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Will Surface Effects Dominate in Quasi Two Dimensional Gallium Oxide for Electronic and Photonic Devices?

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ABSTRACT

There is currently great interest in ultra-wide bandgap semiconductors for their applicability in power switching electronics with improved efficiency compared to current technologies and also to solar-blind UV detection. One of the most promising materials is Ga₂O₃, available in large area bulk crystals and as exfoliated nano-layers (nanobelts, nanomembrane, and nanosheets). One aspect of this material that has not widely been recognized is the sensitivity of its surface to ambient. The goal of this brief focus article is to provide some insight into the mechanisms and defects that underlie this effect and explain inconsistencies in the literature.

Gallium oxide has a number of polymorphs, with the β - and α -variants emerging as next-generation ultra-wide bandgap semiconductors (bandgap 4.6-4.8 eV for β and \sim 5.2 eV for α -polymorphs) ⁽¹⁻³⁾. These have the potential for transformative impact on the energy economy, due to their more efficient power switching capabilities in power control and conversion applications such as automotive, data center power management, grid control, industrial and

locomotive traction control ⁽¹⁻⁴⁾. However, a high density of crystal defects in the bulk of Ga₂O₃ is significantly hindering the progress in realizing many of the beneficial attributes of power electronics systems ⁽⁴⁾. What is less commonly realized is that surface effects are also very important with Ga₂O₃, a factor that has been overlooked in lists of research challenges ⁽⁴⁾.

Despite not being a van der Waals material and having highly strong ionic bonding, the more commonly studied β -phase can be mechanically cleaved and exfoliated easily along favorable surfaces ^(2,3). The large anisotropy of the lattice constant in this monoclinic phase allows a facile cleavage ($a[100] = 12.225 \text{ \AA}$, $b[010] = 3.039 \text{ \AA}$, and $c[001] = 5.801 \text{ \AA}$) along (100) and (001) faces. From bond strength calculations, the surface covalent bonds in these planes are up to 6 eV per bond stronger than the perpendicular bonds, which break easily ⁽⁵⁾. Thin quasi-2D nano-layers (nanobelts, nanomembrane, and nanosheets) can be exfoliated from bulk crystals, similar to true 2D materials such as graphene or MoS₂ ^(6,7) and used to fabricate transistor and photodetector structures ⁽⁸⁾. In addition, single-crystalline ultrathin quasi-hexagonal (010) β -Ga₂O₃ nanosheets can be homoepitaxially grown on nanowire seeds ⁽⁹⁾. The fabrication of devices using β -Ga₂O₃ nanobelts has several advantages over conventional bulk single crystals, including absence of strain and the fact the heat management problem related to the low thermal conductivity ⁽¹⁻³⁾ of β -Ga₂O₃, can be mitigated by applying it to substrates with high thermal conductivity.

The surface termination, relaxation and surface energies for different faces of β -Ga₂O₃ have been reported by Bermudez ⁽¹⁰⁾ and provide insight into the structure of nanosheets. Figure 1 shows schematics of the ideally terminated (100)-A and (100)-B surfaces. The A and B refer to two possible terminations, corresponding to stoichiometric unit cells with non-polar surfaces. A refers to termination in rows of O(II)s lying along the [0 1 0] direction with each

O(II) back-bonded to two Ga(I)s. B refers to termination in nearest-neighbor rows of Ga(II) and O(III) atoms, each singly-unsaturated, with full coordination of Ga (I) and O(I) atoms at the surface. Hartree-Fock calculations show the (100)-B surface has the lowest surface energy in both the ideal ($0.96 \text{ J}\cdot\text{m}^{-2}$) and relaxed ($0.68 \text{ J}\cdot\text{m}^{-2}$) conditions, while the (010) surface has the highest surface energy in the relaxed condition ⁽¹⁰⁾. Experimental data shows Ga_2O_3 powders have equivalent surface and bulk compositions, without favorable surface stabilization ^(11,12). Collins et al. ⁽¹¹⁾ reported that H_2 is dissociatively chemisorbed, following reaction pathways involving endothermal dissociation occurs over surface Ga sites at $T > 450 \text{ K}$, producing Ga–H(I) bonds (heat and entropy of this hydrogen adsorption were $\Delta h_1 = 155 \pm 25 \text{ kJ mol}^{-1}$ and $\Delta s_1 = 0.27 \pm 0.11 \text{ kJ mol}^{-1} \text{ K}^{-1}$). An additional pathway was present at low-temperatures, involves surface Ga–O–Ga species, producing GaO–H and Ga–H(II) bonds stable to after heating under evacuation at $T > 650 \text{ K}$. It has long been established that oxygen vacancies are produced during heating in vacuum ⁽¹²⁾, where the relationship between conductivity (σ) and O_2 partial pressure ($p\text{O}_2$) is given by $\sigma \sim (p\text{O}_2)^{-0.25}$.

