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Metal-phenolic networks as tuneable spore coat mimetics†

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Bacillus subtilis are important probiotic microbes currently formulated for delivery as spores, but their ability to germinate in the gut remains debatable. To optimize their application, cells should be delivered in their vegetative state, but the sensitivity of *B. subtilis* prevents this. Through the application of self-assembled metal-phenolic network (MPN) cellular coatings, *B. subtilis* are protected from lyophilization stresses. These MPNs are an important class of self-assembled materials comprised of polyphenols and metal ions, and the efficacy of MPN protection was found to be dependent on the MPN components used for assembly. Both the size of the polyphenol and stability of the metal-phenol coordination were important factors that influenced their cellular protection; the smallest polyphenol, gallic acid, and the most stable chelated ion, Fe^{III}, were found to provide the highest level of protection. Further, delivery to the gut involves exposure to acidic conditions in the form of stomach acid and intestinal fluid. MPN coatings rapidly disassemble upon mild acid treatment but were found to protect *B. subtilis* from the negative impacts of the acid. Overall, optimized MPNs were found to protect vegetative *B. subtilis* cells from lyophilization stress and enable a more complete understanding of the role of each component in MPNs.

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Introduction

Metal-phenolic networks (MPNs) are a promising class of materials for a variety of human health applications, ranging from contrast agents and drug delivery vehicles to sources of both prokaryotic and eukaryotic cell protection.^{1–3} Broadly, these materials are comprised of metal ions and polyphenols that self-assemble through coordination-based interactions. The diversity of potential applications is tied to the ease of formation of MPNs, allowing for convenient and simple addition to a variety of substrates.^{4–7}

Specifically, there has recently been growing interest in the protection of beneficial microbes as both biotherapeutics and agricultural additives.^{8,9} However, in many cases, non-model microbes remain challenging to produce at scale due to processing stresses. Traditional lyophilization methods can lead to less than 0.1% cell survival for delicate microbes, many of which are promising candidates for biotherapeutics.^{10,11} We recently reported the protection of *Bacterioides thetaiotaomicron*, an important potential living biotherapeutic, from lyophilization and subsequent aerobic exposure using iron-containing MPNs.¹² Others have reported the protection of live *Escherichia coli* from UV exposure following MPN coating.¹³ These results are encouraging for MPN suitability as a general protective coating for microbes. The achievement of relatively high post-processing viability will allow the development and stable manufacturing of exciting new microorganisms.¹⁴



MPNs can be generated from many different polyphenols and metal ions, but the majority of work with these systems has centered on the application of tannic acid (TA) and iron as MPN components.² Yet, MPNs derived from alternative polyphenols and metal ions have significantly variable physicochemical properties that could be beneficial for different applications and may offer different protective properties based on the microbial strain employed. Thus, when designing these systems for specific strains and production stressors, the physical characteristics of the MPNs are important to consider. The most common mechanical metric reported is the Young's modulus, generally acquired through atomic force microscopy (AFM). The Young's modulus has been found to heavily depend on the molecular weight of the polyphenols and the pH of the buffer used; alkaline buffers and polyphenols with lower molecular weights lead to higher Young's modulus.¹⁵ MPN network size has additionally been shown to be dependent on the ionic strength of the assembly solution: increasing ionic strength results in thicker and rougher iron-TA films, measured by scanning electron microscopy (SEM).¹⁶ This controlled variation in the physical properties of MPNs based on variation of the metal ion and assembly solution is anticipated to be critical for their efficacy in microbial protection.

Our initial studies centered on the gram-negative, facultative anaerobe *B. thetaiotaomicron* that is challenging to manufacture.^{17,18} We reported the evaluation of three polyphenols (tannic acid (TA), epigallocatechin gallate (EGCG), and gallic acid (GA)) combined with Fe^{III} to generate MPNs on living microbes to protect them from harsh processing stressors, including oxygen exposure and lyophilization.¹² Our results demonstrated that the components used to form MPNs impacted the resultant protection afforded by the coatings. *Bacillus subtilis*, in contrast, are gram-positive and spore forming, enabling their production and delivery through sporulation.¹⁹

