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Isocyanate-based multicomponent reactions

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Since their discovery, multicomponent reactions have attracted significant attention due to their versatility and efficiency. This review aims to explore the latest advancements in isocyanate-based multicomponent reactions and the sophisticated chemical opportunities they present for generating molecules of interest. The added value of the methodologies described, supported by mechanism schemes, as well as scopes of application, will be discussed. These developments will be organised as the main accessible chemical functions and sorted according to their type of MCR (3, 4 or 5-MCR).

1. Introduction

Since the pioneer work of Strecker^{1,2} in 1850, multicomponent reactions (MCRs) have garnered significant interest in synthetic organic chemistry due to their widespread applications in various fields such as drug discovery, medicinal chemistry, total synthesis, pharmaceuticals and agrochemical industry. These one-pot reactions incorporate more than two reactants into a single final product, providing access to complex molecular architectures, with high molecular diversity.³ Regarded as a convenient and effective approach, MCRs present several advantages over traditional one or two-component reactions or in one-pot stepwise synthesis. In one-pot stepwise synthesis, multiple reactions occur sequentially in a single vessel. In contrast, a multicomponent reaction involves the simultaneous combination of three or more reactants to form a product in a single reaction step. MCRs simplify synthetic routes by reducing the requirement for numerous sequential steps and the number of purification operations, leading frequently to better yields. Additionally to expand the chemical space, the atom-economical nature of MCRs and their alignment with green chemistry principles⁴ have highlighted their environment-friendly aspects.

Hantzsch,⁵ Biginelli,⁶ Passerini⁷ and Ugi⁸ among notable contributors have paved the way for the development of MCRs. For decades, these reactions remained largely underexplored before being used in the synthesis of complex and biologically active compounds. Over time, the scope of these reactions expanded using a combination of serendipity and rational design. Several strategies have been developed for the rational design of new MCRs:⁹ (i) single reactant replacement (SRR), (ii) modular reaction sequences (MRS), (iii) adjusting conditions towards divergent MCRs. Some of these strategies have been used in tandem with innovative technologies such as flow

chemistry,¹⁰ microwave chemistry,¹¹ mechanochemistry¹² and photochemistry¹³ to further enhance the applications of MCRs. A final strategy implemented by different research teams consists in combining multiple known MCRs to achieve higher-order MCRs that can involve six or more reactants.¹⁴ An example of the diversification of known-MCRs is the widely applied Ugi-deprotection–cyclization synthetic strategy^{15,16} or post-Ugi cyclization,¹⁷ which both largely contributed to the construction of diverse heterocyclic and drug-like compounds.

These developments have significantly impacted many domains and applications, such as material science for the preparation of covalent organic frameworks¹⁸ (COFs), combinatorial chemistry providing an efficient chemistry tool for the preparation of asymmetrical product libraries,¹⁹ or for the development of biocompatible reactions^{20,21} through tailored MCR methodologies. Polymer science has also benefited from MCRs, enabling efficient and cost-effective polymer construction and post-polymerization modifications.²² Due to their ability to serve as a starting point for diversity orientating synthesis and provide access to a myriad of diverse chemical scaffolds, MCRs have also gained considerable significance in the pharmaceutical industry, and more particularly in drug discovery.³ MCR-derived products exhibit high atom density, enhancing their interaction potential with biological targets. This often leads to a superior hit ratio compared to non-MCR compounds, particularly in the context of specific targets like protein–protein interactions.²⁰ Moreover, they can generate libraries of complex small molecules that are easily screened in direct-to-biology approaches.²³ Some notable applications in the pharmaceutical industry include their utilization for synthetic routes optimization, facilitating the transition to industrial scale production, and ultimately resulting in faster and more cost-effective synthesis processes. For instance, penicillin, telaprevir, nifedipine and praziquantel syntheses involve MCR steps within their industrial manufacturing processes.³

As one of the most widely recognized reagent in multicomponent reactions, isocyanides have been playing a crucial role in many fields, notably due to the high structural diversity it

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reaches. Carbonyl compounds and amines are fundamental components in Hantzsch and Biginelli reactions, among others MCRs. Other reagents have emerged as very useful building blocks in MCRs, such as sodium azide and isocyanates. However, even if recent reviews reported the use of different chemical class such as isocyanide^{24–26} or sodium azide²⁷ in MCRs, none of them focused on the use of isocyanates. Reviews on isocyanate chemistry^{28,29} or their use in specific transformations of interest as amide bond formation³⁰ or urethane/urea bonds formation³¹ have been reported. This review will discuss the interest of isocyanates in another prism: it will spotlight the potential of isocyanates as versatile building blocks in 3, 4, or 5-component reactions (MCRs) for the efficient synthesis of structurally diverse and complex molecules (Fig. 1). We delve into the various strategies employed by researchers, exploring the use of isocyanates as both pre-formed building blocks and those generated *in situ* within diverse synthetic transformations.

2. Synthesis and reactivity of isocyanates

2.1. Discovery and synthesis of isocyanates

Isocyanates are very commonly used building blocks in many chemical applications since their discovery by Wurtz in 1848.³² They have emerged as remarkably useful for the synthesis of essential compounds across diverse fields, including pharmaceuticals, agrochemicals, materials science, and fine chemicals. The global market of isocyanates is increasing by 5% every year, mostly due to their valuable industrial use³¹ and to their highly efficient and cost-effective large scale production by phosgenation of amines,³³ or by cleavage of urethanes. Besides these techniques, many lab-scale reactions have been developed including the Curtius rearrangement, but also Hofmann, Lossen and Schmidt rearrangements involving an isocyanate intermediate. These methods provide access to a diverse range of isocyanates, more or less sensitive to hydrolysis and prone to dimerization or trimerization. Other preparation methods have been described, such as the reaction of amides and amines with oxalyl chloride or the use of isocyanides with carboxylic derivatives. Recent advancements have introduced milder methods like the Staudinger-Aza Wittig (SAW) sequence. This approach utilizes an azide and CO₂, resulting in a less toxic, more



Fig. 2 Several examples of reactions producing isocyanates.

environmentally friendly reaction that exhibits high tolerance towards functional groups sensitive to phosgene (Fig. 2).³⁴ The abundance of efficient synthetic methods, coupled with the development of *in situ* isocyanate generation, facilitates the utilization of a vast library of alkyl and aryl isocyanates in the construction of diverse scaffolds.

2.2. Fundamental mechanisms of reactions involving isocyanate

Isocyanates can be described as strained linear structures characterized by an electrophilic carbon surrounded by a nitrogen atom and an oxygen atom. The resonance structure of the isocyanate reinforces the electrophilicity of the carbon (Fig. 3).

