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Terahertz spectroscopy of the helium endofullerene He@C₆₀

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We studied the quantized translational motion of single He atoms encapsulated in molecular cages by terahertz absorption. The temperature dependence of the THz absorption spectra of ³He@C₆₀ and ⁴He@C₆₀ crystal powder samples was measured between 5 and 220 K. At 5 K there is an absorption line at 96.8 cm⁻¹ (2.90 THz) in ³He@C₆₀ and at 81.4 cm⁻¹ (2.44 THz) in ⁴He@C₆₀, while additional absorption lines appear at higher temperature. An anharmonic spherical oscillator model with a displacement-induced dipole moment was used to model the absorption spectra. Potential energy terms with powers of two, four and six and induced dipole moment terms with powers one and three in the helium atom displacement from the fullerene cage center were sufficient to describe the experimental results. Excellent agreement is found between potential energy functions derived from measurements on the ³He and ⁴He isotopes. One absorption line corresponds to a three-quantum transition in ⁴He@C₆₀, allowed by the anharmonicity of the potential function and by the non-linearity of the dipole moment in He atom displacement. The potential energy function of icosahedral symmetry does not explain the fine structure observed in the low temperature spectra.

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1 Introduction

The highly symmetric fullerene, C₆₀, has attracted considerable attention since its discovery in 1985.¹ One of the fascinating features is its ability to hold an atom or small molecule within a hollow molecular cavity, forming a complex known as an endofullerene. The discovery of the first endofullerene, La@C₆₀, occurred soon after by the same group of researchers, when they were able to capture lanthanum by a laser ablation technique.² Endofullerenes provide a rare opportunity to examine the behavior of guest species under conditions that have never been studied before such as the confinement of guest species inside a nano-scale cage which is non-polar, homogeneous and in the case of C₆₀, highly symmetric.^{3,4}

Different methods have been used to synthesize endofullerenes. Endofullerenes may be synthesized by the arc discharge method in which carbon is evaporated at high temperature in the presence of a metal⁵ or an inert gas.^{6,7} Insertion of atoms into the fullerene cage is possible by exposing the fullerene to high temperatures and high pressures of inert gas⁸ or by ion bombardment.^{9–11} However, these methods are not suited to capture small molecules and the production yield is small. In a

major development, the group of Komatsu¹² synthesized H₂@C₆₀ endofullerenes by a process known as molecular surgery.¹³ Following in the footsteps of Komatsu, the group of Murata successfully encapsulated a water molecule inside C₆₀.¹⁴ Molecular surgery has been the most successful process to produce high-yield endofullerenes until today. A large number of endofullerenes are now available in macroscopic quantities, such as¹² and its isotopologues,¹⁵ H₂O@C₆₀,¹⁴ HF@C₆₀¹⁶ and CH₄,¹⁷ and atoms like Ar,¹⁸ He^{19,20} and Ne.²⁰

Endofullerenes are of great interest due to quantum effects which are more evident at cryogenic temperatures. The encapsulation isolates atoms and molecules from each other preventing them from forming liquid or solid condensates even at the lowest temperatures. This provides a unique opportunity to study their internal degrees of freedom at the lowest temperatures. The translational motion becomes quantized due to the confining potential. The confinement also prompts the coupling of translational and rotational motions of the encapsulated molecule. The combination of small molecular mass and tight confinement inside the nano cage gives rise to a discrete and well separated translation-rotational energy level structure. The most studied endohedral complex is H₂@C₆₀, which laid the basis for understanding the quantum dynamics of the isolated systems.^{4,21} Different spectroscopic techniques have been used to demonstrate the effect of the highly symmetric trapping potential on the quantum dynamics of the hydrogen

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molecule, including infrared spectroscopy (IR)^{22–24} and inelastic neutron scattering (INS).^{25–28}

There is a growing interest in extending spectroscopic studies to noble gas endofullerenes. The research interest in noble gas endofullerenes is evident from the theoretical studies that have been done so far.^{29–36} The incarceration of large noble gas atoms results in the structural and electronic distortion of which has been examined by IR and Raman,³⁴ NMR,³⁷ X-ray³⁸ and electronic spectroscopy.³⁹

The first experimental evidence of a noble gas endofullerene was He@C₆₀. It was spotted by mass spectrometry when the 4 atom was incorporated in C₆₀ as the highly accelerated C₆₀⁺ ions struck with helium gas⁹ and later found in fullerenes produced by arc discharge in the He gas.⁶ Despite that, and the 2010 synthesis of He@C₆₀ by molecular surgery,¹⁹ the He endofullerene potential energy function study was restricted to theoretical explanations^{31,33,40} and the experimental data were inadequate for supporting the available information. Recently, the translational energies of He@C₆₀ were determined by inelastic neutron scattering and THz spectroscopy studies and the experimentally derived potential was compared to estimates from quantum chemistry calculations and from sums of empirical two-body potentials.⁴¹

Here, as compared to ref. 41, we present a more detailed temperature-dependence study of THz absorption by endohedral 3 and 4 isotopes between 5 and 220 K. Also, we obtained a better fit of line intensities by including an r^3 term in the expansion of the dipole moment in the helium atom displacement r . The temperature dependence and the new fit of line intensities confirms the assignment of the He atom translational energy levels and the accuracy of the derived potential energy surface reported in our previous paper.⁴¹

2 Theory

The potential energy surface of the He atom stems from its interaction with the interior of the C₆₀ molecule. Although the molecule has icosahedral symmetry we approximate it by spherical symmetry. Within this approximation we may ignore the coupling of the C₆₀ rotations and librations to the He motion. Since the masses of C₆₀ and He are different by two orders of magnitude, $M_{C60} \gg M_{He}$, we assume that the C₆₀ center of the mass position is not affected by the motion of C₆₀ the He atom. Also, we ignore the effect of translational motion of in the crystal lattice and its molecular vibrations. Thus, in this model the He atom moves in a spherically symmetric rigid potential well. This motion can be described by the harmonic spherical oscillator model^{42–44} where the coordinate $\mathbf{r}$ is the displacement of He center of mass from the C₆₀ center of mass. To obtain agreement with the observed THz spectra, we add anharmonic corrections to the harmonic potential function. This makes the separation of energy levels non-equidistant.

In general, a free He atom does not absorb electromagnetic radiation at THz frequencies due to the lack of an electric dipole moment. However, similar to the encapsulated hydrogen

molecule,^{15,22,23} He atoms acquire a dipole moment from the interaction with the inner surface of C₆₀. This interaction is modulated by the translational motion of the He atom causing the translational modes to become THz-active.

