Chemical Science



EDGE ARTICLE

View Article Online



Cite this: Chem. Sci., 2024, 15, 19851

d All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 2nd July 2024 Accepted 30th October 2024

DOI: 10.1039/d4sc04361h

rsc.li/chemical-science

Total synthesis of atropodiastereomers of heterodimeric Amaryllidaceae alkaloids: narcipavline and narcikachnine†

Souvik Pal, Da Satyajit Majumder, Da Sovan Niyogi, Db Pranay Shyamal, Db Debabrata Mondal, b Bishnu Das b and Alakesh Bisai ** *ab

We report the first asymmetric total synthesis of recently isolated heterodimeric Amaryllidaceae alkaloids, narcipavlines A (1a) and B (1b), and narcikachnines A (2a) and B (2b), thereby confirming their absolute stereochemistry. These alkaloids showcase a unique heterodimeric structure, amalgamating two distinct types of Amaryllidaceae alkaloids: the cis-hydrodibenzofuran containing tetracyclic galantamine core (6a) and the galanthindole core (7) featuring a biaryl axis. The presence of this biaryl axis, coupled with the substantial galantamine core (6a) at the ortho substituents, imposes constraints on free rotation around the C-C axis, resulting in atropisomerism, an exceedingly rare phenomenon in nature. Key steps in the synthesis encompass the utilization of a one-pot double reductive amination approach for the establishment of C-N-C bonds to merge both the galantamine (6a) and galanthindole (7) cores. Additionally, the Mitsunobu reaction and intramolecular Heck cyclization have emerged as pivotal techniques for crafting the tricyclic hydrodibenzofuran core [(-)-13], incorporating an all-carbon quaternary stereogenic center.

Introduction

The Amaryllidaceae family, particularly plants in the Narcissus genus, have a rich history in traditional and Western medicine.1 Dating back to the 4th century BC, famous Greek physician Hippocrates used extracted oil from *Narcissus poeticus* L. to treat uterine tumors. These plants are renowned for producing structurally and biologically significant Amaryllidaceae alkaloids² (AAs). The identification of lycorine in 1877 marked the start of an extraordinary voyage, leading to the discovery of over 600 AAs since then, highlighting a wide array of biological activities.2,3 Of particular interest are AAs with a cis-hydrodibenzofuran structure, featuring tetracyclic skeletons with vicinal quaternary and tertiary stereogenic centers, which show potential medicinal properties such as acetylcholinesterase (AChE) inhibition, 4a crucial in Alzheimer's disease (AD) treatment, as well as anti-malarial, anti-infective, and anti-cancer²

effects. Amaryllidaceae plants naturally synthesize these AAs from L-tyrosine through a series of oxidation and reduction reactions.3 Galantamine (6a), among the most well-known Amaryllidaceae alkaloids, has been utilized to treat mild to moderate Alzheimer's disease⁵ under the trade name Razadyne©, approved by the FDA in 2001. AD, a chronic and progressive neurodegenerative disorder, represents a significant global health concern affecting over 36 million individuals.6

Given the urgent need for effective therapies, research into novel treatments for AD is imperative. Currently all treatments of AD are only symptomatic aiming to boost acetylcholine levels through the inhibition of acetylcholinesterase using specific cholinesterase inhibitors like galantamine (6a), donepezil, and rivastigmine.⁷ An enzyme called butyrylcholinesterase (BuChE) is also able to hydrolyze AChE.8 Studies have indicated that in later stages of AD, patients experience a significant increase in butyrylcholinesterase levels by up to 50-70% while acetylcholinesterase expression decreases. This suggests that butyrylcholinesterase could be a promising therapeutic target, not only in restoring brain acetylcholine levels but also in serving as a disease-modifying agent in the prodromal stages of the disease.

Recently, a series of intricate heterodimeric Amaryllidaceae alkaloids have been freshly isolated (Fig. 1), showcasing atropodiastereomeric relationships. narcipavline (1) and narcikachnine (2) were isolated in 2018 from Narcissus poeticus ev. Pink Parasol, 9a existing as 1:1 and 1:1.2 diastereomeric

^aDepartment of Chemistry, Indian Institute of Science Education and Research Bhopal, Bhopal Bypass Road, Bhopal 462 066, Madhya Pradesh, India. E-mail: alakeshb@ gmail.com

^bDepartment of Chemical Sciences, Indian Institute of Science Education and Research Kolkata, Mohanpur Campus, Nadia, Kalyani, 741 246, West Bengal, India. E-mail: alakesh@iiserkol.ac.in

[†] This work is dedicated respectfully to Professor Srinivasan Chandrasekaran, IISc Bangalore for his constant encouragement and inspiration.

