

Cite this: *Chem. Sci.*, 2022, 13, 4103

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 21st January 2022

Accepted 10th March 2022

DOI: 10.1039/d2sc00419d

rsc.li/chemical-science

Enantioselective [1,2]-Stevens rearrangement of thiosulfonates to construct dithio-substituted quaternary carbon centers†

Linfeng Hu, Jinzhao Li, Yongyan Zhang, Xiaoming Feng and Xiaohua Liu *

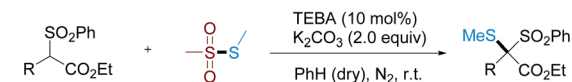
An enantioselective [1,2] Stevens rearrangement was realized by using chiral guanidine and copper(I) complexes. Bis-sulfuration of α -diazocarbonyl compounds was developed through using thiosulfonates as the sulfonylating agent. It was undoubtedly an atom-economic process providing an efficient route to access novel chiral dithioacetal derivatives, affording the corresponding products in good yields (up to 90% yield) and enantioselectivities (up to 96 : 4 er). A novel catalytic cycle was proposed to rationalize the reaction process and enantiocontrol.

Introduction

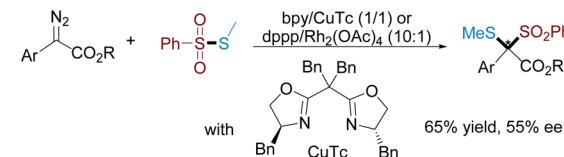
Organosulfur compounds, including thioethers and the oxidized derivatives such as sulfoxides or sulfones, have attracted significant attention owing to their wide occurrence in natural products and biologically active molecules.¹ Furthermore, they could also serve as useful reagents and ligands in organic synthesis.² Thiosulfonate is considered the line of defense against cyanide intoxication,³ and could be used for both sulfonylation and sulfenylation.⁴ For example, sulfonylation of α -sulfonyl carboxylic esters with *S*-methyl methanethiosulfonate resulted in dithioacetals containing two sulfur substitutions with different oxidation states at the same carbon center^{4a} (Scheme 1a). The carbene insertion reaction of α -diazocarbonyl compounds to the S–S bond of thiosulfonates could simultaneously introduce two different sulfur-groups into the stereogenic carbon center in the presence of the bpy/CuTc^{4b} or dppp/Rh₂(OAc)₄ (ref. 4c) catalyst (Scheme 1b). The reaction occurred via a [1,2]-Stevens rearrangement of sulfonium ylide,^{5,6} but enantiocontrol remains difficult. Xu's group explored the possibility of an asymmetric version by screening a variety of bisoxazoline ligands, and the highest ee of 55% was obtained.^{4b} Tang and co-workers demonstrated an efficient catalytic asymmetric [1,2]-Stevens rearrangement of sulfonium ylide using side-armed bisoxazoline/Cu(OTf)₂ as the catalyst (Scheme 1c),^{6a} but the stereogenic center lied in an O-based carbon center rather than the carbon of metal carbenoid species.⁷

Sigmatropic rearrangements of ylides represent a powerful set of reactions for the construction of heterocycles and interesting small molecules with functional groups. Different from asymmetric [2,3]-sigmatropic rearrangements and other transformations of onium ylides which have been well studied,^{8,9} the enantioselective [1,2]-Stevens rearrangement of onium ylides is limited and challenging.¹⁰ The intricacy mainly lies in the unclear mechanism where diradical pairs, concerted or ion pair propositions have an unpredictable influence on the

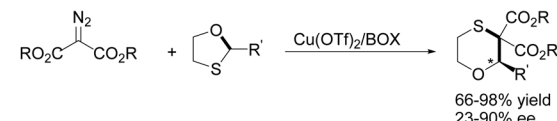
(a) Synthesis of dithioacetals via sulfonylation of α -sulfonyl carboxylic esters



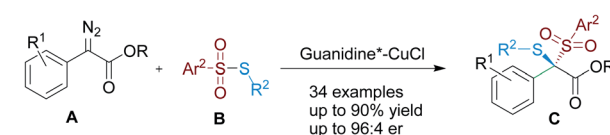
(b) Synthesis of dithioacetals via [1,2]-Stevens rearrangement from α -diazocarbonyl compounds



(c) Asymmetric intramolecular [1,2]-Stevens rearrangement of sulfur ylides



(d) Chiral guanidine/copper complex catalyzed asymmetric [1,2]-Stevens rearrangement of thiosulfonates (This work)



Scheme 1 Synthesis of dithioacetals from thiosulfonates and asymmetric catalytic [1,2]-Stevens rearrangements of sulfonium ylides.

