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1	Current trends in colorimetric biosensors using nanozymes for detecting biotoxins
2	(bacterial food toxins, mycotoxins, and marine toxins)
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Abstract:

Biotoxins, predominantly bacterial food toxins, mycotoxins, and marine toxins, have emerged as major threats in seafood, food, feed, and medicine fields. They have potential teratogenic, mutagenic, and carcinogenic effects on humans, occasionally triggering high morbidity and mortality. One of the apparent concerns relates to the increasing consumption of fast food the demand for processed food without adequate consideration of the toxins they may contain. Therefore, developing improved methods for detecting biotoxins is of paramount significance. Nanozymes, a type of nanomaterials exhibiting enzyme-like activity, are increasingly being recognized as viable alternatives to natural enzymes owing to their benefits, such as customizable design, controlled catalytic performance, excellent biocompatibility, and superior stability. The remarkable catalytic activity of nanozymes has led to their broad utilization in the development of colorimetric biosensors. This has emerged as a potent and efficient approach for rapid detection, enabling the creation of innovative colorimetric sensing methodologies through the integration of nanozymes with colorimetric sensors. In this review, recent development in nanozyme research and its application in colorimetric biosensing of biotoxins is examined with an emphasis on their characteristics and performance. The study particularly focused on the peroxidase (POD) activity. oxidase (OXD) activity, superoxide dismutase (SOD), and catalase (CAT) activity of nanozymes in colorimetric biosensors. Ultimately, the challenges and future prospects of these assays are explored.

Keywords: Nanozymes; Colorimetric biosensing; Biotoxins; Food safety

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1. Introduction

Nowadays, there is an emerging concern regarding the rapid monitoring of biotoxin presence in food products^{1, 2}. Biotoxins are frequently found in bacteria, viruses, fungi, protozoa, rickettsiae, and infectious substances international polluting feed, food, condiments, seafood, and so forth. Biotoxins pose a risk to public health through the food chain by inducing both chronic and acute toxicity, as well as exhibiting carcinogenic, teratogenic, and mutagenic impacts on human health³ ⁵. Certain biotoxin carriers and producers menace human health and pollute the environment; they can lead to crop failures and diminish the quality of agricultural products. Marine are categorized based on their carriers as ciguatoxins, shellfish toxins, mytilotoxin, tetrodotoxins, and others. There exists a wide range of marine toxins, exceeding 1000 different types, with several dozen having been effectively characterized. These toxins have the potential to infiltrate the food chain, leading to human toxicosis and potentially fatal outcomes. For instance, an estimated 750-7500 individuals die globally because of shellfish poisoning annually ⁶. Bacterial toxins represent a distinct category of biotoxins capable of inducing foodborne illnesses through the inhibition of protein synthesis, leading to neurotoxic effects. The diseases are linked to the impact of bacterial toxins on tissues and are associated with distinct clinical symptoms. The most lethal bacterial toxin associated with food consumption is botulinum toxin, which is created by the clostridium botulinum. It has been reported that 100 ng of this toxin can be fatal to humans 7. Mycotoxins are important class of biotoxins that produced as secondary metabolites by fungi, like Fusarium, Aspergillus, Penicillium, Claviceps, Alternaria, Trichoderma, Stachybotrys, Verticimonosporium, Chaetomium, and so forth, under favorable environmental conditions⁸. The pollution can be happening throughout processing, harvesting, transportation, and storage. The International Agency for Cancer Research has classified mycotoxins into distinct categories according to their

capacity to trigger cancer in humans. These classifications encompass Group 1, Group 2A, Group 2B, Group 3, and Group 4 carcinogens. For example, AFB1 (Group 1 carcinogen aflatoxin B1) is the most toxic and abundant group of mycotoxins. It has been shown to trigger lung carcinoma, hepatocellular carcinoma, colon carcinoma, and gallbladder carcinoma in humans. Liver cancer is caused via AFB1 in around 28.2% of individuals 9. In recent times, biosensors have emerged as a novel method of detection with a broad spectrum of applications for the timely quantitative/qualitative assessment of various biotoxins present in food products¹⁰. A biosensor is a diagnostic tool that integrates three components: a biorecognition unit (e.g., aptamer, enzyme, antibody, phage, cell, etc.), a transducer and a signal conversion element¹¹, 12. The biorecognition unit is employed to specifically identify the target molecule. The interaction between the target molecule and the recognition unite initiates a series of physicochemical interactions. This reaction is characterized by changes such as light absorption and electrical signals, which form the basis for further signal transduction processes. The transducer, being the most crucial component of the biosensor, is capable of converting the above physicochemical alterations into quantifiable signals, thereby facilitating signal transduction. There exist various types of sensors that can be classified into four main groups based on distinct signal transduction methods: optical sensors, electrochemical sensors, magnetoelectric sensors, and piezoelectric sensors ^{13, 14}. Among them, the optical sensor plays a crucial role in analyzing the optical signal produced during the integration of the target and the recognition unite, and then through the realtime conversion and signal amplification of the transducer, it will be converted into readable data to realize the quantitative sensing of the target substance. Colorimetric sensing technology is a quantitative optical method that relies on the correlation between the color change of a solution and the concentration of the target analyte. Compared with other optical sensors, the colorimetric

 sensor has the benefits of visualization, low cost, simple operation, and could be coupled with portable substrate, so the practical use is more extensive. This method provide naked-eye sensing abilities (qualitative) and could be incorporated with smartphone imaging (quantitative), making them well-suited for biotoxin presence detection in various foodstuffs ^{15, 16}. Enzymes are regarded as one of the earliest and most frequently utilized biometric components in biosensors. They have a dual role in recognizing the target substance and promoting electron transfer between the substrate, thereby catalyzing the chemical reaction of the specific substrate to induce corresponding signal alterations ¹⁷. Enzymes have the capacity to act as indicators for the identification of particular substances, facilitating the transformation quantities of the target substance into detectable signals for analytical objectives ¹⁸. As a result, the effectiveness of biosensor detection is somewhat dependent on the particular enzymes employed and their individual characteristics. The defects of natural enzymes like difficulty in preservation, high cost, and poor stability limit their more use in biosensors. The rapid advance of enzyme-like nanomaterials (also known as nanozymes) affords an innovative horizon for the choice of suitable signal markers for analyte identification. Nanozymes demonstrate catalytic properties like natural

efficiency of nanozymes allows them to enhance the color change of platforms, resulting in the
generation of light signals upon the colorimetric platforms introduction. The colorimetric
biosensor that utilizes nanozymes primarily depends on the catalytic capabilities of the nanozymes
to imitate the natural enzymes catalytic functions. This mechanism entails the conversion of a
colorless substrate into a colored product through oxidation, enabling visual detection. In contrast

enzymes, allowing them to catalyze effectively even under extreme situations. This attribute

expands the potential uses of nanozymes in biosensing applications ^{19, 20}. The exceptional catalytic

to colorimetric biosensors lacking nanozymes, the integration of nanozymes with colorimetric

 biosensors could serve as a means of signal amplification to a certain extent, thereby enhancing the determination capabilities of biosensors ²¹. Nanozymes have the potential to serve as recognition units for target analytes in biosensing systems, and their specific surface area offers more active sites to bind more targets, which could further enhance the sensitivity of colorimetric sensors. Furthermore, nanozymes do not possess inherent light-absorbing properties in colorimetric biosensors. However, upon the addition of a colorimetric substrate, nanozymes exhibit various mimetic enzyme activities that facilitate the catalysis of the substrate reaction, thereby initiating a light signal. This process contributes to signal amplification in colorimetric sensors²². As well, the surface properties and specific structure of nanozymes enable them to function as adsorbents for selectively adsorbing target molecules. This capability can enhance the selectivity of the colorimetric sensor. By now, nanozyme-enabled colorimetric biosensors have been broadly employed in food safety analysis owing to the advantages of naked-eye visibility, high sensitivity, easy operation, and portability. However, the applications of nanozyme-enabled colorimetric biosensors in the field of biotoxins analysis have not been specially summarized. Therefore, in this review, we focus on the classification of nanozymes according to the different nanomaterials and offer a detailed description of each type of nanozymes. Moreover, we tried to discuss the construction approach of nanozymes in colorimetric biosensor in terms of catalytic activity, and affords a comprehensive review of the recent research progress of nanozymes in the colorimetric biosensing various biotoxin (mycotoxins, marine toxins, and bacterial food toxins) in food products. Finally, main

challenges and prospects of nanozyme-enabled colorimetric biosensors are also deliberated.

2. Classification of nanozymes

Based on their functional activities, nanozymes can be classified into two primary groups: the oxidoreductase family and the hydrolase family. The oxidoreductases predominantly participate in redox reactions and exhibit various activities, including catalase (CAT), oxidase (OXD), superoxide dismutase (SOD), and peroxidase (POD) like functions. Hydrolases play a crucial role in catalyzing hydrolysis reactions and exhibit functions that are analogous to those of proteases, nucleases, and phosphatases, etc²³. The constituents of nanozymes primarily consist of metals, metal oxides, sulfides, salts, metal-organic frameworks (MOFs), carbon materials, and metal-carbon hybrid nanocomposites ²⁴⁻²⁶. Recently, studies on nanozymes have increasingly concentrated on nanomaterials that possess intrinsic catalytic properties, rather than enzymes or catalysts immobilized on nanomaterials. Currently, the majority of published research regarding the nanozymes classification has focused on categorizing them according to their catalytic activity. In order to offer further classification methods for nanozymes, this study emphasis on categorizing them using different nanomaterials and analyzes the fabrication approaches of colorimetric assay based on the various catalytic nanozymes activities.

