



Cite this: *RSC Adv.*, 2018, 8, 19456

Combining transition metals and transient directing groups for C–H functionalizations

Trisha Bhattacharya,^a Sandeep Pimparkar^{ab} and Debabrata Maiti  ^{*ab}

In the domain of synthetic chemistry, C–H bond activation has always remained in the spotlight for researchers over the last few decades. Although different strategies have been employed to chemically trigger unactivated C–H bonds, transition metal catalyzed directing group (DG) aided C–H bond activation is the most explored pathway of all because of its ability to perform diverse site selective functional metamorphosis. Despite its popularity, tedious synthetic methodology requiring additional steps for the installation and removal of DGs from the target substrate diminishes its efficacy. However, replacement of directing groups by transient directing groups (tDGs) reduces the hurdle to a greater extent without compromising the product yield and selectivity. In this report we have depicted the intense journey of transient directing groups with three (Rh, Ru, and Pd) prevalent second row transition metals.

Received 15th April 2018

Accepted 14th May 2018

DOI: 10.1039/c8ra03230k

rsc.li/rsc-advances

1 Introduction

Carbon–hydrogen bonds are nature's most ubiquitous functional unit. Regardless of their easy accessibility, C–H bonds are quite inert due to a very low polarity difference between the two constituent atoms and a high C–H bond dissociation energy. Therefore direct conversion of C–H bonds into C–C or C–heteroatom bonds poses a great challenge to the synthetic community. Therefore, transition metal catalyzed C–H activation has become one of the most exciting topics in the last couple of decades in the domain of organometallic chemistry.^{1–3}

Even with its indisputable evolution, the major complication lies in achieving position-selective C–H functionalization.⁴ Irrespective of different chemical environments, all C–H bond energy values more or less coincide in the same range, hence it becomes difficult to perform a regioselective C–H bond transformation. Although a series of different strategies have been employed to address this concern, the use of directing groups (DGs) is still an irreplaceable solution.⁵ Directing groups are none other than a variety of σ -coordinating groups that can easily bind to the metal center forming a stable metallacycle which then selectively reaches the proximal C–H bond through its tuned orientation. Employing appropriate directing groups, versatile functionalizations have been achieved at *ortho*-,^{6,7} *meta*-,^{8–11} and recently at *para*-^{12–15} positions of different aromatic rings. This strategy is also useful for selective functionalizations

^aDepartment of Chemistry, IIT Bombay, Powai, Mumbai-400076, India. E-mail: dmaiti@chem.iitb.ac.in

^bIITB-Monash Research Academy, IIT Bombay, Powai, Mumbai-400076, India



Trisha Bhattacharya was born in 1993 in the state of West Bengal, India. She obtained her B.Sc. degree in chemistry from Asutosh College, University of Calcutta in 2015. Then she completed her M.Sc in chemistry at the Indian Institute of Technology, Hyderabad, in 2017 following which she joined Prof. Debabrata Maiti's group at the Indian Institute of Technology, Bombay. Presently, she is

pursuing her PhD on the topic of transition metal catalyzed distal C–H activation.



Sandeep was born and raised in Nasik, Maharashtra. After his M.Sc in Organic Chemistry he joined Prof. Maiti's group in Jan 2015 as a collaborative PhD student between IIT Bombay and Monash University, Australia. His research interest is in Catalytic Chemistry for the development of sustainable and eco-friendly methodologies for the synthesis of important classes of bio-active motifs.



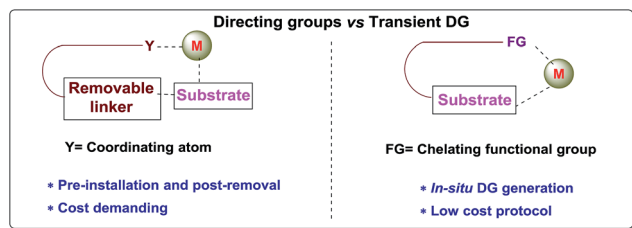


Fig. 1 Directing groups vs. transient directing groups.

at various positions (α , β , γ and δ) of different aliphatic systems.^{16–22} Although enormous progress has been observed in DG assisted C–H bond activation, the major drawback of this approach is the installation of the DG in the substrate and also the removal at the end, and these often demand tedious synthetic procedures. To resolve this problem, two alternative solutions have been applied: (i) use of a transient mediator²³ in a stoichiometric ratio and (ii) implementation of a traceless directing group.^{24,25} As well as these two approaches, an alternative way is to develop a functionally capable directing group which can reversibly connect with the substrate as well as to a metal. The comparison between the concepts of covalently linked directing groups and transient directing groups is shown in Fig. 1.

The concept of transient directing groups relies on the reversible binding of an organocatalyst with a substrate containing a particular functional group. The newly generated entity effectively coordinates with the metal center in a monodentate or bidentate fashion to generate the metallacycle which upon reaction with the coupling partner, furnishes the site selective functionalization (Fig. 2). Previously, detailed reviews on transient DG mediated C–H activation have been reported by Besset²⁶ and Ackermann²⁷ independently from two different perspectives. This work reviews the momentous evolution of transition metal catalyzed C–H activation mediated by transient modifiers. In this mini review we have categorized the entire theme into three major parts which are (a) rhodium catalyzed, (b) ruthenium catalyzed, and (c) palladium catalyzed transient



Fig. 2 Transient directing group mediated C–H activation.

DG assisted C–H activation, with detailed mechanisms and substrate scopes as well as their limitations.

