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Recent progress in metal complexes functionalized nanomaterials for photodynamic therapy

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Metal complexes have shown promise as photosensitizers for cancer diagnosis and therapeutics. However, the vast majority of metal photosensitizers are not ideal and associated with several limitations including pharmacokinetic limitations, off-target toxicity, fast systemic clearance, poor membrane permeability, and hypoxic tumour microenvironments. Metal complex functionalized nanomaterials have the potential to construct multifunctional systems, which not only overcome the above defects of metal complexes but are also conducive to modulating the tumour microenvironment (TME) and employing combination therapies to boost photodynamic therapy (PDT) efficacy. In this review, we first introduce the current challenges of photodynamic therapy and summarize the recent research strategies (such as metal coordination bonds, self-assembly, π - π stacking, physisorption, and so on) used for preparing metal complexes functionalized nanomaterials in the application of PDT.

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1. Introduction

About 100 years ago, Tappeiner determined that the phenomenon of PS combined with light-killing cells was oxygen dependent, and proposed the concept of the photodynamic effect for the first time to describe this oxygen-dependent photosensitization reaction.^{1,2} The emergence of this concept has attracted extensive attention from scientists and researchers and has greatly promoted the development of PDT.³ Nowadays, PDT has been widely used in the treatment of many diseases^{4–8} including malignant tumours^{9,10} due to its minimal invasiveness, temporospatial specificity, and controllable systemic toxicity. Typically, the mechanism of photodynamic action is complex and is not fully understood. With light stimulation, the PS molecule transits to the triplet excited state *via* intersystem crossing, entering into a photochemical reaction of type I or II PDT (Fig. 1(B)). In type I, excited PS molecules interact directly with TME-related biomolecules, forming intermediate radical products that then react with oxygen, which leads to the

formation of various highly active substances, primarily active forms of oxygen, for further redox reactions. In this case, peroxide radicals ($\cdot\text{OOH}$), superoxide anions ($\cdot\text{O}_2^-$), and hydroxyl radicals ($\cdot\text{OH}$) are formed, lipid peroxidation is activated, and cell membranes are damaged which interferes with their functions. In type II, excited PS molecules react first with oxygen, converting it into highly active singlet oxygen ($^1\text{O}_2$). It interacts with the proteins, nucleic acids, and lipids of cell membranes, causing their death by necrosis or apoptosis.¹¹

Photofrin was the first PS to receive approval all over the world for the treatment of cancer. Subsequently, 5-aminolevulinic acid (Levulan), Temoporfin (Foscan), Verteporfin (Visudyne), Telaporfin (Foscan), LUZ111 (Redaporfin) *etc.* appeared successively and acted as PSs for use in clinical PDT. Owing to their shared structural features, the majority of these PSs have similar drawbacks, including tedious synthesis/purification, poor water solubility and photostability, poor cancer selectivity and slow body clearance causing photosensitivity,^{12,13} which significantly reduces ROS generation efficiency during PDT processes. Thus, there is an urgent need for the improvement of these compounds and the development of novel PS scaffolds.

With the unremitting efforts of researchers, many metal complexes have started to be reported as PSs in various treatments, including PDT.¹⁴ Studies have found that many metal complexes possess excellent properties in the fields of therapeutics and bioimaging, including structural flexibility, excellent photostability, large Stokes shifts, long-lived emission,

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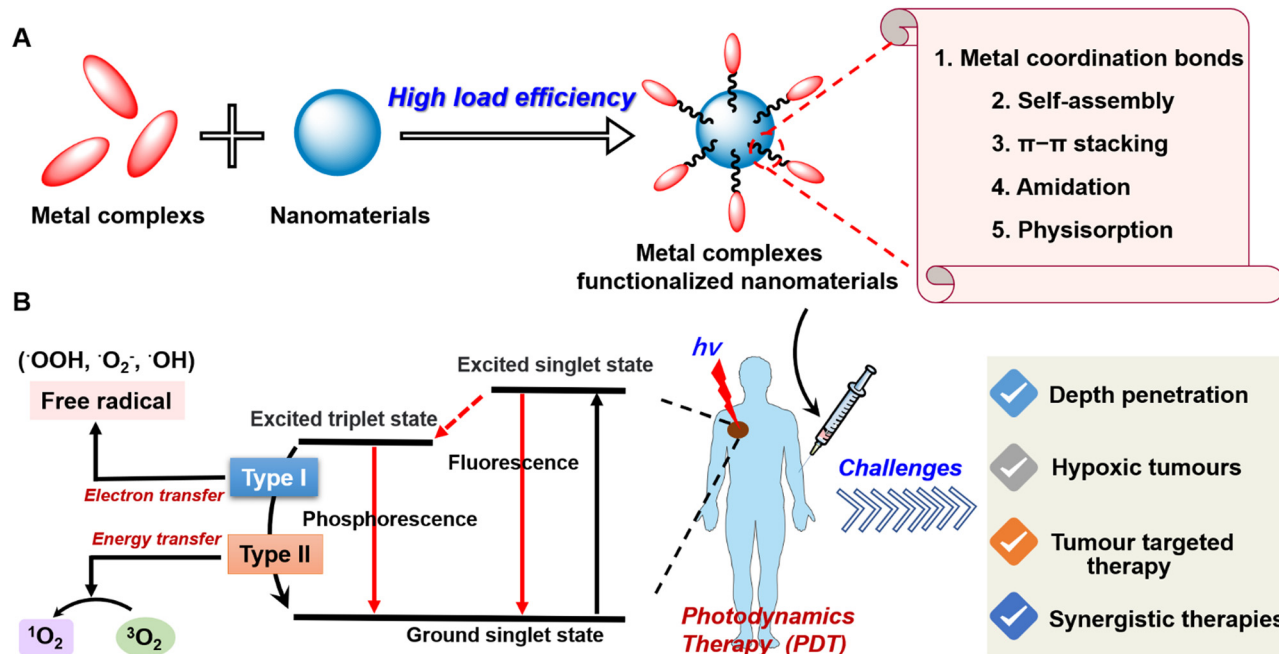


Fig. 1 (A) The framework of metal complexes functionalized nanomaterials. (B) The photophysical mechanism and existing challenges of photodynamic therapy.

high ¹O₂ yield, large two-/multi-photon absorption, remarkable cellular uptake, and organelle-targeting properties, attracting much attention.^{15–19} For instance, commonly used metal complexes-based PSs in clinical development PDT include metal (Sn, Lu, Pd)-containing tetrapyrrolic PSs^{20–23} and the ruthenium polypyridine complex TLD-1433.²⁴ Moreover, most organic molecules (methylene blue, protoporphyrin IX, chlorin e6 *etc.*) have been reported as PSs, which are advantageous for PDT due to their generating a high ROS yield. However, the therapeutic effect of organic PSs for PDT is much lower than that of metal PSs, which is mainly attributed to the complex synthesis, low yield, poor structure stabilization and short fluorescence lifetime of organic PSs.^{25,26} In a word, these PSs are hydrophobic, lack targeting, and have low bioavailability, so PSs *in vivo* are prone to self-aggregation, fluorescence quenching, fast systemic clearance and poor membrane permeability, and even because of the hypoxic tumour microenvironment, they usually fail to achieve the desired antitumor effect of PDT.^{27–29} In recent years, with the development of nanotechnology, supramolecular nanoparticles,^{30–32} liposomes,^{33–35} metal-organic frameworks,^{26,36} and 2D nanosheets,^{37–39} have been widely used for improving the safety and effectiveness of PSs, capitalizing on their advantages of high biocompatibility, effective targeting, controlled release, and improved pharmacokinetic effect. Based on this, metal complexes functionalized nanomaterials can possess the properties of metal complexes and nanoparticles or new functions. In particular, they have the enormous advantage of increasing solubility, prolonged circulation, accumulation in the tumour site, overcoming tumour hypoxia, enhancing ROS yield, minimizing off-target toxicity to healthy cells and increasing therapeutic efficacy. Hence, it is increasingly evident that new nanomedicines for PDT should

overcome the limitations of traditional metal PSs, such as poor water solubility and low efficacy in hypoxic tumours, using novel strategies.

In this review, we first elucidate the challenges of photodynamic therapy and related solutions to the strategy. Moreover, novel design strategies such as metal coordination bonds, self-assembly, π - π stacking, physisorption, and so on to construct metal complexes functionalized nanomaterials for enhancing PDT are also introduced in this review (Fig. 1). Finally, the expected future direction of research into metal complexes functionalized nanomaterials is also discussed.

2. Challenges of photodynamic therapy

Compared to other treatment methods, PDT presents several advantages over conventional therapies because it enables the selective destruction of tumour tissues. This promising therapy combines three components: photosensitizers, light, and oxygen; however, the lack of synergy between these is generally considered to be the main factor causing PDT inefficiency. Compared to existing standard cancer treatments such as surgery or chemotherapy, PDT faces the following challenges, which require unique strategies to overcome.

2.1 Light-penetration depth: short wavelength excitation limits the depth of tissue penetration and damages healthy tissue

Light penetration depth into the skin is one of the preconditions for a PDT procedure. Ideally, a PDT treatment should only cause a therapeutic effect towards cancer tissue while not

damaging healthy, underlying tissue. Thus, the wavelength used in a medical procedure is chosen based on the tumour's depth and shape. However, one of the disadvantages of PDT is the short penetration depth of incident light. Generally, the UV-visible light used for activating the photosensitisers has a penetration of about 1–5 mm, and the NIR light-based lasers can penetrate up to 1 cm tissue depth.⁴⁰ As such, some of the specific nanotechnology, such as upconversion nanotechnology has been developed for converting the lower energy NIR light to UV-vis light to activate the loaded photosensitisers and dyes, including metal complexes for a higher penetration depth.⁴¹ Even so, the short penetration depth of the NIR light hindered the application of PDT in deep-seated tumours.