Key Laboratory of Green Chemistry & Technology, Ministry of Education, College of Chemistry, Sichuan University, Chengdu 610064, China. E-mail: liuxh@scu.edu.cn

† Electronic supplementary information (ESI) available: ¹H, ¹³C{¹H} and ¹⁹F{¹H} NMR, HPLC spectra (PDF). X-ray crystallographic data for **D1**. CCDC **D1** 2113309. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/d2sc00419d

stereoselectivity, regardless of chiral metal-bound or free ylide intermediates.

Recently, our group has realized several asymmetric catalytic rearrangements^{9b,d,e} and carbene-insertion reactions^{12d,e,13a} based on α -diazo carbonyl compounds. Guanidines as a kind of multi-nitrogen-containing compound have versatile functionality as organocatalysts and ligands for asymmetric catalytic reactions.¹¹ Previously, we utilized chiral amino acids to construct bifunctional acyclic guanidine-amide compounds for organocatalysis^{12a-c} and metal complex catalysis.¹³ Their combination with copper(i) salt¹³ showed an obvious accelerating effect in carbenoid insertion of terminal alkynes^{13a} or HCN,^{13e} enabling enantioselective construction of both axial and center chirality. In view of these studies and the characteristics of readily modified subunits of guanidine-amides, we explored their application in [1,2]-Stevens rearrangement between thiosulfonates and α -diazoesters (Scheme 1d). Herein we describe the results of enantioselective synthesis of dithio-substituted carbon centers *via* [1,2]-Stevens rearrangement to simultaneously introduce two C–S bonds with a chiral guanidine/CuCl catalyst.

Results and discussion

Initially, we carried out the reaction with *S*-methyl benzene-sulfonothioate **B1** as the dithio-source, and (*R*)-2-pipecolic acid derived guanidine-amide **G5** and CuCl as the catalyst, and the selected results are listed in Table 1. The reaction with α -diazoesters **A1–A4** tethered to different ester groups could be performed with moderate to good yield and enantioselectivity in CH₂Cl₂ at 40 °C (Table 1, entries 1–4). Especially, 4-fluorobenzyl 2-diazo-2-phenylacetate **A4** could get better results with 75% yield and 91.5 : 8.5 er (entry 4). In comparison with the weak transformation in the presence of CuCl salt itself, the addition of guanidine also dramatically increased the reactivity. The use of other metal salts, such as CuBr, CuI, or CuTc, resulted in sluggish reactivity even with the assistance of guanidine **G5** (entries 5–7). The dithio-substituted racemic product **C4** was generated with low yield in the presence of metal salts such as Rh₂(OAc)₄, Pd₂(dba)₃, or AgOTf (entry 8; see the ESI† for details). Moreover, it was found that the steric hindrance of the substitution on the amide unit of guanidines had a dramatic influence on the enantioselectivity. Only the racemic product **C4** was isolated if aniline-based **G1** was used as the ligand (entry 9). The enantioselectivity gradually enhanced as the steric hindrance of substitutions at the 2,4,6-positions of anilines increased (**G1–G5**, entries 9–12 *vs.* entry 4). Using isopropyl groups instead of cyclohexyl groups at the *N*-substituents of the amidine part of guanidine afforded similar er values but slightly lower yields (**G6** *vs.* **G5**; entry 13 *vs.* entry 4). The amino acid backbone was also critical, for tetrahydroisoquinoline-3-carboxylic acid-based guanidine **G7** decreased the enantioselectivity a lot (entry 14). A screening of other chiral guanidines with variation at different subunits identified that the combination of CuCl and guanidine **G5** was the optimal catalyst for the reaction (see the ESI† for details). Increasing the catalyst loading to 15 mol% led to

Table 1 Optimization of reaction conditions^a

Entry	R; CuX; G*	C/Yield (%)	er
1	Me; CuCl; G5	C1 ; 66	77.5 : 22.5
2	Et; CuCl; G5	C2 ; 80	80 : 20
3	Bn; CuCl; G5	C3 ; 71	90 : 10
4	4-FC ₆ H ₄ CH ₂ ; CuCl; G5	C4 ; 75	91.5 : 8.5
5	4-FC ₆ H ₄ CH ₂ ; CuBr; G5	C4 ; trace	—
6	4-FC ₆ H ₄ CH ₂ ; CuI; G5	C4 ; N.R.	—
7	4-FC ₆ H ₄ CH ₂ ; CuTc; G5	C4 ; trace	—
8	4-FC ₆ H ₄ CH ₂ ; Rh ₂ (OAc) ₄ ; G5	C4 ; 32	50 : 50
9	4-FC ₆ H ₄ CH ₂ ; CuCl; G1	C4 ; 32	50 : 50
10	4-FC ₆ H ₄ CH ₂ ; CuCl; G2	C4 ; 76	76.5 : 23.5
11	4-FC ₆ H ₄ CH ₂ ; CuCl; G3	C4 ; 81	87 : 13
12	4-FC ₆ H ₄ CH ₂ ; CuCl; G4	C4 ; 52	86 : 14
13	4-FC ₆ H ₄ CH ₂ ; CuCl; G6	C4 ; 63	90 : 10
14	4-FC ₆ H ₄ CH ₂ ; CuCl; G7	C4 ; 78	75 : 25
15 ^b	4-FC ₆ H ₄ CH ₂ ; CuCl; G5	C4 ; 78	92 : 8

