

A review from the laboratories of Professor Judith MacManus-Driscoll at the University of Cambridge, UK, Professor Rui Wu at the South China University of Technology, Guangzhou, China, and Professor Weiwei Li at the Nanjing University of Aeronautics and Astronautics, China.

Interface-related phenomena in epitaxial complex oxide ferroics across different thin film platforms: opportunities and challenges

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As featured in:



See Judith L. MacManus-Driscoll, Rui Wu and Weiwei Li, *Mater. Horiz.*, 2023, 10, 1060.



Cite this: *Mater. Horiz.*, 2023, 10, 1060

Interface-related phenomena in epitaxial complex oxide ferroics across different thin film platforms: opportunities and challenges

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Interfaces in complex oxides give rise to fascinating new physical phenomena arising from the interconnected spin, lattice, charge and orbital degrees of freedom. Most commonly, interfaces are engineered in epitaxial superlattice films. Of growing interest also are epitaxial vertically aligned nanocomposite films where interfaces form by self-assembly. These two thin film forms offer different capabilities for materials tuning and have been explored largely separately from one another. Ferroics (ferroelectric, ferromagnetic, multiferroic) are among the most fascinating phenomena to be manipulated using interface effects. Hence, in this review we compare and contrast the ferroic properties that arise in these two different film forms, highlighting exemplary materials combinations which demonstrate novel, enhanced and/or emergent ferroic functionalities. We discuss the origins of the observed functionalities and propose where knowledge can be translated from one materials form to another, to potentially produce new functionalities. Finally, for the two different film forms we present a perspective on underexplored/emerging research directions.

Received 15th December 2022,
Accepted 16th February 2023

DOI: 10.1039/d2mh01527g

rsc.li/materials-horizons

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1. Introduction

Strongly correlated oxides are fascinating and widely explored functional materials, with an unrivalled range of physical properties. Their most perfect and applicable form is as epitaxial films. To exploit emergent and strain-related interface effects in epitaxial films, typically either superlattice (SL) or vertically aligned nanocomposite (VAN) film forms are studied, each offering different phenomena.



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While there have been many important reviews on emergent functional properties of SL¹ and VAN,² a broader article comparing the phenomena in these different structures has not been presented. The two fields have typically followed quite separate paths by quite separate groups. At first glance, this seems somewhat surprising since the tools for creating the films, the functional materials being studied, and the desired multifunctionalities being aimed for are either the same or are similar. On the other hand, there are subtle differences in approach and knowledge-sets for creating each film form which has precluded overlap between them.

In this review we focus on ferroic properties in the two materials as these are particularly exciting and because they exemplify the salient effects offered by each platform. We aim to present the key differences in the functionalities achieved by exploiting interface effects in the two thin film forms, comparing the merits and drawbacks of each and the reasons for those differences. We also aim to give insight into what can be learned from one form for the benefit of the other. We highlight novel phenomena and promising areas for future studies. We first compare the basic growth and structural features of each form and then we compare their exemplar ferroic properties. After that, we then explore in more detail the individual ferroelectric, ferromagnetic and multiferroic properties of SLs and follow this by exploring the same three phenomena for VANs. After each SL ferroics section, we discuss what can be learned for VAN from the SL field. Where appropriate, we do this also for each VAN section also. Finally, we propose future perspectives for each, including underexplored/emerging research directions.

Ferroics are fascinating exemplar materials to explore for our different film forms as they exhibit very wide ranging and tunable properties with huge future applications prospects. Ferroics refer to materials presenting at least one intrinsic long-range ordering such as magnetization, electric polarization, *etc.* The first application of a ferroic was in lodestone, which was used as a simple magnet and later as a compass, more than 2000 years ago. An understanding of the basic origin

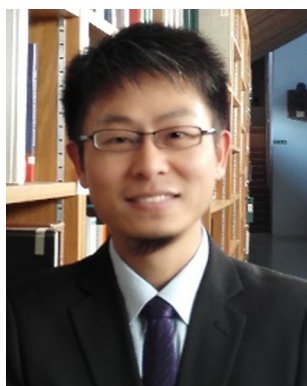
of the ferroics was not achieved until the beginning of the last century with the help of quantum mechanics. Over the past century, and in particular in the last decades, many new materials, of metallic, semiconducting, insulating and even organic materials classes, have been added to the ferroics collection. This endows them with very important applications in a range of current electronics devices.

Ferroelectrics have a spontaneous electric polarization that can be switched by an external electric field,³ or by mechanical or optical stimuli.⁴ They are used in a wide range of radio-frequency and microwave devices, as well in thermal sensors, actuators, and transducers, *etc.*,⁵ and there is also strong interest for non-volatile memory.⁶ Magnetic thin films are used in magnetic recording media, the leading technology for mass data storage.⁷ Also, magnetic random-access memory (MRAM) could be the winning technology for next-generation 'universal' memory.⁸ Another important ferroic form is multiferroics where two ferroic properties coexist, giving promise of magnetoelectricity for multi-state memory in which there is control of magnetism with an electric field. Magnetoelectric random access memory (MeRAM) possesses all advantages of MRAM, but crucially at much lower energy consumption.⁹

For the field of ferroic electronics, thin films of transition metal oxides (TMOs) are of particular interest as they show a very wide range of ferroic properties which are linked to the d orbitals in TM ions. These orbitals, in the presence of a strong ligand field from oxygen ions bonded to the TM ions, produce a wide range of excited states, polarisation effects, and electronic properties. These effects are highly tunable owing to the strong correlation between charge, orbital and structural degrees of freedom of the TM–O bond. Of the TMOs, oxide perovskites have been most widely studied owing to their wide ranging functionalities, ease of growth by pulsed laser deposition (PLD) and compatibility with widely available perovskite single crystals.

If one adds an interface between different TMOs, then symmetry breaking effects, strain and strain-relieving defects⁶ at the interface add yet another handle of tunability, coupling and complexity to TMO functionality. Interfaces and the consequent coupling between two different TMOs can be accomplished effectively in either SL or VAN forms. Conventional SLs have interfaces parallel to the substrate and are grown one layer at a time using vacuum deposition methods. There are a couple of notable exceptions where SLs form by self-assembly from a combined target material, *i.e.* TiO₂/VO₂¹⁰ and YBa₂Cu₃O_{7-x}/BaZrO₃¹¹ the first formed by spinodal decomposition and the second by phase separation.

The schematics of SL and VAN epitaxial thin film forms, grown mainly by PLD, but less often also by molecular beam epitaxy (MBE) for SL films, are shown schematically in Fig. 1, with structural characteristics and interface-related phenomena shown. For far-from-equilibrium processing methods (including PLD but also sputtering), while vertical columnar grains form in standard planar films, these are different to the VAN columnar grains. For VAN columnar grains, their growth is controlled by vertical epitaxy with the matrix phase in the film



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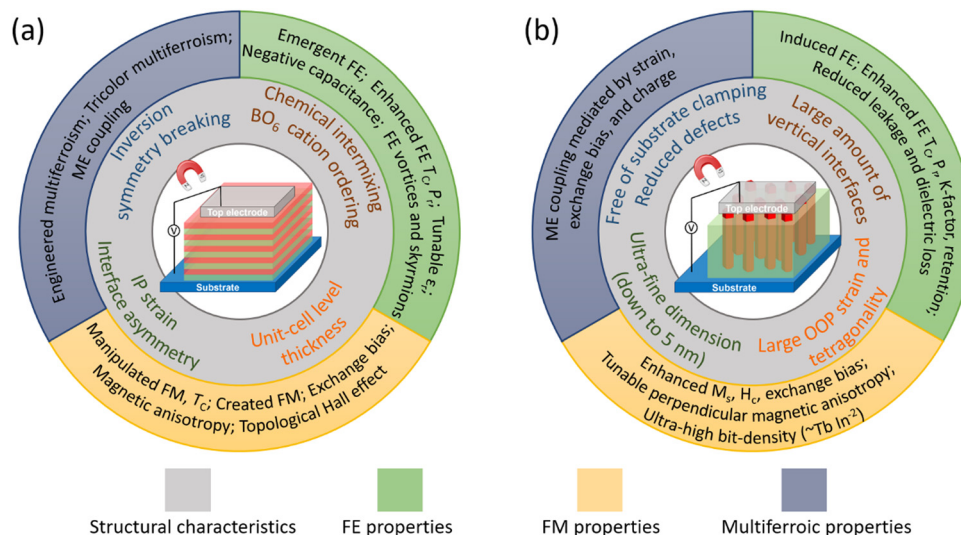


Fig. 1 Schematic diagrams showing structural characteristics (inner ring) and observed interface-related phenomena (outer ring) in epitaxial ferroic thin films formed from (a) SL structures and (b) VAN structures. Tricolor multiferroism is a kind of multiferroism induced in a three-component SL. See Fig. 2–8 for more details about how both structural and physical effects link to the observed interface-related phenomena.

and there is vertical epitaxial strain associated with this, as well as potential for interface phenomena at these interfaces, whereas for standard planar films such effects do not occur. The remaining figures in the paper go into more depth than Fig. 1, explaining in detail the origins of the interface phenomena in both VAN and SL films.

We note that multilayer (ML) films are similar to SL films, except they contain fewer layers ($< \sim 5$) than SL ($> \sim 5$), and the individual film layers are typically thicker than for SL (*e.g.* > 15 – 30 unit cells for ML, and < 15 unit cells for SL). In this paper, MLs are discussed where SLs have not yet been presented.

Both SLs and VANs have large area interfaces which enable physical phenomena to be both precisely tuned, *e.g.*

using external stimuli such as magnetic or electric fields, and probed, *e.g.* using spectroscopic tools. Table 1 compares the growth and structural features of the two different materials forms.

The overall drive in research on VANs has been more applied owing to simplicity of growth, and the ability to grow relatively thick, highly strained films resulting from a high density of vertical interfaces, whereas for SL films the focus has been on basic science, enabled by the growth of well-defined 2D interfaces. The growth of SLs is more complex than VAN, requiring reflection high energy electron diffraction (RHEED) and precise substrate heating to carefully control the thickness and qualities of the individual layers. On the other hand, SL growth produces much more well-defined 2D interfaces than VAN,

Table 1 Comparison of growth and structural features of SL and VAN epitaxial thin film forms made by PLD or also (although less often) by MBE for SL films

	SL	VAN
Complexity of growth process?	Higher	Lower
Very wide variety of materials combinations possible?	Y	N
Intermixing at interfaces?	Few unit cell level	Unit cell level ^g
Substrate control of strain	Y	N ^a
Easy and uniform control of in-plane strain	Y (in films < 20 nm)	N
Easy and uniform control of out-of-plane strain	N	Y (in films > 20 nm) ^b
Misfit dislocations at interfaces	N ^c	Y ^d
Dimensionality of interfaces	2D	Non-ideal 2D
Deviation from dimensionality	Very low ^e	Wide ^f

^a The substrate influences strain only for ≤ 20 nm film thickness. Above this thickness, the vertical strain dominates.¹² Hence, strain can be 3-dimensional with auxetic-like effects possible.¹³ ^b > 1000 nm thicknesses possible. ^c Insufficient time in SL to nucleate and grow dislocations between layers so other defects form to alleviate strain. ^d Growth of vertical interfaces slow, enabling misfit dislocations to form.¹⁴ ^e Interfaces become less 2D-like as the number of superlattice layers increases above ~ 10 – 20 layers and then the layers roughen. ^f Interface directions can deviate from the vertical direction, and hence can be tortuous. ^g In VAN, there can be chemical segregation of species at an atomic level right at the interface.¹⁵ This is not the same as chemical intermixing which occurs by reaction between two materials at either side of the interface. This is the case for SL where you have two separate materials grown from separate targets and which can then react at the interface during growth. In VAN, if intermixing is thermodynamically favoured it will have already occurred when the materials are all mixed together in the ceramic target. It means there is not additional chemical reaction/chemical intermixing at the interfaces between the two materials.



making them ideally suited to exploring emergent interface effects.

VANs grow in a one-shot process from a single ceramic target, and the VAN structure forms epitaxially by self-assembly in the film. The growth of VANs does not need RHEED or very precise control of growth conditions as for epitaxial SLs where each film layer must be very smooth. However, to create new interface-coupled effects in well-aligned VANs, one needs to predict which phases will self-assemble in a given VAN film, starting from the target of all the mixed cations. This requires a greater knowledge of the stabilising forces of epitaxy with the substrate and thermodynamics of the possible phases that can arise from the global cation mix. A simple example is as follows: you may want to grow a VAN of NiO and Fe₃O₄ on a standard single crystal substrate. Depending on the Ni:Fe ratio in the target, and assuming standard growth conditions, you will achieve either NiO + NiFe₂O₄ or NiFe₂O₄ + Fe₃O₄, and hence not the desired NiO and Fe₃O₄ mix. This example does not mean you will always achieve a ternary phase and a binary phase when the aim is for two binaries. VANs of binaries are indeed possible where intermediate ternaries do not form, e.g. TiO₂ + VO₂.¹⁶

The basic ground rules for growing VANs have been set out in comprehensive reviews, the first in 2010¹⁷ with others building on the understanding subsequently.¹⁸ Suffice for this article is to note that in VAN films a matrix phase forms in the film with an inclusion phases. The inclusion phase should be structurally dissimilar to the matrix phase to prevent ready mixing of the two materials during growth. This is a different

situation to SLs where structurally similar phases are required so that the artificial interfaces that form between them are more perfect. If there is a sufficiently high-volume fraction of the inclusion, it typically forms as nanopillars which grow embedded in the matrix phase. The nanopillars are fine, typically <20 nm in diameter and ~5–50 nm pitch, controllable by selecting the appropriate growth parameters.^{2b}

In VAN films the vertical interfaces formed with the matrix can be faceted semi-coherent interfaces or cylindrical surface incoherent interfaces. We term both types of interfaces non-ideal 2D. Owing to the fine pillar dimensions and pitch, it is possible for large vertical and uniform strains to be induced in the matrix from the pillar scaffolds.^{17,19,20} More details about the large vertical strains for VAN films are discussed in Section 6. The most perfect VANs are achieved when the matrix material is structurally the same or very similar to the substrate.^{15b} Here, a single interface type (lowest lattice mismatch one) will form, enabling the high density, highly vertical interfaces to be created. For this situation, strain control to >1 micron film thickness has been shown.^{14,21} On the other hand, when the matrix material is lattice mismatched to the substrate and also to the second phase in the VAN film, the films grow less perfectly.²² Here, the interfaces have higher energy than when films are isostructural. To minimise the vertical interface energies, a range of different interfaces with similar interfacial energy can form, which leads to meandering of the vertical interfaces. Examples are shown in later figures in this paper. More work needs to be done to give a more quantitative description of VAN microstructures in relation to the properties

Table 2 Comparison of maximum or novel ferroic properties created using SL and VAN films

SL	VAN
<p>FE In 10–200 nm thick films T_C of BaTiO₃/SrTiO₃ increased by several hundred K²³ $P_r \sim 16.5 \mu\text{C cm}^{-2}$, the value is doubled in SrTiO₃/BaTiO₃/CaTiO₃ <i>cf.</i> plain BaTiO₃ film²⁴ Tunability values ~40% at 200 kV cm⁻¹ and 300 K in SrTiO₃/BaTiO₃/SrO²⁶ which is >3 higher than for BaTiO₃ plain films</p> <p>Negative capacitance observed in Ba_{0.8}Sr_{0.2}TiO₃/LaAlO₃ and PbTiO₃/SrTiO₃ at 300 K²⁸ Polar vortices observed in PbTiO₃/SrTiO₃^{28b,29}</p>	<p>In ~micron thick films T_C of BaTiO₃-Sm₂O₃ VAN increased by several hundred K²¹ $P_r \sim 12.1 \mu\text{C cm}^{-2}$, the value is trebled in (Ba,Sr)TiO₃-Sm₂O₃ VAN film ~ <i>cf.</i> plain (Ba,Sr)TiO₃ film²⁵ Tunability values are ~75% at 200 kV cm⁻¹ and 300 K in (Ba,Sr)TiO₃-Sm₂O₃ VAN film which is ~40% higher than for (Ba,Sr)TiO₃ plain film.²⁷ Also, loss scales with tunability, which is opposite to plain films. Negative capacitance not yet observed</p> <p>Polar vortices not yet observed</p>
<p>FM In-plane exchange bias as high as 3 kOe induced in CaRuO₃/CaMnO₃ at 10 K³⁰</p> <p>High interfacial magnetization of up to 2.5 μ_B (396 emu cm⁻³) achieved in LaFeO₃/LaCrO₃³² High T_C ferromagnetism of up to 375 K created at interfaces from non-magnetic parent materials in LaFeO₃/LaCrO₃³² Perpendicular magnetic anisotropy (K_{eff}) of 4×10^6 erg cm⁻³ (~13 kOe) realized in La_{1-x}Sr_xMnO₃/SrIrO₃ at 10 K³⁵ Topological Hall effect up to 75 μΩ cm observed in LaMnO₃/SrIrO₃ at 10 K³⁷ DME coefficient α_{ME} up to 55 V cm⁻¹ Oe⁻¹ at 300 K observed in BaTiO₃/BiFeO₃³⁸</p>	<p>Perpendicular exchange bias of ~0.91 kOe at room temperature obtained in NiO-NiFe₂O₄ VAN²² <i>cf.</i> ~0.1 kOe in NiO-NiFe₂O₄ bilayer at room temperature.³¹ Saturation magnetisation in Fe₃O₄-BiFeO₃ VAN film ~900 emu cm⁻³ <i>cf.</i> ~300 emu cm⁻³ in pure Fe₃O₄ film with similar thickness³³ Ferro/ferrimagnets with several-hundred-K T_C created from AFM parent material ZnFe₂O₄³⁴ Perpendicular magnetic anisotropy of 85 kOe induced in BaTiO₃-CoFe₂O₄ at 300 K <i>via</i> strain³⁶ No topological Hall effect observed</p> <p>DME coefficient α_{ME} up to 20 V cm⁻¹ Oe⁻¹ at 300 K measured in BiFeO₃-CoFe₂O₄³⁹</p>
<p>ME No sizable CME yet observed</p>	<p>Self-biased CME effect at room temperature with ME coefficient of ~1.4 × 10⁻⁹ s m⁻¹ at 300 K achieved in Na_{0.5}Bi_{0.5}TiO₃-NiO-NiFe₂O₄⁴⁰</p>

FM – ferromagnetism. FE – ferroelectricity, ME – magnetoelectricity. DME = direct magnetoelectric coefficient, CME = converse magnetoelectric coefficient.



of the two materials in the film and the growth conditions, and the consequences of the nature of the meandered microstructure on the strain levels in the films and thus the functional properties.

The kinetics of interface formation are very different for SL and VAN films which influences the perfection of the interfaces (and regions in the vicinity of the interfaces). For VAN films the vertical interfaces grow slowly, allowing misfit dislocations to form to partially relieve the strain, whereas for SLs 'forced' artificial interfaces form very rapidly, which can lead to the formation of point-like defects to relieve any interfacial strain effects.¹⁴ Also, since the VAN interfaces are formed at equilibrium and are self-assembled, they are not chemically intermixed as SL film interfaces often are.^{1a} This, however, comes at the expense of a lower possible number of materials permutations that can be combined together.

The most enhanced or novel ferroic properties from SL and VAN films, *i.e.* the maximum possible values or novel ferroic properties achieved for each materials form, are summarised in Table 2. Different materials pairs between SL and VAN are compared because different pairs are compatible for each thin film form. Hence for VAN, one typically requires structurally dissimilar materials to prevent a single mixed phase from forming, while for SL structurally similar materials are required to enable high quality interfaces to form. From Table 2, it is clear that the two materials forms offer quite different maximum or novel effects. An important question is what can be learned from one materials platform and translated to another? For example, could a topological Hall effect be achieved in VAN similar to SL, or could a self-biased magnetoelectric effect be achieved in SL similar to VAN?

The rest of the paper expands on the information in Table 2, discussing the different ferroic phenomena of ferroelectricity, ferromagnetism and multiferroicity observed in SL films (part A) and VAN films (part B), with potential synergies between the two film forms noted. Finally, an overall summary, perspective and conclusions are presented (part C).

