



Cite this: *Phys. Chem. Chem. Phys.*, 2024, 26, 15268

Contactless analysis of surface passivation and charge transfer at the TiO₂–Si interface†

Ramsha Khan,^{‡a} Xiaolong Liu,^{‡b} Ville Vähänissi,^{‡b} Harri Ali-Löytty,^{‡c} Hannu P. Pasanen,^a Hele Savin^{*b} and Nikolai V. Tkachenko^{‡*a}

Transition metal oxides are pivotal in enhancing surface passivation and facilitating charge transfer (CT) in silicon based photonic devices, improving their efficacy and affordability through interfacial engineering. This study investigates TiO₂/Si heterojunctions prepared by atomic layer deposition (ALD) with different pre-ALD chemical and post-ALD thermal treatments, exploring their influence on the surface passivation and the correlation with the CT at the TiO₂–Si interface. Surface passivation quality is evaluated by the photoconductance decay method to study the effective carrier lifetime, while CT from Si to TiO₂ is examined by transient reflectance spectroscopy. Surprisingly, the as-deposited TiO₂ on HF-treated n-Si (without interfacial SiO_x) demonstrates superior surface passivation with an effective lifetime of 1.23 ms, twice that of TiO₂/SiO_x/n-Si, and a short characteristic CT time of 200 ps, tenfold faster than that of TiO₂/SiO_x/n-Si. Post-ALD annealing at temperatures approaching the TiO₂ crystallization onset re-introduces the SiO_x layers in HF-treated samples and induces chemical and structural changes in all the samples which decrease passivation and prolong the CT time and are hence detrimental to the photonic device performance.

Received 6th March 2024,
Accepted 3rd May 2024

DOI: 10.1039/d4cp00992d

rsc.li/pccp

1. Introduction

Efficiently harnessing renewable energy is crucial, and silicon's (Si) abundance on Earth, coupled with its favorable 1.1 eV band gap, allows it to operate effectively within the visible to NIR range of the solar spectrum.^{1–3} As a result, it has become a fundamental component in photovoltaic (PV) cells^{4,5} and photoelectrochemical (PEC) cells.^{6,7} Among these, crystalline silicon (c-Si) based devices have emerged as highly attractive and promising devices for cost-effective, green, and sustainable renewable energy production. However, the performance of state-of-the-art c-Si-based PV and PEC devices is often limited by recombination of their surfaces and interfaces which leads to a loss in their photogenerated charge carrier collection and thus a reduction in their energy conversion efficiency. The reduction of such recombination, commonly referred to as surface passivation, can be achieved by the combined effect

of the reduction in the trap states present at the surface *via* a chemical component, and the reduction in the charge carriers available for a recombination process *via* a field effect component.⁸ Passivating layers have demonstrated significant potential in enhancing the performance of c-Si solar cells, enabling their theoretical efficiency limit to approach approx. 29.43%.^{9–11}

Transition metal oxides (TMOs) including ZnO, MoO₃, WO₃, V₂O₅, TiO₂, and CuO have attracted great interest as potential passivating layers for silicon surfaces and interfaces, offering tunable optoelectronic properties with cost-effective deposition methods.^{12–16} Sahoo *et al.* deposited 1 nm CuO as a tunneling oxide passivating layer over Si and improved the minority carrier lifetime of the Si-based solar cells.¹⁴ Gerling *et al.* deposited MoO₃, WO₃ and V₂O₅, over n-Si and improved the lifetime of carriers up to 240, 142, and 4.5 μs, respectively.¹⁷ Loo *et al.* deposited a 75 nm ZnO film over n-Si by ALD and improved the carrier lifetime up to 312 μs.¹⁸ However, among them one promising material is TiO₂ which has excellent chemical stability, dielectric properties, and high transparency that reduce problems such as surface parasitic absorbance.^{19,20} Atomic layer deposition (ALD) has emerged as a powerful method to grow TiO₂ thin films on Si substrates for passivation application.^{21–23} It has been employed extensively since it allows growth of pinhole-free high quality TiO₂ thin films at mono layer thickness with controllable and favorable optical properties.^{19,21,24,25} The properties of ALD TiO₂ thin films can

^a Photonic Compounds and Nanomaterials Group, Faculty of Engineering and Natural Sciences, Tampere University, Tampere University, P.O. Box 541, FI-33014, Finland. E-mail: nikolai.tkachenko@tuni.fi

^b Department of Electronics and Nanoengineering, Aalto University, Tietotie 3, Espoo 02150, Finland. E-mail: hele.savin@aalto.fi

^c Surface Science Group, Faculty of Engineering and Natural Sciences, Tampere University, Tampere University, P.O. Box 692, FI-33014, Finland

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d4cp00992d>

‡ Both of the authors contributed equally to this article.



be affected by substrate pre-treatments, precursor choice, growth parameters and post-deposition treatments. Recent research has demonstrated that by depositing a TiO₂ surface passivation layer on Si by ALD using TiCl₄ and H₂O precursors, the effective minority carrier lifetimes increase above 10 ms with decreasing ALD growth temperature, which is due to the increasing amount of ALD precursor traces (Cl and H) within TiO₂ films as the growth temperature decreases.²⁶ Previous studies show that the quality of surface passivation by amorphous TiO₂ thin film is sensitive to Si wafer pre-treatments and thus can be significantly optimized and increased by applying precisely conditioned annealing, particularly through a pre-ALD growth of a SiO_x interlayer.^{20,27–29} However, post-ALD deposition, the increased annealing temperature and duration can lead to a noticeable decline in surface passivation quality due to TiO₂ crystallization and associated breaking of Si–O–Ti bonds at the TiO₂–Si interface.^{28,29}

In addition to surface passivation, TMOs can also be applied over Si as corrosion protection layers. In that case, meticulous attention should be paid to the ultra-fast charge transfer (CT) (in the ps–ns time domain) from the semiconductor photoabsorber to the TMO coating, as it can significantly impact the performance of PV devices. In high-efficiency PV cells with passivating contacts, majority carriers traverse the passivating TMO layer to reach the metal contact, and similarly, in PEC cells, carriers navigate through the TMO layer to access the reactive surface. Since TiO₂ exhibits hole-blocking and electron-transporting capabilities, it can be used simultaneously for surface passivation and CT in both PV and PEC devices.^{30–32} However, earlier studies often show a trade-off between the surface passivation and CT at the TiO₂–Si interface. For example, the presence of a thin SiO_x interfacial layer can improve the passivation quality but it limits the CT process.^{33–35} Therefore, investigating and understanding the CT dynamics of photoexcited charges from Si to TiO₂^{33,36} across the TiO₂–Si interface is crucial for optimizing the design and performance of the devices that rely on these materials. Femtosecond transient reflectance (TR) spectroscopy is a time-resolved technique used to extract information about the population of photogenerated charge carriers. This technique allows us to map out charge-transfer pathways by identifying decay features in the spectra. Consequently, information about the quenching and transfer of charge carriers, along with their corresponding lifetimes, which is relevant to photonic applications, can be extracted within femtosecond to nanosecond timescales.^{37–39} Herein, we have employed TR spectroscopy to get a complete picture of the phenomena related to ultrafast changes in reflectivity, diffusion of carriers, and particularly the CT process occurring at the interface in our TiO₂/Si heterojunctions.

