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## Nano-silicon/reduced graphene oxide composite anodes for high performance all solid-state batteries†

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**Nano-Si (n-Si) encapsulated with  $\text{SiO}_x$  shells and anchored onto reduced graphene oxide (rGO) via hydrothermal self-assembly is demonstrated as a promising solid-state battery anode. Compared to simple n-Si, this composite anode exhibited improved rate capability and cycle life, enabled by robust Si–O–C bonding, mechanical reinforcement, and rapid electron transport.**

Sulfide-based all solid-state batteries (SSBs) have emerged as prominent next-generation energy storage devices due to their potential for enhanced safety and energy density compared to conventional lithium-ion batteries.<sup>1,2</sup> Despite significant progress, sulfide-based SSBs still face critical challenges at the anode/solid electrolyte (SE) interface, especially when using Si anodes.<sup>3</sup> While Si offers high theoretical capacity (4200 mAh g<sup>-1</sup> for Li<sub>4.4</sub>Si) compared to traditional graphite anodes (372 mAh g<sup>-1</sup>), its practical application has been hindered by electro-chemo-mechanical instabilities.<sup>4</sup> Large volume expansions (>300%) of the Si anode during cycling induce severe mechanical stresses at the anode/SE interface, leading to rapid capacity fading, SE cracking, and premature cell failure.<sup>5</sup> Resolving this interfacial instability is critical, as the adoption of Si anodes for SSBs enables substantial improvements in energy density, driving range, fast-charging capability, and overall safety, which are key factors for the widespread adoption of electric vehicles (EVs).

Addressing the electro-chemo-mechanical instability of Si in SSBs remains challenging due to the complex interactions at the nanoscale interfaces, a limited understanding of the underlying degradation mechanisms, and the lack of effective *in situ*

characterization tools.<sup>6</sup> The most common approach has been replacing micron-sized Si particles with nano-sized Si (n-Si) particles, which can alleviate stress accumulation and mitigate subsequent mechanical degradation.<sup>7</sup> However, the current approach to stabilize n-Si in sulfide-based SSBs remains insufficient. A major issue lies in the intrinsically low electronic conductivity and limited Li<sup>+</sup> diffusivity of n-Si based anodes, which further exacerbate their electro-chemo-mechanical instability.

To solve the Si/SE interfacial problems, we introduced n-Si particles coated with a conformal SiO<sub>x</sub> shell, which are chemically anchored onto reduced graphene oxide (rGO) nanoplatelets. While the enhanced performance of this chemically modified anode (hereafter referred to as n-Si/G) has been demonstrated in Li-ion batteries,<sup>8–10</sup> its impact in sulfide-based SSBs has yet to be fully explored. In this work, n-Si/G powder samples were prepared *via* one-pot hydrothermal reaction. The SiO<sub>x</sub> shell offers moderate Li-ion conductivity (~1  $\mu\text{S cm}^{-1}$ )<sup>11</sup> due to its amorphous structure and serves as a chemically and mechanically stable interfacial layer that mitigates direct contact between the Si core and the argyrodite-type Li<sub>6</sub>PS<sub>5</sub>Cl<sub>0.5</sub>Br<sub>0.5</sub> SE.<sup>10,12</sup> The covalent Si–O–C bonding between the Si@SiO<sub>x</sub> nanoparticles and rGO nanoplatelets ensures strong interfacial adhesion, enhancing structural integrity and effectively accommodating volumetric changes during cycling.<sup>13</sup>

Fig. 1a displays the uniform dispersion of n-Si particles onto the rGO supports, as observed by scanning electron microscopy (SEM). Its transmission electron microscopy (TEM) image (Fig. 1b) reveals an ~3 nm thick SiO<sub>x</sub> shell surrounding the n-Si particles. High-angle annular dark field scanning transmission electron microscopy (HAADF-STEM) images (Fig. 1c and 1d) reveal n-Si particles, with diameters ranging from 30 to 50 nm, anchored on the rGO layers. Fig. 1c and e–g show energy-dispersive X-ray spectroscopy (EDS) mapping images displaying the elemental distribution of Si, C, and O on the surface of the n-Si/G particles. The SiO<sub>x</sub> nanoparticles are uniformly distributed on the rGO layers.

