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# A bio-orthogonal functionalization strategy for site-specific coupling of antibodies on vesicle surfaces after self-assembly†

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Attaching targeting ligands on the surface of self-assembled drug delivery systems is the key request for a controlled transport of the drug to a desired location. Most commonly, the amphiphilic molecules (block-copolymers, lipids etc.) are therefore pre-functionalized before the self-assembly takes place. However, this strategy cannot be applied, if it interferes with the self-assembly process, if the introduced functional groups react with loaded cargo or if natural carriers like extracellular vesicles should be functionalized. Here, we present the site-specific coupling of antibodies to the surface of amino group-terminated liposomes *via* bio-orthogonal copper-free click chemistry after liposome formation. The present primary amino groups were functionalized with a linker carrying a strained alkyne group for a bio-orthogonal strain-promoted alkyne-azide cycloaddition (SPAAC) reaction where Cu(I) as a catalyst can be avoided. Antibodies were site-specifically functionalized with azide moieties along the Fc region to avoid interference with the antigen binding sites. The liposome surface functionalization reaction was optimized by precisely analyzing the number of available functional groups (both amine and alkyne), which often represents a challenge for self-assembled systems. By finally confirming the successful antibody coupling, we provide a facile and robust functionalization strategy, which can be applied to a wide range of self-assembled systems and desired targeting antibodies maintaining physiological conditions throughout the procedure.

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## 1. Introduction

Self-assembled systems of amphiphilic molecules are nowadays applied as drug delivery vehicles as they are biocompatible and can transport hydrophilic as well as hydrophobic cargos. Typically, such self-assembled systems can be categorized into the main groups of polymer-based structures, lipid-based structures and more recently introduced natural vesicles.<sup>1</sup> Polymer-based self-assembled systems consist of amphiphilic block-copolymers, which typically form double-layers and assemble into polymersomes or micelles.<sup>2–4</sup> Regarding lipid-based structures, liposomes, which are small artificial vesicles consisting of one or more phospholipid bilayers, were first introduced by Bangham *et al.* in the 1960s.<sup>5</sup> They are still

of high interest as their composition is similar to natural cellular membranes and as such have an extraordinary biocompatibility.<sup>6–9</sup> Basically, there are four major types of liposomes: (1) MLVs, multilamellar vesicles with several lamellar lipid phase bilayers, (2) SUVs, small unilamellar vesicles with one lipid bilayer,<sup>10–14</sup> (3) LUVs, large unilamellar vesicles, and (4) cochleate vesicles.<sup>15</sup>

Compared to these conventional structures, “second-generation” vesicles are chemically modified to obtain specific properties such as a prolonged blood circulation time or targeting capability. More specifically, vesicles are functionalized with various targeting ligands such as antibodies, peptides, proteins or aptamers, with a variety of surface engineering techniques.<sup>14,16–18</sup> These ligands are mostly chemically attached to the vesicles through interaction of reactive groups on the liposomes’ or polymersomes’ surface and specific groups present in the ligand. For this purpose, the pre-modification of various types of block-copolymers, lipids or cholesterol is often applied to introduce functional groups.<sup>19–24</sup> One example is the commercially available DBCO modified liposome formulation (Immunosome®-DBCO). In this case, the functional DBCO groups were directly introduced *via* the phos-

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pholipid before the liposomes' self-assembly. In a following reaction, azide containing peptides or proteins can be immobilized in the liposomes' surface.

However, direct lipid or polymer functionalization might not be applicable when the self-assembly of the vesicle structure is disturbed by the chemical modification of one component. Similarly, pre-functionalization is not advisable when the functional groups should only be present on the outer vesicle surface to avoid reactions with sensitive cargos. Additionally, the functional moieties, which are introduced prior to self-assembly, might partially be buried in the hydrophobic environment of the formed membrane, as it was demonstrated for polymersomes.<sup>20</sup> Ultimately, also for pre-assembled natural structures like extracellular vesicles<sup>25–27</sup> a different functionalization strategy is required. Therefore, a robust way is needed to enable bio-orthogonal and highly specific functionalization of the vesicle surfaces after self-assembly. Additionally, the proposed functionalization strategy should allow for reliable surface characterization and precise reaction control, which is challenging for preformed self-assembled systems.

For a convenient and reliable coupling of two different functional moieties, click chemistry is widely used due to its unique reaction properties: the procedure is very efficient, mild and can be carried out in aqueous reaction conditions. Click chemistry is rapidly becoming a popular tool to functionalize the surface of biomacromolecules, including viruses, DNA, peptides, antibodies, liposomes, micelles, and nanoparticles with a wide variety of conjugates.<sup>28–30</sup> There are two well established bio-orthogonal coupling approaches: (1) copper-catalyzed azide–alkyne cycloaddition (CuAAC) “click” reactions, which are attractive as the reaction kinetics are favorable and azide-containing reagents are widely available<sup>11,29,31,32</sup> (2) the strain-promoted alkyne–azide cycloaddition (SPAAC) where the use of Cu(I) as a catalyst is avoided.<sup>33</sup> This bio-orthogonal copper-free click chemistry offers a great advantage regarding the known deleterious effects associated with copper in cell-based studies and applications.<sup>34</sup>

In this work, we describe the formation of liposomes based on lipid thin-film hydration followed by extrusion and subsequent surface functionalization of lipid headgroups (Scheme 1). To enable chemical modification *via* the SPAAC



**Scheme 1** Schematic step-by-step illustration of liposome synthesis, surface functionalization and site-selective antibody attachment.