2 Transient DG assisted C–H activation with second row transition metals

2.1 Transient DG assisted C–H activation with rhodium

Rhodium is the most exploited transition metal in terms of transient directing group assisted C–H activation. A series of functionalizations have been reported so far which use rhodium based catalysts. The journey started back in 1979 when Suggs reported a detailed study on the hydroacylation of aromatic aldehydes in three steps where Wilkinson's catalyst [(PPh₃)₃RhCl] (5 mol%) was used to activate the aldehyde C–H bond.²⁸ Of these three steps, one key step was the formation of aldimines from the corresponding aldehydes and amines *via* iminal formation. Starting from 2-amino-3-picoline **A** three different aldimines were prepared, which after reaction with different olefins resulted in the desired hydroacylated products mediated by (iminoacyl)rhodium(III) hydrides. However, the protocol was restricted to very limited variants of aromatic aldehydes with no α -hydrogens. Later, in 1997, Jun and co-workers developed a one-step hydroacylation of aldehydes **2** with different terminal olefins **1** through the *in situ* generation of 3-methyl-2-aminopyridyl aldimines (Fig. 3i).²⁹ The method was not only step-reducing, but a variety of aliphatic aldehydes were also incorporated along with aromatic aldehydes.

The outline of the reaction mechanism was proposed as follows: (1) generation of aldimine **a** from the corresponding aldehyde **2** and 2-amino-3-picoline **A** which acts as a co-catalyst, (2) formation of (iminoacyl)rhodium(III) hydride **b** by C–H activation, (3) olefin coordination at the Rh center forming intermediate **c**, and (4) β -migratory insertion to generate an (iminoacyl)rhodium(III) alkyl complex **d** followed by (5) reductive elimination to form the linear ketone **3** with regeneration of the catalyst again (Fig. 3iii). The reaction was further modified through the use of a catalytic amount of benzoic acid along with aniline as an additive where they both triggered the aldimine formation in the initial step (Fig. 3ii).³⁰ Later, in 2002 the same analogy was applied by Jun *et al.* to obtain the *ortho*-alkylated aromatic ketones starting from a masked ketone *i.e.* ketimine **4** with a variety of unactivated olefins.³¹ The protocol covered



Debabrata Maiti received his PhD from John Hopkins University (USA) in 2008 under the supervision of Prof. Kenneth D. Karlin. After postdoctoral studies at the Massachusetts Institute of Technology (MIT) with Prof. Stephen L. Buchwald (2008–2010), he joined the Department of Chemistry at IIT Bombay in 2011, where he is currently an Associate Professor. His research interests are focused on the

development of new and sustainable synthetic and catalytic methods.





Fig. 3 Intermolecular hydroacylation (i) by Jun and co-workers in 1997 (ii) further modification of Jun's work in 2000 (iii) proposed mechanistic outline for intermolecular hydroacylation.

a reasonable variety of ketones apart from acetone. The same functionalization was further carried out with activated olefins **5** in 2004 by the same group (Table 1).³² The reaction was amenable for a wide range of activated olefins covering linear **5a-b** as well as branched acrylates **5e** as efficient alkylating precursors. Phenyl vinyl sulfone **5c** and acrylonitrile **5e** also provided the *ortho*-alkylated acetophenones but the yields were not satisfactory. However, in this case, the ketone partner was mostly limited to acetophenone.

Table 1 *Ortho*-alkylation of masked ketones with activated olefins



Jun's catalytic system was further revised with an additional phosphine tether which overall offered bidentate coordination to the metal center and subsequently promoted the C–H activation step to provide intermolecular hydroacylated products (Table 2).³³

In the acylated products, different functional groups like ester **3a**, internal alkene **3b**, alcohol **3c**, and acid **3d** were well-tolerated. After the successful intermolecular hydroacylation, the authors intended to extend the scope of this work further to achieve intramolecular hydroacylation, for which they chose *o*-vinylbenzaldehydes **7a-d** in the presence of 5 mol% of [Rh(COD)₂]BF₄ catalyst and 5 mol% of ligand **B** (Table 3). Different functional groups including carboxylic ester **8a**, free phenol **8b**, nitro group **8c**, and chloride **8d** were well-suited to the reaction conditions, earning excellent yields.

Influenced by all of these developments, Dong and co-workers developed an α -alkylation of 1,2-diketones with different types of aromatic as well as aliphatic olefins in the presence of stoichiometric amounts of 2-aminopyridine **C** (Table 4).³⁴

The reaction continued through an enamine mediated C–H activation. A range of electron rich as well as electron deficient olefin coupling sources were equally fruitful. Regardless of

Table 2 Intermolecular hydroacylation with an *in situ* generated bidentate tDG

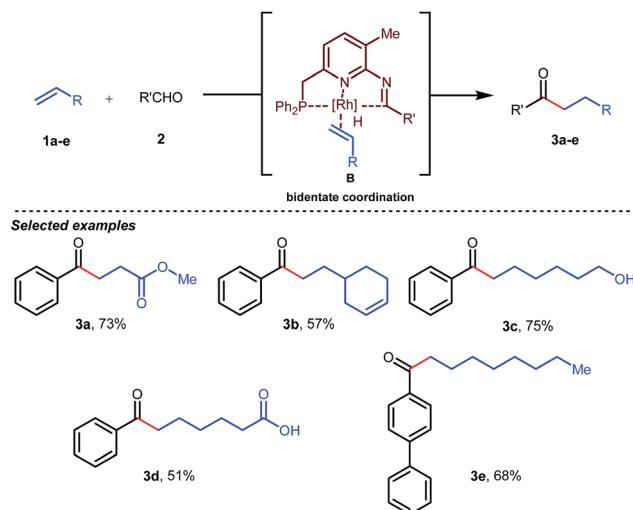
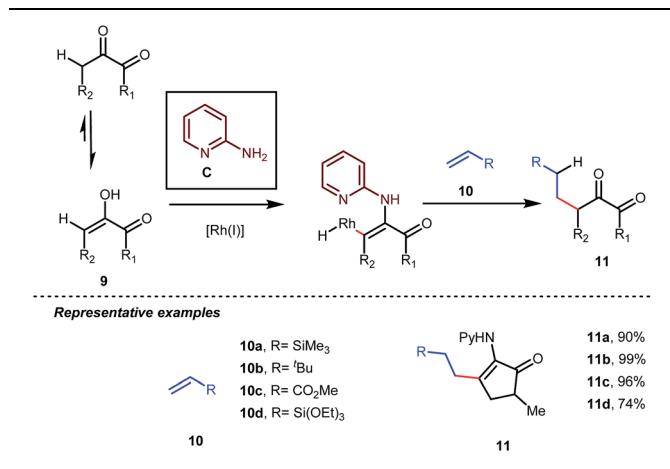


