



Cite this: *CrystEngComm*, 2020, 22, 8236

Single crystal growth of GdB_6 by the optical floating-zone technique

Shu-yu Ning,^{ab} Xinyu Yang,^{ID} ^{*ab} Yan Wang,^{ab}
Zhongwen Zhu^{ab} and Jiuxing Zhang^{*ab}

A high-quality and large-sized GdB_6 single crystal was successfully prepared by using the optical floating zone method to accurately control the temperature and composition of the molten zone. A GdB_6 [100] single crystal 25 mm in length and 5 mm in diameter was obtained. The full width at half-maximum (FWHM) of the X-ray rocking curves was 0.227° , which indicated that the as-grown GdB_6 single crystal had a high quality. The maximum thermionic emission current density of the GdB_6 single crystal was 6.73 A cm^{-2} at $U = 4000 \text{ V}$ and $T = 1873 \text{ K}$, and the average work function was 2.99 eV . Combining first-principles calculations and experiments, the work function was found to monotonically increase as the size of the GdB_6 single crystal changed from nanostructure (1D) to bulk (3D).

Received 24th August 2020,
Accepted 22nd October 2020

DOI: 10.1039/d0ce01236j

rsc.li/crystengcomm

Introduction

Gadolinium hexaboride (GdB_6) single crystal has attracted considerable research interest due to its unique and plentiful physical properties such as magnetic transport, electronic emission, high optical absorption or transmittance coefficient and so on.^{1–10} GdB_6 , with a half-filled 4f electron structure and known as a member of the rare-earth hexaboride family, has a simple cubic CaB_6 -type structure with octahedra of B in the cube corners and Gd in the cube center,¹ which is believed to be responsible for two magnetic transitions at low temperatures,^{2–5} high neutron absorption,⁶ strong light absorption in the near infrared region⁷ and low work function.⁸ In particular, the high brightness and metal-like conductivity ($\sim 10^{-5} \Omega \text{ cm}$) make the GdB_6 single crystal a promising cathode material.^{9,10} Studies showed that the cold field electron emission capability of a GdB_6 single nanowire was more than 5 times larger than that of a LaB_6 nanowire under the same conditions.^{11–13}

However, from the aspect of thermionic electron emission, GdB_6 single crystals with a large size ($>5 \text{ mm}$ in the diameter) have been investigated insufficiently and controversially up to now. The work function of GdB_6 single crystals with a large size is reported to widely range from 3.1 eV to 3.8 eV (ref. 14–17) while the work function of single-crystalline GdB_6 nanowires is only 1.5 eV ,¹¹ which is considered to have the lowest work function among the rare

earth hexaboride family. Therefore, further investigations of its thermionic emission properties including the work function are needed. It is well known that the thermionic emission properties of cathodes are sensitive to the quality of single crystals and the preparation method. Although the preparation of GdB_6 single crystals has been reported since the 1980s, the GdB_6 single crystals are always grown accompanied with GdB_4 impurities. According to the phase diagram shown in Fig. 1, GdB_6 melts incongruently by a peritectic reaction. Moreover, the composition range and melting temperatures of GdB_4 and GdB_6 are close to each other. Additionally, the volatility of Gd and B at high temperature may change the composition of the molten zone.^{14,18} All of the above show that GdB_6 is a metastable phase and easily decomposes into GdB_4 impurity during crystal growth, and this impurity can seriously weaken the quality of the as-grown single crystal. Even if a GdB_6 single crystal is cut from the as-grown crystal, the impurities have already damaged the quality of the crystal.¹⁵ Another way to avoid the formation of GdB_4 impurities is the addition of excess boron into a polycrystalline rod, which can effectively

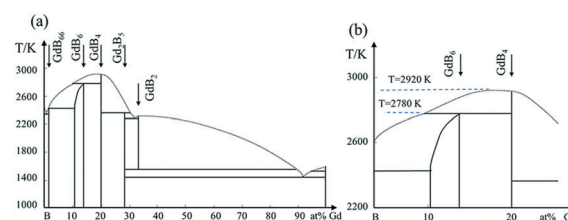


Fig. 1 (a) Phase diagram of the B-Gd binary system. (b) Schematic diagram in detail.

^a Engineering Research Center of High Performance Copper Alloy Materials and Processing, Ministry of Education, Hefei University of Technology, Hefei 230009, China. E-mail: xyinuang@hfut.edu.cn, zjiuxing@hfut.edu.cn

^b School of Materials Science and Engineering, Hefei University of Technology, Hefei 230009, China

avoid GdB₄ impurities, but excess boron can also weaken the thermionic emission properties of the GdB₆ single crystal, with the work function of which being as high as 4.2 eV.¹⁷ Therefore, the preparation of high-purity GdB₆ single crystals without GdB₄ impurity becomes worthy and necessary.

