

## PAPER

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## Optimized electrical and optical properties of Ag micro-meshes by self-generated cracks for transparent electrodes†

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Transparent conductive electrodes (TCEs) are essential components in the development of various transparent devices. In this study, a Ag micro-mesh network was investigated as an indium-free, non-rigid TCE for applications in flexible devices or solar cells. Instead of a complex and expensive process, such as photolithography with a shadow mask to fabricate a micro-grid, the Ag micro-mesh electrode was fabricated using self-generated cracked templates. The structure of the metallic network was controlled by varying the spin-coating speed and silica solution concentration so that a highly interconnected metallic network can be obtained. Using these techniques, various widths and densities of cracks were obtained and analyzed to compare their optical and electrical properties. Finally, the figure of merit (FoM) for the transparent electrode was calculated based on the measurement results, which revealed that the Ag micro-mesh electrode exhibited the highest performance when the crack widths were mostly distributed in the range of 1–2  $\mu\text{m}$ , regardless of the fabrication conditions. The optimum Ag micro-mesh electrode achieved 88% transmittance at 550 nm wavelength and exhibited  $20 \Omega \text{ sq}^{-1}$  sheet resistance. These optical and electrical properties are comparable to those of indium-doped tin oxide (ITO) and/or aluminum-doped zinc oxide (AZO) that are used as TCE materials.

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## Introduction

Transparent conductive electrodes (TCEs) are widely used in various transparent devices, such as touch-screen displays,<sup>1,2</sup> solar cells,<sup>3,4</sup> smart windows,<sup>5,6</sup> and sensors.<sup>7,8</sup> Indium-doped tin oxide (ITO) is a representative material for the TCE of most devices because of its excellent optical and electrical properties.

Despite its advantages, various alternative materials for ITO have been explored because of the scarcity of indium and the diffusion of metal atoms from ITO into other layers, which induces the degradation of devices.<sup>9</sup> Alternative materials for ITO should have high transparency and low sheet resistance as well as the flexibility to be applied in solar cells or devices. Accordingly, various materials, such as metal grids,<sup>10,11</sup> silver nanowires (Ag NWs),<sup>12,13</sup> carbon nanotubes,<sup>14,15</sup> and graphene,<sup>16,17</sup> have been extensively studied and have shown good performance as TCEs. However, these materials have limitations regarding their properties and disadvantages. Ag NW-based electrodes have conductivity issues due to the cross-bar junctions of NWs; therefore, the Ag NWs must be deposited with a high density, which causes a decrease in the transmittance.<sup>14</sup> The applications of carbon nanotubes are limited owing to their high sheet resistance and low stability in air,<sup>18</sup> and similarly, graphene has a sheet resistance limitation of  $30 \Omega \text{ sq}^{-1}$ .<sup>19</sup> In the case of metal grids, there is no limitation on the properties of the electrode, but a shadow mask is required to proceed with photolithography, rendering it difficult to control the width and density of the metal grid.

In contrast to these materials, metal-mesh electrodes are effective candidates to substitute ITO because of the

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straightforward fabrication process and the use of a removable cracked template to replace the shadow mask. The crack templates are simply fabricated using a dried silica or acrylic solution, which is inexpensive compared with the aforementioned materials. Additionally, the optical and electrical properties of the electrode can be easily controlled by changing the fabrication conditions of the cracked template to rearrange the structure of the metallic networks.<sup>20,21</sup> To optimize their performance, metal-mesh electrodes have been extensively investigated to elucidate the formation process of cracks on a solution film and the relationship between crackle patterns and sheet resistances.<sup>12,22,23</sup> In addition, metallic networks have outstanding electrical and optical properties, enabling their successful application to flexible devices and solar cells.<sup>4,20,24–26</sup>

Building on the findings from various studies, an interesting point emerged: nearly all metal-mesh electrodes that demonstrated the best performance had widths predominantly in the 1–2  $\mu\text{m}$  range. This observation led us to hypothesize that the width of the electrode's metallic mesh might play a crucial role in determining its overall performance. As a result, in this study, we focused on investigating the correlation between the width of the micro-mesh grid and the electrode's performance, particularly its figure of merit (FoM). To determine the optimum crack width, we prepared samples with varying crack widths by adjusting the spin-coating speed and colloidal silica concentration, allowing us to systematically evaluate the impact of crack width on the electrode's electrical and optical properties. Additionally, the potential of this Ag micro-mesh was confirmed by applying it to CIGS solar cells.

## Experimental

The Ag micro-mesh electrode was fabricated on a glass substrate following the process described in Fig. 1. The soda-lime glass substrate ( $25 \times 25 \times 0.7 \text{ mm}^3$ ) was sequentially cleaned with acetone, ethanol, and deionized (DI) water for 30 min each using an ultrasonic cleaner. After the cleaning process, hydrophilic treatment was performed on the surface of the glass to improve adhesion and wettability, which affects the cracking behavior.<sup>27</sup> For the hydrophilic treatment, the glass was immersed in a piranha solution composed of sulfuric acid (95 wt%) and hydrogen peroxide (30 wt%) in a ratio of 3 : 1 at 75 °C for 2 h. After completing the treatment, colloidal silica (Ludox HS-30, Sigma-Aldrich) was coated on the hydrophilic surface of the glass substrate using a spin coater (WS-400-6NPP-LITE, Laurell). To prepare cracks with various widths and densities, silica solutions with different concentrations were prepared and coated at different spin-coating speeds (2000–6000 rpm) and a spin-coating time of 10 s. Different concentrations of silica solutions were prepared by diluting the colloidal silica to  $0.363 \text{ g mL}^{-1}$  (S363) with DI water. Consequently, three different concentrations of colloidal silica, 0.242 (S242), 0.272 (S272), and  $0.363 \text{ g mL}^{-1}$  (S363), were prepared to form the crack templates used in this study. The

