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Synthesis of polysubstituted cyclopenta[b]indoles via relay gold(ı)/Brønsted acid catalysis†

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An efficient relay catalytic process involving Au(I)/Brønsted acid to access various polysubstituted cyclopentannulated indoles from easily accessible 1-(2-aminophenyl)prop-2-ynols and readily available 1,3-dicarbonyls has been developed. In an unprecedented event, the intermediate 2-indolylmethyl cations undergo the cation-Ene reaction with various 1,3-dicarbonyls followed by an intramolecular Friedel–Crafts-type reaction generating functionalized cyclopenta[b]indoles.

Indoles and indolines are considered to be privileged structures due to their widespread occurrence in Nature with intricate structural diversity often associated with impressive bioactivities, and in a pharmaceutical sense due to their drug-like properties.¹ Among indole derivatives, cyclopenta[*b*]indoles are especially attractive due to their presence in numerous biologically active natural products, for example, paspalines, terpendoles, emindoles, polyveolines, spiroindimicins, fischerindoles, yuehchukene, *etc.*, in addition to medicinally important compounds such as MK-0524, Fig. 1.² The presence of complex molecular architectures coupled with impressive pharmacological properties prompted several research groups to contribute

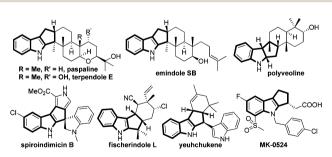
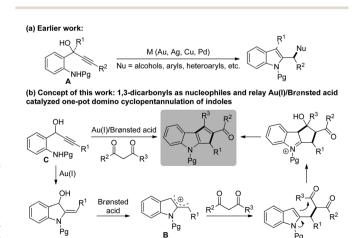


Fig. 1 Representative examples of bioactive cyclopenta[b]indoles.

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† Electronic supplementary information (ESI) available: Complete experimental procedures and characterization data of all new compounds. CCDC 1029309. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/ c4cc08174a significantly to the construction of cyclopentannulated indole derivatives.³ However, the quest for the development of a simple and efficient method to access this class of compounds from readily available starting materials still remains an area of active research.

On the other hand, activation of π -systems of alkynes and alkenes via gold catalysis for the synthesis of a wide range of natural products and complex molecules in an efficient and predictable manner has received significant attention during the past decade.⁴ Especially, relay catalytic processes involving gold were demonstrated to have great potential to rapidly assemble complex chemical structures often associated with pot, step and atom economy.⁵ Motivated by the pioneering studies by Chan⁶ for the synthesis of indole derivatives starting from propargylic tertalcohols of the type A, and with our experience in the chemistry of heteroaryl carbinols,⁷ we initiated a program to develop an efficient and general methodology towards the synthesis of a novel series of cyclopentannulated indoles and to evaluate their biological efficacy, Scheme 1. Prior to commencing our investigation, a detailed literature survey revealed that the prevailing 2-indolylmethyl cation intermediate **B** was routinely trapped by nucleophiles such as alcohols, aryls, heteroaryls, etc.⁸ However, to



Scheme 1 Our strategy for the synthesis of cyclopentannulated indoles.

our surprise, no attempt was ever made to employ readily available 1,3-dicarbonyl compounds as nucleophiles.⁹

Herein, we delineate our efforts towards the generation of indolylmethyl cations from 2-aminophenyl propargylic secondary alcohols C,¹⁰ their reactions with a variety of 1,3-dicarbonyl compounds and subsequent intramolecular aldol-type reactions for the synthesis of 1,2,3-trisubstituted cyclopenta[*b*]indoles. Accordingly, we initiated optimization studies towards identifying an effective catalytic system and other reaction parameters. For this, amino alcohol $1a^{6b,11}$ and acetylacetone 2a were chosen as substrates for the model reaction. The screening results are compiled in Table 1.