The floating zone melting technique is a widely employed method to prepare large-sized single crystals with a high melting point and extreme melt reactivity.^{19–22} Compared to conventional crystal growth methods such as the Bridgman and Czochralski methods, the primary advantages of this crucible-free method are it has no contamination or expansion/contraction problems.²³ The vertical inductive zone melting technique has been the main method used to prepare large-sized GdB₆ single crystals for decades.²⁴ Although this method has high efficiency for purifying the impurities such as oxygen, iron, carbon and so on, impurities caused by melting incongruently are difficult to avoid during crystal growth and are always formed on the periphery of crystals due to the instability of the floating zone.^{25,26} Unlike the vertical inductive zone melting technique, optical floating zone with radiation heating has high stability of the melting zone. By employing an additional technique, the optical floating zone method can make impurities completely disappear for the growth of metastable single crystals.^{27–29} However, the details of preparing GdB₆ single crystals without GdB₄ impurities by the optical floating zone method have not been given so far.

Therefore, we present the successful growth of high-quality GdB₆ single crystals without impurities by the optical floating-zone technique by strictly controlling the length of the molten zone and the lamp power in this work. The characteristics and thermionic emission properties of the as-grown single crystal were studied. The maximum emission current density of GdB₆ is 6.73 A cm^{−2} measured at 1873 K under an applied voltage of 4000 V. The average work function is 2.99 eV. Combining with first-principles calculations, the wide difference in the work function of large-sized GdB₆ single crystals and nanowires can be revealed, which is helpful to understand the difference in the work function of rare earth hexaborides with different sizes.

Experimental

Materials and methods

Commercial GdH₂ (purity 99.9%, mean particle size 35 μm, Hunan Rare Metal Material Research Institute, Changsha, China) and B powders (purity 99.5%, mean particle size 20 μm, Jiangxi Ketai Advanced Materials Co., Ltd. Gaoan, China) were firstly mixed in a stainless steel jar, which was placed in a glovebox with an Ar atmosphere and O content lower than 10 ppm. The homogenization of the powder mixtures was carried out in a planetary high energy ball mill (QM-QX2) under the process parameters of rotation rate $R = 350 \text{ r min}^{-1}$ and time $t = 3 \text{ h}$. The powder mixtures were then put into a hollow graphite die with an inner diameter of 20.4 mm in

the glovebox. After prepressing, the die was covered with porous graphite felt to maintain thermal insulation and then placed into a spark plasma sintering furnace (SPS) (LABOX-350, Sinter Land Inc., Japan). Bulk polycrystalline GdB₆ was synthesized at a sintering temperature $T = 973\text{--}1773 \text{ K}$, heating rate $R = 30\text{--}100 \text{ K min}^{-1}$, axial pressure $P = 1\text{--}12.6 \text{ kN}$ and dwell time $t = 5 \text{ min}$. After that, the furnace was then cooled to room temperature. During sintering, the temperature was monitored using an infrared thermometer.

A GdB₆ single crystal was grown by the optical floating zone melting method under an Ar atmosphere, and the furnace (FZ-T-20000-X-I-VPO-PC, Crystal Systems Inc, Japan) was equipped with $4 \times 5 \text{ kW}$ xenon lamps focused by four polished elliptical mirrors, as shown in Fig. 2. A polycrystalline GdB₆ rod 6 mm in diameter and 40 mm in length was in the upper rotary shaft, and a [100] LaB₆ single crystal with the same diameter was used as the seed crystal located on the lower rotary shaft. Thereafter, a quartz tube with a 40 mm internal diameter was loaded.

The quality and structure of the GdB₆ single crystals were characterized by inductively coupled plasma mass spectrometry (ICP-MS, X Series 2, Thermo Fisher, USA), high-resolution X-ray diffraction (XRD, X Pert Pro MPD, PANalytical B V, Holland) and X-ray Laue diffraction (JF-3, Liaodong, China). The microstructures and the component of the crystal were characterized using scanning electron microscopy (SEM, Sigma, Zeiss, Germany) with energy dispersive spectroscopy (EDS, X-MaxN, Oxford, England). Cu-K α radiation was used in high-resolution X-ray diffraction and X-ray Laue diffraction.

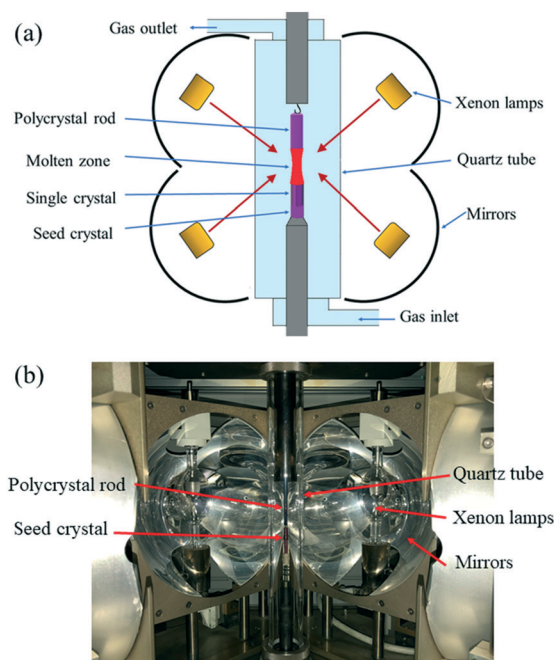


Fig. 2 (a) Schematic diagram and (b) photograph of the optical floating zone growth furnace.

