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Fast tracking of adulterants and bacterial contamination in food *via* Raman and infrared spectroscopies: paving the way for a healthy and safe world†

Raj Kumar Sen,^{ab} K. Karthikeyan,^{id c} Priyanka Prabhakar,^{ab} Jeet Vishwakarma,^{ab} Gaurav Gupta,^{ab} S. N. Mishra,^{ab} Alka Mishra,^{ab} J. P. Chaurasia,^{ab} S. A. R. Hashmi,^{ab} D. P. Mondal,^{ab} Pratima R. Solanki,^d A. K. Srivastava,^{id ab} Chetna Dhand^{id *ab} and Neeraj Dwivedi^{id *ab}

Food adulteration adversely affects human health. Therefore, its fast and accurate detection is essentially required to alleviate health concerns. While there are several methods available for the detection of food adulterants, some of them are slow and require large amounts of samples for analysis. Optical spectroscopy methods, especially Raman and infrared (IR), are emerging techniques owing to their fast and accurate detection abilities as well as their ability to examine a very small amount of sample. This paper critically discusses the advancements in the detection of food adulterants using Raman and IR spectroscopies, with a major focus on identifying adulterants in milk, honey, olive oil, coconut water, clove essential oil, and spices (turmeric and saffron). We also discuss the role of Raman and IR spectroscopies in detecting bacterial contamination in food. The fundamentals and application of several Raman and IR-based techniques, namely, surface-enhanced Raman spectroscopy (SERS), Fourier transform (FT)-Raman, point-scan, portable Raman and IR, near and mid-IR, Fourier transform infrared (FTIR), and different associated types of chemometric methods are substantially discussed for the aforementioned applications. Finally, we also propose ways to further advance food security applications *via* these optical spectroscopy tools.

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1. Introduction

Humans' highest hopes are to live a disease-free life and consume high-quality food, but in today's world, these are difficult. More than half of the groceries in the kitchen are adulterated.¹ As per the 2018–2019 report released by the Food Safety and Standards Authority of India (FSSAI), 28% of the food samples tested were found to be contaminated or sub-standard in India.² If humans continued to consume such adulterated food, their health would be severely affected. Adulteration of food is described as the intentional or unintentional substitution with inferior foreign particles in food or the extraction of a value-added food substitute from the main food component. Food adulteration involves

injecting harmful chemicals into food, which lowers the quality of the food.^{3,4}

Among the techniques available, optical spectroscopies are a powerful tool owing to their fast, easy, and accurate detection of food adulterants and bacterial contamination in food. Ultraviolet-visible (UV-vis) spectroscopy, fluorescence spectroscopy, IR spectroscopy, Raman spectroscopy, and nuclear magnetic resonance have shown huge potential in this regard. Table 1 shows a brief overview of Raman and IR techniques and their advantages. In this review, we will focus on Raman and IR spectroscopy for the detection of food adulterants and bacterial contamination in food. We will review the fundamentals of Raman and IR spectroscopies, critically review the progress of these techniques for the diagnosis of food adulterants and contaminants, and discuss future prospects.

2. Fundamentals of Raman and IR spectroscopy

Initially introduced in the early twentieth century, vibrational spectroscopy methods such as IR spectroscopy⁸ and Raman

^a CSIR-Advanced Materials and Processes Research Institute, Bhopal-462026, India.
E-mail: chetna.dhand@ampri.res.in, chetnachem24@gmail.com,
neeraj.dwivedi@ampri.res.in, neerajdwivedi6@gmail.com

^b Academy of Scientific and Innovative Research (AcSIR), Ghaziabad-201002, India

^c Kongunadu Arts and Science College (KASC), Coimbatore, Tamilnadu-641029, India

^d Special Centre for Nanoscience, Jawaharlal Nehru University, New Delhi-110067, India

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Table 1 Raman and IR spectroscopies and their advantages

Techniques	Advantages	Disadvantages
Raman spectroscopy		
1. Dispersive Raman spectroscopy	Ideal for aqueous samples, samples measured at a higher temperature, <i>etc.</i> ⁵	More time-consuming than FT-Raman
2. FT-Raman spectroscopy	Reduces fluorescence, simplifies FTIR spectrometer work, and has a high spectral resolution	Black samples can strongly absorb, heat up, and produce intense background emissions
3. SERS	High sensitivity, resolution, and specificity	A short-range chemical amplification mechanism results in signal suppression ⁶
IR spectroscopy	Fast, non-destructive, and reliable ⁷	Some of the bonds, mainly hydrogen bonds such as -C-H, -S-H, -N-H, -O-H, are less intense or overlap

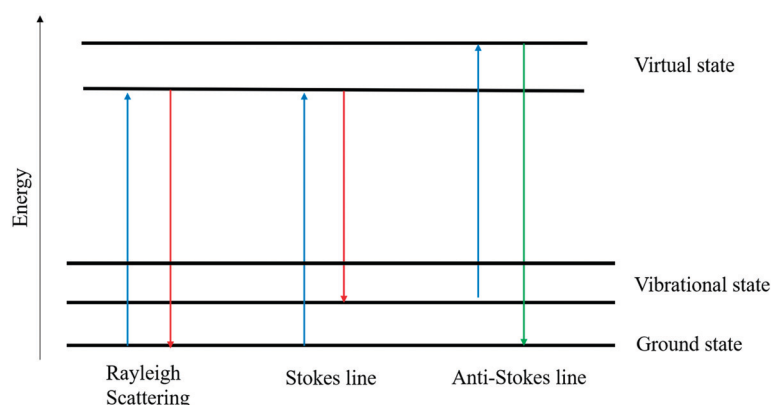
spectroscopy⁹ have quickly emerged as speedy and non-destructive tools in a variety of sectors. Raman spectroscopy is based on the inelastic scattering of photons by phonons. For molecular vibrations to be Raman active, they must cause a shift in polarizability, a parameter that describes how easily electrons respond to an electric field of light. When a laser beam falls on a sample, some of the photons are inelastically scattered. As a result, the molecule's vibrational energy is altered, and the dispersed radiation is transferred to a new wavelength. A Raman shift is defined as the difference in frequency between scattered and incident radiation. Scattered photons are transferred to longer wavelengths, resulting in a Stokes line in the Raman spectrum.⁹ Otherwise, the wavelengths are shifted to shorter wavelengths, which leads to an anti-Stokes line. Fig. 1 depicts a Raman scattering energy level diagram. Scattered light frequency shifts can be analyzed and interpreted as spectra. The spectra can serve as a fingerprint for a specific substance, making it easier to analyze this compound in a variety of samples and providing a foundation for structural and qualitative research.⁵ In recent years, Raman spectroscopy has gained interest in biological measurements because of its rapid and non-destructive analysis, and the requirement of a small amount of sample. Lately, Raman spectroscopy has been upgraded in terms of applications, for example, resonance Raman spectroscopy, surface-enhanced Raman spectroscopy (SERS), tip-enhanced Raman spectroscopy (TERS), *etc.*¹⁰ In SERS, the

Raman scattering signals can be enhanced by several orders of magnitude compared to the common Raman scattering. SERS can be employed in food analysis for the detection of thiram, melamine, and ciprofloxacin.

