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Synthesis, molecular docking analysis and in vitro evaluation of new heterocyclic hybrids of 4-aza-podophyllotoxin as potent cytotoxic agents†

Ha Thanh Nguyen, **D*** Ket Tran Van, **D** Hai Pham-The, **D** Julien Braire, **D** Phuong Hoang Thi, **Tuan Anh Nguyen, **Quynh Giang Nguyen Thi, **Tuyet Anh Dang Thi, **D** Giang Le-Nhat-Thuy, **D** Tu Anh Le Thi, **Doan Vu Ngoc** and Tuyen Nguyen Van **D** Anh **D** Tu Anh Le Thi, **Doan Vu Ngoc** Tuyen Nguyen Van **D** Tu Anh Le Thi, **Doan Vu Ngoc** Tuyen Nguyen Van **D** Tuyen Nguyen Nguyen

Two different synthetic approaches to novel heterocyclic hybrid compounds of 4-azapodophyllotoxin were investigated. The obtained products were characterized by infrared spectroscopy, nuclear magnetic resonance spectroscopy, and high-resolution mass spectrometry. MTT protocol was then performed to examine the cytotoxic activity of these products against KB, HepG2, A549, MCF7, and Hek-293 cell lines. The cytotoxic assessment indicated that all products displayed moderate to high cytotoxicity against all tested cancer cell lines. The most active compound 13k containing the 2-methoxypyridin-4-yl group exhibited selective cytotoxicity against KB, A549, and HepG2 cell lines with the IC50 values ranging from 0.23 to 0.27 μ M, which were between 5- to 10-fold more potent than the positive control ellipticine. Compounds 13a (HetAr = thiophen-3-yl) and 13d (HetAr = 5-bromofuran-2-yl) displayed high cytotoxic selectivity for A549 and HepG2 cancer cell lines when compared to the other cancer cell lines and low toxicity to the normal Hek-293 cell line. Molecular docking study was conducted to evaluate the interaction of new synthesized compounds with the colchicine-binding-site of tubulin. Besides that, physicochemical and pharmacokinetic properties of the most active compounds 13h,k were predicted.

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Introduction

Heterocycles, a class of cyclic organic compounds containing at least one ring hetero atom, have played an important role in pharmaceutical development due to their variety of biological activities such as anti-fungal,¹ antimicrobial,² anti-inflammatory,³,⁴ anti-diabetic,⁵ cytotoxic,⁶,⁻ anti-tumor, anti-cancer,⁶,⁰ anti-viral,¹⁰ acetylcholinesterase inhibitory,¹¹ and SARS-CoV2 inhibitory activities.¹² To date, heterocycles have presented themselves as the basic core of most marketed drugs. For examples, pyridine, a nitrogen-based heterocycle, has been known as the structural unit of several targeted cancer drugs such as sorafenib,¹³ regorafenib,¹⁴ and crizotinib.¹⁵ Ritonavir (1), a sulfur-based heterocycle drug, is an FDA-approved HIV

Podophyllotoxin, an important plant-derived natural product, has several semisynthetic derivatives such as etoposide, teniposide and etophos, which have been employed in chemotherapy for various cancer types.³⁰ Although podophyllotoxin and its derivatives have been known as tubulin polymerization or DNA topoisomerase II inhibitors, they are too toxic for therapeutic use and causes some side effects. To overcome such problems, great interest has been paid to the synthesis of 4-azapodophyllotoxin derivatives in order to obtain the new anticancer agents with little side effects (Fig. 2). 4-Aza-

protease inhibitor which can act as a potential cancer therapeutic agent (Fig. 1).^{16,17} Amiodarone (2), a benzofuran derivative, is an antiarrhythmic medication used in the treatment of heart rate problems, ventricular fibrillation and ventricular tachycardia (Fig. 1).¹⁸ The FDA has approved Jelmyto (mitomycin C, 3), a pyrrole-fused quinone molecule, for treatment of adult patients with low-grade upper tract urothelial cancer (Fig. 1).¹⁹ The quinone moiety of mitomycin C is enzymatically reduced into an active hydroquinone intermediate, which has an extraordinary ability to crosslink DNA with high efficiency.²⁰ Owing to the significant role of heterocyclic ring systems in drug discovery and development, many studies have been undertaken to develop biologically active heterocyclic compounds.^{21–29}

^aInstitute of Chemistry, Vietnam Academy of Science and Technology, 18 Hoang Quoc Viet, Cau Giay, Hanoi, Vietnam. E-mail: hathanhnguyen1512@gmail.com

^bGraduate University of Science and Technology, Vietnam Academy of Science and Technology, 18 Hoang Quoc Viet, Cau Giay, Hanoi, Vietnam

University of Science and Technology of Hanoi, Vietnam Academy of Science and Technology, 18 Hoang Quoc Viet, Cau Giay, Hanoi, Vietnam

^aMilitary Technical Academy, 236 Hoang Quoc Viet, Bac Tu Liem, Hanoi, Vietnam

^{*}Université de Rennes 1, 2 Av. du Professeur Léon Bernard, 35042 Rennes, France † Electronic supplementary information (ESI) available. See DOI: https://doi.org/10.1039/d3ra07396c

Fig. 1 Chemical structure of FDA approved drugs containing heterocyclic ring system.

Mitomycin C

Amiodarone

podophyllotoxins have exhibited a broad spectrum of biological activities including cancer cell growth inhibition,31 caspase-3 dependent apoptosis,32 cell cycle arrest in the G2/M phase,33 inhibition of tubulin polymerization and cellular microtubule disassembly³⁴⁻³⁶ and vascular disruption effect.³⁷ Several hybrid compounds of 4-aza-podophyllotoxin with heterocycles have been developed and evaluated the anticancer activity.38-41 For instance, Kamal et al., has prepared a series of 4-azapodophyllotoxin with heterocycles including benzothiazole-

hybrids 6, pyrimidine-podophyllotoxin podophyllotoxin hybrids 7, and, indole-podophyllotoxin hybrids 8 which have inhibited tubulin polymerization, induced cell cycle arrest at G2/M phase and caspase-3 dependent apoptotic cell death in non-small lung A549 cell line.³³ Pyrazole-podophyllotoxin hybrids which have been synthesized by Magedov et al., have showed apoptosis induction in cancerous Jurkat cells even after a short 24 h exposure. Azaanthraquinone-podophyllotoxin hybrids 11, which have been designed by replacing of γ -butyrolactone ring D of 4-aza-podophyllotoxin with naphthoquinone, have exhibited medium cytotoxic activity against hepatoma carcinoma HepG2 and Hela cell lines. 42 Interestingly, azaanthraquinone-podophyllotoxin hybrids 12, reported in our previous study,43,44 have possessed excellent cytotoxic activity against hepatoma carcinoma HepG2 and non-small lung SK-Lu-1 cell lines with $IC_{50} < 0.04 \mu M$ (Fig. 2). Hybrid compounds 12 were demonstrated to exhibit cytotoxicity by inducing cell cycle arrest at G2/M phase, activating caspase-3/7 activation, and promoting apoptosis in a concentration-dependent manner. In spite of the many researches focused on the synthesis of azapodophyllotoxin hybrids, the replacement of ring E with heterocycles has not attracted enough attention so far. Accordingly, in the view of the above mentioned facts and as a continuation of our efforts on the synthesis of biologically active compounds, 45-49 herein, we have focused our research interest toward the synthesis of novel azaanthraquinonepodophyllotoxin hybrids with heterocyclic ring E 13 via

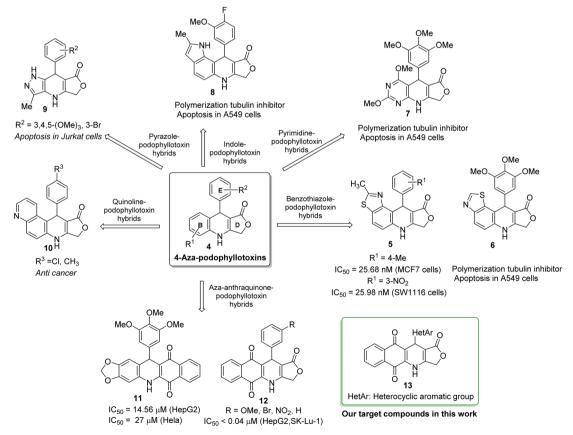


Fig. 2 Hybrid compounds of 4-aza-podophyllotoxin with heterocycles.

microwave-assisted multicomponent reactions (Fig. 2), and evaluation of their cytotoxic activity against various cancer cell lines by using MTT colorimetric method. Besides, synthesized compounds 13 were docked into the binding sites of colchicine in tubulin and their binding energies as well as physicochemical and pharmacokinetic properties were determined.