approach, the lipid bilayer was composed out of cholesterol and two different phospholipids – 2-dioleoyl-*sn*-glycero-3-phosphoethanolamine (DOPE) and 1- $\alpha$ -phosphatidylcholine (egg PC) – with a molar ratio of 1 : 1 : 1. The formed liposomes were characterized thoroughly to determine the exact number of available surface primary amine groups ( $-\text{NH}_2$ ). Amino groups were further coupled with dibenzylcyclooctyne-PEG<sub>4</sub>-NHS ester to introduce a strained alkyne (DBCO group) to the liposomes' surface under different reaction pH. Successful modification with DBCO groups was proven by an anthracene azide (Anth-N<sub>3</sub>) assay which was carried out for a precise DBCO group quantification as well as analysis of the stability in the aqueous solution. With this, we provide a universal functionalization strategy for any NH<sub>2</sub>-terminated nanocarrier. In addition, we give a detailed protocol on methods to monitor each functionalization step. Next to this, the antibody CD11c was site-specifically modified with azide groups by transferase and UDP-*N*-azidoacetylgalactosamine (UDP-GalNAz). Subsequently, the azide-functionalized antibody was coupled to the DBCO-functionalized liposomes *via* bio-orthogonal copper-free click chemistry (SPAAC). The successful coupling was confirmed by a flow cytometry based binding assay using fluorescently labeled secondary antibodies.

## 2. Experimental

### 2.1. Materials

All chemicals and materials were used as received. 1,2-Dioleoyl-*sn*-glycero-3-phosphoethanolamine (DOPE) with  $\geq 99.0\%$  purity (10 mg mL<sup>-1</sup> in CHCl<sub>3</sub>), 1- $\alpha$ -phosphatidylcholine (egg PC) with purity  $\geq 99.0\%$  (100 mg mL<sup>-1</sup> in CHCl<sub>3</sub>), cholesterol (powder, BioReagent,  $\geq 99\%$ , Germany) and dimethylsulfoxide (DMSO) were purchased from Sigma Aldrich ( $>99\%$  purity, Germany). The dye DiI Stain (1,1'-dioctadecyl-3,3,3',3'-tetramethylindocarbocyanine perchlorate) for labelling liposomes was purchased from Thermo Fisher (catalog number: D282, Germany). Dibenzylcyclooctyne (DBCO)-PEG<sub>4</sub>-NHS ester (cat. no. CLK-A134-100) was purchased from Jena Bioscience (purity:  $>95\%$  (HPLC), Germany). Phosphate buffered saline (DPBS-buffer,  $-\text{Mg}^{2+}$ ,  $-\text{Ca}^{2+}$ ) was procured from Sigma-Aldrich (USA), while ethanol (99.5%) was acquired from Carl Roth, Germany. A MilliQ device (Merck Millipore, Germany) was used to obtain the demineralized water. Demineralized water was used for all experiments. The antibodies anti-mouse CD11c (clone N418) and the FITC-labeled goat anti-hamster (Armenian, clone Poly4055) was purchased from Biolegend (0.5 mg mL<sup>-1</sup>). The SiteClick™ Antibody Azido Modification Kit was obtained from Thermo Fisher (catalog number: S20026, America).

### 2.2. Liposome preparation

**Preparation of the thin film.** Amine group-terminated liposomes were prepared from cholesterol (Chol) and two different phospholipids: 1,2-dioleoyl-*sn*-glycero-3-phosphoethanolamine (DOPE) and 1- $\alpha$ -phosphatidylcholine (egg PC) with a molar ratio of egg PC : DOPE : Chol = 1 : 1 : 1 by film hydration fol-

lowed by extrusion. Briefly, both lipids and cholesterol stock solutions in chloroform were prepared at a concentration of 10 mg mL<sup>-1</sup>. Then, 835  $\mu\text{L}$ , 767  $\mu\text{L}$  and 398  $\mu\text{L}$  of the egg PC, DOPE and Chol solutions, respectively, were added into a 50 mL round-bottomed flask with an additional 2 mL of a mixture of chloroform with 1 vol% EtOH. The mixture was dried with a rotary evaporator at a reduced pressure of 450 mbar and 3 mbar each for half an hour at 42 °C. Subsequently, the flask was placed in a vacuum oven for 1 hour to remove organic solvent residues (images of the film formation procedure, see Fig. S1†).

**Hydration of the thin film and extrusion.** A volume of 4 mL PBS buffer (0.1 M, pH = 7.4) was added into the flask and the mixture was stirred over night at 550 rpm (Fig. S1†). Then the mixture was sonicated in a water bath for 20 min. In order to obtain small unilamellar liposomes, extrusion was performed. The liposome solutions were extruded through two polycarbonate membranes with pore sizes of 800 nm, 400 nm and 200 nm a number of 11 times at each step. Afterwards, the liposome samples were stored at 4 °C until further use.

### 2.3. Liposome characterization

**Dynamic light scattering (DLS).** Generally, the average size and the size distribution of all liposome samples was determined with diluted dispersions (10  $\mu\text{L}$  sample with a solid content of 0.35 wt% were diluted in 200  $\mu\text{L}$  PBS buffer) with a Zetasizer Nano S90 (Malvern Panalytical GmbH, Germany) at 20 °C. In order to obtain also the angular dependent hydrodynamic radius ( $R_h$ ), radius of gyration ( $R_g$ ) and structure parameter of non-functionalized liposomes, multi-angle DLS was performed with an ALV spectrometer (ALV-GmbH, Germany). The set-up consisted of a goniometer and an ALV/LSE-5004 multiple-tau full-digital correlator with 320 channels. As a light source a He-Ne laser was used at a wavelength of 632.8 nm. The samples were filtered through Millex-LCR 0.45  $\mu\text{m}$  syringe filters (Merck, Germany) into cylindrical quartz cuvettes (18 mm diameter, Hellma, Germany). The cuvettes were cleaned in an acetone fountain prior to usage for removing dust. Liposome samples were prepared in a concentration of 0.003 mg mL<sup>-1</sup> in PBS. The CONTIN algorithm<sup>35</sup> was used for data analysis.

