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Chemical and biological studies on the soft coral *Nephthea* sp.[†]

Omnia Hesham Abdelhafez,^{‡a} John Refaat Fahim,^{ID ‡b} Ramy R. El Masri,^c M. Alaraby Salem,^d Samar Yehia Desoukey,^b Safwat Ahmed,^e Mohamed Salah Kamel,^{ab} Sheila Marie Pimentel-Elardo,^f Justin R. Nodwell,^{ID f} and Usama Ramadan Abdelmohsen,^{ID *ab}

Soft corals belonging to the family Nephtheidae have been appreciated as marine sources of diverse metabolites with promising anticancer potential. In view of that, the current work investigates the anti-proliferative potential of the crude extract, different fractions, and green synthesized silver nanoparticles (AgNPs) of the Red Sea soft coral, *Nephthea* sp. against a panel of tumor cell lines. The metabolic pool of the soft coral under study was also explored via an LC-HR-ESI-MS metabolomics approach, followed by molecular docking analysis of the characterized metabolites against the target proteins, EGFR, VEGFR, and HER2 (erbB2) that are known to be involved in cancer cell proliferation, growth, and survival. Overall, the *n*-butanol fraction of *Nephthea* sp. exhibited the highest inhibitory activities against MCF7 (breast cancer) and A549 (lung cancer) cell lines, with interesting IC₅₀ values of 2.30 ± 0.07 and 3.12 ± 0.10 μg ml⁻¹, respectively, whereas the maximum growth inhibition of HL60 (leukemia) cells was recorded by the total extract (IC₅₀ = 2.78 ± 0.09 μg ml⁻¹). More interestingly, the anti-proliferative potential of the total soft coral extract was evidently improved when packaged in the form of biogenic AgNPs, particularly against A549 and MCF7 tumor cells, showing IC₅₀ values of 0.72 ± 0.06 and 9.32 ± 0.57 μg ml⁻¹, respectively. On the other hand, metabolic profiling of *Nephthea* sp. resulted in the annotation of structurally diverse terpenoids, some of which displayed considerable binding affinities and molecular interactions with the studied target proteins, suggesting their possible contribution to the anti-proliferative properties of *Nephthea* sp. via inhibition of tyrosine kinases, especially the EGFR type. Taken together, the present findings highlighted the relevance of *Nephthea* sp. to future anticancer drug discovery and provided a base for further work on the green synthesis of a range of bioactive NPs from marine soft corals.

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1. Introduction

Cancer is one of the major contributing diseases to human morbidity and mortality worldwide.¹ About 21 million new cancer cases and 13 million deaths are estimated by 2030, which will

exacerbate global health, social, and economic burdens.² Yet, conventional anticancer therapies, including surgery, chemotherapy, and radiation have not satisfactorily brought the anticipated efficacy, showing both controversial safety and undesirable side effects, *e.g.* alopecia, liver and kidney dysfunction, anemia, and immunosuppression.^{3,4} Additionally, a wide variety of malignant tumors has been reported to develop multi-drug resistance, leading to chemotherapy failure, which reflects the unremitting need for other effective and advanced treatment options.⁵ In this framework, marine ecosystems have been acknowledged as potent supplies of anticancer leads because of their both inconceivable biodiversity and capacity to afford therapeutically useful molecules, some of which are in current clinical use.^{6,7} Among various marine invertebrates, soft corals belonging to the family Nephtheidae have received attention as special providers of several chemical compounds with anti-fouling, antibacterial, antiviral, anti-diabetic, and anti-inflammatory activities.^{8–10} Likewise, a group of assorted metabolites with praiseworthy potential against human cancer cells has been described from various Nephtheidae species,^{8,9,11–14} which

^aDepartment of Pharmacognosy, Faculty of Pharmacy, Deraya University, 61111 New Minia, Egypt

^bDepartment of Pharmacognosy, Faculty of Pharmacy, Minia University, 61519 Minia, Egypt. E-mail: usama.ramadan@mu.edu.eg; Fax: +20-86-2369075; Tel: +20-86-2369075

^cDepartment of Pharmaceutical Chemistry, Faculty of Pharmacy, October University for Modern Sciences and Arts (MSA), Giza, Egypt

^dSchool of Life and Medical Sciences, University of Hertfordshire hosted by Global Academic Foundation, New Administrative Capital, Cairo, Egypt

^eDepartment of Pharmacognosy, Faculty of Pharmacy, Suez Canal University, 41522 Ismailia, Egypt

^fDepartment of Biochemistry, University of Toronto, MaRS Centre West, Toronto, ON, Canada

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[‡] Authors have equally contributed to this work.



valorizes the promising exploration of these prolific organisms in the realm of nature-inspired anticancer drug discovery.

Recent advancements in nanotechnology have blossomed into a myriad of applications that hold the potential to revolutionize modern medicine, encompassing diagnosis, prevention, and remediation of cancer.¹⁵ In this respect, nanoparticles (NPs) have gained considerable interest over the last decade thanks to their privileged capacity to refine the compatibility, bioavailability, and efficacy of many synthetic drugs and natural products in the management of chronic disorders, including cancer.^{16,17} These particles show unique and markedly different chemical, optical, magnetic, mechanical, and biological properties to their larger material counterparts owing to their small sizes (1–100 nm) and high surface-to-volume ratio.^{18,19} Interestingly, a considerable number of NP-based remedies are now in clinical use, including the chemotherapeutic agents 5-fluorouracil, paclitaxel, and doxorubicin, which endorses the impactful role of this approach in boosting the therapeutic potential of bioactive agents.²⁰ In this connection, metal-based NPs have been introduced as multi-functional tools with valued theranostic abilities and biomedical applications.^{17,21} This particular type of nanomaterials can act as new therapeutic agents or drug carriers with the advantage of eliciting more targeted activities that help avoid many unwelcome side effects.^{16,21} Out of various noble metal NPs, silver nanoparticles (AgNPs) have been reported to display remarkable chemical stability, catalytic activity, biocompatibility, and biological activities, including anti-inflammatory, antimicrobial, anti-diabetic, and wound healing properties, among others.^{18,21,22} They have also been broadly investigated in cancer research because of their notable cytotoxic and antitumor potential.^{17,21} Among the most striking advantages of AgNPs is their ability to release active ingredients to the anticipated site in a gradual and sustainable manner, leading to boosted efficacies; thus, their production is thought to accelerate in the next few years.^{18,22}

On the other hand, despite their chemical and biological prolificacy, the biogenic synthesis of AgNPs using marine invertebrates has been scarcely explored,²³ encompassing our recent report on the anti-inflammatory potential of the bio-synthesized AgNPs from the soft coral *Nephthea* sp., which revealed greater anti-COX-2 properties compared with its bulk extracts.²⁴ Therefore, in continuation of our interest in marine organisms and their bioactive NPs,^{24–28} the current study investigates the chemical complexity and anti-proliferative activity of the total extract and different fractions of *Nephthea* sp., along with the green synthesis of potential anti-proliferative AgNPs using the total soft coral extract. Molecular docking analysis was also considered in order to predict the possible binding of the identified metabolites to a number of cancer-related cellular proteins.

2. Materials and methods

2.1. Soft coral material

The animal material used in this work was collected from the Egyptian coasts of the Red Sea at Sharm El-Sheikh using scuba diving at a depth of 10 m by Dr Safwat Ahmed, Professor of Pharmacognosy, Faculty of Pharmacy, Suez Canal University,

Ismailia, Egypt. The collected material was immediately frozen and kept at $-20\text{ }^{\circ}\text{C}$ until investigation. The soft coral biomass was identified by Dr Tarek Temraz, Marine Science Department, Faculty of Science, Suez Canal University, Ismailia, Egypt. A voucher specimen was deposited in the herbarium section of Pharmacognosy Department, Faculty of Pharmacy, Suez Canal University under the registration number SAA-26.

