


 Cite this: *RSC Adv.*, 2022, **12**, 10062

¹⁹F-centred NMR analysis of mono-fluorinated compounds†

 Alan J. R. Smith, Richard York, Dušan Uhrin  and Nicholle G. A. Bell *

Addressing limitations of the existing NMR techniques for the structure determination of mono-fluorinated compounds, we have developed methodology that uses ¹⁹F as the focal point of this process. The proposed ¹⁹F-centred NMR analysis consists of a complementary set of broadband, phase-sensitive NMR experiments that utilise the substantial sensitivity of ¹⁹F and its far reaching couplings with ¹H and ¹³C to obtain a large number of NMR parameters. The assembled ¹H, ¹³C and ¹⁹F chemical shifts, values of J_{HF} , J_{HH} , and J_{FC} coupling constants and the size of ¹³C induced ¹⁹F isotopic shifts constitute a rich source of information that enables structure elucidation of fluorinated moieties and even complete structures of molecules. Here we introduce the methodology, provide a detailed description of each NMR experiment and illustrate their interpretation using 3-fluoro-3-deoxy-D-glucose. This novel approach performs particularly well in the structure elucidation of fluorinated compounds embedded in complex mixtures, eliminating the need for compound separation or use of standards to confirm the structures. It represents a major contribution towards the analysis of fluorinated agrochemicals and (radio) pharmaceuticals at any point during their lifetime, including preparation, use, biotransformation and biodegradation in the environment. The developed methodology can also assist with the investigations of the stability of fluoroorganics and their pharmacokinetics. Studies of reaction mechanisms using fluorinated molecules as convenient reporters of these processes, will also benefit.

 Received 2nd November 2021
 Accepted 15th March 2022

DOI: 10.1039/d1ra08046f

rsc.li/rsc-advances

Introduction

Fluorine's unique properties, such as high electronegativity, strength of a single fluorine-carbon bond and small atomic radius, impart significant benefits to fluorinated organic molecules.¹ Fluorination has been shown to enhance potency and/or specificity of molecular interactions, increase membrane permeability, modulate metabolism, moderate the pK_a of proximal functionalities, influence conformation, stabilise inherently reactive functionalities and produce viable bioisosteres.^{2,3} Currently, about 20% of the commercial pharmaceuticals contain fluorine and the proportion of newly approved fluoro-pharmaceuticals is rising steadily.^{4,5} The proportion of fluoro-agrochemicals is even larger; 53% of all active agrochemicals registered during 1998–2020 belong to this category.⁶ Similarly, ¹⁸F is the most frequently used radioisotope in positron emission tomography radiopharmaceuticals.⁷ Fluorination also has the potential to become a useful tool for improving properties of fragrance and semiochemical molecules.⁸

To capitalise on the ability of fluorine to improve molecular properties, there is a drive to design efficient and

environmentally-safe chemical,^{9,10} enzymatic¹¹ and chemo-enzymatic^{12–14} fluorination methods. To assist these efforts, efficient analytical methods for the characterisation of fluorinated molecules are required. ¹⁹F NMR spectroscopy plays a prominent role in this area due to the favourable properties of ¹⁹F, such as its high sensitivity, 100% natural abundance, large chemical shift dispersion, large and far-reaching spin-spin interactions and ¹³C induced ¹⁹F isotopic shifts.

The lack of background ¹⁹F signals, due to the scarcity of fluorinated endogenous compounds, makes ¹⁹F NMR perfect for the analysis of mixtures produced by chemical or chemo-enzymatic reactions with minimum clean-up steps or compound separation required. 1D ¹⁹F NMR spectroscopy has been widely used in studies of biodegradation and biotransformation of fluorinated compounds^{15–21} mostly relying on the use of known standards¹⁵ or tabulated ¹⁹F chemical shifts. In a similar manner, ¹⁹F NMR has also been used for probing the mechanism and kinetics of chemical reactions, where fluorine is a convenient reporter of the processes taking place.^{22,23}

In support of such wide ranging activities, we have developed a ¹⁹F-centred NMR approach for the analysis of mono-fluorinated compounds, taking ¹⁹F NMR beyond recording simple 1D NMR spectra. Put together, the information obtained allows the structure elucidation of fluorine-containing molecular moieties and complete structure determination of small fluorine-containing molecules. It is well suited for the studies of complex mixtures.

EaStCHEM School of Chemistry, University of Edinburgh, David Brewster Rd, Edinburgh, EH9 3FJ, UK. E-mail: Nicholle.Bell@ed.ac.uk

† Electronic supplementary information (ESI) available. See DOI: 10.1039/d1ra08046f



The ^{19}F -centred NMR shares similarities to the “NMR spy” approach developed for the analysis of complex mixtures of soil organic matter, where $-\text{O}^{13}\text{CH}_3$ tags are introduced to a subset of molecules.^{24–26} Nevertheless, there are significant differences between the two approaches. Firstly, fluorinated molecules already contain ^{19}F and therefore do not require additional chemical modifications. Secondly, the fluorine atom is typically closer to the protons and carbons of an organic molecule than are the nuclei of the $-\text{O}^{13}\text{CH}_3$ group which, when combined with far reaching ^{19}F couplings, allows to inspect parts of the molecule that are more remote from the ^{19}F “tag.” The FESTA family of NMR experiments^{27–29} that relies on selective manipulation of individual ^1H and ^{19}F resonances illustrated this approach and provided ^1H – ^{19}F chemical shift correlations and ^1H – ^{19}F coupling constants when such spin manipulations were possible.