[‡] Electronic supplementary information (ESI) available. CCDC 2333542, 2363948 and 2384521. For ESI and crystallographic data in CIF or other electronic format see DOI: https://doi.org/10.1039/d4sc04361h

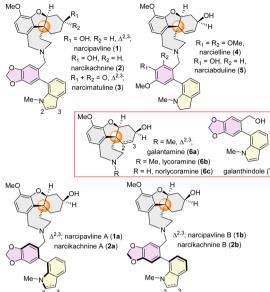


Fig. 1 Dimeric atropodiastereomeric *Amaryllidaceae* alkaloids (1–5) and monomeric congeners (6–7) sharing a *cis*-hydrodibenzofuran core.

mixtures respectively. They are a combination of two structural types of Amaryllidaceae alkaloids: cis-hydrodibenzofuran containing lycoramine (6b) and bi-aryl axis containing galanthindole9c (7) core (Fig. 1). Narcimatuline (3) was isolated from Narcissus pseudonarcissus bulbs,96 existing as a 1:1.1 diastereomeric mixture, featuring a keto functionality at the cishydrodibenzofuran skeleton and N-methylindoline at the atropodiastereomeric position. Similar structural type narcielline (4) and narciabduline (5), isolated from Zapharynthis citrina^{10a} and Narcissus pseudonarcissus 10b respectively, in 2021, also exist as inseparable diastereomeric mixtures. While the stereochemistry and atropodiastereomerism were confirmed for narcielline (4) and narciabduline (5) by isolation chemists, 10a they still remained unclear for narcipavline (1) and narcikachnine (2). Therefore, we have proposed the structures of the diastereomeric mixture of narcipavline (1) as narcipavlines A (1a) and B (1b), and narcikachnine (2) as narcikachnines A (2a) and B (2b), considering them as naturally occurring atropodiastereomers. Importantly, these newly isolated alkaloids exhibit excellent BuChE inhibition, with narcipavlines (1) showing an IC_{50} of 24.4 \pm 1.2 μ M, and narcimatulines (3) exhibiting an IC_{50} of 5.90 \pm 0.87 μ M. Narcipavlines [*i.e.* narcipavlines A (1a) and B (1b)] also demonstrate antiplasmodial activity against malaria parasites, suggesting their potential for the development of new drugs. Biological activity for narcikachnines 2 [i.e. narcikachnines A (2a) and B (2b)] is yet to be discovered.

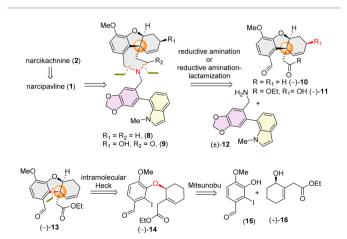
Structurally, these alkaloids possess three chiral centers, a *cis*-hydrodibenzofuran core, along with an all-carbon quaternary stereogenic center¹¹ and a chiral bi-aryl axis which also leads to complicated rare atropisomerism, makes them impressively challenging structures.¹² Atropisomerism, a form of dynamic chirality, occurs due to restricted rotation between the single bond between two sp²-hybridized atoms, yielding

atropisomers. These are classified into three categories, based on the rotational barrier ($\Delta E_{\rm rot}$) and half-life ($t_{1/2}$), class I ($\Delta E_{\rm rot}$ < 20 kcal mol⁻¹, $t_{1/2} \approx$ ns-ms) with rapidly equilibrating conformers considered as achiral; class II ($\Delta E_{\rm rot} \approx 20$ –30 kcal mol⁻¹, $t_{1/2} \approx$ min-hours per days), considered high-risk in drug development due to moderate stability and class III high rotational barrier isomers ($\Delta E_{\rm rot} > 30$ kcal mol⁻¹, $t_{1/2} \approx$ years), highly stable and often developed as single atropisomers, although very challenging to synthesize. Only very few syntheses have been reported for other atropodiastereomeric natural products. Biosynthesis could be hypothesized from a combination of two different *Amaryllidaceae* alkaloids *norlycoramine* (6c) and galanthindole (7) *via* enzymatic alkylation, where both can be derived from L-tyrosine.

