RSC Advances



View Article Online

View Journal | View Issue

PAPER

Check for updates

Cite this: RSC Adv., 2018, 8, 9718

Received 22nd January 2018 Accepted 1st March 2018

DOI: 10.1039/c8ra00659h

rsc.li/rsc-advances

Introduction

Compounds with R-S-S-R structures, where the R groups are alkyl, vinyl or aryl, are known as symmetrical disulfides if the R groups are the same. A large number of unsymmetrical disulfides, in which the R groups are different, are also well known. In the literature, these compounds are often called organic disulfides; however, the IUPAC recommended nomenclature is disulfanes.¹ The name disulfide should only be applied to ionic compounds, such as sodium disulfide (Na₂S₂). Moreover, the term disulfane is more widely applicable than disulfide because it facilitates naming even when the R groups are acyl and/or phosphoryl groups.

The formation of unsymmetrical disulfanes is an important transformation in organic synthesis and medicinal chemistry.² Recent developments in disulfide bond formation reactions have been reviewed.3 Although many different methods exist for the preparation of unsymmetrical disulfanes, the most common approach involves substitution of a sulfenyl derivative with a thiol or thiol derivative. To date, the most commonly utilized sulfenyl derivatives are sulfenyl chlorides,4 S-alkyl thiosulfates and S-aryl thiosulfates (Bunte salts),⁵ S-alkylsulfanylisothioureas,6 benzothiazol-2-yl disulfanes,7 benzotriazolyl sulfanes,8 dithioperoxyesters,9 (alkylsulfanyl)dialkylsulfonium salts,10 2-pyridyl disulfanes and derivatives,11 N-alkyltetrazolyl disulfanes,12 sulfenamides,13 sulfenyldimesylamines,14 sulfenyl thiocyanates,15 4-nitroarenesulfenanilides,16 thiolsulfinates and thiosulfonates,17 sulfanylsulfinamidines,18 thionitrites,19 sulfenyl thiocarbonates,20 thioimides,21 and thiophosphonium salts.²² Other practical procedures involve the reaction of a thiol with a sulfinylbenzimidazole,23 a rhodium-catalyzed disulfide exchange,24 an electrochemical method,25 the ring opening of

Convenient and efficient synthesis of functionalized unsymmetrical *Z*-alkenyl disulfanes[†]

M. Musiejuk, J. Doroszuk and D. Witt[®]*

We developed a simple and efficient method for the synthesis of functionalized unsymmetrical Z-alkenyl disulfanes under mild conditions in moderate to good yields. The designed method is based on the reaction of Z-alkenyl thiotosylates with thiols in the presence of base. The developed method allows the preparation of unsymmetrical Z-alkenyl disulfanes bearing additional hydroxy, carboxy, or amino functionalities.

an aziridine using tetrathiomolybdate in the presence of a symmetrical disulfane,²⁶ or the use of diethyl azodicarboxylate (DEAD)²⁷ or a solid support²⁸ in a sequential coupling of two different thiol groups. Recently, the oxidation of a mixture of two different thiols by 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ) to produce an unsymmetrical disulfane has also been reported.²⁹

Earlier studies demonstrated the preparation of functionalized unsymmetrical molecules, such as dialkyl disulfanes,³⁰ alkyl-aryl disulfanes,³¹ 'bioresistant' disulfanes,³² the unsymmetrical disulfanes of L-cysteine and L-cystine,³³ and diaryl disulfanes³⁴ based on the readily available 5,5-dimethyl-2thioxo-1,3,2-dioxaphosphorinane-2-disulfanyl derivatives. These disulfanyl derivatives of phosphorodithioic acid were convenient for the preparation of α -sulfenylated carbonyl compounds,³⁵ functionalized phosphorothioates,³⁶ and unsymmetrical alkynyl sulfides³⁷ as well as symmetrical³⁸ and unsymmetrical³⁹ trisulfanes.

