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The importance of tyrosines in multimers of cyclic RGD nonapeptides: towards $\alpha\text{v}\beta 6$ -integrin targeted radiotherapeutics†

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In a recent paper in this journal (*RSC Med. Chem.*, 2023, 14, 2429), we described an unusually strong impact of regiospecific exchange of phenylalanines by tyrosines in 10 gallium-68-labeled trimers of certain cyclic RGD peptides, c[XRGDLAXp(NMe)K] (X = F or Y), on non-specific organ uptakes. We found that there was, in part, no correlation of liver uptake with established polarity proxies, such as the octanol-water distribution coefficient ($\log D$). Since this observation could not be explained straightforwardly, we suggested that the symmetry of the compounds had resulted in a synergistic interaction of certain components of the macromolecules. In the present work, we investigated whether a comparable effect also occurred for a series of 5 tetramers labeled with lutetium-177. We found that in contrast to the trimers, liver uptake of the tetramers was well correlated to their polarity, indicating that the unusual observations along the trimer series indeed was a unique feature, probably related to their particular symmetry. Since the Lu-177 labeled tetramers are also potential agents for treatment of a variety of $\alpha\text{v}\beta 6$ -integrin expressing cancers, these were evaluated in mice bearing human lung adenocarcinoma xenografts. Due to their tumor-specific uptake and retention in biodistribution and SPECT imaging experiments, these compounds are considered a step forward on the way to $\alpha\text{v}\beta 6$ -integrin-targeted anticancer agents. Furthermore, we noticed that the presence of tyrosines in general had a positive impact on the *in vivo* performance of our peptide multimers. In view of the fact that a corresponding rule was already proposed in the context of protein engineering, we argue in favor of considering peptide multimers as a special class of small or medium-sized proteins. In summary, we contend that the performance of peptide multimers is less determined by the *in vitro* characteristics (particularly, affinity and selectivity) of monomers, but rather by the peptides' suitability for the overall macromolecular design concept, and peptides containing tyrosines are preferred.

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Introduction

Peptide multimers have gained a firm place in the canon of compound classes used in radiopharmacy.¹ This is because the combination of multiple copies of the same binding motif for a biological target in a single molecule increases

avidity for that target—a principle that nature has always exploited in the form of regular antibodies, which are nothing else than dimers of targeting proteins.² A particularly important component of a synthetic multimer is the scaffold, because its valency, geometry, and symmetry determine multiplicity, distance, and interaction between the conjugated functional monomers.^{3,4} The spatial distribution and motility of the comprised monomeric ligands can furthermore be adjusted using linkers of different lengths and rigidity to optimize avidity,^{5,6} the latter being considered a cumulative effect of individual target binding events.⁷ This strategy has been extensively exploited in connection with perhaps the most widely used class of ligands for $\alpha\text{v}\beta 3$ -integrin, the cyclic pentapeptides c(RGDxX) (x = f, y; X = K, E).⁸ Clinical data have suggested that radiolabeled c(RGDxK) dimers⁹ outperform the respective monomers for $\alpha\text{v}\beta 3$ -integrin targeted positron emission tomography (PET)

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diagnostics in patients.¹⁰ Comparative studies revealed that multimers with more than two c(RGDxK) units possess even higher avidities than monomers and dimers,^{11–13} which frequently also resulted in higher target-specific uptakes *in vivo*.^{14–16} This design principle has been applied to various targeting moieties with high relevance in radiopharmacy, such as fibroblast activation protein inhibitors (FAPi)¹⁷ or prostate-specific membrane antigen (PSMA) inhibitors,^{18–20} although it did not always lead to improved *in vivo* properties.²¹ However, the concept generally works well for radiopharmaceuticals based on peptides and peptidomimetics targeting integrins,²² such as $\alpha\text{v}\beta 3$,^{14–16} $\alpha 5\beta 1$,^{23,24} $\alpha\text{v}\beta 8$,^{25,26} or $\alpha\text{v}\beta 6$.^{27,28} The approach certainly is a valuable addition to the radiopharmaceutical toolbox,¹ as it has been shown that even peptides whose affinities are too low for the radiolabeled monomers to be used as radiotracers can be useful for effective *in vivo* addressing when their target-specific binding is boosted by multimerization.^{26,28,29}

However, multimerization does not only enhance avidity to the desired target but may also amplify unknown weak interactions.³⁰ Peptide multimers therefore can exhibit altered target specificity or unexpectedly bind to sites not recognized by the respective monomers, which makes the successful design of multimeric probes and therapeutics a challenging endeavour. In addition, they differ from oligopeptides in that they have several times the molecular weight, which puts them on the borderline of proteins. Together with their sometimes altered polarity, they can therefore exhibit significant differences in pharmacokinetic parameters such as tissue penetration, elimination pathways and kinetics, and plasma half-life.³¹ Boehmer *et al.* have furthermore pointed out that multimers are characterized by a higher complexity that needs to be better understood to enable a rational design of clinically applicable agents.³² The increased complexity of multimers apparently leads to a higher degree of indeterminacy in the structure–activity relationships, because the components of complex systems can interact in unpredictable ways and thus produce completely new properties, which is known as emergence.³³ Concerning multimers, this means that such emergent properties cannot in principle be calculated or otherwise derived straightforwardly from data for the respective monomers. Comprehensive data collection for libraries of multimers seems to be the only way to close this gap in understanding.³⁴

One aspect of multimerization that is particularly difficult to predict is nonspecific accumulation in non-target organs and tissues, which can emerge unexpectedly and render an agent with otherwise promising properties, such as high avidity and target-specific uptake, virtually useless.³⁵ Although most binding motifs are well characterized in terms of their specificity for their respective pharmacological target, usually only a small fraction of the multitude of possible interactions with other cell types, surface proteins, transporters, enzymes, or other biological contact sites in a



Fig. 1 Structures of compounds discussed in the text.

living organism is known, which of course applies all the more the larger the respective molecule and thus the greater the number of possible contact sites.

