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Stereodivergent entry to β -branched β -trifluoromethyl α -amino acid derivatives by sequential catalytic asymmetric reactions†

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Currently, conventional reductive catalytic methodologies do not guarantee general access to enantioenriched β -branched β -trifluoromethyl α -amino acid derivatives. Herein, a one-pot approach to these important α -amino acids, grounded on the reduction – ring opening of Erlenmeyer–Plöchl azlactones, is presented. The configurations of the two chirality centers of the products are established during each of the two catalytic steps, enabling a stereodivergent process.

Introduction

β-Branched α-amino acids (AAs) carrying different β-substituents – thus bearing two vicinal chirality centres, at the α - and β carbons – are important yet challenging synthetic targets. The stereocontrolled preparation of these compounds has been tackled and realised with different (catalytic) methods.2-7 A noncomprehensive list of examples includes enantioselective conjugate additions and alkylations of glycinate imines,2 palladium catalysed β-C(sp³)-H alkylation and arylation of AAs,³ aziridine ring-opening,4 an engineered tryptophan synthase,5 multi-enzymatic β-methylation of AAs,6 and catalytic asymmetric hydrogenation of α,β -dehydro-amino acids (DHAAs), in its implementation with tetra-substituted substrates.7 This latter approach (Scheme 1(a)) has disclosed diastereo- and enantioselective entries to various classes of β-branched AAs, including^{7d} less common yet relevant β-trifluoromethyl AA derivatives.8,9 Given the common stereospecificity of the hydrogenation, the E/Z geometry of the substrate dictates the relative configuration of the product (Scheme 1(a)). Thus, a diastereoisomer is obtainable only if the parent olefin isomer can be prepared. This constraint can have negative implications. For example, a straightforward synthesis of a preclinical drug candidate via asymmetric hydrogenation of the corresponding β-aryl-β-trifluoromethyl DHAA was envisioned (Scheme 1(b)).94 However, the required E-olefin isomer could

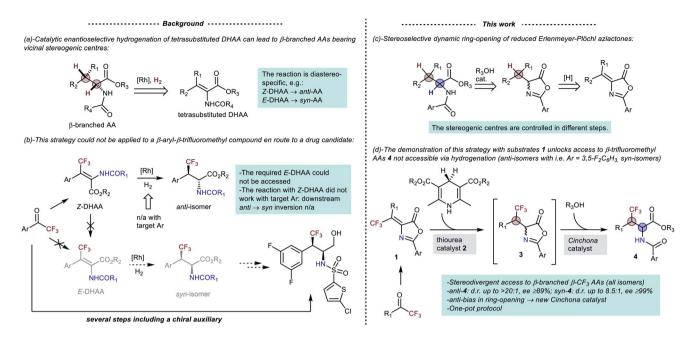
not be accessed with sufficient selectivity. Furthermore, enantioselective hydrogenation of tetrasubstituted olefins can occasionally be challenging. In fact, the β -trifluoromethyl *Z*-DHAAs required for the target was found to be reluctant to asymmetric hydrogenation, ^{9 α} preventing application of Turner's formal stereodivergent reduction of DHAAs. ¹⁰ This ingenious chemoenzymatic approach, which unlocks access to the isomer not obtainable by hydrogenation *via* downstream inversion of the α -amino chirality centre in the reduced DHAA, is in fact established only for β -aryl- β -methyl AAs. Eventually, to obtain the target *syn*-trifluoromethylated amino alcohol,§ Alimardanov *et al.* resorted to a longer, yet effective and scalable, route, encompassing a chiral auxiliary and introduction of the amino functionality at a late stage. ^{9 α}

With this background in mind, we envisioned an original stereodivergent¹¹ entry to β-branched AAs grounded on the dynamic stereoselective ring-opening of enantioselectively reduced Erlenmeyer-Plöchl azlactones (Scheme 1(c)). In contrast with catalytic hydrogenation, this formal hydrogenation of the azlactone olefin (a DHAA derivative) fixes the configurations of the two hydrogenated centres in different steps, lending itself to stereodivergency. Herein, we present the first demonstration of this strategy by its application to the onepot preparation of β-trifluoromethyl AA derivatives 4 (Scheme 1(d)). In more detail, enantioselective transfer hydrogenation of readily available substrates 1 (ref. 9a) with Hantzsch esters12 sets the absolute configuration of the trifluoromethylated β-centre.13 Subsequent dynamic alcoholytic ring-opening¹⁴ of intermediates 3 fixes the absolute configuration of the α -carbon. In the ring-opening reaction, the substrates 3 feature a considerable bias towards the formation of anti-isomers 4. Such bias was readily leveraged with conventional Cinchona catalysts to obtain a range of anti-4 products with very high stereoselectivities (including compounds not suited for hydrogenation). Conversely, the development of the syn-selective process was

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Scheme 1 (Left) (a) catalytic asymmetric hydrogenation approach to β -branched AAs, and (b) its failure with a β -trifluoromethyl drug candidate. (Right) (c) the underlying concept of this work and (d) its demonstration with β -trifluoromethyl substrates 1.

less straightforward, requiring a peculiar ammonia-derived squaramide catalyst to afford the *syn-4* isomers with variable diastereoselectivities.

