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# Sustainability-driven photocatalysis: oxygen-doped g-C<sub>3</sub>N<sub>4</sub> for organic contaminant degradation

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Due to its distinct electrical structure and environmental compatibility, graphitic carbon nitride (g-C<sub>3</sub>N<sub>4</sub>) has become a viable photocatalyst for various applications. Significant initiatives are currently being implemented to enhance the photocatalytic activity of g-C<sub>3</sub>N<sub>4</sub> by adding oxygen dopants to its structure. The unique characteristics of oxygen-doped g-C<sub>3</sub>N<sub>4</sub> (O@g-C<sub>3</sub>N<sub>4</sub>), including enhanced charge carrier mobility and changed electronic structure, make it especially appealing for photocatalytic applications. The synthetic techniques used to create O@g-C<sub>3</sub>N<sub>4</sub> are thoroughly examined in this paper, along with the structural changes brought on by oxygen doping and the processes underpinning its increased photocatalytic activity. The methods for adding oxygen atoms to the g-C<sub>3</sub>N<sub>4</sub> lattice are covered in the synthesis section, including solid-state processes, chemical vapor deposition, hydrothermal synthesis, co-precipitation, and post-treatment procedures. Using these techniques, the type and density of oxygen functional groups may be precisely controlled, allowing O@g-C<sub>3</sub>N<sub>4</sub>'s photocatalytic characteristics to be tailored. O@g-C<sub>3</sub>N<sub>4</sub>'s characteristics are discussed, emphasizing its modified electronic band structure, better surface reactivity, and enhanced light absorption abilities. Recent developments in the field are also presented, exhibiting cutting-edge techniques, including heteroatom doping, nanostructuring, and co-catalyst integration that further enhance O@g-C<sub>3</sub>N<sub>4</sub>'s photocatalytic capabilities. As a versatile photocatalyst, O@g-C<sub>3</sub>N<sub>4</sub> has been extensively reviewed in this study, emphasizing its synthesis processes, structural characteristics, and current developments in improving its photocatalytic activity. Our understanding of O@g-C<sub>3</sub>N<sub>4</sub> is deepened by this review's insights, which also open the door for future research into using the compound in environmentally friendly and sustainable technology.

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## Sustainability spotlight

The mounting challenges caused by the rapid development of urbanization and industry drive the urgent demand for wastewater treatment. As long as these trends continue, the volume of wastewater generated will rise, bringing with it additional toxins, heavy metals, and organic contaminants. The release of untreated or inadequately treated wastewater endangers ecosystems owing to pollution, eutrophication, and public health concerns. This paper demonstrates a dedication to sustainability in materials science and environmental remediation. The focus of the investigation on oxygen-doped g-C<sub>3</sub>N<sub>4</sub> (O@g-C<sub>3</sub>N<sub>4</sub>) yields intriguing advances with evident implications for material sustainability and ecological preservation. The importance of O@g-C<sub>3</sub>N<sub>4</sub> in the development of sustainable solutions cannot be overstated. This material demonstrates its potential to reduce environmental pollution and dependency on energy-intensive, non-renewable resources by harnessing sunlight for photocatalytic breakdown of organic pollutants. Its catalytic characteristics provide an environmentally benign alternative to traditional pollution removal technologies, leading to a greener, more sustainable future. Furthermore, the synthesis strategies mentioned in this article underline the significance of resource efficiency. Researchers are leveraging on the ubiquity of carbon and nitrogen supplies by altering g-C<sub>3</sub>N<sub>4</sub> with oxygen, decreasing the requirement for rare or expensive elements. This strategy is consistent with sustainability goals, emphasizing resource stewardship while providing efficient pollution removal. The structural features of O@g-C<sub>3</sub>N<sub>4</sub> improve its sustainability credentials even further. The material's stability and durability provide long-term efficacy in environmental cleanup, decreasing the need for regular replacements and minimizing waste. Its chemical stability under varied climatic circumstances promotes the concept of sustainability by allowing it to perform over long periods of time. In summary, the study described in this paper on O@g-C<sub>3</sub>N<sub>4</sub> constitutes a significant step toward sustainability in materials science. This material provides an eco-friendly solution for the elimination of organic pollutants by utilizing renewable energy sources, optimizing resource utilization, and boosting long-term performance. Its ability to reduce environmental harm while also contributing to a more sustainable future emphasizes its critical role in tackling global ecological concerns.