To this end, we recently investigated the influence of small structural variations on the biodistribution of the $\alpha\text{v}\beta 6$ -integrin targeted positron emission tomography (PET) radiopharmaceutical ⁶⁸Ga-Trivehexin (Fig. 1).³⁶ This molecule comprises three copies of the cyclic nonapeptide c[RGDLAYp(NMe)K], which are attached to a central triazacyclononane-triphosphinate (TRAP, see Fig. 1)³⁷ structure that serves as both a highly efficient chelator for ⁶⁸Ga³⁸ and a C₃-symmetrical trimeric scaffold.³⁹ ⁶⁸Ga-Trivehexin thus comprises 6 tyrosines, which we regioselectively and consecutively substituted by phenylalanines.³⁴ The influence of these substitutions on the biodistribution proved to be counterintuitive to some extent. For example, although a high non-specific liver uptake of the all-Phe derivative was reduced by 50% upon introduction of a single Tyr, and by almost three quarters by two Tyr's, there was no concordant change in polarity proxies, such as log*D*. We concluded that non-specific tissue uptake is most likely an emergent phenomenon arising from synergistic interaction of phenyl groups in these highly symmetric molecules.³⁴ One question that arose in the course of this study was whether the observed correlations are primarily caused by the highly ordered triangular symmetry of



the compounds, or whether comparable effects can also be reproduced within a similar set of structures that exhibit a different symmetry.

In order to shed more light on this question, the present study takes a closer look at comparable structures based on the DOTPI framework (Fig. 1).⁴⁰ This tetrameric scaffold is a larger congener of TRAP and can be functionalized in a similar manner, *i.e.*, by virtue of the copper(i)-catalyzed alkyne–azide cycloaddition (CuAAC) using the respective azide-functionalized derivative,^{41,42} which provides access to symmetrically and asymmetrically substituted peptide tetramers in practical amounts.⁴³ DOTPI rapidly forms kinetically inert complexes with many radiometal ions that emit particle radiation, such as the alpha emitter ²¹³Bi^{III},⁴⁴ or trivalent radiolanthanides and especially the popular beta emitter ¹⁷⁷Lu^{III}.⁴⁰ We herein report synthesis and preclinical data (*ex vivo* biodistribution and SPECT imaging in tumor-bearing mice) for tetrameric conjugates of $\alpha\beta 6$ -integrin binding cyclic nonapeptides with the sequence cyclo[XRGDLAXp(NMe)K] (X = F or Y), which also constitutes the next step towards $\alpha\beta 6$ -integrin targeted radiotherapeutics on the basis of these peptides. Such agents are currently considered of high clinical relevance^{45,46} because $\alpha\beta 6$ -integrin is upregulated in various malignant cancers,⁴⁷ especially in pancreatic ductal adenocarcinoma (PDAC),⁴⁸ oral squamous cell carcinoma (OSCC),⁴⁹ ovarian⁵⁰ and cervical cancer,⁵¹ as well as in primaries and brain metastases⁵² of non-small cell lung cancer (NSCLC).⁵³

Results

Peptide conjugation *via* CuAAC

Monomeric peptide building blocks **FF**,²⁹ **FY**,³⁴ and **YY**³⁶ (Fig. 2) were synthesized as described previously by coupling 4-pentynoic acid to the Lys side chains of the three cyclopeptides cyclo[FR(Pbf)GD(*t*Bu)LAFp(NMe)K], cyclo[FR(Pbf)GD(*t*Bu)LAY(*t*Bu)p(NMe)K], or cyclo[Y(*t*Bu)R(Pbf)GD(*t*Bu)LAY(*t*Bu)p(NMe)K], followed by acidic deprotection. With regard to our preceding study on comparable trimeric peptides,³⁴ we found that the increase in possible Phe → Tyr exchange positions and the change in symmetry leads to an even greater number of possible isomers, rendering a full-scale *in vivo* evaluation unfeasible. The imperative to reduce animal experiments to a reasonable minimum forced us to limit this investigation to the most relevant examples. We therefore refrained from including the fourth possible monomer **YF** (in which R^A is OH and R^B is H, see Fig. 2) into this investigation because in the preceding study,³⁴ we observed that replacement of **FF** by **YF** invariantly had the same influence on biodistribution as **FY**, albeit to a slightly lower extent. In addition, **FY** showed higher affinity and selectivity for $\alpha\beta 6$ -integrin than **YF**,³⁴ which is why **FY** seemed to be generally more relevant for this type of study.