2.2. Chemicals and reagents

Solvents used in this work were of analytical grade and distilled before use. They were obtained from El-Nasr Company for Pharmaceuticals and Chemicals, Egypt. Acetonitrile and methanol of HPLC grade were obtained from SDFCL SD Fine-Chem Limited, Mumbai, India. Both silver nitrate (AgNO_3 ; purity $\geq 99.5\%$) and ion exchange resin were purchased from Sigma-Aldrich, Germany. Staurosporine (Sigma-Aldrich, Germany) was used as a standard drug in cytotoxicity studies.

2.3. Extraction and fractionation

The freeze-dried soft coral was exhaustively extracted with methanol-methylene chloride (1 : 1)^{24,29,30} and the combined extracts were concentrated under reduced pressure until dryness. The obtained extract (24.0 g) was suspended in distilled water and extracted successively with petroleum ether, ethyl acetate, and *n*-butanol. The organic phase in each step was separately concentrated under vacuum to provide three corresponding fractions: I (10.0 g), II (3.0 g), and III (3.0 g). The remaining mother liquor was desalted using an ion exchange resin, followed by re-extraction with acetone. The latter was also concentrated under vacuum to afford fraction IV (200.0 mg). The crude extract and its fractions were kept at $4\text{ }^{\circ}\text{C}$ for metabolomics and biological analyses.

2.4. Metabolic profiling

Metabolic profiling of the total extract and different fractions of *Nephthea* sp. was carried out according to Abdelmohsen *et al.*³¹ using an Acquity Ultra Performance Liquid Chromatography system combined with a Xevo G2S-qToF quadrupole time-of-flight hybrid mass spectrometer (Waters, Milford, USA). Chromatographic separation was performed using a reversed phase BEH C18 column ($2.1 \times 100\text{ mm}$, $1.7\text{ }\mu\text{m}$ particle size; Waters, Milford, USA) connected to a guard column ($2.1 \times 5\text{ mm}$, $1.7\text{ }\mu\text{m}$ particle size; Waters, Milford, USA). The mobile phase used for chromatographic separation was composed of purified water (A) and acetonitrile (B); each containing 0.1% formic acid. Elution was started in the gradient manner at a flow rate of $200\text{ }\mu\text{L min}^{-1}$ with 10% B linearly increasing to 100% B within 30 min, and then kept isocratic for additional 5 min before linearly returning to 10% B for further 1 min. The injection volume was $2\text{ }\mu\text{L}$, while the column temperature was adjusted at $40\text{ }^{\circ}\text{C}$. MSConvert software was employed to convert raw data into separate positive and negative ionization files, which were then exported to the data mining software, MZmine 2.10 for peak picking, deconvolution, deisotoping, alignment, and formula prediction.³² Dereplication of compounds was accomplished by



comparison with the Dictionary of Natural Products (DNP), METLIN, and Marinlit databases.^{33–35}

2.5. Synthesis and characterization of AgNPs

The green synthesized AgNPs were prepared by adding 3 ml of the 0.0005% crude extract of *Nephthea* sp. in DMSO to 100 ml of 1 mM silver nitrate solution at room temperature. Formation of nanoparticles was observed by the color change from colorless to a brown color and confirmed by measuring the UV-Vis spectrum of the reaction at 200–600 nm using a double beam V-630 spectrophotometer (Jasco, Japan). The shape and size of the resulting AgNPs were determined using a Transmission Electron Microscope (TEM; Jeol model JEM-1010, USA) by adding a drop on a copper grid covered with a carbon support film and left to completely dry at room temperature. Fourier-Transform Infrared Spectroscopy (FT-IR) analysis of AgNPs was performed using a FT-IR-8400S spectrophotometer (IR Prestige-21, IR Affinity-1, Shimadzu, Japan) to determine AgNPs associated biomolecules.^{24,36} Particle size distribution and polydispersity index (PDI) of the prepared NPs was measured in a disposable cell at 25 °C using a Zeta-sizer Nano ZS (Malvern instruments, UK) and the results were then analyzed by Zeta-sizer 7.01 software, UK.

2.6. Anti-proliferative activity

The anti-proliferative potential of the total extract, different fractions, and AgNPs of *Nephthea* sp. was evaluated against three human tumor cell lines, namely breast cancer (MCF7), lung cancer (A549), and leukemia (HL60) cells *via* the MTT assay.³⁷ Cell lines were obtained from the American Type Culture Collection (Manassas, VA, USA). Cells were cultured using DMEM (Invitrogen/Life Technologies, USA) supplemented with 10% FBS (Hyclone, USA), 10 µg ml^{−1} of insulin (Sigma-Aldrich, Germany), and 1% penicillin-streptomycin. In 96-well plates, cells (at a density of $1.2\text{--}1.8 \times 10^4$ cells per well) were prepared in a volume of 100 µl per well of each of the complete growth medium and the tested sample for 24 h before testing. The MTT solution was reconstituted with 3 ml of the medium or a balanced salt solution without phenol red or serum and added in an amount equal to 10% of the culture medium volume. Cultures were incubated for 2–4 h depending on cell type and maximum cell density (an incubation period of 2 h was generally adequate, but was lengthened for either low cell densities or cells showing little metabolic activities). After incubation, the resulting formazan crystals were dissolved by adding a volume of DMSO equal to that of the original culture medium. The absorbance of each plate was then measured spectrophotometrically at 570 nm using an ELISA plate reader (Model 550, Bio-Rad, USA). Three independent experiments were performed. IC₅₀ values, representing the concentration responsible for 50% inhibition of cell growth, were calculated by GraphPad Prism 5 (Version 5.01, GraphPad Software, San Diego, CA, USA).

2.7. Molecular docking

Three crystal structures were selected to study the potential antitumor activity of the characterized compounds from *Nephthea* sp., including PDB ID: 1M17 (for the epidermal growth

factor receptor (EGFR) tyrosine kinase), PDB ID: 4asd (for the vascular endothelial growth factor receptor (VEGFR) tyrosine kinase), and PDB ID: 3PP0 (for the kinase domain of human HER2 (erbB2)) using erlotinib, sorafenib, and SYR127063 as the co-crystallized ligands, respectively. For all dockings, a grid box with dimensions of 50 grid points and spacing of 0.375 was centered on the given co-crystallized ligand. Docking was performed *via* Autodock4 implementing 100 steps of genetic algorithm while keeping the default setting provided by Autodock Tools.³⁸ Visualization was achieved using a discovery studio.³⁹ In total, compounds (1–8) were docked into the three active sites in comparison with the three co-crystallized ligands that served as reference molecules.

