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Molybdenum as a versatile dopant in SnTe: a promising material for thermoelectric application

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The key to enhancing the thermoelectric performance of SnTe is to engineer its electronic structure by doping. It is essential that the beneficial features are exhibited at the Fermi level so as to exploit the benefits without the use of a co-dopant. We report molybdenum as such a versatile dopant in SnTe. The first-principles calculations reveal that Mo is able to introduce resonance levels as well as increase the band gap in SnTe. It not only causes the convergence of light and heavy hole valence sub-bands but also does so in the conduction band. The unique feature is the Rashba splitting of the conduction bands, leading to multiband transport. The transport properties calculated using Boltzmann transport equations predict the dual nature of the resonant dopant with a promising ZT of ~ 1.84 and ~ 1.1 as a p- and an n-type dopant, respectively, in SnTe at 800 K.

Introduction

Tin telluride is considered as a promising thermoelectric material whose performance can be further improvised in converting waste heat into useful electricity.^{1–4} For enhancing the figure of merit (ZT) of a material, it is essential to increase the power factor (σS^2) and decrease the thermal conductivity (κ) simultaneously.^{5,6} Moreover, choosing the right dopant is essential as electrical conductivity (σ) and the Seebeck co-efficient (S) are interrelated in a contrary way *via* carrier concentration. In addition to having an optimum carrier concentration, a material should also have an optimum band gap to act as a good thermoelectric. This is due to the fact that an appropriate band gap will prevent the bipolar diffusion of charge carriers due to thermal excitation at higher temperatures.⁷ Further, a band gap value higher than $6k_B T$ is preferred, which is ~ 0.41 eV at 800 K.⁸ SnTe has a band gap of 0.18 eV at 300 K. It is well known that at higher

temperatures the band gap of SnTe increases due to the decrease in the valence band dispersion or bandwidth due to the decrease in the orbital overlap interaction as a consequence of the bond length increase due to thermal expansion. Also, the emphanitic behavior of the Sn^{2+} in SnTe decreases the energy of s^2 lone pair states near the valence band edge, contributing further to the increasing band gap at higher temperatures.⁹ With increase in temperature the band gap of SnTe increases at a rate of $\sim 4 \times 10^{-4}$ eV K^{-1} , which translates to a value of around 0.38 eV at 800 K, still lower than the required value and hence shows poor performance at high temperature. At high temperatures, the bipolar effect due to the low band gap is detrimental to the ZT as the total ' S ' value decreases due to the offsetting of the ' S ' values contributed by the majority carriers by the ' S ' values of opposite signs contributed by the minority carriers.^{7,10} It also causes an increase in ' κ ' values due to the heat carried by the charge carrier during ambipolar diffusion across the band gap.^{7,10} Hence, dopants like Ag, Ca, Cd, Mg, and V which can open the band gap help in improving the high temperature ZT values.^{11–18} In addition, these dopants are also known to cause convergence of valence sub-bands, resulting in increased density of states effective masses of carriers and in turn increased ' S ' values.^{13,16,18,19} Dopants which can distort the density of states near the Fermi level, *i.e.*, resonant dopants, are known to improve the performance at lower temperatures. Bi, In, Zn, V and Mn are known to introduce resonance levels in SnTe.^{20–24} A record high room temperature ZT of 0.3 (@300 K) for lead free SnTe was reported by co-doping SnTe with Bi and Zn resonant dopants.²⁴ To decrease the thermal conductivity, dopants like Cu, Sb and Pb are known to be very effective.^{2,25–29} Co-doping and multi-doping with elements are also implemented to synergistically extract different benefits and increase the ZT for a broad range of temperatures.^{2,12,13,16,20,23–31} Although resonant dopants improve the ' S ' value, they decrease the ' σ ' value due to carrier scattering and decreased mobility and hence exhibit poor power factor values.³² Many of the currently implemented techniques are good at improving the high temperature power factor, leading to a record value of $\sim 47 \mu\text{W cm}^{-1} \text{K}^{-2}$ in SnTe

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co-doped with Ca and resonant dopant In, but exhibit poor values at room temperature.³⁰ A ZT of ~ 1.85 at 823 K was achieved in SnTe by multidoping with Ca, In and Cu but the average ZT still remained at 0.67.³¹ A record high room temperature ZT of 0.35 (@300 K) and a high average ZT of ~ 0.9 between 300 K and 840 K were reported previously in SnTe by co-doping with Pb and resonant dopant Zn with a ZT maximum of ~ 1.66 at 840 K.²⁸ Hence, it is the need of the day to find resonant dopants which can exhibit beneficial features to keep the power factor high enough both at low and at high temperatures, aiding the development of lead free thermoelectric materials.