The synthesis of the homo-tetramers **Y0**, **Y4**, and **Y8** was done by CuAAC reactions as shown in Fig. 2, using **FF**, **FY**, and **YY**, respectively, as building blocks. **Y3** was synthesized in the same way, applying a combinatorial reaction scheme

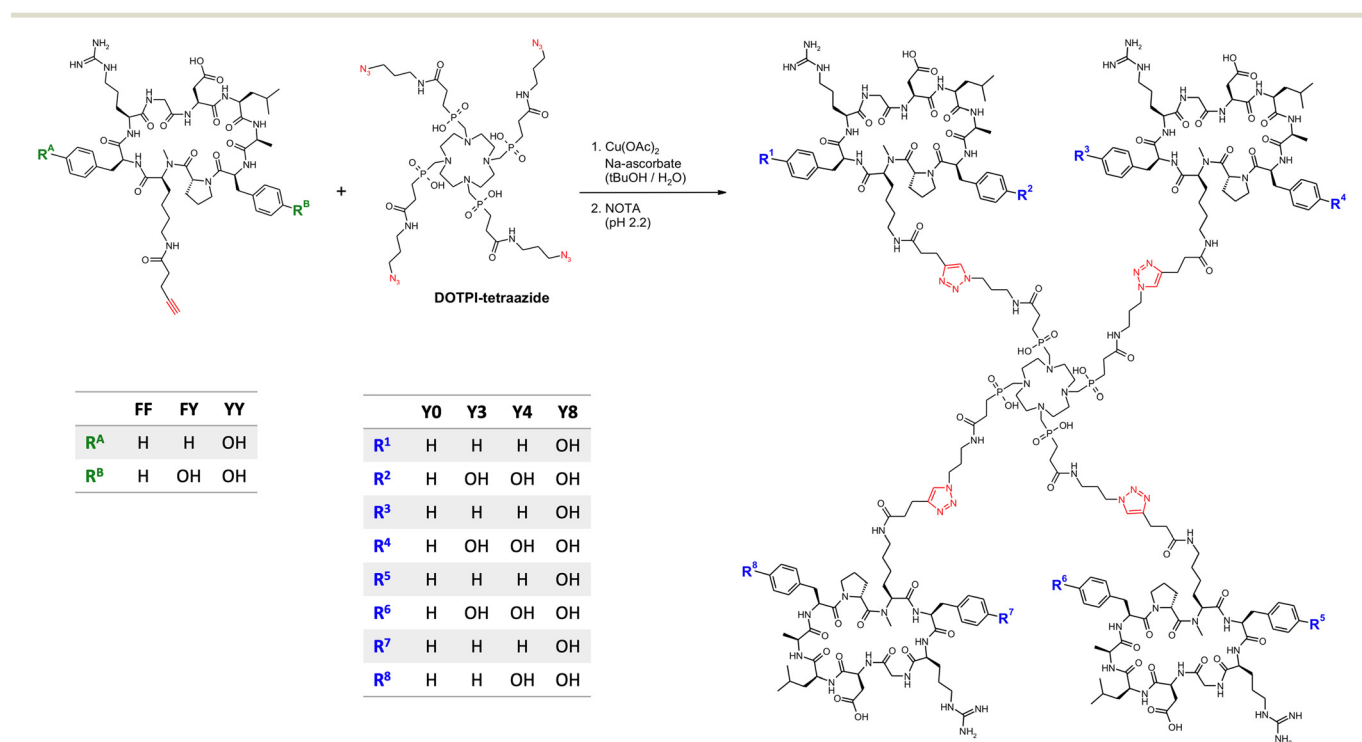


Fig. 2 Synthesis of tetrameric conjugates of $\alpha\beta 6$ -integrin binding peptides. Synthesis of **Y0**, **Y4**, and **Y8** was done by reacting DOTPI-tetraazide with the peptides **FF**, **FY**, and **YY**, respectively. The asymmetrical conjugate **Y3** was isolated from the CuAAC reaction of DOTPI-tetraazide with a mixture of **FF** and **FY** by means of preparative HPLC. Functional groups related to the CuAAC are highlighted in red.





Fig. 4 Biodistribution data for ^{177}Lu -labeled peptide tetramers in H2009 (human lung adenocarcinoma) xenografted mice. For details and numerical data in tabular form, see ESI.†

compound performed best in terms of tumor-to-pancreas and particularly tumor-to-intestines ratios. Since unwanted uptake in the intestines has proven the most critical issue



Fig. 5 Biodistribution in H2009 tumor mice, and resulting tumor-to-organ ratios for the ^{177}Lu -labeled tetramer Y8 at different time points after injection. For details and numerical data in tabular form, see ESI.†

hampering translation of various $\alpha\text{v}\beta 6$ -integrin targeted radiopharmaceuticals,⁵⁵ we considered ^{177}Lu -Y8 to be the most promising compound in the series, despite the overall ambiguous data situation.