**Zeta potential.** A Zetasizer Nano Z (Malvern Panalytical GmbH, Germany) with disposable folded capillary cells was used to determine the zeta-potential ( $\zeta$ -potential) of the various liposome samples. Basically, 10  $\mu\text{L}$  of a dispersion of each liposome sample (3.5  $\mu\text{g mL}^{-1}$ ) were diluted with 1 mL of a 1 mM potassium chloride (KCl) solution. The measurement was performed at 25 °C after 2 min of equilibration. Each measurement was repeated in triplicate and mean values as well as standard deviations were calculated.

**Cryogenic transmission electron microscopy (cryo-TEM).** For the visualization of the morphology of synthesized liposomes, cryo-TEM measurements were performed. The liposome sample (dispersed in PBS) was placed onto a 400 mesh copper grid covered with lacey film, which was treated with oxygen plasma to make it hydrophilic, and immobilized using high



pressure freezing (Engineering Office M. Wohlwend GmbH, Switzerland). The specimen (sapphire discs with cells) was enclosed and protected in a small volume between two specimen carriers and locked inside the specimen pressure chamber by blotting two times for 3 seconds each. Liquid nitrogen was used as cooling medium. After the preparation, the samples were carefully transferred into the liquid nitrogen for further imaging.

**Quantification of primary amine groups on the surface of liposomes.** The amount of  $\text{NH}_2$  groups present on the liposomes' surface was determined based on a fluorescamine assay (FA assay). Hexylamine, which contains primary amine groups, was selected as a reference for establishing the standard calibration curve. For the assay, 250  $\mu\text{L}$  fluorescamine stock solution (concentration of  $0.3 \text{ mg mL}^{-1}$ ) and 25  $\mu\text{L}$  sample solution ( $\text{H}_2\text{O}$  as the control) as well as 725  $\mu\text{L}$  borate buffer (0.1 M,  $\text{pH} = 9.5$ ) were added into a 2 mL Eppendorf tube. The mixture was vortexed (Heidolph REAX2000 at maximum speed) for 30 seconds and then immediately analyzed in a plate reader (Tecan AG, Switzerland) at 25  $^\circ\text{C}$  by exciting at 410 nm and detecting the fluorescence emission at 470 nm. All fluorescence measurements were repeated three times ( $3 \times 100 \mu\text{L}$  in a well of a 96-well-plate).

#### 2.4. Conjugation of DBCO groups to liposome surface via NHS ester reaction

Firstly, a dibenzylcyclooctyne (DBCO)-PEG<sub>4</sub>-NHS ester ( $M_w = 388.37 \text{ g mol}^{-1}$ ) stock solution was prepared by dissolving 10 mg DBCO-PEG<sub>4</sub>-NHS ester in 250  $\mu\text{L}$  of dry DMSO to obtain a final concentration of  $40 \text{ mg mL}^{-1}$ . Secondly, in order to compare the DBCO conjugation efficiency, different reaction parameters were under consideration: (1) the molar ratio of added DBCO-PEG<sub>4</sub>-NHS ester to amine groups ( $-\text{NH}_2$ ) was varied to be 1 : 1, 3 : 1 and 6 : 1; (2) the reaction pH was varied from 7.1 to 7.6 to 8.2. To achieve the different coupling ratios, 126  $\mu\text{L}$ , 63  $\mu\text{L}$  or 21  $\mu\text{L}$  of DBCO stock solution (concentration:  $40 \text{ mg mL}^{-1}$ ) were added into 500  $\mu\text{L}$  of liposome sample (solid content: 0.35 wt%), separately. The different reaction pH was adjusted by addition of 0.5 M NaOH or 0.1 M HCl solution. All the NHS ester conjugation reactions were carried out by firstly vortexing for 20 s and then stirring (700 rpm) over night at room temperature. Afterwards, the liposomes were purified from excess functionalization reagent and organic solvent by centrifugation at 20 000g and 4  $^\circ\text{C}$  for 1 h. The liposomes were then redispersed in PBS. This washing procedure was performed for a total of three times (see Fig. S2†).

#### 2.5. Quantification of the DBCO groups on the liposomes' surface

An anthracene-azide (Anth-N<sub>3</sub>) assay was performed to determine the DBCO conjugation efficiency. For the Anth-N<sub>3</sub> stock solution (needs to be freshly prepared before the assay): 1.8 mg Anth-N<sub>3</sub> yellow powder was added into 1.05 mL dry DMSO and covered with aluminum foil. The modified liposomes (solid content: 0.35–0.42 wt%,  $\sim 3.5 \text{ mg mL}^{-1}$ ) were dispersed in PBS buffer (0.1 M,  $\text{pH} = 7.4$ ). The mixture solutions