Results and discussion

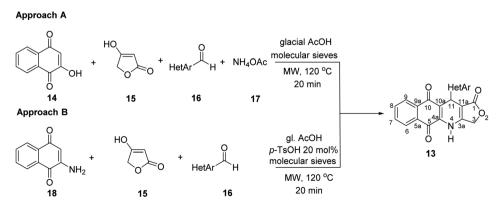
Chemistry

Two convenient multicomponent synthetic approaches have been adopted to synthesize the target compounds 13 (Scheme 1). In the approach A, a range of structurally diverse heteroaromatic aldehydes 16 was subjected to the four-component reactions with 2-hydroxy-naphthoquinone (14), tetronic acid (15) and ammonium acetate (17) in glacial acetic acid (gl. AcOH) at $120\,^{\circ}$ C in $20\,$ min under microwave irradiation, leading to products 13 in low yield ranging from 20 to 34% (Table 1). The formation of products 13 possibly started from Knoevenagel condensation of 2-hydroxy-1,4-naphthoquinone (14) with aldehydes, followed by Michael addition to 4-aminofuran-2(5H)-one 19, produced from the reaction of tetronic acid 15 with ammonia. The adduct 21 underwent tautomerization, intramolecular cyclization and dehydration to form products 13 (Scheme 2).

In the approach B, three-component synthesis of target products 13 from 2-amino-naphthoquinone, heteroaromatic aldehydes, and tetronic acid was investigated. In our preliminary studies, we indicated that the use of p-toluenesulfonic acid (p-TsOH) as an acid catalyst and gl. AcOH as a solvent promoted the three-component reactions to generate products in higher yields. Thus, p-TsOH (20 mol%) was supplemented to the reaction mixtures in gl. AcOH, and the reactions were then stirred at 120 °C, under microwave irradiation (150 W). The reactions proceeded rapidly to completion within 20 min. As shown in Table 1, under these optimized conditions, products 13 were afforded in good yields (60-74%). The synthetic approaches proceeded via a sequential steps including Knoevenagel condensation of 2amino-1,4-naphthoquinone (14) with heteroaromatic aldehydes, Michael addition to tetronic acid, tautomerization, intramolecular cyclization and dehydration (Scheme 2).

It is worth mentioning that the approach B (synthesis by three-component reactions) is likely to be more efficient for the synthesis of target products 13 in comparison with the approach A (synthesis by four-component reactions). In approach B, the effect of heterocyclic substituents on aldehyde on the outcome of reactions could be observed more clearly. For instance, pyridinecarboxaldehyde derivatives 16k-m displayed the highest reactivity to give the desired products 13k-m in high yield (72–74%). The fusion of a benzene ring with five-membered aromatic heterocyclic ring of aldehydes improved the reaction yields compared to five-membered heterocyclic aldehydes. Benzo[b]thiophene-3-carbaldehyde (16e) furnished 70% yield of the product 13c, whereas, thiophene-3-carbaldehyde (16a) afforded 13a in 64% yield.

The structure of obtained compounds were examined by IR, ¹H and ¹³C NMR, HSQC, HMBC, DEPT, and HRMS. Compound 13a was focused for illustrating the 1H and 13C assignments. In general, the ¹H-NMR spectrum of **13a**, presented a typical singlet at $\delta_{\rm H}$ 10.60 ppm corresponding to an –NH group, two doublets of doublets at δ_H 8.05 (1H, dd, J = 1.2, 7.8 Hz, H6), and 7.93 (1H, dd, J= 0.6, 7.8 Hz, H9), two triples of doublets at $\delta_{\rm H}$ 7.85 (1H, td, J = 1.2, 7.2 Hz, H8), 7.81 (1H, td, J = 1.2, 7.2 Hz, H7) corresponding to four aromatic protons in naphthoquinone ring. Three aromatic protons in thiophene ring appeared at $\delta_{\rm H}$ 7.38 (1H, dd, J=3.0, 4.8 Hz, H-5'), 7.24 (1H, dd, J = 0.6, 3.0 Hz, H-2'), 7.07 (1H, dd, J =1.2, 4.8 Hz, H-4'). Moreover, the typical singlet at 5.11 ppm was referred to the H-11 proton, two doublets were observed at 4.94 ppm and 4.90 ppm with a coupling constant of 16.8 Hz assignable to the 3-CH₂ protons in γ -butyrolactone ring. On the other hand, the ¹³C NMR spectrum of product 13a displayed 19 carbon resonances, which were fully assigned with DEPT, ¹H-¹³C HSQC and ¹H-¹³C HMBC support. The DEPT-135 shows 8 positive peaks and 1 negative peak. The negative peak at 66.1 ppm corresponded to methylene (CH2) carbon C-3. Proton signal (H-11) at 5.11 ppm showed proton-carbon couplings to C-10 at 182.1 ppm, C-1 at 171.2 ppm, C-3a at 156.1 ppm, C-4a at 139.2 ppm, C-3' at 145.0 ppm, C-4' at 127.7 ppm, C-2' at 122.4 ppm, C-10a at 118.0 ppm, and C-11a at 101.6 ppm (Fig. 3). Moreover, under positive HR-ESI-MS conditions, $[M + H]^+$ at m/z 350.0484, which confirms the molecular formula for compound 13a is $C_{19}H_{11}NO_4S$.



Scheme 1 Synthesis of compounds 7 under microwave irradiation.

Table 1 Synthesis of compounds 13 under microwave irradiation

		Yield (%)		
Park		Approach	Approach	
Entry	Product	A	В	
1	13a	29	64	
2	13b	25	62	
3	13c	20	60	
4	13 d	20	62	
5	13e	33	70	
6	13f	32	68	
7	13g	33	69	
8	13h	28	64	
9	13i	27	63	
10	13j	20	61	
11	13k	34	74	
12	13 l	32	72	
13	13m	33	72	

Cytotoxicity of synthesized compounds

All of these products were further studied cytotoxicity evaluaagainst cultured A549 (human lung vitro

adenocarcinoma), MCF7 (human breast adenocarcinoma), KB (human mouth epidermal carcinoma), HepG2 (human hepatocellular carcinoma) and Hek-293 (human embryonic kidney

Scheme 2 Plausible mechanism for the formation of obtained products 13.

Fig. 3 Key correlations of protons H-11 (black arrows), H-3 (red arrows), H-9 (blue arrows), and H-5' (green arrows) in 1 H- 13 C HMBC spectrum of compound **13a**.

293) cell lines, in comparison to ellipticine, the positive control, by using MTT assay (Table 2). The selective index (SI) of products was calculated as the ratio of the IC50 value in Hek-293 (human embryonic kidney 293) cell line to the IC₅₀ value in cancer cell line (Table 3). As illustrated in Table 2, all products showed high and moderate cytotoxic activity to all tested cancer cell lines with IC_{50} values ranging from 0.16 to 14.22 μM . Cytotoxic activity of compounds 13a,h,k was better than that of ellipticine against 4 tested cancer cell lines, a fact supporting their anti-cancer activity. Besides, products 13f (HetAr = benzofuran-3-yl) exerted almost 1.8-fold higher toxicity to KB and HepG2 cells when compared with ellipticine. Notably, product 13k (HetAr = 2-methoxypyridin-4-yl) exhibited the significantly potent and selective cytotoxicity against KB, A549, and HepG2 cell lines with the IC50 values ranging from 0.23 to 0.27 μM. The cytotoxic activity of product 13k toward KB, A549, and HepG2 cancer cells was approximately 10 times higher in relation to normal Hek-293 cells (Table 3). Product 13h (HetAr = 2-chlorothiazol-5-yl) showed the highest growth inhibitory activity against A549 and MCF7 cancer cell lines and normal Hek-293 cells with IC₅₀ values of 0.16, 1.07, and 1.38 μ M, respectively. Compound 13a (HetAr = thiophen-3-yl) displayed higher cytotoxic selectivity for A549 cancer cells when compared to the other cancer cell lines and low toxicity to normal Hek-293 cell line with $IC_{50} = 22.90 \mu M$, what can indicate their specificity

to cancer cell lines. In particularly, compound 13d (HetAr = 5-bromofuran-2-yl) exhibited selective cytotoxicity against HepG2 cancer cells with IC $_{50}$ value of 0.30 μ M, which was 122-fold higher than its IC $_{50}$ value for Hek-293 cells (Table 3).

Molecular docking study

Based on experimental assessments, it can be initially identified that compounds 13a-13m are good cytotoxic agents. Due to crucial function in mitosis and cell division, microtubules, which are produced through the polymerization of heterodimers of α and β -tubulins, have been a desirable target in the development of novel anticancer medicines. Meanwhile, podophyllotoxin has long been regarded as a potent microtubule destabilizing agent that binds to the tubulin colchicine site, inhibiting tubulin polymerization and suppressing the production of microtubules. Therefore, in this study, we used molecular docking to evaluate the interaction of novel synthesized substances with this molecular target. The natural tubulin substrate colchicine will be used as a positive control.

