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Haemoglobin(β K120C)–albumin trimer as an artificial O₂ carrier with sufficient haemoglobin allostery†

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The allosteric O₂ release of haemoglobin (Hb) allows for efficient O₂ delivery from the lungs to the tissues. However, allostery is weakened in Hb-based O₂ carriers because the chemical modifications of the Lys- and Cys- β 93 residues prevent the quaternary transition of Hb. In this paper, we describe the synthesis and O₂ binding properties of a recombinant Hb [rHb(β K120C)]–albumin heterotrimer that maintains sufficient Hb allostery. The rHb(β K120C) core, with two additional cysteine residues at the symmetrical positions on its protein surface, was expressed using yeast cells. The mutations did not influence either the O₂ binding characteristics or the quaternary transition of Hb. Maleimide-activated human serum albumins (HSAs) were coupled with rHb(β K120C) at the two Cys- β 120 positions, yielding the rHb(β K120C)–HSA₂ trimer, in which the Cys- β 93 residues were unreacted. Molecular dynamics simulation demonstrated that the HSA moiety does not interact with the amino acid residues around the haem pockets and the $\alpha_1\beta_2$ surfaces of the rHb(β K120C) core, the alteration of which retards Hb allostery. Circular dichroism spectroscopy demonstrated that the quaternary transition between the relaxed (R) state and the tense (T) state of the Hb core occurred upon both the association and dissociation of O₂. In phosphate-buffered saline solution (pH 7.4) at 37 °C, the rHb(β K120C)–HSA₂ trimer exhibited a sigmoidal O₂ equilibrium curve with the O₂ affinity and cooperativity identical to those of native Hb (p_{50} = 12 Torr, n = 2.4). Moreover, we observed an equal Bohr effect and 2,3-diphosphoglycerate response in the rHb(β K120C)–HSA₂ trimer compared with naked Hb.

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Introduction

Allosteric regulation plays a crucial role in controlling the protein function in biological systems through the binding of a small-molecule effector that induces a conformational change. Haemoglobin (Hb) is a well-known allosteric protein encapsulated in red blood cells (RBCs),^{1–5} and its allostery has been extensively studied to understand its mechanism.^{6–12} O₂ binding converts the quaternary structure of Hb from a deoxy tense (T) state with a low O₂ affinity to a relaxed (R) state with a high O₂ affinity.^{1–4} The homotropic allosteric effect, which is known as cooperativity, allows Hb to efficiently transport O₂ from the lungs to the peripheral tissues. As a heterotropic allosteric effect, 2,3-diphosphoglycerate (DPG) stabilizes the T-state conformation by creating salt bridges between the two β subunits to reduce O₂ affinity.^{13,14} Furthermore, O₂ affinity

decreases with decreasing pH (the Bohr effect), shifting the equilibrium toward the T state.^{11,15,16}

Artificial O₂ carriers are currently required as RBC substitutes for transfusion therapy because of worldwide shortages in blood supply.^{17,18} However, the stroma-free Hb splits to an $\alpha\beta$ dimer ($\alpha_2\beta_2 \rightarrow 2\alpha\beta$) in the bloodstream and is immediately expelled into urine. Moreover, Hb diffuses into the interstitial spaces between the endothelial cells and smooth muscle cells, causing hypertension by scavenging nitric oxide (NO; endothelial-derived vasodilator).^{19,20} Therefore, naked Hb is not available for use in transfusion therapy. Since the 1980s, several types of Hb-based O₂ carriers have been developed, including cross-linked Hb,^{21,22} polymerized Hb,^{23,24} and PEGylated Hb.^{25–27} However, these chemical modifications have markedly reduced cooperativity, the Bohr effect, and the DPG effect,¹⁸ and a modified Hb that maintains the original allosteric effects of Hb has never been reported.¹⁸ We previously synthesized a protein cluster as an RBC substitute consisting of an Hb molecule wrapped covalently in human serum albumins (HSAs), called an Hb–HSA_{*m*} cluster ($m = 2–4$, average $m = 3.0 \pm 0.2$).^{28,29} This protein cluster shows a long circulation lifetime in the bloodstream and a superior biocompatibility.^{30,31} The

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Fig. 1 (A) Recombinant Hb variant [rHb(β K120C)] with Lys- β 120 residues replaced by cysteines at the symmetrical positions. The mutation point was designed to avoid the haem pockets and $\alpha_1\beta_2$ interfaces. (B) Synthetic routes of maleimide-activated HSA (MA-HSA) and the rHb(β K120C)-HSA₂ trimer. The Cys-34 of the HSA was reacted with 1,6-bis(maleimido)hexane (BMH), resulting in MA-HSA. The two Cys- β 120 residues at symmetrical positions were reacted with MA-HSA to yield the rHb(β K120C)-HSA₂ trimer.

quaternary structural analysis of the Hb-HSA_m cluster revealed that the chemical modification of the Lys and Cys- β 93 residues of Hb by *N*-succinimidyl 3-maleimidopropionate as a cross-linker prevents the R-T quaternary transition and maintains the Hb core close to the R state.³² Thus, the Hb-HSA_m cluster demonstrated a low cooperativity, a low Bohr effect, and a low DPG effect. If the HSA shells are linked at appropriate positions on the Hb surface so that they do not affect the R-T quaternary motion, the resulting Hb-HSA_m conjugate maintains the original O₂ binding properties of the native Hb. The genetically introduced cysteine could act as a cross-linking point for maleimide-activated HSA (MA-HSA). We assume that the mutation of Hb residues, excluding the mutation of (i) the $\alpha_1\beta_2$ interfaces, including the N- and C-terminals (which play important roles in the R-T quaternary transition), and (ii) the haem pockets (which directly affect O₂ affinity), would allow the resulting Hb variant to maintain its original O₂ binding capability. Based on this hypothesis, the Lys- β 120 residues were chosen as the symmetrical mutation points (Fig. 1A). In this study, we describe a novel recombinant Hb variant, rHb(β K120C), in which

Lys- β 120 is replaced with Cys at symmetrical positions on the Hb surface, and we also present the rHb(β K120C)-HSA₂ heterotrimer, in which MA-HSAs are linked at the Cys- β 120 positions (Fig. 1B). Ultraviolet-Visible (UV-vis) and circular dichroism (CD) spectroscopies revealed that the quaternary structure of the rHb(β K120C)-HSA₂ trimer is reversibly convertible between the R and T states. We also observed identical O₂ affinity and allosteric effects in the rHb(β K120C)-HSA₂ trimer compared with naked Hb.