In this study, we prepare high-quality ALD TiO₂ thin films at 120 °C using TiCl₄ and H₂O precursors on both p- and n-type Si substrates, with different pre-ALD chemical and post-ALD thermal treatments. This ALD TiO₂ coating is amorphous and contains an elevated amount of precursor traces (~1 at% H and Cl).⁴⁰ The post-ALD thermal treatments were carefully selected to control the diffusion of H, Cl, and O within the

TiO₂ while maintaining the amorphous phase. The resulting surface passivation quality of the TiO₂/Si heterojunctions is studied using the contactless quasi steady state photoconductance (QSSPC) method to measure the minority carrier lifetime. Furthermore, to gain insights on the CT dynamics across the TiO₂–Si interface and transport of electrons in the TiO₂ layer, contactless TR spectroscopy is utilized. Additionally, a Kelvin-probe is applied to study the Si surface barrier with TiO₂ and how it contributes to the surface passivation and the CT in TiO₂/Si heterojunctions. The topography of the samples is analyzed with AFM to examine and understand the structural and compositional changes occurring within the TiO₂ thin films as a result of annealing and further how these changes contribute to the surface passivation and the CT process. Our results provide new insights on engineering effective surface passivation with optimized CT at the TiO₂–Si interface, thus, paving the way for achieving highly efficient Si-based PV and PEC devices.

2. Synthesis

Boron-doped (p-Si samples) and phosphorus-doped (n-Si samples) single-side-polished float-zone c-Si wafers (100 mm diameter, (100) orientation, 1–5 Ω cm resistivity, and 280 μm thickness) were used as substrates. Initially, all the wafers underwent thermal oxidation at 1050 °C to deactivate bulk defects.^{34,41} After removing the thermal oxide layer by a 1% HF dip, the wafers were subsequently cleaned using a standard process consisting of RCA1, HF dip, and RCA2. This resulted in a chemical SiO_x surface layer⁴² with a thickness of 1.8 nm determined by ellipsometry, assuming a refractive index of 1.465 at 632.8 nm. However, this thickness is slightly overestimated and actual thickness should be thinner than <1 nm.⁴³ Half of the samples were subjected to a second HF dip after the RCA2 step just before the ALD growth of TiO₂ to prepare TiO₂/Si samples without an interfacial SiO_x layer.^{44,45} Saari *et al.* have previously found out by XPS that the TiO₂/Si interface remains free from SiO_x for the thermal ALD TiO₂ process using TDMAT and H₂O precursors at 100 °C.⁴⁴

Right after the chemical treatments, the samples were immediately transferred to the ALD chamber to deposit the TiO₂ thin films on both sides of the wafers. The deposition was done by using a thermal ALD tool (Beneq TFS-500), which was pumped to vacuum, heated to 120 °C and stabilized for 1 hour before starting deposition. TiCl₄ was employed as the Ti precursor and water (H₂O) was employed as the oxidant.⁴⁰ A single ALD cycle consisted of a 200 ms TiCl₄ precursor pulse and a 150 ms H₂O oxidant pulse, each followed by a 3 s N₂ purge. The deposition rate was estimated to be 0.43 Å per cycle and in total 1200 cycles were used for depositing 54 nm thick films, as determined by ellipsometry and confirmed by SEM (Fig. S1, ESI†).

The thickness of the TiO₂ films was optimized for TR measurements, in order to acquire a suitable interference pattern to increase the signal strength of transferred carriers while retaining the same interface and CT rate. The steady-state reflectance spectra of all the as-deposited TiO₂/Si heterojunctions



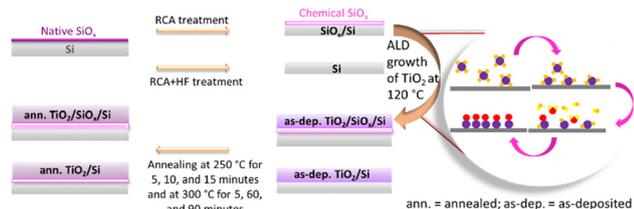


Fig. 1 Synthesis of TiO_2 on Si substrates by ALD using TiCl_4 and H_2O as precursor materials.

are shown in the ESI,[†] Fig. S2 confirming similarity of the samples, *i.e.*, the HF-treatment had only little effect on the TiO_2 film thickness.³³ Full wafers were cut to quarters and then the samples were thermally annealed in a tube furnace (ATV PEO-601) under N_2 ambience at 250 °C for 5, 10, and 15 minutes. Despite the HF treatment, some interfacial SiO_x grows between Si and TiO_2 during the thermal annealing process.²⁹ Saari *et al.* have shown that the thickness of interfacial SiO_x increases linearly with temperature in the range of 200–550 °C.⁴⁴ However, one quarter from each wafer was left without the thermal annealing of the as-deposited (as-dep.) TiO_2/Si heterojunctions, to study how the thermal annealing affects both the surface passivation and CT properties. A schematic of sample preparation is shown in Fig. 1. Similarly, to study the effect of increasing the annealing temperatures and durations, TiO_2/Si heterojunctions were annealed at 300 °C for 5, 60, and 90 minutes while one sample was left without annealing (as-deposited).

3. Results and discussion

3.1. Surface passivation provided by TiO_2

To study the passivation effect of TiO_2 thin films on Si, effective carrier lifetimes, τ_{eff} , of both n-type and p-type silicon wafers with and without HF treatment are measured. Assuming that the bulk substrate has infinite lifetime, the upper limit of the surface recombination velocity (SRV), $S_{\text{eff,max}}$, has been estimated. It can be defined as a rate at which carriers recombine at the surface and in the case of symmetrical samples it can be calculated as $S_{\text{eff,max}} \approx \frac{W}{2\tau_{\text{eff}}}$, where W is the wafer thickness.⁴⁶

Fig. 2a depicts typical injection-dependent effective lifetime results and the determined $S_{\text{eff,max}}$ for all the as dep. TiO_2/Si and $\text{TiO}_2/\text{SiO}_x/\text{Si}$ samples. Interestingly, the HF-treated as dep. samples exhibit higher lifetimes than their counterparts, the $\text{TiO}_2/\text{SiO}_x/\text{Si}$ samples. Among all the samples, the $\text{TiO}_2/\text{n-Si}$ sample shows the best passivation performance having a maximum lifetime of approximately 1.23 ms. At a carrier injection level of $1.0 \times 10^{15} \text{ cm}^{-3}$, the corresponding $S_{\text{eff,max}}$ is around 11.4 cm s^{-1} . This result is comparable to the values reported earlier for ALD TiO_2 passivated silicon.²⁶ Table S1 (ESI[†]) summarizes the lifetimes and the corresponding $S_{\text{eff,max}}$ for all the as-dep. TiO_2/Si and $\text{TiO}_2/\text{SiO}_x/\text{Si}$ samples.