Table 3 Selective index for active compounds

		Selective index			
Entry	Compound	KB	A549	HepG2	MCF7
1	13a	16.00	47.90	16.00	11.12
3	13b	1.37	4.47	1.12	1.09
4	13c	1.02	4.75	3.88	1.20
5	13d	13.19	18.00	122.40	3.58
6	13e	1.02	1.12	1.07	0.85
7	13f	17.98	2.81	15.10	1.56
8	13g	1.31	4.18	1.03	1.16
9	13h	1.23	8.55	1.71	1.29
10	13i	1.14	5.40	1.57	1.18
11	13j	0.63	0.69	0.86	1.16
12	13k	9.30	9.30	10.81	1.86
13	13l	2.18	4.74	1.85	1.39
14	13m	3.61	5.92	2.87	2.31
15	Ellipticine	2.88	3.94	2.65	2.68

Table 2 Cytotoxicity of the products 7 against KB, HepG2, A549, MCF7, and Hek-293 cell lines

	${ m IC}_{50},\mu{ m M}$					
Product	КВ	A549	HepG2	MCF7	Hek-293	
13a	1.43 ± 0.14	$\textbf{0.48} \pm \textbf{0.03}$	1.43 ± 0.15	2.06 ± 0.66	22.90 ± 2.18	
13b	$\textbf{10.05} \pm \textbf{1.17}$	3.09 ± 0.39	12.36 ± 1.29	12.63 ± 1.14	13.80 ± 1.41	
13c	14.22 ± 1.23	3.06 ± 0.27	3.75 ± 0.69	12.15 ± 1.14	14.55 ± 1.35	
13d	2.81 ± 0.32	2.06 ± 0.22	$\textbf{0.30} \pm \textbf{0.06}$	10.36 ± 1.29	37.12 ± 3.30	
13e	$\textbf{10.94} \pm \textbf{1.40}$	9.94 ± 0.73	10.42 ± 1.18	13.07 ± 1.60	11.17 ± 0.85	
13f	1.10 ± 0.09	7.02 ± 0.91	1.30 ± 0.13	12.65 ± 1.20	19.69 ± 1.70	
13g	9.94 ± 0.65	3.13 ± 0.20	12.67 ± 1.40	11.27 ± 1.78	13.07 ± 1.70	
13h	$\textbf{1.12} \pm \textbf{0.10}$	$\textbf{0.16} \pm \textbf{0.02}$	$\textbf{0.81} \pm \textbf{0.11}$	$\textbf{1.07} \pm \textbf{0.10}$	$\textbf{1.38} \pm \textbf{0.18}$	
13i	10.74 ± 1.27	2.27 ± 0.22	7.80 ± 0.53	10.41 ± 0.80	12.26 ± 1.13	
13j	6.00 ± 0.63	5.43 ± 0.39	4.35 ± 0.57	3.24 ± 0.36	3.75 ± 0.69	
13k	$\textbf{0.27} \pm \textbf{0.24}$	$\textbf{0.27} \pm \textbf{0.21}$	$\textbf{0.23} \pm \textbf{0.02}$	1.34 ± 0.16	2.48 ± 0.32	
13l	4.69 ± 0.58	2.15 ± 0.25	5.52 ± 0.86	7.34 ± 0.94	10.21 ± 1.24	
13m	$\textbf{4.63} \pm \textbf{1.04}$	2.83 ± 0.54	5.83 ± 0.86	7.26 ± 1.29	16.73 ± 1.94	
Ellipticine	1.95 ± 0.12	1.42 ± 0.08	2.11 ± 0.16	2.09 ± 0.16	5.60 ± 0.37	

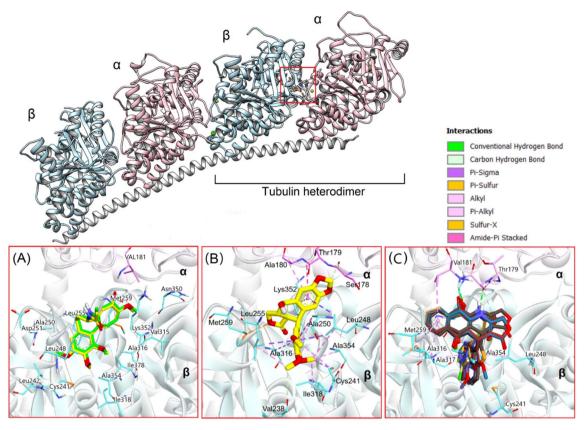


Fig. 4 (A) Redocked poses (green) and original poses (yellow) of colchicine in tubulin; (B) docking pose of podophyllotoxin, and (C) docking poses of 13 synthesized compounds in colchicine binding site

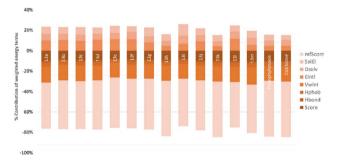


Fig. 5 Contribution (%) of the different energy terms to the binding affinity (score) of tubulin-ligands: the van der Waals interaction energy (VwInt), Hydrogen Bond energy (Hbond), the hydrophobic energy (Hphob), internal conformation energy (Eintl), the desolvation of exposed H-bond donors and acceptors (Dsolv), the solvation electrostatics energy change upon binding (SolEl), the potential of mean force score (mfScore).

Firstly, colchicine were redocked into the active site in tubulin to validate the docking process. As the results, the redocked conformers tightly bound to the colchicine binding site of tubulin with a very high matching with the co-crystallized structure (RMSD = 0.3339 Å). Moreover, it preserves the key interactions such as hydrogen bonds with Val181, Ala180... of the α subunit through its methoxytropone ring, hydrophobic interactions with Leuβ242, Cysβ241, Alaβ354, and Alaβ316 (Fig. 4A),⁵² suggesting the validity of established docking protocol.

In the next step, podophyllotoxin, which is considered reference compound was also docked into the binding site of tubulin. The results were similar to those previously determined.44 As can be seen in Fig. 4, this compound exhibited a large interaction network with residues in the binding site. Some of the key interactions should be mentioned here are the hydrogen bonds with Val178, Ala180... of the α subunit and numerous hydrophobic interactions with Cys241, Ala250, Leu248, Leu255, Ala354, and Ala316 of the β subunit (Fig. 4B). The binding energy was determined to be $-18.68 \text{ kcal mol}^{-1}$, and hydrogen bonding and hydrophobic stacking interactions the most contributed components (Fig. 5). The results obtained confirmed the similarity and accuracy of the docking protocol.

After validating the docking protocol, all 13 synthesized compounds (13a-13m) were docked into the binding site of colchicine in tubulin, using the above validated protocol. The binding energies and the main interactions between the compounds and tubulin in comparison with podophyllotoxin are summarized in Table 4 and Fig. 4-6.

Overall, the docking results of 13 substances 13a-13m were similar to colchicine and those previously published.44 They are all capable of binding to the active site of colchicine in tubulin. The colchicine binding site was divided into three zones including zones 1, 2, and 3 according to Massaroti et al.53 zone 1 is surrounded by the residues Valα181, Serα178, Metβ259, Asn α 101, and is situated at the α subunit interface. Located in the β subunit, zone 2 is an accessory hydrophobic pocket made

Table 4 Binding energy between compounds 13a-13m and active site of tubulin compared to co-crystallized compounds

