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Born–Oppenheimer approximation in optical cavities: from success to breakdown

In this study, we discuss the applicability of the Born–Oppenheimer approximation (BOA) in optical cavities for polyatomic molecules. The formaldehyde molecule serves as a showcase example which does not exhibit any natural nonadiabatic phenomena in the studied energy region. By calculating the one-dimensional absorption spectrum of the molecule, we could clearly identify the cavity-induced nonadiabatic fingerprints, which appear solely due to light–matter coupling, demonstrating that contrary to expectation, BOA may even fail in a one-dimensional model. (Image by Gy. Bujdosó)

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Born–Oppenheimer approximation in optical cavities: from success to breakdown†

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The coupling of a molecule and a cavity induces nonadiabaticity in the molecule which makes the description of its dynamics complicated. For polyatomic molecules, reduced-dimensional models and the use of the Born–Oppenheimer approximation (BOA) may remedy the situation. It is demonstrated that contrary to expectation, BOA may even fail in a one-dimensional model and is generally expected to fail in two- or more-dimensional models due to the appearance of conical intersections induced by the cavity.

1 Introduction

The dynamics initiated in a molecule by the absorption of a photon is usually treated within the framework of the Born–Oppenheimer (BO) or adiabatic approximation¹ where the fast-moving electrons are separated from the slow nuclear degrees of freedom (dofs). In this picture the nuclei move on a single potential energy surface (PES) created by the fast-moving electrons. Although several chemical processes can be rationalised by considering a single BO PES, there are indeed a number of situations where the BO approximation (BOA) breaks down. These are called nonadiabatic processes which involve nuclear dynamics proceeding on at least two coupled PESs, leading to the formation of so-called conical intersections (CIs).^{2–7} Nonadiabatic phenomena are ubiquitous in photochemistry, photophysics, particularly in molecular fragmentation, proton transfer, isomerization or radiationless deactivation processes of excited states as the CI can provide a very efficient channel for ultrafast interstate crossing on the femtosecond timescale.^{8–21}

Conical intersections and avoided crossings (ACs) can be created both by classical and quantum light, as well. To form a CI, the molecule must have at least two independent nuclear

dofs. In diatomics having only one nuclear dof, natural CIs can never be formed, only ACs can arise. If the system interacts with light, either light-induced avoided crossings (LIACs) or light-induced conical intersections (LICIs)^{22–24} can emerge. LICIs can be created even in diatomics where the second dof (either rotation or translation), needed to form a LICI, comes into play due to the light-matter interaction. Moreover, LICIs are ubiquitous and become multidimensional in the nuclear coordinate space in polyatomic molecules due to the presence of several vibrational dofs.^{25,26}

Recently, efforts have been made to study light-induced nonadiabatic phenomena in optical or microwave cavities.^{27–46} It has been successfully demonstrated that describing the photon–matter interaction with the tools of cavity quantum electrodynamics (cQED)^{47–50} can provide an alternative way to study the quantum control of molecules with light. In this framework nonadiabatic dynamics arises due to the strong coupling between the molecular dofs and the photonic mode of the radiation field which can alter the molecular energy levels by controlling the dynamics of basic photophysical and photochemical processes. The molecular vibrational modes which are strongly coupled to the electronic and photonic dofs are taken into account resulting in a new set of “cavity-induced” or “polariton” surfaces in the molecular Hamiltonian. These polariton surfaces are expected to form LIACs or LICIs.

Numerous works deal with quantum-light-induced nonadiabatic effects within a single molecule. In most of the studies diatomic or polyatomic organic molecules are treated as reduced-dimensional two-level systems by taking into account only one vibrational and photonic dof.^{27–29,34,35,42} As is already clear from classical light, quantum LICI situations can also only occur if, in addition to the only vibrational dof, the rotational angle between the molecular axis and the polarization vector of the electric field in the cavity is also accounted for (in case of diatomics)^{37,38} or at least two vibrational dofs are considered in

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mode and the three lowest polaritonic (adiabatic) PESs that emerge due to the light–matter coupling are depicted in Fig. 1. The field-free vibrational eigenstates of H₂CO were computed with the GENIUSH program package^{54–56} for both electronic states treating all six vibrational dofs in a numerically exact way. The rotational dofs are omitted from the present computational protocol and the molecule is imagined to be fixed with respect to the external electric field. Further information on the structure and normal modes of H₂CO as well as technical details of the computations are provided in the ESI.†

