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A covalent organic cage compound acting as a supramolecular shadow mask for the regioselective functionalization of C_{60}

A trigonal-bipyramidal covalent organic cage compound readily forms highly stable 1:1 complexes with fullerenes C_{60} and C_{70} . These capsules serve as nano-sized shadow masks to guide the exohedral functionalization of encapsulated C_{60} . Prato reactions are limited to threefold addition and the preferred *trans*-3, *trans*-3, *trans*-3 addition pattern perfectly matches the symmetry and spatial arrangement of the three cage windows.

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A covalent organic cage compound acting as a supramolecular shadow mask for the regioselective functionalization of C₆₀†‡

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A trigonal-bipyramidal covalent organic cage compound serves as an efficient host to form stable 1 : 1-complexes with C₆₀ and C₇₀. Fullerene encapsulation has been comprehensively studied by NMR and UV/Vis spectroscopy, mass spectrometry as well as single-crystal X-ray diffraction. Exohedral functionalization of encapsulated C₆₀ via threefold Prato reaction revealed high selectivity for the symmetry-matched all-*trans*-3 addition pattern.

Introduction

Fullerenes act as spherically arranged electron-deficient polyolefins.¹ Their unique electronic structures facilitate potential applications in medicine,² photovoltaics³ and organic electronics.⁴ Synthetically, addition reactions, *e.g.*, Diels–Alder,⁵ Bingel,⁶ and Prato⁷ reactions or other dipolar cycloadditions,⁸ exploit the inherent strain energy of the curved double bonds, thus leading to a multitude of exohedrally functionalized derivatives.⁹ I_h symmetrical C₆₀ as the most abundant fullerene gives only one monoadduct. However, the number of potential regioisomers quickly raises to 8 (ref. 9d) and 46 (ref. 9a and e) for bis and trisaddition, respectively. The precise spatial organization of the individual addends identifies such multiple adducts as highly promising building blocks for 3D molecular architectures, *e.g.*, dendritic systems¹⁰ or extended metal-organic¹¹ and supramolecular¹² frameworks. However, regioselectivity is generally governed by a combination of statistical and kinetic factors.¹³ Thereby, the second addition step usually serves as a bottle neck for the selective formation of higher adducts and individual regioisomers have to be purified, if at all possible, by tedious HPLC chromatography. In some cases, specific addition patterns such as C₆₀Cl₆,¹⁴ C₆₀Ph₅H,¹⁵ or

octahedral hexakisaddition^{9c,16} have been obtained in surprisingly high purity under thermodynamic control.

Following a tether-directed remote functionalization approach that was introduced by the Diederich group,¹⁷ all seven sterically possible bisadducts for methanofullerenes have been obtained by means of a covalent fixation of the two reactive sites in suitable tether systems.^{9a} For higher adducts however, only a few tethers for privileged addition patterns are so far available.¹⁸ Elaborate tether synthesis and limitations in post-synthetic modification further hamper this approach. In contrast, any supramolecular prealignment of the reactants or an *in situ* activation of specific double bonds is much more tempting. Seminal work by Guldi, Torres and coworkers on Prato reactions for a phthalocyanine aldehyde showed the highly selective formation of *cis*-1 bisadducts mediated by π–π interactions between the chromophores.¹⁹ More recently, von Delius and coworkers reported the preferred formation of *trans*-bisadducts for Bingel reactions on C₆₀@[10]cyclo-paraphenylene.²⁰ When applying more sophisticated cage structures with a spatially precise orientation of multiple pore windows as hosts, the regioselective formation of higher adducts appears feasible (Fig. 1).

In recent years, a large variety of organic and metallosupramolecular cage receptors for fullerenes have been reported.²¹ Selective binding was utilized for the separation of fullerene mixtures²² and electron transfer was studied for dye-attached complexes.²³ Nitschke and coworkers reported on the selective encapsulation of Diels–Alder bisadduct mixtures within the cavity of an Fe^{II}L₆ cage. However, the regioselectivity for the encapsulation has not been investigated.²⁴ The Clever group used a bowl-shaped [Pd^{II}L₃(MeCN)₂]⁴⁺ cage as a supramolecular protecting group for the selective monofunctionalization of encapsulated C₆₀.²⁵ The square-planar arrangement of the four cage windows for a metallosupramolecular fullerene sponge was recently utilized for the exclusive formation of all-*e* Bingel tetrakisadducts of C₆₀, even