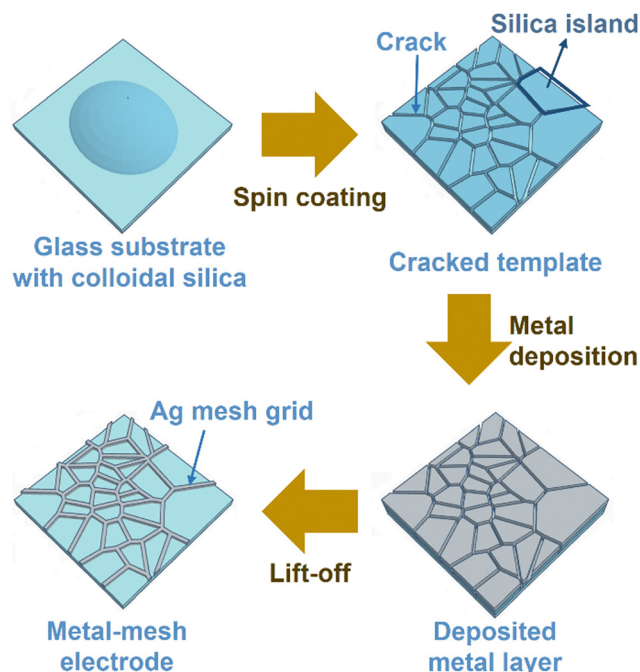


Fig. 1 Schematic of the fabrication process of the Ag micro-mesh electrode using self-generated cracked templates.

diluted solution, colloidal silica, and DI water were mixed, ultrasonicated for 30 min, and left overnight for stabilization. Before proceeding with the spin-coating process, the diluted and stabilized solution was ultrasonicated again to ensure a uniform concentration. After the spin-coating process, the coated silica film was naturally dried at room temperature at 20–30% relative humidity (RH). As shown in Fig. S1a (ESI<sup>†</sup>), cracks with silica islands are well developed on the glass substrate, forming a cracked template that can be used to fabricate a micro-mesh structure. Ag thin films, with thicknesses of 80, 120, and 160 nm, were deposited on the cracked template *via* radio frequency (RF) magnetron sputtering at a power of 30 W. After the sputtering process, the silica islands on the glass were removed *via* ultrasonic cleaning. Thereafter, the Ag micro-mesh was left on the glass substrate to act as a transparent electrode (Fig. S1b, ESI<sup>†</sup>). The scanning electron microscopy (SEM) image confirms that the Ag micro-mesh electrodes, with the shape of the silica island, was well fabricated on the glass substrate, as illustrated in Fig. 1. Field-emission scanning electron microscopy (FE-SEM; SU8220, Hitachi) was performed to obtain the surface images and crack widths of the samples. The crack density per sample was calculated based on the literature.<sup>28</sup> The detailed calculations are described in Fig. S2 (ESI<sup>†</sup>). Sheet resistance was measured using a four-point-probe surface resistivity meter (RC2175, EDTM), and transmittance was measured using a spectrophotometer (Spectro 22RS, Labomed) at a wavelength of 550 nm.

## Results and discussion

The SEM images of the S363 samples prepared at different spin-coating speeds are shown in Fig. 2b–f. As the spin-coating



