### Materials Advances



PAPER View Article Online
View Journal | View Issue



Cite this: *Mater. Adv.*, 2020, 1, 2862

# Tris(triazolo)triazine-based emitters for solution-processed blue thermally activated delayed fluorescence organic light-emitting diodes†‡

Fabian Hundemer, \$\overline{\mathbb{D}}\\$^a\$ Ettore Crovini, \$\overline{\mathbb{D}}\\$^b\$ Yoshimasa Wada, \$\cap \text{Hironori Kaji}, \$\overline{\mathbb{D}}\\*^c\$ Stefan Bräse\*\*\* and Eli Zysman-Colman \$\overline{\mathbb{D}}\\*^b\$

Received 31st August 2020, Accepted 13th October 2020

DOI: 10.1039/d0ma00659a

rsc.li/materials-advances

We report a new emitter 3,4,5-3TCz-TTT based on a tris(triazolo)triazine acceptor that shows thermally activated delayed fluorescence and cross-compare its performance with the recently reported analogue, 3DMAC-TTT. These compounds show blue emission and delayed fluorescence with delayed lifetimes on the order of milliseconds. Solution-processed organic light-emitting diodes achieving maximum external quantum efficiencies,  $EQE_{max}$ , of 5.8% for 3,4,5-3TCz-TTT and 11.0% for 3DMAC-TTT.

### Introduction

Organic light-emitting diodes (OLEDs) generate light by the recombination of electrically generated holes and electrons to form excitons that subsequently radiatively decay. Due to spin-statistics, 75% of the generated excitons are triplets, while 25% are singlet excitons. Thermally activated delayed fluorescence (TADF) is considered one of the most promising mechanisms for efficient electroluminescent devices as 100% of the excitons can be converted to light. A small energy gap between the lowest singlet (S<sub>1</sub>) and triplet (T<sub>1</sub>) excited states ( $\Delta E_{\rm ST}$ ) enables triplet excitons to be converted to singlet excitons by reverse intersystem crossing (RISC) at ambient temperatures, which can subsequently relax *via* the fluorescence channel. Typically, a small  $\Delta E_{\rm ST}$  is achieved by the spatial separation of the highest occupied molecular orbital (HOMO)

By employing solution-based processing techniques, 19-21 large-area OLEDs becomes simpler and more cost-efficient to fabricate. However, solution-processed devices often show inferior performance compared to the vacuum-deposited counterparts.<sup>22-25</sup> For solution-processed devices to become competitive, their performance must improve to rival vacuumsublimed devices. Typically, solution-processed devices rely on polymer emitters as the use of polymers produces high-quality, defect-free amorphous films. TADF polymers have shown promise as emitters in solution-processed devices. 23,26-28 However, the synthesis of polymers will lead to a mixture of compounds with different molecular weight, each of which will have a slightly different set of photophysical properties that will result in a broadened emission spectrum.<sup>29</sup> Moreover, the photophysical properties of TADF polymers usually suffer due to quenching via intra- and inter-molecular charge transfer between the TADF units on the polymer. 30,31 Dendrimeric TADF emitters also show good solubility and generate high-quality amorphous films. OLED performance, however, tends to fall off rapidly with increasing generation size of the dendron and thus far OLEDs using dendrimer emitters have underperformed compared to their polymer and small molecule emitter counterparts.32 The use of small molecule emitters has to date produced the highest performance solution-processed OLEDs.

and the lowest unoccupied molecular orbital (LUMO). The classic design strategies are based on the combination of donor and acceptor groups and their small electronic coupling either *via* (1) a twisted intramolecular charge transfer<sup>6-11</sup> transition, (2) a through-space charge-transfer transition, <sup>7,8,12</sup> (3) a spiroconjugation charge transfer transition <sup>13,14</sup> or (4) an alternating HOMO/LUMO distribution in p,n-doped polycyclic aromatic hydrocarbons, termed a multi-resonance TADF transition. <sup>2,15-18</sup>

<sup>&</sup>lt;sup>a</sup> Institute of Organic Chemistry, Karlsruhe Institute of Technology (KIT), Fritz-Haber-Weg 6, 76131 Karlsruhe, Germany. E-mail: braese@kit.edu; Fax: (+49)-721-6084-8581; Tel: (+49)-721-6084-2903

b Organic Semiconductor Centre, EaStCHEM School of Chemistry, University of St Andrews, St Andrews, Fife, KY16 9ST, UK

E-mail: eli.zysman-colman@st-andrews.ac.uk; Web: http://www.zysman-colman.com; Tel: +44 (0)1334 463826

<sup>&</sup>lt;sup>c</sup> Institute for Chemical Research Kyoto University Uji, Kyoto 611-0011, Japan. E-mail: kaji@scl.kyoto-u.ac.jp

<sup>&</sup>lt;sup>d</sup> Institute of Biological and Chemical Systems – Functional Molecular Systems (IBCS-FMS), Karlsruhe Institute of Technology (KIT),

Hermann-von-Helmholtz-Platz 1, D-76344, Eggenstein-Leopoldshafen, Germany † The research data supporting this publication can be accessed at https://doi.org/10.17630/586490c1-7c86-48aa-ad7c-f90809626afb.

<sup>‡</sup> Electronic supplementary information (ESI) available. See DOI: 10.1039/d0ma00659a

<sup>§</sup> Equal contributions.

Materials Advances Paper

Fig. 1 Molecular structures and performances of the discussed solution-processed materials.