The HF-treated as-dep. $\text{TiO}_2/\text{p-Si}$ shows a higher recombination rate of 20.3 cm s^{-1} as compared to the as-dep. $\text{TiO}_2/\text{n-Si}$, which shows 11.4 cm s^{-1} . For reference, for HF-treated (Si-H terminated/H passivated Si surface)⁴⁷ n- and p-Si, the $S_{\text{eff,max}}$ values were 636.4 and 241.4 cm s^{-1} , respectively. This means that with the deposition of TiO_2 thin films, the S_{eff} values are minimized for both n- and p-Si. The difference in S_{eff} between n- and p-Si can be accounted for by different surface passivation properties, as the bulk of both substrates has been proven to be of high quality (τ_{eff} above 10 ms) in separate experiments using Al_2O_3 surface passivation.^{48,49} Therefore, the dopant type in Si is found to have a minor influence on the achievable passivation levels. Loo *et al.* also observed a minor influence of the dopant type on the achievable passivation levels.¹⁸ Moreover, the results reveal a consistent pattern where the presence of RCA2 oxide increases the SRV for both p-type and n-type wafers underscoring the detrimental impact of this oxide on the passivation performance, as listed in Table S1 (ESI[†]). This observed trend is consistent with prior research, *e.g.*, carried out by Mochizuki *et al.*²⁰ and Liao *et al.*,²⁹ which attributes the increase in S_{eff} , in the case of amorphous TiO_2 coatings, to the increase in the interfacial defect density upon the formation of SiO_x at the TiO_2 -Si interface already during the annealing at 250 °C. The SiO_x layer can either introduce defect states and recombination, or it affects the total charge of the thin film and thus weakens field-effect passivation. This is contrary to the literature where the presence of a thin SiO_x layer has been shown to improve the passivation.^{27,28}

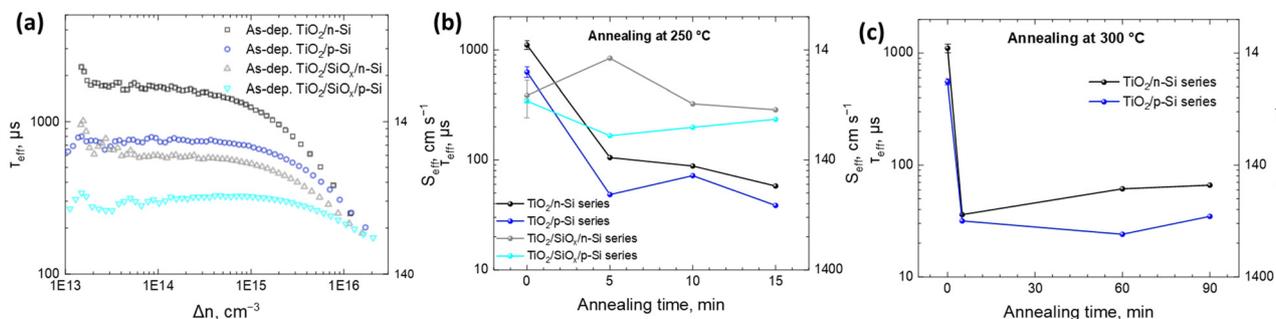


Fig. 2 (a) Injection-level dependent effective lifetimes for Si wafers passivated by 54 nm thin films of (a) as-deposited TiO_2 , (b) annealed TiO_2 films at 250 °C, and (c) annealed TiO_2 films at 300 °C for different durations. The corresponding surface recombination velocities (S_{eff}) are also shown on the right y-axis.



Earlier studies (summarized in Table S2, ESI[†]) have shown that the Si surface passivation quality by ALD TiO₂ thin films depends on the post-deposition thermal treatment. Therefore, the impact of the annealing on the S_{eff} of the samples is studied. The samples are annealed at 250 °C (Fig. 2b) and 300 °C (Fig. 2c) for different durations. The results show that the as-dep. TiO₂ grown on both HF-treated p- and n-Si substrates results in the lowest SRV while the annealing significantly enhances the SRV for all the samples. However, with annealing, a thermally introduced interfacial SiO_x layer forms in between the TiO₂ and Si which seems to prevent the further degradation of the lifetime. This leads to a situation where actually after the annealing, the SRV for samples that initially had RCA2 SiO_x at the interface is lower than the SRV for samples that initially had no SiO_x at the TiO₂-Si interface as shown in Fig. 2b. Furthermore, after annealing at 300 °C for extended durations up to 90 minutes, both HF-treated TiO₂/n-Si and TiO₂/p-Si samples show a similar increment in the SRV as shown in Fig. 2c. This observed trend is consistent with prior research, *e.g.*, carried out by Mochizuki *et al.*²⁰ and Liao *et al.*,²⁹ which attributes the increase in the SRV in the case of amorphous TiO₂ coatings to the increase in the interfacial defect density upon the formation of SiO_x at the TiO₂-Si interface already during thermal annealing at 250 °C. Therefore, it is evident that the surface passivation quality started to degrade already at the thermal treatment temperatures that were not sufficiently high to induce TiO₂ crystallization but at a temperature where diffusion of the dopants and impurities typically precedes the phase transition. This was confirmed by the Raman spectra of the samples after ALD and after 300 °C 90 min annealing (Fig. S10, ESI[†]) where no TiO₂ crystalline peaks (anatase or rutile) are observed. It is suggested that impurities in the ALD TiO₂ film originating from the ALD precursors, especially H can have a significant effect on the surface passivation.^{29,40}

Fig. 3 presents measurements of the Si surface barrier height (V_{sb}) using the Kelvin-probe method after varying annealing temperatures and durations. In TiO₂/n-Si sample series, both with and without HF-treatment, a strong negative V_{sb} is obtained which remains stable even when the samples are annealed at 300 °C up to 90 minutes. Conversely, for TiO₂/p-Si sample series, both with and without an HF-treatment, V_{sb} is slightly negative initially and then increases to slightly positive values as the annealing duration is increased. These differences

in V_{sb} values reflect the different electronic structures in p- and n-type Si.⁵⁰ This happens as the Fermi-levels are closely aligned for p-Si and TiO₂ which causes small V_{sb} , whereas the alignment of Fermi-levels for n-Si and TiO₂ causes a notable upward band bending, as indicated by the high negative V_{sb} values. The notable upward band bending should be harmful for electron transport into TiO₂ from Si and even result in a depletion of electrons at the n-Si surface, forming a hole-rich layer. The lowest SRV observed in Fig. 2 for HF-treated as-dep. TiO₂/n-Si can be attributed to the strong field effect (inversion) passivation, whereas there is nearly no surface band bending on p-Si. This, coupled with the measured favorable lifetimes and low SRVs, indicates that all the as-dep. TiO₂ thin films have a good chemical passivation quality for both p- and n-Si substrates. The values without any annealing in Fig. 3a and b are slightly different because they are from different runs of experiments and the differences are within the given error limits. However, the measured V_{sb} remains at a similar level after the thermal treatment with increased temperature and duration. The increase of absolute V_{sb} values observed in certain cases, such as the annealing of the TiO₂/n-Si sample for 5 minutes, suggests an enhancement in field-effect passivation. However, despite this improvement, the SRV experiences a significant increment. This implies that the increased SRV may stem from thermally induced interfacial defects rather than alterations in field-effect passivation.