Cpd	Binding energ (kcal mol ⁻¹)	y Interaction with residues
13a	-16.38	H-bond: Val 181 α , Thr 179 α , Ala 180 α
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β
13b	-13.79	H-bond: Val 181α, Thr 179α, Ala 180α, Ala 354β, Ala 317β
		Hydrophobic: Leu 248β, Ala 316β, Lys 352β, Met 259β
13c	-14.55	H-bond: Val 181α, Thr 179α, Ala 180α
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β
13d	-15.29	H-bond: Val 181α, Thr 179α, Ala 180α
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β, Ile 318β, Cys 241β
13e	-14.09	H-bond: Val 181α, Thr 179α, Ala 180α
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β, Leu 255β, Cys 241β
13f	-16.45	H-bond: Val 181α, Thr 179α, Ala 180α, Asn 101α, Leu 255β, Ala 250β
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β, Leu 255β, Asn 258β, Ala 250β
13g	-15.08	H-bond: Val 181α, Thr 179α, Ala 180α, Leu 255β, Lys 352β
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β, Leu 255β, Asn 258β, Ala 250β, Cys 241β
13h	-17.86	H-bond: Val 181α, Thr 179α, Ala 180α
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β, Ile 318β, Cys 241β
13i	-13.83	H-bond: Val 181α, Ala 180α, Asn 101α, Ala 317β
		Hydrophobic: Ala 354β, Leu 248β, Leu 255β, Ala 316β, Lys 352β, Met 259β, Ile 318β
13j	-15.64	H-bond: Val 181α, Thr 179α, Ala 180α, Ala 317β, Lys 352β
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β
13k	-18.45	H-bond: Val 181α, Thr 179α, Ala 180α, Ala 317β
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β, Ile 318β, Cys 241β
13l	-15.78	H-bond: Val 181α, Thr 179α, Ala 180α, Ala 317β, Lys 352β
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β
13m	-16.8	H-bond: Val 181α, Ala 180α, Ala 317β, Ala 316β, Asn 101α, Asp251β, Ala 250β
		Hydrophobic: Ala 354β, Leu 248β, Ala 316β, Lys 352β, Met 259β, Leu255β, Thr 353β
Podophyloto	oxin -18.68	H-bond: Ser 178α, Ala 180α, Ala 317β, Ala 316β, Asn 101α, Asp251β, Ala 250β
		Hydrophobic: Val 238β, Cis 241β, Leu 248β, Ala 316β, Ala 318β, Lys 352β, Met 259β, Leu255β, Thr 353β
Colchicin	-19.18	H-bond: Val 181 α , Ala 180 α , Ala 250 β , Asp251 β , val 315 β , asn 350 β
		Hydrophobic: Leu 255β, Leu 248β, Leu 242β, Cys 241β, Ile 318β, Ala 354β, Ile 378β, Ala 316β, Met 359β, Lys 352β

up of the residues Lysβ352, Valβ318, Alaβ317, Alaβ316, Leuβ255, Alaβ250, Leuβ248, Leuβ242, and Cysβ241 zone 3, which is buried deeper inside the β subunit, is formed by the residues Thrβ239, Valβ238, Tyrβ202, Gluβ200, Pheβ169, Asnβ167, Glnβ136, and Ileβ4... Therein, these studied conformations with rigid structures are capable of stretching into the deeper zone 2 of the β -subunit. Almost all the synthesized compounds conserved these key interactions such as the hydrogen bond between the carbonyl group of the quinone ring and Valα181, Alaα180; the hydrophobic interactions which are mainly pi-alkyl stacking with most residues of zone 2 like Lysβ352, Alaβ317, Alaβ316, Leuβ255, Alaβ250, and so on. Moreover, all compounds formed hydrogen bonds between N and Thrα179 that do not exist in colchicine, always it bind to the active site more strongly.

The docking poses and binding energies also showed good ability to discriminate the good ligands from the bad ones. 13b and 13c were determined to be two of the weakest ligands of tubulin. As can be seen in Fig. 6, these compounds were quite different from 13k and 13h in alignment with podophyllotoxin. In fact, fewer interactions were observed in 13b and 13c, being several stacking interactions with HetAr with residues in the β subunit, such as Asn101, Ala250, Leu248, and Ala316 the most important ones. In particular, the orientation of HetAr in 13b and 13c were different from 13k and 13h, suggesting the change in their target interactions.

The relative binding energies were determined based on a GBSA/MM-type function implemented into ICM pro.66 In addition, other energy terms extracted from the final complex, including Hbond, Vwin. Accordingly, compound 13k has the strongest binding energy at -18.05 kcal mol⁻¹. Besides general interactions, it also formed hydrogen bonds between the lactone ring and Asn101α of zone 1, and between the carbonyl group of the quinone ring and Leuβ255 of zone 2. Additionally, it interacts better with more residues of zone 2 of colchicine binding site through its substituent. Compound 13h had good binding energy of -17.86 kcal mol⁻¹ resulted by key hydrogen bonds with residues of zone 1 and hydrophobic interactions which are similar to compound 13k (Fig. 6). The results also highlighted the highest values of Hbond and Hphob energies of 13h and 13k compared to the other derivatives, which correlated well with the order ranking of all the compounds (Fig. 5). There is a concurrence between the binding energy and the ICM docking mfScore which refers to the strength of the inhibitortubulin interaction. The mfScore values of 13h and 13k were -102.91 and -103.1 kcal mol⁻¹, which are similar to podophyllotoxin. Interestingly, Hbonds appear to have small impact on the binding ability of synthesized compounds. While hydrogen bonding energies of 13b and 13c are quite high, their overall energies computed by GBSA/MM-type and mfScore functions showed significantly lower than podophyllotoxin and colchicine (see Table S1†). For the generated docking

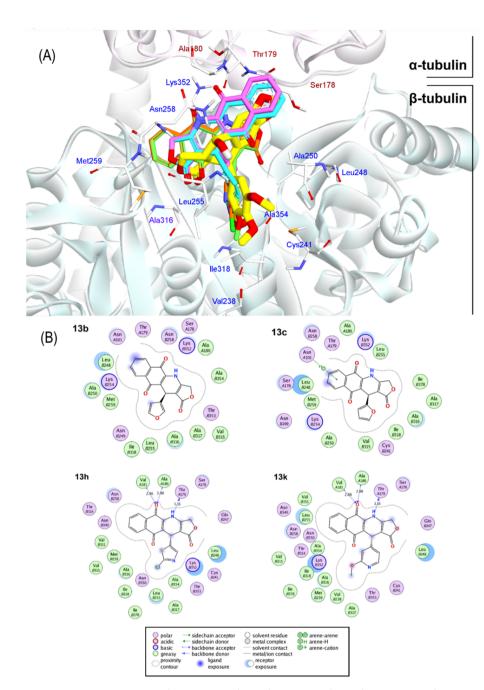


Fig. 6 (A) Superimposition of docking poses of 13b (purple carbon), 13c (cyan carbon), 13h (orange carbon), and 13k (green carbon) against podophyllotoxin (yellow carbon). (B) Topological interactions of 13b, 13c, 13h, and 13k with the residues in the active sites of α and β -tubulin domains (labeled with A and B prefixes).

conformations, the hydrophobic and van der Waals interaction appear to be the driving forces for binding ability of tested compounds. Over podophyllotoxin all, the docking results are in consistency with our previous findings, and in turn match well with experimental ranking using the cytotoxicity tests. ^{67,69}

Physicochemical and pharmacokinetic properties prediction

The above experimental test revealed that all 13 compounds exhibit good activity against a variety of cancer cell lines. In which, the two chemicals 13h and 13k have the most potential.

So, we performed assessments related to physicochemical and pharmacokinetic feature as a criteria for drug-likeness evaluation.

First, radar plot of the physicochemical properties in Fig. 7 showed that both compounds met most of the criteria, excepting logP and logS. This issue may affect to their membrane permeability. However, according to the prediction of SwissADME, the LogP_{O/W} value of them are less than 5.0 that is suitable for absorption. ^{55,56} In terms of solubility, based on ESOL topological model, they classified as moderately soluble

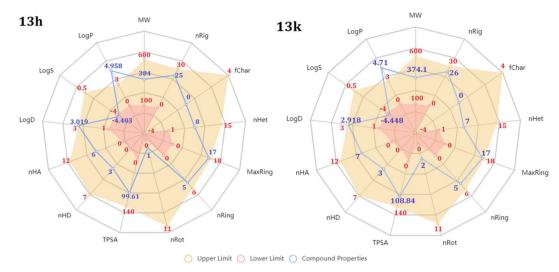


Fig. 7 Radar diagram illustrating the physicochemical characteristics of 13h and 13k. MW: molecular weight, nRig: number of rigid bonds, fChar: formal charge, nHet: number of heteroatoms, MaxRing: number of atoms in the biggest ring, nRing: number of rings, nRot: number of rotatable bonds, TPSA: topological polar surface area (\mathring{A}^2), nHD: number of hydrogen bond donors, nHA: number of hydrogen bond acceptors, logD: log of octanol partition coefficient at physiological pH 7.4, logS: log of aqueous solubility (mol L⁻¹), and logP: log of octanol partition coefficient.

chemicals (Table 4). In addition, 13h and 13k were predicted to have zero alert according to the Pan Assay Interference Compounds (PAINS) which is desired to avoid promiscuous behavior and nonselective reactivity with proteins.