^a Unless otherwise noted, all reactions were carried out with **A** (0.20 mmol), **B1** (0.10 mmol), CuX/guanidine (1 : 1, 10 mol%), and 4 Å MS (60 mg) in CH₂Cl₂ (1.0 mL) at 40 °C for 7 h. Isolated yields. The er was determined by HPLC analysis on a chiral stationary phase.

^b CuCl/**G5** (1 : 1, 15 mol%). N.R. = no reaction.

a slightly higher yield and enantioselectivity (entry 15, 78% yield with 92 : 8 er).

With the optimized reaction conditions in hand, the substrate scope was then explored (Table 2). The established optimal reaction conditions exhibited similar tolerance to phenyl substituted α -diazo compounds with substituted benzyl esters (**C5–C8**), or more sterically hindered benzhydryl ester (**C20**). However, the electronic nature of substitution at the *ortho*-, *meta*-, or *para*-positions of the aromatic rings of α -diazoesters had a dramatic influence on the enantioselectivity. Products with electron-deficient aryl groups (**C12–C18**, and **C20–C28**) were universally generated in moderate to good yields (65–90%) with higher enantioselectivities (90 : 10–96 : 4 er), nevertheless, the products containing electron-donating substituents (**C9–C11**, **C29** and **C30**) were accessible with high yields (75–86%) but with decreased enantioselectivities (only 58.5 : 41.5 er for 4-MeO substituted **C30**). The electronic effect of substitutions on the enantioselectivity was uniform no matter what the ester group, implying that the stability of the sulfonium ylide intermediate was significant to the enantio-control. Additionally, 2-naphthyl substituted dithioketal **C19** was obtained with a decreased er value due to the steric



Table 2 Substrate scope for thiosulfonates and α -diazoesters^a

Product	Yield (%)	er
C4: X = 4-F	78	92:8
C5: X = 4-Cl	71	90:10
C6: X = 4-Br	83	90.5:9.5
C7: X = 3,5-F ₂	90	91.5:8.5
C8: X = 3,5-Cl ₂	90	92.5:7.5
C9: R ¹ = 3-Me	78	90:10
C10: R ¹ = 4-Me	78	89:11
C11: R ¹ = 3,4-Me ₂	75	80:20
C12: R ¹ = 3-F	83	90:10
C13: R ¹ = 4-F	81	90.5:9.5
C14: R ¹ = 3-Cl	71	90:10
C15: R ¹ = 4-Cl	83	91.5:8.5
C16: R ¹ = 3,4-Cl ₂	68	88:12
C17: R ¹ = 3-Br	72	90:10
C18: R ¹ = 4-Br	89	90:10
C19	78	80:20
C20: R ¹ = H	81	95.5:4.5
C21: R ¹ = 3-F	80	95.5:4.5
C22: R ¹ = 4-F	70	94:6
C23: R ¹ = 3-Cl	74	95.5:4.5
C24: R ¹ = 4-Cl	74	94:6
C25: R ¹ = 3-Br	90	95.5:4.5
C26: R ¹ = 4-Br	75	94:6
C27: R ¹ = 4-I	83	91:9
C28: R ¹ = 3-CF ₃	65	96:4
C29: R ¹ = 4-Me	86	83.5:16.5
C30: R ¹ = 4-MeO	84	58.5:41.5
C31: Ar ² = 4-MeC ₆ H ₄	78	89:11
C32: Ar ² = 4-FC ₆ H ₄	72	90.5:9.5
C33: Ar ² = 4-ClC ₆ H ₄	90	90:10
C34: Ar ² = 4-BrC ₆ H ₄	81	88:12
C35: Ar ² = 2-naphthyl	89	89:11
C36	81	73:27
C37	47	62:38