A. Ferroics in superlattice films

2. Ferroelectric phenomena in superlattice films

Ferroelectric polarization in oxide perovskites of the ABO₃ formula is induced by polar displacement in B-site cations at the centres of BO₆ octahedral cages, or displacement of A-site (lone-pair active) cations which sit between the octahedra. In SL films, a wide range of fascinating ferroelectric properties have been observed, *e.g.* ferroelectricity induced at the interfaces between two non-ferroelectric systems, enhanced T_C s and higher remanent polarization values, improved dielectric tunability, negative capacitance effects, and polar skyrmions/vortices. In the following, we provide details of exemplar recent systems where each property has been observed and, in some

cases, also manipulated, and we explain how the strain, interfaces, and defects in the SLs contribute to these properties. In Section 6, we compare the same effects in VAN films.

2.1 Induced ferroelectricity in superlattice films

Remarkably, in several systems SLs formed of two different non-ferroelectric perovskite layers have given rise to emergent ferroelectricity.⁴³ For example, in SLs of non-ferroelectric (NdMnO₃)_n/(SrMnO₃)_n/(LaMnO₃)_n, ferroelectricity was created below 40 K (Fig. 2a), tunable by varying SL periodicity.⁴¹ The emergent ferroelectricity was ascribed to broken space inversion symmetry and the Mn³⁺/Mn⁴⁺ mixed valency.

No such emergent interface FE has been reported in VAN films to date. While it is certainly harder to grow such perfect vertical interfaces of desired compositions in VAN films, it is possible in some systems where the matrix film is highly similar in both structure and lattice parameter to the substrate.⁴⁴ In fact, the effect has likely never been searched for, and is a fruitful avenue of future research, particularly in manganite-based VAN systems where mixed Mn valence is induced by the presence of oxygen vacancies. Indeed, interface structures in VAN systems predicted using DFT calculations show strongly distorted oxygen sublattices and higher oxygen vacancy concentrations.^{15b} Hence, in a SrMnO₃/CeO₂ VAN film, interface-induced oxygen vacancies are anticipated. These would reduce some of the Mn⁴⁺ to Mn³⁺. Hence, it is possible that emergent FE could arise from this mixed valency in VAN films, just as in SL films. The signal would, in theory, be much higher than in SL films, owing to the higher density of interfaces. We note that a reduced overall cation valence at interface structures in VAN films would not be limited to manganites. Other transition metal oxides would also likely give mixed valent cation interfaces and hence wide-ranging novel interfacial properties. This is a rich area for future exploration.

Another proven route to inducing ferroelectricity in SLs is to induce octahedral rotations into a SL layer from another SL layer. This leads to hybrid improper ferroelectricity.⁴⁵ It has been shown in perovskite SLs with the same B ions but different A (A and A') ions, hence A/A' SLs, *e.g.* LaGaO₃/YGaO₃. The octahedral rotations induce a sizeable polarization of 0.65 μC cm⁻². This same effect would likely not be possible in VAN films, owing to the larger lateral pitch of the vertical structures, *i.e.* in the range 5–50 nm, rather than <1–3 nm as required here. We note also that ferroelectricity across the interfaces in the SLs is predicted to originate from charge transfer,⁴⁶ charge order,⁴⁷ and element intermixing.⁴⁸ Similar effects are likely also taking place in VAN structures but, as far as known, have not been probed yet. This would be an interesting area of future research.

2.2 Enhanced ferroelectric T_C and remanent polarization, P_r , in superlattice films

(BaTiO₃)_n/(SrTiO₃)_m SL with $n = 1$ (single unit cell thick BaTiO₃) and $m = 30$ grown on (001) SrTiO₃, with 2.3% lattice mismatch in the BaTiO₃,^{23,49} gives a ferroelectric T_C of 250 K,²³ *i.e.* lower than the bulk value of 403 K (Fig. 2b), which means the SrTiO₃



Ferroelectric phenomena in SL films

Structural and physical effects responsible for observed interface-related phenomena

- Broken symmetry and mixed valency
- In-plane (IP) strain and multiple domains
- Broken symmetry, IP strain and interface structure
- Chemical pressure and IP strain
- IP strain
- IP strain and interface confinement



Fig. 2 Ferroelectric phenomena in SL films. (a) P - E hysteresis loops measured from $(\text{NdMnO}_3)_n/(\text{SrMnO}_3)_n/(\text{LaMnO}_3)_n$ superlattices. Reproduced with permission.⁴¹ Copyright 2012 Springer Nature. (b) FE T_c in $(\text{BaTiO}_3)_n/(\text{SrTiO}_3)_m$ superlattices as a function of n and m layer numbers. Reproduced with permission.²³ Copyright 2006 AAAS. (c) Left panel: STEM image and diagram of $(\text{SrTiO}_3)_x/(\text{BaTiO}_3)_y/(\text{CaTiO}_3)_z$ superlattices. Right panel: P - E hysteresis loops measured from $(\text{SrTiO}_3)_2/(\text{BaTiO}_3)_4/(\text{CaTiO}_3)_2$ and $(\text{SrTiO}_3)_{10}/(\text{BaTiO}_3)_{10}/(\text{CaTiO}_3)_{10}$ superlattices. Reproduced with permission.²⁴ 2005 Springer Nature. (d) Left: Schematic of crystal structure of $(\text{SrTiO}_3)_{n-1}/(\text{BaTiO}_3)_1/\text{SrO}$ superlattices and STEM images the Ti, Ba, Sr and (Ba + Sr) elements. Middle: Real and imaginary parts of dielectric constant (K , the ratio of the material's permittivity ϵ to the permittivity of vacuum ϵ_0) as a function of frequency for the $n = 6$ superlattice film. The inset shows the film loss tangent. Right: The ratio of $K(E_{\text{bias}})/K(0)$ (without applied electrical field) (left-hand axis) and relative tunability (right-hand axis) of the $n = 6$ superlattice film. Reproduced with permission.²⁶ 2020 Springer Nature. (e) Temperature dependent capacitance for superlattices grown on SrTiO_3 , DyScO_3 , and GdScO_3 substrates. Reproduced with permission.^{28a} Copyright 2014 American Chemical Society. (f) Left panel: STEM image of vortex structures formed in a $(\text{SrTiO}_3)_{10}/(\text{PbTiO}_3)_{10}$ superlattice. Reproduced with permission.^{29a} 2016 Springer Nature. Right panel: Optical and thermal field control of the coexistence phase and superlattice phase. Reproduced with permission.⁴² 2019 Springer Nature.

is effective in straining the BaTiO_3 and the atomically thick nature of the BaTiO_3 does not seriously degrade the ferroelectricity. Furthermore, by tuning n and m , the T_c of the BaTiO_3 is increased up to 638 K. The higher-than-bulk T_c arises from the strain effect applied to the BaTiO_3 layers by the SrTiO_3 (Fig. 2b).²³ A similar approach, using thin ferroelectric and non-ferroelectric perovskite layers, has been applied to other FE

SL combinations,⁵⁰ both bilayers and even trilayers. A trilayer example is shown in Fig. 2c. Here x unit cells of SrTiO_3 , y unit cells of BaTiO_3 , and z unit cells of CaTiO_3 ($\text{S}_x\text{B}_y\text{C}_z$) were grown on (001) SrTiO_3 substrates.²⁴ The strain in BaTiO_3 layers is fully retained when its thickness does not exceed the combined thickness of the other layers. As a consequence, an overall 50% enhancement of the SL global remanent polarization, P_r , is



achieved (Fig. 2c). Apart from the strain effect, the broken inversion symmetry and specific interface structure play unexpected roles in the enhancement of ferroelectric polarization. The broken inversion symmetry in three-component SLs has been used for investigating the ferroelectric ground states, the dielectric properties,⁵¹ and multiferroicity, as we discuss more in Section 4.

As we show later, the T_C in VAN films is highly tunable, and, just as for SL films, several hundred K T_C enhancements are readily achievable. The biggest differences with SLs is the ease of growth of VAN films combined with the ability to maintain vertical strain to > micron thicknesses,^{21,25} whereas in SLs the total film thicknesses are only several 10's unit cells maximum. VANs are therefore more suited to actuator/energy harvesting/dielectric applications where larger volumes of material are needed, whereas SL films are more suited to Si-based electronics where charges at planar interfaces play an important role in switching for processor or memory applications.

2.3 Dielectric property enhancements in superlattice films

Owing to temperature/frequency-stable permittivity and dielectric tunability, dielectric materials are of interest in applications where device miniaturization and high-energy-density storage are needed.⁵² (Ba,Sr)TiO₃ is an established compound that has been used in the development of tuneable dielectrics in microwave filters. There is no ferroelectric transition for pure SrTiO₃ as it is paraelectric in bulk, but the ferroelectric T_C is 403 K for BaTiO₃. The T_C is in between these values by varying the Ba: Sr ratio. However, charged pointed defects in (Ba, Sr)TiO₃ thin films, notably oxygen vacancies, lead to increased dielectric loss.⁵³

(SrTiO₃)_{*n*-1}(BaTiO₃)₁SrO SLs with $n = 6$ (single unit cell thick BaTiO₃) (Fig. 2d) have lower dielectric loss than standard films. This is because the single unit cell thick BaTiO₃ in the SL is under chemical pressure from the SrTiO₃ layers. This likely reduces oxygen loss at the same time as increasing the tetragonality and hence also the ferroelectric polarization. Overall, at a temperature $T = 300$ K and an electric field $E = 400$ kV cm⁻¹, a dielectric constant (K) tunability, *i.e.* the ratio of $K(E)/K(0)$ (right panel in Fig. 2d), of around 46%, and a loss tangent ($\tan \delta$) of ~ 0.06 were achieved at a 110 GHz frequency.²⁶ Higher tunability values were achieved in much thicker ($\sim \mu\text{m}$) VAN films (Section 6), *i.e.* dielectric constant tunability of up to 75% and $\tan \delta$ of 0.01,^{27,54} although the frequency studied so far was lower than for SL films, at ~ 1 MHz.

Another form of ferroelectricity in SLs has been demonstrated *via* improper ferroelectricity. In (PbTiO₃)_{*n*}(SrTiO₃)_{*m*} SLs with very short periods of $n = 2$ and $m = 3$, improper ferroelectricity⁵⁵ is achieved resulting from strong competition between ferroelectric PbTiO₃³⁹ and antiferrodistortive SrTiO₃⁵⁶ distortions across the interface, causing antiferrodistortive rotations of the oxygen atoms in PbTiO₃. In turn, a relatively high dielectric constant of 600 is achieved at 300 K and, from 300 K to 550 K, the dielectric constant is independent of temperature. No improper ferroelectricity has been observed in VAN films to date but there is the possibility to observe it in

the most perfect VAN structures. The PbTiO₃/SrTiO₃ VAN system would be a good starting point. While phase intermixing between PbTiO₃ and SrTiO₃ could occur in the VAN films, there is precedent for lack of mixing between similarly lattice matched perovskite VAN materials.⁵⁷ Thus for PbTiO₃/SrTiO₃ VAN, a similar situation could come about. Less (or no) mixing compared to the bulk thermodynamic case related to the role of epitaxy at the substrate film interface. Hence pure SrTiO₃ on SrTiO₃ is the lowest interfacial energy situation and may prevent intermixing with Pb. More experiments and calculations are highly required to confirm this possibility.

2.4 Negative capacitance and emergent polar vortex in superlattice films

The negative capacitance effect in ferroelectrics occurs when there is a negative differential capacitance around a zero charge level. Hence, an increase in charge in the ferroelectric occurs when the voltage is decreased. This effect could reduce the subthreshold swing in a conventional transistor below the otherwise minimum limit of 60 mV decade⁻¹.^{28a,58} Negative capacitance in SL films of ferroelectric Ba_{0.8}Sr_{0.2}TiO₃/dielectric LaAlO₃^{28a} has been shown with the capacitance enhancement level tuned by epitaxial strain *via* the growth of SLs on three different substrates (Fig. 2e). Furthermore, in PbTiO₃/SrTiO₃ SLs a direct measurement of steady-state negative capacitance has shown that local regions of negative capacitance emerge at domain walls, where the energy density is larger than that in the bulk of the domains and the polarization is suppressed.^{58a}

Another novel phenomenon observed recently in PbTiO₃/SrTiO₃ SLs is the formation of polar vortices in PbTiO₃. In fact, ferroelectric vortex states were theoretically predicted in nanostructures,⁵⁹ but not observed in thin films until 2016. Long-range and chiral ferroelectric vortex structures have been observed in PbTiO₃ in (PbTiO₃)_{*n*}/(SrTiO₃)_{*n*} SLs with $n = 10$ (left in Fig. 2f).^{29a} A combination of substrate induced strain and confinement from the SrTiO₃ layers produces the effect leading to a clockwise vortex-like pattern, paired with every other vortex in an anti-clockwise manner. The coexistence phase of ferroelectric a_1/a_2 domains and vortex states can be transferred to a single 3D SL phase by optical and thermal manipulation (right in Fig. 2f).⁴² Room-temperature polar-skyrmion bubbles have further been observed in these SL systems,^{28b} giving the possibility of highly efficient skyrmion memory.

There have been no such reports thus far of polar vortices in VAN films. However, the ~ 5 nm length scales, strain and confinement effects involved in the above SL example, can also be readily achieved in VAN films, even in the PbTiO₃/SrTiO₃ system. Hence, PbTiO₃ pillars should form in a SrTiO₃ matrix when the film is grown on SrTiO₃. The Pb and Sr would be unlikely to mix based on the differing ion sizes and charges, and also because it would not be preferred because pure SrTiO₃ forms a perfect epitaxial match on SrTiO₃. Thus, it may be possible to achieve few nm sized nanopillars of PbTiO₃ as can be achieved in other VAN systems⁶⁰ and thus create vertical polar skyrmions throughout the whole film thickness.



3. Magnetic phenomena in superlattice films

Similar to ferroelectric oxide thin films, magnetic oxides, notably transition metal oxide (TMO) perovskites, have potential applications for data storage and non-volatile random-access memory. Strong magnetization, high T_C , and perpendicular anisotropy are highly desired for such applications. A broad range of functionalities arising from the interplay between the interconnected degrees of freedom of the magnetic spins, lattice structure, bonding orbitals and ionic charges, mediated by corner-shared BO_6 units octahedra which can tilt around three orthogonal axes⁶¹ can be exploited and manipulated in SLs.

At SL interfaces, the following effects have been shown: the enhancement of T_C , modified electronic conduction, induced ferromagnetism from non-ferromagnetic layers, exchange bias, magnetic anisotropy, and topological Hall effect, *etc.* (Fig. 3). In

Section 7, we discuss the comparative magnetic effects in VAN films.

3.1 Magnetic interactions across interfaces involving 3d TMO superlattice films

Exchange bias (EB) has been induced in SLs *via* two different effects. First, charge transfer (to equilibrate the differing chemical potentials of carriers in the materials on either side of the interface) occurs in many oxide SLs and leads to new electronic and magnetic states.⁶⁶ For example, in $(\text{CaRuO}_3)_3/(\text{CaMnO}_3)_N$ SLs composed of the antiferromagnetic insulator CaMnO_3 and the paramagnetic metal CaRuO_3 , an exchange bias as high as 3 kOe is induced at 10 K (Fig. 3a).^{30,67} Here, a ferromagnetic interface is created in one unit cell, because of the double exchange interaction in the CaMnO_3 at the interface where Mn^{4+} is partially reduced to Mn^{3+} because of electron transfer from CaRuO_3 to CaMnO_3 . The consequent

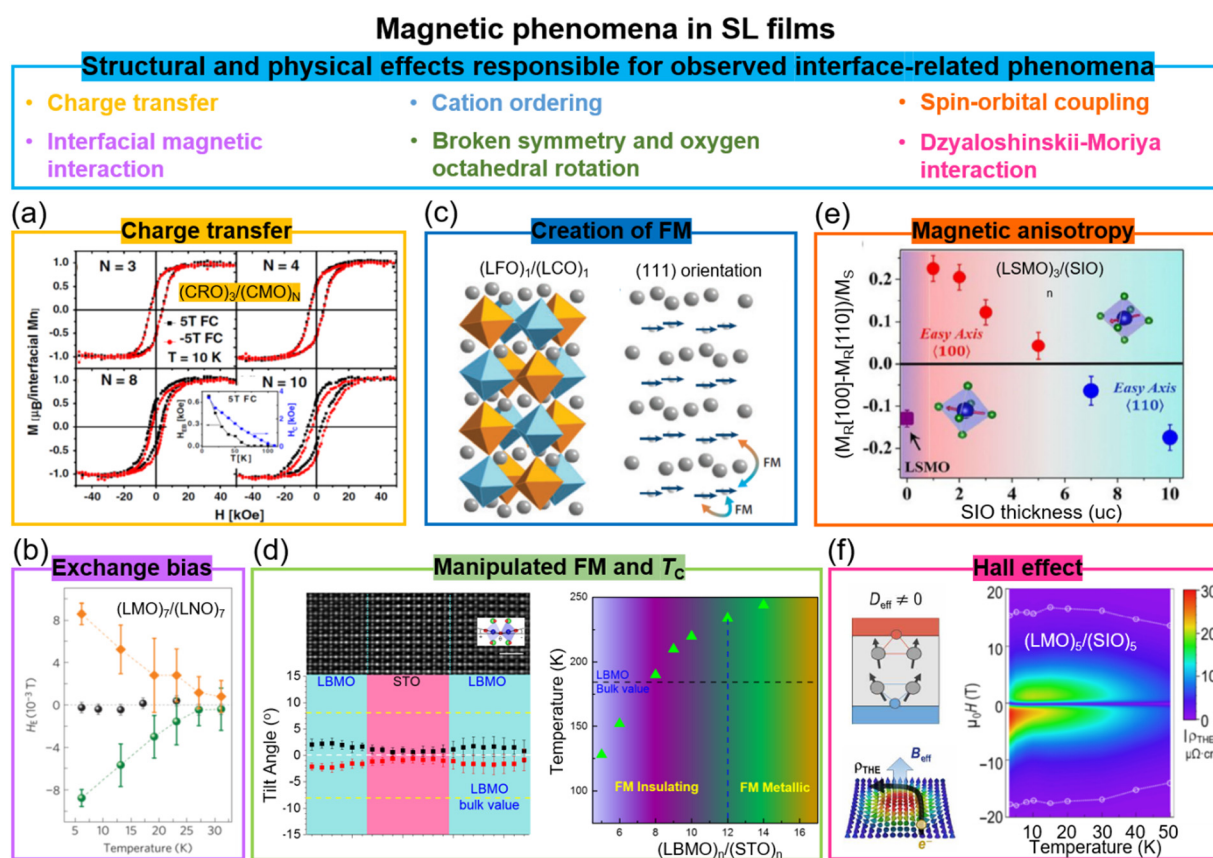


Fig. 3 Magnetic phenomena in superlattice films. (a) Magnetic hysteresis loops of $(\text{CaRuO}_3)_3/(\text{CaMnO}_3)_N$ superlattices. Reproduced with permission.³⁰ Copyright 2012 American Physical Society. The inset shows the temperature dependent H_C and H_{EB} for $(\text{CaRuO}_3)_3/(\text{CaMnO}_3)_8$. (b) Temperature dependent H_E for $(\text{LaNiO}_3)_7/(\text{LaMnO}_3)_7$ superlattices after field-cooling in a +0.4 T field (green circles) and in a -0.4 T field (orange diamonds). Reproduced with permission.⁶² Copyright 2012 Springer Nature. (c) Schematic diagram of $(\text{LaFeO}_3)_1/(\text{LaCrO}_3)_1$ superlattices grown on (111)-oriented substrates and the resultant spin structures. Reproduced with permission.⁶³ Copyright 2014 Annual Reviews. (d) Left panel: STEM image (top) and octahedral tilt angles (bottom) of $(\text{La}_{0.9}\text{Ba}_{0.1}\text{MnO}_3)_8/(\text{SrTiO}_3)_8$ superlattice. Right panel: Ferromagnetic T_C as a function of layer thickness in $(\text{La}_{0.9}\text{Ba}_{0.1}\text{MnO}_3)_n/(\text{SrTiO}_3)_n$ superlattices. Reproduced with permission.⁶⁴ Copyright 2020 Wiley-VCH. (e) Magnetic anisotropy of $(\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3)_3/(\text{SrIrO}_3)_n$ superlattices as a function of SrIrO_3 layer thickness. Reproduced with permission.⁶⁵ Copyright 2016 PNAS. (f) Left panel: Schematic diagram of the topological Hall effect originating from the Berry phase and a chiral spin texture. Right panel: The topological Hall resistivity of the $(\text{LaMnO}_3)_5/(\text{SrIrO}_3)_5$ superlattice. Reproduced with permission.³⁷ Copyright 2020 AAAS.



ferromagnetic/antiferromagnetic coupling across the interface creates the exchange bias effect.