As with many functional groups, the reactivity of isocyanates depends on their substituents. An electron-withdrawing group increases carbon electrophilicity and enhances reactivity. Conversely, any electron-donating group lowers the reactivity of the isocyanate (Fig. 3). Due to the electrophilic carbon, isocyanates can undergo nucleophilic attacks from various nucleophiles such as amines, alcohols or thiols, where the commonly accepted mechanism involves a first step of nucleophilic attack followed by a proton transfer (Fig. 4).



Fig. 3 Reactivity of isocyanates.

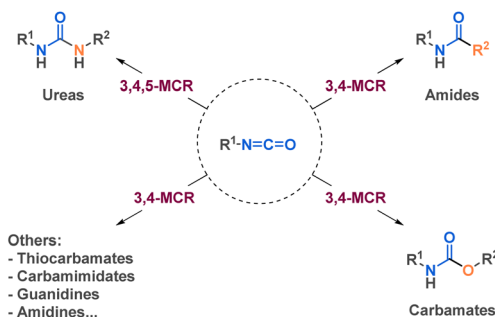


Fig. 1 General representation of possible MCRs with isocyanates.

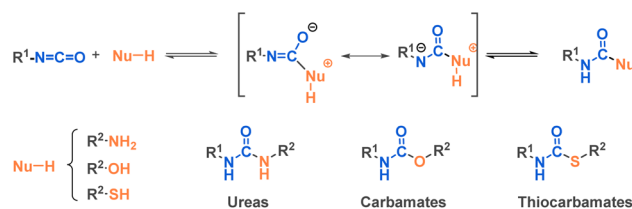


Fig. 4 Main classes of products resulting from nucleophilic attack on isocyanates.



2.3. Reactivity and stability of isocyanates

Isocyanates are known for their high reactivity but also for their inherent toxicity particularly since the 1984 Bhopal tragedy that led to the release of 40 tons of methyl isocyanates in the atmosphere killing thousands of citizens. Isocyanates toxicity come from its ability to react with nucleophilic biological molecules such as DNA bases or glutathione, leading to the production of carbamoyl metabolites that could react with other macromolecules in the surrounding tissues.³⁵ In this context, driven by the goal of sustainable development and the reduction of hazardous chemical use, there has been a trend toward substituting certain substances, such as isocyanates. One area that has received significant attention is the development of non-isocyanate polyurethanes (PUs) and polyureas (PUAs), like CO₂-based PUs/PUAs.^{36,37} This is particularly significant considering that PUs are currently ranked as the 6th most widely sold polymer worldwide.

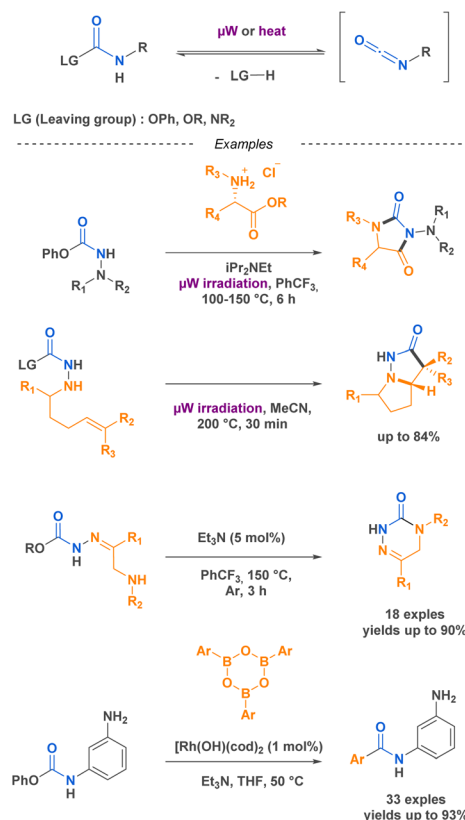
In certain cases, their high reactivity can also cause complications in handling and storage. As an example, isocyanates are known to be prone to polymerization when bearing a nucleophilic functional group. The blocked-isocyanate strategy elegantly bypasses this limitation by providing the *in situ* generation of an isocyanate from a precursor under activation. This controlled release allows for direct reaction with specific substrates. For instance, the research team led by Beauchemin^{38–42} devised several methodologies to generate isocyanates *in situ*. One of these strategies implies the functionalisation of the electrophilic carbon of the isocyanate by a leaving group, which is then cleaved upon microwave irradiation or heating, thus generating the corresponding isocyanate *in situ*. This innovative approach provides on-demand formation of isocyanates with precise control over their reactivity. Beauchemin successfully synthesized *N*-isocyanates, *O*-isocyanates, and isocyanates with nucleophilic functional groups that readily react with desired substrates to yield targeted products (Scheme 1).

Due to their high reactivity and interesting properties, isocyanates have been used in multicomponent reactions as developed in the following sections.

3. Isocyanate-based multicomponent reactions

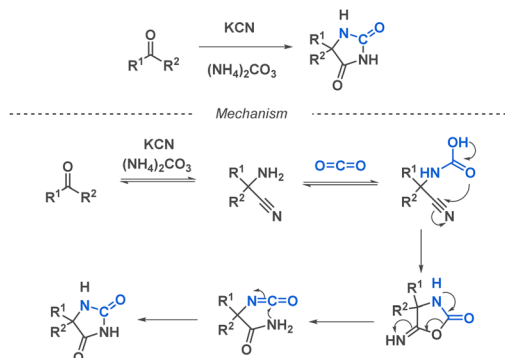
3.1. Urea derivatives

Ureas hold tremendous biological significance, sparking a wealth of research into their synthesis. This structural motif plays a vital role in numerous drugs used to treat a wide range of diseases, such as antitumoral and antiparkinsonian applications.^{43,44} The extensive biological applications of ureas notably stem from their ability to form stable hydrogen bonds with the target molecules, thereby improving ligand–receptor interactions. The substantial interest in ureas has pushed multiple scientific teams to dedicate significant efforts towards developing linear substituted ureas or cyclic ureas using diverse synthetic strategies, including multi-component reaction approaches.



Scheme 1 Examples of *in situ* generation of isocyanates by the team of Beauchemin.

3.1.1. 3-MCR. One of the earliest isocyanate-based MCR is the Bucherer–Bergs reaction which is also one of the most efficient and simple methods to prepare 5-substituted and 5,5-disubstituted hydantoins. This reaction combines readily accessible starting compounds including aldehydes or ketones, potassium cyanide and ammonium carbonate, which facilitates its applicability. As presented in Scheme 2, the proposed mechanism starts with the condensation of potassium cyanide on the carbonyl, followed by the substitution of the alcohol by the ammonia to yield an alpha-aminonitrile intermediate. Then, the corresponding amine attacks a carbon dioxide



Scheme 2 General scheme and a proposed mechanism of the Bucherer–Bergs reaction.