## 1. Introduction

The urgent need for wastewater cleanup is motivated by the growing difficulties brought on by the fast expansion of

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urbanization and industrialization.<sup>1</sup> As long as these trends continue, there will be an increase in the amount of wastewater produced, full of new toxins, heavy metals, and organic pollutants.<sup>2</sup> Ecosystems are seriously threatened by the discharge of untreated or insufficiently treated wastewater due to pollution, eutrophication, and public health issues. Adequate wastewater treatment is crucial to solving these urgent problems. Innovative technologies are being created to address the increasing complexity of wastewater compositions and to support sustainable water resource management.<sup>3</sup> These technologies include enhanced oxidation processes, membrane filtration, and biological treatments.<sup>4</sup> In addition to removing pollutants, these methods seek to conserve energy, recover priceless resources, and lessen the environmental effects of wastewater management. Researchers are making critical efforts to secure freshwater supplies, preserve ecosystems, and assure the welfare of populations worldwide by advancing wastewater cleanup.<sup>5</sup>

As a critical area of environmental research and technology, photodegradation is essential for reducing the harmful effects of organic pollutants and toxins on ecosystems and human health.<sup>6,7</sup> Advanced photocatalytic materials have drawn much attention because they provide effective and sustainable methods for removing resistant organic molecules from various environmental matrices.<sup>8,9</sup> As a competitor among these materials, oxygen-doped graphitic carbon nitride (O@g-C<sub>3</sub>N<sub>4</sub>) has shown great promise in photodegradation.<sup>10</sup> With a thorough examination of its synthesis, characteristics, current developments, and applications in the photodegradation of organic pollutants, this review paper digs into the complex world of O@g-C<sub>3</sub>N<sub>4</sub>. It gives a thorough overview of the crucial role O@g-C<sub>3</sub>N<sub>4</sub> plays in tackling today's environmental concerns by illuminating the structural complexities and functional improvements made possible by oxygen doping. Additionally, this review thoroughly analyzes the most recent developments in the industry, providing insightful information on the material's advancing capabilities and its transformative potential to create a cleaner, more sustainable future.

## 2. Synthesis techniques for fabrication of O@g-C<sub>3</sub>N<sub>4</sub>

The promising material O@g-C<sub>3</sub>N<sub>4</sub> has prospective uses in various industries, including photocatalysis and energy storage.<sup>11</sup> Oxygen doping can improve an object's electrical characteristics and performance in multiple tasks. The following are some typical techniques for producing O@g-C<sub>3</sub>N<sub>4</sub>:

### 2.1. Solid state reaction

This process includes combining an oxygen source, such as oxalic acid dihydrate, with a precursor containing carbon and nitrogen, often melamine, and heating the mixture to high temperatures in a controlled environment.<sup>12</sup> During the heating step, *i.e.*, up to 550 °C for 4 hours at a ramp rate of 20 °C min<sup>-1</sup>, the oxygen source's oxygen is integrated into the g-C<sub>3</sub>N<sub>4</sub> lattice to form O@g-C<sub>3</sub>N<sub>4</sub>. The fabricated O@g-C<sub>3</sub>N<sub>4</sub> by Mishra *et al.*<sup>12</sup>

was used to photodegrade methylene blue, rhodamine B, and methyl orange under visible light irradiation. Moreover, using ascorbic acid as the source, Zhang *et al.*<sup>13</sup> manufactured O@g-C<sub>3</sub>N<sub>4</sub> and used the photocatalyst to remove chromium. As can be seen in Fig. 1a,<sup>13</sup> the concentration of oxygen rises in O@g-C<sub>3</sub>N<sub>4</sub>, and new C–O and N–C–O bond was observed compared to pristine g-C<sub>3</sub>N<sub>4</sub> (Fig. 1b–d).<sup>13</sup>

### 2.2. Chemical vapor deposition (CVD)

In CVD, as carried out by Chubenko *et al.*,<sup>14</sup> a suitable substrate and carbon, nitrogen, and oxygen precursor gas are injected into a reaction chamber. The gas-phase precursors break down and deposit O@g-C<sub>3</sub>N<sub>4</sub> onto the substrate at a regulated temperature and pressure. The oxygen doping resulted in the compression of the g-C<sub>3</sub>N<sub>4</sub> lattice, which could be ascribed to the slightly nonstoichiometric composition of the material.