Single crystal GdB_6 samples were cut into $1 \times 1 \times 3$ mm. After treatment with standard metallographic techniques, the specimen was cleaned with saturated NaOH solution and dilute nitric acid solution for 30 min, respectively. After drying, thermionic emission current tests were performed in a vacuum chamber with a pressure of 5×10^{-5} Pa. The GdB_6 was heated at 1873 K for 3 h to clean the surface before measuring the current density. The variation of thermionic emission current with the applied voltage is recorded at a temperature ranging from 1673 K to 1873 K, which was monitored using a disappearing filament optical pyrometer (WGG2-201N, SAIL, Shanghai) within $\pm 2\%$ accuracy. The distance between the anode and cathode is 0.8 mm.

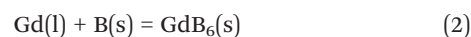
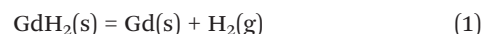
Computational methods

The calculations of the work function of GdB_6 were performed based on density functional theory (DFT) implemented in the VASP code. Ultra-soft pseudopotentials were employed together with the generalized gradient approximation (GGA) in the Perdew–Burke–Ernzerhof (PBE) exchange–correlation and a 400 eV energy cut off was used in our calculations.³⁰ An effective on-site Coulomb repulsion $U_{\text{eff}} = 8.0$ eV with the method of self-interaction correction (SIC) for the Gd-4f orbitals was applied. As shown in Fig. 3(a), 5 layer supercells with a 20 Å vacuum corresponding to the [100] lattice vectors were applied for the calculations of bulk GdB_6 . For the calculations of GdB_6 nanowires, the structures were $n \times n \times \infty$ ($n = 1, 2$, and 3) quasi 1-D nanowires with a more than 24 Å vacuum along the a and b axes, as shown in Fig. 3(b–d). $1 \times 16 \times 16$ and $1 \times 1 \times 16$ k -points were adopted for the bulk and nanowire calculations, respectively. Geometry optimization was employed since the structural distortion on the surface was not ignorable.

Results and discussion

Polycrystalline sample preparation and single crystal growth

Fig. 4 shows the preparation process of polycrystalline GdB_6 using SPS. The preparation of polycrystalline GdB_6 can be divided into two stages: dehydrogenation and liquid-phase synthesis. In the first stage, GdH_2 is decomposed into Gd metal and H_2 at a sintering temperature $T = 973$ – 1373 K and axial pressure $P = 1$ kN. An obvious peak of the gas (blue line) is observed in Fig. 4. In the second stage, the Gd metal starts to melt and form polycrystalline GdB_6 at $T = 1773$ K and $P = 12.6$ kN. The chemical reactions can be written as follows:



The XRD patterns of polycrystalline GdB_6 rods prepared at different temperatures (Fig. 5) show that a suitable temperature is key, which can make the Gd metal completely melt and react with B powders. Otherwise, GdB_4 impurities will form.

The biggest challenge in preparing a GdB_6 single crystal by the optical floating-zone technique is how to avoid the formation of GdB_4 . Firstly, the preparation process of rare-earth hexaboride single crystals such as LaB_6 and CeB_6 is employed to grow the GdB_6 single crystal, as presented in Table 1. It is prepared by a two-step process. A polycrystalline rod is firstly grown for purification with a growth rate of 20 mm h^{-1} . For the second step, a growth rate of 8 mm h^{-1} is used to obtain the high-quality single crystal. The counter rotation rate between the feed and seed rods is 15 rpm. The growth atmosphere is Ar with high purity ($\geq 99.999\%$). The total power of the lamps for stable crystal growth is 9.62 kW. However, it is found that the surface of the single crystal becomes rough during crystal growth because of irregular impurities making the surface look lacklustre (Fig. 6a). Silver-grey impurities are observed in the exterior surface and the core of the as-grown crystal, respectively, as shown in Fig. 6b, d and e. The grey impurities are demonstrated to be the GdB_4 phase by analysing the XRD patterns of the powders

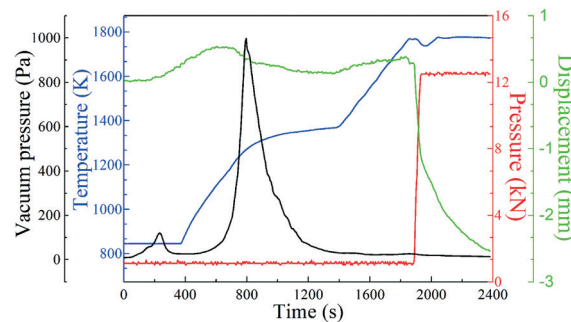


Fig. 4 Preparation process of polycrystalline GdB_6 by the SPS method.