Scheme 3 Synthesis of hydantoin by 3-MCR methodologies.

molecule and forms an intermediate undergoing ring-closing steps. In the final step, a rearrangement occurs leading to a reactive isocyanate intermediate that finally yields to the formation of the hydantoin product.⁴⁵

One of the limitation of this reaction is that there is only one point of diversity based on the starting carbonyl, limiting the broadness of the scope. Another strategy to prepare hydantoin from isocyanates was proposed by Bellucci *et al.* using α -azido ester through a pseudo 3-MCR (Scheme 3a).⁴⁶ Pseudo-MCRs are one-pot processes involving combinations of at least three reactants, in which two of them are identical.⁴⁷ This had the advantage to provide another simple and efficient route to other type of hydantoin increasing also the diversity in their scaffold. Another example was reported by Alizadeh *et al.* where they successfully combined three compounds in the presence of triphenylphosphine: an amine, an arylsulfonyl isocyanate and an alkyl propiolate or a dialkyl acetylenedicarboxylate (Scheme 3b) providing complex hydantoin structures in very good yields but with a limited scope of 5 examples.⁴⁸

Rigby *et al.* reported a dual-copper catalyst system that enables the reaction between aryl isocyanates and dimethoxycarbene for the preparation of hydantoin derivatives.⁴⁹ In contrast to the previous methodologies, this methodology could be regarded as a pseudo-3-MCR given that two components are identical. Some of the isocyanates used for the scope were derived from a Curtius rearrangement (Scheme 4). Importantly, while activated aromatic rings led to excellent yield up to 84%, electron-donating groups on the aromatic ring resulted in moderated yield (range of 50%). Other multicomponent reactions involving more than three reactants have been developed to prepare hydantoin derivatives and will be described in the following sections.

Scheme 4 Synthesis of hydantoin derivatives by Rigby *et al.*

Sharing a close structural relationship with the hydantoin core, uracils are heterocyclic molecules featuring a six-membered ring system, which are prevalent in numerous bioactive molecules, including uracil in RNA, and are found in various compounds exhibiting promising anticancer, antiviral, and antiparasitic properties. These molecules play crucial roles in biological processes and offer potential for therapeutic applications.⁵⁰ Due to the attention paid to uracil in medicinal chemistry, different synthetic routes have been developed to prepare these compounds.

Similarly to the work of Rigby *et al.* on the synthesis of hydantoin derivatives *via* a pseudo-MCR methodology, two teams have independently developed innovative pseudo-MCR methods for the synthesis of uracil derivatives, including dihydrouracils. Fernández de Trocóniz *et al.* reported a pseudo 3-MCR using two isocyanates allowing the preparation of fluorinated triazinane-2,4-diones from two phenyl isocyanate precursors and an unsaturated imine at room temperature (Scheme 5a).⁵¹ At the moment of the report, it was the first synthesis of such fluorinated compounds. Morimoto *et al.* also developed a pseudo 3-MCR but employed a nickel-catalysed strategy to afford the uracil derivatives (Scheme 5b).⁵² Dekamin *et al.* capitalized on the high reactivity and versatility of isocyanates to provide an efficient cyclotrimerisation approach tailored for alkyl and aryl isocyanates.^{53,54} The innovative aspect of these approaches lie in the use of organocatalysts under mild and solvent free reaction conditions, either tetra-butylammonium phthalimide-*N*-oxyl or *tetra*-ethylammonium 2-(carbamoyl) benzoate or sulphate anions (Scheme 5c and d). These catalytic systems present the most popular and industrial

Scheme 5 Synthesis of uracil and aza-uracil derivatives *via* pseudo-MCR methodologies.

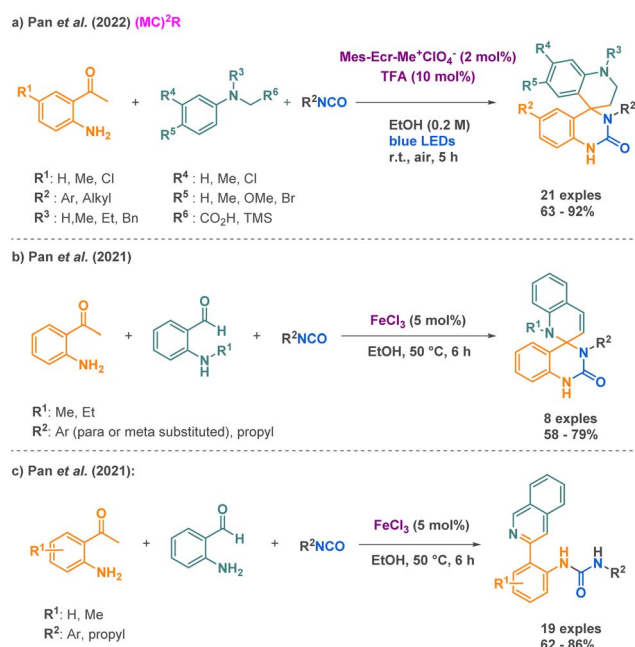
route for the cyclotrimerisation of isocyanates to access isocyanurates, as other methods described remain laborious.

Beyond hydantoins and uracil derivatives, cyclic ureas are present in several FDA-approved drugs, including avibactam, trimetaphan, and azlocillin. They also have garnered significant attention due to their potent biological activities as selective neurokinin-1 receptor antagonists⁵⁵ and HIV-1 protease inhibitors.⁵⁶ Cyclic urea can usually be prepared by cycloaddition, condensation, Curtius rearrangement of acyl azide, phosgenation followed by cyclisation, or ring-closing metathesis. These applications pushed forward several research groups to develop other methodologies to provide cyclic ureas using MCR methodologies.