The spores of these microbes are highly tolerant to environmental stressors, but spores are dormant. Thus, there has been significant debate over the efficacy of spore-based probiotics including *B. subtilis* because germination must occur in the gastrointestinal (GI) tract for most host benefits. Though *Bacillus* spores do yield a host immune response, efficacy as a probiotic is linked to their germination and conversion to active, vegetative cells.²⁰ Studies have shown *Bacillus* strains to be capable of germination in the human gut, but the sample size was small and the data is recent.²¹ Thus, we sought to employ MPN coatings as "synthetic spore coats" on non-sporulated cells to enable their rapid recovery in the GI tract. The importance of a healthy gut microbiome and the negative health effects of an unbalanced microbiome are well known, necessitating improved probiotics for more effective gut implantation.¹⁴

Surprisingly, we observed that for *B. subtilis* preservation, the canonical MPN comprised of TA and Fe^{III} was not the optimal coating. Thus, we evaluated the impact of varying the polyphenol and the metal ion on the cellular protection effects of MPNs and developed an optimized coating recipe to protect

B. subtilis from the freeze-drying process. We demonstrate that MPNs are readily self-assembled on silica bead substrates using different metal ions (Fe^{III}, Mn^{II}, Zn^{II}, and Al^{III}) and phenols (TA, EGCG and GA) for spectroscopic characterization. The coating process was found to be biocompatible and showed negligible adverse effects on microbial viability. Furthermore, Fe^{III}-TA- and Fe^{III}-GA-coated *B. subtilis* in cryoprotectant-free buffer showed faster recovery following lyophilization. Remarkably, after a simple L-ascorbic acid treatment (which mimics the pH of the GI tract), MPN-coated *B. subtilis* reached exponential phase after incubation for 16 h, whereas no growth was observed for the uncoated control. Overall, our results demonstrate that MPN coatings protect different microbes from lyophilization stresses, but the components of these coatings must be optimized based on the strain of interest. Specifically, we have shown the protection of non-sporulated *B. subtilis* during processing with MPN components different from the conventional composition as well as the ready subsequent removal of the coating in a GI environment.

Results and discussion

Spectroscopic characterization of MPN assemblies

The radii of metal ions, size of phenols, and chelation strength have all been reported to contribute to the difference in stiffness of MPNs.¹⁵ Thus, by simply mixing a series of metal ions and polyphenols, we could gain access to a library of MPN coatings with tunable mechanical properties.^{16,22} Biocompatible and biorelevant metal ions (Fe^{III}, Mn^{II}, Zn^{II}, and Al^{III}) and phenols (TA, EGCG, and GA) were selected for this evaluation (Fig. 1). In addition to the structure of the phenols and the valency of the metal ions, the pH of the coordination solution is also important for MPN formation, as alkaline conditions are necessary for effective chelation between the metal ion and hydroxyl groups.³ To evaluate MPN formation using various metals and polyphenols, solutions containing each were mixed. An equal volume of 3-morpholinopropane-1-sulfonic acid (MOPS) buffer (20 mM, pH 7.5) was added to the pre-complexed MPN solution, with a final MOPS concentration of 10 mM when in solution with the MPNs. Interestingly, upon addition of the buffer, a weak ligand-to-metal charge transfer (LMCT) band between 400 nm to 700 nm was observed in the UV-Vis spectrum, indicating the formation of the MPN complex (Fig. 1). These complexes are known to form under a variety of conditions and on diverse substrates due simply to the presence of a nucleation site and the correct pH.^{16,23-25}

To evaluate the impact of micron-scale substrates for MPN formation, we evaluated the encapsulation of negatively-charged, micron-scale silica particles. Excitingly, upon addition of silica to the MPN solution, followed by pH adjustment, stronger absorbance bands were observed, indicating the deposition of MPNs on silica particles (Fig. 1). Some metal-phenol pairs (Fe^{III}-TA, Fe^{III}-EGCG, Fe^{III}-GA, Al^{III}-EGCG and Zn^{II}-EGCG) yielded strong LMCT bands, while others (Zn^{II}-TA, Zn^{II}-GA, Mn^{II}-GA) featured weaker bands, highlighting the