Herein we report the realization of both p and n type thermoelectric properties by doping Mo into SnTe. We show here that Mo can significantly modify the electronic structure by introducing resonance states, which is essential for improving the low temperature performance. It also enlarges the band gap of the material and causes valence and conduction sub-band convergence essential for high temperature performance.¹³ The interesting feature is the multi-band transport resulting from the Rashba splitting of the conduction bands, leading to enhanced

performance of the material. The transport property calculations based on the electronic structure predicted the material to be a potential high-performance material worth experimental investigation.

Computational details

The electronic structures of pristine SnTe and Mo doped SnTe were simulated using the open-source software Quantum ESPRESSO package.³³ A $\sqrt{2} \times \sqrt{2} \times 2$ supercell of a rock salt structure containing 32 atoms was used for the first-principles calculations. Pseudopotentials with a generalized gradient approximation to exchange–correlation energy with the Perdew, Burke, and Ernzerhof (PBE) functional type were used to treat the valence and semi-core electronic states.³⁴ To involve the spin–orbit coupling interaction in the determination of their electronic structures, fully relativistic pseudopotentials considering Sn ($4d^{10}5s^25p^2$), Te ($4d^{10}5s^25p^4$) and Mo ($4s^24p^64d^55s^1$) as valence electrons were used. It needs to be noted that the conventional