Typically, solubilizing groups are decorated around a known emitter core to improve the film morphology of the emissive layer in solution-processed OLEDs. An example of this can be found in the work of Cho et al. where they investigated the effect that solution-processing and vacuum deposition have on the performance of OLEDs, employing three related TADF emitters: 4CzIPN, m4CzIPN and t4CzIPN (Fig. 1).33 The solution-processed device of 4CzIPN shows a substantial drop in performance compared to the vacuum-deposited OLED (maximum external quantum efficiency, EQE<sub>max</sub>, decreasing 26.0% to 8.1%). The incorporation of methyl groups in m4CzIPN has no effect on the efficiency of the solutionprocessed device, leading again to inferior efficiencies than the evaporated OLED (from 19.6% to 8.2%). By contrast, the use of tert-butylcarbazole in t4CzIPN resulted in a much-enhanced solubility of the emitter, stabilizing the morphology of the solution-processed film, which translated into a solutionprocessed OLED with comparable efficiency to the vacuumdeposited device (17.1% for the vacuum-deposited device and 18.3% for the solution-processed device). In a similar vein, Chen et al. reported that the device efficiency could be significantly improved from 8.0% to 19.1% by replacement of carbazole with tert-butycarbazole in triarylborane-based TADF emitters.<sup>34</sup> Acridine-based emitters have also been successfully employed in several solution-processed devices. For instance, Wada et al. reported a solution-processed OLED using triazinebased TADF emitter 3ACR-TRZ, which showed an EQEmax of 18.6%.<sup>35</sup> The same group reported a high-performance solution-processed blue OLED (CIE: 0.15, 0.19) using MA-TA as the emitter, which achieved an EQE<sub>max</sub> of 22.1%.36

1,2,4-Triazoles<sup>37</sup> and 1,3,5-triazines<sup>2,38</sup> have been widely applied as acceptor motifs in TADF materials due to their relatively weak electron-withdrawing character (the LUMO levels are 0.43 eV for 4H-1,2,4-triazole and -1.80 eV for 2,4,6triphenyl-1,3,5-triazine).<sup>39</sup> The combination of these two motifs into a single acceptor in the [1,2,4]-triazolo-[1,3,5]-triazine core (TTT) was synthesized by annulation of three triazole heterocycles onto a central triazine. This motif was first reported in 1911 by Hofmann and Erhardt, 40,41 while Wystrach and co-workers investigated the structure of a triamine TTT derivative in 1953. 42 Then, in 1961, Huisgen et al. were the first to synthesize the first TTT core substituted with phenyl groups by reacting cyanuric chloride with 5-phenyl tetrazole. 43 In 2008, Longo et al. substituted the TTT core with peripheral flexible alkyl chains, endowing the molecule with liquid crystalline character.44 The disc-like core with the peripheral aliphatic side chains led to luminescent and chargetransporting materials known to self-assemble into columnar superstructures driven by  $\pi$ -stacking.

Concurrent with our work, Pathak et al. recently reported the first two examples of emitters employing the TTT motif within a TADF emitter design: TTT-PXZ and TTT-DMAC (Fig. 2). 45 These emitters possess moderate photoluminescence quantum yields  $(\Phi_{\rm PL})$  of 39.5% and 21.4%, respectively, accompanied by short delayed fluorescence lifetimes of 4.2 µs and 4.6 µs, respectively. The emission of TTT-PXZ is expectedly considerably red-shifted given the stronger donor employed, with a  $\lambda_{PL}$  of 522 nm compared to 468 nm for TTT-DMAC. The improved device performance of TTT-PXZ is in part due to the smaller  $\Delta E_{\rm ST}$ 0.07 eV compared to 0.20 eV for TTT-DMAC. Unfortunately, the devices in this study were poor with EQEmax of 6.2 and 1.9% for the OLEDs with TTT-PXZ and TTT-DMAC, respectively, and associated severe efficiency roll-off.

Wang et al.46 expanded upon this work by improving the efficiency of the DMAC-based OLED (they named the TADF **Materials Advances** 

Fig. 2 Molecular structure and OLED performance metrics of known TTT emitters.

emitter as TTT-Ph-Ac in their work) to 9.73%. They also introduced two new TTT-based emitters: TTT-Ph-Cz and TTT-Ph-BAc (Fig. 2). TTT-Ph-Cz did not present a delayed component, reflected in the low  $EQE_{max}$ . TTT-Ph-BAc shows a delayed lifetime of 50.7  $\mu$ s, but the  $\Phi_{PL}$  is significantly lower at 32%,

which translated to a lower  $EQE_{max}$  for the OLED using this emitter than in the device using TTT-Ph-Ac.

Here, we introduce a new TTT-based emitter (Fig. 3), 3,4,5-3TCz-TTT, and cross-compare its performance with the recently reported compound 3DMAC-TTT (aka TTT-DMAC/TTT-Ph-Ac).

Fig. 3 Molecular structures of 3,4,5-3TCz-TTT and 3DMAC-TTT.

Synthesis

Paper

The synthesis for the two emitters is shown in Scheme 1. Starting from the 4-fluorobenzonitrile precursors, the DMAC and *tert*-butylcarbazole (TCz) donor groups were attached under basic S<sub>N</sub>Ar conditions. A 1,3-dipolar cycloaddition reaction with sodium azide and ammonium chloride in DMF at elevated temperatures led to the tetrazole precursors. Following a slightly modified procedure to the literature, <sup>43,47</sup> the TTT core was assembled through a condensation between trichlorotriazine and three equivalents of the tetrazole, yielding 3DMAC-TTT and 3,4,5-3TCz-TTT in very good yields. The identity and purity of the two emitters were determined by <sup>1</sup>H NMR, <sup>13</sup>C NMR, Elemental Analysis (EA) and High-Resolution Mass Spectrometry (HRMS).