2.1 Anharmonic spherical oscillator

The Hamiltonian $\hat{H}$ for one helium trapped in a spherical potential well is given as

$$\hat{H} = \hat{H}_0 + \hat{V}', \quad (1)$$

where

$$\hat{V}' = V_4 r^4 + V_6 r^6 \quad (2)$$

is the perturbation (anharmonic term) added to the harmonic spherical oscillator Hamiltonian⁴⁴

$$\hat{H}_0 = \frac{p^2}{2M_{He}} + V_2 r^2. \quad (3)$$

The harmonic Hamiltonian has eigenstates

$$|n, \ell, m\rangle = R_{n\ell}(r) Y_{\ell m}(\theta, \phi), \quad (4)$$

where $r \in \{0, \infty\}$, $\theta \in \{0, \pi\}$ and $\phi \in \{0, 2\pi\}$. The angular part of each wavefunction m is an ortho-normalized spherical harmonic,

$$\int_0^{2\pi} d\phi \int_0^\pi \sin \theta d\theta [Y_{\ell m'}(\theta, \phi)]^* Y_{\ell m}(\theta, \phi) = \delta_{\ell\ell'} \delta_{mm'}, \quad (5)$$

with the Condon and Shortley phase convention⁴⁵

$$Y_{\ell m}^*(\theta, \phi) = Y_{\ell m}(\theta, -\phi) = (-1)^m Y_{\ell, -m}(\theta, \phi). \quad (6)$$

The radial wavefunction is⁴²

$$R_{n\ell}(r) = C(n, \ell, \xi) e^{-\frac{\xi r^2}{2}} \left(\xi r^2 \right)^{\frac{\ell}{2}} L_{n-\ell}^{\ell+\frac{1}{2}} \left[\xi r^2 \right], \quad (7)$$

where $L_n^\alpha[x]$ is generalised Laguerre polynomial and $\xi = M_{He} \omega_T \hbar^{-1}$. The radial part of the wavefunction is normalised by the constant

$$C(n, \ell, \xi) = \sqrt{\frac{2 \left(\frac{n-\ell}{2} \right)!}{\left(\frac{n+\ell+1}{2} \right)!}} \xi^{\frac{3}{4}}. \quad (8)$$

The eigenenergies of the harmonic Hamiltonian (3) are

$$E_n = \hbar \omega_T \left(n + \frac{3}{2} \right), \quad (9)$$

where the angular frequency is

$$\omega_T = \sqrt{\frac{2V_2}{M_{He}}}. \quad (10)$$

The problem can be solved by diagonalizing the Hamiltonian (1) using the basis set of functions (4) where $n \in \{0, 1, \dots, n_{\max}\}$. The angular momentum quantum number ℓ is given by $\ell \in \{0, 2, \dots, n\}$ for even n and $\ell \in \{1, 3, \dots, n\}$ for odd n .



It is practical to use the reduced matrix elements, $\langle \ell_f \| T_k \| \ell_i \rangle$, of the spherical tensor operator T_{kq} of rank k , $q \in \{-k, -k+1, \dots, k\}$, which are independent of m and q .⁴⁶ The matrix element of T_{kq} is

$$\langle \tau_f, \ell_f, m_f | T_{kq} | \tau_i, \ell_i, m_i \rangle = (-1)^{\ell_f - m_f} \begin{pmatrix} \ell_f & k & \ell_i \\ -m_f & q & m_i \end{pmatrix} \langle \tau_f, \ell_f \| T_k \| \tau_i, \ell_i \rangle, \quad (11)$$

where the angular part of the reduced matrix element is

$$\langle \ell_f \| T_k \| \ell_i \rangle = (-1)^{\ell_f} \sqrt{\frac{(2\ell_f + 1)(2k + 1)(2\ell_i + 1)}{4\pi}} \begin{pmatrix} \ell_f & k & \ell_i \\ 0 & 0 & 0 \end{pmatrix}. \quad (12)$$

The six symbols in the brackets denote the Wigner $3j$ -symbol. The $3j$ -symbol is zero if conditions $|\ell_f - \ell_i| \leq k \leq \ell_f + \ell_i$ and $m_f + q + m_i = 0$ are not satisfied; if $m_f = q = m_i = 0$, then the $3j$ -symbol is not zero if the sum $\ell_f + k + \ell_i$ is an even number.⁴⁶

For the isotropic spherical potential, eqn (2), the rank of potential spherical operator is $k = 0$. Therefore, the states with different ℓ and m are not mixed. Perturbation $V_N r^N$ mixes states with different n and the eigenstate of Hamiltonian (1), $|\tau, \ell, m\rangle$, is a linear superposition of states $|n, \ell, m\rangle$, eqn (4). Evaluation of matrix elements of the radial part of harmonic oscillator wavefunction, $\langle R_{n_f, \ell_f} | r^N | R_{n_i, \ell_i} \rangle$ with $\ell_f = \ell_i$, shows that non-zero elements are $|n_f - n_i| \leq N$ where n_i and n_f must have the same parity. Because each energy level of the anharmonic Hamiltonian (1) is $(2\ell + 1)$ -fold degenerate in m , it is sufficient to include only the $|n, \ell, 0\rangle$ functions in the basis set. This reduces the number of states by a factor $2\ell + 1$ for each ℓ .

2.2 THz absorption line strengths of the spherical oscillator

We write the dipole moment in spherical harmonics $Y_{1q}(\theta, \phi)$ as a function of He displacement keeping the first two terms, linear and cubic in r ,

$$d_{1q} = \sqrt{\frac{4\pi}{3}} (A_{1,1q}r + A_{3,1q}r^3) Y_{1q}(\theta, \phi), \quad q \in \{-1, 0, +1\}. \quad (13)$$

Only the odd powers of r are present because the dipole moment is a polar vector. $A_{1,1q}$ and $A_{3,1q}$ are the amplitude coefficients. In the spherical symmetry the amplitudes do not depend on q , and we use a simplified notation, $A_{1,1q} \equiv A_1$ and

$A_{3,1q} \equiv A_3$. The length of the dipole moment vector is $|\mathbf{d}| =$

$$\sqrt{\sum_{q=-1}^{+1} (d_{1q})^2} = \sqrt{A_1^2 + 3A_3^2}, \text{ as follows from the normalization of spherical harmonics, } \sum_{m=-\ell}^{\ell} Y_{\ell m}^*(\theta, \phi) Y_{\ell m}(\theta, \phi) = \frac{2\ell + 1}{4\pi}.$$