### 2.3. Hydrothermal synthesis

In this process, a precursor solution, including the carbon, nitrogen, and oxygen sources, is enclosed in a high-pressure autoclave and cooked to a high temperature. The oxygen source in this approach is often hydrogen peroxide or urea peroxide. This technique facilitates the development of O@g-C<sub>3</sub>N<sub>4</sub> nanoparticles. According to Chen *et al.*,<sup>15</sup> the obtained g-C<sub>3</sub>N<sub>4</sub> powder (1 g) was combined with 100 mL of a 30 vol% H<sub>2</sub>O<sub>2</sub> solution using ultrasonic dispersion for 20 min. The mixture was then put into a Teflon-sealed autoclave and kept at 120 °C for 6 h. The finished product underwent centrifugation, washing, and 60 °C drying overnight. Fig. 1e (ref. 15) shows that the (002) diffraction peak of O@g-C<sub>3</sub>N<sub>4</sub> moved from 27.3° to 27.6° and was significantly sharper, demonstrating the relationship among the adjacent layers was bolstered, and the interplanar layered distance was reduced due to oxygen atoms doping. As further evidence that the O<sub>2</sub> atoms are successfully incorporated into the g-C<sub>3</sub>N<sub>4</sub> framework, a fresh faint peak at 1208 cm<sup>-1</sup> originating from the tensile vibration of C–O is seen in the O@g-C<sub>3</sub>N<sub>4</sub> sample (Fig. 1f).<sup>15</sup>

### 2.4. Co-precipitation method

This process entails the co-precipitation of metal ions with a source of carbon, nitrogen, and oxygen. After that, the precipitate is thermally processed to create oxygen-doped g-C<sub>3</sub>N<sub>4</sub>.<sup>16</sup> Solgi *et al.*<sup>16</sup> fabricated O@g-C<sub>3</sub>N<sub>4</sub> using the same process and used oxalic acid as the oxygen source.

### 2.5. Ion exchange method

This process involves soaking pre-synthesized g-C<sub>3</sub>N<sub>4</sub> in an oxygen-ion-containing solution, such as one containing nitrate or sulfate ions. The oxygen ions swap out some nitrogen particles in the g-C<sub>3</sub>N<sub>4</sub> lattice through ion exchange, leading to oxygen doping. Researchers carried out similar work, as illustrated in Fig. 1g.<sup>17</sup> HNO<sub>3</sub> was used as the oxygen-ion-containing solution, and g-C<sub>3</sub>N<sub>4</sub> was submerged and hydrothermally treated at 80 °C to form oxygen-doped g-C<sub>3</sub>N<sub>4</sub>.