In addition, these two compounds were subjected to druglikeness prediction. Table 5 demonstrates that 13h and 13k successfully accomplished all the drug-likeness rules with no violations but failed to pass lead-likeness rule due to their molecular weight. 57

Second, the properties related to Absorption, Distribution, Metabolism and Excretion (ADME) were analyzed. According to the absorption, the predicted values showed that both

Table 5 Drug-like and pharmacokinetic properties of 13h and 13k

Predicted parameters	13h	13k
Drug likeness		
Lipinski	Accept	Accept
Goshe	Accept	Accept
Veber	Accept	Accept
Egan	Accept	Accept
Muegge	Accept	Accept
Bioavailability score	0.55	0.55
Lead-likeness	No, $(MW > 350)$	No, $(MW > 350)$
PAINS	0 alert	0 alert
Absorption		
Gastrointestinal absorption	High	High
Pgp-substrate	No	No
$Log K_p$ (skin permeation)	-6.72 cm s^{-1}	$-6.72~{\rm cm~s^{-1}}$
Distribution		
Plasma protein binding	100.0% (not optimal)	97.87% (not optimal)
Volume distribution	0.402	0.413
Blood-brain barrier (BBB) penetration	No	No
Metabolism		
CYP interaction*	CYP1A2 inhibitor (0.94)	CYP1A2 inhibitor (0.92)
	CYP2C9 inhibitor (0.66)	CYP2C9 inhibitor (0.75)
	,	CYP2D6 substrate (0.58)
Excretion		
Clearance (CL)	$3.043 \text{ mL min}^{-1} \text{ kg (low)}$	$3.317 \text{ mL min}^{-1} \text{ kg (low)}$
Half-life $(T_{1/2})$	0.066 h (short)	0.194 (short)

7 7 6 6 9 PGF 13h 13h 12h 12PA

Fig. 8 BOILED-Egg diagram of two potent compounds 13h and 13k. Abbreviation: BBB: blood-brain barrier; HIA: human intestinal absorption; PGP+: P-glycoprotein substrate; PGP-: P-glycoprotein non-substrate.

compounds have high human intestinal absorption.58 According to the BOILED-Egg plot (Fig. 8), they were not glycoprotein-P substrate and shas high intestinal absorption percent. 59,60 Prediction of skin permeability showed that they are less permeable across the skin. In addition, 13f and 13h could not across the blood-brain barrier, suggesting a little effect on the central nervous system (CNS). These compounds were estimated to have high plasma protein binding which may narrow their therapeutic index. Both compounds showed optimal volume of distribution. In accordance with the first-pass metabolism, hepatic cytochrome plays a pivotal role in catalytic reactions. Our predictions showed that 13h and 13k mainly interacted with CYP1A2 and CYP2C9, suggesting potential drug-drug interactions with some medications and food.61 These compounds were also estimated to have low clearance together with a short half-life in human body.

ADMETlab 2.0 was subsequently used to anticipate toxicity of 13h and 13k (Table 6).⁶² Accordingly, both compounds are not hERG blockers, suggesting low cardiotoxicity effect.⁴⁴ Besides, they did not show human hepatoxicity, mutagenic effect, and eye corrosion. However, 13k may cause reactions with the skin and respiratory tract. Meanwhile 13h showed no interaction with most organs excepting the respiratory system.

In general, these derivatives provided suitable physicochemical and ADMET profiles to be considered as good anticancer compounds for further hit-to-lead optimization stages.

Experimental

Materials and methods

All chemicals were used as received without any further purification and obtained from Aldrich or Merck. Microwave irradiation experiments were performed using an Anton Paar Microwave Synthetic Reactor Monowave 400. Merck silica gel 60 F254 plates and Merck silica gel 60 (240–400 mesh) were used for thin-layer chromatography and column flash chromatography, respectively. Melting points were measured in open capillaries on a Buchi melting point B-545 apparatus (Buchi Instrument, Switzerland) and the values reported are uncorrected. A SCIEX X500 QTOF mass spectrometer in ESI (+) or ESI (–) mode was used to calculate the HRMS spectra. IR spectra have been recorded as KBr pellets, with a PerkinElmer Spectrum Two FT-IR spectrometer. NMR experiments were acquired using a Bruker Avance III spectrometer (600 and 125/150 MHz).

General procedure for synthesis of compounds 13a-m

Approach A: Under microwave irradiation (150 W), a mixture of 2-hydroxy-1,4-naphthoquinone (14, 1 mmol), tetronic acid (15, 1 mmol), heteroaromatic aldehyde (16a–m, 1 mmol), and NH₄OAc (17, 3.0 mmol) in glacial acetic acid (3 mL) was stirred for 20 min at 120 °C. The reaction mixture was then poured into water (20 mL) and extracted with dichloromethane (3 \times 20 mL), washed with brine (3 \times 10 mL). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated *in vacuo*, and purified by column chromatography using a dichloromethane-acetone eluent (20:1, 25:2, v/v) to furnish products 13a–m.

Approach B: Under microwave irradiation (150 W), a mixture of 2-amino-1,4-naphthoquinone (18, 1 mmol), tetronic acid (15, 1 mmol), heteroaromatic aldehyde (16a–m, 1 mmol), and p-TsOH (0.02 mmol) in glacial acetic acid (3 mL) was stirred for 20 min at 120 °C. The reaction mixture was then poured into water (20 mL), extracted with dichloromethane (3 × 20 mL), and washed with brine (3 × 10 mL). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated *in vacuo*, and

Table 6 Toxicity predicted using ADMETlab 2.0 of the two most potential compounds

Toxicity ^a	13h	13k
hERG blockers	Inactive (0.002)	Inactive (0.009)
Human hepatotoxicity	Negative (0.425)	Negative (0.12)
AMES toxicity	No (0.324)	No (0.273)
Rat oral acute toxicity	No (0.412)	No (0.436)
FDA maximum (recommended) daily	Negative (0.11)	Negative (0.235)
dose		- , ,
Skin sensitization	No (0.497)	Yes (0.747)
Respiratory toxicity	Yes (0.687)	Yes (0.893)
Eye irritation	No (0.263)	No (0.135)

^a The values in parentheses display the probability of being toxic.

purified by column chromatography using a dichloromethane-acetone eluent (20:1, 25:2, v/v) to furnish products **13a-m**.

11-(Thiophen-3-yl)-4,11-dihydrobenzo[g]furo[3,4-b]quinolin-**1,5,10(3H)-trione (13a).** Yield 101 mg (29% - approach A), 224 mg (64% - approach B), orange solid, mp. 298-299 °C. IR (KBr) $\nu_{\text{max}}/\text{cm}^{-1}$ 3238, 3086, 2967, 2930, 2890, 1709, 1657, 1636 1606, 1589, 1500, 1395, 1351, 1334, 1303, 1240, 1191, 1070, 1015, 932, 838, 783, 727. ¹H NMR (DMSO- d_6 , 600 MHz): δ 10.60 (1H, s, -NH), 8.05 (1H, dd, J = 1.2, 7.8 Hz, H-6), 7.93 (1H, dd, J = 1.2, 7.8 Hz, H-6)0.6, 7.8 Hz, H-9), 7.85 (1H, td, J = 1.2, 7.2 Hz, H-8), 7.81 (1H, td, J)= 1.2, 7.2 Hz, H-7), 7.38 (1H, dd, J = 3.0, 4.8 Hz, H-5'), 7.24 (1H, dd, J = 3.0, 4.8 Hz, H-5')dd, J = 0.6, 3.0 Hz, H-2'), 7.07 (1H, dd, J = 1.2, 4.8 Hz, H-4'), 5.11 (1H, s, H-11), 4.97 (1H, d, J = 16.8 Hz, H^a-3), 4.87 (1H, d, J = 16.8 Hz, H^a 16.8 Hz, H^b-3). ¹³C NMR (DMSO- d_6 , 125 MHz) δ 182.1 (C-10), 179.6 (C-5), 171.2 (C-1), 156.1 (C-3a), 145.0 (C-3'), 139.2 (C-4a), 134.9 (C-8), 133.4 (C-7), 131.9 (C-9a), 130.3 (C-5a), 127.7 (C-4'), 126.0 (C-6), 125.8 (C-9), 125.8 (C-5'), 122.4 (C-2'), 118.0 (C-10a), 101.6 (C-11a), 66.1 (C-3), 29.7 (C-11). HRMS (ESI): Found m/z $350.0484 [M + H]^+$, calcd. for $[C_{19}H_{12}NO_4S]^+$: 350.0482.