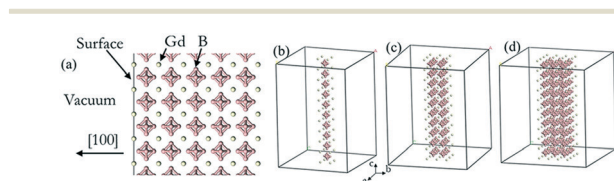


Fig. 3 (a) Gd terminations of the (100) surface of GdB_6 . (b)–(d) The structures of $n \times n \times \infty$ ($n = 1, 2$, and 3) quasi 1-D nanowires of GdB_6 .

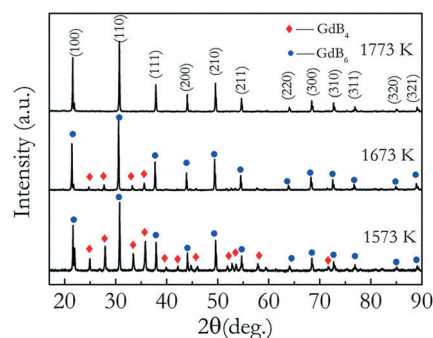


Fig. 5 XRD patterns of polycrystalline GdB_6 at different sintering temperatures.

Table 1 The process parameters for the GdB₆ single crystal grown by optical zone melting

	Growth rate	Rotation rate	Atmosphere	Length of molten zone	Lamp power
GdB ₆ crystal with GdB ₄ impurities	First step: 20 mm h ⁻¹ ; second step: 8 mm h ⁻¹	15 rpm counter rotation rate between the feed and seed rods	Argon (≥99.999%)	5.1 mm	9.62 kW
GdB ₆ single crystal without GdB ₄ impurities				4.5 mm	8.24 kW

of the crushed crystal (Fig. 6c). From Fig. 6e, the EDS results (presented in atomic ratio) further prove that the impurities

are the GdB₄ phase. The microcracks around the GdB₄ phase have a serious influence on the quality of the as-grown crystal. GdB₄ forms at the beginning and middle parts. When the crystal growth ends, B is enriched at the end of the crystal, and GdB₄ disappears. A further attempt is to increase the rate of rotation from 15 rpm to 30 rpm, but GdB₄ impurities are still obvious.

From Fig. 1, GdB₆ will easily decompose into GdB₄ and the liquid will be enriched with boron once the temperature of the melting zone is higher than 2780 K. In order to reduce the GdB₄ phase as much as possible, suitable temperature and composition uniformity of the stable molten zone are necessary. To decrease the temperature of the molten zone and maintain the stabilization of the molten zone, strict control of the lamp power were carried out. Firstly, we carefully changed the position of the four lamps and reduced the spot size of the light source to as low as possible to shorten the length of the stable molten zone (Fig. 7a), which can decrease the diffusion path of the atoms and improve the composition uniformity of the molten zone. Meanwhile, the decrease in the spot size can increase the temperature of the molten zone since the light energy per unit area is higher. So, the light power can be further reduced to 8.24 kW under the premise that the molten zone is stable. The strict control of lamp power is very important to effectively limit the volatilization of the elements and ensure the stoichiometric composition of the compound. Studies showed that if the lamp power was increased to about 9 kW, GdB₄ would form. Finally, the optimized growth conditions for the GdB₆ single crystal are listed in Table 1.