Results and discussion

Expression and physicochemical properties of rHb(β K120C)

We first expressed rHb(β K120C) using *Pichia* yeast (*Pichia pastoris* GS115) according to our previous reports.^{33,34} The expressed rHb(β K120C) was purified using cation- and anion-exchange chromatographies (CEC and AEC). The sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE) analysis of rHb(β K120C) depicted two clear bands corresponding to the α - and β -subunits (Fig. 2A). The α - and β -chain bands demonstrated near-identical mobility to those of native Hb. The matrix-assisted laser desorption/ionization-time of flight (MALDI-TOF) mass spectrum analysis of rHb(β K120C) exhibited molecular-related ion peaks at 15 122 and 15 841 Da. These were almost identical to the simulated masses of the amino acid sequences (15 127 Da for the α -chain and 15 842 Da for the β -chain), indicating that Lys- β 120 was replaced with cysteine. The size-exclusion chromatography (SEC) chromatogram of rHb(β K120C) exhibited a peak with the same elution volume as that of native Hb, indicating that the tetramer ($\alpha_1\alpha_2\beta_1\beta_2$) was formed in the rHb(β K120C) (Fig. 2B; elution volume = 1.9 mL). Fortunately, the oligomeric rHb(β K120C) species created *via* the disulfide bonds of Cys- β 120 was not observed in the SEC chromatogram. The fact that rHb(β K120C) and native Hb had identical CD spectra (λ = 200–250 nm) indicates that their secondary structures were equivalent (Fig. 2C). The isoelectric point (pI) value of rHb(β K120C) was determined to be 6.8, which was slightly lower than that of native Hb (pI = 6.9) because the mutation (Lys- β 120 \rightarrow Cys) had negatively shifted the net surface charge (Fig. S1, ESI[†]). The results of the sulfhydryl group assay using 4,4'-dithiodipyridine (4,4'-DTP)³⁵ revealed the number of reactive sulfhydryl groups in rHb(β K120C) and native Hb as 4.0 and 2.0, respectively. This result strongly indicates that the two Lys- β 120 residues of the β_1 and β_2 chains were replaced by Cys and that the introduced Cys- β 120 residues were reactive with 4,4'-DTP, as they are located at the molecular surface of rHb(β K120C).

Preparation of the rHb(β K120C)-HSA₂ trimer

To prepare MA-HSA, we used 1,6-bis(maleimido)hexane (BMH) as the cross-linking agent. One of the two maleimide groups in BMH was reacted with the free sulfhydryl group of the Cys-34 in HSA (which is the only reduced-form of cysteine in HSA), yielding MA-HSA. The formation of the HSA dimer was not observed because of electrostatic repulsion between the negatively charged HSA surfaces present under our experimental conditions. The





Fig. 2 (A) SDS-PAGE results of native Hb and rHb(β K120C). (B) SEC profiles of rHb(β K120C) and native Hb. (C) CD spectra of the rHb(β K120C) and native Hb in PBS at 25 °C ([protein] = 0.2 μ M).

sulfhydryl group assay using 4,4'-DTP revealed the complete disappearance of the reduced form.

Capping of the Cys- β 93 residues of native Hb with maleimide derivatives causes a decline in O₂ binding cooperativity, the Bohr effect, and the influence of the allosteric effectors. The Cys- β 93 residues of native Hb were able to react with maleimide derivatives, such as *N*-ethyl maleimide (NEM) and *N*-succinimidyl 3-maleimidopropionate.^{32,36} We found that the Cys- β 93 residues had not reacted with MA-HSA because of the steric hindrance between the native Hb and MA-HSA. In contrast, genetically introduced Cys- β 120 located at the molecular surface of rHb(β K120C) had reacted with MA-HSA. The SEC chromatogram of the reaction mixture [rHb(β K120C) + MA-HSA] revealed a new peak in the high-molecular-weight region (Fig. 3A; elution volume = 1.4 mL), and native PAGE analysis also revealed

a new band above that of rHb(β K120C) (Fig. 3B). The high-molecular-weight component was purified using AEC followed by gel filtration chromatography (GFC). The ion peak of the β -chain at 15.8 kDa in the MALDI-TOF MS disappeared after the reaction of rHb(β K120C) with MA-HSA. A new peak then appeared at 82.5 kDa, a value almost identical to the simulated mass of a β -chain + MA-HSA (82.3 kDa), indicating that the HSA moiety was covalently linked to the β -chain of rHb(β K120C). Based on the [total protein]/[Hb unit] assays, the average HSA/Hb ratio of the product was determined to be 2.0. The heme loss was not observed during the preparation. We denoted this hybrid protein as rHb(β K120C)-HSA₂ (M_w : 198 kDa). The CD spectrum of the rHb(β K120C)-HSA₂ trimer coincided perfectly with the sum of the rHb(β K120C) spectrum and a two-times-enlarged HSA spectrum (Fig. 3C). This result demonstrates that the secondary structures



Fig. 3 (A) SEC profiles of the resultant reaction mixture [rHb(β K120C) + MA-HSA] and the purified rHb(β K120C)-HSA₂ trimer. (B) Native PAGE patterns of rHb(β K120C), HSA, the reaction mixture, and the purified rHb(β K120C)-HSA₂ trimer. (C) CD spectra of rHb(β K120C), HSA, rHb(β K120C), and rHb(β K120C)-HSA in PBS at 25 °C ([protein] = 0.2 μ M). (D and E) The energy-minimized structure of the rHb(β K120C)-HSA₂ trimer obtained by a molecular dynamics (MD) calculation using the Desmond software.