To begin with, cyclization of amino alcohol **1a** to indoline **3a** was explored. The reaction of **1a** in the presence of Au(i) or Ag(i) salts alone or a combination of Au(i) and silver based Lewis acids failed to deliver indoline **3a** (Table 1, entries 1–4). Interestingly, amino alcohol **1a** generated indoline **3a** when a combination of Au(i) and base was employed;¹² however, subsequent transformation of indoline **3a** to the desired indole **5a** was not observed upon treatment with a variety of Lewis acids, only the acetylacetone adduct **4a** was isolated

Table 1 Optimization of reaction parameters ^a						
OH Ia NHTs Ph		$ \begin{array}{c} \text{catalyst (2 mol\%)} \\ \text{so-catalyst (2 mol\%)} \\ \text{solvent, 60 °C} \\ \text{step-I} \end{array} \left[\begin{array}{c} \text{OH} \\ \text{N} \\ \text{3a}^{Ts} \\ \text{Ph} \end{array} \right] $		acetylacetone 2a (1.1 equiv) acid (20 mol%) step-II		
	$\rightarrow \left[$	NTS 4a Ph			Ph 5a	
Entry	Catalyst	Co-catalyst	Acid	Solvent	Time (h)	Yield ^b (%)
1	AuCl	_	_	DCE	48	_
2	PPh ₃ AuCl	_		DCE	48	_
3	AuCl	AgOTf		DCE	48	_
4	PPh ₃ AuCl	AgOTf	_	DCE	48	_
5 ^c	AuCl	K ₂ CO ₃	Sc(OTf) ₃	$MeNO_2$	48	_
6 ^{<i>c</i>}	AuCl	K_2CO_3	In(OTf) ₃	$MeNO_2$	48	_
7 ^c	AuCl	K_2CO_3	AgOTf	$MeNO_2$	48	_
8	AuCl	K_2CO_3	Bi(OTf) ₃	$MeNO_2$	26	74
9	AuCl	K_2CO_3	TMSOTT	DCE	20	82
10	AuCl	K_2CO_3	TFA	DCE	13	_
11	AuCl	K_2CO_3	H_3PO_4	DCE	16	51
12	AuCl	K_2CO_3	$HClO_4$	DCE	15	72
13	AuCl	K_2CO_3	H_2SO_4	DCE	18	70
14	AuCl	K_2CO_3	TfOH	DCE	13	83
15	AuCl	K_2CO_3	TfOH	$MeNO_2$	24	74
16	AuCl	K_2CO_3	TfOH	Toluene	13	80
17	AuCl	NaHCO ₃		DCE	40	—
18	AuCl	Na_2CO_3	TfOH	DCE	35	53
19	AuCl	Et ₃ N	TfOH	DEC	27	72
20^d	AuCl	K ₂ CO ₃	TfOH	DCE	18	90
$21^{c,e}$	AuCl	K ₂ CO ₃	TfOH	DCE	48	_
22^d	PPh ₃ AuCl	K_2CO_3	TfOH	DCE	42	76
23 ^d	AuCl	K_2CO_3	TMSOTf	DCE	36	85
24^d	AgOTf		TfOH	$MeNO_2$	34	63

^{*a*} Reaction conditions: a 5 mL glass vial was filled with **1a** (0.1 mmol), catalyst (2 mol%), co-catalyst (2 mol%) in an appropriate solvent (1 mL) and stirred at 60 °C, upon disappearance of the starting compound (**1a**), acetylacetone **2a** (0.11 mmol) and an acid (20 mol%) were introduced and stirring was continued at 60 °C until indoline **3a** and acetylacetone adduct **4a** disappeared. ^{*b*} Isolated yields after silica gel column chromatography. ^{*c*} Intermediate **3a** formed and only the acetylacetone adduct **4a** was isolated. ^{*d*} 10 mol% TfOH was employed. ^{*e*} Step-II was carried out at room temperature.

(Table 1, entries 5–7). Nevertheless, an approach for indoline **3a** *via* a catalytic sequence involving Au(I) and a *base* could be established for the first time.¹³

Upon further screening, Lewis acids such as Bi(OTf)₃ and TMSOTf gratifyingly generated the cyclopentannulated indole **5a** in very good yields (Table 1, entries 8 and 9). We were delighted especially because, as per our hypothesis, the intervening indolylmethyl cation formed under Lewis acidic conditions successfully underwent the cation-Ene reaction with acetylacetone **2a**, and furthermore the 1,3-dicarbonyl adduct **4a** *via* an intramolecular Friedel–Crafts-type reaction furnished 1,2,3-trisubstituted cyclopenta[*b*]indole **5a**. It is worth mentioning that the current method also constitutes a potential alternative to the traditionally employed methods for indole cyclopentannulations such as [3+2]-cycloadditions, Nazarov cyclizations, *etc.*¹⁴ Furthermore, synthesis of cyclopentannulated indoles *via* 2-indolylmethyl cation intermediates is underexplored.^{3h}