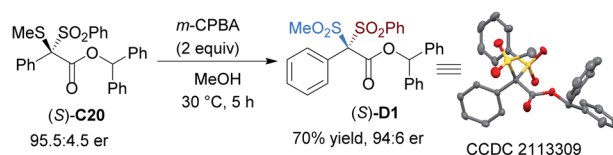
^a Unless otherwise noted, all reactions were performed with **A** (0.20 mmol), **B** (0.10 mmol), **G5**/**CuCl** (1 : 1, 15 mol%) in CH₂Cl₂ (0.1 mL) at 40 °C. Isolated yield. The er was determined by HPLC analysis on a chiral stationary phase.

hindrance. Unfortunately, the reaction of ethyl 2-diazo-propanoate delivered the desired product with 42% yield as a racemate, and other alkyl α -diazoesters resulted in poor reactivity under the current conditions (see the ESI† for details).

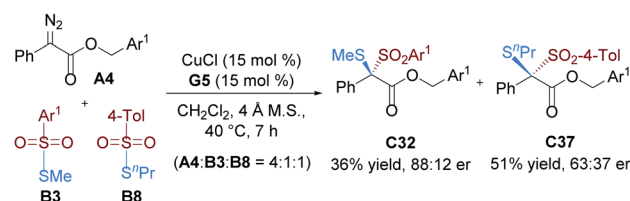
Subsequently, we turned our attention to the scope of thiosulfonates. Not surprisingly, benzenesulfonyl groups with substitutions, such as 4-Me, 4-F, 4-Cl, and 4-Br, were tolerated well in the reaction to yield the corresponding products smoothly (**C31**–**C34**, 72–90% yields, and 88 : 12–90.5 : 9.5 er). The product **C35** deriving from 2-naphthyl thiosulfonate could also be generated in 89% yield with 89 : 11 er. However, *S*-ethyl and *S*-propyl-substituted thiosulfonates delivered the related products **C36** and **C37** with unsatisfactory results, highlighting the steric bias of sulfur substitution on enantioselectivity.

The oxidation of the dithioketal derivative **C20** led to the construction of the derivative **D1** (Scheme 2) with a disulfonyl substituted quaternary center, whose absolute configuration was determined to be *S* based on X-ray single crystal analysis.¹⁴ Accordingly, the absolute configuration of the product **C20** was assigned as the same. Next, we carried out a cross-over experiment by subjecting α -diazoester **A4** and two kinds of thiosulfonate to the same catalytic system. It was found that two direct rearrangement products were detected, and there was no scrambling of the substituents in dithioketals. Primary theoretical calculations about the natural bond orbital charges of thiosulfonate **B1** showed that the sulfonyl center has a strong positive charge while the other thio-atom has a negative charge (Scheme 3), indicating that the homolytic cleavage of the S–S bond to generate a diradical intermediate is difficult. We proposed that the reaction is likely to proceed in a concerted ionic pathway. The above results are in consist with the electronic effect of aryl substituents of diazoesters on the reactivity. The electron-donating aryl group is advantageous for the final rearrangement to increase the yield but disadvantageous for the initial ylide formation step, which might determine the enantioselectivity.

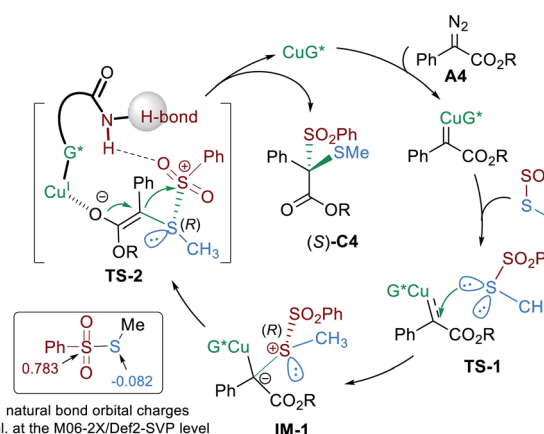
a) Transformation of the product into optically active disulfonyl derivative



b) cross-over experiment



Scheme 2 Transformation of the products and the control experiment.



Scheme 3 Possible asymmetric catalytic process.

In connection with the coordination ability of the guanidine functional group, we propose that it might act as a bifunctional ligand to form a chiral copper complex catalyst. As shown in Scheme 3, upon the addition of α -diazoester **A**, chiral copper carbenoid species are generated that undergo nucleophilic attack by thiosulfonate **B**. Different from the oxonium ylides, there is a S-based central chirality in the sulfonium ylide intermediate, generating from the discrimination of the two lone pair electrons of thiosulfonate which is extremely challenging, similar to asymmetric Sommelet–Houser rearrangement and 2,3-Stevens rearrangements.^{9b,d} Moreover, if the reaction follows a stepwise 1,2-sigmatropic rearrangement or free ylide intermediate, the stereo-chemistry at sulfur disappears or racemizes, leading to a new, difficult to reconstruct stereogenic center at carbon.