of the individual proteins were unaltered by the protein coupling. The dynamic light scattering measurements revealed that the hydrodynamic diameter of the rHb(β K120C)–HSA₂ trimer (12.4 ± 2.8 nm, P.I. = 0.010) was significantly larger than that of rHb(β K120C) (6.9 nm \pm 1.9 nm, P.I. = 0.004). The pI value of the trimer (pI = 5.4) was close to that of HSA (pI = 4.9), which further indicates the coupling of rHb(β K120C) (pI = 6.7) with HSAs (Fig. S1, ESI[†]). The net negative surface charge and the large molecular size could allow the rHb(β K120C)–HSA₂ trimer to circulate in the bloodstream for long periods of time and prevent the undesirable vasopressor response through NO depletion.^{19,20} Furthermore, the sulfhydryl group assay using 4,4'-DTP³⁵ demonstrated that the number of reactive sulfhydryl groups in the rHb(β K120C)–HSA₂ trimer was 1.8, indicating that the Hb core maintained the free Cys- β 93 residues that are important for O₂ binding ability, as mentioned above.³⁶ We could not observe the dissociation of the $\alpha_2\beta_2$ tetramer into two $\alpha\beta$ dimers during the synthesis, purification, and measurements of rHb(β K120C)–HSA₂, implying that the $\alpha_2\beta_2$ tetramer was sufficiently stable the same as native Hb under our experimental conditions.

An MD simulation was used to predict the interactions between the rHb(β K120C) and HSA moieties. The Cys- β 120 and Cys- β 212 residues exist at symmetrical positions of the core Hb; therefore, a sufficient distance is maintained between the two HSA parts in the rHb(β K120C)–HSA₂ trimer. We performed MD simulation using a half model [$\alpha\beta$ (K120C)–HSA] constructed using the crystal structures of native Hb (PDB ID: 2DN1)³⁷ and HSA (PDB ID: 1A06)³⁸ (Fig. S2, ESI[†]). The main chains (C α and amide atoms) of the $\alpha\beta$ (K120C) moiety were frozen for the duration of a 50 ns simulation. The root-mean-square deviation (RMSD) and the distances of the salt bridges and hydrogen bonds between the $\alpha\beta$ (K120C) and HSA residues reached a plateau within 25 ns (Fig. S3, ESI[†]). The MD simulation structure was represented in the rHb(β K120C)–HSA₂ form which was prepared by overlapping the two $\alpha\beta$ (K120C) moieties of the models with the crystal structure of native Hb ($\alpha_1\beta_1\alpha_2\beta_2$; Fig. 3D and Fig. S2B, ESI[†]). The distance between the sulfide atoms of Cys- β 120 [rHb(β K120C)] and Cys-34 (HSA) was measured to be 13.8 Å, meaning that the BMH was long enough to cross-link the Cys residues (Fig. 3E). We found that the following salt bridges and hydrogen bonding interactions between rHb(β K120C) and HSA helped stabilize the conformation of the rHb(β K120C)–HSA₂ trimer: (i) Glu- α 23, Glu- α 27, and His- α 112 of rHb(β K120C)–Arg-81 of HSA and (ii) Ser- α 49 and Asp- α 47 of rHb(β K120C)–Gln-94 of HSA. The HSA moiety did not interact with the amino acid residues around the haem pockets and the $\alpha_1\beta_2$ surfaces of the rHb(β K120C) core.

Quaternary structure analysis of the Hb core

We investigated the quaternary structure of the Hb core using UV-vis absorption and CD spectroscopies. The UV-vis spectral patterns and the absorbances of the oxy, deoxy, and carbonyl rHb(β K120C) as well as the rHb(β K120C)–HSA₂ trimer were indistinguishable from those of native Hb in PBS solution (pH 7.4; Fig. 4A and Table S1, ESI[†]). Generally, the chemical modification of the Lys and Cys- β 93 residues of Hb decreases



Fig. 4 (A) UV-vis absorption spectra of the carbonyl (gray), oxy (red), and deoxy (blue) rHb(β K120C)–HSA₂ trimer (solid line) and native Hb (dotted line). (B) CD spectra (240–320 nm) of the oxy (red) and deoxy (blue) Hb components of the rHb(β K120C)–HSA₂ trimer [rHb(β K120C)–HSA₂ – twofold HSA] (solid line) and rHb(β K120C) (dotted line). The measurements were performed in PBS solution (pH 7.4) at 25 °C ([protein] = 3 μ M).

the Soret band absorption of the deoxy form, because the deoxy Hb remains near the R-state quaternary structure.³² In contrast, the deoxy rHb(β K120C) and the rHb(β K120C)–HSA₂ trimer revealed the same Soret band absorption as that of deoxy native Hb, indicating that the identical T-state quaternary structures were formed. CD spectroscopy was also used to analyse the quaternary structure of the Hb core in the rHb(β K120C)–HSA₂ trimer. Generally, deoxy Hb (T state) exhibits a negative CD band at 287 nm, which is known as the T-state marker.^{39–41} At the same time, this band is not usually observed in the R-state quaternary structure of oxy Hb. The CD spectral patterns and intensities of oxy and deoxy rHb(β K120C) both resembled those of native Hb, implying that the oxy and deoxy forms of these proteins are comparable (Fig. 4B).³⁹ CD also revealed similar spectral changes in the rHb(β K120C)–HSA₂ trimer upon deoxygenation. CD spectra for the oxy and deoxy Hb cores in the rHb(β K120C)–HSA₂ trimer were obtained by subtracting the two-times-enlarged HSA spectrum from the corresponding rHb(β K120C)–HSA₂ trimer spectra (Fig. 4B). The CD spectra of the oxy and deoxy Hb cores coincided with those of rHb(β K120C). These spectral analyses revealed that the mutation Lys- β 120 \rightarrow Cys and coupling of HSA at the Cys- β 120 positions did



not influence the quaternary structural transition between the R and T states.