### Theoretical calculations

To assess the optoelectronic properties of these two compounds, we performed a combination of Density Functional Theory (DFT) calculation and Time-Dependent DFT within the Tamm-Dancoff approximation (TDA-DFT<sup>48</sup>) in the gas phase using the PBE0 functional<sup>49</sup> and the 6-31G(d,p) basis set.<sup>50</sup> These calculations permitted an assessment of the energies and electron density distributions of the frontier molecular orbitals and the nature and energies of the lowest-lying singlet and triplet excited states (Fig. 4). The ground-state optimization of 3,4,5-3TCz-TTT did not converge due to the large size of the molecule given a large number of tert-butyl groups and we thus instead modelled a methyl-substituted analogue (3,4,5-3MCz-TIT), which we hypothesized would present the same electronic structure. The calculated HOMO and LUMO values for 3,4,5-**3MCz-TTT** are -5.39 eV and -2.14 eV, respectively, while those of 3DMAC-TTT are -5.39 eV and -1.93 eV, respectively. The different LUMO levels, despite the same acceptor, reflect the different extent of the conjugation present across the TTT core that is governed by the relative sterics of the donor groups. The  $\Delta E_{\rm ST}$  for 3,4,5-3MCz-TTT is 0.16 eV while for 3DMAC-TTT it is 0.01 eV, reflecting the more orthogonal conformation of the donor with respect to the phenylene bridge (DFT calculated torsions are 57.1° for 3,4,5-3TCz-TTT versus 99.5° for 3DMAC-TTT). Given the presence of three donor groups about the central TTT core, there is significant degeneracy of the low-lying triplet states.

For 3,4,5-3MCz-TTT we can also observe the presence of 5 other intermediate triplet states below the  $S_1$  level, where  $T_1$ ,  $T_2$ ,  $T_3$  and  $T_4$ ,  $T_5$ ,  $T_6$  each have almost identical energies. The large density of triplet states is a desirable trait that can facilitate RISC. <sup>51–54</sup> For 3DMAC-TTT, there are three degenerate triplet states present. The oscillator strength (f) values for the associated CT transitions from the  $S_1$  state vary markedly at 0.1 and 0.0014 for 3,4,5-3MCz-TTT and 3DMAC-TTT, respectively.

### Optoelectronic properties

The electrochemistry of the two emitters was studied in degassed acetonitrile (Fig. 5). The CV of **3,4,5-3TCz-TTT** presents three irreversible oxidation waves at 1.15 V, 1.34 V, 1.51 V  $\nu s$ . SCE (peaks from DPV), which are assigned to the sequential oxidation of the three TCz groups on each of the arms of the TTT core, values in line for the oxidation potentials of DTCz-containing compounds. <sup>55–59</sup> **3DMAC-TTT** shows one irreversible anodic oxidation wave at 0.91 V  $\nu s$ . SCE; the value reported in the literature is 1.15 V  $\nu s$ . Fc/Fc<sup>+</sup> in MeCN. <sup>45</sup> Both emitters present a weak reduction wave at -1.76 V and -1.90 V for **3,4,5-3TCz-TTT** and **3DMAC-TTT**, respectively, assigned to the TTT core. The corresponding HOMO and LUMO values for **3,4,5-3TCz-TTT** and **3DMAC-TTT** are -5.57 eV, -2.66 eV and -5.33 eV, -2.52 eV, respectively.

The UV-vis absorption spectrum in DCM (Fig. S23, ESI‡) for 3DMAC-TTT coincides with the one measured by Pathak *et al.* 45 Both molecules present a low intensity, low energy absorption band around 370 nm that is assigned to an ICT transition from the donor groups to the TTT acceptor. The band at 290 nm is a  $\pi$ - $\pi$ \* locally excited (LE) transition of the DMAC donor. There are four additional absorption bands for 3,4,5-3TCz-TTT which, by TD-DFT analysis, are assigned to different charge transfer states involving the carbazoles and the bridging phenyl. The absorption spectra for both compounds are not affected by changes in solvent polarity (Fig. 6). By contrast, the broad and unstructured emission red-shifts with increasing solvent polarity, indicating an emissive state that is charge-transfer in nature.

A comparison of the excitation and steady-state emission spectra of the two emitters in DCM is shown in Fig. 6a. Due to the use of stronger DMAC donors, the emission of **3DMAC-TTT** 

Scheme 1 Synthetic route to 3,4,5-3TCz-TTT and 3DMAC-TTT.

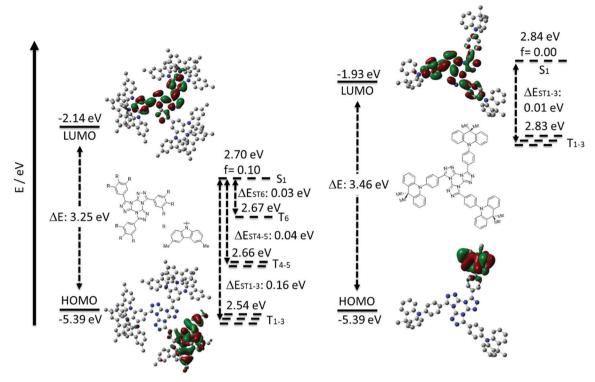


Fig. 4 HOMO and LUMO electron density distributions and energy levels (in eV), excited state energy levels (in eV) **3,4,5-3MCz-TTT** (right) and **3DMAC-TTT** (left) (PBE0/6-31G(d,p)).

