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Efficient synthesis of *P*-chiral biaryl phosphonates by stereoselective intramolecular cyclization[†]

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A series of *P*-chiral biaryl phosphonates were efficiently synthesized from diaryl 2-bromo arylphosphonates in high yields (up to 92%) and good enantioselectivities (up to 88% ee) through a palladium-catalyzed asymmetric cyclization with a novel *P*-chiral biaryl monophosphorus ligand. The *P*-chiral biaryl phosphonate can be rapidly transformed to both antipodes of a *P*-chiral dialkyl biaryl monophosphorus structure. The method provides a convenient access to various *P*-chiral biaryl monophosphines.

Since Knowles first introduced P-chiral phosphines CAMP and DIPAMP for rhodium-catalyzed asymmetric hydrogenation almost half a century ago,¹ P-chiral phosphorus ligands have played significant roles in the rapid development of the asymmetric catalysis area.² Efficient construction of P-chiral phosphorus compounds has become a hot subject of research.³ Various efficient methods were developed including chemical resolutions,⁴ asymmetric synthesis by using chiral auxiliaries or reagents,⁵ and recently catalytic asymmetric methods.⁶ Because of the increasing applications of P-chiral biaryl monophosphorus ligands in organic synthesis,⁷ we propose to develop a general and efficient synthetic method for P-chiral biaryl monophosphorus ligands from a P-chiral biaryl phosphonate A through two consecutive stereospecific substitutions at the phosphorus center (Fig. 1). The challenge is whether the P-chiral biaryl phosphonate A can be efficiently synthesized from the readily accessible ortho-bromo arylphosphonate B through an enantioselective palladium-catalyzed desymmetric

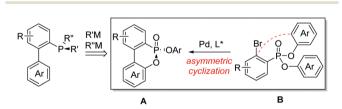


Fig. 1 A new strategy for the synthesis of *P*-chiral biaryl monophosphorus ligands.

State Key Laboratory of Bio-Organic and Natural Products Chemistry, Shanghai Institute of Organic Chemistry, 345 Ling Ling Rd, Shanghai 200032, P. R. China. E-mail: tangwenjun@sioc.ac.cn intramolecular cyclization.⁸ Herein we disclose our study on this asymmetric cyclization and its transformations toward *P*-chiral biaryl monophosphorus ligands.

We chose diphenyl(2-bromophenyl)phosphonate (1a) as the substrate for study. As shown in Table 1, the palladium-catalyzed asymmetric cyclization of 1a proceeded smoothly at 80 °C in toluene with KOAc as the base to afford the cyclization product 1b in excellent yields in the presence of a P-chiral monophosphorus ligand. Among the several P-chiral biaryl monophosphorus ligands employed (entries 1-5),⁹ the newly developed ligand L3 with a tetrahydrobenzodifuran moiety provided an excellent yield (93%) and a good enantioselectivity (77% ee) with potassium acetate as the base. Apparently, the substituents on the low aryl ring of the P-chiral biaryl ligands exert significant influence on the enantioselectivity. Moderate ees were achieved with acyclic or cyclic alkoxy moieties such as methoxy substituents, furans and dioxolanes (entries 1, 3 and 4). In contrast, AntPhos (L5) proved to be ineffective (entry 5). Ligand L2 with a methyl group at the 2 position of the oxophosphole ring also provided a diminished ee (entry 2). When L3 was employed for further optimization, a dramatic base effect was observed. A more hindered base KOPiv afforded an inferior yield and ee value (entry 6). Meanwhile, 1-AdCOOK could provide comparable enantioselectivity to KOAc but with lower yield (entry 7). When PhCOOK was employed as a base, a higher ee value (88%) was achieved, albeit with a low yield (34%, entry 8). The low yield could be largely due to its relatively weak basicity. We thus employed PhCH₂COOK as the base. Although the cyclization yield was comparable to that with KOAc, its enantioselectivity was slightly inferior (entry 9). With Ph₂CHCOOK as the base, we obtained a similar yield to that with KOAc, but with a slightly better ee value (entry 10). When the reaction temperature was reduced to 70 °C, the ee value of 1b was improved to 82% (entry 11). Change of the solvent to cyclohexane, 1,4-dioxane, THF, and 1,2-dichloroethane (DCE) did not enhance the enantioselectivity (entries



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[†]Electronic supplementary information (ESI) available: Detailed procedures of cross-coupling reactions, characterization data, and spectra. CCDC 1062715. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/ c5q000142k