Table 3 Intramolecular hydroacylation with an *in situ* generated bidentate transient DG



Table 4 α -Alkylation of diketones using 2-aminopyridine as a transient modifier



several difficulties with the ketimine formation, in 2014, Mo and Dong reported α -alkylation of cyclic ketone derivatives with a modified rhodium(I) complex $[\text{Rh}(\text{coe})\text{Cl}]_2$ using simple ethylene gas where 7-azaindoline acted as an effective transient directing group (Fig. 4).³⁵ The transformation was highly regioselective for the less crowded 5-position of the cyclic ketone, overriding the electronic bias of the substituents present on the ketones. In particular, this work excluded the possibility of forming other side products. The use of the Rh(I) dimer complex and ligand **D** was found to be essential in this case. The enamine formation in the initial step was promoted by 10 mol% TsOH·H₂O which again supported the proof-of-concept. To accelerate the oxidative addition of enamine C–H bonds, a sterically hindered electron rich NHC ligand (IMes) was employed (5 mol%). A plausible mechanistic outline for this reaction is shown in Fig. 5. At the very beginning, an *in situ* enamine **f** was formed, followed by oxidative addition of the Rh complex, generating **g**. In the next step, olefin **13** coordinated with the metal to form complex **h**, which after reductive elimination generated **i**. Hydrolysis of this intermediate accomplished the desired ketones **14**.



Fig. 4 α -Alkylation of cyclic ketones with the 7-azaindoline tDG precursor.

Apart from imine based transient directing groups, Bedford reported Rh(I) catalyzed *ortho*-arylation of phenols *via* a phosphinite based transient directing group (Fig. 6).³⁶ Previous methods for the preparation of *ortho*-arylated phenols were limited to Suzuki or Stille coupling reactions which demand

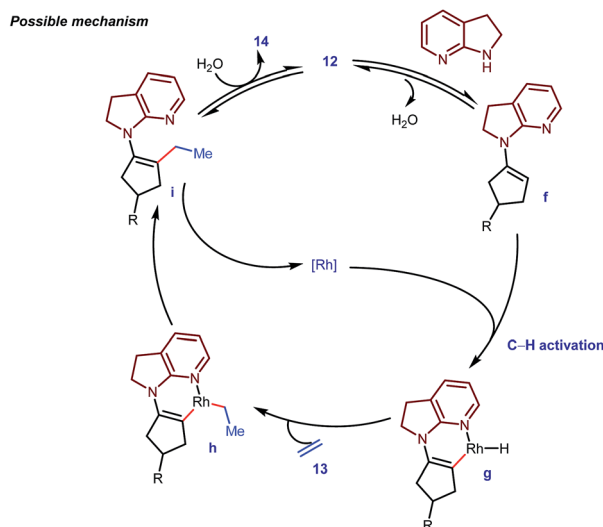


Fig. 5 Proposed mechanism for α -alkylation of cyclic ketones using 7-azaindoline as a tDG.



Fig. 6 Rh-catalyzed *ortho*-arylation of phenols using a phosphite tDG.

stoichiometric quantities of coupling partners like boronic acid derivatives or organotin compounds. On the other hand, the incorporation of aryloxy groups into phosphite or phosphinite systems assured the formation of thermodynamically and kinetically stable low-strain five membered metallacycles for the *ortho*-C–H activation of phenols. The authors envisaged two intertwined catalytic pathways involving directed *ortho*-C–H arylation of phenol assimilated with a $\text{PR}_2(\text{OAr})$ organocatalyst, followed by transesterification of the phosphinite ligand, to release substituted phenol (Fig. 6). Bedford's elegant work was inspired by Lewis's work in 1985 where they elegantly demonstrated ruthenium catalyzed *ortho*-ethylation and deuteration of phenols involving *in situ* reversible transesterification of triarylphosphite ligands with phenol. In 2008, Breit *et al.* used the same phosphinites as temporary directing groups to execute hydroformylation of homoallylic alcohols **17** to furnish γ -lactones and γ -lactols with highly branched regioisomers.³⁷



Early methodologies were selective for the linear hydroformylation of alkenes whereas the use of phosphine or phosphite ligands revised the regiochemistry in favour of branched isomers (Fig. 7). Indeed, both *E* and *Z* alkenes provided the same selective branched lactols **18** under the optimized reaction conditions. Interestingly, Tan and co-workers also achieved selective branched hydroformylation of homoallylic alcohols through an amino phosphine based ligand.³⁸ In 2012, Douglas excellently exploited a similar approach for the design of seven membered cyclic benzoketones *via* intramolecular hydroacylation.³⁹ The report was influenced by Dong's work in 2009, where the presence of a heteroatom in the ketone ring was essential (Fig. 8).⁴⁰

The use of a catalytic amount of ligand **E** accelerated the rate of hydroacylation by reducing decarbonylation of the ketones. The same Rh(I) catalyst was then implemented by Dong and co-workers to furnish α -alkylated ketones, where 2-aminopyridine acted as the key component for the intermediate imine formation. A catalytic quantity of 5-methyl-2-benzoic acid played a crucial role in improving enamine formation.⁴¹ In addition, α -alkenylation of cyclic ketones **23** was achieved with internal alkynes **24** using the same 7-azaindoline **D** as for the transient modifier. The methodology enabled the production of both the α,β - as well as the β,γ -enones in moderate to excellent yields (Fig. 9).⁴² Thereafter, in 2015, Rovis and Piou successfully

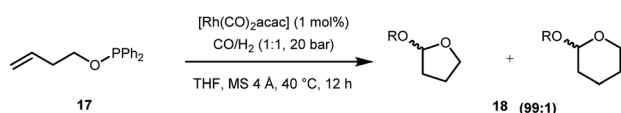


Fig. 7 Rh-catalyzed selective branched lactone synthesis by Breit and co-workers.