O₂ binding affinity and cooperativity

Hb binds with O₂ in the lungs, where there is a high O₂ partial pressure (p_{O_2}), and releases O₂ into the peripheral tissues, where there is a low p_{O_2} . To evaluate the O₂ binding properties, O₂ affinity (p_{50} ; p_{O_2} where Hb is half-saturated with O₂) and the Hill coefficient (n ; degree of cooperativity) of Hb were determined from the O₂ equilibrium curve (OEC). As shown in Fig. 5A, a sigmoidal OEC was observed for native Hb in PBS solution (pH 7.4) at 37 °C ($p_{50} = 12$ Torr; $n = 2.4$). Upon the release of O₂, the quaternary structure of Hb converts from the R state with a high O₂ affinity to the T state with a low O₂ affinity. The cooperative O₂ release allows Hb to efficiently deliver O₂ from the lungs to the tissues. Nevertheless, chemically

modified Hbs have demonstrated a low cooperativity because the quaternary transition of the Hb core is prevented.³² Such modified Hbs include PEG-Hb (71% reduction),^{18,42} polymerized Hb (41% reduction),^{18,42} and Hb-HSA_{*m*} clusters (71% reduction).^{29,42} In contrast, the OECs of rHb(βK120C) and the rHb(βK120C)-HSA₂ trimer were found to have a sigmoidal shape completely consistent with that of native Hb (Fig. 5A). The O₂ binding parameters of rHb(βK120C) ($p_{50} = 12$ Torr, $n = 2.3$) and the rHb(βK120C)-HSA₂ trimer ($p_{50} = 12$ Torr, $n = 2.4$) were also equivalent to those of native Hb (Table 1). Both rHb(βK120C) and the rHb(βK120C)-HSA₂ trimer were able to maintain identical O₂ affinity and cooperativity because the quaternary structure of the Hb units was interconvertible between the R and T states upon O₂ association and dissociation, as in native Hb. The capping of Cys-β93 with NEM in native Hb (NEM-Hb) reduced the p_{50} and n values ($p_{50} = 8$ Torr, $n = 1.9$). The NEM-rHb(βK120C)-HSA₂ trimer demonstrated O₂ binding parameters similar to those of NEM-Hb, clearly indicating that the Cys-β93 residues in the rHb(βK120C)-HSA₂ trimer were in a reduced form. This result is supported by the sulfhydryl group assay mentioned above. We concluded that the selective coupling of MA-HSA occurred at the two Cys-β120 positions of the rHb(βK120C), whereas it did not take place at the Cys-β93 residues because of steric hindrance. On the basis of these findings, we also concluded that the Lys-β120 → Cys mutation and the selective conjugation of MA-HSA at the Cys-β120 positions were impervious to O₂ affinity and cooperativity. To the best of our knowledge, this is the first example of a chemically modified Hb showing an identical O₂ affinity and cooperativity to those of native Hb.¹⁸

The autoxidation rate constant (k_{ox}) of the oxygenated Hb core in the rHb(βK120C)-HSA₂ trimer was measured in PBS solution (pH 7.4) at 37 °C. The k_{ox} value of the rHb(βK120C)-HSA₂ trimer was ascertained as 0.027 h⁻¹, which was almost similar to the data of native Hb (0.020 h⁻¹). The oxy form of the internal Hb maintains good stability even after being covered with HSAs.

Bohr effect on rHb(βK120C)-HSA₂ trimer

A high metabolic activity leads to an increase in the concentration of CO₂ and a decrease in the pH value within the tissues. Under low-pH conditions, the protonation of Hb occurs and the T-state quaternary structure is stabilized, reducing O₂ affinity (the Bohr effect). Thus, the OEC of native Hb was right-shifted to a p_{50} value of 28 Torr at pH 6.2 compared with $p_{50} = 12$ Torr at



Fig. 5 OECs and Bohr plots of Hb (dotted line) and the rHb(βK120C)-HSA₂ trimer (solid line) at 37 °C ([protein] = 5 μM). (A) OECs in PBS solutions at pH 7.4 (black) and pH 6.2 (red). (B) The Bohr plots of Hb (black circle) and the rHb(βK120C)-HSA₂ trimer (red triangle) in PBS solutions (pH 6.2, 6.5, 7.0, 7.4, and 8.0). (C) OECs in 50 mM Tris-HCl buffer solution with 1 mM DPG (blue) and without DPG (black).

Table 1 O₂ binding parameters and Bohr coefficients of haemoproteins in PBS solution at 37 °C

Haemoprotein	pH 7.4		pH 6.2		Bohr coefficient ^a
	p_{50} (Torr)	n (—)	p_{50} (Torr)	n (—)	
Hb	12	2.4	28	1.9	0.29
rHb(βK120C)	12	2.3	29	1.9	0.29
rHb(βK120C)-HSA ₂	12	2.4	28	1.9	0.29

^a The Bohr coefficients were derived from the slopes obtained from plotting $-\Delta \log p_{50} / \Delta \text{pH}$ between pH 6.2 and 8.0 (Fig. 5B).