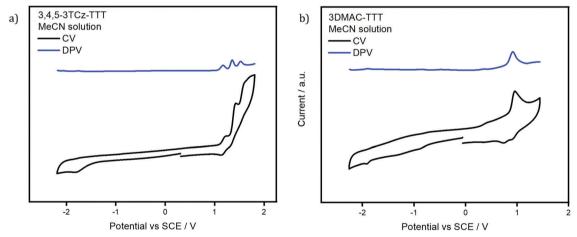


Fig. 5 Cyclic voltammogram (CV) and differential pulse voltammogram (DPV) of (a) **3,4,5-3TCz-TTT** and (b) **3DMAC-TTT** versus saturated calomel electrode (SCE) in MeCN and (scan rate: 0.1 V s<sup>-1</sup>).

with  $\lambda_{PL}$  at 551 nm is red-shifted compared to that of 3,4,5-3TCz-TTT ( $\lambda_{PL}=516$  nm); the excitation spectra mirror the corresponding absorption spectra. The time-resolved PL decays in degassed DCM are shown in Fig. 7b. Both compounds show mono-exponential decay kinetics (Fig. S25a, ESI‡) with prompt lifetimes,  $\tau_p$ , of 21.8 ns and 21.6 ns for 3,4,5-3TCz-TTT and 3DMAC-TTT, respectively, and delayed lifetimes,  $\tau_d$ , of 89.7  $\mu$ s for 3,4,5-3TCz-TTT and 3.6  $\mu$ s for 3DMAC-TTT. Upon exposure to air, the delayed emission was completely suppressed, which

is evidence of accessible triplet states that are quenched by oxygen (Tables 1 and 2).

We next identified CzSi as an appropriate host matrix for OLEDs based on its high triplet energy of 3.02 eV.<sup>2</sup> A doping concentration of 15 wt% was chosen as  $\Phi_{\rm PL}$  was highest (Table S6, ESI‡). Both compounds present sky-blue emission, at 485 nm and 475 nm (Fig. 8b) and high  $\Phi_{\rm PL}$  under N<sub>2</sub> of 80% and 79% for 3,4,5-3TCz-TTT and 3DMAC-TTT, respectively. Both emitters present prompt fluorescence (Fig. S25b, ESI‡),

Materials Advances **Paper** 

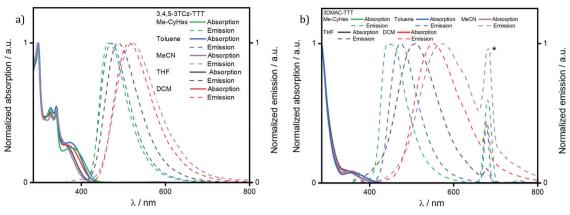


Fig. 6 Solvatochromism study of (a) 3,4,5-3TCz-TTT and (b) 3DMAC-TTT. ( $\lambda_{\rm exc}$  = 340 nm. \* = second harmonic band of the excitation source).

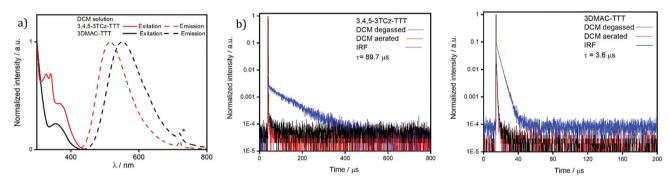


Fig. 7 (a) Emission and excitation spectra of **3,4,5-3TCz-TTT** and **3DMAC-TTT** (DCM  $10^{-5}$  M solutions,  $\lambda_{\text{exc}} = 360$  nm, \* = second harmonic of the excitation source). Time-resolved spectroscopy of degassed and aerated solutions of (b) 3,4,5-3TCz-TTT and (c) 3DMAC-TTT (DCM  $10^{-5}$  M solutions,  $\lambda_{\rm exc} = 378$  nm).

Table 1 Solution-state photophysical data

Compound	$\lambda_{\mathrm{PL}}{}^{a}/\mathrm{nm}$	$\Phi_{ m PL}{}^b/\%$	$E_{ m gap}^{c}/{ m eV}$	$\tau_{\mathrm{p}},~\tau_{\mathrm{d}}{}^d/\mathrm{ns};~\mu\mathrm{s}$	$\mathrm{HOMO}^e/\mathrm{eV}$	LUMO <sup>e</sup> /eV
3,4,5-3TCz-TTT	516	23	2.88	21.8; 89.7	-5.57	-2.66 $-2.52$
3DMAC-TTT	551	13	3.01	21.6; 3.6	-5.33	

<sup>a</sup> In degassed 10<sup>-5</sup> M DCM solution. <sup>b</sup> Obtained via the relative method (see ESI), quinine sulfate in H<sub>2</sub>SO<sub>4</sub> (aq) was used as the reference  $(\Phi_{\rm r} = 54.6\%)$ ,  $^{60}$   $\lambda_{\rm exc} = 360$  nm.  $^{c}$  Obtained from the intersection of the normalized absorption and emission spectra in DCM.  $^{d}$   $\tau_{\rm p}$  (prompt lifetime) and  $\tau_d$  (delayed lifetime) were obtained from the transient PL decay of 3,4,5-3TCz-TTT and 3DMAC-TTT in DCM. e HOMO and LUMO were obtained from the redox potentials from the DPV,  $E_{\text{HOMO/LUMO}} = -(E_{\text{ox/red}} + 4.8)$  where  $E_{\text{ox/red}}$  were taken from DPV scan corrected  $\nu s$ . Fc/Fc<sup>+</sup>.