3.2. Charge transfer through the TiO₂/Si interface

For optimal device performance, most of the minority carriers should remain in Si and there should be only majority charge carrier transfer across the TiO₂-Si interface. Therefore, to carefully analyze the charge transfer from Si to TiO₂ across the interface, contactless TR spectroscopy is utilized. Using this technique, one can measure the change in reflectance of the films upon photoexcitation. The samples were excited at 500 nm (with an energy density of 200 μJ cm⁻²) from the polished side of the samples to selectively generate the photo-carriers in Si since TiO₂ does not absorb light in the visible region of the solar spectrum. The photogenerated carrier density can be calculated using $N_0 = I \times \alpha \times e^{-\alpha x}$, where I is the intensity of the incoming light, α is the absorption coefficient, and x is the thickness at which the photocarriers are generated.³³ Therefore, at 500 nm excitation, approximately 4.8×10^9 e⁻/h⁺ pairs are generated at a depth of 1 μm in Si. The total measured signal in the case of this study is given by $\text{TR}_{\text{total}} = \text{TR}_{\text{Si}} + \text{TR}_{\text{TiO}_2}$.

However, at 500 nm excitation, only Si is excited and any TR_{TiO₂} signal must originate from the carriers transferred from Si to TiO₂, and hence is termed as the ‘‘CT signal’’. This signal is thus calculated by subtracting the estimated Si signal from the measured total signal. To examine how the charge carriers behave in Si upon photoexcitation at 500 nm, the temporal evolution of the TR spectra is observed for both pure Si substrates and TiO₂/Si samples. Since the electron transfer from Si to the TiO₂ layer is investigated, we focus mainly on samples with the n-Si substrate where electrons are the

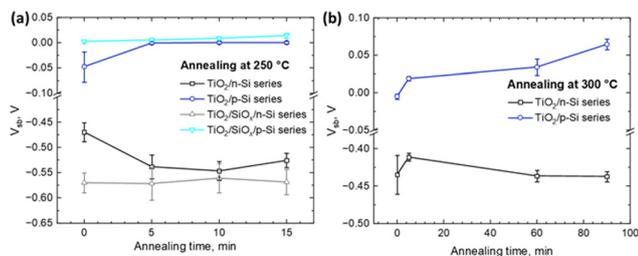


Fig. 3 Surface barrier height of different series of TiO₂/Si heterojunctions deposited at 120 °C and further annealed at (a) 250 °C and (b) 300 °C for different time durations.



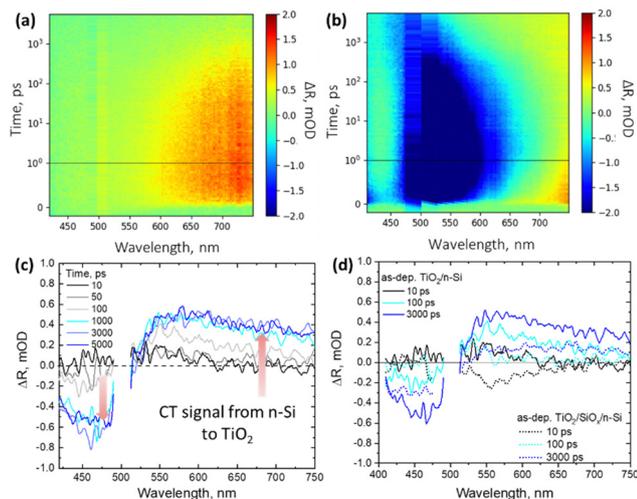


Fig. 4 Excitation of samples at 500 nm with $200 \mu\text{J cm}^{-2}$ to obtain (a) TR spectra of the HF-treated bare n-Si substrate at different delay times, (b) TR spectra of the HF-treated as-dep. $\text{TiO}_2/\text{n-Si}$ sample at different delay times, (c) extracted CT spectra in the HF-treated as-dep. $\text{TiO}_2/\text{n-Si}$ sample where the arrows indicate electrons moving from n-Si to TiO_2 , and (d) comparison of the extracted CT spectra for the HF-treated as-dep. $\text{TiO}_2/\text{n-Si}$ sample and as-dep. $\text{TiO}_2/\text{SiO}_x/\text{n-Si}$ at different delay times.

majority carriers. Hence, the TR spectra of both HF-treated pure n-Si substrate and as-dep. $\text{TiO}_2/\text{n-Si}$ samples are measured with excitation at 500 nm and the results are presented in the form of 2D maps as shown in Fig. 4a and b, respectively. For bare n-Si substrate, the TR spectra exhibit signals only in the positive range which decay as carriers diffuse from the surface to the bulk. However, with the addition of a 54 nm thick TiO_2 thin film on Si, the interference pattern changes which in turn changes the TR spectra. Therefore, in the case of $\text{TiO}_2/\text{n-Si}$ samples, at longer delay times, > 500 ps, the negative TR signal recovers partially and shows a blue shift while the positive TR signal starts evolving in the red region (600–750 nm) from 100 ps–5 ns in as-dep. $\text{TiO}_2/\text{n-Si}$.

To conduct a detailed analysis of the CT, we extract the TR spectra only related to the changes in TiO_2 as a result of the CT

from Si to TiO_2 . For this, any potential TR signal originating from Si is removed from the TiO_2/Si samples. As employed in our previous study,³³ we assume that the signal from 1 to 10 ps is entirely due to free carriers in Si which then decays as the photoinduced carriers diffuse away from the surface deeper into the bulk Si.

The carrier diffusion is assumed to be similar for both bare and TiO_2 coated Si allowing us to estimate how the Si signal would behave if there was no CT to TiO_2 . Therefore, in this case, the CT features are extracted from the TR spectra of as-dep. $\text{TiO}_2/\text{n-Si}$ (Fig. 4b) and the bare n-Si (Fig. 4a). The extracted TR signal directly corresponds to the CT process and the strength of the TR signal, ΔR , from TiO_2 correlates directly with the number of transferred electrons, *i.e.*, an increase in the positive signal in the red region (600–750 nm) and an increase in the negative signal in the blue region (420–480 nm) after 100 ps are indicative of the transfer of charge carriers from n-Si to as-dep. TiO_2 as shown in Fig. 4c. However, it can be observed that intensity of the TR signal reduces to nearly half, 0.25 mOD at 550 nm, in the case of as-dep. $\text{TiO}_2/\text{SiO}_x/\text{Si}$ samples as compared to 0.5 mOD at 550 nm for HF-treated as-dep. TiO_2/Si samples as shown in Fig. 4d. It means that nearly half of the electrons get transferred in SiO_x containing Si samples as compared to SiO_x -free Si samples. The extracted CT spectra for all the other as-deposited samples (namely as-dep. $\text{TiO}_2/\text{p-Si}$, as-dep. $\text{TiO}_2/\text{SiO}_x/\text{p-Si}$, and as-dep. $\text{TiO}_2/\text{SiO}_x/\text{n-Si}$) are analyzed similarly as shown in Fig. 4c and are presented in ESI,[†] Fig. S3–S5. Moreover, the extracted CT dynamics at different probe wavelengths enable us to compare the ultrafast responses of the different TiO_2/Si heterojunctions that experienced different pre-ALD chemical and post-ALD thermal treatments. Therefore, similar CT extraction analysis is applied to attain the decay dynamics of all the TiO_2/Si samples, and we see that the decay of bare n-Si clearly deviates from that of the $\text{TiO}_2/\text{n-Si}$ samples after 100 ps which can be again attributed to the CT feature from Si to TiO_2 . Fig. 5a shows the extracted CT dynamics of the HF-treated as-dep. $\text{TiO}_2/\text{n-Si}$ sample at different probe wavelengths.