Although there are many elegant approaches for the synthesis of galantamine (6a) and related tetracyclic scaffolds containing a *cis*-hydrodibenzofuran core, ¹³⁻¹⁵ there has been no synthesis reported for the newly isolated complex heterodimeric *Amaryllidaceae alkaloids* to date. We hypothesized that classical reductive amination could be a useful tool to construct the C-N-C bond of the seven membered azepine ring in one pot and will be worth exploring in this regard. Due to their impressive biological activity and structurally challenging arrays, we envisioned a catalytic enantioselective unified approach *via* reductive amination to synthesize narcipavline (1) [*i.e.* the atropodiastereomers narcipavlines A (1a) and B (1b)] and narcikachnine (2) [*i.e.* the atropodiastereomers narcikachnines A (2a) and B (2b)] along with galantamine (6a) and lycoramine (6b).

Results and discussion

Retrosynthetically, narcikachnine (2) could be derived from narcipavline (1) via reduction of the indole ring. Narcipavline (1) consists of two parts: the northern part is the lycoramine core and southern part is a galanthindole (7) core. Initially, we envisioned accessing atropodiastereomers of narcipavline (1) [i.e. (1a) and (1b)] from an advanced intermediate 8 via allylic oxidation¹⁶ and hydrogenation (Scheme 1). To construct the



Scheme 1 Retrosynthetic analysis of narcipavline (1) and narcikachnine (2).

Edge Article Chemical Science

Scheme 2 Synthesis of galanthindolyl benzyl amine.

seven membered azepine core, a one-pot double reductive amination of galanthindolyl benzylamine (12) with tricyclic dialdehyde 10 was postulated (Scheme 1). The all-carbon quaternary stereogenic center¹¹ of tricycle 13 bearing an aldehyde and ester could be constructed *via* intramolecular Heck cyclization of alkyl aryl ether 14, which further could be accessed *via* Mitsunobu reaction of the phenol derivative 15 with enantioenriched α -substituted cyclohex-2-en-1-ol 16 (Scheme 1).

Our journey commenced with the preparation of the galanthindole (7) core and its corresponding benzylic amine 12. To this end, we synthesized the known biaryl aldehyde 17, and galanthindole (7) following a modified procedure of Hsieh's protocol (see ESI‡ for details). Subsequently, we transformed the biaryl aldehyde 17 into the corresponding benzylamine (\pm) -12 via a reductive amination using ammonium formate (Scheme 2).

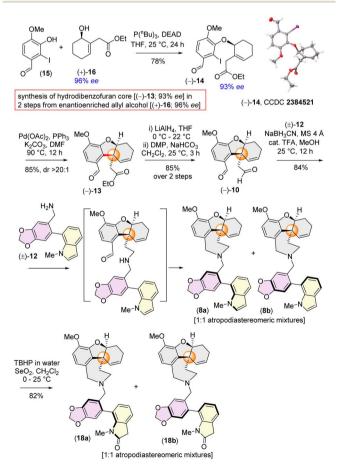
The isolation chemists did not provide any optical rotation data for galanthindole (7). The wever, due to its structural features, including a biphenyl axis and *ortho*-substituents, it can exhibit axial isomerism. Indeed, when subjected to HPLC analysis, galanthindole showed a 50:50 mixture of these two atropisomers, confirming its racemic nature [denoted as (\pm)-7]. For the synthesis of the galantamine core, we began with the synthesis of enantioenriched (96% ee) α -substituted cyclohex-2-en-1-ol¹⁸ 16 *via* a catalytic enantioselective Corey–Bakshi–Shibata (CBS) reduction¹⁹ of a known Trost's enone²⁰ [see ESI‡ for details].

Inspired by Banwell's protocol, ^{14d} we explored the Mitsunobu reaction for the synthesis of enantioenriched **14**. Extensive optimization with different phosphines and azodicarboxylates revealed that allyl alcohol (+)-**16** (96% ee) and 2-iodophenol **15** afforded (-)-**14** in 93% ee in the presence of diethyl azodicarboxylate (DEAD) and tributylphosphine ("Bu₃P) [see ESI‡ for detailed optimization]. Gratifyingly, the stereochemistry of (-)-**14** was confirmed by single crystal X-ray analysis [CCDC 2384521].

With enantioenriched (-)-14 (93% ee) in hand, we moved forward towards the intramolecular Heck cyclization to construct the *cis*-fused tricyclic hydrodibenzofuran core (Scheme 3). To our delight, charging iodoarene (-)-14 under Heck coupling conditions using catalytic Pd(OAc)₂ smoothly afforded tricyclic product (-)-13 with excellent dr (dr > 20:1), with generating the all carbon quaternary stereogenic centre. By employing our strategy, we have been able to achieve the hydrodibenzofuran core in 93% ee (see ESI‡ for HPLC analysis) in 4 steps from commercially available cyclohex-2-en-1-one which is superior to the previous reports. ^{13b-d,14d} Further, lithium aluminium hydride reduction of the ester group of 13