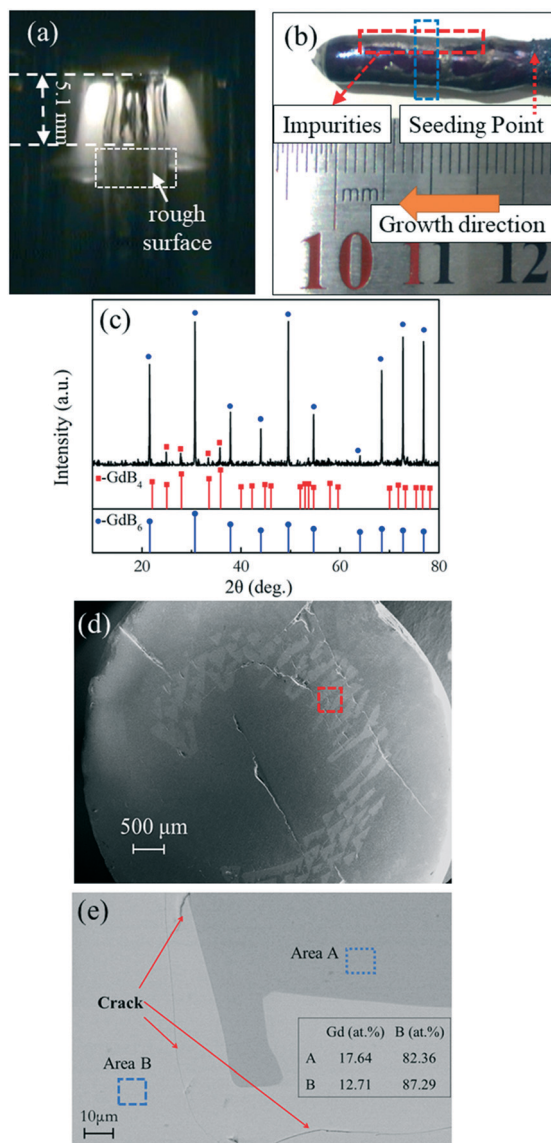


Fig. 6 (a) Photograph of the molten zone during the growth, (b) morphology, (c) XRD patterns of crushed powders, (d) secondary electron image from SEM and (e) backscattered electron image from SEM with EDS results (presented in atomic ratio) of the as-grown GdB₆ crystal with impurities. The blue dotted lines in Fig. 6(b) indicate the part for the XRD and SEM measurements. The analysed area is indicated by red dotted lines in Fig. 6(d). The areas of EDS analysis are indicated by blue dotted lines in Fig. 6(e).

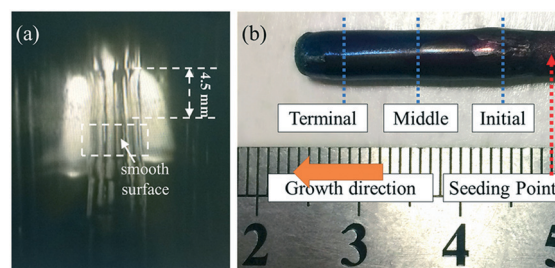


Fig. 7 (a) Photograph of the molten zone during the growth and (b) morphology of the as-grown GdB₆ single crystal. The growth direction and the seeding point are presented. The blue dotted lines indicate the parts for the SEM and ICP measurements.

Characterization of the single crystal

As shown in Fig. 7, a smooth surface of the GdB_6 single crystal is formed (Fig. 7a), and the as-grown GdB_6 single crystal 25 mm in length and 5 mm in diameter is successfully prepared (Fig. 7b). The sample is purple blue and no impurity is observed on the surface.

The elemental content at different parts of the as-grown single crystal is shown in Table 2. The ratio of B/Gd = 4.93 at the initial part of the sample. After stable growth, the ratios of B/Gd are 5.77 and 5.82, respectively, which are very close to the theoretical value of B/Gd = 6, indicating that only a GdB_6 single crystal ($\text{GdB}_{5.77}$ and $\text{GdB}_{5.82}$) forms. The SEM images of the transverse section of different parts of the as-grown GdB_6 single crystal, corresponding to the ICP measurements, are presented in Fig. 8a–c. Some GdB_4 impurities and cracks are observed in the core of the crystal at the beginning while at the middle and terminal parts, there are no impurity phases, cracks and defects. GdB_4 impurities only form at the initial stage of crystal growth because of the unstable composition in the melting zone. When the melting zone becomes stable, the pure GdB_6 single crystal is obtained.

The crystal orientation of the GdB_6 single crystal is characterized by X-ray Laue diffraction, as shown in Fig. 8b. The diffraction spot of the (100) crystal plane is in the center of the black circle, proving that the crystal orientation of the as-grown GdB_6 single crystal is (100). The diffraction spots are independent, clear and have no splitting, which shows that the as-grown GdB_6 has a good quality. The XRD pattern of the transverse section of the single crystal also shows that only the characteristic peaks of a GdB_6 single crystal with (100), (200), and (300) are identified, and no other phase is detected (Fig. 8c). The full width at half-maximum (FWHM) of the X-ray rocking curves is usually used to evaluate the quality of the crystal, and a lower value of the FWHM represents a lower dislocation density of the as-grown crystal. A FWHM value of 0.227° for the GdB_6 single crystal indicates that its dislocation density is very low (Fig. 8d). Rietveld refinement of the XRD data measured on the powders obtained from crushed single crystals was carried out using the GSAS-II program. As shown in Fig. 9, the lattice parameters obtained are $a = b = c = 4.1096(4) \text{ \AA}$ and $\alpha = \beta = \gamma = 90^\circ$ with the space group $Pm\bar{3}m$ (2θ range, $10\text{--}120^\circ$; GOF = 2.16; $R_{\text{wp}} = 3.897\%$; $R_F = 3.69\%$; $R_F^2 = 6.278\%$), which are in agreement with previous structural data.^{3,31,32}

Thermionic emission properties

Fig. 10a shows the current density of the GdB_6 single crystal tested at applied voltages $U = 300 \text{ V}$ up to 4000 V and various