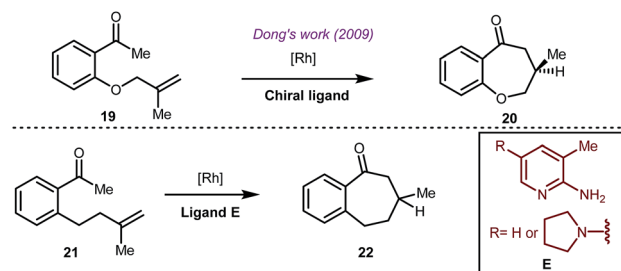


Fig. 8 Rh-catalyzed intramolecular hydroacylation leading to seven membered cyclic ketones.

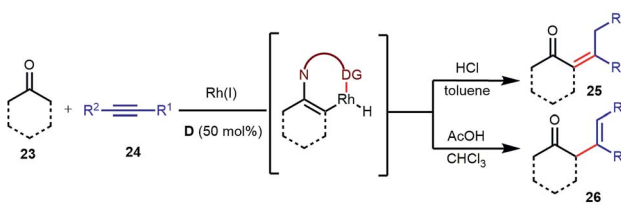


Fig. 9 Rh-catalyzed α -alkenylation of cyclic ketones using 7-azaindoline as a transient modifier.

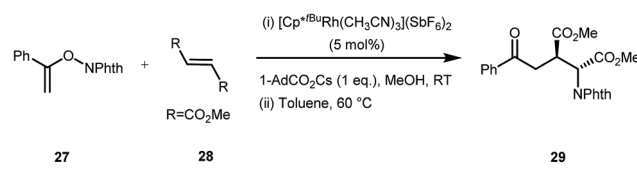


Fig. 10 Rh-catalyzed *syn*-carboamination using enoxyphthalimides as the amine source.

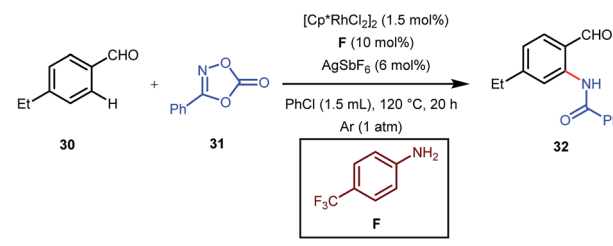


Fig. 11 Rh-catalyzed *ortho*-amination of benzaldehydes.

implemented the *syn*-carboamination of doubly substituted alkenes in the presence of a Rh(III) catalyst with enoxyphthalimides as the amination source (Fig. 10).⁴³ Very recently, Jiao and co-workers demonstrated an imine based transient DG assisted amidation of benzaldehyde derivatives (Fig. 11).⁴⁴ They used 3-phenyl-1,4,2-dioxazol-5-one as the potential aminating source and 4-trifluoromethyl aniline **F** proved its efficacy as a catalytic modifier.

2.2 Transient DG assisted C–H activation with ruthenium

A notable advancement in this domain was achieved by Lewis and co-workers. They reported Ru-catalysed *ortho*-deuteration of phenol using a phosphite transient DG under a D₂ atmosphere. The authors discovered that the presence of a catalytic amount of KOPh can speed up the transesterification process between phenol and triphenylphosphites *via* an active metal-lactone **G** (Fig. 12).⁴⁵ Ruthenium(II) catalyzed *ortho*-arylation of benzaldehydes between imines (masked benzaldehydes) **34** and aryl bromides **35** was also reported by Inoue *et al.* (Fig. 13).⁴⁶ Recently, Rasheed and Zhang synthesized *ortho*-(sulfonylamino)benzaldehydes in the presence of 3 mol% [Ru(*p*-cymene)Cl₂]₂ as the catalyst using organic azides as the source of *N*-functionalities in conjunction with a catalytic quantity of aniline **H** (Fig. 14).⁴⁷

5 mol% AgSbF₆ was found to be the best chloride ion scavenger in this case. With the optimized reaction conditions in

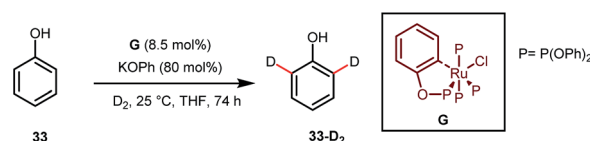


Fig. 12 *Ortho*-deuteration of phenol using a phosphite tDG.



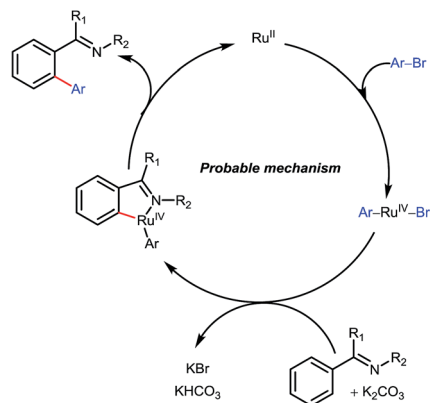
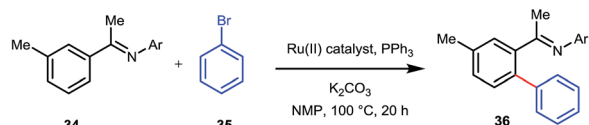


Fig. 13 Mechanistic cycle of Ru-catalyzed *ortho*-arylation of benzaldehydes using aryl bromides as coupling partners.