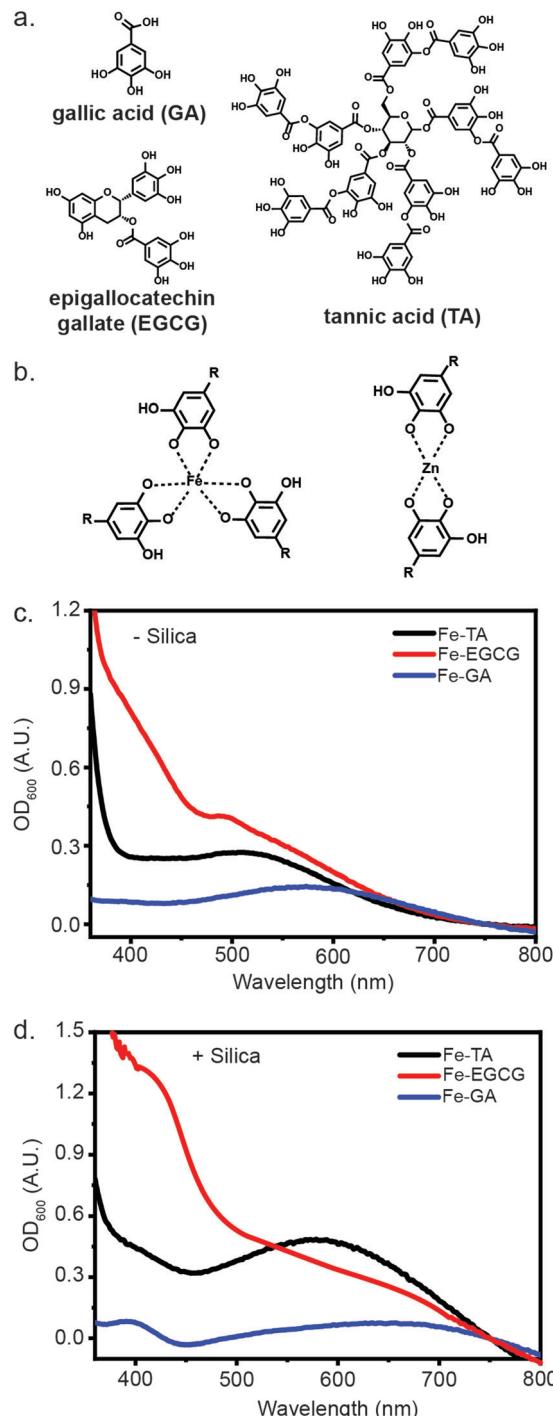


Fig. 1 Abiotic characterization of metal-phenolic networks (MPNs). (a) Polyphenol components used in the MPN assemblies. (b) Example chelation of metal ions (Fe^{III} and Zn^{II}) to polyphenols. (c) UV-Vis absorbance of $\text{Fe}^{\text{III}}\text{-TA}$, $\text{Fe}^{\text{III}}\text{-GA}$, and $\text{Fe}^{\text{III}}\text{-EGCG}$ complexes in MOPS buffer (10 mM, pH 7.5). (d) $\text{Fe}^{\text{III}}\text{-TA}$, $\text{Fe}^{\text{III}}\text{-GA}$, and $\text{Fe}^{\text{III}}\text{-EGCG}$ complexes formed on the surface of silica particles (1 mg mL⁻¹).

differences in d-d transition strength from the metal and the polyphenol (Fig. S1, ESI[†]).²⁶ Based on previous research into the physical differences between coatings derived from different metal ions and polyphenols combined with the

spectroscopic characterization, these assemblies differ significantly in their ligand–metal coordination.^{26,27} We therefore hypothesized that these differences could impact cellular protection capabilities.

Microbial coating and subsequent viability assessment

To apply MPNs as protective agents, these coatings must readily form on microbial substrates with the same ease as the silica particles. Following our established protocol, $\text{Fe}^{\text{III}}\text{-TA}$ assemblies were readily deposited on the surface of *B. subtilis*. Surface coating was confirmed through observation of the metals on the surface of the bacteria by transmission electron microscopy (TEM) (Fig. 2(b)). As can be seen in the image of the uncoated cells, the edges of the uncoated and unstained cells appear smooth and ill-defined. In contrast, the well-defined surface of the MPN-coated cells is visible due to the presence of the metal ions. Following confirmation of coating formation on the cells, we evaluated the impact of the MPNs on cell growth and viability. Bacterial growth was monitored by optical density measurements at 600 nm (OD_{600}), and a delay of 15 hours was observed for the MPN-coated cells to reach exponential phase as compared to the uncoated cells (MPN-coated: 20 hours *vs.* uncoated controls: 5 hours) (Fig. S2, ESI[†]). However, cell viability, which was assessed by live/dead staining coupled with fluorescence microscopy, showed minimal bacterial death for MPN-coated *B. subtilis* (85% viable), as compared to the uncoated control (92% viable) (Fig. 2(a)). Thus, we believe the lag in reaching exponential phase for the coated cells is due to the rigidity of the coating inhibiting cell division, rather than the MPNs impacting viability.