It is worth stressing that β -branched β -trifluoromethyl AAs, and derivatives thereof, have found widespread interest in medicinal chemistry (Fig. 1). β -Trifluoromethylated analogues of natural AAs have been incorporated into peptides, wherein the trifluoromethyl group can give unique effects on stability, acidicy/basicity, folding behaviour, hydrophobicity, and ultimately biological activity. The β -trifluoromethyl AA framework can also be found in less canonical structures. Besides the drug candidate mentioned in Scheme 1(b), which showcases inhibition of β -amyloid production, another β -trifluoromethyl AA structure of medicinal interest is an analogue of thalidomide, which feature enhanced configurational stability compared to

F HO CF₃ OH HO NH₂ OH NH₂ OH NH₂ (3R)-4³-F₃Glu (3S)-4³-F₃Val (3R)-4³-F₃Jall OH NH₂ OH NH

Fig. 1 Medicinal chemistry relevant β -branched β -trifluoromethyl AAs.

thalidomide. An additional example is represented by the heterocyclic compound derived from a 3-(trifluoromethyl)azetidine carboxylic acid, shown in Fig. 1. This compound is a member of a library of related heterocycles, investigated for their activity as inhibitors of phosphoinositide 3-kinases.¹⁶

Results and discussion

At the outset, with transfer hydrogenation to β,β-disubstituted nitroalkenes promoted by Jacobsen-type catalysts as a lead, 17 we explored different variables in the reduction of β-phenyl-β-trifluoromethyl Erlenmeyer-Plöchl substrates 1 with HEs.¶ Preliminary screening of different catalysts and HEs identified the conditions outlined in Table 1, entry 1, as promising starting point. In more detail, using 20 mol% of catalyst 2a in combination with the isobutyl HE, in dichloromethane (reagent grade) as solvent at low temperature $(-30 \, ^{\circ}\text{C})$, the 2-phenyl azlactone 1a could be reduced with full conversion and promising enantioselectivity (70% ee). Products 3 of the transfer hydrogenation reaction are relatively unstable. Thus, for CSP HPLC analysis they are converted to the ultimate products 4 by ring-opening with allyl alcohol using achiral tertiary amine promoters, furnishing the two diastereoisomers 4. These two isomers feature comparable ees, validating this method for the evaluation of the enantioselectivity of the reduction step. Thus, a systematic variation of the modules of this type of Jacobsen catalyst18 (AA portion, double H-bond donor, amide portion, terminal N-group), of C2 substituents of the azlactone, and of reaction conditions, was undertaken. Exploration of catalysts based on double hydrogen bond donors other than thiourea (ureas and squaramides), and another AA portion (L-valine), confirmed the L-tert-leucine derived thiourea as the most efficient catalyst core. Whereas variation of solvent and/or dilution

Table 1 Screening of azlactones 1 and catalysts 2 in the transfer hydrogenation reaction. Representative results a

$$\begin{array}{c} \text{Ph} \\ \text{O} \\ \text{O} \\ \text{Ia-d} \\ \text{Ar} \\ \text{J} \\ \text{BuO}_2\text{C} \\ \text{D} \\ \text{C} \\ \text{H} \\ \text{H} \\ \text{H} \\ \text{H} \\ \text{C} \\ \text{O} \\ \text{C} \\ \text{H} \\ \text{E} \\ \text{O} \\ \text{C} \\ \text{C} \\ \text{E} \\ \text{I} \\ \text{B} \\ \text{U} \\ \text{O} \\ \text{C} \\ \text{C} \\ \text{F} \\ \text{I} \\ \text{E} \\ \text{I} \\ \text{O} \\ \text{I} \\ \text{C} \\ \text{C} \\ \text{C} \\ \text{F} \\ \text{I} \\ \text{I$$

entry	1 : Ar	catalyst 2	4 -ee (%) ^b	
1	1a : Ph	2a	4a -70/70	
2	1a : Ph	2b	4a -78/75	
3	1a : Ph	2c	4a -60/64	
4	1a : Ph	2d	4a -72/71	
5	1a : Ph	2e	4a -55/50	
6	1a : Ph	2f	4a -60/50	
7	1b : 1-naphthyl	2b	4b -72/63	
8	1c : 4-BrC ₆ H ₄	2b	4c -75/73	
9	1d : 4-MeOC ₆ H₄	2b	4d -83/84	
10	1d : 4-MeOC ₆ H ₄	2g	4d -70/70	
11	1d : 4-MeOC₀H₄	2h	4d -83/84	
12	1d : 4-MeOC ₆ H₄	2i	4d -78/80	
13	1d : 4-MeOC₀H₄	2j	4d -75/76	
14	1d : 4-MeOC₀H₄	2k	4d -85/85	

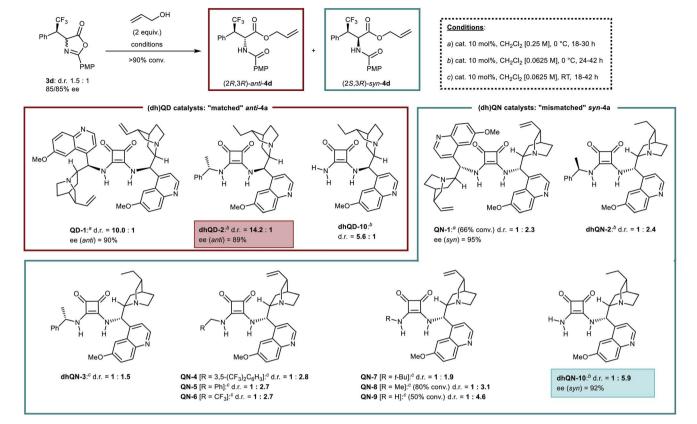
 a Conditions: substrate 1 (0.05 mmol), catalyst 2 (0.01 mmol, 20 mol%), HE (0.075 mmol), CH₂Cl₂ (200 μL), -30 °C, 48 h. All reactions gave >90% conversion of 1 (19 F NMR). Filtration on silica, evaporation, then CH₂Cl₂ (0.5 mL), allyl alcohol (0.1 mmol), Et₃N (1 drop), RT, 24–48 h. Enantiomeric excess of *anti-*4 and *syn-*4, respectively, determined by CSP HPLC after chromatographic purification on silica gel.

was not fruitful, investigation of different terminal N-groups in the thiourea (2a-f) indicated that the 4-(trifluoromethyl)phenyl residue (2b) could provide a distinct improvement, compared to the common 3,5-bis(trifluoromethyl)phenyl (2a) and other aryl/alkyl groups (2c-f) (entries 1–6). With catalyst 2b, three differently C2 substituted Erlenmeyer–Plöchl azlactones 1b-d were screened (entries 7–9). The 4-methoxyphenyl (PMP) derivative 1d gave better results, and was thus selected for further catalyst development, which focussed on exploring various N-substituents on the amide (entries 10–14). Amongst catalysts 2g-k, the N-benzyl-N-methyl amide derivative 2k performed best, affording the transfer hydrogenation product 3d with a respectable 85% ee, measured on the corresponding ring-opened derivative 4d.