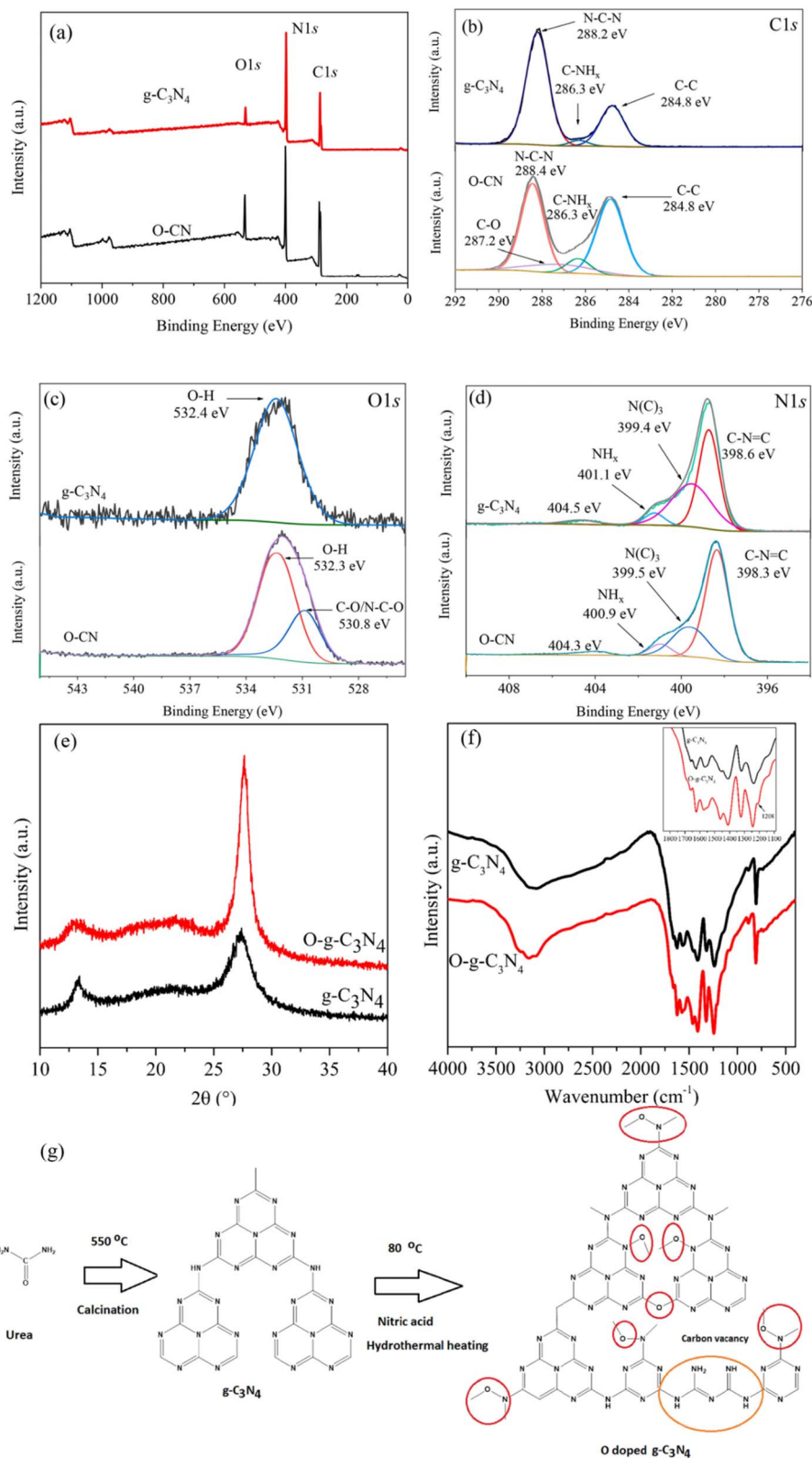


Fig. 1 (a) XPS survey spectra and short scan spectra of oxygen-doped  $g\text{-C}_3\text{N}_4$  with elements (b) C 1s, (c) O 1s, and (d) N 1s<sup>15</sup>; (e) XRD and (f) FTIR spectra of fabricated oxygen-doped  $g\text{-C}_3\text{N}_4$  (ref. 15); (g) ion exchange fabrication of oxygen-doped  $g\text{-C}_3\text{N}_4$ .<sup>17</sup>





