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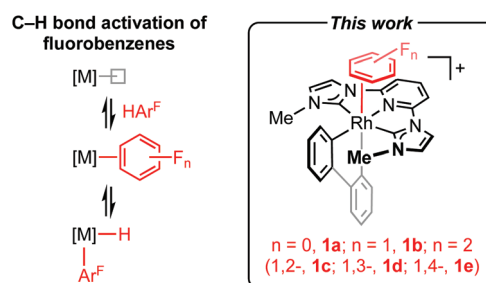
Isolation and structural characterisation of rhodium(III) η^2 -fluoroarene complexes: experimental verification of predicted regioselectivity†

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The isolation and solid-state characterisation of complexes featuring partially coordinated benzene, fluorobenzene and all three isomers of difluorobenzene are described. Supported by a DFT analysis, this well-defined homologous series demonstrates the preference for η^2 -coordination of fluoroarenes *via* the HC=CH sites adjacent to a fluorine substituent.

Partially fluorinated benzenes are chemically robust and weakly coordinating substrates, for which there is a paucity of late transition metal π -complexes.¹ Whilst well-defined examples can be found in the literature they are almost exclusively limited to half sandwich formulations, where the arene adopts an η^6 -coordination mode.^{1,2} The formation of η^2 -arene complexes is notably invoked in C–H bond oxidative addition of partially fluorinated benzenes to late transition metals (Scheme 1),^{3,4} but to the best of our knowledge isolation of mononuclear species of this nature is limited to coinage metal examples.⁵ Computational studies indicate a coordination site preference in the order HC=CH > HC=CF > FC=CF, with the strongest η^2 -arene complexes formed at the HC=CH positions adjacent to a fluorine substituent.⁶ Building on our work employing the high *trans*-influence 2,2'-biphenyl (biph) ancillary ligand,^{7,8} we herein present the synthesis and solid-state characterisation of rhodium(III) pincer complexes [Rh(CNC-Me)(biph)(η^2 -arene)]⁺ (Scheme 1; arene = C₆H₆, **1a**; FC₆H₅, **1b**; 1,2-F₂C₆H₄, **1c**; 1,3-F₂C₆H₄, **1d**; 1,4-F₂C₆H₄, **1e**) that corroborate this conclusion experimentally.

To enable systematic synthesis of the target complexes, [Rh(CNC-Me)(biph)(κ^1 -ClCH₂Cl)]⁺ **2** was ultimately identified



Scheme 1 Intermediacy of η^2 -adducts in the C–H bond activation of fluoroarenes. [B(3,5-(CF₃)₂C₆H₃)₄][−] anion omitted for clarity.

as the most convenient well-defined precursor and prepared using a silver-based transmetalation procedure involving reaction of [Ag(CNC-Me)]⁺ with [Rh(biph)Cl(*t*Bu₂PCH₂P*t*Bu₂)] and halide abstraction in CH₂Cl₂ (80% yield; see ESI† for solid-state structure, Rh–Cl = 2.5932(7) Å).† Dichloromethane is labile and not retained on dissolution of **2** in CD₂Cl₂ or neat fluoroarene, with the organometallic displaying time averaged C_{2v} symmetry at 298 K consistent with formulation as a five-coordinate complex in solution and rapid pseudorotation of the biphenyl ligand on the NMR time scale ($\Delta H^\ddagger$ = 75 ± 1 kJ mol^{−1}, $\Delta S^\ddagger$ = +80 ± 5 J K^{−1} mol^{−1}, $\Delta G_{298K}^\ddagger$ = 52 ± 3 kJ mol^{−1} in CD₂Cl₂).⁶ In the latter case, selective removal of CH₂Cl₂ *in vacuo* and subsequent recrystallisation from the neat fluoroarene enabled isolation of the corresponding η^2 -arene complexes **1b–e** in 61–81% yield. Benzene is a poor solvent for cationic species of this nature, but **1a** was prepared in a similar manner using a 1:1 molar mixture of benzene – 1,2-difluorobenzene in 77% yield. Crystals suitable for analysis by X-ray diffraction were obtained in all cases (Fig. 1), with bulk purity confirmed using a combination of combustion analysis, solid-state ¹⁹F MAS NMR spectroscopy and dissolution in CD₂Cl₂; with one equivalent of the respective free arene observed by ¹H and ¹⁹F NMR spectroscopy (see ESI).†