were vortexed for 20 s with maximum speed after addition of 5 different sample types into 1.5 ml Eppendorf tubes: (a–b) 17.3  $\mu\text{L}$  Anth-N<sub>3</sub>/32.6  $\mu\text{L}$  DMSO; (c–d) 25  $\mu\text{L}$  liposome dispersion/17.3  $\mu\text{L}$  Anth-N<sub>3</sub>/7.63  $\mu\text{L}$  DMSO; (e) 25  $\mu\text{L}$  liposome dispersion/25  $\mu\text{L}$  DMSO. All the samples had the same total volume of 50  $\mu\text{L}$ , were covered with aluminum foil and placed on the shaker over night at 25  $^\circ\text{C}$ . For the fluorescence intensity measurement by the plate reader (Ex: 370 nm, Em: 414 nm), all the mixtures were separated into two groups: 10 times dilution and 100 times dilution by DMSO. Namely, 10  $\mu\text{L}$  of the samples were added into 90  $\mu\text{L}$  DMSO and 1  $\mu\text{L}$  of the samples added into 99  $\mu\text{L}$  DMSO. All the measurement data were calculated according to a previously reported procedure.<sup>24</sup>

#### 2.6. Site-specific, enzymatic modification of antibodies

Anti-mouse CD11c antibodies (clone N418, Biolegend,  $0.5 \text{ mg mL}^{-1}$ ) were modified site-specifically on the heavy chain Fc parts with an azide using the Site Click™ Antibody Azido Modification Kit from Thermo Fisher. In the first step, a buffer exchange was performed. Therefore, 250  $\mu\text{g}$  of the anti-mouse CD11c antibody (500  $\mu\text{L}$ ,  $0.5 \text{ mg mL}^{-1}$ ) were loaded onto an antibody concentrator (supplemented with the kit) and centrifuged for 6 min at 5000g. The flow-through was discarded and 450  $\mu\text{L}$  of antibody preparation buffer (supplemented with the kit) were added and centrifuged for 8 min at 5000g. For antibody recovery, the concentrator was inverted, placed into a fresh lo-bind Eppendorf tube (2 mL) and centrifuged (3 min, 1000g) to obtain a concentrated antibody solution (80  $\mu\text{L}$ ,  $2 \text{ mg mL}^{-1}$ , 160  $\mu\text{g}$ ). Afterwards, galactosidase (supplemented with the kit, 10  $\mu\text{L}$ ) was added to the antibody solution and incubated overnight at 37  $^\circ\text{C}$  to remove the galactose residues on the carbohydrate domain at the Fc region. Further, the azide group was enzymatically attached to the carbohydrate modified-antibody. Therefore, 75  $\mu\text{L}$  of  $\text{H}_2\text{O}$ , 12.5  $\mu\text{L}$  of 20× Tris buffer ( $\text{pH} = 7$ ), 25  $\mu\text{L}$  of buffer additive and 80  $\mu\text{L}$  of GalT (Y289L) enzyme as well as the prepared carbohydrate modified antibody were added to a tube containing UDP-GalNAz (all components supplemented with the kit) and incubated overnight at 30  $^\circ\text{C}$ . To remove excess UDP-GalNAz, the azide-modified antibody was purified with a large antibody concentrator (supplemented with the kit). Therefore, the large antibody concentrator was pre-washed with 1 mL of 1× Tris ( $\text{pH} = 7$ ) via centrifugation (10 min, 1200g). Afterwards, the azide-modified antibody was loaded onto the large antibody concentrator and 1.6 mL of 1× Tris ( $\text{pH} = 7$ ) was added. The concentrator was centrifuged (6 min, 1200g) and the flow-through was discarded. This step was repeated three times. Finally, the concentrator was inverted, placed into the conical collection tube and centrifuged for 3 min at 1000g. The azide-modified antibody ( $\sim 40 \mu\text{L}$ ,  $2.6 \text{ mg mL}^{-1}$ ) was recovered, transferred to a fresh lo-bind Eppendorf tube (1.5 mL) and stored at 4  $^\circ\text{C}$ .

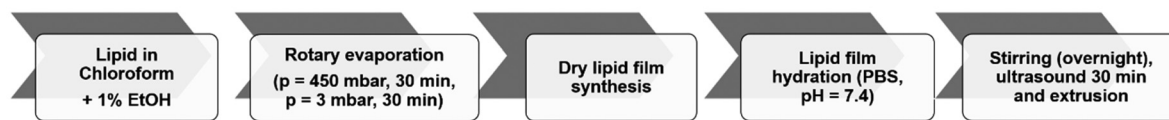
#### 2.7. Liposome surface functionalization with antibody via copper-free click chemistry

167  $\mu\text{L}$  DBCO-liposome ( $\sim 3 \text{ mg mL}^{-1}$  assumption = 500  $\mu\text{g}$ ) and 6.7  $\mu\text{L}$  of azide-modified anti-mouse CD11c-antibody





## A Liposome synthesis



## B Physicochemical Characterization

Synthesized liposomes	Parameters
$R_h$ / nm	$97 \pm 10$
PDI	0.141
Zeta potential / mV	$-16 \pm 7$
Membrane thickness $d$ / nm	$4.0 \pm 0.5$
Liposome number / mL	$(1.0 \pm 0.2) \times 10^{13}$
NH <sub>2</sub> groups / liposome	$(1.6 \pm 0.3) \times 10^5$
Solid content / wt%	0.32–0.38
Composition and molar ratio	Egg PC: DOPE: Cholesterol = 1:1:1

## C



## D



Fig. 1 (A) Schematic overview of liposome preparation. (B) Physicochemical characterization of synthesized liposomes. C, (D) cryo-TEM images of liposomes. From images with high magnification (D) the membrane thickness of the liposomes  $d$  was determined.

sequent coupling reactions. According to the fluorescamine assay (reaction scheme see Fig. S6<sup>†</sup>), the synthesized egg PC/DOPE/Chol liposomes contained  $(1.68 \pm 0.01) \times 10^{18}$  NH<sub>2</sub> groups mL<sup>-1</sup> corresponding to  $(1.6 \pm 0.3) \times 10^5$  NH<sub>2</sub> groups per liposome.