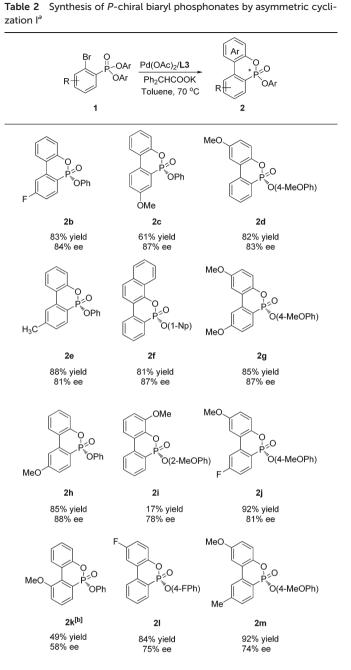
Table 1 Intramolecular asymmetric cyclization of diphenyl(2-bromophenyl)phosphonate (1a)

Br O P(OPh) ₂ Pd(OAc) ₂ /L* Base, Toluene P [*] OPh						
	1a		1b			
MeO + OMe + O + O + O + O + O + O + O + O + O +						
						-
Entries ^a	L*	Base	Solvent	$T(^{\circ}C)$	$\operatorname{Yield}^{b}(\%)$	% ee ^c
1	L1	KOAc	Toluene	80	91	71
2	L2	KOAc	Toluene	80	91	16
3	L3	KOAc	Toluene	80	93	77
4	L4	KOAc	Toluene	80	93	66
5	L5	KOAc	Toluene	80	81	1
6	L3	KOPiv	Toluene	80	70	70
7	L3	1-AdCOOK	Toluene	80	76	77
8	L3	PhCOOK	Toluene	80	34	83
9	L3	PhCH ₂ COOK	Toluene	80	94	75
10	L3	Ph ₂ CHCOOK	Toluene	80	93	78
11	L3	Ph ₂ CHCOOK	Toluene	70	70	82
12	L3	Ph ₂ CHCOOK	CyHex	70	88	76
13	L3	Ph ₂ CHCOOK	Dioxane	70	26	37
14	L3	Ph ₂ CHCOOK	THF	70	19	74
15	L3	Ph ₂ CHCOOK	DCE	70	97	74
16^d	L3	Ph ₂ CHCOOK	Toluene	70	83	88

^{*a*} Unless otherwise specified, the reactions were performed at the designated reaction temperature in organic solvent (1 mL) with aryl bromide (0.2 mmol) under nitrogen for 24 h in the presence of Pd(OAc)₂ (5 mol%), L* (6 mol%), and base (0.3 mmol), the absolute configuration of **1b** was assigned by analogy according to the X-ray crystal structure of **2f**. ^{*b*} Isolated yield. ^{*c*} ee values were determined by chiral HPLC on a chiralcel AD-H column. ^{*d*} Pd(OAc)₂ (4 mol%), L3 (8 mol%).

12–15). When the mole ratio of Pd/L3 increased from 1/1.2 to 1/2 (4 mol% Pd), a better ee value (88%) was achieved along with an acceptable yield (entry 16). Other bases were also tested, but no further improvement of the ee value was achieved.¹⁰

We then investigated the substrate scope of this asymmetric cyclization under optimized conditions (Table 2). Thus, a series of substituted diphenyl *ortho*-bromo phenylphosphonates (**1b**, **e**, **h**, **c**) were successfully cyclized to provide the corresponding *P*-chiral phosphonates in high yields and good enantioselectivities with **L3** as the ligand. Substituents such as methyl, methoxy, and fluoro groups at the *meta-* or *para*-position were well tolerated. A substrate with a methoxy substituent adjacent to the bromine atom **1k** provided the corresponding cyclization product **2k** in only 27% ee and 52% yield. However, an improved ee (58%) value was achieved when **L6** was employed as the ligand. In addition, various di(substituted aryl)*ortho*-bromo phenylphosphonates were also applicable to provide the corresponding cyclization products (**2d**, **2f-g**, **2j**,



^{*a*} Unless otherwise specified, the reactions were performed in toluene (1 mL) at 70 °C under nitrogen for 24 h with aryl bromide (0.2 mmol), Pd(OAc)₂ (4 mol%), L3 (8 mol%), and Ph₂CHCOOK (0.3 mmol); isolated yields; ee values were determined by chiral HPLC. The absolute configuration of **2f** was determined by X-ray crystallography, others were assigned by analogy. ^{*b*} L6 as a ligand.