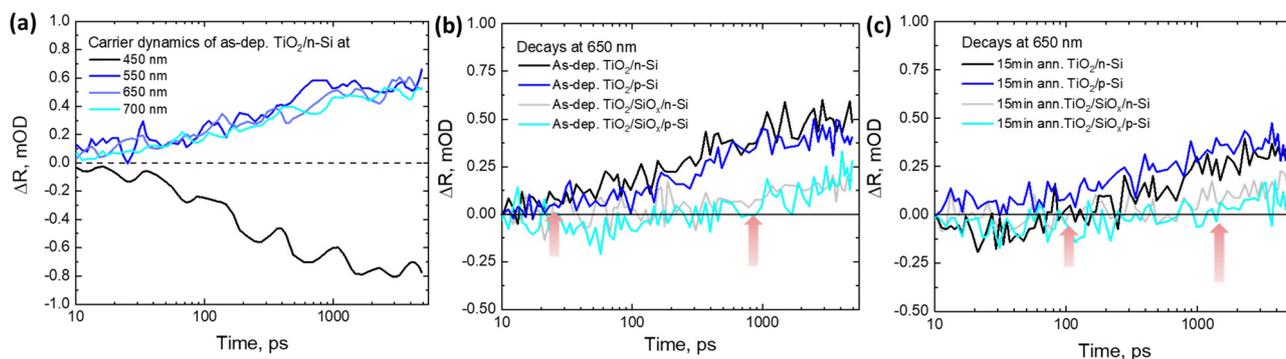


Fig. 5 (a) Extracted charge transfer dynamics after excitation of the HF-treated as-dep. $\text{TiO}_2/\text{n-Si}$ sample at 500 nm (with $200 \mu\text{J cm}^{-2}$ energy density) where the increase of the signal at different wavelengths is shown as a function of time which corresponds to the charge transfer signal increase at different wavelengths, and comparison of the extracted charge transfer signal increase at 650 nm probe wavelength for all the (b) HF-treated as-dep. TiO_2/Si samples, and (c) 15 minute annealed TiO_2/Si samples. The red arrow shows the charge transfer onset time where the charge transfer signal from Si to TiO_2 first appears above the noise range.



To compare the CT dynamics of all the TiO₂/Si samples, the CT signal increase is compared at 650 nm probe wavelength as presented in Fig. 5b and c. The results show that the HF-treatment reduces the CT time from Si to TiO₂ for both as-dep. TiO₂/n-Si and as-dep. TiO₂/p-Si samples and that the observed characteristic CT time (average time from where the CT signal increases from zero OD to the maximum OD on the 6 ns time scale of the TR measurement) is approx. 200 ps. However, for the as-dep. TiO₂/SiO_x/Si samples, the ones with chemical SiO_x at the interface, the CT time prolongs ten fold, *i.e.*, almost > 2 ns, as shown in Fig. 5b. Previously, we reported a somewhat faster CT time of 600 ps in as-dep. TiO₂/SiO_x/Si samples while a similar fast CT time of 200 ps was observed for a HF-treated TiO₂/Si samples.³³ The results also agree with the surface barrier height results as there is a larger barrier for electrons when there is a SiO_x layer. The photodynamics results also show that the CT signal does not keep increasing and saturates after approx. 1 ns. This occurs because charges accumulate near the TiO₂ side of the interface due to the low carrier mobility in TiO₂, taking approximately 1 μs to reach the surface,³³ shown in Fig. 6.

Hence, eventually, the CT process from Si to TiO₂ is inhibited. In practical devices, this problem is typically addressed by making TiO₂ thin enough (a few nm only) so that the carriers can reach the front surface despite the low mobility. As mentioned earlier, the samples were also annealed at 250 °C and 300 °C to investigate their thermal stability and its impact on the surface passivation. It can be observed that when the TiO₂ films grown on both HF-treated p- and n-Si are annealed for 15 minutes at 250 °C, the characteristic CT time prolongs to almost 1.5 ns, as shown in Fig. 5c. The annealing induces the formation of interfacial SiO_x that was previously shown to increase the CT time to 370 ps with low-defect anatase TiO₂ coating.³³ The prolonged CT time here can be rationalized by the differences between amorphous and crystalline TiO₂. While in crystalline low-defect TiO₂ electrons transfer to the conduction band, in amorphous TiO₂ electrons can transfer to the impurity mediated defect states that are strongly modified by the annealing, shown in Fig. 6.

This prolonged CT time in annealed samples is therefore suggested to originate from the introduction of thermally originated interfacial SiO_x at the TiO₂-Si interface which hinders the CT process, and also from a reduction in the amount of available defect states within annealed TiO₂ as compared to the as-deposited amorphous TiO₂.³³

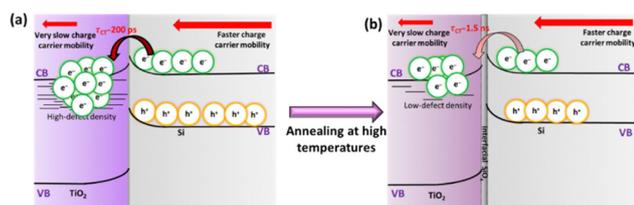


Fig. 6 Schematic of photoinduced interfacial charge transfer in (a) as-dep. TiO₂/n-Si and (b) 15 min annealed TiO₂/n-Si samples.

The increase in the CT time could be linked to the decreasing tunneling probability with increasing SiO_x barrier width. However, our optical instruments are not accurate enough to provide an estimate of the barrier thickness with sufficient accuracy for calculating the tunneling probability, which depends exponentially on the barrier thickness. The CT dynamics were also examined for HF-treated TiO₂/n-Si samples annealed at 300 °C for even longer durations, *i.e.*, for 60 and 90 minutes, to see the effect of higher temperature thermal treatment on the CT time. The results are shown in Fig. S6b in the ESI,[†] where it can be seen that the CT time prolongs even more, *i.e.*, > 5 ns.

Hence, it can be inferred that the samples exhibit limited thermal stability, consequently impacting the CT time and potentially impeding the efficiency of solar-to-electric conversion in Si-based PV devices. The characteristic CT times for all the HF-treated and untreated TiO₂/n-Si heterojunctions (both as-dep. and annealed at 250 °C for different durations) are summarized in Table 1 and the results for TiO₂/p-Si heterojunctions are presented in Table S3 in the ESI.[†] The results show that the CT times are strongly affected by the Si substrate pre-treatments, *i.e.*, the Si surface, and sample post-treatments, *i.e.*, a “thermally generated SiO_x” layer at the TiO₂-Si interface. However, the effect of doping on the CT time cannot be observed clearly in the case of our study as signals are weak and the differences are in the noise range.

3.3. Impact of the annealing on TiO₂

To investigate possible annealing induced morphological changes in TiO₂ thin films, AFM images of TiO₂/n-Si samples subjected to various annealing conditions were taken and the results for all the samples are shown in ESI,[†] Fig. S9. The results indicate that when the annealing is done at 250 °C for 5 to 15 minutes, the surface roughness increases only < 0.1 nm compared to the as-deposited surface. Annealing at 300 °C increases the surface roughness a bit more but the topography remains to depict amorphous characteristics as shown in Fig. 7a and b, thus indicating some degree of structural changes in the TiO₂ films presumably due to the desorption of impurity atoms introduced due to precursors during deposition.