X-ray photoelectron spectroscopy (XPS) was performed to analyze the chemical states of the n-Si/G composite. As shown in Fig. 1h, the deconvoluted C 1s spectrum of the n-Si/G

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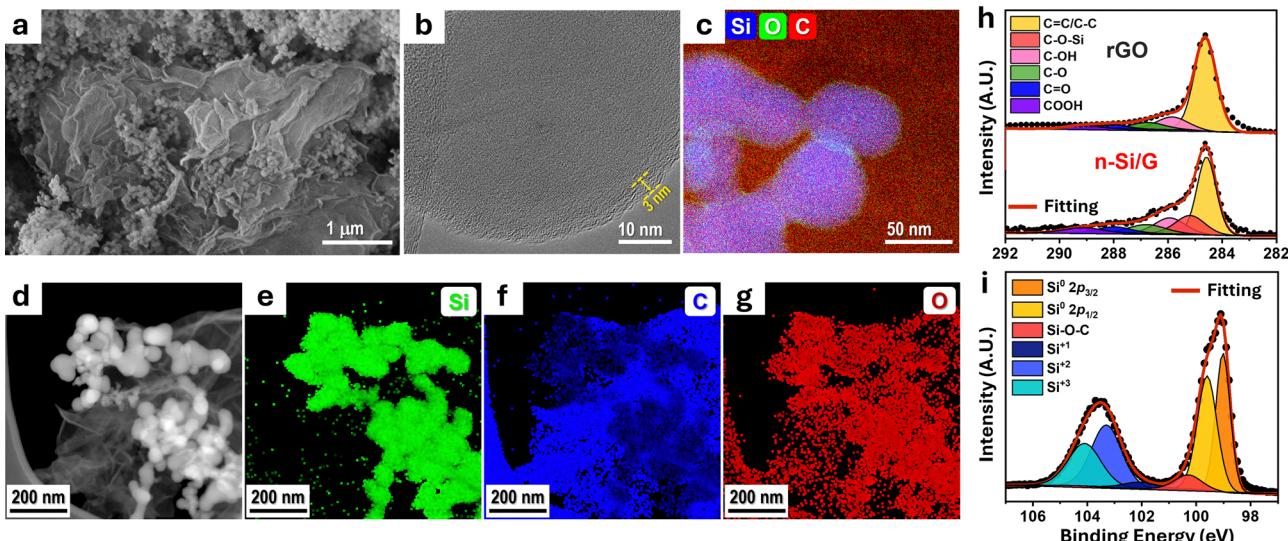


Fig. 1 (a) SEM, (b) TEM, and (c)–(g) STEM-EDS images of the n-Si/G samples. The TEM image focused on a single n-Si particle within the n-Si/G. (h) C 1s XPS spectra of the rGO and n-Si/G sample. (i) Si 2p XPS spectra of the n-Si/G sample (circular dots: experimental data, solid lines: fitting curves).

composite exhibits six peaks at 284.6, 285.2, 285.9, 286.7, 288.0 and 289.1 eV, corresponding to C=C/C-C, C-O-Si, C-OH, C-O, C=O, and COOH bonds, respectively (Fig. 1h). The peak at 285.2 eV is attributed to the new formation of Si-O-C bonds arising from the chemical bonding between the  $\text{SiO}_x$  and rGO layers. Meanwhile, the intensity of the C=C/C-C peak, originating from the rGO, decreased as a result of bonding with  $\text{SiO}_x$  in the n-Si/G sample. In the deconvoluted Si 2p spectrum of the n-Si/G composite (Fig. 1i), distinct peaks were observed corresponding to metallic silicon: Si 2p<sub>3/2</sub> at 99.0 eV and Si 2p<sub>1/2</sub> at 99.6 eV. In addition, a peak at 100.3 eV is attributed to Si-O-C bonding, whereas peaks at 102.0, 103.3, and 104.0 eV correspond to oxidized Si species with oxidation states of Si<sup>1+</sup>, Si<sup>2+</sup>, and Si<sup>3+</sup>, respectively, further confirming the presence of the  $\text{SiO}_x$  outer shell. This result is consistent with the TEM data (Fig. 1b and c), which showed  $\text{SiO}_x$  shells surrounding the n-Si particles, while metallic Si remains predominantly confined to the particle cores.

