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A review on synthesis of furonaphthoquinones through lawsone derivatives annulation reactions and their biological properties

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Furonaphthoquinones and their dihydro derivatives have attracted significant attention due to their diverse pharmacological activities. These compounds can be derived from natural sources, including various plants, or synthesized through chemical methods, resulting in a wide variety of structures with distinct biological properties. As a result, numerous methods have been developed over the past decades for the preparation of these compounds, particularly utilizing 2-hydroxy-1,4-naphthoquinone derivatives as key precursors. Considering these concepts, this review aims to offer a comprehensive overview of the chemical synthesis of linear and angular furonaphthoquinones, along with their dihydro derivatives derived from 2-hydroxy-1,4-naphthoquinones annulation reactions, and to explore their diverse biological activities.

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1. Introduction

2-Hydroxy-1,4-naphthoquinone, commonly referred to as lawsone or hennotannic acid, is a naturally occurring naphthoquinone compound that has attracted considerable attention in the field of synthetic organic chemistry. Among the

simplest naturally occurring naphthoquinones, it is best known for its presence in the leaves of the henna plant (*Lawsonia inermis*), where it contributes to the characteristic red-orange coloration.^{1,2} Henna extracts containing lawsone have been used for thousands of years as hair, nails, wool, cotton and skin dyes, highlighting its historical and cultural significance.³⁻⁶ In

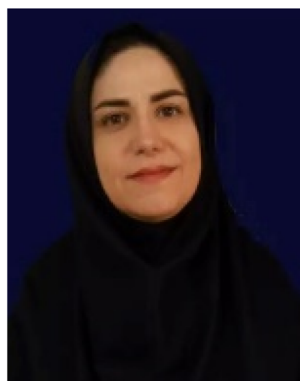
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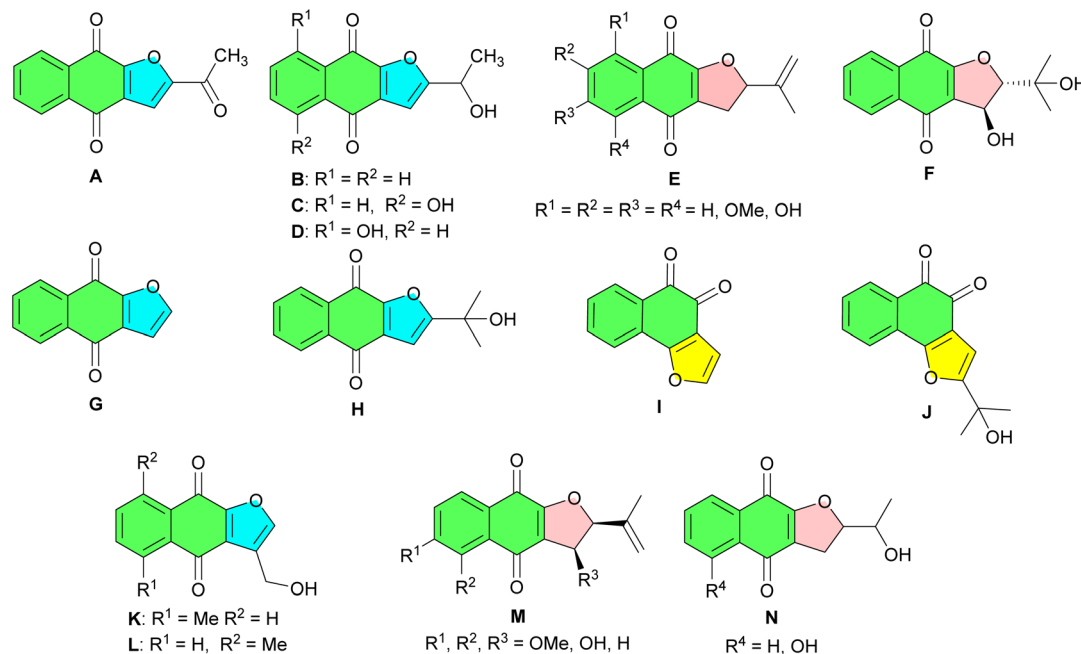


Fig. 1 Some natural, bioactive furonaphthoquinones and their dihydro derivatives.

addition to its well-known dyeing capabilities, 2-hydroxy-1,4-naphthoquinone demonstrates a variety of biological activities, including anti-inflammatory, antibacterial, antiviral, antifungal, and antineoplastic effects.^{7,8} In organic synthesis, lawsone has been extensively utilized in various reactions, playing a pivotal role in constructing diverse molecular frameworks such as furonaphthoquinones and their dihydro derivatives. They constitute an important group of oxygen heterocycles. Many members of this group are widely distributed in nature and isolated from various plants.⁹ For example, as indicated in Fig. 1, furonaphthoquinones **A–C** isolated from *Tabebuia cassinoides*, exhibit anticancer activity.¹⁰ Kigelonone **D**, isolated from the fruits of *Kigelia pinnata*, exhibits antibacterial, antifungal, and antiviral activities.¹¹ Compounds **E** were isolated from *Crescentia cujete*, and showed interesting mutagenic

activities.¹² Avicequinones **F–H** have been isolated from the stem bark of *Avicennia alba*, while naphthoquinones **I** and **J** have been obtained from *Avicennia marina*.¹³ Maturone **K** and Maturinone **L**, isolated from *Cacalia decomposita*, are derived from its root extract, which has been traditionally used for the treatment of diabetes.¹⁴ Naphthofuroquinones **M** and **N**, isolated from the dried wood chips of *Crescentia cujete*, exhibit moderate but selectivity against the repair-deficient rad52 yeast strain, suggesting their role as DNA-damaging agents.¹²

Furonaphthoquinones exhibit a diverse biological activities, including antitumor effects,¹⁵ antiviral activity against the Japanese encephalitis virus¹⁶ and Vero cells,¹¹ and inhibition of human keratinocyte hyperproliferation.¹⁷ They also demonstrate antibacterial,¹⁸ anti-inflammatory,¹⁹ antiallergic,²⁰ anti-apoptotic,²¹ and antileishmanial properties²² along with inhibitory activity against receptor tyrosine kinases.²³ Furthermore, they function as antipsoriatic²⁴ and antileukemic agents¹⁰ as well as antiprotozoan activity against *Trypanosoma cruzi*.²⁵ Additionally, these compounds possess antifungal properties²⁶ and other cytotoxic effects.²⁷ Due to their significant role in pharmaceutical research and drug discovery, substantial efforts have been directed toward developing synthetic strategies for furonaphthoquinone ring systems. Considering the importance of furonaphthoquinone derivatives and as part of our ongoing research into synthesizing organic compounds from lawsone and review articles,²⁸ this review highlights the key synthetic methodologies used for preparing linear and angular furonaphthoquinone derivatives derived from 2-hydroxy-1,4-naphthoquinones. Emphasis is placed on advancements in reaction design, catalyst development, reaction mechanisms, and the discovery of new reaction pathways. Additionally, we explore the applications of these synthesized compounds in



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medicinal chemistry, highlighting their promising potential in drug development.

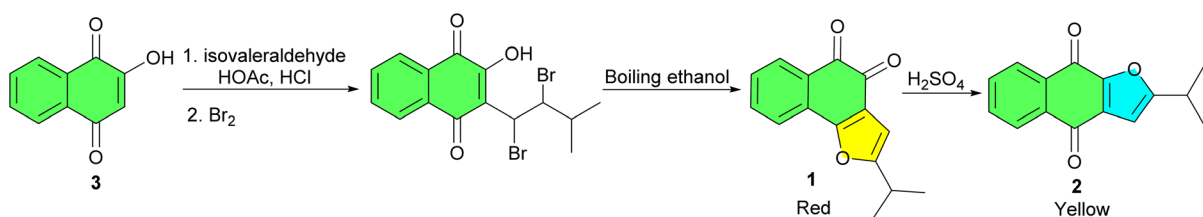
2. Synthesis of furonaphthoquinones

In 1896, Hooker reported the first synthesis of naphthofuroquinone. 2-Iso-propylnaphtho[1,2-*b*]furan-4,5-dione (**1**) and 2-iso-propylnaphtho[2,3-*b*]furan-4,9-dione (**2**) were synthesized from 2-hydroxy-1,4-naphthoquinone (Lawsone) (**3**) through a multi-step process, as outlined in Scheme 1.²⁹

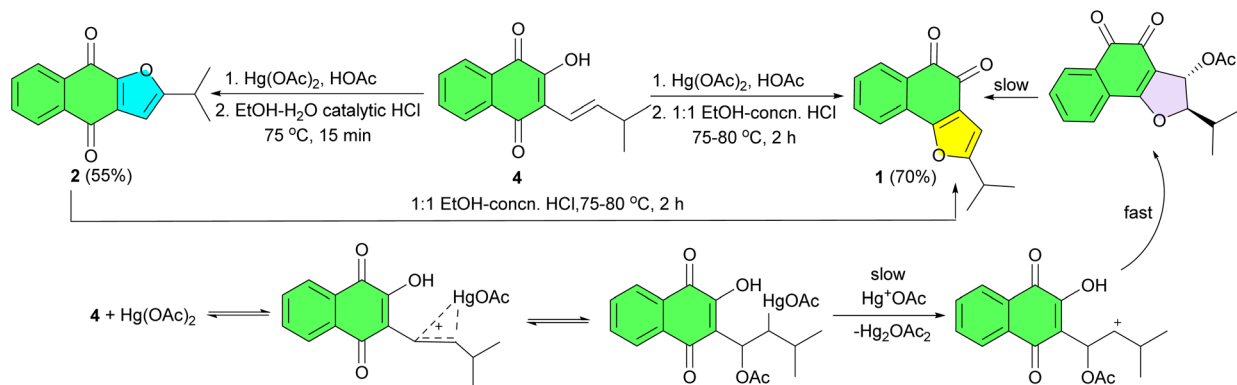
In 1967, Dudley and Waynemill reported that the mercuric acetate oxidation of iso-lapachol (**4**) proved to be an efficient synthetic method, yielding angular furanonaphthoquinone **1** with a 70% yield and linear furanonaphthoquinone **2** with a 55% yield. As seen in Scheme 2, the only differences in the

experimental procedures involved the time of heating and amount of mineral acid employed in the second step. A plausible mechanism for the preparation of **1** is shown in Scheme 2.³⁰

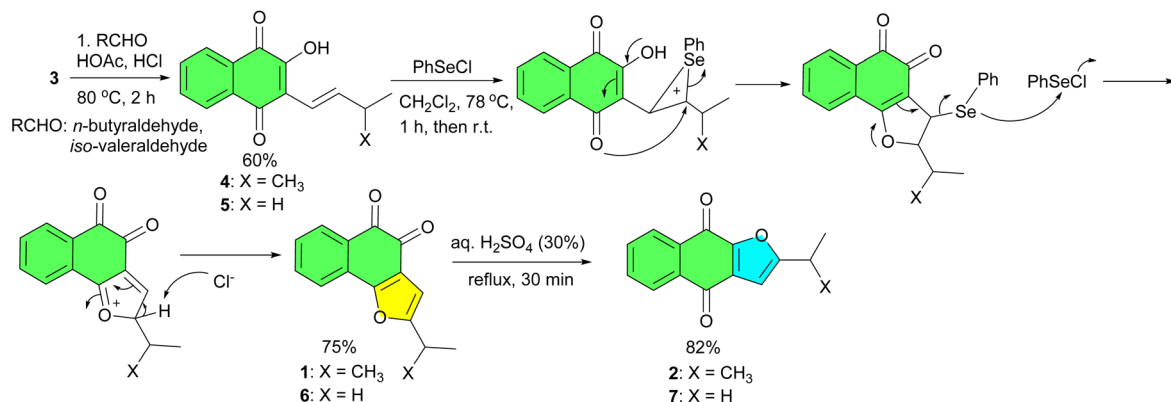
In 1990, phenylselenoetherification was used to synthesize furanonaphthoquinones from lawsone. When derivatives **4** and **5** prepared from **3** according to Hooker's procedure, was treated under the same condition, the reaction was incomplete and requires 2 equivalents of benzeneselenyl chloride to give to completion, giving naphtho[1,2-*b*]furan-4,5-diones **1** or **6** in 75% yield, and dibenzenediselenide. The orthoquinones **1** and **6** could be isomerized with acid to lead to the corresponding naphtho[2,3-*b*]furan-4,9-diones **2** or **7** in 82% yield. In the proposed mechanism, the intermediate selenide was not isolated. A nucleophilic displacement reaction at Se(II) with the



Scheme 1 Synthesis of naphthofuroquinones 1–2.

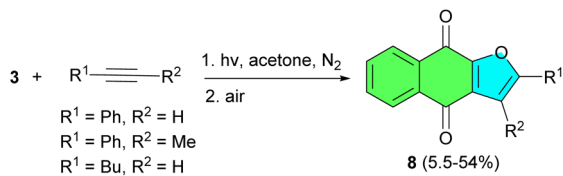


Scheme 2 $\text{Hg}(\text{OAc})_2$ oxidation of iso-lapachol (**4**) to furanonaphthoquinones 1–2.



Scheme 3 Selenium reagent in the synthesis of furanonaphthoquinones 1–2, 6–7.





Scheme 4 Regioselective synthesis of naphtho[2,3-*b*]furan-4,9-diones **8**.

participation of the oxygen atom of the dihydrofuran ring may be invoked to explain this process (Scheme 3).³¹

Next, the Sugimoto group demonstrated naphtho[2,3-*b*]furan-4,9-diones **158** can directly and exclusively be formed by an unprecedented regioselective [3 + 2] photoaddition of **3** with various alkynes in acetone for 7.5–30 h in 5.5–54% yields (Scheme 4). The photoaddition of hydroxynaphthoquinone with alkynes leading to **8** may proceed in a manner parallel to the photoaddition of alkenes and may also involve vinyl cations.³²

In 1995, one-step synthesis of furanonaphthoquinones **9** reported in 15–35% yields by a [3 + 2] photoaddition of 2-hydroxy-1,4-naphthoquinones with 2-naphthol in acetone at room temperature for 30–100 h (Scheme 5).³³

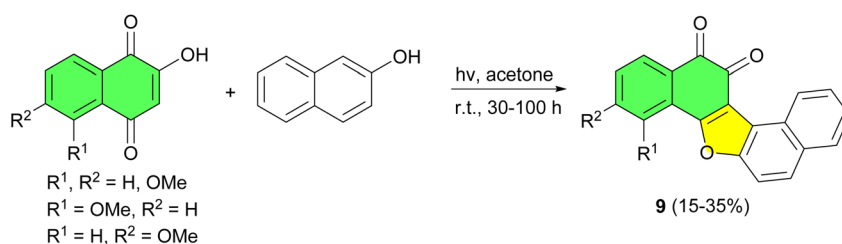
In 1997, Kobayashi and co-workers described treatment of 2-hydroxy-3-iodo-1,4-naphthoquinone (**10**) or 3-phenyliodonio-1,2,4-trioxo-1,2,3,4-tetrahydronaphthalenides **11** with terminal acetylenes in the presence of a bis(triphenylphosphine)palladium chloride-cuprous iodide catalyst at room temperature or cuprous oxidem *N*-methylpiperidine or pyridine at 80 °C, respectively, by sequential coupling/ring closure reactions furnished the corresponding 2-substituted naphtho[2,3-*b*]furan-

4,9-diones **12** in 20–66% yields. In the proposed mechanism, the reactions using the iodonium ylides **11** proceed through formation of the corresponding 2-hydroxy-3-iodo-1,4-naphthoquinones as outlined in Scheme 6. Their formation, followed by coupling with a terminal acetylene, gives rise to the alkynylated hydroxy quinones **13**, which subsequently undergo intramolecular ring closure to give **12**.³⁴

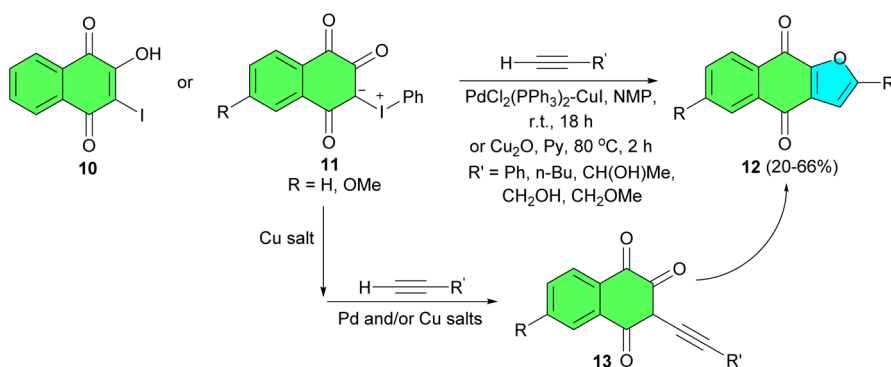
The reaction of 2-hydroxy-1,4-naphthoquinones with enamines, derived from ketones, in refluxing toluene afforded the corresponding 2,3-disubstituted naphtho[2,3-*b*]furan-4,9-dione derivatives **14** after 7–24 h in moderate to good yields. The probable pathways leading to the formation of **14** are outlined in Scheme 7. Initially, the nucleophilic attack of the enamine to the quinone at both of the 2- and 3-positions to give the intermediate adducts **15** or **16**. The 2-oxygen of **15** intramolecularly adds to the immonium carbons, resulting in the formation of a dihydrofuran ring. The dihydrofuran intermediate **17** gives rise to **14** through elimination of the amine and oxidation. On the other hand, the intermediate **16** rearranges to the amination intermediate **18** through elimination of the hydroxide, which gives **19** *via* the dihydrofuran intermediate **20**. When *R* equals *R'*, compounds **14** and **19** are identical.³⁵

After that, Estevez and co-workers described synthesis of benzofuronaphthoquinone **21** in 71–76% yields by the reaction of 3-hydroxy-2-phenyl-1,4-naphthoquinones **22** using CuO, K₂CO₃ in pyridine under reflux conditions for 1.5–4 h (Scheme 8).³⁶

Further, the same group reported synthesis of benzofuronaphthoquinone **23** in 92–99% yields by refluxing a solution of **24** in the presence of cuprous oxide, potassium carbonate and dry deoxygenated pyridine under argon for 2 h (Scheme 9).³⁷

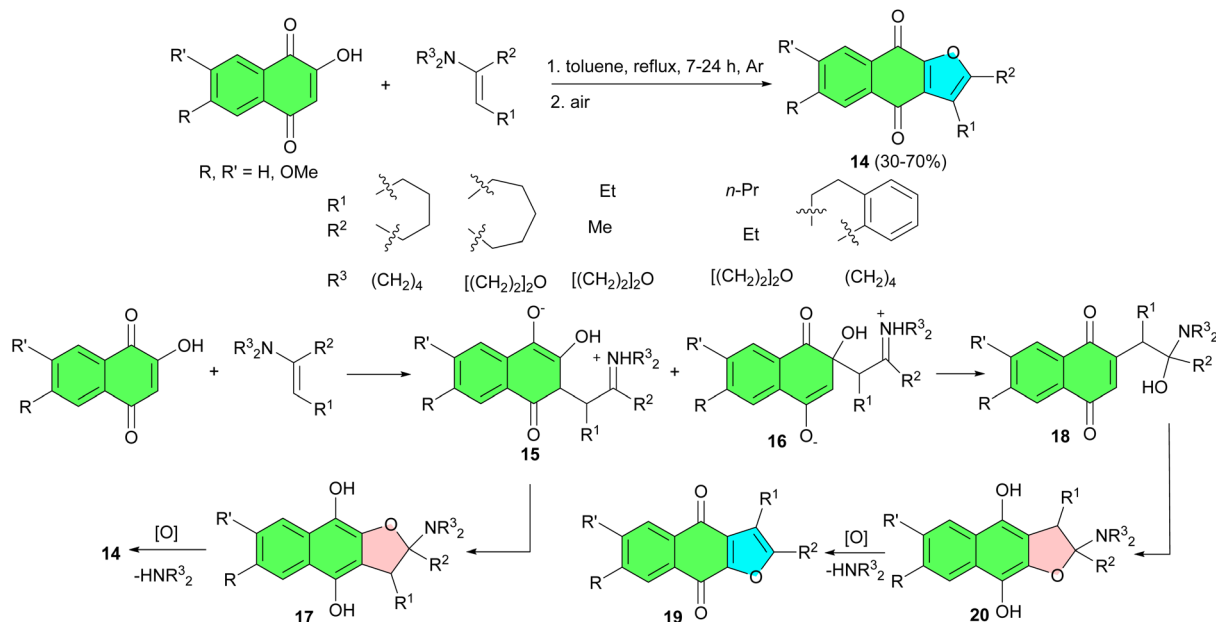
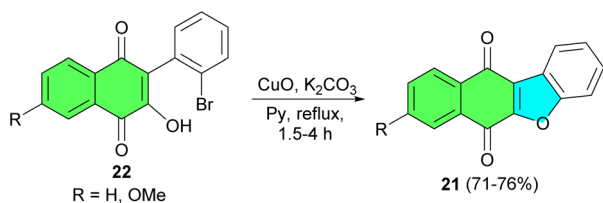


Scheme 5 One-step synthesis of furanonaphthoquinones **9**.



Scheme 6 Transition metal-mediate synthesis of naphtho[2,3-*b*]furan-4,9-diones **12**.



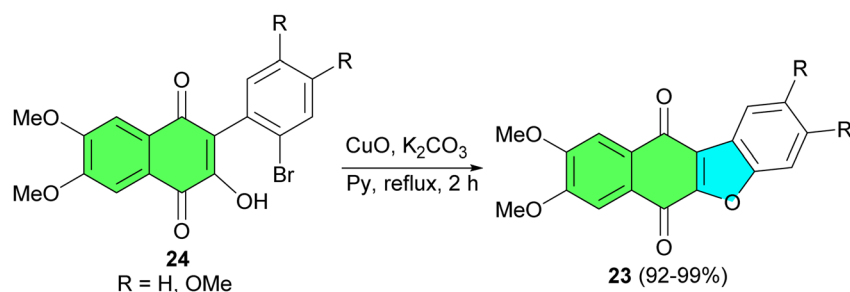
Scheme 7 Preparation of naphtho[2,3-*b*]furan-4,9-diones **14**.Scheme 8 CuO catalyzed preparation of benzofuronaphthoquinone **21**.