Moreover, we observe a correlation between the surface roughness and the S_{eff} values as shown in Fig. 7c where it can be seen that the S_{eff} increase is the most significant during the initial annealing, *i.e.*, when the as-dep. samples are thermally

Table 1 Characteristic charge transfer time from Si to TiO₂ observed in TiO₂/n-Si heterojunctions annealed at 250 °C for different durations. The samples are excited at 500 nm with an excitation density of 200 μJ cm⁻²

| TiO ₂ /n-Si (HF-treated series) | Characteristic CT time | TiO ₂ /SiO _x /n-Si (RCA2-treated series) | Characteristic CT time |
|--|------------------------|--|------------------------|
| As-deposited | 200 ps | As-deposited | 2.5 ns |
| 5 min annealed | 450 ps | 5 min annealed | 4 ns |
| 10 min annealed | 1 ns | 10 min annealed | > 5 ns |
| 15 min annealed | 1.5 ns | 15 min annealed | > 5 ns |



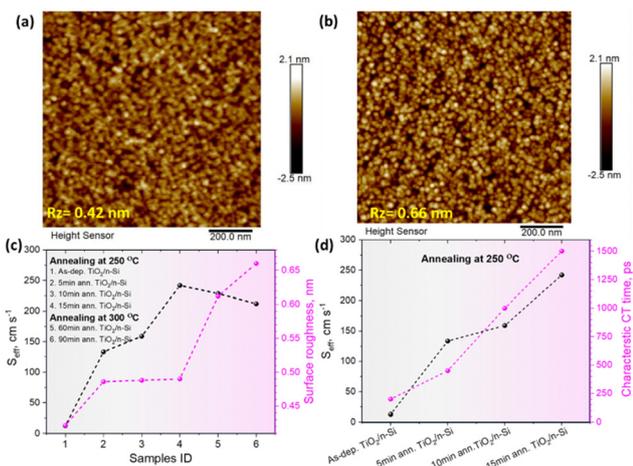


Fig. 7 AFM images of (a) as-dep. TiO₂/n-Si, (b) 90 min annealed TiO₂/n-Si at 300 °C, (c) correlation of surface recombination velocity and surface roughness of the samples, and (d) correlation of surface recombination velocity and characteristic charge transfer times for HF-treated TiO₂/n-Si heterojunctions annealed at 250 °C up to 15 minutes.

treated even for 5 minutes and after that it increases gradually for all annealed samples. It has been pointed out that the presence of H and Cl atoms in the as-deposited samples is favorable for surface passivation.²⁹ In our study, Cl originates from the TiCl₄ precursor during the ALD growth at low temperature while H originates from both the H₂O precursor during ALD growth and HF dip prior to the ALD treatment. Also, the chemical SiO_x contains hydrogen in the form of hydroxyl groups. However, upon the annealing, the H and Cl atoms can desorb from the samples which decreases the passivation quality of all the annealed TiO₂ p- and n-Si samples.¹²

Furthermore, to observe the effect of thermal annealing on the charge carrier dynamics in TiO₂ films solely, the samples were excited at 320 nm by TR spectroscopy to selectively photoexcite the carriers in TiO₂.⁵¹ The photodynamics of the HF-treated TiO₂/n-Si sample series is compared at 600 nm probe wavelength (Fig. S8, ESI†) and it can be seen that the lifetime of photocarriers in TiO₂ does not improve when the samples are annealed for 5 to 15 minutes at 250 °C. However, annealing of the samples at 300 °C for 60 and 90 minutes slightly enhances the lifetime of the photogenerated carriers to ~0.1 ns, but this is still less than the carrier lifetimes in crystalline TiO₂ (~4 ns, ref. 51).

Nevertheless, this enhancement in the lifetimes of photocarriers in annealed TiO₂ films is rather minor and might be attributed to the slight reduction in the recombination centers (less defects and intra-band gap states, Fig. 6b) in them as compared to the as-dep. TiO₂ films.⁵¹ This also suggests a slight reduction in the impurity related defects within TiO₂ while maintaining their amorphous character as shown by the AFM images. To summarize, thermal annealing was found to improve the lifetimes of charge carriers generated within TiO₂ films but were detrimental to the lifetime of carriers photogenerated within Si. The effect should be even more pronounced for the annealed samples that induce crystallization of TiO₂ films.

3.4. Correlation of the surface passivation with the CT time through the TiO₂/Si interface

All the as-dep. TiO₂ films grown on HF-treated n-Si substrates show the minimum CT time of 200 ps as shown in Fig. 6. TiO₂ is commonly used as an electron transport layer in various optoelectronic devices which means that it should keep accepting electrons from Si. However, our measurements show that the CT signal does not keep increasing and saturates after approx. 1 ns as shown in Fig. 5b and c. This happens as the transferred electrons fill up all the available energy states in the CB of the TiO₂ and instead of moving to the TiO₂ surface, the charges start accumulating at the TiO₂ side of the interface because of the low carrier mobility in TiO₂.³³ This leads to inhibition of further acceptance of electrons in TiO₂ from Si and thus, the CT stops. This contributes to the field effect passivation in TiO₂-Si interfaces and the best passivation quality was obtained for samples showing the highest carrier saturation level. Although the Kelvin-probe found similar band bending regardless of the annealing time for TiO₂/n-Si samples, these measurements are carried out under intense illumination and may saturate the TiO₂ layer even with a slow CT process. Also, the as-dep. samples are rich in H and Cl impurities which further improves the chemical surface passivation at the TiO₂-Si interface. However, it should be noted that the CT dynamics is sensitive to even subtle differences at the TiO₂-Si interface and within TiO₂, and a direct correlation is observed between the SRV and the CT process as a result of annealing, as shown in Fig. 7d. For instance, annealing the HF-treated samples at 250 °C even for 15 minutes introduces interfacial SiO_x and induces desorption of the precursor traces and impurities from the amorphous TiO₂ coating which in turn prolongs the CT time to 1.5 ns. Since passivation decreases and the CT time gets prolonged, these annealed samples are not suitable for achieving efficient performance in solar-to-energy conversion devices.

4. Conclusions

This study investigates TiO₂/Si heterojunctions, exploring the impact of pre-ALD chemical and post-ALD thermal treatments on interfacial SiO_x layer formation, the TiO₂ film composition, and the structure. This study examines surface passivation through effective carrier lifetime measurements and investigates CT dynamics using ultrafast transient reflectance spectroscopy. Photocarriers are selectively generated in Si at 500 nm excitation to analyze CT across the TiO₂-Si interface. Findings reveal that HF-treated SiO_x-free n-Si with as-deposited TiO₂ exhibits the highest effective lifetime of 1.23 ns due to the bifunctional passivation of the surface combining both chemical and field effect passivation mainly caused by the precursors especially H impurities. Also, this sample exhibits the fastest CT time from Si to TiO₂, within 200 ps, which is mainly due to the absence of interfacial SiO_x which causes hindrance in the CT. Moreover, annealing increases the surface recombination velocity, possibly due to structural changes and precursor impurity desorption. The presence of the SiO_x layer,



whether chemically or thermally induced, increases the CT time and decreases the number of transferred carriers, adversely affecting the device performance. This research suggests potential for developing TMO-based passivating layers with improved charge transport properties and provides a framework applicable to other semiconductor surfaces and interfaces for understanding time-resolved photoinduced changes and photodynamics.