FTIR spectroscopy is a type of vibrational spectroscopy that uses electromagnetic radiation to interact with functional groups in analyzed samples, resulting in vibrational energy levels. FTIR spectroscopy has also emerged as a fast and non-destructive technique for authentication analysis, particularly when using attenuated total reflectance (ATR) and chemometric models. Based on FTIR spectral characteristics, this technique can qualitatively and quantitatively distinguish between authentic and adulterated foods. The IR radiation covers wavenumbers (ν) from 14 000–50 cm^{-1} with near IR (NIR) from 14 000 to 4000 cm^{-1} , mid-IR (MIR) from 4000 to 400 cm^{-1} , and far IR (FIR) from 400 to 50 cm^{-1} . MIR is the most commonly used method for the detection of functional group authentication among these three spectral regimes.⁷ The combination of spectroscopy and chemometrics could provide a relatively inexpensive, fast, and non-destructive method for testing food microbiological safety. Fig. 2 shows the basic instrumentation for FT-IR spectroscopy.

3. SERS

SERS has emerged as a popular tool for analyzing food adulterants and biological samples.¹¹ Jeanmaire and Van

**Fig. 1** Light-matter interaction and associated elastic and inelastic scattering.

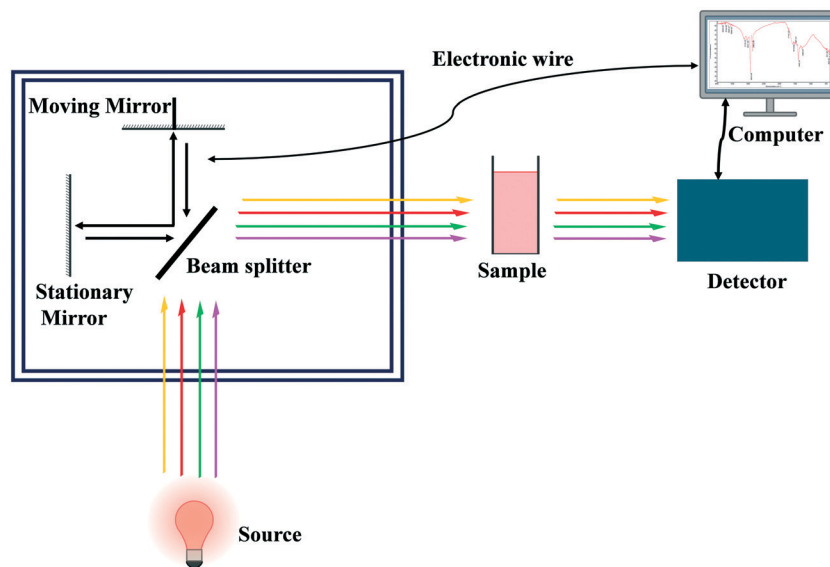


Fig. 2 Schematic diagram of FT-IR spectroscopy.

Duyn¹² conducted pyridine spectroscopy analysis around four decades ago and discovered that using a SERS substrate significantly improved the Raman signals. Generally, analytical approaches that are used to deal with SERS either identify or quantify molecules or both.¹³ A SERS characteristic is a vibrational fingerprint of the species being studied that can be used to identify it directly.¹⁴ Its high sensitivity makes SERS capable of detecting analytes even when their concentration in solution is very low. Recent advances in nanotechnology have made it possible to design, fabricate, and synthesize nanostructures at a breakneck pace.¹⁵ In SERS research, the substrate reproducibility problem has also been attempted to be minimized.^{15,16} Researchers have presented several approaches for substrate preparation, and each of them has its pros and cons. Now, commercial SERS substrates are also available on the market.¹⁶ The functioning of SERS substrates and the purity of materials are ensured using Raman peaks.¹⁷ SERS is now employed in food analysis for the detection of thiram,¹⁸ melamine,¹⁹ and ciprofloxacin.²⁰

4. Detection of food adulterants

Adulteration is a major problem worldwide; it is complex to analyze it because adulterated foods are similar to authentic materials.²¹ Traditional techniques for investigating food adulteration include gas chromatography-mass spectrometer (GS-MS), gas chromatography (GS), high-performance liquid chromatography (HPLC), capillary electrophoresis, ion chromatography, hydrophilic interaction liquid chromatography (HILIC), electrospray ionization mass spectrometry (ESIMS), *etc.* These traditional methods require large amounts of solvents and reagents and sample pretreatment, and affect the sample in the process of detection. Additionally, many of these methods are complex,

slow, labor-intensive, costly, and not conducive to large-scale applications. By contrast, Raman and IR spectroscopies have shown promise as simple, fast, and non-invasive tools for the identification of adulterants. In this section, we will comprehensively discuss the detection of adulterants in essential food products such as milk, honey, olive oil, coconut water, clove essential oil, and different spices using Raman and IR spectroscopic methods. We will also discuss the applications of Raman spectroscopy, IR spectroscopy, and surface plasmon resonance (SPR) techniques for the detection of bacterial contamination in food.