3. Results and discussion

To date, a variety of physical and chemical strategies has been described for the preparation of metal NPs; however, the release of hazardous chemicals and toxic byproducts during these practices have raised several environmental and health concerns.^{18,22} In contrast, biological methods of synthesis have emerged as less toxic, eco-friendly, and non-destructive techniques, with time- and cost-effective nature. Such green approaches have also efficiently allowed the incorporation of many plant extracts, bacteria, and fungi into various metal NPs in order to explore their biological potential.^{17,18,22} In this regard, the wide metabolic and chemical diversities of the aforementioned organisms have been reported to mediate the successful reduction, capping, and conversion of metal ions to metallic NPs with particular characteristics.^{18,22} In recent years, marine flora such as algae, marine bacteria, fungi, finfish, and sponges have been also researched as potential biological precursors for the fabrication of bioactive metal NPs owing to their content of various reductants and stabilizing agents.^{40,41} Nevertheless, a few reports exist in the literature on the synthesis of biogenic AgNPs using marine invertebrates, which have been mostly studied for their antimicrobial potential, exemplified by oyster (*Saccostrea cucullata* Born.),⁴² marine polychaete,²³ and sponges,

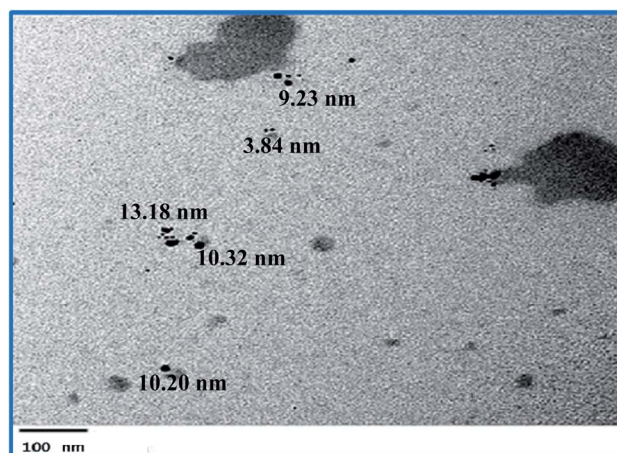


Fig. 1 A TEM photo indicating the shape and size of the formed AgNPs using the total extract of *Nephthea* sp.



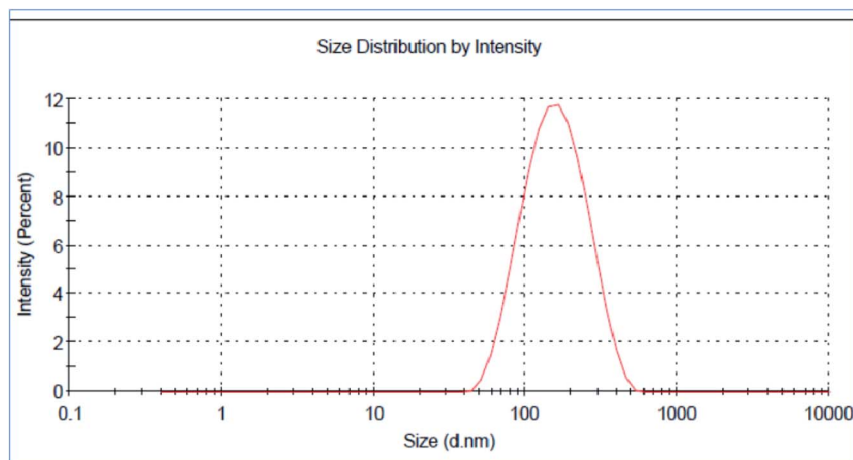


Fig. 2 Size distribution of the green synthesized AgNPs using the total extract of *Nephthea* sp.

e.g. *Acanthella elongate* Dendy,^{43,44} *Callyspongia diffusa* Ridley,⁴⁵ *Haliclona* spp.,^{46,47} *Axinella sinoxia* Alvarez & Hooper (Hamed *et al.*, 2017),⁴⁸ and *Amphimedon* sp.²⁸ Moreover, we have also earlier reported the first biosynthesis of soft coral-based AgNPs using the petroleum ether and acetone fractions of *Nephthea* sp., along with their interesting anti-COX-2 activities.²⁴ Inspired by this, the present work considered the green synthesis of AgNPs using the total extract of *Nephthea* sp., which were further tested for their anti-proliferative potential for the first time among various metal NPs derived from marine invertebrates.

3.1. Synthesis and characterization of AgNPs

The successful synthesis of AgNPs using the total *Nephthea* sp. extract was evident by the development of dark brown color, while the control (1 mM AgNO₃) solution did not show any color change (ESI Fig. S1†).³⁶ The gradual color change was observed after 24 h of incubation and the fully dark brown color appeared after 48 h. Such change in color is attributable to the excitation of surface plasmon, its vibrations, as well as changes in the

metal oxidation state (*i.e.* the reduction of Ag⁺ ions to Ag⁰ by various biomolecules present in the soft coral extract). Therefore, the intensity of the obtained brown color gradually increases as the reaction proceeds.^{18,22,49}

3.1.1. TEM characterization of AgNPs. The morphology and dimensions of the synthesized AgNPs were initially characterized using TEM analysis, which revealed the formation of spherical particles with an average size ranging between 3.84 and 13.18 nm (Fig. 1).

3.1.2. Dynamic light scattering (DLS) analysis of AgNPs. The z-average mean (*d* nm) of the nano-extract was 139.5 nm with a PDI of 0.255 (Fig. 2). Noteworthy, the mean particle size obtained by the TEM was noticeably smaller than that shown in the DLS analysis, which could be attributed to the aggregation of some minute particles or the adhesion of either water molecules or some organic stabilizers from the used extract on NPs' surface.^{50,51}

3.1.3. Zeta potential of AgNPs. The measurement of zeta potential denotes the degree of electrostatic charge repulsion or attraction between particles in a suspension; thus, it is

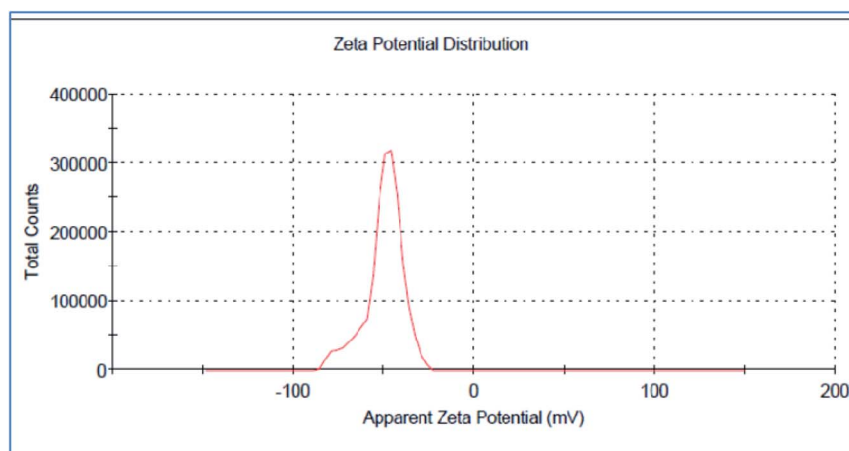


Fig. 3 Zeta potential analysis of the green synthesized AgNPs using the total extract of *Nephthea* sp.

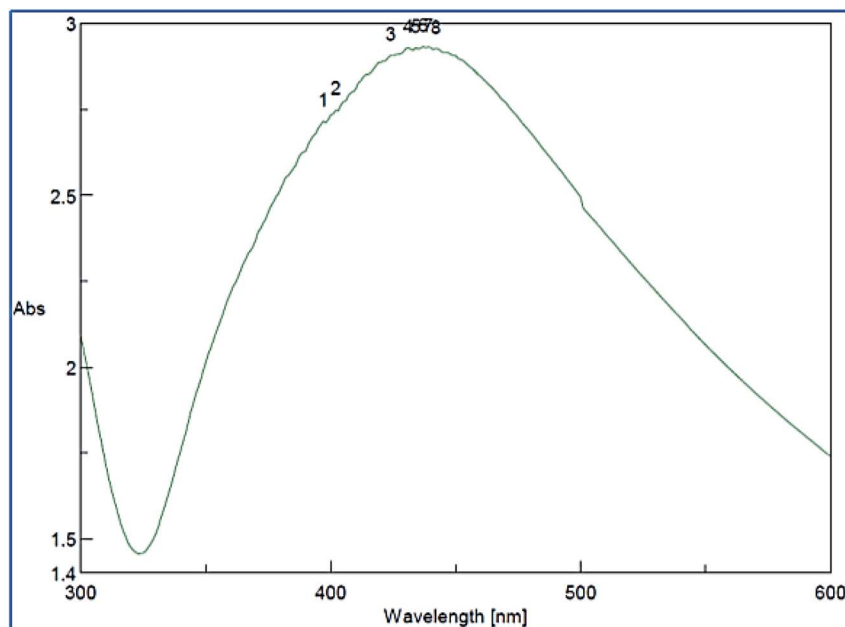


Fig. 4 UV-Vis spectral analysis of the green synthesized AgNPs using the total extract of *Nephthea* sp.

