

Nanoscale Advances

Volume 6
Number 1
7 January 2024
Pages 1–304

rsc.li/nanoscale-advances



ISSN 2516-0230

Cite this: *Nanoscale Adv.*, 2024, 6, 79

Chiraltube, rolling 2D materials into chiral nanotubes†

José M. de Albornoz-Caratozzolo  and Felipe Cervantes-Sodi *

Carbon nanotubes (NTs) are graphene sheets rolled into a 1D material, with a specific chirality that defines its structure and properties. Graphene has triggered the development of thousands of 2D materials, which in principle could also be rolled into 1D NTs. However, most of these NTs have not been proposed due to difficulties in the generation of atomic coordinates for chiral NTs from 2D materials with a non-hexagonal lattice or multi-layered materials. In this paper we present Chiraltube, an open-source Python code that allows the quick generation of a complete NT with any chirality from the unit cell of its original 2D material. We explain the inner workings of the code as well as the theoretical background on which it is built, generalizing concepts from the construction of chiral and achiral carbon NTs to work on any other 2D material. We show various examples of the resulting chiral NT structures built from phosphorene, MoS₂ and Ti₃C₂, and present some analysis on the interatomic distortion in the outermost layers of these NTs, as well as the results of *ab initio* electronic structure calculations on a set of phosphorene NTs generated by the program, showing the immediate practicality and usefulness of the program. We also explore some limitations and details of the tool as well as further work to be done.

Received 5th May 2023
Accepted 30th September 2023

DOI: 10.1039/d3na00301a

rsc.li/nanoscale-advances

1 Introduction

During the last 30 years an immense effort has been devoted to the study of first nanotubes (NTs) and then two dimensional (2D) materials. Their archetypes are carbon based nanostructures: carbon nanotubes (CNTs)^{1–3} and graphene,^{4,5} respectively. CNTs are commonly described as “rolled” sheets of graphene and are defined by their chiral angle (θ) and chiral vector (**C**)⁶ (Fig. 1).

Each CNT has its chirality^{6–8} defined by the angle at which a graphene sheet is wrapped to form a specific nanotube (NT). It is a geometrical classification of CNTs and a topological feature that governs all CNTs' properties, like electronic and optical ones.^{6,9,10} A minimal difference between two CNTs' chiralities may differentiate one as metallic and the other as a semiconductor.

Chirality is related to CNTs' symmetry; chiral and achiral NTs are non-symmorphic and symmorphic respectively. Zigzag and armchair CNTs are both achiral, with their structure being identical to their mirror.¹⁰ In contrast, chiral CNTs mirror's images are not superposed with the original ones.¹⁰ Chiral tubes

are called this way from the chemical nomenclature, axially chiral, presenting a spiral symmetry.

CNTs have been studied in depth, with a huge amount of theoretical^{6,7,10–14} and experimental work^{15–22} devoted to CNTs wrapped in armchair, zigzag or any other chirality, which in turn have completely different electronic, optoelectronic, mechanical and catalytic properties. This diversity hinders the experimental endeavour of obtaining a specific chirality CNT for its applications, but it has also opened an endless path for potential applications.

Generating a supercell of chiral CNTs, with the atomic coordinates of each atom, is easy, and in fact, there are many tools and freely available resources to generate sets of coordinates of virtually any chiral/achiral CNT,^{8,23–25} which can then be the input of further studies, like density functional theory (DFT) calculations to obtain the geometrically relaxed structure, electronic properties, *etc.* While these tools could be adapted for other materials, they are made for the specific hexagonal shape in graphene's unit cell.

As with carbon and graphene, NTs of other elements or compounds have also been studied, especially their achiral zigzag and armchair structures, in part because their coordinates are easier to obtain. Some examples are silicene NTs (SiNTs),^{26,27} phosphorene NTs (PNTs),^{28–30} boron nitride NTs (BNNTs),³¹ GaAs NTs (GaAsNTs)^{32,33} transition metal dichalcogenide (TMD) NTs (TMDNTs) (MoS₂,³⁴ WS₂,³⁵ and MoSSe³⁶), MXene NTs (MXene-NTs),³⁷ and others.^{38–40}

However, for most of these materials, their chiral counterparts present higher complexity in the construction of their

Universidad Iberoamericana, Physics and Mathematics Department, Prol. Paseo de la Reforma 880, Lomas de Santa Fe, Ciudad de México, Mexico. E-mail: felipe.cervantes@ibero.mx; Tel: +52 55 59504275

† Electronic supplementary information (ESI) available: A document with examples of input and output files and a zoomed-in figure into a small region of manuscript Fig. 8. A zip file with the code in Python and the special-xyz input files of the 4 materials used in the manuscript. The code is available at <https://github.com/Chema-dac/chiraltube>. See DOI: <https://doi.org/10.1039/d3na00301a>



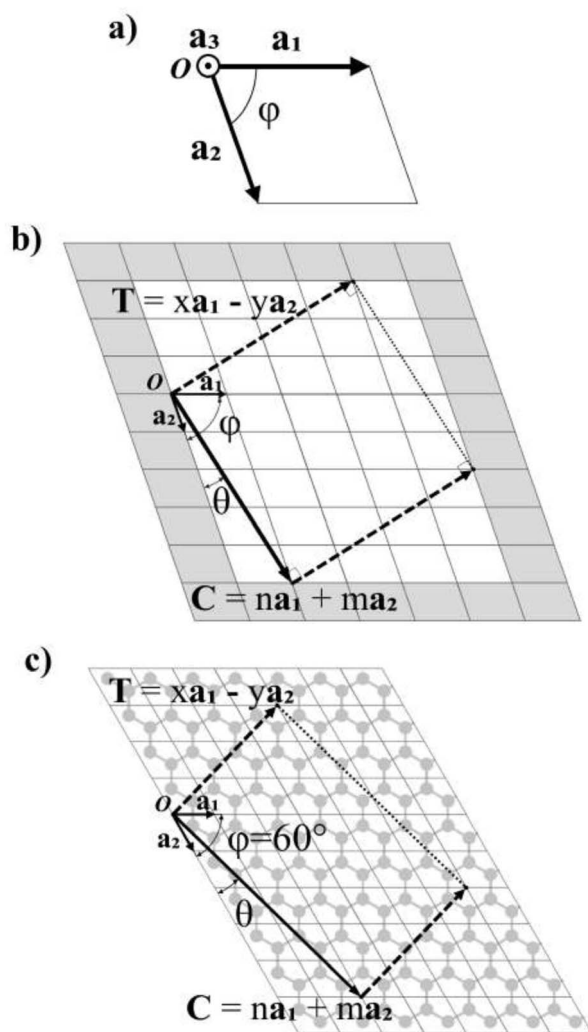


Fig. 1 (a) Unit cell of an arbitrary 2D material. \mathbf{a}_1 , \mathbf{a}_2 and \mathbf{a}_3 are the vectors of the unit cell. φ is the angle formed between \mathbf{a}_1 and \mathbf{a}_2 . (b) Lattice of a general 2D material with the nanoribbon delimited by the chiral vector \mathbf{C} and the translational vector \mathbf{T} . (c) Graphene lattice with a nanoribbon defined by \mathbf{C} and \mathbf{T} .

supercell and the generation of their atomic coordinates. These supercells are usually bigger in size, constituted by hundreds of atoms, making their simulations and theoretical studies more difficult.

Although there are a few examples of theoretical studies in the literature dealing with chiral NTs (*i.e.* a molecular dynamics (MD) study of PNTs self-assembled from phosphorene ribbons,⁴¹ a study on spin orbit effects on chiral MoS₂ NTs⁴² or electronic structure calculations of NTs made out of two dimensional (2D) materials with a unit cell similar to that of graphene or a simple monolayer as BNNTs^{43,44} and TMD-NTs⁴⁵), chiral NTs are still an almost virgin field to be explored. Examining the characteristics of chiral nanotubes will reveal changes in their properties attributable to their unique chirality, as it occurs with CNTs. Such distinctions hold promise for potential applications across various scientific domains, where nanotubes and 2D materials have already made significant contributions.

With the rise of 2D materials,⁴⁶ there is an uncountable number of NT possibilities to be addressed.⁴⁷ The broadening of the spectrum to include not only the achiral, but also the chiral NTs, would be possible with a tool that could generate the atomic coordinates of the NTs with the only input being the unit cell of the 2D material in question (cell parameters and atomic positions can be found elsewhere^{47–49}).

In this paper we present Chiraltube,⁵⁰ a concise code written in Python that allows for the generation of not only CNTs, but also NTs from any 2D material, with any unit cell and in any chirality desired.