Using the Fermi golden rule we write the THz absorption line area for the radiation polarized linearly in the z direction, which couples to the dipole moment z component, $d_z = d_{10}$. Since the energy E of Hamiltonian (1) is degenerate in initial and final state quantum numbers m we perform the summation over m_i and m_f for a given transition frequency,

$\omega_{fi} = (E_f - E_i)/\hbar c_0$. The absorption line area S_{fi} is

$$S_{fi} = \int_{\omega_{fi}} \alpha_{fi}(\omega) d\omega = \mathcal{N} f_0 \frac{2\pi^2}{\hbar \epsilon_0 c_0 \eta} \left(\frac{\eta^2 + 2}{3} \right)^2 \omega_{fi} (p_i - p_f) \times \sum_{m_i, m_f} |\langle \tau_f, \ell_f, m_f | d_{10} | \tau_i, \ell_i, m_i \rangle|^2, \quad (14)$$

where the integral is taken over the frequency range ω_{fi} spanning the transition $|i\rangle \rightarrow |f\rangle$. c_0 is the speed of light in a vacuum, ϵ_0 is the permittivity of vacuum, $\hbar$ is the Planck and k_B is the Boltzmann constant. Factor $(\eta^2 + 2)/3$ is the enhancement of radiation electric field at the molecule embedded in dielectric⁴⁷ and η is the index of refraction (for $\eta = 2$, ref. 48). $\mathcal{N} = 1.48 \times 10^{27} \text{ m}^{-3}$ is the number density of molecules in the solid and $f_0 = \mathcal{N}_\bullet / \mathcal{N}$ is the filling factor; $\mathcal{N} = \mathcal{N}_\bullet + \mathcal{N}_\circ$ where $\mathcal{N}_\bullet$ is the number density of filled and $\mathcal{N}_\circ$ is the number density of empty cages. All units are SI and the transition frequency, ω_{fi} , is measured in wavenumbers, number of waves in one meter, $[\omega_{fi}] = \text{m}^{-1}$. p_i and p_f are the thermal Boltzmann populations of the initial and final states:

$$p_i = \frac{e^{-E_i/k_B T}}{\sum_j (2\ell_j + 1) e^{-E_j/k_B T}}, \quad (15)$$

where $2\ell_j + 1$ is the degeneracy of state $|\tau_j, \ell_j, m_j\rangle$ in quantum number m_j .

The sum over m_i and m_f in eqn (14) is

$$\sum_{m_i, m_f} |\langle \tau_f, \ell_f, m_f | d_{10} | \tau_i, \ell_i, m_i \rangle|^2 = \frac{1}{3} |\langle \tau_f, \ell_f \| d_1 \| \tau_i, \ell_i \rangle|^2, \quad (16)$$

where $\langle \tau_f, \ell_f \| d_1 \| \tau_i, \ell_i \rangle$ is the reduced matrix element of d_{1q} with the angular part given by eqn (12). Since for the dipole moment $k = 1$, a selection rule of THz absorption by a spherical oscillator, $\ell_f = \ell_i \pm 1$, follows from the properties of the $3j$ -symbol. The selection rule for n comes from the matrix element of the radial part of harmonic oscillator wavefunction, $\langle R_{n_f, \ell_f} | r^N | R_{n_i, \ell_i} \rangle$. For the linear term in eqn (13) it is $n_f = n_i \pm 1$ and for the cubic term it is $|n_f - n_i| \leq 3$, where n_i and n_f must have different parities.

3 Experimental

³He@C₆₀ and ⁴He@C₆₀ were synthesised using a solid-state process for the critical step, as described in ref. 20. The initial filling factors were 30% to 50%. The samples were further purified by recirculating HPLC on Cosmosil Buckyprep columns to remove trace impurities of H₂O@C₆₀ and empty C₆₀. Finally, the ⁴He@C₆₀ sample had a filling factor of $f_0 = (88.2 \pm 0.5)\%$ while the ³He@C₆₀ had a filling factor of $f_0 = (97.2 \pm 0.5)\%$, as determined by ¹³C NMR. Samples were sublimed under vacuum.

The sublimed powdered sample was put inside the 3 mm diameter hole of cylindrical brass frame and pressed under the vacuum into a pellet. The mass and thickness of the ³He@C₆₀ pellet were 28 mg and 2.16 mm and of the ⁴He@C₆₀ pellet were 21 mg and 1.72 mm. The brass frame with the pellet was



inserted into a sample chamber with two thin-film polypropylene windows and with a vacuum line for pumping and filling with helium heat exchange gas. The sample chamber was in a thermal contact with the cold finger of the cryostat. The cryostat was placed inside the interferometer Vertex 80v sample compartment. The cold finger with the sample chamber was moved up and down by letting the beam through the sample chamber or through a reference hole with 3 mm diameter. The transmission spectra were recorded up to 300 cm^{-1} using a Hg lamp, a $6\text{ }\mu\text{m}$ Mylar beam splitter, and a 4 K bolometer. The spectral resolution was 0.2 cm^{-1} which was found to be smaller than the width of the He absorption lines or their sub-components.

The absorption spectra were calculated from $\alpha(\omega) = -d^{-1} \ln[\mathcal{T}(\omega)\mathcal{R}(\omega)^{-1}]$ where $\alpha(\omega)$ is the absorption coefficient, $\mathcal{T}(\omega)$ is the transmission, and $\mathcal{R}(\omega)$ is the amount of light lost in reflections from the pellet surfaces and the sample chamber windows. The measured transmission is $\mathcal{T}(\omega) = I_s(\omega)/I_r(\omega)$, where $I_s(\omega)$ is the beam intensity at the bolometer with the sample chamber in the beam and $I_r(\omega)$ is the beam intensity at the bolometer with the reference hole in the beam. The reflection losses and the absorption in the sample chamber windows affect the background absorption but do not affect the absorption line areas of helium.

4 Results

The THz absorption spectra of $^3\text{He}@C_{60}$ and $^4\text{He}@C_{60}$ at different temperatures in a spectral range between 75 and 150 cm^{-1} are shown in Fig. 1. The baseline of the 5 K spectrum was fitted and then subtracted. The same procedure but covering only the spectral range around the line no. 1 was applied to the spectra of higher temperatures. For the rest of the spectra above the line no. 1 the average of 5 and 15 K spectra was used for the baseline subtraction. The broad line at about 140 cm^{-1} in the 5 K spectrum is the absorption feature of the mylar windows on the sample cell.