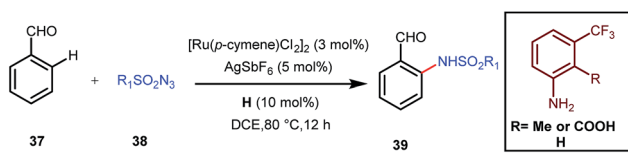


Fig. 14 Ru-catalyzed *ortho*-amination of benzaldehydes using tosyl azide.

hand, the authors explored the generality of this reaction. Both electron donating groups and electron withdrawing groups were well-tolerated. The catalytic cycle for this reaction included the generation of a cationic ruthenium species after ion exchange with AgSbF_6 (Fig. 15). The active catalyst $[\text{Ru}(\text{p-cymene})](\text{SbF}_6)_2$ was thus generated. Eventually, the ruthenium catalyst was bound to the *in situ* formed imine **j** which then underwent C–H activation to form complex **k**. Furthermore, the N center of tosyl azide was ligated to the metal center in an oxidatively additive manner to produce **l**. The exclusion of nitrogen and the concomitant rearrangement led to the complex **m** which upon treatment with acid afforded the *ortho*-aminated product **39**.

2.3 Transient DG assisted C–H activation with palladium

Although Rh-catalyzed transient directing group assisted C–H functionalizations have been well-explored, no such reports with palladium metal catalysts were found in the literature until the work of Yu and co-workers. They disclosed the arylation of $\text{C}(\text{sp}^3)\text{--H}$ bonds of *ortho*-alkyl benzaldehydes in 2016 (Fig. 16).⁴⁸ However, the traditionally-used iminopyridine or iminoxazoline directing groups failed to show adequate directing abilities in this case due to their strong coordination with the Pd metal center, thus disfavoring the targeted C–H bond cleavage. However, a catalytic amount of amino acids solved the problem

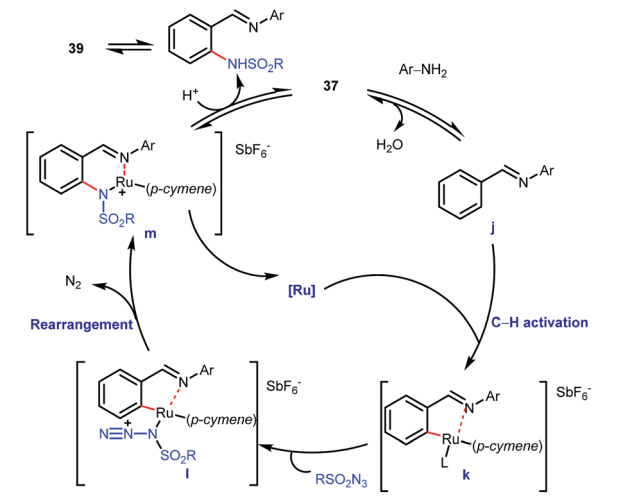


Fig. 15 Ru-catalyzed *ortho*-amination of benzaldehydes.

in the presence of a mixture of protic solvents. A proportionate (3 : 1) mixture of 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP) and acetic acid (AcOH) improved the product formation in the case of the aryl system while a combination of AcOH and water in a ratio of 9 : 1 enhanced the scope for β -arylation of aliphatic ketones.

This functionalization was further extended for the synthesis of fluorenones by the Sorensen group using low-cost anthranilic acid as a transient DG.⁴⁹ In addition to benzaldehydes, *ortho*-arylation of aromatic ketones *via* a glycine transient DG was also reported by Jin *et al.* in 2017.⁵⁰ More recently, Yu *et al.* revealed a series of Pd-catalyzed functionalizations at the *ortho*-position of the benzaldehydes using a bidentate amino acid based transient DG, followed by an Ir(III) catalyzed C–H amidation with monodentate aniline **k** as an effective transient directing group (Fig. 17).⁵¹ A library of functionalized compounds was synthesized with a wide substrate scope and remarkable applications. In a similar way, Hu,⁵² Yu,⁵³ Bull,⁵⁴ and Li and Ge⁵⁵ also achieved β -arylation of unactivated $\text{C}(\text{sp}^3)\text{--H}$ bonds where

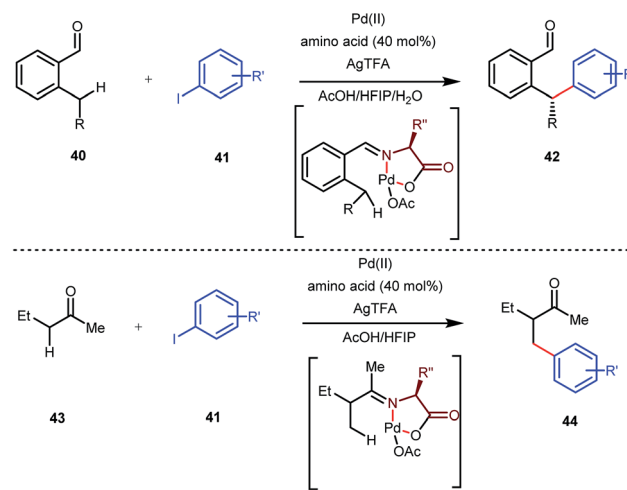


Fig. 16 Pd-catalyzed β -arylation of aromatic and aliphatic ketones.