A closer inspection at the alcoholytic ring-opening step of compounds 3 promoted by achiral tertiary amines, indicated that the ring-opened products 4 were obtained with higher anti/syn ratios (5.5:1 for 4d) than the parent azlactones 3 (ca. 1.5:1). Thus, the alcoholytic process was dynamic, and biased

towards the *anti*-isomer. We initially surmised that such substrate-bias, hindering access to *syn-4*,¹⁹ would be circumscribed to the tertiary amine promoted alcoholytic ring-opening. Since a variety of DKRs of azlactones by ring-opening reactions, using different nucleophiles and catalytic approaches (Lewis bases, Lewis acids, Brønsted acids, enzymes), are available in the literature,¹⁴ we hoped that one of these could be subdued to our aims.

However, a preliminary screening of many of these methods suggested the squaramide Cinchona catalyzed alcoholytic ring opening²⁰ as most promising option, despite its resemblance with the biased achiral amine promoted reaction. State-of-theart Cinchona squaramide dimeric catalysts derived from quinidine (QD-1) and quinine (QN-1) (Scheme 2) were initially employed with enantioenriched azlactone 3d under standard conditions (dichloromethane, allyl alcohol, 0 °C). While, unsurprisingly, "matched" QD-1 increased the anti/syn ratio to a high 10.0: 1 value, compared to an achiral tertiary amine (ca. 5.5:1), we were pleased to observe that the corresponding "mismatched" **ON-1** could reverse the selectivity of the process, forcing the ring-opening reaction towards a moderate preference (1:2.3) for syn-4d. The products 4d displayed a higher enantiomeric excess than 3d, in accordance with the Horeau effect.21 Adjusting the reaction conditions and testing additional QD and dihydroquinidine (dhQD) derived structures led to a highly anti-selective protocol. Catalyst dhQD-2 improved in fact the diastereomeric ratio of the product 4d up to 14.2:1 in favour of the anti-isomer. However, application of its quasienantiomeric derivative dhQN-2 did not result in the expected improvement in the syn-selectivity, providing a result similar to **QN-1** (1:2.4 vs. 1:2.3). This result emphasized that the transition states leading to the anti and syn-products are "intrinsically" diastereomeric, 19h due to the presence of the chiral (R)configured β-branched chain of azlactone 3d. On these grounds, different (i.e. non quasi-enantiomeric) catalyst structures may be required for anti- and syn-selective processes. Thus, a range of (dh)QN derived squaramide catalysts and reaction conditions were examined (see also ESI†). The diastereomeric catalyst dhQN-3 (from (S)-α-methylbenzylamine instead of (R)-α-methylbenzylamine of dhQN-2) was tested first, giving however a poor result. Subsequent catalyst screening, performed at RT, suggested that the main factor affecting the stereoselectivity is the bulkiness of the squaramide portion. While catalysts QN-4, 5, 6, wherein the squaramide bears a methylene group, gave slight improvements compared to dhQN-2 (1:2.7-2.8 vs. 1:2.4), the more bulky tert-butyl substituted QN-7 provided a lower d.r. (1:1.9). The similar performances of the prototypical²² 3,5-bis(trifluoromethyl)benzyl catalyst QN-4 and the simple benzyl derivative QN-5 point to a negligible influence of the electronics of this group on selectivity. Thus, aiming at reducing bulkiness, catalyst QN-8 derived from methylamine was applied, providing indeed a rewarding improvement (anti/ syn = 1:3.1). A further reduction in bulkiness could be achieved only by entirely removing the N-substituent, which was finalized preparing and testing the ammonia derived catalyst QN-9. Pleasingly, this peculiar and unprecedented structure was able to afford syn-4d with a notable 1: 4.6 selectivity. The very

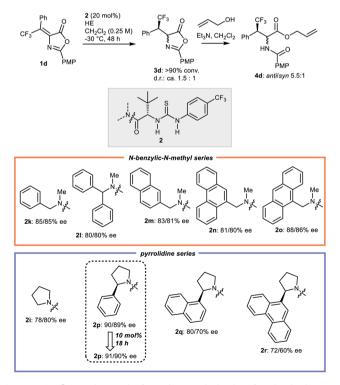


Scheme 2 Optimization of the alcoholytic ring-opening step: selected results (d.r. values refer to anti/syn ratios)

poor solubility of QN-9 in CH₂Cl₂ resulted however in sluggish reactivity, with only 50% conversion after 42 h at RT. Such shortcoming was overcome by switching to the more soluble dihydroquinine derivative dhQN-10, which gave 4d with >90% conversion, even by performing the reaction at 0 °C, and in a 1:5.9 diastereomeric ratio favouring the syn-isomer.23 Enantiomeric excess was found to be good (92%), as expected. At this stage, additional experiments indicated the unique requirements of the syn-selective reaction: catalyst dhQD-10, quasienantiomeric of dhQN-10, gave an anti-selectivity in the process comparable to an achiral tertiary amine (ca. 5:1).