In the body of this paper, first, we generalize the notion of chirality and related concepts (*i.e.*, chiral vector, wrapping angle, nanoribbons (NRs), translational vectors, symmetry vector, rotation angle, and NT reciprocal lattice unit vectors), that have been used for CNTs since the 90s,¹⁰ to any type of 2D material rolled into a chiral NT. This is integrated into an open-source code capable of dealing with materials with an asymmetrical unit cell and a width of either single layer or multiple layers, such as phosphorene, TMDs and MXenes. Second, we present an explanation of the general methodology of the code in Python, which generates a NR from a unit cell of a 2D material, and later wraps it into a NT of the chosen chirality. Later, in the results and discussion, some families of NTs are presented, analysing geometrical aspects related to the generated coordinates. As a proof of concept, the atomic coordinates of chiral phosphorene nanotubes were used as input in *ab initio* calculations, specifically geometry optimization ones, proving the capability of the coordinate generator code. Finally, the discussion includes some scenarios of applicability using examples in the literature, followed by the conclusions.

2 Methodology

In Fig. 1a, the vectors of an arbitrary unit cell are defined as \mathbf{a}_1 , \mathbf{a}_2 , and \mathbf{a}_3 , and it's important to note the third vector \mathbf{a}_3 , pointing out of the page, as it will be important for the folding of multilayered materials such as MoS₂. The angle formed between \mathbf{a}_1 and \mathbf{a}_2 is defined as φ and goes from 0° to 90°. This angle has a value $\varphi = 60^\circ$ for graphene (shown in Fig. 1c). The chiral vector \mathbf{C} is defined as $\mathbf{C} = n\mathbf{a}_1 + m\mathbf{a}_2$, where n and m are the chiral indices, as for CNTs.^{3,6,7,11,21} In fact, several of the cells, ribbons and nanotube parameters presented here are a generalization of those for CNTs in Table 3.3 of ref. 10.

Fig. 1b shows the crystalline lattice of a general 2D material with arbitrary φ , only the outline of the unit cells is shown, and the unit cell could contain any number of atoms in any configuration. The chiral vector \mathbf{C} defines the angle θ , which will be measured from the \mathbf{a}_2 axis in the counter-clockwise direction (Fig. 1b), and goes from 0 to φ . The chiral vector defines the rolling direction of the 2D material to form a NT. Thus, the nanoribbon (NR) which will be rolled into a NT is formed between the chiral vector and the vectors perpendicular to it, called the translational vector \mathbf{T} (Fig. 1b and c). The translational vector is defined as $\mathbf{T} = x\mathbf{a}_1 - y\mathbf{a}_2$, where x and y are positive integers, called from now on the translational indices.



The minus sign is there in order to keep y positive. \mathbf{T} also defines the NT axis.

2.1 *Ab initio* calculations

The electronic structure calculations were performed with the DFT plane wave code, Quantum Espresso.⁵¹ Ultrasoft pseudo-potential type Rappe Rabe Kaxiras Joannopoulos⁵² were used with cutoff energies of 544 eV, and the Perdew–Burke–Ernzerhof (PBE)⁵³ functional was chosen for exchange correlation approximation. The convergence energy for self-consistent calculation was set to 0.001 eV. The Brillouin zone of the supercells was sampled by using Monkhorst-Pack⁵⁴ grids of $P \times 1 \times 1$, with P such that the density of k -points in the periodic direction (along the nanotube axis) was not larger than 0.1 \AA^{-1} . The supercells were built with the periodic direction of the nanotube along the z axis and allowing a free space of at least 6.5 \AA between adjacent nanotubes in the xy plane. Geometry optimization of the phosphorene nanotubes was performed with the Broyden–Fletcher–Goldfarb–Shanno algorithm. The thresholds in the Hellmann–Feynman forces and the stress along z were set to 0.02 eV \AA^{-1} and 0.05 GPa respectively.

3 Description of the program

The program's main functionality is to return a set of atom coordinates for a rolled NT made out of a 2D material. These coordinates are generated and calculated geometrically from the unit cell of the 2D material, the only input to the program, which can be either in special-xyz format or as an input file for Quantum Espresso⁵¹ (an example for MoS_2 is included in Section 2 of the ESI†).

Firstly, the unit cell of the 2D material is read from the input file, the unit vectors are defined and the angle φ is calculated. The atomic positions within the unit cell are stored. The user then inputs the (n, m) chiral indices of the desired NT, which is used by the program to build the chiral vector.

The program looks for a \mathbf{T} that satisfies (1) being perpendicular to \mathbf{C} and (2) having integer components x and y (in such a way that the translational vector, when folded, lands exactly on the origin vertex of another unit cell). For a general lattice with angle φ , the following relation was found between x and y , satisfying both requirements for a given chiral angle θ :

$$\frac{y}{x} = \frac{n|\mathbf{a}_1|^2}{m|\mathbf{a}_2|^2} \frac{\sin(\varphi - \theta)\cos(\varphi - \theta)}{\sin(\theta)\cos(\theta)} \quad (1)$$

The program then finds integer solutions for x and y . Since eqn (1) has an infinite number of solutions, the program orders them according to the magnitude of the corresponding \mathbf{T} vector and uses the (x, y) pair that minimizes $|\mathbf{T}|$. For zigzag and armchair nanotubes where n or m might be zero, the following relations should be used:

- When $n = 0$:

$$\frac{y}{x} = \frac{|\mathbf{a}_1|}{|\mathbf{a}_2|} \cos(\varphi) \quad (2)$$

- When $m = 0$:

$$\frac{y}{x} = \frac{|\mathbf{a}_1|}{|\mathbf{a}_2|\cos(\varphi)} \quad (3)$$

In general, for arbitrary φ and $|\mathbf{a}_1| \neq |\mathbf{a}_2|$, eqn (1) does not have exact integer solutions, so the program finds approximate integer solutions with an error e_T which can be tuned by the user (default $e_T = 0.025$). Variations of e_T lead to finding an (x, y) pair faster or slower, yielding bigger or smaller (x, y) pairs for a given pair of chiral indices (n, m) .

However, for cells where $|\mathbf{a}_1| = |\mathbf{a}_2|$ and $\varphi = 60^\circ$ (E.g.: CNTs and MoS_2 NTs), eqn (1) does have exact integer solutions, and in fact the relation simplifies greatly to the well known equation:³

$$\frac{y}{x} = \frac{2n + m}{2m + n} \quad (4)$$

In orthorhombic unit cells like phosphorene or biphenylene,^{40,55} $\varphi = 90^\circ$ and thus eqn (1) also simplifies neatly to produce the following relation:

$$\frac{y}{x} = \frac{n|\mathbf{a}_1|^2}{m|\mathbf{a}_2|^2} \quad (5)$$

For generality, the program always uses eqn (1)–(3), which yield the same results.

After calculating the translational vector, the program creates an array of atoms following the distribution of the unit cells in the lattice. This array is of dimensions $(n + x + 2)|\mathbf{a}_1|$ in the \mathbf{a}_1 direction, and $(m + y + 2)|\mathbf{a}_2|$ in the \mathbf{a}_2 direction, corresponding to the shaded area in Fig. 1b. It would be sufficient for this array to only have dimensions $(n + x)|\mathbf{a}_1|$ and $(m + y)|\mathbf{a}_2|$ (unshaded area in Fig. 1b), as the ribbon fits exactly inside this area, but one more cell is added in each direction for redundancy in case the unit cell has atoms very close to or on the boundary of the cell.

The array is rotated $(\varphi - \theta)^\circ$ counter-clockwise so that \mathbf{C} lays horizontally and then the 2D nanoribbon is cut out of this array, with dimensions specified by $|\mathbf{C}|$ and $|\mathbf{T}|$ (Fig. 1b). The chiral vector's magnitude $|\mathbf{C}|$ also defines the peripheral length of the NT, so that the radius r_c is given by:

$$r_c = \frac{|\mathbf{C}|}{2\pi} \quad (6)$$

After generating the NR and calculating the radius, the program rolls the NR into a NT. There is also an option within the program to just build the NR, which is useful for studies focusing on the properties of nanoribbons. This is a fairly simple process for 2D materials with only one layer, like CNTs or BNNTs, but for multi-layered materials like PNTs or MoS_2 NTs, we have to take into account the distribution of atoms along the \mathbf{a}_3 direction. For multi-layered materials, atoms are shifted so that they are distributed symmetrically about a central plane (blue dashed line in Fig. 2c). This plane is defined by $z = \frac{|\mathbf{a}_3|}{2}$. This central plane will become the central





Fig. 2 (a) Example of a chiral nanoribbon $(n, m) = (1, 12)$ made out of phosphorene, and the ribbon has already been rotated so that \mathbf{C} is horizontal. (b) Parameters of the chiral NT rolled from the nanoribbon. Z is the height of the nanotube which equals the magnitude of its translational vector. X and Y represent the size of the box surrounding the nanotube. (b) Supercell of the chiral NT, where Z is its height, equal to $|\mathbf{T}|$. $X = Y$ are the parameters of the supercell's base. (c) Side view of the unfolded nanoribbon (top) and cross section view of the resulting NT (bottom) showing the central plane (blue dashed line) and the central circumference (blue solid circle), respectively.

circumference in the NT (blue solid circle in Fig. 2c). This means that anything in the central plane will have the 'proper' radius given by eqn (6). The radius corresponding to each atom r_i is given by its z coordinate (z_i):

$$r_i = r_c + (z_i - z_c) \quad (7)$$

where z_c is the chosen value of z for the central plane (defaults to $z_c = \frac{|\mathbf{a}_3|}{2}$), so that any atom that lies above the central plane will have a larger radius when the NT is generated, and any atom below the central plane will have a lower radius, as shown in Fig. 2c. Eqn (7) works so that the distance between layers in the rolled NT is the same as the distance between layers for the 2D material. This also has the effect of 'spreading out' the atoms on the outer layers and compressing the inner layers. The value of z_c can be chosen by the user so that certain layers of the 2D materials or certain regions of it have the proper radius, while the rest of it is expanded/compressed.