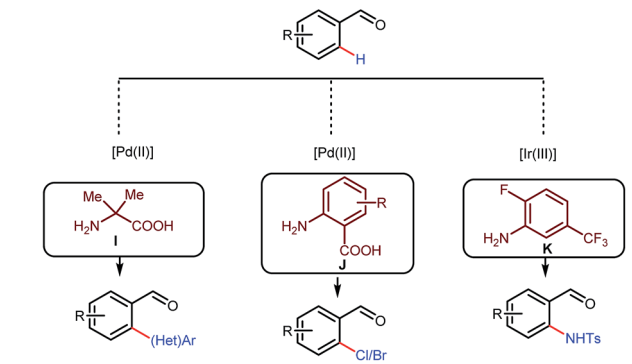
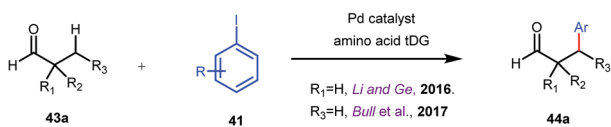


Fig. 17 Pd and Ir-catalyzed *ortho*-functionalizations of benzaldehydes using mono- and bidentate transient ligands.

aceto-hydrazide and β -alanine turned out to be some of the most efficient active transient modifiers (Fig. 18i).

It is worth mentioning that these studies involved arylation of both methyl as well as methylene $C(sp^3)$ -H bonds of the aliphatic carbonyl system. Even though transient DG assisted C-H activation is widespread using Rh and Pd metal catalysts, in 2016, Shi *et al.* first described a transient auxiliary mediated C-H amidation of aryl aldehydes using tosyl azide as the amidating source.⁵⁶ Initially they started with simple aldehyde **45** as their model substrate which failed to give the desired aminated product. Then, they replaced the aldehyde with imine **46** which successfully led to the product in 91% yield. Realizing the importance of the iminyl group, the authors explored the use of aldehydes directly as the substrate, with 3-trifluoromethylaniline in a catalytic amount to carry out the reaction *via* intermediate imine formation which furnished *ortho*-aminated aryl aldehydes as expected, in excellent yields (Fig. 18ii). Until 2016, all *in situ* generated imine based transient templates were utilized to functionalize different aromatic and aliphatic aldehydes or ketone substrates where a catalytic quantity of amine derivatives was always added to the reaction mixture to form the

(i) Seminal examples



(ii) *Ortho*-amination of benzaldehyde

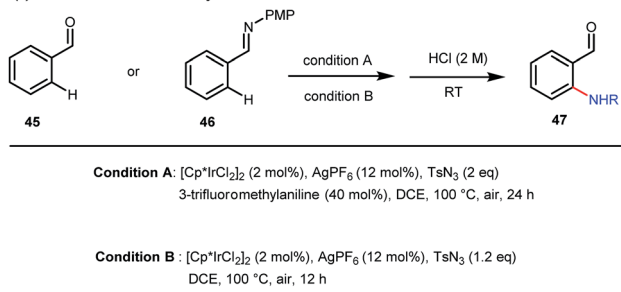


Fig. 18 (i) Seminal examples of Pd-catalyzed β -arylations of aliphatic aldehydes; (ii) *ortho*-amination of benzaldehydes using tosyl azide in the presence of a Pd catalyst.

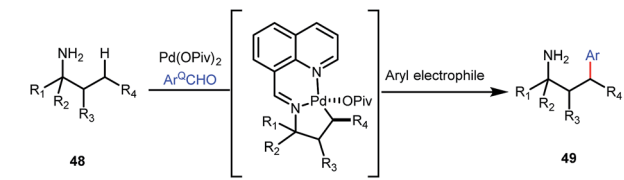


Fig. 19 Pd-catalyzed γ -arylation of free amines.

intermediate imine DG. In 2016, Dong and co-workers first thought to switch ligand *i.e.* they chose free amines as their substrates and added quinoline-8-carbaldehyde in trace amounts to obtain the imine DG.⁵⁷ The *exo* DG enabled efficacious arylation at the γ -position of the free amines (Fig. 19). Later, Ge *et al.* also discovered γ -C-H arylation of free amines with different aryl iodides as aryl coupling partners.⁵⁸ The use of glyoxylic acid as the temporary directing group furnished diverse arylated free amines. A similar strategy was implemented thereafter by the Yu group in 2016 with Boc protected amines to accomplish γ - $C(sp^3)$ -H arylation of free primary amines using 2-hydroxynicotinaldehyde as the transient DG.⁵⁹ In 2017, Shi proposed an atropselective synthesis of biaryls with axial chirality *via* palladium catalyzed olefination using a chiral transient DG.⁶⁰ This strategy involved commercially obtainable *tert*-leucine as the catalytic chiral auxiliary which enabled the formation of chiral biaryls with outstanding enantioselectivity (95–99% ee) (Fig. 20).

Notably, in 2018, Sorensen and co-workers developed reaction conditions for Pd-catalyzed *ortho*-methylation and fluorination of benzaldehydes **53** utilizing potassium methyltrifluoroborate as the methyl source or 1-fluoro-2,4,6-trimethylpyridinium triflate as

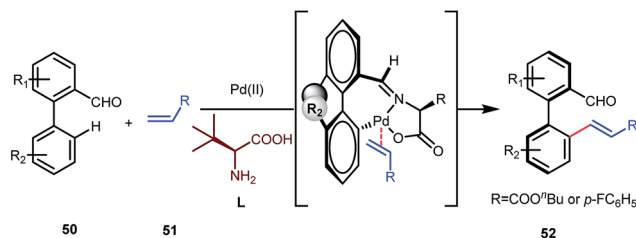


Fig. 20 Pd-catalyzed olefination of biaryls using a chiral amino acid transient DG.