The spectra from the two isotopes share the same pattern of absorption spectra but with the ^4He lines shifted to higher energy relative to the ^3He lines. The line intensities of the two isotopes are similar. For ^3He , a line at 97 cm^{-1} is the only line present at 5 K and its intensity decreases with increasing temperature. This indicates that it corresponds to a transition from the ground state. With the rise in temperature, energy levels above the ground state become thermally populated and we observe six more absorption lines: 106, 114, 120, 122, 128 and 138 cm^{-1} .

In the case of $^4\text{He}@C_{60}$, we observed two transitions from the ground state, at 81 cm^{-1} and at 284 cm^{-1} . By increasing the temperature additional absorption lines become visible at 88, 95, 101, 106, 111, 114 and 126 cm^{-1} . The line no. 9 starts from the ground state as is confirmed by its T dependence, Fig. 7 in the Appendix. This line cannot be detected in $^3\text{He}@C_{60}$ THz spectra because of the strong absorption by the fullerene at its resonance frequency, 340.6 cm^{-1} in Table 4, but was observed

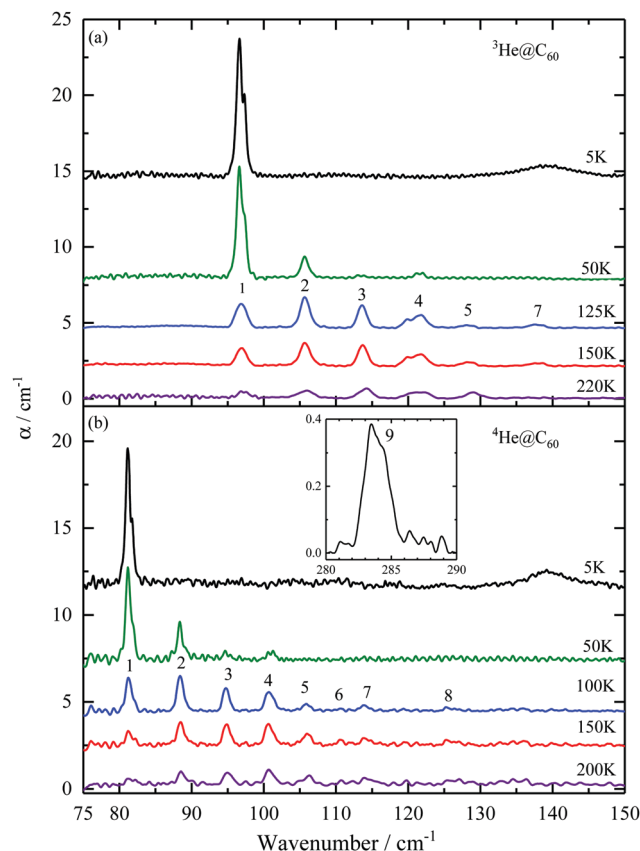


Fig. 1 Temperature dependence of THz absorption spectra of $^3\text{He}@C_{60}$ (a) and $^4\text{He}@C_{60}$ (b). Absorption lines are numbered according to the transitions between the energy levels as shown in Fig. 3. Inset to (b) shows the $^4\text{He}@C_{60}$ line no. 9 at 5 K. The line intensities are not corrected for the filling factor f_0 , see eqn (14).

in the inelastic neutron scattering spectrum, line no. e in ref. 41.

The lowest energy peak has a two-component structure which is visible at low temperatures. The separation of components is approximately 0.8 cm^{-1} for $^3\text{He}@C_{60}$ and 0.7 cm^{-1} for $^4\text{He}@C_{60}$, Tables 4 and 5 in the Appendix. We did not see any additional structure at the best resolution we used, 0.2 cm^{-1} . The possible origin of line no. 1 and no. 9 splitting is discussed in Section 5.

The synthetic spectra were fitted with a spherical oscillator model, as described in the Appendix, and the result is shown in Fig. 2. The best spherical oscillator model fit parameters are listed in Table 2. As compared to our previous fit⁴¹ the extra term, A_3r^3 , was added to the dipole moment expansion, eqn (13). This gives a better fit of the line intensities. The spherical oscillator fit revealed that line number 4 consists of three closely spaced transitions, and lines 5 and 6 of two transitions. The energy diagram together with the numbered transitions is shown in Fig. 3 and the energies of corresponding levels in Table 2. The list of line frequencies and intensities calculated with the spherical oscillator model best fit parameters for ^3He (125 K) and ^4He (100 K) is given in the Appendix, Tables 4 and 5.



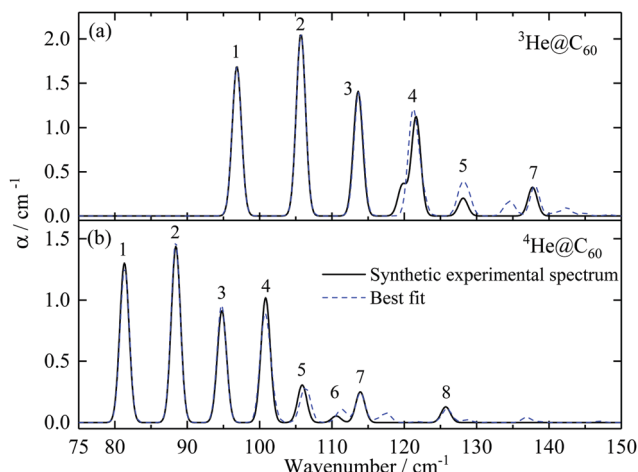


Fig. 2 Synthetic experimental THz spectrum (solid black line) and the best fit result (blue dashed line) of $^3\text{He}@C_{60}$ at 125 K (a) and $^4\text{He}@C_{60}$ at 100 K (b).

In Fig. 4 the areas of the first five THz absorption lines of $^3\text{He}@C_{60}$ and $^4\text{He}@C_{60}$ are compared with the temperature dependence of line areas derived from the spherical oscillator model fit. The line areas at each temperature are calculated using the parameters $\{V_2, V_4, V_6, A_1, A_3\}$ obtained from the anharmonic oscillator model fit of 125 K and 100 K spectra of ^3He and ^4He , respectively. For each line the line areas are normalized to the maximum of the theoretical intensity *versus* temperature curve. Same normalization factors are applied to each experimental line respectively.

