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Antimicrobial textiles based on photocrosslinked poly(ethylene-co-acrylic acid)[†]

Yimin Zeng  and Michael O. Wolf  *

Photocrosslinking of a series of amine-containing poly (ethylene-co-acrylic acids) (PEAAs) on textile surfaces by reaction of photogenerated singlet oxygen at the amine groups to form imines is reported. The materials are hydrophilic and show antimicrobial activity. Modified PEAAs with 3 different lengths of side chains are synthesized to study the effect of varying side chain length on polymer photooxidative cross-linking and antimicrobial activity. Materials with longer side chains show elimination of *E. coli* and MRSA by contact lysis which is enhanced under irradiation with green light.

Introduction

Viral and bacterial infections can be transmitted by surface contamination.¹ Bacterial infections are the second leading cause of death worldwide, affecting individuals in all age groups. In 2019, an estimated 7.7 million deaths were attributed to 33 bacterial pathogens, constituting 13.6% of all worldwide fatalities.² Generally, disinfectants are used to sanitize surfaces to prevent infection by bacteria, viruses, and other pathogens. This labour-intensive process involves repeatedly wiping or spraying disinfectants to ensure continuous sanitization. In the process, volatile and toxic disinfectants can be released into the environment possibly resulting in ecological damage. To address this issue, modified surfaces that kill or inhibit the growth of microbes to reduce or eliminate disease transmission have been developed.³ Modification of textiles with coatings that have inherent antimicrobial properties that can prevent the spread of diseases is one approach that is being explored. By applying an antimicrobial coating to textiles, the spread of pathogens can be mitigated in health care and other high-risk settings, for example on hospital gowns, bedding, and clothing.^{4,5}

Antimicrobial surfaces can be prepared by functionalization with primary amine or phosphonium containing polymers, quaternary ammonium salts (QASs), or metal nanoparticles.^{6–10} Surfaces modified with these functionalities can act to kill bacteria through passive pathways, such as contact lysis or ion release, where the bacterial cell membranes are disrupted. Hydrophobic polymer coatings can also prevent

moisture accumulation and reduce microbial adhesion on surfaces.¹¹ Additionally, photosensitizers can be used to generate singlet oxygen (¹O₂, the first excited state of oxygen) or free radical ions that destroy microbial membranes, a process called antimicrobial photodynamic inactivation (aPDI).^{12–14} Coatings based on QASs and metal nanoparticles are passive while aPDI gives the possibility of active photosensitization utilizing light energy to destroy pathogens *via* singlet oxygen generation. Typically, the photosensitizer is excited to a singlet state with visible light and then undergoes intersystem crossing to a triplet state, followed by triplet-triplet energy transfer to ground state oxygen, forming ¹O₂.^{15,16}

Our group recently developed dual-functional antimicrobial polymers, which employ primary amine-containing amino-propylmethylsiloxane (6–7 wt%)-dimethylsiloxane copolymers (PDMS-NH₂) with a covalently attached photosensitizer, rose bengal (RB), that generate ¹O₂ on irradiation with green light.¹⁷ Others have trapped RB in the pores of organic polymers on cotton surfaces and demonstrated biocidal activity.¹⁸ Our PDMS-NH₂ polymer is photocrosslinked by reaction with ¹O₂ at the amine groups to form imines. By forming covalent bonds between polymer chains a robust coating on textile surfaces results which prevents polymer leaching into the aqueous environment in real-world applications.^{11,19} Textiles can be modified by soaking with a solution of the polymer followed by photocrosslinking resulting in the formation of a durable antimicrobial coating.^{17,20}

Photocrosslinking PDMS-NH₂ results in a hydrophobic coating on textile substrates attributed to the presence of the alkyl side chains on the polysiloxane backbone. Although a hydrophobic polymer coating on fabric can reduce surface bacterial growth, there are disadvantages associated with the use of hydrophobic materials, such as poor durability in clothing applications.²¹ Moreover, silicones are not biodegradable or recyclable raising sustainability concerns. Here we report

Department of Chemistry, University of British Columbia, 2036 Main Mall, Vancouver, British Columbia, V6T 1Z1, Canada. E-mail: mwolf@chem.ubc.ca