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† Electronic supplementary information (ESI) available: Full experimental and computational details, including NMR and IR spectra and ETS-NOCV deformation density plots (PDF), and optimised geometries (XYZ). CCDC 1988128–1988133. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/d0dt01137a

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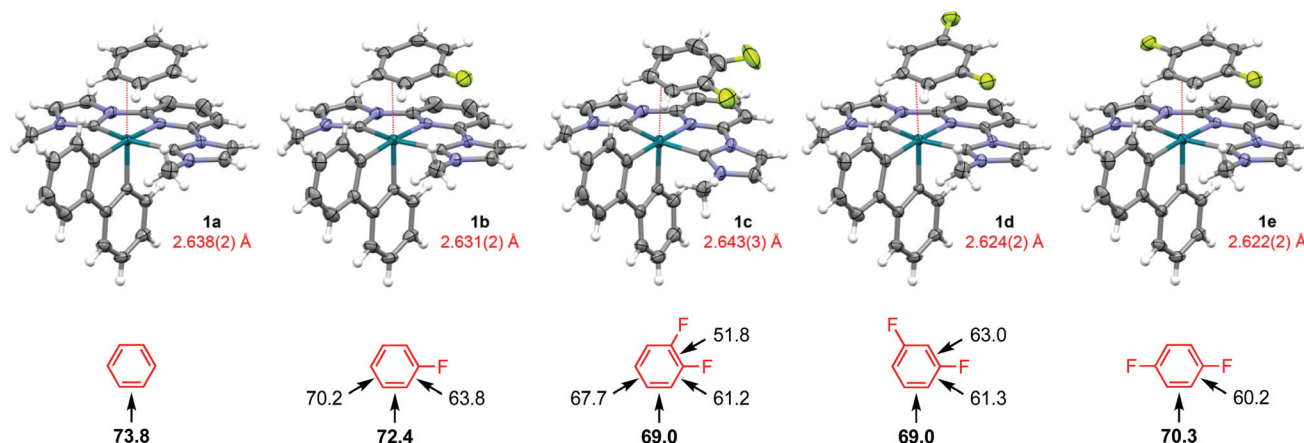


Fig. 1 Solid-state structures of η^2 -arene complexes **1a–e** (top) and calculated binding energies (kJ mol^{-1} , bottom). Solid-state structures drawn with thermal ellipsoids at 50% probability, minor disordered components (local symmetry equivalent F atom in **1b**) and $[\text{B}(\text{3,5-(CF}_3)_2\text{C}_6\text{H}_3)_4]^-$ anion omitted for clarity.

The rigid chelates of the biph and CNC pincer ligands provide a framework for pseudo-octahedral metal geometries in **1a–e**, where η^2 -arene coordination [$\text{Rh}-(\text{C}=\text{C}) = 2.622(2)–2.643(2) \text{ \AA}$] completes the coordination sphere and enables attainment of 18 VE configurations. The observed selectivity for coordination of the fluoroarenes *via* the $\text{HC}=\text{CH}$ sites adjacent to a fluorine substituent, notably vindicates computational trends in binding energy previously established for neutral rhenium cyclopentadienyl fragments⁶ and those determined as part of this study for **1a–e** at the $\omega\text{B97X-D3/def2-TZVP(f)}$ level of theory (Fig. 1). The absolute magnitudes of the calculated arene binding energies are considerably lower than the corresponding rhenium systems ($69.0–73.8$ vs. $87.0–99.3 \text{ kJ mol}^{-1}$ for the lowest energy regioisomers), consistent with the cationic nature of **1** and reconciling the entropically unfavourable coordination inferred in solution. Moreover, the relative binding energies of **1a/c** are supported by the aforementioned (competition) experiment involving dissolution of **2** in a 1 : 1 molar mixture of benzene – 1,2-difluorobenzene, yielding exclusively **1a**. DFT-based energy decomposition analysis of the metal-arene bonding interactions using the ETS-NOCV method, as implemented in ORCA 4.1.2,⁹ suggests these interactions are dominated by arene to metal σ -donation with only minor metal to arene π -backbonding contributions (see ESI).[†] The former are sufficient to explain the observed regioselectivity for all but **1c**, where subtle differences in π -backbonding are decisive.