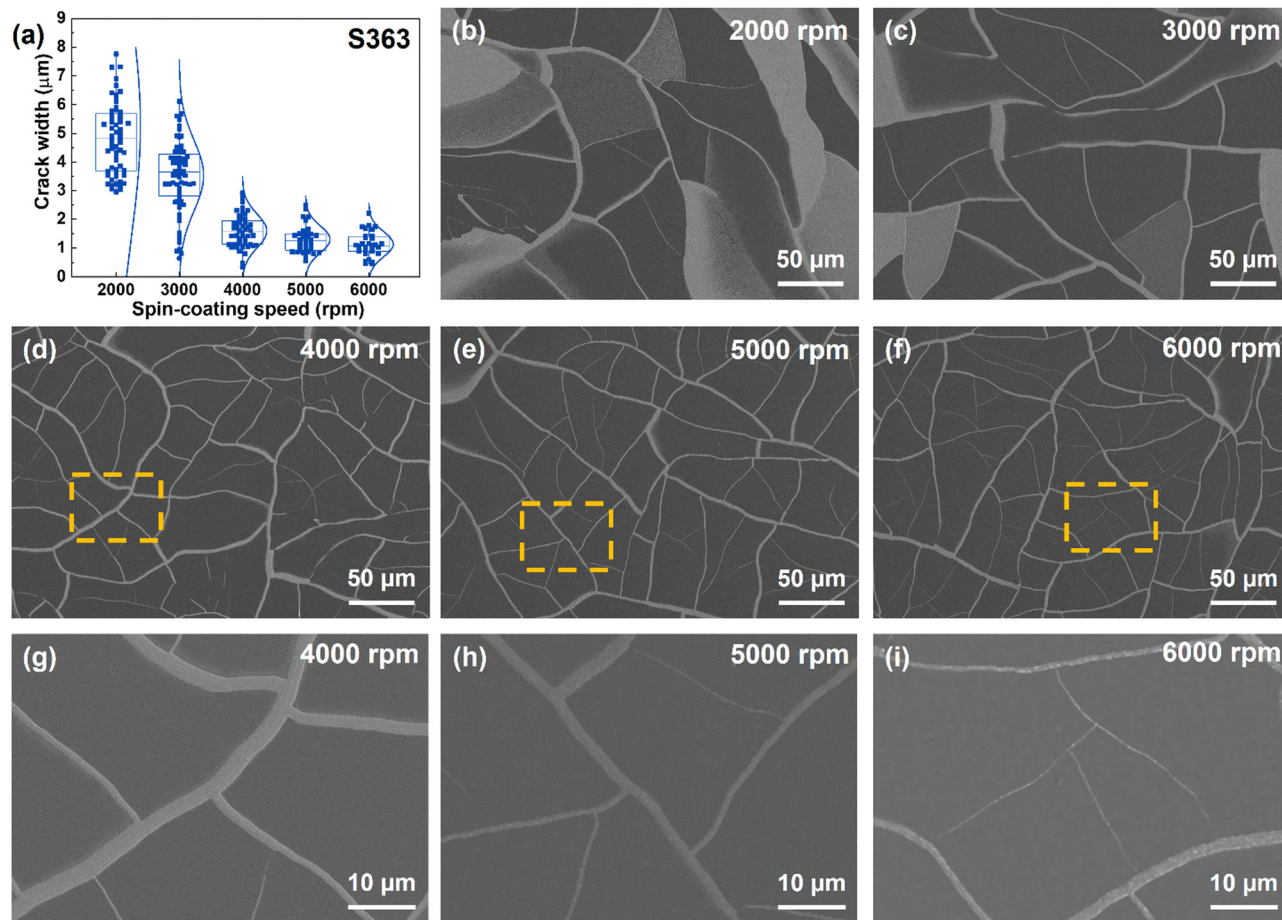


Fig. 2 (a) Crack width distributions of the Ag micro-mesh electrodes fabricated at different spin-coating speeds. FE-SEM images of the S363 samples coated at (b) 2000, (c) 3000, (d) 4000, (e) 5000, and (f) 6000 rpm. Magnified images of the yellow-squared areas in (d)–(f) for the S363 samples coated at (g) 4000, (h) 5000, and (i) 6000 rpm.

speed increased from 2000 to 6000 rpm, the crack widths and silica island size decreased, which indicates an increase in the crack density. The reduction in the silica island size can be identified by observing the dark areas in Fig. 2b–f. As noted in Fig. S1 (ESI<sup>†</sup>), the dark areas indicate the locations of the silica islands. Lines with a bright gray color in the SEM images signify the Ag micro-mesh electrode that originates from the cracks. Fig. 2 presents that crack density increases as spin-coating speed increases. This phenomenon can be attributed to tensile stress. Previous research<sup>29–31</sup> indicates that the crack density increases with decreasing film thickness due to a reduction of tensile stress. In general, a higher spin-coating speed leads to a thinner colloidal silica film, inducing the increase in crack density. Enlarged FE-SEM images of the partial areas in Fig. 2d–f marked in yellow are shown in Fig. 2g–i. Faint lines with thinner crack widths gradually appeared as the spin-coating speed increased from 4000 to 6000 rpm. In Fig. 2b and c, large areas with relatively bright colors are observed, which indicate the detached silica islands during the drying and/or sputtering process; therefore, the Ag thin film was also deposited in these areas. The exfoliated silica islands appear owing to delamination stress from the outside of the liquid

region into the wet solid region during the drying process.<sup>29</sup> As the crack templates fabricated at 2000 and 3000 rpm are thicker than the others owing to the slower spin-coating speeds,<sup>29,30</sup> the silica films coated at 2000 and 3000 rpm were subjected to higher delamination stress, causing the silica islands to be easily detached and/or delaminated. Therefore, the Ag micro-mesh electrode was successfully formed only in the S363 samples coated at 4000–6000 rpm.

To determine the correlation between the crack widths and spin-coating speed, we measured the distribution of the crack widths, as shown in Fig. 2a. The distribution of the crack widths clarified that the average crack widths decreased along with a decrease in the interquartile range, which is represented by the rectangle in Fig. 2a, as the spin-coating speed increased from 2000 to 6000 rpm.

The performances of the Ag micro-mesh electrodes were compared by calculating the FoM for a transparent electrode using the following equation:<sup>31,32</sup>

$$\text{FoM} = \frac{188.5}{R_s \left( \frac{1}{\sqrt{T}} - 1 \right)} \quad (1)$$





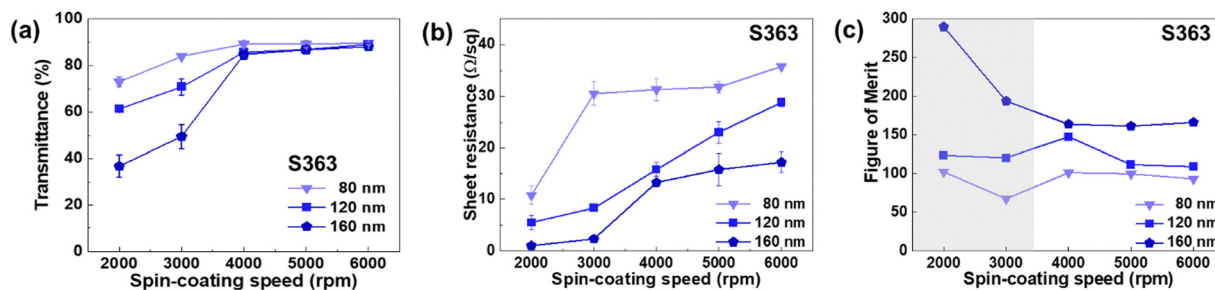


Fig. 3 (a) Transmittance, (b) sheet resistance, and (c) FoM of the transparent electrodes with different thicknesses of the Ag thin film (80, 120, and 160 nm) as a function of the spin-coating speed.