Having described the working formulae and the molecular system, we move on to the discussion of the absorption spectrum of the field-free H₂CO molecule and first present results with a numerically-exact six-dimensional (6D) quantum-dynamical treatment. We consider the high-energy part of the spectrum consisting of spectral lines corresponding to transitions from |X0⟩ (vibrational ground state of the electronic state X) to |A \bar{i} ⟩ (vibrational states of the electronic state A). Fig. 2 presents the 6D field-free spectrum of H₂CO, showing favourable agreement with results reported in ref. 57. The 6D field-free spectrum exhibits progressions of lines that are mainly associated with the ν_4 (out-of-plane bend) and ν_2 (C=O stretch)

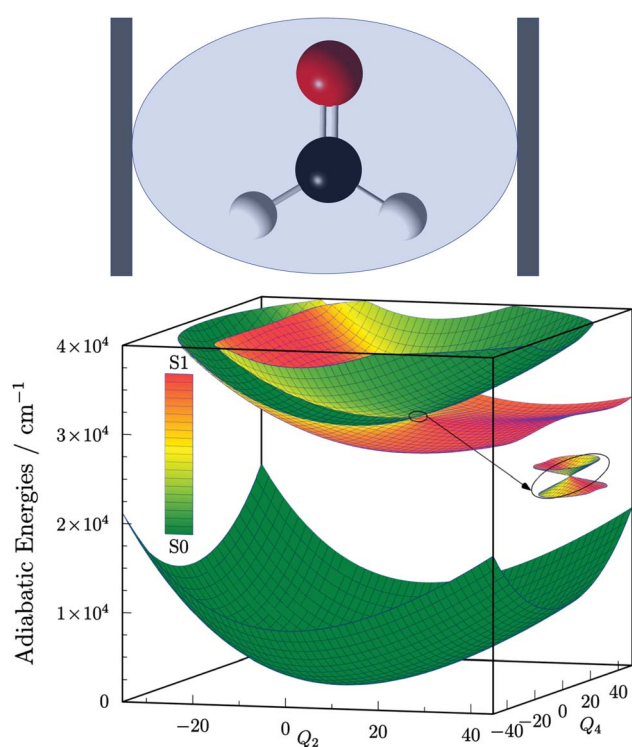


Fig. 1 Structure of the H₂CO (formaldehyde) molecule (upper panel) and the three lowest polaritonic (adiabatic) surfaces of H₂CO coupled to a single cavity mode (lower panel). Q_2 and Q_4 are the normal coordinates of the ν_2 (C=O stretch) and ν_4 (out-of-plane bend) vibrational modes. The cavity wavenumber and coupling strength are chosen as $\omega_c = 29\,000\text{ cm}^{-1}$ and $g = 5.97 \times 10^{-2}\text{ a.u.}$, respectively. The light-induced conical intersection between the second and third polaritonic surfaces is shown in the inset on the right-hand side of the lower panel. The character of the polaritonic surfaces is indicated by different colours (see the legend on the left).

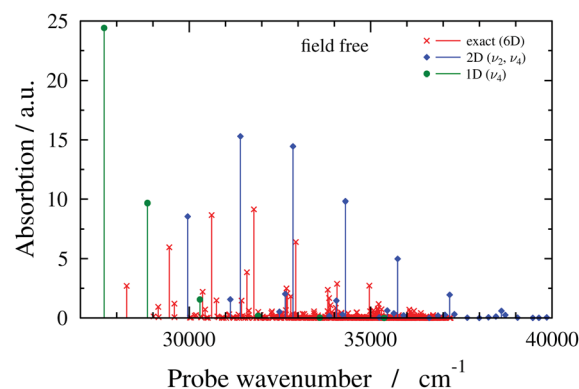


Fig. 2 Absorption spectra of field-free (*i.e.*, no cavity) formaldehyde obtained with different quantum-dynamical models (6D, 2D(ν_2, ν_4) and 1D(ν_4)) models, as indicated in the figure).

vibrational modes. Displacements along the ν_4 vibrational mode produce nonzero TDM and since the C=O equilibrium bond length in the excited electronic state is considerably larger than the corresponding ground-state value, the ν_2 vibrational mode adds complexity to the absorption spectrum by giving rise to several nonzero Franck–Condon factors. Therefore, any reduced-dimensional quantum-dynamical model of H₂CO should incorporate at least the ν_4 and ν_2 vibrational modes. However, if one still aims to define a one-dimensional model, the ν_2 vibrational mode should be omitted as explained in the next paragraph.