The electrochemical performance of the n-Si/G anodes was evaluated using SSB full-cells. For cathodes, single-crystal  $\text{LiNi}_{0.6}\text{Mn}_{0.2}\text{Co}_{0.2}\text{O}_2$  (NMC622) was coated with 1 wt%  $\text{LiNbO}_3$  (LNO) to passivate the cathode/SE interface.<sup>1</sup> In this work, all the SSB cells were cycled under a constant stack pressure of 5 MPa. Fig. 2a presents the long-term cycle life of SSB full-cells at room temperature (RT). The full cell with the n-Si/G anode delivered an initial discharge capacity of 169  $\text{mAh g}^{-1}$  at C/10, with a 1st cycle coulombic efficiency (CE) of 75%, compared to 172  $\text{mAh g}^{-1}$  and 71% for the full cell with the simple n-Si anode. Between the 4th and 50th cycles at C/5, the n-Si/G anode offered improved capacity retention (93%) relative to the n-Si anode (87%). Fig. 2b and c shows the voltage profiles of the full-cells at the 1st, 5th, and 50th cycles. The simple n-Si anode showed larger voltage drops ( $\Delta V = I \cdot \Delta R$ ) with cycle number, indicating greater cell-resistance growth ( $\Delta R$ ) compared to the n-Si/G anode. This could be due to the possible microstructural degradation of the n-Si anodes.

The impact of n-Si/G anodes on the rate capability of SSB full-cells was examined across various C-rates ranging from C/10 to 1C at RT (Fig. 2d). The full-cell with the simple n-Si anode had 169  $\text{mAh g}^{-1}$  at C/10 (5th cycle), 144  $\text{mAh g}^{-1}$  at C/5 (10th cycle), 122  $\text{mAh g}^{-1}$  at C/3 (15th cycle), 99  $\text{mAh g}^{-1}$  at C/2 (20th cycle), 67  $\text{mAh g}^{-1}$  at 1C (25th cycle), and 151  $\text{mAh g}^{-1}$  at C/10 (30th cycle). The cell exhibited an overall capacity retention of 86% at C/10 after 30 accelerated rate tests. In contrast, the full cell with the n-Si/G anode exhibited significantly improved rate capability, achieving an overall capacity retention of 96% at C/10 after 30 accelerated rate tests. For example, it delivered 165  $\text{mAh g}^{-1}$  at C/10 (5th cycle), 149  $\text{mAh g}^{-1}$  at C/5 (10th cycle), 132  $\text{mAh g}^{-1}$  at C/3 (15th cycle), 116  $\text{mAh g}^{-1}$  at C/2 (20th cycle), 97  $\text{mAh g}^{-1}$  at 1C (25th cycle), and 161  $\text{mAh g}^{-1}$  at C/10 (30th cycle). The n-Si/G anode exhibited a 44.8% increase in capacity at 1C charging compared to the n-Si anode.