Our methodology utilises the far reaching ^1H – ^{19}F and ^{19}F – ^{13}C couplings to obtain ^1H and ^{13}C chemical shifts of nuclei multiple bonds away from the ^{19}F atom, provides accurate values of numerous J_{HF} , J_{FC} , and J_{HH} coupling constants and ^{13}C induced ^{19}F isotopic shifts from several purposely designed nonselective 2D NMR experiments. Their advantages over similar existing NMR experiments are highlighted. The ^{19}F -centered approach is illustrated using 3-fluoro-3-deoxy-D-glucose, **1**, which can be characterized as a simple mixture of two ^{19}F -containing molecules. Application of this methodology to a very complex mixture

of compounds produced by chloramination of a single fluorinated molecule is presented elsewhere.³⁰

Experimental

The sample of 3-fluoro-3-deoxy-D-glucose (30 mg), **1**, was dissolved in 600 μL of D_2O (Merck, 99.9 atom% D) and placed into a 5 mm NMR tube. Spectra involving ^{19}F were acquired at 300 K on a 400 MHz Bruker Avance III NMR spectrometers equipped with a TBO BB-H/F-D probe. A 1D ^1H spectrum was acquired on an 800 MHz Bruker Avance III NMR spectrometer equipped with a TCI 5 mm probe. Parameters of the performed NMR experiments are presented in Table S1† and Bruker pulse sequences compatible with TopSpin 3 can be found in the ESI† (pp. 1–6).

The following symbols are used to depict the pulse sequences in Fig. 1–6: the thin and thick filled rectangles represent high power 90° (^1H , $p1$ or ^{19}F , $p3$) and 180° (^1H , $p2$) pulses, respectively. 1 ms adiabatic CHIRP pulses with a peak power of 10.3 kHz ($p44$, shaded trapezoid with an inclined arrow) were applied to ^{19}F . A 20 ms 60 kHz CHIRP ^1H pulse with a peak power of 2294 Hz ($p32$, trapezoid with inclined arrow) was used as part of the z-filter. A 500 μs CHIRP pulse ($p14$) and 2 ms composite CHIRP pulse ($p24$) were applied to ^{13}C with a peak power of 9800 Hz. Unless stated otherwise, the r.f. pulses were applied from the x-axis. The 100% pulsed field gradient strength corresponds to 53.5 G cm^{-1} .



Fig. 1 (a) 400 MHz 1D ^1H spectrum of **1** (structure in the inset) with HOD suppression and resonance assignments; (b) ^1H -coupled 1D ^{19}F spectrum of **1**. Expansions of ^{19}F multiplets of both anomeric forms from resolution-enhanced spectra produced using Lorentzian to Gaussian line shape conversion (LB = -2.0 Hz , GB = 0.5) are given in the inset.



Fig. 2 (a) Pulse sequence of a ^{19}F -detected z-filtered 2D ^1H , ^{19}F HETCOR. In a non z-filtered experiment, the part within the dashed rectangle is not included. For explanation of symbols used for pulses see Experimental. The NMR parameters used are given in Table S1.† The delays used were as follows: $\Delta_1 = p44$; $\Delta_2 =$ one half of the J_{HF} evolution; $t_1(0)$, the initial t_1 evolution delay time = $0.5 \times \text{in}0$, where $\text{in}0$ is the t_1 increment. The gradient strengths were as follows: $G_0 = 3\%$; $G_1 = 17\%$; $G_2 = 31\%$; $G_3 = 24\%$; $G_4 = 10.0\%$. The following phase cycling was used: $\varphi_1 = x, -x$; $\varphi_2 = 4x, 4(-x)$; $\varphi_3 = 2y, 2(-y)$; $\Psi = x, 2(-x), x$. States-TPPI protocol was used for sign discrimination in F_1 with the phase φ_1 incremented by 90° . Purging of ^{19}F magnetisation at the beginning of the pulse sequence by a composite 90° ^{19}F pulse and pulsed field gradients (PFGs) minimises the cancellation artefacts. (b) An overlay of the ^{19}F -detected 2D ^1H , ^{19}F HETCOR spectra with (blue/turquoise) and without the z-filter (red/magenta). For clarity, the spectrum acquired without a z-filter was offset horizontally to the right. Insets show 1D F_1 traces taken at positions indicated by arrows. 1D ^1H and ^{19}F spectra are shown along the left and top, respectively.





Fig. 3 (a) Pulse sequence of a 2D ^1H , ^{19}F TOCSY–HETCOR. For explanation of symbols used for pulses see Experimental. The NMR parameters used are given in Table S1.† The delays were as follows: $\Delta_1 = p44$; $\Delta_2 =$ one half of the J_{HF} evolution; $t_1(0)$ is the initial t_1 evolution delay time = $0.5 \times \text{in}0$, where $\text{in}0$ is the t_1 increment. The gradient strengths were as follows: $G_0 = 5\%$; $G_1 = 17\%$; $G_2 = 31\%$; $G_3 = 24\%$. The following phase cycling was used: $\varphi_1 = x, -x$; $\varphi_2 = 4x, 4(-x)$; $\varphi_3 = 2y, 2(-y)$; $\Psi = x, 2(-x), x, -x, 2x, -x$. States-TPPI protocol was used for sign discrimination in F_1 with the phase φ_1 incremented by 90° . Purging of ^{19}F magnetisation after the z-filter by a composite 90° ^{19}F pulse followed by the G_2 PFG minimises the cancellation artefacts. (b) An overlay of the ^{19}F -detected 2D ^1H , ^{19}F TOCSY–HETCOR spectrum (blue/turquoise) and a z-filtered VT ^{19}F -detected 2D ^1H , ^{19}F HETCOR spectrum (red/magenta, horizontally offset to the right) of **1** acquired with the pulse sequence shown in (a) and Fig. 2a, respectively. Vertical traces of the two spectra as indicated by arrows are shown in the inset. Exclusive/stronger TOCSY cross peaks are labelled in blue. 1D ^1H and ^{19}F spectra are shown along the left and the top, respectively.