followed by oxidation using Dess-Martin periodinane (DMP) afforded di-aldehyde (-)-10 in 85% yield over 2 steps (Scheme 3).21 A one step reduction using DIBAL-H proved to be difficult as it is associated with over-reduced diol in 35% yield in addition to the desired dialdehyde 10 (see ESI‡ for details).21 Next, reacting equimolar amounts of dialdehyde (-)-10 and benzylamine (\pm) -12 in the presence of sodium cyanoborohydride (NaBH₃CN) and a catalytic amount of trifluoroacetic acid afforded the desired product (-)-8 in 84% yield (Scheme 3). Interestingly, ¹H NMR analysis of the pure product revealed the presence of two diastereomers in a \sim 1:1 ratio, attributable to atropisomerism around the biphenyl axis, resulting in two distinct configurations of atropodiastereomers 8a and 8b (Scheme 3). The separation of these atropodiastereomers was a tedious job, as the TLC shows a single spot, making them inseparable in column chromatography.

At this stage, only a diastereoselective allylic oxidation was required to complete the total synthesis of narcipavlines (1). However, all efforts to perform allylic oxidation were unsuccessful and seemed to be very challenging (Scheme 3). When the mixture of 8a and 8b was charged under allylic oxidation conditions using selenium dioxide and TBHP, it was observed that the indole ring of 8 is highly susceptible to oxidation compared to the allylic position. Even under mild conditions using selenium dioxide and *tert*-butyl hydrogen peroxide



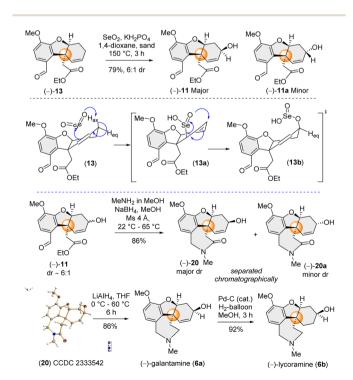
Scheme 3 Forward synthesis towards narcipayline.

Chemical Science

(TBHP), we could observe the formation of $\sim 1:1$ atropodiastereomers of 2-oxindoles, **18a** and **18b** (Scheme 3). Therefore, we recognized the necessity of installing the allylic alcohol early in the tricyclic core (–)-**13**.

Therefore, employing Trost's procedure, ^{13c} we successfully executed the allylic oxidation of (-)-13 using selenium dioxide (SeO_2) with sand as a solid support in refluxing 1,4-dioxane to afford a mixture of 11 and 11a (79% yield, dr \sim 6:1 in favour of 11) (Scheme 4). The diastereoselectivity observed in case of allylic oxidation was attributed to the approach of SeO_2 from the non-traditional more hindered concave face of tricycle (-)-13 (Scheme 4). The mechanism involves an Ene-type reaction and the stereoelectronic requirements of the Ene reaction dictate specific spatial arrangements, where approach through the pseudo-axial H rather than the pseudo-equatorial H of cyclohexene results in the formation of 11 as the major diastereomer (Scheme 4).

Next, we focused on the one-pot reductive amination-lactamization sequence of aldehyde and ester groups present in (–)-11 (Scheme 4). It was observed that (–)-11 (\sim 6:1 dr) in the presence of MeNH₂ in MeOH followed by reduction using NaBH₄ afforded lactams (–)-20 (as the major dr) and 20a (see ESI‡ for details). Gratifyingly, these two diastereomers were separable in column chromatography. The structure of major diastereomer 20, having the hydroxyl group above the plane, was confirmed by X-ray single crystal analysis (CCDC 2333542). Next, with tetracycle (–)-20 in hand, we completed the total synthesis of naturally occurring (–)-galantamine [(–)-6a] just by LiAlH₄ reduction (86% yield). Further, hydrogenation of (–)-6a completed the total synthesis of (–)-lycoramine [(–)-6b].



Scheme 4 Diastereoselective allylic oxidation and synthesis of (–)-galantamine (6a) and (–)-lycoramine (6b).

Scheme 5 Failed attempt towards reductive amination lactamization with galanthindolyl benzylic amine.

We next turned our attention to the synthesis of the azepine core of narcipavline (1) via a similar one-pot reductive amination-lactamization cascade. However, this proved to be very challenging with the sterically demanding benzylamine (\pm)-12 (Scheme 5).