with similar lifetimes reflecting tri-exponential decay kinetics, of 14.5 ns [ $\tau_1$  = 26.14 ns (12.35%),  $\tau_2$  = 12.97 ns (63.43%),  $\tau_3$  = 22.71 (24.22%)] for 3,4,5-3TCz-TTT and bi-exponential decay kinetics 11.9 ns  $[\tau_1 = 6.95 \text{ ns } (34.7\%), \tau_2 = 14.6 (65.3\%)]$  for 3-DMAC-TTT. Both compounds also show very long delayed PL with bi-exponential decay kinetics and lifetimes of 3.1 ms  $[\tau_1 = 1.17 \text{ ms } (54.27\%), \tau_2 = 5.30 \text{ ms } (45.73\%)] \text{ for } 3,4,5-3\text{TCz-TTT}$ and 4.7 ms  $[\tau_1 = 1.16 \text{ ms } (46.84\%), \tau_2 = 7.84 \text{ ms } (53.16\%)]$ for 3-DMAC-TTT (Fig. 8a). After investigation at longer time-windows compared to the reference paper, 45 we found that the two emitters present delayed fluorescence with lifetimes of the order of milliseconds, which is the result of

Table 2 Photophysical properties of 15 wt% 3,4,5-3TCz-TTT and 3DMAC-TTT doped films in CzSi

Compound	$\lambda_{\mathrm{PL}}{}^{a}/\mathrm{nm}$	$\Phi_{\mathrm{PL}}$ air; $\mathrm{N_2}^b/\%$	$\tau_{\rm p},~\tau_{\rm d}{}^c/{\rm ns};~{\rm ms}$	S <sub>1</sub> <sup>d</sup> /eV	T <sub>1</sub> <sup>e</sup> /eV	$\Delta E_{\rm ST}^f/{\rm eV}$
3,4,5-3TCz-TTT	485	47; 80	14.5; 3.1	2.91	2.70	0.21
3DMAC-TTT	475	28; 79	11.9; 4.7	2.92	2.65	0.27

<sup>&</sup>lt;sup>a</sup> Measured at room temperature. <sup>b</sup>  $\lambda_{\rm exc}$  = 340 nm. <sup>c</sup>  $\tau_{\rm p}$  (prompt lifetime) and  $\tau_{\rm d}$  (delayed lifetime) were obtained from the transient PL decay of doped film,  $\lambda_{\rm exc} = 378$  nm.  $^d$  S<sub>1</sub> was obtained from the onset of the prompt emission measured at 77 K,  $\lambda_{\rm exc} = 343$  nm.  $^e$  T<sub>1</sub> was obtained from the onset of the phosphorescence spectra measured at 77 K,  $\lambda_{\rm exc} = 343$  nm.  $^f$  Obtained from the difference of S<sub>1</sub> and T<sub>1</sub>.

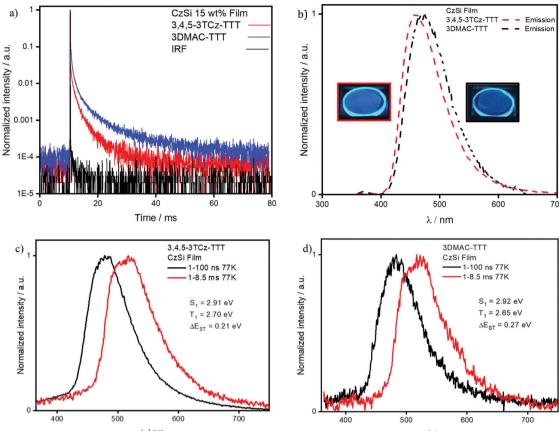


Fig. 8 (a) Time-resolved PL decay of **3,4,5-3TCz-TTT** and **3DMAC-TTT** in spin-coated CzSi film (15 wt%); (b) emission spectra of **3,4,5-3TCz-TTT** and **3DMAC-TTT** in CzSi doped film (15 wt%) ( $\lambda_{\text{exc}}$  = 340 nm); prompt fluorescence spectra at 77 K, and phosphorescence spectra at 77 K in 15 wt% CzSi of (c) **3,4,5-3TCz-TTT** and (d) **3DMAC-TTT**, phosphorescence obtained at 77 K at a delay of 1 ms for 8.5 ms,  $\lambda_{\text{exc}}$  = 343 nm.

suppression of non-radiative decay in the film, compared to solution and is reflective also in the enhanced photoluminescence quantum yield in the film. Temperaturedependent time-resolved decay curves (Fig. 9) reveal the increase in the intensity of the delayed emission with increasing temperature, which is a hallmark of compounds showing TADF. The  $\Delta E_{\rm ST}$  values of 0.21 eV for 3,4,5-3TCz-TTT and 0.27 eV for 3DMAC-TTT were determined from the onset of the prompt fluorescence and phosphorescence spectra on the CzSi films measured at 77 K (Fig. 8c and d). The latter result is slightly higher compared to that measured by Pathak *et al.*<sup>45</sup> ( $\Delta E_{\rm ST}$  = 0.20 eV) in the same host matrix.