4.1 Probing the adulterants in milk

Milk is a commonly consumed food with significant nutritional (protein, fat, carbohydrate, vitamins, and minerals) and economic values. The reconstitution of milk is a process that partially or fully replaces raw milk with milk powder. Adulteration is possible in both raw milk and milk powder, requiring fast, easy, and automated methods for detecting the adulterants.²² The major milk adulterants include urea, melamine, boric acid, ammonia, formalin, hydrogen peroxide, benzoic acid, sucrose, dicyandiamide, and ammonium sulfate.²³ Kuanglin *et al.* used a point-scan Raman spectroscopy device with a 16-bit charge-coupled device (CCD) camera to detect melamine in milk.²⁴ The light source for the analysis of the milk sample was a laser (785 nm) with a maximum power output of 430 mW; Raman spectroscopy successfully identified the melamine in the milk sample.²⁴ Zhang *et al.* reported the application of SERS to detect melamine in milk, whereby a silver-based SERS substrate caused several-fold enhancement of the Raman signals.²⁵ The limit of detection of this method was 0.01 $\mu\text{g mg}^{-1}$ for melamine samples. The results of this method are as accurate and reliable as those predicted by liquid



chromatography/tandem mass spectrometry. Rajapandiyar *et al.* and Mecker *et al.* also reported the application of SERS for the detection of melamine in liquid milk and milk powder, whereby the signal was enhanced by employing silver nanoparticles (AgNPs).^{26,27} Hu *et al.* reported the detection of melamine in whole milk using a novel biosensor that combines molecularly imprinted polymers with SERS (MIPs-SERS).²⁸ MIPs were found to be effective in separating melamine from whole milk. The signal was collected using a silver dendrite nanostructure that was SERS-active. The Raman signatures of whole milk samples with different melamine concentrations were separated by principal component analysis (PCA) and hierarchical cluster analysis (HCA). The intensity of the melamine peak (at 703 cm^{-1}) was found to have a strong linear relationship with the concentration of melamine from 0.005 to 0.05 mmol L^{-1} . The detection and quantification limits were 0.012 and 0.039 mmol L^{-1} , respectively, demonstrating the biosensor's high sensitivity in detecting melamine in whole milk, as illustrated in Fig. 3.

Almeida *et al.* employed a combination of Raman spectroscopy and chemometric modeling for evaluating the consistency of milk powder samples.²⁹ Separation methods and other stages of sample preparation are not needed for this approach. As a result, the analysis can be completed easily and without producing waste. Du *et al.* reported the partial least-squares-discriminant analysis (PLS-DA) model that yielded adequate separation of spectral fingerprints, indicating that FTIR spectroscopy has great potential for

controlling milk fraud.²² Jawaidd *et al.* reported the use of the single bounce (SB)-ATR method of FTIR to detect and quantify melamine in both liquid and powdered milk.³⁰ PLS models were developed for the correlation of melamine concentration with spectral results. Cassoli *et al.* collected samples of milk from 100 farms for the construction of calibration systems.³¹ Three separate adulterants that are widely used in raw milk adulteration were found in the samples: sodium citrate (SC), sodium bicarbonate (SB), and cheese whey (W), and three different concentrations of each adulterant were studied. The results show that the FTIR methodology can be used to determine the concentration of sodium citrate, sodium bicarbonate, and whey in raw milk. Nicolaou *et al.* demonstrated that in binary and tertiary milk mixtures, FTIR spectroscopy combined with chemometric models, such as linear, PLS, and nonlinear Kernel PLS, can provide accurate, simple, and rapid detection of adulteration of sheep, goat, and cow milk.³² For all milk species, the average errors were in the range of 5 to 8%. Due to the high speed of analysis (30 s per sample in batches of 96 or 384) and good accuracy, it was concluded that FTIR spectroscopy can substitute less effective and more time-consuming techniques for detecting milk adulterants in the food industry. Furthermore, He *et al.* used IR techniques in conjunction with a two-dimensional (2D) correlation study to review and analyze adulterants in milk.³³ They first recorded the spectra of fresh raw milk and noted characteristic peaks. Subsequently, the adulterated milk samples were tested and their spectra were compared to those of fresh milk, as shown

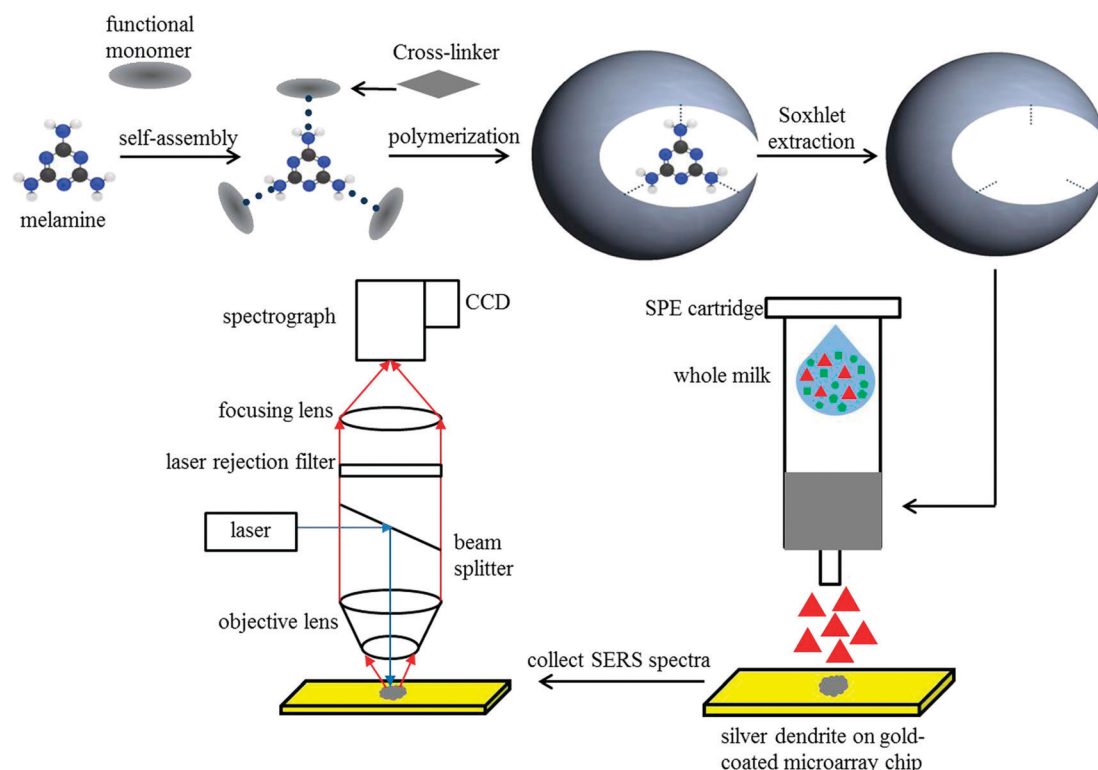


Fig. 3 MIPs-SERS biosensor schematic diagram of melamine detection in whole milk. Reproduced with permission from ref. 28.