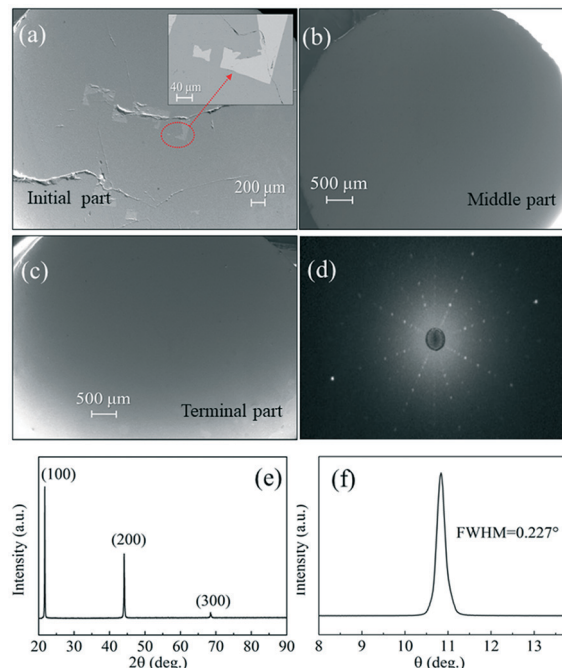


Fig. 8 (a)–(c) SEM images of the initial part, middle part and terminal part, (d) X-ray Laue diffraction photograph, (e) XRD patterns and (f) X-ray rocking curves of the GdB_6 single crystal.

cathode temperatures from 1673 K to 1873 K . The largest current density is 6.73 A cm^{-2} at $U = 4000 \text{ V}$ and $T = 1873 \text{ K}$. The cathode was cut from the middle part of the GdB_6 single crystal.

The relationship between the effective work function Φ_e and thermionic emission current density can be described by the Richardson–Dushman equation:³³

$$\Phi_e = -kT(J_0/AT^2) \quad (3)$$

where k is the Boltzmann constant, $k = 8.62 \times 10^{-5} \text{ eV K}^{-1}$, $A = 120.4 \text{ A cm}^{-2} \text{ K}^{-2}$, and J_0 is the zero-field current density and obtained from the Schottky equation:³⁴

$$\lg J = \lg J_0 + 0.192U^{0.5}T^{-1} \quad (4)$$

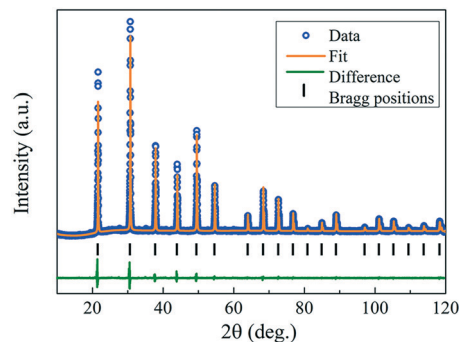


Fig. 9 Rietveld refinement results of XRD data measured on the powders obtained from the crushed as-grown GdB_6 single crystals.

Table 2 Content of gadolinium and boron measured by ICP-MS

Element	Initial part	Middle part	Terminal part
Gd	$70.02 \mu\text{g mL}^{-1}$	$42.72 \mu\text{g mL}^{-1}$	$60.42 \mu\text{g mL}^{-1}$
B	$24.03 \mu\text{g mL}^{-1}$	$16.81 \mu\text{g mL}^{-1}$	$23.92 \mu\text{g mL}^{-1}$
Ratio of B and Gd	4.93	5.77	5.82

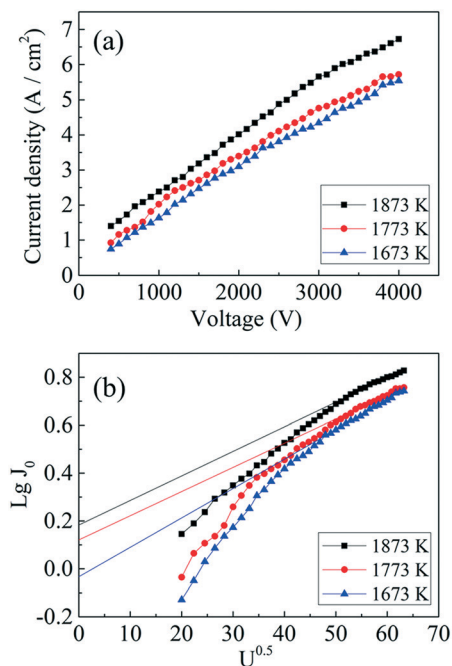


Fig. 10 (a) Thermionic emission properties and (b) Schottky curves of the GdB₆ cathode at different temperatures.

According to eqn (4), J_0 can be obtained at $U^{0.5} = 0$, as shown in Fig. 10b. J_0 is calculated to be 0.92 A/cm^2 at 1673 K, 1.32 A/cm^2 at 1773 K, and 1.52 A/cm^2 at 1873 K, respectively, and the corresponding work functions of GdB₆ are 2.84 eV at 1673 K, 2.98 eV at 1773 K, and 3.14 eV at 1873 K, respectively. The average work function of GdB₆ is 2.99 eV, which is much lower than those from previous research studies,^{14–17} indicating that the pure GdB₆ single crystals grown by the improved process have higher emission properties.