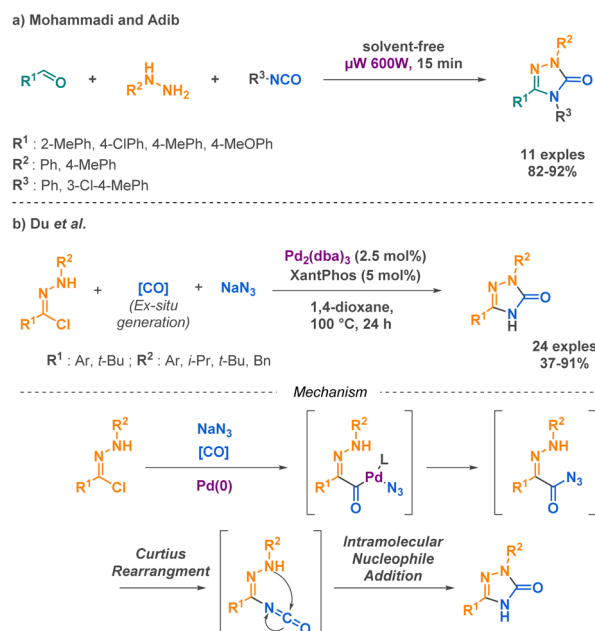
As a peculiar example, Pan *et al.* put forward an innovative approach using visible-light photoredox catalysis for the synthesis of spirocyclic quinazolin-2-ones in a single step, compounds for which a very limited number of methods are describe, compared to their analogues spirocyclic quinazolin-4-ones.⁵⁷ Behind the advantages of being a new strategy to access these complex structures, this methodology proceeds under mild conditions with the assistance of metal-free catalysts, visible light as energy activation, and air serving as a natural oxidant, enhancing its overall sustainability. The synthesis of the desired products involves the use of isocyanates, α -anilinosilanes, and 2-aminoacetophenones (Scheme 6a). With a scope composed by more than 30 examples, it however has to be pointed out that it remains mostly applicable to aryl isocyanates (one example with benzyl isocyanate, no conversion with *tert*-butyl isocyanate). The proposed mechanism involves both a Brønsted acid mediated catalytic cycle using TFA and a photoredox catalytic cycle using a catalytic amount of mesitylmethylacridinium perchlorate to obtain the quinazolinone core. Just a year before, they investigated a similar

multicomponent reaction using iron-III catalysis for the synthesis of spiroquinazolin-2-ones combining an isocyanate, a 2-aminoacetophenone, and a 2-amino-benzaldehyde (Scheme 6b).⁵⁸ In terms of scope limitations, if most examples report aryl isocyanates, one desired spirocyclic quinazolin-2-one from a propyl isocyanate shows a decent yield of 59%. Importantly, the aminoacetophenone needs to be N-Me or N-Et functionalized, exclusively, to avoid rearomatization of the quinolone motif. The reported mechanism starts with a first iron catalysed [4 + 2] annulation between 2-aminoacetophenone and an isocyanate, followed by an intramolecular cyclization, affording 4-methylene-quinazolinone. Then, a second iron catalysed [4 + 2] annulation followed by a cyclization afforded the desired product. It is worth mentioning that the presence of a hydrogen atom at the R¹ position leads to an additional aromatization of the spiroquinazolinone and the consequent formation of quinolinureas (Scheme 6c).

More complex urea derivatives such as semicarbazides and semicarbazones represent an attractive category of compounds that are already present on the market, such as naftazone which is used for the treatment of varicose veins or haemorrhoids. Ongoing studies on the stimulation of the dopamine D4 receptor by APH199,⁵⁹ implied in various psychiatric disorders further highlight the promising nature of these compounds for future applications. To meet the challenge of synthesizing these compounds, Mohammadi and Adib developed one of the few examples of microwave-assisted MCRs.⁶⁰ In their work, the authors demonstrated a highly efficient method for the synthesis of 1,2,4-triazole-3-one derivatives. By combining aldehydes, phenylhydrazines, and phenyl isocyanates, they achieved excellent yields (around 90%) under their reaction conditions. This represents a significant improvement over conventional heating methods, which typically afford low yields



Scheme 6 Synthesis of urea derivatives by Pan *et al.*



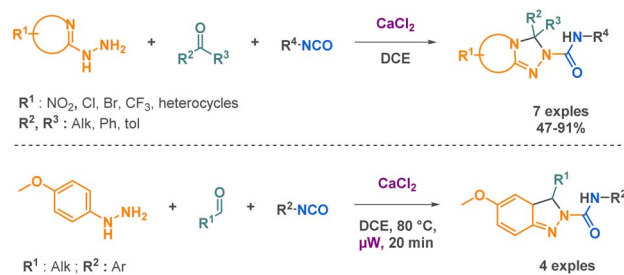
Scheme 7 Synthesis of cyclic semicarbazones by Mohammadi and Adib and Du *et al.*

for similar transformations. However, the substrate scope of this method was limited to aryl-substituted compounds (Scheme 7a). Du *et al.* successfully obtained similar products using a palladium-catalysed methodology involving hydrazonoyl chlorides, sodium azide and carbon monoxide surrogate as starting materials.⁶¹ The advantage of this method relies on the use of azide as substrate, leading to a free nitrogen in position N1 in the desired compounds. The proposed mechanism involves the carbonylation of hydrazonoyl chloride and the insertion of the azide under palladium catalysis. The acyl azide intermediate undergoes a Curtius rearrangement, resulting in a reactive isocyanate intermediate that facilitates intramolecular nucleophilic addition (Scheme 7b).

Concerning the preparation of cyclic semicarbazides, a highly effective and practical method was developed by Soeta *et al.*, directly inspired by the Ugi reaction.⁶² They substituted a carboxylic acid by an isocyanate to be reacted with a nitrile imine, an isocyanide and an isocyanate resulting in the formation of a complex 1,2,4-triazinedione scaffold (Scheme 8). The key challenge of this work was to replace carboxylic acids with isocyanates. Carboxylic acids play a critical role in the Ugi reaction mechanism by activating the imine intermediate, which enables the addition of an isocyanide and the formation of a nitrilium ion. This ultimately leads to the formation of the final product through an acyl group migration. However, to expand the chemical space of the reaction and access a wider range of products, it was necessary to explore alternative reaction partners such as isocyanates.

Martinez-Ariza *et al.* also described the preparation of semicarbazones using MCRs. They showcased various multicomponent strategies for this purpose. Among them, two involve the use of an isocyanate. The most studied one consists in the combination of a ketone, an isocyanate and 2-hydrazinoazine to afford triazolo-carboxamide leading to fused 6,6-bicyclic pyrido [2,1-*c*][1,2,4]triazines with enhanced sp^3 character over Groebke–Blackburn–Bienaymé derived scaffolds. It is the very first use of 2-hydrazinoazines in this particular MCR transformation, allowing access to a built-in H-bond donor–acceptor warhead property, highly advantageous for many biological targeting phenomena. Four examples of the scope of the reaction developed by the group involve an aldehyde, an isocyanate and a hydrazine instead of 2-hydrazinoazine, leading to the synthesis of indazole-carboxamides with correct yields around 50% (Scheme 9).⁶³

Among more than 40 linear urea examples, Babin *et al.* described the preparation of several linear semicarbazides through a Staudinger-Aza-Wittig (SAW) procedure able to generate isotopically labelled isocyanates *in situ* directly from carbon dioxide (¹³C, ¹⁴C, ¹¹C) and azides in the presence of

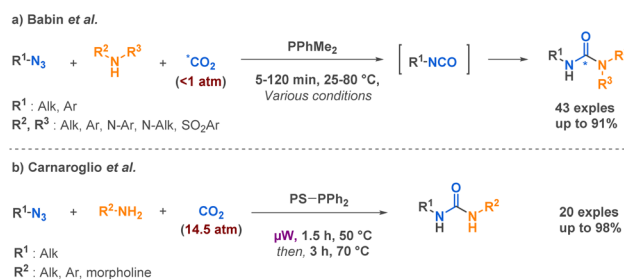


Scheme 9 Synthesis of triazolo-carboxamides and indazole-carboxamides by Martinez-Ariza *et al.*