We rationalize that the steric hindrance between the sulfonate group of thiosulfonate **B** and ester group of α -diazoester **A** would lead to selective attack of sulfur *via* **TS-1** to set up the chiral sulfonium ylide **IM-1**. Meanwhile, the hydrogen-bond between the amide of **G5** and sulfonate unit might also direct the approach of thiosulfonate. Next, in view of the instability of the SO_2Ph cation intermediate and the cross-over experiment, a stepwise process is ruled out. Thus, a concerted rearrangement occurs with the shift of sulfonate from the *Si*-face of the enolate *via* **TS-2** during which the chiral copper-bond enolate might prevent racemization of the sulfur center and assist the approach of the sulfonyl from the rear to construct the second C–S bond. Finally, the desired (*S*)-dithioketal derivative **C** is generated to recycle the chiral copper catalyst.

Conclusions

In summary, we have reported an efficient asymmetric catalytic [1,2]-Stevens rearrangement by using a type of chiral guanidine-copper(i) complex. The catalytic system enabled the direct introduction of two C–S bonds in an enantioselective manner, yielding various chiral dithioketal derivatives in good yield and enantioselectivity, shedding a shaft of light on the asymmetric 1,2-sigmatropic rearrangement. More research about the [1,2]-Stevens rearrangement and the application of the chiral guanidine/copper(i) catalytic system are undergoing.

Data availability

Further details of experimental procedure, ^1H , $^{13}\text{C}\{^1\text{H}\}$ and $^{19}\text{F}\{^1\text{H}\}$ NMR, HPLC spectra, X-ray crystallographic data for **D1** are available in the ESI.†

Author contributions

L. F. H. performed the experiments. J. Z. L. repeated data. X. M. F. and X. H. L. supervised the project. X. M. F., X. H. L., Y. Y. Z. and L. F. H. co-wrote the manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

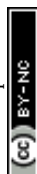
We appreciate the National Natural Science Foundation of China (21625205) and Sichuan University (2020SCUNL204) for financial support. The authors also acknowledge Dr Yuqiao Zhou (Sichuan University) for his assistance in X-ray crystallographic analysis and calculations.