2.1. Carbon-based nanozymes

Carbon-derived nanozymes, as a significant constituent of the nanozymes category, refer to carbon-based nanomaterials having enzyme-like catalytic function. Owing to the benefits of environmental friendliness, low cost, good biocompatibility, and easy modification, carbon-based nanozymes are broadly employed in the fields like biosensing, environmental protection, food safety, and disease diagnosis. Nevertheless, unlike other kinds of nanozymes, the synthesis of carbon-based nanozymes mainly relies on a trial-and-error approach. This methodology results in a notable level of uncontrollability in the modulation of the catalytic activity of carbon-based

nanozymes. Furthermore, carbon-based nanozymes exhibit a lower degree of substrate specificity compared to other types of nanozymes, primarily due to the absence of substrate binding pockets. Consequently, there is a necessity to advance rational design approaches aimed at addressing this limitation. Extensive study into the characteristics of carbon-based nanomaterials has revealed their significant potential for uses related to catalytic features. The concept of carbon-based nanozymes was simultaneously introduced. The majority of carbon-based nanozymes are documented to own POD-like or OXD-like function. Additionally, to enhance the catalytic properties of these carbon-based nanomaterials, the incorporation of other elemental dopants is often necessary. For instance, N-doped nanocarbon may hold the exceptional catalytic performance²⁷. Nevertheless, the synthesis of N-doped carbon nanozymes with a high nitrogen content presents challenges due to the instability of the N element at high temperatures. In 2019, Wei and college utilized polyethyleneimine (PEI) as a source of carbon and nitrogen, alongside montmorillonite (MMT) as a template, to synthesize a carbon-based nanozyme characterized by a high nitrogen content ²⁸. MMT was dispersed in an aqueous solution, after which the supernatant was subjected to incubation with a polyethyleneimine (PEI) solution. The obtained solution underwent freeze-drying to yield the assembled powder (MP), which was subsequently subjected to carbonization and etching processes to produce nitrogen-doped carbon nanomaterials. This study identified the critical factor, specifically the N doping content, that influences the catalytic performance of carbon-based nanozymes, and thus was of great importance for the development of carbon-based nanozymes with enhanced catalytic performance. Regardless of the significant progress, the most of existing carbon-based nanozymes exhibit only a singular enzymatic activity. Consequently, the development of methodologies to impart dual or multiple enzymatic activities to carbon-based nanozymes is of considerable interest. In light of that, Gao's team used Pluronic

 F127 as the soft template, and phenol, formalin and melamine as raw materials, to prepare a polymer, which consequently was pyrolyzed to acquire the N-doped carbon ²⁹. The N-doped carbon N-PCNSs3, prepared under optimal conditions, demonstrated a porous nanosheet morphology, so enabling the substrates to diffuse into the pore to increase reaction performance. Furthermore, the nitrogen content in N-PCNSs3 is significantly greater than that of other comparable materials, which enhances its multi-enzymatic activities and overall catalytic efficiency. As estimated, N-PCNSs3 had the POD-, OXD-, SOD-, and CAT-like activities. It is evident that the catalytic function of carbon-based nanozymes is rest upon the precursor. Consequently, altering the precursor may represent a viable approach for the development of highperformance carbon-based nanozymes. In 2019, Choi and colleagues identified the photoresponsive glucose oxidase (GOx) and POD-like activities of carbon nitride by carbonizing melamine in the KCl and KOH existence ³⁰. Graphene, a subclass of carbon materials, features high conductivity, large surface area, strong thermal stability, and excellent transparency and thus, is deliberated a favorable candidate for preparing carbon-based nanozymes ³¹. In 2015, Qu's team found the POD-like function of graphene quantum dots (GQDs) containing of hydroxyl, carboxylic, and carbonyl elements, and thoroughly explore the catalytic mechanism. Following the activity test, it was determined that the carbonyl group serves as the active site and is responsible for determining catalytic activity. The carboxylic group functions to interact with the substrate, while the hydroxyl group is opposite to activity ³². To the best of our knowledge, this study explained the disturbance of surface functional elements on catalytic function of carbon-based nanozymes for the first time, and thereby will lead further researcher to progress the study of preparation and design of high-performance carbonbased nanozymes. Nevertheless, graphene alone demonstrates minimal catalytic function.

 Graphene, in isolation, demonstrates minimal catalytic activity³³. To enhance this activity, it is necessary to employ a doping strategy involving other elements such as N, P, B and S. Furthermore, it is essential to optimize the doping methodology to achieve the most effective coordination effects. In 2019, Lee and his colleagues introduced a series of N- and B-doped reduced graphene oxide, like BN-rGO, N-rGO, B-rGO (reaction between melamine and B-rGO), NB-rGO (reaction between H₃BO₃ and N-rGO), and h-BN-rGO (reaction between H₃BO₃, melamine, and rGO)³⁴. As expected, the synthesized rGO derivatives exhibited a nanosheet morphology, with the doped elements being uniformly distributed throughout the rGO matrix. This phenomenon may be attributed to the synergistic interaction between N and B, which notably enhanced the electron transmission rate during POD-assisted reaction.

2.2. Metal-based nanozymes

Metal-based nanozymes are one of the most typical nanozymes owing to their easy synthesis and stable structure. In general, metal-based nanozymes can be categorized into two primary types: single-metal nanozymes and metal alloy nanozymes, which consist of multiple metal constituents. Furthermore, various metal-doped materials, including metal core/shell nanostructures, have been successfully synthesized ³⁵. These nanozymes show diverse shapes, like nanowires, nanoflowers, nanoparticles (NPs), nanosheets, and nanospheres. Various shapes can exhibit distinct catalytic characteristics. Among them, precious metal NPs like gold, silver, platinum, and palladium are commonly exploited in the determination of bacterial food toxins, mycotoxins, and marine toxins. The synthesis techniques of metal-based nanozymes comprise photochemical, high-temperature reduction, and mediated growth methods. In the experiments concerning nanozyme-assisted analytes sensing, the reduction technique is primarily utilized. For instance, PtNPs were produced using polyvinylpyrrolidone (PVP) as a stabilizing agent sodium and citrate as a reducing agent,

 whereas AgNPs were prepared using potassium hydroxide and L-tyrosine as raw materials, which all have excellent peroxidase function^{36, 37}. In addition to exhibiting single-enzyme-like activity, certain metal-based nanozymes may also demonstrate multi-enzyme-like activity. For instance, peroxidase-like and oxidase-like IrNPs were made-up based on PVP and IrCl₃-3H₂O as raw materials through alcohol reduction technique^{38, 39}. The combined influence of these two activities enhanced the sensitivity and selectivity of the nanozyme. Due to the synergistic interactions among each component, bimetallic nano-alloys frequently exhibit enhanced catalytic behavior. Recently, bimetallic nanocomposites received significant attention owing to their distinctive synergistic effect and multifunction, and are extensively exploited in catalytic field. The Au-Pt nanozyme demonstrated superior peroxidase-like activity compared to the Au nanozyme alone and exhibited greater stability over time than horseradish peroxidase (HRP). This phenomenon can be attributed to the disparity in electronegativity between Pt and Au, which facilitates the migration of electrons from Pt to Au. This transfer enhances the surface electron density of Au, thereby augmenting its catalytic activity 40. To mitigate costs, the copper (Cu) as an element was employed in the synthesis of bimetallic nanocomposites. Ramanathan et al. investigated the process of electroless deposition to facilitate the in-situ growth of Cu NPs on the surface of cotton fabrics 41. The obtained cotton fabrics were then submerged in water solution having AgNO₃ or HAuCl₄ or PdCl₂ or H₂PtC₁₆ to prepare bimetallic nanoparticles, like Cu-Ag, Cu-Au, Cu-Pd and Cu-Pt. Cu-Pt NPs displayed the excellent POD-like function and highest catalytic rate. Despite the extensive research conducted on metal-based nanozymes, the tendency to aggregate and the toxicity associated with certain heavy metals pose significant limitations to their potential applications ⁴².

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2.3. Metal oxidize-based nanozyme

As well to metal-based nanozymes, newly, metal oxide/sulfide/salt-derived nanozymes have gathered higher considerable attention because they enjoyed the merits of simple preparation steps. low cost, and distinctive magnetic/dielectric/optimal features ^{43, 44}. Considering our understanding, the frequently applied techniques mainly included sol-gel, hydrothermal reaction, atomic layer deposition and air pyrolysis. Through adjusting the surface groups, structures, and categories, the catalytic behavior of nanozymes along with functionality and stability can be readily controlled. The pioneering research on nanozymes was documented via Yan in 2007. This study revealed that Fe₃O₄ NPs exhibited POD-like activity for the first time⁴⁵. Engaging TMB, OPD and diaminobenzidine (DAB) as substrates, Fe₃O₄ NPs catalyzed oxidation of them producing blue, brown and orange color, on the basis of which Fe₃O₄ NPs revealed POD-like function. In addition to Fe₃O₄ NPs, V₂O₅ nanocomposites were also typically used to synthesis nanozymes, and till now, V₂O₅ enjoying OXD-, POD-, and dual-enzyme (GOx and POD)-like function with 1D and 2D morphologies are introduced. The primary instance of V₂O₅ with POD-like performance was documented by Tremel's team research 46. Initially, KBrO₃ and VOSO₄ were combined, after which HNO₃ was slowly added to achieve a pH of 2.0. Following this, thermal treatment was conducted at 180 °C for 24 h to synthesize V₂O₅ nanocomposites. In another study, Doong et al exploited the bulk V₂O₅ powder as the raw material, and using DMF to separate bulk V₂O₅⁴⁷. The V₂O₅ nanosheets demonstrated significant oxidative activity, effectively catalyzing the oxidation of TMB to oxTMB. In 2023, Li et al. utilized the reaction between KBrO₃ and VOSO₄ to synthesize 2D V₂O₅ by optimizing the reaction conditions ⁴⁸. The resulting 2D V₂O₅ exhibited superior POD-like activity compared to V₂O₅ with other morphologies.