A condensation reaction of (–)-11 with benzylic amine 12, followed by reduction using NaBH₄ didn't afford the required azepine core; instead, we could isolate benzyl alcohol 21 and ε -lactone 22 (through acid-catalysed cyclization) in a few specific cases (Scheme 5). This fact is probably attributed to the strongly electron-donating nature of the *p*-methoxy group on (–)-11 which could diminish the electrophilicity of benzaldehyde hindering the crucial imine formation (Scheme 5).

So, we first protected the secondary alcohol group of 11 as TBS-ether to access (-)-23 and (-)-23a (Scheme 6). Delightfully, we could separate the major diastereomers 23 via column chromatography. Next, the ester and the aldehyde group both were reduced and again re-oxidized using Dess-Martin periodinane (DMP) to afford dialdehyde (-)-24 in 85% yield over 2 steps (Scheme 6). Subsequently, a one-pot sequential reductive amination was successfully conducted with (-)-24 and galanthindolyl benzylamine (±)-12 in the presence of NaBH₃CN and catalytic trifluoroacetic acid to furnish 25 as a mixture of two atropodiastereomers, 25a and 25b with dr \sim 1:1 (see ESI‡ for detailed characterization). Next, deprotection of the silyl ether group was achieved using tetra-butyl ammonium fluoride (TBAF) yielding dehydronarcipvaline (26a and 26b) as 1:1 atropodiastereomers. It is pertinent to mention that, a racemic crystal structure of (±)-26b (CCDC 2363948) (prepared from (\pm) -24 in a different sequence) unequivocally proved the allbond connections present in dehydronarcipvaline 26. With a few milligrams of crystals in hand, a detailed ¹H-NMR analysis was undertaken. It is interesting to note that the ¹H-NMR of (\pm) -26b showed the signals for a single atropodiastereomer (see ESI‡ for details). However, when 13 C-NMR of (\pm)-26b was recorded after a day (\sim 24 h), the signal of each carbon appeared to be doubled, thereby confirming the rotation along the axis happening at room temperature. Further, the ¹H-NMR of the same sample after \sim 24 h provided the signals for \sim 1:1 atropodiastereomers $[(\pm)-26a$ and $(\pm)-26b]$. Thus, it may be concluded that the atropodiastereomer (\pm)-26b has a very short half-life period and could be easily converted to (\pm) -26a, leading

Edge Article Chemical Science

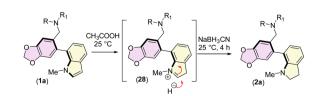
Scheme 6 Total synthesis of atropodiastereomers narcipavline A, B and narcikachnine A, B

to the mixtures of atropodiastereomers at room temperature (see ESII for details).

Next, the atropodiastereomers of dehydronarcipavline 26 $(\sim 1:1 \text{ dr})$ were hydrogenated using Pd-C at 1 atm. H₂-gas afforded the first total synthesis of atropodiastereomers of novel heterodimeric Amaryllidaceae alkaloids, narcipavlines A (1a) and B (1b) as \sim 1:1 dr (Scheme 6). Efforts towards the separation of the atropodiastereomers proved to be unsuccessful at room temperature using column chromatography. Nevertheless, all the corresponding peaks of ¹H- and ¹³C-NMR are in good agreement with the isolation report of narcipavlines (1a: **1b** ∼1:1) by Cahlíková *et al.*^{9a} The structure was confirmed through 2D NMR analysis, which revealed a negative specific rotation value for the diastereomeric mixture. Notably, compound 1 and lycoramine (6b) exhibited similar NMR spectra, particularly in the region of the secondary alcohol within the cis-hydrodibenzofuran core, and displayed negative optical rotation values. Further, HPLC analysis using a Chiralpak OD-H column (40% PrOH/n-hexane) revealed two pairs of enantiomers [93% ee] corresponding to the atropodiastereomers narcipavlines A (1a) and B (1b) [\sim 50:50 ratio] (see ESI‡ for details). These findings strongly support the proposed structure of atropodiastereomers 1a and 1b, resembling (-)-lycoramine (**6b**) as a *cis*-hydrodibenzofuran core, where the secondary alcohol group resides above the plane (Fig. 1).

We next turned our attention to the total synthesis of atropodiastereomeric alkaloids, narcikachnines A (2a) and B (2b) (Scheme 6). Their structures are nearly identical, with the sole difference being an indoline ring in narcikachnine (2) instead of an indole in narcipavline (1). After a quick optimization, we were pleased to see that the indole ring of narcipavline (1) could be reduced with NaBH3CN in AcOH to form indolines of narcikachnines A (2a) and B (2b) in 72% yield in ~1:1 dr (Scheme 6). A plausible mechanism of reductive amination of indole to indoline is shown in Scheme 7.

