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An enzyme-activatable probe liberating AIEgens: on-site sensing and long-term tracking of β -galactosidase in ovarian cancer cells†

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Development of fluorescent probes for on-site sensing and long-term tracking of specific biomarkers is particularly desirable for the early detection of diseases. However, available small-molecule probes tend to readily diffuse across the cell membrane or remain at the activation site but always suffer from the aggregation-caused quenching (ACQ) effect. Here we report an enzyme-activatable aggregation-induced emission (AIE) probe QM- β gal, which is composed of a hydrophilic β -galactosidase (β -gal)-triggered galactose moiety and a hydrophobic AIE-active fluorophore QM-OH. The probe is virtually non-emissive in aqueous media, but when activated by β -gal, specific enzymatic turnover would liberate hydrophobic AIE luminogen (AIEgen) QM-OH, and then highly fluorescent nanoaggregates are *in situ* generated as a result of the AIE process, allowing for on-site sensing of endogenous β -gal activity in living cells. Notably, taking advantage of the improved intracellular retention of nanoaggregates, we further exemplify QM- β gal for long-term (~12 h) visualization of β -gal-overexpressing ovarian cancer cells with high fidelity, which is essential for biomedicine and diagnostics. Thus, this enzyme-activatable AIE probe not only is a potent tool for elucidating the roles of β -gal in biological systems, but also offers an enzyme-regulated liberation strategy to exploit multifunctional probes for preclinical applications.

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Introduction

On-site sensing and long-term tracking are increasingly important for real-time observation of dynamic cell migration, division, and lysis to gain deep insights into the pathogenesis of tumors in diagnostics.^{1–5} Small-molecule fluorescent probes have been broadly utilized for monitoring of biomarkers in living systems,^{6–13} but unfortunately, the modality of on-site sensing and long-term imaging with high resolution is still rarely realized by available probes. The main obstacles are that these small-molecule probes often tend to diffuse across the cell membrane and could be extruded out to adjacent tissues (diffusion away from detection sites) on activation,^{14–18} due to the concentration

gradient of newly freed fluorochromes between the cellular interior and exterior. A conventional solution to this issue would be to adopt bioorthogonal reaction-based strategies, such as azide-alkyne click reaction^{19,20} and quinone methide chemistry.²¹ However, covalent attachment of fluorophores to the cellular proteins possibly disrupts their folding patterns and causes dysfunction, thus leading to undesirable effects on metabolism of the cells or tissues. In particular, even the remaining probes at the activation site may still suffer from the notorious aggregation-caused quenching (ACQ) effect,^{22–27} compromising spatiotemporal resolution and failing to persistently provide *in situ* information at the molecular level. Therefore, a new generation of molecular probes capable of on-site measurement and long-term visualization of targeted biomolecules or cells in live biological samples is of high demand.

With this in mind, we envisioned that aggregation-induced emission (AIE) materials may be a kind of promising candidate to tackle the aforementioned intractable issues. In contrast to conventional ACQ fluorescent dyes, AIE luminogens (AIEgens) are non-emissive as an isolated molecule but become highly fluorescent in the aggregate state because of the restriction of intramolecular motion.^{28–37} In particular, the more the AIEgens aggregate, the brighter their emission becomes, making them suitable for long-term tracking of biomolecules with high fidelity in living systems.^{38–43} Indeed, despite significant advances of AIEgens as fluorescent materials in biosensing and

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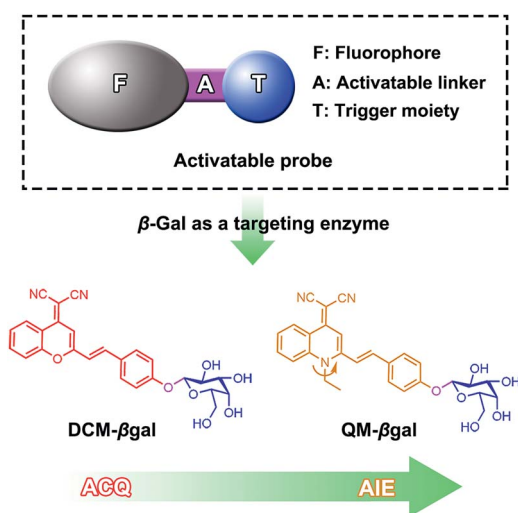




Scheme 1 Schematic illustration of an enzyme-regulated liberation strategy for on-site sensing and long-term tracking.

bioimaging, probes with simultaneous on-site sensing and long-term tracking mode by using AIEgens are scarcely reported.