### 3.2. Conjugation of DBCO groups to liposomes via NHS ester reaction

The DBCO-PEG<sub>4</sub>-NHS ester was selected for liposome surface modification to introduce active alkyne groups, which react specifically and efficiently with a primary amine ( $-\text{NH}_2$ ) at pH 7.0–8.5 to form a covalent bond (see Fig. 2A). This is especially desired since amine groups can also easily be introduced into *e.g.* polymersomes and are present on other natural carrier systems such as extracellular vesicles, where they mostly appear in the form of membrane proteins. Like this, the functionalization strategy could directly be adapted for all kinds of self-assembled structures. The hydrophilic polyethylene glycol (PEG) spacer arm imparts water solubility and provides a flexible connection that minimizes steric hindrance involved with ligation to complementary azide-containing molecules. This spacer can further be varied in length to later provide optimal accessibility of the coupled compound.

At this point, it is worth to point out that the liposomes (including all other self-assembled structures comprised of amphiphilic molecules) are sensitive towards organic solvents, which can cause liposome aggregation. Thus, the DBCO-PEG<sub>4</sub>-NHS ester was dissolved in DMSO in a very high concentration ( $40 \mu\text{g} \mu\text{L}^{-1}$ ) so that only a minimal addition of organic solvent DMSO was required to carry out the functionalization

reaction successfully. With this low amount of organic solvent, no aggregation of liposomes was observed.

The reaction of NHS esters with amines is strongly pH-dependent: on one hand at a low pH, the amino groups on the liposomes' surface will be protonated and therefore the reactivity is decreased. On the other hand, if the reaction is performed at higher pH, hydrolysis of the NHS ester will be very fast and thus the functionalization yield of liposomes will be reduced. Therefore, optimization of the pH value for this reaction is crucial. In our work, different reaction pH values (7.1, 7.6 and 8.2) as well as different molar ratios of DBCO/NH<sub>2</sub> groups (1:1, 3:1 and 6:1) were compared carefully with regards to the modification efficiency.

### 3.3. Quantification of DBCO groups via an anthracene-azide (Anth-N<sub>3</sub>) assay

For further antibody click reaction, it was important to determine the total number of DBCO groups that could participate in a further click reaction. Therefore, a click reaction of Anth-N<sub>3</sub> (9-(azidomethyl) anthracene) to the Lipo-DBCO samples was performed for the quantification of the surface alkyne groups. This procedure was described in literature by Baier *et al.*<sup>31</sup> The Anth-N<sub>3</sub> click reaction mechanism is shown in Fig. 2.B. Anth-N<sub>3</sub> itself is already fluorescent ( $I_0$ , in eqn (1)) (Ex: 370 nm, Em: 414 nm). However, after the click reaction with the Lipo-DBCO the quantum yield increases by a factor of 48 resulting in an increase of the overall fluorescence intensity ( $I$ , in eqn (1)) due to a photoinduced electron transfer (PET) effect. A 48-fold increased quantum yield was used for our calculation for the following reason: the quantum yield of the azide is 0.02 and the newly formed triazole has a quantum





**Fig. 2** Quantification of DBCO groups on the liposomes' surface. Schematic illustration of (A) the functionalization procedure with a DBCO-PEG<sub>4</sub>-NHS ester. (B) The anthracene-azide assay: the coupling of anthracene provides fluorescence emission proportional to the number of DBCO groups. (C) Anthracene-azide assay quantification results for different reaction conditions (varied DBCO/NH<sub>2</sub> group ratio and different reaction pH). (D) DBCO group quantification after different storage times.

yield of 0.95–0.97 based on literature reports.<sup>37</sup> Therefore, the ratio and increase of the quantum yield is  $0.96/0.02 = 48$ . The amount of DBCO groups on the liposome surface could further be quantitatively calculated *via* eqn (1) and (2). In the equations,  $\Omega$  represents the fluorescent enhancement and  $n_{(\text{react-Azido})}$  the total amount of reacted azide groups, which can be calculated. The obtained values are as shown in Fig. 2C.

$$\Omega = \frac{I}{I_0} \quad (1)$$

$$n_{(\text{react-Azido})} = \left(\frac{\Omega}{48}\right)n_{(\text{Anth-N}_3)} \quad (2)$$

In the Anth-N<sub>3</sub> assay, the quantitative determination<sup>38–40</sup> of DBCO groups on the liposomes' surface was performed for the three different DBCO/NH<sub>2</sub> group ratios 1 : 1, 3 : 1 and 6 : 1 with three different reaction pH values (7.1, 7.6 and 8.2) (as shown in Fig. 2C). From the measured fluorescence, the number of DBCO groups per liposome were calculated. Additionally, the DBCO conjugation stability after the reaction with the liposomes was determined (Fig. 2C and D). We found that the



highest modification efficiency is observed for a 3-fold molar ratio of DBCO to liposomes for the same reaction pH value. In this case, the maximum conjugation efficiency is  $\sim 20\%$ . In addition, the conjugation efficiency increased slightly along with the increasing reaction pH for all reaction molar ratios between DBCO/NH<sub>2</sub>. Next to this, the stability of the DBCO groups after the liposome modification was investigated (Fig. 2D). The value of DBCO groups per liposomes decreased by more than 50% after 7 days compared to the fresh samples. Moreover, some floccules started to appear within the Lipo-DBCO suspension, especially for the molar ratio DBCO/NH<sub>2</sub> = 6 : 1 after 7 days. This means that further coupling by click chemistry ideally should be performed within the same day.