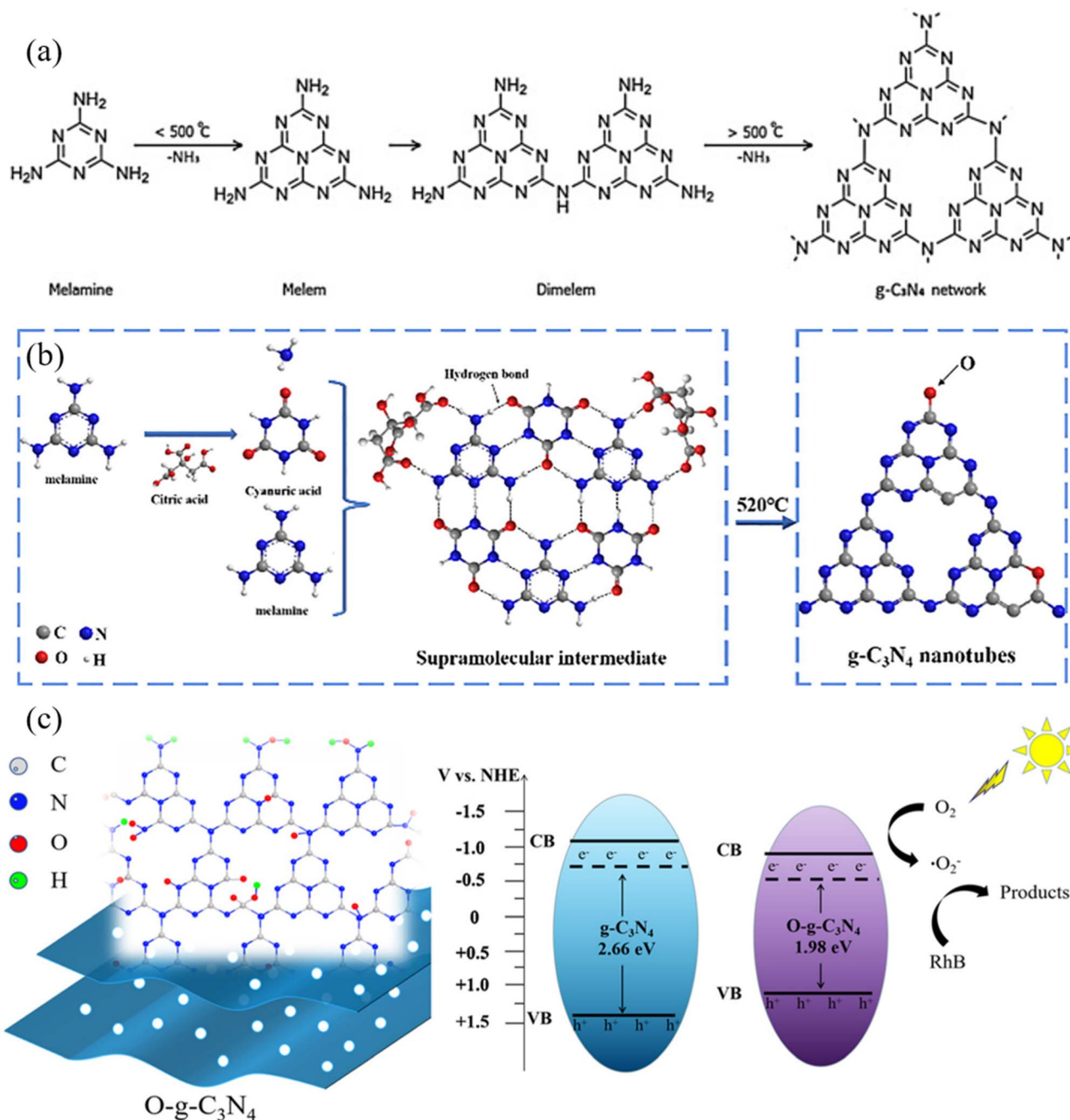


Fig. 2 Schematic illustration of structural changes during the synthesis of (a) g-C<sub>3</sub>N<sub>4</sub> (ref. 20) and (b) O@g-C<sub>3</sub>N<sub>4</sub> (ref. 23); (c) decrease in bandgap and increase in light absorption capacity due to oxygen doping onto g-C<sub>3</sub>N<sub>4</sub>.<sup>15</sup>

#### 4. Recent advancements in the photodegradation performance of O@g-C<sub>3</sub>N<sub>4</sub>

In the field of photodegradation, O@g-C<sub>3</sub>N<sub>4</sub> has recently become a compelling and adaptable photocatalyst.<sup>30</sup> These developments have considerably improved the material's ability to break down harmful organic contaminants and pollutants, revolutionizing the area. The sorts of oxygen functional groups

included in the lattice and the oxygen doping levels can now be precisely controlled thanks to advancements in synthesis techniques. O@g-C<sub>3</sub>N<sub>4</sub> now exhibits increased catalytic activity, expanded spectrum responsiveness, and better charge carrier separation, all of which have improved photocatalytic performance.<sup>31</sup> To further enhance O@g-C<sub>3</sub>N<sub>4</sub>'s photodegradation abilities, synergistic techniques have been investigated, including coupling it with co-catalysts or heteroatom doping. These recent developments highlight O@g-C<sub>3</sub>N<sub>4</sub> as a critical component in the search for environmentally friendly and