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‡ C₆₀⊂1: C₆₀⊂C₁₀₆H₇₂B₆O₄·(C₇H₁₁BrO₄)₂, *M* = 2801.23 g mol⁻¹, monoclinic, *C*2/*c*, *a* = 23.3258(10), *b* = 22.5169(10), *c* = 31.3889(10) Å, α = 90, β = 91.177(2), γ = 90°, *V* = 16 482.7(12) Å³, *Z* = 4, ρ_{calc.} = 1.129 g cm⁻³, μ_(CuKα) = 1.129 mm⁻¹, *T* = 100(2) K; 168 505 independent measured reflections. *F*² refinement, *R*₁ = 0.1264, *wR*₂ = 0.4437 (observed), 16 220 independent observed reflections (*R*_{int} = 0.0337) [*F*₀] > 4σ(|*F*₀|), 2θ ≤ 144.72°, 1100 parameters, 490 restraints.





Fig. 1 (a) Supramolecular complexation of fullerenes C₆₀ and C₇₀ within the cavity of covalent organic boronate ester cage **1** (molecular structures of inclusion complexes are derived from semiempirical PM6 modeling); (b) schematic representation for the use of C₆₀-cage complexes as shadow masks for a regioselective exohedral C₆₀ functionalization.

under catalytic conditions.²⁶ This masking strategy provides a significant improvement over the multistep orthogonal transposition approach²⁷ as the so far only practicable synthetic route for this addition pattern. Despite these initial examples, no cage templates to control the inherently less selective Prato reaction have been reported so far.

Results and discussion

Recently, we utilized the unique orthogonal arrangement of the three aromatic wings in C_{3v} symmetrical hexahydroxy tri-benzotriquinacenes (TBTQs)²⁸ for the design and synthesis of covalent organic cage compounds^{21f,29} *via* dynamic covalent boronate ester formation. By varying the bite angle of the diboronic acid counterpart, a series of cages³⁰ with trigonal-bipyramidal, tetrahedral or cubic shape and sizes ranging from 1.9 to 3.5 nm (solvodynamic diameters derived from DOSY NMR) were obtained. As fullerene binding was already shown for some π -extended TBTQ derivatives,³¹ we tested the host properties of these cages. After saturation with either C₆₀ or C₇₀ by standing over pristine fullerenes for 24 hours at room temperature, fullerene uptake was analysed by UV/Vis absorption spectroscopy. No additional absorption in the range of 350 to 600 nm, which would indicate fullerene complexation, was observed for CHCl₃ solutions of a methoxy-protected TBTQ precursor and a large cubic cage (Fig. S5 and S6[†]). In case of trigonal-bipyramidal [2 + 3] cage **1** however, a linear correlation between the host concentration and additional absorption features characteristic for C₆₀ and C₇₀, respectively, clearly indicated uptake of both fullerenes (Fig. S7[†]).

Further evidence for encapsulation was obtained by MALDI-TOF MS. The isotope patterns for the major signals at $m/z = 2323.57$ and 2442.67 are in excellent agreement with the respective 1 : 1 host-guest complexes C₆₀⊂**1** and C₇₀⊂**1** (Fig. 1). Remarkably, no signals for the empty cages besides

small peaks for pristine C₆₀ or C₇₀ are detected for 1 : 1 mixtures (Fig. 2a and b). Semiempirical molecular modeling on the PM6 level revealed that the two TBTQ units in **1** are indeed perfectly preorganized for efficient encapsulation of one fullerene molecule. The distance of the two TBTQ *closo*-atoms in **1**, C₆₀⊂**1** and C₇₀⊂**1** only slightly changes from 1.55 to 1.66 and 1.69 nm (Fig. S11[†]). In the case of the larger cubic assemblies, only empty cages and no complexes were detected in MALDI measurements (Fig. S10[†]). Here, the increased distance between two TBTQs does not allow any favourable interaction of one fullerene with two TBTQs, thus preventing complexation.

For a quantitative analysis, CHCl₃ solutions of fullerenes C₆₀ and C₇₀ were titrated with a stock solution of **1** and absorption changes were plotted against cage concentration (see ESI[†] for details). Global fitting to a 1 : 1 binding model revealed complexation constants of $6.3 \pm 0.4 \times 10^5 \text{ M}^{-1}$ and $5.3 \pm 0.4 \times 10^5 \text{ M}^{-1}$ for C₆₀ and C₇₀, respectively (Fig. 2c and d). Apparently, there is no significant preference for either C₆₀ or C₇₀, which is presumably attributed to opposing effects of better size and shape matching for C₆₀ and larger dispersion interactions for C₇₀.