Since during this reaction optimization step it was found that the ratio of DBCO/NH<sub>2</sub> = 6 : 1 did not result in further increase of the reaction yield, but rather in a significant decrease, subsequent coupling of the chosen antibody was only performed for the 1 : 1 and 3 : 1 ratio conditions, which were coupled at a reaction pH of 8.2.

### 3.4. Antibody functionalization and coupling to liposomes via copper-free click reaction

Most widely in literature, antibodies are functionalized *via* NHS-chemistry.<sup>41–43</sup> In this case, the functional group is attached on the lysine residues or the N-termini of the antibodies. This functionalization strategy is highly unspecific as



**Fig. 3** Antibody functionalization and attachment to surface-modified liposomes. (A) Enzymatic removal of galactose from the Fc-part of the CD11c-antibody with galactosidase. (B) Site-specific, enzymatic attachment of an azide functionality using UDP-N-azidoacetylgalactosamine (UDP-GalNAz) to the sugar residues on the Fc-part of the CD11c-antibody (C) bio-orthogonal copper-free click reaction to attach the azide-modified antibody on the DBCO-modified liposome surface.



lysine residues are distributed over the whole antibody sequence including the antigen-binding site (Fab region).<sup>44</sup> Next to this, the two N-termini of the antibody are also located within the Fab region. Therefore, introducing functional groups along those groups can cause a loss of the antibody functionality and decreased cellular recognition.<sup>36</sup> Therefore, we chose a site-selective antibody functionalization strategy to overcome the loss of binding specificity and antibody functionality. Monoclonal antibodies have two conserved *N*-glycosylation sites on the heavy chain of the fragment crystallizable region (Fc-part).<sup>45</sup> It is known that *N*-glycosylation occurs at the amino acid asparagine (Asn) in the position Asn-297.<sup>46</sup> Therefore, here we used a functionalization strategy, which allows the site-selective attachment of a maximum of four azide groups (two per heavy chain) at the Asn-297 position to the CD11c antibody. It is a two-step protocol, where in the first step galactose is enzymatically removed from the sugar-tree and an *N*-acetylglucosamine residue is exposed (Fig. 3.A). In a second step, an azide-labeled sugar derivate (GalNAz, *N*-azidoacetylglucosamine-tetraacylated) is enzymatically attached *via* the galactose-specific transferase ( $\beta$ 1,4-galactosyltransferase mutant Y289L-Gal-T1) to the glycans of the antibody (Fig. 3.B). This method was first developed for the site-selective

radiolabeling of antibodies.<sup>47</sup> In this case, the azide-modified antibody was conjugated with a desferrioxamine-modified dibenzocyclooctyne (DIBO) in a copper-free click reaction and <sup>89</sup>Zr was further added for radiolabeling. The same antibody functionalization strategy was also applied for engineering site-specific antibody–drug-conjugates.<sup>48,49</sup>

To investigate the structural and functional integrity of the azide-modified antibody compared to the native antibody, nanoDSF measurements, SDS-PAGE (Fig. 4) and a cell binding assay (Fig. S7†) were performed. nanoDSF is a method to determine protein stability based on the intrinsic fluorescence intensity of a protein upon heating. The characteristic melting temperature ( $T_M$ ) is defined at the point, where 50% of the protein is unfolded. For the native anti-mouse CD11c antibody (blue) and the azide-modified antibody (orange), the thermal behavior determined by nanoDSF was highly comparable (Fig. 4A). Next to this, the melting temperature before and after functionalization was about 67 °C, which is in accordance with literature reports for other antibodies.<sup>50</sup> This indicates that the azide modification of the antibody had no major impact on the structural properties of the antibody.

Next to this, the unmodified (1) and modified (2) antibody was applied to SDS-PAGE (Fig. 4C) to investigate molecular



**Fig. 4** Characterization of the azide-modified antibody. (A) and (B) nanoDSF measurements of the unmodified and azide-modified anti-mouse CD11c antibody (10  $\mu\text{L}$ ,  $\sim 0.5 \text{ mg mL}^{-1}$ ). The antibody was heated up ( $1 \text{ }^\circ\text{C min}^{-1}$ ) from 20  $^\circ\text{C}$  to 95  $^\circ\text{C}$  and the intrinsic fluorescence intensity at 330 nm and 350 nm was recorded. The ratio of the fluorescence intensity (350/330 nm, A) and the first derivation (B) is plotted. The melting temperature  $T_M$  is determined from the maximum of the 1<sup>st</sup> derivation of the ratio. (C) SDS-PAGE of unmodified (1), azide modified CD11c (2) and PEGylated CD11c antibody (3). Proteins (3  $\mu\text{g}$ ) are reduced, applied to an SDS-PAGE and protein bands are visualized by Silver Staining.



weight differences and possible degradation products after antibody functionalization. We could observe no significant difference for the azide-modified antibody compared to the unmodified one. To verify the successful azide functionalization of the antibody, a model reaction with a DBCO-PEG-linker (5 kDa) was carried out and the resulting conjugate was applied to SDS-PAGE. In lane 3 (Fig. 4C) there are two additional protein bands visualized, which correspond to the heavy chain (HC) fragment of the antibody (50 kDa) modified with one DBCO-PEG<sub>5 kDa</sub> (+5 kDa) and thus one azide group or an HC fragment conjugated with two DBCO-PEG<sub>5 kDa</sub> (+10 kDa) and thus two azide groups per HC fragment. This means for the full antibody (having two HC fragments) the labeling efficiency is between 2–4 azides per antibody. This is also in agreement with literature reports, stating a labeling degree of  $2.7 \pm 0.2$  GalNAz residues per antibody.<sup>47</sup>