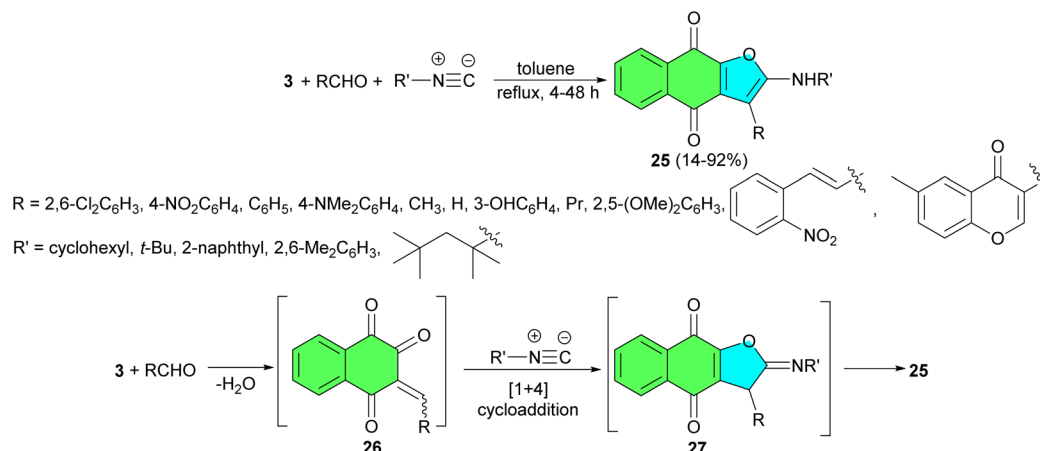
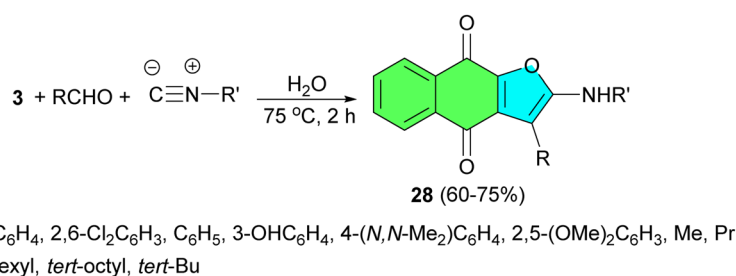
In 2007, the Teimouri group described regioselective one-pot three-component condensation reaction of lawsone with isocyanides in the presence of a variety of aldehydes in refluxing toluene for 4–48 h afforded linear naphtho[2,3-*b*]furan-4,9-dione derivatives **25** in 14–92% yields. The elucidation of regiochemistry accomplished by X-ray determination. In the proposed mechanism (Scheme 10), at first, a conjugated electron-deficient enone **26** obtained by a Knoevenagel condensation of **3** and the aldehyde. The next step of this mechanism could involve a [4 + 1] cycloaddition reaction of the

electron-deficient heterodyne moiety of adduct **26** with the isocyanide to afford an iminolactone intermediate **27**. The subsequent isomerization of iminolactone **27** leads to formation of product **25**.³⁸

After that, an efficient regioselective and clean green synthesis of highly substituted linear naphtho[2,3-*b*]furan-4,9-dione derivatives **28**, starting from lawsone, alkyl isocyanides and a variety of aliphatic and aromatic aldehydes containing electron-withdrawing groups and electron-donating groups, is described. This [3 + 1 + 1] furannulation strategy affords furanonaphthoquinones in moderate to high yields, using water at 75 °C after 2 h (Scheme 11). The proposed mechanism is similar to that described in Scheme 10.³⁹

In 2011, a three-component domino reaction of lawsone, aromatic aldehydes and a pyridinium salt in the presence of ammonium acetate, under microwave irradiation (at 130 °C and 150 W) and using water as solvent, furnished a library of 2-arylcarbonyl-3-aryl-4,9-dihydronaphtho[2,3-*b*]furan-4,9-diones **29** in 74–86% yields after 20 min. A plausible mechanism for the formation of **29** is depicted in Scheme 12. Presumably, the intermediate **30** could arise in two ways: (i) *via* the Mannich

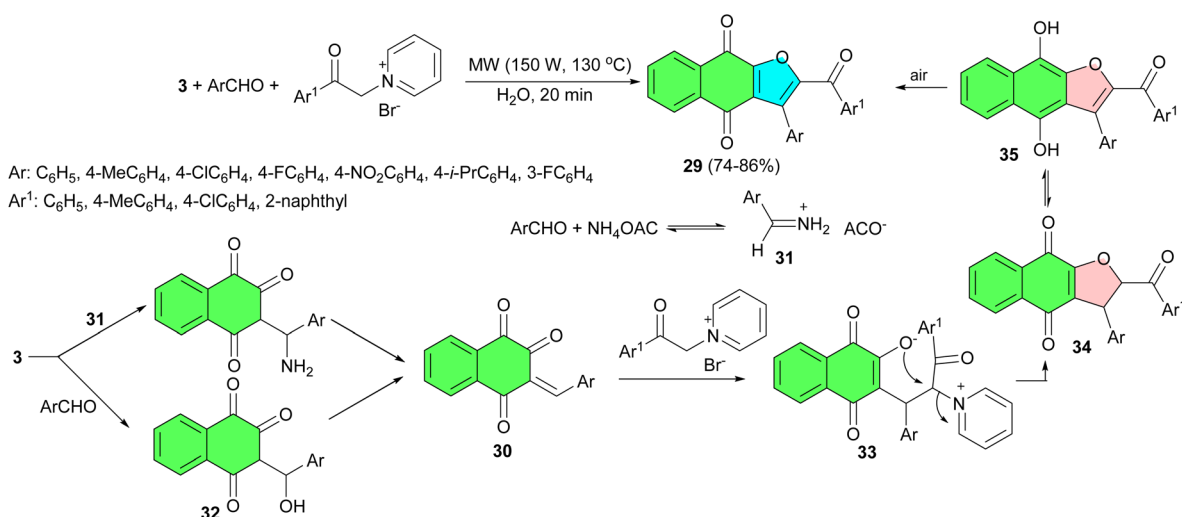
Scheme 9 CuO catalyzed synthesis of benzofuronaphthoquinone **23**.

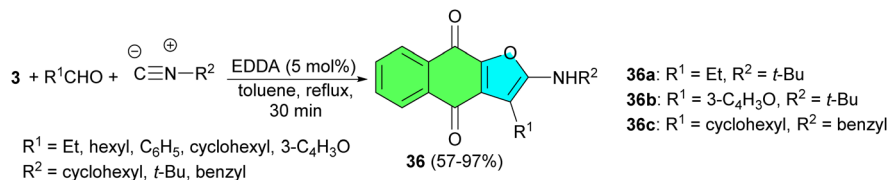
Scheme 10 Regioselective synthesis of linear naphtho[2,3-*b*]-furan-4,9-diones **25**.Scheme 11 Synthesis of highly substituted linear naphtho[2,3-*b*]-furan-4,9-dione derivatives **28**.

reaction of lawsone with an iminium ion **31** generated from aldehyde and ammonium acetate, followed by elimination of ammonia, or (ii) the ammonium acetate-catalyzed reaction of lawsone with the starting aldehyde to afford aldol **32**, which would undergo dehydration. Then, the Michael addition of pyridinium ylide to **30** presumably affords the pyridinium enolate **33**, which subsequently undergoes annulation *via*

displacement of pyridine to give **34**, which is probably in tautomeric equilibrium with the corresponding hydroquinone **35**. Ultimately, these intermediates are transformed into the final products **29** *via* air oxidation.⁴⁰

After that, 2-amino-naphtho[2,3-*b*]furan-4,9-diones **36** were synthesized in 57–97% yields using a one-pot three-component reaction of lawsone with various aldehydes and aromatic or

Scheme 12 Microwave-assisted synthesis of functionalized naphtho[2,3-*b*]furan-4,9-diones **29**.

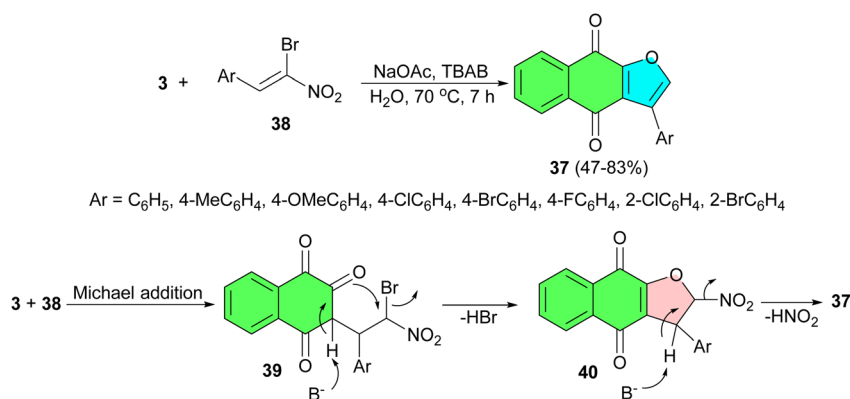
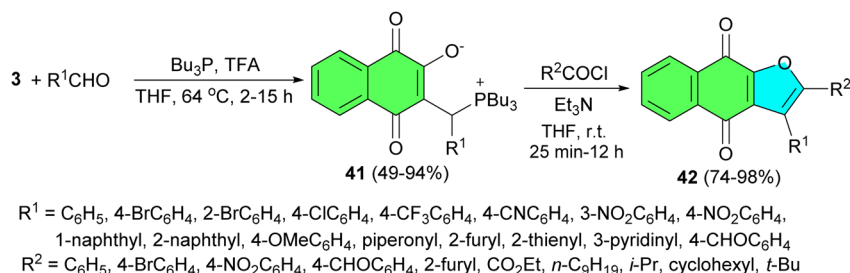
Scheme 13 EDDA catalyzed synthesis of 2-amino-naphtho[2,3-*b*]furan-4,9-diones 36.

aliphatic isocyanides in the presence of catalytic amounts of EDDA (ethylenediamino diacetate) under refluxing in toluene for 30 min under nitrogen atmosphere (Scheme 13). These compounds were tested for cytotoxicity against several human solid tumor cell lines (MCF7, MCF7/BUS, and SK-Br-3). Compounds **36a**, **36b**, and **36c** exhibit the highest cytotoxicities. Compound **36a** shows selectivity toward the hormone-dependent cell line MCF7/BUS with a GI₅₀ value of 9.2 μM. Compounds **36b** and **36c** present good cytotoxicities against MCF7, MCF7/BUS, and SK-Br-3 tumor cell lines. The best GI₅₀ was achieved by compound **36b** against the SK-Br-3 cell line, with GI₅₀ = 1.6 μM. Moreover, the electronic properties of these aromatic donor-acceptor derivatives were analyzed by means of their redox potentials and solvatochromic properties.⁴¹

Further, a simple and efficient protocol developed for the synthesis of 3-phenylnaphtho[2,3-*b*]furan-4,9-diones **37** in 47–83% yields by domino reaction of α-bromonitroalkenes **38** to lawsone using NaOAc and TBAB (tetrabutylammonium bromide) in H₂O at 70 °C for 7 h. A mechanistic rationalization

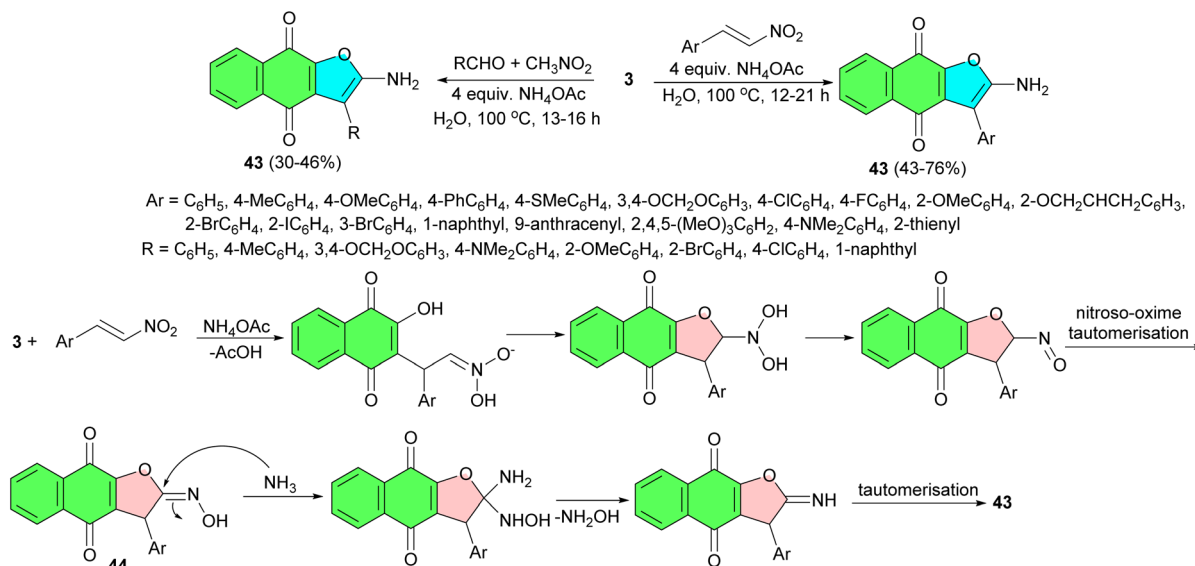
for this reaction is provided in Scheme 14. The domino reaction of α-bromonitroalkenes with **3** gives the Michael addition product **39** catalyzed by NaOAc. Then, the enolate anion was formed under the basic conditions and the subsequent intramolecular nucleophilic displacement of **39** affords intermediate **40**. Subsequently, elimination of the nitro group leads to the formation of the desired product **37**. The absorption characteristics of the compounds were examined by UV-vis spectra and fluorescence spectroscopy. All compounds were fluorescent in solution emitting at blue light (432–433 nm), green light (512–536 nm), or yellow light (591 nm).⁴²

The Lin group described an efficient and attractive synthesis of a series of poly-functionalized phosphorus zwitterions **41** in 49–94% yields was achieved *via* three-component reactions of lawsone, aldehydes, and Bu₃P in the presence of acidic promoter in THF at 64 °C for 2–15 h. These polysubstituted zwitterions could regioselectively undergo further transformations to synthetically important furanonaphthoquinones **42** in 74–98% yields *via* the intramolecular Wittig reaction in

Scheme 14 Preparation of 3-phenylnaphtho[2,3-*b*]furan-4,9-diones 37.

Scheme 15 Regioselective synthesis of polysubstituted furanonaphthoquinones 42.





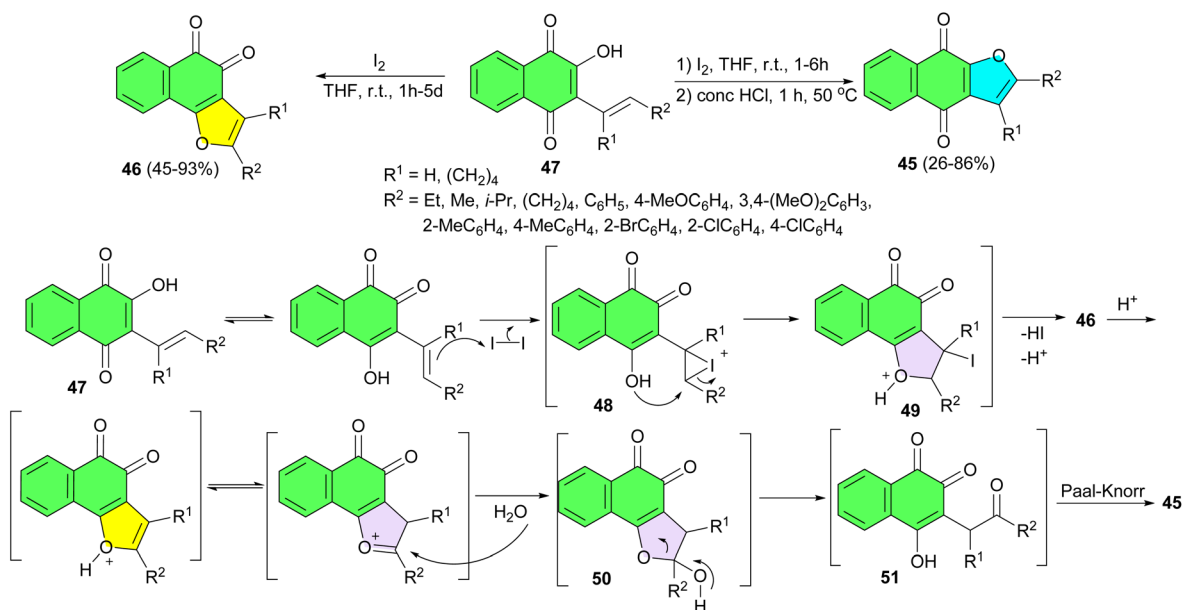
Scheme 16 Synthesis of 3-substituted 2-aminonaphtho[2,3-b]furan-4,9-diones 43.

THF at room temperature for 25 min to 12 h (Scheme 15). The outstanding features of the synthetic methodology are the flexibility to synthesize furanonaphthoquinone derivatives with various substituents at the 2- and 3-positions of the furan segment, and a wide selection of substituents that can be easily obtained from commercially available aldehydes and acid chlorides.⁴³

In 2013, Yao and co-workers demonstrated an efficient base-catalyzed synthesis of 3-substituted 2-aminonaphtho[2,3-b]furan-4,9-dione derivatives 43 in 43–76% yields from lawsone and nitroalkenes under aqueous conditions at 100 °C for 12–21 h. Moreover, synthesis of 43 could be achieved directly in a three-component reaction of lawsone, aldehyde, and

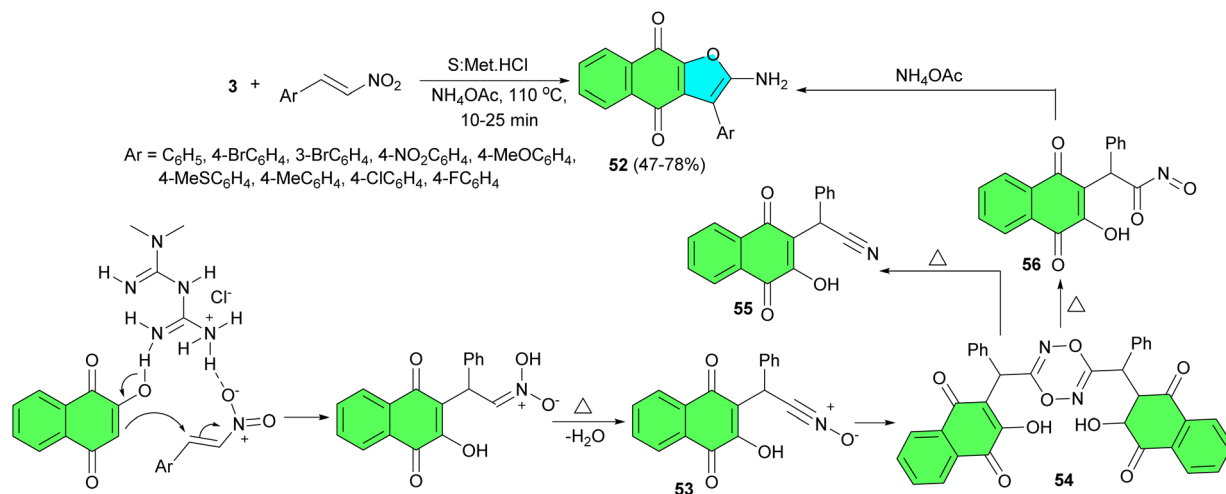
nitromethane under the same conditions. The most important feature of this methodology is the conversion of the nitro group into an amino group without any reducing agent. The mechanism was thought to involve the formation of a Michael adduct from the reaction of a nitroalkene and lawsone, which undergoes cyclization in the presence of ammonium acetate to produce the oxime derivative 44. The oxime derivative is then transformed into 43 by the nucleophilic addition of ammonia to the oxime followed by the elimination of hydroxylamine (Scheme 16).⁴⁴

In 2015, a facile method for highly regioselective synthesis of both linear and angular naphthofuroquinones 45 and 46 in 26–93% yields were developed *via* iodine mediated cyclization of 2-



Scheme 17 Iodine-mediated regioselective synthesis of naphthofuroquinones 45–46.

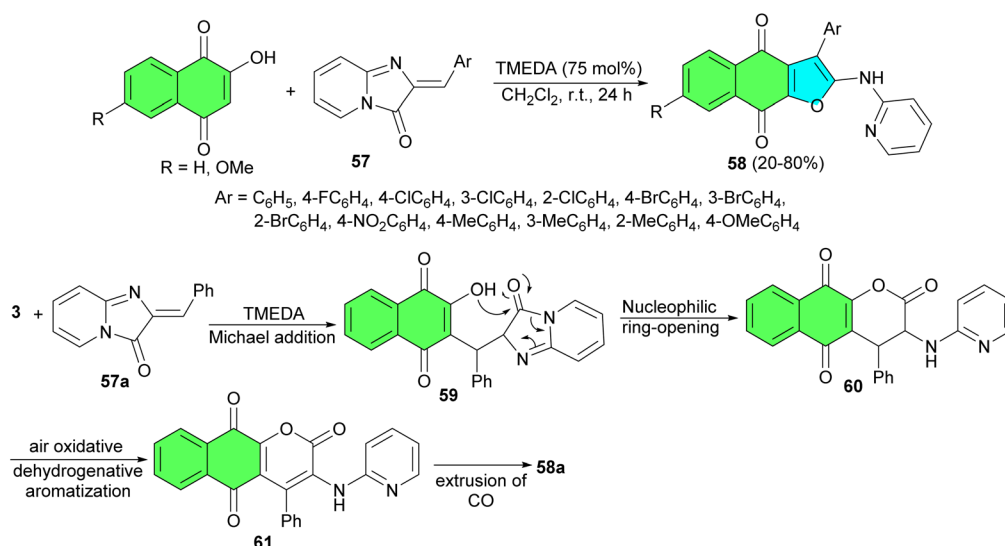


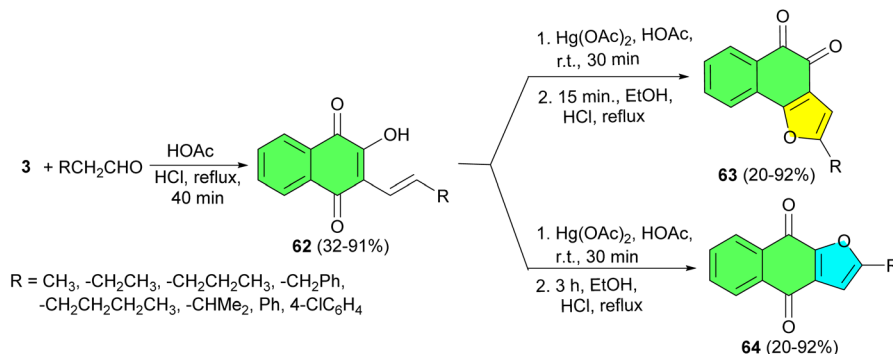
Scheme 18 Synthesis of 2-amino-3-arylnaphtho[2,3-*b*]furan-4,9-diones **52**.

hydroxy-3-substitutedvinyl-1,4-naphthoquinones **47** in THF at room temperature. The possible reaction mechanism is proposed in Scheme 17. The reaction was supposed to be initiated by the nucleophilic attack of the olefinic double bond to iodine to form the iodonium ion **48**. The following decomposition of **48** might involve the attack from the neighboring hydroxyl group to form the unstable iodide intermediate **49** which would automatically eliminate one molecular of HI to give the angular naphthofuroquinone **46**. Under acidic condition, hemiketal **50** could be formed through the nucleophilic attack of H₂O to the C-2 position of the protonated angular naphthofuroquinone **46**. The following decomposition of hemiketal **50** would give the 1,4-diketone intermediate **51**, which would finally afford the linear naphthofuroquinone **45** through a classic Paal-Knorr process. Considering the very mild reaction condition, high yields, high angular/linear selectivity, and non-involvement of transitional metals, this method might

find great application in the synthesis of naphthofuroquinones.⁴⁵

In 2016, an efficient and environmentally benign one-pot method reported for the diverse synthesis of 3-substituted 2-aminonaphtho[2,3-*b*]furan-4,9-diones **52** in 47–78% yields *via* the domino reaction between lawsonone, β -nitrostyrenes and ammonium acetate in a new deep eutectic solvent made of sorbitol and metformin HCl at 110 °C for 10–25 min. A pronounced positive solvatochromism was observed for the electron donor–acceptor conjugated system of these products, resulting in bathochromic and hyperchromic shifts of their visible absorption band in polar and protic solvents. A plausible mechanism is outlined in Scheme 18. The progress of the reaction involves thermal dehydration of the Michael adducts that occurs at temperatures above 90 °C to give the intermediate nitrile-*N*-oxides **53**. These zwitterionic intermediates undergo a dipolar dimerization reaction to yield the intermediate 1,4,2,5-

Scheme 19 Synthesis of naphtho[2,3-*b*]furan-4,9-diones **58** under aerobic condition.