Ajoene was first isolated by Block⁴⁰ in 1984 as an E/Z-mixture of a rearrangement product of allicin produced from freshly crushed garlic. It was established to be an allyl sulfoxide containing an unusual vinyl disulfane functionality, which is rarely seen in the structures of natural products. *Z*-Ajoene is more active than its *E*-isomer as an anti-thrombotic agent,⁴¹ and some studies on anticancer treatments have focused primarily on the *Z*-isomer.⁴²

Although many different synthetic methods exist for the preparation of unsymmetrical disulfanes, the preparation of unsymmetrical alkenyl disulfanes can be achieved by only two methods. The first method is based on the reaction of sulfenyl bromide with trityl-alkenyl sulfide.⁴³ The second method involves the low temperature hydroxide-promoted cleavage of an alkenyl thioester followed by sulfenylation with an appropriate *S*-alkylated *p*-toluenethiosulfonate to afford vinyl disulfide in high yield after chromatography.⁴⁴ Unfortunately, the methods provide exclusively *E* or a mixture of *Z*/*E* alkenyl disulfanes, respectively. In this context, we set out to investigate

Department of Organic Chemistry, Faculty of Chemistry, Gdansk University of Technology, Narutowicza 11/12, 80-233 Gdansk, Poland. E-mail: chemwitt@pg.gda. pl; Fax: +48 58 3472694

[†] Electronic supplementary information (ESI) available. See DOI: 10.1039/c8ra00659h

the feasibility of a more convenient and experimentally practical diastereoselective method to exclusively access *Z*-alkenyl disulfanes.

Results and discussion

Paper

Our synthetic strategy included the preparation of *E*-alkenylboronic acid⁴⁵ **2** from terminal alkyne **1** followed by its conversion to appropriate *E*-alkenyliodonium salt⁴⁶ **3** by known methods. Further reaction with sodium *p*-toluenethiosulfonate provided *Z*-1-octenyl *p*-toluenethiosulfonate **4** with inversion of configuration (Scheme 1).

The reaction of *Z*-1-octenyl *p*-toluenethiosulfonate **4** with a variety of thiols in the presence of NEt₃ provided *Z*-alkenyl disulfanes **6** in good or very good yield (Table 1). All compounds have been fully characterized by ¹H and ¹³C NMR spectroscopy (see the ESI†). *E/Z*-Stereochemistry was assigned based on the vinyl coupling constants in the ¹H NMR spectra; 15 Hz was indicative of the *E*-isomer and 10 Hz for the *Z*-isomer.

The reaction proceeded via the nucleophilic substitution of the thiolate anion (generated from 5) at the sulfur atom of thiotosylate 4, and the *p*-toluenesulfinate anion served as the leaving group, which is why the Z geometry of the alkene remained unchanged. The thiolate anion can also be generated in situ from the corresponding thioacetate and sodium methoxide in methanol (entries 11-12). Such an approach is very convenient when a high-purity or stable thiol is not readily available. The developed method seems to be very versatile. The presence of additional functional groups including carboncarbon multiple bonds (entries 10-12) and hydroxy (entry 2), ester (entry 3), azide (entry 4), amino (entry 5) aryl or heteroaryl (entries 6-9 and 15) moieties did not interfere with the formation of Z-alkenyl disulfanes 6. Arylthiol 5 - disulfane Z-6 exchange reaction was responsible for the formation of corresponding diaryl disulfane and moderate yield of 6f and 6m (entries 6 and 15). The exchange reaction can be limited by the excess of Z-4, what resulted in higher yield of 6f and 6m respectively (entries 7 and 16). L-Cysteine derivatives were also converted to the corresponding Z-alkenyl disulfanes 6k and 6l (entries 13-14). The biological activities of these compounds are expected to be higher than their E-isomer analogs.43,44



Scheme 1 Preparation of Z-1-octenyl p-toluenethiosulfonate 4.



Entry		R	х	Solvent	$\operatorname{Yield}^{b}(\%)$
1	5a	$-C_{12}H_{25}$	Н	CH_2Cl_2	6a (90)
2	5b	-(CH ₂) ₁₁ OH	н	CH_2Cl_2	6b (82)
3	5c	$(CH_2)_{10}CO_2Me$	н	CH_2Cl_2	6c (88)
4	5d	$-(CH_2)_{11}N_3$	Н	CH_2Cl_2	6d (70)
5	5e	$-(CH_2)_{11}NH_2$	н	CH_2Cl_2	6e (77)
6	5f	4-MeC ₆ H ₄ -	н	CH_2Cl_2	6f (40)
7	5f	4-MeC ₆ H ₄ -	н	CH_2Cl_2	6f $(62)^c$
8	5g	2-Furyl-CH ₂ -	н	CH_2Cl_2	6g (71)
9	5h	4-Py-	н	CH_2Cl_2	6h (52)
10	5i	CH ₂ =CHCH ₂ -	н	CH_2Cl_2	6i (51)
11	5i	CH2=CHCH2-	Ac	MeOH	6i (80)
12	5j	$HC \equiv CCH_2$ -	Ac	MeOH	6j (78)
13	5k	H-CysOEt	н	CH_2Cl_2	6k (60)
14	51	BocCysOEt	н	CH_2Cl_2	6l (59)
15	5m	4-MeOC ₆ H ₄ -	Н	CH_2Cl_2	6m (43)
16	5m	4-MeOC ₆ H ₄ -	Н	CH_2Cl_2	6m $(62)^c$