Exchange bias is also achieved in (111)-oriented $\text{LaMnO}_3/\text{LaNiO}_3$ (ferromagnetic/paramagnetic) SLs (Fig. 3b). Here, interfacial interactions, *e.g.* quantum confinement effects,⁶⁸ induce a complex magnetic structure in a non-magnetic material. Hence, magnetism resembling a spin density wave in the paramagnetic LaNiO_3 layers in contact with the ferromagnetic LaMnO_3 is created.⁶² An exchange bias as high as 90 Oe is induced at 5 K. Large vertical exchange bias effects have been demonstrated at much higher temperatures in VAN films, namely at room temperature, and the reasons for this are discussed in Section 7.

Similar to the $(\text{CaRuO}_3)_3/(\text{CaMnO}_3)_n$ SLs, in $(\text{LaMnO}_3)_{2n}/(\text{SrMnO}_3)_n$ SL, a metallic ferromagnetic state emerges at the interface from the parent materials which are not ferromagnetic, namely an antiferromagnetic Mott insulator (LaMnO_3),⁶⁹ and a band insulator with G-type antiferromagnetism (SrMnO_3).⁷⁰ Again, interfacial charge transfer between Mn^{3+} and Mn^{4+} cations is responsible for emergent FM, *i.e.* electrons 'leak' from the Mn^{3+} in LaMnO_3 into the Mn^{4+} in SrMnO_3 .^{70,71} The metallic state is apparent for $n \leq 2$ (when the occupied $\text{Mn } e_g$ electrons spread over the entire lattice), but for higher n , the bulk of the film (insulating antiferromagnetic) dominates the properties.

Emergent ferromagnetism and metallicity should also be easily achievable in VAN films, as charge transfer across interfaces is not restricted to planar interfaces. To date, in several VAN systems interesting additional magnetic and conducting effects have been observed (separate to the expected contributions from the bulk properties to the film).^{15a,72} However, the origins of these effects, whether they be from charge transfer effects, orbital reconstructions, *etc.*, have not been widely explored. Further studies of such effects in perfectly aligned VAN structures, including understanding of the phenomena using high resolution microscopic and scanning probe tools to probe the structures and interface charge states is a fruitful area of future research. Also, since the vertical interfaces in VAN films intersect the film surface, the conducting properties of VAN interfaces are readily measurable giving an advantage over SL where interfaces are buried.⁷³

Another mechanism to induce ferromagnetism at interfaces is by tuning the relative fractions of B–O–B, B–O–B', and B'–O–B' bonds. In $(\text{ABO}_3)/(\text{AB}'\text{O}_3)$ SLs, there are different magnetic exchange interactions couplings across the interfaces for different orientation films. Hence, different spin structures have been observed for $(\text{LaFeO}_3)_1/(\text{LaCrO}_3)_1$ SLs grown on (001)-, (110)-, and (111)-oriented SrTiO_3 substrates.^{32a,74} As shown in Fig. 3c, based on Anderson, Goodenough, and Kanamori rules, ferromagnetic exchange interactions across Fe–O–Cr bonds can be realized in the (111)-oriented SL films. A high T_C of 375 K and high interfacial magnetization of $2.5\mu_B$ were achieved in (111)-oriented $(\text{LaFeO}_3)_1/(\text{LaCrO}_3)_1$ SL.

A further interfacial phenomenon which induces modified interfacial properties in magnetic perovskite SLs is orbital reconstruction. This occurs when TMO thin films are strained

leading to different BO_6 octahedral rotations.^{61,64} For example, the ferromagnetic T_C in $\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3$ is significantly enhanced from 370 K in bulk $\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3$ to 650 K in $(\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3)_3/(\text{BaTiO}_3)_3$ SLs owing to straining of the $\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3$ by BaTiO_3 .⁷⁵ The change of the Mn orbital occupation from $d_{x^2-y^2}$ within the film to $d_{3z^2-r^2}$ at the interface is responsible.

A further mechanism for tuning the magnetic coupling within SL layers containing magnetic moments is *via* non-trivial oxygen octahedral coupling (OOC) across the heterointerfaces which produces different BO_6 octahedral rotations in the layer of interest.^{61c,76} For instance, in non-magnetic $(\text{LaMnO}_3)_n/(\text{SrTiO}_3)_n$ SLs, magnetisation was induced in the interfacial layers in LaMnO_3 with a maximal saturation magnetization of $2.97\mu_B$ per Mn for the $n = 6$ SL film.⁷⁷ It should be noted that while interactions between layers in the films lead to the modulation of BO_6 octahedral rotations, epitaxial strain also play an important role. To largely eliminate the strain effect, $(\text{La}_{0.9}\text{Ba}_{0.1}\text{MnO}_3)_n/(\text{SrTiO}_3)_n$ SLs were studied,⁶⁴ where there is only 0.2% strain between the $\text{La}_{0.9}\text{Ba}_{0.1}\text{MnO}_3$ and SrTiO_3 . It was possible to engineer the BO_6 octahedral rotations up to 12 unit cells, larger than previous reports by 6 unit cells. Consequently, the out-of-plane Mn $d_{3z^2-r^2}$ orbitals became occupied, the Mn e_g bandwidth reduced, a ferromagnetic insulating state in $\text{La}_{0.9}\text{Ba}_{0.1}\text{MnO}_3$ created, and T_C enhanced to 235 K from 185 K in bulk (Fig. 3d).

As already noted, in SL films the substrate plays a role in controlling the perovskite octahedral rotations and oxygen octahedral coupling. The substrate exerts strain on the film which induces octahedral rotations in the film, but there is also coupling of the film octahedra to substrate octahedra. The question is whether similar interface coupling effects could be created in VAN films but without the interference from the substrate as is the case of SL films. In fact, in VAN the substrate influence is weak above around 20 nm film thickness and since micron thick films are relatively easy to grow and indeed preferred for VAN, any coupling effects across the vertical interfaces in VAN are largely independent of the substrate. On the other hand, as shown in Table 1, the challenge for VAN is creating highly faceted interfaces which are also perpendicular to the substrate without meandering. As already noted, this is highly possible when the matrix material is the same as the same or a very closely lattice and structurally matched material as the substrate material, *e.g.* SrTiO_3 -based VAN film on SrTiO_3 substrate,⁵⁴ and hence there is wide scope for exploring new interface coupling effects in VAN films with the most ideal interfaces.

3.2 Magnetic interactions across interfaces involving in 5d TMO superlattice films

SLs containing perovskites with strong spin-orbit coupling have been investigated to further explore new interface coupled effects and exotic magnetic reconstructions. For example, in $\text{SrMnO}_3/\text{SrIrO}_3$ superlattices,⁷⁸ where SrMnO_3 is the 3d anti-ferromagnetic insulator and SrIrO_3 is a 5d paramagnetic metal,



there is charge transfer of ~ 0.5 electrons per unit cell from SrIrO_3 to SrMnO_3 . Hence, there is induced ferromagnetism in the SrMnO_3 which is normally antiferromagnetic.⁷⁹ In addition, the strong spin-orbit coupling also leads to perpendicular magnetic anisotropy (PMA) and a large anomalous Hall effect. Furthermore, in $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3/\text{SrIrO}_3$ (ferromagnetic metal/paramagnetic metal) SLs,⁶⁵ magnetic coupling across the interfaces produces an emergent weak ferromagnetism in the SrIrO_3 layers. Also, since magnetocrystalline anisotropy is closely linked to spin-orbit coupling, the in-plane magnetic anisotropy of $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ was changed by varying the thickness of the SrIrO_3 (Fig. 3e). A PMA constant (K_{eff}) of $4 \times 10^6 \text{ erg cm}^{-3}$ was further optimised by increasing the doping level, x , in the $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3/\text{SrIrO}_3$ SLs.³⁵

Broken inversion symmetry and strong spin-orbit coupling in 3d-5d interfaces can also lead to large chiral Dzyaloshinskii-Moriya interactions (DMIs), inducing non-trivial spin textures, *i.e.* topological phenomena. In $\text{LaMnO}_3/\text{SrIrO}_3$ SLs (Fig. 3f), a highly robust chiral magnetic phase with large topological Hall effect (THE) of $75 \mu\Omega \text{ cm}$ was observed.³⁷ The present research on the interfaces with iridate oxide systems is still in infancy, and many open and intriguing questions still remain to be explored.

Finally, we note that the skyrmion structures formed in SrIrO_3 superlattices incorporate insulating or metallic layers are formed in films of only 2–8 unit cells (or 1–4 nm) thickness.^{37,80} Thus, as mentioned for the polar skyrmions in Section 2.4, it is highly possible to grow VAN structures of this dimension and potentially grow, by self-assembly, skyrmion SrIrO_3 columns in a film matrix, the route for growing perovskite nanocolumns in VAN films having already been demonstrated in several systems, *e.g.* BiFeO_3 , BaTiO_3 , and $\text{Ba}_{1-x}\text{Sr}_x\text{TiO}_3$ systems.²⁵ Indeed, it is possible that SrIrO_3 would grow more readily in VAN films compared to plain epitaxial films. This is because the phase could be stabilised using vertical epitaxy with a matrix phase in the VAN film.^{19b} Also, to maintain stoichiometry of the SrIrO_3 , a matrix material containing either or both Sr and Ir species could be used so as to produce a source material to prevent loss of these volatile cations during growth.

4. Multiferroism and magnetoelectricity in superlattice films

Materials that exhibit simultaneous order in their electric and magnetic ground states are promising for enabling electric field control of magnetism *via* the CME effect. Such materials are rare, however, due to the contradicting requirements of empty d orbitals *via* cation off-centering required for ferroelectricity and partially filled d orbitals for magnetism.⁸¹ Creating composite systems where the properties of two different systems are coupled across their interface is a potential way to create artificial multiferroics. Oxide SLs are one form of composite system. As we discuss below, when two systems are coupled *via* an interface in SLs comprising magnetic and ferroelectric

layers, proximity effects, broken inversion symmetry, strain and defects, and geometric frustration and strain engineering can give rise to either multiferroism (at least two ferroic properties in *one* materials systems) and/or artificial magnetoelectricity (electric field control of magnetism or *vice versa*). We note that while there have been many predictions of the aforementioned effects in a range of SLs⁸² not many have been experimentally verified. On the other hand, there have been interesting experimental observations in SLs not predicted by theory, typically using combinations of a perovskite which exhibits ferroelectricity, with another which exhibits some form of magnetic order.

Two key examples of multiferroism induced in SLs are in $\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3/\text{BaTiO}_3$ (non-polar-tilt ferromagnet/polar ferroelectric) and $\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3/\text{BiFeO}_3$ (non-polar-tilt ferromagnet/ferroelectric antiferromagnet). In the first case (Fig. 4a), due to proximity effects a structural phase transition from non-polar to unconventional polar (hence giving both ferroelectricity and ferromagnetism, thus multiferroicity) was induced in ultrathin 4-unit-cell $\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3$.⁸³ In the second case, magnetism (up to 200 K) was induced in ultrathin 5-unit-cell ferroelectric BiFeO_3 ,⁸⁷ to a level of $1.83 \pm 0.16 \mu_B$ per Fe, increased from $0.03 \mu_B$ per Fe in bulk BiFeO_3 , and attributed to strong orbital reconstruction between Fe and Mn across the interfaces.

The creation of ferroelectricity, multiferroicity, and a DME effect has been achieved through simultaneous breaking of time-reversal and spatial-inversion symmetries in “tricolour” ABCABC SLs composed of three sequential kinds of perovskite layers⁸⁸ (top plot in Fig. 4b). The polarization at the AB (or BC) interfaces has the same sign leading a non-vanishing toroidal moment equivalent to a built-in vector potential under the presence of spin-orbit coupling, thereby producing ME coupling. For example, a ferroelectric T_C of 40 K was achieved in $\text{NdMnO}_3/\text{SrMnO}_3/\text{LaMnO}_3$ SLs, even though individual layers have no polar moment. Also, ME coupling was achieved by magnetic modulation of the polarization (bottom plot in Fig. 4b).⁴¹ Density functional calculations show that the combination of broken space inversion symmetry and the presence of $\text{Mn}^{3+}/\text{Mn}^{4+}$ mixed valence, arising from cationic asymmetry and interfacial polar discontinuity, produce emergent ferroelectric and ME coupling. However, the interface-induced ME coupling is very weak and occurs below 55 K.

Other works on creating magnetoelectricity in SLs have centred around a wide range of SL combinations, *e.g.* in $\text{BaTiO}_3/\text{BiFeO}_3$ (ferroelectric/ferroelectric and antiferromagnet) SLs a direct ME coefficient α_{ME} up to $55 \text{ V cm}^{-1} \text{ Oe}^{-1}$ at 300 K was measured,^{38b} higher than in single-phase BiFeO_3 ($\alpha_{\text{ME}} = 4.2 \text{ V cm}^{-1} \text{ Oe}^{-1}$), which was explained by the diffusion of Ba and Ti into the BiFeO_3 layers. Hence, the influence of unwanted chemical intermixing across the interface and impurities buried within superlattices is uncertain.

Another different mechanism to achieve multiferroicity and ME coupling was demonstrated using symmetry breaking across a SL interface of $(\text{LuFeO}_3)_m/(\text{LuFe}_2\text{O}_4)_1$. Rumpling (periodic buckling of lattice planes) in LuFeO_3 ⁸⁹ imposed a local distortion in one unit cell LuFe_2O_4 , thus removing the mirror



Multiferroic phenomena in SL films

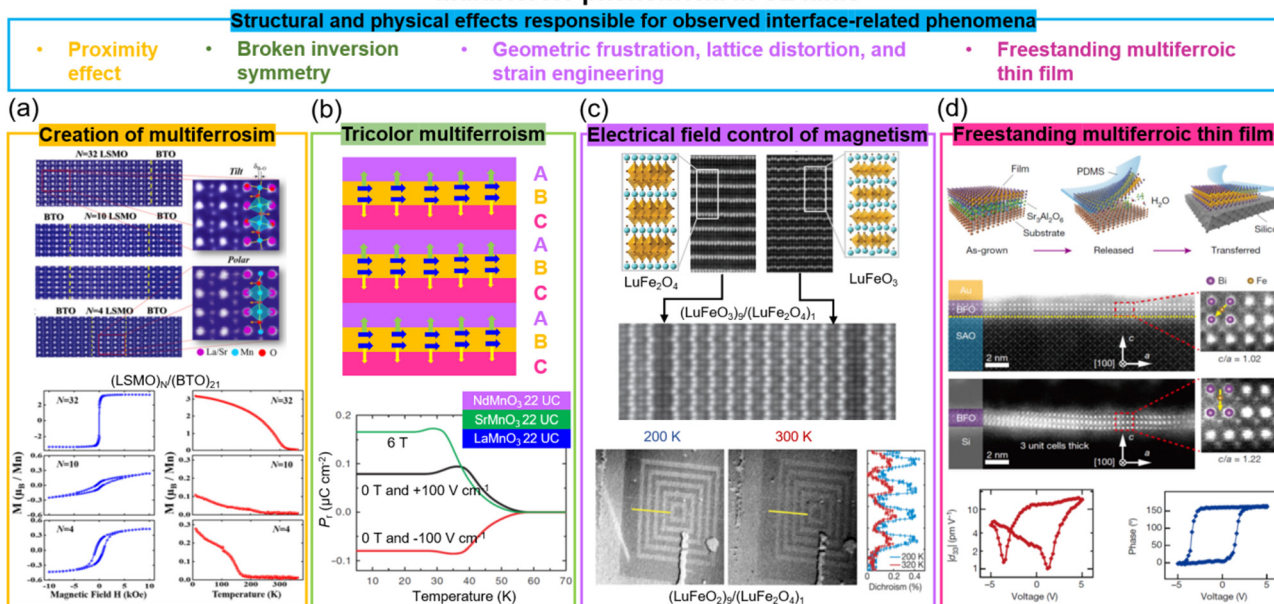


Fig. 4 Multiferroic phenomena in SL films. (a) Top panel: STEM images of $(\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3)_n/(\text{BaTiO}_3)_{21}$ superlattices. Bottom panel: M vs. H loops (left) and M vs. T curves (right) for $(\text{La}_{2/3}\text{Sr}_{1/3}\text{MnO}_3)_n/(\text{BaTiO}_3)_{21}$ superlattices. Reproduced with permission.⁸³ Copyright 2017 PNAS. (b) Top panel: A tricolor superlattice for designing the multiferroicity. Reproduced with permission.⁸⁴ Copyright 2007 Elsevier. Bottom panel: Temperature dependent polarization measured under electrical and magnetic field for $\text{NdMnO}_3/\text{SrMnO}_3/\text{LaMnO}_3$ superlattices. Reproduced with permission.⁴¹ Copyright 2012 Springer Nature. (c) Top panel: STEM images of end-members LuFe_2O_4 (left) and LuFeO_3 (right); middle panel: $(\text{LuFeO}_3)_9/(\text{LuFe}_2\text{O}_4)_1$ superlattice. Bottom panel: XCMD PEEM images of $(\text{LuFeO}_3)_9/(\text{LuFe}_2\text{O}_4)_1$ superlattice measured at 200 K and 320 K. Reproduced with permission.⁸⁵ Copyright 2016 Springer Nature. (d) Top panel: Schematic of growth and transfer of a freestanding thin films. Middle panel: STEM images of 3-unit-cell BiFeO_3 film before and after etching $\text{Sr}_3\text{Al}_2\text{O}_6$ layer. Bottom panel: Piezoelectric properties of 4-unit-cell BiFeO_3 film after etching $\text{Sr}_3\text{Al}_2\text{O}_6$ layer. Reproduced with permission.⁸⁶ Copyright 2019 Springer Nature.

symmetry in the LuFe_2O_4 and enabling ferroelectricity to emerge in this ferrimagnetic phase⁸⁵ (Fig. 4c). The ferrimagnetic T_C of the LuFe_2O_4 was increased from 240 K to 281 K because of a local distortion applied by LuFeO_3 . Since the ferroelectric order couples to the ferrimagnetism, electrical field control of magnetism, as measured using X-ray magnetic circular dichroic photoemission electron microscopy (XCMD PEEM), was achieved at 200 K (Fig. 4c). The clamping of the films to the substrate or leakage effects possibly limit macroscopic measurements using electrodes applied across the whole film.

Finally, we note that there has been more success in achieving direct ME (DME) in all-metallic systems⁹⁰ than in oxides. An example is $(\text{Ta}/\text{Cu}/\text{IrMn}/\text{FeCoSi})_n/\text{AlN}/\text{Si}$,^{90c} where a relatively large direct ME effect of $96 \text{ V cm}^{-1} \text{ Oe}^{-1}$ was observed at 300 K. However, the requirement to use precious metal Ir, the need to use a relatively complex stack of layers, and no observation of CME, are limiting factors for this system. Hence, arguably, there is more promise for oxide systems, although controlled interface composition and structure, and the elimination of substrate clamping effects are important areas of future study. The latter effect prevents intrinsic displacement of ferroelectric and magnetostrictive materials.

Very promisingly, in recent years much progress has been made with lift-off and transfer of films to practical substrates.^{86,91} As far as known, the transfer process has not

been applied to multiferroic SL films but it has been used for single layer films and it is expected that multiferroic SL films would behave similarly. The advantage of lifting off a multiferroic SL film is that it would be free from substrate clamping. This could enable strain coupled magnetoelectricity to be realised. Fig. 4d shows an example of using water-soluble $\text{Sr}_3\text{Al}_2\text{O}_6$ as the sacrificial buffer layer to produce freestanding BiFeO_3 multiferroic thin films with a high crystalline quality that can be fabricated down to one unit cell. Also, robust and switchable ferroelectric polarization was demonstrated for a four unit-cell, transferred BiFeO_3 film on Si, as confirmed by STEM and piezoelectric measurements, respectively. We note that VAN films have also been successfully lifted-off and transferred to Si. Owing the nanoscale nature of VAN films, they appear to be more resistant to crack propagation yielding a higher success rate of film transfer compared to plain films.⁹² More work is needed to explore the performance of lifted-off ferroic and multiferroic VAN films.