2l-2m) in good yields and enantioselectivity. Di(*ortho*-methoxyphenyl)*ortho*-bromo phenylphosphonate (**1i**) also provided a decent ee value (78%) albeit with a low yield of **2i**. The absolute configuration of **2f** was determined as R by X-ray crystallographic analysis.¹¹

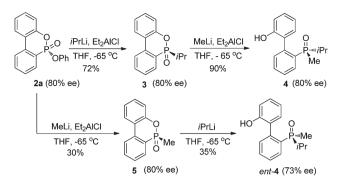
zation II^a L1: 93% yield, 21% ee L2: 90% yield, 20% ee -0 =0 L3: 85% yield, 30% ee L5: 94% yield, 77% ee 0(4-FPh) L6: 83% yield, 88% ee 2n 2r 64% yield 87% ee Me MeO 0 -0 0 O(2-MePh) O(4-MeOPh) ′О(1-Np) 20 2p 2a 88% yield 62% yield 65% yield 87% ee 75% ee 76% ee

 Table 3
 Synthesis of P-chiral biaryl phosphonates by asymmetric cycli

^{*a*} Unless otherwise specified, the reactions were performed for 24 h under nitrogen at 70 °C in toluene (1 mL) with naphthyl bromide (0.2 mmol), $Pd(OAc)_2$ (5 mol%), L6 (6 mol%), and KOAc (0.3 mmol); isolated yields; ee values were determined by chiral HPLC; the absolute configurations were assigned by analogy.

Interestingly, when diphenyl (1-bromo-2-naphthyl)phosphonate (1n) was employed for cyclization under similar reaction conditions, the cyclization product 2n was formed in only 30% ee and in 85% yield. In order to obtain a better enantioselectivity, we further screened the P-chiral biaryl monophosphorus ligands in our laboratory. As can be seen in Table 3, ligands L1-3 all provided very poor enantioselectivities. To our surprise, AntPhos (L5) formed the cyclization product in 77% ee. L6 with a methyl substituent on the oxophosphole ring deriving from L5 afforded the cyclization product in 88% ee and 83% yield. It was thus chosen as the ligand for this series of substrates. By using these conditions, various di(substitutedaryl) (1-bromo-2-naphthyl)phosphonates (10-r) were also subjected to the cyclization and the corresponding cyclization products (2o-r) were formed in good yields and high enantioselectivities. The di(*para*-methoxy)phosphonate substrate **1p** and di(1-naphthyl)phosphonate substrate 1q afforded the corresponding products 2p and 2q in slightly lower ee values, respectively.

The *P*-chiral phosphonates **2a–q** can be envisioned as useful precursors for a variety of *P*-chiral biaryl phosphorus ligands. Because both aryloxy substituents of the phosphonate can be displaced stereospecifically by different alkyl lithium or Grignard reagents sequentially, both antipodes of a *P*-chiral biaryl structure could be prepared from a single *P*-chiral phosphonate product. In order to demonstrate this utility (Scheme 1), the *P*-chiral biaryl phosphonate **2a** was treated first with isopropyllithium in the presence of Et₂AlCl to form isopropyl substituted product **3** without erosion of enantioselectivity. Subsequent treatment of **3** with methyllithium stereospecifically provided *P*-chiral dialkyl biarylphosphine



Scheme 1 Stereospecific transformation of *P*-chiral phosphonate 2a to *P*-chiral biaryl phosphine oxides 4 and *ent*-4.

oxide **4**.¹² Alternatively, treatment of **2a** (80% ee) with methyllithium and isopropyllithium sequentially provided *ent*-**4** in an unoptimized yield with light erosion of the ee value (73% ee). Stereospecific reduction of **4** and *ent*-**4** with a reported procedure¹³ could provide both antipodes of a *P*-chiral dialkyl biaryl phosphine, respectively.

In summary, we have developed an efficient Pd-catalyzed desymmetric intramolecular cyclization of diaryl *ortho*-bromo aryl phosphonates that have led to a series of *P*-chiral biaryl phosphonates in high yields (up to 92%) and good enantio-selectivities (up to 88% ee) under very mild conditions. The *P*-chiral biaryl phosphonates have been demonstrated as excellent precursors to both antipodes of *P*-chiral dialkyl biaryl monophosphines. This method has provided convenient access to various *P*-chiral biaryl monophosphine ligands, which should have increasing applications in the area of asymmetric catalysis.