To elucidate the superior rate capability and capacity retention of the n-Si/G anode relative to the simple n-Si anode, electrochemical impedance spectroscopy (EIS) combined with distribution of relaxation times (DRT) analysis was performed on the full-cells. Fig. 3a shows the Nyquist plots of SSB full-cells with simple n-Si and n-Si/G anodes, measured at 50% state-of-charge (SOC) and RT over a frequency range of 5 MHz and 50 mHz. Fig. 3b and c illustrate their corresponding DRT profiles for each cell. The resulting Nyquist plots were further processed via the distribution of relaxation time (DRT) technique MATLAB code developed by Wan *et al.*<sup>14</sup> All subsequent analyses derived from the Nyquist and DRT plots were based on interpretations established in prior studies of SSBs.<sup>15–17</sup> The high frequency intercept ( $f > 3$  MHz) in the Nyquist plot corresponds to the bulk resistance ( $R_B$ ) of the SE. The first semi-circle, corresponding to the P1 peak in the DRT profile at  $\sim 0.5$  MHz is attributed to the grain boundary resistance ( $R_{GB}$ ). The second large semi-circles (between 200 kHz and 1 Hz) in the Nyquist plots were deconvoluted using DRT into contributions from chemical contact resistance ( $R_C$ ) for P2, and

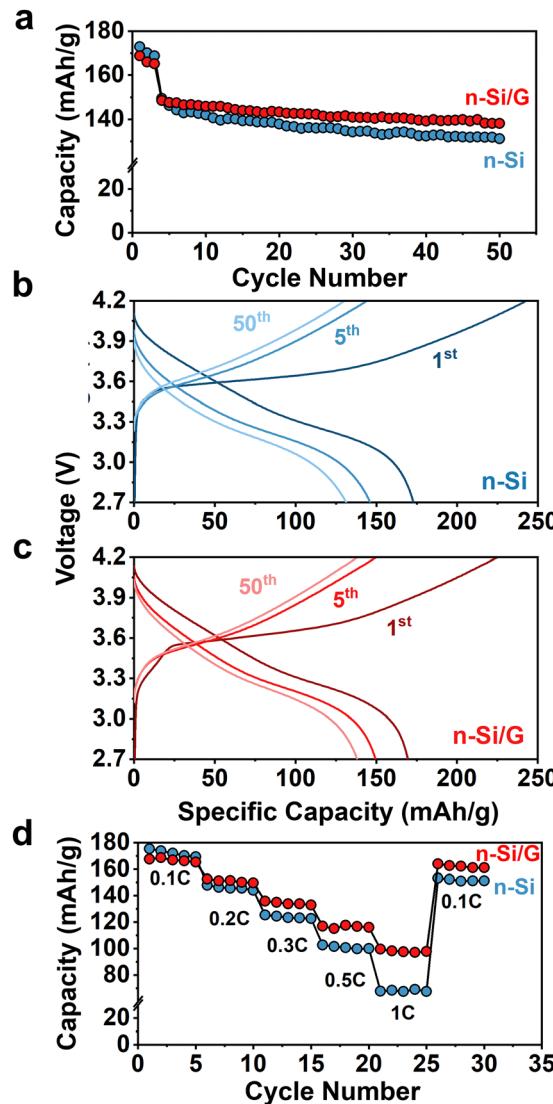


Fig. 2 (a) Cycle life and voltage profiles of (b) simple n-Si and (c) n-Si/G anodes in SSB full-cells paired with NMC622 cathodes. (d) Rate capability test of the full cells at RT, where both charging and discharging C-rates were varied from 0.1C to 1C.

mechanical contact resistance ( $R_M$ ) for P3 at the electrode/SE interface. The P4 peak group, which arises between 25 and 0.1 Hz, corresponds to the charge transfer resistance ( $R_{CT}$ ) from both the cathode and anode. Finally, the Warburg ( $R_W$ ) impedances observed in the Nyquist plots at  $f < 10$  mHz correspond to the P5 peaks in the DRT profiles, which is attributed to solid-state  $\text{Li}^+$  diffusion.