Contrasting with much of the SL work describe above, artificial multiferroic effects in VAN films have not focused on purely interfacial phenomena, but on more macroscopic effects caused by strain coupling of phenomena across vertical interfaces which take effect in the whole body of film. An advantage of VAN films over SL is the lack of a substrate clamping effect. On the other hand, leakage currents can be problematic if the right combinations of materials are not used.



Strategies for overcoming the leakage while also boosting interface coupling are discussed for VAN multiferroics in Section 8.

5. Summary of ferroic phenomena in superlattice films

Overall, artificial SLs of complex oxides for ferroics are systems which provide an excellent platform for fundamental research, allowing the validity of theoretical predictions to be probed, new science to be revealed, and the realization of new materials with tailor-made properties. On the other hand, SLs also have challenges as interfaces can be subject to intermixing effects and defect formation which are hard to probe and identify. Promisingly, improvements in tools to understand interfaces are improving rapidly, there are also continuous advances in growth tools. Hence, SL films continue to have huge promise for next generation electronics.

B. Vertically aligned nanocomposite (VAN) films

6. Ferroelectric phenomena in VAN films

In SL films (discussed in part A above), it has been demonstrated that ferroelectricity can be induced from interfacing non-FE materials, and also that fascinating effects such as negative capacitance and polar vortices can be created. As already noted, such phenomena have not yet been observed in VAN films (Table 2). On the other hand, a number of other structural effects (Table 1) give rise to fascinating phenomena in VAN films. In particular, novel strain effects play a strong role in determining functional properties.

We recall that in standard planar SL ferroelectric films, epitaxial strain from the underlying substrate is often used to tune properties.^{16,24,93} However, this substrate-induced interfacial strain can only extend to tens of nanometers thickness and hence the T_C enhancement is largely confined to the interfacial region. It has been shown in epitaxial perovskites that from above the first tens of nanometers thickness up to ~ 100 nm, the strain is relieved by a range of crystalline defects to compensate the strain.⁹⁴ Such defects can degrade the functionality of epitaxial perovskite devices which exploit the functionality of the whole film rather than just the interface, e.g. a strain-enhanced- T_C ferroelectric actuator. In VAN films, on the other hand, by incorporating stiff nanopillars into a ferroelectric matrix large vertical strains (of $>1\%$) can be induced throughout the whole film and hence the thickness-limiting substrate strain effect is not relied upon.²¹ This enables wide-ranging ferroelectric property tuning in micron thick films. Since the lateral distance between the nanopillars is typically only ~ 20 nm, the strain is not relaxed to a large extent as for epitaxial films, and so the film perfection and

crystallinity remains high for 100s nm thick VAN films.^{13,95} Hence, enhanced T_C , remnant polarization, retention, dielectric tunability, and reduced loss tangent and leakage current are achieved.^{25,27,54,96}

In section A, we presented several examples of VAN compositions that could be explored to try and mimic and also new opportunities the exciting emergent effects achieved in SL films. In the following sections we focus on VAN films. Before, we go into details of each of the enhanced properties, explaining how the strain in VAN films contributes to these properties, we first consider how research on SLs could benefit from VAN research. A key processing advantage of the VAN approach over SL is it is relatively fast and simple for exploring new combinations of materials and their interfaces. Hence, VAN can serve as fast screening method to predict which phases and which ratios of phases could also give rise to novel interface related functionalities in SL structures. For example, in Section 8, we show how a triple VAN system enables a self-biased magnetoelectric to be achieved. For SL films, it would take longer to explore the right thicknesses of the layers to use, to be sure there are no interface mixing effects, and also to know if the magnetoelectric effect were present (as it could be obscured by substrate clamping effects). Hence, with the knowledge gained from VAN, SL films of the demonstrated compositions could now be tested, and films lifted off to remove any clamping effects.

6.1 Enhancement of ferroelectric T_C , remanent polarization, P_r , piezoelectric d_{33} and retention in VAN films

The first demonstration of enhanced ferroelectricity using VAN was in ~ 2 micrometre-thick BaTiO₃-based VAN films embedded with stiff Sm₂O₃ pillars yielding a $>2\%$ vertical extension in the BaTiO₃. While many BaTiO₃ applications are limited by its relatively low T_C (~ 396 K),⁹⁸ here in the VAN films a T_C up to ~ 970 K (higher also than the 638 K achieved in BaTiO₃/SrTiO₃ SL⁹⁹) can be achieved which opens up many more applications possibilities.²¹ Other BaTiO₃-based and SrTiO₃-based VAN films have also shown strong enhancements with T_C enhanced (Fig. 5a) and leakage current reduced.^{54,96c,100} We note that reduced leakage is not universal for VAN films but depends on whether the interface is in tension or compression,^{72b} the nature of the interface structure(s) that form in the VAN film to minimise interfacial energy,^{73b} the nature of the transition metal cations and their redox equilibria, and whether a rectifying p-n junction forms between the film and substrate.⁴⁰

As shown in Table 2, similar ferroelectric property enhancements are only achievable in planar or SL films if they are ~ 10 's nm thick or less.^{93a} Such high T_C 's in micron thick films are necessary for advanced device applications, such as the electrocaloric devices,¹⁰¹ actuators,¹⁰² and microwave tuneable filters,^{53a} and cannot be achieved in planar films.

The remnant polarization (P_r) has also been systematically increased using (Ba,Sr)TiO₃-based VAN structures (Fig. 5b), both by increasing the fraction, x , of nanopillars and increasing film thickness (to $\sim \mu\text{m}$'s),²⁵ both of which increase the film



Ferroelectric Properties in VAN Films

Structural and physical effects responsible for observed interface-related phenomena

- Large Vertical Strain
- Uniform strain throughout film and reduction of substrate clamping
- Reduced defects and improved crystalline structure
- Multiphase boundaries (MPB)
- Large amount of vertical interfaces



Fig. 5 Enhanced ferroelectric properties achieved using VAN structures. (a) Cross-sectional TEM image of SrTiO₃–Sm₂O₃ VAN; (b) remnant polarization of (Ba,Sr)TiO₃–Sm₂O₃ VAN film as a function of the tetragonality of (Ba,Sr)TiO₃ phase. Reproduced with permission.²⁵ Copyright 2017 Wiley-VCH. (c) Giant piezoelectric Na-deficient NaNbO₃–NaNbO₃ (NPR–NNO) VAN film. Reproduced with permission.⁹⁷ Copyright 2020 AAAS. (d) Comparison of normalized retained polarization versus temperature for a pure (Ba,Sr)TiO₃ film and for (Ba,Sr)TiO₃–Sm₂O₃ VAN films with different Sm₂O₃ contents (x). Reproduced with permission.²⁵ Copyright 2017 Wiley-VCH. (e) The evolution of polarisation retention in a FE domain of a BiFeO₃ (111) mesocrystal with time. Reproduced with permission.^{96b} Copyright 2016 Springer Nature. (f) High tunable relative permittivity, ϵ_r , as shown by the large change in capacitance value with electric field in a (Ba,Sr)TiO₃–Sm₂O₃ VAN film, measured at 1 MHz. An enhanced commutation quality factor (CQF) is also achieved. Reproduced with permission.²⁷ Copyright 2012 American Chemical Society. (g) Reduced leakage current of a BiFeO₃–Sm₂O₃ VAN film compared to plain BiFeO₃ or plain Sm₂O₃ films. Reproduced with permission.^{96a} Copyright 2008 AIP Publishing.

tetragonality. The P_r values are about $\sim 3\times$ larger compared to (Ba,Sr)TiO₃ planar films (Table 1). A near doubling of P_r has also been shown in BaTiO₃–NiO VAN films.¹⁰³ A giant enhancement of piezoelectric response is also possible in VAN films (Fig. 5c). Liu *et al.*⁹⁷ showed in nanopillar-like sodium-deficient NaNbO₃ VAN a piezoelectric coefficient d_{33} of $\sim 1098 \text{ pm V}^{-1}$, which compares to $\sim 22 \text{ pm V}^{-1}$ for stoichiometric NaNbO₃ films. The T_C of the VAN film is $\sim 723 \text{ K}$, which is similar to that of the stoichiometric NaNbO₃ film. Local heterogeneity of

sodium-deficient NaNbO₃ and stoichiometric NaNbO₃ phases was deemed to be responsible.

Temperature-dependent and time-dependent changes of the polarization state of ferroelectric films, *i.e.* thermal and temporal retention, are of great importance for non-volatile nanoelectronic devices based on ferroelectrics,^{25,96b,104} *e.g.* for above-room temperature-operation, high-stability memory devices. The retention parameters are sensitive to film thickness and microstructure, electrodes, *etc.*¹⁰⁵ Much enhanced thermal and



temporal polarization retention has been demonstrated in (Ba,Sr)TiO₃-Sm₂O₃ and CoFe₂O₄-BiFeO₃ VAN films (Fig. 5d and e) compared to the pure film counterparts of (Ba,Sr)TiO₃ and BiFeO₃, respectively. Ultra-dense vertical interfaces lead to a strong tetragonal distortion and vertical ferroelectric domains^{25,96b} which are 'pinned', thus retarding relaxation processes.

6.2 Dielectric property enhancements and reduced leakage in VAN films

Stemming from the increased tetragonality induced in the ferroelectric phase, dielectric properties are enhanced in VAN films.²⁷ In (Ba,Sr)TiO₃-based VAN films, a tunability of 75% (200 kV cm⁻¹ field), and a low dielectric loss of 0.01 was achieved at room temperature, in contrast to the respective values of ~52% and 0.06 in pure (Ba,Sr)TiO₃ films²⁷ (Fig. 5f). Critically, the loss is reduced when the tunability is enhanced which is opposite to planar films and results from the unusual strain states in VAN films.¹³ The combination of high tunability and low loss lead to a high communication quality factor (CQF) value. In SrTiO₃-based VAN films, a record communication quality factor (CQF) of up to 3300 was achieved which gives the system strong potential for tuneable RF applications.⁵⁴

Leakage is a long-standing problem in ferroelectric oxide thin films, commonly caused by oxygen deficiency and the presence of point or line defects. As already mentioned above, leakage can be reduced using VAN films. Taking the example of BiFeO₃, a promising candidate for multifunctional non-volatile memory devices because of the coexistence of both the anti-ferromagnetic order with Néel temperature (T_N) of 643 K¹⁰⁶ and ferroelectric order with T_C of 1123 K,¹⁰⁷ leakage from a range of defects is common and can give either p-type or n-type behaviour.¹⁰⁸ In BiFeO₃-based VAN films, the leakage was reduced by several orders (Fig. 5g).^{72a,96a} This was ascribed to the compressive out-of-plane (OOP) strain of -1.46% that was induced in the BiFeO₃, induced by the embedded nanopillars. It was postulated that such a compression would reduce the oxygen vacancy concentration and hence carriers associated with these vacancies.¹⁰⁹ However, some chemical intermixing, leading to doping of Sm into the BiFeO₃ could also contribute to the lowering of the leakage. Of course, not all VAN films can have judicious compression engineered into them as for this BiFeO₃ example.

6.3 Exploiting both bulk and surface effects for property enhancements using VAN

One key advantage of VAN films over SL or standard films is the ability to create very high surface area structures from them, and at the same time to retain strong ferroic properties even after treating them by chemical etching. Such etching removes the pillar phase in the VAN rendering a mesoporous film. There are many potential applications which could benefit from mesoporous films with tubular pores from the etched nanopillars, having high surface area for chemical, gas, or biological sensing, or even filtration (especially because the pore size is tunable *via* growth parameters and VAN film composition).

Adding in ferroicity to the mesoporous film could enable greater sensitivity or tunability of the device performance/operation.

One example of the benefits of mesoporous VAN is the doped/undoped SrTiO₃-MgO VAN film system. Here, the MgO is chemically etched out of the film, leaving a mesoporous SrTiO₃ matrix (Fig. 6a) with the out-of-plane tensile strain in the SrTiO₃ matrix largely intact. More understanding of the mechanisms and universality of matrix strain retention after etching pillars out of the VAN films is required. This mesoporous matrix leads to photoelectrochemical (PEC) water splitting compared to literature (Fig. 6b and c),¹¹⁰ arising from (a) an order of magnitude larger surface area for light absorption, (b) optimised carrier concentration from Ag and Nb of the doped SrTiO₃,¹¹¹ and (c) above-room-temperature ferroelectricity from the vertical strain effect induced by the VAN structure, which produced a reduced flat-band potential and enhanced charge extraction (Fig. 6d-f)^{110,111}

Another area which has been unexplored so far is the filling of epitaxial mesoporous films with different phases, *e.g.* using atomic layer deposition or solution processing. This can produce new composites with metallic or polymeric phases which can couple to the mesoporous matrix either to achieve multifunctional properties, or to simply exploit large surface areas and reduced dimensionality (important for strong light absorption and high charge extraction in optoelectronic systems, for example). These areas remain largely unexplored but there is strong potential for new multifunctional systems to emerge.

7. Magnetic phenomena in VAN films

While novel magnetic phenomena involving a range of interactions have been widely explored at interfaces in SL films, direct interactions at interfaces in VAN films has been limited to exchange bias coupling.¹¹² However, a wide range of other interactions may be present at VAN interfaces, just as at SL interfaces, *e.g.* charge transfer, orbital reconstructions, and magnetic exchange interactions. However, so far, they remain unexplored. On the one hand, while non-planar interfaces in VAN might frustrate some of these interactions, on the other VAN films have perfectly clean interfaces and this is beneficial in terms of not degrading the interactions.

A key benefit of VAN over SL resides in the potential for high density magnetic memory and storage applications. Radically new approaches are required to increase magnetic recording densities beyond 1 Tbit in⁻² and to lower energy consumption.¹¹³ While self-assembled magnetic nanoparticles, such as the self-assembled FePt nanoparticle arrays shows a high area density up to 50 Tbit in⁻² for <4 nm particle sizes,¹¹⁴ difficulties exist for anchoring and ordering these nanoparticles. Self-assembled VAN films, with ~Tbit in⁻² density (~10 nm sized features with ~10 nm spacing) represent a promising approach. The vertical alignment of magnetic nanocolumns anchored in a non-magnetic matrix gives exchange decoupled, well-separated perpendicular magnetic recording



Photoelectrochemical properties of VAN films

Structural and physical effects responsible for observed interface-related phenomena

- Ability to make large surface area mesoporous film
- Large vertical strain



Fig. 6 Doped/undoped SrTiO₃-based VAN porous ferroelectric films for water splitting applications. (a) SEM image of the surface of an undoped mesoporous SrTiO₃ film. Reproduced with permission.¹¹⁰ Copyright 2016 American Chemical Society. (b) Schematic of the water-splitting by the undoped mesoporous SrTiO₃ film. Reproduced with permission.¹¹⁰ Copyright 2016 American Chemical Society. (c) Enhanced incident photon to current efficiency (IPCE) in the undoped mesoporous SrTiO₃ film compared to both a continuous SrTiO₃ film and the Nb-doped SrTiO₃ substrate. Reproduced with permission.¹¹⁰ Copyright 2016 American Chemical Society. (d) The photoelectrochemical (PEC) response of a mesoporous (Ag,Nb)-co-doped SrTiO₃ film at different polarization states. Reproduced with permission.¹¹¹ Copyright 2019 American Chemical Society. (e and f) Schematic of the modified band alignment at the interfaces in the mesoporous (Ag, Nb)-co-doped SrTiO₃ film and electrolyte solution with different polarization states and the impact on the PEC. Reproduced with permission.¹¹¹ Copyright 2019 American Chemical Society.

bits. A range of other enhanced magnetic properties in VAN films also have potential benefit for magnetic storage/memory applications, as outlined below.

7.1 Enhanced magnetization in VAN films

Magnetization can be enhanced significantly in VAN films. In BiFeO₃ (90%)–Fe₃O₄ (10%) nanocomposite films, enhanced saturation magnetization values of 900 emu cm⁻³ were obtained compared to 300 emu cm⁻³ for pure Fe₃O₄ films. The result was explained by a highly strained (OOP, 7%), ultra-thin Fe₃O₄ phase in which ferromagnetism rather than ferrimagnetism was favoured.³³

Similar magnetisation enhancement was found in Pr_{0.5}Ba_{0.5}MnO₃–CeO₂ VAN films¹¹⁵ due to the enhanced tetragonality in Pr_{0.5}Ba_{0.5}MnO₃ from the stiff CeO₂ nanopillars, although other structural effects could be at play. However, in La_{0.7}Sr_{0.3}MnO₃–MgO VAN films, the magnetization was suppressed, possibly because of octahedral tilting/rotations.¹¹⁶

Antiferromagnetic-to-ferromagnetic property switching has also been demonstrated in different VAN systems, *e.g.* in spinel ZnFe₂O₄ based VAN films,³⁴ a *T_C* of ~ 500 K was achieved with 4.8μ_B per Fe moment, and moderate coercivity (Fig. 7a (bottom panel)), overwhelmingly larger than the *T_N* (~ 10 K) of bulk ZnFe₂O₄ (Fig. 7a) (top panel). The VAN films induced stronger intersite A–B exchange interactions of the Fe³⁺ cations than the intrasite interactions at the A- or B-sites. Mössbauer spectra (Fig. 7b) showed that half of the Fe³⁺ ions go into the A sites in the ZnFe₂O₄ phase. This is similar to the case for nanocrystalline powder form of ZnFe₂O₄.¹¹⁷

Similar antiferromagnetic-to-ferromagnetic property switching has also been observed in other VAN films, *e.g.* SmMnO₃–Bi₂O₃,^{96d} (Sr,Sm)MnO₃–Sm₂O₃,¹¹⁸ and EuTiO₃–Eu₂O₃.¹¹⁹ In all cases, large vertical strains were induced, *e.g.* in the latter case the value was ~ 3.15% in the EuTiO₃ phase. The strain decreases the Eu–Ti–Eu bond angle along <111> up to 1°, leading to a weakening of the antiferromagnetic interactions



Magnetic properties in VAN films

Structural and physical effects responsible for observed interface-related phenomena

- High density of vertical interfaces
- Highly epitaxial growth
- Large aspect ratio pillars
- Well-separated/decoupled pillars