Fig. 4 (a) Pulse sequence of a 2D ^{19}F , ^1H CP-DIPSI3–DIPSI2. For explanation of symbols used for pulses see Experimental. The NMR parameters used are given in Table S1.† The dashed line indicates signal acquisition before an optional ^1H – ^1H spin-lock. For description of pulses see Experimental. The delays were as follows: $\delta_1 = 20 \mu\text{s}$; $\delta_2 = \delta_1 + (2/\pi) \times p3$; $\delta_3 = p2$; $t_1(0)$ is the initial t_1 evolution delay time = $0.5 \times \text{in}0$, where $\text{in}0$ is the t_1 increment. The gradient strengths were as follows: $G_0 = 5\%$; $G_1 = 17\%$; $G_2 = 31\%$; $G_3 = 66\%$. The following phase cycling was used: $\varphi_1 = y, -y$; $\varphi_2 = 4x, 4(-x)$; $\varphi_3 = 2y, 2(-y)$; $\Psi = x, 2(-x), x$. The states-TPPI protocol was used for sign discrimination in F_1 with the phase φ_1 incremented by 90° . Purging of ^{19}F magnetisation at the beginning of the pulse sequence by a composite 90° ^{19}F pulse and PFGs minimises the cancellation artefacts. (b) An overlay of two 2D ^{19}F , ^1H CP-DIPSI3–DIPSI2 spectra acquired with 20 ms $^{19}\text{F} \rightarrow ^1\text{H}$ cross-polarisation (CP) only (red) and an additional 50 ms $^1\text{H} \rightarrow ^1\text{H}$ spin-lock (blue) using the pulse sequence shown in (a). The red spectrum was offset vertically to facilitate visualisation of the cross peaks. The two insets show overlaid 1D F2 traces through ^{19}F resonances of α - and β -D-glucose from both spectra. Twice as many scans were acquired for the blue spectrum as for the red spectrum. 1D ^{19}F and ^1H projections of the blue spectrum are shown along the left and top, respectively.

Results and discussion

1D ^1H and ^{19}F spectra of 3-fluoro-3-deoxy-D-glucose, **1**

A 400 MHz 1D ^1H spectrum of **1** with the suppression of the HOD signal shows considerable overlap of ^1H resonances (Fig. 1a). A ^1H -coupled 1D ^{19}F spectrum **1** (Fig. 1b) contains two ^{19}F signals belonging to α - and β -anomeric forms of **1**. The insets highlight numerous ^1H – ^{19}F coupling constants of **1**. NMR parameters of **1** obtained using the developed experiments, including those involving ^{13}C , are presented in Table S2.†

^{19}F -centred NMR experiments – novelty and hardware requirements

Although a number of NMR experiments exist that correlate ^{19}F chemical shifts with those of other nuclei,³¹ the majority of

existing techniques yield magnitude mode spectra.^{32,33} Acquisition of pure-phase absorption signals in a phase-sensitive manner is much preferred, as it provides higher sensitivity and allows for accurate determination of coupling constants, including identification of the active coupling constants. Some existing phase-sensitive experiments yield complicated cross peak structures that lower their sensitivity.^{34,35}

The optimal performance of experiments constituting the ^{19}F -centred NMR approach across a range of ^{19}F frequencies, is ensured by the use of adiabatic inversion pulses.^{36,37} The experiments provide pure phase multiplets with simple structure afforded by ^1H or ^{19}F decoupling and were designed to minimise the effect of passive spins; they do not use refocusing intervals, which maximises their sensitivity. NMR hardware capable of pulsing simultaneously on ^1H and ^{19}F frequencies is



Using ^{19}F as the directly detected nucleus, the 2D ^1H , ^{19}F HMBC has the highest sensitivity, but yields mixed-phased multiplets. 2D ^1H , ^{19}F hetero-COSY can be implemented with either nucleus being sampled in the directly detected (F_2) dimension. Nevertheless, sampling ^{19}F in the F_2 dimension has a distinct advantage of acquiring spectra with the high digital resolution required for the identification of active and passive J_{HF} coupling constants and potentially also for their measurements. A disadvantage of COSY type spectra is the antiphase nature of their cross peaks (particularly in F_1) and their large footprint.

Choosing to obtain the ^1H - ^{19}F correlations using a phase-sensitive ^{19}F -detected 2D ^1H , ^{19}F HETCOR experiment (Fig. 2a) retains the advantages of ^{19}F detection. Its uniform performance across a large ^{19}F chemical shift range is guaranteed by the use of broadband inversion CHIRP pulses³⁸ arranged in a double inversion adiabatic sweep (Fig. S1†), a feature applied in several experiments presented here to eliminate phase evolution of the transverse magnetisation during pulses.^{39–41} This allows the use of such pulses not only for spin inversion but also refocusing.

The structure of cross peaks in HETCOR spectra is simplified by the application of a 180° ^{19}F pulse in the middle of the t_1 interval, reducing the probability of signal overlap in spectra of complex mixtures. A drawback of this experiment is the evolution of ^1H - ^1H couplings during the defocusing interval $2\Delta_2$, which competes with the evolution of ^1H - ^{19}F couplings, decreasing its sensitivity. This decrease can often be tolerated because of the 100% natural abundance of both nuclei.

Due to diverse sizes of J_{HF} coupling constants, no attempt was made to refocus ^{19}F magnetisation prior to detection and ^1H decoupling was not applied during t_2 . Preserving the antiphase character of cross peaks is important, as it allows the identification of active couplings. Nevertheless, if a ^1H -coupled ^{19}F 1D spectrum is overlap free, it is advised to read the coupling constants from this spectrum, where accurate values are readily obtained (see Fig. 1b).