considered an important parameter for confirming the stability of NPs.⁵¹ Therefore, the surface charge of the biosynthesized AgNPs was determined by measuring their zeta potential, which was found to be -49.7 mV (Fig. 3). Such high values of zeta potential indicate the high electrical charge on the surface of NPs, leading to a strong electrostatic repulsion that prevents their agglomeration, supports their stability, and helps control their shape and size.^{51,52} Moreover, it was also reported that NPs that show zeta potentials of more than $+30$ mV or less than -30 mV are regarded as stable colloidal suspension;⁵² thus, in

view of the measured zeta potential, the prepared AgNPs of the total extract of *Nephthea* sp. represent an adequately stable colloidal system. Besides, the zeta potential of the AgNPs was also negative, suggesting the contribution of negatively charged functional groups from the total extract to the colloidal stability of the formed AgNPs.¹⁸

3.1.4. UV-Vis characterization of AgNPs. The synthesis of AgNPs was approved by measuring the UV-Vis spectrum of the colloidal reaction medium at 200–600 nm (Fig. 4). An absorbance band appeared at 417 nm in agreement with the reported

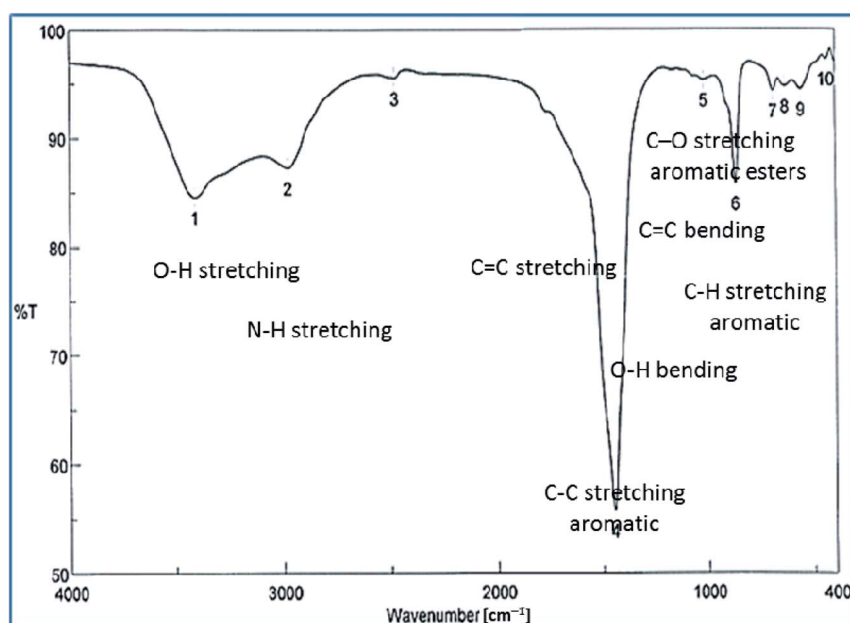


Fig. 5 FT-IR spectrum of the green synthesized AgNPs using the total extract of *Nephthea* sp.



literature data,^{22,36} which was due to the plasmon resonance electrons present on the surface of AgNPs, indicating their successful formation.^{18,22}

3.1.5. FT-IR characterization of AgNPs. FT-IR analysis was performed to identify different functional groups present in various biomolecules of the crude extract and their possible involvement in the synthesis and stabilization of AgNPs. The obtained spectrum (Fig. 5) revealed the presence of multiple peaks at 3424.96, 2375.87, 1955.47, 1651.73, 1573.63, 1444.42, 972.912, 877.452, 694.248, and 647.965 cm^{-1} , of which that at 3424.96 cm^{-1} is characteristic for O–H stretching in alcohols with strong hydrogen bonds and N–H stretching in primary amines. Additionally, the peak observed at 2375.87 cm^{-1} corresponds to O–H stretching of acids, whereas those at 1955.47, 1651.73, and 1573.63 cm^{-1} are consistent with C–H bending in aromatic compounds and C=C stretching vibrations of alkene groups. The peak at 1444.42 cm^{-1} is also assignable to C–C stretching in aromatic rings and O–H bending in carboxylic acids and alcohols, while the stretching of C–H in aromatic rings is responsible for those emerging at 877.452, 694.248, and 647.965 cm^{-1} . Likewise, the peak at 972.912 cm^{-1} is indicative of C=C bending of alkenes. Such data reflect the implication of different biomolecules in both the reduction of Ag^+ ions and stabilization of the synthesized AgNPs, such as terpenoids, steroids, peptides, proteins, polysaccharides, and pigments.^{22,24,36,53}

3.2. Metabolic profiling

Metabolic profiling of *Nephthea* sp. using LC-HR-ESI-MS for dereplication purposes led to the annotation of a number of structurally varied secondary metabolites, among which, sesquiterpenoids were noticed to prevail (Fig. 6; ESI Table S1 and Fig. S2†). Identification of these metabolites was achieved *via* coupling MZmine with some common databases such as DNP,

METLIN, and Marinlit. As a result, the mass ion peak at m/z 265.143, corresponding to the molecular formula $\text{C}_{15}\text{H}_{20}\text{O}_4$, was characterized as philippinlin E (1). This 4,5-seconeolemnane sesquiterpene was previously obtained from the soft coral *Lemnalia philippinensis* that belongs to the family Nephtheidae.⁵⁴ Moreover, the mass ion peak at m/z 267.159 in consonance with the suggested molecular formula $\text{C}_{15}\text{H}_{24}\text{O}_4$ was annotated as 5,8-epidioxy-11-hydroperoxy-6-eudesmen (2); a sesquiterpenoid that was earlier isolated from *Nephthea erecta* Kükenthal.⁵⁵ Another sesquiterpenoid of the nardosinan-type was also identified as paralemnolin L (3) in agreement with both the observed mass ion peak at m/z 276.196 and the molecular formula $\text{C}_{17}\text{H}_{24}\text{O}_3$. Noteworthy, this molecule was previously described among the natural metabolites of the family Nephtheidae, namely the soft coral *Paralemnalia thyrsoides* Ehrenberg.⁵⁶ In the same context, the mass ion peak at m/z 289.141 was annotated as laevinone A (4) in harmony with the molecular formula $\text{C}_{17}\text{H}_{22}\text{O}_4$. Laevinone A belongs to neolemnane sesquiterpenoids and was formerly reported from *Lemnalia laevis* Thomson and Dean.⁵⁷ Furthermore, two nardosinane-type sesquiterpenoids with the molecular formulas $\text{C}_{17}\text{H}_{22}\text{O}_4$ and $\text{C}_{16}\text{H}_{24}\text{O}_5$ were characterized as nardosinanol F (5) and nardosinanol I (6) based on the observed mass ion peaks at m/z 291.158 and 297.151, respectively. Both compounds were identified from some members within the Nephtheidae family, including *Paralemnalia clavata* Verseveldt and *Lemnalia africana* May, respectively.⁵⁸

Aside from the abovementioned metabolites, two structurally different terpenoids were also described comprising dendronpholide N (7); a cembrane type-diterpene and the polyhydroxylated sterol, dendronsterol B (8), which were annotated from the mass ion peaks at m/z 469.262 and 494.361 and the corresponding molecular formulas $\text{C}_{25}\text{H}_{42}\text{O}_8$ and $\text{C}_{29}\text{H}_{50}\text{O}_6$, respectively. Both compounds (7) and (8) were earlier reported