 MnO₂ was also typically used to develop nanozymes with CAT-, POD-, OXD-, GOx- and SOD-like activities. For instance, Han and colleagues used BSA as soft template to controllably produce 2D MnO₂ nanozyme⁴⁹. Actually BSA with rich-NH₂ and -COOH elements could efficiently fixe Mn²⁺ in its molecular structure to generate Mn²⁺@BSA. Upon the addition of NaOH, Mn²⁺ present in the Mn²⁺@BSA complex was initially converted into MnO(OH). This intermediate subsequently underwent oxidation by O₂, ultimately resulting in the formation of MnO₂. As anticipated, the synthesized MnO₂ under alternative conditions exhibited an irregular flocculent morphology, highlighting the significant influence of BSA.

To sum up, metal oxide-based nanozymes, like peroxidase-like V₂O₅, GeO₂, TiO₂, Fe₃O₄, and oxidase-like MnO₂ are also employed in the determination of various toxins. Nevertheless, the unmodified metal oxide-derived nanozymes might display poor stability and other challenges ⁵⁰. Consequently, some researchers will alter specific components within the metal oxide to enhance its detection efficacy.

2.4. MOF-based nanozymes

MOFs are crystalline materials characterized by a periodic network structure that arises from the self-assembly of organic components, typically organic ligands such as pyridine or carboxylic acids, in conjunction with metal clusters or metal ions, predominantly comprising transition metal ions like Fe²⁺, Zn²⁺, and Cu^{2+ 51, 52}. MOFs are a class of nanomaterials that have garnered significant interest in recent years due to their advantageous characteristics, including a high specific surface area, a porous network structure, and tunable chemical properties ⁵³. The poor selectivity is prevalent among numerous nanozymes; therefore, it is essential to conduct a thorough investigation into the factors contributing to the high selectivity observed in natural enzymes and to incorporate these insights into the development of nanozymes. Similarly, when utilizing MOF-

based nanozymes in vivo, it is imperative to assess their cytotoxicity and biosafety to confirm that they do not pose any harm to the organism. According to their synthesis approaches, MOF-based nanozymes could be divided into chemically modified MOF, pristine MOF, MOF derivatives and MOF-based composites ⁵⁴.

The coordination binding sites of the metal centers within MOFs are often obstructed by the organic moieties, leading to diminished catalytic activity. Therefore, it is imperative to explore novel strategies to address this issue and enhance the catalytic performance of the unmodified MOF. Strategies have been established to augment their catalytic performance, like assembly of metal nanoparticles, surface modification of MOFs, metal oxides, and other constituents in MOF 55. The Fe₃O₄@MIL-100(Fe) composites were effectively synthesized using an in-situ growth technique, which involved the incorporation of Fe₃O₄ NPs within the MIL-100 framework. The catalytic potential of these MOF composites for photo-Fenton reactions was subsequently investigated ⁵⁶. Research has also been conducted on the synthesis of AuNPs@MIL-101 composites through the hydrothermal deposition of AuNPs onto the MIL-101 MOF. This composite material exhibits catalytic properties for the oxidation of ascorbic acid, which can be utilized in conjunction with an electrochemical sensor for the detection of microcystin-LR in water samples ⁵⁷. Both rGO and TiO₂ have been revealed to couple with MIL MOFs to improve their catalytic function^{58, 59}. In addition, MOFs could also be applied as templates or precursors to provide a series of MOF derivatives, for example carbon materials, metal oxides, metals, and so on, owing to their modifiable structures. For instance, by calcining the precursors of Fe-ZIF-67 and Fe-ZIF-8 at high temperatures, a Fe-N-C nanozyme with POD-like function was achieved, which exhibited good performance ⁶⁰.

3. Colorimetric biosensor using nanozymes activity

As is well known, colorimetric biosensing relies on the correlation between the extent of color alteration in the platform and the level of the analyte substance to enable quantitative determination. The alteration in substrate color is dependent on the catalytic function of the enzyme. Nanozymes typically facilitate the conversion of a colorless substrate into a colored one through their inherent enzyme-like properties. For instance, nanozymes exhibit the ability to induce color changes in various substrates, including OPD (o-phenylenediamine), TMB, ABTS, and other chromogenic substrates, enabling colorimetric recognition. Recently, the most research endeavors concerning the catalytic capabilities of nanozymes have primarily concentrated on imitating oxidoreductase characteristics, specifically those associated with OXD, POD, CAT, and SOD functionalities ⁶¹. Some previous studies have explored hydrolytic enzyme mimics, however, the focus has shifted towards the oxidoreductase-imitating capabilities of nanozymes, particularly in the context of colorimetric sensing. This section primarily examines the development strategies and advancements in employment colorimetric nanoprobes that leverage the oxidoreductase-like properties of nanozymes. In recent years, there has been a growing interest in the integration of biosensors with innovative nanomaterials, and nanozymes are broadly exploited in colorimetric recognizing platforms owing to their superior stability and exceptional catalytic characteristics. From one perspective, nanozymes have the potential to serve as biorecognition units for analytes in biosensing applications, as they offer a large specific surface area that can render numerous active sites for binding multiple targets, which in turn can effectively boost the colorimetric sensors sensitivity ^{62, 63}. Furthermore, nanozymes lack inherent light-absorbing characteristics in colorimetric nanoprobes. However, upon the addition of a colorimetric platform, nanozymes exhibit diverse simulated enzyme performances that facilitate the catalysis of the surface reaction,

 thereby initiating a light signal. This mechanism enables nanozymes to contribute to signal enhancement within colorimetric probes ²². Additionally, the distinct building and interface characteristics of the nanozymes enable them to function as adsorbents for selectively capturing the desired molecules. This capability could enhance the colorimetric assay selectivity. More importantly, nanozymes play a crucial role in inducing a color change in the substrate as part of the catalytic process. This transformation enables the conversion of the concentration or amount of the desired molecules into visible hue indicators, facilitating the naked eye detection of signal variations through colorimetric sensors.

2.1. POD-like activity

Peroxidase could catalyze the reaction in the peroxides existence like ROOH and as H_2O_2 redox substrates and electron acceptors, changing them to oxidation and H_2O products ⁶⁴. The nanozymes' POD mimetic system exhibits ping-pong mechanism and Michaelis-Menten kinetics similar to those observed in the catalytic behavior of natural peroxidase enzymes ⁶¹. The ping-pong mechanism is defined by the enzyme's alternating transition between its initial conformation and an altered state. This process involves the enzyme releasing the initial product upon binding with the initial substrate in its base state, transitioning to the altered form, and reverting to the original conformation upon binding with the second substrate to release the subsequent product ⁶⁵. In contrast to the OXD mimetic activity, the POD mimetic system requires the presence of H_2O_2 . H_2O_2 serves as a natural scaffold toward POD, and the nanozymes facilitate the breakdown of the O-O bond in H_2O_2 into (·OH) with significant oxidation potential. The resulting ·OH species can subsequently oxidize various colorimetric substrates, including OPD, ABTS, TMB, and other similar compounds ²². The production of ·OH in processes mimicking peroxidase enzyme activity has been documented through two pathways. One pathway involves the Fenton reaction, where