Importantly, we further observed that a stable MPN assembly is critical for protection. Without the addition of MOPS buffer at a slightly alkaline pH, which serves to ensure chelation of the polyphenols to the metal ions and to rigidify the MPN capsule, a significant loss in the viability of MPN-coated *B. subtilis* (54% viable) was shown (Fig. S3, ESI[†]). This significant decrease in viability in the absence of MOPS buffer addition is attributed to the protonated polyphenols and unchelated metal ions in solution. Indeed, tannic acid has demonstrated inhibitory effects on the growth of some microbes.²⁸ However, a stable MPN assembly under neutral pH yielded cells with equivalent viability to uncoated cells. Though simply coating cells is a first step in the development of non-sporulated *B. subtilis* for probiotics, lyophilization is necessary for formulation and delivery of these microbes to the gut. Thus, we next investigated the impact of lyophilization on *B. subtilis* viability.

Role of the phenol in MPN microbial protection

Previously, we discovered that $\text{Fe}^{\text{III}}\text{-TA}$ MPNs protected living microbes (*E. coli* and *B. thetaiotaomicron*) from stresses caused by the freezing and lyophilization processes, significantly increasing their viability post-reconstitution. We hypothesized that this protection could be extended to *B. subtilis*. We previously observed the most significant protection with TA-containing MPNs. We therefore began our evaluations of the impact of MPN coating on post-lyophilization viability of