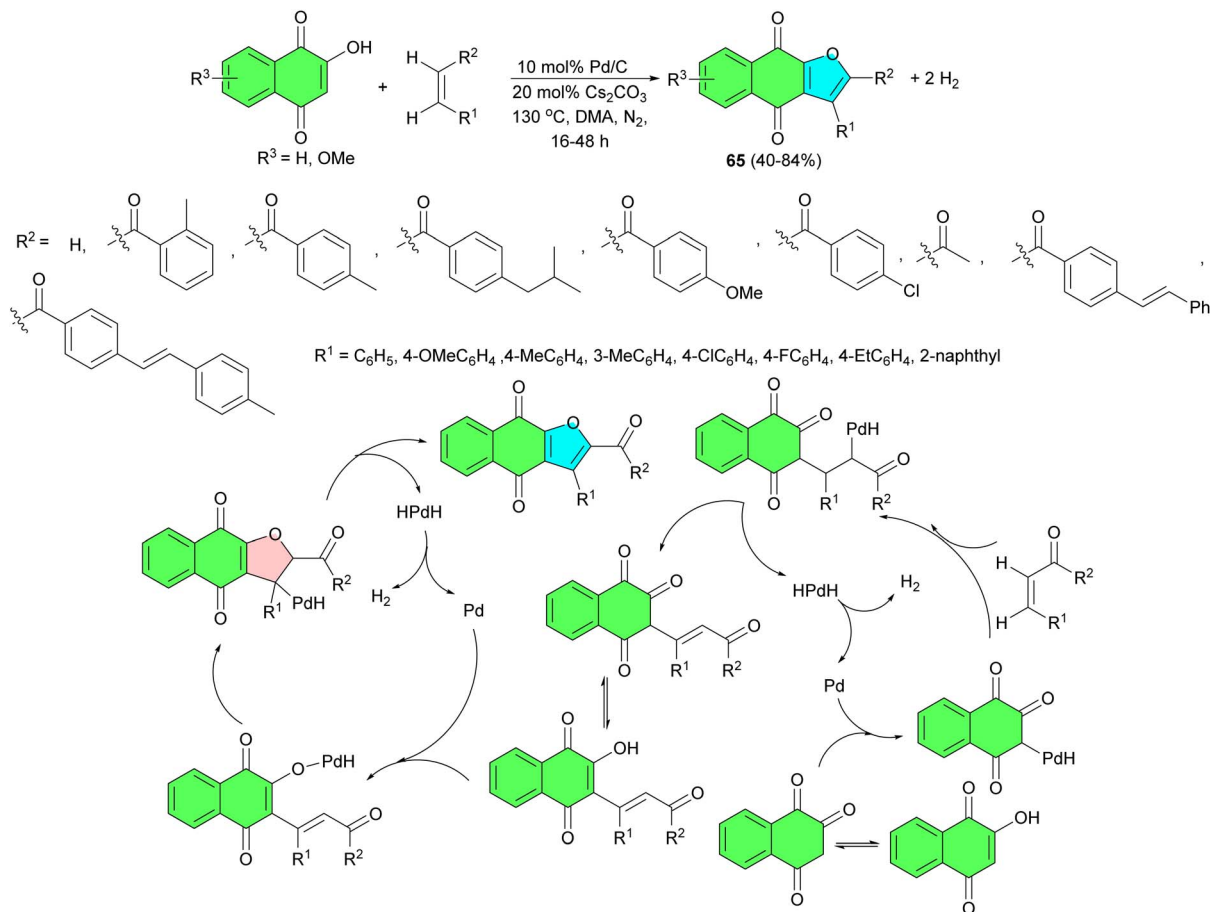


Scheme 20 Synthesis of furanonaphthoquinones 63–64.

dioxadiazines **54** followed by a chain of thermal demanding reactions involving electrocyclic scission of the dioxadiazine ring and conversion of the resulting two fragments into naphtho[2,3-*b*]furan-4,9-diones. Upon heating, the fragmented intermediate **55** is presumed to undergo a direct cyclization while its twin **56** cyclocondenses with ammonia to deliver the product **52**.⁴⁶

In 2017, Zhou *et al.* developed an unprecedented base-promoted oxidative coupling of lawsone derivatives with (*Z*)-2-ylideneimidazo[1,2-*a*]pyridin-3(2*H*)-ones **57**, which provide

a convenient approach to access naphtho[2,3-*b*]furan-4,9-dione derivatives **58** with a 2-aminopyridine moiety using TMEDA in 20–80% yields under mild reaction conditions. These compounds in ethanol showed the characteristic intense charge-transfer bands (π - π^* transitions) occurring in the visible region. Compound **58a** exhibited selective response to Hg²⁺ and Pd²⁺ *via* a chelate-binding module and can be developed as a sensitive chromogenic sensor for Hg²⁺ in the presence of a range of competing cations in aqueous media. A tentative mechanism is postulated in Scheme 19. With the aid of the base

Scheme 21 Palladium-catalyzed synthesis of naphtho[2,3-*b*]furan-4,9-diones **65**.

catalyst, Michael addition of **3** to **57a** afforded an adduct intermediate **59**. Then, a suffered an intramolecular nucleophilic ring opening of the imidazo[1,2-*a*]pyridin-3(2*H*)-one moiety to provide a 3,4-dihydrobenzo[*g*]chromene-2,5,10-trione intermediate **60**. Under aerobic conditions, benzo[*g*]chromene-2,5,10-trione **61** was formed through the air oxidative dehydrogenative aromatization of **60**. Finally, extrusion of carbon monoxide from intermediate **61** resulted in the generation of thermodynamically more stable naphtho[2,3-*b*]furan-4,9-dione **58a**.⁴⁷

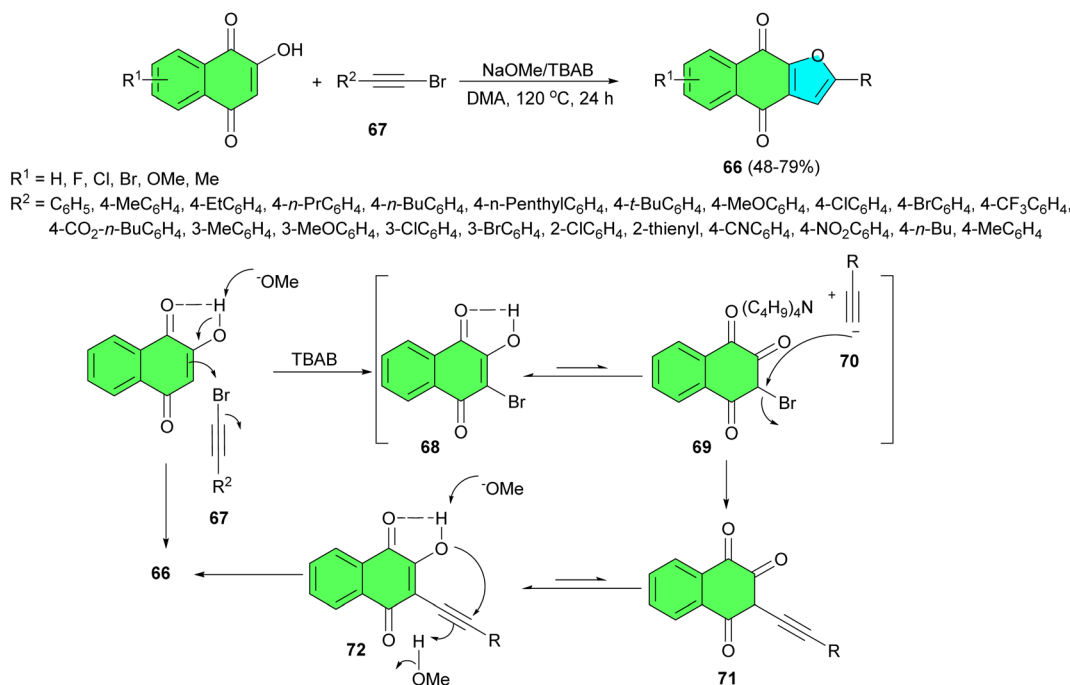
Next, the HNQs **62** synthesized in 32–91% yields by aldol condensation between lawsone and appropriate aldehydes in glacial acetic acid, followed by concentrated HCl under reflux conditions for 40 min. The FNQs **63** and **64** were prepared by oxidative cyclization of the corresponding **62** with Hg(OAc)₂ in glacial acetic acid at room temperature for 30 min and at 80 °C for 15 min. All compounds disclosed higher *in vitro* antiplasmodial activity than lapachol. *Ortho*- and *para*-naphthoquinones with a furan ring fused to the quinonoid moiety were more potent than 2-hydroxy-3-(1'-alkenyl)-1,4-naphthoquinones, while *ortho*-furanonaphthoquinones were more cytotoxic (Scheme 20).⁴⁸

In 2019, a reverse hydrogenolysis process developed for two-site coupling of 2-hydroxy-1,4-naphthoquinones with olefins to produce naphtho[2,3-*b*]furan-4,9-diones **65** in 40–84% yields and hydrogen. The reaction is catalyzed by commercially available Pd/C (10 mol%) without oxidants and hydrogen acceptors in DMA at 130 °C for 16–48 h. The results showed that enones with a variety of groups such as methyl, ethyl, isobutyl, and methoxyl could be directly coupled with 2-hydroxy-1,4-naphthoquinone to give moderate to high isolated yields. Fluorine- and chlorine-containing enones were unaffected.

Non-aromatic enones could also be smoothly coupled with the 2-hydroxy-1,4-naphthoquinones to afford the corresponding products with moderate isolated yields. The proposed mechanism is illustrated in Scheme 21.⁴⁹

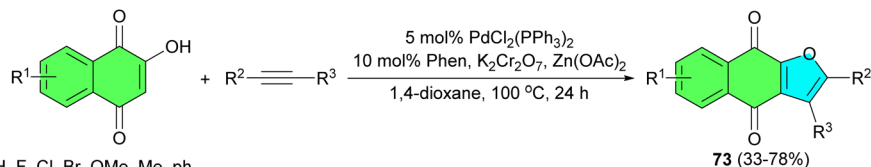
In 2020, Peng *et al.* explored a transition-metal-free route for tandem one-pot synthesis of naphthoquinonefuran derivatives **66** in 48–79% yields from 2-hydroxynaphthoquinones and arylethynyl bromides **67** using NaOMe/TBAB in DMA at 120 °C for 24 h. A plausible reaction pathway is outlined in Scheme 22. A sodium methoxide-promoted deprotonation of 2-hydroxynaphthoquinones produced sodium enolate, which was then brominated at the 3-position to form the intermediate **68** using the alkynyl bromide **67** as the source of Br⁺. The tautomeric isomer **69** in the keto-form from **68** underwent a nucleophilic attack by the produced arylacetylide **70** to give the alkynated intermediate **71**, which was transformed into 3-alkynated hydroxynaphthoquinone **72** through a keto-enol tautomeric process. A sodium methoxide-promoted deprotonation and subsequent intramolecular nucleophilic addition of enolate anion to conjugated ynone furnished the cyclized product **66**.⁵⁰

The Peng group developed a modular approach for rapid syntheses of diverse naphtho[2,3-*b*]furan-4,9-dione derivatives **73** through Pd-catalyzed oxidative annulations of 2-hydroxynaphthalene-1,4-diones with readily accessible unactivated internal alkynes in 1,4-dioxane at 100 °C for 24 h. The combination of Zn(OAc)₂ and K₂Cr₂O₇ was found to be essential for the efficient formation of furonaphthoquinones in 33–78% yields. This synthetic method exhibits a broad substrate scope with good yields and excellent regioselectivity for aryl, alkyl substituted alkynes. Two competitive catalytic cycles for the synthesis of **73** are proposed in Scheme 23. In the presence of



Scheme 22 Transition-metal-free synthesis of naphthoquinonefurans **66**.

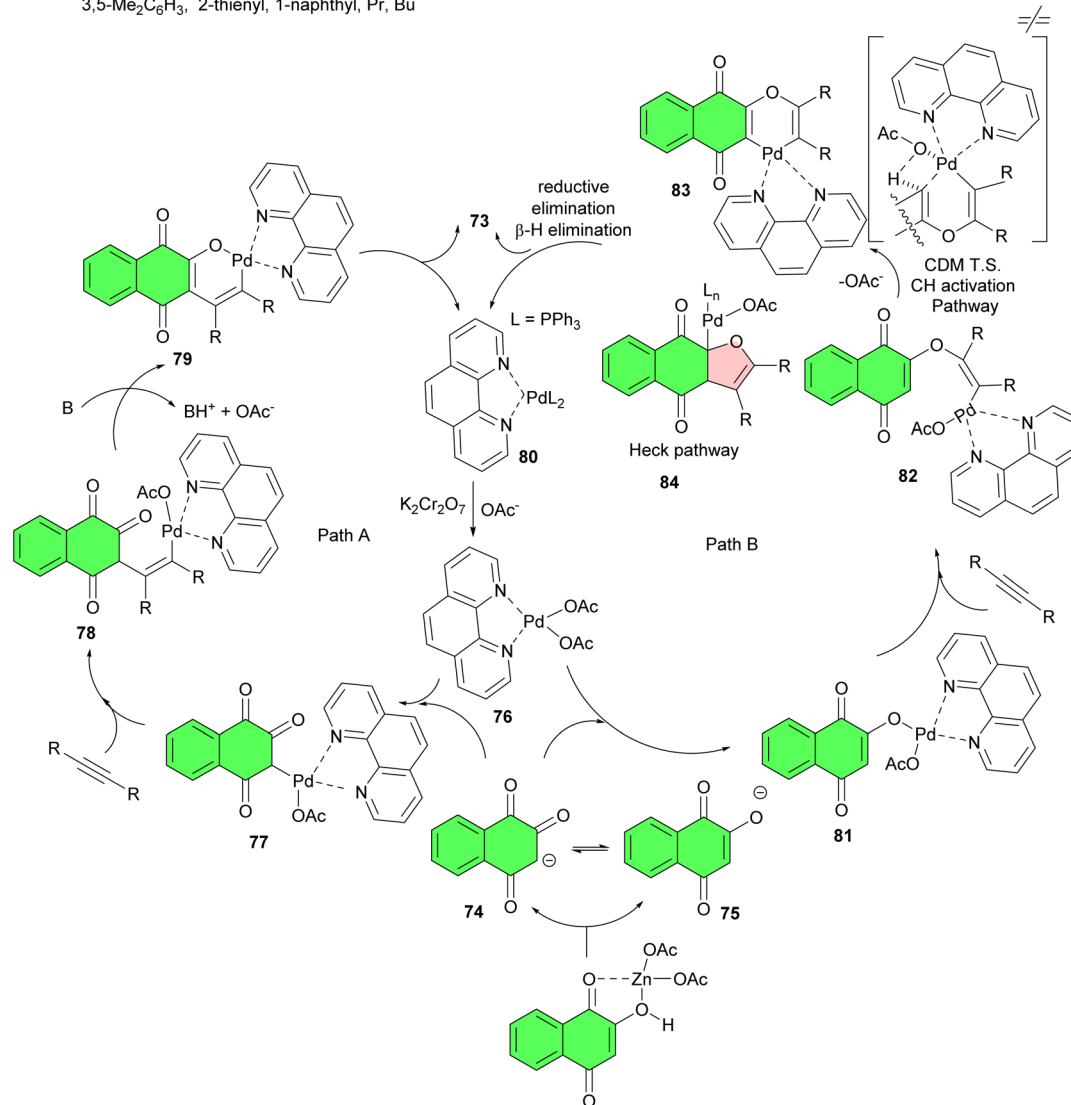




$\text{R}^1 = \text{H, F, Cl, Br, OMe, Me, Ph}$

$\text{R}^2 = \text{C}_6\text{H}_5, 4\text{-MeOC}_6\text{H}_4, 4\text{-MeC}_6\text{H}_4, 4\text{-ClC}_6\text{H}_4, 4\text{-BrC}_6\text{H}_4, 4\text{-FC}_6\text{H}_4, 3\text{-MeC}_6\text{H}_4, 3\text{-MeOC}_6\text{H}_4, 3\text{-ClC}_6\text{H}_4, 2\text{-ClC}_6\text{H}_4, 2\text{-MeC}_6\text{H}_4, 2\text{-MeOC}_6\text{H}_4, 3,5\text{-Me}_2\text{C}_6\text{H}_3, 2\text{-thienyl, 1-naphthyl, Pr}$

$\text{R}^3 = \text{C}_6\text{H}_5, 4\text{-MeOC}_6\text{H}_4, 4\text{-MeC}_6\text{H}_4, 4\text{-ClC}_6\text{H}_4, 4\text{-BrC}_6\text{H}_4, 4\text{-FC}_6\text{H}_4, 3\text{-MeC}_6\text{H}_4, 3\text{-MeOC}_6\text{H}_4, 3\text{-ClC}_6\text{H}_4, 2\text{-ClC}_6\text{H}_4, 2\text{-MeC}_6\text{H}_4, 2\text{-MeOC}_6\text{H}_4, 3,5\text{-Me}_2\text{C}_6\text{H}_3, 2\text{-thienyl, 1-naphthyl, Pr, Bu}$

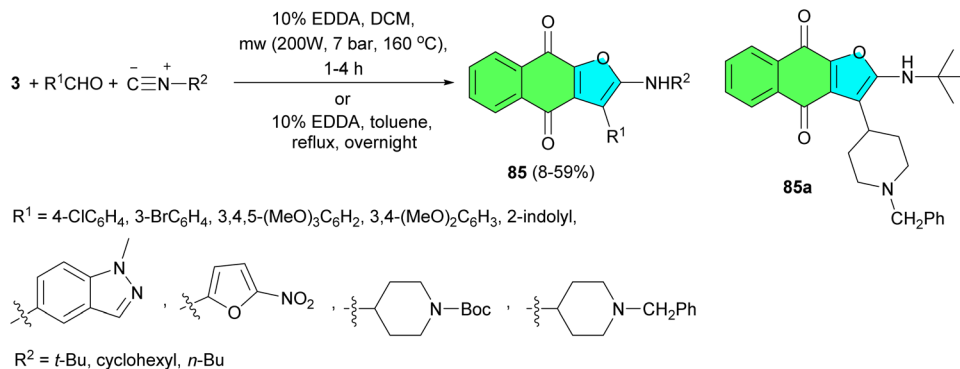


Scheme 23 Palladium(II)-catalyzed synthesis of naphtho[2,3-*b*]furan-4,9-diones **73**.

$\text{Zn}(\text{OAc})_2$ as the base, the deprotonation of 2-hydroxynaphthalene-1,4-dione yielded two tautomeric anions, which existed as either carbanion **74** in the keto-form or as oxygen anion **75** in the enol-form. At this stage, attack of anion onto the electrophilic Pd(II) species may occur in two different ways (paths **A** and **B**). Path **A** involved the attack of carbanion **74** onto catalytic active species **76** to form alkyl-Pd(II) species **77**. Subsequently, the coordination of the internal alkyne to **77** would induce its carbopalladation to afford an alkenyl palladium(II) complex **78**. Base-assisted further deprotonation of the

ketone α -carbon of **78** led to O-Pd bond formation, affording intermediate **79**. Palladacycle **79** underwent C-O reductive elimination to afford the desired product **73** and a Pd(0) species **80**, which was oxidized by $\text{K}_2\text{Cr}_2\text{O}_7$ to regenerate the active Pd(II) species **76** for the next catalytic cycle. On the other hand, mechanistic cycle **B** was initiated by the attack of enol anion onto the electrophilic Pd(II) species **76**, giving the enol-type palladium(II) **81**. The coordination followed by syn migratory insertion of internal alkyne into the O-Pd bond then afforded alkenyl-Pd(II) species **82**. The final product **73** can be formed

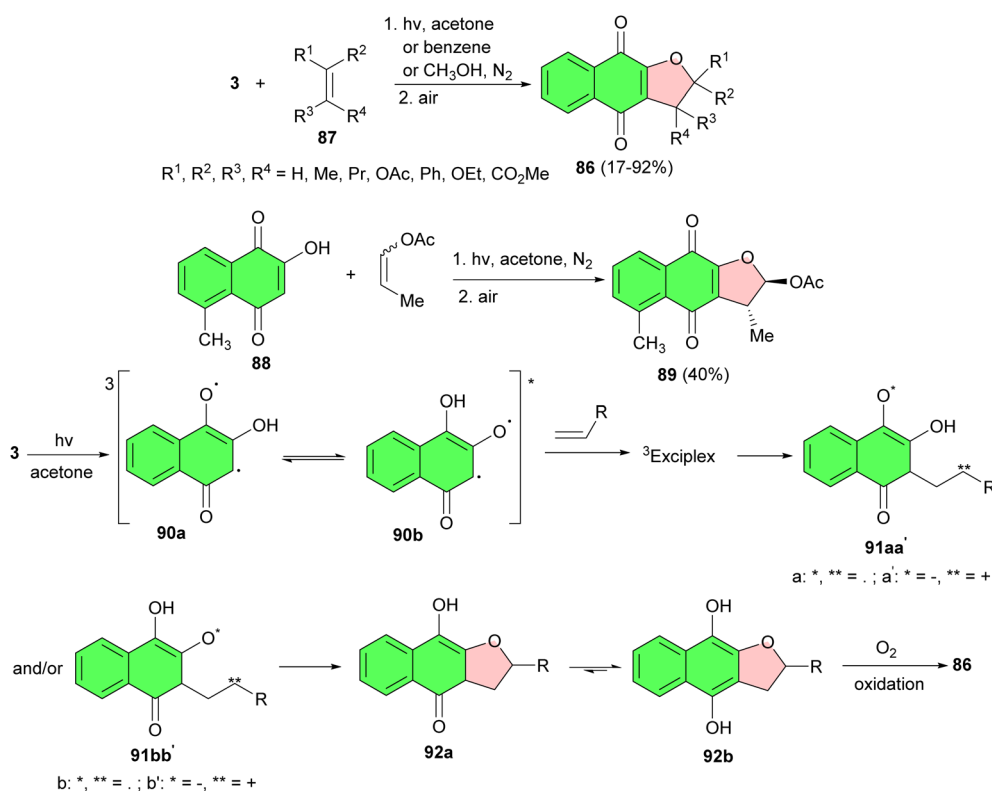


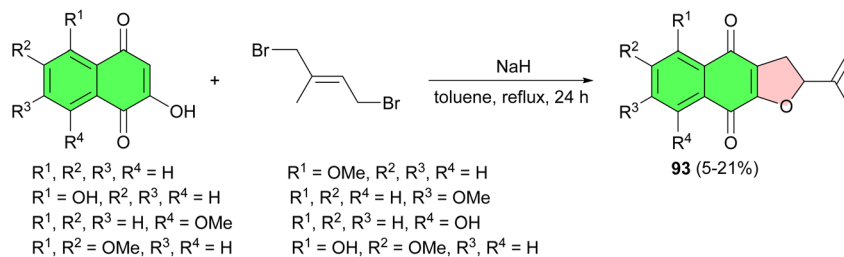
Scheme 24 EDDA catalyzed synthesis of naphthofuroquinones **85**.

possibly through two distinct pathways (C–H activation or Heck pathway). The C–H activation pathway involved a concerted metalation deprotonation (CMD) transition state of alkene to form the palladacycle **83**, which underwent C–C bond forming reductive elimination to afford the desired furonaphthoquinone **73** and regenerate a Pd(0) species **80**. The Heck pathway involved an intramolecular *syn* migratory insertion into the olefin moiety of **82** and then was followed by an isomerization process to give the σ -alkyl-palladium(II) acetate **84** with β -hydrogen in a *syn* position relative to the palladium atom. A *syn* β -hydride elimination afforded **73** and hydrido-palladium(II) acetate, which underwent a reversible reductive elimination to regenerate Pd(0) complex **80**. Finally, Pd(0)

resulting from an elimination process was oxidized to Pd(II) by $K_2Cr_2O_7$.⁵¹

In 2022, the García-Sosa reported synthetic approach to naphthofuroquinones **85** in moderate to good yields *via* a reaction involving lawsone, various aldehydes, and three isocyanides using EDDA in dichloromethane under microwave irradiation for 1–4 h at 160 °C and conventional method in the presence of EDDA in toluene under reflux overnight. In addition, for less-reactive aldehydes, two naphtho-enaminodione quinones were obtained. All compounds were evaluated for their anti-infectious activities. Among the naphthofuroquinone series, **85a** exhibited comparatively the best activity against *P. falciparum* ($IC_{50} = 2.5 \mu\text{M}$) and *M. tuberculosis* ($MIC = 9 \mu\text{M}$)

Scheme 25 One-step synthesis of 2,3-dihydonaphtho[2,3-*b*]furan-4,9-diones **86**.