 Fe²⁺ reacts with H₂O₂ to produce ·OH and hydroxides (OH⁻). The Haber-Weiss reaction as a second pathway can catalyze superoxide (·O²-) via iron ions and H₂O₂ leads to the generation of ·OH 66. The emerging trend in analytical chemistry involves utilizing nanozymes with inherent peroxidaselike activity in conjunction with colorimetric biosensors for the purpose of detecting specific target substances. This approach is using catalytic color development concept through nanozymes. Graphite-like carbon nitride (g-C₃N₄) nanosheets exhibit properties similar to POD enzymes with notable stability. Notably, the functionalization of g-C₃N₄ with heteroatoms has been shown to substantially improve its POD-like activity. In this regard, Fu's group activated g-C3N4 with Pd NPs for the improvement of its enzyme-catalyzed performance and loaded the heteroatomactivated g-C₃N₄ with Fe₃O₄ ⁶⁷. A magnetic platform consisting of Fe₃O₄/Pd NPs supported on g-C₃N₄ was developed to enhance POD-like activity and utilized for immobilizing different natural enzymes, aiming to achieve material reusability and mitigate optical nanozymes interference caused by scattering in colorimetric sensors. Using Fe₃O₄/Pd NPs/g-C₃N₄/GOx, a colorimetric probe was developed for the determination of glucose. There was no detectable evidence at 652 nm once TMB, glucose, and Fe₃O₄/Pd NPs/g-C₃N₄ without GOx were existent, indicating that Fe₃O₄/Pd NPs/g-C₃N₄ does not exhibit inherent GOx-like function. The alteration in TMB hue was observed only when Fe₃O₄/Pd NPs/g-C₃N₄/GOx, TMB, and target were co-present. In another study, Li et al. ⁶⁸ produced S- and N-doped carbon-loaded POD-like FeCoZn triatomic catalysts stemmed from ZIF-8 by taking benefit of two properties, specifically, that MOFs could

catalysts stemmed from ZIF-8 by taking benefit of two properties, specifically, that MOFs could efficiently mitigate the issue of single-atom catalyst agglomeration and the potential of dispersing multiple metal atoms to boost the efficiency of single-atom nanozymes. A colorimetric nanozyme using dual-channel assay was developed by integrating FeCoZn-TAC/SNC with a sensor array,

 aimed at differentiating food preservatives (Figure 1). The FeCoZn-TAC/SNC nanozymes exhibit POD-like activity, facilitating the color development reactions of TMB and OPD to yield green and yellow products, respectively. The food preservatives can adhere to the nanozyme surface via hydrogen bonding and π - π stacking interactions. The diverse levels of interaction noted between FeCoZn-TAC/SNC and different food preservatives led to a reduction in the catalytic efficiency of the nanozymes. Consequently, this occurrence resulted in varying degrees of color signals, which serve as the fundamental mechanism for discriminating among the preservatives. According to this principle, the colorimetric reaction profiles of various preservatives exhibit variations. Through linear discriminant analysis, a 100% accuracy rate was attained during the crossvalidation of seven food preservatives. This result indicates that the sensor array adeptly distinguished seven varieties of food preservatives even at low concentrations.

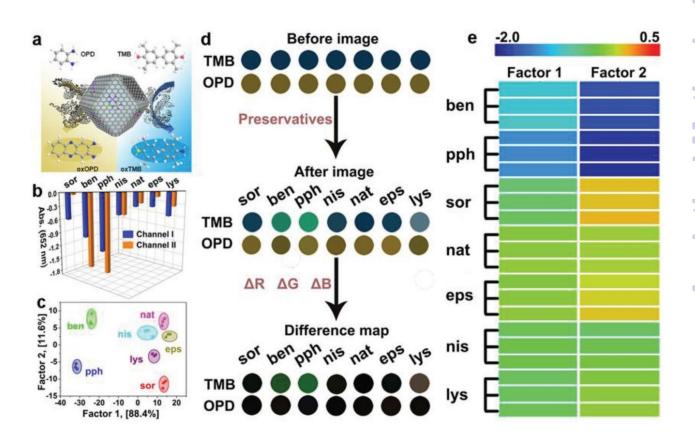


Figure 1. Schematic drawing of fabricating a colorimetric biosensor using two-channel array on the basis of the POD-like function of nanozymes for discriminating food preservatives. Reprinted with permission from Ref ⁶⁸.

2.2. Oxidase-like activity

Oxidases have the ability to facilitate the substrates oxidation (such as electron donors) in the existence of molecular oxygen or other oxidants (such as electron acceptors), leading the production of oxidation products along with H₂O, H₂O₂, or O²-. Currently, a variety of nanozymes have been identified to exhibit oxidase-like activity. Massimiliano and colleagues discovered that gold nanoparticles (Au NPs), even when not loaded with protective agents, exhibited enhanced catalytic activity. Bare Au NPs exhibited the capability to catalyze the production of H₂O₂ and gluconate from glucose in the O₂ presence. In contrast, colloidal particles of Ag, Cu, Pt, and Pd did not demonstrate comparable activity under identical conditions ⁶⁹. Thereafter covalent organic frameworks, metal-organic frameworks, carbon-derived nanozymes, and other nanozymes have been identified to possess OXD-like activity ⁷⁰⁻⁷³. The colorimetric biosensors can be developed by utilizing the color change resulting from the catalytic process of oxidases, where the in situ production of superoxide radicals and hydrogen peroxide oxidizes colorless substrates into colored products ⁷⁴. To some extent, the OXD mimics are considered more appropriate for biochemical analysis compared to the POD mimics due to their ability to function without the need for H₂O₂ in the catalytic procedure. Additionally, the reaction situations for the enzyme mimics are simpler and more direct. Nevertheless, Singh and colleagues determined that nanozymes possessing OXDlike activity have the capability to trigger molecular oxygen conversion into reactive oxygen species like singlet oxygen, oxygen radicals, and hydroxyl radicals during the catalytic mechanism, which can be effectively utilized for the oxidation of diverse substrates⁷⁵⁻⁷⁷. This results in a lack

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 of specificity in the catalytic activity of the nanozymes during the substrate process. In order to address this issue, they developed a biomimetic approach utilizing MOF materials. Through the manipulation of MOF crystal growth in the Z direction, researchers successfully synthesized ultrathin nanosheets of Mn-UMOF using benzene dicarboxylic acid and triethylamine. The presence of the robust donor ligand triethylamine and the bridging ligand benzene dicarboxylic acid in Mn-UMOF leads to an increased density of active sites and enhanced substrate binding properties, particularly for electron unsaturated, when compared to the bulk Mn-BMOF. Additionally, Mn-UMOF could better catalyze the substrates oxidation like amplex red (AR), ABTS, and TMB without the any external oxides addition ⁷⁸. In another study, Zhang's team established an innovative technique that can selectively identify thiophanate-methyl ⁷⁹. A Cu-doped carbon nanozyme, denoted as Cu@NC, was developed with Cu serving as the active center site. The thiocarbamide-like and ethylenediamine-like buildings in the target analyte, thiophanate-methyl exhibit a robust affinity for metal ions. This property enables thiophanate-methyl to selectively interact with Cu@NC in the existence of other pesticides, leading to a significant reduction in the catalytic performance of the nanozyme and facilitating colorimetric detection (Figure 2). The researchers also examined the specificity of the colorimetric sensor and revealed that thiophanate-methyl had a direct and specific inhibitory effect on the OXD performance of Cu@NC nanozymes. To examine the mechanism through which the nanozymes activity is inhibited, experiments were carried out. These experiments revealed a reduction in the nanozymes catalytic activity after pre-incubating the target molecule, thiophanate-methyl, with Cu@NC. Subsequently, the thiophanate-methyl was introduced to the chromogenic substrate. Upon the introduction of thiophanate-methyl into the conventional system consisting of TMB and Cu@NC, it was observed that the catalytic activity remained unaffected. This suggests that the

decrease in Cu@NC activity caused by thiophanate-methyl is a result of its direct interaction with Cu@NC, rather than from enhancing the reduction of oxTMB. Additional inhibitory elements were subsequently examined, and it was lastly found that TM can be fixed onto the Cu@NC interface via π - π stacking interactions and attached to its metal sites to suppress its catalytic function.

Cu Cu Wavelength (nm)

Direct and Specific Nanozyme Inhibition

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 CeO₂ nanoparticles demonstrate remarkable mimetic activity similar to oxidase enzymes, enabling the catalysis of TMB color progress in the absence of hydrogen peroxide. In this regard, Jia's group developed a multifunctional incorporated portable colorimetric detecting substrate for the determination of bisphenol A 80. As depicted in Figure 3, the experimental configuration utilized the CeO₂@ZIF-8/Apt nanoprobe both the signal generation unit and recognition component. Sodium alginate hydrogel tubes were employed to contain the TMB and catalytic reaction buffer as the reaction medium. The hydrogel containing the TMB platform was attached to the top of the tube, while the signal probe was placed at the base of the tube. Following the introduction of the sample and subsequent mixing with the signal probe, the Apt molecules on the signal probe dissociated and attached to the target, which resulted in the triggering of the OXD activity from the CeO₂@ZIF-8. Upon completion of the reaction process, photographs of the reaction solution were taken using a smartphone subsequently uploaded to a color analysis based on the RGB app. This method was employed to quantitatively evaluate the target object.

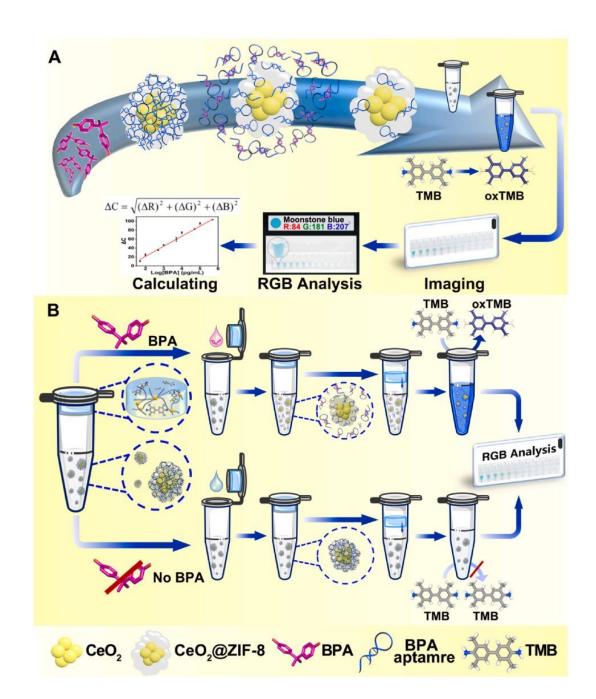


Figure 3. (A) Diagram description of the fabrication and the OXD-like properties of CeO₂@ZIF-8/Apt nanoprobes. (B) The sensing mechanism of one-pot portable testing substrate for bisphenol A determination. Reprinted with permission from Ref ⁸⁰.

