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[4.2.1] lactones†

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rearrangement: stereocontrolled access to bridged

Metal-free alkene carbooxygenation following

tandem intramolecular alkoxylation/Claisen

Alkene carbooxygenation has attracted considerable attention over the past few decades as this approach provides an efficient access to various oxygen-containing molecules, especially the valuable O-heterocycles. However, examples of catalytic alkene carbooxygenation *via* a direct C–O cleavage are quite scarce, and the C–O cleavage in these cases is invariably initiated by transition metal-catalyzed oxidative addition. We report here a novel Brønsted acid-catalyzed intramolecular alkoxylation-initiated tandem sequence, which represents the first metal-free intramolecular alkoxylation/Claisen rearrangement. Significantly, an unprecedented Brønsted acid-catalyzed intramolecular alkene insertion into the C–O bond *via* a carbocation pathway was discovered. This method allows the stereocontrolled synthesis of valuable indole-fused bridged [4.2.1] lactones, providing ready access to biologically relevant scaffolds in a single synthetic step from an acyclic precursor. Moreover, such an asymmetric cascade cyclization has also been realized by employing a traceless chiral directing group. Control experiments favor the feasibility of a carbocation pathway for the process. In addition, biological tests showed that some of these newly synthesized indole-fused lactones exhibited their bioactivity as antitumor agents against different breast cancer cells, melanoma cells, and esophageal cancer cells.

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Introduction

Bridged [4.2.1] lactones are widely distributed heterocycles found in various natural products such as hushinone and citrinovirin, and bioactive compounds (Fig. 1).¹ However, such bicyclic frameworks bearing both a medium ring and bridged unit are regarded as difficult skeletons to construct due to entropic effects and the ring strain factor,² and very few methods have been reported to date.¹,³ Hence, novel and stereocontrolled synthesis of bridged [4.2.1] lactone motifs allowing structurally diverse modification is in great demand in both organic and medicinal chemistry.

Difunctionalization of unactivated olefins in a single operation is one of the most valuable transformations in organic chemistry. Among them, alkene carbooxygenation is particularly attractive as this approach provides an efficient access to various oxygen-containing molecules, especially the valuable Oheterocycles, and various synthetic methods have been developed. However, examples of catalytic alkene carbooxygenation *via* a direct C–O cleavage are quite scarce, and the C–O cleavage in these cases is invariably initiated by transition metal-catalyzed

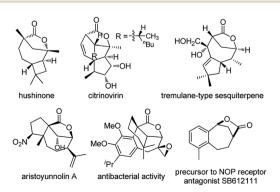


Fig. 1 Selected bioactive molecules containing bridged [4.2.1] lactones.

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oxidative addition (Scheme 1a).^{5,6} For example, Douglas *et al.* reported an elegant protocol for the rhodium-catalyzed intramolecular alkene oxyacylation reaction *via* an acyl C–O bond activation.^{6a,b} In 2012, Nakao *et al.* disclosed an intramolecular oxycyanation of alkenes by palladium/BPh₃ catalysis.^{6c} Thus, the development of an alternative approach for the catalytic cleavage of the C–O bond and the addition reaction to the alkenes is highly desirable.

Because of their high bond-forming efficiency and atom economy, catalytic tandem intramolecular alkoxylation/Claisen rearrangements have received significant attention.7 In particular, such a cascade cyclization of alkynyl allyl ethers could lead to the formation of various valuable O-heterocycles, which is also established by Hashmi, Liu, Gagosz, Miyata, and others (Scheme 1a).8-10 While these achievements are impressive, almost all of these approaches rely on the use of noble metals such as gold and platinum as catalysts. In our recent study on the catalytic tandem reactions of ynamides for heterocycle synthesis,11,12 we realized the first metal-free intramolecular alkoxylation/Claisen rearrangement of indole-linked ynamideallyl ethers (Scheme 1b). Interestingly, the resulting sixmembered lactone intermediate further underwent an unprecedented Brønsted acid-catalyzed intramolecular carbooxygenation of olefins by C-O bond cleavage involving ring opening, carbocation rearrangement, and then ring closing. This Brønsted acid catalysis led to the highly efficient and stereocontrolled formation of valuable indole-fused bridged [4.2.1] lactones. Furthermore, such an asymmetric cascade cyclization was also realized by employing a traceless chiral directing group. The mechanistic rationale for this cascade reaction is strongly supported by a variety of control experiments. In this paper, we wish to report the results of our detailed investigations of this novel cascade cyclization, including the substrate

a) C-O cleavage initiated by transition metal-catalyzed oxidative addition (general protocol)

$$R^1$$
 R^3 R^4 R^4 R^3 R^4 R^4 R^3 R^4 R^3 R^4 R^3 R^4

b) C-O cleavage initiated by proton acid-catalyzed carbocation formation (this work)

$$\begin{array}{c}
R^1 \\
R^2
\end{array}$$

$$\begin{array}{c}
R^3 \\
R^2 \\
R^3
\end{array}$$

Scheme 1 Catalytic alkene carbooxygenation via a C-O cleavage

scope, synthetic applications, biological tests and mechanistic studies.