^{*a*} Performed with 4 (0.67 mmol), 5 (0.61 mmol), NEt₃ (0.61 mmol) in solvent (5 mL), 15 min. ^{*b*} isolated yield. ^{*c*} Performed with 4 (1.22 mmol), 5 (0.61 mmol), NEt₃ (0.61 mmol) in solvent (5 mL), 15 min.

Conclusions

We have developed the first simple and efficient diastereoselective method for the synthesis of functionalized unsymmetrical *Z*-alkenyl disulfanes under mild conditions in moderate to good yields. The developed method allows the preparation of unsymmetrical *Z*-alkenyl disulfanes bearing additional hydroxy, carboxy, or amino functionalities. Anti-fungal and anti-cancer activity studies of the *Z*-alkenyl L-cysteine disulfane derivatives are in progress.

Experimental

A typical procedure for the preparation of *Z*-alkenyl disulfanes 6 and representative analytical data

A compound Z-4 (0.67 mmol, 200 mg) was dissolved in dry DCM (3 mL) in the round bottom flask. Then a solution of thiol 5 (0.61 mmol) and NEt₃ (0.61 mmol) in dry DCM (2 mL) was added. Reaction was stirred for 15 min. After this time solvent was evaporate and Et₂O (10 mL) was added. Slurry was washed with water (10 mL) and aqueous phase was extracted 2 times with Et₂O (2 × 10 mL). Organic layers were dried with MgSO₄ and evaporated. The residue was purified by column chromatography (SiO₂).

(Z)-1-(dodec-1-yldisulfanyl)-oct-1-ene Z-6a

Chromatography, PE, $R_f = 0.6$; a colorless oil, yield 188 mg (90%) IR (ATR): 510 (s), 625 (m), 1490 (w), 2875 (m),

2900 (m) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.09 (dt, J = 9.3, 1.4 Hz, 1H), 5.65 (dt, J = 9.3, 7.4 Hz, 1H), 2.72 (t, J = 7.3 Hz, 2H), 2.19 (qd, J = 7.4, 1.3 Hz, 2H), 1.70 (dt, J = 14.9, 7.2 Hz, 2H), 1.47-1.24 (m, 26H), 0.93-0.88 (m, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 132.66, 129.24, 39.07, 31.93, 31.67, 29.66, 29.64, 29.60, 29.51, 29.36, 29.23, 29.08, 28.91, 28.85, 28.80, 28.46, 22.70, 22.62, 14.13, 14.09. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₀H₄₁S₂: 345.2650; found: 345.2655.

(Z)-11-(1-octen-1-yldisulfanyl)-undecan-1-ol Z-6b

Chromatography: PE:DCM 1:1; $R_f = 0.4$; a colorless oil, yield 173 mg (82%) IR (ATR): 755 (w), 1100 (w), 1500 (w), 1650 (w), 2875 (s), 2900 (s), 3300 (br) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.09 (dt, J = 9.3, 1.3 Hz, 1H), 5.65 (dt, J = 9.3, 7.4 Hz, 1H), 3.66 (t, J = 6.6 Hz, 2H), 2.72 (t, J = 7.4 Hz, 2H), 2.19 (qd, J = 7.4, 1.3 Hz, 2H), 1.72–1.60 (m, 2H), 1.60–1.50 (m, 2H), 1.44–1.23 (m, 23H), 0.91 (t, J = 6.8 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 132.68, 129.22, 63.10, 39.06, 32.81, 31.67, 29.58, 29.51, 29.48, 29.42, 29.21, 29.07, 28.90, 28.84, 28.80, 28.44, 25.74, 22.62, 14.09. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₉H₃₉OS₂: 347.2442; found: 347.2438.