Fig. 3d compares the individual contribution and evolution of resistance from different sources in the SSB full-cells over 50 cycles for the simple n-Si and n-Si/G anodes. Overall, all full-cells exhibited relatively larger  $R_M$  and  $R_{CT}$  components compared to other resistance contributions, indicating that the electrode/SE interface is the dominant source of cell resistance. After 50 cycles, the n-Si/G anode exhibited a lower total resistance increase of 24%, compared to 37% for the simple n-Si anode. Notably, the full-cell with the n-Si/G anode showed smaller increases of 18% in  $R_M$  and 52% in  $R_{CT}$ , compared to 28% in  $R_M$  and 66% in  $R_{CT}$  for

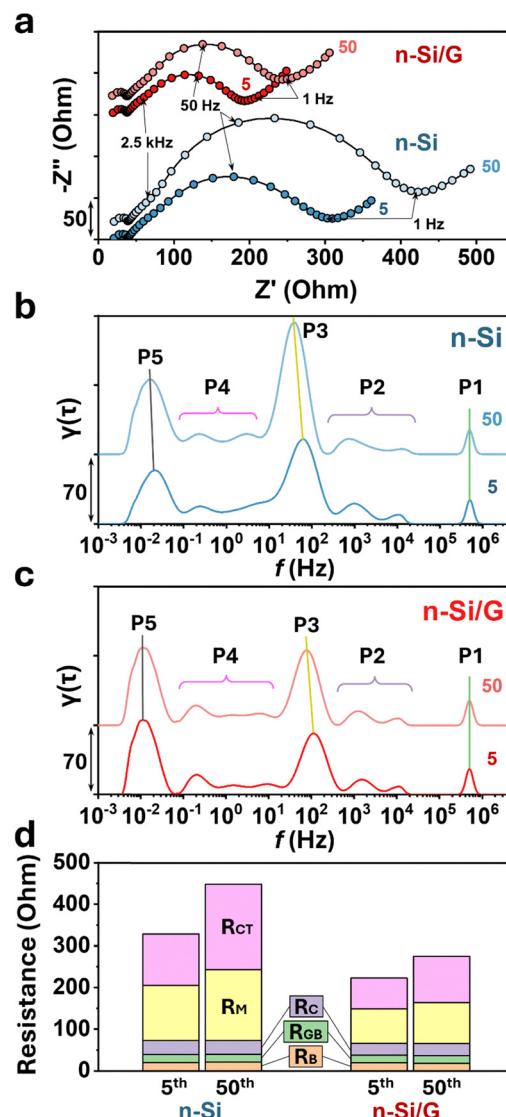


Fig. 3 (a) Nyquist plots (5<sup>th</sup> vs. 50<sup>th</sup> cycles) and corresponding DRT profiles of (b) simple n-Si and (c) n-Si/G anodes in SSB full-cells paired with NMC622 cathodes at 50% SOC and RT. (d) Individual contribution from different resistance sources in SSB full-cells and their evolution over 50 cycles, including bulk ( $R_B$ ), grain boundary ( $R_{GB}$ ), chemical contact ( $R_C$ ), mechanical contact ( $R_M$ ), and charge-transfer ( $R_{CT}$ ) resistances.

the simple n-Si anode. In contrast, the  $R_B$ ,  $R_{GB}$ , and  $R_C$  values remained stable over 50 cycles. It is important to note that the SSB full-cells in this work employed an  $\sim 700$   $\mu\text{m}$  thick SE layer, which negatively impacts the overall rate capabilities (Fig. 2d). However, the resistance contributions from the thick SE (*i.e.*,  $R_B + R_{GB}$ ) are substantially lower than other sources (*e.g.*,  $R_M$  and  $R_{CT}$ ), highlighting the superior interfacial stability between the n-Si/G anode and SE, relative to that of the simple n-Si anode.

The microstructures of the cycle-aged anodes in the SSB full-cells are shown in Fig. 4. After 50 cycles, the simple n-Si anodes exhibited large voids, up to 1  $\mu\text{m}$  in size, resulting from the substantial volumetric changes of the n-Si particles. These voids degrade the contact between the n-Si and SE, leading to an increase in  $R_M$ , and subsequently hindering the charge-transfer reactions (*i.e.*,

