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Palladium-catalyzed asymmetric allylic amination: enantioselective synthesis of chiral α -methylene substituted β -aminophosphonates[†]

中国化学学

Xubin Wang,^a Xiaoming Wang,^a Zhaobin Han,^a Zheng Wang^{*a} and Kuiling Ding^{*a,b,c}

Received 7th October 2016, Accepted 25th November 2016 DOI: 10.1039/c6qo00597g rsc.li/frontiers-organic Spiroketal backbone based diphosphine ligands (SKP) were disclosed to be highly efficient and enantioselective (94 \rightarrow 99% ee) in the palladium catalyzed asymmetric allylic amination of 2-diethylphosphonate-substituted allylic acetates, affording a series of chiral β -aminophosphonates bearing an α -methylene functionality in high yields with excellent regioselectivities.

Amino phosphonic acids¹ are phosphorus analogues of the corresponding amino acids, in which the planar and less bulky carboxylic acid group is replaced by a tetrahedral phosphonic acid functionality. In this context, β -amino phosphonic acids and their derivatives, as the isosteres of β -amino acids, have been revealed to possess a diverse range of medicinal properties including anti-bacterial,² enzyme inhibitors,³ antibiotics,⁴ anti-HIV,⁵ and anti-inflammatory⁶ activities (Fig. 1). Given their growing importance in pharmaceutical applications, it is not surprising to see that the synthesis of β -amino phosphonic acid derivatives has attracted considerable interest of the synthetic community.⁷ Whereas a number

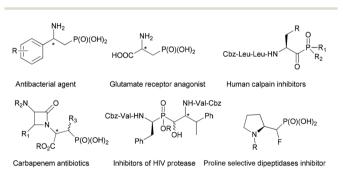


Fig. 1 Selected examples of bioactive $\beta\text{-aminophosphonic}$ acid derivatives.

of useful methods have been developed in general, there remains a dearth of efficient and versatile methodologies for the asymmetric synthesis of optically active β-amino phosphonic acid derivatives.8 Since the initial reports on the catalytic asymmetric synthesis of β -amino phosphonates *via* the amino hydroxylation of unsaturated phosphonates in the late 1990s,⁹ several types of catalytic systems have been documented to date, including the catalytic asymmetric hydrogenation of β-amidovinylphosphonates,¹⁰ asymmetric Mannich reactions,¹¹ and the desymmetrization of aziridines by phosphites.¹² Alternatively, optically enriched β-amino phosphonates have also been obtained indirectly via catalytic asymmetric nitroaldol¹³ or phospha-Michael additions¹⁴ followed by reduction of the resulting β -nitroethylphosphonates. Despite these notable advances, catalytic stereoselective protocols that can provide an efficient direct access to chiral β-aminophosphonic acid derivatives are still scarce. Herein, we report a highly enantioselective catalytic asymmetric amination of 2-(diethylphosphonyl)substituted allylic acetates, to afford a range of *a*-methylene- β -aminophosphonates in excellent optical purities.

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Recently, our group has reported the development of spiroketal-based chiral diphosphine ligands (SKP),¹⁵ a new class of diphosphines with sterically well-defined spiro backbones.¹⁶ The SKP ligands were found to be highly efficient in the Pd catalyzed asymmetric allylic amination¹⁷ of racemic ethyl 2-(acetoxy(phenyl)methyl)acrylates, a type of Morita–Baylis– Hillman (MBH) adduct,¹⁸ to give the corresponding β -arylamino acid esters with high regio- and enantioselectivities.¹⁹ Kinetic and mechanistic studies indicated that the unusual long distance of the two P atoms in the SKP ligand allows for its unique role in the reaction, *i.e.* the ligand adopts an organo- and organometallic bifunctional mode in the cooperative catalysis.²⁰ Encouraged by these results, we sought to extend the catalytic system to the asymmetric allylic amination of 2-(diethylphosphonyl)-substituted allylic acetates. The

^aState Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, China. E-mail: kding@mail.sioc.ac.cn; Fax: +(21)-6416-6128

