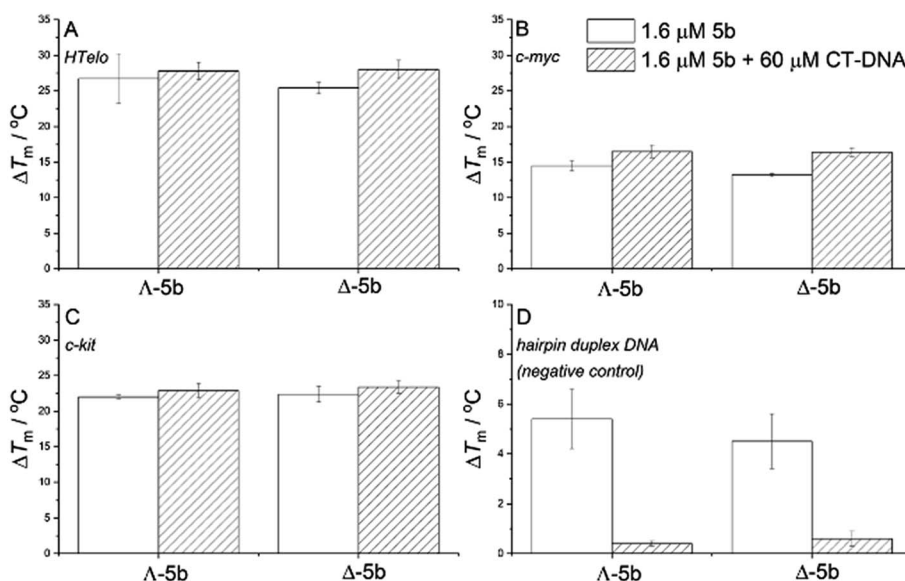


Fig. 8 Selective stabilization of DNA G-quadruplexes over duplex DNA. FRET melting assay of fluorophore-labelled quadruplex-forming oligonucleotides (A) HTelo, (B) *c-myc*, (C) *c-kit*, and (D) the labelled oligonucleotide F26T<sub>ds</sub>-hairpin, which does not form a quadruplex structure (negative control).  $\Delta T_m$  values determined by FRET upon addition of **5b** ( $1.6 \mu\text{M}$ ) in the absence and in the presence of  $60 \mu\text{M}$  CT-DNA show strong preferential quadruplex binding in the presence of excess duplex CT-DNA ( $\Delta T_m$  unchanged).





enantiomers, as indicated by the change in melting temperatures ( $\Delta T_m$ ) of ca. +26, 22 and 14 °C respectively, and that this was not affected by addition of 60  $\mu$ M ctDNA. This indicates substantial selectivity. In Fig. 8D we see also that while the thermal stability of the control hairpin duplex was increased by both enantiomers, the effect disappeared in the presence of 60  $\mu$ M ctDNA. Interestingly we subsequently demonstrated that *in vitro* **A-5b** did not affect the endonuclease cleavage of plasmid pSP73KB DNA by EcoRI or DNA transcription (data not shown). However, the compound was shown to inhibit both relaxation and supercoiling of pBR322 plasmid DNA by *E. coli* topoisomerase I and DNA gyrase respectively (Fig. S11†).

## Conclusions

In this work we present a range of self-assembled helical metallo-organic compounds able to selectively kill – at low doses – a range of bacterial pathogens, including clinically problematic species such as members of the ESKAPE panel of microbes. A particularly useful property of this class of molecule is the ability to change the handedness, size, shape and charge-distribution in the structure by selection of sub-components for the self-assembly. Many other avenues are available including changes to substituent groups for modification of lipophilicity and functionality. This will also facilitate investigation of the basis of the intriguing specificities observed.

The cationic nature of the metallohelices would predispose them, as for CAMPs, to interaction with bacterial membranes, which are negatively charged. Given the size and charge of **A-5b** we may have expected to see it accumulate across the outer surface of the membrane or cause membrane disruption, like some CAMPs. However, membrane integrity is maintained and bacterial localisation experiments revealed that **A-5b'** was able to enter the cell cytoplasm, partially accumulating at distinct foci, typically at the poles of cells.

Given that microdomains of cardiolipin and chromosome partitioning machinery is known to accumulate at polar foci in cells,<sup>67–69</sup> it is reasonable to suggest that **A-5b** could be interacting with either of these components. The latter would be consistent with our DNA binding studies. We showed that **A-5b** has a very high affinity for binding G-quadruplex DNA over double stranded DNA. The role of tertiary structure and the conformation of DNA during replication is highly complex<sup>78</sup> and so it is difficult at this juncture to posit a specific point at which the metallohelix may be interfering. With this in mind we note this compound is able to inhibit the activity of both DNA gyrase and topoisomerase I *in vitro*. DNA topoisomerase enzymes play crucial roles in the supercoiling and relaxation of DNA, which is essential for both chromosomal replication and partitioning. It is perhaps relevant here to recall that the compound only shows activity against exponentially growing cells. Furthermore, **A-5b** also caused the curing of the pO157 plasmid in one sub-class of the tolerance mutants. pO157 is a single copy number plasmid, and like the chromosome itself, it relies upon a ParAB-like partitioning system (SopAB), which is an inner membrane associated mechanism.