5. Materials and methods

5.1. Scanning electron microscopy

For the thickness estimation of TiO₂ thin films deposited by ALD, cross-sectional SEM images of the samples were taken by using a FEG Zeiss Gemini 500 SEM microscope in secondary and back scattered electron modes with an accelerating voltage of 3 kV and a working distance of 5 mm.

5.2. AFM

To assess the topography of TiO₂ thin films, atomic force microscopy images of the samples were taken by using AFM Bruker Icon in tapping mode.

5.3. Steady state reflectance spectroscopy

The steady state reflectance spectra of the TiO₂/Si samples were measured using a Shimadzu UV-3600 UV-Vis-NIR spectrophotometer. The measurements were carried out in the reflectance mode using a 'Specular Reflectance Attachment for 5° Incidence Angle' accessory with the spectrophotometer setup. Aluminium coated mirrors were used as reference mirrors. The spectra were recorded from 300 to 1200 nm.

5.4. The injection-dependent effective lifetime τ_{eff}

The injection dependent effective lifetime was measured from the quarters by the contactless quasi steady-state photoconductance (QSSPC) method by using Sinton WCT-120 at room temperature. The software version for measurement and analysis is v5.72.0.10. For high lifetime (>100 μs) samples short-flash and transient analysis modes are used, while for low lifetime (<100 μs) samples long flash and generalized analysis modes are used. The reported lifetime was under an injection level of $1 \times 10^{15} \text{ cm}^{-3}$.

5.5. Surface barrier height

The surface barrier height was determined using a Semilab PV-2000A to execute the Kelvin-probe method. Measurements of the contact potential difference between the wafer and a reference electrode were conducted under intense illumination ($V_{\text{CPD}}^{\text{light}}$) and in complete darkness ($V_{\text{CPD}}^{\text{dark}}$), and the surface barrier height ($V_{\text{sb}} = V_{\text{CPD}}^{\text{dark}} - V_{\text{CPD}}^{\text{light}}$) was determined based on the difference between the two lighting conditions. It was averaged from an area of $20 \times 20 \text{ cm}^2$ with a measurement step of 5 mm.

5.6. Transient reflectance spectroscopy

Ultra-fast time-resolved pump-probe spectroscopy was used in reflectance mode to determine the optical properties of the TiO₂/Si samples. The fundamental laser pulses were generated via a Ti:Sapphire laser (Libra F, Coherent Inc., 800 nm, approx. 100 fs pulse width, repetition rate 1 kHz). 90% of the fundamental beam was directed onto an optical parametric amplifier (Topas C, Light conversion Ltd) to produce the desired excitation wavelength (320 nm and 500 nm in our case), approximately 1 mm beam diameter at the sample, attenuated up to 200 $\mu\text{J cm}^{-2}$. 10% of the fundamental laser was delivered to the motorized stage (delay line) and then to a 2 mm cuvette with water to generate continuum white light for probe pulses. The probe light was split into the reference and signal beams. The absorbance change was measured in chopper mode which is synchronized with fundamental laser pulses. The samples were excited at 320 nm and 500 nm to selectively excite TiO₂ and Si, respectively.³³ The spectra were averaged over 3000 pulses for each delay time.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The authors acknowledge the financial support of the Research Council of Finland (formerly Academy of Finland) via the Flagship on Photonics Research and Innovation "PREIN" #346511 (Tampere University) and #346529 (Aalto University). R. K. thanks Kaute foundation and Tampere University doctoral school for the funding. X. L. acknowledges the financial fundings from Business Finland (#7479/31/2019) and Research Council of Finland (#354199). V. V. acknowledges the financial funding from Research Council of Finland (#331313). Part of the research was performed at the OtaNano - Micronova Nanofabrication Centre of Aalto University.

Notes and references

- 1 S. Knez, S. Štrbac and I. Podbregar, *Energy Sustainability Soc.*, 2022, **12**, 1.
- 2 S. Tagliapietra, G. Zachmann, O. Edenhofer, J.-M. Glachant, P. Linares and A. Loeschel, *Energy Policy*, 2019, **132**, 950–954.
- 3 S. Park, J. Ha, M. F. Khan, C. Im, J. Y. Park, S. H. Yoo, M. A. Rehman, K. Kang, S. H. Lee and S. C. Jun, *ACS Appl. Electron. Mater.*, 2022, **4**, 4306–4315.
- 4 N. Asim, K. Sopian, S. Ahmadi, K. Saeedfar, M. A. Alghoul, O. Saadatian and S. H. Zaidi, *Renewable Sustainable Energy Rev.*, 2012, **16**, 5834–5847.
- 5 T. G. Allen, J. Bullock, X. Yang, A. Javey and S. De Wolf, *Nat. Energy*, 2019, **4**, 914–928.
- 6 R. Fan, Z. Mi and M. Shen, *Opt. Express*, 2019, **27**, A51–A80.