Because the \mathbf{a}_3 component is usually unimportant in the study of 2D materials and usually left as a big number so that other layers don't interact with each other, the unit cell given as input for the program does not have to be centred around the $z = |\mathbf{a}_3|/2$ plane. The program will automatically centre the atoms in the cell around that plane.

After rolling the nanotube, the program outputs the coordinates of each atom in the special-xyz format (see example in Section 2 of the ESI†). It also outputs the total number of atoms and the parameters of the nanotube: its central diameter, outer and inner diameters, its height Z (the magnitude of its translational vector) and X and Y parameters, which indicate the size of the box surrounding the nanotube central diameter, shown in Fig. 2. Note that Z may not be equal to the z coordinate of the highest atom, but rather it is the height of the NT unit cell.

Since the program generates the translational vector's indices $(x, y) \rightarrow \mathbf{T}$, it is also able to compute easily other relevant parameters and quantities such as the number of lattice unit cells (the number of hexagons for CNTs) N , the surface area of the nanotube A_{NT} , and the symmetry vector \mathbf{R} . Since the ribbon

is delimited by the chiral vector \mathbf{C} and the translational vector \mathbf{T} , which are perpendicular, its area and thus the area of the NT will be given by:

$$A_{\text{NT}} = |\mathbf{C} \times \mathbf{T}| = |\mathbf{C}||\mathbf{T}| \quad (8)$$

Note that this area represents the surface area corresponding to the central plane of the ribbon/central diameter of the NT, and the actual surface area of the complete NT may be bigger because of the expansion of outer layers.

The number of unit cells N in the ribbon, and therefore in the NT, is given by the ratio of the total area of the ribbon A_{NT} divided by the area of the unit cell. This ratio is given by:

$$N = \frac{|\mathbf{C} \times \mathbf{T}|}{|\mathbf{a}_1 \times \mathbf{a}_2|} = \frac{|(n\mathbf{a}_1 + m\mathbf{a}_2) \times (x\mathbf{a}_1 - y\mathbf{a}_2)|}{|\mathbf{a}_1 \times \mathbf{a}_2|} \rightarrow N = mx + ny \quad (9)$$

Any of the N unit cells forming the nanotube can be denoted by a vector from the origin to its origin vertex. One such vector which has the smallest component in the \mathbf{C} direction is called the symmetry vector \mathbf{R} , shown in Fig. 2a, and is expressed in terms of the lattice unit vectors:¹⁰

$$\mathbf{R} = p\mathbf{a}_1 + q\mathbf{a}_2 \quad (10)$$

where p and q are integers and do not share a common divisor.

By the requirements imposed on \mathbf{R} (minimizing $\mathbf{R} \cdot \mathbf{C}$ and pointing directly to a unit cell within the ribbon), p and q are uniquely determined¹⁰ by

$$xq + yp = 1, (1 < mp - nq \leq N) \quad (11)$$

The program uses eqn (11) to find p and q and thus have the components of the symmetry vector \mathbf{R} . The position of any other unit cell within the NT can be expressed in terms of a multiple of the symmetry vector $i\mathbf{R}$, where i is an integer from 1 to N . When $i\mathbf{R}$ goes out of the ribbon supercell, the vector is shifted to lie within the supercell using a translation by an integer number



of **C** or **T** vectors, using periodic boundary conditions. As such, the vector $N\mathbf{R}$ returns to the origin and $(N + k)\mathbf{R} = k\mathbf{R}$.

Viewed from another perspective, if the NT is extended beyond its unit cell through the **T** direction, then the multiples $i\mathbf{R}$ will point to the i th lattice unit cell, possibly on another NT supercell. Thus, the vector **R** consists of a combination of a rotation ψ around the NT axis and a translation τ in the axial direction (the **T** direction), as shown in Fig. 3a. Therefore, applying N times **R** results in a full rotation around the NT combined with a translation of an integer number M of unit cells along the axial direction, pointing to a point in the NT identical to the origin, as shown in Fig. 3b.

The symmetry vector can therefore be completely defined by its indices (p, q) , like **C** and **T**, or by the parameters (τ, ψ) .

The translation τ across the **T** direction is given by the projection of **R** on **T**, and it can be shown that:

$$\tau = \frac{\mathbf{R} \cdot \mathbf{T}}{|\mathbf{T}|} = \frac{|\mathbf{R} \times \mathbf{C}|}{|\mathbf{C}|} = \frac{(mp - nq)|\mathbf{T}|}{N} \quad (12)$$

It can easily be seen from eqn (12) and the restrictions on p and q from eqn (11) that τ will always be a fraction of $|\mathbf{T}|$, as expected.

The angle of rotation ψ around the NT axis is proportional to the projection of **R** on the chiral vector **C**, which means that eqn (11) also minimizes ψ . This angle is given by:

$$\psi = \frac{2\pi}{|\mathbf{C}|} \frac{\mathbf{R} \cdot \mathbf{C}}{|\mathbf{C}|} = \frac{2\pi}{|\mathbf{C}|} \frac{|\mathbf{R} \times \mathbf{T}|}{|\mathbf{T}|} = \frac{2\pi(xq + yp)|\mathbf{a}_1 \times \mathbf{a}_2|}{|\mathbf{C} \times \mathbf{T}|}$$

And using eqn (9) and (11),

$$\psi = \frac{2\pi}{N} \quad (13)$$

Therefore, $N\psi = 2\pi$, which shows that repeating the symmetry vector N times returns you to the origin, only displaced vertically by $N\tau = M|\mathbf{T}|$ (as shown in Fig. 3b), where M is an integer given

by $M = mp - nq$. As stated earlier, since the program generates the translational vector's indices (x, y) , it can print on-demand the NT parameters A_{NT} , N , $(p, q) \rightarrow \mathbf{R}$ and $(\tau, \psi) \rightarrow \mathbf{R}$ using eqn (8), (9) and (11)–(13). All of these parameters are useful for the theoretical prediction of cutting lines and therefore the electronic properties of NTs and their characteristics in reciprocal space, which we propose as future work.

4 Results

First, we use black phosphorene nanotubes as an example to illustrate the main attributes of the code. The input file containing the unit cell of phosphorene is in the special-xyz format following its description in ref. 56:

```
4
#### 1 row with text####
P 0.0000 0.00 3.93
P 1.1088 3.30 6.07
P 2.5872 1.65 6.07
P 3.1416 1.65 3.93
alat
1.0
mass P 30.97
supercell
4.62 0.00 0.00
0.00 3.30 0.00
0.00 0.00 10.00
cartesian coordinates
```

where the lattice parameters come from DFT calculations in ref. 41. These are the only input lines that the program needs in order to build any chiral or achiral NT desired from this particular material. All the parameters are given in Å.

Fig. 4a includes an armchair phosphorene NR of indices $(n, m) = (10, 0)$, and its cell has 40 atoms and $|\mathbf{T}| = 3.3$ Å; however, for visualization purposes, the NT shown in Fig. 4a is 5 times the repetition of the unit cell along the NT axis. This PNT has in total 200 atoms, a height of $Z = 16.5$ Å and a central radius of $r_c = 7.35$ Å. The generation of the coordinates of 5 unit cells is also a feature of the program. Fig. 4b presents a zigzag PNT with indices $(0, 10)$, and its PNT unit cell has 40 atoms as well, with a height of $|\mathbf{T}| = 4.62$ Å, and it is repeated 3 more times along its axis to give a total of 160 atoms, a height of $Z = 18.48$ Å and a central radius of $r_c = 5.25$ Å. Since these are armchair/zigzag and the phosphorene unit cell is orthorhombic, eqn (1) has exact integer solutions and so the error $e_T = 0$.