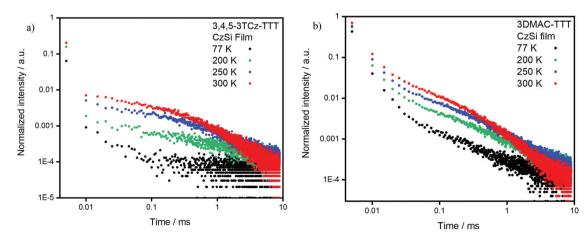


Fig. 9 (a) 3,4,5-3TCz-TTT and (b) 3DMAC-TTT delayed fluorescent decay data measured at different temperatures in 15 wt% doped CzSi film ( $\lambda_{\rm exc} = 378$  nm).

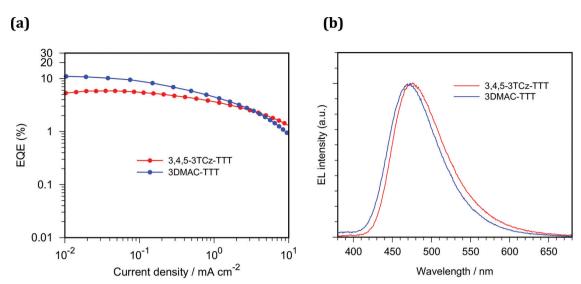


Fig. 10 (a) EQE-current-density characteristics and (b) EL spectra of OLEDs at the current density of 1 mA cm<sup>-2</sup>. Performances of OLEDs containing **3,4,5-3TCz-TTT** and **3DMAC-TTT** as emitters are displayed in red and blue, respectively.

## Solution-processed organic light-emitting diodes

We fabricated solution-processed OLEDs using 3,4,5-3TCz-TTT and 3DMAC-TTT as emitters. The device structures consist of: indium tin oxide (ITO)/poly(3,4-ethylenedioxythiophene)-poly (styrenesulfonate) (PEDOT:PSS)/poly(9-vinylcarbazole) (PVK)/15 wt% Emitter:CzSi/dibenzo[b,d]furan-2,8-diylbis(diphenylphosphine oxide) (PPF)/1,3,5-tris(1-phenyl-1*H*-benzo[*d*]imidazol-2-yl)benzene (TPBi)/lithium quinolin-8-olate (Liq)/Al. PEDOT:PSS, PVK and emitter layers were each deposited by spin-coating and other layers were vacuum-deposited. Here, we inserted the PVK layer to prevent electron leakage from the emitter layer into PEDOT:PSS layer and to confine triplet exciton within the emitter layer. 3,4,5-3TCz-TTT and 3DMAC-TTT exhibited blue emission with CIE coordinates of (0.17, 0.28) and (0.16, 0.23) and EQE<sub>max</sub> of 5.8% and 11.0%, respectively (Fig. 10). Compared to the reported work, 45 3DMAC-TTT-based OLED exhibited improved EQE<sub>max</sub> where the same CzSi host was used. This is probably because better exciton confinement within emitter layer was achieved by inserting PVK layer. Unfortunately, 3,4,5-3TCz-TTT exhibited lower EQE than that of 3DMAC-TTT, although the device structures are the same. As shown in Fig. S26 (ESI‡), different current density-voltage characteristics were observed, which can be attributed to the different carrier transporting and/or recombination behavior of emitters. From the PLQY and PL lifetime measurements, higher EQE for 3,4,5-3TCz-TTT OLEDs can be expected when the device structure is further optimized.

### Conclusions

We report two examples of TADF molecules using the TTT as a central acceptor. The two compounds emit in the blue and show a long-lived delayed fluorescence with similar lifetimes of 3.1 ms for the TCz-based emitter and 4.7 ms for the DMAC-containing

emitter in CzSi. High photoluminescence quantum yield values of 80% and 79% were observed in CzSi for 3,4,5-3TCz-TTT and 3DMAC-TTT, respectively. Both emitters were thus used in solution-processed OLEDs, showing EQE $_{\rm max}$  of 5.8% and 11% for the 3,4,5TCz-TTT and 3DMAC-TTT based-devices, respectively. The performance of the 3DMAC-TTT based is improved compared to previous reports owing to improved exciton management.

### Conflicts of interest

There are no conflicts to declare.

### Acknowledgements

The KIT authors are part of the Cluster 3D Matter Made To Order under Germany's Excellence Strategy (3DMM2O-EXC-2082-390761711), which has been funded by the German Research Foundation. This work was supported, in part, by the Cooperative Research Center (Sonderforschungsbereich SFB 1176) "Molecular Structuring of Soft Matter". This work was partly supported by JSPS KAKENHI grant no. 17H01231 and 17J09631, and by the International Joint Usage/Research Centre (iJURC) at the Institute for Chemical Research (ICR), Kyoto University, Japan. The St Andrews authors thank EU Horizon 2020 Grant Agreement No. 812872 (TADFlife) for funding. EZ-C acknowledges the International Collaborative Research Program of iJURC/ICR, Kyoto University (Grant No. 2020-37).

### References

1 M. A. Baldo, D. F. O'Brien, Y. You, A. Shoustikov, S. Sibley, M. E. Thompson and S. R. Forrest, *Nature*, 1998, **395**, 151–154.

- 2 M. Y. Wong and E. Zysman-Colman, *Adv. Mater.*, 2017, 29, 1605444.
- 3 X. Cai and S. J. Su, Adv. Funct. Mater., 2018, 28, 1-33.