In a basic HETCOR experiment,³² the evolution of ^1H - ^1H couplings during the ^1H - ^{19}F defocusing interval, $2\Delta_2$, leads to the appearance of mixed phase proton multiplets in F_1 – a feature that is masked by the magnitude mode presentation of spectra. This issue was resolved in the proposed phase-sensitive experiment by inserting a z-filter⁴² after the t_1 period, which separates the evolution of ^1H - ^1H couplings during the t_1 and the $2\Delta_2$ defocusing interval. Providing the $t_{1\text{max}}$ is kept short (<30 ms), the cross peaks appear as singlets in F_1 . The described features of the experiment are illustrated on a 2D ^1H , ^{19}F HETCOR spectrum of **1** (Fig. 2b), where correlations with many ^{19}F coupled protons are observed.

Protons not coupled by a sizable (>1.0 Hz) coupling constant to a ^{19}F , but which are part of a spin system containing at least one ^1H coupled to a ^{19}F , are detected in a 2D ^1H , ^{19}F TOCSY-HETCOR experiment (Fig. 3a). Here, the ^1H chemical shifts are labelled before their magnetisation is spread through the network of J_{HH} coupled spins by a DIPSI-2 spin-lock.⁴³ Part of the magnetisation that has reached the ^{19}F -coupled protons is then transferred to ^{19}F for detection in a subsequent HETCOR step. An overlay of the 2D ^1H , ^{19}F HETCOR and 2D ^1H , ^{19}F TOCSY-

HETCOR spectra (Fig. 3b) revealed several protons with a J_{HF} close to zero, which were not detected by the HETCOR experiment. Other protons of both anomeric forms of **1** coupled with small coupling constants to ^{19}F showed increased intensities.

In addition to J_{HF} coupling constants, J_{HH} coupling constants provide important structural information that for complex mixtures is inaccessible by standard 2D experiments, but can be retrieved when some form of ^{19}F editing is used. In principle, ^1H - ^1H couplings modulate cross peaks in the F_1 dimension of the 2D (TOCSY-)HETCOR experiments discussed above but in practice, typical t_1 acquisition times used to record such spectra are too short to resolve them. The ^1H - ^1H couplings are more likely to be resolved in the F_2 dimension of ^1H -detected experiments considering a non-refocused 2D ^1H -detected ^{19}F , ^1H HETCOR, this experiment shows F_2 multiplets with J_{HF} and J_{HH} coupling constants as anti-phase and inphase splitting, respectively, complicating access to J_{HH} coupling constants (data not shown).

The J_{HH} coupling constants can be measured more effectively from inphase proton multiplets acquired in the presence of ^{19}F decoupling. Developed for simple mixtures of fluorinated compounds, this reasoning has led to the design of FESTA experiments.^{27–29} These 1D selective experiments require that both ^{19}F and ^1H multiplets are amenable to selective inversion, which is rarely the case for complex mixtures; experiments that do not rely on selective manipulations of spins are more robust.

A suitable alternative involving the use of $^{19}\text{F} \rightarrow ^1\text{H}$ cross-polarisation (CP) that produces inphase ^1H multiplets was already proposed in the form of a 3D CP ^{19}F , ^1H heteronuclear TOCSY experiment.⁴⁴ We did not find it necessary to label the ^1H chemical shifts after the initial $^{19}\text{F} \rightarrow ^1\text{H}$ magnetisation transfer and present here a 2D version of this experiment in the form of a 2D ^{19}F , ^1H CP-DIPSI3-DIPSI2 (Fig. 4a). Here, the signal acquisition can start immediately after the z-filter⁴² that follows the CP step. Note that signals of protons not coupled to ^{19}F can appear in the spectrum even at this point due to the ^1H - ^1H TOCSY transfer that takes place simultaneously with the heteronuclear CP step.

This pulse sequence can be extended by a dedicated ^1H - ^1H DIPSI-2 spin-lock propagating the magnetisation transfer to more remote parts of the spin system. Application of two z-filters and ^{19}F decoupling ensures that pure inphase ^1H multiplets are eventually acquired. DIPSI-3,⁴⁵ using $40 \mu\text{s}$ $^{19}\text{F}/^1\text{H}$ pulses, was applied for the CP step covering a ± 4 kHz frequency range with >75% efficiency. A slight improvement was achieved with the FLOPSY-16 mixing scheme⁴⁶ covering ± 4.7 kHz, *i.e.* 25 ppm of ^{19}F resonances on a 400 MHz NMR spectrometer with >65% efficiency relative to the on-resonance signal (Fig. S2†). Further improvements, not explored here, can be achieved by using broadband pulses during the CP step.⁴⁷

An overlay of two 400 MHz 2D ^{19}F , ^1H CP-DIPSI3-DIPSI2 spectra acquired with a 20 ms $^{19}\text{F} \rightarrow ^1\text{H}$ cross-polarisation (red) and an additional 50 ms $^1\text{H} \rightarrow ^1\text{H}$ spin-lock (blue) using the pulse sequence of Fig. 4a is presented in Fig. 4b. Both spectra are suitable for the determination of the J_{HH} coupling constants. The former spectrum contains pure in phase multiplets of protons H2, 3 and 4 of **1**, while the latter spectrum also shows all their



other protons. Note the dominance of the H3 signals in the red spectra caused by an effective CP *via* large J_{H3F3} (~ 50 Hz).