- 7 F. Decker and S. Cattarin, in *Encyclopedia of Electrochemical Power Sources*, ed. J. Garche, Elsevier, Amsterdam, 2009, pp. 1–9.
- 8 R. S. Bonilla and P. R. Wilshaw, *Appl. Phys. Lett.*, 2014, **104**, 232903.
- 9 A. Tyagi, K. Ghosh, A. Kottantharayil and S. Lodha, *IEEE Trans. Electron Devices*, 2019, **66**, 1377–1385.
- 10 H. Lin, M. Yang, X. Ru, G. Wang, S. Yin, F. Peng, C. Hong, M. Qu, J. Lu, L. Fang, C. Han, P. Procel, O. Isabella, P. Gao, Z. Li and X. Xu, *Nat. Energy*, 2023, **8**, 789–799.
- 11 M. A. Rehman, S. Park, M. F. Khan, M. F. Bhopal, G. Nazir, M. Kim, A. Farooq, J. Ha, S. Rehman, S. C. Jun and H.-H. Park, *Int. J. Energy Res.*, 2022, **46**, 11510–11522.
- 12 M. M. Shehata, G. Bartholazzi, D. H. Macdonald and L. E. Black, *Adv. Energy Mater.*, 2023, **13**, 2300251.
- 13 B. S. Richards, J. E. Cotter, C. B. Honsberg and S. R. Wenham, Conference Record of the Twenty-Eighth IEEE Photovoltaic Specialists Conference, 2000, pp. 375–378.
- 14 G. S. Sahoo, C. Harini, N. Mahadevi, P. S. Nethra, A. Tripathy, M. Verma and G. P. Mishra, *Silicon*, 2023, **15**, 4039–4048.
- 15 A. Rehman, M. F. Khan, M. A. Shehzad, S. Hussain, M. F. Bhopal, S. H. Lee, J. Eom, Y. Seo, J. Jung and S. H. Lee, *ACS Appl. Mater. Interfaces*, 2016, **8**, 29383–29390.
- 16 A. Richter, S. W. Glunz, F. Werner, J. Schmidt and A. Cuevas, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **86**, 165202.
- 17 L. G. Gerling, C. Voz, R. Alcubilla and J. Puigdollers, *J. Mater. Res.*, 2017, **32**, 260–268.
- 18 B. W. H. van de Loo, B. Macco, J. Melskens, W. Beyer and W. M. M. Kessels, *J. Appl. Phys.*, 2019, **125**, 105305.
- 19 S. Acharyya, S. Sadhukhan, T. Panda, D. K. Ghosh, N. C. Mandal, A. Nandi, S. Bose, G. Das, S. Maity, P. Chaudhuri, H. Saha and D. Banerjee, *Surf. Interfaces*, 2022, **28**, 101687.
- 20 T. Mochizuki, K. Gotoh, A. Ohta, S. Ogura, Y. Kurokawa, S. Miyazaki, K. Fukutani and N. Usami, *Appl. Phys. Express*, 2018, **11**, 102301.
- 21 K. M. Gad, D. Vössing, A. Richter, B. Rayner, L. M. Reindl, S. E. Mohney and M. Kasemann, *IEEE J. Photovolt.*, 2016, **6**, 649–653.
- 22 B. Liao, B. Hoex, A. G. Aberle, D. Chi and C. S. Bhatia, *Appl. Phys. Lett.*, 2014, **104**, 253903.
- 23 Z.-P. Yang, H.-E. Cheng, I.-H. Chang and I.-S. Yu, *Appl. Sci.*, 2016, **6**, 233.
- 24 J. Cui, T. Allen, Y. Wan, J. Mckeon, C. Samundsett, D. Yan, X. Zhang, Y. Cui, Y. Chen, P. Verlinden and A. Cuevas, *Sol. Energy Mater. Sol. Cells*, 2016, **158**, 115–121.
- 25 G. S. Park, S. Lee, D.-S. Kim, S. Y. Park, J. H. Koh, D. H. Won, P. Lee, Y. R. Do and B. K. Min, *Adv. Energy Mater.*, 2023, **13**, 2203183.
- 26 T.-C. Chen, T.-C. Yang, H.-E. Cheng, I.-S. Yu and Z.-P. Yang, *Appl. Surf. Sci.*, 2018, **451**, 121–127.
- 27 K. Gotoh, T. Mochizuki, T. Hojo, Y. Shibayama, Y. Kurokawa, E. Akiyama and N. Usami, *Curr. Appl. Phys.*, 2021, **21**, 36–42.
- 28 M. M. Plakhotnyuk, N. Schüler, E. Shkodin, R. A. Vijayan, S. Masilamani, M. Varadharajaperumal, A. Crovetto and O. Hansen, *Jpn. J. Appl. Phys.*, 2017, **56**, 08MA11.
- 29 B. Liao, N. Dwivedi, Q. Wang, R. J. Yeo, A. G. Aberle, C. S. Bhatia and A. Danner, *IEEE J. Photovolt.*, 2021, **11**, 319–328.
- 30 Y. Chang, J. R. Yates and C. E. Patrick, *ACS Omega*, 2023, **8**, 20138–20147.
- 31 H. Liu, J. Tang, I. J. Kramer, R. Debnath, G. I. Koleilat, X. Wang, A. Fisher, R. Li, L. Brzozowski, L. Levina and E. H. Sargent, *Adv. Mater.*, 2011, **23**, 3832–3837.
- 32 H. Hu, B. Dong, H. Hu, F. Chen, M. Kong, Q. Zhang, T. Luo, L. Zhao, Z. Guo, J. Li, Z. Xu, S. Wang, D. Eder and L. Wan, *ACS Appl. Mater. Interfaces*, 2016, **8**, 17999–18007.
- 33 R. Khan, H. P. Pasanen, H. Ali-Löyty, H. M. Ayedh, J. Saari, V. Vähänissi, M. Valden, H. Savin and N. V. Tkachenko, *Surf. Interfaces*, 2023, **38**, 102871.
- 34 J. Ott, T. P. Pasanen, P. Repo, H. Seppänen, V. Vähänissi and H. Savin, *Phys. Status Solidi A*, 2019, **216**, 1900309.
- 35 C. Luderer, D. Kurt, A. Moldovan, M. Hermle and M. Bivour, *Sol. Energy Mater. Sol. Cells*, 2022, **238**, 111412.
- 36 H. P. Pasanen, P. Vivo, L. Canil, H. Hempel, T. Unold, A. Abate and N. V. Tkachenko, *J. Phys. Chem. Lett.*, 2020, **11**, 445–450.
- 37 R. Khan, H. Ali-Löyty, J. Saari, M. Valden, A. Tukiainen, K. Lahtonen and N. V. Tkachenko, *Nanomaterials*, 2020, **10**, 1567.
- 38 H. P. Pasanen, R. Khan, J. A. Odutola and N. V. Tkachenko, *J. Phys. Chem. C*, 2024, **128**, 6167–6179.
- 39 N. V. Tkachenko and R. Khan, in *Tailored Functional Oxide Nanomaterials*, eds. C. Maccato and D. Barreca, Wiley, 1st edn, 2022, pp. 193–228.
- 40 O. M. E. Ylivaara, A. Langner, X. Liu, D. Schneider, J. Julin, K. Arstila, S. Sintonen, S. Ali, H. Lipsanen, T. Sajavaara, S.-P. Hannula and R. L. Puurunen, *Thin Solid Films*, 2021, **732**, 138758.
- 41 N. E. Grant, V. P. Markevich, J. Mullins, A. R. Peaker, F. Rougieux, D. Macdonald and J. D. Murphy, *Phys. Status Solidi A*, 2016, **213**, 2844–2849.
- 42 W. Kern, *Handbook of Silicon Wafer Cleaning Technology*, 3rd edn, 2018, p. 978-0-323-51084-4.
- 43 D. Rouchon, N. Rochat, F. Gustavo, A. Chabli, O. Renault and P. Besson, *Surf. Interface Anal.*, 2002, **34**, 445–450.
- 44 J. Saari, H. Ali-Löyty, M. Honkanen, A. Tukiainen, K. Lahtonen and M. Valden, *ACS Omega*, 2021, **6**, 27501–27509.
- 45 R. Methaapanon and S. F. Bent, *J. Phys. Chem. C*, 2010, **114**, 10498–10504.
- 46 H. Liu, T. P. Pasanen, O. Leiviskä, J. Isometsä, T. H. Fung, M. Yli-Koski, M. Miettinen, P. Laukkanen, V. Vähänissi and H. Savin, *Appl. Phys. Lett.*, 2023, **122**, 191602.
- 47 N. E. Grant, A. I. Pointon, R. Jefferies, D. Hiller, Y. Han, R. Beanland, M. Walker and J. D. Murphy, *Nanoscale*, 2020, **12**, 17332–17341.
- 48 X. Liu, B. Radfar, K. Chen, E. Pälkkö, T. P. Pasanen, V. Vähänissi and H. Savin, *IEEE Photonics Technol. Lett.*, 2022, **34**, 870–873.
- 49 X. Liu, B. Radfar, K. Chen, T. P. Pasanen, V. Vähänissi and H. Savin, *Adv. Photonics Res.*, 2022, **3**, 2100234.
- 50 A. G. Scheuermann, C. E. D. Chidsey and P. C. McIntyre, *J. Electrochem. Soc.*, 2015, **163**, H192.
- 51 R. Khan, H. Ali-Löyty, A. Tukiainen and N. V. Tkachenko, *Phys. Chem. Chem. Phys.*, 2021, **23**, 17672–17682.