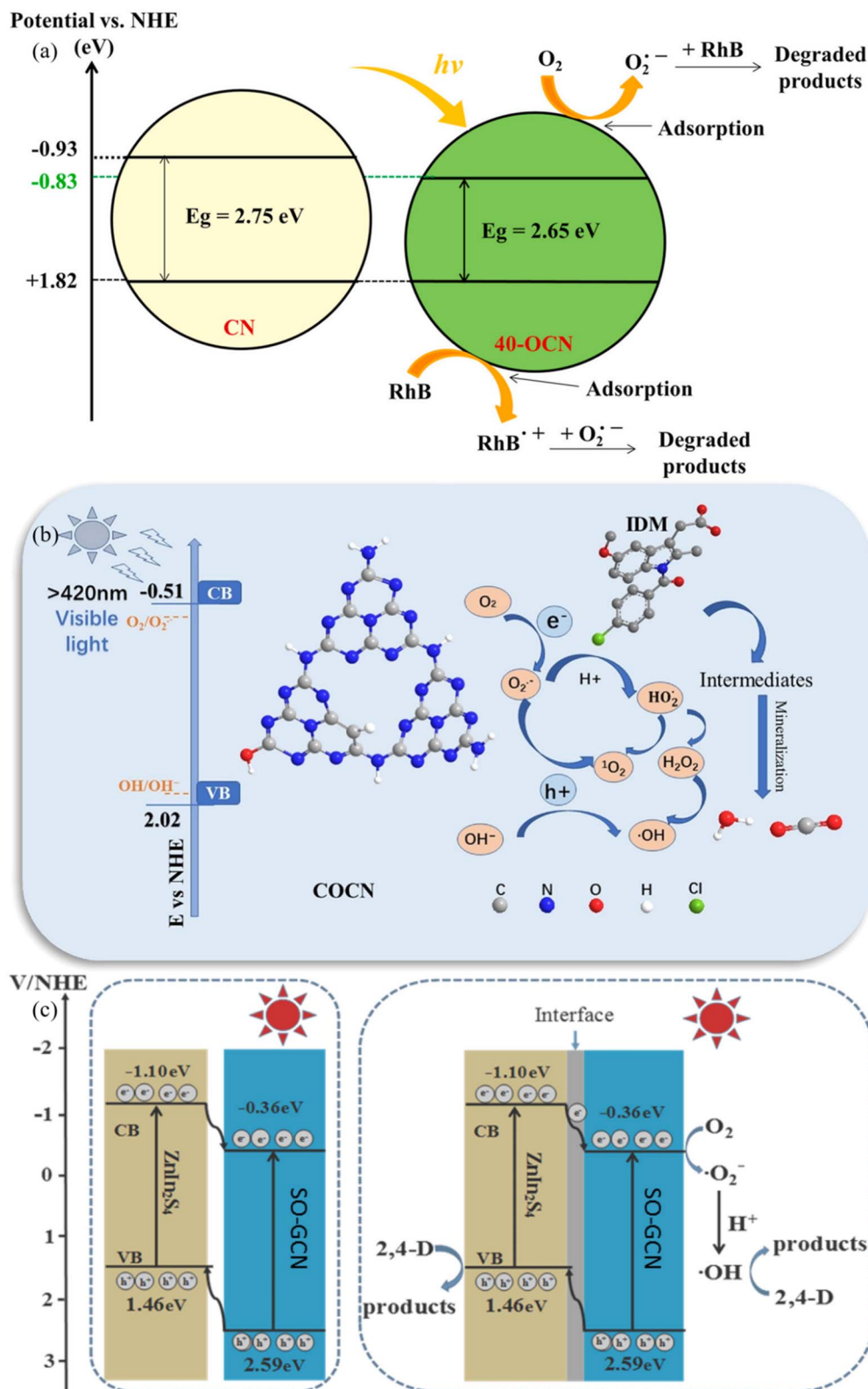


Fig. 3 (a) Degradation mechanism of pristine (a) O@g-C<sub>3</sub>N<sub>4</sub> (ref. 32), (b) O@g-C<sub>3</sub>N<sub>4</sub> developed on a 2D surface<sup>33</sup>, and (c) O@g-C<sub>3</sub>N<sub>4</sub>-based nanocomposite.<sup>34</sup>



Table 1 Summary of the photocatalytic performance of O@g-C<sub>3</sub>N<sub>4</sub> and O@g-C<sub>3</sub>N<sub>4</sub>-based nanocomposite

Photocatalyst	Pollutants	Catalyst dosage (g L <sup>-1</sup> )	Pollutant conc. (mg L <sup>-1</sup> )	Time (min)	Efficiency (%)	Ref.
O@g-C <sub>3</sub> N <sub>4</sub>	Methylene blue	0.2	20	30	95.05 ± 2.17	12
O@g-C <sub>3</sub> N <sub>4</sub>	Rhodamine B	0.2	20	30	97.79 ± 1.58	12
O@g-C <sub>3</sub> N <sub>4</sub>	Methyl orange	0.2	20	30	89.77 ± 2.19	12
O-g-C <sub>3</sub> N <sub>4</sub>	Rhodamine B	1	1.0 × 10 <sup>-5</sup> mol L <sup>-1</sup>	70	90	15
O@g-C <sub>3</sub> N <sub>4</sub>	Tetracycline	0.4	30	120	87.2	29
O@g-C <sub>3</sub> N <sub>4</sub>	Rhodamine B	0.5	1.0 × 10 <sup>-5</sup> mol L <sup>-1</sup>	15	98	35
O@g-C <sub>3</sub> N <sub>4</sub>	Rhodamine B	1	30	140	94	32
O@g-C <sub>3</sub> N <sub>4</sub>	2,4-Dinitrophenol	0.3	10	75	100	24
O@g-C <sub>3</sub> N <sub>4</sub>	Bisphenol A	0.4	10	180	63	36
O@g-C <sub>3</sub> N <sub>4</sub>	Methylene blue	—	—	180	81	25