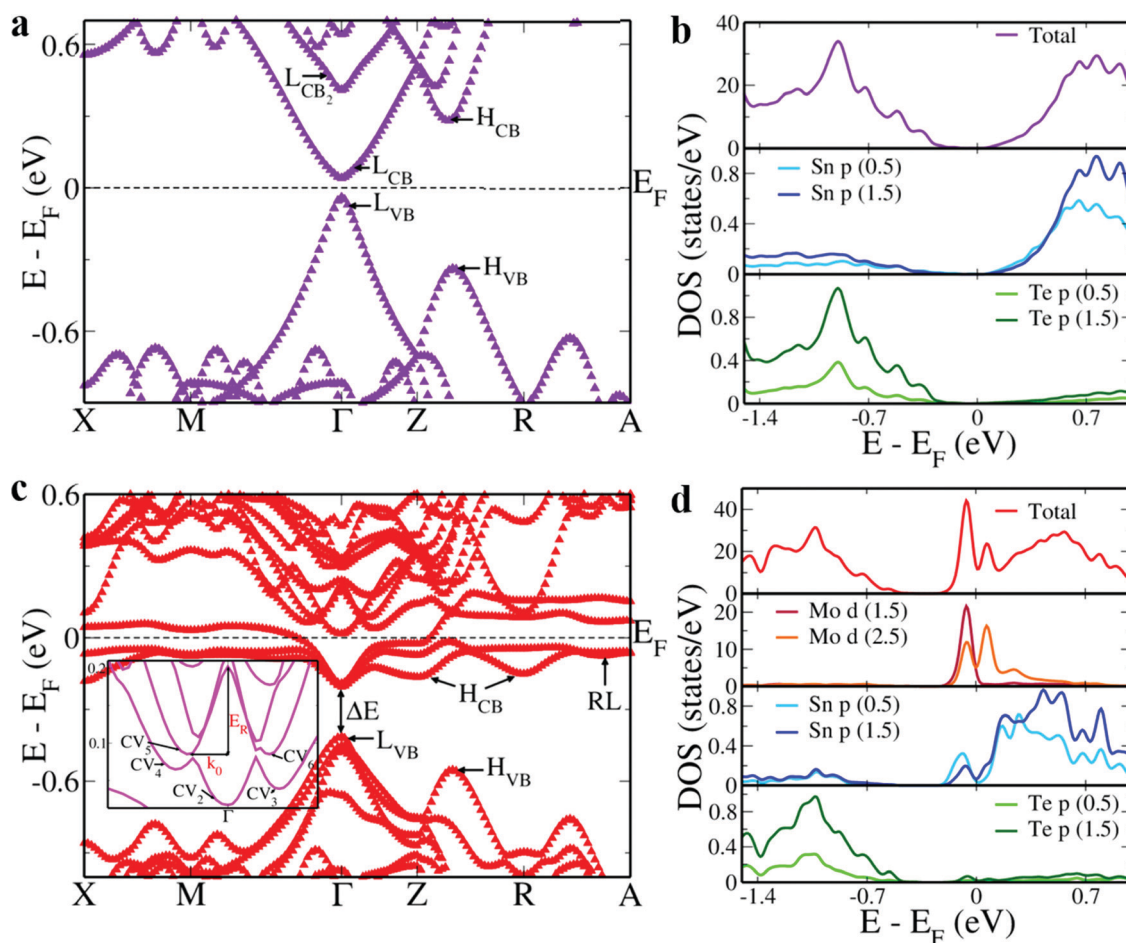


Fig. 1 Electronic structures and PDOS of $\text{Sn}_{16}\text{Te}_{16}$ (a and b) and $\text{Sn}_{15}\text{MoTe}_{16}$ (c and d). L_{VB} , L_{CB} , H_{VB} , and H_{CB} are the light hole valence band, light electron conduction band, heavy hole valence band and heavy electron conduction band respectively. L_{CB_2} refers to the second set of eight-fold degenerate light electron conduction bands at the Γ point in $\text{Sn}_{16}\text{Te}_{16}$. RL refers to the resonance level. The inset of (c) shows the Rashba splitting occurring in the conduction band as a result of Mo doping. CV_2 , CV_3 , CV_4 , CV_5 , and CV_6 indicate various valleys in the conduction band, potentially contributing to the transport properties. E_{R} refers to the Rashba energy and the momentum offset is indicated by k_0 .



GGA underestimates the band gap of the materials while incorporation of the Hubbard U factor or hybrid functionals can give better band gap estimates.^{35,36} Nevertheless, the trend in the band gap predicted by the current method used is accurate. The total energy was determined using a k -mesh of $14 \times 14 \times 10$ points, and an energy and charge density cutoff of 50 Ry and 400 Ry, respectively.

Results and discussion

The electronic structure largely influences the power factor due to the competing traits of ' σ ' and ' S ' and hence plays a crucial role in improving the performance of thermoelectric materials.^{2,10,13} We see an increase in the principal band gap (ΔE) of SnTe after Mo doping at the Γ point from 0.08 to 0.21 eV (Fig. 1). The appearance of a band gap at the Γ point due to the folding of the L point in the supercell and the underestimation of the band gap in DFT based calculations are well documented in previous reports.^{13,22,24,28,37} The large energy separation between the heavy hole (H_{VB}) and light hole (L_{VB}) valence bands of 0.30 eV observed in SnTe is a matter of concern.²⁴ As the Fermi level increases in energy with increase in temperature, it leads to the suppression of the contribution from the heavy hole valence band at the $Z + \delta$ point in the $Z \rightarrow R$ direction, resulting in decreased ' S ' values.¹⁵

We observe that substitutional doping with Mo leads to a decrease in this energy offset to 0.14 eV, leading to an increase in the number of degenerate valleys, contributing to higher ' S ' values.³⁰ Such band convergence has a very little influence on carrier mobility and hence keeps the power factor high.¹⁹ Similarly, we estimate a decrease in energy offset in the conduction band from 0.24 eV to 0.04 eV and 0.05 eV considering the heavy electron conduction sub-bands (H_{CB}) at the Z point and R point, respectively. This shift in the position of H_{CB} from the $Z + \delta$ point to the Z and R points is a consequence of alterations taking place in the higher energy conduction bands which is explained later. Another heavy carrier band is also generated at the R point 0.1 eV above the Fermi level. The density of states (DOS) reveals that in both doped and undoped SnTe the conduction bands are formed from Sn ' p ' orbitals and the valence bands from Te ' p ' orbitals. The doped configuration in addition shows a well defined peak near the Fermi level formed due to the hybridization of Mo ' d ' orbitals with Sn ' p ' orbitals, indicating the formation of resonance levels. While in the case of V doped SnTe we see the contribution of the d orbitals with both j values 1.5 and 2.5 at one point, here we see the d orbitals of Mo with a j value of 2.5 giving rise to two peaks like a doublet.¹⁸ The eight-fold degenerate conduction band loses its degeneracy with substitution of Mo in the Sn site and splits into four sets of doubly degenerate bands. The lowermost three sets form the