(Z)-Methyl 11-(1-octen-1-yldisulfanyl)undecanoate Z-6c

Chromatography: PE:DCM 3:1; $R_f = 0.4$; a colorless oil, yield 201 mg (88%): IR (ATR): 500 (s), 625 (m), 1200 (w), 1490 (w), 1750 (m), 2875 (w), 2990 (m) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.09 (dt, J = 9.3, 1.3 Hz, 1H), 5.65 (dt, J = 9.3, 7.4 Hz, 1H), 3.69 (s, 3H), 2.75–2.70 (m, 2H), 2.32 (t, J = 7.6 Hz, 2H), 2.19 (qd, J = 7.4, 1.3 Hz, 2H), 1.70–1.52 (m, 4H), 1.46–1.24 (m, 20H), 0.91 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 174.32, 132.67, 129.23, 51.45, 39.05, 34.11, 31.66, 29.43, 29.37, 29.23, 29.18, 29.14, 29.07, 28.89, 28.84, 28.79, 28.43, 24.95, 22.61, 14.09. HRMS (ESI): $m/z [M + H]^+$ calcd for C₂₀H₃₉O₂S₂: 375.2391; found: 375.2396.

(Z)-1-(11-azidoundec-1-yldisulfanyl)-oct-1-ene Z-6d

Chromatography: PE; $R_{\rm f} = 0.45$; a colorless oil, yield 159 mg (70%) IR (ATR): 510 (s), 625 (m), 1260 (w), 1500 (w), 2110 (m), 2875 (m), 2900 (m)) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.09 (dt, J = 9.3, 1.3 Hz, 1H), 5.65 (dt, J = 9.3, 7.4 Hz, 1H), 3.28 (t, J = 7.0 Hz, 2H), 2.73 (t, J = 7.4 Hz, 2H), 2.19 (qd, J = 7.4, 1.3 Hz, 2H), 1.75–1.55 (m, 4H), 1.47–1.24 (m, 22H), 0.91 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 132.68, 129.22, 51.50, 39.05, 31.67, 29.46, 29.20, 29.15, 29.07, 28.89, 28.85, 28.80, 28.43, 26.72, 22.62, 14.09. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₉H₃₈N₃S₂: 372.2507; found: 372.2511.

(Z)-1-(11-aminoundec-1-yldisulfanyl)-oct-1-ene Z-6e

Chromatography: DCM: MeOH 14:1; $R_f = 0.3$; a colorless oil, yield 162 mg (77%) IR (ATR): 510 (s), 625 (s), 800 (w), 1100 (w), 1490 (w), 1510 (w), 2875 (m), 2900 (s), 3500 (br) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.09 (dt, J = 9.3, 1.3 Hz, 1H), 5.65 (dt, J = 9.3, 7.4 Hz, 1H), 2.87–2.81 (m, 2H), 2.75–2.70 (m, 2H), 2.18 (dt, J = 7.4, 4.2 Hz, 2H), 1.75–1.55 (m, 4H), 1.47–1.24 (m, 24H), 0.91 (t, J = 6.8 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 132.66, 129.25,

39.75, 39.07, 31.67, 29.50, 29.41, 29.25, 29.08, 28.92, 28.84, 28.48, 27.91, 26.53, 22.62, 14.11. HRMS (ESI): $m/z [M + H]^+$ calcd for C₁₉H₄₀NS₂: 346.2602; found: 346.2604.

(Z)-1-(p-tolyldisulfanyl)-oct-1-ene Z-6f

Chromatography: PE; $R_{\rm f} = 0.8$; a colorless oil, yield 65 mg (40%) IR (ATR): 500 (s), 625 (m), 780 (w), 1500 (w), 2875 (w), 2990 (m) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 7.43 (d, J = 8.1 Hz, 2H), 7.15 (d, J = 8.1 Hz, 2H), 6.16 (dt, J = 9.3, 1.3 Hz, 1H), 5.75–5.65 (m, 1H), 2.36 (s, 3H), 2.19 (m, 2H), 1.46–1.25 (m, 8H), 0.91 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 137.52, 134.02, 129.74, 129.12, 127.92, 31.66, 29.08, 29.02, 28.89, 28.82, 22.61, 21.08, 14.09. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₅H₂₃S₂: 267.1241; found: 267.1243.