In Fig. 4c, a $(n, m) = (9, 6)$ chiral PNT is presented, corresponding to a chiral angle of $\theta = 64.54^\circ$. This specific chirality was used because it has an approximate solution of $x = 1, y = 3$ for the translational indices, with an error of $e_T = 0.02$. Therefore, the NT unit cell is sufficiently small, with only 135 atoms and a height of $Z = 10.92$ Å. Note that this is not in agreement with eqn (9), which predicts that the NT supercell has $N = mx + ny = 6 + 27 = 33$ unit cells. Since the unit cell has 4 atoms, this would mean there are $33 \cdot 4 = 132$ atoms. However, this doesn't take into account the error found in eqn (1) when calculating

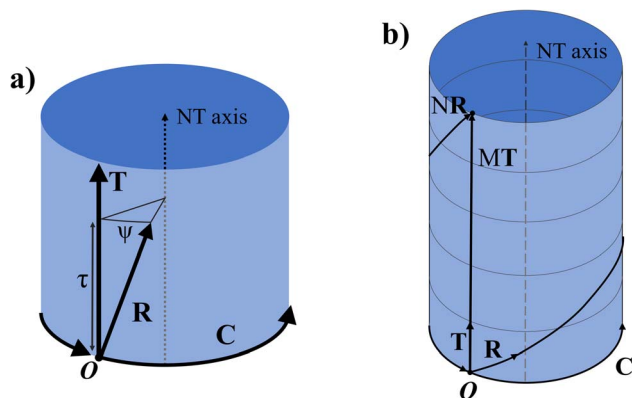


Fig. 3 (a) Schematics of a NT unit cell showing the folded nanoribbon and its now circular chiral vector **C** along with the translational vector **T** parallel to the NT axis. The symmetry vector **R** is shown with its component in the **T** direction τ and the rotation around the NT axis ψ . (b) M nanotube unit cells joined together through the shared NT axis. The **R** vector is shown extended N times so that it wraps around the NT, ending up on a point equivalent to the origin O .



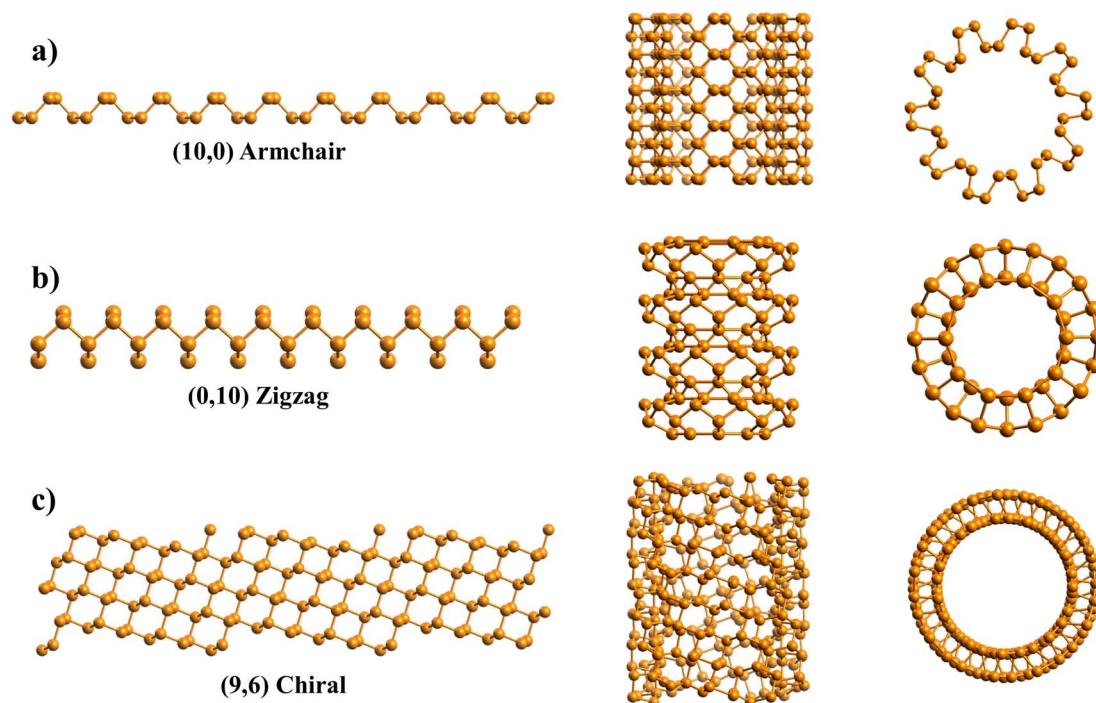


Fig. 4 Schematics of three black phosphorene NTs are shown through orthographic projections in transversal (center) and longitudinal views (right) along with their respective nanoribbons (left): (a) an armchair PNT with indices (10,0), (b) a zigzag PNT (0,10), and (c) a chiral PNT with indices (9,6).

the values for x and y . A function within the program was used to look for this NT; the function searches for chiral NTs, the dimensions of which are small enough so that they can be easily studied. Like for the previous ones, in Fig. 4c, the NT is repeated 1 time along its axis to yield a total of 270 atoms, with $Z = 21.85$ Å and a central radius of $r_c = 7.33$ Å. This chirality was also chosen because it has a similar radius to the other two PNTs.

With an in-built function of the program, one can look for NTs of a selected radius or diameter, instead of looking for a specific (n, m) pair. The program goes through all combinations of n and m (up to a user-customised limit) and calculates their radii with eqn (6). It then prints out those combinations that give a radius similar to the wanted radius, up to an error e_r .

In Table 1 we present the translational vector and the NT parameters of a few chiral PNTs to illustrate the capabilities of the program.

As explained before, since eqn (1) in general does not have exact integer solutions, the program looks for approximate solutions up to an error e_T . To be more specific, e_T represents the maximum accepted difference between a non-integer x (or y) value that satisfies eqn (1) and the closest integer to it. It therefore has units relative to the size of the unit cell vectors, e.g., if $e_T = 0.2$, then the translational vector would be off by a maximum value of $|\mathbf{a}_1| \cdot e_T \approx 0.9$ Å. e_T is a user-customisable parameter, which defines the allowed error, and thus increasing the error will yield more solutions (and smaller solutions), but those solutions will in general be farther from being exact. The two columns in Table 1 include two different values of e_T . It is clear that decreasing the accepted error from $e_T = 0.025$ to $e_T = 0.01$ makes several (n, m) pairs not suitable for study due to the gigantic proportions of the corresponding NT unit cells. This is better illustrated in Fig. 5, where four chiral PNTs were

Table 1 Central diameter (d_c), chiral angle (θ), number of unit cells in the NT supercell (N), height of the NT unit cell (Z) and translational vector (x, y) for a small set of chiral PNTs

C(n, m)		NT parameters		NT size ($e_T = 0.025$)		T(x, y) ($e_T = 0.025$)	NT size ($e_T = 0.01$)		T(x, y) ($e_T = 0.01$)
n	m	d_c (Å)	θ (deg.)	N	Z (Å)	x, y	N	Z (Å)	x, y
6	7	11.49	50.19	51	21.55	3,5	427	180.42	25,42
9	7	15.14	60.95	237	18.91	2,5	2968	237.83	25,63
10	4	15.29	74.05	218	17.13	1,5	2120	168.17	10,49
10	13	20.07	47.12	226	13.54	2,3	7300	441.32	65,98
14	9	22.66	65.34	205	10.92	1,3	4137	221.49	20,61



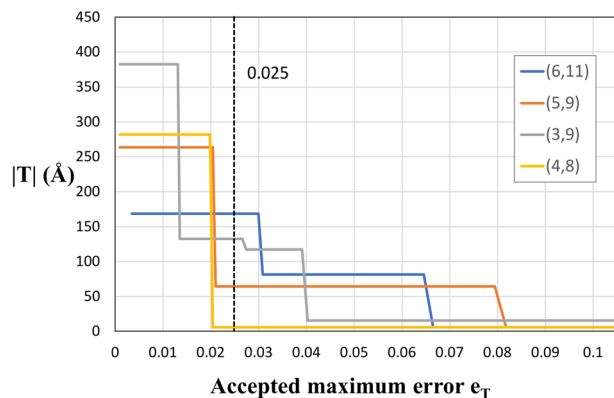


Fig. 5 Plot of the size of nanotubes (its height, $|T|$) vs. the error e_T for several chiral PNTs. The dashed line represents the default value of the program $e_T = 0.025$.

generated with different errors e_T . One can see clearly that increasing the error decreases the size of the NT. The program has $e_T = 0.025$ as default, yielding moderately sized NTs with high precision. However, as explained, this setting is user-customisable and it is recommended to tune it for every material and purpose. For example, we recommend decreasing the accepted error when working with CNTs, because these have suitable exact solutions for (x, y) and having a large e_T returns approximate answers, especially for large $n + m$ and for near-armchair CNTs where $(n, m) = (n, n \pm 1)$. Setting the error to

$e_T = 0.01$ yields results that reproduce exactly those of Maruyama⁸ in the translational indices, chiral angle, diameter, and number of hexagons of hundreds of chiral and achiral CNTs. In fact, the agreement with Maruyama's results was an important benchmark to achieve in order to validate the code.

Further comparison with Maruyama's results⁸ was done by looking at the relation between the nanotube size (the number of atoms or the number of unit cells in its supercell) and radius. The results are presented in Fig. 6, where there are 4 graphs showing the number of atoms (on a logarithmic scale) in every nanotube (chiral and achiral) with $n, m < 30$, ordered by their radius for different materials, including CNTs, PNTs, MoS₂ NTs and Ti₃C₂ NTs. In Fig. 6a, we were able to reproduce exactly the patterns from the literature⁸ for CNTs. The results for the number of atoms were produced through completely general methods, using eqn (1) and the steps explained above to create the NT, and then counting how many atoms were in the final structure. Setting $e_T = 0.01$, these results matched perfectly with theoretical results from eqn (4) and (9), and produced the patterns shown in Fig. 6a.