While the understanding of possible thickness-dependent bandgap and mobility at the nanoscale limit ⁽¹³⁾ have not been well understood in Ga_2O_3 , the nanobelts can also be integrated with other low-dimensional materials, taking advantage of quasi-2D structures, including h-BN, to fabricate metal-insulator-semiconductor field effect transistors ^(8,14-17). Zhou et al. ⁽¹⁶⁾ fabricated nanobelt devices with various thicknesses and found the threshold voltage (V_T) shifted from negative to positive as the thickness was decreased. Enhancement-mode (E-mode) $\beta\text{-Ga}_2\text{O}_3$ nanobelt FETs demonstrated breakdown voltage of 185 V with negligible hysteresis and high on/off ratio, promising for future power devices ⁽⁸⁾. There are also extensive studies of solar-blind UV photodetectors fabricated on Ga_2O_3 nanobelts ⁽⁸⁾.

There are also more speculative potential applications, including resistive-RAM (ReRAM or RRAM), a non-volatile memory gaining attention because it can be utilized to build neuromorphic computing chips⁽¹⁸⁾. Many metal oxides have been investigated in this regard because the oxygen ions and oxygen vacancies can easily drift under an electric field. The motion of oxygen ions and oxygen vacancies needs to be investigated in β -Ga₂O₃ nano-layer, but memristic behavior has been observed⁽¹⁹⁻²¹⁾. Heterojunction Bipolar Transistors (HBTs) using nano-layer β -Ga₂O₃ as the emitter have potential because the physical stacking of the exfoliated layers can offer the ideal structure of HBTs without considering the lattice matching⁽²²⁾. Finally, ultra-thin β -Ga₂O₃ can offer a more robust radiation-hard platform because of the lower chance for damage accumulation. Most high energy particles which commonly cause lethal damage to electronic devices can penetrate through the active area of the nano-scale devices, which is advantageous over the conventional thin-film devices. Monolithic integration of β -Ga₂O₃ -base diodes, transistors, gas sensors, solar-blind photodetectors, HBTs, inverters, logic devices, resistive random access memory (ReRAM) can be demonstrated in the β -Ga₂O₃ nanodevices, which will be robust under a harsh environment⁽²³⁾.

However, it is not widely appreciated that the surface of β -Ga₂O₃ is strongly affected by exposure to gaseous or plasma ambients and the influence of changing conductivity and role of surface states in oxidizing or reducing ambients is not established. This is despite the fact that gallium oxide diodes are known to be sensitive detectors of hydrogen⁽²³⁾. Swallow et al.⁽²⁴⁾ reported that as-grown (2⁻01) single crystals exhibit electron accumulation at the surface, producing downward band bending, ascribed to negatively charge acceptor-type intrinsic surface states. Removal of OH species from the surface shifted the valence band maximum by roughly

0.5 eV and electrons accumulated, causing upward band bending. Other electronic oxides, including ZnO and SnO₂ can also exhibit either surface electron accumulation or depletion and this has been ascribed to the differences in cation/anion size and electronegativity^(25,26). Photoemission experiments in Ga₂O₃ have typically suggested surface electron accumulation⁽²⁷⁾, but the role of cleaning and annealing, processes inherent to actual device processing, indicate a more complex behavior^(24,28). For example, F plasma exposure leads to compensation of near-surface donors by F⁻ ions and significant changes in Fermi level pinning⁽²⁸⁾.