Aiming at streamlining the overall process $(1d \rightarrow 4d)$ by implementing a one pot procedure, thus circumventing the problematic purification of azlactone intermediate 3d, it was found that an excess of HE in the transfer hydrogenation reaction has to be avoided, since this species inhibits the basic squaramide catalyst used in the ring-opening step (see ESI†). In contrast, the other components of the transfer hydrogenation (thiourea catalyst 2k and pyridine co-product) do not interfere with the second step. Fortunately, it was possible to drive the transfer hydrogenation reaction to completion even by using just 1.1 equiv. of HE. Ultimately, this modification was sufficient to develop an efficient one-pot procedure.

Then, in line with the notion that enhancing the enantiopurity of 3d would result in additional improvement of synselectivity, 19 an additional round of optimization of the catalyst used in the hydrogen transfer step was undertaken (Scheme 3).



Scheme 3 Second round of catalyst optimisation for the hydrogen transfer step: identification of optimal catalyst 2p.

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Fig. 2 Tentative transition state model for the transfer hydrogenation

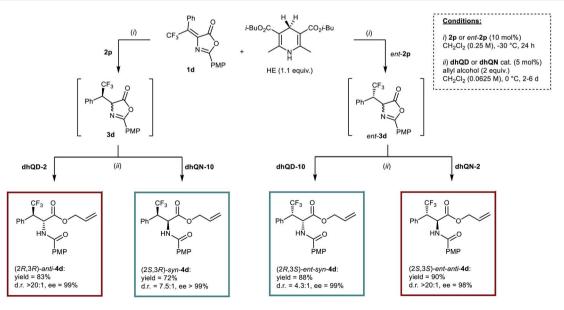
Different N-benzylic derivatives 21-o related to 2k, and more elaborated Jacobsen catalysts bearing chiral 2-aryl pyrrolidin-1yl amides²⁴ 2p-r were applied, promptly leading to improvements. Indeed, compared to N-benzyl catalyst 2k, the related 9anthracenylmethyl derived structure 20 and the (2R)-2-phenylpyrrolidine amide 2p afforded azlactone 3a with higher enantioselectivities, with catalyst 2p providing better results (91/90% ee), even at lower catalyst loading (10 vs. 20 mol%), and in shorter reaction time (18 vs. 48 h).

A tentative transition state picture can be built from a computationally validated model for the transfer hydrogenation of nitroalkenes with HEs catalysed by Jacobsen-type thiourea catalysts (2),17c complemented with recognition studies of lactones by a thiourea.25 Coordination of the acidic thiourea hydrogens to the lactone moiety, possibly assisted by its N-aryl group, and simultaneous stabilisation of the positive charge on the HE by the amide oxygen, are the key interactions between catalyst and substrates (Fig. 2). The tert-butyl group serves to "lock" the conformation of the catalyst as shown, thus leading to a match between the catalyst polar functionalities and a transition state leading to (3R)-3d. While this model does not

help rationalizing the subtle effects of the amide and the thiourea aryl groups on the enantioselectivity of the reaction, it reconciles with the observed comparably high, but opposite, sense of enantioinduction exerted by catalyst 2p on the two isomeric olefins (i.e. Z-1 \rightarrow (3R)-3, and E-1 \rightarrow (3S)-3, see ESI†). From the experimental results shown in Scheme 3 (compare 2p with 2q and 2r), the often encountered positive relationship between the extension of the π -system of the 2-substituent of the pyrrolidine and the enantioselectivity24b is not apparent. Stabilising cation- π interactions might not be helpful to selectivity in this case.

The stage was thus set for the full unravelment of the stereodivergent methodology (Scheme 4). It is clear from the results reported that the improvement in enantioselectivity provided by catalyst 2p in the first step was indeed beneficial to the diastereoselectivities of the whole processes. Its combination with catalyst dhQD-2 furnished anti-4d in good yield and in essentially diastereo- and enantiopure form, while use of "mismatched" dhQN-10 in the second step afforded syn-4d in 72% vield and >99% ee, with a notable 7.5:1 diastereomeric ratio. These results are to be compared with the 14.2:1 d.r. and 89% ee for anti-4d, and the 5.9: 1 d.r. and 92% ee for syn-4d obtained when catalyst 2k was used in the first step (Scheme 2). Scheme 4 shows also how the different combinations of catalysts (2p and ent-2p, dhQD-2 and dhQN-2, dhQD-10 and dhQN-10) could permit the obtainment of the full set of stereoisomeric products 4d with moderate to excellent results. Moreover, although the one pot protocol required longer reaction times (2-6 days instead of 24 h), for the ring opening step, the Cinchona loading could be lowered to 5 mol% without affecting the selectivity of the process.

The scope of the one-pot procedure was then studied (Table 2), by first applying a range of β-trifluoromethyl Erlenmeyer– Plöchl azlactones 1d-j, bearing electron-donating (1g, i) or



Scheme 4 Diastereodivergent, enantioselective synthesis of the whole set of stereoisomers of 4d by applying different catalysts combinations in the one-pot process.