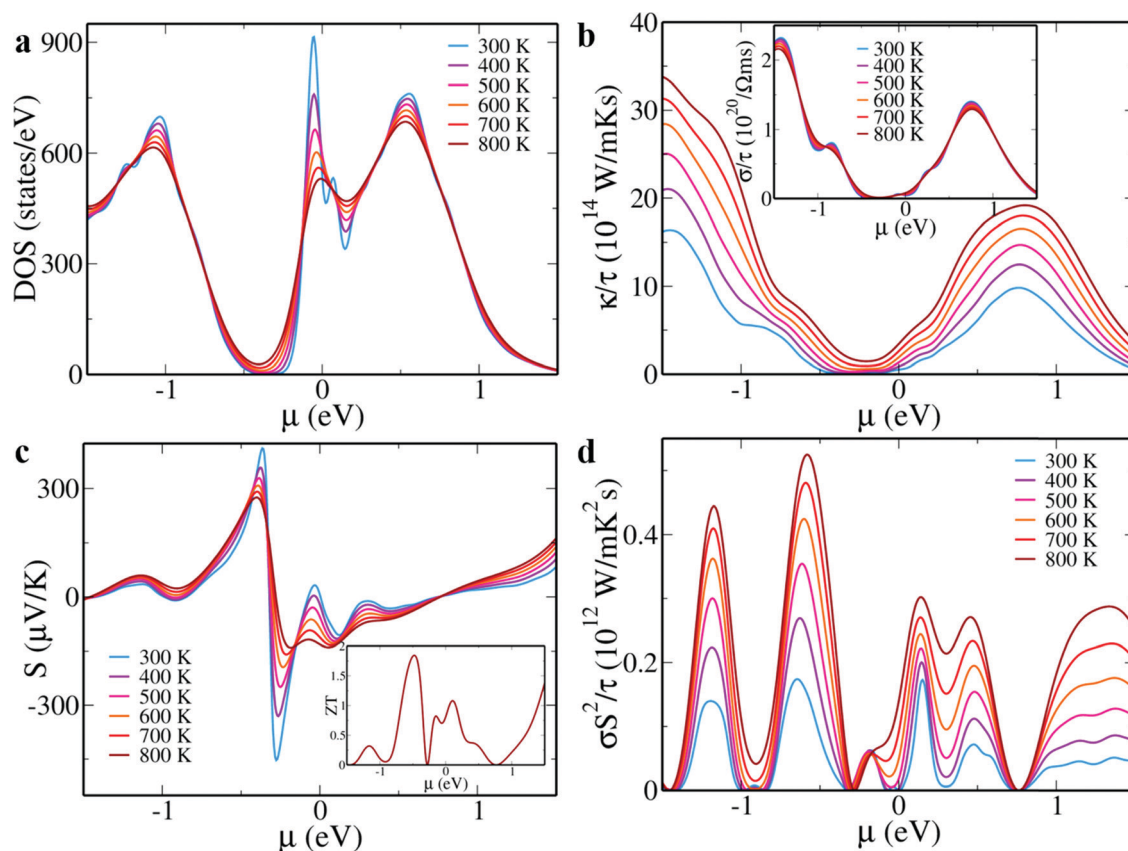


Fig. 2 (a) DOS; (b) thermal conductivity (inset: electrical conductivity); (c) Seebeck co-efficient (inset: $ZT@800$ K) and (d) power factor of $\text{Sn}_{15}\text{MoTe}_{16}$ as functions of chemical potential (μ) at various temperatures. Conductivity and power factor values are reported by scaling them with τ .



We determined the DOS, ' σ ', ' S ', ' σS^2 ' and ' κ ' as functions of temperature and chemical potential ' μ ' using a rigid band approximation (Fig. 2). BoltzTraP code was used to implement the semiclassical theory under a constant scattering time approximation.⁴² At ' μ ' = 0, the DOS revealed a peak due to the resonance level, the intensity of which decreased with the increase in temperature.⁴³ While ' σ ' and ' κ ' showed similar variations with change in ' μ ' from the p- to n-type, the ' κ ' values showed prominent increases in the values with increase in temperature.⁴⁴ The sign of ' S ' changed from positive to negative as we moved from the negative potential region corresponding to hole doping to the positive potential region corresponding to electron doping.⁴³ Large positive and negative ' S ' values indicate the success of tuning the electronic structure by doping Mo. The power factor shows 5 prominent peaks, with higher values in the p-type region. Assuming a lattice thermal conductivity of $0.4 \text{ W m}^{-1} \text{ K}^{-1}$ we estimated ZT values of ~ 1.84 and ~ 1.1 at 800 K in the negative and positive potential regions, respectively. This indicates that the material can act as both p and n-type by tuning its chemical potential.¹⁸ The predicted p-type ZT is much higher than the recently predicted ZT of 1.66 for V doped SnTe exhibiting similar features.¹⁸ Implementing various strategies to reduce the lattice thermal conductivity further may lead to increases in the ZT values, resulting in a promising thermoelectric material.^{19,45,46}

In this work we have reported the versatile nature of molybdenum in improving the thermoelectric properties of SnTe.

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