Fluorine–carbon correlation

Structure determination of sparsely protonated fluorinated molecules, such as heavily substituted aromatic rings, based only on ^1H and ^{19}F chemical shifts and coupling constants could be problematic. Thanks to the far-reaching ^{19}F – ^{13}C couplings ($^nJ_{\text{FC}}$, $n = 1$ – 5), many ^{19}F -coupled ^{13}C atoms can be identified by 2D ^{19}F , ^{13}C correlated experiments such as HMBC or HSQC, making structure determination of such molecules possible. A 2D ^{19}F , ^{13}C HSQC experiment^{33,41} was not considered in this study mainly because of a larger complexity of the double INEPT transfer. For small molecules, the slower relaxation of single-quantum (HSQC) relative to multiple-quantum (HMBC) coherences does not make a substantial difference to their sensitivity and for mono-fluorinated compounds F_1 singlets are produced by both experiments.

As the $^1J_{\text{FC}}$ coupling constants are large (~ 150 – 250 Hz), while the $^{n>1}J_{\text{FC}}$ typically range from 0 to 50 Hz,⁴⁸ the one-bond (Fig. S3†) and long-range correlation (Fig. 5a) experiments are best performed separately. A single long-range optimised experiment can also yield one-bond correlations if multiple rotations of the ^{19}F magnetisation vectors during the evolution interval fall outside of even multiples of $0.5/{}^1J_{\text{FC}}$. This approach can only be used when values of $^1J_{\text{FC}}$ coupling constants are known, and if dealing with mixtures, their spread is narrow. Values of $^1J_{\text{FC}}$ coupling constants required for such optimisation can be obtained from 1D ^1H -decoupled ^{19}F spectra acquired with a sufficient S/N ratio. Alternatively, accordion optimisation⁴⁹ can be used to obtain simultaneously both types of correlations. Both experiments perform best when ^1H decoupling is applied during most of the pulse sequence. Such decoupling removes splitting of cross peaks by ^1H – ^{13}C couplings in F_1 and by J_{HF} in F_2 . Resulting F_1 singlets and F_2 anti-phase doublets split by ^{19}F – ^{13}C interactions (Fig. 5b) allow accurate measurement of J_{FC} coupling constants that provide valuable structural information.

A comparison of ^{19}F chemical shifts of ^{13}C isotopomers obtained from 2D ^{19}F , ^{13}C HMBC spectra with the ^{19}F signal in a 1D ^1H -decoupled ^{19}F spectrum yields ^{13}C induced ^{19}F isotopic shifts (see a large isotopic shift of C3 resonances in Fig. 5b). In aliphatic systems these decrease with the number of bonds separating the two atoms and are generally measurable to up to four bonds separating ^{13}C and ^{19}F . A careful alignment of the one-bond correlation trace from the pure phase HMBC spectrum and the satellites from the 1D ^{19}F spectrum is required to obtain accurate values of these isotopic shifts.

Proton–carbon–fluorine correlation

^1H – ^1H and ^1H – ^{13}C interactions are the cornerstone of NMR structure determination of small molecules. For fluorinated compounds, the existence of ^1H – ^{19}F and ^{19}F – ^{13}C couplings makes this process even more robust. However, for complex mixtures, mapping of these interactions separately, can compromise identification of the nuclei belonging to individual molecules.

This ambiguity can be avoided by correlating all three spin types in a dedicated HCF experiment. There are numerous possibilities for how such an experiment can be designed. Inspired by the 3D HNCA, a pulse sequence for assigning protein backbone resonances,⁵⁰ a ^1H -detected 3D triple-resonance ^1H , ^{13}C , ^{19}F experiment has been proposed previously.^{51,52} This out-and-back 3D experiment contains J_{FC} defocusing and refocusing intervals, samples ^{19}F and ^{13}C chemical shifts indirectly and applies simultaneous ^{13}C and ^{19}F decoupling during the direct detection of ^1H . We prefer to use a unidirectional polarisation transfer pathway and direct detection of ^{19}F ; both of these features are well suited for molecules with a large spread of coupling constants, as is typical for ^{19}F – ^{13}C interactions. The pulse sequence of such an experiment starts with a one-bond ^1H – ^{13}C correlation step followed by a ^{13}C , ^{19}F long-range transfer step. It incorporates a reduced dimensionality approach^{53–55} and samples ^{13}C chemical shifts simultaneously with the indirect labelling of ^1H resonances. The resulting 2D experiment is referred to as (3, 2)D $\text{H}^1\text{C}^n\text{F}$, where the superscripts indicate the type of ^{13}C and ^{19}F interactions (1-one-bond, n -long-range) mediating the polarisation transfer (Fig. 6a).

In the (3, 2)D $\text{H}^1\text{C}^n\text{F}$ experiment, the ^1H chemical shifts are recorded first, while suppressing the evolution of ^1H – ^1H and ^1H – ^{19}F couplings by a BIRD^{r,x} pulse^{56,57} and a 180° ^{19}F pulse applied in the middle of the t_1 period, respectively.

The magnetisation is then transferred in an INEPT step to ^{13}C *via* one-bond ^1H – ^{13}C couplings, where it is refocused before starting ^1H decoupling. During the subsequent evolution interval, the ^{19}F – ^{13}C anti-phase magnetisation is developed while the central 180° ^{13}C and ^{19}F pulses move simultaneously with the t_1 incrementation. This causes modulation of ^1H chemical shifts by ^{13}C offsets, $\Omega_{13\text{C}} (= \delta(^{13}\text{C}) - ^{13}\text{C}$ r.f. carrier frequency) of their directly bonded ^{13}C , splitting the signals into doublets centred at the ^1H chemical shift. The size of ^{13}C doublets can be scaled down relative to the t_1 evolution (κ factor), keeping the F_1 spectral width small and without any limitations for setting the length of the constant-time ^{19}F – ^{13}C coupling evolution interval, $2\Delta_5$.