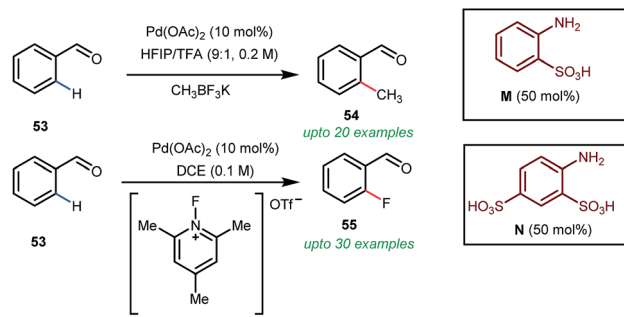


Fig. 21 Pd-catalyzed *ortho*-methylation and fluorination of benzaldehydes.



the effective fluorinating agent in the presence of orthanilic acid **M** (45 mol%) or aniline-2,4-disulfonic acid **N** (50 mol%) respectively (Fig. 21).⁶¹

For the first time, orthanilic acid has been used as a potential transient DG for C–H functionalization. Interestingly, employment of one extra sulfonic acid group in **N** facilitated the C–F reductive elimination step by increasing the electrophilicity of the Pd(IV) center.

This increased electrophilicity then attenuated the polarity difference of the Pd(IV)–F bond to execute a facile reductive elimination. Notably, the reaction has covered a broad substrate scope irrespective of electron rich and electron poor systems. For further mechanistic interest, the authors were able to trap the C–H activated metallacycle in crystal form.

3 Conclusions

In conclusion, transition metal catalyzed functionalization of C–H bonds heralds the commencement of a new horizon for the synthesis of potential complex molecular scaffolds. However, transient directing group mediated C–H activation is one of the recent advances in this field which can equally assure site-selective functional transformation of C–H bonds whilst over-throwing the technical drawbacks of directed C–H activation. This strategy has already been recognized as a significant tool for different functional alterations of C–H bonds which include arylation, amination, halogenation, alkylation, and many more. Despite its hassle-free synthetic pathway, the protocol is solely restricted to different aromatic and aliphatic aldehydes, ketones, amines and phenol derivatives. This strategy is not only functional group specific but it is also restricted to proximal C–H bonds of the respective functional groups. Systematic design of transient DGs can further extend the scope of distal C–H activation. On top of that, the strategy mainly relies on 4d transition metals like rhodium, palladium, ruthenium and iridium. Therefore experiments with 3d transition metals are highly anticipated due to their abundance as well as their eco-friendly nature. Although the approach is yet not generalized for all possible functional transformations, further remodelling will definitely shape the field of C–H activation.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This research is supported by (BRNS-37(2)/14/06/2018) India. Financial support was received from UGC-India (TB) and IITB-Monash Research Academy (SP).

References

- S. M. A. Shakoor, S. Kumari, S. Khullar, S. K. Mandal, A. Kumar and R. Sakhuja, *J. Org. Chem.*, 2016, **81**, 12340–12349.
- B. Gopalakrishnan, S. Mohan, R. Parella and S. A. Babu, *J. Org. Chem.*, 2016, **81**, 8988–9005.
- R. Manikandan and M. Jeganmohan, *Chem. Commun.*, 2017, **53**, 8931–8947.
- T. W. Lyons and M. S. Sanford, *Chem. Rev.*, 2010, **110**, 1147–1169.
- O. Daugulis, H. Do and D. Shabashov, *Acc. Chem. Res.*, 2009, **42**, 1074–1086.
- A. Deb, S. Bag, R. Kancherla and D. Maiti, *J. Am. Chem. Soc.*, 2014, **136**, 13602–13605.
- S. Rakshit, C. Grohmann, T. Besset and F. Glorius, *J. Am. Chem. Soc.*, 2011, **133**, 2350–2353.
- J. Yang, *Org. Biomol. Chem.*, 2015, **13**, 1930–1941.
- A. Modak, T. Patra, R. Chowdhury, S. Raul and D. Maiti, *Organometallics*, 2017, **36**, 2418–2423.
- S. Bag, R. Jayarajan, U. Dutta, R. Chowdhury, R. Mondal and D. Maiti, *Angew. Chem., Int. Ed.*, 2017, **56**, 12538–12542.
- U. Dutta, A. Modak, B. Bhaskararao, M. Bera, S. Bag, A. Mondal, D. W. Lupton, R. B. Sunoj and D. Maiti, *ACS Catal.*, 2017, **7**, 3162–3168.
- S. Bag, T. Patra, A. Modak, A. Deb, S. Maity, U. Dutta, A. Dey, R. Kancherla, A. Maji, A. Hazra, M. Bera and D. Maiti, *J. Am. Chem. Soc.*, 2015, **137**, 11888–11891.
- A. Dey, S. Maity and D. Maiti, *Chem. Commun.*, 2016, **52**, 12398–12414.
- T. Patra, S. Bag, R. Kancherla, A. Mondal, A. Dey, S. Pimparkar, S. Agasti, A. Modak and D. Maiti, *Angew. Chem., Int. Ed.*, 2016, **55**, 7751–7755.
- A. Maji, S. Guin, S. Feng, A. Dahiya, V. K. Singh, P. Liu and D. Maiti, *Angew. Chem., Int. Ed.*, 2017, **56**, 14903–14907.
- S. Guin, A. Deb, P. Dolui, S. Chakraborty, V. K. Singh and D. Maiti, *ACS Catal.*, 2018, 2664–2669.
- Y. Mu, X. Tan, Y. Zhang, X. Jing and Z. Shi, *Org. Chem. Front.*, 2016, **3**, 380–384.
- L. V. Desai, K. L. Hull and M. S. Sanford, *J. Am. Chem. Soc.*, 2004, **126**, 9542–9543.
- H. Y. Thu, W. Y. Yu and C. M. Che, *J. Am. Chem. Soc.*, 2006, **128**, 9048–9049.
- A. Deb, S. Singh, K. Seth, S. Pimparkar, B. Bhaskararao, S. Guin, R. B. Sunoj and D. Maiti, *ACS Catal.*, 2017, **7**, 8171–8175.
- X. C. Wang, W. Gong, L. Z. Fang, R. Y. Zhu, S. Li, K. M. Engle and J. Q. Yu, *Nature*, 2015, **519**, 334–338.
- P. X. Shen, X. C. Wang, P. Wang, R. Y. Zhu and J. Q. Yu, *J. Am. Chem. Soc.*, 2015, **137**, 11574–11577.
- N. Della Ca, M. Fontana, E. Motti and M. Catellani, *Acc. Chem. Res.*, 2016, **49**, 1389–1400.
- J. Luo, S. Preciado and I. Larrosa, *J. Am. Chem. Soc.*, 2014, **136**, 4109–4112.
- Q. Lu, S. Grefies, S. Cembellin, F. J. R. Klauck, C. G. Daniliuc and F. Glorius, *Angew. Chem., Int. Ed.*, 2017, **56**, 12778–12782.
- Q. Zhao, T. Poisson, X. Pannecoucke and T. Besset, *Synthesis*, 2017, **49**, 4808–4826.
- P. Gandeepan and L. Ackermann, *Chem*, 2018, **4**, 199–222.
- J. William Suggs, *J. Am. Chem. Soc.*, 1979, **101**, 489.