(Z)-1-(furan-2-ylmethyldisulfanyl)-oct-1-ene Z-6g

Chromatography: PE; $R_f = 0.7$; a colorless oil, yield 111 mg (71%) IR (ATR): 600 (m), 740 (m), 770 (w), 900 (m), 1010 (m), 1125 (m), 1500 (w), 2875 (w), 2950 (m) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 7.40 (dd, J = 1.8, 0.8 Hz, 1H), 6.34 (dd, J = 3.2, 1.8 Hz, 1H), 6.28 (d, J = 0.6 Hz, 1H), 5.84 (dt, J = 9.3, 1.3 Hz, 1H), 5.61 (dt, J = 9.3, 7.4 Hz, 1H), 3.94 (s, 2H), 2.20–2.12 (m, 2H), 1.42–1.15 (m, 8H), 0.91 (t, J = 6.7 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 150.16, 142.46, 133.11, 128.24, 110.64, 108.97, 35.72, 31.66, 29.05, 28.84, 28.76, 22.61, 14.08. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₃H₂₁OS₂: 257.1034; found: 257.1031.

(Z)-1-(pyridin-4-yldisulfanyl)-oct-1-ene Z-6h

Chromatography: DCM; $R_{\rm f} = 0.4$; a colorless oil, yield 80 mg (52%) IR (ATR): 510 (s), 625 (m), 740 (m), 770 (m), 1450 (w), 1490 (w), 1600 (s), 2875 (w), 2990 (w) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 8.50 (d, J = 6.1 Hz, 2H), 7.41 (dd, J = 4.6, 1.6 Hz, 2H), 6.00 (dt, J = 9.2, 1.3 Hz, 1H), 5.81 (dt, J = 9.2, 7.5 Hz, 1H), 2.32 (qd, J = 7.4, 1.3 Hz, 2H), 1.52–1.23 (m, 8H), 0.93 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 149.50, 148.78, 135.62, 125.93, 120.14, 31.65, 29.09, 29.00, 28.89, 22.63, 14.09. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₃H₂₀NS₂: 254.1037; found: 254.1033.

(Z)-1-(allyldisulfanyl)-oct-1-ene Z-6i

Chromatography: PE; $R_{\rm f} = 0.6$; a colorless oil, yield 106 mg (80%) IR (ATR): 510 (s), 650 (s), 990 (w), 1510 (w), 2875 (w), 2990 (m) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.08 (dt, J = 9.3, 1.3 Hz, 1H), 5.87 (ddt, J = 17.1, 10.0, 7.3 Hz, 1H), 5.66 (dt, J = 9.3, 7.4 Hz, 1H), 5.25–5.14 (m, 2H), 3.37 (dd, J = 7.3, 0.9 Hz, 2H), 2.19 (qd, J = 7.4, 1.3 Hz, 2H), 1.52–1.15 (m, 8H), 0.91 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 133.07, 132.99, 128.74, 118.66, 41.95, 31.66, 29.05, 28.85, 28.83, 22.61, 14.08. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₁H₂₁S₂: 217.1085; found: 217.1088.

(Z)-1-(propargyldisulfanyl)-oct-1-ene Z-6j

Chromatography: PE; $R_{\rm f} = 0.55$; a colorless oil, yield 102 mg (78%) IR (ATR): 625 (s), 1250 (w), 1500 (w), 2875 (w), 2990 (m), 3240 (w) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.21 (dt, J = 9.3, 1.4 Hz, 1H), 5.73 (dt, J = 9.3, 7.4 Hz, 1H), 3.49 (d, J = 2.6 Hz, 2H), 2.32 (t, J = 2.6 Hz, 1H), 2.21 (qd, J = 7.4, 1.3 Hz, 2H), 1.46–1.25

View Article Online RSC Advances

(m, 8H), 0.91 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 134.23, 127.62, 79.33, 72.41, 31.64, 29.04, 28.82, 28.76, 27.14, 22.61, 14.09. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₁H₁₉S₂: 215.0928; found: 215.0932.