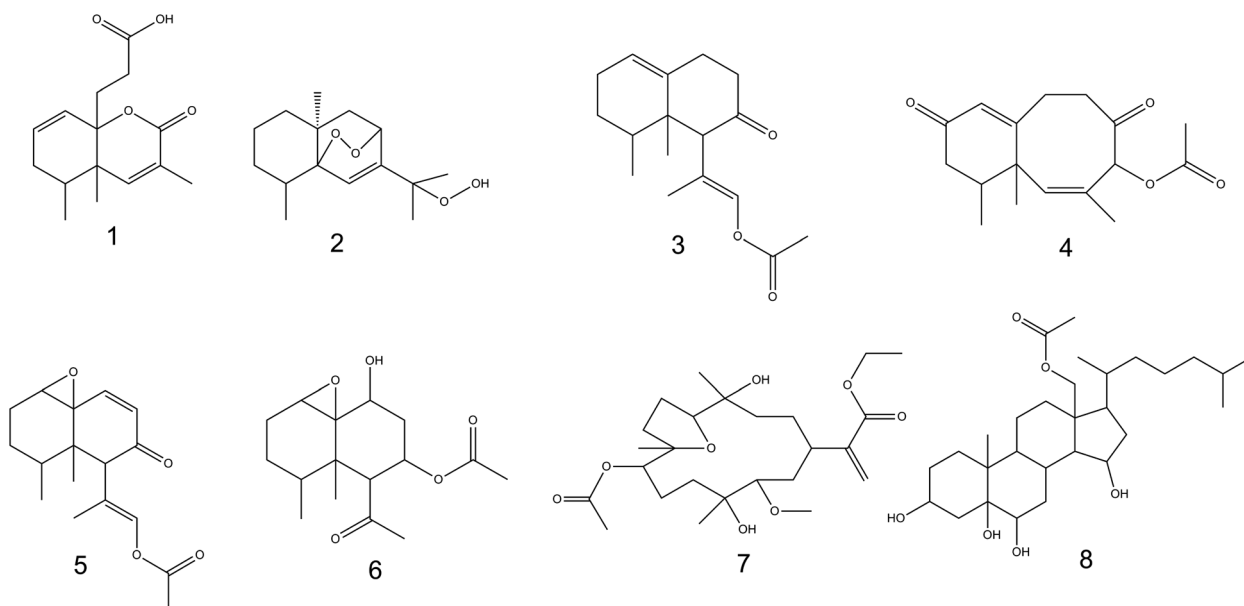


Fig. 6 Chemical structures of the characterized metabolites from *Nephthea* sp.

from the two Nephtheidae soft corals, *Dendronephthya* sp. and *Dendronephthya gigantea* Verrill, respectively.^{59,60} According to the reported literature,^{8,9} all the characterized metabolites are firstly reported herein from soft corals of the genus *Nephthea*, except for compound (2).

3.3. Anti-proliferative activity

Marine soft corals are endowed with the ability to accumulate several bioactive molecules that can serve as anticancer drug leads, exemplified by (18S)-18-O-acetyl nephthoacetal; a pentacyclic sterol isolated from *Nephthea* sp., which showed potent *in vitro* anti-proliferative activity against human cervical cancer (HeLa) cells ($IC_{50} = 10.1 \mu\text{g ml}^{-1}$).¹³ Sclerosteroid A is another pregnane-type steroid identified from *Scleronephthya gracillimum* Kükenthal with notable inhibitory potential towards human liver (HepG2) and breast (MDA-MB-231) cancer cells ($IC_{50} = 19.5$ and $15.8 \mu\text{M}$, respectively).¹² Lemnaphilin A was also reported from the Formosan soft coral, *Lemnalia philippinensis* May as potent cytotoxic sesquiterpenoid against a number of tumor cells, namely HepG2, MDA-MB231, and A549 cell lines, with IC_{50} values of 15.99, 16.31, and $15.81 \mu\text{g ml}^{-1}$, respectively.¹⁴ In the same context, members of the genus *Nephthea* have provided a range of diversified metabolites, mostly dominated by steroids and terpenoids that have been described as the major contributors to their promising anticancer properties.^{8,61} In view of that, it was of interest to evaluate the anti-proliferative potential of the total extract and different fractions of *Nephthea* sp. using the MTT assay in comparison with the reference drug, staurosporine. Overall, the tested samples revealed varying *in vitro* growth inhibitory potencies against MCF7, A549, and HL60 tumor cells, showing IC_{50} values in the range of 2.30 ± 0.07 to $141.8 \pm 0.46 \mu\text{g ml}^{-1}$ (Table 1). According to the US NCI (National Cancer Institute) screening guidelines, the majority of the tested extracts from *Nephthea* sp. exhibited IC_{50} values less than $20 \mu\text{g ml}^{-1}$ (Table 1), indicating their potent anti-proliferative effects.^{25,54} As depicted in Table 1, the *n*-butanol fraction exerted the greatest inhibitory activity against MCF7 cells, followed by the ethyl acetate fraction, total extract, and acetone fraction, with interesting IC_{50} values of 2.30 ± 0.07 , 9.96 ± 0.32 , 13.95 ± 0.45 , and $15.25 \pm 0.50 \mu\text{g ml}^{-1}$, respectively, which were even more potent than staurosporine ($IC_{50} = 20.30 \pm 0.6 \mu\text{g ml}^{-1}$). The petroleum ether fraction, on

the other hand, showed weaker cytotoxicity against MCF7 cells ($IC_{50} = 27.45 \pm 0.90 \mu\text{g ml}^{-1}$). Similarly, the *n*-butanol fraction displayed higher anti-proliferative effects against A549 cells ($IC_{50} = 3.12 \pm 0.10 \mu\text{g ml}^{-1}$) compared with staurosporine ($IC_{50} = 27.30 \pm 0.80 \mu\text{g ml}^{-1}$), followed by the acetone and petroleum ether fractions ($IC_{50} = 9.21 \pm 0.30$ and $11.32 \pm 0.30 \mu\text{g ml}^{-1}$, respectively), whereas both the ethyl acetate fraction and the total extract had much weaker potential against A549 cells ($IC_{50} = 66.40 \pm 2.18$ and $84.87 \pm 2.70 \mu\text{g ml}^{-1}$, respectively). On the other hand, the total soft coral extract exhibited the maximum growth inhibitory potential against HL60 cells ($IC_{50} = 2.78 \pm 0.09 \mu\text{g ml}^{-1}$), followed by staurosporine ($IC_{50} = 18.17 \pm 0.59 \mu\text{g ml}^{-1}$), while the ethyl acetate and petroleum ether fractions were moderately active ($IC_{50} = 23.74 \pm 0.78$ and $31.22 \pm 1.02 \mu\text{g ml}^{-1}$, respectively). Contrary to this, both the *n*-butanol and acetone fractions showed very weak activities towards the HL60 cell line ($IC_{50} = 80.45 \pm 2.60$ and $141.8 \pm 0.46 \mu\text{g ml}^{-1}$, respectively) (Table 1).