**Materials Advances** 

- 4 Y. Tao, K. Yuan, T. Chen, P. Xu, H. Li, R. Chen, C. Zheng, L. Zhang and W. Huang, Adv. Mater., 2014, 26, 7931–7958.
- 5 Y. Liu, C. Li, Z. Ren, S. Yan and M. R. Bryce, *Nat. Rev. Mater.*, 2018, 3, 18020.
- 6 G. Méhes, H. Nomura, Q. Zhang, T. Nakagawa and C. Adachi, *Angew. Chem., Int. Ed.*, 2012, **51**, 11311–11315.
- 7 K. Kawasumi, T. Wu, T. Zhu, H. S. Chae, T. Van Voorhis, M. A. Baldo and T. M. Swager, *J. Am. Chem. Soc.*, 2015, 137, 11908–11911.
- 8 E. Spuling, N. Sharma, I. D. W. Samuel, E. Zysman-Colman and S. Bräse, *Chem. Commun.*, 2018, 54, 9278–9281.
- 9 A. Pershin, D. Hall, V. Lemaur, J. C. Sancho-Garcia,
  L. Muccioli, E. Zysman-Colman, D. Beljonne and
  Y. Olivier, *Nat. Commun.*, 2019, 10, 3-7.
- 10 H. Kaji, H. Suzuki, T. Fukushima, K. Shizu, K. Suzuki, S. Kubo, T. Komino, H. Oiwa, F. Suzuki, A. Wakamiya, Y. Murata and C. Adachi, *Nat. Commun.*, 2015, 6, 2–9.
- 11 D. H. Ahn, S. W. Kim, H. Lee, I. J. Ko, D. Karthik, J. Y. Lee and J. H. Kwon, *Nat. Photonics*, 2019, **13**, 540–546.
- 12 X. Wang, S. Wang, J. Lv, S. Shao, L. Wang, X. Jing and F. Wang, *Chem. Sci.*, 2019, **10**, 2915–2923.
- 13 J. Lu, Y. Zheng and J. Zhang, RSC Adv., 2015, 5, 18588-18592.
- 14 L. Gan, Z. Xu, Z. Wang, B. Li, W. Li, X. Cai, K. Liu, Q. Liang and S. J. Su, *Adv. Funct. Mater.*, 2019, **29**, 1808088.
- 15 X. Liang, Z. L. Tu and Y. X. Zheng, *Chem. Eur. J.*, 2019, 25, 5623–5642.
- 16 T. Hatakeyama, K. Shiren, K. Nakajima, S. Nomura, S. Nakatsuka, K. Kinoshita, J. Ni, Y. Ono and T. Ikuta, *Adv. Mater.*, 2016, **28**, 2777–2781.
- 17 X. Liang, Z. P. Yan, H. B. Han, Z. G. Wu, Y. X. Zheng, H. Meng, J. L. Zuo and W. Huang, *Angew. Chem., Int. Ed.*, 2018, 57, 11316–11320.
- 18 S. Madayanad Suresh, D. Hall, D. Beljonne, Y. Olivier and E. Zysman-Colman, *Adv. Funct. Mater.*, 2020, 1908677.
- 19 K. M. O. Nuyken, S. Jungermann, V. Wiederhirn and E. Bacher, *Monatsh. Chem.*, 2006, 137, 811–824.
- 20 M. Singh, H. M. Haverinen, P. Dhagat and G. E. Jabbour, *Adv. Mater.*, 2010, **22**, 673–685.
- 21 D. V. A. Arjona-Esteban, *Highly Efficient OLEDs: Materials Based on Thermally Activated Delayed Fluorescence*, 2018, pp. 543–572.
- 22 L. Duan, L. Hou, T. W. Lee, J. Qiao, D. Zhang, G. Dong, L. Wang and Y. Qiu, *J. Mater. Chem.*, 2010, **20**, 6392–6407.
- 23 T. Huang, W. Jiang and L. Duan, *J. Mater. Chem. C*, 2018, **6**, 5577–5596.
- 24 Y. Zou, S. Gong, G. Xie and C. Yang, *Adv. Opt. Mater.*, 2018, **6**, 1–25.
- 25 J. L. Wu, Y. T. Lee, C. T. Chen and C. T. Chen, *J. Chin. Chem. Soc.*, 2018, **65**, 87–106.
- 26 C. Li, Z. Ren, X. Sun, H. Li and S. Yan, *Macromolecules*, 2019, 52, 2296–2303.
- 27 J. Hu, Q. Li, S. Shao, L. Wang, X. Jing and F. Wang, *Adv. Opt. Mater.*, 2020, 1902100.