(Z)-Ethyl (R)-2-amino-3-(oct-1-ene-1-yldisulfanyl)propanoate Z-6k

Chromatography: DCM: MeOH 14:1; $R_f = 0.35$; a colorless oil, yield 107 mg (60%) IR (ATR): 510 (s), 625 (m), 1010 (w), 1200 (w), 1500 (w), 1750 (m), 2875 (w), 2990 (w), 3500 (br) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.11 (dt, J = 9.3, 1.3 Hz, 1H), 5.70 (dq, J = 8.9, 7.5 Hz, 1H), 4.22 (q, J = 7.1 Hz, 2H), 3.81 (dd, J = 8.0, 4.5 Hz, 1H), 3.15 (dd, J = 13.7, 4.5 Hz, 1H), 2.89 (dd, J = 13.7, 8.0 Hz, 1H), 2.18 (qd, J = 7.4, 1.3 Hz, 2H), 1.57–1.07 (m, 8H), 0.89 (t, J = 6.8 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 173.66, 134.02, 127.86, 61.35, 53.43, 43.90, 31.63, 29.01, 28.85, 28.80, 22.58, 14.18, 14.07. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₃H₂₆NO₂S₂: 292.1399; found: 292.1403.

(*Z*)-Ethyl (*R*)-2-((*tert*-butoxycarbonyl)amino)-3-(oct-1-ene-1-yldisulfanyl)propanoate *Z*-6l

Chromatography: PE: DCM 1:2; $R_f = 0.5$; a colorless oil, yield 141 mg (59%) IR (ATR): 500 (s), 625 (m), 1010 (m), 1200 (s), 1375 (m), 1625 (m), 1750 (m), 2875 (w), 2990 (w), 3000 (w) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 6.11 (dt, J = 9.3, 1.3 Hz, 1H), 5.72 (dt, J = 9.3, 7.4 Hz, 1H), 5.35 (d, J = 7.3 Hz, 1H), 4.61 (d, J = 6.1 Hz, 1H), 4.28–4.21 (m, 2H), 3.19 (ddd, J = 19.8, 14.0, 5.2 Hz, 2H), 2.17 (qd, J = 7.4, 1.2 Hz, 2H), 1.48 (s, 9H), 1.44–1.25 (m, 11H), 0.90 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 170.70, 155.06, 133.93, 127.97, 80.14, 61.79, 53.43, 53.07, 41.53, 31.64, 29.04, 28.86, 28.85, 28.31, 22.59, 14.13, 14.08. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₈H₃₄NO₄S₂: 392.1929; found: 392.1934.

(Z)-1-(4-methoxylphenyldisulfanyl)-oct-1-ene Z-6m

Chromatography: PE; $R_{\rm f} = 0.3$; a colorless oil, yield 74 mg (43%) IR (ATR): 523 (w), 823 (m), 1031 (m), 1244 (s), 1490 (s), 1589 (m), 2853 (s), 2923 (s), 3333 (br) cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 7.52–7.47 (m, 2H), 6.91–6.83 (m, 2H), 6.22 (dt, J = 9.3, 1.3 Hz, 1H), 5.71 (dt, J = 9.2, 7.4 Hz, 1H), 3.83 (s, 3H), 2.19–2.07 (m, 2H), 1.44–1.17 (m, 8H), 0.90 (t, J = 6.9 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 159.86, 134.14, 132.49, 128.34, 127.90, 114.61, 55.38, 31.64, 29.00, 28.83, 28.80, 22.59, 14.09. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₅H₂₃OS₂: 283.4725; found: 283.4729.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We gratefully acknowledge the National Science Centre (NCN) for financial support (grant no. 2015/19/B/ST5/03359).