Scheme 26 One-step preparation of 2-isopropenyl-2,3-dihydronaphtho-[2,3-*b*]furan-4,9-diones **93**.

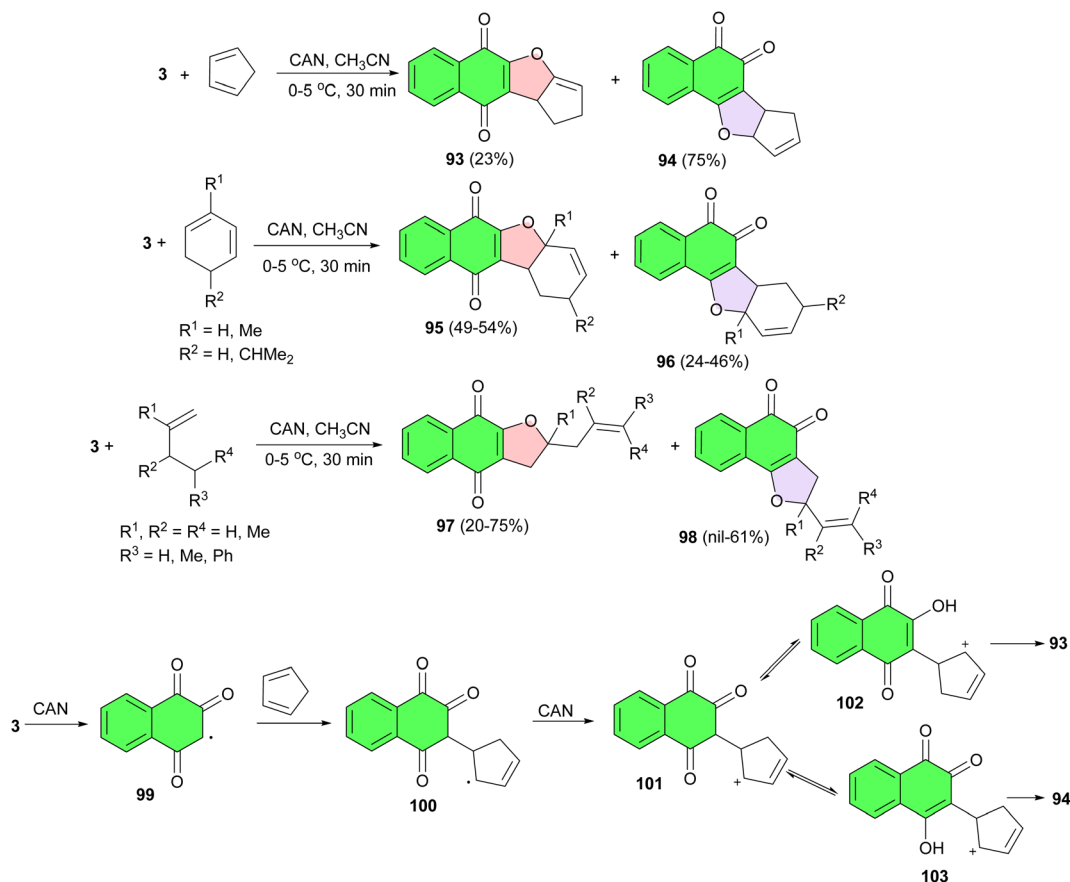
with better (*P. falciparum*) or equivalent (*M. tuberculosis*) values to already-known naphthofuroquinone compounds (Scheme 24).⁵²

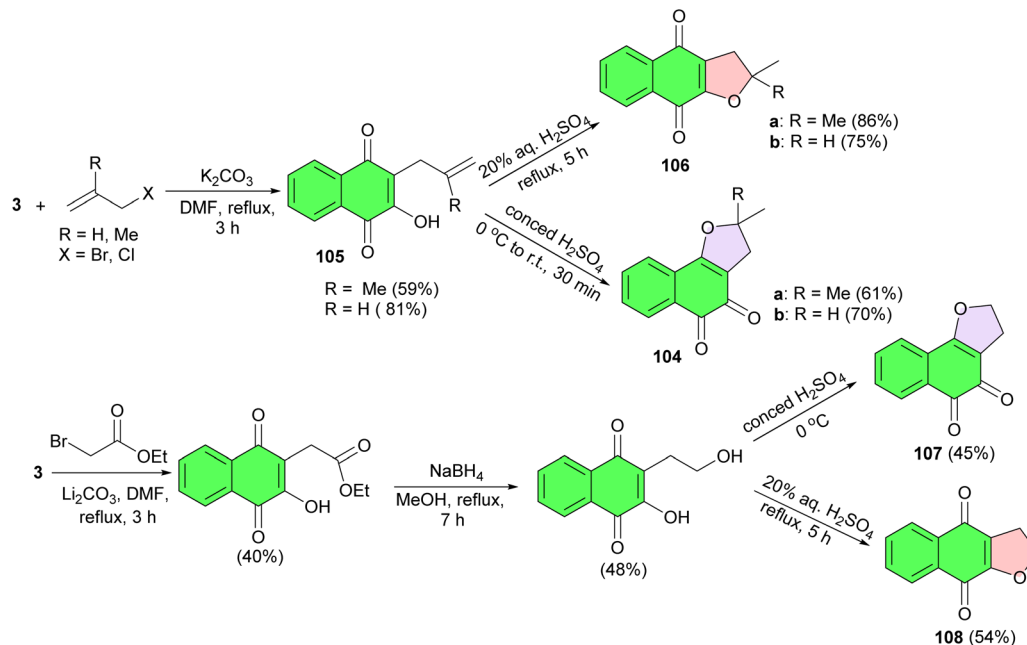
3. Synthesis of dihydronaphthofuroquinones

In 1991, Suginome *et al.* synthesized a series of 2,3-dihydronaphtho [2,3-*b*]furan-4,9-diones **86** in 17–92% yields by a new [2 + 3] type regioselective photoaddition of 2-hydroxy-1,4-naphthoquinones with a variety of alkenes **87** in acetone or benzene after 4–24 h. The yield of adduct decreased appreciably when the photoaddition was conducted in methanol. Moreover, photoaddition of 2-hydroxy-*methyl*-1,4-naphthoquinone (**88**)

with 1-propenyl acetate in acetone gave *trans*-2-acetoxy-2,3-dihydro-3,5-dimethylnaphtho[2,3-*b*]-furan-4,9-dione (**89**) in 40% yield. Mechanistically, in this process, irradiation of **3** in acetone or benzene generates tautomeric excited triplets (**90a**) and (**90b**), which react with an alkene through a triplet exciplex to give biradical (**91aa'**) and/or (**91bb'**). Intramolecular cyclization of the intermediate gives hydroquinones (**92a**) and (**92b**). 2,3-Dihydronaphthofuran-4,9-dione **86** is then formed by air oxidation of the hydroquinone during the workup and isolation procedures (Scheme 25).⁵³

In 2001, Yamaguchi *et al.* one-step preparation of some 2-isopropenyl-2,3-dihydronaphtho-[2,3-*b*]furan-4,9-diones **93** in 5–21% yields described by the reaction of lawsone derivatives with 1,4-dibromo-2-methylbut-2-ene in dry toluene in the

Scheme 27 CAN mediated synthesis of linear and angular naphthofurandiones **93–98**.

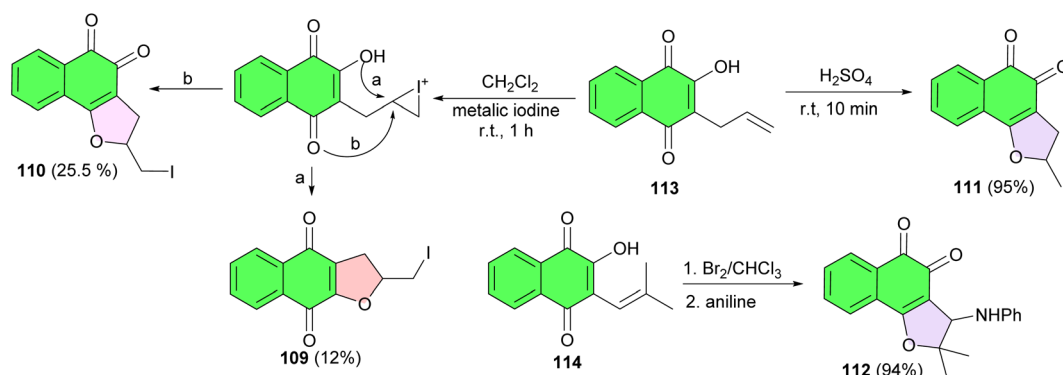
Scheme 28 Synthesis of dihydrofuranonaphthoquinones **104**, **106–108**.

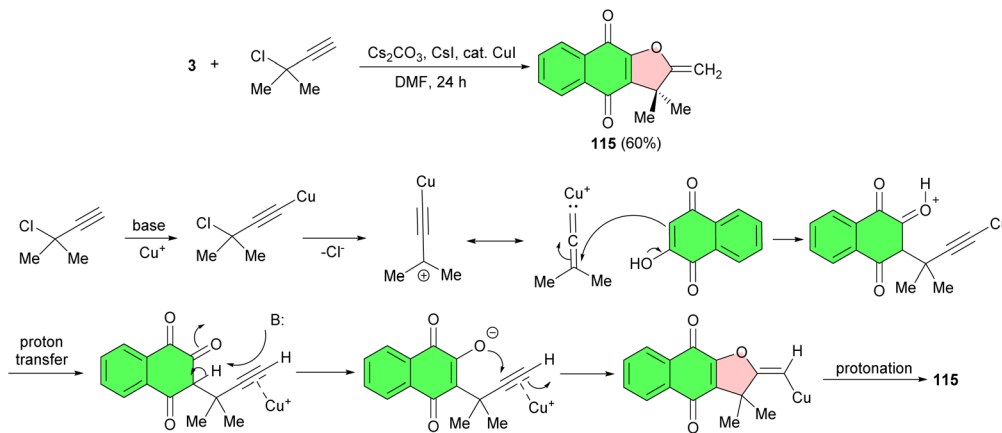
presence of sodium hydride under reflux condition for 24 h (Scheme 26).⁵⁴

Lawsonone undergoes can mediated oxidative addition to various dienes followed by ring closure in CH₃CN at 0–5 °C for 30 min yielding linear and angular naphthofurandiones **93–98** in 20–75% yields. The proposed mechanism is outlined in Scheme 27. Oxidation of lawsonone by CAN lead to the radical **99**, which is trapped by the diene to yield the reactive intermediate **100**. The latter further oxidized by CAN to the cation **101**, which in turn undergoes rearrangement yielding **102** and **103**. The cyclization of **102** leads to **93** whereas **103** affords **94**.⁵⁵

Dihydrofuranonaphthoquinones **104** were synthesized in only two steps (C-allylation and cyclization) from lawsonone which could be isolated from the shrub, *Lawsonia inermis*. C-Allylation of lawsonone using K₂CO₃ in DMF under reflux conditions for 3 h gave lawsonone derivatives **105** in 59 and 81% yields. Compounds **105** were cyclized to furano-1,2-naphthoquinone **104a–b** by concentrated sulfuric acid at 0 °C to room temperature for

30 min in 61 and 70% yields. With respect to cyclization of olefinic alcohols **105**, using 20% aqueous sulfuric acid under reflux conditions for 5 h, it was found that the 1,4-naphthoquinone products **106a–b** were obtained in 75 and 86% yields. It seemed that the mechanism of using concentrated sulfuric acid is different from using aqueous sulfuric acid. Also, furanonaphthoquinones **107** and **108** were accessed in 45 and 54% yields from C-allylation of lawsonone by α-bromoacetate ethyl ester followed by reduction with NaBH₄ in MeOH and then cyclization by sulfuric acid. The mechanism of concentrated H₂SO₄ action may be mediated by tautomerization of protonated naphthoquinone whereas using aqueous H₂SO₄, protonation of the double bond at the side chain may be faster than protonation of the carbonyl group of naphthoquinone and then the hydroxy group at C-3 position attacked the carbocation straight away (Scheme 28). Compound **108** had significant cytotoxicity against HeLa cancer cell line (IC₅₀ value of 9.25 mM) while it showed no toxic to vero cell.⁵⁶

Scheme 29 Synthesis of naphthofuranquinones **109–112** with activity against *Trypanosoma cruzi*.

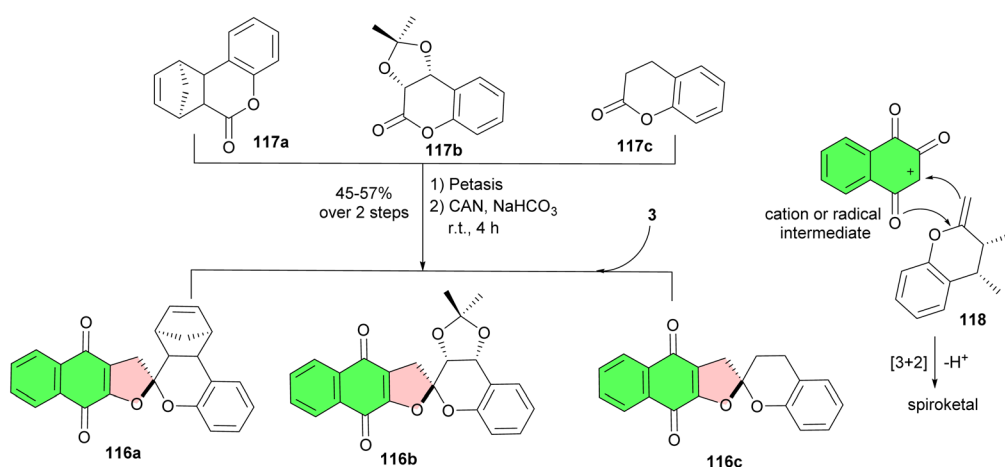
Scheme 30 Synthesis of dehydro- α -dunnione **115**.

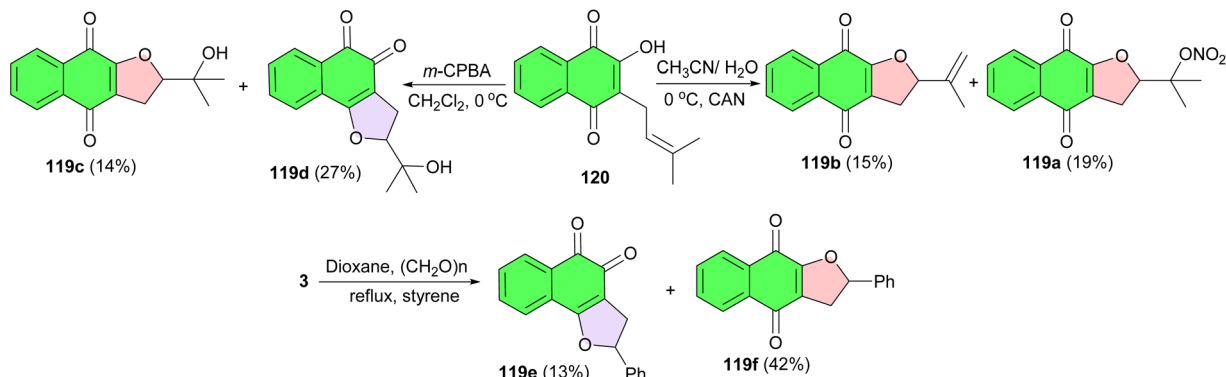
In 2006, The de Castro group disclosed four naphthofuranquinones **109–112**, obtained from 2-hydroxy-3-allylnaphthoquinone (**113**) and nor-lapachol (**114**) and their structures established by physical and X-ray analysis. Compounds **109** and **110** were obtained by addition of iodine to **113** followed by cyclization generating a furan ring and separated by column chromatography. Compound **111** was obtained through the acid-catalyzed reaction by dissolution of **113** in sulfuric acid. Compound **112** was synthesized by addition of bromine and aniline to **114**. The $IC_{50}/24\text{ h}$ for **109–112** in assays with *T. cruzi* trypomastigotes was between 157 and 640 μM , while the value for crystal violet was $536.0 \pm 3.0\ \mu\text{M}$. Compounds **109–111** also inhibited epimastigote proliferation (Scheme 29).⁵⁷

The Cooke group reported synthesis of dehydro- α -dunnione **115** in 60% yield by the reaction of lawsone with 3-chloro-3-methyl-1-butyne *via* one-pot, formal [3 + 2] cyclization. The reaction was carried out in the presence of CuI as catalyst in DMF as solvent for 24 h. The reduction of the exocyclic double bond in compound **115** for the synthesis of α -dunnione, using Pd/C and pressurized H_2 gas in MeOH, showed no progress even after several days or weeks. A plausible mechanism is illustrated

in Scheme 30. They suggested that under the conditions of the coupling, a zwitterion-vinyl carbene intermediate is formed. This species couples with lawsone at C-3. Proton transfer (intramolecular or intermolecular) provides the terminal alkyne, π -activated by the copper ions present. Cyclization by the nucleophilic oxygen at C-2 on the activated alkyne thus forms the exocyclic enol vinylcuprate, which is protonated *in situ* or during the aqueous work-up.⁵⁸

Reich and co-workers described a highly selective oxidative [3 + 2] cycloaddition of chiral enol ethers and hydroxynaphthoquinone. This convergent strategy is amenable to an enantioselective synthesis of naphthoquinone spiroketals **116a–c**. In this reaction, initially, the carbonyl functionality in chromone compounds **117a–c** was individually subjected to methylenation using the Petasis reagent. After purification, the corresponding exocyclic enol ether **118** was subjected to oxidation by cerium ammonium nitrate in the presence of the 2-hydroxy-1,4-naphthoquinone (Scheme 31). Naphthoquinone spiroketal **116a** was found to inhibit DNA-polymerase and telomerase in a manner resembling α -rubromycin and β -rubromycin.⁵⁹

Scheme 31 Synthesis of naphthoquinone spiroketals **116** by diastereoselective oxidative [3 + 2] cycloaddition.

Scheme 32 Synthesis of dihydronaphthofuranquinone derivatives **119a–f**.