^bUniversity of Chinese Academy of Sciences, Beijing 100049, China

^cCollaborative Innovation Center of Chemical Science and Engineering,

Tianjin 300071, China

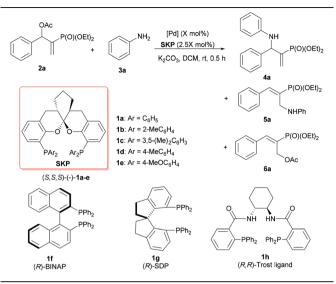
[†]Electronic supplementary information (ESI) available. CCDC 1012761. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/ c6q000597g

expected amination products would be enantioenriched β -aminophosphoric acid derivatives which can be viewed as the bioisosteres of α -methylene- β -amino acids that have very recently been found to be a key unnatural amino acid unit in a new class of endomorphin-1 analogues with potent anti-nociceptive activity.²¹ Furthermore, the olefin functionality present in the amination products may constitute a useful handle for further synthetic manipulation, thus giving ready access to a wider array of β -amino phosphoric acid derivatives.

The study was initiated by a survey of the reaction conditions, including variations in palladium sources and SKP ligands, catalyst loadings, solvents, and bases, for the amination of 2-(diethylphosphonyl)-substituted allylic acetate (2a) with aniline (3a) as the nucleophile. The reactions were generally conducted at room temperature for 0.5 h, using the complex generated in situ from a SKP ligand [(S,S,S)-1a-e] and a palladium precursor as the catalyst. The effects of solvents and bases on the reaction of 2a and 3a were examined in the presence of the $[Pd(\eta^3-C_3H_5)Cl]_2$ (1.0 mol%)/(S,S,S)-1a (2.5 mol%) catalyst, indicating that both parameters have a significant impact on the reactivity as well as the chemo-, regio-, and enantioselectivities (for details, see Table S1 in the ESI[†]). In these cases, the reaction was found to be best performed in dichloromethane in the presence of two equivalents of anhydrous K₂CO₃ as the base, to afford the chiral allylic amination product 4a in 95% yield with excellent chemo-, regio-, and enantioselectivities (4a/5a/6a = >98/<2/0, >99% ee, entry 1 in Table 1). Under these optimized reaction conditions, the effects of catalyst compositions and loadings on the reaction of 2a and 3a were further evaluated, using SKP ligands (S,S,S)-1a-e with subtle variations in their aryl substituents at the P atoms and a couple of Pd precursors. The results are summarized in Table 1. With $[Pd(\eta^3-C_3H_5)Cl]_2$ (1.0 mol%) as the palladium precursor, a sharp difference in catalytic behavior was observed among the SKP ligands (S,S,S)-1a-e (entries 1-5). For example, high activity and excellent regio-/ enantioselectivities were obtained using ligand 1a, 1c, or 1e with phenyl, 3,5-xylyl or 4-methoxyphenyl substituents, respectively, affording the targeted product 4a in high yields (92-95%) with 98-99% ee values (entries 1, 3, and 5). In contrast, ligand 1b possessing 2-tolyl moieties on the P atoms obviously deteriorates the reactivity and regioselectivity, leading to only very poor conversion (5%) and a modest branched/linear regioselectivity (4a/5a = 2/3) under otherwise identical conditions (entry 2). Intriguingly, 4-tolyl-bearing ligand 1d, with structural features analogous to both 1a and 1e, afforded much inferior chemoselectivity albeit with a 98% ee for 4a (entry 4), presumably as a result of incomplete amination of the isomerization product 6a within 0.5 h. Intriguingly, the reaction results with some privileged chiral ligands, 16 e.g., (R)-BINAP, (R)-SDP or (R,R)-Trost ligand, were less satisfactory under the otherwise identical conditions, affording incomplete conversions and moderate chemo-, regioand enantioselectivities (entries 6-8). These facts clearly indicated that SKP ligands demonstrate unique performance in the catalysis of this type of asymmetric transformation. With