Once again, all the corresponding peaks of ¹H- and ¹³C-NMR are in good agreement with the isolation report of narcikachnines (2a:2b \sim 1:1) by Cahlíková et al.9a The diastereomeric mixture of narcikachnines (2a:2b) shows a negative specific rotation value. The cause of the doubling of some signals in NMR was found to be due to axial isomerism through the sterically hindered single bond rotation. Accordingly, VT NMR analysis was performed to demonstrate the atropodiastereomerism at different temperatures. Few milligrams of (2) and (26) were dissolved in DMSO-d₆ having a sufficiently high boiling point. At the highest experimental temperature, (up to



Scheme 7 Probable mechanism for the reduction of the indole ring

Chemical Science Edge Article

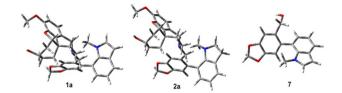
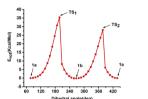


Fig. 2 Energy minimized structure of 1a, 2a and 7 using the B3LYP/6-31G(d) method.



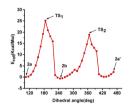


Fig. 3 Rotational energy barrier diagram for narcipavline ${\bf 1}$ and narcikachnine ${\bf 2}$.

70 °C for 2, and 125 °C for 26), the proton resonances coalesced for both diastereomers (see ESI‡ for details).

Further, we performed DFT calculations to understand the rotational barriers of these naturally occurring atropodiastereomers (Fig. 1). In particular, the rotational energy barriers of narcipavline 1, narcikachnine 2, and galanthindole 7 were analyzed to understand their conformational dynamics by simulating a 360-degree rotation around their biaryl C-C bond in 36 steps.²² For narcipavline 1, the energy profile reveals that conformations 1a and 1b are different, indicating that these conformations are atropodiastereomers of each other. The high energy barriers of 35.35 kcal mol⁻¹ at TS₁ and 28.09 kcal mol⁻¹ at TS2 indicate significant intramolecular forces and steric hindrance during rotation. This substantial energy requirement suggests that 1 is structurally rigid, making it probably less likely to undergo spontaneous conformational changes. This rigidity could facilitate the distinct separation of conformations 1a and 1b under special circumstances (Fig. 2).

In contrast, narcikachnine 2 exhibits a low energy barrier at TS_1 (25.08 kcal mol^{-1}) and TS_2 (18.90 kcal mol^{-1}), indicating greater rotational flexibility. The energy profile shows that conformations 2a, 2b, and 2a' are all different. This suggests that the molecule undergoes significant structural changes throughout the rotation. The lower energy peaks at TS_1 and TS_2 suggest weaker intramolecular interactions, allowing 2 to transition between conformations with less resistance, which could lead to interconversion between individual conformers more easily as compared to conformations 1a and 1b (Fig. 3).

Galantindole 7 presents an intermediate energy barrier, with TS_1 at 25.09 kcal mol^{-1} and TS_2 at 30.78 kcal mol^{-1} , reflecting a balance between flexibility and rigidity. The energy peaks at TS_1 and TS_2 indicate that 7 experiences moderate steric hindrance during rotation. This intermediate behaviour may allow for the possibility of selectively separating the distinct conformers (S)-7, (R)-7, and (S)-7', however under very special circumstances.

Conclusion

In summary, a concise catalytic enantioselective (93% ee) first total synthesis of heterodimeric atropodiastereomeric *Amaryllidaceae* alkaloids narcipavlines A (1a) and B (1b) [\sim 1:1 dr, 27% overall yield in 9 LLS] and narcikachnines A (2a) and B (2b) [\sim 1:1 dr, 19% overall yield in 10 LLS] has been achieved from enantioenriched allyl alcohol 16 (96% ee). The synthesis involves sequential reductive amination with galanthindolyl benzylamine (12) as the key steps. Through this synthesis we also confirmed the absolute structure of narcipavline A (1a) and narcipavline B (1b) as well as narcikachnine A (2a) and narcikachnine B (2b) as a mixture of atropodiastereomers. This effort also culminated in the protecting group free total synthesis of (-)-galantamine (6a) and (-)-lycoramine (6b), in six and seven steps, respectively.

Data availability

Experimental details and spectral analysis are available free of charge from the ESI‡ available with this article.