Herein, we describe an enzyme-regulated liberation strategy to *in situ* generate AIEgen nanoaggregates with improved intracellular retention for high-fidelity bioimaging. As depicted in Scheme 1, our design rationale is based on the fact that highly efficient enzyme-catalyzed reaction would liberate hydrophobic AIEgens, and then they can *in situ* form nanoaggregates at the activation site with intensive emission, facilitating diffusion-resistant on-site sensing of enzyme activity in living cells. Further, the retention effect of nanoaggregates enables long-term tracking of enzyme activity in living systems. Specifically, an enzyme-activatable AIE fluorescent probe, named QM- β gal (Scheme 2), is elaborated, in which a hydrophobic AIE-active fluorophore QM-



Scheme 2 Enzyme-activatable probes for β -gal activity sensing.

OH is utilized as a signal reporter and a hydrophilic galactose moiety as the β -galactosidase (β -gal)-triggered unit. Meanwhile, the reported ACQ probe DCM- β gal was used as a control.²⁵ β -Gal, a glycoside hydrolase, was chosen as a targeting enzyme because it is an important biomarker for primary ovarian cancers and cell senescence.^{44,45} By virtue of *in situ* enzyme-catalyzed conversion and intracellular retention characteristics of nanoaggregates, QM- β gal is able to perform on-site sensing and long-term trapping of β -gal activity in ovarian cancer cells. Our results demonstrate that this probe provides an ideal molecular tool to better understand the biological roles of β -gal, making a breakthrough in capturing high-fidelity information about intracellular enzyme activity, which is unattainable by using traditional small-molecule probes.

Results and discussion

Design of an enzyme-activatable probe liberating AIEgens

Quinoline-malononitrile (QM) derivatives are one kind of typical AIE-active material, which have been widely employed for fluorescent sensors, imaging agents, optical waveguides, and drug delivery applications.^{31,46–50} In this work, we report a simple but efficient enzyme-activatable AIE-active probe based on AIEgen QM-OH with a *para*-position phenolic group protected by a β -galactopyranoside unit. Our design rationale is depicted in Scheme 1. QM- β gal consists of two parts: (1) a hydrophobic AIEgen QM-OH as a signal reporter and (2) a β -galactopyranoside unit endowing the probe with water solubility as a β -gal-responsive moiety. QM- β gal is a highly water-soluble species to render it almost nonfluorescent in aqueous solution. After being specifically hydrolyzed by β -gal, hydrophobic AIE-active QM-OH can be released and aggregated with a remarkable light-up fluorescent signal, which enables diffusion-resistant on-site localization and long-term trapping of endogenous β -gal activity in living cells. The molecular structures of the probe and intermediates were further confirmed to be right by ^1H and ^{13}C NMR and high-resolution mass spectroscopy (HRMS, in the ESI $^+$).

Photophysical properties of QM- β gal and QM-OH

The photophysical properties of both QM- β gal and QM-OH were initially evaluated in water/dimethyl sulfoxide (DMSO) mixtures with different water fractions (f_w). The absorption of both compounds covers the range of 300–500 nm with two peaks at 355 and 434 nm, respectively. We observed that the absorbance of QM- β gal always stays relatively unchanged with increasing f_w (Fig. 1A), but that of QM-OH apparently descends once the f_w is more than 70 vol% (Fig. 1D), which is caused by the scattering effect of nanoaggregates. As expected, QM- β gal endowed with good water solubility is virtually nonfluorescent in different water/DMSO mixtures (Fig. 1B and C), with a quantum yield (Φ) of 0.05 using fluorescein as a reference, because of its free motions in the molecularly dissolved state; while the emission of QM-OH increases quickly and monotonously when the f_w is higher than 70 vol%, yielding a luminous yellow signal with an emission maximum at \sim 560 nm.