We determined the biodistribution for ^{177}Lu -Y8 for another time point, 24 h p.i., which revealed that the compound rapidly cleared from the blood stream but was retained in many organs (Fig. 5). We noted that uptake and retention was somewhat higher for the $\alpha\text{v}\beta 6$ -integrin positive H2009 tumor and stomach and furthermore observed a similar clearance pattern, which substantiates the interpretation that the prolonged tissue accumulation was target-specific. The long-term tumor retention of ^{177}Lu -Y8 was further illustrated by single-photon emission computed tomography (SPECT) imaging (Fig. 6). The images showed an uneven tumor uptake, which corresponds well to the generally heterogeneous $\alpha\text{v}\beta 6$ -integrin expression in this model.^{28,29} Volume-of-interest (VOI) based quantification of tumor uptakes did not quite correspond to biodistribution, but nevertheless corroborated a tumor retention of ^{177}Lu -Y8 over days. However, the images also showed a high uptake in the kidneys, which did not decrease over time, consistent with the biodistribution data shown in Fig. 5a.





Fig. 6 Single-photon emission computed tomography (SPECT) of a SCID mouse bearing a human H2009 tumor xenograft on the right shoulder, recorded at 4 time points (1 h, 3 h, 24 h, and 72 h) after i.v. injection of 36.5 MBq of ^{177}Lu -Y8, scan time 1 h. Tumor position is indicated by white dotted circle. Adjacent figures denote ROI-based tumor uptakes in $\% \text{ IA mL}^{-1}$.

Discussion

Towards $\alpha\text{v}\beta 6$ -integrin targeted radiotherapeutics

In a series of 10 different ^{68}Ga -labeled peptide trimers featuring different combinations of $\alpha\text{v}\beta 6$ -integrin targeting cyclic nonapeptides monomers of the generic formula cyclo[XRGDLAXp(NMe)K] (X = Phe or Tyr), the trimer comprising only tyrosines showed the lowest degree of nonspecific organ accumulation in preclinical models.³⁴ This radiotracer, referred to as ^{68}Ga -Trivehexin (see Fig. 1),³⁶ has convincingly demonstrated its utility for PET imaging of cancers and metastases,⁴⁵ particularly of pancreatic carcinoma.⁵⁶ Its low intestinal uptake has proven beneficial in a clinical setting, because it resulted in a better delineation of metastases than radiotracers based on other $\alpha\text{v}\beta 6$ -integrin binding peptides.^{46,55} The next step towards improved clinical treatment of cancer patients was, of course, the development of corresponding therapeutic $\alpha\text{v}\beta 6$ -integrin radioligands to enable a theranostic approach.^{46,57}

We previously evaluated the potential of various ^{177}Lu labeled 1:1 chelator-peptide conjugates of cyclo[FRGDLAFp(NMe)K] but found that such monomers are generally not suitable for radioligand therapy due to low uptake and retention in tumors ($<0.8\% \text{ IA g}^{-1}$, 90 min p.i.).⁵⁸ The novel peptide tetramers ^{177}Lu -Y0, ^{177}Lu -Y3, ^{177}Lu -Y4, and ^{177}Lu -Y8 showed a markedly improved tumor uptake (range 3.2 to 6.9% IA g^{-1} , 90 min p.i.) and a retention over days (range 1.7 to 2.1% IA g^{-1} , 3 d p.i.; see Fig. 4, Fig. 6, and ESI† Tables S1–S11). However, one cannot fail to notice a high and persistent kidney uptake of these agents, which is expected to limit their clinical applicability because of a high risk of radiation-induced renal damage.⁵⁹ This problem has been observed for many radiolabeled peptides and proteins, and is most likely caused by a long residence time of the radiometabolites generated by lysosomal proteolysis of the radiopeptides in renal tubular cells.⁶⁰

Different approaches have been pursued to solve this issue. One strategy aims at selective breakdown of radiopharmaceuticals that have accumulated in the kidneys. This can be achieved by introducing peptidic linkers that are stable in blood plasma but are cleaved by enzymes present on the brush border membrane of renal tubules, to selectively generate small radiometabolites in the kidneys that are rapidly excreted *via* the urine.⁶¹ One such linker is the amino acid sequence GFK that is recognized by neutral endopeptidase (NEP) located in the renal brush border.⁵⁴ We consequently equipped one of the tetramers, Y4, with this sequence in all peptide-chelator linkages, resulting in the conjugate GFK-Y4 (Fig. 3). Unfortunately, the approach did not result in the expected substantial improvement of the tumor-to-kidney ratio. The renal retention of ^{177}Lu -GFK-Y4 was still high after 3 d, and the compound furthermore showed lower tumor uptake and faster elimination from the blood stream than the other tetramers (Fig. 4). Such an *in vivo* behavior suggests that the GFK sequence is actually cleaved before reaching the brush border, leading to premature decomposition and excretion of a large fraction of the radiopharmaceutical before it can bind to its target on the tumor cells. The desired reduction in renal uptake while maintaining tumor uptake could therefore not be achieved in this explorative experiment. Further linker variations or other established methods to reduce the radiation dose of the kidneys, such as the simultaneous infusion of renal protection agents,⁶² shall therefore be evaluated in the future.