Recent years have witnessed an increasing amount of research on the anticancer perspective of nano-sized natural products, with those packaged as AgNPs revealed a promising potential against various tumor cells both *in vitro* and *in vivo*, including cervical, breast, liver, lung, nasopharyngeal, colorectal, and prostate cancers, among others.^{17,21,62} These natural product-based AgNPs have also been proven to exert advanced pharmacological and cytological effects in comparison with the chemical entities they contain thanks to their superior physicochemical and surface properties, considerable loading ability of natural compounds, and significant cellular interactions, which reflect their possible application in nano-chemoprevention and nano-chemotherapy.^{17,18} In line with the reported literature,^{17,21,62} our results unveiled the noteworthy anti-proliferative potential of the synthesized AgNPs from the total extract of *Nephthea* sp. as inferred from their higher growth inhibitory activities against A549 and MCF7 cell lines compared with the bulk total extract and staurosporine; showing IC_{50} values of 0.72 ± 0.06 and $9.32 \pm 0.57 \mu\text{g ml}^{-1}$, respectively (Table 1). These findings therefore imply the possible exceptional role of metallic NPs in augmenting the anticancer potential of *Nephthea* sp. Thus far, several mechanisms have been proposed to mediate the cytotoxic effects of biogenic

Table 1 *In vitro* anti-proliferative activities of the total extract, different fractions, and AgNPs of *Nephthea* sp.

Sample	IC_{50} ($\mu\text{g ml}^{-1}$)		
	A549	MCF7	HL60
Total extract	84.87 ± 2.70	13.95 ± 0.45	2.78 ± 0.09
Petroleum ether fraction	11.32 ± 0.30	27.45 ± 0.90	31.22 ± 1.02
Ethyl acetate fraction	66.40 ± 2.18	9.96 ± 0.32	23.74 ± 0.78
<i>n</i> -Butanol fraction	3.12 ± 0.10	2.30 ± 0.07	80.45 ± 2.60
Acetone fraction	9.21 ± 0.30	15.25 ± 0.50	141.8 ± 0.46
AgNPs of the total extract	0.72 ± 0.06	9.32 ± 0.57	10.07 ± 0.71
Staurosporine	27.30 ± 0.80	20.30 ± 0.60	18.17 ± 0.59

Table 2 Docking scores of compounds (1–8) in the active sites of the three selected proteins

Compound	1M17	3pp0	4asd
Co-crystallized ligand	−7.8	−10.0	−11.8
Philippinlin E (1)	−7.0	−7.4	−7.0
5,8-Epidioxy-11-hydroperoxy-6-eudesmen (2)	−7.7	−8.3	−8.1
Paralemnolin L (3)	−7.7	−7.4	−7.5
Laevinone A (4)	−7.9	−8.7	−7.7
Nardosinanol F (5)	−7.7	−7.1	−6.4
Nardosinanol I (6)	−7.1	−7.3	−6.6
Dendronpholide N (7)	−7.3	−7.5	−7.3
Dendronsterol B (8)	−6.7	−6.3	−6.9



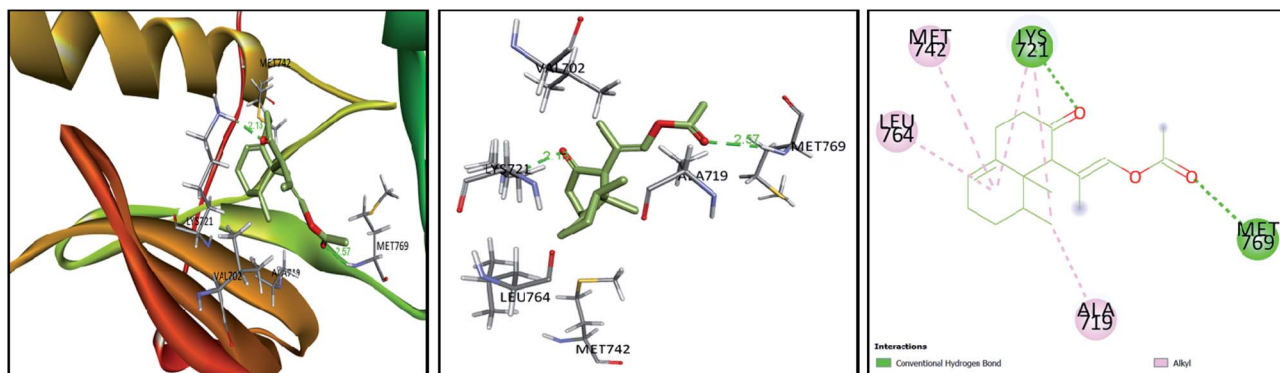


Fig. 7 3D and 2D plots of the poses of compound 3 at the active site of EGFR tyrosine kinase (PBD:1M17).

AgNPs on cancerous cells, encompassing activation of apoptotic pathways by destroying the ultrastructure of tumor cells, disruption of mitochondrial functions *via* stimulating the production of reactive oxygen species, and suppression of ATP synthesis, leading to DNA damage. Their ability to target some cellular enzymes and signaling pathways has been also testified in a variety of tumor cells.^{17,21,62}

3.4. Molecular docking

Tyrosine kinases are pivotal mediators of intracellular signaling, showing several key roles in different cellular events, such as growth, differentiation, metabolism, and apoptosis. Though their activity is strictly controlled in normal cells, the over-activation of these entities due to the expression of mutated kinase genes can embolden carcinogenesis.⁶³ Hence, a series of studies have reported the implication of tyrosine kinases in cancer pathogenesis and their potential as propitious anticancer drug targets.⁶⁴ Members of the EGFR family, such as EGFR and HER2 (erbB2) are currently among the most targeted tyrosine kinases owing to their regulatory roles in some complex signal transduction networks.^{63,65} Both EGFR and HER2 are also known to be amplified in many cancers, including breast, lung, and colorectal tumors. Such overexpression can halt apoptosis, resulting in unregulated cell cycle, proliferation, invasion, and neovascularization.^{63,66} In the interim, the proangiogenic factor, VEGF plays an important role in inducing and regulating angiogenesis, which underwrites tumor survival and metastasis. The overexpression of VEGFR tyrosine kinase was also seen to hasten the onset and development of cancer; consequently, blocking of angiogenesis *via* VEGF suppression represents another attractive target in cancer therapy.⁶⁷ For this, a range of anticancer agents has been developed to counteract the actions of EGFR, HER2, and VEGF; however, multiple therapeutic downsides and off-target effects have hampered their efficacies.⁶³ In this context, accumulated data have shown the capacity of natural products to modulate or inhibit protein kinases, providing a cornucopia of antitumor drug candidates.⁶⁸ In light of this as well as the observed anti-proliferative potential of *Nephthea* sp., the current docking approach was considered to investigate the possible interactions of compounds (1–8) with the above-mentioned cellular proteins in comparison with

a number of specific kinase inhibitors. As shown in Table 2, the docked compounds formed considerably stable complexes with 1M17 with markedly comparable binding scores (-6.7 to -7.9 kcal mol⁻¹) to that of the co-crystallized ligand, erlotinib (-7.8 kcal mol⁻¹). Among them, compounds (2, 3, 4, and 5) revealed the highest binding affinities within the active site of 1M17 (Table 2). Erlotinib is an EGFR inhibitor that is used for pancreatic cancer, non-small cell lung cancer, and some other tumor types.⁶⁹ This drug is known to interact with the gate-keeper Thr766 *via* a water-bridged H-bond, in addition to hydrogen bonding with the amino acid Met769 in the hinge region of EGFR tyrosine kinase.⁷⁰ Interestingly, some of the tested metabolites herein displayed analogous interactions to those of erlotinib, exemplified by compounds (3) and (5) that formed hydrogen bonds with Met769. Likewise, some of the hydrophobic contacts exhibited by erlotinib were also reproduced by our compounds, including the interactions of compounds (3) and (5) with Ala719 and those of (3) with Leu764 as shown in Fig. 7. On the other hand, the binding energy scores of compounds (1–8) in case of 3pp0 and 4asd were relatively lower than the co-crystallized ligands, SYR127063 (-10.0 kcal mol⁻¹) and sorafenib (-11.8 kcal mol⁻¹), respectively (Table 2), of which compounds (2) and (4) showed the highest binding aptitudes in terms of their energy scores (-8.3 and -8.7 kcal mol⁻¹ for 3pp0 and -8.1 and -7.7 kcal mol⁻¹ for 4asd, respectively). These results collectively suggest the contribution of the characterized metabolites to the observed anti-proliferative activities of *Nephthea* sp. that might involve the inhibition of tyrosine kinases, particularly the EGFR type. In this connection, different receptor–ligand interactions displayed by compounds (1–8) could also offer better insights to design alternative anticancer agents with fewer unwanted effects.