where  $R_s$  and  $T$  are the sheet resistance and transmittance at 550 nm, respectively. Generally, the transmittance of Ag micro-mesh mainly depends on the coverage area, while the sheet resistance is mainly affected by the thickness of Ag micro-mesh.<sup>28</sup> For the calculation, the transmittance and sheet resistance were measured according to the spin-coating speed as well as the thickness of the Ag micro-mesh, as shown in Fig. 3. Based on the results obtained for the S363 samples coated at 4000–6000 rpm, the transmittance values slightly increased from 84.7% to 89.6% (Fig. 3a), although the crack density increased. The reason for this can be found in the decrease in the width of Ag micro-mesh as shown in Fig. 2(a). Due to the decrease in coverage area of Ag micro-mesh by the reduction of crack width, the transmittance can slightly increase. Meanwhile the sheet resistance values significantly increased from 13.3 to 35.8  $\Omega \text{ sq}^{-1}$  (Fig. 3b), regardless of the thickness of the Ag micro-mesh. In particular, for the S363 sample with a thickness of 120 nm, the transmittance slightly improved from 85.6% to 89.0% (Fig. 3a), whereas the sheet resistance rapidly increased from 15.8 to 28.8  $\Omega \text{ sq}^{-1}$  (Fig. 3b). Accordingly, the highest value of FoM was observed for the samples coated at 4000 rpm

because of the lower sheet resistance values than those obtained at 5000 and 6000 rpm. Similarly, as the thickness of the Ag micro-mesh increased from 80 to 160 nm, the FoM values at 4000 rpm improved from 101 to 164 because the sheet resistance decreased from 31.3 to 13.3  $\Omega \text{ sq}^{-1}$ . This indicates that the FoM value is more affected by the sheet resistance than the transmittance.

The highest FoM value of 289.7 was obtained for the S363 sample coated at 2000 rpm with an Ag micro-mesh thickness of 80 nm (Fig. 3c). The highest FoM value originates from the detached silica island during the drying and/or sputtering process, so that the deposited area of the Ag thin film is larger than the others, inducing the lowest sheet resistance and transmittance. Hence, the FoM values for the S363 samples coated at 2000 and 3000 rpm (gray-shaded area in Fig. 3c) were not compared with those of the other samples. The silica solution concentration also influences the formation of a crack structure that determines the size of the silica island, crack width, and crack density. The SEM images of the Ag micro-mesh electrodes fabricated at a spin-coating speed of 4000 rpm using different silica solution concentrations of 0.242 (S242),

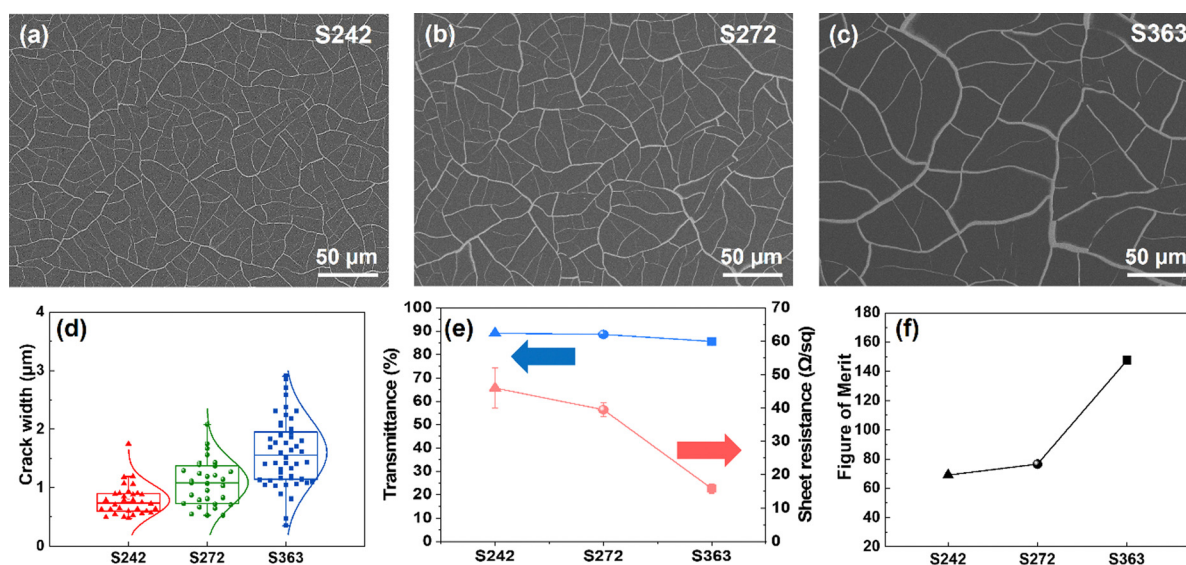


Fig. 4 FE-SEM images of the micro-mesh grid obtained using the silica solution concentrations of (a) 0.242 (S242), (b) 0.272 (S272), and (c) 0.363 g mL<sup>-1</sup> (S363). (d) Distribution of crack widths, (e) transmittance and sheet resistance, and (f) FoM of the transparent electrodes as a function of the silica solution concentration.

0.272 (S272), and 0.363 g mL<sup>-1</sup> (S363) are shown in Fig. 4a–c. The size of the silica islands decreased as the solution concentration decreased from 0.363 to 0.242 g mL<sup>-1</sup>. Calculating the crack density using a previously reported method,<sup>28</sup> the crack density increases from 15.6 (S363) to 47.4 (S242) per 1000  $\mu\text{m}$ , as shown in Fig. S2 (ESI<sup>†</sup>). Although the crack density increased rapidly, the transmittance improved slightly from 85.6% (S363) to 89.1% (S242), as indicated by the blue line in Fig. 4e. Based on an interquartile range of the boxplot in Fig. 4d, the covered areas of the Ag micro-mesh on the sample surface became similar as the crack widths gradually reduced from 1.14–1.95  $\mu\text{m}$  (S363) to 0.60–0.90  $\mu\text{m}$  (S242). The tendency for the smaller silica island to appear along with the shrinking crack widths as the solution concentration decreases is similar to that reported in the literature.<sup>33,34</sup> In contrast to the transmittance, the sheet resistance significantly increases from 15.8 (S363) to 46  $\Omega \text{ sq}^{-1}$  (S242) as the solution concentration decreased from 0.363 to 0.242 g mL<sup>-1</sup>. This is because as the crack widths of the Ag micro-mesh decreased, the thickness of the Ag micro-mesh remained the same, which increases the sheet resistance. This phenomenon implies that the crack widths determine the sheet resistance, and low sheet resistances are observed when the crack widths are mainly distributed in the range of 1.14–1.95  $\mu\text{m}$ . Consequently, the FoM value also decreased from 148 (S363) to 69 (S242) owing to the increasing sheet resistance, which reduces the performance of the Ag micro-mesh electrode.