 Table 1
 Catalytic asymmetric allylic amination of 2a with aniline 3a

 catalyzed by various palladium complexes of bisphosphine ligands^a



Entry	$[\mathrm{Pd}]^b (X \operatorname{mol}\%)$	Ligand ^b	Conv. ^c (%)	4a/5a/6a ^c	Yield ^d (%)	ee ^e (%)
1	$[Pd(C_{3}H_{5})Cl]_{2}(1)$	1a	>99	>98/<2/0	95	>99
2	$[Pd(C_3H_5)Cl]_2(1)$	1b	5	2/3/0	_	_
3	$[Pd(C_3H_5)Cl]_2(1)$	1c	>99	93/2/5	92	>99
4	$[Pd(C_3H_5)Cl]_2(1)$	1d	>99	68/2/30	63	98
5	$[Pd(C_3H_5)Cl]_2(1)$	1e	>99	>98/<2/0	94	98
6	$[Pd(C_3H_5)Cl]_2(1)$	1f	34	31/69/0	9	43
7	$[Pd(C_3H_5)Cl]_2(1)$	1g	16	70/30/0	10	4
8	$[Pd(C_3H_5)Cl]_2(1)$	1ĥ	57	56/34/10	28	10
9	$Pd_2(dba)_3(1)$	1a	>99	>98/<2/0	95	98
10^{f}	$Pd(OAc)_2(2)$	1a	36	36/0/64	12	88
11^f	$Pd(CH_3CN)_2Cl_2(2)$	1a	>99	44/6/50	43	95
12^g	$[Pd(C_3H_5)Cl]_2(0.5)$	1a	>99	>98/<2/0	95	98
13^g	$[Pd(C_3H_5)Cl]_2(0.1)$	1a	87	89/0/11	82	97
14^g	$Pd_2(dba)_3(0.2)$	1a	>99	>98/<2/0	95	98
15^g	$Pd_2(dba)_3(0.1)$	1a	>99	>98/<2/0	94	98
16^g	$Pd_2(dba)_3 (0.05)$	1a	64	94/0/6	57	98

^{*a*} Unless otherwise noted, the reaction was performed with **2a** (0.2 mmol) and **3a** (0.4 mmol), K_2CO_3 (0.4 mmol) in CH_2Cl_2 (2 mL) at rt for 0.5 h. ^{*b*} The molar percent of the Pd salt relative to that of **2a**. In each case, the loading of the SKP ligand was 1.25 equiv. relative to that of Pd. ^{*c*} Determined by ¹H NMR spectroscopy. ^{*d*} Yield of the isolated **4a**. ^{*e*} The ee value of **4a** was determined by chiral HPLC. ^{*f*} The loading of **1a** was 2.5 mol% relative to that of **2a**. ^{*g*} The reactions were run for 3 h.

1a as the ligand, the use of different palladium precursors also resulted in distinct catalytic activities and selectivities (entries 1 and 9–11). While the use of $Pd_2(dba)_3$ delivers excellent results nearly identical to those of $[Pd(\eta^3-C_3H_5)Cl]_2$ (entries 9 vs. 1), $Pd(OAc)_2$ or $Pd(CH_3CN)_2Cl_2$ turns out to be much less efficient, realizing only partial conversion of **2a** (entry 7) and lower yields of **4a** (entries 10 and 11), or a substantial amount of the unreacted isomerization product **6a** (entry 11). Further trials to lower the catalyst loadings were thus performed using either $[Pd(\eta^3-C_3H_5)Cl]_2$ or $Pd_2(dba)_3$ along with ligand **1a** as the catalyst, and the reaction times were prolonged to 3 h (entries 12–16). Under these conditions, the loading of $[Pd(\eta^3-C_3H_5)Cl]_2$ was lowered to 0.5 mol% without loss of either yield of selectivities (entry 12), whereas further decreasing the

Table 2 Catalytic asymmetric allylic amination of MBH adducts 2a-h with various amines 3a-j catalyzed by Pd/(S,S,S)- $1a^a$