When applied at sub-MIC levels, we observe a transcriptomic response consistent with the normal response of *E. coli* to a natural CAMP. The cell must be able to sense the influence of the metallohelix, possibly on membrane properties, leading to the activation of the various two-component sensor/regulator pathways, acid response pathways and subsequent attempts by the cell to lower the net negative charge of the surface.

The Gram negative outer membrane can act as an efficient intrinsic permeability barrier to the egress of certain antimicrobial agents such as vancomycin.<sup>79</sup> In addition, surface charge modification, as indicated here, is a well-established mechanism of resistance to cationic antimicrobial peptides, such as the action of the ArnB enzyme on LPS to reduce the interaction of polymyxin.<sup>82</sup> Furthermore, alterations in membrane permeability have also been established as a more general mechanism to control the influx of antibiotics.<sup>80</sup>

These defence responses by *E. coli* to the metallohelices are nevertheless insufficient at the MIC. Our inability to isolate fully resistant “target site” mutants also strongly suggests that the compound impacts on multiple structures or pathways.

The metallohelices thus represent functional emulators of key properties of their more exquisitely structured natural counterparts. Notably they are of similar size to short CAMPs but remain folded thereby exhibiting “patchy” hydrophobic regions on the surface. We propose that as highly stable and fully non-natural chemical entities, containing no amino acids or peptide bonds, the evolution or acquisition of novel enzyme functions that could mitigate the action of the metallohelix may well be rare events or entail high fitness costs for bacteria.

We note that the first report of antimicrobial activity of metal complex ions appeared in 1952,<sup>81</sup> and arguably the most promising developments in this area<sup>82</sup> have been the closely related ruthenium-based intercalators and oligonuclear ruthenium systems.<sup>83–85</sup> Perhaps the ability to make optically pure compounds without the need for resolution, and the absence of toxic metals, further adds to attractiveness of the Fe-based metallohelices in this report.

Taken as a whole our findings suggest that the development of peptide-inspired scaffolds that emulate the structural features of CAMPs but have lower cost of synthesis and perhaps better pharmacological properties, is a realistic approach for the discovery of new antimicrobial drugs.

## Experimental section

### Synthesis of water soluble metallohelices

The appropriate optically pure diamine (3.0 eq.) and 2-pyridinecarboxaldehyde (6.0 eq.) were dissolved in methanol (50 ml) and stirred for 2 h at ambient temperature. Anhydrous iron(II) chloride (2.0 eq.) was added, and an immediate color change to deep purple was observed. The solution was then heated at reflux (80 °C) for 48 hours. After filtering, the solvent was removed under reduced pressure to give the desired product as a dark purple solid, which was dried overnight at 50 °C *in vacuo*. Detailed synthesis and characterization methods can be found in the ESI and methods.†



### DNA quadruplex binding FRET melting assay

0.4  $\mu\text{M}$  annealed double-FRET-labelled synthetic oligonucleotides (Eurofins Genomics, Ebersberg, Germany) were mixed with 1.6  $\mu\text{M}$  metallohelix in the absence of CT-DNA, and in the presence of 60  $\mu\text{M}$  CT-DNA. Experiments with F21T\_HTelo and F26T\_ds-hairpin were performed at 40 mM  $\text{K}^+$  concentration (additional 30 mM KCl). Measurements were performed on a real-time PCR instrument RotorGene 6000 (Corbett Research)  $\lambda_{\text{exc}} = 470 \pm 10 \text{ nm}$ ,  $\lambda_{\text{em}} = 510 \pm 5 \text{ nm}$ , at intervals of  $0.7 \text{ }^\circ\text{C min}^{-1}$ .