Carbon doped O@g-C <sub>3</sub> N <sub>4</sub>	Indometacin	0.4	4	60	98.9	33
Carbon doped O@g-C <sub>3</sub> N <sub>4</sub>	Bisphenol A	1	10	240	97.85	37
OCN/Py	Tetracycline	0.16	10	60	97.37	38
Polyoxometalates/O@g-C <sub>3</sub> N <sub>4</sub>	Sulfosalicylic acid	1	40	35	99	39
Protonated g-C <sub>3</sub> N <sub>4</sub> /O-g-C <sub>3</sub> N <sub>4</sub>	Deoxyvalenol	—	10	150	86.6	40
O@g-C <sub>3</sub> N <sub>4</sub> /NaBiS <sub>2</sub>	Methylene blue	1	5	90	81	41
O@g-C <sub>3</sub> N <sub>4</sub> /NaBiS <sub>2</sub>	Rhodamine B	1	5	240	90	41
In <sub>2</sub> O <sub>3</sub> /O@g-C <sub>3</sub> N <sub>4</sub>	Bisphenol A	1	—	180	91	42
β-Bi <sub>2</sub> O <sub>3</sub> /O@g-C <sub>3</sub> N <sub>4</sub>	Bisphenol A	0.5	15	180	98.7	43
O@g-C <sub>3</sub> N <sub>4</sub> /CdS	2-Chlorophenol	—	10	60	100	44
O-doped CN-M/CN-U	Bisphenol A	—	5	90	75	45
POCN/anatase TiO <sub>2</sub>	Enrofloxacin	1	10	60	98.5	26
ZnIn <sub>2</sub> S <sub>4</sub> /SO-GCN	2,4-Dichloro phenoxyacetic acid	0.4	—	180	92	34
O@g-C <sub>3</sub> N <sub>4</sub> /g-C <sub>3</sub> N <sub>4</sub> /TiO <sub>2</sub>	Gatifloxacin	1	10	120	91.7	46

sustainable photodegradation solutions, holding great promise for addressing urgent environmental challenges like detoxifying industrial wastewater and removing persistent organic pollutants from water.