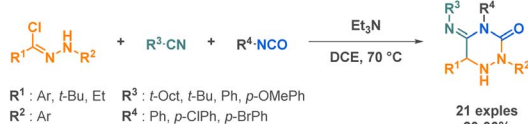
a phosphine.⁶⁴ This technique opens up new possibilities for radiolabeling, which holds a place of choice in pre-clinical studies. It is worth mentioning that the reaction conditions are influenced by the substituents on both the azide and the amine. However, a key advantage of this approach lies in its utilization of carbon dioxide in near-stoichiometric quantities (Scheme 10a). Another SAW-MCR was previously reported as a microwave assisted SAW procedure by Carnaroglio *et al.* in 2006 (Scheme 10b).⁶⁵ Despite employing supported phosphines to streamline purification, their method consumed over ten times more CO₂ than that developed by Babin *et al.* and exhibited a narrower scope of application.

Other teams have developed MCR-methods for the synthesis of linear urea derivatives. For example, Yavari and Nematpour's work offers an innovative solution to this complex challenge through their copper-catalysis method (Scheme 11).⁶⁶

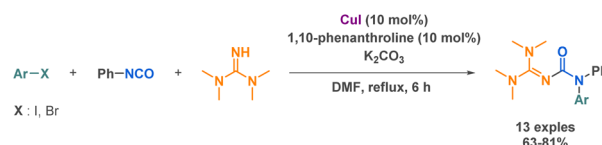
3.1.2. 4-MCR and 5-MCR. The greater the number of components, the more challenging the task becomes. Nevertheless, several teams tackled the challenge and developed methodologies for the preparation of urea derivatives using four or more components. To prepare uracil derivatives, Perrone *et al.* introduced a novel four-component reaction (4-MCR) in



Scheme 10 Synthesis of linear ureas derivatives by SAW procedures by Babin *et al.* and Carnaroglio *et al.*



Scheme 8 Ugi-type synthesis of cyclic semicarbazides by Soeta *et al.*



Scheme 11 Synthesis of linear ureas derivatives by Yavari *et al.*





Scheme 12 Synthesis of cyclic urea derivatives by Perrone *et al.*, Groenendaal *et al.* and Vugts *et al.*

2015, based on the simultaneous formation of an urea and the palladium-catalysed carbonylation of α -chloroketones.⁶⁷ The resulting products combined to form various uracil analogues. It is worth mentioning that depending on the substituents present on the urea moiety, the final product can either exhibits regioselectivity or exists as a mixture of both isomers (Scheme 12a). Groenendaal *et al.* also worked on the synthesis of uracils and developed a pseudo-5-MCR involving a phosphonate, a nitrile, an aldehyde and two isocyanates.⁶⁸ The mechanistic sequence is based on a Horner–Wadsworth–Emmons reaction that generates an azadiene intermediate, which subsequently undergoes nucleophilic attack on the isocyanate leading to the triazinane diones (Scheme 12b).

The same team implemented a comparable approach two years before to obtain dihydropyrimidines. Their strategic approach also involves the use of a Horner–Wadsworth intermediate, but instead of engaging in reactions with two isocyanates, the intermediate subsequently reacts with one isocyanate in an aza-Diels–Alder reaction (Scheme 12c).⁶⁹

The latest example of cyclic urea synthesis *via* a 4-MCR was reported by Messa *et al.*, who described the palladium catalysed formation of a pyrimidine-dione from an α -chloroketone, an isocyanate, an aniline and carbon monoxide (Scheme 13).⁷⁰ The



Scheme 13 4-MCR synthesis of a cyclic urea example by Messa *et al.*

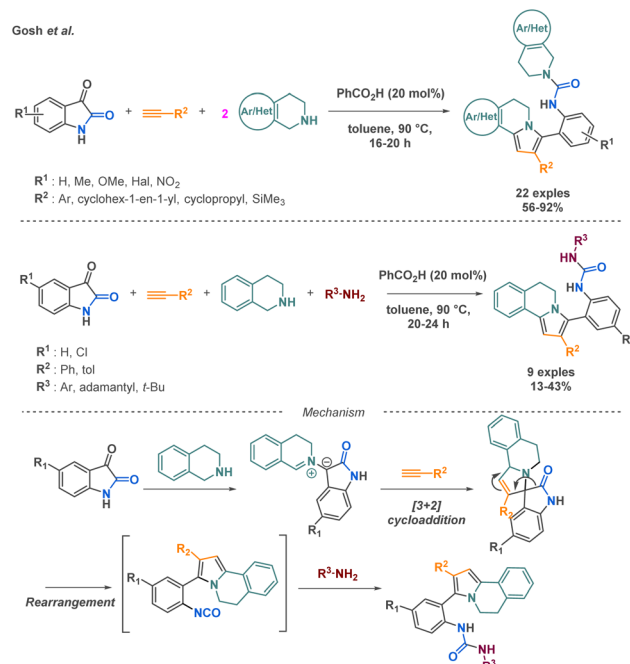
authors proposed a concerted mechanism where urea formation occurs simultaneously with the α -chloroketone carbonylation. These intermediates then cyclize to provide the desired product in 73% yield.

Sulfonylureas a subclass of linear urea derivatives, are particularly attractive due to their diverse applications across various fields, notably, in the treatment of type 2 diabetes. Zhao *et al.* developed a palladium-catalysed reaction combining sulfonyl azides with carbon monoxide to generate an isocyanate intermediate that undergoes a nucleophilic attack from a primary or secondary amine (Scheme 14).⁷¹ They successfully extended their methodology on a scope of 37 compounds with yields higher than 90% and applied it for the gram scale synthesis of Glibenclamide, an antidiabetic drug.