11-(Furan-3-yl)-4,11-dihydrobenzo[g]furo[3,4-b]quinolin-**1,5,10(3H)-trione (13b).** Yield 83 mg (25% – approach A), 207 mg (62% - approach B), yellow-brown solid, mp. 345-346 °C. IR (KBr) $\nu_{\text{max}}/\text{cm}^{-1}$ 3199, 3118, 2961, 2927, 2857, 1712, 1658, 1628, 1604, 1590, 1503, 1396, 1357, 1335, 1303, 1194, 1071, 1014, 934, 874, 794, 727. ¹H NMR (DMSO- d_6 , 600 MHz): δ 10.57 (1H, s, – NH), 8.06 (1H, dd, J = 1.2, 7.2 Hz, H-6), 7.93 (1H, dd, J = 1.2, 7.2 Hz, H-9), 7.86 (1H, td, J = 1.2, 7.2 Hz, H-8), 7.82 (1H, td, J =1.2, 7.2 Hz, H-7), 7.49 (1H, t, I = 1.2 Hz), 7.47 (1H, t, I = 0.6 Hz), 6.40 (1H, dd, J = 0.6, 1.8 Hz), 4.98 (1H, d, J = 16.2 Hz, H^a-3), 4.94 (1H, s, H-11), 4.87 (1H, dd, J = 1.2, 16.2 Hz, H^b-3). ¹³C NMR (DMSO- d_6 , 125 MHz) δ 182.1 (C-10), 179.5 (C-5), 171.2 (C-1), 156.3 (C-3a), 143.0, 140.2, 139.2 (C4a), 134.9 (C8), 133.3 (C-7), 131.9 (C-9a), 130.2 (C-5a), 128.5, 125.9 (C6), 125.7 (C9), 117.6 (C-10a), 110.5, 101.0 (C-11a), 66.1 (C-3), 25.3 (C-11). HRMS (ESI): Found m/z 334.0693 [M + H]⁺, calcd. for $[C_{19}H_{12}NO_5]^+$: 334.0710.

11-(Furan-2-yl)-4,11-dihydrobenzo[g]furo[3,4-b]quinolin-1,5,10(3*H*)-trione (13c). Yield 67 mg (20% – approach A), 200 mg (60% – approach B), orange solid, mp. 304–305 °C. IR (KBr) ν_{max} / cm⁻¹ 3202, 3153, 2965, 2932, 1716, 1659, 1631, 1604, 1590, 1504, 1396, 1352, 1335, 1303, 1195, 1070, 1014, 933, 727. ¹H NMR (DMSO- d_6 , 600 MHz): δ 10.70 (1H, s, NH), 8.06 (1H, dd, J = 1.2, 7.2 Hz, H-6), 7.96 (1H, dd, J = 1.2, 7.2 Hz, H-9), 7.87 (1H, td, J = 1.2, 7.2 Hz, H-8), 7.83 (1H, td, J = 1.2, 7.2 Hz, H-7), 7.46 (1H, dd, J = 0.6, 1.8 Hz), 6.32 (1H, q, J = 1.8 Hz), 6.19 (1H, d, J = 3.0 Hz), 5.14 (1H, s, H-11), 4.99 (1H, d, J = 16.8 Hz, H^a-3), 4.88 (1H, dd, J = 1.2, 16.8 Hz, H^b-3). ¹³C NMR (DMSO- d_6 , 125 MHz) δ 182.0, 179.5, 170.9, 156.6, 155.5, 141.9, 139.3, 135.1, 133.5, 131.8, 130.1, 126.1, 125.8, 115.9, 110.7, 106.6, 99.6, 66.1, 28.5. Found m/z 356.0513 [M + Na]⁺, calcd. for [C₁₉H₁₁NNaO₄]⁺: 356.0529

11-(5-Bromofuran-2-yl)-4,11-dihydrobenzo[*g*]furo[3,4-*b*]quinolin-1,5,10(3*H*)-trione (13d). Yield 82 mg (20% – approach A), 256 mg (62% – approach B), orange-yellow solid, mp. 371–372 ° C. IR (KBr) $\nu_{\text{max}}/\text{cm}^{-1}$ 3249, 1726, 1662, 1599, 1592, 1499, 1476, 1459, 1439, 1400, 1332, 1298, 1192, 1159, 1134, 1114, 1068, 1012, 960, 926, 779, 727. ¹H NMR (DMSO- d_6 , 600 MHz): δ 10.74 (1H, s, NH), 8.06 (1H, dd, J = 0.8, 7.8 Hz, H-6), 7.96 (1H, dd, J =

1.2, 7.8 Hz, H-9), 7.87 (1H, td, J = 1.2, 7.8 Hz, H-8), 7.83 (1H, td, J = 1.8, 7.8 Hz, H-7), 6.42 (1H, d, J = 3.0 Hz), 6.28 (1H, d, J = 3.0 Hz), 5.11 (1H, s, H-11), 5.02 (1H, d, J = 16.8 Hz, H^a-3), 4.91 (1H, dd, J = 1.2, 16.8 Hz, H^b-3). ¹³C NMR (DMSO-d₆, 125 MHz) δ 181.9, 179.4, 170.8, 158.0, 156.6, 139.5, 135.0, 133.5, 131.7, 130.2, 126.1, 125.8, 119.6, 115.2, 112.7, 109.8, 99.1, 66.2, 28.9. Found m/z 411.9832 and 413.9772 [M + H]⁺, calcd. for $[C_{19}H_{11}BrNO_5]^+$: 411.9816 and 413.9795.

11-(Benzo[*b*]thiophen-3-yl)-4,11-dihydrobenzo[*g*]furo[3,4-*b*] quinolin-1,5,10(3*H*)-trione (13e). Yield 132 mg (33% – approach A), 279 mg (70% – approach B), red-brown solid, mp. 295–296 ° C. IR (KBr) ν_{max} /cm⁻¹ 3445, 3212, 3070, 2923, 2853, 1728, 1661, 1595, 1501, 1397, 1332, 1299, 1194, 1135, 1068, 1010, 922, 786, 762, 722. ¹H NMR (DMSO-*d*₆, 600 MHz): δ 10.72 (1H, s, NH), 8.24 (1H, d, *J* = 7.8 Hz), 8.08–8.06 (1H, m), 7.91 (1H, d, *J* = 7.8 Hz), 7.84–7.79 (3H, m), 7.55 (1H, s), 7.46 (1H, td, *J* = 1.2, 7.8 Hz), 7.38 (1H, td, *J* = 0.6, 7.8 Hz), 5.40 (1H, s), 4.97 (1H, d, *J* = 16.8 Hz), 4.91 (1H, d, *J* = 16.8 Hz). ¹³C NMR (DMSO-*d*₆, 125 MHz) δ 182.0, 179.6, 171.1, 155.6, 141.21, 139.4, 139.2, 137.5, 134.8, 133.4, 131.8, 130.3, 126.4, 126.0, 125.7, 124.3, 124.0, 122.8, 122.5, 118.3, 102.0, 66.1, 27.9. HRMS (ESI): Found *m/z* 422.0429 [M + Na]⁺, calcd. for [C₂₃H₁₃NNaO₄S]⁺: 422.0458.

11-(Benzofuran-3-yl)-4,11-dihydrobenzo[g]furo[3,4-b]quino-lin-1,5,10(3H)-trione (13f). Yield 123 mg (32% – approach A), 261 mg (68% – approach B), brown solid, mp. 295–296 °C. IR (KBr) $\nu_{\text{max}}/\text{cm}^{-1}$ 3215, 3097, 1749, 1734, 1664, 1640, 1597, 1506, 1451, 1399, 1341, 1301, 1196, 1110, 1066, 1009, 750, 725. ¹H NMR (DMSO- d_6 , 600 MHz): δ 10.77 (1H, s), 8.08–8.05 (1H, m), 7.91 (1H, s), 7.90–7.89 (1H, m), 7.83 (1H, td, J = 1.8, 7.2 Hz), 7.80 (1H, td, J = 1.8, 7.2 Hz), 7.78 (1H, d, J = 1.2, 6.6 Hz), 7.50 (1H, d, J = 1.2, 7.2 Hz), 7.28 (1H, td, J = 1.8, 7.2 Hz), 7.26 (1H, td, J = 1.8, 7.2 Hz), 5.24 (1H, s), 4.99 (1H, d, J = 16.8 Hz), 4.92 (1H, dd, J = 1.2, 16.8 Hz), 3.73 (3H, s). ¹³C NMR (DMSO- d_6 , 125 MHz) δ 182.1, 179.5, 171.1, 156.4, 154.7, 144.4, 139.5, 134.9, 133.4, 131.9, 130.3, 126.3, 126.0, 125.7, 124.2, 123.9, 122.7, 120.3, 117.1, 111.2, 100.9, 66.1, 24.6. HRMS (ESI): Found m/z 384.0846 [M + H] $^+$, calcd. for $[C_{23}H_{14}NO_5]^+$: 384.0866.