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2.3. Catalase-like activity

The catalase enzyme facilitates the degradation of hydrogen peroxide into oxygen and water. The catalase-like properties of NPs were first reported in amine-terminated PAMAM dendrimers that encapsulated gold nanoclusters, which were observed in both physiological and acidic conditions⁸¹. Likewise, several nanozymes-like manganese oxide (Mn₃O₄) nanoparticles, platinum nanoparticles, and cerium oxide nanoparticles are defined to display the catalase-like activity. The molecular-level investigation of the catalase-like behavior of nanozymes explores mechanisms involving bi-hydrogen peroxide association, base-like dissociation, or acid-like dissociation 82. The bi-hydrogen peroxide mechanism has been identified as the most appropriate explanation for the catalase-like activity in nanozymes, particularly in the case of cobalt (II, III) oxide (Co₃O₄) nanoparticles. The catalase-like activity of cerium oxide nanoparticles involves the oxidation of hydrogen peroxide on the nanoparticles' surface to form oxygen. This process leads to the reduction of cerium oxide to H₂-ceric oxide. Subsequently, H₂-ceric oxide reacts with another hydrogen peroxide molecule, resulting in the production of water⁸³. Further, Zhang et al. illustrated the catalase-like function of iron-based single atom nanozymes (Fe-SANzymes) considered via obviously exposed edge-hosted defective Fe\N₄ atomic sites⁸⁴. The mechanistic investigation demonstrates that defects facilitate a substantial charge transfer from the Fe atom to the carbon matrix. This process activates the central Fe atom, enhancing its interaction with hydrogen peroxide and simultaneously weakening the O\\O bond.

2.4. Multi-enzyme-like activity

Nanozymes with CAT-like function are frequently employed to eliminate extra naturally happening reactive oxygen hydrogen peroxide, which is somewhat close to POD enzymes. Nevertheless, nanozymes typically lack the capability to oxidize substrates to enable hue advance.

 Consequently, it is less frequent to depend solely on the nanozymes in terms of the CAT activity for the fabrication of colorimetric nanoprobes. Superoxide dismutase is a metalloenzyme with antioxidant properties found in various organisms, capable of enabling the dismutation of reactive superoxide anion radicals into O₂ and H₂O₂. SOD enzymes like CAT enzymes were frequently employed to neutralize surplus reactive oxygen classes, playing a pivotal role in body's oxidation and antioxidant equilibrium. The catalytic mechanism of SOD enzymes primarily entails the protonation of superoxide anion (O2⁻⁻), which in turn protonated by water to produce OH⁻ and HO2 radicals. Nanozymes with O2 scavenging capability are deliberated to be favorable substitutes to natural SOD enzymes. If nanozymes demonstrate multiple simulated enzyme functions simultaneously, like SOD, OXD, and POD, the dominant activity is expected to be the SOD enzyme activity. This activity is influenced by factors like environmental pH, surface ions, and nanomaterial structure 85. Currently, the majority of reported nanozyme activities focus on POD and OXD activities, with limited research on mimicking SOD activity. Moreover, given that the primary purpose of the SOD enzyme is to preserve redox equilibrium in cells and mitigate oxidative stress, a significant portion of research endeavors focusing on nanozymes with SODmimetic characteristics are primarily directed towards mitigating inflammation. In contrast, there is a conspicuous lack of research investigating the employment of SOD nanozymes in colorimetric recognizing methodologies. Like CAT enzymes, the utilization of SOD-like nanozymes in colorimetric probes typically depends on the multifunctional enzyme properties of the nanozymes.

4. The application of nanozyme based on the colorimetric biosensor for biotoxins detection

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4.1. Mycotoxins detection

Mycotoxins, as one of the most alarming food and feed contaminants, are carcinogenic and highly toxic secondary metabolites produced by specific fungi, predominantly molds, which have the potential to contaminate a broad spectrum of crops (including nuts, grains, and legumes) and this, in turn, can be transferred to various food products. These naturally occurring toxins pose a substantial risk to both human, with exposure capable of inducing numerous adverse effects such as acute poisoning, chronic ailments, and potentially cancerous consequences. Mycotoxins are typically prevalent in cereals, grains, nuts, spices, coffee, cocoa, dried fruits, and animal-sourced products such as milk and meat. Prominent and extensively researched mycotoxins comprise aflatoxins, ochratoxins, fumonisins, trichothecenes, and zearalenone 86, 87. Therefore, the implementation of effective and innovate mycotoxin analytical detection methods, and alongside that, novel nanomaterials has become pivotal for safeguarding humans against health dangers and risks. This section has been conducted to review the major mycotoxins, including aflatoxin B₁ (AFB1) and ochratoxin A (OTA), detection based on nanozymes.

4.1.1. Aflatoxin

Aflatoxins are hazardous secondary metabolites primarily synthesized by *Aspergillus fungi*. These toxins reveal significant mutagenic, carcinogenic, and teratogenic potency in both human and animal subjects. Corn, rice, peanuts, dried fruits, spices, and dairy products can be considered the most important source of these toxins. Among different aflatoxins, AFB₁ introduced the most

 perilous member of the aflatoxin group, as a group 1 carcinogen, responsible for major of all aflatoxins-associated feed and food contamination ^{88, 89}. In this regard, the maximum allowable limits of AFB₁ in foodstuffs were set a 1.0–20 µg/kg. The field of biosensors based on various nanozymes including single-atomic nanozymes ⁹⁰, MOFs ⁵⁵, and metal nanoparticles ¹⁹ is rapidly evolving, with ongoing research focused on understanding their mechanisms of action and increasing their catalytic efficiency. In favor of the application of metal nanoparticles as nanozymes, two metal components demonstrate better peroxidase catalytic performance than monometallic nanozymes. For example, Zhao et al. 91, reported a surface-enhanced Raman scattering (SERS) sensor based on gold-mercury nanoparticles (Au@HgNPs) coupled with carbon dots (CDs) for AFB₁ determination. In this study, the poor colloidal stability and low enzyme-like activity of the AuNPs were distinctly improved by using Hg²⁺. This modification could be beneficial for the oxidase-mimicking activity of the AuNPs, attributing to Hg2+ reduced to the metallic (Hg⁰) forming Au@HgNPs. Under optimal conditions, the presence of the target inhibited the aggregation of Au@HgNPs particles, through oxygen atoms in the carbonyl group, which caused SERS intensity. In 2022, Lai et al. 92, focused on introducing a simple and rapid synthesis method of nanozyme by simply mixing Cu(II) and K₃[Fe(CN)₆] for presenting a copper hexacyanoferrate nanoparticles (CHNPs) in terms of AFB₁ detection. Elaborately, in contrast to the common nanozyme synthesis methods which rely on re-synthesized nanomaterials, the nanozyme was designed in this biosensing platform without requiring the tedious process and preparation of nanomaterials. Therefore, this biosensing approach can broaden our horizon about another important factor in terms of the application nanozyme in detection, which is tedious nanozyme preparation, by offering a simple, rapid, and accessible way to generate the nanozyme on-demand. Interestingly, bimetal nanozymes demonstrate better catalytic efficacy compared to

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 their single-metal counterparts, owing to the synergistic interplay between the two metal varieties. Most recently, Wu and co-workers ⁹³, exploited mesoporous SiO₂/gold-platinum (Au-Pt), m-SAP, in the structure of colorimetric biosensing device for AFB₁ detection. In this protocol, the mesoporous of SiO₂ nanospheres were loaded with Au-Pt, enjoying high catalase like activity. In the absence of the target, the complementary DNA conjugated m-SAP was captured by aptamermagnetic nanoparticles and facilitated the 3,3',5,5'-tetramethylbenzidine (TMB)/H₂O₂ coloring system. The detection limit of the developed colorimetric aptassay was 5 pg/mL, 600-fold lower than that of traditional ELISA. In addition, during interference testing, AFB₁ could be differentiated from six other interfering substances. In order to achieve more a reliable and highly sensitive nanozyme based on two metal varieties, Jiang and colleagues ⁹⁴, exploited the advantages of bimetallic MoS₂/Au nanozyme, as a substrate for immobilization of aptamer, in developing of sandwich-based colorimetric aptasensor of AFB₁ (Figure 4). In this work, the synergic effect between MoS₂ and Au increased catalysis activity and stability. In detail, the low Michaelis constant (Km) and the high maximum reaction rate (Vmax) of MoS₂/Au in comparison of singlecomponent MoS₂ or Au nanomaterials demonstrated that stronger binding affinity and superior catalytic efficiency, respectively. Importantly, the specific structure, the flower-like MoS₂/Au composite, introduced numerous surface-active sites through its unique multilayered and porous structure which was considered an excellent immobilization substrate of aptamers. Furthermore, the high surface area and porous structure of silica aerogel modified stainless-steel mesh (SiO₂/SSM) could immobilization of the negatively charged aptamer2. Under normal circumstances, the construction of sandwich aptassay was conducted by capturing AFB1 via SiO₂/SSM, followed by binding of the MoS₂/Au/aptamer1. This structure was successfully

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 employed to quantify AFB₁ in different real food samples including peanut, corn, and wheat, with negligible matrix effects (90.84%-106.11%) and recoveries of 88.52%-113.40%.