Fig. 1 Spectral properties of QM- β gal and QM-OH with different water fractions (f_w) in a mixture of water/DMSO. (A) Absorption spectra, (B) emission spectra, and (C) I/I_0 plots of QM- β gal (10 μ M); (D) absorption spectra, (E) emission spectra, and (F) I/I_0 plots of QM-OH (10 μ M), where I is the fluorescence intensity at 560 nm and I_0 is the fluorescence intensity of QM- β gal or QM-OH in 0% water, $\lambda_{ex} = 434$ nm. Inset: images of (C) QM- β gal and (F) QM-OH with various f_w under UV lamp illumination.

Specifically, at $f_w = 95$ vol%, the fluorescence intensity of QM-OH is *ca.* 52-fold higher than its intrinsic fluorescence intensity in the benign solvent DMSO ($\Phi = 0.70$, Fig. 1E and F). The impressive fluorescence is highly linked with the formation of QM-OH aggregates, which is confirmed by dynamic light scattering (DLS, Fig. S1 \dagger) and scanning electron microscope (SEM, Fig. S2 \dagger) measurements. The effective diameter of QM-OH in water/DMSO ($f_w = 95$ vol%) is ~ 150 nm. In contrast, no DLS signal was detected for QM- β gal solution, implying that it has good water solubility. These results indicate that the fluorescence of QM-OH can be revitalized in the aggregation state.

In situ generation of QM-OH nanoaggregates by β -gal activation

Subsequently, we focused on *in situ* generation of nanoaggregates and investigated the optical response of the as-prepared probe to β -gal in aqueous solution (PBS/DMSO = 95 : 5, v/v, 50 mM, pH = 7.4) at 37 $^{\circ}$ C. After addition of 6 U β -gal, the absorbance of QM- β gal decreased obviously (Fig. 2A), in accordance with that of QM-OH nanoaggregates (Fig. 1D). As shown in Fig. 2B, a quick and steady fluorescence increment at 560 nm was observed, with a Stokes shift up to *ca.* 125 nm. Significantly, a *ca.* 15-fold higher fluorescence intensity than the intrinsic fluorescence of the probe was obtained. Meanwhile, we found that increased fluorescence intensity has an apparently positive correlation with the concentration of β -gal (Fig. 2C), which is an indication of higher extent of enzymatic turnover and *in situ* generation of AIEgen nanoaggregates.

The fluorescence intensity has a good linear relationship with β -gal in 0–6 U ($R^2 = 0.992$, inset of Fig. 2C) and the detection limit of QM- β gal for β -gal was calculated to be as low

as 1.0×10^{-3} U mL $^{-1}$ (Fig. S3 \dagger). Moreover, the enzyme response rate was tested using time-course fluorescence profiles. After incubation with β -gal, the fluorescence intensity of the probe progressively increases in the early stage and gradually levels off at around 30 min (Fig. 2D), indicative of the rapid and efficient conversion of the substrate by β -gal. For the ensemble system of QM- β gal and β -gal, we also observed *in situ* generation of about 100 nm nanoaggregates by DLS and SEM characterization (Fig. 2E and F).

We speculated that the β -galactopyranoside as a leaving group could be specifically cleaved by β -gal, accompanied by *in situ* release of AIEgen QM-OH. To prove this, HPLC and HRMS analyses were systemically performed. As illustrated in Fig. 2G, the retention time of free QM- β gal and QM-OH is 4.9 and 6.5 min, respectively. After reaction with β -gal for 20 min, QM- β gal exhibited an intense peak with a retention time at 6.5 min, suggesting that QM- β gal is a substrate of β -gal and the sensing product is exclusively QM-OH. Actually, the product peak has a longer retention time of 6.5 min, demonstrating that its hydrophobicity is increased as compared to that of QM- β gal (4.9 min). Moreover, the cleavage product was further unambiguously confirmed by HRMS analysis. In the ensemble system of QM- β gal and β -gal, the peaks of QM- β gal and its cleavage product were found at m/z 524.1580 and 338.1221, respectively (Fig. 2H). All these results collectively substantiate that enzyme-catalyzed reaction *in situ* produces QM-OH nanoaggregates with bright fluorescence.