Ghosh *et al.* reported two multicomponent reaction (MCR) approaches, a pseudo 4-MCR and a standard 4-MCR, for the synthesis of 5,6-dihydropyrrolo[2,1-*a*]isoquinolines. The pseudo



Scheme 14 4-MCR for the preparation of sulfonylureas and the proposed mechanism by Zhao *et al.*



Scheme 15 4-MCR Synthesis of 5,6-dihydropyrrolo[2,1-*a*]isoquinolines by Ghosh *et al.*

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Scheme 17 Synthesis of sulfonylamide derivatives by Alizadeh *et al.* and Huang *et al.*

protocol operates under mild conditions, enabling the synthesis of complex molecules with high yields. Sulfonyl isocyanates have also been used by Huang *et al.* in combination with diazooxindoles and internal alkenes to generate complex scaffolds *via* MCR. Their innovative approach relies on the use of zinc catalysis, which facilitates $[2 + 2 + 1]$ annulation that produces spirooxindole derivatives (Scheme 17c).⁸¹ This method employs an abundant and inexpensive catalyst to enable the efficient synthesis of spirooxindole derivatives with yields up to 96% across 38 examples. The broad substrate scope and high catalytic efficiency of this approach represent a significant improvement over previous methods, which typically exhibit limitations in terms of both substrate compatibility and catalytic activity. The proposed mechanism involves an initial step where the isocyanate group activated by the zinc, goes through a nucleophilic attack of the diazooxindole leading to cyclization and the subsequent elimination of nitrogen gas. Further coordination between the zinc, the resulting intermediate and the alkene leads to a $[3 + 2]$ cycloaddition yielding the final product.

Imidazoles are nitrogen-containing heterocycles found in diverse biological contexts and found significant interest for their anticancer, anti-inflammatory, or antimicrobial properties.⁸² Three teams have proposed 3-MCR synthetic routes for their functionalisation with cyclic or linear amide moieties. First, Adib *et al.* reported a methodology involving imidazoles and isocyanates. This 3-MCR occurs with acetylenic esters as the third reactant. A mixture of the three compounds at room

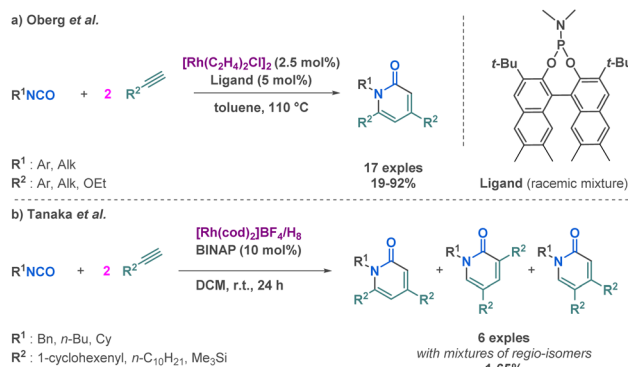
temperature in dichloromethane affords tetrahydroimidazo $[1,2-a]$ pyrimidines in one hour (Scheme 18a).⁸³ Although the scope evaluation is limited (6 examples), this method is an efficient, fast and easy strategy to access such ring systems having hydrogen at bridgehead carbon atom. Shen *et al.* reported the C2-functionalisation of nitrogen-substituted imidazoles. This reaction involves acetylenic ketones and isocyanates in combination with imidazoles (Scheme 18b).⁸⁴ The 3-MCR methodology, which operates under catalyst-free conditions, offers a straightforward approach to the synthesis of moderately complex molecules. The method exhibits excellent functional group tolerance and provides good to excellent yields (up to 98%) across a broad range of 21 substrates. The functionalisation at the C2 position can also be achieved using isocyanates and cyanophenylacetylene as reported by Belyaeva *et al.* (Scheme 18c) in a very similar approach.⁸⁵ The transformation allows the formation of *N*-(*Z*)-alkenylimidazole-2-carboxamides at room temperature, without catalyst and without solvent and goes through the *in situ* formation of imidazole carboxamide zwitterions. A further advantage of this reaction is its ability to proceed with complete *Z*-stereoselectivity, which the authors explained by the authors by the simultaneous cleavage of the C–N bond and the formation of the bond between the Nitrogen of the anionic carboxamide and the corresponding carbon (Scheme 18b, mechanism).

Since their development, several examples of MCR forming cyclic amide derivatives have been published. At the end of the 70's, Hong and Hoberg already investigated, respectively, cobalt



Scheme 18 Synthesis of imidazole derivatives by 3-MCR by Adib *et al.*, Shen *et al.* and Belyaeva *et al.*



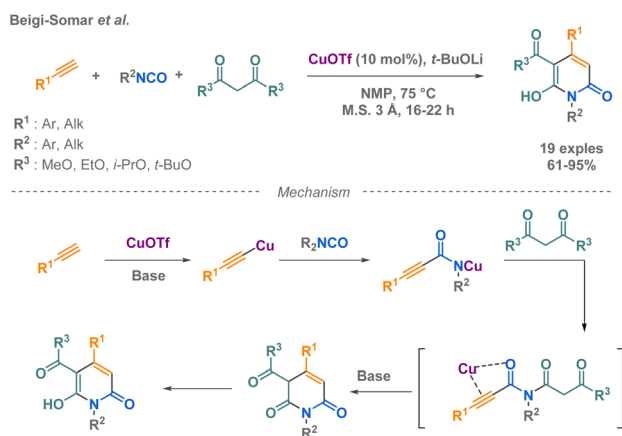


Scheme 19 Synthesis of amide derivatives by Oberg *et al.* and Tanaka *et al.*

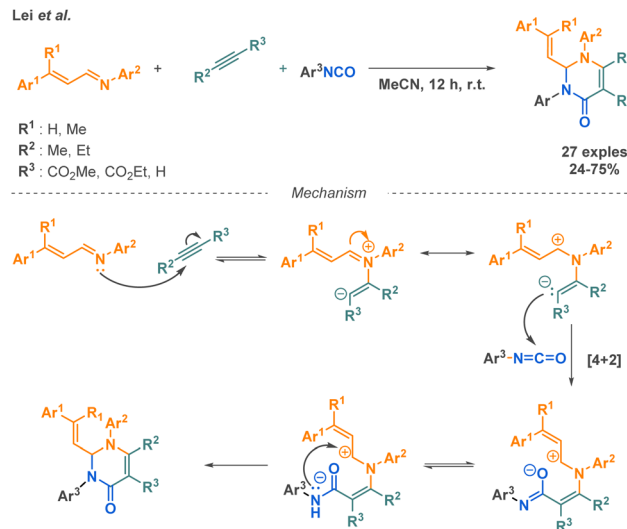
and nickel catalysis *via* a pseudo 3-MCR involving isocyanates and 2 equivalents of alkyne to prepare pyridone.^{86–89} Later on, Oberg *et al.* employed rhodium catalysis to achieve cycloadditions between terminal alkynes and isocyanates *via* another pseudo 3-MCR since it involves two equivalents of the alkyne (Scheme 19a).⁹⁰ This reaction came after the development of very similar rhodium catalysed pseudo 3-MCR in milder conditions by Tanaka *et al.* providing substituted 2-pyridones (Scheme 19b).⁹¹ Both studies have reported similar catalytic systems, pioneered by Tanaka *et al.* but Oberg *et al.* demonstrated the versatility of this approach by carefully selecting a ligand/metal complex that could direct the reaction toward different regioselectivities, depending on the nature of the starting material. Those two studies are highly complementary in regards of the synthesis of 2-pyridones by [2 + 2 + 2] cycloaddition.