Results and discussion

Inspired by our previous work on the indolyl ynamide chemistry, 12d-f we chose an indole-tethered ynamide 1a as the model substrate for the initial study. As shown in Table 1, indole-fused lactone 2aa was obtained in the presence of most non-noble metals (Table 1, entries 1-4), with Cu(OTf)2 giving the best yield of the desired 2aa (Table 1, entry 4). Different from Hashmi's protocol, 8b N-methyl methanesulfonamide here not only serves as the directing group to achieve regioselective attack at the Nterminus of alkyne, but also can be removed spontaneously and regarded as a traceless directing group. Surprisingly, indolefused bridged [4.2.1] lactone 2a was detected as the main product by employing In(OTf)₃ or Fe(OTf)₃ as catalysts (Table 1, entries 5-6). Of note, typical gold catalysts, such as Ph₃PAuNTf₂ and IPrAuNTf₂, were not effective in promoting this reaction and the decomposition of 1a was observed in these cases (Table 1, entries 7-8). Various Brønsted acids were also evaluated, but typical organic acids (e.g., TFA, MsOH, and TsOH) were not capable of catalyzing the reaction.13 Gratifyingly, HOTf and HNTf₂ (Table 1, entries 9-10) could effectively catalyze this cascade cyclization,14 and the bridged lactone 2a was obtained in 81% yield in the latter case (Table 1, entry 10). The reaction proved to be less efficient when it was performed at 60 °C (Table 1, entry 11) or in other solvents. 13 The observed excellent efficiency with HNTf2 as the catalyst can be explained by its high Brønsted acidity combined with the low nucleophilicity of its counterion.15,16

Table 1 Optimization of reaction conditions

Entry	Catalyst	Reaction conditions	Yield ^b (%)		
			2a	2aa	1a
1	Y(OTf) ₃	DCE, 40 °C, 48 h	<1	20	75
2	Yb(OTf) ₃	DCE, 40 °C, 48 h	<1	50	30
3	$Zn(OTf)_2$	DCE, 40 °C, 48 h	<1	70	25
4	Cu(OTf) ₂	DCE, 40 °C, 24 h	<1	84	<1
5	$In(OTf)_3$	DCE, 40 °C, 48 h	63	15	<1
6	$Fe(OTf)_3$	DCE, 40 °C, 48 h	74	<5	<1
7 ^c	$Ph_3PAuNTf_2$	DCE, rt, 10 h	<1	<1	<1
8 ^c	$IPrAuNTf_2$	DCE, rt, 10 h	<1	<1	<1
9	HOTf	DCE, 40 °C, 24 h	72	<1	<1
10	$HNTf_2$	DCE, 40 °C, 24 h	81	<1	<1
11	$HNTf_2$	DCE, 60 °C, 18 h	67	<1	<1

 $[^]a$ Reaction conditions: 1a (0.1 mmol), catalyst (0.02 mmol), DCE (2 mL), 40–60 °C, in vials. b Measured by 1 H NMR using diethyl phthalate as the internal standard. c 5 mol% of catalyst was used.

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With the optimal reaction conditions in hand (Table 1, entry 10), the scope of this novel tandem reaction was explored (Table 2). This metal-free cascade cyclization proceeded efficiently to furnish a series of indole-fused bridged [4.2.1] lactones in mostly good to

Table 2 Reaction scope for the construction of bridged [4.2.1] lactones 2^a

2x, R = CI, 61%^d

2z, 67%^d

excellent yields. For instance, various indolyl-substituted ynamides bearing both electron-donating and -withdrawing groups could be readily converted into the desired indole-fused bicyclic skeletons 2a-2p with yields ranging from 60% to 86%. In particular, the functional groups, such as CF₃, CN and CO₂Me, were well tolerated under this Brønsted acid catalysis. Further investigation of N-protecting groups demonstrated that the Bs- and Ms-protected substrates 1q-1r gave slightly improved yields. In addition, an ynamide with a methyl group $(R^1 = Me)$ was also a suitable substrate for this tandem reaction to afford the corresponding 2s in 63% yield (dr. 4:1). This result clearly indicated that [3,3] rearrangement, but not [1,3] rearrangement, 17 was presumably involved in this multiple cascade sequence.¹³ Furthermore, the reaction also occurred smoothly with various aryl- or methylsubstituted ynamides ($R^2 = \text{aryl}$, Me), and the desired 2t-2z containing a quaternary carbon center could be formed in 60-75% yields by employing 30 mol% of HNTf2 as the catalyst. Importantly, excellent diastereoselectivity (>20:1) was achieved in all cases except for the substrate 1s. The molecular structures of 2a, 2s and 2u were confirmed by X-ray diffraction. 18 Thus, this metal-free protocol provides a highly convenient and practical route for the preparation of valuable bridged [4.2.1] lactones.