Notes and references

- For selected reviews and examples, see: (a) S. W. Kaldor, V. J. Kalish, J. F. Davies, B. V. Shetty, J. E. Fritz, K. Appelt, J. A. Burgess, K. M. Campanale, N. Y. Chirgadze, D. K. Clawson, B. A. Dressman, S. D. Hatch, D. A. Khalil, M. B. Kosa, P. P. Lubbehusen, M. A. Muesing, A. K. Patick, S. H. Reich, K. S. Su and J. H. Tatlock, *J. Med. Chem.*, 1997, **40**, 3979; (b) T. Nakazawa, J. Z. Xu, T. Nishikawa, T. Oda, A. Fujita, K. Ukai, R. E. P. Mangindaan, H. Rotinsulu, H. Kobayashi and M. Namikoshi, *J. Nat. Prod.*, 2007, **70**, 439; (c) A. Y. Kovalevsky, A. K. Ghosh and I. T. Weber, *J. Med. Chem.*, 2008, **51**, 6599; (d) A. Z. Rys, I. A. Abu-Yousef and D. N. Harpp, *Tetrahedron Lett.*, 2008, **49**, 6670; (e) C. S. Jiang, W. E. G. Müller, H. C. Schröder and Y. W. Guo, *Chem. Rev.*, 2012, **112**, 2179; (f) E. K. Davison and J. Sperry, *J. Nat. Prod.*, 2017, **80**, 3060; (g) D. H. Scharf, J. D. Dworschak, P. Chankhamjon, K. Scherlach, T. Heinekamp, A. A. Brakhage and C. Hertweck, *ACS Chem. Biol.*, 2018, **13**, 2508; (h) E. O. N'Nang, G. Bernadat, E. Mouray, B. Kumulungui, P. Grellier, E. Poupon, P. Champy and M. A. Beniddir, *Org. Lett.*, 2018, **20**, 6596; (i) E. O. N'Nang, P. L. Pogam, T. N. Mba, C. S. Obiang, E. Mouray, P. Grellier, B. Kumulungui, P. Champy and M. A. Beniddir, *J. Nat. Prod.*, 2021, **84**, 1409.
- For selected reviews and examples, see: (a) M. Mellah, A. Voituriez and E. Schulzas, *Chem. Rev.*, 2007, **107**, 5133; (b) G. Sipos, E. E. Drinkel and R. Dorta, *Chem. Soc. Rev.*, 2015, **44**, 3834; (c) F. L. Lam, F. Y. Kwong and A. S. C. Chan, *Chem. Commun.*, 2010, **46**, 4649; (d) W. Y. Qi, T. S. Zhu and M. H. Xu, *Org. Lett.*, 2011, **13**, 3410; (e) Q. Y. Zhao, S. K. Li, K. X. Huang, R. Wang and X. M. Zhang, *Org. Lett.*, 2013, **15**, 4014; (f) Y. Li and M. H. Xu, *Chem. Commun.*, 2014, **50**, 3771; (g) B. J. Wang, C. Lin, Y. Liu, Z. L. Fan, Z. X. Liu and Y. H. Zhang, *Org. Chem. Front.*, 2015, **2**, 973; (h) Y. Wei, L. Q. Lu, T. R. Li, B. Feng, Q. Wang, W. J. Xiao and H. Alper, *Angew. Chem., Int. Ed.*, 2016, **55**, 2200; (i) I. V. Chuchelkin, K. N. Gavrilov, V. K. Gavrilov, S. V. Zheglov, I. D. Firsin, A. M. Perepukhov, A. V. Maximychev, N. E. Borisova, I. A. Zamilatskov, V. S. Tyurin, C. Dejoie, V. V. Chernyshev, V. S. Zimarev and N. S. Goulioukina, *Organometallics*, 2021, **40**, 3645.
- (a) J. P. Weidner and S. S. Block, *J. Med. Chem.*, 1964, **7**, 671; (b) E. Block, *Angew. Chem., Int. Ed. Engl.*, 1992, **31**, 1135; (c) M. L. Birsá, M. Cherkinsky and S. Braverman, *Tetrahedron Lett.*, 2002, **43**, 9615; (d) M. L. Alcaraz, S. Atkinson, P. Cornwall, A. C. Foster, D. M. Gill, L. A. Humphries, P. S. Keegan, R. Kemp, E. Merifield, R. A. Nixon, A. J. Noble, D. O'Beirne, Z. M. Patel, J. Perkins, P. Rowan,