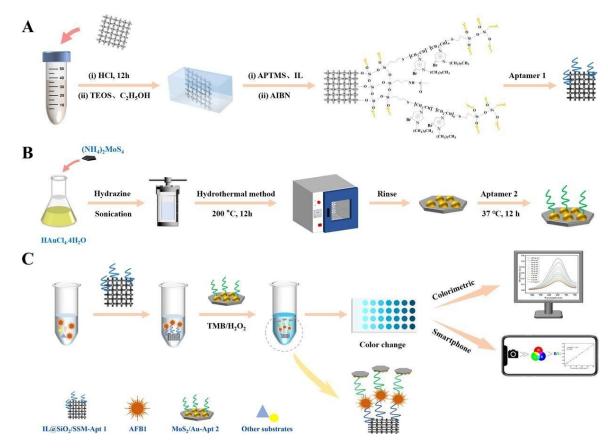


Figure 4(A-C). Representation of colorimetric aptasensor based on MoS₂/Au for AFB₁ determination. Reproduced with permission from ref ⁹⁴ Copyright Elsevier Science, 2024.

Another excellent example of this concept was implemented in colorimetric and photothermal dual-mode immunoassay using peroxidase-like activity of Pt supported on nitrogen-doped carbon for AFB₁ quantification ⁹⁵. As shown in Figure 5, the immune competition conducted between glucose oxidase (GOx)-labeled AFB₁-bovine serum albumin (BSA) and AFB₁, therefore releasing GOx to catalyze the glucose production of H₂O₂. Under normal condition, the colorimetric signal was produced due to oxidization of TMB to TMBOx. Along with the colorimetric signal, a thermal

 signal was achieved when TMBOx underwent photothermal conversion under 808 nm laser irradiation. The fabricated biosensor was able to detect AFB₁ with a LOD of 0.22 and 0.76 pg/mL.

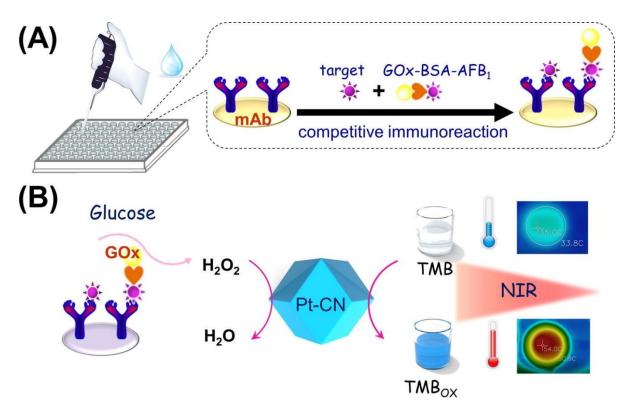


Figure 5. (A) Competitive-based immunosensor exploiting GOx-labeled AFB₁-BSA conjugate as the tag. (B) Illustration colorimetric and photothermal measurement based on H₂O₂-responsive peroxidase-like activity of Pt-CN. Reproduced with permission from ref 95. Copyright Elsevier Science, 2019.

The application of multimodal biosensors based on nanozyme in AFB₁ determination can introduce different signals, extending the linear range of quantification. In addition, these signals can verify each other to improve the accuracy of biosensing approaches ^{96, 97}. Another dual-mode approach for AFB₁ quantification based on Ag@Au IP6 bifunctional nanozyme, with peroxidaselike activity and SERS effect, was reported by Tan and colleagues 98. For this purpose, the surface of magnetic particles was decorated with AFB₁ aptamers along with a trigger probe. In the presence

 of the target, the conjugation of targets with specific aptamers led to the trigger probe and this, in turn, initiated a hybridization chain reaction (HCR) which introduced alkaline phosphatase (ALP), catalyzing the self-assembly of the Ag@Au IP6 nanozyme. The constructed nanozyme revealed increased peroxidase-like activity, improving the oxidation of TMB to the blue-colored TMBOx, additionally, the core-shell structure of that also enabled strong SERS enhancement of the TMBOx signal.

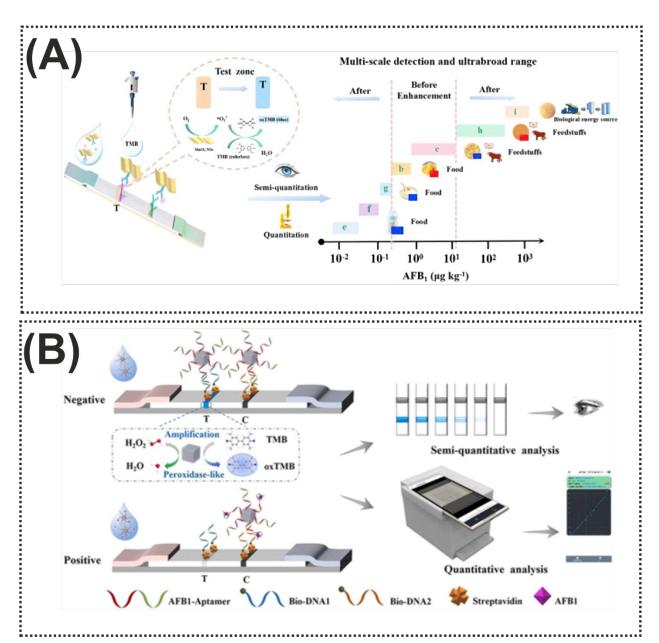
Particularly, in light of economic discussion, substituting precious metals, which contribute to nanozymes, with more affordable transition metals, using in the structure of nanozymes can markedly diminish the expenditure associated with nanozymes. For instance, Cai et al. 99, designed a novel colorimetric nanozyme-based lateral flow assay (LFA) based on MnO₂ nanosheets (MnO₂ NSs) as an oxidase mimics and catalytic label for AFB1 determination. As shown in Figure 6A, the test (T) line of the strip was decorated with anti-AFB₁ antibody-conjugated MnO₂ NSs for capturing AFB₁. Attractively, thanks to the properties of MnO₂ NSs in catalyzing the oxidation of TMB, TMB solution was added onto the T-line. When the target was added to the immunosensing device, the oxidation could provide a visual color signal which was inversely proportional to the AFB₁ concentration in the sample. The reported nanozyme-strip bioassay revealed a LOD of 15 pg/mL for AFB₁, over 100-fold lower than the maximum limit set by the European Union. On the other hand, both of the instability of the colloidal nature of nanozymes and their complicated interactions with bioreceptors can limit nanozyme exploitation in LFA. In this light, it is crucial to rationally design nanozyme-based signal labels with features that facilitate easy functionalization, good dispersibility, distinctly visible color, and enzymatic activity ¹⁰⁰. These features are pivotal for increasing the applicability of nanozymes in LFA applications. A good example of this concept was prepared in another transition metal, CuCo, which was coated by polydopamine (PDA) with

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excellent biocompatibility, good adhesion, and rich functional groups (chinone, imine, and amine). leading to good hydrophilicity and binding ability with biomolecules 101. In this protocol, the carboxyl-functionalized aptamers were conjugated with CuCo@PDA nanozyme via amide condensation reactions. The fabricated probe was used on the surface of the T line and the difference of color with/without the presence of the target was the principle of detection (Figure 6B). Further, the ability of the nanozyme to catalyze the oxidation of TMB-H₂O₂ could amplify the color change on the T-line. To illustrate, the visual LOD was reduced to 0.1 ng/mL by the TMB-H₂O₂ catalytic amplification. In 2024, Fan and colleagues ¹⁰², designed another functionalized nanozyme, flower-like L-cysteine-functionalized FeNi bimetallic nanoparticles (L-Cys-FeNiNPs), with excellent peroxidase-like catalytic activity in colorimetric aptasensor for detection of AFB₁. In the reported nanozyme, the peroxidase-like activity was attributed to the generation of superoxide radicals (•O₂-) and holes (h⁺) that were produced through the catalysis, and alongside that, the high selectivity of probe was described as the conjugation of specific aptamer with L-Cys-FeNiNPs via EDC/NHS chemistry and streptavidin-biotin interaction. Indeed, like previous modification, the decoration of FeNiNPs with L-Cys not only can help increase the dispersibility and stability of the FeNiNPs, but also the biocompatibility and functionalization capability of nanozyme significantly improved. In the presence of the target, the reduction of the active sites of the L-Cys-FeNiNPs suppressed the TMB oxidation and reduced the color signal.

 Along with metallic-based nanozymes, the application of metal-organic frameworks (MOFs) can be considered as a high potential nanozyme for quantification of AFB₁. Among different MOFs,

Figure 6A. Schematic of nanozyme-strip bioassay based on MnO₂ NSs for AFB₁ detection.