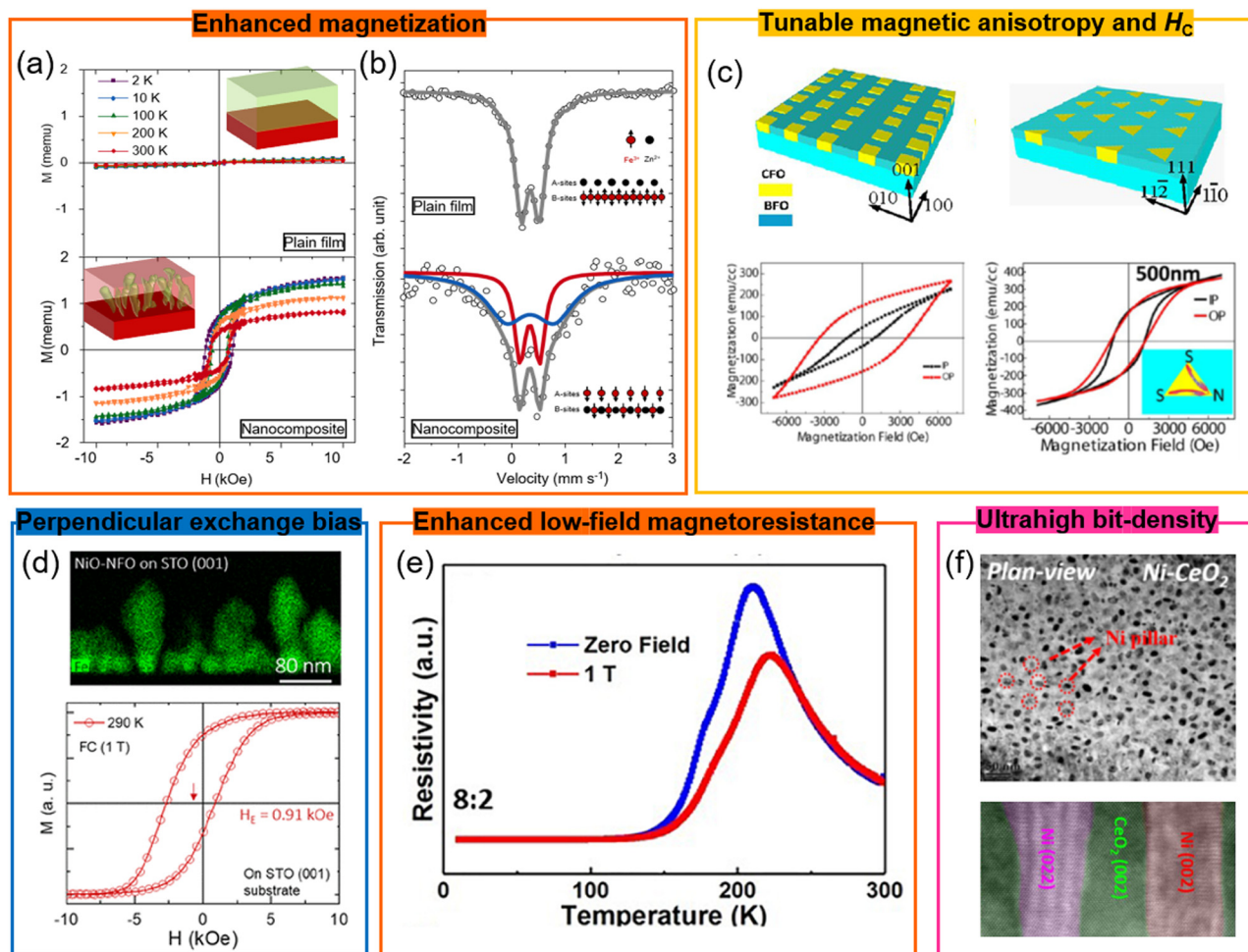


Fig. 7 Enhanced magnetic properties achieved using VAN structures. (a) The magnetic hysteresis loops of a ZnFe₂O₄ plain film and a ZnFe₂O₄-SrTiO₃ VAN film grown on a SrTiO₃ (001) substrate, measured at different temperatures. The magnetic field was applied in the OOP direction. Reproduced with permission.³⁴ Copyright 2018 Wiley-VCH. The insets show schematics of a plain ZnFe₂O₄ film and a ZnFe₂O₄-SrTiO₃ VAN film, respectively. (b) Mössbauer spectra of the Fe element in the ZnFe₂O₄ plain film and the ZnFe₂O₄-SrTiO₃ VAN film. The insets show schematics of the magnetic structures of the ZnFe₂O₄ phases. Reproduced with permission.³⁴ Copyright 2018 Wiley-VCH. (c) Tunable magnetic anisotropy and coercivity in BiFeO₃-CoFe₂O₄ VAN films grown on SrTiO₃ substrates with magnetic field applied either in-plane (IP) or OOP. Reproduced with permission.¹²⁰ Copyright 2013 American Chemical Society. (d) Enhanced perpendicular exchange bias effect in a NiO-NiFe₂O₄ VAN film with magnetic field applied OOP. Reproduced with permission.²² Copyright 2018 American Chemical Society. (e) Temperature dependence of the resistivity measured at different magnetic fields applied OOP. Reproduced with permission.¹²¹ Copyright 2019 AIP Publishing. (f) Plan-view (top panel) and cross-sectional (bottom panel) high resolution TEM images of a CeO₂-Ni VAN film. Reproduced with permission.¹²² Copyright 2016 Royal Society of Chemistry.

and switching from antiferromagnetic to ferromagnetic behaviour.

7.2 Highly tunable magnetic anisotropy and coercivity in VAN films

The magnetic anisotropy defines the direction of the magnetization in magnetic materials. In continuous magnetic thin films, the anisotropy is usually dominated by shape anisotropy,

which makes the in-plane (IP) direction the easy plane for the magnetization to minimize the demagnetization energy. However, perpendicular magnetic anisotropy (PMA) is required to achieve ultra-high-density data recording.¹²³ In traditional media, PMA is obtained in the ultrathin films and films with noble metals or rare-earth elements with strong spin-orbit coupling, such as Co/Pt multilayers¹²⁴ and DyCo₅,¹²⁵ *etc.* In magnetic oxide films, PMA is typically generated using epitaxial



strain-induced magnetostrictive anisotropy¹²⁶ or it is driven by symmetry mismatch¹²⁷ at the interfaces which leads to distinct orbital reconstructions. However, due to the interfacial nature of PMA, in standard planar films (including SL), the effect degrades with film thickness above a few nm to tens of nm depending on the mechanism for the PMA.

In VAN films the total anisotropy is controlled by three kinds of magnetic anisotropy, *i.e.* magnetocrystalline, shape and magnetostrictive. While magnetocrystalline anisotropy is an inherent property of the material, shape and magnetostrictive anisotropy are highly tunable in VAN structures. Because of the large aspect ratio of the magnetic nanopillars which grow in VAN films, a large shape anisotropy is produced with an easy axis along the OOP direction, producing the PMA. A shape anisotropy field $H_{\text{shape}} = 2.0$ kOe was estimated in the BaTiO₃-CoFe₂O₄ VAN films, where the aspect ratio of magnetic pillars was around 10.³⁶ The shape anisotropy is controlled by substrate orientation (as well as other growth parameters). The substrate orientation effect on the magnetic anisotropy and coercivity has been demonstrated for CoFe₂O₄ pillars in BiFeO₃-CoFe₂O₄ VAN films (Fig. 7c) as well as also for other VAN systems.^{120,128} Since BiFeO₃ is an antiferromagnet, it has been proposed that spin-flop exchange coupling between the compensated BiFeO₃ and the CoFe₂O₄ spins at the interfaces might contribute to the observed PMA in BiFeO₃-CoFe₂O₄ VAN films.¹²⁹

Strain-induced stress anisotropy in VAN films also plays a significant role *via* the magnetoelastic effect. Here, the magnetoelastic energy depends on both the magnetostriction coefficient of the magnetic material as well the strain in it. CoFe₂O₄ has a large magnetostrictive coefficient with a large strain-induced magnetoelastic effect. In BaTiO₃-CoFe₂O₄ VAN films, a compressive strain in the CoFe₂O₄ columns of $\sim 0.8\%$ is achieved, with corresponding anisotropy field of $H_{\text{stress}} = 34$ kOe, matching the coercivity of the measured magnetic hysteresis.¹³⁰ The magnetostrictive and magnetocrystalline anisotropies have been systematically tuned by varying the chemical composition of the Co_xNi_{1-x}Fe₂O₄ column phases in the Co_xNi_{1-x}Fe₂O₄-BiFeO₃ VAN system.¹³¹ Large tunability of magnetic anisotropy and coercivity has also been achieved in La_{0.7}Sr_{0.3}MnO₃-MgO VAN, where a switch from the easy-plane magnetic anisotropy to PMA occurs as the volume ratio of MgO increases, thus leading to different strain values in the La_{0.7}Sr_{0.3}MnO₃ matrix.^{60a}

Potential systems for high-density perpendicular recording media have also been shown in ferromagnetic metal pillar-oxide VAN systems.¹³² Favourable properties such as ultra-high bit densities of $\sim \text{Tb in}^{-2}$ (Fig. 7f shows an example for a CeO₂-Ni film),¹³³ uniaxial anisotropy, strong OOP magnetization, high coercivities, anisotropic electric conductivity,^{132b} and higher T_{C} s and lower intrinsic resistances than ferromagnetic oxides^{133,134} have all been demonstrated.

7.3 High perpendicular exchange bias in VAN films

Perpendicular exchange bias (PEB) is appealing for applications in perpendicular magnetic tunneling junctions (MTJs).

Owing to PMA in VAN systems, PEB is also expected. Another advantage of VAN compared to traditional FM/AFM bilayers or SLs, where a $H_{\text{E}} \sim 1/t_{\text{F}}$ law (where H_{E} and t_{F} are the EB field and film thickness) is obeyed, is that exchange bias (EB) scales with t_{F} , because the coupled vertical interfacial area scales with t_{F} .

The first trial PEB system was a (BiFeO₃)_{0.9}-(Fe₃O₄)_{0.1} composite film (Fe₃O₄ nanoparticles in a BiFeO₃ film). A PEB field of $H_{\text{E}} \sim 0.04$ kOe was obtained at room temperature.^{112b} Later, larger PEB values were obtained in LaFeO₃-La_{0.7}Sr_{0.3}MnO₃ and BiFeO₃-La_{0.7}Sr_{0.3}MnO₃ VAN systems with EB fields up to ~ 1 kOe,^{112c,d} although this was achieved at a relatively low temperatures (< 10 K). Most recently, a large PEB of ~ 0.91 kOe at room temperature was obtained in the NiO-NiFe₂O₄ VAN system (Fig. 7d).²² With the benefits of both high T_{C} and high T_{N} of the FM and AF materials as well as $T_{\text{C}} > T_{\text{N}}$, the blocking temperature (T_{B}) in this system reached above 400 K.

8. Multiferroics and the magnetoelectric effect in VAN films

The key aims for non-volatile memory devices are large ME coupling coefficient at room temperature, and self-biased characteristics (electric field control of magnetism and no applied magnetic field). While, there are many reviews on both SL and VAN for creating artificial multiferroics, *i.e.* use of coupled systems rather than an intrinsic single phase material,¹³⁵ there has been insufficient information discussing the problems of achieving 'practical' systems using two different materials forms and how to overcome these. We cover these points below.

A strong advantage of VAN artificial multiferroics over SL are minimal clamping from the underlying substrate. Clamping prevents effective vertical strain and strain coupling of a ferroelectric with a magnetostrictive material to produce a change in magnetisation with applied voltage to give the converse magnetoelectric effect (CME), or change in polarisation with change in magnetisation to give the direct magnetoelectric effect (DME).

BaTiO₃-CoFe₂O₄ was the first VAN film explored for achieving the ME effect.¹³⁰ However, owing to poor electrical insulation of most ferromagnetic materials in multiferroic nanocomposites, only thermally-driven ME coupling was demonstrated. On the other hand, it has been shown in many different VAN systems *e.g.* BaTiO₃-CoFe₂O₄,¹³⁰ Ni_{0.5}Zn_{0.5}Fe₂O₄-BaTiO₃¹³⁶ and Y₂Fe₅O₁₂-BaTiO₃¹³⁷ (Fig. 8a). However, thermal ME provides only indirect evidence of ME coupling, as the strain is not induced by an electric field.

In terms of DME, the BaTiO₃-CoFe₂O₄ system is a key exemplar system.¹³⁸ Here, magnetic field induced strain has been observed using different methods, X-ray linear dichroism (XLD) (Fig. 8b),^{138a} second-harmonic generation,^{138b} and by measuring the modified PFM amplitude upon applying a magnetic field along both IP and OOP directions



Magnetoelectric properties in VAN films

Structural and physical effects responsible for observed interface-related phenomena

- Large vertical strain coupling
- Free of substrate clamping
- High density of vertical interfaces
- Reduced leakage
- Function-oriented composition design



Fig. 8 Enhanced magnetoelectric properties achieved using VAN structures. (a) Thermally-driven ME effect in $\text{Y}_3\text{Fe}_5\text{O}_{12}$ - BaTiO_3 VAN films detected by the FMR. Reproduced with permission.¹³⁷ Copyright 2017 American Chemical Society. (b) XLD asymmetry study of the lattice distortion induced by magnetic field in a BaTiO_3 - CoFe_2O_4 VAN film, the upper plot showing magnetic field applied OOP and the lower plot showing the magnetic field applied IP. Reproduced with permission.^{138a} Copyright 2013 Springer Nature. (c) The effect of the application of IP magnetic field on the amplitude-voltage butterfly loop (DME effect) in BaTiO_3 - CoFe_2O_4 VAN films. Reproduced with permission.^{138b} Copyright 2019 Wiley-VCH. (d and e) Charge mediated ME to give the CME effect in $\text{La}_2\text{CoMnO}_6$ - ZnO VAN film. Reproduced with permission.¹⁴² Copyright 2013 American Chemical Society. (d) Cross-sectional (false-colour) TEM image of the $\text{La}_2\text{CoMnO}_6$ - ZnO VAN film. (e) Electric field dependence of the magnetic hysteresis loops of $\text{La}_2\text{CoMnO}_6$ - ZnO VAN film where the magnetic field is applied IP. (f) *In situ* electric field control of magnetization (CME effect) in $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ - CoFe_2O_4 VAN film where the magnetic field is applied OOP. Reproduced with permission.¹⁴³ Copyright 2018 American Chemical Society. The inset shows a schematic of the application of an *in situ* electric field. (g) The magnetoelectric coefficient of the $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ - CoFe_2O_4 VAN film calculated from the magnetic hysteresis loops given in (f). (h) Cross-sectional TEM and EDX image of a 50 nm-thick $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ - NiO - NiFe_2O_4 film. Reproduced with permission.⁴⁰ Copyright 2021 Springer Nature. (i) *In situ* electric field control of magnetization (CME effect) in a 200 nm-thick $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ - NiO - NiFe_2O_4 VAN film where the magnetic field is applied OOP. The inset shows the magnetoelectric coefficient, calculated from the magnetic hysteresis loops. Reproduced with permission.⁴⁰ Copyright 2021 Springer Nature.

(Fig. 8c).^{138b,139} However, possibly owing to leakage problems, a direct electric readout of DME has not been reported.

The first direct measurement of the CME effect was shown in BiFeO_3 - CoFe_2O_4 VAN.¹⁴⁰ Here, magnetic hysteresis loops

were measured to detect the change of magnetization upon applying *ex situ* electric field pulses. However, this system suffered leakage as revealed by conductive atomic force microscopy (c-AFM) and cross-sectional scanning tunnelling microscopy (XSTM).^{15a,141}



In $\text{La}_2\text{CoMnO}_6\text{-ZnO}$ VAN films, *in situ* measurements showed the existence of the CME effect as well as tunability of magnetization and coercivity (Fig. 8d and e).¹⁴² It was determined that voltage-controlled charge trapping-detrapping in the ZnO at the high-density vertical interfaces led to a modified transition metal ion valence of Mn or Co in the $\text{La}_2\text{CoMnO}_6$ and hence to a change in the magnetic state of the ferrimagnetic $\text{La}_2\text{CoMnO}_6$. However, the low T_C of $\text{La}_2\text{CoMnO}_6$ (<210 K) in the thin films meant that the CME was observed only at a relatively low temperature (120 K).

The first direct measurement of a large CME effect at room temperature was observed in a VAN system of $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3\text{-CoFe}_2\text{O}_4$. By using the wide bandgap (3.3 eV), high T_C (603 K) $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ ferroelectric instead of the commonly studied, lower T_C and lower bandgap BaTiO_3 and BiFeO_3 ferroelectrics, a large-scale *in situ* electric field control of the magnetic anisotropy was obtained at room temperature (Fig. 8f).¹⁴³ The relatively low leakage was linked to (a) the lower leakage ferroelectric, (b) reduction of VAN vertical interface leakage which can arise when mixed valence transition metal cations are present in the system, and most importantly (c) the formation of a rectifying interface between the film and the substrate, enabling current blockage. An CME coefficient of around 1.25×10^{-9} S m^{-1} , at a bias field of ~ 0.3 T, was achieved (Fig. 8g).

Taking the above result one step further, a function-oriented composition design approach was used to reduce leakage even more. Hence, an antiferromagnetic (AFM) phase was incorporated into a similar VAN composition as discussed above,⁴⁰ making a triple composite system. Here, a triple VAN 3-1 structure of FE-AFM-FM $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3\text{-NiO-NiFe}_2\text{O}_4$ was created (Fig. 8h). We note that AFM materials have so far played an important role in traditional spintronic devices as pinning layers,¹⁴⁴ and as emerging AFM spintronic devices, offering ultrafast (\sim THz) dynamics, high stability to external fields, and absence of stray fields.¹⁴⁵ However, they have largely been overlooked in multiferroic VANs. In the triple VAN system, the high T_C (603 K) of $\text{Na}_{0.5}\text{Bi}_{0.5}\text{TiO}_3$ was again exploited, as above. But in addition, the soft magnetism of NiFe_2O_4 , the high Néel temperature (525 K) of NiO, and the large exchange coupling between the FM and AFM material was also capitalised upon.

A large, self-biased (zero applied magnetic field) CME effect at room temperature with a ME coefficient of up to 1.38×10^{-9} s m^{-1} was achieved (Fig. 8i). The self-biased CME effect is ascribed to 3 critical features introduced by the addition of NiO: the leakage is strongly suppressed by the introduction of p-type NiO; the antiferromagnetic spins are rotated by the strain due to the large magnetostrictive effect of NiO, as indicated by the electric controllable EB effect; and the exchange coupling between the AFM NiO spins and the FM NiFe_2O_4 spins enhance the overall ME effect. This one-shot growth film, which does not rely on precise control of composition since if one phase has a slightly different ratio than another it will not be of detriment to the coupling effect, holds much promise for wide ranging applications relating to non-volatile memory and magnetic sensors.¹⁴⁶ The next steps for this work are to demonstrate growth on TiN/Si.

C. Summary, perspectives and conclusions

9. Summary of ferroic phenomena in VAN films

VAN films formed from complex oxides provide an excellent platform for achieving enhanced ferroic properties in thin films by a one-shot process, *i.e.* the creating of self-assembled composite films. So far, most of the VAN systems have focused on strain coupling effects across the vertical interfaces, which give rise to 3D strain tuning of the bulk of the film, thus tuning the film properties up to micron thickness levels. The anisotropic nature of the nanopillars in VAN films can enable magnetic anisotropy effects to be tuned. VANs are different to the SL films where most emphasis has been on emergent interface properties and much thinner films. There has been much less emphasis on emergent interface effects in VAN films and there is a strong opportunity to explore more in this direction. The challenges for VAN are that there are limitations on what materials can be combined together, that the interfaces are non-ideal 2D, and that there may be more than one interface type in a given system. Overall, VAN films have strong promise for enhancing properties of ferroic films which are required to be relatively thick (*e.g.* actuator/energy harvesting), anisotropic (*e.g.* for perpendicular magnetic recording), or unclamped (*e.g.* multiferroic).

10. Future perspectives

Throughout the paper, we highlighted the different scientific discoveries and ferroic functionalities arising in oxide superlattice (SL) and self-assembled vertically aligned nanocomposite (VAN) structures. Using exemplars, we also discussed how and why the two systems offer different and often complementary effects. Below, we highlight some underexplored/emerging research directions for ferroic properties in both oxide superlattices and VAN systems.

10.1 Future perspective for superlattice films

1. Although many SL examples discussed have great potential for the discovery of new physical phenomena, one must always be aware of the many subtleties in dealing with these intrinsically nanoscale systems. As summarized in Fig. 1 and Table 1, in terms of tuning the functionalities of superlattice films, defects (cationic intermixing, point defect segregation, structural inhomogeneities, oxygen vacancies and so on) all play important, often poorly understood and interlinked roles. More work employing high intensity spectroscopic tools should be employed to understand the nature of the interface structures. Also, in order to minimise the role of interfacial strain which often produces a defective interface, focus on systems with minimal interfacial mismatch should be explored. This can be done by using new single crystal substrate compositions. There



has been some excellent work in this direction in recent years.¹⁴⁷

Also, to minimise strain and defect formation, the choice of more closely lattice matched layers in the SL films should be explored, where possible. To minimise cation non-stoichiometry in films, growth under lower oxidation conditions should be studied.¹⁴⁸ To minimise defects, films could be grown higher temperatures (*i.e.* at above 1000 °C rather than the conventional 700–800 °C).¹⁴⁹ However, considering a key driver is to make oxide SLs CMOS compatible, other strategies to minimise defects at ~500 °C and below must also be considered, *e.g.* atomic layer growth methods where epitaxial perovskite films have been grown at <300 °C,¹⁵⁰ as well as novel ways to enhance growth kinetics and also carefully control composition.¹⁴

2. Most ferroic SLs have been grown mainly on perovskite substrates with (001)-orientation. Epitaxial integration of ferroic superlattices with different crystal orientation other than (001) should be explored. This field has grown rapidly in the past decade but there are many more materials combinations to be explored. Non-(001) orientations give different surface/interface terminations, neutral charge states and 2D-like structures (*e.g.* (111) orientations giving buckled honeycombs). Also, growth on a range of buffered semiconductor substrates beyond the well-studied Si, *e.g.* on Ge, Si–Ge, GaAs, GaN, and for flexible electronics, flexible substrates, *e.g.* mica¹⁵¹ or analogous crystals needs to be explored.