To overcome the limited penetration depth in traditional PDT systems, several strategies have been applied. (1) Light transducers can be used as energy amplifiers, which can absorb light in the NIR region and emit it in the visible region, thereby activating PSs in the vicinity.⁴² Such examples can be found in two-photon light in addition to up-conversion light as mentioned above.^{43,44} (2) Bioluminescence resonance energy transfer (BRET) systems combining bioluminescent luciferase and quantum dots (QDs) could allow for *in situ* production of light and internal activation of the PSs because of the unique optical properties of QDs.⁴⁵ (3) X-ray as a light source has shown great promise in PDT applications with no tissue penetrating limitations. This strategy employs nano scintillators to convert X-rays into visible light and in turn activates the nearby PSs.⁴⁶ (4) Ultrasound (US) as a mechanical wave with a frequency beyond human hearing (> 20 kHz) can penetrate deep soft tissues up to 10 cm due to the unique advantage of minimal tissue scattering, offering considerable potential for activating sonosensitizers in deep tissues.⁴⁷ Among the many strategies, metal complexes functionalized nanomaterials have been widely reported for two-photon PDT (TP-PDT) using NIR light irradiation with a deep tissue penetration depth. The Mao group reported a nanohybrid Ru1@CDs composed of CDs modified with a Ru(II) complex (Ru1, **1**, Fig. 3) for lysosome-targeted imaging and PDT. Ru1@CDs can photoinduce ROS generation to the lysosomal membrane to obtain one- and two-photon PDT therapy in 2D cells and a 3D multicellular tumour spheroid (MCTS). Using the two-photon characteristics, their *in vivo* toxicity and capacity for deep imaging were investigated using a zebrafish model.⁴⁸

2.2 Hypoxic tumours: low oxygen level hampers the generation of therapeutically required ROS

A hypoxic tumour core is one of the main hallmarks of solid tumours.^{49–51} This tumour microenvironment generally results from the imbalance between the uptake and consumption of oxygen, mainly due to the development of abnormal tumour blood vessels, poor blood flow, and the intensified proliferation of cancer cells.^{40–49} Statistically, O₂ concentrations vary with location in solid tumours, with some interior regions having very low levels (partial pressures of O₂ < 5 mmHg, corresponding to 7 μm).^{50,51} One of the intrinsic limitations of PDT treatments toward aggressive and/or drug-resistant tumours is

the low oxygen concentration in hypoxic areas. From the analysis of the action mechanism of PDT, since the more common operation is the type II pathway, most existing PDT systems are highly O₂-dependent and involve a dramatic consumption of O₂. Based on this, hypoxic solid tumours severely reduce the effectiveness of type II PDT in clinical tumour treatment. Therefore, effective tumour oxygenation is of great significance to promoting this anti-cancer method. To generate a phototoxic effect despite hypoxic conditions, different strategies have been developed to enhance PDT under hypoxia, including (1) reducing the oxygen consumption rates of tumours, (2) O₂-evolving synergistic chemoradiotherapy, (3) artificial blood to transport oxygen to the hypoxic core and (4) combining oxygen-enriched gases with vasodilators.^{52–55}

Besides regulation of the oxygen concentration at the tumour site, research efforts have also been devoted to the development of PSs which act by an oxygen-independent mechanism, also referred to as the remote-controlled release of ¹O₂. Aromatic compounds including naphthalene, anthracene, *etc.*, which capture and store singlet oxygen to generate endoperoxides (EPOs),⁵⁶ have been widely investigated and employed in material science⁵⁷ and chemical synthesis.⁵⁸ After light irradiation or thermal reduction, the physical quenching pathway through intersystem crossing enables the exciplexes to transform into excited triplet complexes, ultimately generating reactive radical products before decomposition into starting materials. A novel Ir(III) complex EPOs (2-O-IrAn, **2**, Fig. 3) was recently investigated by the Chao group.⁵⁹ 2-O-IrAn includes an endoperoxides anthracene as the ¹O₂ supplier, overcoming the problem of PDT in hypoxia. Interestingly, upon two-photon irradiation under hypoxia, 2-O-IrAn was a photosensitizer with low cytotoxicity and released the trapped ¹O₂ with the generation of highly cytotoxic 2-IrAn and alkoxy radicals. Overall, 2-O-IrAn exhibited mitochondria targeting synergistic PDT/PACT effects to treat solid tumours in a mouse model. Besides the generation of ¹O₂ or ROS, other types of radical species could also be generated upon light irradiation, including carbon radicals,⁶⁰ chlorine radicals,⁶¹ nitric oxides,⁶² or sulfate radicals.⁶³ The discovery of potent oxygen-independent mechanisms of action for PDT under hypoxic tumour microenvironment-relevant conditions is ongoing. Using the above strategies to overcome hypoxia will be very beneficial for photodynamic therapy under hypoxia in the future.

2.3 Tumour targeted therapy: avoiding toxic side effects by tumour selective delivery of the photosensitizer

Despite the many positive features of PDT cancer therapy, this form of treatment is still not always fully adapted to clinical settings.⁶⁴ On the one hand, most PSs often suffer from poor solubility and cancerous affinity, which leads to difficult administration and unsatisfying tumour accumulation.^{65,66} Generally, conventional PSs rely heavily on passive accumulation into tumour sites (due to EPR). On the other hand, the remainder of the PSs inevitably accumulating in the tumour's surrounding tissue would also be exposed to light irradiation due to light scattering effects as well as practical challenges to irradiate only

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the tumour site, causing systemic phototoxicity and immunological barriers destruction.⁶⁷ Additionally, with only a short half-life, the radius of action of $^1\text{O}_2$ is at a small extent ($\leq 0.02 \mu\text{m}$).⁶⁸ This means only ROS generated in the vicinity of PS directly affects cancerous tissue. Based on this point, the overall extent of PDT-induced cytotoxicity and photodamage is highly dependent on the PSs' bioavailability, as well as their extracellular and intracellular localization.⁶⁹

To solve the above issues, efforts have been made to develop various drug delivery vehicles to improve the accumulation of PS in tumours, the main strategies include active or passive tumour targeting. For an active tumour targeting strategy, the specific interaction of a targeting moiety with cancer cells is utilized, such as antibodies,⁷⁰ oligosaccharides,⁷¹ oligonucleotides,⁷² proteins,⁷³ vitamins,⁷⁴ and signal peptides.⁷⁵ A passive tumour targeting approach mainly adopts nanocarriers such as inorganic nanoparticles,⁷⁶ micelles,⁷⁷ polymeric nanoparticles,⁷⁸ liposomes⁷⁹ or metal-organic frameworks^{28,36} to deliver PSs into tumours, capitalizing on the leaky, permeable vasculature and lymphatic properties of the tumour tissue known as the enhanced permeability and retention (EPR) effect. However, the high internal osmotic pressure in solid tumours also leads to a greater distribution of nanomedicines at the tumour periphery.⁸⁰ These natural barriers seriously limit the extravasation of cytotoxic nanomedicines into the tumour tissues by passive transport. Despite the fact that extravasation of nanomedicine into the tumour through the EPR effect has been widely regarded as an advantage for the successful targeted delivery of nanomedicines into tumours,⁸¹ the low success rate of clinical translation of nanomedicine in cancer therapy causes researchers to begin to question the contribution of the EPR effect in the treatment of tumors.⁸²

Besides, from the literature, we know that surface potential, particle size, and targeted polypeptide modification affect cellular distribution.⁸³ Lipophilic cations accumulate in the mitochondria and the nucleus. Most nuclear-targeted nanoparticles are less than 50 nm in diameter. These factors together affect the distribution of nanoparticles in cells. It is worth mentioning that metal complexes have structural flexibility and possess the above conditions, as organelle-targeted PSs are also prevalent fields of research.⁸⁴ It follows that the rational design of metal complexes and nanomaterials with controlled size has important meaning for organelle precisely targeted therapy. Furthermore, the metal complex functionalized with nanoparticles not only acts as a photosensitizer for one- and two-photon PDT but also as an organelles-targeting and imaging agent for tracking the NPs. Meanwhile, the nanosystems can effectively induce organelle dysfunction *via* organelles-targeting to enhance the sensitivity of PSs intracellular tumour, eventually triggering a combination of multiple cell death pathways. In a typical example, Chao's group designed a self-assembly of thiol-functionalized Ir(III) complexes into biodegradable coordination polymeric nanoparticles (IrS NPs, 3, Fig. 3).⁸⁵ The nanomaterials were truncated by intracellular GSH and released the molecular Ir(III)

complexes with mitochondrial-targeting and decreased endogenous GSH levels, causing amplifying oxidative stress in the cell. Upon two-photon irradiation, IrS NPs can generate a mixture of $^1\text{O}_2$ and superoxide anion radicals ($\cdot\text{O}_2^-$) and induce cell death by a combination of apoptosis and ferroptosis pathways.

2.4 Synergistic therapies: overcoming the limitations of a single therapeutic method for PDT

Admitting that PDT as a temporospatial-selective and minimal-invasive modality plays an important role in clinical cancer therapeutics, the therapeutic efficacy of PDT alone against several deep or hypoxic solid tumours is limited due to its inherent drawbacks and the clinical challenges of metastasis, recurrence, and resistance of cancer therapeutics.^{86,87} The main mechanisms of PDT resistance include the limited penetration of the therapeutic light source, the monotonous activation of signalling pathways, and inadequate efficacy of oxygen-independent phototoxicity. To optimize this, much research effort has been dedicated to the development of PDT-combined strategies. Several studies have reported on the combination of PDT with photothermal therapy,⁸⁸ sonodynamic therapy,⁸⁹ chemodynamic therapy,⁹⁰ chemotherapy,⁹¹ starvation therapy,⁹² radiotherapy⁹³ or immunotherapy,⁹⁴ and gas therapy.⁹⁵ From this perspective, by combining PDT with other current cancer modalities, one may be able to exploit the strengths and bypass the weaknesses of different therapies to achieve the goal of mutual benefit (Fig. 2). Moreover, the combination of PDT with imaging techniques has always been the focus of research, which is attributed to precise discrimination between the cancerous and the healthy tissue by using imaging-guided theranostics. The ongoing efforts on the combination of PDT with luminescent imaging,⁹⁶ positron emission tomography,⁹⁷ magnetic resonance imaging,⁹⁸ computed tomography,⁹⁹ photoacoustic imaging,¹⁰⁰ and ultrasonography¹⁰¹ are promising in clinical trials in the future.