- 28 J. Rao, X. Liu, X. Li, L. Yang, L. Zhao, S. Wang, J. Ding and L. Wang, *Angew. Chem., Int. Ed.*, 2020, **59**, 1320–1326.
- 29 A. Menon, H. Dong, Z. I. Niazimbetova, L. J. Rothberg and M. E. Galvin, *Chem. Mater.*, 2002, **14**, 3668–3675.
- 30 Q. Wei, Z. Ge and B. Voit, *Macromol. Rapid Commun.*, 2019, 40, 1800570.
- 31 B. Zhang and Y. Cheng, Chem. Rec., 2019, 19, 1624-1643.
- 32 T. Jiang, Y. Liu, Z. Ren and S. Yan, *Polym. Chem.*, 2020, **11**, 1555–1571.
- 33 Y. J. Cho, K. S. Yook and J. Y. Lee, *Adv. Mater.*, 2014, **26**, 6642–6646.
- 34 X. L. Chen, J. H. Jia, R. Yu, J. Z. Liao, M. X. Yang and C. Z. Lu, *Angew. Chem., Int. Ed.*, 2017, **56**, 15006–15009.
- 35 Y. Wada, K. Shizu, S. Kubo, K. Suzuki, H. Tanaka, C. Adachi and H. Kaji, *Appl. Phys. Lett.*, 2015, **107**, 183303.
- 36 Y. Wada, S. Kubo and H. Kaji, *Adv. Mater.*, 2018, **30**, 1705641.
- 37 J. Lee, K. Shizu, H. Tanaka, H. Nomura, T. Yasuda and C. Adachi, *J. Mater. Chem. C*, 2013, **1**, 4599–4604.
- 38 Z. Yang, Z. Mao, Z. Xie, Y. Zhang, S. Liu, J. Zhao, J. Xu, Z. Chi and M. P. Aldred, *Chem. Soc. Rev.*, 2017, 46, 915–1016.
- 39 Y. Im, M. Kim, Y. J. Cho, J. A. Seo, K. S. Yook and J. Y. Lee, Chem. Mater., 2017, 29, 1946–1963.
- 40 O. E. K. A. Hofmann, Ber. Dtsch. Chem. Ges., 1911, 44, 2713–2717.
- 41 O. E. K. A. Hofmann, Ber. Dtsch. Chem. Ges., 1912, 45, 2731–2740.
- 42 D. W. Kaiser, G. A. Peters and V. P. Wystrach, *J. Org. Chem.*, 1953, **18**, 1610–1615.
- 43 R. Huisgen, Chem. Ber., 1961, 94, 1555-1562.
- 44 R. Cristiano, H. Gallardo, A. J. Bortoluzzi, I. H. Bechtold, C. E. M. Campos and R. L. Longo, *Chem. Commun.*, 2008, 5134–5136.
- 45 S. K. Pathak, Y. Xiang, M. Huang, T. Huang, X. Cao, H. Liu, G. Xie and C. Yang, RSC Adv., 2020, 10, 15523–15529.
- 46 S. Wang, X. Wang, K. H. Lee, S. Liu, J. Y. Lee, W. Zhu and Y. Wang, *Dyes Pigm.*, 2020, **182**, 108589.
- 47 A. G. Dal-Bó, G. G. L. Cisneros, R. Cercena, J. Mendes, L. M. da Silveira, E. Zapp, K. G. Domiciano, R. da Costa Duarte, F. S. Rodembusch and T. E. A. Frizon, *Dyes Pigm.*, 2016, **135**, 49–56.
- 48 S. Grimme, Chem. Phys. Lett., 1996, 259, 128-137.
- 49 V. Adamo and C. Barone, J. Chem. Phys., 1999, 110, 6158–6170.
- 50 T. H. Dunning, J. Chem. Phys., 1989, 90, 1007-1023.
- 51 P. L. Santos, J. S. Ward, P. Data, A. S. Batsanov, M. R. Bryce, F. B. Dias and A. P. Monkman, *J. Mater. Chem. C*, 2016, 4, 3815–3824.
- 52 T. Hosokai, H. Matsuzaki, H. Nakanotani, K. Tokumaru, T. Tsutsui, A. Furube, K. Nasu, H. Nomura, M. Yahiro and C. Adachi, *Sci. Adv.*, 2017, 3, 1603282.
- 53 H. Noda, H. Nakanotani and C. Adachi, *Sci. Adv.*, 2018, **4**, 1–8.
- 54 P. K. Samanta, D. Kim, V. Coropceanu and J. L. Brédas, J. Am. Chem. Soc., 2017, 139, 4042–4051.

**Paper** 

55 N. Sharma, E. Spuling, C. M. Mattern, W. Li, O. Fuhr,

- Y. Tsuchiya, C. Adachi, S. Bräse, I. D. W. Samuel and E. Zysman-Colman, Chem. Sci., 2019, 10, 6689-6696.
- 56 D. Chen, P. Rajamalli, F. Tenopala-Carmona, C. L. Carpenter-Warren, D. B. Cordes, C. M. Keum, A. M. Z. Slawin, M. C. Gather and E. Zysman-Colman, Adv. Opt. Mater., 2020, 8, 1901283.
- 57 P. L. Dos Santos, D. Chen, P. Rajamalli, T. Matulaitis, D. B. Cordes, A. M. Z. Slawin, D. Jacquemin, E. Zysman-Colman and I. D. W. Samuel, ACS Appl. Mater. Interfaces, 2019, 11, 45171-45179.
- 58 Z. Li, W. Li, C. Keum, E. Archer, B. Zhao, A. M. Z. Slawin, W. Huang, M. C. Gather, I. D. W. Samuel and E. Zysman-Colman, J. Phys. Chem. C, 2019, 123, 24772-24785.
- 59 M. Y. Wong, S. Krotkus, G. Copley, W. Li, C. Murawski, D. Hall, G. J. Hedley, M. Jaricot, D. B. Cordes, A. M. Z. Slawin, Y. Olivier, D. Beljonne, L. Muccioli, M. Moral, J. C. Sancho-Garcia, M. C. Gather, I. D. W. Samuel and E. Zysman-Colman, ACS Appl. Mater. Interfaces, 2018, 10, 33360-33372.
- 60 G. A. Crosby and J. N. Demas, J. Phys. Chem., 1971, 75, 991-1024.