Table 2 Effect of DPG on hemoproteins in 50 mM Tris-HCl buffer solution (pH 7.4) at 37 °C containing 1 mM DPG

Hemoprotein	Without DPG		With DPG	
	p_{50} (Torr)	n (—)	p_{50} (Torr)	n (—)
Hb	9	2.3	27	2.4
rHb(β K120C)	9	2.3	26	2.1
rHb(β K120C)-HSA ₂	9	2.3	26	2.1

pH 7.4 (Table 1 and Fig. 5A). The Bohr effect allows for Hb high O₂-transport efficiency to the tissues. However, chemical modifications have tended to notably reduce the Bohr effect, such as in PEG-Hb (87% reduction)¹⁸ and polymerized Hb (75% reduction).¹⁸ We observed that the O₂ affinities of rHb(β K120C) and the rHb(β K120C)-HSA₂ trimer were reduced under lower pH conditions, similar to native Hb (p_{50} : 12 → 29 Torr (pH 7.4 → 6.2)). The p_{50} and n values of both rHb(β K120C) and the rHb(β K120C)-HSA₂ trimer were nearly identical to those of native Hb at pH 6.2 (Table 1). The Bohr coefficients were derived from the slopes obtained from plotting $-\Delta \log p_{50} / \Delta \text{pH}$ between pH 6.2 and 8.0 (Fig. 5B and Table S2, ESI[†]). The Bohr coefficients of both rHb(β K120C) and the rHb(β K120C)-HSA₂ trimer were identical to those of native Hb (Table 1). The value measured for native Hb, however, was somewhat weaker than the result reported by Ho *et al.*⁴³ This difference is probably attributable to several factors, such as the difference in temperature (37 °C *vs.* 29 °C) and the difference in solvents (PBS with low phosphate and high NaCl concentrations *vs.* 0.1 M sodium phosphate buffer (PB)). Altogether, we conclude that neither the Lys- β 120 → Cys mutation nor the HSA conjugation affected the Bohr effect.

Effects of DPG on O₂ binding affinity

Allosteric effectors bind to proteins, altering their activity. The most important allosteric regulator of Hb in RBCs is DPG, which reduces the O₂ binding affinity (the DPG effect). An X-ray crystallography analysis has previously revealed that a single DPG molecule binds to the cleft of Hb β chains, known as the DPG binding site.^{13,14} The anionic groups of DPG form salt bridges with the cationic groups of the β -chains, including the His- β 2, Lys- β 82, and His- β 143 residues, stabilizing the T-state conformation and reducing O₂ affinity. The T → R conformational change induced by oxygenation leads to the closure of the binding site and weakens the DPG binding affinity to oxy Hb. This DPG response allows Hb to efficiently deliver O₂ from the lungs to the tissues. We measured the OEC of native Hb in PBS (pH 7.4) solution containing 1 mM DPG at 37 °C, but a reduction in O₂ affinity was not observed because of the fact that phosphate anions in the PBS solution interrupt the binding of DPG to Hb. Thus, we used 0.1 M Tris-HCl buffer for DPG response experiments. Indeed, native Hb in Tris-HCl buffer demonstrated a higher O₂ affinity (p_{50} = 9 Torr) than in PBS solution because of the exclusion of the Hb-phosphate anion interactions (Table 2). The O₂ affinity of the native Hb was also reduced upon the addition of DPG to Tris-HCl buffer. Interestingly, rHb(β K120C) (p_{50} = 26 Torr) and the rHb(β K120C)-HSA₂ trimer (p_{50} = 26 Torr) demonstrated identical DPG responses to

native Hb (p_{50} = 27 Torr; Fig. 5C). These results indicate that the rHb(β K120C)-HSA₂ trimer maintained a DPG response similar to that of native Hb and that its O₂ affinity is controllable *via* the allosteric effector.

Conclusions

A novel Hb variant (Lys- β 120 → Cys) was designed to prepare the rHb(β K120C)-HSA₂ trimer as an artificial O₂ carrier. The mutation did not affect the O₂ binding properties of Hb. Two MA-HSA molecules were selectively linked to Cys- β 120 in rHb(β K120C), resulting in the rHb(β K120C)-HSA₂ trimer in which Cys- β 93 remained free because of steric repulsion between the rHb(β K120C) core and the MA-HSA shell. The CD spectra revealed that the R-T quaternary motion of the Hb core in the rHb(β K120C)-HSA₂ trimer occurred upon O₂ association and dissociation. Thus, the rHb(β K120C)-HSA₂ trimer demonstrated sufficient allosteric O₂ binding properties with respect to cooperativity, the Bohr effect, and the DPG effect. The rHb(β K120C)-HSA₂ trimer is a unique artificial O₂ carrier with sufficient allostery to be an RBC substitute.

Materials and methods

Materials and apparatus

1,6-Bis(maleimido)hexane (BMH) was purchased from Tokyo Chemical Industry Co., Ltd. Human serum albumin (HSA, albumin 25%, Benesis) was purchased from the Japan Blood Products Organization. The other special-grade chemicals were used without additional purification unless otherwise noted. Water was deionized (18.2 M Ω cm) using two water purification systems (Elix UV and Milli Q Reference; Millipore Corp.). The SDS-PAGE and native PAGE analyses were performed using a 15% or 5–12% poly(acrylamide) precast gel (SuperSep Ace 15%; SuperSep Ace 5–12%; Fujifilm Wako Pure Chemical Corp.). Isoelectric focusing (IEF) was performed using a pH 3–10 IEF protein gel (Novex; Thermo Fischer Scientific Inc.). The UV-vis absorption spectra were obtained using a UV-Visible spectrophotometer (8543; Agilent Technologies Inc. or V-650; Jasco Corp.). CD spectra were recorded using a CD spectrometer (J-820; Jasco Corp.) at 25 °C.

Expression and preparation of rHb(β K120C)

The expression plasmid for the rHb(β K120C) variant [pPHIL-D2-rHb(β K120C)] was constructed according to the standard protocol of the QuickChange XL Site-Directed Mutagenesis Kit (Agilent Technologies Inc.) using a pPHIL-D2-rHb(α - and β -chains) vector³⁴ and an oligonucleotide primer set (forward [5'-CTCGCGCATCACTTCGGCTGTGAGTTTACACCGCCAGTTC-3'] and reverse [5'-GAACTGGCGGTGTAACTCACAGCCGAAGTGATGCGCGAG-3']). The obtained pPHIL-D2-rHb(β K120C) vector was linearized with *SalI* and used to transform the GS115-strain *Pichia pastoris* (Thermo Fisher Scientific K.K.) *via* electroporation.