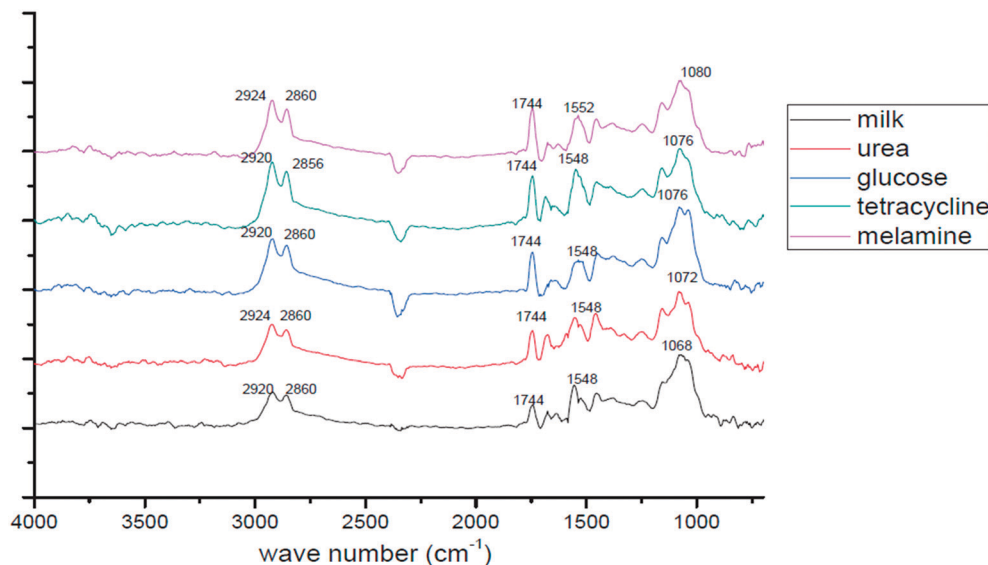


Fig. 4 FTIR absorbance spectra of milk without and with adulterants. Some of the figures were reproduced with the permission from ref. 33.

in Fig. 4. 2D correlation spectroscopy based on the intensity, position, and shape of the absorption peaks may be utilized to identify the adulterants in the milk sample.

With the help of the PLS algorithm, Raman spectroscopy was utilized for the quantitative identification of urea in milk powder.³⁴ The findings reveal higher accuracy (greater than 90%) and the ability to detect urea levels below the suggested cut-off limit (70 mg dl⁻¹). For the analysis of milk powder samples, Limm *et al.* used portable MIR spectroscopy as well as NIR spectroscopy combined with non-targeted chemometrics (SIMCA).³⁵ Melamine, dicyandiamide, ammonium sulfate, and sucrose are all detectable in milk using portable Raman spectrometers.³⁶ The capability of low-cost and compact Raman spectrometers for quick screening of sucrose and N-rich molecules provides an efficient way to monitor milk adulterants from farm to factory.³⁷ The excessive use of tetracycline antibiotics in cattle causes contamination in milk; the high level of tetracycline in milk affects human health.³⁷ With the help of the PCA method, Marques *et al.* used a paper-based SERS platform for the one-step detection of tetracycline.³⁸ They used an aluminum layer deposited with silver nanoparticles as the substrate for the SERS platform. Fig. 5 displays the process for the detection of adulterants using Raman and FTIR spectroscopies.³⁹ Table 2 represents Raman and IR-based spectroscopies for the detection of adulterants in milk.

4.2 Detection of adulterants in honey

Honey is a nutrient-dense food with many medicinal properties, including antibacterial, anti-inflammatory, antioxidant, and cell-growth properties;^{45,46} it is easy to contaminate it with other sugar syrups and hence the honey loses its purity. Honey has 80% carbohydrate content (glucose, fructose, and sucrose) along with vitamins,

minerals, proteins, and amino acids.⁴⁷ Because of its complexity, honey comes in a wide variety across the world.⁴⁸ Raman spectroscopy identifies precise bands that characterize a molecular mixture, allowing for high-quality measurements. Özbacı *et al.* reported the application of Raman spectroscopy to assess the amount of glucose, fructose, sucrose, and maltose in aqueous solutions and commercial honey, and spectral data were analyzed using PCA, PLS, and artificial neural network (ANN) methods.⁴⁹ These findings show that using Raman spectroscopy in conjunction with suitable multivariate techniques, it is possible to quantitatively evaluate the amounts of glucose, fructose, sucrose, and maltose in honey samples without the use of any treatment or chromatographic methods. Mignani *et al.* used Raman spectroscopy with an excitation wavelength of 1064 nm to analyze honey adulterants and nutraceutical indicators such as amino acids, proteins, enzymes, and other minerals.⁵⁰ Here PLS regression was used for analyzing the concentration of the sugars and potassium from Raman spectral bands while PCA was used for data dimensionality reduction and explorative analysis. Oliveira *et al.* reported the use of FT-Raman spectroscopy to analyze the liquid and crystalline states of honey; they showed that crystallized samples have well-resolved distinctive spectral bands compared to liquid samples.⁵¹ Pierna *et al.* also employed FT-Raman spectroscopy, working with PCL-DA and support vector machines (SVM), for the detection of honey adulterants.⁵² Li *et al.* reported the application of Raman spectroscopy to detect the high fructose corn syrup (HFCS) and maltose syrup (MS) in honey; the results yielded a total accuracy of detection of about 84.4%.⁵³ NIR spectroscopy was also used for qualitative and quantitative detection of honey adulterated with HFCS or MS.⁵⁴ The main variables were chosen using competitive adaptive reweighted sampling (CARS). The adulterated honey samples were classified using