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Results and discussion

PEAA

DCC,
1,3-diaminopropane

PEAA-3C-NH₂

neat,
405 nm

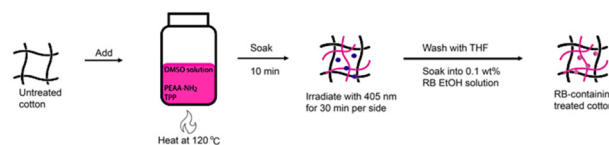
TPP
O₂

TPP

Absorbs at 405 nm

Crosslinked PEAA-3C-NH₂

The average water contact angle of cotton treated with 1 wt% PDMS-NH₂ is $134 \pm 9^\circ$, while for all PEA-NH₂ samples water is absorbed, and a contact angle cannot be measured on the cotton surface (Fig. S10[†]). This indicates that while the



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Fig. 2 A comparison of FTIR spectra of PEEA, PEEA-8C-NH₂, cross-linked PEEA-8C-NH₂, and PEEA-8C-NH₂ irradiated without photosensitizer. Similar spectra were observed for PEEA-3C-NH₂, PEEA-6C-NH₂, and PEEA-8C-NH₂ between 1100 and 2100 cm⁻¹.

1 wt% PDMS-NH₂ solution-treated cotton is hydrophobic, the PEEA-NH₂ samples are very hydrophilic. Tensile tests (fabric stretching) were performed to probe if the physical properties of the modified fabrics are changed from prior to treatment. The load average of untreated plain cotton is 16 ± 2 N and 17 ± 2 N for 1 wt% PEEA-8C-NH₂ solution-treated cotton (Table S2†). Thus, polymer treatment results in a minimal tensile strain change.

Antimicrobial tests were carried out with *Escherichia coli* (*E. coli*) and methicillin resistant *Staphylococcus aureus* (MRSA) with a concentration of 10^5 bacteria per mL. Identical sized pieces of coated cotton were used to compare the bacterial killing efficiency between PDMS-NH₂ and the three PEEA-NH₂ derivatives. To allow generation of ¹O₂ under irradiation, all coated fabrics were soaked in a 0.1 wt% RB solution for 3 min. The average loadings of RB on fabrics treated with PEEA-3C-NH₂, PEEA-6C-NH₂ and PEEA-8C-NH₂ were determined by measuring the absorption of extracted RB at 560 nm (Table S4†). Unmodified cotton was also treated with RB under the same conditions for comparison. The average weight percentages (wt%) of RB on solution treated fabric were determined to be $3.7 \pm 0.1\%$ for plain cotton, $3.3 \pm 0.2\%$ for PEEA-3C-NH₂ solution-treated cotton, $3.0 \pm 0.2\%$ for PEEA-6C-NH₂ solution-treated cotton, and $3.3 \pm 0.4\%$ for PEEA-8C-NH₂ solution-treated cotton. The PEEA coatings result in only a minor change in the RB loading relative to plain cotton.

Contact lysis experiments were carried out by soaking the polymer solution-treated cotton sample in a bacterial culture solution in the dark. Photosensitization was conducted by irra-

diating a bacterial culture containing coated fabrics with a 15 W 530 nm green light. In both experiments, the bacterial cultures were assessed over 150 min to monitor bacterial growth under treatment. No reduction in the number of bacteria was observed when bacterial solutions or untreated textiles were irradiated with green light (Fig. S11†). Both 0.5 wt% PEEA-8C-NH₂ and 0.5 wt% PDMS-NH₂ solution-treated cotton behaved the same by gradually reducing the number of *E. coli* colony-forming units (CFUs) over 150 min (Fig. S12†). In this case, the 0.5 wt% PEEA-8C-NH₂ solution-treated cotton showed an average of 85% of *E. coli* elimination and 30% killing of MRSA through contact lysis (Fig. 3 and Fig. S13†). All bacteria were killed under 10 min irradiation with green light. This indicates that changing the hydrophobicity of the polymer coating significantly does not change the dual-functional antimicrobial capability. The bacterial killing behaviour under irradiation is dominated by ¹O₂ produced by excitation of the RB. Identical treatment and antimicrobial assays were carried out with PEEA-3C-NH₂ and PEEA-6C-NH₂ in comparison to PEEA-8C-NH₂ to study the impact of amine chain