5 Discussion

The overlap of potential energy curves of two He isotopes proves the reliability of our fit as the difference of potential curves is only about $\pm 0.5 \text{ cm}^{-1}$ over the energy scale probed by THz spectroscopy, Fig. 5(b). The potential is strongly anharmonic. It can be seen that for the displacement $r > 75 \text{ pm}$ the anharmonic contribution to the potential energy, $V_4 r^4 + V_6 r^6$, starts to dominate over the harmonic part, $V_2 r^2$. The significance of anharmonic correction is also evidenced by the composition of the oscillator wave function: as the energy increases the single n -component of a wavefunction becomes less dominant meaning that the components with other n values are mixed in Table 2. States with different ℓ values are not mixed by the isotropic potential and each level is $2\ell + 1$ degenerate in the quantum number m .

Experimentally determined translational energies of $A@C_{60}$ endohedral complexes are summarized in Table 3. Although the anharmonic contributions to the potential have been determined experimentally,^{15,22,23} a more detailed comparison with is not meaningful as firstly, has translation-rotation coupling terms in the potential and secondly, it misses the V_6 term in the potential fit. The shape of the potential curve is not known for other A species. We approximate the translational energy, ω_T , as the difference of energies between the ground

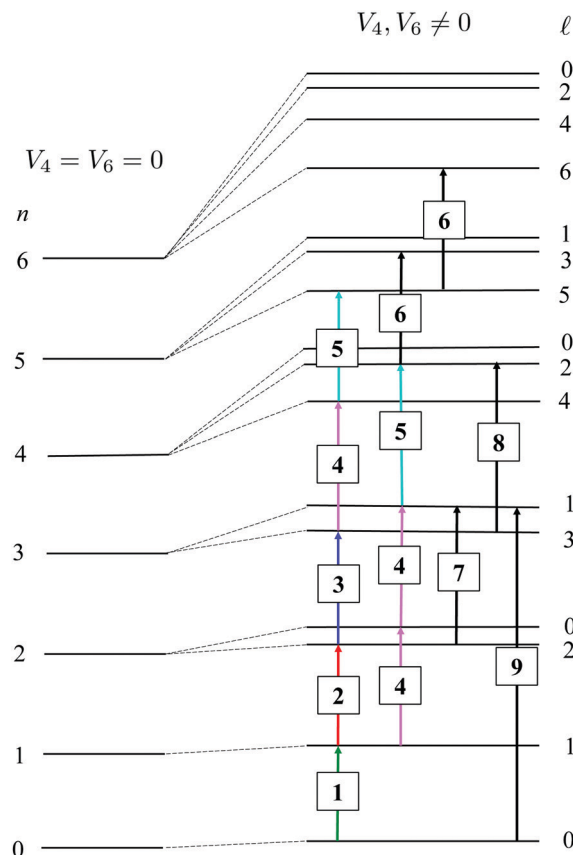


Fig. 3 Energy level diagram and THz transitions of $\text{He}@C_{60}$. The principal translational quantum number is n and the angular momentum quantum number is ℓ . The energy levels of a spherical harmonic oscillator are degenerate in ℓ , shown on left. The anharmonic terms in the potential, and V_6 , split the energy levels with different ℓ and each energy level has the unique ℓ value within the spherical symmetry, on the right. The mixing of states with different n by V_4 - and V_6 -term is not shown on the diagram. The THz transitions between the translational energy levels observed in the experiment are labeled with numbers, see Fig. 1 and Tables 4 and 5, and the translational energies are in Table 2.

state and the first excited translational state,. Assuming harmonic approximation, $V_2^A \approx M_A \omega_{01}^2/2$, we scaled the potential of other species, V_2^A , relative to hydrogen, $V_2^{\text{H}_2}$. The last column demonstrates that $1/\sqrt{M_A}$ scaling of translational frequency of different A species has no predictive power. In general, the interaction of neutral A with can be separated into repulsive interaction and electrostatic interaction expanded in induction and dispersion terms.³³ Since has no electric dipole nor quadrupole moment the induction terms are zero. The absence of induction terms may only partially explain the “softness” of potential because the dominant interactions are repulsion and dipole-dipole dispersion for endohedral atoms and molecules.³³ In general, modern quantum chemistry calculations of $\text{He}@C_{60}$ give a good description of $V(r)$.⁴¹

To further validate the potential parameters of obtained from the fit of single high temperature spectra we compare the temperature dependence of line intensities of measured and calculated spectra, Fig. 4. The parameters determined from