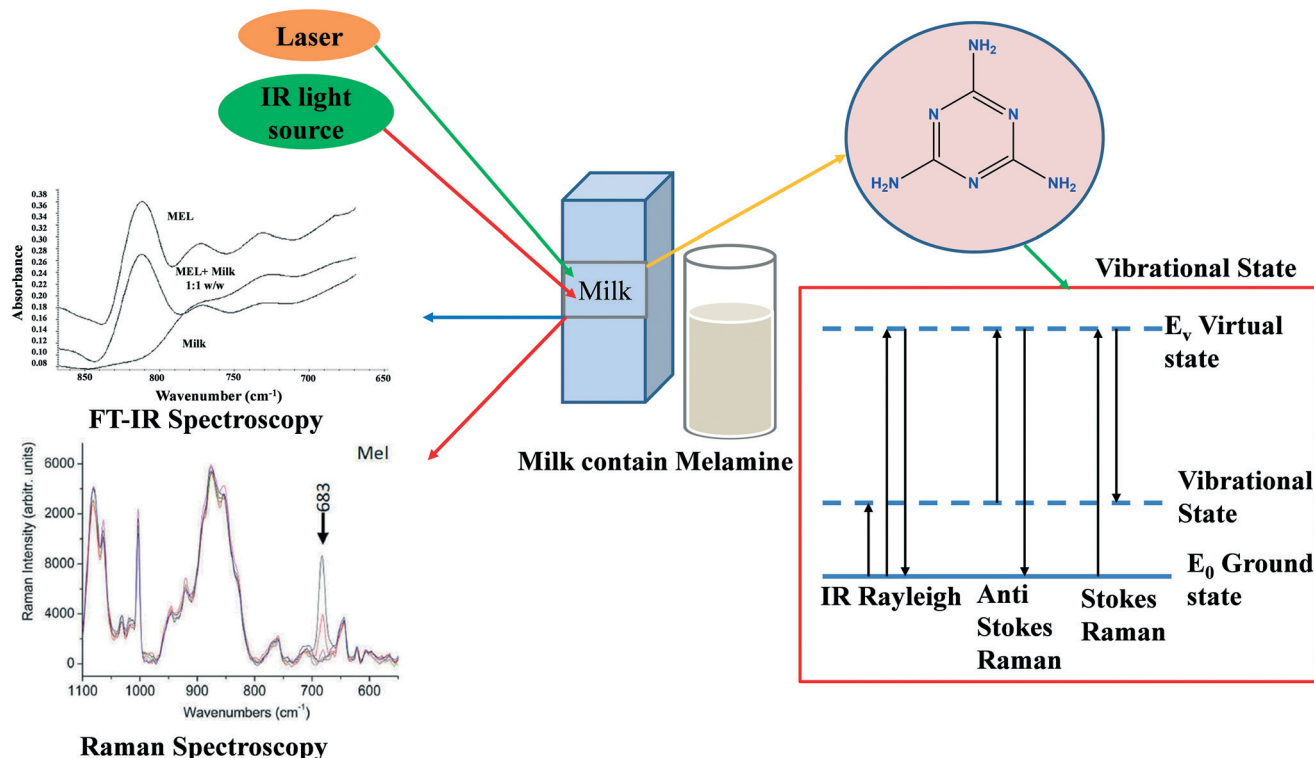


Fig. 5 The schematic diagram for milk adulteration detection using Raman and FT-IR. Reproduced with permission from ref. 39.

Table 2 Raman and IR-based spectroscopies for the identification of milk adulterants

Adulterant	Method for detection	Chemometrics	Accuracy/LOD	Ref.
Addition of sucrose	Normal mid-infrared spectra	For classification: PCA and SIMCA Quantification using polymerase chain reaction (PCR) and PLS	100%	40
Addition of water, starch, sodium citrate, formaldehyde, and sugar	Mid-infrared-spectra in the range of 4000–600 cm ⁻¹	PLS-DA	0.5–10% w/v	41
Addition of melamine	Normal NIR spectra	One class partial least squares (OCPLS)	89%	42
Chloride, citrate, hydroxide, formaldehyde, hydrogen peroxide bicarbonate, carbonate, hypochlorite, starch, sucrose, and water	MIR from 1000 to 4000 cm ⁻¹	PCA was used to visualize the sample distribution, and SIMCA was used to classify milk	93.10	43
Tetracycline's residue (tetracycline, chlortetracycline, and oxytetracycline)	MIR (FTIR) from 4000 to 550 cm ⁻¹	SIMCA for classification, PLS, and PCR for quantification of tetracycline residue	99%	44
Sodium hypochlorite	FTIR spectra in the range of 4000–650 cm ⁻¹	SIMCA for classification	56.7%	43
Sweet whey	Raman spectra from 800 to 1800 cm ⁻¹	ANN for quantification and PLS for corrected prediction	—	—

PLS-linear discriminate analysis (LDA). The accuracy of the CARS-PLS-LDA models was 86.3% for honey *versus* contaminated honey with HFCS, and 96.1% for honey *versus* contaminated honey with MS. Oroian *et al.* employed Raman spectroscopy combined with PLS-LDA, PLS regression (PLSR), and PCR to detect honey adulterants.⁵⁵ Guelpa *et al.* used NIR spectroscopy combined with PLS-DA and found 90% accuracy in the detection of honey adulterants.⁵⁶ Subari *et al.* compared the LDA and PCA methods for identifying adulterants in honey and found that the LDA method performs better than the PCA method.⁵⁷ Basar and Ozdemir investigated honey adulterants (corn syrup, beet sugar, and

water) through ATR-FTIR employing genetic algorithm-based inverse least squares (GILS) and PLS analytical techniques and observed that GILS provides a better model compared with PLS.⁵⁸ Magdas *et al.* also used Raman spectroscopy with SIMCA and machine learning (ML) approaches to determine the authenticity of honey.⁵⁹ Fig. 6 shows the procedure for the detection of adulterated honey through typical Raman and FTIR spectroscopy with various analysis approaches. Csilla *et al.* suggest a new method for the preparation of green samples for the detection of honey adulterants using Raman spectroscopy.⁶⁰ It was proposed that honey dilution with distilled water during sample preparation helps to get

reliable and reproducible spectra. Table 3 shows a summary of the results of detection of adulterants in honey using Raman and IR-based spectroscopies.

4.3 Detection of adulterants in olive oil

Olive oil is made from olives. Saturated fat makes up 14% of the oil, while polyunsaturated fats like omega-6 and omega-3 fatty acids make up 11%. Olive oil's most abundant fatty acid is oleic acid, a monounsaturated fat that accounts for 73% of the total oil content. Extra virgin olive oil (EVOO) is a nutrient-dense oil that is also high in antioxidants. However, a lot of adulteration agents are mixed with olive oil, so pure virgin olive oil is seldom available. An accurate and fast method for identifying the adulterants in olive oil is needed within an hour. Combining FT-Raman spectroscopy with multivariate statistical methods can be a rapid and accurate technique for the detection of adulterants in virgin olive oil. Olive oil is generally contaminated with soybean, corn oil, and olive residue (olive pomace). The PCR and stepwise linear discriminate analysis (SLDA) methods of FT-Raman can be used to distinguish between authentic and adulterated samples.^{66,67} Similarly, NIR

with PCA has been used to accurately identify the adulterants in olive oil.⁶⁸ Raman spectra of olive oil samples were analyzed on-site and in real-time with a portable Raman spectrometer.⁶⁹ Authentic olive oil can be separated from olives having 5% or more of other edible oils, based on Raman bands. Also, some researchers have been able to rapidly screen and assess the quality of olive oil with the help of low-resolution portable Raman spectroscopy.^{70,71} NIR spectroscopy was used for the detection of adulterants in EVOO (adulterated with corn oil in an amount ranging from 2.7 to 25 w/w%).⁷² Many researchers have used FTIR, FT-Raman, and Raman spectroscopy and chemometric methods to analyze EVOO adulterated with various concentrations of soybean, peanut, olive pomace, and sunflower oil.^{73–78} These studies demonstrated the potential of optical spectroscopy in detecting olive oil adulterants. Waste cooking oil adulterant for olive oil was identified using Raman spectroscopy and associated chemometric methods such as interval partial least squares (IPLS) and synergy interval partial least squares (SIPLS).⁷⁹ The root mean square errors of calibration and validation are found to be 0.0503 and 0.0485, respectively. This method is a fast and easy way to predict adulteration in olive oil. In another study, Joshi *et al.* also