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 porphyrin (PCN)-based organic linkers were employed for self-assembly with metal nodes, leading in the development of MOFs with different structures and functions. On the other hand, accessing the active sites within MOFs remains challenging. One of the efficient solutions is based on hybrid nanomaterials, which have been attributed to using platinum nanoparticles Pt NPs on two-dimensional support substrates ^{103, 104}. As example, Zhang et al. ¹⁰⁵, reported a novel colorimetric approach exploiting Pt@PCN-222 nanozyme, with oxygen vacancies, for AFB₁ detection. The detection principle of this study was operated according to the oxidization of the 2,2'-azino-bis(3-ethylbenzthiazoline-6-sulfonic acid (ABTS) substrate to produce a blue-green colored product. In the presence of the target, the binding of Pt@PCN-222 with the target caused inhibition which decreased the number of active Pt@PCN-222 conjugates available for the ABTS oxidation reaction. Another technique for addressing accessibility to the active sites of MOFs is operated based on a synthesis technique, NanoMOFs, can accelerate substrate diffusion in catalytic MOF material and this, in turn, provides greater external surface area and lower diffusion barriers. For instance, Peng and co-workers ¹⁰⁶, developed a sensitive, reproducible, and accurate colorimetric immunoassay based on NanoPCN-223(Fe) with high peroxidase-like activity and excellent dispersion for AFB1 determination (Figure 7). In detail, the implantation of NanoPCN-223(Fe) in the structure of colorimetric technique which produced color by catalyzing the oxidation of the colorless substrate TMB in the presence of H₂O₂. To illustrate, through the catalyzing process, the nanozyme generated hydroxyl radicals (•OH), from H₂O₂, which oxidize the TMB substrate, converting the colorless TMB to the blue ox-TMB. Under normal conditions, the conjugation of the nanozyme and the target inhibited catalyzed oxidation of TMB which reduced the intensity of color.

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 All things considered, the application of nanozymes based on metal nanoparticles and MOFs can be considered efficient materials for aflatoxins detection. The investigation of these materials is based on two metals varieties, bimetallic and affordable materials, and alongside that, the amplification techniques can broaden our horizon about the performance of nanozymes in colorimetric approaches for the detection of aflatoxins.

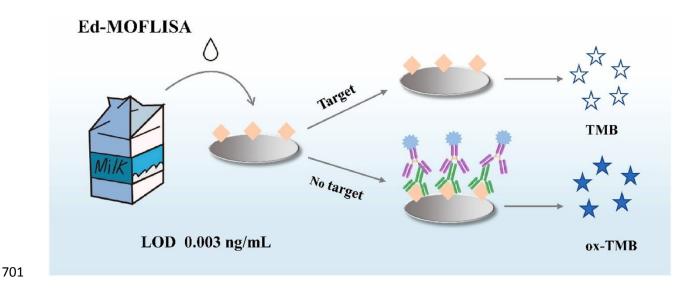


Figure 7. Illustration of the application of NanoMOFs for colorimetric detection of AFB₁. Reproduced with permission from ref ¹⁰⁶. Copyright Elsevier Science, 2022.

4.1.2. Ochratoxin A

According to the International Agency for Research on Cancer, Ochratoxin A, as a possibly carcinogenic to humans (Group 2B), is classified as one of the significant mycotoxins. To elaborate, high chemical stability against heat treatments and hydrolysis through food processing makes it one of the most dangerous poisons for human. This mycotoxin originates from the species of fungi including Penicillium verrucosum, Aspergillus carbonarius, Aspergillus ochraceus, and Aspergillus niger 107, 108. Over the last decades, the advent of nanozyme has revolutionized multiple domains in detection of OTA. Interestingly, the investigation of common structures of reported

 nanozymes can open new doors in terms of quantification of OTA. One of the common structures is spinels which form huge families, and they consist of one or more metal elements. Among them, spinel-type metal oxides, with the formula of AB_2O_4 , enjoy superior promoting nanozyme performance over other metal oxides due to controllable structure, composition, valence, and morphology $^{109,\,110}$, For example, Huang and co-workers 111 , employed OTA-specific aptamer on the surface of manganese cation substituted cobalt oxide $(MnCo_2O_4)$ for OTA determination. As shown in Figure 8, the attachment of aptamer on the surface of the nanozyme inhibited the nanozyme activity of spinel $MnCo_2O_4$ through aptamer-target complex, presenting a new colorimetric aptasensor. Under normal conditions, the developed aptassay could able to detect OTA with a LOD of 0.08 ng/mL.

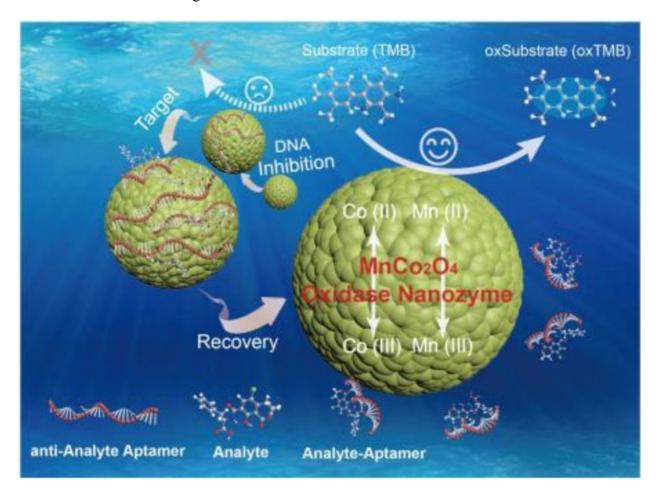


 Figure 8. Schematic of using MnCo₂O₄ in the structure of colorimetric aptasensor for OTA determination. Reproduced with permission from ref ¹¹¹. Copyright Elsevier Science, 2018.

The distribution of cations among the octahedral and tetrahedral sites in the crystal structure can introduce inverse spinel structure, with a formula of $B(AB)O_4$, which demonstrates different electronic and magnetic features compared to the spinel structure 112. Most recently, Liu and colleagues 113, improved the performance of anti-spinel structure by using Au and Pt in the structure of Fe₃O₄, through the ionic liquid (IL) as the cross-linker, for capturing synergistic interaction between the alloy atoms. In this work, the surface of designed nanozyme was modified with complementary DNA for conjugating with stainless steel mesh-aptamer. In the presence of the OTA, the binding of the target with aptamer caused to separate signaling probe (AuPt@IL@Fe₃O₄) and this, in turn, was used in a tube containing a tube with H₂O₂ and TMB for observe the color change. All in all, the inverse spinel and spinel structures introduce the effective bioreceptor immobilization and high catalytic activity, increasing the performance of biosensors. However, the limited surface area and non-optimal electronic properties of spinel structure can restrict their application. In terms of inverse spinel, although magnetic features and high catalytic activity improve their performance, a complex synthesis processes can limit their nanozyme-based application.

Along with these structures, tetragonal crystals and nanocubes structures are other structures used in the development of nanozyme for OTA detection. To illustrate, tetragonal crystal systems have been widely exploited in biosensors based on nanozyme owing to several advantages such as high surface area to volume ratio, stability, electronic features, high catalytic activity, and versatile functionalization ¹¹⁴. Currently, one of the excellent examples of using nanozyme-based tetragonal crystal system for detection of OTA was developed by Tian and co-workers ¹¹⁵. In this protocol,

the principle of this study was measuring the oxidase-mimicking activity of MnO₂ nanosheets. For this purpose, the decoration specific aptamer on the surface of aptamer was exploited for capturing OTA, leading to the production of the alkaline phosphatase-modified complementary DNA and the cascade reaction is triggered by the product (ascorbic acid) of alkaline phosphatase catalysis. The ascorbic acid reduced the oxidase-mimicking activity of MnO₂ nanosheets, leading to a pale color of the enzyme catalytic substrate. Another structure is layer structure, with unique benefits, which can provide efficient nanozymes in quantification of OTA. In 2020, Zhu et al. 116, fabricated a novel colorimetric immunosensor based on cobalt hydroxide nanoparticle (Cu₂O) nanocubes, with specific properties such as high surface area and high catalytic activity, for OTA quantification (Figure 9). In detail, the decoration microwell plate with dopamine was modified with OTA and ab₁ and, followed, Co(OH)₂-Ab₂ bounded to the prepared substrate. Under optimal circumstances, the difference of the color changes of TMB from Co(OH)₂ nanocage in the absence of H₂O₂ could present an efficient biosensing platform with a linear range and detection limit of 0.5 ng/L to 5 µg/L and 0.26 ng/L, respectively. Despite the high surface area and catalytic activity of nanocubes structure, the stability issue in some conditions and aggregation of cubes be considered important disadvantages of this structure. In addition, tetragonal crystal systems suffer from limited surface area.

Figure 9. Illustration of colorimetric immunoassay based on Cu₂O nanocubes for detection of OTA. Reproduced with permission from ref ¹¹⁶. Copyright Elsevier Science, 2020.