Ultimately, single crystals suitable for X-ray diffraction were obtained during reactivity studies (see below) for the C₆₀ complex. C₆₀⊂**1** crystallizes in the monoclinic space group C2/c. Fig. 3b shows an ORTEP representation of the 1 : 1 inclusion complex. Whereas the host could be nicely refined, much higher disorder was observed for the guest indicating a very low barrier for internal C₆₀ rotation.

Specific shifting of cage protons in ¹H NMR spectra of **1** in CDCl₃ in the presence of 3 equivalents of solid C₆₀ revealed instantaneous complex formation (Fig. 3a). The rather small shifts might be explained by the assumption that spherical C₆₀ is freely rotating within the cavities and the fact, that the para



Fig. 2 MALDI-TOF MS (TCNQ, CHCl₃) for 1 : 1 mixtures of **1** and (a) C₆₀ and (b) C₇₀ (insets show comparison of measured and simulated isotope patterns); UV/Vis titration experiments for complex formation between **1** and (c) C₆₀ (black **1**, purple C₆₀, dotted red **1** + C₆₀, red C₆₀⊂**1** derived from global fit to 1 : 1 model, inset shows fit to 1 : 1 model at $\lambda = 329 \text{ nm}$) and (d) C₇₀ (black **1**, brown C₇₀, dotted blue **1** + C₇₀, blue C₇₀⊂**1** derived from global fit to 1 : 1 model, inset shows fit to 1 : 1 model at $\lambda = 379 \text{ nm}$).





Fig. 3 (a) $^1\text{H-NMR}$ spectra (400 MHz, CDCl_3 , rt) for complexation of **1** with C_{60} , (b) ORTEP representation of single-crystal X-ray structure of $\text{C}_{60} \subset \mathbf{1}$ (thermal ellipsoids were set at 50% probability, C = grey, O = red, B = purple, H atoms were omitted for clarity) and (c) packing of $\text{C}_{60} \subset \mathbf{1}$ in the solid state indicating the accessible C_{60} surface in the three cage windows.

and diamagnetic ring currents within the five- and six-membered rings of C_{60} are almost cancelling each other out. Directly after mixing, residual signals that can be attributed to free boronic acid and catechol moieties (Fig. 3a) suggest that encapsulation is achieved *via* partially opened cages and not *via* direct slipping. After one hour, cages **1** are again fully closed and quantitative complex formation is observed in case of equimolar mixtures or excess C_{60} . For substoichiometric fullerene concentrations, the appearance of two separate sets of signals for both **1** and $\text{C}_{60} \subset \mathbf{1}$ suggests a rather high kinetic stability for the assemblies, at least on the NMR time scale. For $\text{C}_{70} \subset \mathbf{1}$, chemical shifts are more prominent, with the significant low-field shift for the aromatic TBQ protons being particularly apparent (Fig. S2 \dagger). Therefore, we speculate that C_{70} is complexed in a more rigid fashion with the benzenoid hexagons of the equatorial belt being in close proximity to the axial TBQ units and the fivefold rotational axes located along one of the cage windows. This model is also in accordance with semiempirical PM6 calculations (Fig. S1 \dagger). However, the exchange of the encapsulated C_{70} between the three windows should be fast, at least on the NMR time scale, as we did not observe any splitting of the signals for **1** in $^1\text{H-NMR}$. In the $^{13}\text{C-NMR}$ spectrum of $\text{C}_{60} \subset \mathbf{1}$ (Fig. S3d \dagger), a slight upfield shift from 143.24 to 141.65 ppm was observed for C_{60} upon complexation (Fig. S3e \dagger). Due to the lower solubility of C_{70} , no $^{13}\text{C-NMR}$ spectrum could be measured in CDCl_3 . For $\text{C}_{70} \subset \mathbf{1}$ however,

five signals at 129.01, 143.72, 146.18, 146.63 and 149.63 ppm appeared, which are in very good agreement with values reported for C_{70} in C_6D_6 (130.28, 144.77, 146.82, 147.52, 150.07).³²

Based on these analytical data, we concluded that the 1 : 1 complex $\text{C}_{60} \subset \mathbf{1}$ forms with very high thermodynamic and kinetic stability. A space filling model from the X-ray structure (Fig. 3c) further reveals that the aromatic scaffold of **1** shields a significant part of the C_{60} surface apart from the three cage windows arranged in a trigonal planar fashion. Therefore, we speculated on the potential of $\text{C}_{60} \subset \mathbf{1}$ acting as a supramolecular template for exohedral C_{60} functionalization. Assuming that each window can only accommodate one addend due to steric constraints, a limitation to trisaddition was anticipated. Furthermore, *trans*-3 and neighbouring addition patterns, which approach the 120° angle between two individual cage windows, should be favoured (Fig. 6 and S16 \dagger).