4. Conclusion

The current study highlighted the anti-proliferative potential of the total extract and different fractions of the soft coral *Nephthea* sp., which showed notable inhibitory activities against MCF7, A549, and HL60 tumor cells. Such effects are likely underlain by the availability of a range of compounds, mostly

terpenoids that were mined with the help of LC-MS-based metabolomics. Comparative docking screening of the characterized metabolites also revealed their ability to interact with the active sites of EGFR, HER2, and VEGF, denoting their probable contribution to the anti-proliferative potential of *Nephthea* sp. as tyrosine kinase inhibitory molecules, particularly compounds (2) and (4). Additionally, our results provided evidence for the interesting role of biogenic AgNPs in enhancing the anticancer properties of *Nephthea* sp., which were addressed herein for the first time among marine invertebrates. The obtained findings drew attention to the potential of *Nephthea* sp. to provide the current therapeutic arsenal against cancer with alternative agents of natural origin, offering a good starting point for future research on the development of nanoparticle-based chemotherapies using marine soft corals.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 J. T. McDaniel, K. Nuhu, J. Ruiz and G. Alorbi, *Glob. Health Promot.*, 2019, **26**, 41–49.
- 2 L. A. Torre, R. L. Siegel, E. M. Ward and A. Jemal, *Cancer Epidemiol. Biomark. Prev.*, 2016, **25**, 16–27.
- 3 V. Schirrmacher, *Int. J. Oncol.*, 2019, **54**, 407–419.
- 4 E. K. Davison and M. A. Brimble, *Curr. Opin. Chem. Biol.*, 2019, **52**, 1–8.
- 5 H. Choudhury, M. Pandey, T. H. Yin, T. Kaur, G. W. Jia, S. L. Tan, H. Weijie, E. K. S. Yang, C. G. Keat and S. K. Bhattamishra, *Mater. Sci. Eng., C*, 2019, **101**, 596–613.
- 6 G. Ercolano, P. De Cicco and A. Ianaro, *Mar. Drugs*, 2019, **17**, 31.
- 7 S.-K. Kim, *Handbook of Anticancer Drugs from Marine Origin*, Springer International Publishing, Switzerland, 2014.
- 8 O. H. Abdelhafez, J. R. Fahim, S. Y. Desoukey, M. S. Kamel and U. R. Abdelmohsen, *Chem. Biodivers.*, 2019, **16**, e1800692.
- 9 J. Hu, B. Yang, X. Lin, X. Zhou, X. Yang, L. Long and Y. Liu, *Chem. Biodivers.*, 2011, **8**, 1011–1032.
- 10 V. T. Sang, T. T. H. Dat, L. B. Vinh, L. C. V. Cuong, P. T. T. Oanh, H. Ha, Y. H. Kim, H. L. T. Anh and S. Y. Yang, *Mar. Drugs*, 2019, **17**, 468.
- 11 M.-E. Hegazy, A. M. Gamal-Eldeen, T. A. Mohamed, M. A. Alhammady, A. A. Hassani, M. A. Shreadah, I. I. Abdelgawad, E. M. Elkady and P. W. Paré, *Nat. Prod. Res.*, 2016, **30**, 1266–1272.
- 12 H.-Y. Fang, C.-C. Liaw, C.-H. Chao, Z.-H. Wen, Y.-C. Wu, C.-H. Hsu, C.-F. Dai and J.-H. Sheu, *Tetrahedron*, 2012, **68**, 9694–9700.
- 13 J. Zhang, L.-C. Li, K.-L. Wang, X.-J. Liao, Z. Deng and S.-H. Xu, *Bioorg. Med. Chem. Lett.*, 2013, **23**, 1079–1082.
- 14 Y.-J. Xio, M.Sc. thesis, National Sun Yat-sen University, China, 2011.
- 15 E. S. Kawasaki and A. Player, *Nanomed. Nanotechnol. Biol. Med.*, 2005, **1**, 101–109.
- 16 B. Klębowski, J. Depciuch, M. Parlińska-Wojtan and J. Baran, *Int. J. Mol. Sci.*, 2018, **19**, 4031.
- 17 V. Rao, D. Nallappan, K. Madhavi, S. Rahman, L. Jun Wei and S. H. Gan, *Oxid. Med. Cell. Longevity*, 2016, **2016**, 3685671.
- 18 P. R. M. Hemlata, A. P. Singh and K. K. Tejavath, *ACS Omega*, 2020, **5**, 5520–5528.
- 19 A. Roy, O. Bulut, S. Some, A. K. Mandal and M. D. Yilmaz, *RSC Adv.*, 2019, **9**, 2673–2702.
- 20 A. C. Anselmo and S. Mitragotri, *Bioeng. Transl. Med.*, 2019, **4**, e10143.
- 21 L. Xu, Y.-Y. Wang, J. Huang, C.-Y. Chen, Z.-X. Wang and H. Xie, *Theranostics*, 2020, **10**, 8996–9031.
- 22 Y. Singh, S. Kaushal and R. S. Sodhi, *Nanoscale Adv.*, 2020, **2**, 3972–3982.
- 23 R. Singh, S. K. Sahu and M. Thangaraj, *J. Nanopart.*, 2014, **2014**, 718240.
- 24 O. H. Abdelhafez, T. F. S. Ali, J. R. Fahim, S. Y. Desoukey, S. Ahmed, F. A. Behery, M. S. Kamel, T. A. Gulder and U. R. Abdelmohsen, *Int. J. Nanomed.*, 2020, **15**, 5345–5360.
- 25 R. F. A. Abdelhameed, E. S. Habib, M. S. Goda, J. R. Fahim, H. A. Hassanean, E. E. Eltamany, A. K. Ibrahim, A. M. Aboulmagd, S. Fayed, A. M. A. El-kader, T. Al-Warhi, G. Bringmann, S. A. Ahmed and U. R. Abdelmohsen, *Mar. Drugs*, 2020, **18**, 354.
- 26 R. F. Abdelhameed, E. S. Habib, N. A. Eltahawy, H. A. Hassanean, A. K. Ibrahim, J. R. Fahim, A. M. Sayed, O. M. Hendawy, U. R. Abdelmohsen and S. A. Ahmed, *Tetrahedron Lett.*, 2021, **72**, 152986.
- 27 Y. Elsayed, J. Refaat, U. R. Abdelmohsen, E. M. Othman, H. Stopper and M. A. Fouad, *Phytochem. Anal.*, 2018, **29**, 543–548.
- 28 N. H. Shady, A. R. Khattab, S. Ahmed, M. Liu, R. J. Quinn, M. A. Fouad, M. S. Kamel, A. B. Muhsinah, M. Krischke and M. J. Mueller, *Int. J. Nanomed.*, 2020, **15**, 3377–3389.
- 29 T. H. Hsiao, C. S. Sung, Y. H. Lan, Y. C. Wang, M. C. Lu, Z. H. Wen, Y. C. Wu and P. J. Sung, *Mar. Drugs*, 2015, **13**, 3443–3453.
- 30 W. A. Tanod, U. Yanuhar, Maftuch, M. Y. Putra and Y. Risjani, *Anti-Inflammatory Anti-Allergy Agents Med. Chem.*, 2019, **18**, 126–141.
- 31 U. Abdelmohsen, C. Cheng, C. Viegelmann, T. Zhang, T. Grkovic, S. Ahmed, R. J. Quinn, U. Hentschel and R. Edrada-Ebel, *Mar. Drugs*, 2014, **12**, 1220–1244.
- 32 Mzmine, <http://sourceforge.net/projects/mzmine/>, (accessed August 2019).