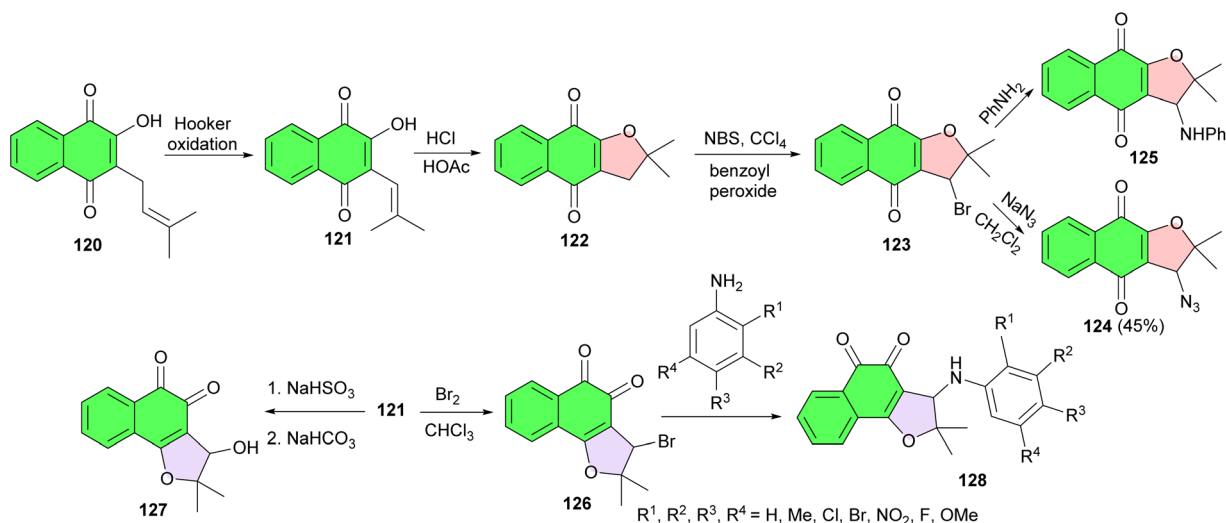
The Estevez-Braun group reported synthesis of naphthofuranquinone derivatives **119a–f** from lapachol (**120**) or lawsone under various conditions (Scheme 32). The synthesized naphthoquinone derivatives tested in human promyelocytic leukemia HL-60 cell line and the computational models have facilitated the identification of structural elements of the ligands that are key for antitumoral properties. The results of the study provided a valuable tool in designing new and more potent cytotoxic analogues.⁶⁰

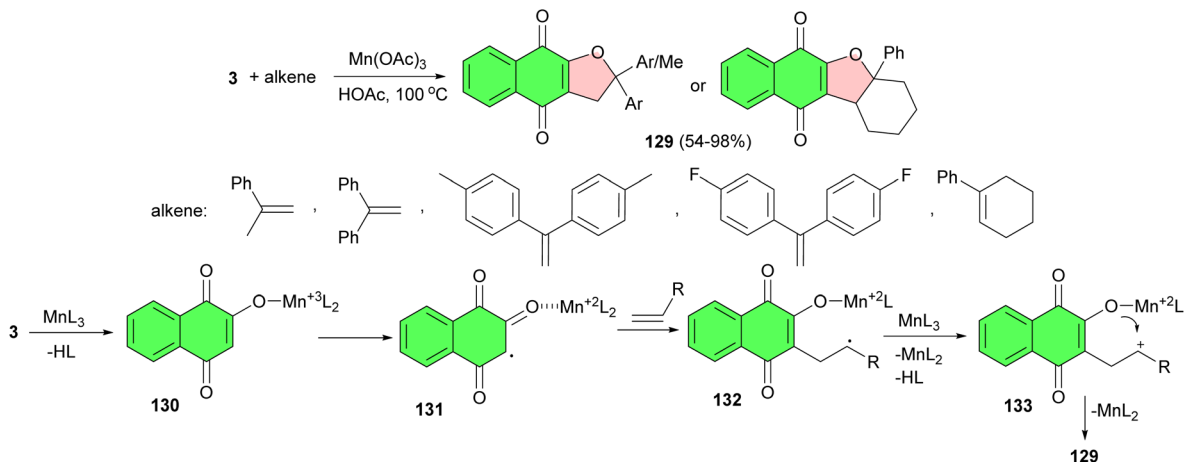
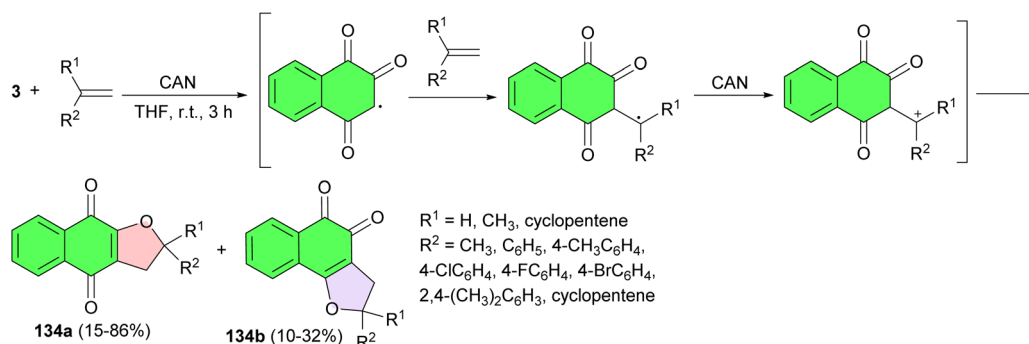
In 2008, the de Castro group reported synthesis of some naphthofuranquinone derivatives from nor-lapachol (2-hydroxy-3-(2-methyl-propenyl)-[1,4]-naphthoquinone) **121** was obtained from lapachol (2-hydroxy-3-(3'-methyl-2-butenyl)-1,4-naphthoquinone) (**120**) by Hooker oxidation.⁶¹ The treatment of **121** with HCl/AcOH produced nor- α -lapachone **122** which was transformed into 3-bromo-nor-lapachone **123**. Reaction of **123** with sodium azide in CH_2Cl_2 and with aniline gave, respectively, the corresponding azide and arylamino derivatives **124** and **125**. The reaction of **121** with bromine in CHCl_3 originated **126**, the starting material for the synthesis of the naphthoquinone **127** and of the *ortho*-arylamino-naphthofuranquinones **128**. The compounds were

rationalized based on hybrid drugs and appear as important compounds against *Trypanosoma cruzi* (Scheme 33).⁶²

Next, the Yılmaz group reported synthesis of 2,3-dihydronaphtho[2,3-*b*]furan-4,9-diones **129** in 54–98% yields in different molar ratios by the radical cyclizations of 2-hydroxy-1,4-naphthoquinone with electron-rich alkenes in the presence of manganese(III) acetate under nitrogen in HOAc at 100 °C. According to the proposed mechanism, $\text{Mn}(\text{OAc})_3$ (MnL_3) and hydroxyenone give manganese(III)-enolate complex **130**, and an α -carbon radical **131** is formed on this structure while Mn^{3+} is reduced to Mn^{2+} . A radical intermediate product **132** is obtained in the addition of the α -carbon radical to alkene. This product is oxidized to carbocation **133** with equivalent of $\text{Mn}(\text{OAc})_3$. The intramolecular cyclization of **133** can produce linear product **129** (Scheme 34).⁶³

In 2010, the Ferreira group described synthesis and antifungal activity of substituted α - and β -dihydrofuran naphthoquinones **134a–b**. The furan naphthoquinones were obtained in 10–86% yields by oxidative [3 + 2] cycloaddition of lawsone, to the alkene, mediated by cerium(IV) ammonium nitrate (CAN) in THF at room temperature for 3 h (Scheme 35). It is noteworthy that all the products were formed regioselectively,

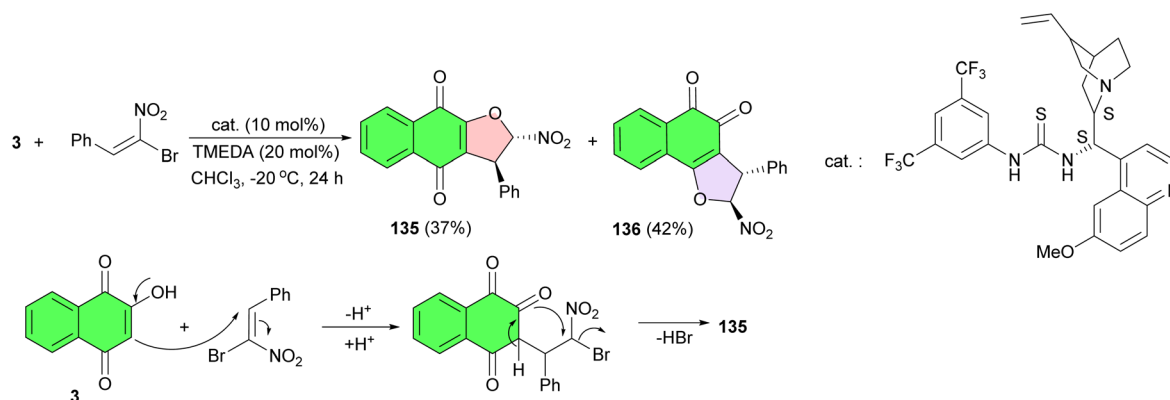
Scheme 33 Synthetic route for preparing the naphthoquinones **122–128** assayed against trypomastigote forms of *T. cruzi*.

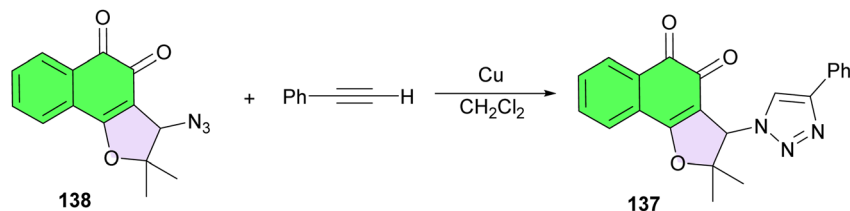
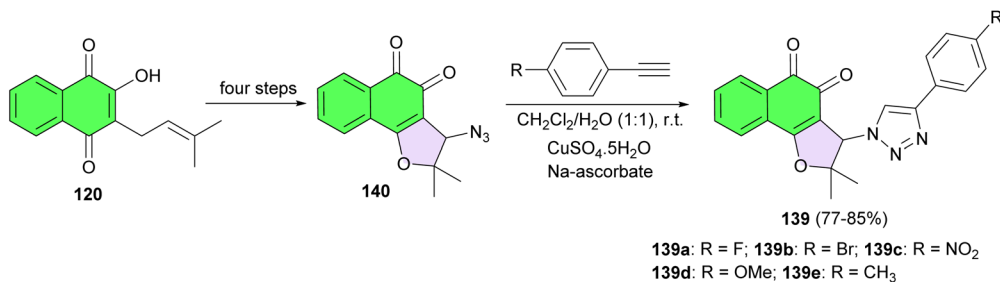
Scheme 34 Mn(OAc)₃ catalyzed synthesis of 2,3-dihydronaphtho[2,3-*b*]furan-4,9-diones **129**.Scheme 35 CAN catalyzed synthetic route used for the preparation of α - and β -furan naphthoquinones **134a–b**.

with respect to the double bond of the alkene. The compounds evaluated against the following six strains of *Candida* (*C*): *C. albicans*, *C. krusei*, *C. parapsilosis*, *C. kefyr*, *C. tropicalis* and *C. dubliniensis*. Some of the α -furan naphthoquinones exhibited potent antifungal activity, with no hemolytic activity or cytotoxic effects.⁶⁴

The Rueping group developed synthesis of dihydrofuranonaphthoquinones **135** and **136** as the two regioisomers in **37**

and 42% yields and with excellent enantioselectivities of 90 and 92% ee, respectively. The reaction was carried out between lawsone and β,β -bromonitrostyrene in the presence of a chiral bifunctional thiourea catalyst in CHCl₃ at –20 °C for 24 h. The proposed reaction sequence would involve the enantiocontrolled Michael addition of lawsone to the (*E*)- β,β -bromonitrostyrene, followed by the diastereoselective cyclization

Scheme 36 Enantioselective synthesis of dihydrofuranonaphthoquinones **135–136**.

Scheme 37 Synthesis of triazolic naphthofuroquinone **137**.Scheme 38 Synthesis of nor- β -lapachone-based 1,2,3-triazoles **139**.

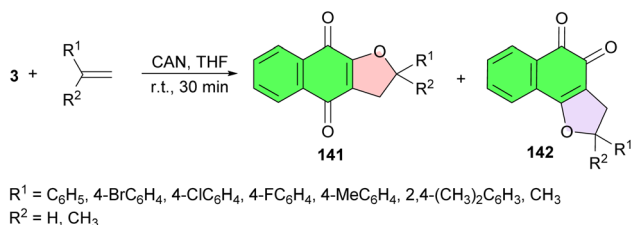
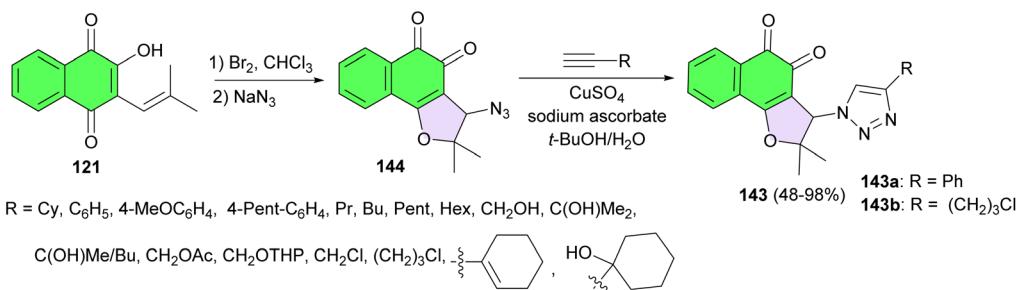
requiring nucleophilic substitution of the bromide to yield the desired polysubstituted dihydrofurans (Scheme 36).⁶⁵

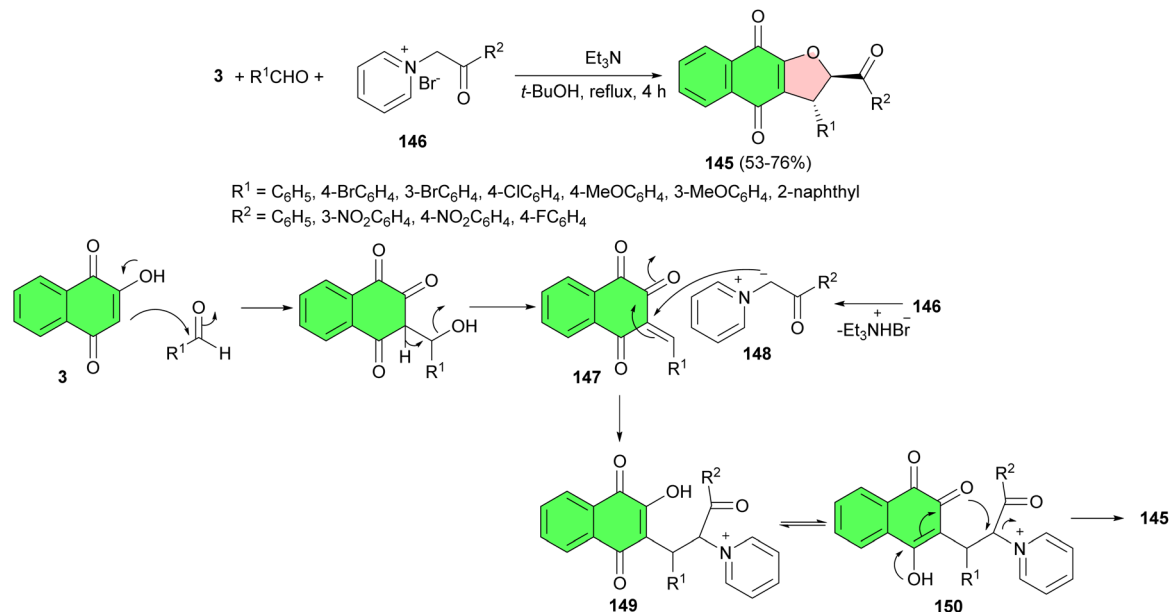
In 2012, Menna-Barreto reported the mechanism of action of the triazolic naphthoquinone (TN; 2,2-dimethyl-3-(4-phenyl-1*H*-1,2,3-triazol-1-yl)-2,3-dihydronaphtho[1,2-*b*]furan-4,5-dione) (**137**), which is the most active compound against *T. cruzi* trypanosomes among a series of naphthofuranquinones. It was synthesized by the reaction between 3-azido-nor- β -lapachone **138** and ethynylbenzene catalysed by Cu. The key intermediate, azidoquinone, was generated by nucleophilic substitution from

3-bromo- β -norlapachone with sodium azide in dichloromethane (Scheme 37).⁶⁶

Next, synthesis of nor- β -lapachone-based 1,2,3-triazoles **139** reported in 77–85% yields by the reaction of substituted alkyne with 3-azido-2,2-dimethyl-2,3-dihydronaphtho[1,2-*b*]furan-4,5-dione **140** in the presence of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ and sodium ascorbate in $\text{CH}_2\text{Cl}_2/\text{H}_2\text{O}$ at room temperature. Compound **140** was previously synthesized from lapachol in a four-step process.⁶⁷ These compounds were evaluated against the infective bloodstream form *Trypanosoma cruzi*, the etiological agent of Chagas disease. All of the compounds were considered potent trypanocidal compounds, with the exception of **139d** ($\text{IC}_{50}/24 \text{ h } \frac{1}{4}$ 359.2 11.1 mM). The triazoles **139a**, **139c** and **139e** were 2.5–5 times more active than the standard drug and represent potential candidates for further testing in preclinical assays of drugs for the treatment of Chagas disease. The insertion of electron withdrawing groups possibly amplifies the redox potential of the naphthoquinoidal structure, leading to an important increase in its biological activity (Scheme 38).⁶⁸

Next, α - and β -dihydrofuran naphthoquinones **141–142**, respectively, synthesized in good yields from readily available

Scheme 39 Synthetic route used for the preparation of α - and β -dihydrofuran naphthoquinones **141–142**.Scheme 40 Preparation of 1,2-furan naphthoquinone triazoles **143**.

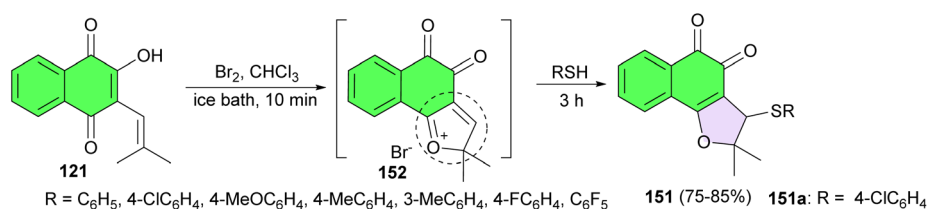
Scheme 41 Stereoselective synthesis of dihydrofuranonaphthoquinones **145**.

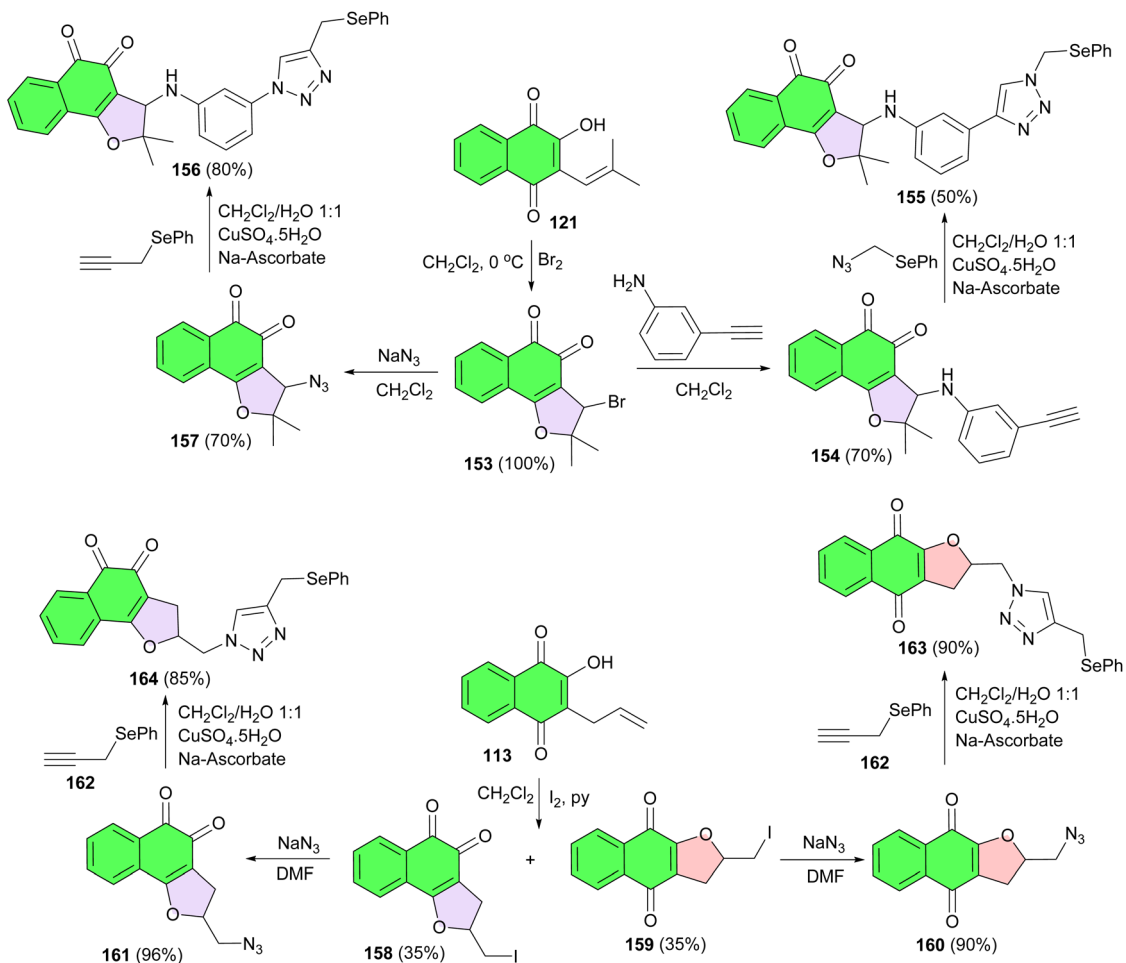
lawsone and olefins in the presence of cerium(IV) ammonium nitrate in THF at room temperature for 30 min. The reaction led to two products: the α - and β -dihydrofuran naphthoquinones which were separated by column chromatography on silica gel (Scheme 39). The antitumor activity of the compounds against 4 human tumor cell lines, HL-60 (leukemia), SF-295 (CNS), HCT-8 (colon) and MDA-MB435 (melanoma), and their electrochemical parameters, in the absence and presence of oxygen were investigated in comparison with their non-substituted precursors. The β -dihydrofuran naphthoquinones were shown to be highly cytotoxic, while their α -isomers were shown to be less active.⁶⁹

In 2014, the da Silva group reported synthesis of 1,2-furanonaphthoquinones tethered to 1,2,3-*H*-triazoles (1,2-FNQT) **143** and investigated their antileukemic activity. At first, nor- β -lapachol was reacted with excess of bromine in chloroform, followed by nucleophilic substitution with sodium azide in dichloromethane to yield **144** (3-azido-2,2-dimethyl-2,3-dihydronaphtho[1,2-*b*]furan-4,5-dione) in 90% yield. This key intermediate was employed in Huisgen 1,3-dipolar cyclization with an appropriate terminal alkyne catalyzed by copper(I) ion, also known as a click reaction, to obtain the desired product **143** in 48–98% yields. Compound **143a** has great potential for further development as an anti-leukemia drug not only because

of its potent and selective cytotoxicity (normal \times cancer cells) but also because of its selectivity against leukemia lymphoid cell lines (approximately 7–19 times more effective than in leukemia myeloid cells). Additionally compound **143b** is also promising due to its high cytotoxic activity against some leukemia cells (IC_{50} ranging from 0.48 to 1.38 μM) and its lower toxicity against normal hematopoietic cells (estimated $\text{IC}_{50} > 10 \mu\text{M}$) (Scheme 40).⁷⁰

After that, Nguyen and co-workers explored an efficient stereoselective synthesis of dihydrofuranonaphthoquinones **145** in 53–76% yields by means of one-pot multicomponent reactions using lawsone, an aromatic aldehyde and pyridinium bromide **146** using Et_3N in *t*-BuOH under reflux conditions for 4 h. A possible mechanistic explanation for this multicomponent reaction starts with a Knoevenagel condensation of lawsone with aromatic aldehydes, followed by dehydration resulting in the formation of 1,2,3,4-tetrahydro-1,2,4-naphthalenetriones **147**. The next step is a Michael addition of pyridinium ylides **148**, formed *in situ* by deprotonation of pyridinium bromides **146** by triethylamine, across Michael acceptors **147**. The obtained naphthoquinones **149/150** undergo a cyclization to produce the desired substituted dihydrofuranonaphthoquinones **145** (Scheme 41).⁷¹

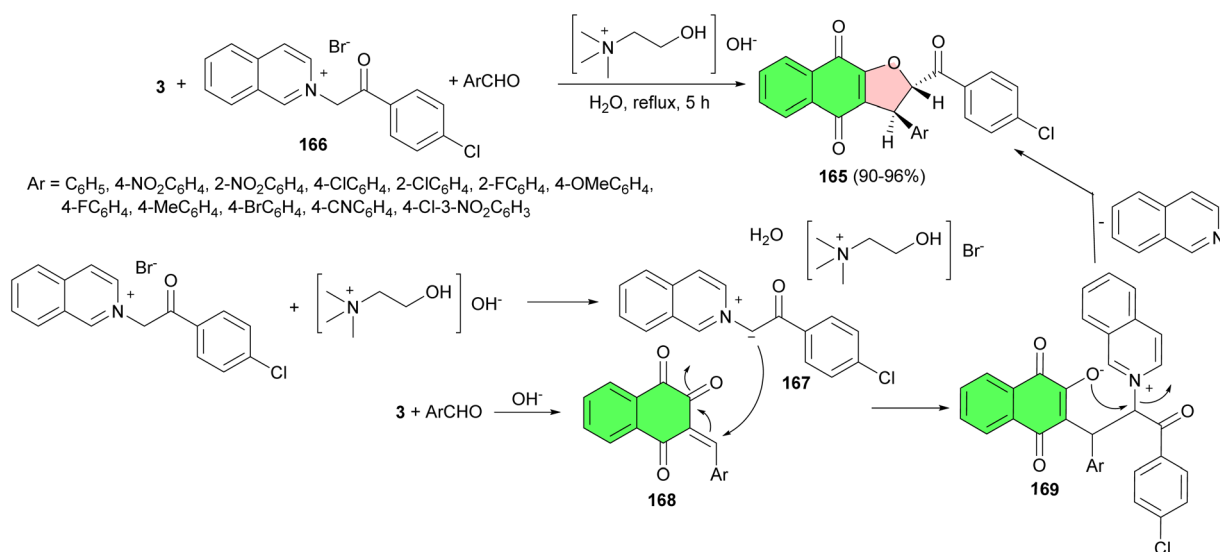
Scheme 42 Synthesis of 3-phenylthio groups linked to 2,3-dihydronaphtho[1,2-*b*]-furan-4,5-diones **151**.