Transformed clone cells were grown in a buffered mineral glycerol complex (BMGY) medium (4 L total) in a shaking



incubator (Bio-Shaker G-BR-200; Taitec Corp.; 200 rpm, 30 °C) and subsequently in a buffered mineral methanol complex (BMMY) medium (1.6 L total) containing 0.3 mM hemin for 5 days. During cultivation, 100% methanol at 1.5% of the medium volume was added every 24 h. The cells were harvested by centrifugation at 3000g for 10 min. The obtained cells were then washed with water (300 mL × 2) and resuspended in 100 mL of sodium PB solution (10 mM, pH 6.0) containing 1 mM phenylmethanesulfonyl fluoride. After the addition of glass beads (150 mL, $d = 0.5$ mm), the cells were lysed using a BeadBeater (Biospec Products, Inc.) with four cycles of disruption (2 min) and incubated by cooling on ice (2 min). After centrifugation at 12 000g for 1 h, the supernatant was equilibrated with CO atmosphere. The solution was loaded onto a cation exchange chromatography column (SP Sepharose Fast Flow, GE Healthcare UK Ltd), which was equilibrated in 10 mM PB (pH 6.0). After washing with the same buffer solution, the target protein was eluted using 20 mM Tris-HCl buffer solution (pH 8.0). Next, the resulting protein solution was subjected to AEC using a Q Sepharose Fast Flow column (GE Healthcare UK Ltd) with 20 mM Tris-HCl (pH 8.0) as the running buffer. After washing with 20 mM Tris-HCl (pH 8.0), rHb(β K120C) was eluted using PBS solution (pH 7.4). The purity was checked using SDS-PAGE analysis and SEC on a high-performance liquid chromatography (HPLC) system (AKTA purify; GE Healthcare UK Ltd). The system was equipped with an SEC column (Superdex 200 Increase, 5/150 GL; GE Healthcare UK Ltd), and PBS solution (pH 7.4) was used as the mobile phase. The concentration of rHb(β K120C) was measured using a protein assay kit (Pierce 660 nm; Thermo Fisher Scientific K.K.). The obtained rHb(β K120C) had a concentration of approximately 100 mg L⁻¹ of media. The sulfhydryl group assay of rHb(β K120C) was conducted by reaction with 4,4'-dithiodipyridine (4,4'-DTP).³⁵

Preparation of MA-HSA

The 25% HSA solution (4 mL) was diluted with PBS solution (14 mL, pH 7.4). A DMSO solution of BMH (7.5 mM, 2 mL) was then added dropwise to the HSA solution (8.4 mM, 18 mL). After stirring for 3 h at 25 °C, the reactant was subjected to a GFC column (Sephadex G25 superfine; GE Healthcare UK Ltd) to remove the unreacted cross-linker. The protein concentration was measured using a protein assay kit (Pierce 660 nm), whereas the sulfhydryl group assay of MA-HSA was conducted by reaction with 4,4'-DTP. The MA-HSA solution was stored at -80 °C.

Preparation of rHb(β K120C)-HSA₂

The MA-HSA and rHb(β K120C) solutions were mixed and concentrated to 6 mL ([MA-HSA] = 0.4 mM, [rHb(β K120C)] = 0.1 mM). The reactant was stirred under dark conditions for 24 h at 4 °C. An aliquot of the reaction mixture was analysed using SEC on an HPLC system (AKTA purify; GE Healthcare UK Ltd) equipped with an SEC column (Superdex 200 Increase, 5/150 GL; GE Healthcare UK Ltd) and using PBS solution (pH 7.4) as the mobile phase. The reaction mixture was purified using AEC using a Q Sepharose Fast Flow column (GE Healthcare UK Ltd). The mixture solution was diluted with the same volume of water

and loaded onto the column equilibrated with PBS solution. The column was then washed using 10 mM PB (pH 7.4) solution. The rHb(β K120C)-HSA₂ trimer was eluted with a 10 mM PB (pH 6.0) + 120 mM NaCl solution. Under this condition, the HSA dimer remained in the column. The collected solution was concentrated and subjected to GFC using a Superdex 200 pg column (GE Healthcare UK Ltd) to isolate the rHb(β K120C)-HSA₂ trimer (yield: 50%). The total protein and Hb concentrations were measured using a protein assay kit (Pierce 660 nm; Thermo Fischer Scientific Inc.) and the extinction coefficient of cyanometHb ($\epsilon_{541} = 4.4 \times 10^4$ M⁻¹ cm⁻¹), respectively. The average [HSA]/[rHb(β K120C) unit] ratio of the product was estimated to be 2.0. The sulfhydryl group assay of MA-HSA was conducted *via* the reaction with 4,4'-DTP.

Molecular dynamics simulations

MD simulations were performed using the Desmond application implemented in a Maestro graphical interface.^{44,45} We used the optimized potentials for liquid simulations (OPLS_2005) force field implemented in the Desmond software for all molecules in the system.⁴⁶ We constructed a half model of rHb(β K120C) [α (β K120C)-HSA model] because rHb(β K120C)-HSA₂ has a C₂ symmetry, and there has to be a sufficient distance between the HSAs for their interactions to be negligible. The crystal structure of native Hb (PDB ID: 2DN1)³⁷ was used for the α (β K120C) model. Lys- β 120 was replaced with Cys using the PyMOL software (PyMOL Molecular Graphics System Version 2.0, Schrödinger, LLC.). For the HSA moiety, we used the crystal structure of HSA (PDB ID: 1AO6).³⁸ The missing atoms of the side chains in the HSA were added to the model using the PyMOL software. The water molecules in the model were removed. The bond orders were assigned, and the hydrogen atoms were added to the models using the Protein Preparation Wizard tool⁴⁷ in the Maestro software. The α (β K120C) and HSA models were covalently linked using the BMH cross-linker [Cys- β 120(α)-BMH-Cys-34(HSA)] and arranged so that sufficient distance was maintained between them (Fig. S2A, ESI†). The Cys- β 120(Hb), Cys-34(HSA), and BMH moieties were minimized using the 3D builder in the Maestro software. The hydrogen networks and the protonated states were optimized to pH 7 using the Protein Preparation Wizard for which the pK_a values of the protein residues were predicted using PROPKA.⁴⁸ The α (β K120C)-HSA model was solvated using explicit TIP3P water⁴⁹ with a buffering distance of 10 Å in an octahedral box. Sodium counterions were added to neutralize the charges. The model was relaxed using a default relaxation protocol implemented in the Desmond software. An isothermal-isobaric MD simulation was performed for 50 ns to maintain the 300 K temperature and 1.01325 bar pressure. The main chains (C α and amide atoms) of the α (β K120C) moiety were frozen during the simulation. The time step was 2 fs, and the information for analysis was printed every 100 ps. A 10 Å cut off was used for non-bonded interactions.