Acknowledgements

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Notes and references

- (a) W. S. Knowles and M. J. Sabacky, *Chem. Commun.*, 1968, 1445;
 (b) W. S. Knowles, M. J. Sabacky, B. D. Vineyard and D. J. Weinkauff, *J. Am. Chem. Soc.*, 1975, **97**, 2567.
- 2 (a) P. C. J. Kamer and P. W. N. M. Van Leeuween, *Phosphorus(m) Ligands in homogeneous Catalysis: Design and Synthesis*, Wiley & Sons, West Sussex, 2012; (b) W. Tang and X. Zhang, *Chem. Rev.*, 2003, **103**, 3029; (c) *P-Stereogenic Ligands in Enantioselective Catalysis*, ed. A. Grabulosa, RSC, Cambridge, 2011.
- 3 For reviews on the synthesis of *P*-chiral phosphines, see:
 (a) K. M. Pietrusiewicz and M. Zablocka, *Chem. Rev.*, 1994,
 94, 1375; (b) A. Grabulosa, J. Granell and G. Muller, *Coord. Chem. Rev.*, 2007, 251, 25; (c) J. S. Harvey and

V. Gouverneur, *Chem. Commun.*, 2010, **46**, 7477; (*d*) O. I. Kolodiazhnyi, *Tetrahedron: Asymmetry*, 2012, **23**, 1.

- 4 For selective examples, see: (a) K. Tani, L. D. Brown, J. Ahmed, J. A. Ibers, M. Yokota, A. Nakamura and S. Otsuka, J. Am. Chem. Soc., 1977, 99, 7876;
 (b) N. K. Roberta and S. B. Wild, J. Am. Chem. Soc., 1979, 101, 6254; (c) T. Imamoto, K. V. L. Crépy and K. Katagiri, Tetrahedron: Asymmetry, 2004, 15, 2213; (d) D. Liu and X. Zhang, Eur. J. Org. Chem., 2005, 646.
- 5 For selective examples, see: (a) O. Korpiun and K. Mislow, J. Am. Chem. Soc., 1967, 89, 4784; (b) D. Gatineau, L. Giordano and G. Buono, J. Am. Chem. Soc., 2011, 133, 10728; (c) O. Berger and J.-L. Montchamp, Angew. Chem., Int. Ed., 2013, 52, 11377; (d) S. Jugé, M. Stephan, J. A. Laffitte and J. P. Genet, Tetrahedron Lett., 1990, 31, 6357; (e) Z. S. Han, N. Goyal, M. A. Herbage, J. D. Sieber, B. Qu, Y. Xu, Z. Li, J. T. Reeves, J.-N. Desrosiers, S. Ma, N. Grinberg, H. Lee, H. P. R. Mangunuru, Y. Zhang, D. Krishnamurthy, B. Z. Lu, J. J. Song, G. Wang and C. H. Senanayake, J. Am. Chem. Soc., 2013, 135, 2474.
- 6 For selective examples, see: (a) J. R. Moncarz, N. F. Laritcheva and D. S. Glueck, J. Am. Chem. Soc., 2002, 124, 13356; (b) V. S. Chan, I. C. Stewart, R. G. Bergman and F. D. Toste, J. Am. Chem. Soc., 2006, 128, 2786; (c) C. Scriban and D. S. Glueck, J. Am. Chem. Soc., 2006, 128, 2788; (d) N. F. Blank, J. R. Moncarz, T. J. Brunker, C. Scriban, B. J. Anderson, O. Amir, D. S. Glueck, L. N. Zakharov, J. A. Golen, C. D. Incarvito and A. L. Rheingold, J. Am. Chem. Soc., 2007, 129, 6847; (e) V. S. Chan, R. G. Bergman and F. D. Toste, J. Am. Chem. Soc., 2007, 129, 15122; (f) C. Scriban, D. S. Glueck, J. A. Golen and A. L. Rheingold, Organometallics, 2007, 26, 1788; (g) B. J. Anderson, M. A. Guino-o, D. S. Glueck, J. A. Golen, A. G. DiPasquale, L. M. Liable-Sands and A. L. Rheingold, Org. Lett., 2008, 10, 4425; (h) V. S. Chan, M. Chiu, R. G. Bergman and F. D. Toste, J. Am. Chem. Soc., 2009, 131, 6021; (i) T. W. Chapp, D. S. Glueck, J. A. Golen, C. E. Moore and A. L. Rheingold, Organometallics, 2010, 29, 378; (j) C. Li, W.-X. Li, S. Xu and W.-L. Duan, Chin. J. Org. Chem., 2013, 33, 799; (k) Y. Huang, Y. Li, P.-H. Leung and T. Hayashi, J. Am. Chem. Soc., 2014, 136, 4865; (l) C. Li, B.-L. Bian, S. Xu and W.-L. Duan, Org. Chem. Front., 2014, 1, 541; (m) Z.-J. Du, J. Guan, G.-J. Wu, P. Xu, L.-X. Gao and F.-S. Han, J. Am. Chem. Soc., 2015, 137, 632.