### Determination of minimum inhibitory concentrations (MIC) and minimum bactericidal concentrations (MBC)

The bacterial strains used in this study are listed in Table S2.† For MIC determination the standard broth microdilution method was employed, in agreement with the Clinical and Laboratory Standards Institute (CLSI) guidelines M07-A9 and M100-S24. Cation-adjusted Mueller Hinton Broth (CAMHB) was used as the media and a  $1280 \mu\text{g ml}^{-1}$  stock solution of each metallohelix was prepared in water containing 10% methanol.  **$\Lambda$ -5b** and  **$\Delta$ -5b** MICs were independently validated using an extended range of  $0.007$ – $256 \mu\text{g ml}^{-1}$ . Determination of MBCs was carried out for strain/compound pairings where an MIC  $\leq 128 \mu\text{g ml}^{-1}$  was determined, immediately following that assay. For each culture with compound concentration in the range  $128 \mu\text{g ml}^{-1}$  to the MIC (inclusive),  $100 \mu\text{l}$  of the bacteria/compound mix was recovered from the microtitre plate for analysis. Cell and/or debris was collected from samples by centrifugation and resuspended in  $100 \mu\text{l}$  sterile PBS, which was streaked onto a sterile antimicrobial-free LB/agar plate. Upon overnight incubation ( $37 \text{ }^\circ\text{C}$ ), plates were inspected and the MBC was determined to be the lowest concentration of compound at which this dilution/culturing assay showed no visible signs of bacterial growth. This was performed at least in duplicate for each pairing of compound concentration and bacterial strain.

### Isolation of tolerant mutants

7 independent biological replicates of EHEC Sakai were grown in CAMHB, and  $25 \mu\text{l}$  of each culture was applied to MH agar containing  $80 \mu\text{g ml}^{-1}$  of  **$\Lambda$ -5b** (a high concentration was used as the compound does not appear to diffuse well in agar) and incubated at  $37 \text{ }^\circ\text{C}$  overnight. Four colonies from each replicate were re-streaked onto MH agar both with and without supplementation with  **$\Lambda$ -5b**. These were used to seed fresh cultures in MH broth to a cell density of  $5 \times 10^5 \text{ cfu ml}^{-1}$ , incubated with  $2 \mu\text{g ml}^{-1}$   **$\Lambda$ -5b** (the Wild Type (WT) MIC). The MIC of the  **$\Lambda$ -5b** for the isolates was measured and the bacteria were studied by whole-genome sequencing.

### Whole genome sequencing

Genomic DNA libraries were prepared using the Illumina Nextera® XT kit using 1 ng of input DNA. Following PCR clean up, the DNA concentration of each library was measured using the Qubit® high-sensitivity dsDNA and the libraries were normalised to 4 nM and pooled together. The pool was prepared for

loading following the Illumina Nextera® XT guidelines. Sequencing was performed using a MiSeq Reagent Kit v2 (500-cycles) on an Illumina MiSeq™ instrument.

### Extraction and purification of microbial RNA

EHEC Sakai were incubated in CAMHB with a sub-MIC dose of metallohelix to a final concentration of  $0.5 \mu\text{g ml}^{-1}$  for 40 min at  $37 \text{ }^\circ\text{C}$ . Samples were added to RNAprotect reagent (Qiagen) and incubated for 5 min at room temperature. Enzymatic lysis was performed using TE buffer with proteinase K and  $1 \text{ mg ml}^{-1}$  lysozyme. RNA was then purified using the miRNeasy kit (Qiagen) with the inclusion of a double on-column DNA digestion. Absence of contaminating DNA was confirmed by PCR using primers for the *E. coli* 16S ribosomal subunit gene. RNA integrity was verified using the RNA 6000 pico kit (Agilent) on the Agilent 2100 Bioanalyzer. To deplete rRNA, the Ribo-Zero™ bacterial rRNA removal kit (Illumina) was used with an input RNA of  $4 \mu\text{g}$  per sample.

### Generation of cDNA libraries and sequencing

rRNA-depleted samples were used as input for cDNA library preparation using the Illumina TruSeq™ stranded mRNA kit with a slightly modified library preparation protocol. cDNA libraries were quantified using the Qubit® High Sensitivity DNA assay kit and fragment sizes were determined using the High Sensitivity DNA kit on the Agilent 2100 Bioanalyzer. The libraries were normalised to 4 nM and pooled together. Paired-end sequencing was performed using two MiSeq reagent kits v3 (150-cycle) on an Illumina MiSeq™ sequencer.

### Bacterial sub-cellular compound localization

Overnight culture of EHEC Sakai was diluted in cation-adjusted Mueller Hinton broth and grown to mid-exponential ( $\text{OD}_{600} = 0.52$ ) in order to image dividing cells. For stationary phase cells, a 24 h culture of EHEC Sakai at an  $\text{OD}_{600}$  of 2.9 was used. The cultures were incubated with  **$\Lambda$ -5b'** for 30 min at  $37 \text{ }^\circ\text{C}$ . Ten minutes before the end of the incubation period,  $5 \mu\text{g ml}^{-1}$  FM™ 4-64FX (Invitrogen) was added to stain the cell membrane. For visualisation of the target compound we used the Click-IT cell reaction buffer kit (Invitrogen) as per the manufacturer's instructions. Finally, cells were stained with  $1 \mu\text{g ml}^{-1}$  DAPI for 1 min, washed with PBS, and imaged using a Leica DMi8 inverted microscope.

## Conflicts of interest

There are no conflicts to declare.

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