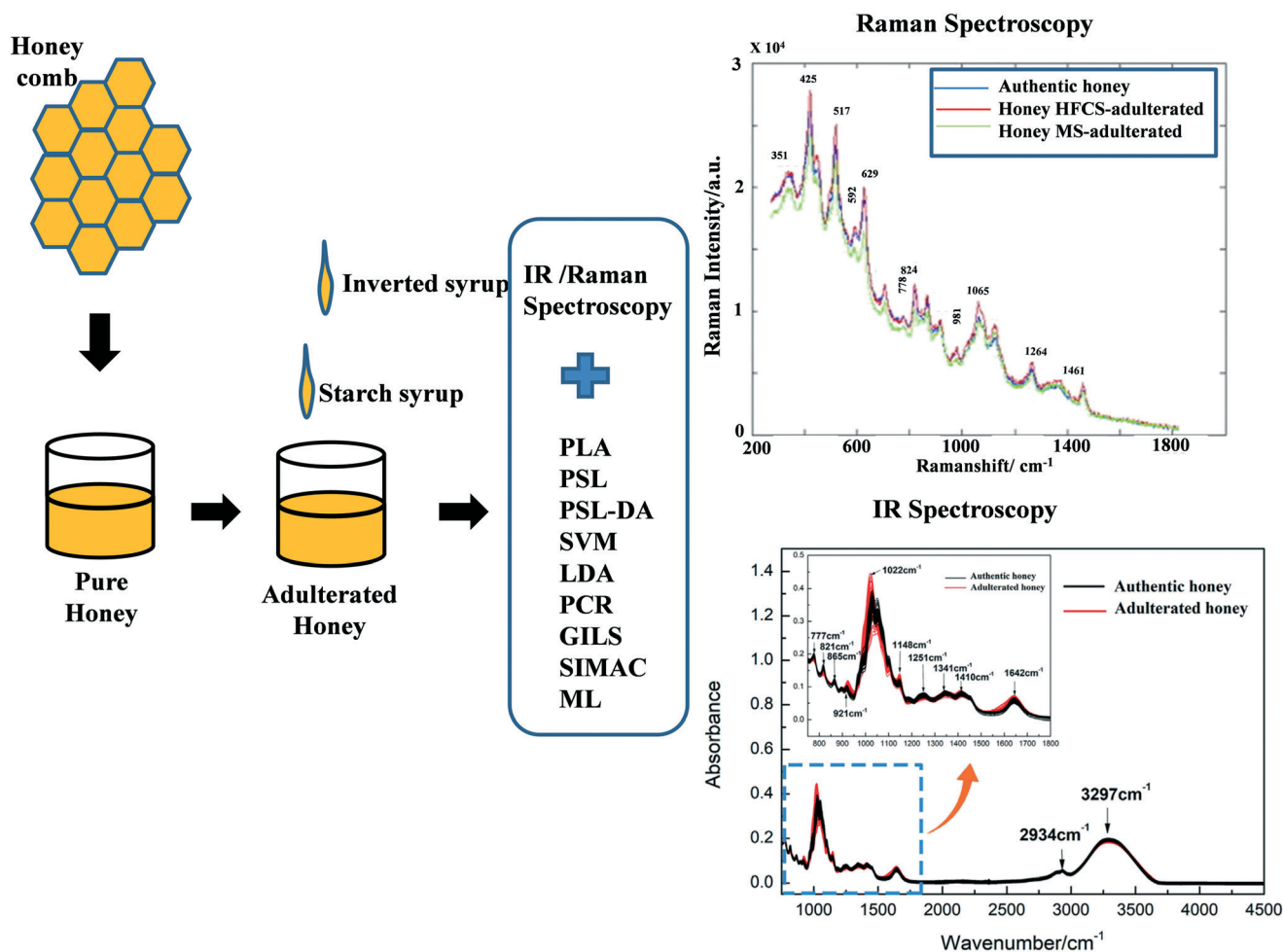


Fig. 6 The schematic diagram shows the procedure for the detection of adulterated honey through typical Raman and FTIR spectroscopy with various analysis methods. Reproduced with permission from ref. 53 and 65.

Table 3 Raman and IR-based spectroscopies for honey adulteration detection

Adulterant	Method for detection	Chemometrics	Accuracy/LOD	Ref.
Sugar syrup	FT-MIR spectroscopy in the range of 4000–600 cm ⁻¹	PCA, HCA, and SVM	100%	61
HFCS	Visible NIR spectroscopy	PCA, HCA, and LDA	100%	62
Sweetener's materials	NIR spectroscopy	PCA, and least squares support vector machine (LS-SVM)	95.1%	63
Beet syrup, high-fructose corn syrup, partial invert cane syrup, dextrose syrup, and beet sucrose	IR spectroscopy with MIR spectra from 800 to 4000 cm ⁻¹	SIMCA, and PLS	96.2%, 97.5%, 95.8%, and 91.7%	64

reported the detection of an olive oil mixture in argan oil by using portable Raman spectroscopy.⁸⁰ The analysis of EVOO combined with grapeseed and canola oils at 0–100% was performed using a short distance standoff Raman detection instrument attached to a 785 nm laser source.⁸¹ Further, Raman spectroscopy with exponential equations has been used to predict the adulteration in olive oil.⁸² Exponential equations are used to measure the volume fraction (μ) of adulterants like rapeseed, corn, and sunflower oils in olive oil. Table 4 shows a summary of the results of detection of adulterants in olive oil using Raman and IR-based spectroscopies.