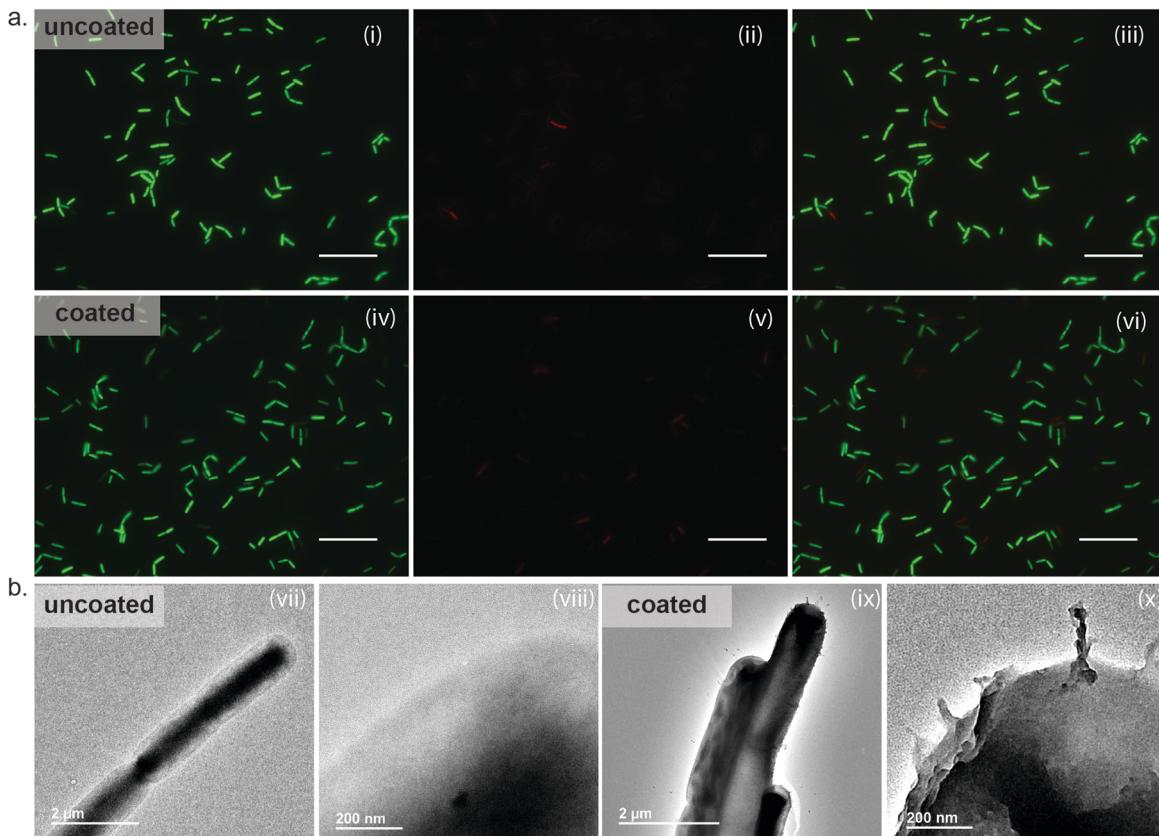


Fig. 2 Fluorescence and TEM images of encapsulated *B. subtilis*. (a) Cells were treated with a live/dead staining kit. Shown are fluorescence images of (i)–(iii) uncoated and (iv)–(vi) Fe^{III} -TA-coated *B. subtilis* in the FITC channel (SYTO-9), and TxRed channel (propidium iodide) and the merged images. Scale bars indicate 20 μm . (b) TEM images of (vii), (viii) uncoated and (ix), (x) Fe^{III} -TA-coated *B. subtilis*.

B. subtilis with Fe^{III} -TA MPNs. Fe^{III} -TA-coated *B. subtilis* were flash-frozen and lyophilized in one of four different buffers: phosphate citrate (PC) or phosphate buffered saline (PB), either in the absence of or presence of 0.1 M trehalose (cryoprotectant-PCT or PBT, respectively). OD_{600} was monitored post-reconstitution in the nutrient broth medium at 30 °C (Fig. 3(a)). Surprisingly, the protection of cells with this coating was found to be highly buffer-dependent. OD_{600} values after 48 hours of post-reconstitution growth was fastest when lyophilized in PCT, with high variability in growth from lyophilization in other buffered conditions (Fig. 3(a)). This result indicates that the protective ability of the MPNs is somewhat strain dependent. Thus, we sought to investigate whether altering the components of the MPN impacted post-lyophilization viability. We expanded the polyphenols investigated to include EGCG and GA. No significant difference was observed in the post-reconstitution growth of Fe^{III} -EGCG-coated *B. subtilis* as compared to the uncoated controls. Yet interestingly, Fe^{III} -GA-coated *B. subtilis* demonstrated higher OD_{600} values post-reconstitution as compared to the uncoated controls in both buffers (PBT and PCT).

Though trehalose is canonically used to maintain biomolecule stability during lyophilization, the addition of this molecule is not ideal for the eventual formulation of the cells for oral delivery.^{29–31} Thus, we additionally investigated the impact of

MPN coating on protection of *B. subtilis* following lyophilization in the absence of trehalose. Remarkably, Fe^{III} -GA- and Fe^{III} -TA-coated *B. subtilis* in cryoprotectant-free buffer (PC) demonstrated 4-fold and 3-fold increases in OD_{600} , respectively. Under the same conditions, Fe^{III} -GA- and Fe^{III} -TA-coated *B. subtilis* reach exponential growth within 8 hours, whereas no growth was observed for either Fe^{III} -EGCG-coated microbes or the uncoated controls (Fig. 3(b)). These results demonstrate the ability of iron-containing MPNs to provide protection for *B. subtilis* post-lyophilization even in absence of cryoprotectant. Furthermore, this protection was found to be phenol-dependent, where the efficacy decreased in the order of GA > TA > EGCG. It was previously reported that the size of the phenol ligand affects the Young's modulus (E_Y) of MPNs, with smaller ligands leading to higher moduli.¹⁵ Thus, we anticipate that the higher protection afforded by Fe^{III} -GA complexes could be attributed to the modulus of this MPN. Meanwhile, our results further demonstrate how critical the lyophilization buffers are in stabilizing microbes during the freeze-drying process. Because of the significant impact that varying the polyphenol was found to have on post-lyophilization stability, we further sought to investigate the impact of varying the metal ion in the MPNs.