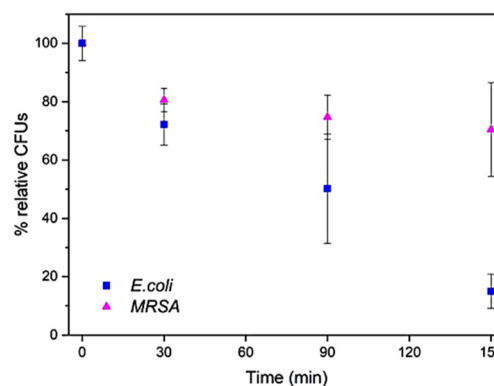


Fig. 3 A comparison of contact lysis for *E. coli* and MRSA on 0.5 wt% PEEA-8C-NH₂ solution-treated cotton over time.

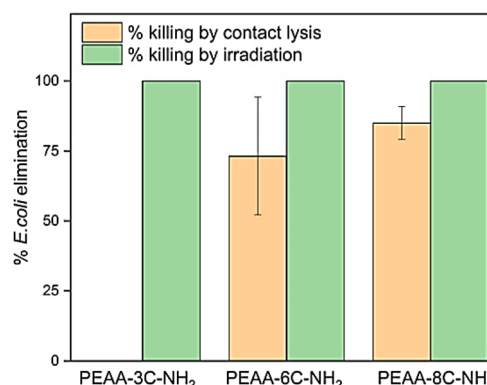


Fig. 4 A comparison of percent *E. coli* elimination by singlet oxygen and contact lysis on 0.5 wt% PEEA-3C-NH₂, PEEA-6C-NH₂, and PEEA-8C-NH₂ solution-treated cotton over time. *E. coli* killing by singlet oxygen was performed under 10 min irradiation with green light, where contact lysis was performed in the dark over 150 min.



length on bacterial killing. All bacteria were killed with 10 min irradiation by PEAA-3C-NH₂, PEAA-6C-NH₂, and PEAA-8C-NH₂. PEAA-6C-NH₂ behaved the same as both PEAA-8C-NH₂ and PDMS-NH₂, while PEAA-3C-NH₂ had no effect on bacteria present by contact lysis, even with a higher amount of amine groups remained after crosslinking (Fig. 4 and Table S1†). This can be rationalized by the short alkyl chain in PEAA-3C-NH₂ which is less flexible for insertion into the bacteria cell membrane compared with the longer analogs. Although PDMS-NH₂ also has amine groups that are 3 carbons away, the siloxane backbone in PDMS-NH₂ may be more flexible than the carbon backbone in PEAA-3C-NH₂.

Conclusions

In summary, we report a series of new dual-functional organic antimicrobial polymer analogs to PDMS-NH₂ to explore the effect of modifying the hydrophilicity of polymer coatings on fabrics while retaining its bacterial killing property. PEAA-3C-NH₂, PEAA-6C-NH₂, and PEAA-8C-NH₂ can all be crosslinked under 405 nm light with an average of 36% amine remained after crosslinking. PEAA-6C-NH₂ and PEAA-8C-NH₂ behave the same as PDMS-NH₂ in antimicrobial tests, where *E. coli* and MRSA were eliminated over 150 min by contact lysis and 10 min under irradiation of green light. The PEAA-NH₂ treatment results in a hydrophilic coating on fabrics while not changing the tensile properties of the textile.

Experimental section

General

PEAA and RB lactone (95%) were obtained from Sigma-Aldrich. DCC, 1,3-diaminopropane, 1,6-diaminohexane, and 1,8-diaminooctane were purchased from Thermo Fisher Scientific. TPP was purchased from STREM. Synthesis of PEAA-NH₂ was carried out under nitrogen with properly dried glassware and solvents. ¹H, HSQC and HMBC NMR experiments were carried out to obtain NMR peak assignments in the NMR lab at UBC. FTIR experiments were carried out in the Shared Instrument Facility (SIF) at UBC. Scheme 1 was followed to synthesize the desired products.