4.4 Detection of adulterants in coconut water and clove oil

Coconut water is a good source of many nutrients, like fiber, protein, magnesium, potassium, sodium, vitamins, and carbohydrate.⁸⁴ It contains antioxidants that protect the cells from damage caused by free radicals. It also improves blood sugar control, increases insulin sensitivity, prevents kidney stones by reducing stone formation, and helps lower blood pressure. Coconut water can be adulterated with normal water, sugar, corn syrup, and other chemical components.⁸⁵ The quantification of adulterants was performed using Raman spectroscopy (785 nm excitation source) with chemometric methods, and three types of adulterants were detected, namely, single sugar, mixed sugar, and HFCS.⁸⁶ FTIR spectroscopy also has the potential to analyze sugars in fruit juices with the help of the PLSR approach.⁸⁷ Clove essential oil is used for its antimicrobial activity, for preventing cavities, and for relieving oral pain, and it may also be used for its anticancer activity.⁸⁸ Clove oil is also adulterated with foreign materials due to its high cost. Handheld Raman spectroscopy has been used to detect benzyl alcohol and vegetable oil adulteration in clove oil.⁸⁹

The FTIR-ATR method was adopted for the detection of adulterants in clove oil.⁹⁰

4.5 Detection of adulterants in spices

Turmeric is an indispensable ingredient in the Asian kitchen and is used not only as a flavoring agent and seasoning but also as a medicine for treating colds, coughs, ulcers, and various diseases and inflammations. It is also known for its antiviral, antibacterial, antifungal, anticancer, and antioxidant properties. This is mainly due to its high curcumin content. Turmeric is mostly contaminated with metanil yellow, starch, Sudan red, lead chromate, and white turmeric. The long-term use of azo dye (metanil yellow, Sudan red) and lead chromate contaminated turmeric can cause tumors, carcinoma, neurotoxicity, liver diseases, etc.^{91–98}

FT-Raman and IR spectroscopies have proved to be potential tools for the detection of turmeric adulterants. Some of the vibrational spectra of IR spectroscopy are not present in Raman spectroscopy or *vice versa*, due to molecular symmetric and asymmetric properties. So, for a complete analysis, both IR and Raman spectra are recorded and analyzed. In the case of the white turmeric adulterant, the Raman spectra have no distinct spectral peaks thus the white and yellow turmeric adulteration mixture cannot be detected using the Raman spectroscopy method. However, IR spectroscopy shows a distinct spectral peak (1078 cm⁻¹) in the white and yellow turmeric mixture sample. The positions of the Raman and IR spectral peaks are used to identify the chemical component, and the spectral peak intensity is used to estimate the concentration of the component. The majority of the spectral peaks of turmeric and adulterants overlapped, but some of them which did not overlap act as a fingerprint of adulterants. These fingerprint peaks are utilized to detect

Table 4 Raman and IR-based spectroscopies for the detection of adulterants in olive oil

Adulterant	Method for detection	Chemometrics	Accuracy/LOD	Ref.
2% (volume percentage) of three other edible oils, <i>i.e.</i> , soybean oil, sunflower seed oil, and corn oil	Raman spectroscopy	External standard method (ESM)	99%	78
Soybean, corn, and raw olive residue (olive pomace) oils at 1, 5, and 10%, respectively	FT-Raman spectroscopy	PCR	91.3%	67
Peanut oil (PEO)	FTIR with ATR sampling	PCR, PLS-R, and LDA	LDA provided 92.3% correct classification for the calibration set; the lowest detection limit for PEO in EVOO was about 0.5% V/V	83



and analyze the additives in turmeric powder.^{91–98} Mostly, FT-Raman and FTIR show the 1406 cm⁻¹ and 1140 cm⁻¹ peaks, respectively, as non-overlapped spectral peaks in turmeric-metanil yellow adulterant samples, which is helpful to detect and analyze the adulterant components and their concentration levels. However, the FTIR fingerprint peak disappears at lower adulterant concentrations, due to the dominance of turmeric. FTIR and Raman show minimal detection levels of 5% and 2%, respectively, of the metanil adulteration.^{91,95} Erasmus *et al.* reported lead chromate adulterant detection through FT-Raman with the help of LDA and the PLSR algorithm. This method of detection achieved a 0.5% detection level of the lead chromate concentration level.⁹³ Dhakal *et al.* proposed Sudan red as an adulterant in turmeric through FTIR with the help of PLSR, and a 748 cm⁻¹ Sudan red fingerprint spectral peak was used to analyze the contamination.⁹⁴ Recently, NIR and FT-NIR have received attention for the detection of mentanil yellow and adulterants in turmeric; they have achieved a 0.3% concentration detection level with the help of PCR and PLSR chemometric algorithms.⁹⁷

Saffron (*Crocus sativus* L.), also known as red gold, is one of the most valuable and expensive spice crops in the world. The majority of consumers place a high value on saffron quality. As a result, assessing the purity and authenticity of a product as well as grading its quality is critical in the saffron trade. Saffron quality is determined by the amount of crocin, picrocrocin, and safranal, which are responsible for saffron's colour, flavor, and aroma.⁹⁹ The higher the concentration of these compounds, the higher the purity of saffron. However, due to high market demand, saffron contamination with various food colorants (tartrazine, sunset yellow, azorubine, quinoline-yellow, allura red, and Sudan II) and other plant materials (safflower, marigold, and turmeric) is very common.^{100–105} Eman Shawky *et al.* employed NIR spectroscopy with the PLSR method to record three different fingerprint spectral regions, which were utilized to achieve the 1% detection limit of different plant adulterants in saffron Table 5.¹⁰⁴

4.6 Detection of bacterial contamination in food

Foodborne diseases cause significant morbidity and mortality worldwide and hence are a significant public health

concern.^{4,107,108} The growth of microorganisms in food can change its appearance, flavour, odour, and other characteristics. Food quality is affected by the degradative processes caused by microorganisms, such as putrefaction, fermentation, and rancidity, as well as the substances they produce, such as pigments and slimes. Bacterial detection methods that have been used over the years are based on growing bacteria in culture media, isolating bacteria using selective media, and finally identifying bacteria based on morphological, physiological, and biochemical characteristics.¹⁰⁹ These conventional methods are labor-intensive and time-consuming, but they produce reliable results. Preliminary identification can take anywhere from 48 to 72 hours, and confirmation of the pathogen species can take up to a week. Because products may degrade while testing results are available, faster tools are necessarily required.^{110,111} Numerous rapid methods for detecting foodborne bacteria have been developed including immunological, nucleic acid-based, and vibrational spectroscopy. IR and Raman spectroscopy can be used to determine the characteristics of whole organisms based on the vibrational energy of the bonds within functional groups. Increasingly popular in microbiology, vibration spectroscopy offers the advantages of high speed, low cost, and high throughput.^{112,113} This method can be used as a rapid screening tool for contaminated liquids because it does not require culturing bacteria or amplifying certain genes. A cost-effective, simple, and accurate approach to microbial identification is needed by the food sector to maintain safe products and diagnose contamination.¹¹⁴ SERS is one of the important Raman methods for the analysis of bacterial contamination in food. For SERS to function, no complicated sample pre-treatment process is needed, and no expensive solvent or column is necessary. In comparison with HPLC and GC, SERS is carried out in a matter of seconds. The SERS approach has a lot of potential for the detection and identification of chemical and microbiological contaminants in foods, as well as food quality assessments.^{115,116} SERS can also be used to detect food-borne microbes, like *Salmonella*, *Staphylococcus aureus*, and *Escherichia coli*, through Raman fingerprinting. The second derivative transformation of the spectra resulted in unique FT-NIR characteristics in the 7000–4000 cm⁻¹ range, allowing PCA to reliably classify the data into