Table 2 Reaction scope of the anti-4 and syn-4 selective processes

			Anti-4 selective process ^a			Syn-4 selective process ^a				
Entry	1: R ₁ , R _f	R_2	Anti-4	Yield ^b (%)	Anti/syn ^c	ee^d (%)	Syn-4	Yield ^b (%)	Syn/anti ^c	ee ^d (%)
1	1d : C ₆ H ₅ , CF ₃	Allyl	anti-4d	83	>20:1	99	syn-4d	72	7.5:1	>99
2	1e : 4-BrC ₆ H ₄ , CF ₃	Allyl	anti- 4e	93	>20:1	98	syn- 4e	72	3.6:1	99
3	1f : 4-ClC ₆ H ₄ , CF ₃	Allyl	anti- 4f	90	>20:1	99	syn- 4f	77	4.4:1	99
4	1g: 3-MeC_6H_4 , CF_3	Allyl	anti- $\mathbf{4g}$	96	>20:1	98	syn- 4g	84	8.5:1	99
5	1h : 4-FC ₆ H ₄ , CF ₃	Allyl	anti- 4h	88	>20:1	98	syn - 4h e	78	4.4:1	99
6	1i : 4-MeOC ₆ H ₄ , CF ₃	Allyl	anti- $\mathbf{4i}^{e,g}$	90	>20:1	98	syn- 4i ^{e,f}	60	3.7:1	>99
7	1j : 3,5-F ₂ C ₆ H ₃ , CF ₃	Allyl	anti- 4j	85	>20:1	99	syn- 4j ^e	87	4.0:1	>99
8	1k : 2-thienyl, CF ₃	Allyl	anti- 4k	78	>20:1	97	syn - 4k g	37	2.1:1	99
9	1l: N-Ts-indol-3-yl, CF ₃	Allyl	anti-41	82	>20:1	96	syn- 4l g	53	2.3:1	>99
10	1m : C ₆ H ₅ , CF ₃ CF ₂ CF ₂	Allyl	anti- 4m	69	>20:1	98	syn -4 $\mathbf{m}^{g,h}$	50	2.0:1	>99
11^{i}	1n: cyclohexyl, CF ₃	Allyl	anti- $4n^e$	98	16.7:1	99	syn- $4n^{e,f}$	65	3.6:1	>99
12^i	10 : Et, CF ₃	Allyl	anti- 40	98	15.3:1	97	$syn-4o^e$	78	2.6:1	>99
13^{i}	1p : Me, CF ₃	Allyl	anti- 4p	97	10.1:1	89	syn - 4p e	85	1.3:1	99
14	1d : C ₆ H ₅ , CF ₃	Me	anti- 4q	91	>20:1	97	syn- 4q	77	5.3:1	>99
15	1d : C_6H_5 , CF_3	Bn	anti- 4r	81	18:1	98	$syn-4\mathbf{r}^{e}$	70	5.9:1	99
16	1d : C ₆ H ₅ , CF ₃	<i>i</i> -Bu	anti- $4s^e$ e	50	>20:1	96	$syn-4s^{e,g}$	50	6.7:1	99

^a Conditions: **1** (0.15 mmol), HE (0.165 mmol, 1.1 equiv.), **2p** (0.015 mmol, 10 mol%), CH₂Cl₂ (0.60 mL), -30 °C, 24-48 h, then CH₂Cl₂ (1.8 mL), **dhQD-2** for *anti-4* or **dhQN-10** for *syn-4* (0.0075 mmol, 5 mol%), R²OH (0.30 mmol, 2 equiv.), 0 °C, 2-8 d. ^b Isolated yield of combined diastereoisomers **4** after chromatography on silica gel. ^c Determined on the crude mixtures by ¹⁹F NMR spectroscopy. ^d Enantiomeric excess of major diastereoisomer, determined by CSP HPLC. ^e In the ring-opening step, after 2-5 d, additional catalyst **dhQD-2** for *anti-4* or **dhQN-10** for *syn-4* (0.0075 mmol) and R²OH (0.15 mmol, 1-2 equiv.), were added. ^f Ring-opening step warmed to RT after 2 d. ^g Two step reaction performed by isolating intermediate 3 by a rapid filtration on silica gel. ^h Reduction step performed at 0 °C. ⁱ Conditions: **1** (0.05 mmol), HE (0.055 mmol, 1.1 equiv.), **2p** (0.01 mmol, 20 mol%), CH₂Cl₂ (0.300 mL), -20 or 0 °C, 24-48 h, then CH₂Cl₂ (0.60 mL), **dhQD-2** for *anti-4* or **dhQN-10** for *syn-4* (0.01 mmol, 10 mol%), allyl alcohol (0.1 mmol, 2 equiv.), 0 °C, 3-6 d.

electron-withdrawing (1e, f, h, j) groups at the β -aryl ring, and β heteroaromatic substituents (1k, l). Entries 1-9 show that these substrates behaved very well in the anti-selective reaction, providing the corresponding anti-4d-l with results comparable to the parent anti-4d, that is, in good yields and outstanding diastereo- and enantioselectivities. The syn-selective processes provided variable results in terms of diastereoselectivities, ranging from a fully satisfactory 8.5: 1 value for product syn-4g to less pleasing ca. 2:1 results for the β -heteroaromatic derivatives syn-4k and syn-4l. The latter results can be ascribed to a very high substrate bias towards anti-4k, I in the ring-opening process (>10:1 employing Et₃N), rather than to poor catalyst dhQN-10 efficiency. Syn-4k and syn-4l were also obtained in lower yields compared to the other compounds. Nevertheless, the enantiomeric excesses of the major syn-4 isomers were found to be excellent in all cases examined (≥99% ee). Substrate **1m** bearing a β-perfluoro residue rendered results similar to the β-heteroaromatic derivatives 1k and 1l, that is, excellent selectivity in the anti-4m isomer, and moderate yield and diastereoselectivity, but with >99% ee, for the syn-4m diastereoisomer (entry 10). The application of β -aliphatic substrates 1n-p required an adjustment to the conditions used in the transfer hydrogenation step, which was performed with higher (20 mol%) catalyst loading and at higher temperatures (-20 °C for the ethyl and methyl derivatives **10** and **1p**, 0 °C for the more hindered cyclohexyl counterpart **1n**). With these adjustments, it was possible to obtain *anti*-**4n**-**p** with good selectivities, while the results for *syn*-**4n**-**p** vary from the satisfactory level of *syn*-**4n**

anti-4j
$$\overrightarrow{i}$$
 \overrightarrow{i} \overrightarrow{i}

Scheme 5 Conversion of the catalytic products 4j to the corresponding amino alcohol hydrochlorides anti-6 and syn-6. (i) NaBH₄, THF/H₂O, 0 °C \rightarrow RT, then column chromatography on silica gel. (ii) HCl, MeOH/H₂O, reflux, then work up and evaporation.