Following the same entirely general procedure, results for MoS₂ NTs and Ti₃C₂ NTs were obtained and are presented in Fig. 6c and d, respectively. They share the same patterns as for CNTs, as expected, since all have hexagonal unit cells. This specific lattice structure distinguishes precisely between those NTs having (n, m) pairs where n and m are coprime, and those where the greatest common divisor (gcd) of n and m is another number. The distinction is clear in Fig. 6a, c and d, where these

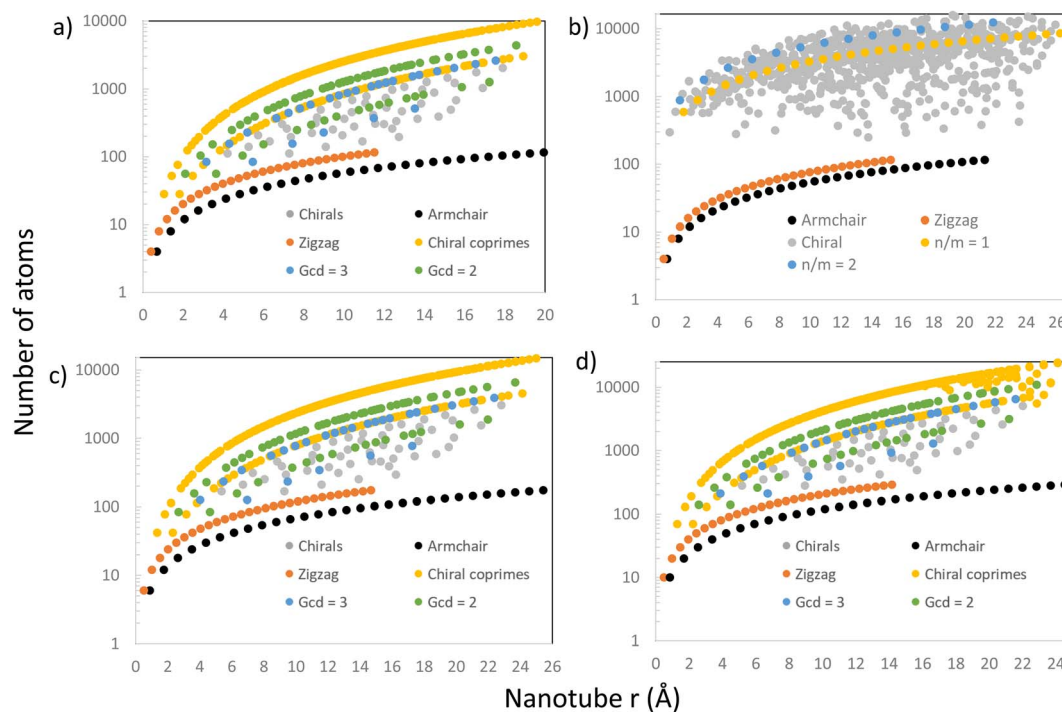


Fig. 6 Plots of the number of atoms for all chiral and achiral nanotubes in the range $n, m < 30$ vs. their radius for 4 different materials: (a) SWCNTs, (b) PNTs, (c) MoS₂ NTs and (d) Ti₃C₂ NTs. In a, c, and d, those NTs with n, m pairs which are coprime are highlighted in yellow, those with $\text{g.c.d} = 2$ in green, and those with $\text{g.c.d} = 3$ in blue. For b, the series of NTs with $n/m = 1$ and $n/m = 2$ are highlighted in yellow and blue, respectively.



NTs have been highlighted in yellow, those with a $\text{gcd} = 2$ in green, and those with a $\text{g.c.d} = 3$ in blue. For hexagonal unit cells, whether $n - m$ is divisible by 3 is also an important distinguishing factor in the electronic properties of NTs, and those NTs that do not have this divisibility criteria (and are coprime) form the upper curve in Fig. 6a, c and d.

However, for PNTs (Fig. 6b) the same pattern does not emerge because they do not share the same type of unit cell. For PNTs, the main feature is n/m , as can be seen directly from eqn (5). This creates a different pattern for the PNTs, exemplified by the yellow and blue curves in Fig. 6b, which represent those NTs with $n/m = 1$ and $n/m = 2$, respectively.

In all four cases, armchair and zigzag NTs are all confined to a relatively small number of atoms compared to chiral NTs. This is due to all zigzag and armchair NTs of a certain material having the same height (same translational vector) regardless of their radius. Therefore, their increase in size comes only from their radius in a linear tendency, and they stay small compared to their chiral counterparts.

To visualise how the number of atoms in a NT supercell depends on the (n, m) pair, in Fig. 7 we present two plots of the number of atoms corresponding to all (n, m) pairs with $n, m < 50$. CNTs (Fig. 7a) show a periodic and size increasing behaviour as n and m increase. The pattern, besides its beauty, allows the line for $n = m$ (armchair) to be distinguished, which stays with a small number of atoms. The plot is perfectly mirrored around the armchair line because of the symmetry in CNTs (and other materials with hexagonal unit cells as MoS_2), where $(n, m) = (m, n)$. Note that those NTs around this line (near-armchair) have a large number of atoms. On the other hand, in Fig. 7b, PNTs do not present a periodic behaviour. However, lines corresponding to $n/m = 1$ and $n/m = 2$ can be distinguished, along with the general increase in NT size as n and m increase.

All of these results were calculated using the code and setting the error to $e_T = 0.01$ ($e_T = 0.001$ for Fig. 7a). This yielded some NTs with thousands of atoms in their supercell, as can be seen in Fig. 6, but gave much more exact results for the desired purpose. Using the default error (0.025) would give much smaller NTs, which are more easily studied using other

computational tools, but would give less accuracy for these results.

The lack of precision in the values of x and y results in a slightly altered geometry on the NTs. Since the purpose of the program is to geometrically generate NT coordinates for several 2D materials, so that these coordinates can be used in further computational studies and simulations, we advise to first relax/optimize the structures obtained by the program, so that these slight inaccuracies will be ultimately negligible. The error also impacts other parameters, which depend on x and y , such as N , \mathbf{R} and its components (p, q) or (τ, ψ) , whose calculated values may not be completely accurate.

Another source of geometric alteration we have talked about is the different radii of different layers of the same material, where atoms above the central plane will be spread out and the ones below the central plane will be compressed. This compression/expansion changes the interatomic distances of atoms in each layer, introducing positive/negative stress within the structure. To exemplify the construction of multilayered NTs (MLNTs) (which is different than the commonly used acronym MWNTs for multiwall CNTs) and features in them, we will use MoS_2 with 3 atomic layers, and the MXene Ti_3C_2 , with 5 atomic layers, and the results will be compared against CNTs and PNTs, with 1 and 2 atomic layers respectively. All the input files used are in Section 1 of the ESI.†

In Fig. 8, we present four graphs with the distance to the first and second neighbours of the outermost layer for all NTs with $n, m < 30$ for different materials, including CNTs, PNTs, MoS_2 NTs and Ti_3C_2 NTs, materials with 1, 2, 3 and 5 atomic layers respectively. Using the program allowed us to include all chiral NTs in this range, as well as the usual zigzag and armchair NTs, making certain patterns more obvious to find. One can see from Fig. 8 that most NTs present some change in the distance to the 1st and 2nd neighbours due to the curvature and that this behaviour depends heavily on the radius of the NT.

NTs generated from materials with more than one atomic layer present an increased distance to the 1st and 2nd neighbours as expected due to the outermost layer being expanded. For these NTs with more than one layer, the increase in the distance depends on the direction towards the 1st and 2nd

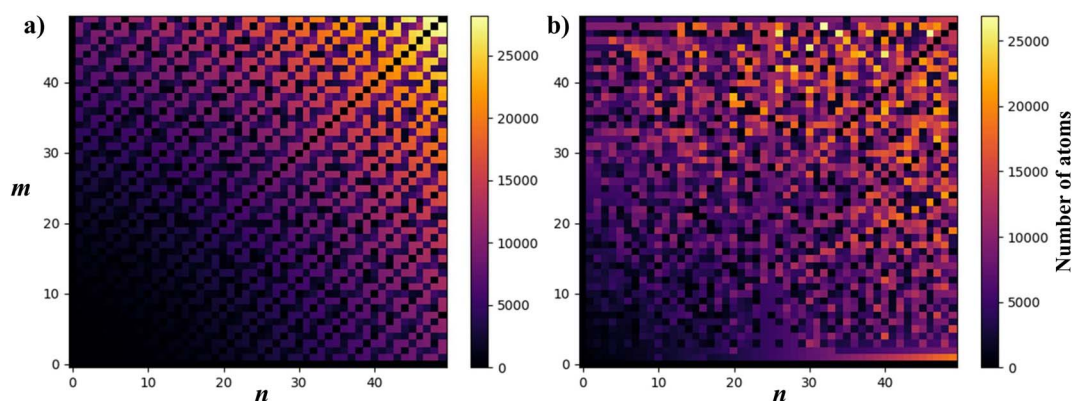


Fig. 7 Plot of the number of atoms corresponding to each (n, m) pair for $n, m < 50$ for (a) CNTs and (b) PNTs. Color represents the number of atoms in the CNT supercells and is plotted in a plane where the x and y axes represent the chiral indices n and m , respectively.