Pristine O-doped g-C<sub>3</sub>N<sub>4</sub> was fabricated by Tran *et al.*<sup>32</sup> for visible light photodegradation of rhodamine B. Compared to pure g-C<sub>3</sub>N<sub>4</sub>, the OCN compounds exhibit better photodegradation of rhodamine B. This is because the photocatalytic activity is improved due to a smaller bandgap energy and a slower recombination rate of generated electrons and holes, as illustrated in Fig. 3a.<sup>32</sup> Furthermore, O@g-C<sub>3</sub>N<sub>4</sub> was developed on carbon surface to photodegrade indometacin (IDM) under visible light irradiation by Zheng *et al.*<sup>33</sup> The improved photocatalytic activity may have been accredited to effective charge separation and a wider visible light absorption zone, as seen in Fig. 3b.<sup>33</sup> The carbon and oxygen-doped g-C<sub>3</sub>N<sub>4</sub> photocatalytic accomplishes excellent mineralization of IDM and offers a new approach to further strategize promising photocatalysts with exceptional wide spectral-responsive characteristics as a promising substitute for conventional water treatment technology. O@g-C<sub>3</sub>N<sub>4</sub>-based nanocomposites, as researched by Uddin *et al.*,<sup>34</sup> demonstrated a 3–5 times faster rate of 2,4-dichlorophenoxyacetic acid degradation under visible light irradiation (*i.e.*, 0.0112/min, by 30%-ZnIn<sub>2</sub>S<sub>4</sub>/SO-GCN), equated to pristine ZnIn<sub>2</sub>S<sub>4</sub> and SO-GCN. This is ascribed to the nanocomposite's harmonious effect in an exclusive 3D/2D heterojunction, which speeds up the e<sup>-</sup>/h<sup>+</sup> separation. The h<sup>+</sup> and ·OH species play a significant part in the breakdown of 2,4-dichlorophenoxyacetic acid over 30%-ZnIn<sub>2</sub>S<sub>4</sub>/SO-GCN, as illustrated in Fig. 3c.<sup>34</sup> Encouraging recent developments in O@g-C<sub>3</sub>N<sub>4</sub> for

photodegradation applications has, in summary, ushered in a new age of environmental remediation and sustainable pollution elimination. The material has evolved into a highly effective and promising photocatalyst, ready to solve the ever-growing ecological pollution and contamination difficulties. Table 1, provided below, summarises these cutting-edge advances in great detail.

## 5. Future perspective and conclusion

Future research and applications in oxygen-doped graphitic carbon nitride (O@g-C<sub>3</sub>N<sub>4</sub>) look promising. More research into precisely adjusting oxygen doping amounts and positions within the g-C<sub>3</sub>N<sub>4</sub> lattice will yield a more excellent knowledge of structure–property correlations. Enhancing photocatalytic effectiveness and expanding the material's uses require developing nanostructuring methods and exploring synergistic codoping with other heteroatoms. Further areas for investigation include O@g-C<sub>3</sub>N<sub>4</sub>'s ability to solve new environmental problems and its incorporation into scalable, affordable technologies for pollutant degradation, energy conversion, and environmental purification. Innovation, practical application, and critical role in long-term solutions for a cleaner and greener world are the hallmarks of O@g-C<sub>3</sub>N<sub>4</sub> research's future. Finally, O@g-C<sub>3</sub>N<sub>4</sub> has become a highly promising photocatalyst with enormous promise for resolving today's ecological and energy concerns. This article has emphasized the crucial significance of oxygen doping in cultivating the photocatalytic activity of O@g-C<sub>3</sub>N<sub>4</sub> by thoroughly studying synthesis techniques, structural alterations, and current developments. The g-C<sub>3</sub>N<sub>4</sub> lattice









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