In 2020, Beigi-Somar and co-workers prepared dihydropyridine-3-carboxylates from terminal alkynes, isocyanates and malonates (Scheme 20b).⁹²

Using a copper(I) catalysis facilitates the reaction by the formation of a copper acetylide *in situ*, which subsequently reacts with the electrophilic carbon of the isocyanate compound. The team hypothesized a mechanism and conducted experiments to test the regioselectivity of the reaction



Scheme 20 Synthesis of cyclic amide derivatives Beigi-Somar *et al.*



Scheme 21 Synthesis of pyrimidine derivatives by Lei *et al.*

using an unsymmetrical malonate, the ethyl methyl malonate. The results showed that 57% of the product formed was ethyl carboxylate, while 31% was methyl carboxylate. Although the scope of malonate remains limited, the efficiency of the reaction with a large scope of isocyanates has been demonstrated (R = alkyl, aryl) with good to excellent yields.

Combined with isocyanates, acetylene dicarboxylates are useful building blocks to access amide derivatives as presented in the above-mentioned reactions. Another example of MCR employing this starting material was described by Lei *et al.*, where they combined acetylene dicarboxylates with aryl isocyanates along with α,β -unsaturated imines to afford pyrimidone derivatives (Scheme 21).⁹³ Although the scope presents the evaluation of 24 new structures with moderate to good yields, the variation of functional groups remains restricted, as for instance only aryl isocyanates and *N*-phenyl imines are tolerated.

3.3. Carbamate derivatives

Carbamates are found in FDA-approved drugs with antitumor or anticonvulsant properties.^{94,95} Additionally, they represent a class of commonly used pesticides and serve as starting materials in the synthesis of polyurethanes.^{96,97} Their widespread use in medicinal chemistry is attributed to their biological compatibility and functional stability. Carbamate groups, acting as bioisosteres of amide bonds, offer improved pharmacokinetic properties and enhanced bioavailability due to their stability against proteases. Consequently, carbamate derivatives are of interest in peptidomimetic synthesis. The nucleophilic attack of alcohols on isocyanates is a common route to carbamate formation, which has been employed by multiple research teams in multicomponent reaction methodologies to access various scaffolds containing cyclic or linear carbamate moieties.

3.3.1. 3-MCR for the synthesis of cyclic carbamate derivatives. In 2008, Liu *et al.* proposed the first example of a 3-MCR yielding cyclic carbamates through the iodo-cyclization of

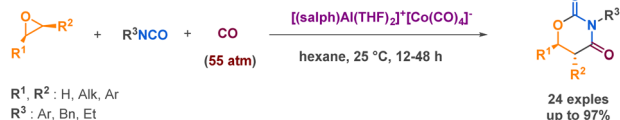


Scheme 22 Synthesis of cyclic carbamate derivatives Liu *et al.*

alkenes and alkynes under sodium nitrite catalysis. During their investigation, they expanded the methodology to the iodoamination of *O*-allylic carbamates. Finally, they tested this method using *in situ* generated carbamates from tosylisocyanate and reported several examples with moderate to good yields (Scheme 22).⁹⁸

Church *et al.*, also explored a catalytic 3-MCR process but this time using an aluminium complex. While investigating its ability to act as a catalyst in a reaction involving an isocyanate, carbon monoxide, and an epoxide, they demonstrated the selective production of 1,3-oxazinane-2,4-diones stereospecifically (Scheme 23).⁹⁹ The harsh reaction conditions (55 atm. of CO) and the complex synthesis of the catalyst limit the practical applicability of this methodology. However, the reaction demonstrates a broad tolerance for various substituents on the epoxide component, with enhanced selectivity observed for electron-deficient isocyanates. Additionally, the use of chiral epoxides enables the synthesis of enantiopure products.

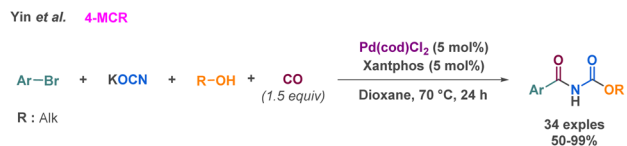
Potuganti *et al.* explored the synthesis of cyclic carbamates *via* a metal-catalysed multicomponent reaction (MCR) approach. They developed a novel strategy to produce spiro oxazolidinones using copper catalysis. The interest of the methods among other strategies to access spiro oxazolidinones relies on the mild and inexpensive conditions. By combining carbonyl compounds, arylacetylenes, and isocyanates, they successfully synthesized 24 compounds with moderate to good yields (up to 88% yields) (Scheme 24).¹⁰⁰

Scheme 23 Synthesis of oxazinanedione derivatives by Church *et al.*Scheme 24 Synthesis of cyclic carbamate derivatives by Potuganti *et al.*

3.3.2. 3-MCR and 4-MCR for the synthesis of linear carbamate derivatives. Linear carbamates are particularly interesting in the preparation of peptidomimetics. The only 3-MCR reported for this purpose was proposed by Abdou *et al.* in 2018, where they prepared two examples of phosphonate derivatives containing a carbamate functional group to investigate their antibacterial and antineoplastic properties (Scheme 25).¹⁰¹

As previously described by Church *et al.* carbon monoxide could be used to prepare cyclic carbamates *via* MCR.⁹⁹ On this base, Yin *et al.* combined aryl bromides and carbon monoxide to perform a 4-MCR palladium-catalysed carbonylation.¹⁰² Following this carbonylation, the potassium cyanate moiety coordinates to the palladium complex formed. Eventually, an alcohol performs a nucleophilic attack on the cyanate to provide a carbamate that couples with the carbonylated intermediate (Scheme 26). The advantage of this method relies on the milder conditions employed compared to Church *et al.* (1.5 equivalent of CO, catalytic amount of commercially accessible metal/ligands). Also, the use of CO allows the access to carbamate derivatives from aryl bromide as building blocks, a vast and

Scheme 25 Synthesis of linear carbamate derivatives by Abdou *et al.*



Scheme 26 Synthesis of linear carbamate derivatives by 4-MCR by Yin *et al.*

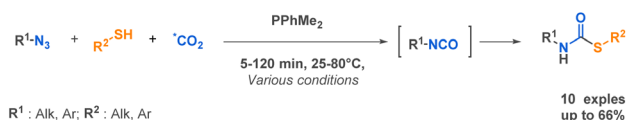
easily accessible feedstock. This is demonstrated by the large scope investigated by the authors (34 examples up to 99% yield).