3. Room-temperature ferroelectricity and ferromagnetism have been widely achieved in superlattice films. However, room-temperature multiferroic superlattices with strong and robust magnetoelectric coupling have still not been realized. Since lift-off procedures for oxide films are now quite well established,^{86,91,152} this allows clamping effects to be overcome and so more work could be directed here to achieve robust, crack-free, non-wrinkled and well adhered films. Also, the vast majority of work on superlattices has focused on *A/B* superlattices. However, there is a largely unexplored area of multi-component superlattices with complex periodicities. These could offer much further potential for achieving interesting ferroic behaviour. An example was shown for tricolor superlattices in Fig. 4, where three constituent materials forming an *ABC* periodicity. With the *ABC* periodicity, inversion symmetry is broken, enabling interesting novel magnetoelectric phenomena to be achieved.

4. There remains much to be understood about emergent phenomena, such as polar skyrmions, vortices, negative capacitance, and so on. Real-space topological textures such as polar skyrmions and vortices have been observed, but we are still in the early stages of the exploration of these phenomena. For example, the dynamics of such skyrmions structures and vortices must be explored in detail. In addition, these topologies demonstrate chirality and if this can be manipulated by an external field (such as an electric field), then this would provide a powerful way to use the chirality as an externally controllable order parameter.

5. The interfaces between 3d TMOs and 5d TMOs provide a new fruitful research area. Again, current research is still in the

early stages and limited to a few systems, so more systematic investigations are highly desired to fully unravel the unique role of 5d TMOs which in turn could provide more candidates to search for novel magnetic textures and topological phenomena. Moreover, owing to the inherent mixture of spin and orbital degrees of freedom in 5d TMOs, these superlattices may open a new pathway to achieve robust electric field control of magnetism at room temperature and above.

10.2 Future perspective for VAN films

1. Just as for SL films, VAN films are subject to a range of interface and bulk point and line defects, such as mutual doping of components.¹⁵³ To further optimise the perfection of these structures, similar application of spectroscopic tools, and the implementation of carefully controlled growth processes combined with a range of theoretical predictions of microstructure evolution, should take place.

2. As highlighted in Table 1, the wide ranging emergent interface phenomena, topological, skyrmion and polar vortices observed in SL films have not been observed in VAN films. Considering the very high epitaxial quality of VAN films, the ultraclean self-assembled interfaces, and the ability to control dimensionalities down to a few nm *via* controlling growth conditions, there are unique opportunities to discover and control these novel effects in VAN.

3. Combining VAN and SLs to form 3D film architectures has not yet been widely studied.¹⁵⁴ However, these structures could offer the chance to combine the complementary advantages of both methods and should be explored further.

4. Pulsed laser deposition is the main method to make VAN films with good quality. However, the method is area-limited. However, promising growth of a limited number of VAN systems by sputtering¹⁵⁵ and chemical methods¹⁵⁶ show the possibility for growth by methods of industrial interest. More work should be focused in these areas.

5. As already noted, compared to SL films, VAN films can exhibit more bulk-like effects, *i.e.* interface-driven effects throughout the whole film, not just right at the interfaces. Their simplicity of fabrication is also an important aspect. However, typically they have a random column distribution. For some applications, such as data storage, a highly ordered arrangement is desired. Templating by a variety of ways can produce the desired order,^{2b,155a,157} recently spontaneous order has been shown.^{2b} More work is needed to determine which methods are most practical for applications.

6. Just as for SLs, mentioned above, most VAN film work has been undertaken on single crystal substrates although progress has been made with STO/TiN-buffered Si¹⁵⁸ and with mica,¹⁵⁹ and lifted-off films onto Si.⁹² More work is required on these 'other substrates' as well as new ways to grow directly on Si, *e.g.* using remote epitaxy.

7. As shown in Section 6, high crystalline quality mesoporous films derived from VAN films give enhanced ferroelectric effects and they also exploit high area surface effects for photocatalysis. There are many new opportunities to exploit the large surface area effects or to create new composites (not



just oxide–oxide but also oxide–metal and oxide–organic systems) by chemical coating of mesoporous films made by selective chemical etching of VAN films. With this broader range of compositions, new coupling effects in wide ranging functional systems can be realised.

11. Conclusions

Over more than three decades, fascinating physical phenomena have been achieved in heteroepitaxial interfaces in complex oxide superlattices (SLs), and over the last two decades in vertically aligned nanocomposites (VANs). Numerous novels or enhanced ferroic phenomena have been achieved which have been highlighted in this review. The subtlety of the different natures of the interfaces in these two materials forms, SL or VAN, enables highly complementary phenomena to be achieved. Notably, SL films with 2D interfaces enable basic emergent physical phenomena to be discovered and explored, for example the creation of ferroelectricity,^{11,12,15,16} ferromagnetism,^{41,42,45} and multiferroism,⁶³ emergent polar skyrmions and vortices,^{25,27} etc. On the other hand, VAN films offer greater simplicity of growth, cleaner self-assembled interfaces, are largely unclamped by the substrate, and can be grown to micron thickness. On the other hand, they have more complex geometric vertical interfaces, and a much narrower range of materials can be grown. VAN are more suited to the engineering of ferroic properties, e.g. to increase the ferroelectric T_C ,^{21,54} to increase the magnetization^{34,115,119} and to enhance the magnetoelectric coupling.^{130,143,160} For both material forms, there are many new avenues to be explored for property discoveries and the creation of practical engineering devices, with several examples highlighted in this review, e.g. topological textures, negative capacitance, integration with silicon and other semiconductor substrates, flexibilization, etc.

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

The authors acknowledge support from Leverhulme Trust grant # RPG-2015-017, EPSRC grants EP/L011700/1, EP/N004272/1, EP/M000524/1, the Royal Academy of Engineering Chair in Emerging Technologies Grant CiET1819\24, and the ERC Advanced Grant EU-H2020-ERC-ADG #882929, EROS. R. W. also acknowledges support from the National Natural Science Foundation of China (Grant No. 12104052) and State Key Lab of Luminescent Materials and Devices, South China University of Technology (Grant No. Skllmd-2022-05). W.-W. L. also acknowledges support from the National Natural Science Foundation of China (Grant No. 52102177), the National Natural Science Foundation of Jiangsu Province (Grant No. BK20210313), Top-notch Academic Programs Project of Jiangsu Higher Education

Institutions (TAPP), and the Jiangsu Specially-Appointed Professor Program.

References

- (a) R. Ramesh and D. G. Schlom, *Nat. Rev. Mater.*, 2019, **4**, 257; (b) S. Das, A. Ghosh, M. R. McCarter, S. L. Hsu, Y. L. Tang, A. R. Damodaran, R. Ramesh and L. W. Martin, *APL Mater.*, 2018, **6**, 100901; (c) J. Zhang and R. D. Averitt, *Annu. Rev. Mater. Res.*, 2014, **44**, 19; (d) P. Zubko, S. Gariglio, M. Gabay, P. Ghosez and J. M. Triscone, *Annu. Rev. Condens. Matter Phys.*, 2011, **2**, 141; (e) D. G. Schlom, L. Q. Chen, C. B. Eom, K. M. Rabe, S. K. Streiffer and J. M. Triscone, *Annu. Rev. Mater. Res.*, 2007, **37**, 589.
- (a) A. Chen, Q. Su, H. Han, E. Enriquez and Q. Jia, *Adv. Mater.*, 2019, **31**, e1803241; (b) X. Sun, J. L. MacManus-Driscoll and H. Y. Wang, *Annu. Rev. Mater. Res.*, 2020, **50**, 229; (c) J. J. Huang, J. L. MacManus-Driscoll and H. Y. Wang, *J. Mater. Res.*, 2017, **32**, 4054.
- M. Dawber, K. M. Rabe and J. F. Scott, *Rev. Mod. Phys.*, 2005, **77**, 1083.
- J. X. Guo, W. W. Chen, H. S. Chen, Y. N. Zhao, F. Dong, W. W. Liu and Y. Zhang, *Adv. Opt. Mater.*, 2021, **9**, 2002146.
- L. W. Martin and A. M. Rappe, *Nat. Rev. Mater.*, 2016, **2**, 16087.
- (a) H. Takasu, *J. Electroceram.*, 2000, **4**, 327; (b) C. S. Hwang and T. Mikolajick, in *Advances in Non-Volatile Memory and Storage Technology*, ed. B. Magyari-Köpe and Y. Nishi, Woodhead Publishing, 2019; (c) N. Setter, D. Damjanovic, L. Eng, G. Fox, S. Gevorgian, S. Hong, A. Kingon, H. Kohlstedt, N. Y. Park, G. B. Stephenson, I. Stolitchnov, A. K. TagansteV, D. V. Taylor, T. Yamada and S. Streiffer, *J. Appl. Phys.*, 2006, **100**, 051606; (d) G. H. Haertling, *J. Am. Ceram. Soc.*, 1999, **82**, 797; (e) J. F. Scott, *Science*, 2007, **315**, 954; (f) A. Kursumovic, W. W. Li, S. Cho, P. J. Curran, D. H. L. Tjhe and J. L. MacManus-Driscoll, *Nano Energy*, 2020, **71**, 104536.
- G. Varvaro and F. Casoli, *Ultra-High-Density Magnetic Recording: Storage Materials and Media Designs*, Pan Stanford Publishing, 2016.
- K. L. Wang, J. G. Alzate and P. K. Amiri, *J. Phys. D: Appl. Phys.*, 2013, **46**, 074003.
- (a) J. F. Scott, *Nat. Mater.*, 2007, **6**, 256; (b) M. Bibes and A. Barthelemy, *Nat. Mater.*, 2008, **7**, 425; (c) S. H. Baek, H. W. Jang, C. M. Folkman, Y. L. Li, B. Winchester, J. X. Zhang, Q. He, Y. H. Chu, C. T. Nelson, M. S. Rzchowski, X. Q. Pan, R. Ramesh, L. Q. Chen and C. B. Eom, *Nat. Mater.*, 2010, **9**, 309.
- Z. Chen, X. Wang, Y. Qi, S. Yang, J. A. Soares, B. A. Apgar, R. Gao, R. Xu, Y. Lee, X. Zhang, J. Yao and L. W. Martin, *ACS Nano*, 2016, **10**, 10237.
- R. Zhao, W. Li, J. H. Lee, E. M. Choi, Y. Liang, W. Zhang, R. Tang, H. Wang, Q. Jia, J. L. MacManus-Driscoll and H. Yang, *Adv. Funct. Mater.*, 2014, **24**, 5240.
- J. L. MacManus-Driscoll, P. Zerrer, H. Wang, H. Yang, J. Yoon, A. Fouchet, R. Yu, M. G. Blamire and Q. Jia, *Nat. Mater.*, 2008, **7**, 314.



- 13 J. MacManus-Driscoll, A. Suwardi, A. Kursumovic, Z. X. Bi, C. F. Tsai, H. Y. Wang, Q. X. Jia and O. J. Lee, *APL Mater.*, 2015, **3**, 062507.
- 14 J. L. MacManus-Driscoll, M. P. Wells, C. Yun, J. W. Lee, C. B. Eom and D. G. Schlom, *APL Mater.*, 2020, **8**, 040904.
- 15 (a) Y. H. Hsieh, J. M. Liou, B. C. Huang, C. W. Liang, Q. He, Q. Zhan, Y. P. Chiu, Y. C. Chen and Y. H. Chu, *Adv. Mater.*, 2012, **24**, 4564; (b) B. Zhu, G. Schusteritsch, P. Lu, J. L. Macmanus-Driscoll and C. J. Pickard, *APL Mater.*, 2019, **7**, 061105.
- 16 Z. Hiroi, H. Hayamizu, T. Yoshida, Y. Muraoka, Y. Okamoto, J. Yamaura and Y. Ueda, *Chem. Mater.*, 2013, **25**, 2202.
- 17 J. L. MacManus-Driscoll, *Adv. Funct. Mater.*, 2010, **20**, 2035.
- 18 (a) A. P. Chen, Z. X. Bi, Q. X. Jia, J. L. MacManus-Driscoll and H. Y. Wang, *Acta Mater.*, 2013, **61**, 2783; (b) W. R. Zhang, A. P. Chen, Z. K. Bi, Q. X. Jia, J. L. MacManus-Driscoll and H. Y. Wang, *Curr. Opin. Solid State Mater. Sci.*, 2014, **18**, 6.
- 19 (a) W. R. Zhang, R. Ramesh, J. L. MacManus-Driscoll and H. Y. Wang, *MRS Bull.*, 2015, **40**, 736; (b) J. L. MacManus-Driscoll, A. Suwardi and H. Wang, *MRS Bull.*, 2015, **40**, 933.
- 20 E. M. Choi, A. Di Bernardo, B. Zhu, P. Lu, H. Alpern, K. H. L. Zhang, T. Shapira, J. Feighan, X. Sun, J. Robinson, Y. Paltiel, O. Millo, H. Wang, Q. Jia and J. L. MacManus-Driscoll, *Sci. Adv.*, 2019, **5**, eaav5532.
- 21 S. A. Harrington, J. Zhai, S. Denev, V. Gopalan, H. Wang, Z. Bi, S. A. Redfern, S. H. Baek, C. W. Bark, C. B. Eom, Q. Jia, M. E. Vickers and J. L. Macmanus-Driscoll, *Nat. Nanotechnol.*, 2011, **6**, 491.
- 22 R. Wu, C. Yun, X. Wang, P. Lu, W. Li, Y. Lin, E.-M. Choi, H. Wang and J. L. MacManus-Driscoll, *ACS Appl. Mater. Interfaces*, 2018, **10**, 42593.
- 23 D. A. Tenne, A. Bruchhausen, N. D. Lanzillotti-Kimura, A. Fainstein, R. S. Katiyar, A. Cantarero, A. Soukiassian, V. Vaithyanathan, J. H. Haeni, W. Tian, D. G. Schlom, K. J. Choi, D. M. Kim, C. B. Eom, H. P. Sun, X. Q. Pan, Y. L. Li, L. Q. Chen, Q. X. Jia, S. M. Nakhmanson, K. M. Rabe and X. X. Xi, *Science*, 2006, **313**, 1614.
- 24 H. N. Lee, H. M. Christen, M. F. Chisholm, C. M. Rouleau and D. H. Lowndes, *Nature*, 2005, **433**, 395.
- 25 O. Lee, A. Kursumovic, Z. Bi, C. F. Tsai, H. Wang and J. L. MacManus-Driscoll, *Adv. Mater. Interfaces*, 2017, **4**, 1700336.
- 26 N. M. Dawley, E. J. Marksz, A. M. Hagerstrom, G. H. Olsen, M. E. Holtz, V. Goian, C. Kadlec, J. Zhang, X. Lu, J. A. Drisko, R. Uecker, S. Ganschow, C. J. Long, J. C. Booth, S. Kamba, C. J. Fennie, D. A. Muller, N. D. Orloff and D. G. Schlom, *Nat. Mater.*, 2020, **19**, 176.
- 27 O. Lee, S. A. Harrington, A. Kursumovic, E. Defay, H. Wang, Z. Bi, C. F. Tsai, L. Yan, Q. Jia and J. L. MacManus-Driscoll, *Nano Lett.*, 2012, **12**, 4311.
- 28 (a) W. Gao, A. Khan, X. Marti, C. Nelson, C. Serrao, J. Ravichandran, R. Ramesh and S. Salahuddin, *Nano Lett.*, 2014, **14**, 5814; (b) S. Das, Y. L. Tang, Z. Hong, M. A. P. Goncalves, M. R. McCarter, C. Klewe, K. X. Nguyen, F. Gomez-Ortiz, P. Shafer, E. Arenholz, V. A. Stoica, S. L. Hsu, B. Wang, C. Ophus, J. F. Liu, C. T. Nelson, S. Saremi, B. Prasad, A. B. Mei, D. G. Schlom, J. Iniguez, P. Garcia-Fernandez, D. A. Muller, L. Q. Chen, J. Junquera, L. W. Martin and R. Ramesh, *Nature*, 2019, **568**, 368.
- 29 (a) A. K. Yadav, C. T. Nelson, S. L. Hsu, Z. Hong, J. D. Clarkson, C. M. Schlepütz, A. R. Damodaran, P. Shafer, E. Arenholz, L. R. Dedon, D. Chen, A. Vishwanath, A. M. Minor, L. Q. Chen, J. F. Scott, L. W. Martin and R. Ramesh, *Nature*, 2016, **530**, 198; (b) P. Shafer, P. Garcia-Fernandez, P. Aguado-Puente, A. R. Damodaran, A. K. Yadav, C. T. Nelson, S. L. Hsu, J. C. Wojdel, J. Iniguez, L. W. Martin, E. Arenholz, J. Junquera and R. Ramesh, *Proc. Natl. Acad. Sci. U. S. A.*, 2018, **115**, 915.
- 30 C. He, A. J. Grutter, M. Gu, N. D. Browning, Y. Takamura, B. J. Kirby, J. A. Borchers, J. W. Kim, M. R. Fitzsimmons, X. Zhai, V. V. Mehta, F. J. Wong and Y. Suzuki, *Phys. Rev. Lett.*, 2012, **109**, 197202.
- 31 B. Negulescu, L. Thomas, Y. Dumont, M. Tessier, N. Keller and M. Guyot, *J. Magn. Magn. Mater.*, 2002, **242–245**, 529.
- 32 (a) K. Ueda, H. Tabata and T. Kawai, *Science*, 1998, **280**, 1064; (b) K. Ueda, H. Tabata and T. Kawai, *J. Appl. Phys.*, 2001, **89**, 2847.
- 33 E. Weal, S. Patnaik, Z. Bi, H. Wang, T. Fix, A. Kursumovic and J. L. M. Driscoll, *Appl. Phys. Lett.*, 2010, **97**, 153121.
- 34 C. Park, R. Wu, P. Lu, H. Zhao, J. B. Yang, B. Zhang, W. W. Li, C. Yun, H. Y. Wang, J. L. MacManus-Driscoll and S. Cho, *Adv. Funct. Mater.*, 2018, **28**, 1706220.
- 35 D. Yi, C. L. Flint, P. P. Balakrishnan, K. Mahalingam, B. Urwin, A. Vaillonis, A. T. N'Diaye, P. Shafer, E. Arenholz, Y. Choi, K. H. Stone, J. H. Chu, B. M. Howe, J. Liu, I. R. Fisher and Y. Suzuki, *Phys. Rev. Lett.*, 2017, **119**, 077201.
- 36 H. M. Zheng, J. Kreisel, Y. H. Chu, R. Ramesh and L. Salamanca-Riba, *Appl. Phys. Lett.*, 2007, **90**, 113113.
- 37 E. Skoropata, J. Nichols, J. M. Ok, R. V. Chopdekar, E. S. Choi, A. Rastogi, C. Sohn, X. Gao, S. Yoon, T. Farmer, R. D. Desautels, Y. Choi, D. Haskel, J. W. Freeland, S. Okamoto, M. Brahlek and H. N. Lee, *Sci. Adv.*, 2020, **6**, eaaz3902.
- 38 (a) M. Lorenz, G. Wagner, V. Lazenka, P. Schwinkendorf, H. Modarresi, M. J. Van Bael, A. Vantomme, K. Temst, O. Oeckler and M. Grundmann, *Appl. Phys. Lett.*, 2015, **106**, 012905; (b) M. Lorenz, D. Hirsch, C. Patzig, T. Hoche, S. Hohenberger, H. Hochmuth, V. Lazenka, K. Temst and M. Grundmann, *ACS Appl. Mater. Interfaces*, 2017, **9**, 18956.
- 39 L. Yan, Z. P. Xing, Z. G. Wang, T. Wang, G. Y. Lei, J. F. Li and D. Viehland, *Appl. Phys. Lett.*, 2009, **94**.
- 40 R. Wu, D. Zhang, T. Maity, P. Lu, J. Yang, X. Gao, S. Zhao, X. Wei, H. Zeng, A. Kursumovic, G. Tian, W. Li, C. Yun, Y. Wang, Z. Ren, Z. Zhou, M. Liu, K. H. L. Zhang, Q. Jia, J. Yang, H. Wang and J. L. MacManus-Driscoll, *Nat. Electron.*, 2021, **4**, 333.
- 41 K. Rogdakis, J. W. Seo, Z. Viskadourakis, Y. Wang, L. F. Qune, E. Choi, J. D. Burton, E. Y. Tsymbal, J. Lee and C. Panagopoulos, *Nat. Commun.*, 2012, **3**, 1064.
- 42 V. A. Stoica, N. Laanait, C. Dai, Z. Hong, Y. Yuan, Z. Zhang, S. Lei, M. R. McCarter, A. Yadav, A. R. Damodaran, S. Das,