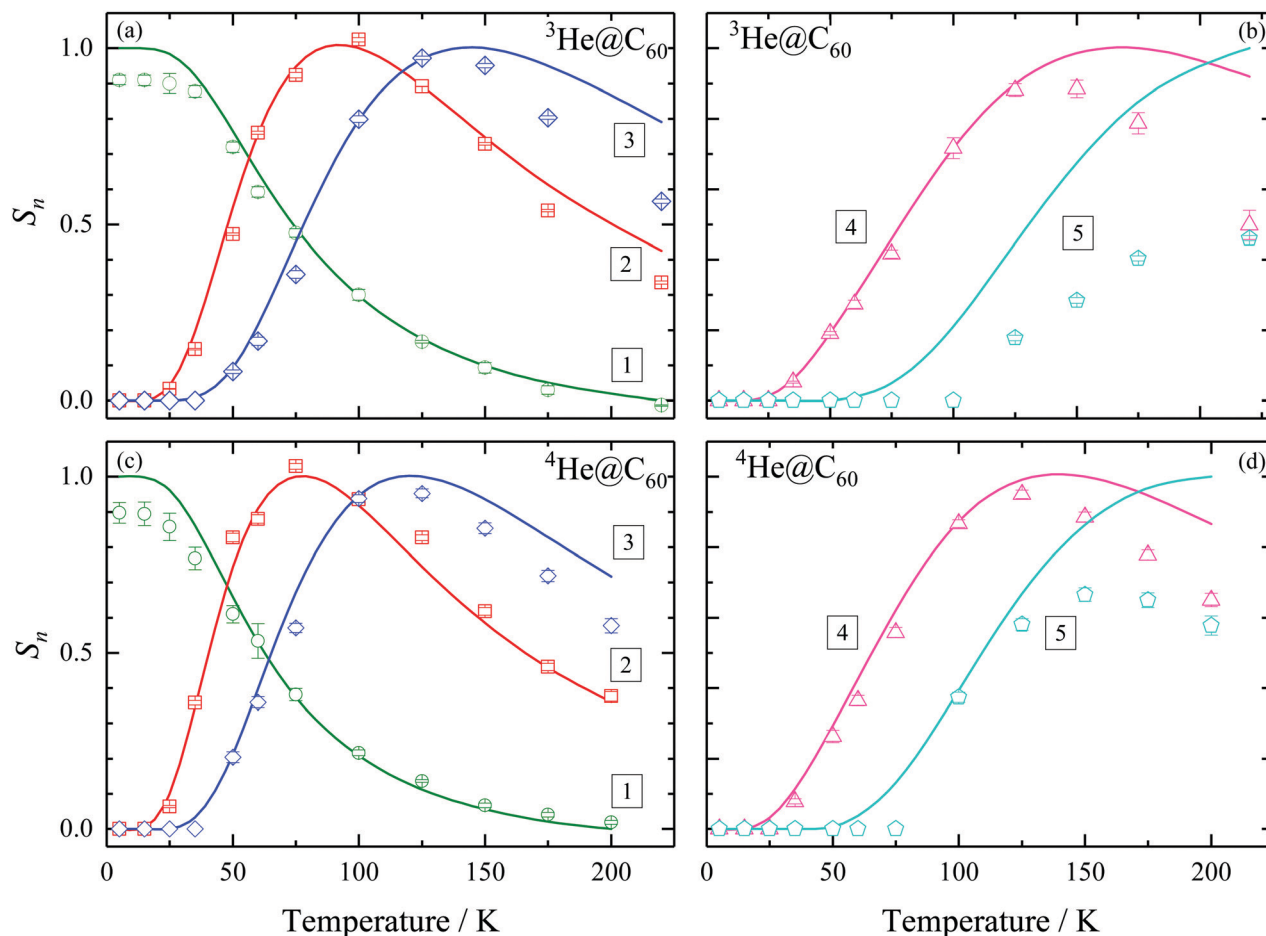


Fig. 4 Temperature dependence of normalized THz absorption line areas S_n of $^3\text{He}@C_{60}$, panels (a) and (b) and of $^4\text{He}@C_{60}$, panels (c) and (d). Symbols are experimental line areas with errors from the fit with Gaussian lines and solid lines are theoretical areas calculated with the parameters from Table 1. Line areas have been normalized to the maximum value in the T dependence of theoretical area for each line. The lines are numbered according to the transitions shown in Fig. 3 and listed in Tables 4 and 5.

the ^3He (^4He) fit at 125 K (100 K) describe the temperature dependence from 5 to 200 K rather well except for the line no. 5. Other lines, no. 6, no. 7, no. 8 and no. 9, are weak and their intensities could not be determined reliably from the experimental spectra and therefore their T dependence was not analyzed. The intensity of the lowest frequency line no. 1, transition from the ground state, is overestimated by the fit at the lowest temperatures. The discrepancy of calculated and measured T dependence could be due the thermal motion of C_{60} not taken into account in our model.

The dipole moment of is induced by the displacement from the C_{60} cage center. The estimate for the displacement of the first transition can be made from the potential energy which is half of the total energy, $V(r) = E_1/2$. Using the data from Table 2, $E_1 = 222.5 \text{ cm}^{-1}$ (^3He) and $E_1 = 187.9 \text{ cm}^{-1}$ (^4He), we obtain the displacement to be about 60 pm. The dipole moment induced by 60 pm displacement is $d = A_1 r + A_3 r^3 = 2.7 \times 10^{-2} \text{ D}$, small compared to permanent dipole moments of molecules. For example, the permanent screened dipole moment of C_{60} endohedral water is about 0.5 D.^{49,50}

Transition no. 9 starts from the ground state with $\ell = 0$ and predominantly with $n = 0$ radial content. The final state is the

$\ell = 1$ state, see Fig. 3 and Table 2, but with predominantly $n = 3$ radial part. It is an allowed electric dipole transition with the selection rule $\Delta\ell = +1$ but three translation quanta are created, $\Delta n = +3$. All other lines, no. 1 to no. 8, are ordinary one quantum transitions, $\Delta n = +1$. The r^3 term in the dipole moment expansion allows the $\Delta n = +3$ excitation by radiation. However, because of the mixing of states by anharmonic potential, $n = 0$ state is mixed with $n = 2$ and $n = 3$ state is the mixed $n = 1$ state, the linear term in r of the dipole moment expansion with the selection rule $\Delta n = +1$ can induce the three-quantum transition. Thus, in there are two factors what activate the three-quantum transition, the anharmonicity of the potential and the r^3 term in the dipole moment expansion.

Two absorption lines, no. 1 and no. 9, show partially resolved fine structure at 5 K, Fig. 1. These lines are the transitions from the ground state with $\ell = 0$ to the state $\ell = 1$, see Fig. 3 and Table 2. In the icosahedral potential the first A_g symmetry representation after fully spherical potential, rank $k = 0$, is a combination of spherical harmonics of rank $k = 6$.⁵¹ Rank 6 term in the potential splits states with $\ell \geq 3$, which cannot explain the fine structure of helium line no. 1 and no. 9.



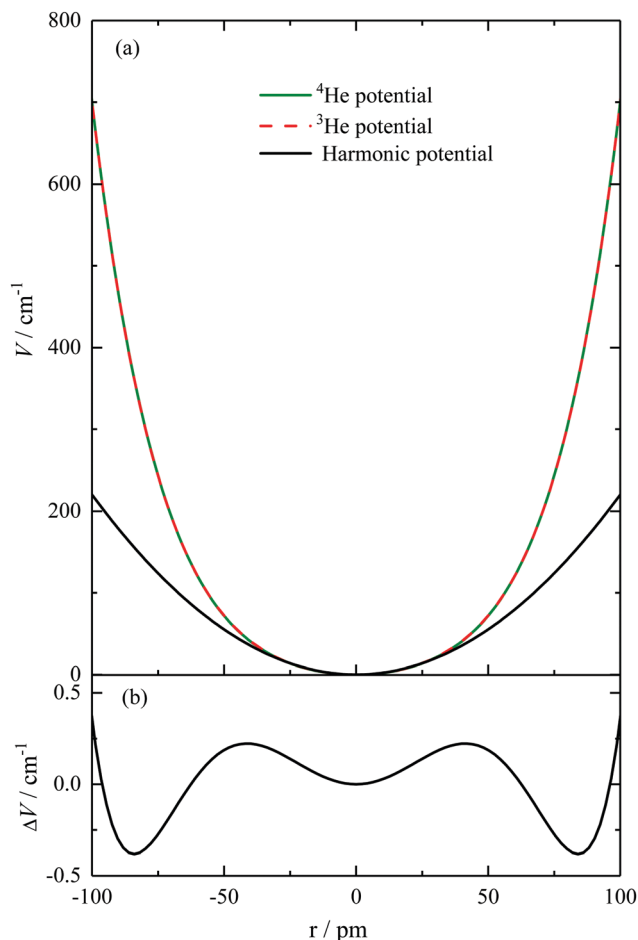


Fig. 5 (a) Potential energy curves of ^3He and ^4He , $V(r) = V_2r^2 + V_4r^4 + V_6r^6$, calculated with parameters from Table 1. The harmonic part of ^3He potential curve, V_2r^2 , is plotted as blue solid line. The potential curves of two isotopes, green solid line of ^3He and red dashed line of ^4He , are indistinguishable in this plot. (b) The difference between the potential curves of ^3He and ^4He , $\Delta V = V_3 - V_4$, is less than $\pm 0.5 \text{ cm}^{-1}$.