- 29 C. H. Jun, H. Lee and J. B. Hong, *J. Org. Chem.*, 1997, **62**, 1200–1201.
- 30 C. H. Jun, D. Y. Lee, H. Lee and J. B. Hong, *Angew. Chem., Int. Ed.*, 2000, **39**, 3070–3072.
- 31 C. H. Jun, C. W. Moon, J. B. Hong, S. G. Lim, K. Y. Chung and Y. H. Kim, *Chem.–Eur. J.*, 2002, **8**, 485–492.
- 32 S. G. Lim, J. A. Ahn and C. H. Jun, *Org. Lett.*, 2004, **6**, 4687–4690.
- 33 N. R. Vautravers, D. D. Regent and B. Breit, *Chem. Commun.*, 2011, **47**, 6635.
- 34 Z. Wang, B. J. Reinius and G. Dong, *J. Am. Chem. Soc.*, 2012, **134**, 13954–13957.
- 35 F. Mo and G. Dong, *Science*, 2014, **345**, 68–72.
- 36 R. B. Bedford, S. J. Coles, M. B. Hursthouse and M. E. Limmert, *Angew. Chem., Int. Ed.*, 2003, **42**, 112–114.
- 37 C. U. Grünanger and B. Breit, *Angew. Chem., Int. Ed.*, 2008, **47**, 7346–7349.
- 38 T. E. Lightburn, M. T. Dombrowski and K. L. Tan, *J. Am. Chem. Soc.*, 2008, **130**, 9210–9211.
- 39 E. V. Beletskiy, C. Sudheer and C. J. Douglas, *J. Org. Chem.*, 2012, **77**, 5884–5893.
- 40 Z. Shen, H. A. Khan and V. M. Dong, *J. Am. Chem. Soc.*, 2008, **130**, 2916–2917.
- 41 H. N. Lim and G. Dong, *Angew. Chem., Int. Ed.*, 2015, **54**, 15294–15298.
- 42 F. Mo, H. N. Lim and G. Dong, *J. Am. Chem. Soc.*, 2015, **137**, 15518–15527.
- 43 T. Piou and T. Rovis, *Nature*, 2015, **527**, 86–90.
- 44 X. Wang, S. Song and N. Jiao, *Chin. J. Chem.*, 2018, **36**, 213–216.
- 45 L. N. Lewis, *Inorg. Chem.*, 1985, **24**, 4433–4435.
- 46 S. Oi, Y. Ogino, S. Fukita and Y. Inoue, *Org. Lett.*, 2002, **4**, 1783–1785.
- 47 O. K. Rasheed, *Synlett*, 2018, **29**, 1033–1036.
- 48 F.-L. Zhang, K. Hong, T.-J. Li, H. Park and J. Q. Yu, *Science*, 2016, **351**, 252–256.
- 49 X. Y. Chen, S. Ozturk and E. J. Sorensen, *Org. Lett.*, 2017, **19**, 1140–1143.
- 50 J. Xu, Y. Liu, Y. Wang, Y. Li, X. Xu and Z. Jin, *Org. Lett.*, 2017, **19**, 1562–1565.
- 51 X. H. Liu, H. Park, J. H. Hu, Y. Hu, Q. L. Zhang, B. L. Wang, B. Sun, K. S. Yeung, F. L. Zhang and J. Q. Yu, *J. Am. Chem. Soc.*, 2017, **139**, 888–896.
- 52 F. Ma, M. Lei and L. Hu, *Org. Lett.*, 2016, **18**, 2708–2711.
- 53 K. Hong, H. Park and J. Q. Yu, *ACS Catal.*, 2017, **7**, 6938–6941.
- 54 S. St John-Campbell, A. J. P. White and J. A. Bull, *Chem. Sci.*, 2017, **8**, 4840–4847.
- 55 K. Yang, Q. Li, Y. Liu, G. Li and H. Ge, *J. Am. Chem. Soc.*, 2016, **138**, 12775–12778.
- 56 Y. F. Zhang, B. Wu and Z. J. Shi, *Chem.–Eur. J.*, 2016, **22**, 17808–17812.
- 57 Y. Xu, M. C. Young, C. Wang, D. M. Magness and G. Dong, *Angew. Chem., Int. Ed.*, 2016, **55**, 9084–9087.
- 58 Y. Liu and H. Ge, *Nat. Chem.*, 2017, **9**, 26–32.
- 59 Y. Wu, Y. Q. Chen, T. Liu, M. D. Eastgate and J. Q. Yu, *J. Am. Chem. Soc.*, 2016, **138**, 14554–14557.
- 60 Q. J. Yao, S. Zhang, B. B. Zhan and B. F. Shi, *Angew. Chem., Int. Ed.*, 2017, **56**, 6617–6621.
- 61 X.-Y. Chen and E. J. Sorensen, *J. Am. Chem. Soc.*, 2018, **140**, 2789–2792.