Preparation of oxy and deoxy forms of the rHb(β K120C)-HSA₂ trimer

The oxy (O₂ complex) rHb(β K120C)-HSA₂ trimer solution (PBS, pH 7.4, 3 μ M, 3 mL) was prepared using our previously reported



technique.²⁹ The solution was transferred to an optical quartz cuvette (10 mm path length) with a rubber septum cap. N₂ gas was blown into the oxy-form solution to yield the deoxy rHb(β K120C)–HSA₂ trimer. The UV-vis absorption and CD spectra of these species were recorded at 25 °C.

O₂ binding parameters

The O₂ affinity (p_{50} ; O₂ partial pressure where Hb is half-saturated with O₂) and Hill coefficient (n) were determined using an automatic recording system for the O₂ equilibrium curve (Hemox Analyzer; TCS Scientific Corp.) at 37 °C. The oxy-form of the rHb(β K120C)–HSA₂ trimer solution (PBS, pH 7.4, approximately 5 μ M, 4 mL) was used for the measurements. The trimers in PBS solution (pH 7.4) were deoxygenated by flushing with N₂ and oxygenated by increasing the O₂ partial pressure.

O₂ complex stability

The oxy form stability of the rHb(β K120C)–HSA₂ trimer was assessed using the first-order autoxidation rate constant (k_{ox}) of the core Hb using our earlier described procedures.⁵⁰

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- M. F. Perutz, H. Muirhead, J. M. Cox and L. C. G. Goaman, *Nature*, 1968, **219**, 131–139.
- M. F. Perutz, *Nature*, 1970, **228**, 726–734.
- J. Baldwin and C. Chothia, *J. Mol. Biol.*, 1979, **129**, 175.
- G. Fermi, M. F. Perutz, B. Shaanan and R. Fourme, *J. Mol. Biol.*, 1984, **175**, 159.
- W. A. Eaton, E. R. Henry, J. Hofrichter and A. Mozzarelli, *Nat. Biol.*, 1999, **6**, 351–358.
- S. Adachi, S. Y. Park, J. R. H. Tame, Y. Shiro and N. Shibayama, *Proc. Natl. Acad. Sci. U. S. A.*, 2003, **100**, 7039–7044.
- E. M. Jones, G. Balakrishnan and T. G. Spiro, *J. Am. Chem. Soc.*, 2012, **134**, 3461–3471.
- C. Viappiani, S. Abbruzzetti, L. Ronda, S. Bettati, E. R. Henry, A. Mozzarelli and W. A. Eaton, *Proc. Natl. Acad. Sci. U. S. A.*, 2014, **111**, 12758–12763.
- N. Shibayama, K. Sugiyama, J. R. Tame and S. Y. Park, *J. Am. Chem. Soc.*, 2014, **136**, 5097–5105.
- E. M. Jones, E. Monza, G. Balakrishnan, G. C. Blouin, P. J. Mak, Q. H. Zhu, J. R. Kincaid, V. Guallar and T. G. Spiro, *J. Am. Chem. Soc.*, 2014, **136**, 10325–10339.
- Y. Yuan, M. F. Tam, V. Simplaceanu and C. Ho, *Chem. Rev.*, 2015, **115**, 1702–1724.
- N. Shibayama, A. Sato-Tomita, M. Ohki, K. Ichiyanagi and S. Y. Park, *Proc. Natl. Acad. Sci. U. S. A.*, 2020, **117**, 4741–4748.
- A. Arnone, *Nature*, 1972, **237**, 146–149.
- V. Richard, G. G. Dodson and Y. Mauguén, *J. Mol. Biol.*, 1993, **233**, 270–274.
- C. Ho and I. M. Russu, *Biochemistry*, 1987, **26**, 6299–6305.
- T.-Y. Fang, M. Zou, V. Simplaceanu, N. T. Ho and C. Ho, *Biochemistry*, 1999, **38**, 13423–13432.
- A. Mozzarelli, L. Ronda, S. Faggiano, S. Bettati and S. Bruno, *Blood Transfus.*, 2010, **8**, s59–s68.
- F. Meng, T. Kassa, S. Jana, F. Wood, X. Zhang, Y. Jia, F. D'Agnillo and A. I. Alayash, *Bioconjugate Chem.*, 2018, **29**, 1560–1575.
- S. C. Schultz, B. Grady, F. Cole, I. Hamilton and K. Burhop, *J. Lab. Clin. Med.*, 1993, **122**, 301–308.
- D. H. Doherty, M. P. Doyle, S. R. Curry, R. J. Vali, T. J. Fattor, J. S. Olson and D. D. Lemon, *Nat. Biotechnol.*, 1998, **16**, 672–676.
- R. Chatterjee, E. V. Welty, R. Y. Walder, S. L. Pruitt, P. H. Rogers, A. Arnone and J. A. Walder, *J. Biol. Chem.*, 1986, **261**, 9929–9937.
- E. Nagababu, S. Ramasamy, J. M. Rifkind, Y. Jia and A. I. Alayash, *Biochemistry*, 2002, **41**, 7407–7415.
- J. S. Jahr, M. Moallempour and J. C. Lim, *Expert Opin. Biol. Ther.*, 2008, **8**, 1425–1433.
- R. Kluger, J. S. Foot and A. A. Vandersteen, *Chem. Commun.*, 2010, **46**, 1194–1202.
- K. D. Vandegriff, A. Malavalli, J. Wooldridge, J. Lohman and R. M. Winslow, *Transfusion*, 2003, **43**, 509–516.
- B. M. Manjula, A. Tsai, R. Upadhyaya, K. Perumalsamy, P. K. Smith, A. Malavalli, K. Vandegriff, R. M. Winslow, M. Intaglietta, M. Prabhakaran, J. M. Friedman and A. S. Acharya, *Bioconjugate Chem.*, 2003, **14**, 464–472.
- D. Li, T. Hu, B. N. Manjula and S. A. Acharya, *Bioconjugate Chem.*, 2009, **20**, 2062–2070.
- D. Tomita, T. Kimura, H. Hosaka, Y. Daijima, R. Haruki, K. Ludwig, C. Böttcher and T. Komatsu, *Biomacromolecules*, 2013, **14**, 1816–1825.
- R. Funaki, T. Kashima, W. Okamoto, S. Sakata, Y. Morita, M. Sakata and T. Komatsu, *ACS Omega*, 2019, **4**, 3228–3233.
- R. Haruki, T. Kimura, H. Iwasaki, K. Yamada, I. Kamiyama, M. Kohno, K. Taguchi, S. Nagao, T. Maruyama, M. Otagiri and T. Komatsu, *Sci. Rep.*, 2015, **5**, 12778.
- H. Iwasaki, K. Yokomaku, M. Kureishi, K. Igarashi, R. Hashimoto, M. Kohno, M. Iwazaki, R. Haruki, M. Akiyama, K. Asai, Y. Nakamura, R. Funaki, Y. Morita and T. Komatsu, *Artif. Cells, Nanomed., Biotechnol.*, 2018, **46**, S621–S629.
- Y. Morita, T. Yamada, M. Kureishi, K. Kihira and T. Komatsu, *J. Phys. Chem. B*, 2018, **122**, 12031–12039.
- Y. Morita, K. Igarashi, R. Funaki and T. Komatsu, *ChemBioChem*, 2019, **20**, 1684–1687.
- R. Funaki, W. Okamoto, C. Endo, Y. Morita, K. Kihira and T. Komatsu, *J. Mater. Chem. B*, 2020, **8**, 1139–1145.