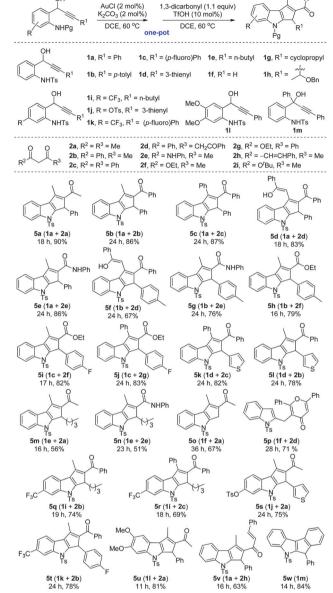
In order to further improve the efficiency of the reaction, we opted to investigate the influence of Brønsted acids in place of Lewis acids in step-II. Among several Brønsted acids explored, trifilic acid mediated reaction in DCE delivered the cyclopentannulated indole 5a in very good yield (Table 1, entries 10–16). Subsequently, influence of the co-catalyst (base) in step-I was also studied. While no desired product was observed when sodium bicarbonate was employed as the co-catalyst, sodium carbonate and an organic base such as triethylamine furnished 5a only in moderate yields (Table 1, entries 17–19). Gratifyingly, further enhancement in the yield was observed when TfOH loading was reduced from 20 mol% to 10 mol% (Table 1, entry 20). Striking temperature dependence was realized when the TfOH reaction at room temperature failed to generate the desired product 5a and only the acetylacetone adduct 4a was isolated (Table 1, entry 21). Further attempts to enhance the yield were not encouraging (Table 1, entries 22-24).

In order to validate the generality of the unprecedented method for the synthesis of polysubstituted cyclopenta[*b*]indoles, a variety of 1-(2-aminophenyl)prop-2-ynols **1a–1m** were synthesized according to literature methods^{6,8} and subjected to the optimized conditions, Table 2. It is noteworthy that the relay Au(i)/Brønsted acid catalyzed tandem transformation is quite general, and a diverse library of annulated indoles can be rapidly accessed in good to excellent yields (Table 2, **5a–5w**). The reaction displays significant tolerance towards various alkynols bearing electrondonating as well as electron-withdrawing aryl groups (for example, *p*-tolyl and *p*-fluorophenyl), heteroaryls such as 2-thienyl, and the alkyl groups. Furthermore, alkyl and aryl-bearing 1,3-diketones, 1,3,5-triketones, β -ketoesters and β -ketoamides are also welltolerated under the reaction conditions.

As can be seen from Table 2, amino alcohol **1a** upon reaction with di- and triketones **2a–2d**, and ketoamide **2e** generated the respective **1**,2,3-trisubstituted cyclopentannulated indoles **5a– 5e** in excellent yields. Similarly, the analogous aryl and heteroaryl amino alcohols **1b–1d** upon reaction with a variety of diketones, ketoesters and ketoamides generated functionalized cyclopenta[*b*]indoles (Table 2, **5f–5l**). Significantly, reaction of

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1,3-dicarbonyls in the relay Au(I)/Brønsted acid catalyzed tandem process^{a,t}



^a Reaction conditions: a mixture of amino alcohol 1 (0.1 mmol), AuCl (2 mol%), K₂CO₃ (2 mol%) and DCE (1 mL) in a 5 mL glass vial was stirred at 60 °C. After complete consumption of the starting compound (1), 1,3-dicarbonyl 2 (0.11 mmol) and TfOH (10 mol%) were added successively and stirring was continued at 60 °C until the complete disappearance of the respective 1,3-dicarbonyl adduct (4). ^b Isolated yields after silica gel column chromatography.

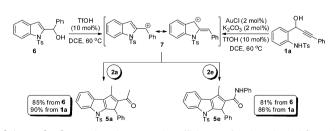
1e, having a pendent alkyl group on the acetylenic carbon centre, with diketones 2a and 2e also furnished the respective annulated indoles 5m and 5n though in moderate yields,¹⁵ but enhanced the scope of this method. Even 1,2-disubstituted cyclopentannulated indoles (such as 50) could be efficiently generated by the reaction of unsubstituted alkynol 1f with diketone 2a. But reaction of 1f with triketone 2d, unexpectedly formed the pyranone indole 5p. Complex indole derivatives bearing electron-withdrawing as well as electron-donating substituents (such as -CF3, -OTs, -OMe) on the indole moiety could also be accessed easily in high yields (Table 2, 5q-5u). However, contrary to our expectation, the amino alcohols 1g and 1h failed to deliver the desired products under the reaction conditions. Presumably, the presence of an acid sensitive cyclopropyl system and a 2-butyne-1,4oxygenated system would have triggered unwanted side reactions. The molecular structure of a representative example 5v, obtained by the reaction of **1a** and **2h**, was unambiguously confirmed by single crystal X-ray diffraction analysis (see the ESI⁺ for details).¹⁶