Fig. 2 also shows field-free spectra obtained with two reduced-dimensional models, treating the ν_2 and ν_4 vibrational modes (2D(ν_2, ν_4) model), or the ν_4 vibrational mode (1D(ν_4) model). Comparing the field-free 2D(ν_2, ν_4) spectrum to its 6D counterpart reveals that although the 2D(ν_2, ν_4) spectrum lacks lines that are present in the 6D spectrum, the overall structures of the 2D(ν_2, ν_4) and 6D field-free spectra are similar. On the contrary, the 1D(ν_4) field-free spectrum does not resemble the 6D field-free spectrum at all. Besides the 1D(ν_4) model, one could also think of testing the performance of the 1D(ν_2) model. Since the reduced-dimensional models are defined by setting the inactive normal coordinates to zero and the ν_2 vibrational mode is totally symmetric, the C_{2v} symmetry of the equilibrium structure is preserved when displacements are made along the ν_2 vibrational mode (see also ESI†). Due to symmetry, the TDM between the electronic states X and A is zero at nuclear configurations of C_{2v} symmetry, therefore, no transitions are allowed if the 1D(ν_2) model is used. Based on the analysis of the field-free spectra presented in this subsection, the simplest model expected to provide sensible results is the 2D(ν_2, ν_4) model.

2.2 One-dimensional results for the 1D(ν_4) model

We define a one-dimensional quantum-dynamical model, called 1D(ν_4) model henceforth, treating only the ν_4 (out-of-plane) vibrational mode of H₂CO (see ESI† for technical details). The 1D(ν_4) model is utilized to assess the performance of a simplified one-dimensional description for H₂CO and test





Fig. 3 1D(v_4) dressed spectra (exact and BOA) with a cavity wavenumber of $\omega_c = 27\,653.3\text{ cm}^{-1}$ for different coupling strength values.

the applicability of the BOA in one dimension, which requires further explanation. Since the frequency of the cavity mode is chosen to be nearly resonant with the frequency of the $X \rightarrow A$ electronic excitation, the dof associated with the “fast” cavity mode is grouped with the electrons and the “slow” nuclear dofs are separated from the electronic and cavity dofs. This allows the construction of the lower ($|-\rangle$) and upper ($|+\rangle$) hybrid light-matter (polaritonic) states that can be approximately described as

$$\begin{aligned} |-\rangle &= a|X\rangle|1\rangle - b|A\rangle|0\rangle \\ |+\rangle &= b|X\rangle|1\rangle + a|A\rangle|0\rangle \end{aligned} \quad (5)$$