In addition, this multiple cascade reaction was also applicable to other electron-rich aromatic ring-substituted ynamides such as benzofuran-, pyrrole-, and alkoxy arene-tethered ynamides 3a-3c, delivering the desired bridged [4.2.1] lactones 4a-4c in serviceable yields with excellent dr values (>20:1), as depicted in eqn (1)-(3). Of note, the use of HFIP as an additive led to a significantly improved yield in the case of 3c (eqn (3)).19 Attempts to extend the reaction to non-terminal alkenesubstituted ynamide 3d, indole-linked ynamide-allyl amine 3e and sulfide 3f led to the formation of complicated mixtures.

2w, 66%^d

^a Reaction conditions: 1 (0.2 mmol), HNTf₂ (0.04 mmol), DCE (4 mL), 40 °C, in vials; yields are those for the isolated products. b 80 °C, 36 h. c 48 30 mol% of HNTf₂ was used, 60 °C, 24 h.

Although our attempts to employ various chiral Brønsted acids such as chiral phosphoric acids and chiral phosphoric amides to catalyze this cascade cyclization failed probably due to the fact that their acidity is not strong enough, the chiral bridged [4.2.1] lactones could be synthesized by employing chiral oxazolidinone instead of N-methyl methanesulfonamide as the directing group.20 As shown in eqn (4), it was found that the tert-butyl substituted oxazolidinone-derived chiral ynamide 5d gave the best enantioselectivity, and the desired chiral bridged [4.2.1] lactone 2a-ent was formed in 55% yield with 95% ee in the presence of 30 mol% of HNTf₂ as the catalyst and 1.5 equiv. of water as the additive. Thus, the chiral auxiliary can be regarded as a traceless directing group to introduce the chirality in a very facile manner. It should be mentioned that significant racemization was observed when prolonging the reaction time.13

The scope of this asymmetric intramolecular alkoxylation-initiated tandem reaction was further examined by using *tert*-butyl substituted oxazolidinone-derived chiral ynamides. As depicted in Table 3, the reaction proceeded smoothly with various chiral ynamides 5, allowing the facile synthesis of the corresponding enantioenriched bridged [4.2.1] lactones 2-ent in serviceable yields with excellent ee values (91–99% ee) and excellent dr values (>20:1). Of note, (*R*)-tert-butyl substituted oxazolidinone-derived chiral ynamide 5d' could also undergo smooth cascade cyclization to deliver the desired 2a-ent' with the opposite enantioselectivity.

The potential synthetic utility of this protocol was then demonstrated by the facile diversification of bridged lactone 2a, and importantly, excellent diastereoselectivity was achieved in all cases (Scheme 2). For example, the Ts group in lactone 2a, prepared on a gram scale in 72% yield, was easily removed by treatment with TBAF to afford the corresponding 2ab in 75% yield. In addition, the lactone part of 2a could be selectively reduced to furnish the hemiacetal 2ac (91%, dr > 10:1) with DIBAL-H. By contrast, the use of Et₃SiH led to the total reduction of the lactone to produce the desired 2ad in almost quantitative yield. Interestingly, 2a could also be oxidized with DDQ to deliver 2ae in 95% yield. Moreover, the ring of lactone 2a was readily opened by employing PhLi, leading to the stereocontrolled construction of cyclohepta[b]indole scaffold 2af, frequently occurring in natural products and bioactive molecules.21 Finally, the chiral 2a-ent could be further transformed into the indole-fused cycloheptanone 2ah with the ee

Table 3 Reaction scope for the construction of chiral bridged [4.2.1] lactones $\mathbf{2}\text{-}ent^a$

^a Reaction conditions: 5 (0.2 mmol), HNTf₂ (0.06 mmol), water (0.3 mmol), dry DCE (2 mL), 30 °C, in vials; yields are those for the isolated products; determined by HPLC analysis. ^b 40 °C, 20 min. ^c 30 min. ^d Using (R)-configured ynamide 5**d**′ as the substrate.

maintained, and its structure was confirmed by X-ray diffraction, 18 which also determined the absolute configuration of 2a-ent 22

Considering the bioactivity reported in the literature for bridged [4.2.1] lactone systems, we also tested the above

Scheme 2 Gram scale reaction and synthetic applications. Reagents and conditions: (i) TBAF (4 equiv.), THF, 65 °C, 6 h; (ii) DIBAL-H (1.5 equiv.), THF, -78 °C, 6 h; (iii) InBr₃ (0.5 equiv.), Et₃SiH (2.5 equiv.), CHCl₃, 60 °C, 10 h; (iv) DDQ (3 equiv.), DCE, 60 °C, 48 h; (v) PhLi (1.2 equiv.), THF, -40 °C, 2 h; -20 °C, 10 h.