- P. Sadler, J. T. Singleton, J. Tornos, A. J. Watts and I. A. Woodland, *Org. Process Res. Dev.*, 2005, **9**, 555; (e) G. G. Liang, M. C. Liu, J. X. Chen, J. C. Ding, W. X. Gao and H. Y. Wu, *Chin. J. Chem.*, 2012, **30**, 1611; (f) Y. Zheng, F. L. Qing, Y. G. Huang and X. H. Xu, *Adv. Synth. Catal.*, 2016, **358**, 3477; (g) Z. Y. Mo, T. R. Swaroop, W. Tong, Y. Z. Zhang, H. T. Tang, Y. M. Pan, H. B. Sun and Z. F. Chen, *Green Chem.*, 2018, **20**, 4428; (h) G. Q. Li, Z. Y. Gan, K. X. Kong, X. M. Dou and D. S. Yang, *Adv. Synth. Catal.*, 2019, **361**, 1808; (i) Y. F. Lv, J. Y. Luo, Y. C. Ma, Q. Dong and L. He, *Org. Chem. Front.*, 2021, **8**, 2461.
- 4 For selected examples, see: (a) B. Wladislaw, L. Marzorati, N. F. Claro Jr and C. D. Vitta, *Synthesis*, 1997, **4**, 420; (b) H. R. Yuan, T. Nuligonda, H. Y. Gao, C. H. Tunga and Z. H. Xu, *Org. Chem. Front.*, 2018, **5**, 1371; (c) C. Q. Rao, S. Y. Mai and Q. L. Song, *Chem. Commun.*, 2018, **54**, 5964; (d) J. Li, S. Y. Wang and S. J. Ji, *J. Org. Chem.*, 2019, **84**, 16147; (e) P. Mampuy, C. R. McElroy, J. H. Clark, R. V. A. Orru and B. U. W. Maesa, *Adv. Synth. Catal.*, 2020, **362**, 3; (f) J. Y. Luo, M. Z. Lin, L. F. Wu, Z. H. Cai, L. He and G. F. Du, *Org. Biomol. Chem.*, 2021, **19**, 9237; (g) J. L. Qi, F. Wei, S. Huang, C. H. Tung and Z. H. Xu, *Angew. Chem., Int. Ed.*, 2021, **60**, 4561.
- 5 For selected examples, see: (a) M. Ioannou, M. J. Porter and F. Saez, *Chem. Commun.*, 2002, 346; (b) K. K. Ellis-Holder, B. P. Peppers, A. Yu. Kovalevsky and S. T. Diver, *Org. Lett.*, 2006, **8**, 2511; (c) A. J. P. Mortimer, A. E. Aliev, D. A. Tocher and M. J. Porter, *Org. Lett.*, 2008, **10**, 5477; (d) X. F. Xu, C. Li, M. T. Xiong, Z. H. Tao and Y. J. Pan, *Chem. Commun.*, 2017, **53**, 6219; (e) R. Fan, C. Tan, Y. G. Liu, Y. Wei, X. W. Zhao, X. Y. Liu, J. J. Tan and H. Yoshida, *Chin. Chem. Lett.*, 2021, **32**, 299.
- 6 For selected examples, see: (a) J. P. Qu, Z. H. Xu, J. Zhou, C. L. Cao, X. L. Sun, L. X. Dai and Y. Tang, *Adv. Synth. Catal.*, 2009, **351**, 308; (b) Z. Song, Y. Z. Wu, T. Xin, C. Jin, X. A. Wen, H. B. Sun and Q. L. Xu, *Chem. Commun.*, 2016, **52**, 6079; (c) R. R. Lin, L. Y. Cao and F. G. West, *Org. Lett.*, 2017, **19**, 552; (d) X. B. Xu, Z. H. Lin, Y. Y. Liu, J. Guo and Y. He, *Org. Biomol. Chem.*, 2017, **15**, 2716; (e) Z. Yang, Y. J. Guo and R. M. Koenigs, *Chem. Commun.*, 2019, **55**, 8410.
- 7 For reviews: (a) T. Ye and M. A. McKerver, *Chem. Rev.*, 1994, **94**, 1091; (b) M. P. Doyle, M. A. McKerver and T. Ye, in *Modern Catalytic Methods for Organic Synthesis with Diazo Compounds*, Wiley, New York, 1998; (c) Z. H. Zhang and J. B. Wang, *Tetrahedron*, 2008, **64**, 6577; (d) H. M. L. Davies and D. Morton, *Chem. Soc. Rev.*, 2011, **40**, 1857; (e) S. F. Zhu and Q. L. Zhou, *Acc. Chem. Res.*, 2012, **45**, 1365; (f) X. Guo and W. H. Hu, *Acc. Chem. Res.*, 2013, **46**, 2427; (g) A. Ford, H. Miel, A. Ring, C. N. Slattery, A. R. Maguire and M. A. McKerver, *Chem. Rev.*, 2015, **115**, 9981; (h) N. R. Candeias, R. Paterna and P. M. P. Gois, *Chem. Rev.*, 2016, **116**, 2937; (i) Y. Xia, D. Qiu and J. B. Wang, *Chem. Rev.*, 2017, **117**, 13810.
- 8 For selected examples about asymmetric rearrangements of onium ylides, see: (a) W. Kirmse and M. Kapps, *Chem. Ber.*, 1968, **101**, 994; (b) M. P. Doyle, W. H. Tamblin and V. J. Bagheri, *J. Org. Chem.*, 1981, **46**, 5094; (c) M. Ma, L. L. Peng, C. K. Li, X. Zhang and J. B. Wang, *J. Am. Chem. Soc.*, 2005, **127**, 15016; (d) D. Yadagiri and P. Anbarasan, *Chem. Eur. J.*, 2013, **19**, 15115; (e) T. H. West, S. S. M. Spoehrle, K. Kasten, J. E. Taylor and A. D. Smith, *ACS Catal.*, 2015, **5**, 7446; (f) X. F. Xu, C. Li, Z. H. Tao and Y. J. Pan, *Green Chem.*, 2017, **19**, 1245; (g) D. Kaldre, B. Maryasin, D. Kaiser, O. Gajsek, L. Gonzalez and N. Maulide, *Angew. Chem., Int. Ed.*, 2017, **56**, 2212; (h) H. Kim, J. Jang and S. Shin, *J. Am. Chem. Soc.*, 2020, **142**, 20788.