The signal is finally transferred to ^{19}F , where it is detected during t_2 under ^1H decoupling as a pure phase doublet in anti-phase with regard to J_{FC} (Fig. 6b).

Interleaved acquisition of two spectra, differing by 90° in the phase of the last 90° ^{13}C pulse of the pulse sequence, generates inphase and anti-phase F_1 doublets, respectively, allowing spectra to be simplified by spectral editing^{58,59} as illustrated in Fig. S5.† A pulsed field gradient assisted echo-antiecho protocol is used to obtain pure phase signals in F_1 .

Overall, the reduced dimensionality experiment retains the full information content of 3D spectra with substantially increased digital resolution. Due to the use of a single $^nJ_{\text{FC}}$ evolution interval, sensitivity is also improved relative to the original 3D HCF experiment.⁵¹ Detecting ^{19}F under ^1H decoupling during t_2 further increases sensitivity of this experiment, while providing values of J_{FC} coupling constants. The (3, 2)D $\text{H}^1\text{C}^n\text{F}$ experiment thus complements the 2D ^{19}F , ^{13}C HMBC technique discussed above and for protonated carbons



correlates unambiguously three atom types, HCF, instead of aiming to achieve the same through a combined interpretation of 2D ^1H , ^{13}C HSQC and 2D ^{19}F , ^{13}C HMBC spectra, which for complex mixtures, is problematic.

Structure determination process in ^{19}F -centred NMR

This process is briefly summarised with the help of a graphical representation in Fig. 7, using the β -anomeric form of **1** as an example. The ^{19}F - ^1H correlations experiments, 2D ^1H , ^{19}F HETCOR and 2D ^1H , ^{19}F TOCSY-HETCOR spectra, together with 1D ^1H -coupled/decoupled ^{19}F spectra provided the parameters summarised in Fig. 7a, while 2D ^{19}F , ^1H CP-DIPS13-DIPS12 experiments extended the identified spin system to protons not directly coupled to fluorine (Fig. 7b).

These experiments thus provide ^{19}F and ^1H chemical shift correlations together with $^nJ_{\text{HF}}$ ($n = 2-4$)⁶⁰ and $^nJ_{\text{HH}}$ ($n = 2-3$) coupling constants, enabling the start of a structure determination process.

Experiments involving ^{19}F - ^{13}C correlations are very informative. Central to these is the 2D ^{19}F , ^{13}C HMBC experiment, which provides long-range ^{19}F - ^{13}C correlations and $^nJ_{\text{FC}}$ coupling constants and in conjunction with a 1D ^1H decoupled ^{19}F spectrum also the ^{13}C induced ^{19}F isotopic chemical shifts (Fig. 7c). The subsequent (3, 2)D $\text{H}^1\text{C}^n\text{F}$ experiment provides correlations of HC pairs, in which the carbon is coupled to ^{19}F , and if present, a distinction between non-protonated and protonated carbons (Fig. 7d).

Occasionally, a 2D ^1H , ^{19}F HOESY experiment^{31,61} can be used to identify protons not accessible by exploring J coupled

networks of spins. In general, at this point, the chemical shift assignment and of numerous ^1H , ^{13}C and ^{19}F resonances, values of J_{HF} , J_{HH} and J_{FC} coupling constants and ^{13}C induced ^{19}F isotopic shifts are known and the structure determination of fluorine containing moieties can be completed.

For larger molecules, which contain spin systems isolated from those containing ^{19}F , the ^{19}F -centered approach provides a starting point by identifying protons and carbons that appear in both the ^{19}F -centered and the standard ^1H - ^1H and ^1H - ^{13}C 2D chemical shift correlated spectra. These resonances can then be used to extend the structures and connect the fluorinated and non-fluorinated parts of molecules, *e.g.* via ^1H - ^1H NOESY experiments or ^1H - ^{13}C HMBC experiments, which can bridge such spin-systems. This approach is particularly beneficial for analyses of mixtures, where the identity of cross peaks belonging to the non-fluorinated parts of the molecule could be difficult to establish.

Although the discussed NMR experiments were developed for mono-fluorinated compounds, they can also be applied to compounds bearing more than one fluorine atom. Nevertheless, the presence of multiple ^{19}F atoms should be taken into account when setting up some of the experiments, as the existence of passive ^1H - ^{19}F (or ^{19}F - ^{13}C) couplings need to be reflected in the parameters used as outlined in Table S3.†

It should be emphasised, that the ^{19}F -centered approach takes full advantage of the high sensitivity of ^{19}F to its environment and minute differences in the ^{19}F chemical shift of the order of few Hz are sufficient to obtain the kind of information illustrated here on a very simple mixture provided by **1**. Application of the ^{19}F -centered approach to a very complex mixture of chloramination by-products is presented elsewhere.³⁰

Conclusions

The described methodology is based on a concerted use of several NMR experiments, nevertheless, these can also be used in their own right. Collectively, these experiments represent the most effective NMR approach for the structure determination of mono-fluorinated compounds, particularly those contained in mixtures.

The ^{19}F -centred approach developed here is applicable at any point during the lifetime of fluorinated compounds, *e.g.* in analysing reaction mixtures during their production, performing mechanistic studies to understand reaction mechanisms and to optimise chemical reactions, investigating their stability, pharmacokinetics, biodegradation and biotransformation and ultimately to follow their fate in the environment.⁶²

Data availability

The spectra obtained in this study are available here: <https://doi.org/10.7488/ds/3422>.

Conflicts of interest

There are no conflicts to declare.