The Bingel⁶ and Prato⁷ reactions are the two most common methods for exohedral functionalization of fullerenes. For the twofold Bingel reaction, there is an intrinsic preference for *trans*-3 and *e*-addition,¹³ whereas *cis*-addition is significantly disfavoured due to the steric demand of the ester substituents at the cyclopropane rings. For higher adducts, any pre-existing addends in *e*-position increasingly favour further *e*-addition, ultimately leading to the highly preferred T_h symmetrical hexakisaddition pattern.³³ Building upon this inherent selectivity, Ribas and coworkers reported on the exclusive formation of the all-equatorial tetrakisadduct for Bingel reactions on a complex of C_{60} within a tetragonal prismatic metallocupramolecular cage.²⁶ For the Prato reaction however, a much broader distribution of all eight possible bisadducts is usually observed (Fig. S12b \dagger).³⁴ Only *trans*-1 and *cis*-1 isomers are formed in lower yields due to statistical (*t1*) and steric (*c1*) factors. Substantial *cis*-additions for smaller Prato addends generally result in very complex mixtures of higher adducts. In contrast to Bingel reactions, the isolation of specific regioisomers has only been reported in rare cases.^{34a}



Fig. 4 (a) Cage-templated synthesis of *N*-methylfulleropyrrolidine trisadducts; (b) relative yields (based on HPLC integration) for unreacted C_{60} (black) and Prato adducts **2** (mono, blue), **3** (bis, green) and **4** (tris, red) for control experiment and cage-templated reactions.





Fig. 6 Accessibility of specific addition patterns in the cage windows for a second addition on $2\text{C}1$ (top, first addend purple) and a third addition on $3\text{-eC}1$ (middle left, e-addends in red), $3\text{-t}4\text{C}1$ (middle right, t₄-addends in blue), $3\text{-t}3\text{C}1$ (bottom left, t₃-addends in green) and $3\text{-t}2\text{C}1$ (bottom right, t₂-addends in yellow).

separation. Finally, the four different trisadducts were isolated as pure regioisomers in a one-pot-reaction starting from pristine C₆₀ with a combined yield of 16%. All products were analyzed by MALDI-TOF MS (Fig. S22†) and comparison of UV/Vis (Fig. S23†) and ¹H-NMR spectra (Fig. S20 and S21†) to literature data.^{34a} When compared to the rather nonselective control reaction, the selectivity for the cage-templated reaction is completely reversed and the isolated yield for $4\text{-t}3\text{,t}3\text{,t}3$ as the main product is increased by a factor of 30 (Fig. S24†). The small pore windows of **1** enforce the formation of the unfavoured t₃,t₃,t₃-addition pattern that would otherwise only be obtained in trace amounts. On the other hand, the cage walls prevent overreaction to higher adducts. In future work, we plan to optimize reaction conditions for a further increase in isolated amounts of pure trisadducts, modify the cage windows to enhance the selectivity and develop more stable cages to perform more sensitive reactions.

Conclusions

In summary, we have demonstrated the efficient fullerene encapsulation within the pores of trigonal bipyramidal covalent organic cage **1**. Due to suitable preorganization of its two TBTQ moieties, strong 1 : 1 complexes C₆₀⊂**1** and C₇₀⊂**1** are formed in organic solvents and binding constants of $6.3 \pm 0.4 \times 10^5$ and $5.3 \pm 0.4 \times 10^5 \text{ M}^{-1}$, respectively, have been determined by

UV/Vis titrations. As evidenced by an X-ray structure for C₆₀⊂**1**, only three distinct parts of the C₆₀ surface are accessible for exohedral functionalization. In prototypical Prato reactions on C₆₀⊂**1**, remarkable selectivity for the symmetry-matched trisadduct $4\text{-t}3\text{,t}3\text{,t}3$, which is formed only in trace amounts in cage-free reactions, was observed. Cage **1** acts as an efficient supramolecular shadow mask, whose small pore windows force the formation of an otherwise unfavoured trisaddition pattern. These exciting findings are the first example for a supramolecular control of the regioselectivity for the intrinsically nonselective Prato reaction. They will pave the way for novel applications of covalent organic cage compounds as effective templates for spatially precise reactions on large spherical π-systems.

Conflicts of interest

There are no conflicts to declare.

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