Scheme 43 Synthesis of selenium-containing furanonaphthoquinones 155–156, 163–164.

Further, Ferreira and co-workers revealed an efficient route to prepare compounds containing 3-phenylthio groups linked to 2,3-dihydronaphtho[1,2-*b*]-furan-4,5-diones 151. This

methodology involved one-step reactions starting with nor-lapachol 121, which upon reaction with bromine in chloroform generates *in situ* cationic *ortho*-quinone methide 152. This

Scheme 44 Choline hydroxide catalyzed synthesis of *trans*-2-(4-chlorobenzoyl)-3-aryl-2,3-dihydronaphtho[2,3-*b*]furan-4,9-diones 165.

intermediate then reacts with phenyl thiols for 3 h to produce the corresponding product in 75–85% yields. The compounds possess a broad range of activity ($IC_{50}/24$ h from 9.2 to 182.7 μ M), higher than the original quinone (391.5 μ M) and four of them higher than standard drug benznidazole (103.6 μ M). The most active was compound **151a** (9.2 μ M), being 11 times active than benznidazole and the less toxic derivative to heart muscle cells (Scheme 42).⁷²

Selenium-containing quinone-based 1,2,3-triazoles synthesized using click chemistry, the copper catalyzed azide–alkyne 1,3-dipolar cycloaddition, and evaluated against six types of cancer cell lines: HL-60, HCT-116, PC3, SF295, MDA-MB-435 and OVCAR-8. These compounds could provide promising new lead derivatives for more potent anticancer drug development and delivery. For the preparation of these compounds, bromo intermediate **153** was synthesized from nor-lapachol in CH_2Cl_2 at 0 °C. Subsequently, arylamino-substituted lapachone **154**, featuring a terminal alkyne group, was prepared from **153** in 70% yield. The selenium-containing 1,2,3-triazole **155** was then obtained from **154** in 50% yield in the presence of $CuSO_4$. Following the same strategy as outlined above, compound **156** was synthesized from azide derivative **157** in 80% yield. Compound **157** itself was prepared from **153** using sodium azide. The iodination of **113** affords compounds **158** and **159** in 68% yield and 1 : 1 ratio, which were easily separated by column chromatography. With these compounds in hand, the respective azide derivatives, compounds **160** and **161**, were synthesized in 90 and 96% yields by the reaction of sodium azide in dimethylformamide, respectively. The respective selenium derivatives **162**, compounds **163** and **164**, were prepared in 90 and 85% yields by Cu-catalyzed azide–alkyne cycloaddition, respectively (Scheme 43).⁷³

Next, a green and highly efficient synthesis of *trans*-2-(4-chlorobenzoyl)-3-aryl-2,3-dihydronaphtho[2,3-*b*]furan-4,9-

diones **165** achieved in 90–96% yields *via* a three-component, one-pot condensation of 2-[2-(4-chlorophenyl)-2-oxoethyl] isoquinolinium bromide **166** with lawsone and an aromatic aldehyde in the presence of catalytic amounts of choline hydroxide in water under reflux conditions for 5 h. A proposed mechanism is shown in Scheme 44. Compound **166** undergoes deprotonation in the presence of aqueous choline hydroxide to give the reactive isoquinolinium ylide **167**. Lawsone reacts with aromatic aldehyde in the presence of choline hydroxide to give the Knoevenagel product **168**. This reacts instantly with the isoquinolinium ylide **167** to form the zwitterionic intermediate **169**, which undergoes cyclisation with the elimination of isoquinoline to give the desired product **165**.⁷⁴

Next, a series of 1,2-naphthoquinones tethered in C2 to 1,2,3-1*H*-triazoles **170** designed, synthesized and their cytotoxic activity evaluated using HCT-116 (colon adenocarcinoma), MCF-7 (breast adenocarcinoma) and RPE (human nontumor cell line from retinal epithelium). The chemical synthesis was performed from C-3 allylation of lawsone followed by iodocyclization with subsequent nucleophilic displacement with sodium azide and, finally, the 1,3-dipolar cycloaddition catalyzed by Cu(i) with terminal alkynes led to the formation of the desired product in 35–70% yields (Scheme 45). Compounds containing aromatic group linked to 1,2,3-triazole ring

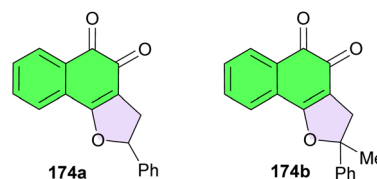
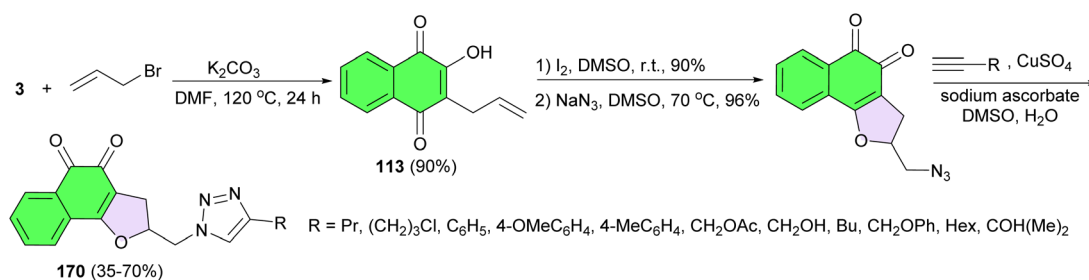
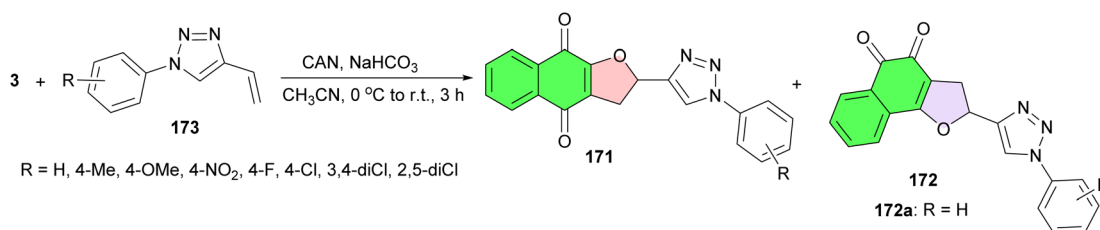


Fig. 2 Structures of 2,3-dihydrofuranaphthoquinones **174a–b** as antifungal.

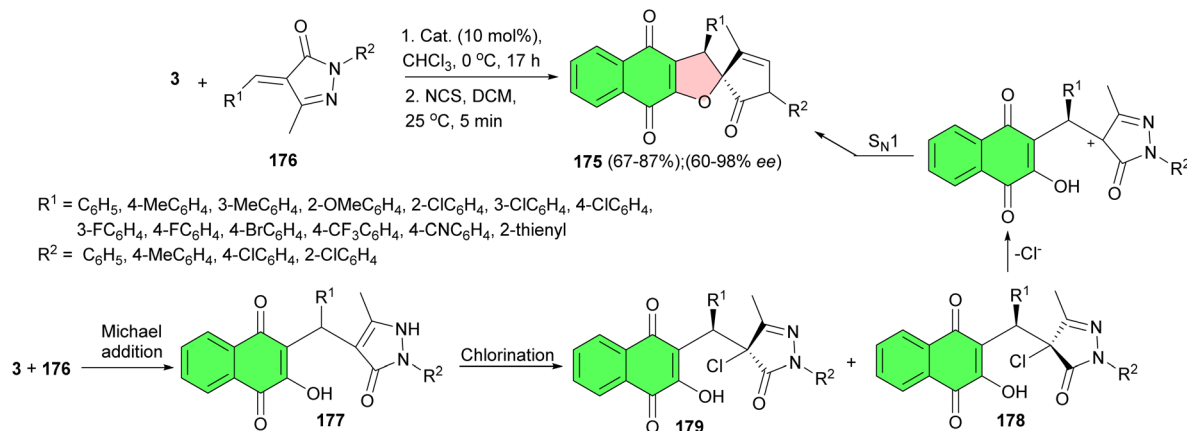


Scheme 45 Synthesis of 1*H*-1,2,3-triazoles-linked to 2,3-dihydronaphtho[1,2-*b*]furan-4,5-dione **170**.



Scheme 46 CAN mediated synthesis of furanaphthoquinones tethered to 1*H*-1,2,3-triazoles **171–172**.



Scheme 47 Asymmetric synthesis of spiro-pyrazolone-fused dihydrofuran-naphthoquinones **175**.

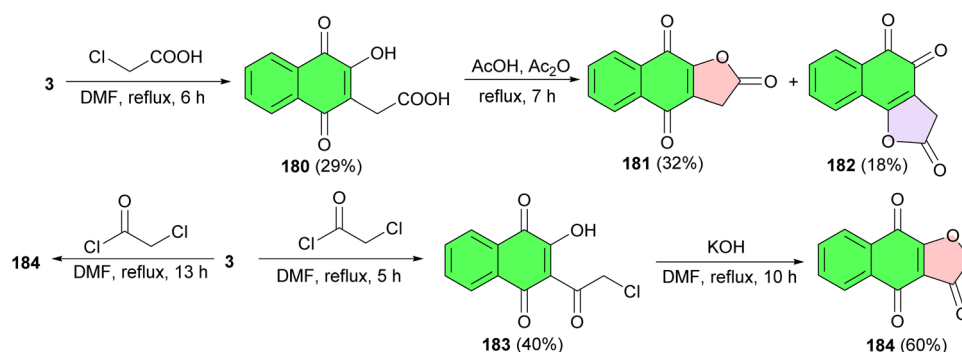
presented superior cytotoxic activity against cancer cell lines with IC_{50} in the range of 0.74 to 4.4 μM indicating that the presence of aromatic rings substituents in the 1,2,3-*1H*-triazole moiety is probably responsible for the improved cytotoxic activity.⁷⁵

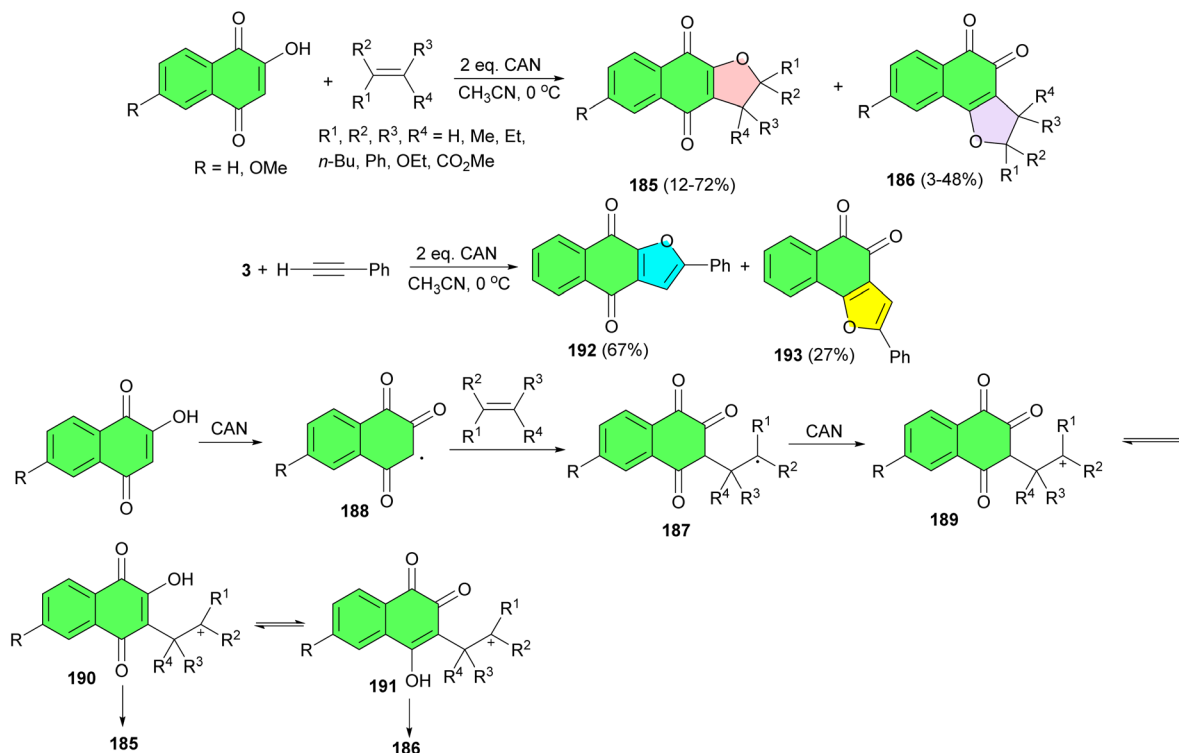
After that, Ferreira *et al.* revealed synthesis and evaluation of the cytotoxic activity of furanaphthoquinones tethered to 1*H*-1,2,3-triazoles **171**–**172** against human tumor cell lines (MDA-MB231, Calu-3 and Caco-2) and healthy cells (Vero). The reaction of lawsone with 4-vinyl-1*H*-1,2,3-triazoles **173** in the presence of CAN in acetonitrile at room temperature for 3 h. The isomers were isolated by column chromatography. The results showed that compound **172a** exhibited the most promising profile due to its selective cytotoxic action against colon adenocarcinoma cells (Scheme 46).⁷⁶

Next, Ferreira *et al.* revealed antifungal activities of substituted α - and β -2,3-dihydrofuran-naphthoquinones against *Sporothrix brasiliensis* and *Sporothrix schenckii*—the main etiological agents of sporotrichosis in Brazil. The results showed that compounds **174a** and **174b** were the most active dihydrofuran-naphthoquinones *in vitro* for both species; in fungi, these compounds induced yeast-hyphae conversion and alteration in the hyphae and conidia structures. Compound **174b** also exhibited a synergistic activity with itraconazole against *S. schenckii*, with a SFIC index value of 0.3 (Fig. 2).⁷⁷

Next, the Song group revealed enantioselective synthesis of spiro-pyrazolone-fused dihydrofuran-naphthoquinones **175** via a Michael addition/chlorination/nucleophilic substitution sequence. The reactions of lawsone and α,β -unsaturated pyrazolones **176** in the presence of the cinchona alkaloid derived hydrogen-bonding catalyst and NCS provide **175** bearing contiguous stereocenters in good to excellent yields (67–87%) and moderate to excellent enantioselectivities (60–98% ee). A plausible reaction mechanism is proposed in Scheme 47. Lawsone is deprotonated by the basic nitrogen atom of a tertiary amine to form a nucleophilic species, and the unsaturated pyrazolone **176** is synergistically activated by the squaramide moiety via two hydrogen bonds. Then Re-face attack occurs favorably in the Michael addition to give the enantioselective adduct **177**. Computational study on the Hartree-Fock level revealed that chlorination of the Michael adduct **177** preferentially produces the thermodynamically stable product intermediate **178** rather than **179**. The intermediate **178** then undergoes intramolecular nucleophilic substitution ($\text{S}_{\text{N}}1$) to finally provide thermodynamically favored isomer **175** with exclusive diastereoselectivity and good enantioselectivity.⁷⁸

Further, Elbana and co-workers reported synthesis of 2-(3-hydroxy-1,4-dioxo-1,4-dihydro-naphthalen-2-yl)acetic acid (**180**) by the reaction of lawsone with chloroacetic acid in DMF under reflux conditions for 6 h. Then, the treatment of **180** with acetic

Scheme 48 Synthesis of naphthofuranquinone derivatives **181**–**182**, **184**.



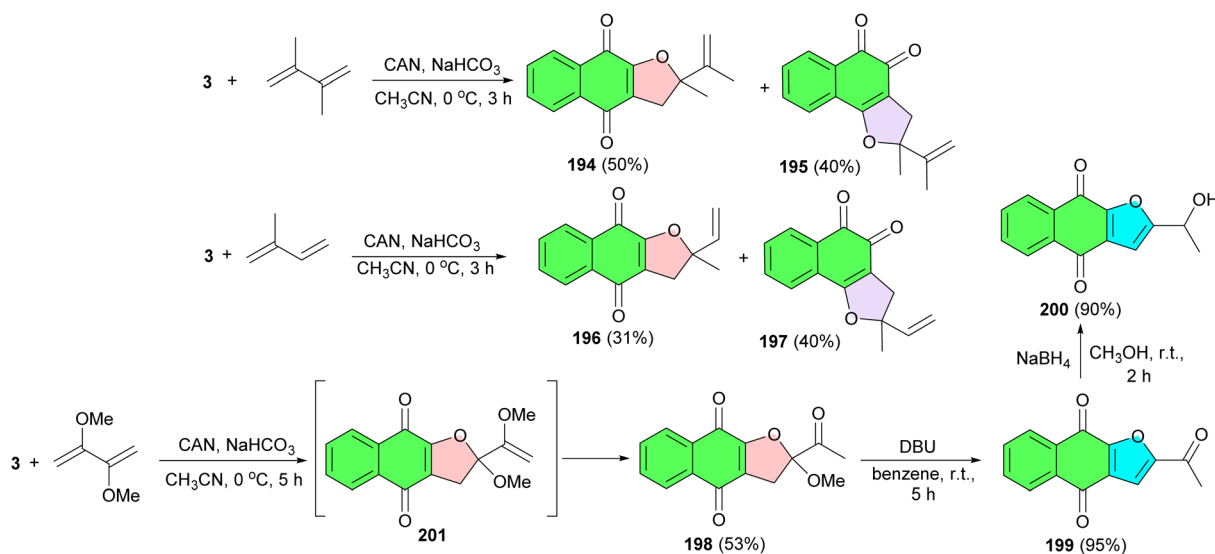
Scheme 49 CAN mediated synthesis of furanonaphthoquinones 185–186, 192–193.

acid and acetic anhydride under reflux for 7 h gave naphthofuran-2,4,9(3*H*)-triones **181** and **182**. Moreover, the reaction of lawsone with chloroacetyl chloride in refluxing DMF for 5 h afforded 2-(2-chloroacetyl)-3-hydroxynaphthalene-1,4-dione (**183**). Also, naphtho[2,3-*b*]furan-3,4,9(2*H*)-trione (**184**) synthesized *via* two methods: refluxing a mixture of lawsone and chloroacetyl chloride in DMF for 13 h in the presence of the catalytic amount of potassium hydroxide and refluxing a mixture of **183** in DMF in the presence of a catalytic amount of potassium hydroxide for 10 h (Scheme 48). The target molecules

were showed easy access to antioxidant and antitumor activities. Geometrical isomers (enol, Keto conformers, and *syn*, anti-conformers) were achieved by DFT that conformed to the spectral analysis of the investigated compounds.⁷⁹

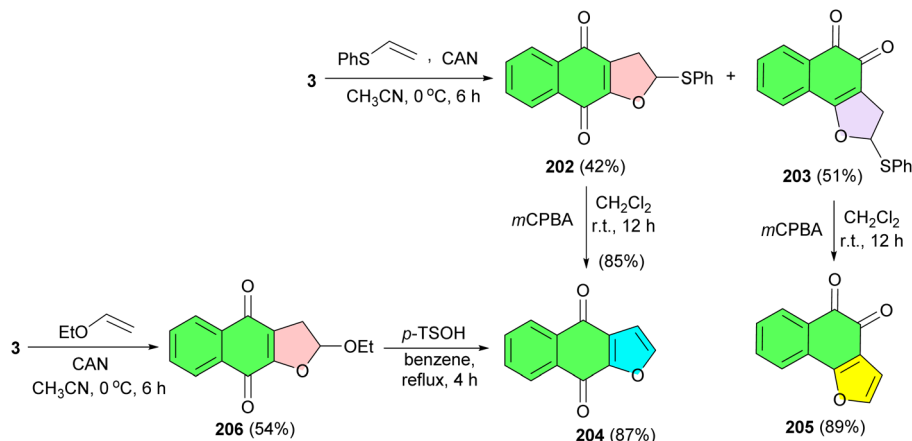
4. Synthesis of naphthofuroquinones together with their dihydro derivatives

In 1996, the Kobayashi group reported a one-pot formation of 2,3-dihydronaphtho[2,3-*b*]furan-4,9-diones **185** in 12–72%



Scheme 50 CAN-mediated synthesis of furonaphthoquinones 194–201.





Scheme 51 CAN-mediated synthesis of furonaphthoquinones 202–205.