PEAA-8C-NH₂

0.5 g of PEAA was mixed with 0.1 g (0.48 mmol) of DCC and 0.8 g (5.5 mmol) of 1,8-diaminooctane. The neat reaction mixture was heated at 120 °C under N₂ for 24 h. The desired amine containing PEAA-NH₂ was precipitated with methanol and washed by centrifugation. The product was dried under vacuum. ¹H NMR (120 °C, 400 MHz, DMSO): δ 2.51 (m, α -H, overlap with DMSO), δ 2.58 (t, amide-H₂).

PEAA-3C-NH₂ and PEAA-6C-NH₂ were prepared using the same procedure with excess 1,3-diaminopropane or 1,6-diaminohexane instead of 1,8-diaminooctane.

Polymer crosslinking

In 100 μ L of DMSO solution, 50 mg of PEAA-NH₂ was mixed with 0.1 mg (0.16 μ mol) of TPP. The reaction mixture was irradiated with a 36 W 405 nm LED bulb for 30 min at 120 °C to crosslink the PEAA-NH₂.

Textile treatment

Cotton pieces were soaked in polymer solutions in DMSO at 120 °C with TPP for 10 min in the dark. DMSO was removed by soaking fabrics in stirring water. The fabrics were air dried and irradiated with a 36 W 405 nm LED bulb for 30 minutes per side. The light bulb was placed 9 cm above the samples. Once the polymer was coated onto cotton, residual TPP was removed by a THF wash. The treated cottons were soaked into a 0.1 wt% RB EtOH solution for 3 min and dried under vacuum.

Loading of RB on treated fabrics

After treating coated fabrics with 0.1 wt% RB EtOH solution, RB was redissolved into 8 mL of EtOH by sonication. The absorption of redissolved RB was measured at 560 nm using a Cary 5000 UV/Vis spectrometer (Agilent Technologies Inc., Santa Clara).

Antimicrobial tests

E. coli and MRSA (level 2, from the Biological Services Laboratory, UBC Chemistry) were used. Samples from the same batch of tensile tests were employed to ensure identical surface area. The cultures were prepared with a concentration of 10⁵ bacteria per mL and 10 mL was transferred into Falcon tubes contain polymer-coated cotton. Samples were treated with or without irradiation for comparison. A 15 W 530 nm LED bulb was set 9 cm above the samples. The optical power at the center of the lamp was 13 mW cm⁻² measured using a thermal power meter (Ophir 3A-P-SH). 20 μ L of sample solution was taken at each time point and incubated on agar gel. Results were collected in triplicate.

Hydrophobicity test

A drop of deionized water (10 μ L) was placed on each fabric sample and measured from the horizon. Photos were taken with an iPhone XR and analysed by ImageJ.

Tensile tests

Polymer coated cottons were cut into identical dog bone shape (10 mm in the course direction and 37 mm in the wale direction) by laser cutting. Tensile measurements on polymer-treated cottons were measured until the break point using an Instron 5980 instrument under a 2 kN load at 20 cm min⁻¹ with a clamp distance of 30 mm. Results were averaged from ten samples for each comparison group.

Ninhydrin test

50 mg of PEAA-NH₂ was mixed with 0.1 mg (0.16 μ mol) of TPP in 100 μ L of DMSO. The reaction mixture was irradiated under



identical conditions to those used to crosslink PEAA-NH₂. For comparison, samples were also prepared the same way without TPP. A ninhydrin solution in ethanol (10 µL, 3.5 mg mL⁻¹) was added to samples while heating. The absorption of ninhydrin after reaction was obtained using an Infinite M1000 Pro plate reader (Tecan Ltd, Morrisville) on a non-binding 96-well plate. Measurements were carried out in triplicate.

Author contributions

The manuscript was written through contributions of all authors.

Data availability

The data supporting this article have been included as part of the ESI.†

Conflicts of interest

The materials disclosed in this paper are included in a provisional patent that has been filed.

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