- 33 *Dictionary of Natural Products*, <http://dnp.chemnetbase.com/faces/chemical/ChemicalSearch.xhtml>, (accessed August 2019).
- 34 *METLIN*, <http://metlin.scripps.edu/index.php>, (accessed August 2019).
- 35 *MarinLit*, <http://pubs.rsc.org/marinlit/>, (accessed August 2019).
- 36 E. G. Haggag, A. M. Elshamy, M. A. Rabeh, N. M. Gabr, M. Salem, K. A. Youssif, A. Samir, A. Bin Muhsinah, A. Alsayari and U. R. Abdelmohsen, *Int. J. Nanomed.*, 2019, **14**, 6217–6229.
- 37 T. Rasheed, M. Bilal, H. M. Iqbal and C. Li, *Colloids Surf. B Biointerfaces*, 2017, **158**, 408–415.
- 38 G. M. Morris, R. Huey, W. Lindstrom, M. F. Sanner, R. K. Belew, D. S. Goodsell and A. J. Olson, *J. Comput. Chem.*, 2009, **30**, 2785–2791.
- 39 T. Mohammad, F. I. Khan, K. A. Lobb, A. Islam, F. Ahmad and M. I. Hassan, *J. Biomol. Struct. Dyn.*, 2019, **37**, 1813–1829.
- 40 N. Asmathunisha and K. Kathiresan, *Colloids Surf. B Biointerfaces*, 2013, **103**, 283–287.
- 41 C. R. Singh, K. Kathiresan and S. Anandhan, *Afr. J. Biotechnol.*, 2015, **14**, 1525–1532.
- 42 S. Umayaparvathi, M. Arumugam, S. Meenakshi and T. Balasubramanian, *Int. J. Sci. Nat.*, 2013, **4**, 199–203.
- 43 D. Inbakandan, C. Kumar, L. S. Abraham, R. Kirubakaran, R. Venkatesan and S. A. Khan, *Colloids Surf. B Biointerfaces*, 2013, **111**, 636–643.
- 44 D. Inbakandan, G. Sivaleela, D. M. Peter, R. Kiurbakaran, R. Venkatesan and S. A. Khan, *Mater. Lett.*, 2012, **87**, 66–68.
- 45 V. Vasanthabharathi, V. Kalaiselvi and S. J. Jayalakshmi, *Egypt. Acad. J. Biol. Sci. B. Zoology*, 2013, **5**, 40–48.
- 46 M. R. Hamed, M. H. Givianrad and A. M. Moradi, *Orient. J. Chem.*, 2015, **31**, 125–130.
- 47 G. Arya, N. Sharma, R. Mankamna and S. Nimesh, Antimicrobial silver nanoparticles: future of nanomaterials, in *Microbial Nanobionics*, Springer, 2019, pp. 89–119.
- 48 M. Hamed, M. Givianrad and A. Moradi, *Indian J. Geomar. Sci.*, 2017, **46**, 125–130.
- 49 S. Gurunathan, J. W. Han, V. Eppakayala, M. Jeyaraj and J.-H. Kim, *BioMed Res. Int.*, 2013, **2013**, 535796.
- 50 K. A. Youssif, E. G. Haggag, A. M. Elshamy, M. A. Rabeh, N. M. Gabr, A. Seleem, M. A. Salem, A. S. Hussein, M. Krischke, M. J. Mueller and U. R. Abdelmohsen, *PLoS One*, 2019, **14**, e0223781.
- 51 K. A. Youssif, A. M. Elshamy, M. A. Rabeh, N. Gabr, W. M. Affi, M. A. Salem, A. Albohy, U. R. Abdelmohsen and E. G. Haggag, *ChemistrySelect*, 2020, **5**, 12278–12286.
- 52 K. Shameli, M. Bin Ahmad, E. A. Jaffar Al-Mulla, N. A. Ibrahim, P. Shabanzadeh, A. Rustaiyan, Y. Abdollahi, S. Bagheri, S. Abdolmohammadi, M. S. Usman and M. Zidan, *Molecules*, 2012, **17**, 8506–8517.
- 53 Y. K. Mohanta, S. K. Panda, R. Jayabalan, N. Sharma, A. K. Bastia and T. K. Mohanta, *Front. Mol. Biosci.*, 2017, **4**, 14.
- 54 Y.-J. Xio, J.-H. Su, Y.-J. Tseng, B.-W. Chen, W. Liu and J.-H. Sheu, *Mar. Drugs*, 2014, **12**, 4495–4503.
- 55 S.-Y. Cheng, C.-F. Dai and C.-Y. Duh, *J. Nat. Prod.*, 2007, **70**, 1449–1453.
- 56 S.-Y. Cheng, E.-H. Lin, J.-S. Huang, Z.-H. Wen and C.-Y. Duh, *Chem. Pharm. Bull.*, 2010, **58**, 381–385.
- 57 A. A. El-Gamal, E.-P. Chiu, C.-H. Li, S.-Y. Cheng, C.-F. Dai and C.-Y. Duh, *J. Nat. Prod.*, 2005, **68**, 1749–1753.
- 58 A. Bishara, D. Yeffet, M. Sisso, G. Shmul, M. Schleyer, Y. Benayahu, A. Rudi and Y. Kashman, *J. Nat. Prod.*, 2008, **71**, 375–380.
- 59 A. Ma, Z. Deng, L. van Ofwegen, M. Bayer, P. Proksch and W. Lin, *J. Nat. Prod.*, 2008, **71**, 1152–1160.
- 60 K. Yoshikawa, S. Kanekuni, M. Hanahusa, S. Arihara and T. Ohta, *J. Nat. Prod.*, 2000, **63**, 670–672.
- 61 Y.-S. Lee, T.-H. Duh, S.-S. Siao, R.-C. Chang, S.-K. Wang and C.-Y. Duh, *Mar. Drugs*, 2017, **15**, 392–401.
- 62 K. C. Hembram, R. Kumar, L. Kandha, P. K. Parhi, C. N. Kundu and B. K. Bindhani, *Artif. Cells, Nanomed., Biotechnol.*, 2018, **46**, S38–S51.
- 63 M. K. Paul and A. K. Mukhopadhyay, *Int. J. Med. Sci.*, 2004, **1**, 101–115.
- 64 P. S. Sharma, R. Sharma and T. Tyagi, *Curr. Pharm. Des.*, 2009, **15**, 758–776.
- 65 R. J. Craven, H. Lightfoot and W. G. Cance, *Surg. Oncol.*, 2003, **12**, 39–49.
- 66 C. T. Guy, R. D. Cardiff and W. J. Muller, *J. Biol. Chem.*, 1996, **271**, 7673–7678.
- 67 I. F. Nerini, M. Cesca, F. Bizzaro and R. Giavazzi, *Chin. J. Canc.*, 2016, **35**, 61.
- 68 M. S. A. Gill, H. Saleem and N. Ahemad, *Curr. Top. Med. Chem.*, 2020, **20**, 1093–1104.
- 69 C. Schettino, M. A. Bareschino, V. Ricci and F. Ciardiello, *Expert Rev. Respir. Med.*, 2008, **2**, 167–178.
- 70 K. Choowongkamon, O. Sawatdichaikul, N. Songtawee and J. Limtrakul, *Molecules*, 2010, **15**, 4041–4054.