In summary, we have exploited a planar NHC-based pincer ligand and the high *trans*-influence 2,2'-biphenyl ancillary to prepare an unprecedented homologous series of rhodium(III) complexes featuring η^2 -coordinated benzene and fluoroarenes. Supported by a DFT analysis, these complexes provide evidence for preferential η^2 -coordination of fluoroarenes *via* the $\text{HC}=\text{CH}$ sites adjacent to a fluorine substituent; an important finding relevant to the selective C–H activation of these valuable fluoroaryl synthons.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 S. D. Pike, M. R. Crimmin and A. B. Chaplin, *Chem. Commun.*, 2017, 53, 3615–3633.
- 2 For recent rhodium examples see: (a) A. I. McKay, J. Barwick-Silk, M. Savage, M. C. Willis and A. S. Weller, *Chem. – Eur. J.*, 2020, **26**, 2883–2889; (b) A. L. Colebatch, A. I. McKay, N. A. Beattie, S. A. Macgregor and A. S. Weller, *Eur. J. Inorg. Chem.*, 2017, 4533–4540; (c) S. D. Pike, I. Pernik, R. Theron, J. S. McIndoe and A. S. Weller, *J. Organomet. Chem.*, 2015, **784**, 75–83; (d) I. Pernik, J. F. Hooper, A. B. Chaplin, A. S. Weller and M. C. Willis, *ACS Catal.*, 2012, **2**, 2779–2786; (e) A. B. Chaplin, J. F. Hooper, A. S. Weller and M. C. Willis, *J. Am. Chem. Soc.*, 2012, **134**, 4885–4897.
- 3 O. Eisenstein, J. Milani and R. N. Perutz, *Chem. Rev.*, 2017, **117**, 8710–8753.
- 4 For spectroscopic evidence see: J. J. Carbó, O. Eisenstein, C. L. Higgitt, A. H. Klahn, F. Maseras, B. Oelckers and R. N. Perutz, *Dalton Trans.*, 2001, 1452–1461.
- 5 For representative examples see: (a) A. Martens, P. Weis, M. C. Krummer, M. Kreuzer, A. Meierhöfer, S. C. Meier,



- J. Bohnenberger, H. Scherer, I. Riddlestone and I. Krossing, *Chem. Sci.*, 2018, **9**, 7058–7068; (b) M. M. D. Roy, M. J. Ferguson, R. McDonald and E. Rivard, *Chem. Commun.*, 2018, **54**, 483–486; (c) R. Ramírez-Contreras and O. V. Ozerov, *Dalton Trans.*, 2012, **41**, 7842–7844; (d) G. Santiso-Quinones, A. Higelin, J. Schaefer, R. Brückner, C. Knapp and I. Krossing, *Chem. – Eur. J.*, 2009, **15**, 6663–6677.
- 6 E. Clot, B. Oelckers, A. H. Klahn, O. Eisenstein and R. N. Perutz, *Dalton Trans.*, 2003, 4065–4074.
- 7 T. M. Hood, B. Leforestier, M. R. Gyton and A. B. Chaplin, *Inorg. Chem.*, 2019, **58**, 7593–7601.
- 8 (a) T. M. Hood, M. R. Gyton and A. B. Chaplin, *Dalton Trans.*, 2020, **49**, 2077–2086; (b) J. Emerson-King, I. Prokes and A. B. Chaplin, *Chem. – Eur. J.*, 2019, **25**, 6317–6319; (c) M. R. Gyton, B. Leforestier and A. B. Chaplin, *Organometallics*, 2018, **37**, 3963–3971; (d) R. C. Knighton, J. Emerson-King, J. P. Rourke, C. A. Ohlin and A. B. Chaplin, *Chem. – Eur. J.*, 2018, **24**, 4927–4938.
- 9 (a) A. Altun, F. Neese and G. Bistoni, *J. Chem. Theory Comput.*, 2019, **15**, 215–228; (b) M. P. Mitoraj, A. Michalak and T. Ziegler, *J. Chem. Theory Comput.*, 2009, **5**, 962–975; (c) F. Neese, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2018, **8**, e1327; (d) F. Neese, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2012, **2**, 73–77.