to the less pleasing *syn*-selectivity for **4p** (entries 11–13). The latter result was ascribed to the deleterious combination of moderate enantioselectivity in the transfer hydrogenation step (*ca.* 70% ee) with high substrate bias in the ring-opening step (*ca.* 10:1). The last three entries 14–16 of Table 2 display the results obtained by applying alcohols other than allyl in the alcoholytic process with substrate **1d**. The peculiarity of the present reaction system makes the tolerance to different primary alcohols, known for the DKR of simple azlactones, ²⁰ less than obvious, especially in the case of the *syn*-selective protocol. However, it was pleasing to observe that results in line with the allyl derivatives **4d** were obtained for the products of methyl, benzyl and isobutyl alcohols **4q–s**, although lower yields were observed in the latter case.

Compounds **4j** were separately subjected to a two-step reduction-hydrolysis sequence (Scheme 5), delivering the corresponding aminoalcohol hydrochlorides **6**, *via* amides **5**. It is worth stressing that neither *syn-***6** – intermediate *en route* to the drug candidate (see Scheme 1(b))^{9a} – nor *anti-***6** can be easily accessed by conventional asymmetric hydrogenation.^{7d}

Conclusions

We have proved that the conceptually new combination of two catalytic processes (transfer hydrogenation - dynamic ringopening) on Erlenmeyer-Plöchl azlactones can provide a new stereodivergent strategy to enantioenriched β-branched AAs. The realization of this tactic with trifluoromethylated substrates has disclosed a one-pot entry to β-aryl-β-trifluoromethyl AA derivatives. Using the appropriate catalyst combination, the anti-bias of the ring-opening reaction was leveraged, giving antiproducts with excellent stereoselectivities (d.r. up to >20:1, ee \geq 89%). The scope of this reaction includes substrates reluctant to enantioselective hydrogenation. A newly designed ammonia derived squaramide catalyst afforded the syn-isomers, not by hydrogenation, with variable eoselectivities (d.r. up to 8.5:1) and high enantioselectivities $(ee \ge 99\%).$

Author contributions

Conceptualization and supervision: VC and LB. Investigation and methodology: all authors. Writing - original draft: LB. Writing - review and editing: VC, MF, LB. Funding acquisition, project administration and resources: MF and LB.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

§ All through the paper, to identify the diastereomers of the β -branched β -trifluoromethyl AA derivatives, we have used Masamune's *syn* and *anti* descriptors, arbitrarily setting the β -aryl/alkyl group of these compounds in the main chain. Using CIP descriptors for the relative configuration (*e.g.* R^* , S^*), although more rigorous, would result in a less clear identification of the diastereoisomeric pairs.

- \P For a more comprehensive list of screening results, see ESI.†
- || For limitations in terms of substrate variations, see ESI.†
- 1 J. Michaux, G. Niel and J.-M. Campagne, *Chem. Soc. Rev.*, 2009, **38**, 2093.
- 2 (a) B.-H. Zheng, C.-H. Ding, X.-L. Hou and L.-X. Dai, Org. Lett., 2010, 12, 1688; (b) J. Wang, S. Zhou, D. Lin, X. Ding, H. Jiang and H. Liu, Chem. Commun., 2011, 47, 8355; (c) W.-D. Chu, L.-F. Zhang, X. Bao, X.-H. Zhao, C. Zeng, J.-Y. Du, G.-B. Zhang, F.-X. Wang, X.-Y. Ma and C.-A. Fan, Angew. Chem., Int. Ed., 2013, 52, 9229; (d) S. Lou, G. M. McKenna, S. A. Tymonko, A. Ramirez, T. Benkovics, D. A. Conlon and F. González-Bobes, Org. Lett., 2015, 17, 5000; (e) X.-Z. Zhang, Y.-H. Deng, X. Yan, K.-Y. Yu, F.-X. Wang, X.-Y. Ma and C.-A. Fan, J. Org. Chem., 2016, 81, 5655; (f) F.-S. He, J.-H. Jin, Z.-T. Yang, X. Yu, J. S. Fossey and W.-P. Deng, ACS Catal., 2016, 6, 652; (g) Y.-C. Gong, Y. Wang, E.-Q. Li, H. Cui and Z. Duan, Adv. Synth. Catal., 2019, 361, 1389; (h) M. Winter, R. Schütz, A. Eitzinger, A. R. Ofial and M. Waser, Eur. J. Org. Chem., 2020, 3812; (i) V. A. Soloshonok, T. Yamada, K. Sakaguchi and Y. Ohfune, Future Med. Chem., 2009, 1, 897.
- 3 (a) S.-Y. Zhang, Q. Li, G. He, W. A. Nack and G. Chen, J. Am. Chem. Soc., 2013, 135, 12135; (b) G. Chen, T. Shigenari, P. Jain, Z. Zhang, Z. Jin, J. He, S. Li, C. Mapelli, M. M. Miller, M. A. Poss, P. M. Scola, K.-S. Yeung and J.-Q. Yu, J. Am. Chem. Soc., 2015, 137, 3338, see also: (c) B. V. S. Reddy, L. R. Reddy and E. J. Corey, Org. Lett., 2006, 8, 3391.
- 4 (a) J. Legters, L. Thijs and B. Zwanenburg, *Recl. Trav. Chim. Pays-Bas*, 1992, 111, 16; (b) S. C. Valdez and J. L. Leighton, *J. Am. Chem. Soc.*, 2009, 131, 14638; (c) A. Armstrong and A. Ferguson, *Beilstein J. Org. Chem.*, 2012, 8, 1747, see also: (d) D. Wilcke, E. Herdtweck and T. Bach, *Chem.-Asian J.*, 2012, 7, 1372.
- 5 C. E. Boville, R. A. Scheele, P. Koch, S. Brinkmann-Chen, A. R. Buller and F. H. Arnold, *Angew. Chem., Int. Ed.*, 2018, 57, 14764.
- 6 C. Liao and F. P. Seebeck, *Angew. Chem., Int. Ed.*, 2020, **59**, 7184.
- 7 (a) M. J. Burk, M. F. Gross and J. P. Martinez, J. Am. Chem. Soc., 1995, 117, 9375; (b) R. S. Haermer, D. Askin, R. P. Volante and P. J. Reider, Tetrahedron Lett., 1998, 39, 3455; (c) M. J. Burk, l. K. M. Bedingfield, W. F. Kiesman and J. G. Allen, Tetrahedron Lett., 1999, 40, 3093; (d) C. Benhaim, L. Bouchard, G. Pelletier, J. Sellstedt, L. Kristofova and S. Daigneault, Org. Lett., 2010, 12, 2008;