Role of the metal in MPN-based microbe protection

Though 18 metal ions have been shown in the literature to form MPNs with polyphenols, only biocompatible metal ions were



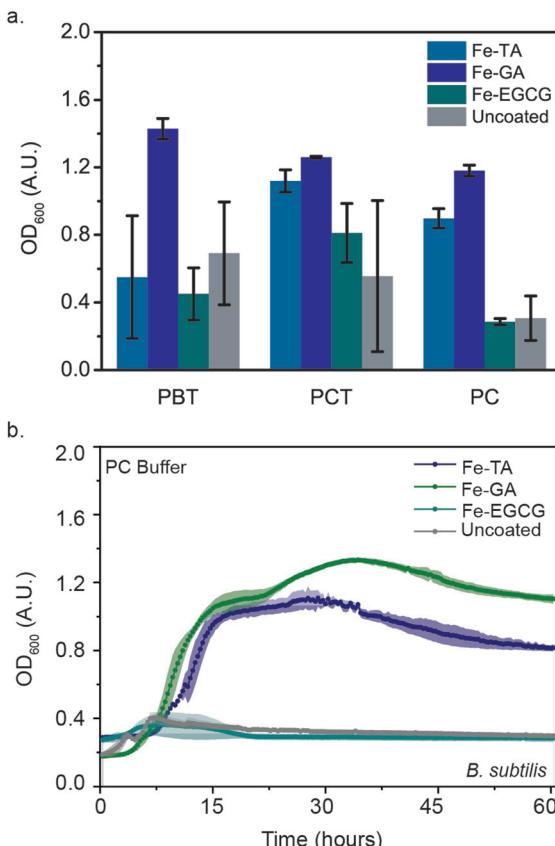


Fig. 3 Fe^{III} -phenol encapsulation protects microbes during lyophilization. (a) OD_{600} of lyophilized *B. subtilis* following 48 h of growth in the presence and absence of Fe^{III} -phenols and cryoprotectants (0.1 M) (PC, phosphate citrate buffer; PB, phosphate-buffered saline; T, trehalose). (b) Growth curves of Fe^{III} -phenols-coated and uncoated *B. subtilis* following lyophilization without cryoprotectants. Error bars and shaded region represent SD for $n = 3$ biological replicates.

investigated in this study, as the eventual goal of coating *B. subtilis* is to enable their oral delivery.²⁶ Specifically, we focused on four ions: Fe^{III} , Mn^{II} , Zn^{II} , and Al^{III} . Because GA demonstrated the highest level of microbial protection of the three polyphenols investigated, the four metal ions were tested as components of MPNs formed with GA. Fe^{III} -GA, Mn^{II} -GA, Zn^{II} -GA, and Al^{III} -GA-coated *B. subtilis* were flash-frozen and lyophilized in either PBT or PCT. OD_{600} was monitored after reconstitution in nutrient broth medium (Fig. 4(a)). No significant difference in OD_{600} values after 48 h was observed between Fe^{III} -GA, Mn^{II} -GA, and Al^{III} -GA-coated *B. subtilis* as compared to the uncoated control. Moreover, no growth was observed for Zn^{II} -GA coated microbes. We attribute this to the protective abilities of the trehalose in the buffer because, upon addition of cryoprotectant-free PC buffer, no growth was observed after 48 h for Mn^{II} -GA-coated *B. subtilis* or the uncoated controls. However, growth was observed for cells coated in MPNs formed with the three other metal ions. Interestingly, the delay in reaching exponential growth (t_{eg}) decreased in the order of $\text{Fe}^{\text{III}} < \text{Al}^{\text{III}} < \text{Zn}^{\text{II}}$, revealing that metal ions also impact the efficacy of MPN cellular protection (Fig. S5, ESI[†]). This trend is