Based on the results in Fig. 3, it is confirmed that the FoM is significantly influenced by the sheet resistance rather than the crack density or transmittance as the transmittance mostly shows similar values in this study. Additionally, in this study, the lowest sheet resistances were obtained when the crack widths were mainly distributed in the range of 1.14–1.95  $\mu\text{m}$ .

To confirm the influence of the crack width on the properties of the Ag micro-mesh electrode, samples with various crack widths were prepared by changing the spin-coating speed from 2000 to 6000 rpm and the solution concentrations from 0.363 to 0.242 g mL<sup>-1</sup>. The crack widths of all samples are in the range of 0.3–8  $\mu\text{m}$ , as shown in Fig. 5a–c. For the S242 samples coated at 5000 and 6000 rpm (Fig. 5a), cracks were not clearly observed on the silica film. The absence of cracks in these two samples was due to the formation of a silica film thinner than the critical thickness during the spin-coating process, in accordance with the literature data.<sup>35,36</sup> The transmittance, sheet resistance, and FoM values of all samples are shown in Fig. 5d–f. The FoM values for the S363 samples coated at 2000 and 3000 rpm and the S272 sample coated at 2000 rpm are excluded in Fig. 5f because the Ag micro-mesh network was unsuccessfully fabricated in these samples, which renders the comparison with other samples difficult. As shown in Fig. S3 (ESI<sup>†</sup>), the Ag thin film was also deposited in the detached areas of the silica islands in these three samples; therefore, the transmittances of the samples decreased more rapidly in comparison with the others.

Similar to the aforementioned results, the FoM values in Fig. 5f considerably depend on the sheet resistance, rather than the transmittance. In particular, the FoM value for the S272 sample decreased almost linearly, whereas the sheet resistance of S272 increased gradually when the spin-coating speed increased from 3000 to 6000 rpm. Notably, the increase in sheet resistance is related to the distribution of the crack widths, in which as the number of crack widths below 1  $\mu\text{m}$  increased, the sheet resistance increased. Fig. 5b shows that the number of crack widths below 1  $\mu\text{m}$  increases with the increase in sheet resistance in Fig. 5e, while the spin-coating speed increased from 3000 to 6000 rpm. When the numbers of crack

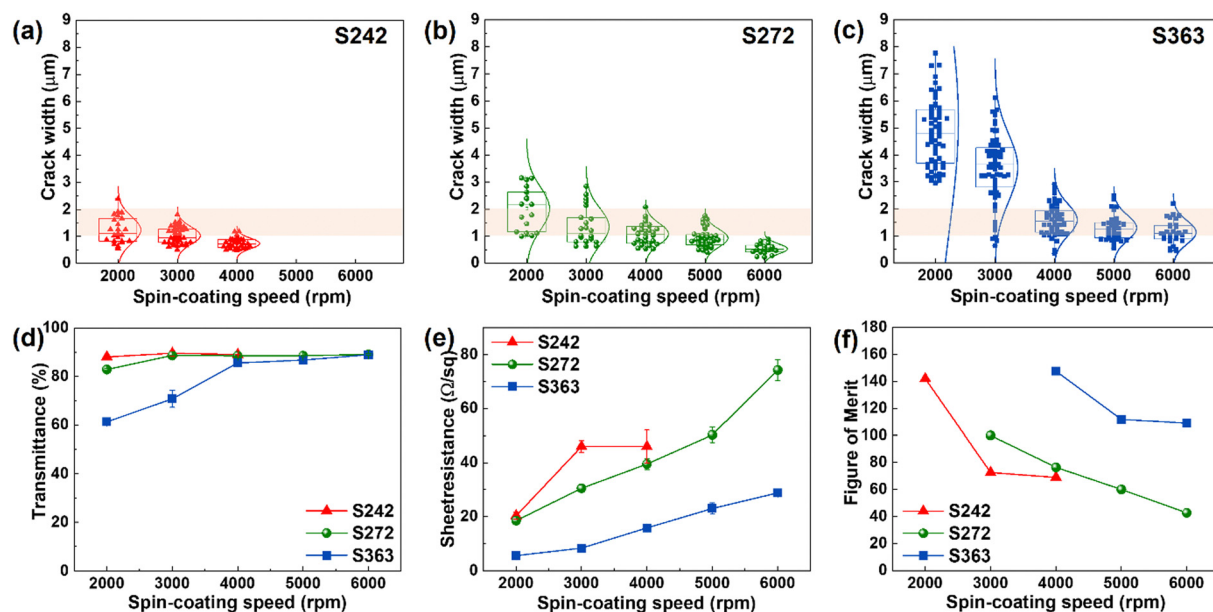


Fig. 5 Distributions of the crack width values of (a) S242 (0.242 g mL<sup>-1</sup>), (b) S272 (0.272 g mL<sup>-1</sup>), and (c) S363 (0.363 g mL<sup>-1</sup>) samples as a function of spin-coating speed. (d) Transmittance, (e) sheet resistance, and (f) FoM of the S242, S272, and S363 samples as a function of spin-coating speed.