3. The design strategies of metal complexes functionalized nanomaterials for PDT

It is no secret that currently the use of nanotechnology is gaining in popularity and is usually easily able to form composite nanosystems with other substances, including metal complexes, molecular drugs, organic probes, *etc.*^{30,102} Expressly, metal complexes functionalized nanomaterials containing nanocarriers and metal complexes PSs have been mostly reported. Metal complexes can modulate the surface charge and photophysical properties of nanoparticles to improve ROS generation, achieving better PDT efficacy. Many nanocarriers have excellent biocompatibility but produce negligible ROS production. Thus, using a combination of a metal complex with nanoparticles can overcome some deficiencies between the metal complex and nanoparticles. Determining

Photodynamic therapy/ Chemotherapy		Photodynamic therapy/ Immunotherapy	
Advantages	Disadvantages	Advantages	Disadvantages
<ul style="list-style-type: none"> ● Improve the safety of chemotherapy ● Avoid multidrug resistance (MDR) ● Improve overall anticancer effect 	<ul style="list-style-type: none"> ■ The inevitable toxicity of chemotherapy drugs 	<ul style="list-style-type: none"> ● Effectively inhibit tumor recurrence and metastasis ● Enhance anti-tumor immune effect 	<ul style="list-style-type: none"> ■ An adverse reaction caused by an overactive immune system
Photodynamic therapy/ Radiotherapy		Photodynamic therapy/ Chemodynamic therapy	
Advantages	Disadvantages	Advantages	Disadvantages
<ul style="list-style-type: none"> ● Achieve deep tumor therapy ● Increase the sensitivity of tumor for radiation detection (RD) ● Achieve abscopal effect 	<ul style="list-style-type: none"> ■ Unavoidable toxic side effects for RD 	<ul style="list-style-type: none"> ● Catalyze the production of multiple ROS ● Rapid chemical reaction rate to enhance antitumor therapy 	<ul style="list-style-type: none"> ■ Insufficient H₂O₂ levels in tumor cells ■ Degradation of the enzyme itself
Photodynamic therapy/ Starvation therapy		Photodynamic therapy/ Gas Therapy	
Advantages	Disadvantages	Advantages	Disadvantages
<ul style="list-style-type: none"> ● Amplify therapeutic efficiency ● Suitable for the treatment of hypoxic tumor 	<ul style="list-style-type: none"> ■ Enzyme's shortcomings, such as poor stability, short half-life and immunogenicity ■ The systemic toxicity of H₂O₂ 	<ul style="list-style-type: none"> ● Oxygen independent to produce cytotoxic gas (CO, NO and SO₂ etc.) and suitable for the treatment of hypoxic solid tumor ● Achieve effective therapy 	<ul style="list-style-type: none"> ■ limited penetration (non-invasive PDT) ■ Uncontrolled release, high toxic side effects

Fig. 2 The advantages and disadvantages of synergistic therapies containing PDT.

how to provide effective strategies to construct metal complexes functionalized nanomaterials is the focus of this section.

3.1 Metal coordination bonds-based nanomaterials

Currently, there are two typical strategies for the fabrication of metal coordination bonds-based nanomaterials, including (1) M–N coordination bonds between metal ion/metal complex precursors and nanomaterials and (2) M–O coordination bonds between metal complex PSs and nanomaterials. These nanomaterials can not only extend the blood circulation time of metal complex PSs and prevent drug leakage but can also enhance PDT by the synergism of nanomaterials and metal complexes in photophysical modulation, achieving the antitumor effect of $1 + 1 > 2$. As a typical example, Chao's group reported the first case of an oxygen self-sufficient photosensitizer (**4**, Fig. 3) produced *via* grafting a metal complex precursor ($[\text{Ru}(\text{bpy})_2]^{2+}$) onto *g*-C₃N₄, according to Ru–N bonding (Fig. 4).¹⁰³ In this system, the *g*-C₃N₄ frame acts as a N[^]N ligand coordinated with the metal center of $[\text{Ru}(\text{bpy})_2]^{2+}$ to obtain stable nano-PSs (Ru-*g*-C₃N₄). The PS show high loading capability, good biocompatibility, and high stability, and can catalyze the O₂ generation from H₂O₂ or H₂O in a hypoxic tumour under visible light irradiation, simultaneously producing multiple cytotoxic ROS ($\bullet\text{OH}$, $\bullet\text{O}_2^-$, and $^1\text{O}_2$). Finally, nano-PSs have red-shifted luminescence which reduces the interference of biological background emission allowing precisely guided PDT to reduce the potential side effects of irradiation. Unfortunately, the nanomaterials are limited to visible light excitation, resulting in poor penetration of deep tissue. To address the problem, the group also developed functionalization of graphitic carbon nitride nanosheets with mitochondria-targeting Ir(III) complexes (**5**, Fig. 3) for oxygen self-sufficient two-photon PDT.¹⁰⁴ This strategy can not only solve the above

problems but also improve the efficacy of two-photon PDT in melanoma tumour models.

The second strategy is with the M–O coordination bond, which is excellent in anti-tumour synergistic therapy. We believe that synergistic therapeutic formulations can make up for the deficiency of PDT in anti-tumour treatment, especially oxygen-dependent PDT. Chao *et al.* prepared a functionalization of black-titanium nanoparticles with iridium complexes (**6**, Fig. 3) proposed by Ti–O bond,¹⁰⁵ which were further encapsulated with cancer cell membranes to perform hierarchical-targeted synergistic photothermal and sonodynamic cancer therapy. Despite the use of ultrasonic and near-infrared-II (NIR-II) region light to improve the light penetration depth, the load rate of the photosensitizer needs to be further improved.

Except for the strategy based on metal coordination bonds between metal complexes and nanomaterials, using metal cations coordinated on nanomaterials to form metal coordination bonds-based nanomaterials has also been the focus of research in recent years.^{106–108} Classically, Qu's group previously reported the coordinate integration of Cu²⁺ and *g*-C₃N₄ nanosheets (Cu²⁺-*g*-C₃N₄) (**9**, Fig. 3), rendering improvement in light-triggered ROS generation as well as the depletion of intracellular GSH levels.¹⁰⁹ In this system, the author verified the generation of redox-active species Cu⁺-*g*-C₃N₄ under illumination, catalyzing the reduction of O₂ to $\bullet\text{OH}$ or $\bullet\text{O}_2^-$, both of which facilitated the generation of ROS to enhance the efficiency of photodynamic therapy. Unfortunately, the system had not been further verified *in vivo*, hampered by the short excitation wavelength. Moreover, single-atom catalysts (SACs) have become one of the hottest subjects of research. SACs with distinct properties, such as precisely located metal centres, identical coordination environments, tailorable composition, and structure, can induce multiple ROS ($\bullet\text{OH}$, $\bullet\text{O}_2^-$, and $^1\text{O}_2$)

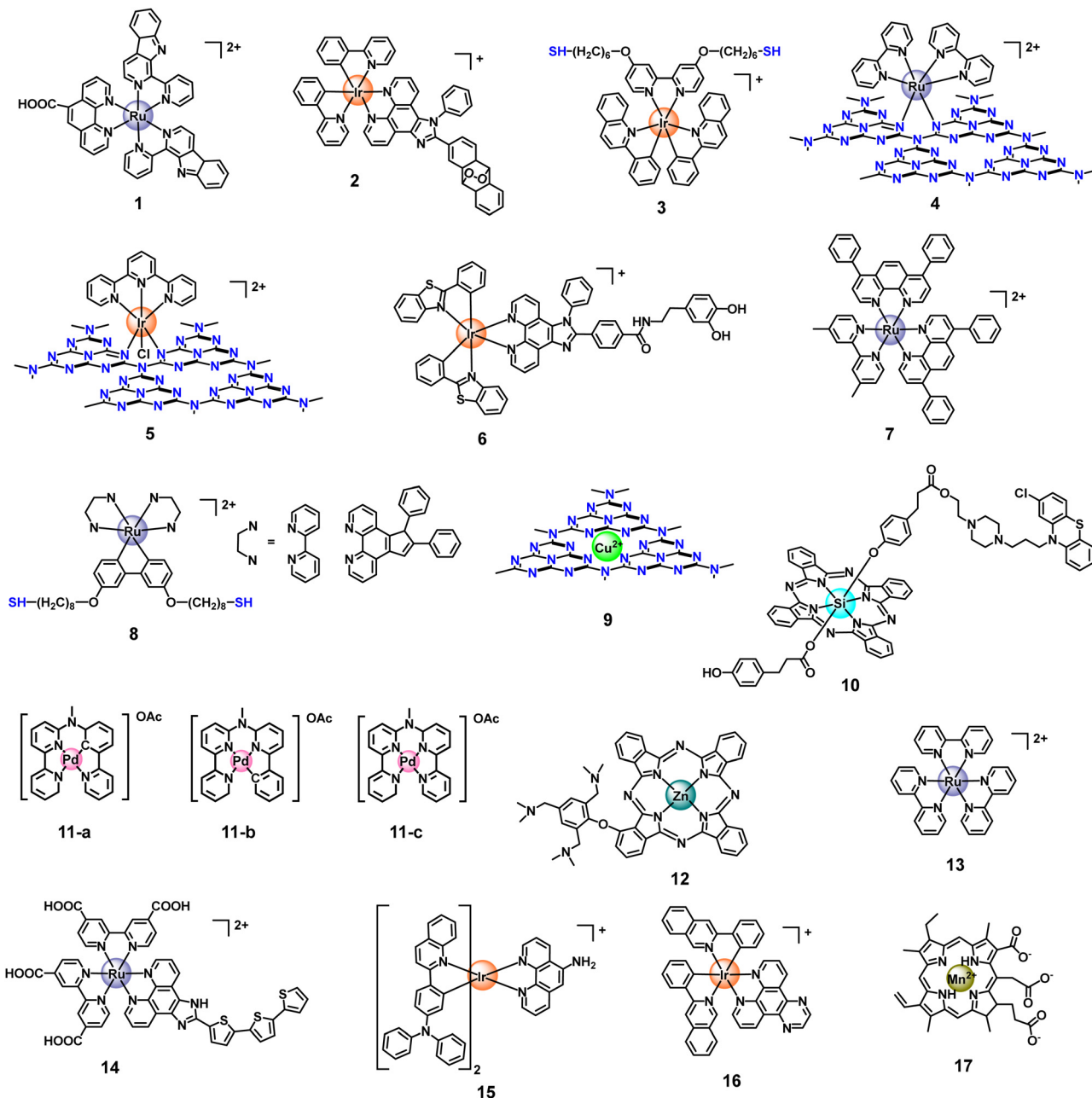


Fig. 3 Structures of Ru, Ir, Cu, Si, Pd, Zn, and Mn-based metal complexes as metal photosensitizers.

generation *via* catalyzing H_2O_2 decomposition, improving PDT in the tumour microenvironment, especially hypoxic tumours.^{110,111} Recently, inspired by the catalase-like activities of Ru nanoparticles, Wang *et al.* developed a multifunctional OxgeMCC-r SAC for efficiently degrading H_2O_2 to O_2 (Fig. 5).¹¹² The OxgeMCC-r SACs with coordination environment and unique structure of six unsaturated Ru-C₆ coordination sites can endow excellent catalytic activity in the decomposition of H_2O_2 with the atomic economy and superior stability, which could highly relieve the hypoxic microenvironment of solid tumours, thus facilitating subsequent photocatalytic $^1\text{O}_2$ generation, and finally augmenting the PDT efficacy.