Theoretical calculation of the work function

First-principles calculation is used to explain the wide difference between GdB₆ single crystals with different sizes, and the work function can be written as:³⁰

$$\Phi = V_{\text{vac}} - E_F \quad (5)$$

where V_{vac} is the average potential in the vacuum of the cathode emission system, and E_F is the Fermi energy level of the cathode emission system. The calculated electrostatic potentials of the (100) surface of the bulk and nanowire GdB₆ along the c axis are shown in Fig. 11a and b. In the vacuum far away from the surface, the potential energy curves turn into flat straight lines rapidly, indicating the potential converging to the vacuum value. The work functions of bulk and nanowire GdB₆ calculated by eqn (5) are 2.71 and 2.11 eV, respectively. GdB₆ with a nanostructure does have a lower work function. To evaluate the influence of the size effect, the work function of nanowire GdB₆ with $n \times n \times \infty$ ($n = 1, 2,$

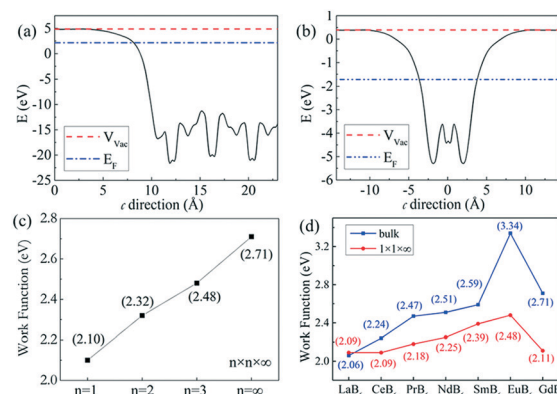


Fig. 11 Calculated electrostatic potential energy of the (100) surface of (a) bulk and (b) nanowire GdB₆. Work function of bulk and $n \times n \times \infty$ ($n = 1, 2,$ and 3) nanowires of (c) GdB₆ and (d) rare earth hexaborides.

3) is calculated, as shown in Fig. 11c. The work function monotonically increases as a function of n . When n tends to be infinite, the model is equivalent to that of the bulk structure, and the work function will gradually converge and approach the value of bulk GdB₆.

The calculated work function of 2.71 eV for the bulk GdB₆ single crystal is lower than the experimental value of 2.99 eV because the GGA method may overestimate the spatial extension of conducting bands, leading to some differences between the calculated and experimental results.³³ To make a comparison, we calculated the work function of other rare earth hexaborides, as shown in Fig. 11d. A relatively high work function compared to those of the rare earth hexaboride family belongs to the bulk GdB₆ single crystal, while the nanostructure presents a lower work function, which is consistent with previous experimental studies.^{11–13}

Conclusions

In summary, pure GdB₆ single crystals with high quality are grown by the optical float zone method. GdB₄ impurities can be effectively avoided by strictly controlling the lamp power. The single crystal with the (100) surface is confirmed to have high quality by X-ray diffraction and X-ray Laue diffraction. The thermionic emission experiment presents the largest current density of 6.73 A/cm^2 and the average work function of GdB₆ is 2.99 eV. The first-principles calculations show that the work function of the single crystal is obviously affected by the size effect, and the work function of the GdB₆ single crystal is relatively high compared to those of the rare earth hexaboride family.

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

This work is supported by the National Natural Science Foundation of China (No. 51501051), the Joint Foundation of

China's Ministry of Education in the Equipment Pre-Research (No. 6141A02022251) and the Fundamental Research Funds for the Central Universities (No. JZ2019HGTB0081).