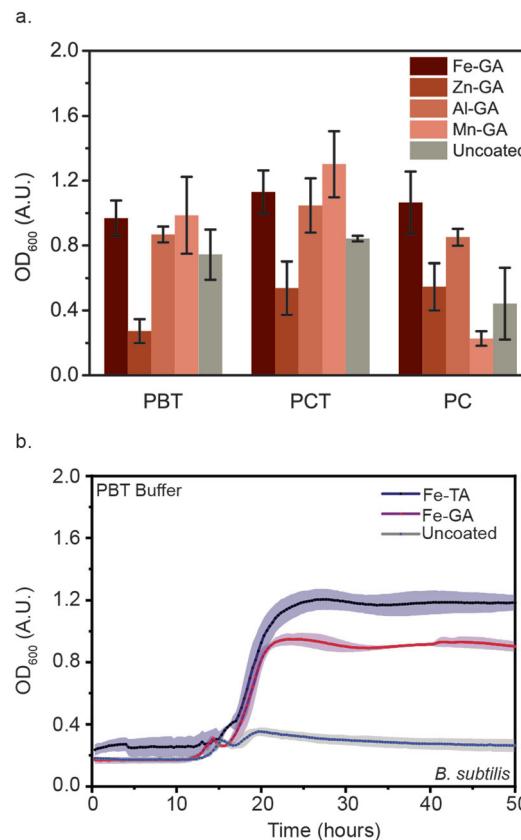


Fig. 4 Metal-GA encapsulation protect microbes during lyophilization. (a) OD_{600} of lyophilized *B. subtilis* following 48 h of growth in the presence and absence of metal-GA and cryoprotectants (0.1 M) (PC, phosphate citrate buffer; PB, phosphate-buffered saline). (b) Growth curves of Fe^{III} -TA, Fe^{III} -GA-coated and uncoated lyophilized *B. subtilis* following dilute L-ascorbic acid (AA) treatment (5 min). Error bars and shaded region represent SD for $n = 3$ biological replicates.

consistent with previous work showing that trivalent metal ions increased the network size of MPN assemblies as compared to those using divalent metal ions with same phenol. Further, the increased recovery rate with Fe^{III} -GA as compared to Al^{III} -GA-coated *B. subtilis* is consistent with the strength of ligand chelation and the size of the metal ionic radius.^{32,33} Taken together, these results confirm that both the polyphenol and the metal ion used to form MPNs impact the ability of the MPNs to protect microbes following lyophilization. Further, as compared to our previously-reported results with *B. thetaiotao-micron*, the coating efficacy seems to be strain-dependent, with optimal MPNs varying based on the microbe being evaluated.

Mild acid treatment leads to faster recovery

Under acidic conditions, hydroxyl groups on the polyphenols in the MPN complexes are protonated, disrupting chelation and leading to disassembly.³⁴ For oral delivery applications, this pH-dependent degradation is important, as it provides an inherent on-demand disassembly for the cells upon their exposure to stomach acid and intestinal fluid. Though uncoated *B. subtilis* demonstrated faster recovery than MPN-coated cells post-reconstitution in nutrient



broth with PBT or PCT (Fig. S4, S5 and S7, ESI†), this appears to be due to the inherent lag in growth consistently observed with MPN-coated *B. subtilis* (Fig S2, ESI†). This hypothesis is supported by the rapid growth of lyophilized MPN-coated *B. subtilis* when cells are reconstituted in 10 mM L-ascorbic acid solution for 5 min prior to transfer to nutrient broth. The pH of the L-ascorbic acid solution used in this study was measured to be pH = 2.5, which is within the range of the pH of stomach acid (pH = 1.5–3.5). Removal of the coating prior to growth studies decreased the delay time to exponential growth (t_{eg}) of Fe^{III}-GA and Fe^{III}-TA-coated cells dramatically. Further, microbial survival following acid treatment was only observed with MPN-coated cells; no growth after 48 hours was observed with the uncoated controls (Fig. 4). These results further demonstrate the utility of MPN protective coatings for oral delivery of microbes, as these cells will necessarily be exposed to acidic conditions in the stomach and gut.

Conclusions

MPNs are an important class of self-assembled materials comprised of polyphenols and metal ions that had previously been shown to protect microbes of interest as next-generation biotherapeutics during lyophilization and processing. However, the universality of these coatings for microbial protection and delivery had not been established. Here, we investigated the protection of *B. subtilis*, an important probiotic strain, from processing stressors during lyophilization. Surprisingly, the efficacy of protection was found to be dependent on the metal and polyphenol used for assembly. Both the size of the polyphenol and stability of the metal-phenol coordination were important factors that influenced their cellular protection; the smallest polyphenol, GA, and the most stable chelated ion, Fe^{III}, were found to provide the highest level of protection. Further, for probiotics and living biotherapeutics, delivery to the gut involves exposure to acidic conditions in the form of stomach acid and intestinal fluid. MPN coatings rapidly disassembled upon mild acid treatment but protected the *B. subtilis* from the negative impacts of the acid. Overall, *B. subtilis* is an important probiotic strain currently formulated for delivery as spores, but their ability to germinate in the gut remains debatable. Thus, the ability to deliver viable *B. subtilis* cells directly is anticipated to improve the efficacy of these strains and serves as an additional study to better understand role of each component in MPNs.

Conflicts of interest

We have filed a provisional patent based on this work.

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