Constructed *via* coordination bonds between metal cluster secondary building units (SBUs) and bridging ligands, nano-scale metal-organic frameworks (nMOFs) have emerged as a new class of hybrid materials with tunable, crystalline, and porous structures. In recent years, nMOFs have been identified as the most promising nano-PSs. Compared to other nano-PSs, the porous and crystalline structures of nMOFs isolate PSs from each other to avoid self-quenching of PS excited states. As a result, nMOFs can deliver porphyrin, chlorin, and bacteriochlorin PSs for PDT without suffering from self-quenching.¹¹³⁻¹¹⁵ The biodegradability of nMOFs alleviates long-term toxicity, whereas tunable compositions and structures allow



Fig. 4 A schematic illustration of an oxygen self-sufficient photosensitizer (Ru-g-C₃N₄) with activated multiple ROS ([•]OH, [•]O₂⁻, and ¹O₂) for the efficient photodynamic therapy of hypoxic tumours. Figure adapted from ref. 103 Copyright: 2021, with permission from Elsevier.

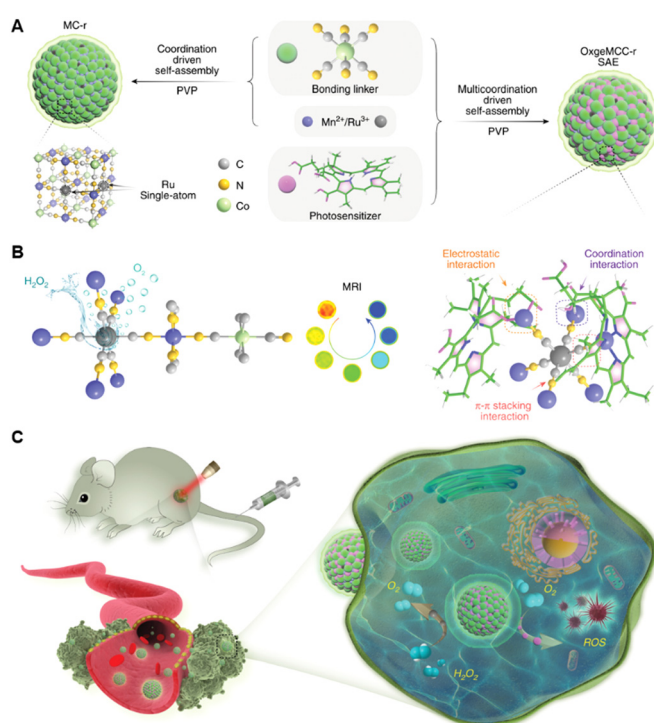


Fig. 5 A schematic illustration of the synthesis of OxgeMCC-r SAE and mechanism of action of catalyzing oxygen generation and ROS production for enhanced PDT of cancer by OxgeMCC-rSAE. Figure adapted from ref. 112 Copyright: 2020, with permission from Springer Nature.

the optimization of nMOF nano-PSs for PDT applications. In 2014, Lin *et al.* first reported a Hf-porphyrin nMOF, DBPHf, as a highly effective photosensitizer for PDT of resistant head and neck cancer.¹¹⁶ nMOFs as PSs have been successively reported for PDT in recent years,^{28,117,118} but nMOFs have not been

explored in depth for PTT monotherapy or other combined therapies.

In brief, metal coordination bonds-based nanomaterials have structural stability and great advantages in anti-hypoxic solid tumour PDT and combination with multiple therapeutic modalities.

3.2 Self-assembling nanomaterials

Self-assembling nanoparticles based on metal complexes deploy a simple yet powerful strategy to construct high-loading rate nanomaterials with improved bioavailability. Aggregates of metal complexes could be effectively conveyed to biological targets while being protected from irrelevant chemical and/or biological degradation. The one strategy is mainly to form nanoparticles through the S-S bond, Se-Se bond, phenyl borate lipid and dithioacetals structure and so on, which can prevent the leakage of photosensitizers or drugs in the drug delivery mechanism and realize the purpose of the most accurate possible release of tumours. Notably, the transformation from nanoparticles to molecules (PSs) by GSH or ROS-responsive controlled release fully uncages the PDT performance of nano-polymers.^{119–121} Based on this point, Chao's group designed a nano-assembled photosensitizer based on thiol-tailed Ru(II) complexes (8, Fig. 3) linked by disulfide bonds for the two-photon PDT of cancer.¹²² Xiao's group reported a nano-polymer (NP(Se)s) from the self-assembly of lipophilic Pt(IV) prodrugs and polymer P1 with Se-Se bonds in its main chain for redox-based therapy.¹²³ Furthermore, using intermolecular interactions to form supramolecular self-assembling nano-polymers could deliver high concentrations of a drug to cancer cells by providing a high drug-loading capacity. Che and Pan reported that the cyclometalated platinum/gold/ruthenium complex forms supramolecular self-assembly complexes *via* intermolecular interactions, producing an increase in drug efficacy.^{124–127}

Another self-assembling strategy is self-assembled polymeric micelles or liposomes with amphiphilic structures for hydrophobic photosensitizers loaded into the hydrophobic core space in aqueous solution. Biocompatible phospholipids or polymers (such as F127 and DSPE-PEG₂₀₀₀ *et al.*)^{128,129} used in self-assembly dissociate from cargo metal complexes after being endocytosed into the target cells and offer no severe toxicity by themselves. For instance, Gasser's group reported a nanoparticle from the encapsulation of a PSs (Ru(II) polypyridine complex, 7, Fig. 3) in an amphiphilic polymer with terminal folate groups (DSPE-PEG₂₀₀₀-folate) for PDT.¹³⁰ The authors found that the Ru(II) complex itself has a cytotoxic effect in the dark, but this drawback was avoided after the encapsulation of polymers. Meanwhile, DSPE-PEG₂₀₀₀-folate provides targeting cancer cells, increasing to more than 20 times higher accumulation of the Ru(II) complexes in cancer cells. Upon 480 or 595 nm irradiation, the nanoparticles were found to generate ROS and showed a high phototoxicity effect in the very low micromolar range in 2D monolayer cancer cells and 3D multicellular tumour spheroids. This interesting work

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is limited to the cellular level attributed to the limited light-penetration depth.

Moreover, researchers found that the polymeric chains could self-assemble into nanoparticles during the transformation process of the aqueous phase. Wei *et al.* designed a novel type of nucleus-targeting Pt(IV) nanoparticles based on a polymeric chain, which contained a chemotherapeutic agent (oxaliplatin), a photosensitizer with aggregation-induced emission (AIE), and cancerous tissue/nucleus targeting peptides R^8K .¹³¹ Within an aqueous solution, the functionalized polymer chains self-assemble into spherical nanoparticles. Upon irradiation, the Pt(IV) centre is reduced to Pt(II) and axially coordinated ligands are released. The PS catalytically generates reactive oxygen species (ROS) causing immunogenic cell death (ICD) and presenting a multimodal treatment by chemotherapy and photodynamic immunotherapy. Using a self-assembling strategy to form novel nano-PSs, which need straightforward fabrication procedures, is a method for preparing the highest PS load rate reported at present. Based on this, only a minimal dose of PS is required to maximize antitumor activity during the PDT process.

3.3 π - π stacking-based nanomaterials

It has been widely recognized that non-covalent interactions between aromatic moieties, which are habitually called π - π stacking interactions, could play a vital role in a wide variety of chemical systems.¹³² π - π stacking interactions refer to the interactions between aromatic rings containing π orbitals.¹³³ Most metal complexes containing ligand structures of aromatic rings, porphyrins, phthalocyanines and their derivatives, regarded as photosensitizers have large conjugated planar aromatic structures, which have also rarely been reported in self-assembly with nanoparticles by π - π stacking interactions in a crowded environment. This not only can maintain the photosensitivity of the original metal complex but can also change the physical properties of the material to enhance the antitumor effect during the PDT process. Classically, Bonnet's group prepared supramolecular nanorods, which utilize monocationic cyclometalated palladium complexes OAc ($N^{\wedge}N^{\wedge}C^{\wedge}N$) and OAc ($N^{\wedge}N^{\wedge}N^{\wedge}C$) (**11**, Fig. 3) *via* metallophilic Pd...Pd interactions and π - π stacking to self-assemble in aqueous solutions (Fig. 6).¹³⁴ The authors found that the nanorods improved cellular uptake of the cyclometalated compounds by endocytosis (*i.e.*, an active uptake pathway). Under blue light irradiation, a better singlet oxygen generation ability was achieved, leading to dramatically enhanced photodynamic properties in multicellular tumour spheroids and in mice tumour xenografts. However, the short wavelength blue light used in the PDT process limited light penetration to the tissue and the antitumor activity of the photosensitizer must not be significantly weakened.

Meanwhile, for PSs molecules, the π - π stacking strategy can make up for the deficiency of traditional PSs, such as poor water solubility, low biocompatibility and highly efficient vibrational relaxation of their electronic excited states. For example, Yoon *et al.* developed the nanostructured Zn(II) phthalocyanine

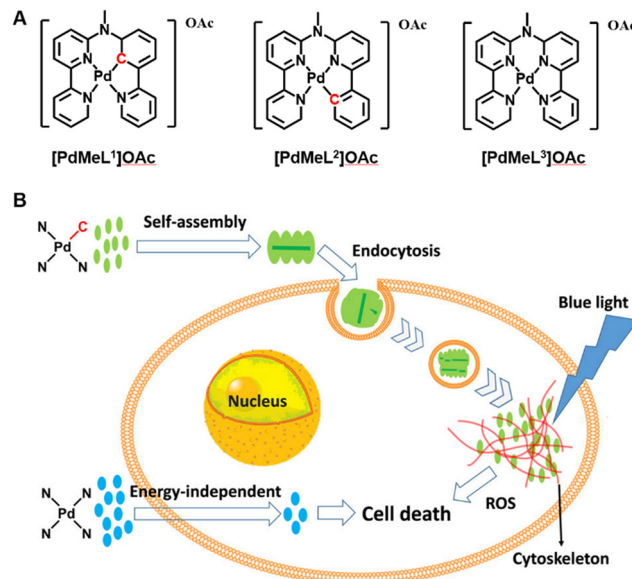


Fig. 6 A schematic illustration of (A) structures of the Pd complexes and (B) synthesis and mechanism of supramolecular nanorods *via* π - π stacking of Pd complexes to enhance photodynamic therapy under blue light. Figure adapted from ref. 134 Copyright: 2020, with the permission from American Chemical Society.