widths below 1  $\mu\text{m}$  are similar, the sheet resistances are also similar, as shown for the S242 (coated at 4000 rpm) and S272 (coated at 5000 rpm) samples, which have the sheet resistance values of  $46 \pm 6.04$  and  $50.3 \pm 2.77 \Omega \text{ sq}^{-1}$ , respectively. Another notable observation is that when the electrode performance was high with a high FoM value, the crack widths were mostly distributed in the range of 1–2  $\mu\text{m}$ . In this study, the S363 sample coated at 4000 rpm (S363\_4000 rpm) and the S242 sample coated at 2000 rpm (S242\_2000 rpm) exhibited the first and second highest values of FoM at 148 and 142, respectively, as shown in Fig. 5f. Meanwhile, the interquartile ranges of crack widths for the S363\_4000 rpm and S242\_2000 rpm samples are 1.14–1.96  $\mu\text{m}$  and 0.84–1.67  $\mu\text{m}$ , respectively. Notably, despite the fact that the crack densities of these two samples differ significantly –23.6 per 1000  $\mu\text{m}$  for S242\_2000 rpm and 15.6 per 1000  $\mu\text{m}$  for S363\_4000 rpm (a 1.51-fold difference), as described in Fig. S4 (ESI<sup>†</sup>) – their FoM values remain nearly identical. This suggests that while crack density variations influence both transmittance and  $R_s$ , these effects compensate for each other, resulting in a similar overall FoM. Thus, our results indicate that crack width is the dominant factor in determining FoM, rather than crack density.

Fig. 6 clearly indicates that the distribution of crack widths in the samples strongly influence not only the FoM but also the sheet resistance. As shown in Fig. 6, the FoM values can be

divided into four sections, which are shaded in different colors, and each section has a similar range of crack width values. The numbers of crack widths indicated by the graphs (Fig. 6(a) and (b)) are designated as the first and third quartile values of the crack width distribution in Fig. 5a–c, which indicate the distribution range of the crack width values in the sample. Furthermore, although the transmittance presents a small difference from 85% to 90%, the sheet resistance significantly changes from 46 to  $15.8 \Omega \text{ sq}^{-1}$ . Meanwhile, the distribution range of the crack width values approaches 1–2  $\mu\text{m}$  (from 0.41–0.69  $\mu\text{m}$  to 0.82–1.95  $\mu\text{m}$ ), and the FoM value increases from 40–50 to 140–150. Fig. S4 (ESI<sup>†</sup>), which describes Fig. 6 in three dimensions, clearly presents the correlation between the FoM, transmittance, and sheet resistance. Based on Fig. S5(a) (ESI<sup>†</sup>), it is clear that the decrease in the sheet resistance is due to the optimization of the crack widths in the samples, which improves the performance of the Ag micro-mesh electrode. To more clearly illustrate the correlation between crack width and FoM, Fig. 6(c) presents the FoM values of each sample plotted against their average crack width. As clearly shown in Fig. 6(c), the FoM increases gradually as the average crack width increases from a few hundred nm to around 1–2  $\mu\text{m}$ . This result indicates that when the crack width remains below 1  $\mu\text{m}$ , the performance as a transparent electrode significantly degrades. To the best of our knowledge, while the FoM for transparent electrodes has conventionally been studied in

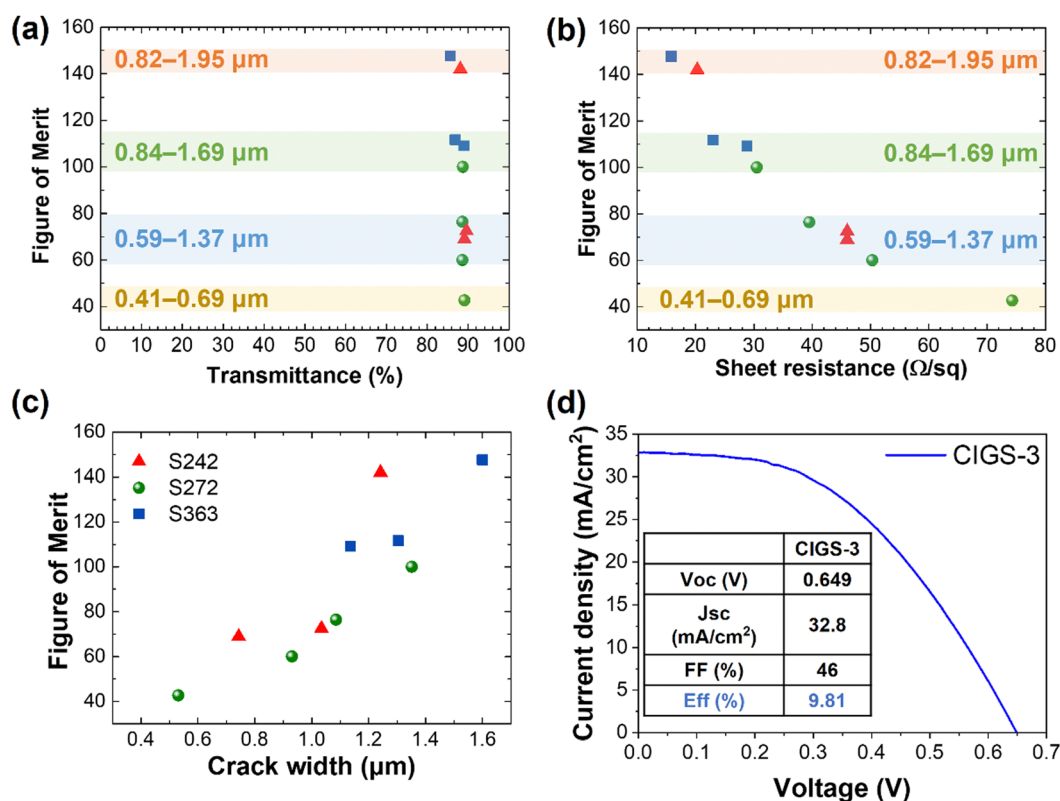


Fig. 6 FoM of the Ag micro-mesh electrodes as a function of (a) transmittance, (b) sheet resistance, and (c) average crack width of samples. The numbers stated in the colored areas of (a) and (b) represent the corresponding crack widths of the samples in the same-colored regions. (d) Current density–voltage ( $J$ – $V$ ) curves of the CIGS solar cell. The inset table provides its solar cell properties.



relation to the coverage of metallic grids, there have been no clear reports directly correlating FoM with the crack width. Fig. S5(b) (ESI†) presents that the Ag micro-mesh electrodes have a high optical transmittance, revealing that those formed by self-generated cracks can be industrially applied *via* a simple and low-cost process.