Table 5 Detection of adulterants in spices

Spices	Spectroscopy and chemometrics method	Adulterants	Accuracy/LOD	Ref.
Turmeric powder	FT-Raman	Metanil yellow	1%	91, 95
	FT-IR	Metanil yellow	5%	91, 95
	FT-Raman-PLSR	Lead chromate	0.5%	93
	FT-IR-PLSR	Sudan red	1%	94
	Handheld NIR	Metanil yellow	0.33%	97
	Handheld FT-IR	Corn starch	2%	106
		Metanil yellow	2%	
		Sudan dye IV	2%	
		Calendula	1.9%	105
		Safflower	2.1%	
Saffron	Diffuse reflectance infrared Fourier transform spectroscopy (DRIFTS) with PLSR–DA	Turmeric	1.0%	
		Plant materials	1%	104
	NIR with PLSR-VIP			



discrete clusters.¹¹⁷ Nicolaou and Goodacre were able to use FT-IR ATR and high throughput (HT) techniques as well as multivariate statistical methods like principal components-discriminant function analysis (PC-DFE) and PLSR, to analyse milk samples and estimate microbial loads accurately and rapidly, in under 30 seconds, with minimal sample preparation.¹¹⁸ Assaf *et al.* experimented with various methods for identifying *Salmonella* spp. in rice, oats, wheat, maize, poultry, and pork using Raman spectroscopy and following ISO standards.¹¹⁹ Wang *et al.* used spectral characteristics of the SERS tag to detect *S. aureus* and *S. enterica* from spinach and peanut butter.¹²⁰ Wu *et al.* employed silver nanorod arrays functionalized with vancomycin substrates to identify *Salmonella* spp., *Escherichia coli*, and *Staphylococcus epidermidis* in mung bean sprouts using Raman spectroscopy.¹²¹ The information acquired from the matrix, such as carbohydrates, proteins, and lipid, is highly linked to bacterial growth behavior, according to the SERS study, and chemical fluctuation during biofilm development can disclose whether Gram-positive or Gram-negative bacteria are present.¹²² Knowing their robustness, their distinctive ability to distinguish isolates based on changes in incubation time and medium, and their accuracy in identifying microbes down to the strain level, FTIR spectroscopy and matrix-assisted laser desorption/ionization time-of-flight (MALDI-TOF) mass spectrometry were examined for their utility and applicability in routine food-related microbiology diagnostics. Wenning *et al.* investigated the accuracy of detection of a large number of food-related bacteria from an in-house strain collection and different fresh isolates from regular food analysis.¹¹⁴ FTIR and MALDI-TOF produce accurate identification of bacterial strains. Yang and Irudayaraj investigated whole apple surfaces non-destructively using FT-Raman spectroscopy. The five separate strains of *E. coli* were distinguished with 100% accuracy. FT-Raman can be used for measuring food products directly from the sample or, by using the focused laser, for obtaining the spectra much quicker than FTIR.¹²³ Klein *et al.* have constructed and demonstrated, with an accuracy of 96.6 percent, a model for discriminating the examined microorganisms at the genus, species, and strain levels.¹⁴² This was accomplished despite the addition of diverse stress conditions and at times after the bacteria had been incubated. Zajac *et al.* successfully showed that rotting chicken meat may be detected using FTIR and Raman spectroscopies.¹²⁴ The Raman spectrum shows that the protein content (bands at 1655 and 1320 cm^{-1}) decreases during spoiling within 10 days, while the amounts of amino acids (band at 1657 cm^{-1}) increase.¹²⁴ By using SERS microscopy and silver nanoparticles paired with PCA, *L. monocytogenes* and *Salmonella* spp. were found in smoked salmon and ham, *Salmonella* spp. was found in eggs, and *Cronobacter* spp. was found in powdered infant formula and mixed herbs.¹²⁵ There are other optical techniques such as surface plasmon resonance (SPR) that have shown potential for the detection of adulterants and bacterial contamination in food (ref. 126–141), as discussed in detail in the ESI.†

5. Conclusion and future prospects

Raman and IR spectroscopies are simple and fast methods for the detection of food adulterants and bacteria. Different analysis methods have been adopted to discriminate between adulterants and bacteria in samples. While these techniques have shown huge potential for fast and accurate detection, they are not widely commercially deployed so far. One of the reasons for this could be the size of the optical apparatus. Nevertheless, in the recent past, handheld SERS-based Raman spectrophotometers have attracted huge attention, which provide fast and accurate results and avoid the size limitation concern.

While unexpected progress has been achieved over the years on SERS for analytical characterization studies including food analysis, many concerns have yet to be resolved for their widespread commercial application. As a technique, SERS is regarded as an irreproducible technique. It has been found that SERS substrates (metallic nanostructured surfaces supporting the effect) are responsible for the lack of reproducibility. Deviations in size, shape, and orientation of nanoparticles reduce the reproducible characteristics of the SERS substrate. Thus, considerable research has to be performed to develop high-quality SERS substrates with reproducible results. Furthermore, hand-held Raman spectrophotometers, as well as portable IR spectrophotometers, have huge potential and can bring lab-scale research to commercial scale products for the identification of adulterants. Next, the addition of big data analysis tools can help to further enhance the accuracy of these optical methods for the authentication of foods. Thus, the future of optical spectroscopy methods, especially Raman and IR, for biological and adulteration detection is bright, and many new analysis algorithms are expected to be discovered for further advancement in this field.

Author contributions

N. D. conceived the idea and planned the review manuscript. R. K. S., K. K., P. P., J. V., C. D., and N. D. wrote the initial draft of the manuscript. R. K. S. took the lead in writing the review. G. G., S. N. M., A. M., J. P. C., S. A. R. H., D. P. M., P. R. S., and A. K. S. critically reviewed the manuscript and gave their inputs to improve the manuscript draft. All the authors thoroughly read the final manuscript draft and gave permission for its submission.

Conflicts of interest

There are no conflicts to declare.

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