(**12**, Fig. 3) assembly (NanoPcA), which serves as a PS for efficient ROS generation *via* the type I mechanism. In their study, Zn(II) phthalocyanine monomers aggregate to form molecular clusters *via* mutual diffusion of water in DMF, furtherly ensuing growth through π -stacking. In biological activity studies, NanoPcA was found to display excellent photodynamic antibacterial activity against both common and antibiotic-resistant bacterial strains.¹³⁵ However, this interesting study was not followed up *in vivo*. Recently, Huang's group prepared a uniform and stable nanosphere (NanoPcAF) through a spontaneous assembly process of Si(IV) phthalocyanine derivatives (**10**, Fig. 3) bearing perphenazine groups. NanoPcAF displayed an efficient type I photoreaction to generate abundant superoxide radicals and a significant vibrational relaxation to induce a relatively high photothermal conversion efficiency, which enabled it to overcome tumour hypoxia in PDT *in vivo*.¹³⁶ This work solves most of the bottlenecks encountered by PDT.

Besides the above method, some multifunctional nanomaterials (such as graphene,¹³⁷ carbon nanotubes (CNTs),¹³⁸ and graphite phase carbon nitride (g - C_3N_4))¹³⁹ also contain aromatic ring structures, which means they can combine with metal complexes to form a new class of materials by π - π stacking interactions. Among them, Li *et al.* prepared a two-photon excited nanocomposite FCRH, which is assembled by π - π stacking between $[Ru(bpy)_3]^{2+}$ (**13**, Fig. 3) and Fe- C_3N_4 , and modified by a copolymer with poly (ethylene glycol) arms (HOP).¹⁴⁰ Under 800 nm two-photon irradiation, the nanocomposite can separate charges and prevent the recombination of electron-hole pairs more efficiently to enhance photocatalytic activity and O_2 generation. The activated photosensitizer

([Ru(bpy)₃]²⁺) promotes ¹O₂ generation to enhance two-photon PDT under hypoxia. Although the nano-platform alleviates the hypoxic tumour environment, the PDT efficacy in hypoxic solid tumours is a limiting attribute to single ¹O₂ generation.

3.4 Amidation-based nanomaterials

Amidation is a readily achievable chemical synthesis process, which can be used to form an amido linkage as a bridge between metal complexes and nanomaterials to enhance PDT and achieve the purpose of controlled release in cells. Zhou *et al.* constructed a hypoxia-adaptive nanocomposite TiO₂@Ru@siRNA for the prevention and treatment of oral squamous cell carcinoma (OSCC) (Fig. 7).¹⁴¹ In this system, the carboxylic Ru complex PSs (14, Fig. 3) and TiO₂ NPs surface with amino-propyltriethoxysilane (APTES) were linked by an amido bond, ulteriorly loading siRNA targeting HIF-1 α . Under visible light (525 nm) irradiation, the nanoparticles can produce ROS through both types I and II PDT, causing lysosomal damage to effectively promote siRNA escape and induce pyroptosis of OSCC cells, which activates multifaceted cancer immune responses. Collectively, the author claims that the nanosystem has outstanding PDT performance and immunomodulatory

function. However, the PDT effect is still limited by its short wavelength light penetration, which is the problem with most metal photosensitizers.

Moreover, as a similar example, Huang's group designed a new type of mesoporous silica-coated lanthanide-doped upconversion nanoparticles, which contain oxygen-sensitive Ir(III) complex (15, Fig. 3) in the outer silica shell.¹⁴² An Ir(III) complex and SiO₂ NPs were bridged by an amido bond. Regrettably, this nanosystem shows negligible PDT effects but only acts as a nanoprobe, monitoring oxygen concentration by reducing auto-fluorescence under both downconversion and upconversion channels.

3.5 Physisorption-based nanomaterials

Physisorption, also known as van der Waals adsorption, is caused by the intermolecular force between the adsorbent and the adsorbent, which is also known as the van der Waals force and is weaker than that in the methods mentioned above. Excitingly, physisorption is one of the most common construction methods of metal complexes and nanomaterials, including electrostatic interaction, hydrophobic interaction, polarity interaction and other non-covalent bonding. However, due to

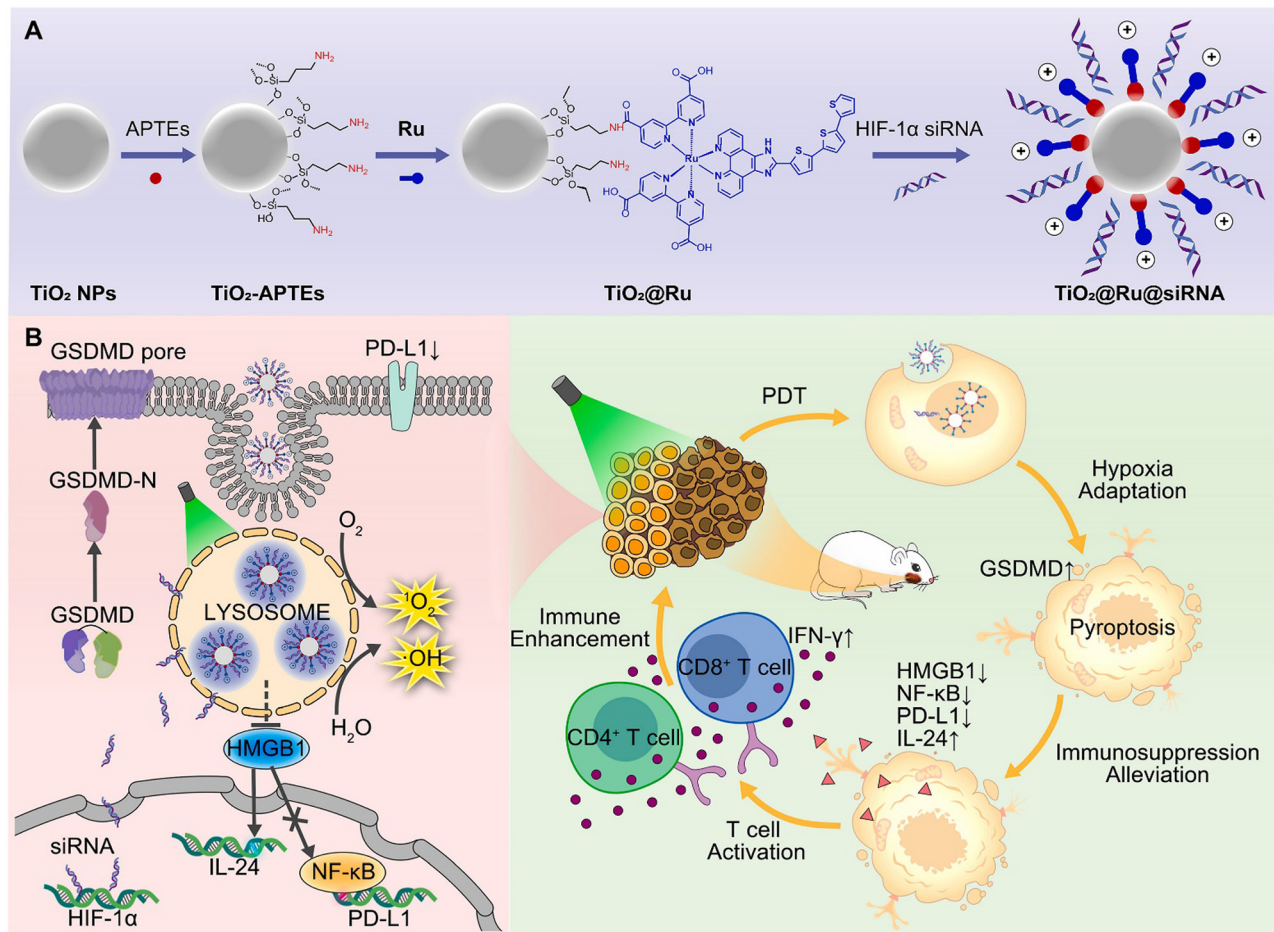


Fig. 7 A schematic illustration of (A) the construction of the nanocomposite TiO₂@Ru@siRNA and (B) the action mechanism of TiO₂@Ru@siRNA. Figure adapted from ref. 141 Copyright: 2022, with permission from Elsevier.

Highlight

the leakage of drugs during nanomaterials drug delivery, the structural stability of this method is still under debate compared to the above strategies.¹⁴³

Using the lipophilic and cationic properties of metal complexes, the researchers designed nanomaterials with electro-negative surfaces, which can combine metal complexes with positive ions to maintain electric neutrality by electrostatic interaction. Based on this point, Gao's group reported a biscyclometalated Ir(III) complex PSs (**16**, Fig. 3) electrostatically loaded on an MOF decorated with a dual-responsive polycationic polymer to form nanoplateforms (Ir@MOF/P NPs). The endo/lysosome escape ability and the pH & ROS-responsive charge reversal function of Ir@MOF/P NPs are demonstrated to be a promising strategy to improve the PDT performance *in vitro* (Fig. 8).¹⁴⁴ The study was limited to the cellular level because of the limited penetration of single-photon light.

Another strategy is using hydrophobic interaction, which makes hydrophobic groups gather, and avoids burst leakage in the hydrophilic medium, minimizing the non-active surface. Most metal photosensitizers are usually not hydrophilic, the synthesis of metal complexes functionalized nanomaterials are usually dispersed in fat and then transferred to the water phase. This process can adsorb metal complexes into nanomaterials by hydrophobic interaction. For instance, Liu *et al.*¹⁴⁵ proposed pH-responsive CaCO₃ nanoparticles with polyethene glycol (PEG) as a multifunctional nano-carrier, following loading both a photosensitizer (Mn²⁺-chelated chlorin e6) (**17**, Fig. 3) and a chemotherapy drug (doxorubicin) to form nanoplateforms (CaCO₃@Ce6(Mn)-PEG(DOX)). The nanoparticles were found to be highly sensitive to pH and would rapidly degrade under slightly acidic solutions, leading to the efficient release of Ce6(Mn) and doxorubicin (DOX). Under the irradiation of light, the nanoplateforms can execute a synergistic photodynamic chemotherapy of cancer *in vivo*. Using a CaCO₃ nanoparticles loading strategy as a smart tumour-pH responsive drug delivery platform is useful for applications in cancer theranostics.