Edge Article Chemical Science

synthesized indole-fused lactones for their biological activity as antitumor agents. The cytotoxic effects of these compounds were evaluated against a panel of cancer cells, including breast cancer cells MDA-MB-231 and MCF-7, melanoma cells A375, and esophageal cancer cells SK-GT-4 and KYSE-450 using cell viability assay. The results revealed that these compounds exhibited differential cytotoxicity and selectivity. Compounds 2k, 2l, and 2o selectively inhibited the cell growth of A375 by more than 50% at a concentration of $20~\mu\text{M}$, and compound 2af inhibited the cell growth of MCF7 by around 70%. While compound 2ae showed broad activity with a cell viability less than 50% against cancer cells MDA-MB-231, A375 and KYSE-450.

To understand the reaction mechanism, several control experiments were conducted. First, we performed deuterium labeling studies and found that no deuterium loss was observed, thus ruling out any possible reaction pathways involving a hydride shift (eqn (5)). Importantly, 2aa was readily converted into 2a in 95% yield in the presence of HNTf₂ while no 2a was formed without an acid catalyst, strongly supporting the notion that 2aa is the key intermediate for this tandem reaction (eqn (6)). In addition, the formation of cyclohepta[b] indole 2ai (dr > 20 : 1) was detected when 2aa was treated with 10 equiv. of MeOH in the presence of HNTf₂, which indicates that the cationic intermediate is presumably involved in such a tandem sequence (eqn (7)). Of note, 2a could not be converted into 2ai in the presence of HNTf₂ and MeOH. Moreover, the cascade cyclization of ynamide 3g or 3h under the standard

1 alkoxylation
$$R^3$$
 R^3 R

Scheme 3 Plausible reaction mechanism.

conditions only led to the formation of the corresponding **4ga** (42%) or **4ha** (85%), and no desired bridged lactone product was observed (eqn (8) and (9)). These results further suggest that the intramolecular alkene insertion into the C–O bond proceeds *via* a carbocation pathway, and the generation of a stable electronrich benzylic carbocation is the key for this lactone expansion process. Finally, it was found that significant incorporation of ¹⁸O (>85%) into the product **2a** was observed in the presence of ¹⁸O-labelled water (10 equiv.), indicating that the oxygen atom on the carbonyl group of **2a** originates from water. ¹³

Based on the above experimental observations, a plausible mechanism to rationalize the formation of indole-fused bridged [4.2.1] skeleton 2 is proposed (Scheme 3). Initially, the alkoxy group attacks the acid-activated ynamide 1 to form the Claisen rearrangement precursor A via a keteniminium intermediate, which undergoes typical [3,3] rearrangement and subsequent trapping by trace water, delivering the indole-fused lactone \mathbf{D}^{23} (that is, 2aa in the case of substrate 1a) along with the generation of sulfonamide. Brønsted acid further promotes the ring opening of lactone D to produce the carbocation intermediate F via C-O bond cleavage. Finally, the carbocation of F is trapped by an intramolecular electron-rich alkenyl group to generate another carbocation intermediate G, which is further captured by the intramolecular carboxylic acid group to afford the final product 2.24 Notably, the low nucleophilicity of Tf2N might be important to maintain the cationic nature and/or reactivity of certain intermediates involved.15

Conclusions

In summary, we have developed a novel Brønsted acid-catalyzed intramolecular alkoxylation-initiated tandem sequence, which represents the first metal-free intramolecular alkoxylation/Claisen rearrangement to the best of our knowledge. Significantly, an unprecedented Brønsted acid-catalyzed intramolecular alkene insertion into the C–O bond *via* a carbocation pathway was discovered, which may serve as an alternative approach for alkene carbooxygenation *via* a direct C–O cleavage. This method enables efficient and stereocontrolled access to

valuable indole-fused [4.2.1] lactones under mild reaction conditions, providing ready access to biologically relevant scaffolds in a single synthetic step from an acyclic precursor. Moreover, such an asymmetric cascade cyclization was also realized by employing a traceless chiral directing group. A mechanistic rationale for this novel tandem reaction is well supported by a variety of control experiments. In addition, our preliminary biological tests showed that some of these newly synthesized indole-fused lactones exhibited their bioactivity as antitumor agents against different breast cancer cells, melanoma cells, and esophageal cancer cells. Thus, we believe that this novel multiple cascade reaction will not only inspire chemists to design new rearrangement processes, but also

encourage them to find their potential usefulness in organic

Conflicts of interest

and medicinal chemistry.

Chemical Science

There are no conflicts to declare.

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