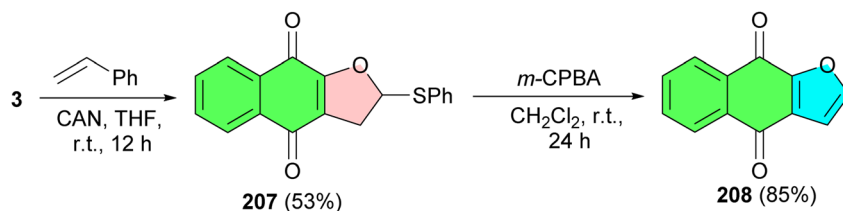
yields and 2,3-dihydronaphtho[1,2-*b*]furan-4,5-diones **186** in 3–48% yields by the ceric ammonium nitrate (CAN) mediated [3 + 2] type cycloaddition of 2-hydroxy-1,4-naphthoquinones with alkenes or phenylacetylene in CH_3CN at 0 °C. The proposed mechanism is outlined in Scheme 49. The formation of each product was completely regioselective. The initial formation of a reactive radical intermediate **187** (through **188**) and its oxidation gives a carbonium ion intermediate **189**, which is intramolecularly trapped with the hydroxyl group of 2-hydroxy-**190** or 4-hydroxy-tautomer **191** to give **185** or **186**, respectively. Analogous cycloaddition can be achieved with phenylacetylene instead of alkenes. Formation of the products **192** and **193** was almost quantitative.⁸⁰

In 2000, Lee and his group reported synthesis of dihydrofuronaphthoquinones **194** (50% yield) and **195** (40% yield) as a mixture of linear and angular regioisomers by the reaction of lawsone and conjugated dienes using ceric ammonium nitrate (CAN) in CH_3CN at 0 °C for 3 h. The mixture was easily separated by column chromatography and the two isomers were assigned by their spectroscopic data. Similarly, with isoprene, two regioisomers, **196** and **197**, were also obtained in 31 and 40% yields, respectively. Treatment of **3** with 2,3-dimethoxy-1,3-butadiene in the presence of 3.0 equiv. of CAN in acetonitrile at 0 °C for 5 h afforded the dihydrofuronaphthoquinone **198** in 53% yield. Reaction of **198** with DBU in benzene at room temperature for 5 h results in furonaphthoquinone **199** in 95% yield. Next, the synthesis of furonaphthoquinone **200** was easily achieved by reduction of **199** with sodium borohydride in methanol in 90% yield. The product **198** probably results from

CAN-mediated oxidative cycloaddition and followed by the methyl group cleavage of intermediate **201** (Scheme 50).⁸¹ Compounds **199** and **200** are isolated from the *Tabebuia cassioides* and they are reported to have significant biological properties such as antileukemic activity and *in vitro* cytotoxicity against KB, K562, and P388 cells.⁸² These furonaphthoquinone derivatives have been also used in traditional medicine as Pau d'Arco, IpeRoxo, Lapacho, and Tahebo for many years in North and South America as anticancer, antifungal, antibacterial, and anti-inflammatory drugs.⁸³

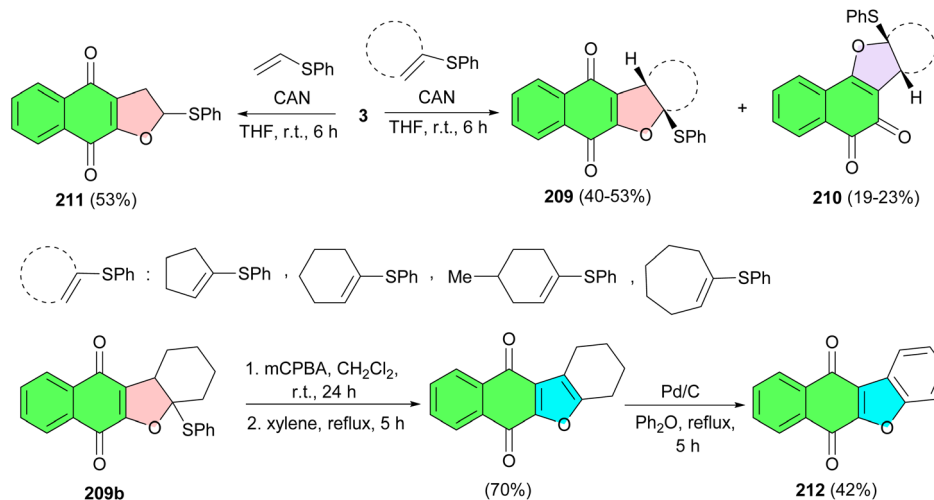
CAN-mediated cycloaddition reaction of lawsone (**3**) with phenyl vinyl sulfide at 0 °C for 6 h in acetonitrile afforded the dihydrofuronaphthoquinone **202** (42% yield) and **203** (51% yield) as a mixture of linear and angular isomers. The mixture was easily separated by column chromatography and the two isomers were assigned by their spectroscopic data. The conversion of compound **202** to the natural product was begun by elimination of the phenyl sulfide group. Treatment of **202** and **203** with *m*-CPBA in CH_2Cl_2 at room temperature for 12 h resulted in avicequinone-B (**204**) (85% yield) and furonaphthoquinone **205** (82% yield), respectively. Treatment of **3** with ethyl vinyl ether in the presence of CAN at 0 °C for 6 h in acetonitrile gave dihydrofuronaphthoquinone **206** (54% yield) as a single compound, without the isolation required of the expected angular regioisomer. The dihydrofuronaphthoquinone **206** was easily converted to **204** in 87% yield by treatment of *p*-TsOH in refluxing benzene (Scheme 51).⁸⁴

Next, Lee and co-workers described synthesis of dihydrofuronaphthoquinone **207** in 53% yield from lawsone and



Scheme 52 Ceric ammonium nitrate(CAN)-mediated synthesis of furonaphthoquinone 207–208.



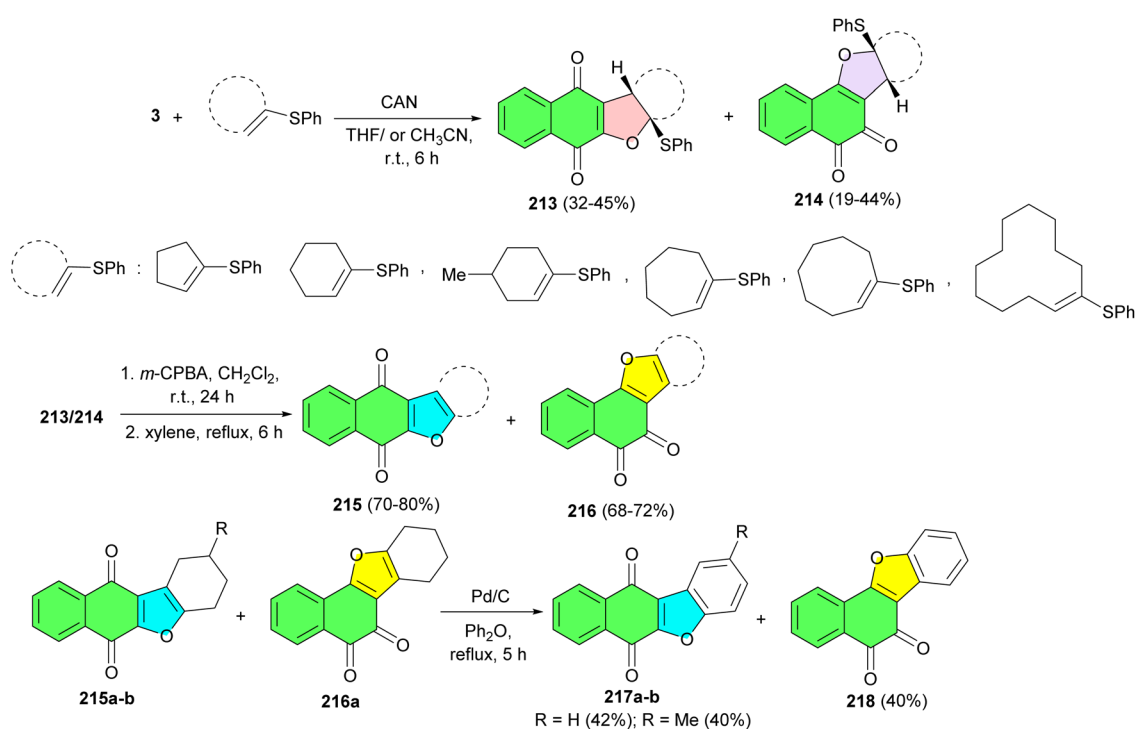


Scheme 53 CAN-Mediated synthesis of furanonaphthoquinones 209–212.

phenyl vinyl sulfide in the presence of CAN at room temperature for 12 h in THF, without formation of any other possible regioisomers. Treatment of **207** with *m*-CPBA in CH_2Cl_2 at room temperature for 24 h afforded avicequinone-B **208** in 85% yield (Scheme 52).⁸⁵ It has been shown to have a great cancer chemopreventive activity against Epstein–Barr virus early antigen (EBV-EA) activation, without showing any cytotoxicity.⁸⁶

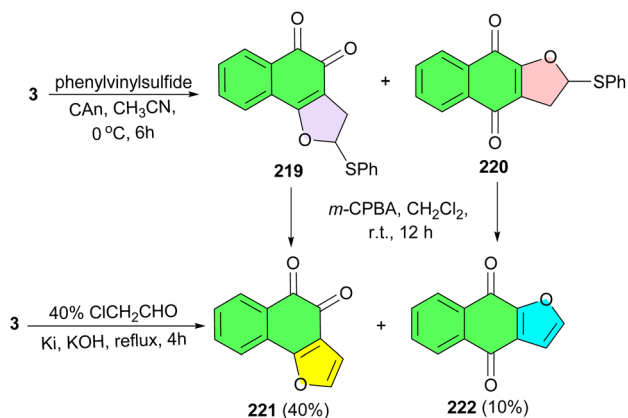
The reaction of 2-hydroxy-1,4-naphthoquinone (**3**) with vinyl sulfides in the presence of ceric(IV) ammonium nitrate in THF at room temperature for 6 h resulted in dihydrofuranonaphthoquinone **209** (40–53%) and **210** (19–23%) as a mixture of

linear and angular regioisomers. The products were easily purified by column chromatography and the structures of the two isomers were determined by their spectroscopic data. However, reaction with vinyl sulfide at room temperature in THF afforded solely dihydrofuranonaphthoquinone **211** in 53% yield, without formation of the other possible regioisomer. Dihydrofuran **209b** can be readily converted to benzofuranonaphthoquinone derivative **212** which has been reported to have significant biological activities such as antipruritic, anti-tumor, topo II-mediate DNA cleavage (Scheme 53).⁸⁷



Scheme 54 CAN mediated synthesis of furanonaphthoquinones 213–218.





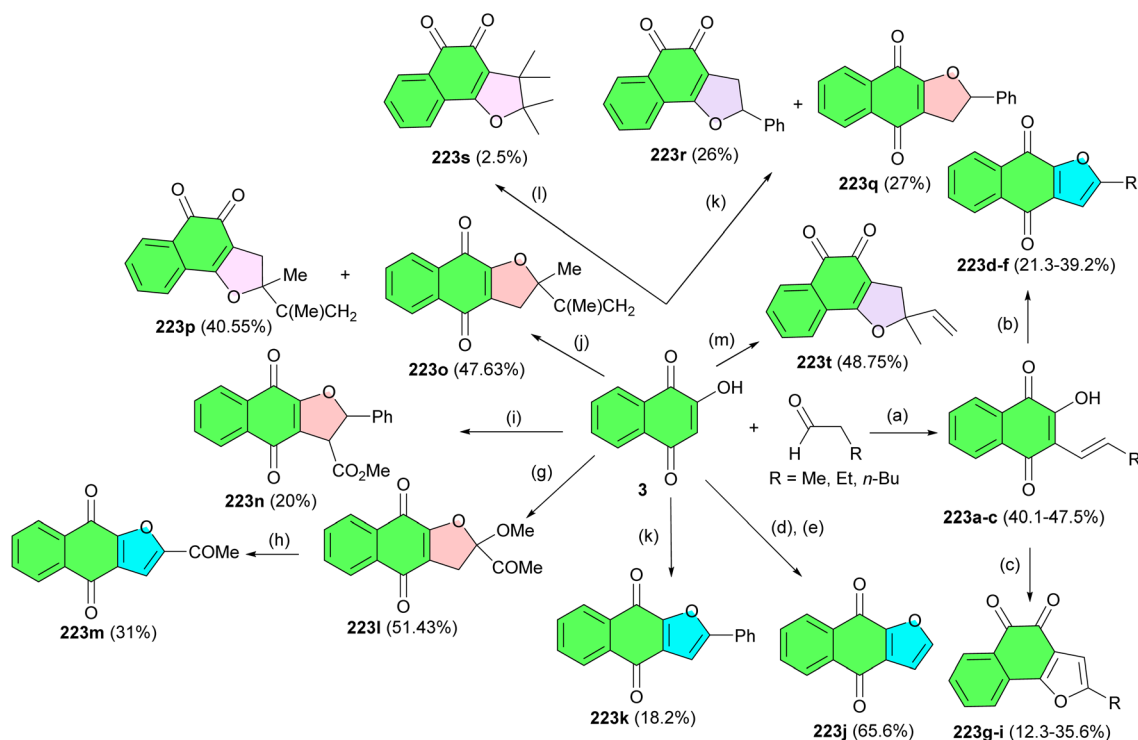
Scheme 55 Preparation of furonaphthoquinones 219–222.

An efficient synthesis of linear and angular dihydrofuranonaphthoquinones **213** and **214** in 19–45% yields, has been carried out starting from 2-hydroxy-1,4-naphthoquinone and a variety of vinyl sulfides using CAN in THF or CH₃CN at room temperature for 6 h. Also, furanonaphthoquinones **215** and **216** were synthesized with yields of 68–80% by reacting dihydrofuranonaphthoquinones with *m*-CPBA in CH₂Cl₂ at room temperature for 24 hours, followed by refluxing in *m*-xylene for 6 h. Moreover, the reaction of furanonaphthoquinones with Pd/

C in diphenyl ether under reflux conditions for 5 h afforded benzonaphthoquinones **217a–b** and **218** in 40–42% yields (Scheme 54).⁸⁸

In 2009, Cheng and co-workers disclosed synthesis of angular dihydrofuranaphthoquinone **219** and its linear isomer **220** by the reaction of lawsonone with phenylvinylsulfide in the presence of ceric ammonium nitrate (CAN) in CH₃CN at 0 °C for 6 h. Oxidative elimination of **219** and **220** with *m*-CPBA in CH₂Cl₂ at room temperature for 12 h afforded angular naphtho[1,2-*b*]furan-4,5-dione (**221**) and linear naphtho[2,3-*b*]furan-4,9-dione (**222**). Also, lawsonone was reacted with chloroacetaldehyde under reflux conditions for 4 h to give a mixture of **221** and **222** in a yield of 40% and 10%, respectively (Scheme 55).⁸⁹

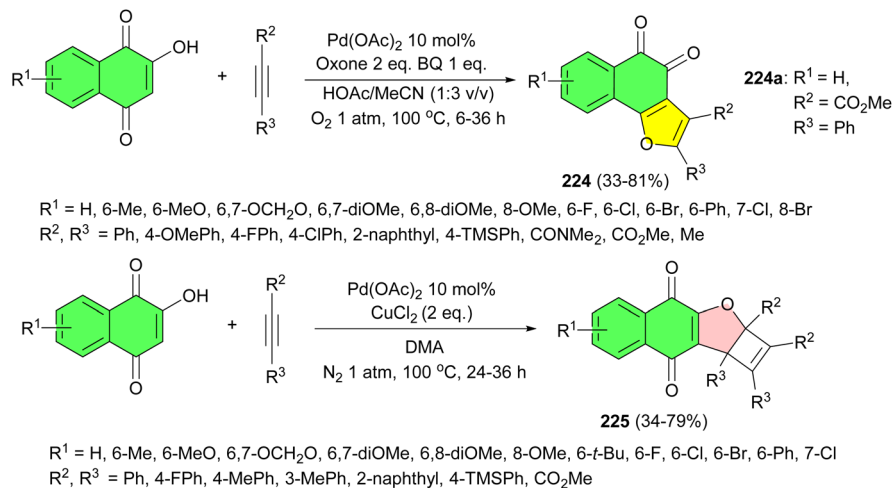
In 2018, the Franco group reported synthesis of a series of furanonaphthoquinone derivatives **223** and evaluated their anti-proliferative activity against the human cell line of colorectal cancer HT-29. Several of these compounds exhibited significant anti-proliferative activity. Condensation of lawsonone with aldehydes in acidic condition was carried out and afforded compounds **223a**, **223b**, and **223c**. These compounds were oxidized and cyclized by mercuric acetate in harsh reaction conditions, concentrated hydrochloric acid, 65 °C and 2 h of reaction time, to afford **223d**, **223e**, and **223f**, respectively. Alternatively, milder reaction conditions, *i.e.* diluted hydrochloric acid, 65 °C and 15 min of reaction time, were used to obtain **223g**, **223h**, and **223i**. Cycloaddition reactions of lawsonone



Reagents and conditions: (a) AcOH, HCl, 60 °C; (b) Hg(AcO)₂, AcOH, 65 °C; EtOH, HCl conc., 65 °C, 2 h; (c) Hg(AcO)₂, AcOH, 65 °C; EtOH, HCl dil., 65 °C, 15 min; (d) vinylacetate, acetone, uv, 7 h; (e) *p*-TsOH, dry benzene, reflux; (f) phenylacetylene, CH₃CN, CAN, 0 °C, 20 min; (g) 2,3-dimethoxy-1,3-butadiene, CH₃CN, CAN, 0 °C, 6 h; (h) benzene, DBU, r.t., 5 h; (i) acetone, methyl (2E) 2-phenylpropionate, uv, 22 h; (j) 2,3-dimethoxy-1,3-butadiene, CH₃CN, CAN, Na₂CO₃, 0 °C, 3 h; (k) styrene, CH₃CN, CAN, 0 °C, 20 min; (l) 2,3-dimethyl-2-butene, CH₃CN, CAN, 0 °C, 20 min; and (m) isoprene, CH₃CN, CAN, 0 °C, 3 h.

Scheme 56 Synthesis of a series of furanonaphthoquinone derivatives 223.





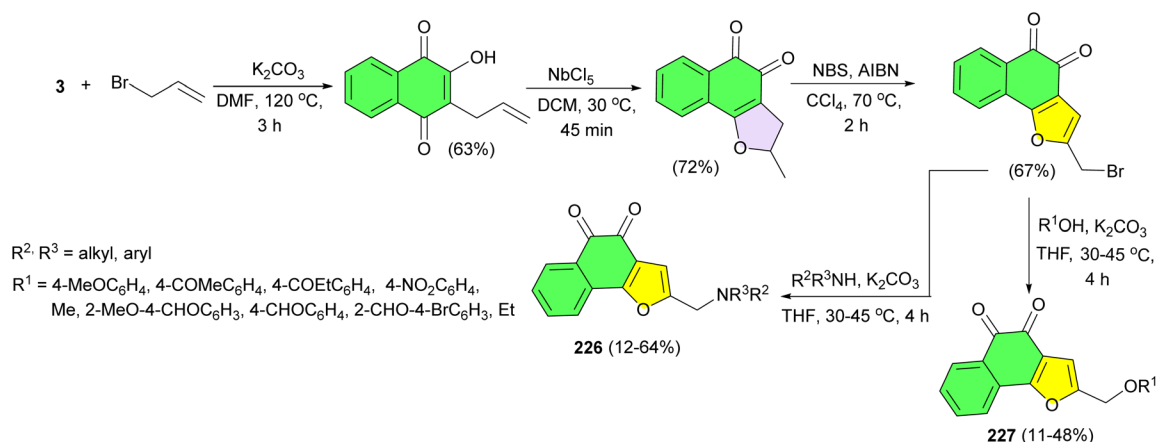
Scheme 57 Palladium-catalyzed synthesis of naphthofuroquinones 224–225.

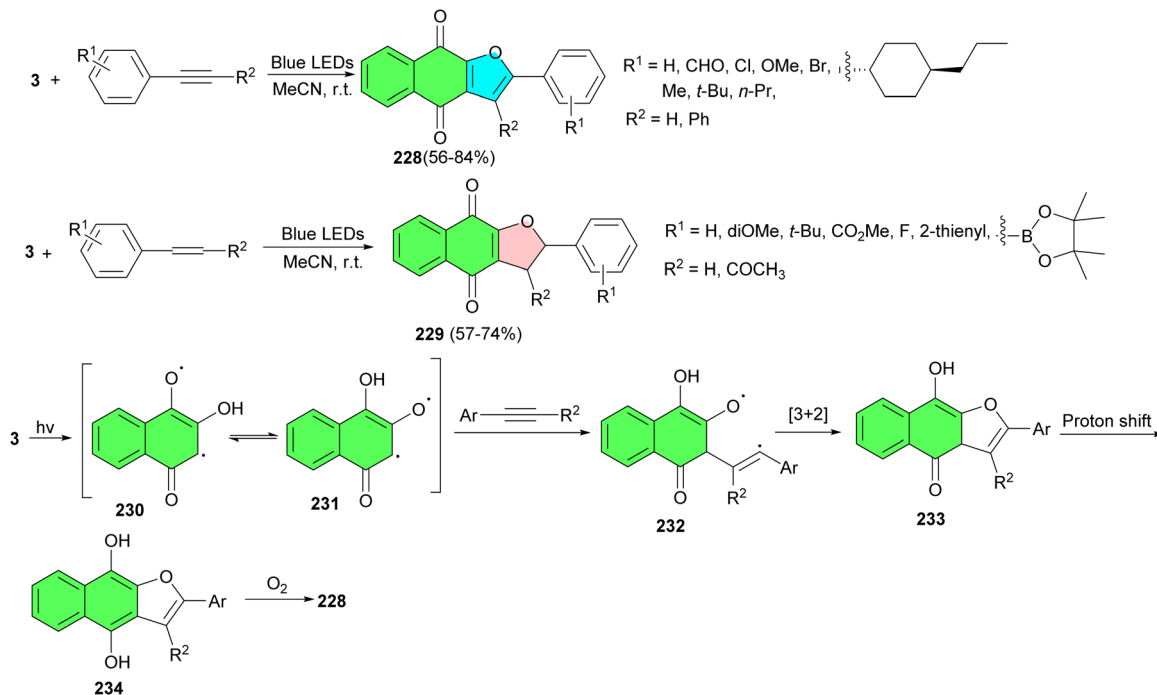
with different reagents were carried out to obtain the compounds **223j**–**223t**. The compound **223m** was obtained by the hydrolysis of **223i** (Scheme 56).⁹⁰

The Wang group developed palladium-catalyzed oxidative switchable annulation of naphthalquinones, bearing electron-neutral, electron-donating, and electron-withdrawing substituents on the aromatic ring, with symmetrical and unsymmetrical alkynes substituted with electron-rich and electron-deficient groups in HOAc/MeCN at 100 °C to assemble a series of biologically relevant functionalized 1,2-naphthofuroquinones **224** in 33–81% yields after 6–36 h and densely functionalized cyclobutene embedded 1,4-naphthofuroquinones **225** in the presence of CuCl₂ in DMA at 100 °C for 24–36 h in 34–79% yields. Also, the bioactivity of the synthesized compounds was evaluated. The results showed that **224a** exhibited a strong endothelial protective effect against oxidized low-density lipoprotein (ox-LDL)-induced human umbilical vein endothelial cell (HUVEC) injury. Additionally, it has no effects on the normal cells (Scheme 57).⁹¹

After that, the Chen group designed and synthesized L-shaped *ortho*-quinone analogs **226**–**227** using a one pot double-radical synthetic strategy followed by removing methyl at C-3 of the furan ring and introducing a diverse side chain at C-2 of the furan ring. Scheme 58 refers to a step-by-step diagram outlining the synthetic pathway used to produce the compounds. Moreover, the synthesized compounds exhibit cytotoxic activity against human leukemia cells K562, prostate cancer cells PC3, and melanoma cells WM9.⁹²

In 2023, Tan and co-workers developed visible-light-mediated [3 + 2] cycloaddition reactions of 2-hydroxy-1,4-naphthoquinones and alkynes and alkenes under irradiation of blue LEDs (460 nm) in the absence of any bases, metals, ligands, or other catalysts in CH₃CN at room temperature. Under environmentally friendly conditions, a variety of naphtho [2,3-*b*]furan-4,9-diones **228** and dihydronaphtho[2,3-*b*]furan-4,9-diones **229** were delivered within 6 h in 56–84% yields. A plausible mechanism is demonstrated in Scheme 59. First, the irradiation of lawsone in MeCN generates tautomeric excited triplets **230** and **231**, which react with an alkyne to give a 1,5-

Scheme 58 Synthesized L-shaped *ortho*-quinone analogs 226–227.