- G. A. Stone, J. Karapetrova, D. A. Walko, X. Zhang, L. W. Martin, R. Ramesh, L. Q. Chen, H. Wen, V. Gopalan and J. W. Freeland, *Nat. Mater.*, 2019, **18**, 377.
- 43 R. Mishra, Y. M. Kim, J. Salafranca, S. K. Kim, S. H. Chang, A. Bhattacharya, D. D. Fong, S. J. Pennycook, S. T. Pantelides and A. Y. Borisevich, *Nano Lett.*, 2014, **14**, 2694.
- 44 S. Cho, C. Yun, S. Tappertzhofen, A. Kursumovic, S. Lee, P. Lu, Q. Jia, M. Fan, J. Jian, H. Wang, S. Hofmann and J. L. MacManus-Driscoll, *Nat. Commun.*, 2016, **7**, 12373.
- 45 J. M. Rondinelli and C. J. Fennie, *Adv. Mater.*, 2012, **24**, 1961.
- 46 H. Zhang, Y. Weng, X. Yao and S. Dong, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2015, **91**, 195145.
- 47 S. Y. Park, A. Kumar and K. M. Rabe, *Phys. Rev. Lett.*, 2017, **118**, 087602.
- 48 S. Ghosh, A. Y. Borisevich and S. T. Pantelides, *Phys. Rev. Lett.*, 2017, **119**, 177603.
- 49 (a) J. Ravichandran, A. K. Yadav, R. Cheaito, P. B. Rossen, A. Soukiassian, S. J. Suresha, J. C. Duda, B. M. Foley, C. H. Lee, Y. Zhu, A. W. Lichtenberger, J. E. Moore, D. A. Muller, D. G. Schlom, P. E. Hopkins, A. Majumdar, R. Ramesh and M. A. Zurbuchen, *Nat. Mater.*, 2014, **13**, 168; (b) W. Tian, J. C. Jiang, X. Q. Pan, J. H. Haeni, Y. L. Li, L. Q. Chen, D. G. Schlom, J. B. Neaton, K. M. Rabe and Q. X. Jia, *Appl. Phys. Lett.*, 2006, **89**, 092905.
- 50 (a) S. S. A. Seo, J. H. Lee, H. N. Lee, M. F. Chisholm, W. S. Choi, D. J. Kim, J. Y. Jo, H. Kim, J. Yu and T. W. Noh, *Adv. Mater.*, 2007, **19**, 2460; (b) X. Wu, K. M. Rabe and D. Vanderbilt, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **83**, 020104.
- 51 (a) S. Berber, Y. K. Kwon and D. Tomanek, *Phys. Rev. Lett.*, 2000, **84**, 4613; (b) X. Wu, M. Stengel, K. M. Rabe and D. Vanderbilt, *Phys. Rev. Lett.*, 2008, **101**, 087601.
- 52 (a) C. C. Homes, T. Vogt, S. M. Shapiro, S. Wakimoto and A. P. Ramirez, *Science*, 2001, **293**, 673; (b) S. Krohns, P. Lunkenheimer, S. Meissner, A. Reller, B. Gleich, A. Rathgeber, T. Gaugler, H. U. Buhl, D. C. Sinclair and A. Loidl, *Nat. Mater.*, 2011, **10**, 899.
- 53 (a) A. K. Tagantsev, V. O. Sherman, K. F. Astafiev, J. Venkatesh and N. Setter, *J. Electroceram.*, 2003, **11**, 5; (b) P. Bao, T. J. Jackson, X. Wang and M. J. Lancaster, *J. Phys. D: Appl. Phys.*, 2008, **41**, 063001.
- 54 A. L. Sangle, O. J. Lee, A. Kursumovic, W. Zhang, A. Chen, H. Wang and J. L. MacManus-Driscoll, *Nanoscale*, 2018, **10**, 3460.
- 55 E. Bousquet, M. Dawber, N. Stucki, C. Lichtensteiger, P. Hermet, S. Gariglio, J. M. Triscone and P. Ghosez, *Nature*, 2008, **452**, 732.
- 56 W. Zhong and D. Vanderbilt, *Phys. Rev. Lett.*, 1995, **74**, 2587.
- 57 X. Gao, D. Zhang, X. Wang, J. Jian, Z. He, H. Dou and H. Wang, *Nanoscale Adv.*, 2020, **2**, 3276.
- 58 (a) A. K. Yadav, K. X. Nguyen, Z. Hong, P. Garcia-Fernandez, P. Aguado-Puente, C. T. Nelson, S. Das, B. Prasad, D. Kwon, S. Cheema, A. I. Khan, C. Hu, J. Iniguez, J. Junquera, L. Q. Chen, D. A. Muller, R. Ramesh and S. Salahuddin, *Nature*, 2019, **565**, 468; (b) S. Salahuddin and S. Datta, *Nano Lett.*, 2008, **8**, 405; (c) P. Zubko, J. C. Wojdel, M. Hadjimichael, S. Fernandez-Pena, A. Sene, I. Luk'yanchuk, J. M. Triscone and J. Iniguez, *Nature*, 2016, **534**, 524; (d) J. Iniguez, P. Zubko, I. Luk'yanchuk and A. Cano, *Nat. Rev. Mater.*, 2019, **4**, 243.
- 59 I. Kornev, H. Fu and L. Bellaiche, *Phys. Rev. Lett.*, 2004, **93**, 196104.
- 60 (a) A. Chen, J. M. Hu, P. Lu, T. Yang, W. Zhang, L. Li, T. Ahmed, E. Enriquez, M. Weigand, Q. Su, H. Wang, J. X. Zhu, J. L. MacManus-Driscoll, L. Q. Chen, D. Yarotski and Q. Jia, *Sci. Adv.*, 2016, **2**, e1600245; (b) R. Zhao, C. Yang, H. Wang, K. Jiang, H. Wu, S. Shen, L. Wang, Y. Sun, K. Jin, J. Gao, L. Chen, H. Wang, J. L. MacManus-Driscoll, P. A. van Aken, J. Hong, W. Li and H. Yang, *Nat. Commun.*, 2022, **13**, 2364.
- 61 (a) M. B. Salamon and M. Jaime, *Rev. Mod. Phys.*, 2001, **73**, 583; (b) J. M. Rondinelli and N. A. Spaldin, *Adv. Mater.*, 2011, **23**, 3363; (c) W. Li, B. Zhu, Q. He, A. Y. Borisevich, C. Yun, R. Wu, P. Lu, Z. Qi, Q. Wang, A. Chen, H. Wang, S. A. Cavill, K. H. L. Zhang and J. L. MacManus-Driscoll, *Adv. Sci.*, 2020, **7**, 1901606.
- 62 M. Gibert, P. Zubko, R. Scherwitzl, J. Iniguez and J. M. Triscone, *Nat. Mater.*, 2012, **11**, 195.
- 63 A. Bhattacharya and S. J. May, *Annu. Rev. Mater. Res.*, 2014, **44**, 65.
- 64 W. W. Li, B. N. Zhu, R. X. Zhu, Q. Wang, P. Lu, Y. W. Sun, C. Cafolla, Z. M. Qi, A. P. Chen, P. Gao, H. Y. Wang, Q. He, K. H. L. Zhang and J. L. MacManus-Driscoll, *Adv. Funct. Mater.*, 2020, **30**, 2001984.
- 65 D. Yi, J. Liu, S. L. Hsu, L. Zhang, Y. Choi, J. W. Kim, Z. Chen, J. D. Clarkson, C. R. Serrao, E. Arenholz, P. J. Ryan, H. Xu, R. J. Birgeneau and R. Ramesh, *Proc. Natl. Acad. Sci. U. S. A.*, 2016, **113**, 6397.
- 66 H. Y. Hwang, Y. Iwasa, M. Kawasaki, B. Keimer, N. Nagaosa and Y. Tokura, *Nat. Mater.*, 2012, **11**, 103.
- 67 K. S. Takahashi, M. Kawasaki and Y. Tokura, *Appl. Phys. Lett.*, 2001, **79**, 1324.
- 68 S. Dong and E. Dagotto, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, **87**, 195116.
- 69 I. Loa, P. Adler, A. Grzechnik, K. Syassen, U. Schwarz, M. Hanfland, G. K. Rozenberg, P. Gorodetsky and M. P. Pasternak, *Phys. Rev. Lett.*, 2001, **87**, 125501.
- 70 A. Bhattacharya, S. J. May, S. G. te Velthuis, M. Warusawithana, X. Zhai, B. Jiang, J. M. Zuo, M. R. Fitzsimmons, S. D. Bader and J. N. Eckstein, *Phys. Rev. Lett.*, 2008, **100**, 257203.
- 71 C. Aruta, C. Adamo, A. Galdi, P. Orgiani, V. Bisogni, N. B. Brookes, J. C. Cezar, P. Thakur, C. A. Perroni, G. De Filippis, V. Cataudella, D. G. Schlom, L. Maritato and G. Ghiringhelli, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2009, **80**, 140405.
- 72 (a) H. Yang, H. Wang, J. Yoon, Y. Wang, M. Jain, D. M. Feldmann, P. C. Dowden, J. L. MacManus-Driscoll and Q. Jia, *Adv. Mater.*, 2009, **21**, 3794; (b) Y. H. Hsieh, E. Strelcov, J. M. Liou, C. Y. Shen, Y. C. Chen, S. V. Kalinin and Y. H. Chu, *ACS Nano*, 2013, **7**, 8627.



- 73 (a) S. Lee, A. Sangle, P. Lu, A. Chen, W. Zhang, J. S. Lee, H. Wang, Q. Jia and J. L. MacManus-Driscoll, *Adv. Mater.*, 2014, **26**, 6284; (b) S. Lee and J. L. MacManus-Driscoll, *APL Mater.*, 2017, **5**, 042304.
- 74 K. Ueda, H. Tabata and T. Kawai, *J. Appl. Phys.*, 2001, **89**, 2847.
- 75 A. Sadoc, B. Mercey, C. Simon, D. Grebille, W. Prellier and M. B. Lepetit, *Phys. Rev. Lett.*, 2010, **104**, 046804.
- 76 (a) L. Qiao, J. H. Jang, D. J. Singh, Z. Gai, H. Xiao, A. Mehta, R. K. Vasudevan, A. Tselev, Z. Feng, H. Zhou, S. Li, W. Prellier, X. Zu, Z. Liu, A. Borisevich, A. P. Baddorf and M. D. Biegalski, *Nano Lett.*, 2015, **15**, 4677; (b) J. M. Rondinelli, S. J. May and J. W. Freeland, *MRS Bull.*, 2012, **37**, 261.
- 77 X. Zhai, L. Cheng, Y. Liu, C. M. Schlepütz, S. Dong, H. Li, X. Zhang, S. Chu, L. Zheng, J. Zhang, A. Zhao, H. Hong, A. Bhattacharya, J. N. Eckstein and C. Zeng, *Nat. Commun.*, 2014, **5**, 4283.
- 78 S. Okamoto, J. Nichols, C. Sohn, S. Y. Kim, T. W. Noh and H. N. Lee, *Nano Lett.*, 2017, **17**, 2126.
- 79 J. Nichols, X. Gao, S. Lee, T. L. Meyer, J. W. Freeland, V. Lauter, D. Yi, J. Liu, D. Haskel, J. R. Petrie, E.-J. Guo, A. Herklotz, D. Lee, T. Z. Ward, G. Eres, M. R. Fitzsimmons and H. N. Lee, *Nat. Commun.*, 2016, **7**, 12721.
- 80 M. A. Marioni, M. Penedo, M. Bacani, J. Schwenk and H. J. Hug, *Nano Lett.*, 2018, **18**, 2263.
- 81 N. A. Hill, *J. Phys. Chem. B*, 2000, **104**, 6694.
- 82 X. He and K. J. Jin, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2016, **93**, 161108.
- 83 H. Guo, Z. Wang, S. Dong, S. Ghosh, M. Saghayezhian, L. Chen, Y. Weng, A. Herklotz, T. Z. Ward, R. Jin, S. T. Pantelides, Y. Zhu, J. Zhang and E. W. Plummer, *Proc. Natl. Acad. Sci. U. S. A.*, 2017, **114**, E5062.
- 84 Y. Tokura, *J. Magn. Magn. Mater.*, 2007, **310**, 1145.
- 85 J. A. Mundy, C. M. Brooks, M. E. Holtz, J. A. Moyer, H. Das, A. F. Rebola, J. T. Heron, J. D. Clarkson, S. M. Disseler, Z. Liu, A. Farhan, R. Held, R. Hovden, E. Padgett, Q. Mao, H. Paik, R. Misra, L. F. Kourkoutis, E. Arenholz, A. Scholl, J. A. Borchers, W. D. Ratcliff, R. Ramesh, C. J. Fennie, P. Schiffer, D. A. Muller and D. G. Schlom, *Nature*, 2016, **537**, 523.
- 86 D. Ji, S. Cai, T. R. Paudel, H. Sun, C. Zhang, L. Han, Y. Wei, Y. Zang, M. Gu, Y. Zhang, W. Gao, H. Huyan, W. Guo, D. Wu, Z. Gu, E. Y. Tsymbal, P. Wang, Y. Nie and X. Pan, *Nature*, 2019, **570**, 87.
- 87 (a) S. Singh, J. T. Haraldsen, J. Xiong, E. M. Choi, P. Lu, D. Yi, X. D. Wen, J. Liu, H. Wang, Z. Bi, P. Yu, M. R. Fitzsimmons, J. L. MacManus-Driscoll, R. Ramesh, A. V. Balatsky, J. X. Zhu and Q. X. Jia, *Phys. Rev. Lett.*, 2014, **113**, 047204; (b) E. J. Guo, J. R. Petrie, M. A. Roldan, Q. Li, R. D. Desautels, T. Charlton, A. Herklotz, J. Nichols, J. van Lierop, J. W. Freeland, S. V. Kalinin, H. N. Lee and M. R. Fitzsimmons, *Adv. Mater.*, 2017, **29**, 1700790.
- 88 (a) Y. Ogawa, H. Yamada, T. Ogasawara, T. Arima, H. Okamoto, M. Kawasaki and Y. Tokura, *Phys. Rev. Lett.*, 2003, **90**, 217403; (b) N. Kida, H. Yamada, H. Sato, T. Arima, M. Kawasaki, H. Akoh and Y. Tokura, *Phys. Rev. Lett.*, 2007, **99**, 197404.
- 89 H. Das, A. L. Wysocki, Y. Geng, W. Wu and C. J. Fennie, *Nat. Commun.*, 2014, **5**, 2998.
- 90 (a) A. T. Chen, H. L. Huang, Y. Wen, W. Y. Liu, S. F. Zhang, J. Kosel, W. D. Sun, Y. G. Zhao, Y. L. Lu and X. X. Zhang, *Mater. Horiz.*, 2020, **7**, 2328; (b) M. Chen, M. G. Ju, H. F. Garces, A. D. Carl, L. K. Ono, Z. Hawash, Y. Zhang, T. Shen, Y. Qi, R. L. Grimm, D. Pacifici, X. C. Zeng, Y. Zhou and N. P. Padture, *Nat. Commun.*, 2019, **10**, 16; (c) E. Lage, C. Kirchhof, V. Hrkac, L. Kienle, R. Jahns, R. Knochel, E. Quandt and D. Meyners, *Nat. Mater.*, 2012, **11**, 523.
- 91 G. Dong, S. Li, M. Yao, Z. Zhou, Y. Q. Zhang, X. Han, Z. Luo, J. Yao, B. Peng, Z. Hu, H. Huang, T. Jia, J. Li, W. Ren, Z. G. Ye, X. Ding, J. Sun, C. W. Nan, L. Q. Chen, J. Li and M. Liu, *Science*, 2019, **366**, 475.
- 92 B. Zhang, C. Yun and J. L. MacManus-Driscoll, *Nanomicro Lett.*, 2021, **13**, 39.
- 93 (a) J. H. Haeni, P. Irvin, W. Chang, R. Uecker, P. Reiche, Y. L. Li, S. Choudhury, W. Tian, M. E. Hawley, B. Craigo, A. K. Tagantsev, X. Q. Pan, S. K. Streiffer, L. Q. Chen, S. W. Kirchoefer, J. Levy and D. G. Schlom, *Nature*, 2004, **430**, 758; (b) K. J. Choi, M. Biegalski, Y. L. Li, A. Sharan, J. Schubert, R. Uecker, P. Reiche, Y. B. Chen, X. Q. Pan, V. Gopalan, L. Q. Chen, D. G. Schlom and C. B. Eom, *Science*, 2004, **306**, 1005; (c) L. Zhang, J. Chen, L. Fan, O. Dieguez, J. Cao, Z. Pan, Y. Wang, J. Wang, M. Kim, S. Deng, J. Wang, H. Wang, J. Deng, R. Yu, J. F. Scott and X. Xing, *Science*, 2018, **361**, 494.
- 94 (a) S. C. Wimbush, M. C. Li, M. E. Vickers, B. Maiorov, D. M. Feldmann, Q. X. Jia and J. L. MacManus-Driscoll, *Adv. Funct. Mater.*, 2009, **19**, 835; (b) V. Nagarajan, C. L. Jia, H. Kohlstedt, R. Waser, I. B. Misirlioglu, S. P. Alpay and R. Ramesh, *Appl. Phys. Lett.*, 2005, **86**, 192910; (c) F. Sandiumenge, J. Santiso, L. Balcells, Z. Konstantinovic, J. Roqueta, A. Pomar, J. P. Espinos and B. Martinez, *Phys. Rev. Lett.*, 2013, **110**, 107206.
- 95 C. Yun, E.-M. Choi, W. Li, X. Sun, T. Maity, R. Wu, J. Jian, S. Xue, S. Cho and H. Wang, *Nanoscale*, 2020, **12**, 9255.
- 96 (a) H. Yang, H. Wang, G. F. Zou, M. Jain, N. A. Suvorova, D. M. Feldmann, P. C. Dowden, R. F. DePaula, J. L. MacManus-Driscoll, A. J. Taylor and Q. X. Jia, *Appl. Phys. Lett.*, 2008, **93**, 142904; (b) Y. H. Hsieh, F. Xue, T. Yang, H. J. Liu, Y. Zhu, Y. C. Chen, Q. Zhan, C. G. Duan, L. Q. Chen, Q. He and Y. H. Chu, *Nat. Commun.*, 2016, **7**, 13199; (c) X. Zhang, R. Xu, X. Gao, Y. Ji, F. Qian, J. Fan, H. Wang, W. Li and H. Yang, *J. Mater. Chem. C*, 2020, **8**, 8091; (d) E. M. Choi, T. Maity, A. Kursumoyic, P. Lu, Z. H. Bi, S. K. Yu, Y. Park, B. N. Zhu, R. Wu, V. Gopalan, H. Y. Wang and J. L. MacManus-Driscoll, *Nat. Commun.*, 2020, **11**, 2207.
- 97 H. Liu, H. Wu, K. P. Ong, T. Yang, P. Yang, P. K. Das, X. Chi, Y. Zhang, C. Diao, W. K. A. Wong, E. P. Chew, Y. F. Chen, C. K. I. Tan, A. Rusydi, M. B. H. Breese, D. J. Singh, L. Q. Chen, S. J. Pennycook and K. Yao, *Science*, 2020, **369**, 292.