Notes and references

1 R. Steudel, Chem. Rev., 2002, 102, 3905.

- 2 (a) R. Cremlyn and J. An, Introduction to Organosulfur Chemistry, Wiley, New York, 1996; (b) S. Oae, Organic Sulfur Chemistry: Structure and Mechanism, CRC Press, Boca Raton FL, 1991; (c) V. M. Vrudhula, J. F. MacMaster, L. Zhengong, D. E. Kerr and P. D. Senter, Bioorg. Med. Chem. Lett., 2002, 12, 3591; (d) Y. Mu, M. Nodwell, J. L. Pace, J. P. Shaw and J. K. Judice, Bioorg. Med. Chem. Lett., 2004, 14, 735.
- 3 (a) I. Shcherbakova and A. F. Pozharskii, in Comprehensive Organic Functional Group Transformations II, ed. A. R. Katritzky, R. Taylor and Ch. Ramsden, Pergamon, Oxford, 2004, vol. 2, pp. 177–187; (b) R. Sato and T. Kimura, in Science of Synthesis, ed. N. Kambe, J. Drabowicz and G. A. Molander, Thieme, Stuttgart-New York, 2007, vol. 39, pp. 573–588; (c) D. Witt, Synthesis, 2008, 2491; (d) B. Mandal and B. Basu, RSC Adv., 2014, 4, 13854; (e) M. Musiejuk and D. Witt, Org. Prep. Proced. Int., 2015, 47, 95.
- 4 (*a*) D. N. Harpp, B. T. Friedlander, C. Larsen, K. Steliou and A. Stockton, *J. Org. Chem.*, 1978, **43**, 3481; (*b*) C. Brown and G. R. Evans, *Tetrahedron Lett.*, 1996, **37**, 9101.
- 5 (*a*) J. M. Swan, *Nature*, 1957, **180**, 643; (*b*) P. Hiver, A. Dicko and D. Paquer, *Tetrahedron Lett.*, 1994, **35**, 9569.
- 6 K. Sirakawa, O. Aki, T. Tsujikawa and T. Tsuda, *Chem. Pharm. Bull.*, 1970, **18**, 235.
- 7 (a) A. L. Ternay, C. Cook and E. Brzezinska, *Phosphorus, Sulfur Silicon Relat. Elem.*, 1994, **95**, 351; (b) A. L. Ternay and E. Brzezinska, *J. Org. Chem.*, 1994, **59**, 8239.
- 8 R. Hunter, M. Caira and N. Stellenboom, *J. Org. Chem.*, 2006, 71, 8268.
- 9 C. Leriverend and P. Metzner, Synthesis, 1994, 761.
- 10 P. Dubs and R. Stuessi, Helv. Chim. Acta, 1976, 59, 1307.
- 11 (a) D. H. R. Barton, C. Chen and M. G. Wall, *Tetrahedron*, 1991, 47, 6127; (b) D. H. R. Barton, A. C. O'Sullivan and M. M. Pechet, *J. Org. Chem.*, 1991, 56, 6697.
- 12 M. Ohtani and N. Narisada, J. Org. Chem., 1991, 56, 5475.
- 13 M. Bao and M. Shimizu, Tetrahedron, 2003, 59, 9655.
- 14 A. Blaschette and M. Naveke, Chem.-Ztg., 1991, 115, 61.
- 15 R. G. Hiskey and B. F. Ward Jr., J. Org. Chem., 1970, 35, 1118.
- 16 L. Benati, P. C. Montevecchi and P. Spagnolo, *Tetrahedron Lett.*, 1986, 27, 1739.
- 17 (a) R. Cragg, J. P. N. Husband and A. F. Weston, J. Chem. Soc., Chem. Commun., 1970, 1701; (b) D. A. Armitage, M. J. Clark and C. C. Tsao, J. Chem. Soc., Perkin Trans. 1, 1972, 680; (c) G. Capozzi, A. Capperucci, A. Degl'Innocenti, R. DelDuce and S. Menichetti, Tetrahedron Lett., 1989, 30, 2995; (d) A. Rajca and M. Wiessler, Tetrahedron Lett., 1990, 31, 6075.
- 18 I. V. Koval, Russ. J. Org. Chem., 2002, 38, 232.
- 19 S. Oae, Y. H. Kim, D. Fukushima and K. Shinhama, J. Chem. Soc., Perkin Trans. 1, 1978, 913.
- 20 S. J. Brois, J. F. Pilot and H. W. Barnum, *J. Am. Chem. Soc.*, 1970, **92**, 7629.
- 21 (a) K. S. Boustang and A. B. Sullivan, *Tetrahedron Lett.*, 1970, 11, 3547; (b) D. H. Harpp, D. K. Ash, T. G. Beck, J. G. Gleason, B. A. Orwig, W. F. VanHorn and J. P. Snyder, *Tetrahedron Lett.*, 1970, 11, 3551; (c) J. Klose, C. B. Reese and Q. Song, *Tetrahedron*, 1997, 53, 14411.
- 22 M. Masui, Y. Mizuki, K. Sakai, C. Ueda and H. Ohmori, J. Chem. Soc., Chem. Commun., 1984, 843.