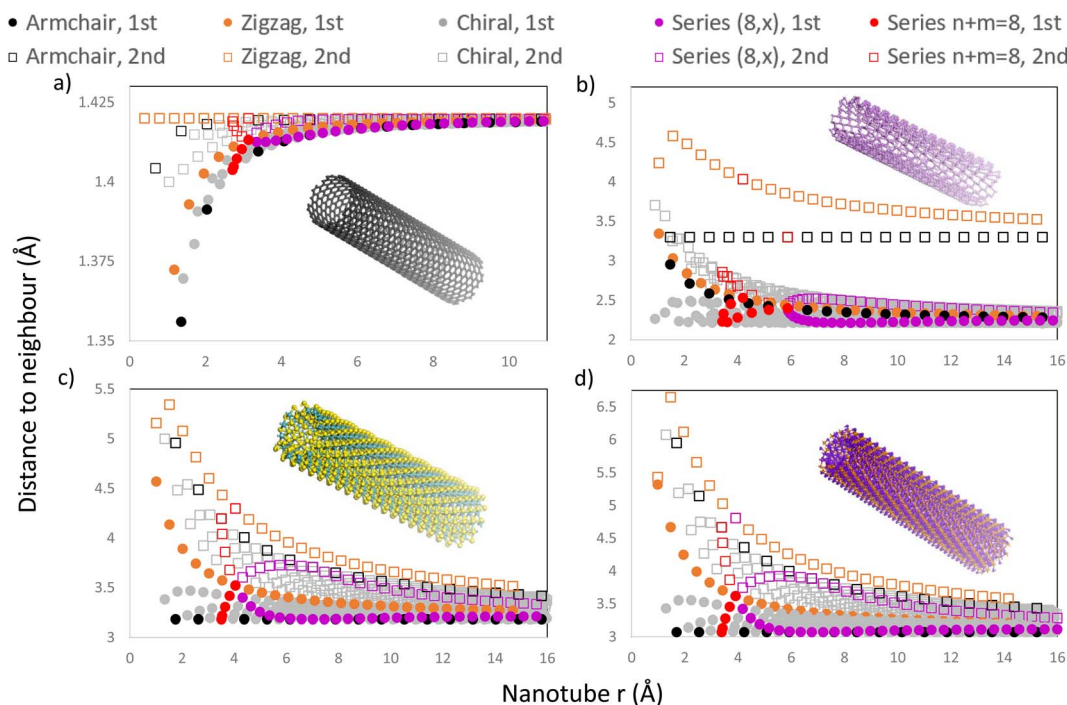


Fig. 8 Plots of the 1st and 2nd neighbours of the outermost atomic layer for all the chiral and achiral nanotubes in the range $n, m < 30$ vs. their radius for 4 different materials: (a) SWCNTs, (b) PNTs, (c) MoS_2 NTs and (d) Ti_3C_2 NTs. In the 4 cases, the nanotube series $(8, m)$ and $n + m = 8$ were highlighted in red and purple, respectively. The insets show atomistic models to illustrate each NT family.

neighbours relative to the NT axis. If the distance to a neighbour is directly in the direction of the NT axis, then the curvature will not change this distance as the curvature is present only in the chiral vector's direction. This behaviour can be seen in the 1st neighbours of armchair MoS_2 NTs and armchair Ti_3C_2 NTs, which remain constant as the radius increases, as shown in Fig. 8c and d, respectively. This also occurs for the 2nd neighbours of armchair PNTs and of zigzag CNTs. SWCNTs, as shown in Fig. 8a, show a different behaviour as expected due to being a single layer, where the distance falls very slightly as the radius decreases, and beyond $r \approx 10 \text{ \AA}$ every NT has a typical interatomic distance for graphene (1.42 \AA).

On all graphs from Fig. 8 patterns emerge for different series of NTs. Those NTs with chiral indices (n, m) sum to the same number, and form trends in a curve with a high slope (e.g. $(8, 0)$, $(4, 4)$, $(3, 5)$, $(2, 6)$ and $(1, 7)$), because they have a similar radius, as shown in Fig. 8, where the $n + m = 8$ families have been highlighted in red. Out of the 4 materials presented, phosphorene is the only one to have a rectangular unit cell with $|\mathbf{a}_1| \neq |\mathbf{a}_2|$, which effectively means that the NT characterized by the chiral indices (n, m) is completely different from the (m, n) NT. This splits the vertical trend into two curves, each corresponding to armchair-like NTs ($n > m$) or to zigzag-like NTs ($m > n$), and they meet at the point where $n = m$ (in this case, where $n = m = 4$), as shown in Fig. 8b.

Another pattern can be found for all four materials for NTs sharing the same n or m . This is exemplified in Fig. 8 where the series $(8, m)$ has been highlighted in purple, forming an almost horizontal curve, which is mirrored around the curve defined by

the 1st neighbours of zigzag NTs (highlighted in orange). We note that all the 1st neighbours of chiral and armchair NTs are situated below the orange curve of zigzag NT 1st neighbours. On the other hand, all 2nd neighbours are located above this orange curve, Fig. 8. A zoomed-in version of these plots is included in Fig. 1 of the ESI,[†] in which the regions of series $n + m = 8$ and $(8, x)$ are amplified.

It should be noted that these results were obtained prior to any relaxation of the structures and were calculated purely from the geometric structure given as output by Chiraltube. The relaxation of the generated structures is briefly explored in the following section for four phosphorene nanotubes. However, a systematic relaxation and calculation of the electronic structure of full nanotube families is proposed as future work.

4.1 Electronic properties of phosphorene nanotubes

As a proof of concept, electronic structure calculations with Quantum Espresso were performed on four phosphorene nanotubes: two with chiral indices $(4, 8)$ and $(6, 6)$, and also one armchair NT $(7, 0)$ and a zigzag NT $(0, 10)$. Black phosphorene was selected for the calculations because its unit cell is not hexagonal and it consists of 2 atomic layers, representing important variations in contrast with conventional cells as that of CNTs and thus representing the utility of the proposed program. These PNTs were selected with a relatively similar radius $5 < r < 6 \text{ \AA}$, and with a relatively small number of atoms (between 28 and 72 atoms) in order to reduce computational costs (again, using built-in functions of the program).



Table 2 Properties of geometry-optimized and electronic structure calculations for phosphorene nanotubes with the chirality $C(n, m)$, central diameter (d_c) and chiral angle (θ), number of atoms (N) and initial height (Z), PNT total energy (E_T), % change of d_c and Z after relaxation, and bandgap (E_g)

$C(n, m)$			Parameters		Initial size		E_T/atom	$\Delta\%$		E_g
Classification	n	m	d_c (Å)	θ (deg)	N	Z (Å)	(eV)	d_c	Z	(eV)
Chiral	4	8	10.26	34.99	48	5.68	-19.0823	5.82	9.02	Metallic
Armchair	7	0	10.29	90	28	3.30	-19.0969	7.68	0.16	0.27
Zigzag	0	10	10.50	0	40	4.62	-19.0895	12.62	7.33	Metallic
Chiral	6	6	10.84	54.46	72	8.06	-19.0915	9.59	-0.09	0.42

The atomic coordinates generated by the program provide the input to the *ab initio* calculations. Firstly, the atomic coordinates and the length of the z dimension of the supercell were allowed to fully relax. Table 2 includes the characteristics of these PNTs before geometry relaxation, and the effect of the relaxation on their diameter and height. PNTs (7,0) and (6,6) present a negligible adjustment of the nanotube height, while NTs (4,8) and (0,10) have an increment of 9% and 7.3%, respectively. All NTs had their (central) diameters increased in the range of 5–12%. The chiral (4,8) PNT had the lowest increase in the diameter of 5.82%, while the zigzag (0,10) had the highest increase, with 12.62%. Interestingly, within this small sample of the 4 PNTs, the zigzag (0,10) and the chiral (4,8) present a higher strain when folded as NTs, and although their original supercell presents high initial stresses along Z , relaxation was successfully achieved in all cases. The four PNTs present a very similar total energy (E_T) per atom, with variations in the thousandth of eV; all nanotube supercells are equally stable.

Secondly, after obtaining the relaxed structures, the band structures were calculated and are presented in Fig. 9. Notice that armchair (7,0) and chiral (6,6) PNTs are semiconducting with indirect bandgaps (E_g) of 0.27 eV and 0.42 eV respectively, while the chiral (4,8) and the zigzag (0,10) PNTs presented a metallic character.



Fig. 9 Band structure of phosphorene nanotubes, with a metallic character for (4,8) and (0,10), and a semiconductor structure for (7,0) and (6,6), and their bandgaps are 0.27 eV and 0.42 eV, respectively.