To explain the spectral line splitting, a potential with symmetry lower than the icosahedral or two different exohedral environments must be invoked.

Fullerene molecules stop rotating below 90 K.⁵² This leads to two sites distinguished by the relative orientation of the central cage and its 12 nearest-neighbor C_{60} .⁵³ In the pentagonal orientation the electron-rich double bonds are facing pentagonal rings and in the hexagonal orientation the electron-rich double bonds are facing hexagonal rings of nearest-neighbor cages. It was proposed by Felker *et al.*⁵⁴ that the orientational order creates electrostatic field that interacts with the quadrupole moment of endohedral molecule. Their model explains the splitting of $J = 1$ rotational state observed in $\text{H}_2@\text{C}_{60}$,^{22,23,28} $\text{HF}@\text{C}_{60}$ ¹⁶ and $\text{H}_2\text{O}@\text{C}_{60}$.^{49,55,56} This mechanism is not applicable to the helium atom because it does not have rotational degrees of freedom nor quadrupole moment. The icosahedral symmetry of $V(r)$ could still be disturbed by the nearest-neighbor C_{60} molecules. A splitting similar to the splitting of line no. 1 was observed in $\text{H}_2@\text{C}_{60}$ for the $\ell = 0 \rightarrow \ell = 1$

Table 1 The best fit parameter values for $^3\text{He}@\text{C}_{60}$ and $^4\text{He}@\text{C}_{60}$. V_2 , V_4 and V_6 are potential parameters, eqn (2) and (3), and A_1 and A_3 are the dipole moment parameters, eqn (13)

κ_i	$^3\text{He}@\text{C}_{60}$	$^4\text{He}@\text{C}_{60}$	Unit
V_2	$(2.500 \pm 0.015) \times 10^{-3}$	$(2.46 \pm 0.04) \times 10^{-3}$	meV pm^{-2}
V_4	$(3.64 \pm 0.03) \times 10^{-7}$	$(3.77 \pm 0.08) \times 10^{-7}$	meV pm^{-4}
V_6	$(2.560 \pm 0.017) \times 10^{-11}$	$(2.46 \pm 0.06) \times 10^{-11}$	meV pm^{-6}
A_1	$(3.83 \pm 0.10) \times 10^{-4}$	$(3.73 \pm 0.22) \times 10^{-4}$	D pm^{-1}
A_3	$(1.7 \pm 0.3) \times 10^{-8}$	$(2.3 \pm 0.6) \times 10^{-8}$	D pm^{-3}

Table 2 Translational energy levels of $^3\text{He}@\text{C}_{60}$ and $^4\text{He}@\text{C}_{60}$ obtained from the fit of THz absorption spectra. Translational energy E , the angular momentum quantum number ℓ (single-valued for each energy level), and the amplitude squared of the main component of eigenstate with the principal quantum number n . The zero point energy 125.7 cm^{-1} (106.5 cm^{-1}) of ^3He (^4He) has been subtracted

$^3\text{He}@\text{C}_{60}$				$^4\text{He}@\text{C}_{60}$			
E/cm^{-1}	ℓ	n	$ \xi_n ^2$	E/cm^{-1}	ℓ	n	$ \xi_n ^2$
0	0	0	0.95	0	0	0	0.95
96.8	1	1	0.87	81.4	1	1	0.88
202.6	2	2	0.77	169.8	2	2	0.79
218.3	0	2	0.64	182.1	0	2	0.68
316.3	3	3	0.65	264.5	3	3	0.69
340.6	1	3	0.43	283.7	1	3	0.48
437.3	4	4	0.54	365.1	4	4	0.58
469.4	2	6	0.39	390.6	2	6	0.39
483.0	0	6	0.40	401.1	0	6	0.40
565.1	5	5	0.43	471.1	5	5	0.47
604.3	3	7	0.36	502.0	3	7	0.38
625.8	1	7	0.31	519.0	1	7	0.35
699.3	6	8	0.37	582.2	6	6	0.38
745.1	4	10	0.29	618.2	4	8	0.31
774.0	2	10	0.30	640.8	2	10	0.30
787.4	0	10	0.30	650.5	0	10	0.31

Table 3 The translational energies of endohedral species A in $A@\text{C}_{60}$ from the ground to the first excited state, $\hbar\omega_{01}$. The scaling of V_2^A with respect to $V_2^{\text{H}_2}$ is found using harmonic approximation, $\omega_T = \omega_{01}$, and eqn (10) where M_A is the mass in atomic mass units, a.u

A	$\hbar\omega_{01}/\text{cm}^{-1}$	$M_A/\text{a.u.}$	$V_2^A/V_2^{\text{H}_2}$
H_2	179.5 ²³	2	1
^3He	96.7 ⁴¹	3	0.44
^4He	81.3 ⁴¹	4	0.41
HF	78.6 ¹⁶	20	1.92
H_2O	110 ⁴⁹	18	3.4

transition of *para*- $\text{H}_2@\text{C}_{60}$ in the $J = 0$ rotational state.^{22,23} Application of pressure changes the relative population of pentagon- and hexagon-oriented molecules.⁵⁷ The analysis of the inelastic neutron scattering spectra of pressure-treated $\text{H}_2@\text{C}_{60}$ shows that the potential and the energy levels of are sensitive to the orientation of neighboring cages.²⁸ Similar THz and infrared spectroscopy experiments are planned to verify the effect of orientational order on the potential energy function of endohedral molecules and atoms in C_{60} .

In summary, with the THz absorption spectroscopy we have determined the energy level structure of quantized



translational motion of single ^3He and ^4He atoms trapped in the C_{60} molecular cages. The fitted potential energy curves and the induced dipole moments of two isotopes overlap with high precision. However, there are deviations between the modeled and measured spectra. Firstly, the fine structure of spectral lines observed in the low temperature spectra cannot be explained by icosahedral symmetry of C_{60} molecule. Secondly, there are some discrepancies between the measured and modeled spectra in the line intensities and their temperature dependence. Both deviations could be due to the orientational order and the thermal motion of C_{60} molecules in the solid.