11-(Benzo[*b*]thiophen-2-yl)-4,11-dihydrobenzo[*g*]furo[3,4-*b*] quinolin-1,5,10(3*H*)-trione (13*g*). Yield 132 mg (33% – approach A), 276 mg (69% – approach B), red-brown solid, mp. 302–303 ° C. IR (KBr) ν_{max} /cm⁻¹ 3466, 3050, 2924, 2853, 2263, 2126, 1773, 1740, 1667, 1637, 1591, 1495, 1431, 1394, 1341, 1296, 1192, 1157, 1137, 1066, 999, 927, 822, 788, 770, 748, 719. ¹H NMR (DMSO-*d*₆, 600 MHz): δ 10.84 (1H, s), 8.08 (1H, dd, *J* = 1.2, 7.2 Hz), 7.97 (1H, d, *J* = 7.2 Hz), 7.87 (1H, td, *J* = 1.2, 7.2 Hz), 7.85 (1H, d, *J* = 1.2, 8.4 Hz), 7.24 (1H, s), 5.38 (1H, s), 5.06 (1H, d, *J* = 16.2 Hz), 4.96 (1H, dd, *J* = 1.2, 16.2 Hz). ¹³C NMR (DMSO-*d*₆, 125 MHz) δ 182.0, 179.4, 170.9, 156.5, 148.1, 139.4, 139.0, 138.9, 135.0, 133.5, 131.8, 130.2, 126.1, 125.9, 124.3, 124.0, 123.4, 122.3, 121.7, 117.2, 100.9, 66.2, 30.3. HRMS (ESI): Found *m/z* 400.0623 [M + H]⁺, calcd. for [C₂₃H₁₄NO₄S]⁺: 400.0638.

11-(2-Chlorothiazol-5-yl)-4,11-dihydrobenzo[*g*]**furo**[3,4-*b*] **quinolin-1,5,10**(3*H*)**-trione (13h).** Yield 108 mg (28% – approach A), 246 mg (64% – approach B), yellow-brown solid, mp. 383–384 °C. IR (KBr) $\nu_{\text{max}}/\text{cm}^{-1}$ 3341, 3092, 2922, 2852, 1774, 1734, 1660, 1602, 1525, 1507, 1478, 1443, 1414, 1397, 1335, 1300,

1267, 1240, 1199, 1159, 1140, 1044, 1019, 927, 878, 826, 776, 721, 705. $^1{\rm H}$ NMR (DMSO- d_6 , 600 MHz): δ 10.86 (1H, s), 8.06 (1H, dd, J=1.2, 7.8 Hz), 7.99 (1H, dd, J=1.2, 7.8 Hz), 7.88 (1H, td, J=1.2, 7.8 Hz), 7.83 (1H, td, J=1.2, 7.8 Hz), 7.48 (1H, d, J=1.2, 7.8 Hz), 5.29 (1H, s), 5.06 (1H, d, J=1.2, 7.8 Hz), 4.94 (1H, dd, J=1.2, 16.8 Hz). $^{13}{\rm C}$ NMR (DMSO- d_6 , 125 MHz) δ 182.1, 179.3, 170.9, 169.7, 157.1, 143.9, 139.7, 139.5, 135.1, 133.6, 131.7, 130.3, 127.0, 125.9, 116.3, 99.9, 66.4, 27.7. HRMS (ESI): Found m/z 385.0026 [M + H] $^+$, calcd. for [C₁₈H₁₀ClN₂O₄S] $^+$: 385.0044.

11-(1,5-Dimethyl-1*H*-pyrazol-4-yl)-4,11-dihydrobenzo[g]furo [3,4-*b*]quinolin-1,5,10(3*H*)-trione (13i). Yield 98 mg (27% – approach A), 228 mg (63% – approach B), red-brown solid, mp. 262–263 °C. IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3439, 3290, 3163, 3077, 2925, 2854, 1722, 1660, 1596, 1490, 1428, 1393, 1331, 1298, 1267, 1191, 1134, 1108, 1065, 1037, 1011, 929, 884, 828, 792, 723.

1H NMR (DMSO- d_6 , 600 MHz): δ 10.52 (1H, s), 8.04 (1H, dd, J = 1.2, 7.8 Hz), 7.89 (1H, dd, J = 1.2, 7.8 Hz), 7.82 (1H, td, J = 1.2, 7.8 Hz), 7.79 (1H, td, J = 1.2, 7.8 Hz), 7.11 (1H, s), 4.94 (1H, d, J = 16.2 Hz), 4.85 (1H, dd, J = 1.2, 16.2 Hz), 4.79 (1H, s), 3.63 (3H, s), 2.30 (3H, s).

13°C NMR (DMSO- d_6 , 150 MHz) δ 182.2, 179.7, 171.3, 155.3, 138.6, 137.2, 134.9, 134.7, 133.3, 131.9, 130.2, 126.0, 125.7, 122.4, 118.5, 102.0, 66.0, 36.0, 24.7, 9.1. HRMS (ESI): Found m/z 362.1118 [M + H]+, calcd. for [C₂₀H₁₆N₃O₄]+: 362.1135.

11-(1H-pyrazol-5-yl)-4,11-dihydrobenzo[g]furo[3,4-b]quino-lin-1,5,10(3H)-trione (13j). Yield 67 mg (20% – approach A), 203 mg (61% – approach B), pink-red solid, mp. 310–311 °C. IR (KBr) $\nu_{\text{max}}/\text{cm}^{-1}$ 3337, 3249, 3207, 3148, 2960, 2927, 1725, 1664, 1630, 1590, 1499, 1398, 1354, 1304, 1199, 1167, 1054, 1011, 933, 784725. ¹H NMR (DMSO- d_6 , 600 MHz): δ 12.39 (1H, br. s, NH), 10.55 (1H, s), 8.05 (1H, d, J = 7.2 Hz), 7.94 (1H, d, J = 7.2 Hz), 7.86 (1H, t, J = 7.2 Hz), 7.81 (1H, t, J = 7.2 Hz), 7.52 (1H, br. s), 6.13 (1H, br. s), 5.11 (1H, br. s), 4.93 (1H, d, J = 16.8 Hz), 4.87 (1H, d, J = 16.8 Hz). ¹³C NMR (DMSO- d_6 , 125 MHz) δ 182.1, 179.6, 171.2, 156.0, 135.0, 133.4, 131.9, 130.1, 126.0, 125.8, 103.2, 66.0, 29.8. HRMS (ESI): Found m/z 334.0808 [M + H][†], calcd. for $[C_{18}H_{12}N_3O_4]^{\dagger}$: 334.0822.

11-(2-Methoxypyridin-4-yl)-4,11-dihydrobenzo[g]furo[3,4-b] quinolin-1,5,10(3H)-trione (13k). Yield 127 mg (34% – approach A), 277 mg (74% – approach B), red-brown solid, mp. 245–246 ° C. IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3455, 3044, 3014, 2984, 2941, 2893, 2740, 1739, 1665, 1630, 1601, 1561, 1505, 1478, 1396, 1341, 1300, 1192, 1167, 1147, 1068, 1045, 1021, 1004, 940, 786, 728. ¹H NMR (DMSO- d_6 , 600 MHz): δ 10.69 (1H, s), 8.07 (1H, dd, J = 1.2, 7.2 Hz), 8.04 (1H, dd, J = 1.2, 5.4 Hz), 7.90 (1H, dd, J = 1.2, 7.2 Hz), 7.86–7.81 (2H, m), 6.99 (1H, dd, J = 1.2, 5.4 Hz), 6.74 (1H, s), 4.99 (1H, s), 4.98 (1H, d, J = 16.8 Hz), 4.90 (1H, dd, J = 16.8 Hz), 3.79 (3H, s). ¹³C NMR (DMSO- d_6 , 125 MHz) δ 182.0, 179.3, 170.9, 163.8, 156.5, 155.2, 146.6, 140.1, 134.8, 133.4, 131.8, 130.3, 126.0, 125.7, 117.1, 116.9, 109.7, 100.8, 66.2, 53.0, 34.7. HRMS (ESI): Found m/z 375.0960 [M + H]⁺, calcd. for [C₂₁H₁₅N₂O₅]⁺: 375.0976.

11-(3-Fluoropyridin-2-yl)-4,11-dihydrobenzo[g]furo[3,4-b] quinolin-1,5,10(3H)-trione (13I). Yield 116 mg (32% – approach A), 261 mg (72% – approach B), grey-red solid, mp. 288–289 °C. IR (KBr) $\nu_{\rm max}$ /cm⁻¹ 3349, 3221, 3081, 2928, 2857, 1747, 1671, 1660, 1630, 1607, 1497, 1442, 1397, 1343, 1303, 1194, 1066,

1005, 803, 724. ¹H NMR (DMSO- d_6 , 600 MHz): δ 10.48 (1H, s), 8.25–8.23 (1H, m), 8.07 (1H, dd, J = 2.4, 8.4 Hz), 7.88–7.84 (1H, m), 7.83–7.78 (2H, m), 7.63–7.58 (1H, m), 7.27 (1H, quint, J = 5.4 Hz), 5.51 (1H, s), 4.92 (2H, t, J = 16.2 Hz). ¹³C NMR (DMSO- d_6 , 125 MHz) δ 181.6, 179.3, 170.4, 156.4, 155.3 (1C, d, J = 256.3 Hz), 149.8 (1C, d, J = 15 Hz), 145.0 (1C, d, J = 6.3 Hz), 139.9, 134.8, 133.2, 131.5, 129.8, 125.8, 125.5, 123.5 (1H, d, J = 3.8 Hz), 122.4 (1H, d, J = 20 Hz), 118.0, 100.3, 65.9, 30.7. HRMS (ESI): Found m/z 363.0760 [M + H]⁺, calcd. for [C₂₀H₁₂FN₂O₄]⁺: 363.0776.