- (e) C. Molinaro, J. P. Scott, M. Shevlin, C. Wise, A. Ménard, A. Gibb, E. M. Junker and D. Lieberman, *J. Am. Chem. Soc.*, 2015, 137, 999.
- 8 (*a*) M. Salwiczek, E. K. Nyakatura, U. I. M. Gerling, S. Ye and B. Koksch, *Chem. Soc. Rev.*, 2012, **41**, 2135; (*b*) J. Moschner, V. Stulberg, R. Fernandes, S. Huhmann, J. Leppkes and B. Koksch, *Chem. Rev.*, 2019, **119**, 10718.
- 9 For another approach to a specific β-aryl-β-trifluoromethyl AA, wherein asymmetric hydrogenation failed: (*a*) A. Alimardanov, A. Nikitenko, T. J. Connolly, G. Feigelson, A. W. Chan, Z. Ding, M. Ghosh, X. Shi, J. Ren, E. Hansen, R. Farr, M. MacEwan, S. Tadayon, D. M. Springer, A. F. Kreft, D. M. Ho and J. R. Potoski, *Org. Process Res. Dev.*, 2009, **13**, 1161, for alternative approaches to β-alkyl-β-trifluoromethyl AAs, see: (*b*) X. Xing, A. Fichera and K. Kumar, *J. Org. Chem.*, 2002, **67**, 1722, correction: *J. Org. Chem.*, 2002, **67**, 8290; (*c*) J. A. Pigza, T. Quach and T. F. Molinski, *J. Org. Chem.*, 2009, **74**, 5510, see also ref. 2h.
- 10 (a) A. Enright, F. R. Alexandre, G. Roff, I. G. Fotheringham, M. J. Dawson and N. J. Turner, *Chem. Commun.*, 2003, 2636; (b) G. J. Roff, R. C. Lloyd and N. J. Turner, *J. Am. Chem. Soc.*, 2004, **126**, 4098, for a stereodivergent approach to β,β-diaryl AAs, see ref. 7e.
- 11 Reviews on diastereodivergent enantioselective catalysis: (a) G. Zhan, W. Du and Y.-C. Chen, Chem. Soc. Rev., 2017, 46, 1675; (b) L. Lin and X. Feng, Chem.-Eur. J., 2017, 23, 6464; (c) M. Bihania and J. C.-G. Zhao, Adv. Synth. Catal., 2017, **359**, 534; (*d*) S. Krautwald and E. M. Carreira, *J. Am. Chem.* Soc., 2017, 139, 5627; (e) I. P. Beletskaya, C. Nájera and M. Yus, Chem. Rev., 2018, 118, 5080, for a landmark stereodivergent olefin functionalisation by sequential reactions, see; (f) B. Simmons, A. M. Walji and D. W. C. MacMillan, Angew. Chem., Int. Ed., 2009, 48, 4349, for recent relevant examples, see: (g) S.-L. Shi, Z. L. Wong and S. L. Buchwald, Nature, 2016, 532, 353; (h) B. M. Trost, C.-I. Hung, T. Saget and E. Gnanamani, Nat. Catal., 2018, 1, 523; (i) X.-J. Liu, S. Jin, W.-Y. Zhang, Q.-Q. Liu, C. Zheng and S.-L. You, Angew. Chem., Int. Ed., 2020, 59, 2039; (j) Masson-Makdissi, L. Prieto, X. Abel-Snape and M. Lautens, Angew. Chem., Int. Ed., 2021, DOI: 10.1002/ anie.202105800, early view; (k) D.-X. Zhu, J.-G. Liu and M.-H. Xu, J. Am. Chem. Soc., 2021, 43, 8583.
- 12 Recent reviews on enantioselective transfer hydrogenation with Hantzsch esters: (a) W. Dong and D. Astruc, *Chem. Rev.*, 2015, **115**, 6621; (b) R. P. Herrera, *Top. Curr. Chem.*, 2016, **374**, 29; (c) A. M. Faisca Phillips and A. J. L. Pombeiro, *Org. Biomol. Chem.*, 2017, **15**, 2307.
- 13 For stereocontrolled formation of trifluoromethylated chirality centres, see: (a) J. Nie, H.-C. Guo, D. Cahard and J.-A. Ma, *Chem. Rev.*, 2011, 111, 455; (b) P. Poutrel, M. V. Ivanova, X. Pannecoucke, P. Jubault and T. Poisson, *Chem.-Eur. J.*, 2019, 25, 15262, and references cited therein.
- 14 (a) A. Berkessel, F. Cleemann, S. Mukherjee, T. N. Müller and J. Lex, *Angew. Chem., Int. Ed.*, 2005, 44, 807, for reviews, see:;
 (b) A.-N. R. Alba and R. Rios, *Chem.-Asian J.*, 2011, 6, 720; (c)
 Z. Rodríguez-Docampo and S. J. Connon, *ChemCatChem*, 2012, 4, 151; (d) P. P. de Castro, A. G. Carpanez and