- 23 D. R. Graber, R. A. Morge and J. C. Sih, *J. Org. Chem.*, 1987, 52, 4620.
- 24 (a) M. Arisawa and M. Yamaguchi, J. Am. Chem. Soc., 2003, 125, 6624; (b) K. Tanaka and K. Ajiki, Tetrahedron Lett., 2004, 45, 5677.
- 25 Q. T. Do, D. Elothmani, G. Le Guillanton and J. Simonet, *Tetrahedron Lett.*, 1997, **38**, 3383.
- 26 (a) D. Sureshkumar, V. Ganesh, R. S. Vidyarini and S. Chandrasekaran, J. Org. Chem., 2009, 74, 7958; (b) D. Sureshkumar, S. M. Koutha and S. Chandrasekaran, J. Am. Chem. Soc., 2005, 127, 12760.
- 27 T. Mukaiyama and K. Takahashi, *Tetrahedron Lett.*, 1968, 9, 5907.
- 28 A. K. Galawde and A. F. Spatola, Org. Lett., 2003, 5, 3431.
- 29 (a) J. K. Vandavasi, W.-P. Hu, Ch.-Y. Chen and J.-J. Wang, *Tetrahedron*, 2011, 67, 8895; (b) R. Smith, X. Zeng,
 H. Müller-Bunz and X. Zhu, *Tetrahedron Lett.*, 2013, 54, 5348; (c) M. Musiejuk, T. Klucznik, J. Rachon and D. Witt, *RSC Adv.*, 2015, 5, 31347.
- 30 S. Lach, S. Demkowicz and D. Witt, *Tetrahedron Lett.*, 2013, 54, 7021.
- 31 S. Antoniow and D. Witt, Synthesis, 2007, 363.
- 32 J. Kowalczyk, P. Barski, D. Witt and B. A. Grzybowski, Langmuir, 2007, 23, 2318.
- 33 M. Szymelfejnik, S. Demkowicz, J. Rachon and D. Witt, *Synthesis*, 2007, 3528.
- 34 S. Demkowicz, J. Rachon and D. Witt, Synthesis, 2008, 2033.
- 35 E. Okragla, S. Demkowicz, J. Rachon and D. Witt, *Synthesis*, 2009, 1720.

- 36 S. Lach and D. Witt, Synthesis, 2011, 3975.
- 37 J. Doroszuk, M. Musiejuk, S. Demkowicz, J. Rachon and D. Witt, *RSC Adv.*, 2016, 6, 105449.
- 38 (a) A. Kertmen, S. Lach, J. Rachon and D. Witt, Synthesis, 2009, 1459; (b) S. Lach and D. Witt, Heteroat. Chem., 2014, 25, 10.
- 39 (a) S. Lach, M. Sliwka-Kaszynska and D. Witt, Synlett, 2010, 2857; (b) S. Lach and D. Witt, Synlett, 2013, 24, 1927.
- 40 E. Block and S. Ahmad, J. Am. Chem. Soc., 1984, 106, 8295.
- 41 E. Block, S. Ahmad, J. L. Catalfamo, M. K. Jain and R. Apitz-Castro, *J. Am. Chem. Soc.*, 1986, **108**, 7045.
- 42 (a) M. Li, J.-R. Ciu, Y. Ye, J.-M. Min, L.-H. Zhang, K. Wang, M. Gares, J. Cros, M. Wright and J. Leung-Track, *Carcinogenesis*, 2002, 23, 573; (b) M. Li, J. M. Min, J. R. Cui, L.-H. Zhang, K. Wang, A. Valette, C. Davrinche, M. Wight and J. Leung-Track, *Nutr. Cancer*, 2002, 42, 241.
- 43 G. Zhang and K. L. Parkin, J. Agric. Food Chem., 2013, 61, 1896.
- 44 (a) R. Hunter, C. H. Kaschula, I. M. Parker, M. R. Caira,
 P. Richards, S. Travis, F. Taute and T. Qwebani, *Bioorg. Med. Chem. Lett.*, 2008, 18, 5277; (b) C. H. Kaschula,
 R. Hunter, N. Stellenboom, M. R. Caira, S. Winks,
 T. Ogunleye, P. Richards, J. Cotton, K. Zilbeyaz, Y. Wang,
 V. Siyo, E. Ngarande and M. I. Parker, *Eur. J. Med. Chem.*, 2012, 50, 236.
- 45 H. C. Brown and J. B. Campbell Jr, *J. Org. Chem.*, 1980, 45, 389.
- 46 M. Ochiai, M. Toyonari, T. Nagaoka, D.-W. Chen and M. Kida, *Tetrahedron Lett.*, 1997, **38**, 6709.