- 9 For selected examples about asymmetric rearrangements of onium ylides, see: (a) Z. K. Zhang, Z. Sheng, W. Z. Yu, G. J. Wu, R. Zhang, W. D. Chu, Y. Zhang and J. B. Wang, *Nat. Chem.*, 2017, **9**, 970; (b) X. B. Lin, Y. Tang, W. Yang, F. Tan, L. L. Lin, X. H. Liu and X. M. Feng, *J. Am. Chem. Soc.*, 2018, **140**, 3299; (c) D. Kaldre, I. Klose and N. Maulide, *Science*, 2018, **361**, 664; (d) X. B. Lin, W. Yang, W. K. Yang, X. H. Liu and X. M. Feng, *Angew. Chem., Int. Ed.*, 2019, **58**, 13492; (e) W. Yang, X. B. Lin, Y. Y. Zhang, W. D. Cao, X. H. Liu and X. M. Feng, *Chem. Commun.*, 2020, **56**, 10002; (f) S. S. Li and J. B. Wang, *J. Org. Chem.*, 2020, **85**, 12343; (g) Z. Y. Liu, X. J. Jin and Y. F. Dang, *ACS Catal.*, 2021, **11**, 691; (h) S. X. Dong, X. H. Liu and X. M. Feng, *Acc. Chem. Res.*, 2022, **55**, 415.
- 10 For selected examples, see: (a) F. G. West, B. N. Naidu and R. W. Tester, *J. Org. Chem.*, 1994, **59**, 6892; (b) K. W. Glaeske and F. G. West, *Org. Lett.*, 1999, **1**, 31; (c) G. Ghigo, S. Cagnina, A. Maranzana and G. Tonachini, *J. Org. Chem.*, 2010, **75**, 3608; (d) B. Xu and U. K. Tambar, *J. Am. Chem. Soc.*, 2016, **138**, 12073; (e) B. Xu, J. A. Gartman and U. K. Tambar, *Tetrahedron Lett.*, 2017, **73**, 4150; (f) D. Baidilov, *Synthesis*, 2020, **52**, 21; (g) V. N. Nair, V. Kojasoy, C. J. Laconsay, W. Y. Kong, D. J. Tantillo and U. K. Tambar, *J. Am. Chem. Soc.*, 2021, **143**, 9016; (h) C. Empel, S. Jana and R. M. Koenigs, *Synthesis*, 2021, **53**, 4567.
- 11 For selected reviews, see: (a) D. S. Leow and C. H. Tan, *Synlett*, 2010, **2010**, 1589; (b) L. L. Zong and C. H. Tan, *Acc. Chem. Res.*, 2017, **50**, 842; (c) S. X. Dong, X. M. Feng and X. H. Liu, *Chem. Soc. Rev.*, 2018, **47**, 8525; (d) X. Y. Cui, C. H. Tan and D. S. Leow, *Org. Biomol. Chem.*, 2019, **17**, 4689; (e) J. Francos and V. Cadierno, *Dalton Trans.*, 2019, **48**, 9021; (f) L. Steuer, E. Kaifer and H. J. Himmel, *Dalton Trans.*, 2021, **50**, 9467.
- 12 (a) S. X. Dong, X. H. Liu, X. H. Chen, F. Mei, Y. L. Zhang, B. Gao, L. L. Lin and X. M. Feng, *J. Am. Chem. Soc.*, 2010, **132**, 10650; (b) S. X. Dong, X. H. Liu, Y. L. Zhang, L. L. Lin and X. M. Feng, *Org. Lett.*, 2011, **13**, 5060; (c) S. X. Dong, X. H. Liu, Y. Zhu, P. He, L. L. Lin and X. M. Feng, *J. Am. Chem. Soc.*, 2013, **135**, 10026; (d) Y. Zhu, X. H. Liu, S. X. Dong, Y. H. Zhou, W. Li, L. L. Lin and X. M. Feng, *Angew. Chem., Int. Ed.*, 2014, **53**, 1636; (e) F. Tan, X. H. Liu, X. Y. Hao, Y. Tang, L. L. Lin and X. M. Feng, *ACS Catal.*, 2016, **6**, 6930.
- 13 (a) Y. Tang, Q. G. Chen, X. H. Liu, G. Wang, L. L. Lin and X. M. Feng, *Angew. Chem., Int. Ed.*, 2015, **54**, 9512; (b) Q. G. Chen, Y. Tang, T. Y. Huang, X. H. Liu, L. L. Lin and



- X. M. Feng, *Angew. Chem., Int. Ed.*, 2016, **55**, 5286; (c) Y. Tang, J. Xu, J. Yang, L. L. Lin, X. M. Feng and X. H. Liu, *Chem*, 2018, **4**, 1658; (d) S. S. Guo, P. Dong, Y. S. Chen, X. M. Feng and X. H. Liu, *Angew. Chem., Int. Ed.*, 2018, **57**, 16852; (e) S. Ruan, X. Zhong, Q. G. Chen, X. M. Feng and X. H. Liu, *Chem. Commun.*, 2020, **56**, 2155.
- 14 CCDC 2113309 (D1) contains the supplementary crystallographic data for this paper.†