- G. W. Amarante, *Chem.-Eur. J.*, 2016, **22**, 10294; (*e*) I. F. S. Marra, P. P. de Castro and G. W. Amarante, *Eur. J. Org. Chem.*, 2019, 5830, for a more comprehensive list of references on DKR of azlactones by ring-opening, see ESI†.
- 15 T. J. Caggiano, K. M. Morris, B. L. Harrison, A. F. Kreft III, D. M. Kubrak and D. M. Springer, US Pat., US20090023801A1, 2009.
- 16 W.-G. Su, G. Dai, H. Jia, Z. Zhang, J. Weng, J. D. Venable, S. D. Bembenek, W. Chai, S. P. Meduna, J. M. Keith, W. Eccles, A. D. Lebsack, W. M. Jones and R. C. Smith, WO2016119707 A1, 2016.
- 17 (a) N. J. A. Martin, L. Ozores and B. List, J. Am. Chem. Soc., 2007, 129, 8976; (b) E. Martinelli, A. C. Vicini, M. Mancinelli, A. Mazzanti, P. Zani, L. Bernardi and M. Fochi, Chem. Commun., 2015, 51, 658; (c) E. Massolo, M. Benaglia, M. Orlandi, S. Rossi and G. Celentano, Chem.-Eur. J., 2015, 21, 3589; (d) T. Hostmann, J. J. Molloy, K. Bussmann and R. Gilmour, Org. Lett., 2019, 21, 10164; (e) L. Bernardi and M. Fochi, Molecules, 2016, 21, 1000, and references cited therein.
- 18 (a) A. G. Wenzel, M. P. Lalonde and E. N. Jacobsen, Synlett, 2003, 1919; (b) R. S. Klausen and E. N. Jacobsen, Org. Lett., 2009, 11, 887; (c) S. J. Zuend, M. P. Coughlin, M. P. Lalonde and E. N. Jacobsen, Nature, 2009, 461, 968; (d) S. J. Zuend and E. N. Jacobsen, J. Am. Chem. Soc., 2009, 131, 15358.
- 19 According to Masamune's multiplicity rule for double asymmetric synthesis, applied to **3a** with 85% ee, even a highly selective chiral catalyst in the ring-opening of simple azlactones (20:1) would give only a moderate 2.6:1 *syn/anti-***4a** ratio, while using enantiopure **3a** the d.r. would increase to 3.6:1 (*a*) S. Masamune, W. Choy, J. S. Petersen and L. R. Sita, *Angew. Chem., Int. Ed.*, 1985, **24**, 1; (*b*) D. Lotter, A. Castrogiovanni, M. Neuburger and C. Sparr, *ACS Cent. Sci.*, 2018, **4**, 656, for additional discussion, see ESI†.
- 20 (a) J. W. Lee, T. H. Ryu, J. S. Oh, H. Y. Bae and C. E. Song, *Chem. Commun.*, 2009, 7224; (b) J.-S. Oh, J.-W. Lee, T. H. Ryu, J. H. Lee and C. E. Song, *Org. Biomol. Chem.*, 2012, 10, 1052; (c) S. Tallon, F. Manoni and S. J. Connon, *Angew. Chem., Int. Ed.*, 2015, 54, 813.
- 21 According to the Horeau effect (or Horeau amplification), in a sequential process, like the present one, the enantiomeric excess of the major diastereomer is higher than the enantiomeric excess of the product obtained in the first step [(ee (int.)]. It can be understood by considering that the minor enantiomer produced in the first step is converted, during the second step, to the minor diastereoisomer. In simplified terms, for a reaction in which the selectivity (s) of the second step is the same on both enantiomers obtained in the first step, the ee of the major diastereoisomer can be derived from the following equation: ee = [ee (int.) + s]/[1 + s \times ee(int.)] However, the reaction under study represents a more complex situation, since the second step occurs with different selectivities (s, s') for (3R)-3d and (3S)-3d, due to substrate bias. For a rigorous treatment, based on: (a) A. M. Harned,

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- *Tetrahedron*, 2008, **74**, 3797; (*b*) S. E. Baba, K. Sartor, J. Poulin and H. Kagan, *Bull. Soc. Chim. Fr.*, 1994, **131**, 525, see ESI†.
- 22 J. P. Malerich, K. Hagihara and V. H. Rawal, *J. Am. Chem. Soc.*, 2008, **130**, 14416.
- 23 For reaction models built on the computational study of the alcoholytic ring-opening of an alanine derived azlactone, reported in: A. Berkessel and K. Etzenbach-Effers, in
- Hydrogen Bonding in Organic Synthesis, ed. P. M. Pihko, Wiley-VCH, Weinheim, 2009, ch. 3, pp. 15–42, see ESI†.
- 24 (a) S. E. Reisman, A. G. Doyle and E. N. Jacobsen, J. Am. Chem. Soc., 2008, 130, 7198; (b) R. R. Knowles, S. Lin and E. N. Jacobsen, J. Am. Chem. Soc., 2010, 132, 5030.
- 25 K. M. Lippert, K. Hof, D. Gerbig, D. Ley, H. Hausmann, S. Guenther and P. R. Schreiner, *Eur. J. Org. Chem.*, 2012, 5919.