- 7 (a) J. Yin and S. L. Buchwald, J. Am. Chem. Soc., 2000, 122, 12051; (b) X. Shen, G. O. Jones, D. A. Watson, B. Bhayana and S. L. Buchwald, J. Am. Chem. Soc., 2010, 132, 11278; (c) W. Tang, N. D. Patel, G. Xu, X. Xu, J. Savoie, S. Ma, M.-H. Hao, S. Keshipeddy, A. G. Capacci, X. Wei, Y. Zhang, J. J. Gao, W. Li, S. Rodriguez, B. Z. Lu, N. K. Yee and C. H. Senanayake, Org. Lett., 2012, 14, 2258; (d) K. Li, N. Hu, R. Luo, W. Yuan and W. Tang, J. Org. Chem., 2013, 78, 6350; (e) G. Xu, W. Fu, G. Liu, C. H. Senanayake and W. Tang, J. Am. Chem. Soc., 2014, 136, 570; (f) K. Du, P. Guo, Y. Chen, Z. Cao, Z. Wang and W. Tang, Angew. Chem., Int. Ed., 2015, 54, 3033.
- 8 During preparation of the manuscript, two examples of palladium-catalyzed enantioselective C-H arylation for the synthesis of *P*-stereogenic phosphinic amides were reported: (a) Z.-Q. Lin, W.-Z. Wang, S.-B. Yan and W.-L. Duan, *Angew. Chem., Int. Ed.*, 2015, 54, 6265; (b) L. Liu, A.-A. Zhang, Y. Wang, F. Zhang, Z. Zuo, W.-X. Zhao, C.-L. Feng and W. Ma, *Org. Lett.*, 2015, 17, 2046.
- 9 For other applications of ligands L1-2 and L5-6 in catalysis, see ref. 7c-f and: (a) W. Tang, A. G. Capacci, X. Wei, W. Li, A. White, N. D. Patel, J. Savoie, J. J. Gao, S. Rodriguez, B. Qu, N. Haddad, B. Z. Lu, D. Krishnamurthy, N. K. Yee and C. H. Senanayake, Angew. Chem., Int. Ed., 2010, 49, 5879; (b) W. Tang, S. Keshipeddy, Y. Zhang, X. Wei, J. Savoie, N. D. Patel, N. K. Yee and C. H. Senanayake, Org. Lett., 2011, 13, 1366; (c) Q. Zhao, C. Li, C. H. Senanayake and W. Tang, Chem. Eur. J., 2013, 19, 2261; (d) C. Li, G. Xiao, Q. Zhao, H. Liu, T. Wang and W. Tang, Org. Chem. Front., 2014, 1, 225; (e) G. Xu, Q. Zhao and W. Tang, Chin. J. Org. Chem., 2014, 34, 1919.
- 10 Ph₃CCOOK, Ph₂CHCOOCs, and potassium 2-(naphthalen-1-yl)acetate were also tested as bases and the highest ee value was 85%.
- 11 CCDC 1062715 contains the supplementary crystallographic data for this paper.
- 12 ¹H NMR and ³¹P NMR spectra showed two atropisomers in a ratio of 2.2/1 at 25 °C.
- 13 For examples of reduction of chiral phosphine oxides, see ref. 5*c*-*e* and the following literatures: (*a*) T. Imamoto, S.-i. Kikuchi, T. Miura and Y. Wada, *Org. Lett.*, 2001, 3, 87; (*b*) K. V. Rajendran and D. G. Gilheany, *Chem. Commun.*, 2012, 48, 817.