- 35 D. R. Grassetti and J. F. Murry Jr., *Arch. Biochem. Biophys.*, 1967, **119**, 41–49.
- 36 Y. Cheng, T. J. Shen, V. Simplaceanu and C. Ho, *Biochemistry*, 2002, **41**, 11901–11913.
- 37 S. Y. Park, T. Yokoyama, N. Shibayama, Y. Shiro and J. R. Tame, *J. Mol. Biol.*, 2006, **360**, 690–701.
- 38 S. Sugio, A. Kashima, S. Mochizuki, M. Noda and K. Kobayashi, *Protein Eng.*, 1999, **12**, 439–446.
- 39 M. F. Perutz, J. E. Ladner, S. R. Simon and C. Ho, *Biochemistry*, 1974, **13**, 2163–2173.
- 40 Y. Aki-Jin, Y. Nagai, K. Imai and M. Nagai, in *New Approaches in Biomedical Spectroscopy*, ed. K. Kneipp, R. Aroca, H. Kneipp and E. Wentrup-Byrne, American Chemical Society, Washington, DC, 1st edn, 2007, ch. 19, vol. 963, pp. 297–311.
- 41 M. Nagai, S. Nagatomo, Y. Nagai, K. Ohkubo, K. Imai and T. Kitagawa, *Biochemistry*, 2012, **51**, 5932–5941.
- 42 The cooperativity reduction was determined using the following equation, cooperativity reduction (%) = $(n_{\text{native Hb}} - n)/(n_{\text{native Hb}} - 1.0) \times 100$, where $n_{\text{native Hb}}$ and n are Hill coefficient of native Hb and chemically modified Hb, respectively, and are measured in the same conditions. When the cooperativity is completely lost, the Hill coefficient becomes 1.0 and the cooperativity reduction reaches 100%.
- 43 M. E. Wiltrout, J. L. Giovannelli, V. Simplaceanu, J. A. Lukin, N. T. Ho and C. Ho, *Biochemistry*, 2005, **44**, 7207–72017.
- 44 *Desmond Molecular Dynamics System*, D. E. Shaw Research, Maestro-Desmond Interoperability Tools, Schrödinger LLC, New York, NY, 2016.
- 45 *Maestro*, Schrödinger Release 2017-3, Schrödinger LLC, New York, NY, 2017.
- 46 J. L. Banks, H. S. Beard, Y. Cao, A. E. Cho, W. Damm, R. Farid, A. K. Felts, T. A. Halgren, D. T. Mainz, J. R. Maple, R. Murphy, D. M. Philipp, M. P. Repasky, L. Y. Zhang, B. J. Berne, R. A. Friesner, E. Gallicchio and R. M. Levy, *J. Comput. Chem.*, 2005, **26**, 1752–1780.
- 47 Schrödinger Suite 2012 *Protein Preparation Wizard*, LLC, New York, NY, 2012; *Impact version 5.8*, Schrödinger, LLC, New York, NY, 2012.
- 48 M. H. M. Olsson, C. R. Søndergard, M. Rostkowski and J. H. Jensen, *J. Chem. Theory Comput.*, 2011, **7**, 525–537.
- 49 W. L. Jorgensen, J. Chandrasekhar, J. D. Madura, R. W. Impey and M. L. Klein, *J. Chem. Phys.*, 1983, **79**, 926–935.
- 50 K. Yamada, K. Yokomaku, M. Kureishi, M. Akiyama, K. Kihira and T. Komatsu, *Sci. Rep.*, 2016, **6**, 36782.