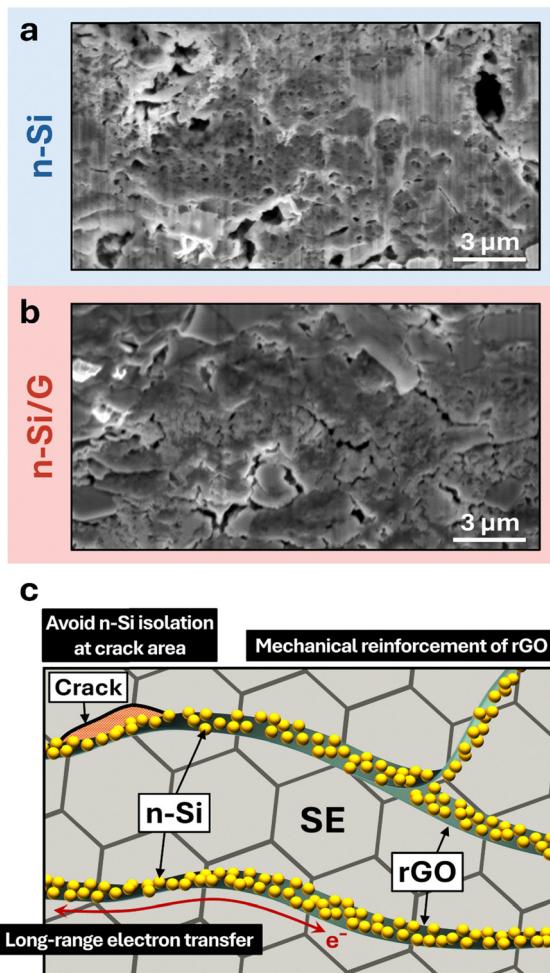


Fig. 4 FIB-SEM cross-sectional images of cycle-aged (a) simple n-Si and (b) n-Si/G anodes, recovered after 50 cycles. (c) Schematic illustration of the improvement mechanisms of n-Si/G anodes in SSB cells.

increase in  $R_{CT}$ ). In stark contrast, the cycle-aged n-Si/G anode showed voids with reduced scales compared to the simple n-Si anode. rGO and graphene have been widely incorporated into various ceramic matrices to improve the mechanical properties of ceramic/rGO composites, such as fracture toughness, strength, and structural integrity.<sup>18</sup> In the n-Si/G composite anodes, rGO nanoplatelets serve as a flexible and robust mechanical scaffold, effectively accommodating the large volumetric changes of n-Si during cycling. This structural reinforcement suppresses interfacial degradation (e.g., large pore formation), thereby preserving the mechanical integrity of the n-Si/G anodes during cycling. Moreover, even in the presence of small cracks, the extended lateral size of the rGO ( $>1\ \mu\text{m}$ ) provides a continuous electron conduction network, preventing electrical isolation of the n-Si particles anchored onto the rGO.

In conclusion, this work demonstrates that the n-Si/G composite can serve as a promising anode material for SSB full-cells. Electrochemical and microscopy analyses suggest that the enhanced rate capability and cycle life, with a capacity retention of 93%, in full-cells with n-Si/G anodes can be attributed to robust interfacial contact between n-Si and rGO through Si–O–C bonding, mechanical reinforcement of the n-Si/G + SE composite structure, and stable long-range

electron transport facilitated by the rGO network. The EIS and DRT results suggest that resistance due to mechanical contact at the electrode/SE interface and charge transfer were the major components contributing towards the performance degradation. The improvement mechanism of the n-Si/G anodes is illustrated in Fig. 4c. Our results highlight the potential of n-Si/G as a promising anode material for high performance SSBs. Future research should aim to optimize the microstructure and electro-chemo-mechanical stability of the n-Si/G anodes, while also advancing scalable processing methods compatible with solid-state battery (SSB) manufacturing. In parallel, enabling stable operation under reduced external pressure could simplify cell design and lower system-level costs, thereby facilitating the broader commercialization of SSBs incorporating n-Si/G anodes.

## Conflicts of interest

There are no conflicts to declare.

## Data availability

The experimental methods for data collection are provided in the ESI.<sup>†</sup>

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