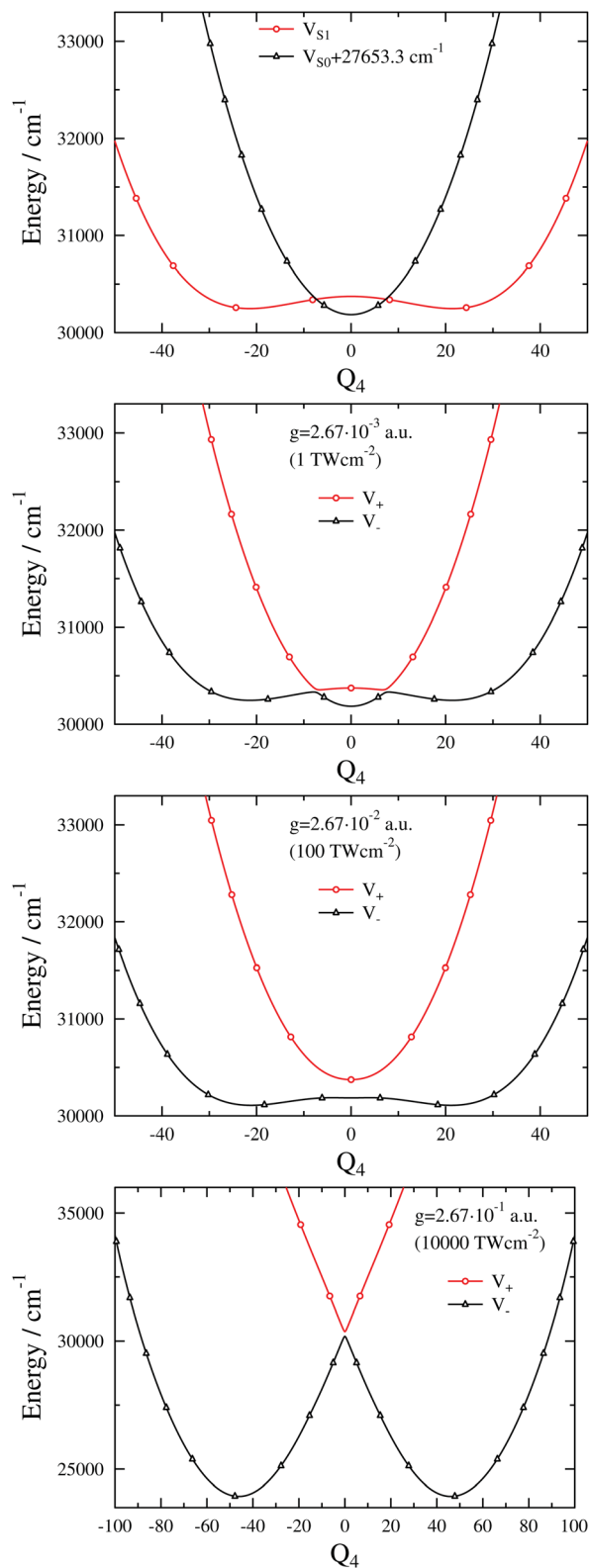


Fig. 4 1D(v_4) diabatic (uppermost panel) and adiabatic potential curves (V_+ and V_-) with a cavity wavenumber of $\omega_c = 27\,653.3\text{ cm}^{-1}$ for different coupling strength values. Note that the range of the energy axis in the last panel ($g = 2.67 \times 10^{-1}\text{ a.u.}$) differs from those of the other three panels.



nonadiabatic effects appearing in the absorption spectrum can be attributed solely to the quantum LICl.

First, as for comparison, we have calculated the field-free spectrum of H₂CO utilizing the full-dimensional (6D) as well as reduced-dimensional (2D(v₂,v₄) and 1D(v₄)) quantum-dynamical models. It has been found that the simplest model which can approximately reproduce the structure of the numerically-exact 6D spectrum is the 2D(v₂,v₄) model. As the one-dimensional 1D(v₂) model leads to a vanishing absorption spectrum, the 1D(v₄) model is the one-dimensional model to be used.

For the reduced-dimensional 1D(v₄) and 2D(v₂,v₄) models, the field-dressed exact and BOA spectra have been computed and compared. A striking finding of our work is that the BOA can fail even for a one-dimensional quantum-dynamical treatment of H₂CO irrespective of the value of the coupling strength. This complements previous results claiming that the BOA can be used in the strong coupling regime when only one vibrational dof is taken into account. Analytical considerations fully corroborate our conclusion and point out that the breakdown of the BOA even in 1D for H₂CO arises due to an intrinsic property of the molecule, namely that the transition dipole moment vanishes at certain geometries due to molecular symmetry. Clearly, one should be careful even when describing a molecule in a cavity using the BOA with only one vibration.

Moreover, we have also shown that the BOA is not applicable for the 2D(v₂,v₄) model. We can expect such a failure whenever there is a LICl and the spectrum extends to energies above the LICl. Consequently, we have no doubt that the BOA fails for the full-dimensional (6D) model as well.

The above-discussed aspects are valid when the system under investigation lacks any inherent nonadiabatic phenomena in the absence of the cavity. If this is not the case, the situation is expected to be even more involved.

Conflicts of interest

There are no conflicts to declare.

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