Selectivity and photostability

The selectivity of QM- β gal was investigated by monitoring its fluorescence response after exposure to various potentially





Fig. 2 (A) Absorption and (B) time-dependent fluorescence spectra of QM-βgal (10 μM) with 6 U β-gal in aqueous solution (PBS/DMSO = 95 : 5, v/v, 50 mM, pH = 7.4) at 37 °C. (C) Fluorescence spectra of QM-βgal (10 μM) upon treatment with increasing concentrations of β-gal (0–6 U). Inset (C): fluorescence intensity $I_{560\text{ nm}}$ as a function of β-gal concentration. (D) Fluorescence intensity $I_{560\text{ nm}}$ as a function of time, $\lambda_{\text{ex}} = 434\text{ nm}$. (E) DLS data and (F) SEM of the ensemble system of QM-βgal and β-gal. (G) Reverse-phase HPLC chromatogram of QM-βgal, QM-βgal reaction with β-gal for 20 min, and QM-OH. The eluent is methanol/H₂O (v/v, 8 : 2) mixed solvent. The flow rate is 0.6 mL min⁻¹, and the detection wavelength is 434 nm. (H) HRMS demonstrating the enzyme-activatable mechanism and showing the cleavage product QM-OH.

interfering species, including enzyme species, amino acids, and biomolecules in aqueous solution (PBS/DMSO = 95 : 5, v/v, 50 mM, pH = 7.4) at 37 °C. As shown in Fig. 3, an approximately 15-fold increase in fluorescence intensity was only observed toward β-gal, while the presence of 100 equiv. of other enzyme



Fig. 3 Fluorescence responses of QM-βgal (10 μM) upon incubation with β-gal (6 U) and various other analytes (100 equiv.) in aqueous solution (PBS/DMSO = 95 : 5, v/v, 50 mM, pH = 7.4) at 37 °C, $\lambda_{\text{ex}} = 434\text{ nm}$. I/I_0 represents the fluorescence intensity ratio at 560 nm, and I_0 is the fluorescence intensity of free QM-βgal.

species (cellulase, reductase, lysozyme, and esterase), amino acids (cysteine and homocysteine), and biomolecules (glutathione, dithiothreitol, hydrogen peroxide, and hydrogen sulfide) could give negligible fluorescence changes as low as the basal level under identical conditions. Considering the ionic strength is one of key factors for probe performance in bio-sensing, the fluorescence intensity of the probe with different concentrations of sodium chloride (NaCl) was measured. Increasing the concentration of NaCl from 0 to 960 mM, almost negligible changes in the fluorescence spectrum of the probe were obtained (Fig. S4A[†]), proving that ionic strength does not affect the fluorescence properties of the probe. Additionally, its fluorescence properties did not change either after incubation with Dulbecco's Modified Eagle Medium (DMEM) and Roswell Park Memorial Institute (RPMI 1640) medium, containing fetal bovine serum, amino acids, salts, glucose, and vitamins (Fig. S4B[†]). These data verify the superior selectivity of QM-βgal for β-gal over other competitive analytes and its potential utility as an activatable bioprobe for detecting β-gal in living cells.

Photostability is another key factor to assess the performance of fluorophores for biomedical applications, especially for long-term tracking and bioimaging in disease diagnostics.^{51,52} Herein, time-dependent photobleaching measurements with lasting irradiation (Hamamatsu, LC8 Lightningcure, 300 W) were carried out for testing the photostability of QM-βgal and QM-OH, and the commercial contrast agent indocyanine