11-(2-Methoxy-5-(trifluoromethyl)pyridin-3-yl)-4,11-dihydrobenzo[g]furo[3,4-b]quinolin-1,5,10(3*H*)-trione (13*m*). Yield 146 mg (33% – approach A), 318 mg (72% – approach B), yelloworange solid, mp. 289–290 °C. IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3285, 3079, 3027, 2997, 2950, 1725, 1663, 1603, 1577, 1495, 1437, 1400, 1322, 1297, 1246, 1195, 1144, 1116, 1093, 1072, 1015, 940, 724. ¹H NMR (DMSO-*d*₆, 600 MHz): δ 10.63 (1H, s), 8.41 (1H, d, *J* = 1.2 Hz), 8.05 (1H, dd, *J* = 1.2, 7.2 Hz), 8.00 (1H, d, *J* = 2.4 Hz), 7.85 (1H, dd, *J* = 1.2, 7.2 Hz), 7.84–7.78 (2H, m), 5.34 (1H, s), 4.95 (1H, d, *J* = 16.2 Hz), 4.89 (1H, dd, *J* = 0.6, 16.2 Hz), 3.96 (3H, s). ¹³C NMR (DMSO-*d*₆, 150 MHz) δ 181.9, 179.4, 170.7, 163.2, 156.8, 142.9 (1C, d, *J* = 4.5 Hz), 140.5, 135.3, 134.9, 133.3, 131.8, 130.1, 127.3, 125.9, 125.7, 124.1 (1C, q, *J* = 270 Hz), 118.8 (1C, q, *J* = 31.5 Hz), 117.0, 100.2, 66.0, 54.2, 30.6. HRMS (ESI): Found *m/z* 443.0836 [M + H]⁺, calcd. for [C₂₂H₁₄F₃N₂O₅]⁺: 443.0849.

MTT assay

The investigated cancer cell lines were purchased from the American Type Culture Collection (ATCC, USA). The cytotoxicity of the newly synthesized series of compounds was studied against non-small lung (A549, CCL-185™), epidermoid carcinoma (KB, CCL-17[™]), breast (MCF7, HTB-22[™]), hepatocellular carcinoma (HepG2, HB-8065[™]), cancer cells, and human embryonic kidney (Hek-293, CRL-1573[™]) cells, which were cultured in DMEM (Dulbeccos Modified Eagle Medium) medium, supplemented with 10% fetal bovine serum, 100 U ml⁻¹ penicillin, and 100 μg mL⁻¹ streptomycin, at 37 °C in humidified atmosphere (95% air and 5% CO2). Solutions of compounds 13a-m in DMSO at different concentratrions (1.00, 0.25, 0.0625, and $0.0156 \mu g mL^{-1}$) were added to human cancer cell lines (3 \times 10⁴ cells per ml). After 3 days-incubation, 10 μ l solution of 5 mg mL⁻¹ MTT in sodium phosphate buffer (PBS, 0.1 M, pH 7.4) was then supplemented, the cells were incubated at 37 °C for 4 h, and then removed medium. The obtained formazan crystals were dissolved by DMSO (150 μl). A Biotek Epoch 2 microplate reader was used to measure the absorbance of the solutions at 540 nm.63-65

Molecular docking study

Molecular docking studies were performed for all the synthesized compounds against the tubulin target. To do so, the tubulin-colchicine complex (PDB ID: 4O2B) was taken from the Protein Data Bank. They were prepared for docking assays following the standard protocol implemented into ICM Pro (x64) software. The chains α and β tubulin were retained, followed by deleting all the water molecules, optimizing all hydrogens and optimizing HisProAsnGlnCys, setting MMFF

forcefield, and predicting the binding site of Colchicine by using ICMpocketfinder.67 The 2D structures of synthesized compounds 13a-13m were generated using ChemDraw 20.1.1, then imported to ICM Pro software and convert to 3D conformations for docking. Before docking synthesized compounds, the co-crystallized ligand colchicine was removed, and redocked into the binding site of Tubulin. Conformational sampling is based on the biased probability Monte Carlo (BPMC) procedure. The ICM scoring function is a GBSA/MM-type,66 which is weighted according to the following parameters (i) internal force-field energy of the ligand, (ii) entropy loss of the ligand between bound and unbound states, (iii) ligand-receptor hydrogen bond interactions, (iv) polar and non-polar solvation energy differences between bound and unbound states, (v) electrostatic energy, (vi) hydrophobic energy, and (vii) hydrogen bond donor or acceptor desolvation. The binding energy (kcal mol^{-1}) is calculated *via* equation: $\Delta G = \Delta E_{\text{IntFF}} + T \Delta S_{\text{Tor}} + \alpha_1$ $\Delta E_{\rm HBond} + \alpha_2 \Delta E_{\rm HBDesol} + \alpha_3 \Delta E_{\rm Solel} + \alpha_4 \Delta E_{\rm HPhob} + \alpha_5 Q_{\rm Size}$. For each ligand, 50 conformations were generated, and the conformations with better binding scores and key interactions similar to colchicine were selected for further studies. The docking results were then visualized using BIOVIA Discovery Studio Visualizer 2021.

Physicochemical and pharmacokinetic properties prediction

Calculations of the physicochemical parameters of synthesized compounds relevant to ADME were performed using SwissAMDE, a free online cheminformatics tool.⁵⁷ Computed parameters are related to the evaluation of drug-likeness, lead-likeness characteristics such as Lipinski, Veber, Goshe, Egan, and Muegge rules.⁶⁸ Additionally, their toxicity was also predicted by another web-based tool – ADMETlab 2.0.^{62,69} The parameters can be rapidly determined by easily inputting the SMILES codes of the compounds into the website.

Conclusions

A series of new heterocyclic hybrid compounds of 4-azapodophyllotoxin were successfully synthesized via microwaveassisted multicomponent reactions. The efficient synthesis of novel hybrid compounds has been expected to facilitate the discovery of numerous classes of bioactive heterocyclic compounds. Obviously, the in vitro cytotoxic assessment and molecular docking study revealed that all products showed cytotoxic activity and they are all capable of binding to the active site of colchicine in tubulin. Compounds 13a (HetAr = thiophen-3-yl), and 13d (HetAr = 5-bromofuran-2-yl) displayed high cytotoxic selectivity for A549 and HepG2 cancer cell lines when compared to the other cancer cell lines and low toxicity to normal Hek-293 cell line with IC $_{50}$ = 22.90 \pm 2.18 and 37.12 \pm 3.30 µM, respectively. The most active compound 13k containing 2-methoxypyridin-4-yl group exhibited significant cytotoxicity against human lung adenocarcinoma cells, human mouth epidermal carcinoma cells, and human hepatocellular carcinoma cells with IC_{50} ranging from 0.23 to 0.27 μM . Compound 13h (HetAr = 2-chlorothiazol-5-yl) displayed the highest toxicity

against A549 cells with $IC_{50} = 0.16 \pm 0.02 \mu M$. Notably, compounds **13h,k**, which have strong binding energy with the active site of tubulin at -17.86, and -18.05 kcal mol⁻¹, respectively, provided suitable physicochemical and ADMET profiles to be considered as good anticancer compounds. Taken together, these potent cytotoxic compounds have potential for investigation into the anticancer activity.

Author contributions

Ha Thanh Nguyen: project administration: conceptual design; writing – original draft. Ket Tran Van, and Phuong Hoang Thi: investigation: synthesis of compounds. Julien Braire and Quynh Giang Nguyen Thi: investigation: characterisation of compounds. Tu Anh Le Thi and Giang Le-Nhat-Thuy: investigation: biological evaluation of compounds. Tuyet Anh Dang Thi and Doan Vu Ngoc: investigation: revised draft. Hai Pham-The and Tuan Anh Nguyen: investigation: molecular docking studies. Tuyen Van Nguyen: supervision: conceptual design; writing – review & editing.

Conflicts of interest

There are no conflicts to declare.

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