R	OAc P(O)(OEt) ₂	~ 1	Pd ₂ (dba) ₃ (X mol%), 1a (2.5X mol%) K ₂ CO ₃ , DCM, rt, 3 h		:0)(OEt) ₂ +	P(0)(OEt) ₂	
:	2a-h	3a-j		R̃ 4a-q		5a-q	
2b: <i>m</i> -CH ₃ 2c: <i>p</i> -CH ₃ 2d: <i>p</i> -OCH ₃ 2e: <i>p</i> -F 2f: <i>m</i> -Br		$\begin{array}{c} R' \\ \hline 3a: Ph \\ 3b: p-CH_3C_0H_4 \\ 3b: p-FC_0H_4 \\ 3d: p-BC_0H_4 \\ 3d: p-BC_0H_4 \\ 3d: m-CH_2C_0H_4 \\ 3f: p-OCH_3C_0H_4 \\ 3f: p-OCH_3C_0H_4 \\ 3f: p-CC_0H_4 \\ 3f: 3f: 3A.5(OCH_{3)3}(C_0H_2) \\ 3f: Bn \end{array}$		$\begin{array}{c} R, R' \\ \textbf{4a:} H, Ph \\ \textbf{4b:} H, \mathcal{p}\text{-}CH_3C_6H_4 \\ \textbf{4c:} H, \mathcal{p}\text{-}FC_6H_4 \\ \textbf{4d:} H, \mathcal{p}\text{-}FC_6H_4 \\ \textbf{4d:} H, \mathcal{m}\text{-}HC_5C_4H_4 \\ \textbf{4e:} H, \mathcal{m}\text{-}HC_5C_6H_4 \\ \textbf{4f:} H, R' = \mathcal{p}\text{-}OCH_3C_6H_4 \\ \textbf{4f:} H, \mathcal{R}' = \mathcal{p}\text{-}OCH_3C_6H_4 \\ \textbf{4f:} H, \mathcal{A}_5C_6H_4 \\ \textbf{4h:} H, \mathcal{A}_5(OCH_3)_3(C_6H_2) \\ \textbf{4h:} H, \mathcal{A}_5(S(OCH_3)_3(C_6H_2) \\ \textbf{4h:} H, \mathcal{A}_5(C_6H_3) \\ \textbf{{4h:} H, \mathcal{A}_5(C_6H_3) \\ \textbf{{4h:} H, \mathcal{A}_5(C_6H_3) \\ {$		4j: H, Bn 4k: m-CH ₃ , Ph 4l: ρ-CH ₃ , Ph 4m: ρ-ρ-CH ₃ , Ph 4m: ρ-F, Ph 4o: m-Br, Ph 4p: ρ-NO ₂ , Ph 4q: ρ-CH ₃ , Ph	
Entry	4		X	$4/5^b$	Yield ^c	(%) ee^{d} (%)	
1	NH NH 4a	P(O)(OEt) ₂	0.1	>98/<2	94	98	
2		P(O)(OEt) ₂	0.1	95/5	91	95	
3	F, NH NH 4c	P(O)(OEt) ₂	0.1	>98/2	94	96	
4	Br NI	P(O)(OEt) ₂	0.1	91/9	87	98	
5	NI 4e	H	0.1	95/5	83	94	
6	MeO	NH P(O)(OEt	0.1	93/7	88	96	
7	NH 4g	P(O)(OEt) ₂	0.1	96/4	70	98	
8		H P(O)(OEt) ₂	0.1	96/4	84	98	
9	MeO MeO		0.1	>98/2	89	98	
10	NH Aj	Bn P(O)(OEt);	0.1	90/10	84	98	
11	NH Ak	P(O)(OEt) ₂	0.5	93/7	75	94	

	Table	2	(Contd.)
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Entry	4	X	$4/5^b$	Yield ^c (%)	$\mathrm{e}\mathrm{e}^{d}\left(\% ight)$
12	NH	0.5	96/4	92	97
	P(O)(OEt) ₂				
13	NH	0.5	94/6	80	94
	MeO 4m				
14	NH	0.5	92/8	70	96
	F 4n P(O)(OEt) ₂				
15	NH	0.5	97/3	84	98
	P(O)(OEt) ₂ Br 40				
16	NH	0.5	98/2	75	>99
	P(O)(OEt) ₂				
17	NH NH	1	61/39	40	94
	P(O)(OEt) ₂				

^{*a*} Unless otherwise noted, the reactions were typically performed at rt with 2 (2.0 mmol) and 3 (4.0 mmol), K_2CO_3 (4.0 mmol) in CH₂Cl₂ (20 mL) for 3 h, in the presence of a specified amount of catalysts Pd₂(dba)₃ and (*S*,*S*,*S*)-1a. ^{*b*} Determined by ¹H NMR spectroscopy. ^{*c*} Yield of the isolated 4a–q. ^{*d*} The ee values of 4a–q were determined by chiral HPLC. The absolute configurations for 4a–q were all determined to be *R* (see text).

loading to 0.1 mol% resulted in partial conversion and declined yield (entry 13). In this context, $Pd_2(dba)_3$ seems to be superior as the palladium precursor, and its loading can be lowered all the way to 0.1 mol% with essentially no changes in yields or ee values of **4a** (entries 14 and 15 *vs*. 9). Further lowering of the $Pd_2(dba)_3$ loading to 0.05 mol%, however, led to a significant decrease in the reactivity albeit still with a 98% ee value for **4a** (entry 16).