Conflicts of interest

There are no conflicts to declare.

Appendix

Fit of anharmonic spherical oscillator and dipole moment parameters

The experimental THz absorption spectrum was fitted using Gaussian line shapes to find the line areas, line widths and frequencies, see Fig. 6. A synthetic experimental spectrum $y(\omega_n)$, the distance between the points in the spectrum $\omega_n - \omega_{n-1} = \Delta\omega/4$, was then generated consisting of lines with equal full width at half maximum, $\Delta\omega = 1.5 \text{ cm}^{-1}$, while keeping the line areas and frequencies of the original experimental lines. The synthetic spectrum approach was needed as the model did not include any line broadening mechanism.

The synthetic spectra were fitted using the anharmonic spherical oscillator model. The reduced basis was limited to

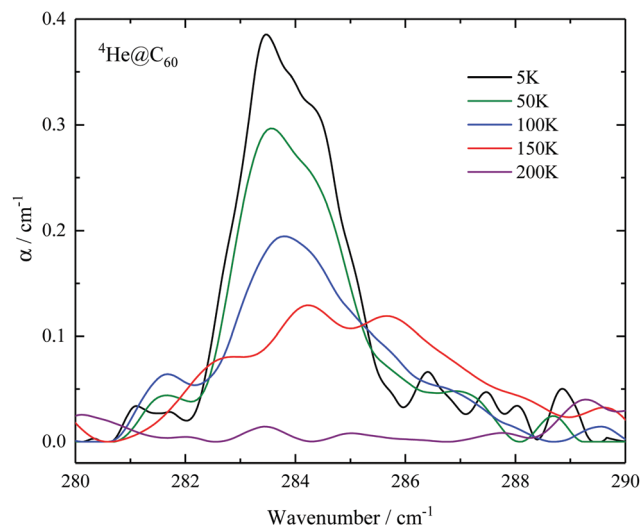


Fig. 7 Temperature dependence of line no. 9 in $^4\text{He}@C_{60}$.

$n_{\text{max}} = 18$ providing 100 states in the reduced basis $|n, \ell\rangle$. For a given model and basis, matrix elements of the Hamiltonian, eqn (1), and the dipole operator, eqn (13), were evaluated analytically in a symbolic form using the Mathematica software. At each step of minimizing chi squared, $\chi^2 = \sum [y - f(\omega_n, \{\kappa\})]^2$, the Hamiltonian was diagonalized numerically. Here $f(\omega_n, \{\kappa\})$ is the theoretical spectrum with the same linewidth and lineshape as the synthetic experimental spectrum; $\{\kappa\} = \{V_2, V_4, V_6, A_1, A_3\}$ is the set of Hamiltonian and dipole operator fit parameters. Errors were calculated using the method described in ref. 49. Adding the V_6 term to V_4 in the potential reduced χ^2 of ^3He and ^4He fit by three and two times, respectively. The frequencies and intensities of experimental spectra and spectra calculated using the best fit parameters are given in Tables 4 and 5.

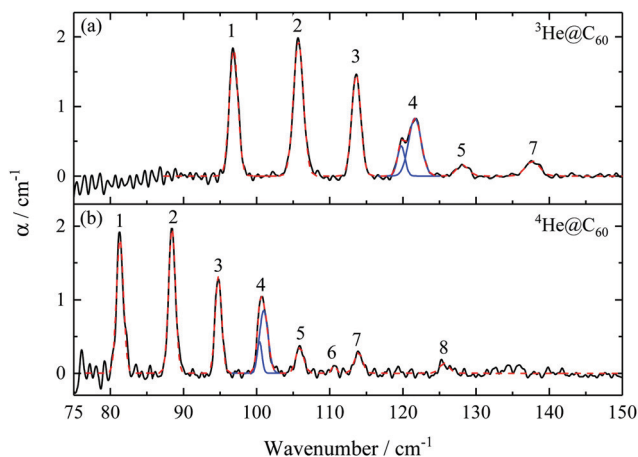


Fig. 6 THz absorption spectrum of ^3He at 125 K (a) and ^4He at 100 K (b), black solid line, and the fit with Gaussian lineshapes, blue solid lines. Red dashed line is the sum of Gaussians. The line intensities are not corrected for the filling factor f_0 , see eqn (14).

Table 4 The line frequencies f and areas S of experimental spectra at 5 and 125 K and the anharmonic spherical oscillator model fit results at 125 K for $^3\text{He}@C_{60}$. The f and S of lines no. 1 and no. 9 at 5 K are calculated using the best fit parameters obtained at 125 K

Line no.	Experiment		Model fit	
	f/cm^{-1}	S/cm^{-2}	f/cm^{-1}	S/cm^{-2}
5 K				
1	96.6	8.5	96.8	11.7
	97.4	2.2	—	—
9	—	—	340.6	0.4
125 K				
1	96.9	2.7	96.8	2.7
2	105.7	3.3	105.7	3.3
3	113.6	2.3	113.7	2.2
4	119.8	0.6	121.0	1.1
	121.6	1.8	121.4	0.8
	—	—	122.4	0.5
5	128.1	0.3	127.8	0.5
	—	—	128.8	0.4
7	137.7	0.5	138.0	0.6



Table 5 The line frequencies f and areas S of experimental spectra at 5 and 100 K and the anharmonic spherical oscillator model fit results at 100 K for $^4\text{He@C}_{60}$. The f and S of lines no. 1 and no. 9 at 5 K are calculated using the best fit parameters obtained at 100 K

Line no.	Experiment		Model fit	
	f/cm^{-1}	S/cm^{-2}	f/cm^{-1}	S/cm^{-2}
5 K				
1	81.2	6.1	81.4	8.0
	81.9	1.2	—	—
9	283.3	0.3	283.7	0.3
	284.3	0.4	—	—
100 K				
1	81.3	2.1	81.4	2.0
2	88.4	2.3	88.4	2.4
3	94.8	1.5	94.8	1.5
4	100.4	0.4	100.6	0.7
	101.0	1.3	100.8	0.6
	—	—	101.6	0.3
5	105.9	0.5	106.0	0.3
	—	—	106.7	0.2
6	110.6	0.1	111.0	0.09
	—	—	111.7	0.10
7	113.9	0.4	113.9	0.4
8	125.7	0.2	125.9	0.2

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