To determine the functional binding properties of the modified antibody, a DIBO-modified fluorescent dye (PE) was incubated with the azide-modified anti-mouse CD11c antibody. Murine dendritic cells (DC2.4), which express CD11c antigen, were incubated with the PE-modified anti-mouse CD11c antibody for 30 min at 4 °C (Fig. S7†). The PE-modified anti-mouse CD11c antibody strongly bound to the DC2.4 cells. To confirm the antibody specificity, cells were pre-treated with

free unlabeled anti-mouse CD11c antibody (blocking assay). In this case, the PE-modified anti-mouse CD11c antibody was not able to bind to the DC2.4 cells proving the receptor specificity of the PE-modified anti-mouse CD11c antibody.

In a last step, the azide-modified anti-mouse CD11c antibodies were incubated with the DBCO-functionalized liposomes overnight at room temperature (Fig. 3C). Via the bio-orthogonal copper-free click reaction (SPAAC), it was possible to covalently immobilize the azide-modified antibodies on the DBCO-functionalized liposomes' surface.

To prove the successful coupling, flow cytometry measurements were performed using our previously established protocol.<sup>36</sup> Here, we chose a fluorescently-labeled secondary antibody (FITC-anti hamster antibody), which specifically binds to the immobilized anti-mouse CD11c antibodies (clone N418, host species Armenian Hamster) on the liposomes' surface by recognition of the Fc-region (Fig. 5A). As a control, liposomes functionalized with DBCO only were incubated with the FITC-labeled secondary antibody.

As a result of the flow cytometry analysis, we detected a high amount of FITC-fluorescence positive liposomes for Lipo-CD11c (~60% to 80%, Fig. 5B). In contrast, there was a very low unspecific binding of the FITC-labeled secondary antibody to the Lipo-DBCO samples (<20%) as expected without any

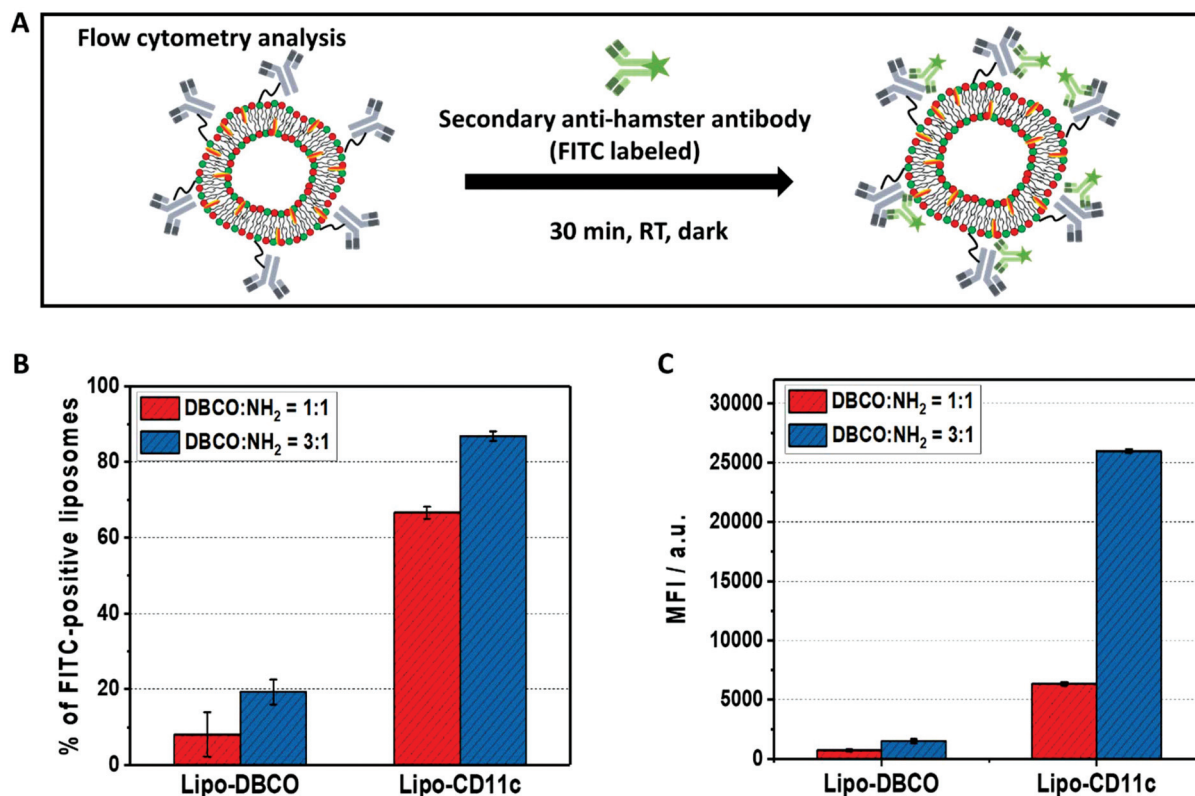


Fig. 5 Verification of antibody presence on the liposomes' surface. (A) Schematic illustration of the flow cytometry measurements for the antibody quantification on the liposome surface. (B) and (C) Flow cytometry measurements for liposomes functionalized with anti-mouse-CD11c antibodies. A secondary labeled FITC-anti-hamster antibody was used to detect the antibodies on the liposome surface. Different reaction conditions (ratio between DBCO/NH<sub>2</sub> groups 1:1 or 3:1) were investigated. The number of FITC-positive liposomes in % (B) or the median fluorescence intensity (MFI) is shown (C).