To evaluate the potential of the Ag micro-mesh network as a functional electrode in devices, it was integrated following the established methods from a previous study.<sup>25</sup> The performance of four solar cells, each equipped with the Ag micro-mesh electrode, was analysed by measuring the current density–voltage (*J*–*V*) characteristics, as shown in Fig. 6d and Fig. S6, Table S1 (ESI†). These cells achieved power conversion efficiencies ranging from 8.92% to 9.81%. Specifically, the best-performing cell exhibited a short-circuit current density (*J*<sub>sc</sub>) of 32.8 mA cm<sup>−2</sup>, an open-circuit voltage (*V*<sub>oc</sub>) of 0.649 V, and a fill factor (FF) of 46%. These results confirm that the Ag micro-mesh electrode effectively functions as a carrier collector in solar cells, demonstrating its viability as a high-performance electrode in device applications.<sup>37</sup>

## Conclusions

In conclusion, the optimum crack width for the Ag micro-mesh electrode was determined to be approximately 1–2 μm. To verify this, samples with various crack widths were prepared by adjusting the spin-coating speed and silica solution concentration. It was observed that increasing the spin-coating speed led to a more uniform distribution of crack widths, with the average crack width decreasing as the speed increased. Similarly, the silica solution concentration played a significant role in determining crack formation. More diluted solutions resulted in the generation of smaller silica islands, which in turn produced thinner crack widths in the micro-mesh network.

One of the most critical findings of this study was the strong correlation between crack width distribution and the sheet resistance. The results showed that as the number of cracks with widths below 1 μm increased, sheet resistance increased substantially. This increase in sheet resistance, however, was not accompanied by a proportional increase in transmittance, which only rose slightly. As a result, the FoM decreased considerably, leading to inferior performance of the Ag micro-mesh electrode. This suggests that the sheet resistance is more sensitive to crack width variations than the transmittance, indicating that maintaining crack widths within a specific range is key to achieving high performance. By clearly demonstrating the correlation between crack width and FoM, this study provides important insights into the optimal width of metallic grids for transparent electrodes, which represents one of the key contributions of this work.

Notably, the highest performance of the Ag micro-mesh electrode, in terms of FoM, was achieved when the crack widths were predominantly distributed in the range of 1–2 μm. In this range, the electrodes exhibited FoM values exceeding 140,

regardless of the processing conditions. Specifically, the S363\_4000 rpm and S242\_2000 rpm samples demonstrated excellent FoM values of 148 and 142, respectively, with transmittance of 88% and 82%, and sheet resistance values of 20 Ω sq<sup>−1</sup> and 15 Ω sq<sup>−1</sup>. These optical and electrical properties are comparable to those of conventional TCEs like ITO, making the Ag micro-mesh electrode a competitive candidate for TCE in various applications. Moreover, applying the Ag micro-mesh network as the metallic electrode on the top layer of a CIGS solar cell resulted in an efficiency of 9.81%, demonstrating the potential of this materials as a high-performance metallic electrode in solar cells.

In summary, the crack width distribution plays a crucial role in determining the electrical and optical properties of the Ag micro-mesh electrode, with the 1–2 μm range proving to be optimal for achieving high FoM values. This work highlights the importance of controlling material microstructure during fabrication to enhance electrode performance, and it opens up new possibilities for the use of Ag micro-mesh electrodes in advanced solar energy technologies.

## Author contributions

Seung Taek Jo, Jin Wook Shin: investigation, visualization, writing – original draft. Min-Soo Kim, Sang-Shik Park: resources. Walter Commerell: formal analysis. Hyesun Yoo, Jinyoung Hwang, Jong Wook Roh: supervision, data curation, writing – review & editing, project administration, resources.

## Data availability

The data supporting this article have been included as part of the ESI.†

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

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## References

- 1 S. Yuan, Z. Fan, G. Wang, Z. Chai, T. Wang, D. Zhao, A. A. Busnaina and X. Lu, *Adv. Sci.*, 2023, **10**, 2304990.
- 2 J. Sengupta and C. M. Hussain, *Nanomaterials*, 2022, **12**, 3146.
- 3 S.-H. Lim, H.-J. Seok, M.-J. Kwak, D.-H. Choi, S.-K. Kim, D.-H. Kim and H.-K. Kim, *Nano Energy*, 2021, **82**, 105703.
- 4 C. P. Muzzillo, M. O. Reese and L. M. Mansfield, *ACS Appl. Mater. Interfaces*, 2020, **12**, 25895–25902.