Fig. 7 NMR parameters obtained by ^{19}F -centred NMR for the β -anomeric form of **1**. Chemical shifts, coupling constants and ^{13}C isotopic shifts are given in ppm, Hz and ppb, respectively. (a) $\delta_{1\text{H}}/{}^nJ_{\text{HF}}$; red and blue colours indicate correlations obtained from 2D ^1H , ^{19}F HETCOR and 2D ^1H , ^{19}F TOCSY-HETCOR spectra, respectively; (b) $J_{\text{H}(i)\text{H}(i+1)}$; red and blue colour indicates correlation from 2D ^{19}F , ^1H CP-DIPS13-DIPS12 without and with the DIPS1-2 extension; (c) ${}^nJ_{\text{FC}}/\Delta^{19}\text{F}(^{13}\text{C})$; (d) $\delta_{1\text{H}}/\delta_{13\text{C}}$; n.d. – not detected.



Acknowledgements

NGAB would like to acknowledge NERC soil security programme NE/N020227/1 for funding. AJRS was supported by Scottish Water and EPSRC grant EP/N509644/1 and RY by the NERC Centre for Doctoral Training, E4 (NE/S007407/1). Instrument support was in part provided by the EPSRC grant EP/R030065/1. The authors would like to thank Juraj Bella and Dr Lorna Murray for maintenance of the NMR spectrometers.

References

- 1 K. Muller, C. Faeh and F. Diederich, *Science*, 2007, **317**, 1881–1886.
- 2 E. P. Gillis, K. J. Eastman, M. D. Hill, D. J. Donnelly and N. A. Meanwell, *J. Med. Chem.*, 2015, **58**, 8315–8359.
- 3 B. M. Johnson, Y. Z. Shu, X. L. Zhuo and N. A. Meanwell, *J. Med. Chem.*, 2020, **63**, 6315–6386.
- 4 M. Inoue, Y. Sumii and N. Shibata, *ACS Omega*, 2020, **5**, 10633–10640.
- 5 J. L. Han, A. M. Remete, L. S. Dobson, L. Kiss, K. Izawa, H. Moriwaki, V. A. Soloshonok and D. O'Hagan, *J. Fluorine Chem.*, 2020, **239**, 109639.
- 6 Y. Ogawa, E. Tokunaga, O. Kobayashi, K. Hirai and N. Shibata, *iScience*, 2020, **23**, 101467.
- 7 O. Jacobson, D. O. Kiesewetter and X. Y. Chen, *Bioconjugate Chem.*, 2015, **26**, 1–18.
- 8 P. T. Lowe and D. O'Hagan, *J. Fluorine Chem.*, 2020, **230**, 109420.
- 9 T. Liang, C. N. Neumann and T. Ritter, *Angew. Chem., Int. Ed.*, 2013, **52**, 8214–8264.
- 10 T. Furuya, A. S. Kamlet and T. Ritter, *Nature*, 2011, **473**, 470–477.
- 11 D. O'Hagan and H. Deng, *Chem. Rev.*, 2015, **115**, 634–649.
- 12 J. Fang, D. Hait, M. Head-Gordon and M. C. Y. Chang, *Angew. Chem., Int. Ed.*, 2019, **58**, 11841–11845.
- 13 T. Hayashi, G. Kehr, K. Bergander and R. Gilmour, *Angew. Chem., Int. Ed.*, 2019, **58**, 3814–3818.
- 14 A. Rentmeister, F. H. Arnold and R. Fasan, *Nat. Chem. Biol.*, 2009, **5**, 26–28.
- 15 C. D. Murphy, *Biotechnol. Lett.*, 2010, **32**, 351–359.
- 16 B. D. Key, R. D. Howell and C. S. Criddle, *Environ. Sci. Technol.*, 1997, **31**, 2445–2454.
- 17 X. J. Zhang, T. B. Lai and R. Y. C. Kong, in *Fluorous Chemistry*, ed. I. T. Horvath, 2012, vol. 308, pp. 365–404.
- 18 M. G. Boersma, T. Y. Dinarieva, W. J. Middelhoven, W. J. H. van Berkel, J. Doran, J. Vervoort and I. Rietjens, *Appl. Environ. Microbiol.*, 1998, **64**, 1256–1263.
- 19 V. S. Bondar, M. G. Boersma, E. L. Golovlev, J. Vervoort, W. J. H. Van Berkel, Z. I. Finkelstein, I. P. Solyanikova, L. A. Golovleva and I. Rietjens, *Biodegradation*, 1998, **9**, 475–486.
- 20 M. Kiel and K. H. Engesser, *Appl. Microbiol. Biotechnol.*, 2015, **99**, 7433–7464.
- 21 R. Natarajan, R. Azerad, B. Badet and E. Copin, *J. Fluorine Chem.*, 2005, **126**, 425–436.
- 22 E. Nieto-Sepulveda, A. D. Bage, L. A. Evans, T. A. Hunt, A. G. Leach, S. P. Thomas and G. C. Lloyd-Jones, *J. Am. Chem. Soc.*, 2019, **141**, 18600–18611.
- 23 R. Wei, A. M. R. Hall, R. Behrens, M. S. Pritchard, E. J. King and G. C. Lloyd-Jones, *Eur. J. Org. Chem.*, 2021, **2021**, 2331–2342.
- 24 N. G. A. Bell, L. Murray, M. C. Graham and D. Uhrin, *Chem. Commun.*, 2014, **50**, 1694–1697.
- 25 N. G. A. Bell, M. C. Graham and D. Uhrin, *Analyst*, 2016, **141**, 4614–4624.
- 26 G. A. Bell, A. A. L. Michalchuk, J. W. T. Blackburn, M. C. Graham and D. Uhrin, *Angew. Chem., Int. Ed.*, 2015, **54**, 8382–8385.
- 27 L. Castanar, P. Moutzouri, T. M. Barbosa, C. F. Tormena, R. Rittner, A. R. Phillips, S. R. Coombes, M. Nilsson and G. A. Morris, *Anal. Chem.*, 2018, **90**, 5445–5450.
- 28 T. M. Barbosa, L. Castanar, P. Moutzouri, M. Nilsson, G. A. Morris, R. Rittner and C. F. Tormena, *Anal. Chem.*, 2020, **92**, 2224–2228.
- 29 G. Dal Poggetto, J. V. Soares and C. F. Tormena, *Anal. Chem.*, 2020, **92**, 14047–14053.
- 30 A. J. R. Smith, R. York, D. Uhrin and N. G. A. Bell, *Chem. Sci.*, 2021, DOI: 10.1039/D1SC06057K.
- 31 J. Battiste and R. A. Newmark, *Prog. Nucl. Magn. Reson. Spectrosc.*, 2006, **48**, 1–23.
- 32 A. A. Marchione, R. J. Dooley and B. Conklin, *Magn. Reson. Chem.*, 2014, **52**, 183–189.
- 33 R. A. Newmark and R. J. Webb, *J. Fluorine Chem.*, 2005, **126**, 355–360.
- 34 K. A. M. Ampt, R. Aspers, P. Dvortsak, R. M. van der Werf, S. S. Wijmenga and M. Jaeger, *J. Magn. Reson.*, 2012, **215**, 27–33.
- 35 R. Aspers, K. A. M. Ampt, P. Dvortsak, M. Jaeger and S. S. Wijmenga, *J. Magn. Reson.*, 2013, **231**, 79–89.
- 36 A. A. Marchione and B. Conklin, *Appl. Magn. Reson.*, 2017, **48**, 485–499.
- 37 J. E. Power, M. Foroozandeh, P. Moutzouri, R. W. Adams, M. Nilsson, S. R. Coombes, A. R. Phillips and G. A. Morris, *Chem. Commun.*, 2016, **52**, 6892–6894.
- 38 J. M. Bohlen, I. Burghardt, M. Rey and G. Bodenhausen, *J. Magn. Reson.*, 1990, **90**, 183–191.
- 39 T. L. Hwang and A. J. Shaka, *J. Magn. Reson., Ser. A*, 1995, **112**, 275–279.
- 40 M. H. Levitt and R. Freeman, *J. Magn. Reson.*, 1981, **43**, 65–80.
- 41 B. Adams, *Magn. Reson. Chem.*, 2008, **46**, 377–380.
- 42 M. J. Thrippleton and J. Keeler, *Angew. Chem., Int. Ed.*, 2003, **42**, 3938–3941.
- 43 S. P. Rucker and A. J. Shaka, *Mol. Phys.*, 1989, **68**, 509–517.
- 44 H. Hu, P. Kulanthaivel and K. Krishnamurthy, *J. Org. Chem.*, 2007, **72**, 6259–6262.
- 45 A. J. Shaka, C. J. Lee and A. Pines, *J. Magn. Reson.*, 1988, **77**, 274–293.
- 46 M. Kadkhodaie, O. Rivas, M. Tan, A. Mohebbi and A. J. Shaka, *J. Magn. Reson.*, 1991, **91**, 437–443.
- 47 A. A. Marchione and E. L. Diaz, *J. Magn. Reson.*, 2018, **286**, 143–147.