attached proteins being present. Along with the quantification of the DBCO groups on the liposomes' surface (Fig. 2B), a higher amount of antibodies was detected for liposomes modified with the DBCO/NH<sub>2</sub> = 3 : 1 ratio (Fig. 5C). This means that, indeed, the number of attached antibodies was also proportional to the number of available DBCO groups on the liposomes and the proposed coupling strategy provides precise control over the obtained liposomal surface. It has to be noted, however, that the exact number of antibodies per liposome cannot be deduced from the fluorescence intensity, as the unspecific binding also has to be taken into account. The theoretical maximum number of antibodies per liposome was calculated to be 42 (see Table S1†).

Finally, the liposome-antibody conjugates were again characterized with regards to their physico-chemical properties as shown in Fig. 6. The size and surface charge of the non-functionalized as well as the functionalized liposomes was compared for both coupling ratios DBCO/NH<sub>2</sub> 1 : 1 and 3 : 1.

As indicated in Fig. 6.A and C, the size of the liposomes increased step-wise after functionalization. Coupling of the DBCO groups and the antibody CD11c slightly enlarged the liposomes hydrodynamic diameter from 194 nm up to 293 and 325 nm (1 : 1 and 3 : 1 DBCO/NH<sub>2</sub> groups, respectively). There are mainly two reasons: (1) for the DBCO conjugation/antibody

click reaction, the DBCO-liposomes and the liposome-antibody conjugates have to be purified from free excess DBCO-PEG<sub>4</sub>-NHS ester and further free unreacted antibodies by centrifuging 3 times at 20 000g. During these washing steps, smaller liposomes could also be removed. (2) Long chain molecule PEG and DBCO groups as well as the attached antibodies result in larger overall liposomes. Furthermore, we observed that the zeta potential of the functionalized liposomes was more negative compared to the non-functionalized surface. After attachment of the CD11c antibodies, the zeta potential approached a value of -25 mV (for the 3 : 1 DBCO/NH<sub>2</sub> ratio), which is typical for nanocarrier surfaces covered with proteins.

Finally, with the performed characterization approaches, we are confident that we have presented a facile functionalization approach for bio-orthogonal and site-specific attachment of antibodies on a liposome surface. By making use of specific quantification techniques, the available surface functional groups can precisely be analyzed so that the click chemistry reaction stoichiometry can be adjusted to yield an optimal efficiency. With the site-specific antibody modification, additionally it can be assured that the final conjugation product carries recognizable antibody units, which is of importance for further application. The availability of the coupled antibody could even be increased if necessary upon introduc-



Sample type:	DBCO:NH <sub>2</sub> = 1:1			DBCO:NH <sub>2</sub> = 3:1		
	Lipo	Lipo-DBCO	Lipo-DBCO-CD11c	Lipo	Lipo-DBCO	Lipo-DBCO-CD11c
<i>D</i> / nm	194 ± 19	249 ± 25	293 ± 29	194 ± 19	272 ± 27	325 ± 33
PDI	0.176	0.005	0.036	0.176	0.096	0.189
ζ-Potential / mV	-16 ± 3	-42 ± 4	-37 ± 4	-16 ± 2	-39 ± 4	-25 ± 3

Fig. 6 Physicochemical characterization of liposomes before and after functionalization with DBCO/antibodies. (A) Size of liposomes before/after DBCO and antibody attachment using 1 : 1 and 3 : 1 DBCO/NH<sub>2</sub> ratios. (B) Surface charge before and after liposome functionalization. (C) Overview of the physicochemical properties of the liposomes for the different reaction conditions.



tion of a longer PEG linker, so that *e.g.* the adsorption of proteins would not cover the targeting structures.

## 4. Conclusion

In this study, we have shown the synthesis of amine group-terminated liposomes *via* lipid dry film rehydration followed by extrusion. In contrast to common vesicle functionalization approaches, which rely on pre-modification of lipids or block-copolymers, we introduced further functional groups and performed antibody coupling *via* bio-orthogonal copper-free click chemistry after liposome self-assembly. For precise reaction control and tunability, a fluorescamine assay (FA assay) and an anthracen-azide (Anth-N<sub>3</sub>) assay were utilized to determine the number of present NH<sub>2</sub> groups as well as DBCO groups after functionalization. The antibody, which was introduced as a possible targeting structure, was modified site specifically at the Fc fragment and coupled to the liposomes using the strain-promoted alkyne-azide click (SPAAC) reaction. This surface functionalization procedure offers the advantage to be transferred to all self-assembled systems bearing primary surface amino groups without damaging the assembled structure because no large amounts of organic solvent are required and the carriers can be kept in physiological conditions. Amino groups can easily be introduced as end groups of block-copolymers and are already present on natural structures such as extracellular vesicles (in the form of membrane proteins). Additionally, this type of antibody coupling is applicable for all antibody types without the disadvantage of potentially covering or modifying the antigen binding sites. If needed, the introduced PEG spacer can also be varied in length or type of building block to provide additional characteristics like enhanced stealth properties or better antibody accessibility. In summary, this versatile vesicle surface functionalization strategy is applicable for a wide range of drug delivery systems as well as straightforward and robust in terms of handling.

## Conflicts of interest

The authors declare no conflict of interest.

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