5 Discussion

The large number of different NT (especially chiral NT) configurations that could be explored having at hand their atomic coordinates could unveil either interesting specific properties of an individual NT with a particular (n, m) , or trends in a physical property of a family of NTs. For example, it has been experimentally found that the broken inversion symmetry of chiral WS_2 NTs endows them with a bulk photovoltaic effect.⁵⁷ An approach to this phenomenon from the theoretical point of view could shed light onto which specific chirality would enhance this effect. Calculating the electronic structure of the relaxed coordinates obtained from our program could unveil similar properties in other TMD NTs. On the other hand, experimental evidence has shown that BNNTs present a chiral distribution that not only includes armchair or zigzag NTs.⁵⁸ This evidence could be corroborated with DFT calculations of the cohesion energy of these nanotubes, hopefully matching the experimental evidence.

Some interesting features could be first studied theoretically, for example, finding the smallest (in radius) NT possible of a certain material through DFT calculations and corroborating them with experiments. Going back to the CNT archetype, SWCNTs, their minimum diameter proposed to be structurally stable and characterised by transmission electron microscopy is the (2,2) armchair SWCNT with a diameter ~ 3 Å.¹³ Thus, relying on a set of atomic coordinates for any 2D material, their NT with the minimum diameter could be predicted.

There are even more options considering functionalised 2D materials leading to functionalised NTs. Theoretical calculations have been done for hydrogenated graphene (graphane)⁵⁹ and hydrogenated BN.⁴⁴ Their respective functionalised NT coordinates are easily obtained by the program presented here, allowing the calculation of any functionalised NT, not only with H, but also with any desired functionalisation.

As graphene, phosphorene is also an interesting and well studied 2D material, existing in one of two allotropes, blue and black phosphorene. The last one presents an $E_g = 1.0$ eV at T^{60} and its tunability through doping has been reported.⁶¹ PNTs in their zigzag and armchair configurations have been reported,^{29,30} but the properties of their chiral NTs remain unknown, presenting an area of opportunity now made readily available through the use of this tool as shown in the short example of Table 2.



Other stars among 2D materials are TMDs. From the early studies of Tenne *et al.*, the issue of chiral NTs, analogous to chiral CNTs, has been stated.⁶² The chirality of experimental TMD NTs is ambiguous in some experimental studies,^{35,63–65} however high resolution transmission electron microscopy and electron diffraction studies have proved the presence of chiral NTs.^{57,66,67} Some remarkable properties have been proven in chiral TMDNTs. For example asymmetric superconducting transport has been proved in chiral WS₂ NTs.⁶⁸ Regarding TMD NTs, some theoretical studies have focused on a tiny sample of them⁴⁵ and also on self-scrolling MoS₂,⁶⁹ which are similar to NTs. It has been suggested that single-walled TMDs have electronic properties that depend on their chirality: zigzag TMD NTs present a direct band gap, resembling their 1H 2D forms, and armchair TMD NTs are similar to the 2H 2D structures;³⁴ however the electronic structure of their chiral configurations is an open problem. Counting with accurate atomic positions of any chiral TMD NT will open the field to high-throughput calculations, allowing experimental data to be proved and expanding the theoretical results that are currently only available for a limited number of NTs.

Perhaps the most promising avenue is the prediction of Van Hove singularities, and the interband energy transition for any nanotube, as has been done with SWCNTs.^{16,70} Knowing the electronic configuration of a 2D material, and their reciprocal lattice geometry, could bring forward a set of Kataura-like plots for many other 2D materials rolled into 1D NTs. Multilayered materials may face E_g shrinkage due to the stress accumulated in the inner and outer layers.

6 Conclusions

Most current theoretical research on nanotubes focuses on the structure of well-known materials and is usually limited to the study of achiral NTs, especially for more complex materials.

The program here presented, Chiraltube, is a tool that allows the expansion of current research of 2D materials into 1D chiral NTs of the same materials. It does this by generalising the concepts learned from the construction of CNTs and applying them to an arbitrary 2D unit cell, thus working with any conceivable 2D material. Knowing the unit cell of a certain 2D material, one can now immediately generate any number of NTs from that material with any chirality desired. The capability of the code was confirmed by comparing the resulting parameters and structures to those in the literature, mostly with CNTs but also with a few achiral examples of PNTs and MoS₂ NTs. Additionally, *ab initio* calculations were performed on generated PNT structures, proving their stability after a successful relaxation simulation and extracting important characteristics from these NTs such as their electronic bandgap.

A user-customised error parameter allows for the generation of shorter nanotube supercells for easier study in further *ab initio* calculations, at the expense of slight geometric alterations which might affect the overall stability of the structure.

One can easily obtain important parameters of generated NTs like their height, radius, translational vector, symmetry vector or the number of unit cells in the NT supercell. This

provides immediate understanding into the trends and main characteristics of NTs of a certain material, exemplified in this work by the analysis of the number of atoms and the interatomic distances of the 1st and 2nd neighbours on the outermost layers of PNTs, MoS₂ NTs, Ti₃C₂ NTs and CNTs.

The main advantages are anticipated to result from the employment of Chiraltube to produce nanotube coordinates that will be a source of information for future research and calculations.

Future work includes calculations of the reciprocal space from the generated structures, in order to give a rough prediction of the electronic properties of the nanotube families purely from a geometric standpoint. Additionally, with the use of Chiraltube, there is an opportunity for a larger project that gives a quick relaxation to the generated NTs and determines their stability to form a massive database of stable nanotube structures of different materials and make it freely available online.

Author contributions

The conceptualisation, funding acquisition, *ab initio* calculations and visualization were done by FCS. The investigation, methodology and writing were shared tasks. The algorithms, coding, and data curation were done by JMAC.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The authors thank J. A. Reyes-Retana for fruitful discussions and encouragement. FCS acknowledges funding from FICSAC and FISMAT, at Universidad Iberoamericana, and also from CONACYT Synergy project 1564464. JMAC acknowledges Universidad Iberoamericana for his *Scholarship to Talent*. Computational resources were provided by FISMAT.

References

- 1 S. Iijima, *Nature*, 1991, **354**, 56–58.
- 2 A. Oberlin, M. Endo and T. Koyama, *J. Cryst. Growth*, 1976, **32**, 335–349.
- 3 A. Jorio, G. Dresselhaus and M. S. Dresselhaus, *Carbon Nanotubes: Advanced Topics in the Synthesis, Structure, Properties and Applications*, Springer, Germany, 2008, vol. 111.
- 4 K. S. Novoselov, A. K. Geim, S. V. Morozov, D.-e. Jiang, Y. Zhang, S. V. Dubonos, I. V. Grigorieva and A. A. Firsov, *science*, 2004, **306**, 666–669.
- 5 A. Geim and K. Novoselov, *Nat. Mater.*, 2007, 183–191.
- 6 R. Saito, M. Fujita, G. Dresselhaus and M. Dresselhaus, *Appl. Phys. Lett.*, 1992, **60**, 2204–2206.
- 7 R. Saito, M. Fujita, G. Dresselhaus and M. S. Dresselhaus, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1992, **46**, 1804.