References

- 1 S. Xu, F. Jia, Y. Yang, L. Qiao, S. Hu, D. J. Singh and W. Ren, Interplay of Electronic, *Phys. Rev. B*, 2019, **100**, 104408.
- 2 H. Hacker, J. Y. Shimada and K. S. Chung, *Phys. Status Solidi A*, 1971, **4**, 459–465.
- 3 S. E. Luca, M. Amara, R. M. Galéra, F. Givord, S. Granovsky, O. Isnard and B. Beneu, *Phys. B*, 2004, **50**, 39–42.
- 4 M. Amara, S. E. Luca, R.-M. Galéra, F. Givord, C. Detlefs and S. Kunii, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2005, **72**, 064447.
- 5 M. Anisimov, V. Glushkov, A. Bogach, S. Demishev, N. Samarin, A. Samarin, N. Shitsevalova, A. Levchenko, V. Filipov, S. Gábani, K. Flachbart and N. Sluchanko, *Acta Phys. Pol., A*, 2017, **131**, 973–975.
- 6 K. Kuwahara, S. Sugiyama, K. Iwasa, M. Kohgi, M. Nakamura, Y. Inamura, M. Arai and S. Kunii, *Appl. Phys. A: Mater. Sci. Process.*, 2002, **74**, 302–304.
- 7 L. Xiao, Y. Su, J. Ran, Y. Liu, W. Qiu, J. Wu, F. Lu, F. Shao, D. Tang and P. Peng, *J. Appl. Phys.*, 2016, **119**, 164903.
- 8 X. H. Ji, Q. Y. Zhang, J. Q. Xu and Y. M. Zhao, *Prog. Solid State Chem.*, 2011, **39**, 51–69.
- 9 S. R. Suryawanshi, N. Kanhe, V. L. Mathe, D. M. Phase, D. J. Late and M. A. More, *ChemistrySelect*, 2017, **2**, 562–566.
- 10 E. S. Zhukova, B. P. Gorshunov, G. A. Komandin, L. N. Alyabyeva, A. V. Muratov, Yu. A. Aleshchenko, M. A. Anisimov, N. Yu. Shitsevalova, S. E. Polovets, V. B. Filipov, V. V. Voronov and N. E. Sluchanko, *Phys. Rev. B*, 2019, **100**, 104302.
- 11 H. Zhang, Q. Zhang, G. Zhao, J. Tang, O. Zhou and L. C. Qin, *J. Am. Chem. Soc.*, 2005, **127**, 13120.
- 12 S. R. Suryawanshi, A. K. Singh, M. Deo, D. J. Late, S. Sinha and M. A. More, *CrystEngComm*, 2015, **17**, 3936–3944.
- 13 K. K. Yadav, M. Sreekanth, Ankush, S. Ghosh and M. Jha, *CrystEngComm*, 2020, **22**, 5473–5480.
- 14 E. K. Storms and B. A. Mueller, *J. Appl. Phys.*, 1981, **52**, 2966–2970.
- 15 K. Ostrovskii, A. A. Taran, A. V. Kovalev, V. F. Tkachenko, E. M. Dudnik and A. A. Matvienko, *Powder Metall. Met. Ceram.*, 1990, **29**, 237–240.
- 16 V. I. Lazorenko, N. I. Siman, A. I. Dmitriev and V. B. Muratov, *Powder Metall. Met. Ceram.*, 1989, **28**, 555–559.
- 17 T. Tanaka, R. Nishitani, C. Oshima, E. Bannai and S. Kawai, *J. Appl. Phys.*, 1980, **51**, 3877–3883.
- 18 E. K. Storms, *J. Appl. Phys.*, 1981, **52**, 2961–2965.
- 19 S. G. Ebbinghaus, H. Krause and F. Syrowatka, *Cryst. Growth Des.*, 2013, **13**, 2990–2994.
- 20 Y. Wang, X. Yang, S. Ning, J. Zhao, B. Xu and J. Zhang, *Vacuum*, 2019, **165**, 157–162.
- 21 S. Y. Ning, T. Iitaka, D. D. Xu, Z. Li, Y. Wang, X. Y. Yang and J. X. Zhang, *Phys. Status Solidi A*, 2019, **216**, 1800706.
- 22 Md. M. Hossain, S. Watauchi, M. Nagaoa and I. Tanaka, *CrystEngComm*, 2014, **16**, 4619–4623.
- 23 Y. B. Paderno, V. I. Lazorenko, N. I. Buryak, A. V. Kovalev, A. A. Matvienko and A. P. Galasun, *Powder Metall. Met. Ceram.*, 1983, **22**, 50–53.
- 24 E. F. Westrum, J. T. S. Andrews, B. H. Justice and D. A. Johnson, *J. Chem. Thermodyn.*, 2002, **34**, 239–250.
- 25 K. Takahashi and S. Kunii, *J. Solid State Chem.*, 1997, **133**, 198–200.
- 26 I. Mazilu, M. Frontzek, W. Löser, G. Behr, A. Teresiak and L. Schultz, *J. Cryst. Growth*, 2005, **275**, 103–107.
- 27 S. M. Koohpayeh, D. Fort, A. Bradshaw and J. S. Abell, *J. Cryst. Growth*, 2009, **311**, 2513–2518.
- 28 Q.-Y. Liu, Z.-Y. Liu, X.-B. Zhou, Z.-G. Liu, M.-X. Huo, X.-J. Wang and Y. Sui, *Cryst. Growth Des.*, 2019, **19**, 730–736.
- 29 Y.-C. Lai, G.-J. Shu, W.-T. Chen, C.-H. Du and F.-C. Chou, *J. Cryst. Growth*, 2015, **413**, 100–104.
- 30 M. A. Uijtewaald, G. A. de Wijs and R. A. de Groot, *J. Phys. Chem. B*, 2006, **110**, 18459–18465.
- 31 K. Sugiyama, Y. Koyoshi, S. Kunii, T. Kasuya and M. Date, *J. Phys. Soc. Jpn.*, 1988, **57**, 1762–1770.
- 32 Y. Takahashi, M. Fujimoto, M. Tsuchiko and K.-I. Ohshima, *J. Appl. Crystallogr.*, 2001, **34**, 208–209.
- 33 S. Y. Ning, T. Iitaka, X. Y. Yang, Y. Wang, J. J. Zhao, Z. Li and J. X. Zhang, *J. Alloys Compd.*, 2018, **760**, 1–5.
- 34 Y. Wang, J. X. Zhang, X. Y. Yang, Z. W. Zhu, J. J. Zhao, B. Xu and Z. Li, *J. Alloys Compd.*, 2018, **769**, 706–712.