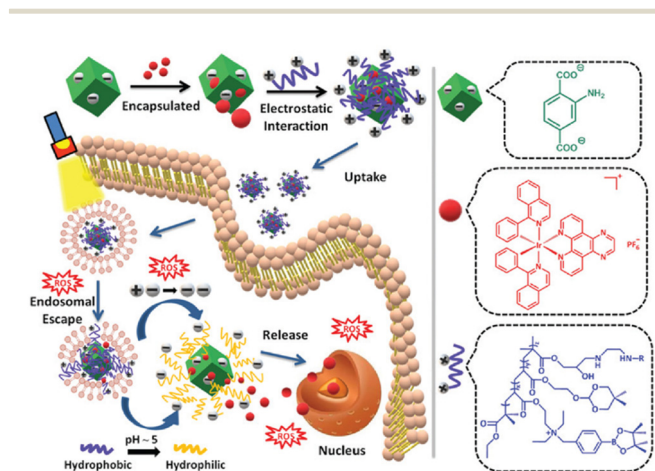


Fig. 8 A schematic illustration of the preparation of Ir@MOF/P NPs and its putative mechanism for intracellular metabolism and ROS-induced tumour inhibition processes. Figure adapted from ref. 144 Copyright: 2020, with the permission from Royal Society of Chemistry.

4. Summary and outlook

Compared to traditional cancer treatments (*i.e.* surgery, radiotherapy, or chemotherapy), photodynamic therapy has emerged as a complementary medical technique to classical procedures for the treatment of cancer. However, the application of PDT is hindered by classic PSs due to their poor water solubility and poor biocompatibility. In addition, this review discusses some of the challenges facing PDT including light-penetration depth, hypoxic tumour environments, targeting therapy for tumours and synergistic therapy. Although strategies designed by researchers can solve some of these problems, further exploration is needed to realize efficient PDT. The application of nanoparticles, especially metal complexes-based nanoparticles, in PDT is a very promising approach for the imminent breakthrough of classic techniques. The nanoparticles can be applied as carriers of hydrophobic PSs and delivered to the sites of different tumours *via* the EPR effect, which can handle the off-target effect of metal complexes.

In this review, the recent developments in metal complexes functionalized nanomaterials for PDT are herein presented. The fabrication strategies between nanomaterials and complexes are discussed, including metal complexes metal coordination bonds, self-assembly, π - π stacking, amidation, and physisorption. We believe in using these strategies to aid the judicious choice of complexes and nanomaterials in future nanoformulation design. We envision that the next generation of metal complex functionalized nanomaterials will achieve the optimization of characteristics such as stability of nanoformulations, physicochemical properties, and energy transfer properties. Attention to these factors will aid the development of new nanoplateforms to overcome the challenges of PDT, achieving higher treatment efficiency and outcomes. Furthermore, the biosafety of metal complexes functionalized nanomaterials needs further evaluation. The long-term effects of the nanoformulations, including biocompatibility, pharmacokinetic properties, systemic toxicities, biological metabolic pathways and immunogenicity must be studied in detail in animal experiments before further clinical development and application. Recently, combining PDT with immunotherapy has provided a powerful approach to treating metastatic cancers. As a local treatment, PDT can also produce a strong inflammatory response by inducing immunogenic cell death.¹⁴⁶ We expect significant amounts of research into metal complexes functionalized nanomaterials-mediated PDT to induce immunotherapy soon. Thus, we anticipate that metal complexes functionalized nanomaterials will find future applications in the diagnosis and treatment of various diseases for cancer treatment in clinic.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- D. Dolmans, D. Fukumura and R. Jain, *Nat. Rev. Cancer*, 2003, **3**, 380–387.
- G. I. Stables and D. V. Ash, *Cancer Treat. Rev.*, 1995, **21**, 311–323.
- M. J. Manyak, A. Russo, P. D. Smith and E. Glatstein, *J. Clin. Oncol.*, 1988, **6**, 380–391.
- X. Zhen, P. Cheng and K. Pu, *Small*, 2019, **15**, 1804105.
- S. G. Perlman, L. H. Gerber, R. M. Roberts, T. P. Nigra and W. F. Barth, *Ann. Intern. Med.*, 1979, **91**, 717–722.
- R. Huang, C. Zhang, Y. Bu, Z. Li, X. Zheng, S. Qiu, J. O. A. Machuki, L. Zhang, Y. Yang, K. Guo and F. Gao, *Biomaterials*, 2021, **277**, 121088.
- E. H. C. van Dijk, S. Fauser, M. B. Breukink, R. Blanco-Garavito, J. M. M. Groenewoud, J. E. E. Keunen, P. J. H. Peters, G. Dijkman, E. H. Souied, R. E. MacLaren, G. Querques, S. M. Downes, C. B. Hoyng and C. J. F. Boon, *Ophthalmol.*, 2018, **125**, 1547–1555.
- Y. Dong, W. Cao and J. Cao, *Nanoscale*, 2021, **13**, 14591–14608.
- J. Nam, S. Son, L. J. Ochyl, R. Kuai, A. Schwendeman and J. J. Moon, *Nat. Commun.*, 2018, **9**, 1074.
- X. Liang, M. Chen, P. Bhattarai, S. Hameed and Z. Dai, *ACS Nano*, 2020, **14**, 13569–13583.
- Y. Wu, S. Li, Y. Chen, W. He and Z. Guo, *Chem. Sci.*, 2022, **13**, 5085–5106.
- B. Myrzakhetov, P. Arnoux, S. Mordon, S. Acherar, I. Tsoy and C. Frochot, *Pharmaceuticals*, 2021, **14**, 138.
- X. Zhao, J. Liu, J. Fan, H. Chao and X. Peng, *Chem. Soc. Rev.*, 2021, **50**, 4185–4219.
- L. Zeng, P. Gupta, Y. Chen, E. Wang, L. Ji, H. Chao and Z.-S. Chen, *Chem. Soc. Rev.*, 2017, **46**, 5771–5804.
- S. Monro, K. L. Colón, H. Yin, J. Roque, III, P. Konda, S. Gujar, R. P. Thummel, L. Lilge, C. G. Cameron and S. A. McFarland, *Chem. Rev.*, 2019, **119**, 797–828.
- K. K.-W. Lo, M.-W. Louie and K. Y. Zhang, *Coord. Chem. Rev.*, 2010, **254**, 2603–2622.
- K. Qiu, H. Zhu, T. W. Rees, L. Ji, Q. Zhang and H. Chao, *Coord. Chem. Rev.*, 2019, **398**, 113010.
- E. Meggers, *Angew. Chem., Int. Ed.*, 2017, **56**, 5668–5675.
- J. Shen, T. W. Rees, L. Ji and H. Chao, *Coord. Chem. Rev.*, 2021, **443**, 214016.
- F. Heinemann, J. Karges and G. Gasser, *Acc. Chem. Res.*, 2017, **50**, 2727–2736.
- A. E. O'Connor, W. M. Gallagher and A. T. Byrne, *Photochem. Photobiol.*, 2009, **85**, 1053–1074.