Structure, polarity, and biodistribution revisited

The other aim of this study was to figure out whether multimers of cyclo[XRGDLAXp(NMe)K] (X = F or Y) that are based on another scaffold than TRAP show the same surprising trends of polarity measures, especially $\log D$,





Fig. 7 Comparison of trends in polarity proxies and liver uptakes, 3 d p.i. Connecting lines between data points are intended only to visualize trends and do not indicate a functional correlation.

and non-specific organ uptakes, especially in the liver. We recall that for the trimer series, replacement of less than 50% of phenylalanines by tyrosines did not change the $\log D$ but reduced liver uptake by up to 72%.³⁴ Such an observation was not made for the tetramers of this work (Fig. 4). We found a consistent decrease of both parameters with gradual Phe $\rightarrow$ Tyr exchange. The first 3 Phe $\rightarrow$ Tyr substitutions (Y0 $\rightarrow$ Y3) still had a larger influence on both parameters than the last 4 exchanges (Y4 $\rightarrow$ Y8), but the overall picture is well in line with the textbook scheme, *i.e.*, a higher degree of polarity reduces liver absorption (Fig. 7).

Fig. 8 further illustrates that the tetramer series did not show the anomaly that was previously observed for the corresponding library of TRAP-based trimers, namely, that polarity proxies such as $\log D$ had no predictive value for certain *in vivo* properties, such as liver uptake.³⁴ For the tetramers, an increase of the $\log D$ always corresponded to an increase of liver uptake (Fig. 8a). It thus appears possible to derive a bijective correlation function, which could be used to approximate the liver uptake of a novel compound in that series based on its $\log D$ (exemplified by the dotted line in Fig. 8a). In lack of a suitable theory, we are however not able to define which type of regression could be appropriate to obtain a fit function for a correlation between the logarithm of the octanol-PBS distribution coefficient and the liver uptake expressed as % IA g⁻¹. In contrast, a fit was not meaningful at all for the trimers,³⁴ which is particularly evident from the fact that in the corresponding plot (see Fig. 8b), five out of ten data points for the trimer series are more or less vertically aligned. A hypothetical correlation function would therefore have to comprise non-bijective sections (indicated by a vertical dotted line in Fig. 8b), which means it does not exist.

Taken together, we found that in contrast to $\alpha\beta\beta 6$ -integrin binding cyclic RGD peptide trimers,³⁴ the tetramers behaved “normally”, *i.e.*, the relationship between polarity and liver absorption adhered to the textbook pattern. Their conformational structure apparently does not promote a synergistic interaction between certain structural elements or the peptide monomers that would lead to non-linear effects in this regard. Therefore, it seems plausible to assume that

for the DOTPI conjugates, *in vitro* polarity proxies such as $\log D$ may be more useful to guide early-stage drug development than in case of TRAP trimers.

The importance of tyrosines in peptide multimers

Regardless of whether or not a synergistic interaction of phenylalanines caused a disproportionately high non-specific



Fig. 8 Plots of liver uptakes against octanol-water distribution coefficients ($\log D$) for a) peptide tetramers (this work) and b) Trivehexin-related trimers reported earlier (*RSC Med. Chem.* 2023, 14, 2429).³⁴ Large-size circles in b highlight the conjugates of the trimer series (Y0, Y2F, Y3F, and Y6, see ref. 34, Fig. 1) which, in terms of percentage and distribution of Phe/Tyr units, are most equivalent to the four tetramers in this work. Color codes of circles correspond to the ones used for the respective compound series; for trimers see ref. 34, Fig. 2a; for the tetramers, see this work, Fig. 4. Note that similar colors in a and b do not necessarily indicate a structural match.



organ uptake, one fundamental observation made for both groups of cyclopeptide multimers studied (^{68}Ga -TRAP trimers or ^{177}Lu -DOTPI tetramers) was the same: more tyrosines improved the *in vivo* performance. In particular, undesired non-target uptake was reduced, while target-specific binding *in vivo* (i.e. tumor uptake) of the conjugates remained more or less unchanged or was even improved. Similar findings have already been forged into a golden rule for the development of synthetic binding proteins. Koide and Sidhu pointed out in 2009 that non-natural engineered proteins showed better binding properties if they consisted predominantly of Tyr, Ser, and Gly.⁶³ These authors described the unique role of Tyr as follows: “The physicochemical properties of tyrosine make it the amino acid that is most effective for mediating molecular recognition, and protein engineers have taken advantage of these characteristics to build tyrosine-rich protein binding sites that outperform natural proteins in terms of affinity and specificity.” However, the same concept does not seem to play a major role for peptides, as there is apparently no systematic tendency to favor tyrosine-containing sequences over comparable peptides without Tyr. To stay with integrin ligands, even a glance into the wealth of RGD peptide literature^{8,57} quickly reveals that there is apparently no clear preference for either of the two most popular $\alpha\beta3$ -integrin binding motifs, c(RGDyK) and c(RGDfK), although the former contains a D-Tyr and the latter does not.

With regard to the present work, it should be mentioned that during the development of the nonapeptides used herein, the sequence cyclo[FRGDIAFp(NMe)K] showed a slightly better $\alpha\beta6$ -integrin affinity and subtype selectivity than its two siblings with one Phe replaced by a Tyr.⁶⁴ Therefore, it was initially chosen as a lead structure for radiopharmaceutical development.^{29,58} However, as long as such peptide monomers are not directly used as drugs (either non-functionalized or equipped with simple tags) but as components of larger constructs, it seems reasonable not to rely exclusively on the “best” peptide, but also to consider their “second best” congeners as long as they contain more tyrosines. Our data convincingly demonstrated that it can be advantageous to follow this rationale in the development of radiopharmaceuticals, as our multimers comprising only cyclo[FRGDIAFp(NMe)K] were outperformed by those with tyrosines, although all corresponding Tyr-containing monomers exhibited less favorable *in vitro* affinities and selectivities.³⁴