Author contributions

A. B. and S. P. designed the research plan. S. P., S. M., and S. N. investigated the key synthetic processes leading to atropodiastereomers of *Amaryllidaceae* alkaloids, narcipavlines A (1a) and B (1b) as well as narcikachnines A (2a) and B (2b). P. S. and D. M. were actively involved in the total synthesis of (—)-galantamine (6a) and (—)-lycoramine (6b). B. D. calculated the rotational energy barrier. A. B. and S. P. wrote the manuscript with contributions from all the authors; all the authors were actively engaged in the editing of the manuscript and gave their approval of the final version.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work is supported by grants from the SERB [SCP/2022/000486], SERB [CRG/2023/000782] and STARS-MoE [STARS/2023/0753]. SP thanks IISER Bhopal for pre-doctoral fellowships; SM and SN thank the CSIR; PS and DM thank the UGC for the research fellowships. AB is a SERB-STAR Fellow and gratefully acknowledges the SERB [STR/2020/000061] for generous support.

Notes and references

- (a) Z. Jin, Nat. Prod. Rep., 2016, 33, 1318–1343; (b) Y. Ding,
 D. Qu, K. M. Zhang, X. X. Cang, Z. N. Kou, W. Xiao and
 J. B. Zhu, J. Asian Nat. Prod. Res., 2016, 19, 53–100.
- 2 (a) S. Kobayashi, M. Kihara, K. Yuasa, Y. Imakura, T. Shingu, A. Kato and T. Hashomoto, *Chem. Pharm. Bull.*, 1985, 33,

Edge Article Chemical Science

- 5258-5263; (b) J. Refaat, M. S. Kamel, M. A. Ramadan and A. A. Ali, Int. J. Res. Pharm. Sci., 2013, 4, 1239-1252.
- 3 Z. Jin and G. Yao, Nat. Prod. Rep., 2019, 36, 1462-1488.
- 4 (a) G. M. Bores and R. W. Kosley, Drugs Future, 1996, 21, 621-635; (b) S. Lilienfeld, CNS Drug Rev., 2006, 8, 159–176.
- 5 (a) R. L. Irwin and H. J. Smith III, Biochem. Pharmacol., 1960, **3**, 147–148; (b) P. Williams, A. Sorribas and M. J. R. Howes, Nat. Prod. Rep., 2011, 28, 48-77.
- 6 E. Nichols, C. Szoeke, S. Vollset, N. Abbasi, F. Abd-Allah, J. Ebro, M. Aichour, R. Akinyemi, F. Alahdab, S. Asgedom, et al., Global, regional, and national burden of Alzheimer's disease and other dementias, 1990-2016: A systematic analysis for the Global Burden of Disease Study 2016, Lancet Neurol., 2019, 18, 88-106.
- 7 M. Paskaski and J. Kalman, Neurochem. Int., 2008, 53, 103-111.
- 8 Y. Nicolet, O. Lockridge, P. Masson, J. C. Fontecilla-Camps and F. Nachon, J. Biol. Chem., 2003, 278, 41141-41147.
- 9 (a) M. Safratová, A. H. álková, D. Hulcová, K. Breiterová, V. Hrabcová, M. Machado, D. Fontnha, M. Pruděncio, J. Kuněs, J. Chlebek, D. Jan, M. Hrabinová, L. Nováková, R. Havelek, M. Seifrtová, L. Opletal and L. Cahlíková, Arch Pharm. Res., 2018, 41, 208-218; (b) D. Hulcová, J. Mařĺková, J. Korábečny, A. Hošťálková, D. Jun, J. Kuněs, J. Chlebek, L. Opletal, A. D. Simone, L. Nováková, V. Andrisano, A. Růžička and L. Cahlíková, Phytochemisty, 2019, 165, 112055; (c) Y. Yang, S. X. Huang, Y. M. Zhao, Q. S. Zhao and H. D. Sun, Helv. Chim. Acta, 2005, 88, 2550-2552.
- 10 (a) E. Kohelová, J. Mařĺková, J. Korábečny, D. Hulcová, T. Kučera, D. Jun, J. Chlebek, J. Jenčo, M. Safratová, M. Hrabinová, A. Ritomska, M. Malanik, R. Perinova, K. Breiterova, J. Kunes, L. Novakova, L. Opletal and Cahlíková, *Bioorg. Chem.*, 2021, **107**, 104567; (b) J. Marikova, A. Al Mamun, L. A. Shammari, J. Koranency, T. Kucera, D. Hulcova, J. Kunes, M. Malanik, M. Vaskova, E. Kohelova, L. Novakova, L. Cahlíková and M. Pour, Molecules, 2021, 26, 1279.
- 11 (a) Quaternary Stereocenters: Challenges and Solutions for Organic Synthesis, ed. J. Christoffers and A. Baro, Wiley-VCH, Weinheim, 2005; (b) M. K. Das, N. Kumar and A. Bisai, Org. Lett., 2018, 20, 4421-4424.
- 12 (a) R. Nandi, S. Niyogi, S. Kundu, V. R. Gavit, M. Munda, R. Murmu and A. Bisai, Chem. Sci., 2023, 14, 8047-8053; (b) G. Bringmann, A. Irmer, D. Feineis, T. A. M. Gulder and H. P. Fiedler, *Phytochemistry*, 2009, **70**, 1776–1786; (c) J. E. Smyth, N. M. Butler and P. A. Keller, Nat. Prod. Rep., 2015, 32, 1562-1583; (d) S. T. Toenjes and J. L. Gustafson, Future Med. Chem., 2018, 10, 409-422; (e) M. Fragkiadakis, Thomaidi, T. Stergiannakos, E. Chatziorfanou,