Subsequently, we proceeded to examine the substrate scope of the catalysis by variation in both the 2-diethylphosphonatesubstituted allylic acetates (**2a-h**) and nucleophilic amines (**3a-j**). The reactions were run under the optimized conditions with a low loading of Pd₂(dba)₃ (0.1–0.5 mol%) and (*S*,*S*,*S*)-**1a**, and the results are summarized in Table 2. Gratifyingly, excellent enantioselectivities (94 \rightarrow 99% ee) were observed in the resultant β -aminophosphonates **4a-q** (entries 1–17). Both electron-donating and electron-withdrawing groups on the phenyl rings, located whether on the aromatic amine or on the allylic acetate, are well tolerated. The regioselectivities for the amination products (**4**/5) are also generally high, ranging from

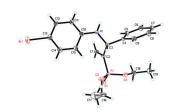


Fig. 2 X-ray single crystal structure of (R)-4d.

90/10 to >98/2 (entries 1-16). The reaction involving substrate 2h was an exception (entry 17), however, giving a much higher content of the linear amination product (4q/5q = 61/39) and a moderate yield (40%) of 4q even at a relatively high loading of the catalyst (1.0 mol%), probably as a result of unfavorable interaction with the Pd catalyst caused by the sterically congested o-tolyl group in 2h. It is also noteworthy that the stereoelectronic properties of the aromatic amines have no obvious influence on the catalysis, as reactions of 2a with a range of anilines (3a-i) gave the corresponding products 4a-i in comparable good yields, high regioselectivities and excellent enantioselectivities (entries 1-9). The amination of 2a also proceeded smoothly with benzylamine 3j, an aliphatic nucleophile, to furnish β -aminophosphonate 4j in 84% yield with a 90:10 branched/linear ratio and 98% ee (entry 10). Finally, the absolute configuration of 4d was unambiguously established to be R by the X-ray crystal diffraction analysis (Fig. 2), while those for other products were deduced to be all R by comparison of their Cotton effects with that of (R)-(-)-4d as shown in the CD spectra (Fig. S2, ESI[†]).

Conclusions

In conclusion, we have developed an efficient asymmetric synthesis of enantioenriched β -aminophosphonates bearing an α -methylene functionality. Using the complex generated *in situ* from the SKP ligand and a palladium precursor as the catalyst, asymmetric allylic amination of 2-diethylphosphonate-substituted allylic acetates proceeded smoothly under mild conditions with various amines as the nucleophiles, affording a series of β -aminophosphoric acid derivatives in good to excellent yields, high regioselectivities, and uniformly excellent enantioselectivities (94 \rightarrow 99% ee). It is noteworthy that the olefin functionality present in the chiral β -aminophosphonate products may provide a useful handle for further synthetic manipulation, and thus may stimulate future explorations to use them as intermediates to access a wider array of β -amino phosphoric acid derivatives.

Experimental

General procedures for SKP/Pd catalyzed asymmetric amination of 2-(diethylphosphonyl)-substituted allylic acetates

Into a Schlenk tube equipped with a magnetic stirring bar were added $Pd_2(dba)_3$ (1.8 mg, 0.005 mmol), (*S*,*S*,*S*)-1a (9.6 mg,

0.0125 mmol) and dichloromethane (5 mL) under a stream of argon. The solution was stirred for 5 min, followed by addition of 2 (0.5 mmol), K_2CO_3 (138 mg, 1.0 mmol) and 3 (1.0 mmol). The mixture was stirred for 3 h at room temperature, and then the solid residue was removed by filtration through a pad of Celite. The branched to linear ratio (4/5) of the amination products was determined by ¹H NMR analysis of an aliquot of the filtrate. The solvent was removed *in vacuo*, and the residue was purified by flash chromatography on silica gel with petroleum ether/EA (1/2) as the eluent to afford optically enriched α -methylene β -amino phosphonate ester 4.

The method can be applied in the Gram-scale preparation of β -aminophosphonates **4a** under a reduced catalyst loading. By following the above mentioned procedure, the reaction of **2a** (4.0 mmol, 1.24 g) with **3a** (740 mg, 8.0 mmol) proceeded smoothly at rt for 8 h in dichloromethane (38 mL) in the presence of Pd₂(dba)₃ (7.3 mg, 0.008 mmol), (*S*,*S*,*S*)-**1a** (13.2 mg, 0.02 mmol), and K₂CO₃ (1.1 g, 8.0 mmol), to give branched amination product **4a** (1.17 g, 85% yield) with 98% ee.

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