- 5 S. Kumar, D. Kang, V. H. Nguyen, N. Nasir, H. Hong, M. Kim, D. C. Nguyen, Y. Lee, N. Lee and Y. Seo, *ACS Appl. Mater. Interfaces*, 2021, **13**, 40976–40985.
- 6 S. C. Karumuthil, M. K. Ganesha, I. Mondal, A. K. Singh and G. U. Kulkarni, *J. Mater. Chem. A*, 2022, **10**, 23265–23273.
- 7 Q. Liu, Z. Liu, C. Li, K. Xie, P. Zhu, B. Shao, J. Zhang, J. Yang, J. Zhang and Q. Wang, *Adv. Sci.*, 2020, **7**, 2000348.
- 8 I. Mondal, M. K. Ganesha, A. K. Singh and G. U. Kulkarni, *Mater. Lett.*, 2022, **312**, 131724.
- 9 T.-H. Han, S.-H. Jeong, Y. Lee, H.-K. Seo, S.-J. Kwon, M.-H. Park and T.-W. Lee, *J. Inf. Disp.*, 2015, **16**, 71–84.
- 10 Z. Chen, R. T. Yin, S. N. Obaid, J. Tian, S. W. Chen, A. N. Miniovich, N. Boyajian, I. R. Efimov and L. Lu, *Adv. Mater. Technol.*, 2020, **5**, 2000322.
- 11 Z. Fakharan and A. Dabirian, *Curr. Appl. Phys.*, 2021, **31**, 105–121.
- 12 A. Kim, Y. Won, K. Woo, S. Jeong and J. Moon, *Adv. Funct. Mater.*, 2014, **24**, 2462–2471.
- 13 D.-S. Um, Y. Lee, T. Kim, S. Lim, H. Lee, M. Ha, Z. Khan, S. Kang, M. P. Kim and J. Y. Kim, *ACS Appl. Mater. Interfaces*, 2020, **12**, 32154–32162.
- 14 A. A. Green and M. C. Hersam, *Nano Lett.*, 2008, **8**, 1417–1422.
- 15 D. A. Ilatovskii, E. P. Gilshtein, O. E. Glukhova and A. G. Nasibulin, *Adv. Sci.*, 2022, **9**, 2201673.
- 16 J. Miao and T. Fan, *Carbon*, 2023, **202**, 495–527.
- 17 M. Zarei, M. Li, E. E. Medvedeva, S. Sharma, J. Kim, Z. Shao, S. B. Walker, M. LeMieux, Q. Liu and P. W. Leu, *ACS Appl. Mater. Interfaces*, 2024, **16**, 6382–6393.
- 18 A. Khan, S. Lee, T. Jang, Z. Xiong, C. Zhang, J. Tang, L. J. Guo and W.-D. Li, *Small*, 2016, **12**, 3021–3030.
- 19 S. Bae, H. Kim, Y. Lee, X. Xu, J.-S. Park, Y. Zheng, J. Balakrishnan, T. Lei, H. Ri Kim and Y. I. Song, *Nat. Nanotechnol.*, 2010, **5**, 574–578.
- 20 B. Han, K. Pei, Y. Huang, X. Zhang, Q. Rong, Q. Lin, Y. Guo, T. Sun, C. Guo and D. Carnahan, *Adv. Mater.*, 2014, **26**, 873–877.
- 21 A. S. Voronin, Y. V. Fadeev, M. O. Makeev, P. A. Mikhalev, A. S. Osipkov, A. S. Provatorov, D. S. Ryzhenko, G. Y. Yurkov, M. M. Simunin and D. V. Karpova, *Materials*, 2022, **15**, 1449.
- 22 P. B. Catrysse and S. Fan, *Nano Lett.*, 2010, **10**, 2944–2949.
- 23 S.-T. Jo, J. W. Shin, W. Commerell, D.-H. Yoo, H. Yoo, J. Hwang, H.-S. Kim, S. Kim and J. W. Roh, *Korean J. Met. Mater.*, 2024, **62**, 464–471.
- 24 K. D. M. Rao, C. Hunger, R. Gupta, G. U. Kulkarni and M. Thelakkat, *Phys. Chem. Chem. Phys.*, 2014, **16**, 15107–15110.
- 25 H. Yoo, H. Van Quy, I. Lee, S. T. Jo, T. E. Hong, J. Kim, D.-H. Yoo, J. Shin, W. Commerell and D.-H. Kim, *Energy Environ. Mater.*, 2024, e12765.
- 26 K. S. Cho, S. Kang, Y.-J. Oh, J. S. Park, S. Lee, J.-S. Wi, J.-H. Park, S. Song, K. Kim, Y.-J. Eo, J. H. Yun, J. Gwak, J.-S. Cho and C.-H. Chung, *ACS Appl. Electron. Mater.*, 2022, **4**, 823–830.
- 27 U. U. Ghosh, M. Chakraborty, A. B. Bhandari, S. Chakraborty and S. DasGupta, *Langmuir*, 2015, **31**, 6001–6010.
- 28 A. Kumar and G. U. Kulkarni, *J. Appl. Phys.*, 2016, **119**, 015102.
- 29 A. Osman, L. Goehring, H. Stitt and N. Shokri, *Soft Matter*, 2020, **16**, 8345–8351.
- 30 D. B. Hall, P. Underhill and J. M. Torkelson, *Polym. Eng. Sci.*, 1998, **38**, 2039–2045.
- 31 B. Dan, G. C. Irvin and M. Pasquali, *ACS Nano*, 2009, **3**, 835–843.
- 32 S. De and J. N. Coleman, *ACS Nano*, 2010, **4**, 2713–2720.
- 33 M. Leang, F. Giorgiutti-Dauphine, L.-T. Lee and L. Pauchard, *Soft Matter*, 2017, **13**, 5802–5808.
- 34 C. P. Muzzillo, M. O. Reese and L. M. Mansfield, *Langmuir*, 2020, **36**, 4630–4636.
- 35 V. Lazarus and L. Pauchard, *Soft Matter*, 2011, **7**, 2552–2559.
- 36 J. W. Hutchinson and Z. Suo, *Adv. Appl. Mech.*, 1991, **29**, 63–191.
- 37 K. D. M. Rao, R. Gupta and G. U. Kulkarni, *Adv. Mater. Interfaces*, 2014, **1**, 1400090.