- 8 S. Maruyama, *Kataura-Plot for Resonant Raman*, <https://www.photon.t.u-tokyo.ac.jp/~maruyama/kataura/kataura.html>.
- 9 V. I. Artyukhov, E. S. Penev and B. I. Yakobson, *Nat. Commun.*, 2014, **5**, 1–6.
- 10 G. Dresselhaus, M. S. Dresselhaus and R. Saito, *Physical Properties of Carbon Nanotubes*, Imperial College Press, London, 1998.
- 11 G. G. Samsonidze, A. R. Saito, D. A. Jorio, E. Pimenta, F. Souza, F. A. Grüneis, D. G. Dresselhaus and M. Dresselhaus, *J. Nanosci. Nanotechnol.*, 2003, **3**, 431–458.
- 12 D. Hedman, H. Reza Barzegar, A. Rosén, T. Wågberg and J. Andreas Larsson, *Sci. Rep.*, 2015, **5**, 1–10.
- 13 X. Zhao, Y. Liu, S. Inoue, T. Suzuki, R. Jones and Y. Ando, *Phys. Rev. Lett.*, 2004, **92**, 125502.
- 14 N. Hamada, S.-I. Sawada and A. Oshiyama, *Phys. Rev. Lett.*, 1992, **68**, 1579.
- 15 T. W. Odom, J.-L. Huang, P. Kim and C. M. Lieber, *Nature*, 1998, **391**, 62–64.
- 16 H. Telg, J. Maultzsch, S. Reich, F. Hennrich and C. Thomsen, *Phys. Rev. Lett.*, 2004, **93**, 177401.
- 17 S. J. Tans, A. R. Verschueren and C. Dekker, *Nature*, 1998, **393**, 49–52.
- 18 A. L. Antaris, J. T. Robinson, O. K. Yaghi, G. Hong, S. Diao, R. Luong and H. Dai, *ACS Nano*, 2013, **7**, 3644–3652.
- 19 S. Pisana, A. Jungen, C. Zhang, A. M. Blackburn, R. Sharma, F. Cervantes-Sodi, C. Stampfer, C. Ducati, A. C. Ferrari and C. Hierold, *J. Phys. Chem. C*, 2007, **111**, 17249–17253.
- 20 M. Irita, T. Yamamoto and Y. Homma, *Nanomaterials*, 2021, **11**, 2309.
- 21 J. W. Wilder, L. C. Venema, A. G. Rinzler, R. E. Smalley and C. Dekker, *Nature*, 1998, **391**, 59–62.
- 22 W. Su, X. Li, L. Li, D. Yang, F. Wang, X. Wei, W. Zhou, H. Kataura, S. Xie and H. Liu, *Nat. Commun.*, 2023, **14**, 1672.
- 23 J. T. Frey and D. J. Doren, *TubeGen 3.4*, 2011, <http://turin.nss.udel.edu/research/tubegenonline.html>.
- 24 W. Humphrey, A. Dalke and K. Schulten, *J. Mol. Graphics*, 1996, **14**, 33–38.
- 25 M. D. Hanwell, D. E. Curtis, D. C. Lonie, T. Vandermeersch, E. Zurek and G. R. Hutchison, *J. Cheminf.*, 2012, **4**, 1–17.
- 26 X. Yang and J. Ni, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2005, **72**, 195426.
- 27 J. Bai, X. C. Zeng, H. Tanaka and J. Zeng, *Proc. Natl. Acad. Sci. U. S. A.*, 2004, **101**, 2664–2668.
- 28 Z. Zhang, M.-Q. Cheng, Q. Chen, H.-Y. Wu, W. Hu, P. Peng, G.-F. Huang and W.-Q. Huang, *Nanoscale Res. Lett.*, 2019, **14**, 1–11.
- 29 J. Hao, Z. Wang and Q. Jin, *Sci. Rep.*, 2019, **9**, 1–8.
- 30 H. Guo, N. Lu, J. Dai, X. Wu and X. C. Zeng, *J. Phys. Chem. C*, 2014, **118**, 14051–14059.
- 31 X. Blase, A. Rubio, S. G. Louie and M. L. Cohen, *EPL*, 1994, **28**, 335.
- 32 C. Ghosh, S. Pal, B. Goswami and P. Sarkar, *J. Phys. Chem. C*, 2007, **111**, 12284–12288.
- 33 R. Fathi and T. Movlaroo, *J. Electron. Mater.*, 2018, **47**, 7358–7364.
- 34 G. Seifert, H. Terrones, M. Terrones, G. Jungnickel and T. Frauenheim, *Phys. Rev. Lett.*, 2000, **85**, 146.
- 35 Y. Zhang, T. Ideue, M. Onga, F. Qin, R. Suzuki, A. Zak, R. Tenne, J. Smet and Y. Iwasa, *Nature*, 2019, **570**, 349–353.
- 36 H.-H. Wu, Q. Meng, H. Huang, C. Liu and X.-L. Wang, *Phys. Chem. Chem. Phys.*, 2018, **20**, 3608–3613.
- 37 X. Zhang, Z. Ma, X. Zhao, Q. Tang and Z. Zhou, *J. Mater. Chem. A*, 2015, **3**, 4960–4966.
- 38 V. Nagarajan and R. Chandiramouli, *Chem. Phys.*, 2017, **495**, 35–41.
- 39 G. B. Pinhal, N. L. Marana, G. S. Fabris and J. R. Sambrano, *Theor. Chem. Acc.*, 2019, **138**, 1–11.
- 40 M. A. Hudspeth, B. W. Whitman, V. Barone and J. E. Peralta, *ACS Nano*, 2010, **4**, 4565–4570.
- 41 D. Pan, T.-C. Wang, C. Wang, W. Guo and Y. Yao, *RSC Adv.*, 2017, **7**, 24647–24651.
- 42 M. Milivojevic, S. Dmitrovic, M. Damjanovic and T. Vukovic, *J. Phys. Chem. C*, 2020, **124**, 11141–11149.
- 43 A. Rubio, J. L. Corkill and M. L. Cohen, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **49**, 5081.
- 44 J. Kroes, A. Fasolino and M. Katsnelson, *Phys. Chem. Chem. Phys.*, 2016, **18**, 19359–19367.
- 45 M. Damjanović, T. Vuković and I. Milošević, *Isr. J. Chem.*, 2017, **57**, 450–460.
- 46 G. R. Bhimanapati, Z. Lin, V. Meunier, Y. Jung, J. Cha, S. Das, D. Xiao, Y. Son, M. S. Strano, V. R. Cooper, *et al.*, *ACS Nano*, 2015, **9**, 11509–11539.
- 47 N. Mounet, M. Gibertini, P. Schwaller, D. Campi, A. Merkys, A. Marrazzo, T. Sohier, I. E. Castelli, A. Cepellotti, G. Pizzi and N. Marzari, *Nat. Nanotechnol.*, 2018, **13**, 246–252.
- 48 P. Miró, M. Audiffred and T. Heine, *Chem. Soc. Rev.*, 2014, **43**, 6537–6554.
- 49 J. Reyes-Retana and F. Cervantes-Sodi, *Sci. Rep.*, 2016, **6**, 1–10.
- 50 J. M. de Albornoz-Caratozzolo and F. Cervantes-Sodi, *Chiraltube*, 2023, <https://github.com/Chema-dac/chiraltube>.
- 51 P. Giannozzi, S. Baroni, N. Bonini, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, G. L. Chiarotti, M. Cococcioni, I. Dabo, *et al.*, *J. Phys.: Condens. Matter*, 2009, **21**, 395502.
- 52 A. M. Rappe, K. M. Rabe, E. Kaxiras and J. Joannopoulos, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1990, **41**, 1227.
- 53 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 54 H. J. Monkhorst and J. D. Pack, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1976, **13**, 5188.
- 55 Q. Fan, L. Yan, M. W. Tripp, O. Krejčí, S. Dimosthenous, S. R. Kachel, M. Chen, A. S. Foster, U. Koert, P. Liljeroth and J. M. Gottfried, *Science*, 2021, **372**, 852–856.
- 56 P. Hirel, *Format: XYZ*, https://atomsk.univ-lille.fr/doc/en/format_xyz.html.
- 57 Y. Zhang, T. Ideue, M. Onga, F. Qin, R. Suzuki, A. Zak, R. Tenne, J. Smet and Y. Iwasa, *Nature*, 2019, **570**, 349–353.
- 58 A. Celik-Aktas, J. Zuo, J. Stubbins, C. Tang and Y. Bando, *Appl. Phys. Lett.*, 2005, **86**, 133110.
- 59 J. O. Sofo, A. S. Chaudhari and G. D. Barber, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **75**, 153401.



- 60 H. Liu, A. T. Neal, Z. Zhu, Z. Luo, X. Xu, D. Tománek and P. D. Ye, *ACS Nano*, 2014, **8**, 4033–4041.
- 61 J. Kim, S. S. Baik, S. H. Ryu, Y. Sohn, S. Park, B.-G. Park, J. Denlinger, Y. Yi, H. J. Choi and K. S. Kim, *Science*, 2015, **349**, 723–726.
- 62 R. Tenne, *Colloids Surf., A*, 2002, **208**, 83–92.
- 63 R. L. Whitby, W. K. Hsu, C. B. Boothroyd, P. K. Fearon, H. W. Kroto and D. R. Walton, *ChemPhysChem*, 2001, **2**, 620–623.
- 64 R. L. Whitby, W. K. Hsu, P. K. Fearon, N. C. Billingham, I. Maurin, H. W. Kroto, D. R. Walton, C. B. Boothroyd, S. Firth, R. J. Clark and D. Collison, *Chem. Mater.*, 2002, **14**, 2209–2217.
- 65 W. K. Hsu, B. H. Chang, Y. Q. Zhu, W. Q. Han, H. Terrones, M. Terrones, N. Grobert, A. K. Cheetham, H. W. Kroto and D. R. Walton, *J. Am. Chem. Soc.*, 2000, **122**, 10155–10158.
- 66 M. B. Sadan, L. Houben, A. N. Enyashin, G. Seifert and R. Tenne, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 15643–15648.
- 67 Y. Chen, H. Deniz and L.-C. Qin, *Nanoscale*, 2017, **9**, 7124–7134.
- 68 F. Qin, W. Shi, T. Ideue, M. Yoshida, A. Zak, R. Tenne, T. Kikitsu, D. Inoue, D. Hashizume and Y. Iwasa, *Nat. Commun.*, 2017, **8**, 1–6.
- 69 Q. Qian, R. Zu, Q. Ji, G. S. Jung, K. Zhang, Y. Zhang, M. J. Buehler, J. Kong, V. Gopalan and S. Huang, *ACS Nano*, 2020, **14**, 13333–13342.
- 70 H. Kataura, Y. Kumazawa, Y. Maniwa, I. Umezumi, S. Suzuki, Y. Ohtsuka and Y. Achiba, *Synth. Met.*, 1999, **103**, 2555–2558.