Scheme 59 Visible-light-mediated synthesis of naphtho[2,3-*b*]furan-4,9-diones 228–229.

biradical intermediate **232**. Subsequently, an intramolecular [3 + 2] cyclization of the intermediate **232** gives hydroquinone intermediate **233**. Upon 1,3-hydrogen transfer, the hydroquinone intermediate **234** is formed, and then, naphtho[2,3-*b*]furan-4,9-diones **228** is produced by air oxidation of the hydroquinone by oxygen in the air. Similarly, the [3 + 2] cycloaddition reaction of lawsone with alkenes leading to product dihydronaphtho[2,3-*b*]furan-4,9-diones **229** may proceed in a manner parallel to the [3 + 2] cycloaddition of alkynes and may also involve biradical intermediates.⁹³

5. Conclusions

Furonaphthoquinone and their dihydro derivatives are a highly important heterocyclic compounds and a well-known pharmacophoric unit present in natural products, drugs, and drug candidates. A great number of naphthofuroquinones have exhibited diverse biological activities. Over the past decades, numerous furonaphthoquinone compounds have been synthesized and studied by researchers, revealing promising biological activities. This review highlights recent strategies for synthesizing various furonaphthoquinones and their dihydro derivatives, along with the exploration of their biological activities. Primarily starting with 2-hydroxy-1,4-naphthoquinones, a variety of synthetic methods have been developed, including multicomponent reactions, CAN-mediated oxidative cycloaddition, photoaddition, thermal cyclization, coupling with olefins or alkynes, one-step cascade approaches involving lawsone, isocyanides, and aldehydes, the Wittig reaction, [3 + 2] cycloaddition, oxidative cyclization/isomerization, Friedel–Crafts acylation/oxidation, nitrogen ylide coupling reactions, and

bromine-mediated intramolecular cyclization under diverse catalytic systems and reaction conditions. The synthesized compounds exhibited diverse biological activities such as anti-tumor, anti-plasmodial, anti-parasitic, anti-bacterial, anti-infectious, anti-oxidant, anti-inflammatory, anti-fungal, anti-cancer, anti-pruritic, anti-leukemia, anti-DNA-polymerase, anti-*Trypanosoma cruzi* and anti-epimastigote. We hope this review highlights the versatility of these compounds and inspires current and future generations of chemists to further explore and advance this field.

Data availability

No new data were generated or analysed for this article.

Author contributions

All authors contributed to the scientific writing of the review article. Each author has reviewed and approved the final version for submission.

Conflicts of interest

We have no conflicts of interest to disclose.

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References

- 1 A. K. Jordao, M. D. Vargas, A. C. Pinto, F. d. C. da Silva and V. F. Ferreira, *RSC Adv.*, 2015, **5**, 67909–67943.
- 2 A. Chaudharya and J. M. Khurana, *Curr. Org. Chem.*, 2016, **20**, 1314–1344.
- 3 D. C. Mcmillan, S. D. Sarvate, J. E. Oatis Jr and D. J. Jollow, *Toxicol. Sci.*, 2004, **82**, 647–655.
- 4 M. Dogru, Z. Baysal and C. Aytakin, *Prep. Biochem. Biotechnol.*, 2006, **36**, 215–221.
- 5 S. Ali, T. Hussain and R. Nawaz, *J. Cleaner Prod.*, 2009, **17**, 61–66.
- 6 J.-Y. Heo, C.-H. Cho, H.-S. Jeon, B.-S. Cheong and H.-G. Cho, *Spectrochim. Acta, Part A*, 2011, **83**, 425–431.
- 7 S. Sharma, N. B. Dutta, M. Bhuyan, B. Das and G. Baishya, *Org. Biomol. Chem.*, 2020, **18**, 6537–6548.
- 8 M. R. Xavier, M. M. S. Santos, M. G. Queiroz, M. S. de Lima Silva, A. J. S. Goes and M. A. De Moraes Jr, *Mol. Biol. Rep.*, 2020, **47**, 1173–1185.
- 9 A. Sendl, J. L. Chen, S. D. Jolad, C. Stoddart, E. Rozhon and M. Kernan, *J. Nat. Prod.*, 1996, **59**, 808–811.
- 10 M. M. Rao and D. G. I. Kingston, *J. Nat. Prod.*, 1982, **45**, 600–604.
- 11 O. A. Binutu, K. E. Adesogan and J. I. Okogun, *Antimicrob. Agents Chemother.*, 1998, **42**, 700–702.
- 12 C. E. Heltzel, A. A. Gunatilaka, E. Glass and G. I. Kingston, *J. Nat. Prod.*, 1993, **56**, 1500–1505.
- 13 D. C. Sutton, F. T. Gillan and M. Susic, *Avicennia marina, Phytochemistry*, 1985, **24**, 2877–2879.
- 14 (a) J. Romo and P. Joseph-Nathan, *Tetrahedron*, 1964, **20**, 2331–2337; (b) J. Correa and J. Romo, *Tetrahedron*, 1966, **22**, 685–691.
- 15 K. Gach, J. Modranka, J. Szymanski, D. Pomorska, U. Krajewska, M. Mirowski, T. Janecki and A. Janecka, *Eur. J. Med. Chem.*, 2016, **120**, 51–63.
- 16 T. Takegami, E. Simamura, K. Hirai and J. Koyama, *Antiviral Res.*, 1998, **37**, 37–45.
- 17 A. Reichstein, S. Vortherms, S. Bannwitz, J. Tentrop, H. Prinz and K. Muller, *J. Med. Chem.*, 2012, **55**, 7273–7284.
- 18 H. R. Nasiri, M. G. Madej, R. Panisch, M. Lafontaine, J. W. Bats, C. R. D. Lancaster and H. Schwalbe, *J. Med. Chem.*, 2013, **56**, 9530–9541.
- 19 J.-C. Lien, L.-J. Huang, C.-M. Teng, J.-P. Wang and S.-C. Kuo, *Chem. Pharm. Bull.*, 2002, **50**, 672–674.
- 20 H. Oku, T. Kato and K. Ishiguro, *Biol. Pharm. Bull.*, 2002, **25**, 137–139.
- 21 (a) D. Gao, M. Hiromura, H. Yasui and H. Sakurai, *Biol. Pharm. Bull.*, 2002, **25**, 827–832; (b) W. S. Hamama, A. E. Hassanien and H. H. Zoorob, *J. Heterocycl. Chem.*, 2018, **55**, 282–290; (c) D. K. Nair, R. F. Menna-Barreto, E. N. da Silva Jr, S. M. Mobin and I. N. Namboothiri, *Chem. Commun.*, 2014, **50**, 6973–6976.
- 22 T. T. Guimaraes, F. R. Maria do Carmo, J. S. Lanza, M. N. Melo, R. L. do Monte-Neto, I. M. M. de Melo, E. B. T. Diogo, V. F. Ferreira, C. A. Camara, W. O. Valença, R. N. de Oliveira, F. Frezard and E. N. da Silva Jr, *Eur. J. Med. Chem.*, 2013, **63**, 523–530.
- 23 K.-I. Lee, Y. Park, S.-J. Park, J.-H. Hwang, S.-J. Lee, G.-D. Kim, W.-K. Park, S. Lee, D. Jeong, J.-Y. Kong, H.-K. Kangd and H. Choc, *Bioorg. Med. Chem. Lett.*, 2006, **16**, 737–742.
- 24 K. Muller, A. Sellmer and W. Wiegrebe, *J. Nat. Prod.*, 1999, **62**, 1134–1136.
- 25 R. Ribeiro-Rodrigues, W. G. dos Santos, A. B. Oliverira, V. Snieckus, C. L. Zani and A. J. Romanha, *Bioorg. Med. Chem. Lett.*, 1995, **5**, 1509–1512.
- 26 B. J. Abbott, J. L. Hartwell, J. Leiter, R. E. Perdue and S. A. Schepartz, *Cancer Res.*, 1967, **27**, 190–345.
- 27 S. Eghbaliferiz, S. A. Emami, Z. Tayarani-Najaran, M. Iranshahi, A. Shakeri, J. Hohmann and J. Asili, *Fitoterapia*, 2018, **128**, 97–101.
- 28 (a) A. Olyaei, M. Shalbafan, F. Rahimi and M. Sadeghpour, *J. Adv. Biomed. Sci.*, 2024, **14**, 222–231; (b) A. Olyaei, M. Shalbafan and M. Sadeghpour, *New J. Chem.*, 2024, **48**, 12841–12848; (c) F. Rahimi, A. Olyaei and H. Ghasemzadeh, *Res. Chem. Intermed.*, 2024, **50**, 239–249; (d) A. Olyaei, N. Ghaleghovandi, F. Moghadami, M. Sadeghpour and S. Abediha, *RSC Adv.*, 2023, **13**, 29393–29400; (e) M. R. Khodabakhshi, A. Adl and A. Olyaei, *Res. Chem. Intermed.*, 2023, **49**, 4759–4770; (f) A. Olyaei and M. Sadeghpour, *RSC Adv.*, 2022, **12**, 13837–13895; (g) A. Olyaei, M. Sadeghpour and M. Khalaj, *RSC Adv.*, 2020, **10**, 30265–30281.
- 29 S. C. Hooker, Constitution of lapachol and its derivatives, *J. Chem. Soc.*, 1896, **69**, 1355–1381.
- 30 K. H. Dudley and H. Waynemill, *J. Org. Chem.*, 1967, **32**, 2341–2344.
- 31 A. B. de Oliveira, D. S. Raslanz and F. Khuong-Huu, *Tetrahedron Lett.*, 1990, **31**, 6873–6876.
- 32 K. Kobayashi, *J. Org. Chem.*, 1993, **58**, 4614–4618.
- 33 H. Suginome, A. Konishi, H. Sakurai, H. Minakawa, T. Takeda, H. Senboku, M. Tokuda and K. Kobayashi, *Tetrahedron*, 1995, **51**, 1377–1386.
- 34 K. Kobayashi, T. Uneda, M. Kawakita, O. Morikawa and H. Konishi, *Tetrahedron Lett.*, 1997, **38**, 837–840.
- 35 K. Kobayashi, K. Tanaka, T. Uneda, K. Maeda, O. Morikawa and H. Konishi, *Synthesis*, 1998, 1243–1245.
- 36 A. Martinez, J. C. Estevez, R. J. Estevez and L. Castedo, *Tetrahedron Lett.*, 2000, **41**, 2365–2367.
- 37 E. Martinez, L. Martinez, M. Treus, J. C. Estevez, R. J. Estevez and L. Castedo, *Tetrahedron*, 2000, **56**, 6023–6030.
- 38 M. B. Teimouri and H. R. Khavasi, *Tetrahedron*, 2007, **63**, 10269–10275.
- 39 M. B. Teimouri and R. Bazhrang, *Monatsh. Chem.*, 2008, **139**, 957–961.
- 40 P. Prasanna, K. Balamurugan, S. Perumal and J. C. Menendez, *Green Chem.*, 2011, **13**, 2123–2129.
- 41 S. Jimenez-Alonso, J. Guasch, A. Estevez-Braun, I. Ratera, J. Veciana and A. G. Ravelo, *J. Org. Chem.*, 2011, **76**, 1634–1643.
- 42 Z. Renzun, X. Dongcheng and X. Jianwu, *Chin. J. Chem.*, 2012, **30**, 1690–1694.



- 43 Z.-Z. Wu, Y.-J. Jang, C.-J. Lee, Y.-T. Lee and W. Lin, *Org. Biomol. Chem.*, 2013, **11**, 828–834.
- 44 M. R. Zanwar, V. Kavala, S. D. Gawande, C.-W. Kuo, T.-S. Kuo, M.-L. Chen, C.-H. He and C.-F. Yao, *Eur. J. Org. Chem.*, 2013, 8288–8298.
- 45 S. Liu, L. Long, D. Xie, L. Liu and D. Ma, *Tetrahedron Lett.*, 2015, **56**, 6730–6733.
- 46 K. Rad-Moghadam, S. A. R. Mousazadeh Hassani and S. Toorchi Roudsari, *RSC Adv.*, 2016, **6**, 13152–13159.
- 47 H. Tang, X. Zhang, X. Zeng and Z. Zhou, *Tetrahedron*, 2017, **73**, 6962–6968.
- 48 T. F. Borgati, M. F. A. do Nascimento, J. F. Bernardino, L. C. O. Martins, A. G. Taranto and A. B. de Oliveira, *J. Trop. Med.*, 2017, 7496934.
- 49 J. Li, J. Zhang, M. Li, C. Zhang, Y. Yuan and R. Liu, *Chem. Commun.*, 2019, **55**, 2348–2351.
- 50 X. Li, P. Sun, K. Xie, D. Zhou, J. Peng, A. Fan, J. Zhang and C. Chen, *J. Org. Chem.*, 2020, **85**, 9313–9320.
- 51 P. Sun, J. Yang, J. Peng, B. Mo, X. Chen, X. Li and C. Chen, *J. Org. Chem.*, 2020, **85**, 6761–6769.
- 52 C. L. Koumpoura, M. Nguyen, C. Bijani, L. Vendier, E. G. Salina, S. Buroni, G. Degiacomi, S. Cojean, P. M. Loiseau, F. Benoit-Vical, A. T. Garcia-Sosa, A. Robert and M. Baltas, *ACS Omega*, 2022, **7**, 35635–35655.
- 53 K. Kobayashi, H. Shimizu, A. Srtsaki and H. Sugimoto, *J. Org. Chem.*, 1991, **56**, 3204–3205.
- 54 S. Yamaguchi, T. Katsuki, H. Yokoyama and Y. Hirai, *J. Heterocycl. Chem.*, 2001, **38**, 511–514.
- 55 V. Nair, P. M. Tressa, D. Maliakal and N. P. Path, *Tetrahedron*, 2001, **57**, 7705–7710.
- 56 (a) N. Kongkathip, B. Kongkathip, P. Siripong, C. Sangma, S. Luangkamin, M. Niyomdech, S. Pattanapa, S. Piyaviriyagul and P. Kongsaree, *Bioorg. Med. Chem.*, 2003, **11**, 3179–3191; (b) P. Skehan, R. Storeng, D. Scudiero, A. Monks, J. McMahon, D. Vistica, J. T. Warren, H. Bokesch, S. Kenney and M. R. Boyd, *J. Natl. Cancer Inst.*, 1990, **82**, 1107–1112.
- 57 R. S. F. Silva, E. M. Costa, U. L. T. Trindade, D. V. Teixeira, M. de Carmo, F. R. Pinto, G. L. Santos, V. R. S. Malta, C. A. De Simone, A. V. Pinto and S. L. de Castro, *Eur. J. Med. Chem.*, 2006, **41**, 526–530.
- 58 A. L. Perez, G. Lamoureux and A. Sanchez-Kopper, *Tetrahedron Lett.*, 2007, **48**, 3735–3738.
- 59 (a) K.-L. Wu, S. Wilkinson, N. O. Reich and T. R. R. Pettus, *Org. Lett.*, 2007, **9**, 5537–5540; (b) M. Brasholz, S. Sorgel, C. Azap and H.-U. Reissig, *Eur. J. Org. Chem.*, 2007, 3801–3814.
- 60 E. Perez-Sacau, R. G. Diaz-Penate, A. Estevez-Braun, A. G. Ravelo, J. M. Garcia-Castellano, L. Pardo and M. Campillo, *J. Med. Chem.*, 2007, **50**, 696–706.
- 61 L. F. Fieser and M. Fieser, *J. Am. Chem. Soc.*, 1948, **70**, 3215–3222.
- 62 E. N. da Silva Jr, M. C. B. V. de Souza, M. C. Fernandes, R. F. S. Menna-Barreto, M. do Carmo, F. R. Pinto, F. de Assis Lopes, C. A. de Simone, C. K. Z. Andrade, A. V. Pinto, V. F. Ferreira and S. L. de Castro, *Bioorg. Med. Chem.*, 2008, **16**, 5030–5038.
- 63 M. Yilmaz, M. Yakut and A. T. Pekel, *Synth. Commun.*, 2008, **39**, 914–927.
- 64 C. P. V. Freire, S. B. Ferreira, N. S. Melo de Oliveira, A. B. J. Matsuura, I. L. Gama, F. de C. da Silva, M. C. B. V. de Souza, E. S. Lima and V. F. Ferreira, *Med. Chem. Commun.*, 2010, **1**, 229–232.
- 65 M. Rueping, A. Parra, U. Uria, F. Besselièvre and E. Merino, *Org. Lett.*, 2010, **12**, 5680–5683.
- 66 M. C. Fernandes, E. N. Dasilva Jr, A. V. Pinto, S. L. De Castro and R. F. S. Menna-Barreto, *Parasitology*, 2012, **139**, 26–36.
- 67 E. N. da Silva Jr, M. C. B. V. Souza, A. V. Pinto, M. C. F. R. Pinto, V. F. Ferreira, R. F. S. Menna-Barreto, R. S. F. Silva, D. V. Teixeira, C. A. de Simone and S. L. de Castro, *Eur. J. Med. Chem.*, 2008, **43**, 1774–1780.
- 68 E. N. da Silva Júnior, I. M. M. de Melo, E. B. T. Diogo, V. A. Costa, J. D. de Souza Filho, W. O. Valença, C. A. Camara, R. N. de Oliveira, A. S. de Araujo, F. S. Emery, M. R. dos Santos, C. A. de Simone, R. F. S. Menna-Barreto and S. L. de Castro, *Eur. J. Med. Chem.*, 2012, **52**, 304–312.
- 69 F. da Rocha Ferreira, S. B. Ferreira, A. J. Araujo, J. D. B. M. Filho, C. Pessoa, M. O. Moraes, L. V. Costa-Lotufo, R. C. Montenegro, F. de C. da Silva, V. F. Ferreira, J. G. da Costa, F. C. de Abreu and M. O. F. Goulart, *Electrochim. Acta*, 2013, **110**, 634–640.
- 70 M. F. C. Cardoso, P. C. Rodrigues, M. E. I. M. Oliveira, I. L. Gama, I. M. C. B. da Silva, I. O. Santos, D. R. Rocha, R. T. Pinho, V. F. Ferreira, M. C. B. V. de Souza, F. de C. da Silva and F. P. Silva-Jr, *Eur. J. Med. Chem.*, 2014, **84**, 708–717.
- 71 T. A. D. Thi, Y. Depetter, K. Mollet, H. T. Phuong, D. V. Ngoc, C. P. The, H. T. Nguyen, T. H. N. Thi, H. H. Nguyen, M. Dhooghe and T. V. Nguyen, *Tetrahedron Lett.*, 2015, **56**, 2422–2425.
- 72 M. F. do Carmo Cardoso, K. Salomao, A. C. Bombaca, D. R. da Rocha, F. de C. da Silva, J. A. S. Cavaleiro, S. L. de Castro and V. F. Ferreira, *Bioorg. Med. Chem.*, 2015, **23**, 4763–4768.
- 73 E. H. G. da Cruz, M. A. Silvers, G. A. M. Jardim, J. M. Resende, B. C. Cavalcanti, I. S. Bomfim, C. Pessoa, C. A. de Simone, G. V. Botteselle, A. L. Braga, D. K. Nair, I. N. N. Namboothiri, D. A. Boothman and E. N. da Silva Jr, *Eur. J. Med. Chem.*, 2016, **122**, 1–16.
- 74 M. Salaria, M. H. Mosslemina and A. Hassanabadi, *J. Chem. Res.*, 2017, **41**, 448–451.
- 75 I. C. Chipoline, E. Alves, P. Branco, L. V. Costalotufo, V. F. Ferreira and F. C. da Silva, *An. Acad. Bras. Cienc.*, 2018, **90**, 1027–1033.
- 76 D. C. S. Costa, G. S. de Almeida, V. W.-H. Rabelo, L. M. Cabral, P. C. Sathler, P. A. Abreu, V. F. Ferreira, L. C. R. P. da Silva and F. de C. da Silva, *Eur. J. Med. Chem.*, 2018, **156**, 524–533.
- 77 P. G. Ferreira, L. P. Borba-Santos, L. L. Noronha, C. D. Nicoletti, M. de Sa, H. Queiroz, F. de C. da Silva, S. Rozental, D. O. Futuro and V. F. Ferreira, *Molecules*, 2019, **24**, 930.
- 78 G. Li, H. Zhang, G. Zhang, B. Wei, B. Liu, H. Song, Q. Li and S. Ban, *J. Org. Chem.*, 2023, **88**, 4809–4813.



- 79 A. E. Hassanien, G. E. Abd EL-ghani and G. G. Elbana, *Polycycl. Aromat. Compd.*, 2023, **43**, 1378–1394.
- 80 K. Kobayashi, M. Mori, T. Uneda, O. Morikawa and H. Konishi, *Chem. Lett.*, 1996, **25**, 451–452.
- 81 Y. R. Lee, B. S. Kim and D. H. Kim, *Tetrahedron*, 2000, **56**, 8845–8853.
- 82 T. Hayashi, F. T. Smith and K.-H. Lee, *J. Med. Chem.*, 1987, **30**, 2005–2008.
- 83 H. Wagner, B. Kreher and H. Lotter, *Helv. Chim. Acta*, 1989, **72**, 659–667.
- 84 Y. R. Lee, B. S. Kim, Y. U. Jung, W. S. Koh, J. S. Cha and N. W. Kim, *Synth. Commun.*, 2002, **32**, 3099–3105.
- 85 Y. R. Lee, G. J. Lee and K. Y. Kang, *Bull. Korean Chem. Soc.*, 2002, **23**, 1477–1480.
- 86 M. Itoigawa, C. Ito, H. T.-W. Tan, M. Okuda, H. Tokuda, H. Nishino and H. Furukawa, *Cancer Lett.*, 2001, **174**, 135–139.
- 87 Y. R. Lee, K. Y. Kang, G. J. Lee and W. K. Lee, *Synthesis*, 2003, 1977–1988.
- 88 Y. R. Lee and B. S. Kim, *Synth. Commun.*, 2003, **33**, 4123–4135.
- 89 C.-H. Tseng, C.-S. Lin, P.-K. Shih, L.-T. Tsao, J.-P. Wang, C.-M. Cheng, C.-C. Tzeng and Y.-L. Chen, *Bioorg. Med. Chem.*, 2009, **17**, 6773–6779.
- 90 J. Acuna, J. Piermattey, D. Caro, S. Bannwitz, L. Barrios, J. Lopez, Y. Ocampo, R. Vivas-Reyes, F. Aristizábal, R. Gaitan, K. Müller and L. Franco, *Molecules*, 2018, **23**, 186.
- 91 S. Lv, H. Liu, J. Kang, Y. Luo, T. Gong, Z. Dong, G. Sun, C. He, X. Sun and L. Wang, *Chem. Commun.*, 2019, **55**, 14729–14732.
- 92 S.-Y. Li, Z.-K. Sun, X.-Y. Zeng, Y. Zhang, M.-L. Wang, S.-C. Hu, J.-R. Song, J. Luo, C. Chen, H. Luo and W.-D. Pan, *Molecules*, 2019, **24**, 4138.
- 93 H. Tan, Z. Qi, Y. Yu, X. Zhang, Y. Xiang, J. Huang, Z. Xu, D. Tang, Z. Chen and B. Wang, *Molecules*, 2023, **28**, 4751.