© 2024 The Author(s). Published by the Royal Society of Chemistry

- A. M. Barakat, C. Stoumpos and Gaidatzi. C. G. Neochoritis, Chem.-Eur. J., 2024, 30, e202401461.
- 13 (a) B. Cheng, Q. Wang, Y. An and F. Chen, Nat. Prod. Rep., 2024, 41, 1060-1090; (b) B. M. Trost and F. D. Toste, J. Am. Chem. Soc., 2000, 122, 11262-11263; (c) B. M. Trost and W. Tang, Angew. Chem., Int. Ed., 2002, 41, 2795-2797; (d) B. M. Trost, W. Tang and F. D. Toste, J. Am. Chem. Soc., 2005, 127, 14785–14803; (e) P. Chen, X. Bao, L. F. Zhang, M. Ding, X. J. Han, J. Li, G. B. Zhang, Y. Q. Tu and C. A. Fan, Angew. Chem., Int. Ed., 2011, 50, 8161-8166.
- 14 (a) V. Satcharoen, N. J. McLean, S. C. Kemp, N. P. Camp and R. C. D. Brown, Org. Lett., 2007, 9, 1867-1869; (b) C. Guillou, J. L. Beunard, E. Gras and C. Thal, Angew. Chem., Int. Ed., 2001, 40, 4745-4746; (c) C. A. Fan, Y. Q. Tu, Z. L. Song, E. Zhang, L. Shi, M. Wang, B. Wang and S. Y. Zhang, Org. Lett., 2004, 6, 4691-4694; (d) J. Nugent and M. G. Banwell, Eur. J. Org Chem., 2016, 19, 5862-5867; (e) T. Venkatesh, P. S. Mainkar and S. T. Chandrasekhar, Org. Biomol. Chem., 2019, 17, 2192-2198.
- 15 (a) L. Li, Q. Yang, Y. Wang and Y. Jia, Angew. Chem., Int. Ed., 2015, 54, 6255–6259; (b) Q. Zhang, F. M. Zhang, C. S. Zhang, S. Z. Liu, J. M. Tian, S. H. Wang, X. M. Zhang and Y. Q. Tu, J. Org. Chem., 2019, 84, 12664-12671; (c) S. Majumder, A. Yadav, S. Pal, A. Khatua and A. Bisai, J. Org. Chem., 2022, 87, 7786-7797; (d) J. Q. Chen, J. H. Xie, D. H. Bao, S. Liu and Q. L. Zhou, Org. Lett., 2012, 14, 2714-2717; (e) Y. Zhang, S. Shen, H. Fang and T. Xu, Org. Lett., 2020, 22, 1244-1248.
- 16 B. N. Kakde, P. Kumari and A. Bisai, J. Org. Chem., 2015, 80, 9889-9899.
- 17 T. T. F. Chiang, H. K. Wang and J. C. Hsieh, Tetrahedron, 2016, 72, 5640-5645.
- 18 A. Mondal, S. Pal, A. Khatua, A. Mondal and A. Bisai, J. Org. Chem., 2024, 89, 12485-12497.
- 19 (a) E. J. Corey, R. K. Bakshi and S. Shibata, I. Am. Chem. Soc., 1987, 109, 5551-5553; (b) S. P. West, A. Bisai, A. D. Lim, R. R. Narayan and R. Sarpong, J. Am. Chem. Soc., 2009, 131, 11187-11194.
- 20 B. M. Trost, W. Tang and F. D. Toste, J. Am. Chem. Soc., 2005, **127**, 14785–14803.
- 21 Reduction of (-)-13 using DIBAL-H afforded 40% yield of (-)-10 along with 35% yield of diol dialdehyde

22 K. Kapłon, O. M. Demchuk and K. M. Pietrusiewicz, Curr. Chem. Lett., 2015, 4, 145-152.