3.4. Miscellaneous

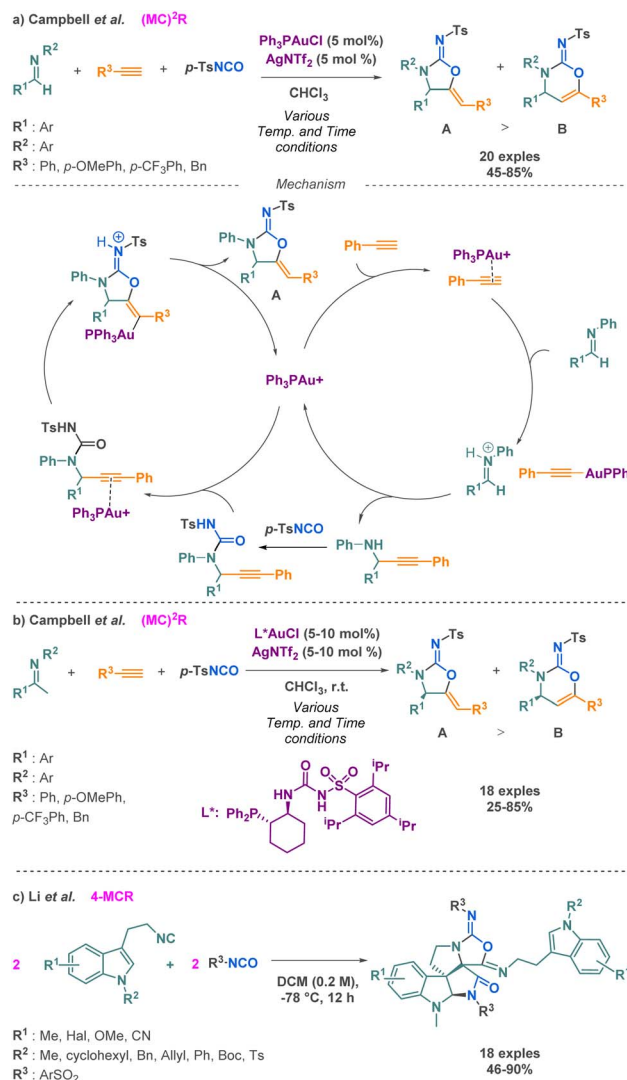
Thiocarbamates represent a class of compounds that can be prepared using an isocyanate and a thiol as nucleophilic agent. This strategy has been employed by Babin *et al.* in a 3-MCR. While they were investigating the robustness of the SAW reaction for linear urea synthesis they found that thiols directly competed with the amine, resulting in the predominant formation of thiocarbamate by-product. Building upon their *in situ* isocyanate generation method, the researchers further applied it to synthesize isotopically labeled thiocarbamates using thiols (Scheme 27).³⁴

Carbamimides is another class of compounds that can be accessed *via* isocyanate-based MCR. Among the three examples described in the literature, Campbell and Toste reported the synthesis of cyclic carbamimides using a (MC)²R approach.¹⁰³ Their technique combines imines, terminal alkynes, and *p*-toluenesulfonylisocyanate (*p*-TSNCO) under the influence of monophosphine gold catalysis. Using unusual ligands, *i.e.*, *trans*-1-diarylphosphino-2-aminocycloalkanes, they were able to propose the sequence with high regio- and enantioselectivities (41–95% ee). The proposed mechanism involves a gold-catalysed addition reaction between an alkyne and an imine, producing a propargyl-amine, which is then trapped with *p*-TSNCO to afford acyclic ureas. Then, a 5-*exo-dig* cyclization occurs *via* a nucleophilic attack of the urea oxygen to afford the desired product. This strategy allows for the conservation of the stereocenter along the synthetic sequence, where traditional biaryl bisphosphine ligands were not competent in the cyclization step (Scheme 28a). Moreover, by changing the monophosphine for a sulfonyl urea ligand, they were able to form product A enantioselectively (Scheme 28b).

In 2018, Li *et al.* conducted a novel pseudo 4-MCR yielding fused pyrocyclic bispiroindolines. These compounds belong to a group of biologically significant molecules, including strychnine. This Ugi-type reaction occurred between two 2-isocyanoethylindoles and two benzylsulfonates isocyanates, leading to the construction of polycyclic bispiroindolines



Scheme 27 Synthesis of thiocarbamates using a 3-MCR by Babin *et al.*



Scheme 28 Synthesis of carbamimide derivatives by Campbell *et al.* and Li *et al.*

(Scheme 28c).¹⁰⁴ This method presents the enormous advantage of a simple protocol to access large molecular complexity, rarely observed in a single-step synthesis.

In recent years, there have been several reports on Biginelli-like reactions incorporating isocyanates. Liang *et al.* showcased a one-pot synthesis of *N*-aryl-5-carboxyl-6-methyl cytidine derivatives utilizing an aromatic isocyanate, ethyl acetoacetate, and urea derivatives. This innovative approach highlights the remarkable potential of isocyanates as versatile building blocks for the synthesis of diverse nitrogen heterocycles (Scheme 29a).¹⁰⁵ Similarly, Elhefny *et al.* reported a 3-MCR affording amidine derivatives. This strategy was implemented to introduce the quinoxaline motif in biologically tested compounds. This scaffold holds significant interest due to its frequent presence in diverse antibiotic molecules (Scheme 29b).¹⁰⁶

To conclude this part on miscellaneous isocyanate-based MCR, one last example reported by Wang *et al.* in 2019 describes an interesting copper-catalyzed MCR to synthesize





Scheme 29 MCR reactions involving isocyanates by Liang *et al.* and Elhefny *et al.*



Scheme 30 MCR reactions involving isocyanates by Wang *et al.*

Boc-protected α -aminals from di-*tert*-butyl peroxide (DTBP), unactivated ethers and isocyanates (Scheme 30).¹⁰⁷

4. Conclusion

In conclusion, isocyanates have proven to be incredibly versatile building blocks in multicomponent reactions (MCRs), playing a key role in the efficient creation of complex and diverse molecules. Their ability to take part in 3-, 4-, and 5-component reactions has opened up exciting possibilities for designing compounds that are not only structurally varied but also biologically relevant. This makes isocyanates a valuable tool for developing new molecules with potential applications in fields like drug discovery and material science. Isocyanates, whether used as pre-formed reagents or generated *in situ*, contribute to the development of novel scaffolds with enhanced interaction potential for drug discovery and material science. The integration of isocyanates into MCRs, coupled with modern techniques like flow and microwave chemistry, has further expanded their applications. As MCR strategies evolve, isocyanates are poised to play an increasingly important role in accelerating the discovery of new functional compounds, particularly in pharmaceutical and agrochemical industries. Their use aligns well with green chemistry principles, making them an environmentally friendly and efficient choice for synthetic transformations.

Data availability

No new data were created or analysed in this study. Data sharing is not applicable to this article.

Author contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Conflicts of interest

There are no conflicts to declare.

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