- R. Baskaran, J. Lee and S.-G. Yang, *Biomater. Res.*, 2018, **22**, 25.
- P.-C. Lo, M. S. Rodriguez-Morgade, R. K. Pandey, D. K. P. Ng, T. Torres and F. Dumoulin, *Chem. Soc. Rev.*, 2020, **49**, 1041–1056.
- J. Karges, *Angew. Chem., Int. Ed.*, 2022, **61**, e202112236.
- R. Bonnett, *Chem. Soc. Rev.*, 1995, **24**, 19–33.
- A. Kamkaew, S. H. Lim, H. B. Lee, L. V. Kiew, L. Y. Chung and K. Burgess, *Chem. Soc. Rev.*, 2013, **42**, 77–88.
- E. Villemin, Y. C. Ong, C. M. Thomas and G. Gasser, *Nat. Rev. Chem.*, 2019, **3**, 261–282.
- Y. Ye, Y. Zhao, Y. Sun and J. Cao, *Int. J. Nanomed.*, 2022, **1**, 2367–2395.
- C.-N. Ko, G. Li, C.-H. Leung and D.-L. Ma, *Coord. Chem. Rev.*, 2019, **381**, 79–103.
- M. Abbas, Q. Zou, S. Li and X. Yan, *Adv. Mater.*, 2017, **29**, 1605021.
- Y. Li, Y. Wang, G. Huang and J. Gao, *Chem. Rev.*, 2018, **118**, 5359–5391.
- C. Healy, L. Hermanspahn and P. E. Kruger, *Coord. Chem. Rev.*, 2021, **432**, 213756.
- W. Chen, E. M. Goldys and W. Deng, *Prog. Lipid Res.*, 2020, **79**, 101052.
- W. Li, J. Yang, L. Luo, M. Jiang, B. Qin, H. Yin, C. Zhu, X. Yuan, J. Zhang, Z. Luo, Y. Du, Q. Li, Y. Lou, Y. Qiu and J. You, *Nat. Commun.*, 2019, **10**, 3349.
- Y. Yang, L. Wang, H. Cao, Q. Li, Y. Li, M. Han, H. Wang and J. Li, *Nano Lett.*, 2019, **19**, 1821–1826.
- Y. Zhang, H. Fu, S. Chen, B. Liu, W. Sun and H. Gao, *Chem. Commun.*, 2020, **56**, 762–765.
- S.-Y. Liu, Y. Xu, H. Yang, L. Liu, M. Zhao, W. Yin, Y.-T. Xu, Y. Huang, C. Tan, Z. Dai, H. Zhang, J.-P. Zhang and X.-M. Chen, *Adv. Mater.*, 2021, **33**, 2100849.
- L. Fu, Z. Yan, Q. Zhao and H. Yang, *Adv. Mater. Interfaces*, 2018, **5**, 1801094.
- N. Ni, X. Zhang, Y. Ma, J. Yuan, D. Wang, G. Ma, J. Dong and X. Sun, *Coord. Chem. Rev.*, 2022, **458**, 214415.
- W. Zhao, Y. Zhao, Q. Wang, T. Liu, J. Sun and R. Zhang, *Small*, 2019, **15**, 1903060.
- R. Zhang, L. Liang, Q. Meng, J. Zhao, H. T. Ta, L. Li, Z. Zhang, Y. Sultanbawa and Z. P. Xu, *Small*, 2019, **15**, 1803712.
- N. M. Idris, M. K. G. Jayakumar, A. Bansal and Y. Zhang, *Chem. Soc. Rev.*, 2015, **44**, 1449–1478.
- O. S. Wolfbeis, *Chem. Soc. Rev.*, 2015, **44**, 4743–4768.
- Y. I. Park, K. T. Lee, Y. D. Suh and T. Hyeon, *Chem. Soc. Rev.*, 2015, **44**, 1302–1317.
- M.-K. So, C. Xu, A. M. Loening, S. S. Gambhir and J. Rao, *Nat. Biotechnol.*, 2006, **24**, 339–343.
- A. Kamkaew, F. Chen, Y. Zhan, R. L. Majewski and W. Cai, *ACS Nano*, 2016, **10**, 3918–3935.
- Y. Zhang, X. Zhang, H. Yang, L. Yu, Y. Xu, A. Sharma, P. Yin, X. Li, J. S. Kim and Y. Sun, *Chem. Soc. Rev.*, 2021, **50**, 11227–11248.
- D.-Y. Zhang, Y. Zheng, H. Zhang, L. He, C.-P. Tan, J.-H. Sun, W. Zhang, X. Peng, Q. Zhan, L.-N. Ji and Z.-W. Mao, *Nanoscale*, 2017, **9**, 18966–18976.
- V. Bhandari, C. Hoey, L. Y. Liu, E. Lalonde, J. Ray, J. Livingstone, R. Lesurf, Y.-J. Shiah, T. Vujcic, X. Huang, S. M. G. Espiritu, L. E. Heisler, F. Yousif, V. Huang, T. N. Yamaguchi, C. Q. Yao, V. Y. Sabelnykova, M. Fraser, M. L. K. Chua, T. van der Kwast, S. K. Liu, P. C. Boutros and R. G. Bristow, *Nat. Genet.*, 2019, **51**, 308–318.
- X. Li, N. Kwon, T. Guo, Z. Liu and J. Yoon, *Angew. Chem., Int. Ed.*, 2018, **57**, 11522–11531.
- L. Larue, B. Myrzakhetov, A. Ben-Mihoub, A. Moussaron, N. Thomas, P. Arnoux, F. Baros, R. Vanderesse, S. Acherar and C. Frochot, *Pharmaceuticals*, 2019, **12**, 163.
- W. Yu, T. Liu, M. Zhang, Z. Wang, J. Ye, C.-X. Li, W. Liu, R. Li, J. Feng and X.-Z. Zhang, *ACS Nano*, 2019, **13**, 1784–1794.
- Z. He, P. Liu, S. Zhang, J. Yan, M. Wang, Z. Cai, J. Wang and Y. Dong, *Angew. Chem., Int. Ed.*, 2019, **58**, 3834–3837.
- Y. Cheng, H. Cheng, C. Jiang, X. Qiu, K. Wang, W. Huan, A. Yuan, J. Wu and Y. Hu, *Nat. Commun.*, 2015, **6**, 8785.
- Z. Ma, X. Jia, J. Bai, Y. Ruan, C. Wang, J. Li, M. Zhang and X. Jiang, *Adv. Funct. Mater.*, 2017, **27**, 1604258.
- L. Wei, Z. Zhang, A. Kumar, S. Banerjee and H. Huang, *Chem. – Eur. J.*, 2022, **28**, e202202233.
- Z. Gao, Y. Han and F. Wang, *Nat. Commun.*, 2018, **9**, 3977.
- A. A. Ghogare and A. Greer, *Chem. Rev.*, 2016, **116**, 9994–10034.
- S. Kuang, F. Wei, J. Karges, L. Ke, K. Xiong, X. Liao, G. Gasser, L. Ji and H. Chao, *J. Am. Chem. Soc.*, 2022, **144**, 4091–4101.
- S. Kuang, L. Sun, X. Zhang, X. Liao, T. W. Rees, L. Zeng, Y. Chen, X. Zhang, L. Ji and H. Chao, *Angew. Chem., Int. Ed.*, 2020, **59**, 20697–20703.
- R. Song, H. Wang, M. Zhang, Y. Liu, X. Meng, S. Zhai, C.-C. Wang, T. Gong, Y. Wu, X. Jiang and W. Bu, *Angew. Chem., Int. Ed.*, 2020, **59**, 21032–21040.
- J. Sun, X. Cai, C. Wang, K. Du, W. Chen, F. Feng and S. Wang, *J. Am. Chem. Soc.*, 2021, **143**, 868–878.
- Y. Liu, W. Zhen, Y. Wang, S. Song and H. Zhang, *J. Am. Chem. Soc.*, 2020, **142**, 21751–21757.
- Y. Song, L. Wang and Z. Xie, *Biotechnol. J.*, 2021, **16**, 1900382.
- E. J. Hong, D. G. Choi and M. S. Shim, *Acta Pharm. Sin. B*, 2016, **6**, 297–307.
- G. M. Calixto, J. Bernegossi, L. M. De Freitas, C. R. Fontana and M. Chorilli, *Molecules*, 2016, **21**, 342.
- M. G. Mokwena, C. A. Kruger, M.-T. Ivan and A. Heidi, *Photodiagn. Photodyn. Ther.*, 2018, **22**, 147–154.
- G. Chen, I. Roy, C. Yang and P. N. Prasad, *Chem. Rev.*, 2016, **116**, 2826–2885.
- H. Abrahamse and M. R. Hamblin, *Biochem. J.*, 2016, **473**, 347–364.
- J. Karges, M. Jakubaszek, C. Mari, K. Zarschler, B. Goud, H. Stephan and G. Gasser, *ChemBioChem*, 2020, **21**, 531–542.