On the other hand, reaction of amino tert-alcohol 1m even in the presence of 2a generated only indole 5w in 84% yield via a Nazarov electrocyclization reaction (Table 2).6a Thus, indoles of the type 5w could now be accessed in excellent yields with less Au(I) catalyst loading and under milder reaction conditions over the existing method.6a

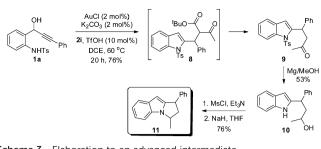
Since alcohol 6 and cationic intermediate 7 are believed to be the intermediates in the transformation of 1a to 5a, we planned to undertake a comparative study between the reactions of amino alcohol 1a and 2-indolyl carbinol 6 under the optimized conditions, Scheme 2. It can be noted that the reaction of amino alcohol 1a with 2a or 2e is found to be efficient in generating 5a or 5e, respectively, when compared to the reaction of alcohol 6 in forming 5a and 5e, thereby clearly demonstrating the advantage of the one-pot tandem process. It is worth mentioning that the direct Friedel-Crafts-type alkylation of unmodified 2-indolyl carbinols and 1,3-dicarbonyls as such is unprecedented¹⁷ and of course the subsequent cyclization cascade as well.

To further illustrate the generality and synthetic utility of this methodology, we considered an elaboration (see Scheme 3). Thus, reaction of 1a with ketoester 2i under the optimized conditions furnished adduct 8, which underwent smooth in situ decarboxylation to form β -branched 4-(2-indolyl)-2-butanone 9 in 76% yield, synthesis of which otherwise would require a multistep sequence. Indole 9 upon reaction with excess Mg in methanol generated alcohol 10 by undergoing simultaneous tosyl deprotection and ketone reduction. Selective O-mesylation and subsequent intramolecular N-alkylation¹⁸ conveniently generated 1,3-disubstituted dihydropyrroloindole 11, an important motif prevalent in a number of pharmaceutically important compounds and natural products.¹⁹

In conclusion, we have developed a general and efficient relay Au(1)/Brønsted acid catalyzed one-pot tandem process for the synthesis of medicinally significant 1,2-di- and 1,2,3-trisubstituted cyclopentannulated indoles from 1-(2-aminophenyl)prop-2-ynols



Scheme 2 Comparison between the efficiency of amino alcohol 1a and indolyl carbinol 6 in forming the same end product. First demonstration of a direct reaction between 2-indolyl carbinols and 1,3-dicarbonyls.



Scheme 3 Elaboration to an advanced intermediate.

and 1,3-dicarbonyls. Key features of this method are its readily accessible starting compounds and atom, step and pot economy. During the course of our investigation, we also developed novel Au(i)/base mediated conditions for the synthesis of indolines starting from 1-(2-aminophenyl)prop-2-ynols. In addition, we have demonstrated for the first time, Friedel–Crafts-type alkylation of unmodified 2-indolyl carbinols and 1,3-dicarbonyls. A study regarding the application of this methodology for the synthesis of biologically active natural products is currently underway in our laboratory and will be communicated shortly.

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- 11 Other NH-protecting groups (Ms, Boc, and Ac) were also evaluated prior to proceeding to optimization. Only *N*-sulfonyl propargyl alcohols generated the desired product (see the ESI[†] for details).
- Step-I was found to proceed very slowly at room temperature. After several attempts, the reaction temperature was optimized to 60 °C.
 Earlier, Au(i)/Lewis acid, ^{6a} Ag(i), ^{6b} Pd, ^{8a,b} Cu, ^{8c} conditions were reported
- 13 Earlier, Au(i)/Lewis acid, ^{6a} Ag(i), ^{6b} Pd, ^{8a,b} Cu, ^{8c} conditions were reported for the conversion of 1 to 3. For Au(i)/base mediated cyclization of 2-(1-hydroxyprop-2-ynyl)phenols to dihydrobenzo-furans, see: H. Harkat, A. Blanc, J.-M. Weibel and P. Pale, *J. Org. Chem.*, 2008, 73, 1620.
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