Conclusion

The overarching message of our two multimerization studies (RSC Med. Chem., 2023, 14, 2429,³⁴ and this work) is that the development of peptide multimers should preferably be guided by the concepts and rules known from protein engineering. This is because peptide multimers are not simply somewhat larger molecules whose properties are the sum of the characteristics of the monomeric peptide

components of which they are composed. Rather, they should be perceived as a special class of small or medium-sized proteins, which have their own special design rules that need to be systematically explored. We conclude that the performance of multimers is not primarily determined by the *in vitro* properties (especially affinity and selectivity) of the peptide monomers, but rather on the peptides' suitability for the overall macromolecular design concept. One lesson we have learned is that tyrosines in the peptide sequence can contribute a great deal to this suitability. This particular aspect has proven to be crucial for the development of $\alpha\beta6$ integrin-targeted radiopharmaceuticals.

Experimental Section

Materials & methods

Analytical HPLC was performed on a Shimadzu (Kyoto, Japan) Prominence HPLC system equipped with either a Dr. Maisch ReproSil Pur C18 AQ 150 $\times$ 4.6 mm, 5 μm , 120 $\text{\AA}$ (AC1) or a Phenomenex Kinetex C18 150 $\times$ 4.6 mm 5 μm column (AC2). Semipreparative HPLC was performed on a Shimadzu HPLC system equipped with either a Dr. Maisch ReproSil Pur C18 AQ 250 $\times$ 10 mm, 5 μm , 120 $\text{\AA}$ column with 5 mL min^{-1} flow (PC1) or a Dr. Maisch ReproSil Pur C18 150 $\times$ 30 mm, 5 μm , 120 $\text{\AA}$ column with 20 mL min^{-1} flow (PC2). Solvents for both systems were water (A) and acetonitrile (B) each supplemented with 0.1% (vol) trifluoroacetic acid (TFA). ESI-MS spectra (positive mode) were recorded on a Shimadzu LC-MS 2020 single quadrupole mass spectrometer. Solvents for MS applications were water (A) and acetonitrile (B) each supplemented with 0.1% (vol) formic acid. The other experimental procedures (determination of $\log D_{7.4}$, cultivation of H2009 human lung adenocarcinoma cells, generation of tumor xenografts in CB17 SCID mice, and biodistribution) were performed as described before.²⁸ All animal experiments have been carried out according to applicable law and institutional guidelines of Technical University of Munich, and were approved by the responsible local authority (Regierung von Oberbayern). The H2009 cells were regularly authenticated and tested for mycoplasma contamination. DOTPI-tetraazide,⁴³ FF (previously termed AvB6),²⁹ FY,³⁴ and YY (also referred to as Tyr2-alkyne)³⁶ were synthesized as described previously.

Syntheses

The synthesis of the Y4 analog GFK-linker functionalized peptide building block GFK-Y4 is outlined in Fig. 9. Corresponding chromatograms and spectra are given in the ESI.†

FY-Suc-TFP. The protected cyclic peptide cyclo[FR(Pbf)GD(tBu)LAY(tBu)p(NMe)K]³⁴ was reacted with succinic acid anhydride according to a published procedure.⁶⁵ 143 mg (111 μmol , 1.0 eq.) of peptide were dissolved in a small amount of dry dichloromethane (DCM) and 12.2 mg (122 μmol , 1.1 eq.) of succinic anhydride were added. The mixture was cooled to 0 $^{\circ}\text{C}$ and triethylamine (16.9 μL , 122 μmol , 1.1 eq.) was added slowly. The ice bath was removed and the reaction was





Fig. 9 Synthesis of the building block **GFK-FY**, a derivative of **FY** featuring the brush border cleavable linker **GFK**.

allowed to run over night with stirring. For workup DCM was removed *in vacuo* and the crude residue was dissolved in H₂O/MeCN, followed by purification *via* semipreparative HPLC (20–60% B in 15 min, PC2). After lyophilization, **FY-Suc** was obtained as a colorless solid (89%, 150 mg, 100 μmol). This intermediate was dissolved in dimethyl formamide (DMF) and 33 mg (33 μmol, 2.0 eq.) of 2,3,5,6-tetrafluorophenol and 21 mg (100 μmol, 2.0 eq.) of dicyclohexyl carbodiimide (DCC) were added to the solution. The product was purified *via* semi-preparative HPLC (20–60% B in 15 min, PC2) and lyophilized, affording a colorless solid (90%, 150 mg, 90 μmol). *m/z*: 1675.7 [$M + H^+$]⁺, 838.4 [$M + 2H^+$]²⁺. **FY-Suc-TFP** RP-HPLC (10–90% B in 15 min, AC2): 11.5 min.