- 48 H. Duddeck and M. R. Islam, *Tetrahedron*, 1981, **37**, 1193–1197.
- 49 G. Bodenhausen and R. R. Ernst, *J. Am. Chem. Soc.*, 1982, **104**, 1304–1309.
- 50 L. E. Kay, M. Ikura, R. Tschudin and A. Bax, *J. Magn. Reson.*, 1990, **89**, 496–514.
- 51 L. Li and P. L. Rinaldi, *Macromolecules*, 1996, **29**, 4808–4810.
- 52 L. L. Li, B. Zhang, F. Wyzgoski, X. H. Li, E. F. McCord and P. L. Rinaldi, *ACS Macro Lett.*, 2013, **2**, 141–145.
- 53 G. Bodenhausen and R. R. Ernst, *J. Magn. Reson.*, 1981, **45**, 367–373.
- 54 Y. Shen, H. S. Atreya, G. H. Liu and T. Szyperski, *J. Am. Chem. Soc.*, 2005, **127**, 9085–9099.
- 55 W. Kozminski and I. Zhukov, *J. Biomol. NMR*, 2003, **26**, 157–166.
- 56 J. R. Garbow, D. P. Weitekamp and A. Pines, *Chem. Phys. Lett.*, 1982, **93**, 504–509.
- 57 D. Uhrin, T. Liptaj and K. E. Kover, *J. Magn. Reson., Ser. A*, 1993, **101**, 41–46.
- 58 N. Brodaczewska, Z. Kostalova and D. Uhrin, *J. Biomol. NMR*, 2018, **70**, 115–122.
- 59 J. Sakas and N. G. A. Bell, *Faraday Discuss.*, 2019, **218**, 191–201.
- 60 Hans Reich's Collection, ¹⁹F NMR Spectroscopy, https://organicchemistrydata.org/hansreich/resources/nmr/?index=nmr_index%2F19F_coupling#f-data08.
- 61 P. L. Rinaldi, *J. Am. Chem. Soc.*, 1983, **105**, 5167–5168.
- 62 M. T. Anaraki, D. H. Lysak, K. Downey, F. V. C. Kock, X. You, R. D. Majumdar, A. Barison, L. M. Liao, A. G. Ferreira, V. Decker, B. Goerling, M. Spraul, M. Godejohann, P. A. Helm, S. Kleywegt, K. Jobst, R. Soong, M. J. Simpson and A. J. Simpson, *Prog. Nucl. Magn. Reson. Spectrosc.*, 2021, **126–127**, 121–180.