- 71 L. N. Lameijer, S. L. Hopkins, T. G. Brevé, S. H. C. Askes and S. Bonnet, *Chem. – Eur. J.*, 2016, **22**, 18484–18491.
- 72 Y. Huang, W. Huang, L. Chan, B. Zhou and T. Chen, *Biomaterials*, 2016, **103**, 183–196.
- 73 P. Kaspler, S. Lazic, S. Forward, Y. Arenas, A. Mandel and L. Lilge, *Photochem. Photobiol. Sci.*, 2016, **15**, 481–495.
- 74 J. Karges, J. Li, L. Zeng, H. Chao and G. Gasser, *ACS Appl. Mater. Interfaces*, 2020, **12**, 54433–54444.
- 75 V. Novohradsky, A. Zamora, A. Gandioso, V. Brabec, J. Ruiz and V. Marchán, *Chem. Commun.*, 2017, **53**, 5523–5526.
- 76 J. Karges, D. Díaz-García, S. Prashar, S. Gómez-Ruiz and G. Gasser, *ACS Appl. Biol. Mater.*, 2021, **4**, 4394–4405.
- 77 W. Sun, Y. Wen, R. Thiramanas, M. Chen, J. Han, N. Gong, M. Wagner, S. Jiang, M. S. Meijer, S. Bonnet, H.-J. Butt, V. Mailänder, X.-J. Liang and S. Wu, *Adv. Funct. Mater.*, 2018, **28**, 1804227.
- 78 N. Soliman, L. K. McKenzie, J. Karges, E. Bertrand, M. Tharaud, M. Jakubaszek, V. Guérineau, B. Goud, M. Hollenstein, G. Gasser and C. M. Thomas, *Chem. Sci.*, 2020, **11**, 2657–2663.
- 79 S. H. C. Askes, A. Bahreman and S. Bonnet, *Angew. Chem., Int. Ed.*, 2014, **53**, 1029–1033.
- 80 Y. Boucher, L. T. Baxter and R. K. Jain, *Cancer Res.*, 1990, **50**, 4478–4484.
- 81 K. Greish, *Methods Mol. Biol.*, 2010, **624**, 25–37.
- 82 H. He, L. Liu, E. E. Morin, M. Liu and A. Schwendeman, *Acc. Chem. Res.*, 2019, **52**, 2445–2461.
- 83 R. Wang, X. Li and J. Yoon, *ACS Appl. Mater. Interfaces*, 2021, **13**, 19543–19571.
- 84 K. Qiu, Y. Chen, T. W. Rees, L. Ji and H. Chao, *Coord. Chem. Rev.*, 2019, **378**, 66–86.
- 85 L. Ke, F. Wei, L. Xie, J. Karges, Y. Chen, L. Ji and H. Chao, *Angew. Chem., Int. Ed.*, 2022, **61**, e202205429.
- 86 D. Van Straten, V. Mashayekhi, H. S. De Bruijn, S. Oliveira and D. J. Robinson, *Cancers*, 2017, **9**, 19.
- 87 Y. Hao, C. K. Chung, Z. Yu, R. V. Huis in 't Veld, F. A. Ossendorp, P. ten Dijke and L. J. Cruz, *Pharmaceutics*, 2022, **14**, 120.
- 88 S. Wang, G. Yu, W. Yang, Z. Wang, O. Jacobson, R. Tian, H. Deng, L. Lin and X. Chen, *Adv. Sci.*, 2021, **8**, 2002927.
- 89 W. Li, Z. Zhang, J. Liu, B. Wang, G. Pu, J. Li, Y. Huang and M. Chu, *Acta Biomater.*, 2022, **146**, 341–356.
- 90 S. Wang, G. Yu, W. Yang, Z. Wang, O. Jacobson, R. Tian, H. Deng, L. Lin and X. Chen, *Adv. Sci.*, 2021, **8**, 2002927.
- 91 X. Huang, L. Tang, L. Xu, Y. Zhang, G. Li, W. Peng, X. Guo, L. Zhou, C. Liu and X.-C. Shen, *J. Mater. Chem. B*, 2022, **10**, 7717–7731.
- 92 Y. Zhu, H. Shi, T. Li, J. Yu, Z. Guo, J. Cheng and Y. Liu, *ACS Appl. Mater. Interfaces*, 2020, **12**, 18309–18318.
- 93 L. M. Chong, D. J. H. Tng, L. L. Y. Tan, M. L. K. Chua and Y. Zhang, *Appl. Phys. Rev.*, 2021, **8**, 041322.
- 94 Z. Liu, Z. Xie, W. Li, X. Wu, X. Jiang, G. Li, L. Cao, D. Zhang, Q. Wang, P. Xue and H. Zhang, *J. Nanobiotechnol.*, 2021, **19**, 160.
- 95 M. Zhang, X. Liu, Y. Mao, Y. He, J. Xu, F. Zheng, W. Tan, S. Rong, Y. Chen, X. Jia and H. Li, *ACS Appl. Mater. Interfaces*, 2022, **14**, 27551–27563.
- 96 F. Zhang, Y. Liu, B. Yang, P. Guan, J. Chai, G. Wen and B. Liu, *J. Mater. Chem. B*, 2021, **9**, 2417–2427.
- 97 C. Xu, J. Nam, H. Hong, Y. Xu and J. J. Moon, *ACS Nano*, 2019, **13**, 12148–12161.
- 98 Z. Feng, T. Zhu, L. Wang, T. Yuan, Y. Jiang, X. Tian, Y. Tian and Q. Zhang, *Inorg. Chem.*, 2022, **61**, 12652–12661.
- 99 Y. Jiang, W. Meng, L. Wu, K. Shao, L. Wang, M. Ding, J. Shi and X. Kong, *Adv. Healthcare Mater.*, 2021, **10**, 2100789.
- 100 Y. Shi, D. Zhu, D. Wang, B. Liu, X. Du, G. Wei and X. Zhou, *Coord. Chem. Rev.*, 2022, **471**, 214725.
- 101 W. Guo, F. Wang, D. Ding, C. Song, C. Guo and S. Liu, *Chem. Mater.*, 2017, **29**, 9262–9274.
- 102 J. Sun, S. Kormakov, Y. Liu, Y. Huang, D. Wu and Z. Yang, *Molecules*, 2018, **23**, 1704.
- 103 F. Wei, S. Kuang, T. W. Rees, X. Liao, J. Liu, D. Luo, J. Wang, X. Zhang, L. Ji and H. Chao, *Biomaterials*, 2021, **276**, 121064.
- 104 F. Wei, J. Karges, J. Shen, L. Xie, K. Xiong, X. Zhang, L. Ji and H. Chao, *Nano Today*, 2022, **44**, 101509.
- 105 J. Shen, J. Karges, K. Xiong, Y. Chen, L. Ji and H. Chao, *Biomaterials*, 2021, **275**, 120979.
- 106 L. Wang, X. Qu, Y. Zhao, Y. Weng, G. I. N. Waterhouse, H. Yan, S. Guan and S. Zhou, *ACS Appl. Mater. Interfaces*, 2019, **11**, 35228–35237.
- 107 L. Jiao, H. Yan, Y. Wu, W. Gu, C. Zhu, D. Du and Y. Lin, *Angew. Chem., Int. Ed.*, 2020, **59**, 2565–2576.
- 108 X. Lu, S. Gao, H. Lin and J. Shi, *Small*, 2021, **17**, 2004467.
- 109 E. Ju, K. Dong, Z. Chen, Z. Liu, C. Liu, Y. Huang, Z. Wang, F. Pu, J. Ren and X. Qu, *Angew. Chem., Int. Ed.*, 2016, **55**, 11467–11471.
- 110 C. Wang, F. Cao, Y. Ruan, X. Jia, W. Zhen and X. Jiang, *Angew. Chem., Int. Ed.*, 2019, **58**, 9846–9850.
- 111 M. Huo, L. Wang, Y. Wang, Y. Chen and J. Shi, *ACS Nano*, 2019, **13**, 2643–2653.
- 112 D. Wang, H. Wu, S. Z. F. Phua, G. Yang, W. Qi Lim, L. Gu, C. Qian, H. Wang, Z. Guo, H. Chen and Y. Zhao, *Nat. Commun.*, 2020, **11**, 357.
- 113 M. Lismont, L. Dreesen and S. Wuttke, *Adv. Funct. Mater.*, 2017, **27**, 1606314.
- 114 P. Gao, Y. Chen, W. Pan, N. Li, Z. Liu and B. Tang, *Angew. Chem., Int. Ed.*, 2021, **60**, 16763–16776.
- 115 T. Luo, G. T. Nash, Z. Xu, X. Jiang, J. Liu and W. Lin, *J. Am. Chem. Soc.*, 2021, **143**, 13519–13524.
- 116 K. Lu, C. He and W. Lin, *J. Am. Chem. Soc.*, 2014, **136**, 16712–16715.
- 117 D. Jin, J. Zhang, Y. Huang, X. Qin, J. Zhuang, W. Yin, S. Chen, Y. Wang, P. Hua and Y. Yao, *Dalton Trans.*, 2021, **50**, 1189–1196.
- 118 L. He, Q. Ni, J. Mu, W. Fan, L. Liu, Z. Wang, L. Li, W. Tang, Y. Liu, Y. Cheng, L. Tang, Z. Yang, Y. Liu, J. Zou, W. Yang, O. Jacobson, F. Zhang, P. Huang and X. Chen, *J. Am. Chem. Soc.*, 2020, **142**, 6822–6832.
- 119 N. Ma, Y. Li, H. Xu, Z. Wang and X. Zhang, *J. Am. Chem. Soc.*, 2010, **132**, 442–443.
- 120 F. Fan, S. Ji, C. Sun, C. Liu, Y. Yu, Y. Fu and H. Xu, *Angew. Chem., Int. Ed.*, 2018, **57**, 16426–16430.
- 121 Y. Liu, N. Wen, K. Li, M. Li, S. Qian, S. Li, T. Jiang, T. Wang, Y. Wu and Z. Liu, *Mol. Pharmaceutics*, 2022, **19**, 805–818.
- 122 L. Ke, F. Wei, X. Liao, T. W. Rees, S. Kuang, Z. Liu, Y. Chen, L. Ji and H. Chao, *Nanoscale*, 2021, **13**, 7590–7599.
- 123 D. Wei, Y. Yu, X. Zhang, Y. Wang, H. Chen, Y. Zhao, F. Wang, G. Rong, W. Wang, X. Kang, J. Cai, Z. Wang, J.-Y. Yin, M. Hanif, Y. Sun, G. Zha, L. Li, G. Nie and H. Xiao, *ACS Nano*, 2020, **14**, 16984–16996.
- 124 C.-N. Lok, T. Zou, J.-J. Zhang, I. W.-S. Lin and C.-M. Che, *Adv. Mater.*, 2014, **26**, 5550–5557.
- 125 J.-J. Zhang, W. Lu, R. W.-Y. Sun and C.-M. Che, *Angew. Chem., Int. Ed.*, 2012, **51**, 4882–4886.
- 126 J. L.-L. Tsai, T. Zou, J. Liu, T. Chen, A. O.-Y. Chan, C. Yang, C.-N. Lok and C.-M. Che, *Chem. Sci.*, 2015, **6**, 3823–3830.
- 127 J. Li, L. Zeng, Z. Wang, H. Chen, S. Fang, J. Wang, C.-Y. Cai, E. Xing, X. Liao, Z.-W. Li, C. R. Ashby Jr, Z.-S. Chen, H. Chao and Y. Pan, *Adv. Mater.*, 2022, **34**, 2100245.
- 128 M. Fornasier, S. Biffi, B. Bortot, P. Macor, A. Manhart, F. R. Wurm and S. Murgia, *J. Colloid Interface Sci.*, 2020, **580**, 286–297.
- 129 J. Chen, A. C. Sedgwick, S. Sen, Y. Ren, Q. Sun, C. Chau, J. F. Arambula, T. Sarma, L. Song, J. L. Sessler and C. Liu, *Chem. Sci.*, 2021, **12**, 9916–9921.
- 130 J. Karges, M. Tharaud and G. Gasser, *J. Med. Chem.*, 2021, **64**, 4612–4622.
- 131 D. Wei, Y. Huang, B. Wang, L. Ma, J. Karges and H. Xiao, *Angew. Chem., Int. Ed.*, 2022, **61**, e202201486.
- 132 E. R. Johnson, S. Keinan, P. Mori-Sánchez, J. Contreras-García, A. J. Cohen and W. Yang, *J. Am. Chem. Soc.*, 2010, **132**, 6498–6506.
- 133 M.-S. Yuan, D.-E. Wang, P. Xue, W. Wang, J.-C. Wang, Q. Tu, Z. Liu, Y. Liu, Y. Zhang and J. Wang, *Chem. Mater.*, 2014, **26**, 2467–2477.
- 134 M. O. Sinnokrot, E. F. Valeev and C. D. Sherrill, *J. Am. Chem. Soc.*, 2002, **124**, 10887–10893.
- 135 X. Li, D. Lee, J.-D. Huang and J. Yoon, *Angew. Chem., Int. Ed.*, 2018, **57**, 9885–9890.
- 136 Y.-Y. Zhao, L. Zhang, Z. Chen, B.-Y. Zheng, M. Ke, X. Li and J.-D. Huang, *J. Am. Chem. Soc.*, 2021, **143**, 13980–13989.
- 137 T. X.-Q. Zhou, M. Xiao, V. Ramu, J. Hilgendorf, X. Li, P. Papadopolou, M. A. Siegler, A. Kros, W. Sun and S. Bonnet, *J. Am. Chem. Soc.*, 2020, **142**, 10383–10399.
- 138 W. Yu, L. Sisi, Y. Haiyan and L. Jie, *RSC Adv.*, 2020, **10**, 15328–15345.
- 139 V. Negri, J. Pacheco-Torres, D. Calle and P. López-Larrubia, *Top. Curr. Chem.*, 2020, **378**, 15.
- 140 R.-Q. Li, C. Zhang, B.-R. Xie, W.-Y. Yu, W.-X. Qiu, H. Cheng and X.-Z. Zhang, *Biomaterials*, 2019, **194**, 84–93.

- 141 J.-Y. Zhou, W.-J. Wang, C.-Y. Zhang, Y.-Y. Ling, X.-J. Hong, Q. Su, W.-G. Li, Z.-W. Mao, B. Cheng, C.-P. Tan and T. Wu, *Biomaterials*, 2022, **289**, 121757.
- 142 W. Lv, T. Yang, Q. Yu, Q. Zhao, K. Y. Zhang, H. Liang, S. Liu, F. Li and W. Huang, *Adv. Sci.*, 2015, **2**, 1500107.
- 143 J. Wang, Q. Ni, Y. Wang, Y. Zhang, H. He, D. Gao, X. Ma and X. J. Liang, *J. Controlled Release*, 2021, **331**, 282–295.
- 144 Y. Zhang, H. Fu, S. Chen, B. Liu, W. Sun and H. Gao, *Chem. Commun.*, 2020, **56**, 762–765.
- 145 Z. Dong, L. Feng, W. Zhu, X. Sun, M. Gao, H. Zhao, Y. Chao and Z. Liu, *Biomaterials*, 2016, **110**, 60–70.
- 146 X. Wei, M. Song, G. Jiang, M. Liang, C. Chen, Z. Yang and L. Zou, *Theranostics*, 2022, **12**, 5272–5298.