GFK-alkyne (brush-border cleavable linker). The GFK-motif was assembled on a CTC-resin. First, Fmoc-L-Lys(Dde)-OH (1.3 g, 2.1 mmol, 1.0 eq.) was loaded on to the resin using 1.1 mL (3.0 eq.) of DIPEA in 20 mL of DMF. After subsequent Fmoc-deprotection, Fmoc-L-Phe-OH (1.6 g, 4.1 mmol, 2.0 eq.) was coupled to the free amine using 1.6 g (4.1 mmol, 2.0 eq.) of HATU, 0.6 g (4.1 mmol, 2.0 eq.) of 1-hydroxybenzotriazol (HOBt) and 1.9 mL (10.3 mmol, 5.0 eq.) of diisopropyl ethylamine (DIPEA). Next, Fmoc-Gly-OH (1.2 mg, 4.1 mmol, 2.0 eq.) was reacted with the unprotected *N*-terminus using the same conditions as mentioned above. Lastly, the Fmoc-deprotected tripeptide was functionalized with 4-pentynoic acid (0.4 g, 4.1 mmol, 2.0 eq.), followed by Dde-deprotection with 5% hydrazine in DMF. GFK-alkyne was cleaved from the resin using a cocktail of trifluoroacetic acid (TFA)/triisopropyl silane (TIPS)/H₂O (95:2.5:2.5) (3 × 4 mL).

After removal of all volatiles *in vacuo*, the residue was purified *via* semipreparative HPLC (10–40% B in 15 min, PC2) and subsequently lyophilized (50%, 430 mg, 1 mmol). *m/z*: 431.3 [$M + H^+$]⁺, 453.3 [$M + Na^+$]⁺. **GFK-alkyne** RP-HPLC (10–90% B in 15 min, AC2): 6.0 min.

GFK-FY (Fig. 9). **GFK-alkyne** (15.0 mg, 33 μmol, 1.0 eq.) was reacted with **FY-Suc-TFP** (84 mg, 50.2 μmol, 1.5 eq.) in DMF (30 mL). The pH of the solution was adjusted to 8–9 with DIPEA (70 μL, 8.0 eq.) and the mixture was stirred at room temperature for 16 h. All volatiles were removed *in vacuo* and the crude residue was treated with a cocktail of TFA/TIPS/H₂O (95:2.5:2.5) (3 mL) for final deprotection. TFA was removed under a stream of air and the peptide was purified *via* semipreparative HPLC (20–60% B in 15-min, PC2). After lyophilization GFK-FY was obtained as a colorless solid (85%, 44.1 mg, 28 μmol). *m/z*: 1576.1 [$M + H^+$]⁺, 788.5 [$M + 2H^+$]²⁺. **GFK-FY** RP-HPLC (10–90% B in 15 min, AC2): 6.2 min.

General procedure for synthesis of DOTPI-tetramers. The synthesis was modified from the literature.⁴³ 4.4 eq. of the respective peptide building block (a mixture was used for synthesis of **Y3**) was added to a solution of DOTPI-tetraazide (1.0 eq.) and copper(II) acetate hydrate (2.0 eq.) in a minimum amount of H₂O or a mixture of H₂O and *t*BuOH to aid dissolution of reagents. Sodium ascorbate (50.0 eq.) dissolved in a minimum volume of H₂O was quickly added to the reaction, the mixture vortexed for 1 min, and placed in a water bath at 60 °C. Upon addition of sodium ascorbate to the reaction mixture, a blue precipitate formed slowly which dissolved after 2–5 min, resulting in a transparent dark blue



solution. The reaction was allowed to progress for 1 h, followed by monitoring reaction progress by HPLC-MS. Cu species were sequestered from the tetramer and the reaction solution by addition of NOTA (30.0 eq.) dissolved in H₂O with adjustment of the pH to 2.2 using concentrated 12 M HCl (typically 5–10 μ L). The Cu transchelation reaction was allowed to progress for 1 h at 60 °C. The metal-free tetramers were then directly purified by RP-HPLC, followed by lyophilization. Corresponding chromatograms and MS spectra are given in the ESI†

Y0. This compound was first synthesized according to the general procedure with 15.1% overall yield. We later tested alternative reaction conditions and found that an optimization of the yield is feasible: **FF***TFA (8.18 mg, 6.6 μ mol) and DOTPI-tetraazide*2TFA (2.0 mg, 1.5 μ mol) in 500 μ L H₂O/*t*BuOH (1 : 1). Copper(II) acetate hydrate (1.49 mg, 7.5 μ mol), sodium ascorbate (17.8 mg, 90 μ mol) and tris(3-hydroxypropyltriazolylmethyl) amine (THPTA, 0.33 mg, 0.75 μ mol) in 500 μ L H₂O/*t*BuOH (1 : 1). Catalyst solution was added to azide/alkyne solution and the mixture was stirred for 10 min at room temperature. Cu species were demetallated as described above with NOTA (6.82 mg, 22.5 μ mol) and bispidol (2.68 mg, 4.5 μ mol) in 1 mL H₂O/ACN pH 2.4 for 1 h at 60 °C. **Y0** was obtained as the TFA salt in form of a colorless solid (46.7%, 4.0 mg, 0.70 μ mol). *m/z*: 1403.1 [M + 4H]⁴⁺, 1122.5 [M + 5H]⁵⁺, 935.3 [M + 6H]⁶⁺. ¹⁷⁷Lu-**Y0** RP-HPLC (0–99% B in 15 min, AC1): *t*_R = 9.4 min.