- 98 K.-i Sakayori, Y. Matsui, H. Abe, E. Nakamura, M. Kenmoku, T. Hara, D. Ishikawa, A. Kokubu, K.-i Hirota and T. Ikeda, *Jpn. J. Appl. Phys.*, 1995, **34**, 5443.
- 99 D. A. Tenne, A. Bruchhausen, N. D. Lanzillotti-Kimura, A. Fainstein, R. S. Katiyar, A. Cantarero, A. Soukiassian, V. Vaithyanathan, J. H. Haeni, W. Tian, D. G. Schlom, K. J. Choi, D. M. Kim, C. B. Eom, H. P. Sun, X. Q. Pan, Y. L. Li, L. Q. Chen, Q. X. Jia, S. M. Nakhmanson, K. M. Rabe and X. X. Xi, *Science*, 2006, **313**, 1614.
- 100 (a) F. Khatkhatay, A. Chen, J. H. Lee, W. Zhang, H. Abdel-Raziq and H. Wang, *ACS Appl. Mater. Interfaces*, 2013, **5**, 12541; (b) X. Y. Zhang, R. X. Xu, X. Y. Gao, M. Li, X. N. Shi, Y. D. Ji, F. J. Qian, J. Y. Fan, H. Y. Wang, W. W. Li and H. Yang, *Appl. Surf. Sci.*, 2020, **516**, 146093; (c) W. Li, R. Zhao, R. Tang, A. Chen, W. Zhang, X. Lu, H. Wang and H. Yang, *ACS Appl. Mater. Interfaces*, 2014, **6**, 5356; (d) W. Li, W. Zhang, L. Wang, J. Gu, A. Chen, R. Zhao, Y. Liang, H. Guo, R. Tang, C. Wang, K. Jin, H. Wang and H. Yang, *Sci. Rep.*, 2015, **5**, 11335.
- 101 M. C. Rose and R. E. Cohen, *Phys. Rev. Lett.*, 2012, **109**, 187604.
- 102 C. A. Randall, A. Kelnberger, G. Y. Yang, R. E. Eitel and T. R. Shrout, *J. Electroceram.*, 2005, **14**, 177.
- 103 M. J. Chen, X. K. Ning, S. F. Wang and G. S. Fu, *RSC Adv.*, 2017, **7**, 38231.
- 104 D. Zhang, D. Sando, P. Sharma, X. Cheng, F. Ji, V. Govinden, M. Weyland, V. Nagarajan and J. Seidel, *Nat. Commun.*, 2020, **11**, 349.
- 105 P. C. Joshi and S. B. Krupanidhi, *Appl. Phys. Lett.*, 1993, **62**, 1928.
- 106 S. V. Kiselev, R. P. Ozerov and G. S. Zhdanov, *Sov. Phys. Dokl.*, 1963, **7**, 742.
- 107 Y. N. Venetsev, G. Zhdanov and S. Solov'ev, *Sov. Phys. Crystallogr.*, 1960, **4**, 538.
- 108 (a) A. Tsurumaki, H. Yamada and A. Sawa, *Adv. Funct. Mater.*, 2012, **22**, 1040; (b) A. Q. Jiang, C. Wang, K. J. Jin, X. B. Liu, J. F. Scott, C. S. Hwang, T. A. Tang, H. B. Lu and G. Z. Yang, *Adv. Mater.*, 2011, **23**, 1277.
- 109 R. U. Chandrasena, W. Yang, Q. Lei, M. U. Delgado-Jaime, K. D. Wijesekara, M. Golarikhani, B. A. Davidson, E. Arenholz, K. Kobayashi, M. Kobata, F. M. de Groot, U. Aschauer, N. A. Spaldin, X. Xi and A. X. Gray, *Nano Lett.*, 2017, **17**, 794.
- 110 A. L. Sangle, S. Singh, J. Jian, S. R. Bajpe, H. Wang, N. Khare and J. L. MacManus-Driscoll, *Nano Lett.*, 2016, **16**, 7338.
- 111 S. Singh, A. L. Sangle, T. Wu, N. Khare and J. L. MacManus-Driscoll, *ACS Appl. Mater. Interfaces*, 2019, **11**, 45683.
- 112 (a) S.-H. Kim, H. Choi, K. No and S.-C. Shin, *J. Phys. D: Appl. Phys.*, 2010, **43**; (b) E. M. Choi, E. Weal, Z. X. Bi, H. Y. Wang, A. Kursumovic, T. Fix, M. G. Blamire and J. L. MacManus-Driscoll, *Appl. Phys. Lett.*, 2013, **102**, 012905; (c) W. Zhang, A. Chen, J. Jian, Y. Zhu, L. Chen, P. Lu, Q. Jia, J. L. MacManus-Driscoll, X. Zhang and H. Wang, *Nanoscale*, 2015, **7**, 13808; (d) M. Fan, W. R. Zhang, J. Jian, J. J. Huang and H. Y. Wang, *APL Mater.*, 2016, **4**, 076105.
- 113 R. L. Stamps, S. Breikreutz, J. Akerman, A. V. Chumak, Y. Otani, G. E. W. Bauer, J. U. Thiele, M. Bowen, S. A. Majetich, M. Klaui, I. L. Prejbeanu, B. Dieny, N. M. Dempsey and B. Hillebrands, *J. Phys. D: Appl. Phys.*, 2014, **47**, 333001.
- 114 S. Sun, C. B. Murray, D. Weller, L. Folks and A. Moser, *Science*, 2000, **287**, 1989.
- 115 L. Shen, C. Ma, S. Cheng, S. Ren, S. Cheng, S. Mi and M. Liu, *J. Mater. Chem. C*, 2016, **4**, 10955.
- 116 A. Vailionis, H. Boschker, W. Siemons, E. P. Houwman, D. H. A. Blank, G. Rijnders and G. Koster, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **83**, 064101.
- 117 (a) C. N. Chinnasamy, A. Narayanasamy, N. Ponpandian, K. Chattopadhyay, H. Guerault and J. M. Greneche, *J. Phys.: Condens. Matter*, 2000, **12**, 7795; (b) F. S. Li, L. Wang, J. B. Wang, Q. G. Zhou, X. Z. Zhou, H. P. Kunkel and G. Williams, *J. Magn. Magn. Mater.*, 2004, **268**, 332; (c) J. T. Wu, N. Li, J. Xu, Y. Q. Jiang, Z. G. Ye, Z. X. Xie and L. S. Zheng, *Appl. Phys. Lett.*, 2011, **99**, 202505.
- 118 A. Suwardi, B. Prasad, S. Lee, E. M. Choi, P. Lu, W. Zhang, L. Li, M. Blamire, Q. Jia, H. Wang, K. Yao and J. L. MacManus-Driscoll, *Nanoscale*, 2016, **8**, 8083.
- 119 Y. Lin, E.-M. Choi, P. Lu, X. Sun, R. Wu, C. Yun, B. Zhu, H. Wang, W. Li, T. Maity and J. MacManus-Driscoll, *ACS Appl. Mater. Interfaces*, 2020, **12**, 8513.
- 120 Z. Wang, Y. Li, R. Viswan, B. Hu, V. G. Harris, J. Li and D. Viehland, *ACS Nano*, 2013, **7**, 3447.
- 121 X. Y. Gao, L. G. Li, J. Jian, J. J. Huang, X. Sun, D. Zhang and H. Y. Wang, *Appl. Phys. Lett.*, 2019, **115**, 053103.
- 122 J. Huang, Z. Qi, L. Li, H. Wang, S. Xue, B. Zhang, X. Zhang and H. Wang, *Nanoscale*, 2018, **10**, 17182.
- 123 B. Dieny and M. Chshiev, *Rev. Mod. Phys.*, 2017, **89**, 025008.
- 124 (a) S. Maat, K. Takano, S. S. Parkin and E. E. Fullerton, *Phys. Rev. Lett.*, 2001, **87**, 087202; (b) P. Borisov, A. Hochstrat, X. Chen, W. Kleemann and C. Binck, *Phys. Rev. Lett.*, 2005, **94**, 117203.
- 125 F. Radu, R. Abrudan, I. Radu, D. Schmitz and H. Zabel, *Nat. Commun.*, 2012, **3**, 715.
- 126 (a) R. Galceran, L. López-Mir, B. Bozzo, J. Cisneros-Fernández, J. Santiso, L. Balcels, C. Frontera and B. Martínez, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2016, **93**, 144417; (b) J. B. Fu, M. X. Hua, X. Wen, M. Z. Xue, S. L. Ding, M. Wang, P. Yu, S. Q. Liu, J. Z. Han, C. S. Wang, H. L. Du, Y. C. Yang and J. B. Yang, *Appl. Phys. Lett.*, 2017, **110**, 202403.
- 127 J. Zhang, Z. Zhong, X. Guan, X. Shen, J. Zhang, F. Han, H. Zhang, H. Zhang, X. Yan, Q. Zhang, L. Gu, F. Hu, R. Yu, B. Shen and J. Sun, *Nat. Commun.*, 2018, **9**, 1923.
- 128 (a) S. C. Liao, P. Y. Tsai, C. W. Liang, H. J. Liu, J. C. Yang, S. J. Lin, C. H. Lai and Y. H. Chu, *ACS Nano*, 2011, **5**, 4118; (b) H. Zheng, Q. Zhan, F. Zavaliche, M. Sherburne, F. Straub, M. P. Cruz, L. Q. Chen, U. Dahmen and R. Ramesh, *Nano Lett.*, 2006, **6**, 1401; (c) O. J. Lee, S. Misra, H. Wang and J. L. MacManus-Driscoll, *APL Mater.*, 2021, **9**, 030904.
- 129 W. Zhang, J. Jian, A. Chen, L. Jiao, F. Khatkhatay, L. Li, F. Chu, Q. Jia, J. L. MacManus-Driscoll and H. Wang, *Appl. Phys. Lett.*, 2014, **104**, 062402.



- 130 H. Zheng, J. Wang, S. E. Lofland, Z. Ma, L. Mohaddes-Ardabili, T. Zhao, L. Salamanca-Riba, S. R. Shinde, S. B. Ogale, F. Bai, D. Viehland, Y. Jia, D. G. Schlom, M. Wuttig, A. Roytburd and R. Ramesh, *Science*, 2004, **303**, 661.
- 131 S. Ojha, W. C. Nunes, N. M. Aimon and C. A. Ross, *ACS Nano*, 2016, **10**, 7657.
- 132 (a) L. Mohaddes-Ardabili, H. Zheng, S. B. Ogale, B. Hannoyer, W. Tian, J. Wang, S. E. Lofland, S. R. Shinde, T. Zhao, Y. Jia, L. Salamanca-Riba, D. G. Schlom, M. Wuttig and R. Ramesh, *Nat. Mater.*, 2004, **3**, 533; (b) J. Huang, L. Li, P. Lu, Z. Qi, X. Sun, X. Zhang and H. Wang, *Nanoscale*, 2017, **9**, 7970.
- 133 J. Huang, Z. Qi, L. Li, H. Wang, S. Xue, B. Zhang, X. Zhang and H. Wang, *Nanoscale*, 2018, **10**, 17182.
- 134 Q. Su, W. Zhang, P. Lu, S. Fang, F. Khatkhatay, J. Jian, L. Li, F. Chen, X. Zhang, J. L. MacManus-Driscoll, A. Chen, Q. Jia and H. Wang, *ACS Appl. Mater. Interfaces*, 2016, **8**, 20283.
- 135 (a) C. A. Vaz, J. Hoffman, C. H. Ahn and R. Ramesh, *Adv. Mater.*, 2010, **22**, 2900; (b) J. Ma, J. Hu, Z. Li and C. W. Nan, *Adv. Mater.*, 2011, **23**, 1062; (c) Y. Wang, J. M. Hu, Y. H. Lin and C. W. Nan, *NPG Asia Mater.*, 2010, **2**, 61; (d) R. Ramesh and S. Manipatruni, *Proc. R. Soc. A*, 2021, **477**, 20200942.
- 136 G. Dong, Z. Zhou, M. Guan, X. Xue, M. Chen, J. Ma, Z. Hu, W. Ren, Z. G. Ye, C. W. Nan and M. Liu, *ACS Nano*, 2018, **12**, 3751.
- 137 G. Dong, Z. Zhou, X. Xue, Y. Zhang, B. Peng, M. Guan, S. Zhao, Z. Hu, W. Ren, Z. G. Ye and M. Liu, *ACS Appl. Mater. Interfaces*, 2017, **9**, 30733.
- 138 (a) C. Schmitz-Antoniak, D. Schmitz, P. Borisov, F. M. de Groot, S. Stienen, A. Warland, B. Krumme, R. Feyerherm, E. Dudzik, W. Kleemann and H. Wende, *Nat. Commun.*, 2013, **4**, 2051; (b) A. Chen, Y. Dai, A. Eshghinejad, Z. Liu, Z. Wang, J. Bowlan, E. Knall, L. Civale, J. L. MacManus-Driscoll, A. J. Taylor, R. P. Prasankumar, T. Lookman, J. Li, D. Yarotski and Q. Jia, *Adv. Sci.*, 2019, **6**, 1901000.
- 139 Z. H. Mo, G. Tian, W. D. Yang, S. Ning, C. A. Ross, X. S. Gao and J. M. Liu, *APL Mater.*, 2021, **9**, 041109.
- 140 F. Zavaliche, H. Zheng, L. Mohaddes-Ardabili, S. Y. Yang, Q. Zhan, P. Shafer, E. Reilly, R. Chopdekar, Y. Jia, P. Wright, D. G. Schlom, Y. Suzuki and R. Ramesh, *Nano Lett.*, 2005, **5**, 1793.
- 141 E. Strelcov, A. Belianinov, Y. H. Hsieh, Y. H. Chu and S. V. Kalinin, *Nano Lett.*, 2015, **15**, 6650.
- 142 T. Fix, E. M. Choi, J. W. Robinson, S. B. Lee, A. Chen, B. Prasad, H. Wang, M. G. Blamire and J. L. Macmanus-Driscoll, *Nano Lett.*, 2013, **13**, 5886.
- 143 R. Wu, A. Kursumovic, X. Gao, C. Yun, M. E. Vickers, H. Wang, S. Cho and J. L. MacManus-Driscoll, *ACS Appl. Mater. Interfaces*, 2018, **10**, 18237.
- 144 I. K. S. J. Nogue, *J. Magn. Magn. Mater.*, 2002, **192**, 203.
- 145 T. Jungwirth, J. Sinova, A. Manchon, X. Marti, J. Wunderlich and C. Felser, *Nat. Phys.*, 2018, **14**, 200.
- 146 R. Wu and J. L. MacManus-Driscoll, *APL Mater.*, 2022, **10**.
- 147 (a) R. Uecker, R. Bertram, M. Brutzam, Z. Galazka, T. M. Gesing, C. Gugushev, D. Klimm, M. Klupsch, A. Kwasniewski and D. G. Schlom, *J. Cryst. Growth*, 2017, **457**, 137; (b) C. Gugushev, D. Klimm, M. Brutzam, T. M. Gesing, M. Gogolin, H. Paik, T. Markurt, D. J. Kok, A. Kwasniewski, U. Jendritzki and D. G. Schlom, *J. Cryst. Growth*, 2020, **536**, 125526.
- 148 J. H. Song, T. Susaki and H. Y. Hwang, *Adv. Mater.*, 2008, **20**, 2528.
- 149 M. Lippmaa, N. Nakagawa, M. Kawasaki, S. Ohashi and H. Koinuma, *J. Electroceram.*, 2000, **4**, 365.
- 150 H. H. Sonstebly, E. Skaar, O. S. Fjellvag, J. E. Bratvold, H. Fjellvag and O. Nilsen, *Nat. Commun.*, 2020, **11**, 2872.
- 151 (a) J. Huang, H. Wang, X. Wang, X. Gao, J. Liu and H. Wang, *ACS Appl. Mater. Interfaces*, 2020, **12**, 39920; (b) T. Amrillah, Y. Bitla, K. Shin, T. Yang, Y. H. Hsieh, Y. Y. Chiou, H. J. Liu, T. H. Do, D. Su, Y. C. Chen, S. U. Jen, L. Q. Chen, K. H. Kim, J. Y. Juang and Y. H. Chu, *ACS Nano*, 2017, **11**, 6122.
- 152 (a) B. Peng, R. C. Peng, Y. Q. Zhang, G. Dong, Z. Zhou, Y. Zhou, T. Li, Z. Liu, Z. Luo, S. Wang, Y. Xia, R. Qiu, X. Cheng, F. Xue, Z. Hu, W. Ren, Z. G. Ye, L. Q. Chen, Z. Shan, T. Min and M. Liu, *Sci. Adv.*, 2020, **6**, eaba5847; (b) D. Pesquera, E. Khestanova, M. Ghidini, S. Zhang, A. P. Rooney, F. Maccherozzi, P. Riego, S. Farokhipoor, J. Kim, X. Moya, M. E. Vickers, N. A. Stelmashenko, S. J. Haigh, S. S. Dhesi and N. D. Mathur, *Nat. Commun.*, 2020, **11**, 3190.
- 153 (a) H.-J. Liu, V.-T. Tra, Y.-J. Chen, R. Huang, C.-G. Duan, Y.-H. Hsieh, H.-J. Lin, J.-Y. Lin, C.-T. Chen, Y. Ikuhara and Y.-H. Chu, *Adv. Mater.*, 2013, **25**, 4753; (b) W. I. Liang, Y. Liu, S. C. Liao, W. C. Wang, H. J. Liu, H. J. Lin, C. T. Chen, C. H. Lai, A. Borisevich, E. Arenholz, J. Li and Y. H. Chu, *J. Mater. Chem. C*, 2014, **2**, 811.
- 154 (a) A. P. Chen, Z. Harrell, P. Lu, E. Enriquez, L. G. Li, B. Zhang, P. Dowden, C. L. Chen, H. Y. Wang, J. L. MacManus-Driscoll and Q. X. Jia, *Adv. Funct. Mater.*, 2019, **29**; (b) Y. Li, Z. Wang, J. Yao, T. Yang, Z. Wang, J. M. Hu, C. Chen, R. Sun, Z. Tian, J. Li, L. Q. Chen and D. Viehland, *Nat. Commun.*, 2015, **6**, 6680.
- 155 (a) T. C. Kim, S. Ojha, G. Tian, S. H. Lee, H. K. Jung, J. W. Choi, L. Kornblum, F. J. Walker, C. H. Ahn, C. A. Ross and D. H. Kim, *J. Mater. Chem. C*, 2018, **6**, 5552; (b) T. C. Kim, S. H. Lee, H. K. Jung, Y. E. Kim, J. W. Choi, D. Yang and D. H. Kim, *J. Magn. Magn. Mater.*, 2019, **471**, 116; (c) S. H. Lee, G. Tian, T. C. Kim, H. K. Jung, J. W. Choi, F. J. Walker, C. H. Ahn, C. A. Ross and D. H. Kim, *Nanotechnology*, 2019, **30**, 105601; (d) D. H. Kim, S. Ning and C. A. Ross, *J. Mater. Chem. C*, 2019, **7**, 9128; (e) T. C. Kim, S. H. Lee, H. K. Jung, H. Lee, J. H. Mun, I. Oh and D. H. Kim, *Ceram. Int.*, 2019, **45**, 12182.
- 156 (a) H. Luo, H. Yang, S. A. Baily, O. Ugurlu, M. Jain, M. E. Hawley, T. M. McCleskey, A. K. Burrell, E. Bauer, L. Civale, T. G. Holesinger and Q. Jia, *J. Am. Chem. Soc.*, 2007, **129**, 14132; (b) S. M. Baber, Q. L. Lin, G. F. Zou, N. Haberkorn, S. A. Baily, H. Y. Wang, Z. X. Bi, H. Yang, S. G. Deng, M. E. Hawley, L. Civale, E. Bauer, T. M. McCleskey, A. K. Burrell, Q. X. Jia and H. M. Luo,



- J. Phys. Chem. C*, 2011, **115**, 25338; (c) M. Saito, M. Hagiwara and S. Fujihara, *Mater. Chem. Phys.*, 2019, **236**, 121762.
- 157 (a) R. Comes, H. Liu, M. Khokhlov, R. Kasica, J. Lu and S. A. Wolf, *Nano Lett.*, 2012, **12**, 2367; (b) S. M. Stratulat, X. Lu, A. Morelli, D. Hesse, W. Erfurth and M. Alexe, *Nano Lett.*, 2013, **13**, 3884; (c) N. M. Aimon, H. K. Choi, X. Y. Sun, D. H. Kim and C. A. Ross, *Adv. Mater.*, 2014, **26**, 3063; (d) M. Faustini, *Nat. Mater.*, 2021, **20**, 8.
- 158 W. Zhang, A. Chen, F. Khatkhatay, C. F. Tsai, Q. Su, L. Jiao, X. Zhang and H. Wang, *ACS Appl. Mater. Interfaces*, 2013, **5**, 3995.
- 159 J. C. Liu, X. J. Wang, X. Y. Gao, H. Wang, J. Jian, J. J. Huang, X. Sun, Z. M. Qi, S. Misra, Z. H. He and H. Y. Wang, *Appl. Mater. Today*, 2020, **21**, 100856.
- 160 N. M. Aimon, D. H. Kim, X. Sun and C. A. Ross, *ACS Appl. Mater. Interfaces*, 2015, **7**, 2263.