A number of authors have shown that annealing Ga₂O₃ in O₂ versus N₂ ambients produces a pronounced reduction in near-surface n-type carrier concentration^(3,29). Annealing of undoped Ga₂O₃ in an oxidizing atmosphere at ≥ 1200 °C for extended periods (>20 h) decreases the bulk free electron concentration by about one order of magnitude, while the surface becomes insulating⁽³⁾. Moreover, the semiconducting behavior of the surface was recovered by annealing in a H₂-containing reducing atmosphere. The surface conductivity could be reversibly changed one order of magnitude by annealing only, while the surface of the bulk crystals could be reversibly switched between semiconducting and insulating by annealing in oxidizing and reducing atmospheres^(3,24).

These large swings in near-surface and bulk conductivity will be prohibitive in achieving stable operation of nanobelt devices unless the defects responsible are understood and effective passivation schemes developed. At a minimum, it is already clear that O₂, N₂, H₂ and F₂ exposure can dramatically alter the near-surface conductivity.

Native point defects in β -Ga₂O₃ include Ga vacancies (V_{Ga}) at tetrahedral (Ga1) and octahedral (Ga2) sites, as well as oxygen vacancies at threefold coordinated sites (O1 and O2) and fourfold coordinated sites (O3)^(30,31). In most metal oxides, the cation vacancies and oxygen

interstitials tend to be acceptors ^(6,7), while the oxygen vacancies and cation interstitials are donors. In n-type Ga₂O₃, the gallium vacancy in Ga₂O₃ is a triple acceptor, the oxygen interstitial (O_i) neutral, and the gallium interstitials (Ga_i) in the +3 charge state ^(30,31). The calculated oxygen vacancy (V_O) formation energies of 2.7-3.6 eV suggests they are deep donors and will be neutral in highly n-type material and do not directly contribute to the electrical conductivity ⁽³¹⁾. Oxygen vacancies occur as neutral (V_O) and double ionized vacancies (V_O²⁻) with V_O dominating at low oxygen partial pressure ⁽³⁾. Gallium vacancies (V_{Ga}) form triple acceptors at -3 charge state, between 1.6 and 1.8 eV below the conduction band minimum for V_{Ga1} and V_{Ga2} ^(30,31), respectively, and are deep acceptors. The V_{Ga} concentration increases with oxygen partial pressure, leading to a conductivity compensation. The influence of ambient on the diffusion and complexing of these defects with other defects or impurities in Ga₂O₃ is not currently understood. Figure 2 shows a schematic of the β-Ga₂O₃ lattice structure, containing Ga and O vacancies and the typical ambient species present during processes such as annealing during device fabrication.

The configurations of hydrogen in the bulk of β-Ga₂O₃ are now clearer ⁽³²⁾, but the expected donor nature of interstitial hydrogen may be modified in the presence of surface oxygen atoms ⁽¹⁹⁾. All of this suggests that variations in process ambient and history between different groups will continue to lead to large variations in reported device performance. Trap spectra measurements show that exfoliated β-Ga₂O₃ inherits the characteristics of the single-crystal β-Ga₂O₃ from which they were separated ⁽³³⁾. Thus the control of defects and impurities in the starting substrate is also key. Optimized passivation layers for β-Ga₂O₃ nanobelts are required to enhance the stability of these devices because the properties are affected by environmental conditions.

In summary, in this Focus Article, we have provided examples of significant changes in the electronic nature of Ga₂O₃ surfaces, and briefly mentioned mechanisms that affect these quantities as well as some techniques to mitigate them. This article should be considered as a short introduction to the vast literature that covers these topics in great depth.

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Figure Caption

Figure 1. Schematic representation of ideally-terminated (1 0 0)-A (top) and (100)-B surfaces of β -Ga₂O₃. (a) View along surface normal-dashed lines show the surface unit cell (b) outermost surface layers, viewed along the [0 1 0] direction, with vectors indicating the displacements occurring during relaxation. The displacement vectors are increased by a factor of 4 relative to the interatomic distances. All displacements are in the plane of the page, i.e., the (0 1 0) plane.(after Bermudez⁽⁹⁾).

Figure 2. β -Ga₂O₃ lattice structure showing presence of O and Ga vacancies and ambient molecules that can affect surface conductivity.