Y3. Reagents: **FF** (1.2 eq., 5.4 μ mol) **FY** (3.6 eq., 16.2 μ mol) DOTPI tetraazide (5.0 mg, 4.5 μ mol, 1.0 eq.) in 350 μ L of H₂O/*t*BuOH (1 : 2), Cu^{II} acetate hydrate (2.0 mg, 10 μ mol, 2.2 eq.), Na ascorbate (36.1 mg, 182 μ mol, 40.0 eq.). A Cu^{II} complex of **Y3** was isolated from the CuAAC reaction mixture by preparative HPLC (32–48% B in 20 min, PC1). Afterwards, Cu species were demetallated as described above, followed by final purification by HPLC. (35% B isocratic, PC1). After lyophilization **Y3** was obtained as the TFA salt in form of a colorless solid (1.7%, 450 μ g, 78 nmol). *m/z*: 1886.0 [M + 3H]³⁺, 1414.8 [M + 4H]⁴⁺, 1132.1 [M + 5H]⁵⁺, 943.6 [M + 6H]⁶⁺. ¹⁷⁷Lu-**Y3** RP-HPLC (0–99% B in 15 min, AC1): *t*_R = 8.6 min.

Y4. Reagents: **FY** (10.49 mg, 9.18 μ mol), DOTPI-tetraazide (2.3 mg, 2.09 μ mol), copper(II) acetate hydrate (833 μ g, 4.18 μ mol) and sodium ascorbate (20.67 mg, 103.32 μ mol) in 410 μ L H₂O. Cu species were demetallated as described above with NOTA (16.71 mg, 55.09 μ mol) in 1 mL H₂O. **Y4** was obtained as the TFA salt in form of a colorless solid (13%, 1.35 mg, 0.23 μ mol). *m/z*: 1891.4 [M + 3H]³⁺, 1418.8 [M + 4H]⁴⁺, 1135.3 [M + 5H]⁵⁺, 946.3 [M + 6H]⁶⁺. ¹⁷⁷Lu-**Y4** RP-HPLC (0–99% B in 15 min, AC1): *t*_R = 8 min.

Y8. Reagents: (9.43 mg, 8.29 μ mol), DOTPI-tetraazide (2.07 mg, 1.88 μ mol), copper(II) acetate hydrate (752 μ g, 3.77 μ mol) and sodium ascorbate (18.66 mg, 94.18 μ mol) in 345 μ L H₂O. Cu species were demetallated as described above with NOTA (16.71 mg, 55.09 μ mol) in 1 mL H₂O. **Y8** was obtained as the TFA salt in form of a colorless solid (34.5%, 3.71 mg, 0.6 μ mol). *m/z*: 1434.56 [M + 4H]⁴⁺, 1147.75 [M + 5H]⁵⁺, 956.77 [M + 6H]⁶⁺. ¹⁷⁷Lu-**Y8** RP-HPLC (0–99% B in 15 min, AC1): *t*_R = 8.4 min.

GFK-Y4. Reagents: **GFK-FY** (6.26 mg, 3.98 μ mol), DOTPI-tetraazide (972 μ g, 0.88 μ mol), copper(II) acetate hydrate (752 μ g, 3.77 μ mol) and sodium ascorbate (18.66 mg, 94.18 μ mol) in 540 μ L H₂O/*t*BuOH (1 : 1.7). Cu species were demetallated as described above with NOTA (7.02 mg, 23.15 μ mol) in 0.5 mL H₂O. **GFK-Y4** was obtained as the TFA salt in form of a colorless solid (22.4%, 1.28 mg, 0.17 μ mol). *m/z*: 1480.52 [M + 5H]⁵⁺, 1234.00 [M + 6H]⁶⁺, 1057.71 [M + 7H]⁷⁺. ¹⁷⁷Lu-**GFK-Y4** RP-HPLC (0–99% B in 15 min, AC1): *t*_R = 8.1 min.

Radiolabeling

Labeling of the tetramers with ¹⁷⁷Lu was performed manually in 300 μ L of a 1 M NaOAc/HOAc-buffer (pH 5.9). 1–7 nmol (1–7 μ L of an 1 mM stock solution) of labeling precursor were incubated with 5–106 μ L (32–106 MBq) of ¹⁷⁷LuCl₃ in 0.04 M HCl (Endolucin Beta, ITM, Germany) for 15–60 min at 95 °C. Completion of the reaction was monitored with radio-TLC or radio-HPLC (for chromatograms see ESI†).

SPECT imaging

SPECT imaging was performed for 60 min per animal on a nanoScan SPECT/CT (Mediso, Budapest, Hungary) at different timepoints (1 h, 3 h, 24 h, 72 h) after intravenous injection of approx. 40 MBq [¹⁷⁷Lu]Lu-**Y8** per animal (*n* = 2). Reconstruction, image analysis and quantification of SPECT data were performed using Nucline and Interview fusion software (both Mediso).

Conflicts of interest

N. G. Q. and J. N. are inventors on patent applications related to α v β 6-integrin binding peptide conjugates and ⁶⁸Ga-Trivehexin. J. N. and J. Š. are CSO and CEO, respectively, and co-founders of TRIMT GmbH (Radeberg, Germany) who has licensed IP from TU Munich. J. N. is furthermore a member of the Scientific Advisory Board of Radiopharm Theranostics LLC (Carlton, Australia) who has licensed IP from TRIMT GmbH. S. K. receives research support from TRIMT GmbH.

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