4.2. Marine toxins detection

Marine toxins, as poisonous substances, can be considered as natural metabolites which are produced by numerous such as bacteria, algae, and many marine invertebrates. Among them, algal toxins (including ciguatera toxin, domoic acid, and saxitoxin), invertebrate toxins, and some

 bacterial toxins may concentrate on various organisms through the food web. The negative consequences of these toxins in both humans and animals are undeniable 117, 118. Importantly, various bioreceptors/receptors can improve the performance of nanozyme-based colorimetric biosensors for determination of marine toxins. Antibodies, as one of the important bioreceptors, have been used in antibody-antigen interaction for presenting sensitivity and specificity detection approaches of marine toxins. Interestingly, the integration of nanozyme in the structure of colorimetric immunosensors can improve analytical signal owing to the catalyzation of enzymatic reaction by nanozyme, which increases the color intensity. For example, Hendrickson and coworkers 119, conjugated Au@Pt nanozyme with an antibody for enhancing label of okadaic acid quantification. The tendency of Au@Pt nanozyme to conjugate with anti-mouse antibodies, rather than anti-okadaic acid antibodies, could provide an excellent situation for an indirect competitive immunoassay format. This phenomenon led to unproductive immune binding without signal change is excluded, resulting in improving sensitivity of immunoassay. When the target was added to the system, the okadaic acid competed with the okadaic acid on conjugate pad for binding with anti-okadaic acid antibodies and were immobilized on the T line of LFA. The lack of chemical and thermal stability of antibodies, in harsh environmental circumnutates, can make them sensitive to degradation and denaturation. Furthermore, the high cost of production and purification of these bioreceptors can be considered another significant issue. In this regard, competitive colorimetric was implanted in aptasensors based on AuNPs nanozyme for sensitive and selective quantification of saxitoxin ¹²⁰. In this study, the competition of saxitoxin in samples with immobilization saxitoxin was a principle of the developed aptassay. To illustrate, in the presence of the target, the separation of specific aptamer from magnetic particles was conducted

and this, in turn, led to triggering the hybridization chain reaction. The colorimetric signal was

 improved by amplifying catalytic activity of the AuNPs nanozymes. Similarly, in many colorimetric aptasensor studies, the adsorption of aptamer could promote the enzyme-like activity of AuNPs for saxitoxin determination ^{121, 122}. Elaborately, this phenomenon could enhance surface negativity of nanozymes which increased adsorption and diffusion of positively charged substrates of TMB, improving of catalytic efficiency. In 2022, Li and colleagues ¹²³, reported a novel aptasensor exploiting AuNPs@Fe²⁺ for multiple diarrheic shellfish poisons detection (Figure 10). Indeed, the performance of nanozyme, in terms of chemical stability and peroxidase-like activity, was improved by using Fe²⁺ in the structure of AuNPs. Furthermore, the high affinity of terminalfixed aptamer (TF-DSP) was used in this study for the simultaneous detection of three diarrheic shellfish poisons. Under normal circumstances, the catalyze of TMB/H₂O₂/acetic acid due to the excellent peroxidase-like activity and the brilliant selectivity of aptamer could introduce a biosensing platform with a linear range and LOD of 0.4688–7.5 nM and 86.28 pM, respectively.

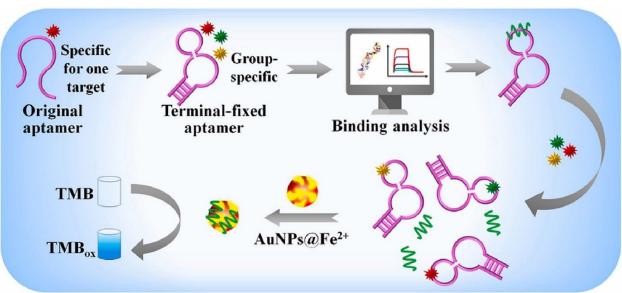


Figure 10. Schematic of colorimetric aptassay using decoration AuNPs with Fe⁺² for multiple diarrheic shellfish poisons detection. Reproduced with permission from ref ¹²³. Copyright Elsevier Science, 2022.

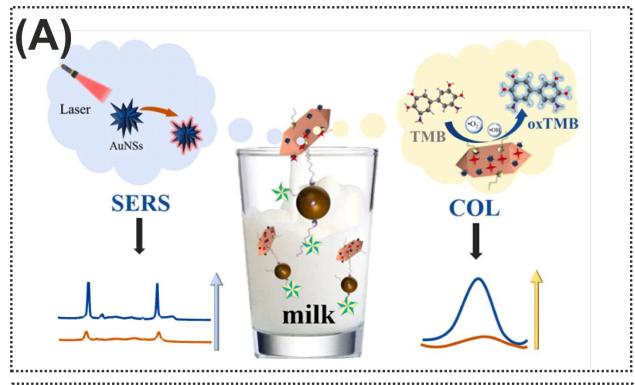
 The complex immobilization process on the surface of nanozyme and the side effect of potential interferences, in real samples, can considered the most important limitation of using aptamers in the nanozyme structure. In addition, in terms of overcoming stability issues of combination of antibodies with nanozyme, molecularly imprinted polymer (MIP), as artificial antibody and antigen systems, attracted considerable attention for marine toxins detection ^{124, 125}. Most recently, Wu et al. ¹²⁶, integrated a MIP with Au-Pt nanoparticles modified Fe₃O₄ magnetic nanozymes for introducing an efficient biosensor of saxitoxin. For this purpose, Au-Pt nanoparticles were loaded in Fe₃O₄ magnetic and, subsequently, thanks to the hydrolysis polymerization reaction, the decoration modified Fe₃O₄ magnetic nanozymes with MIPs were prepared. In the presence of the target, catalyzed the oxidation of TMB enabled the developed biosensing approach for detection saxitoxin with a detection limit of 3.1 nM. Potential interference in signal transduction, durability, and stability issues can be supposed most important limitation of MIPs. In order to address these limitations, scholars must pay special attention to the fabrication optimized and compatible MIPnanozyme platforms. For instance, Cho and colleagues ¹²⁷, used peptide (as both the imprinting template and the signal peptide), instead of antigen or aptamer, for designing competitive colorimetric quantification of saxitoxin. In this light, the exploitation of specific peptides of saxitoxin could overcome difficulty of aptamers and antibodies removal in biosensors based on MIP-nanozyme. In addition, the integration MIP with specific peptide of saxitoxin was measured with AuNP/Co₃O₄@Mg/Al. In the presence of the target, the less specific peptide of saxitoxin was conjugated with MIP, leading to intensity of color change.

4.3. Bacterial food toxins

Nowadays, bacteria food toxin, which is macromolecule mainly of protein origin, is one the main issues in the realm of food safety. These microorganisms can produce toxins in food or once the

 pathogen has colonized the digestive tract. These types of toxins damage in a specific organ of the host. To illustrate, such toxins cause foodborne diseases including vomiting, nausea, abdominal cramps, and diarrhea. ^{128, 129}. Despite the fact that the field of research about detection bacterial toxins is very wide, researchers attempted to use different nanomaterials in the structure of nanozymes for amplification of detection signal. Various forms of Au-based nanomaterials such as AuNPs gold nanostars (AuNSs) have been exploited as one of the important nanomaterials to amplify the signal due to several benefits such as excellent stability, repeatability, and accuracy. These nanomaterials gained great attention due to their ability to couple and integrate with different bioreceptors and nanomaterials for acting as a nanozyme through chemical bonds and electrostatic adsorption. As for labelling nanomaterials, Ren et al. 130, used the seed growth method for assembling AuNSs in the structure of Mn/Fe-MIL(53) for introducing an efficient nanozyme in colorimetric and SERS detection of Shiga toxin type II. As shown in Figure 11A, in terms of construction signal probe which could oxidize colorless TMB into blue color oxTMB, the surface of Mn/Fe-MIL(53) was decorated with AuNSs for providing an excellent for immobilization of SH-complementary DNA, through Au-S bonding. In addition, in favor of capturing probe, streptavidin-labeled magnetic beads were modified with a specific biotin-labeled aptamer. In the presence of the target, the conjugation of aptamer with target the release of signal probes from the complex and bring an enhancement of SERS and colorimetric signals. In the reported dual-mode, the integration of SERS and colorimetric approaches introduced complementary and validated results, improving the reliability of the Shiga toxin type II detection. In another example of Aubased nanozymes, Liang and co-workers 131, designed a novel dual-signal probe (AuPt nanoparticle-loaded single atom nanocomposite, AuPt@Fe-N-C) for Staphylococcus aureus enterotoxin B quantification (Figure 11B). For this purpose, the ab₁ was immobilized on the

 surface of AuPt@Fe-N-C and this was used for sandwich structure. In addition to the target to the system, the ab₁-Staphylococcus aureus enterotoxin B-ab₂ was performed in a 96-well plate which could oxidation TMB from colorless to blue with a detection limit of 0.066 7 pg/mL. Along with Au-based nanozyme, the unique properties and structure of rhodium (Rh), a non-toxic transition metal, was exploited for staphylococcal enterotoxin B determination in milk samples ¹³². The implementation of Rh, with peroxidase-like catalytic activity, on the sensing zone of LFA could able to detect staphylococcal enterotoxin B with a detection limit of as low as 1.2 pg/mL.



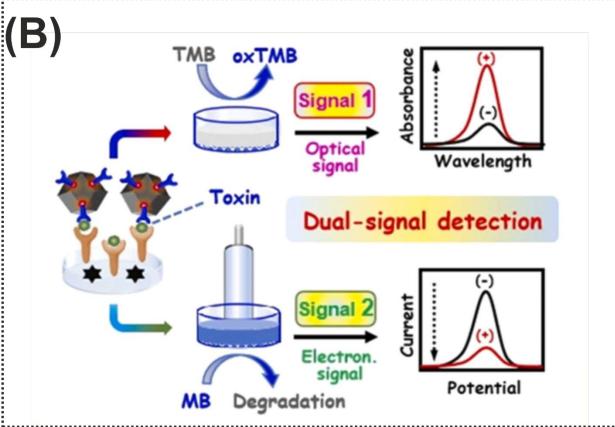


Figure 11. (A) Representation of dual-mode colorimetric and SERS biosensor based on Mn/Fe-MIL(53)@AuNSs for Shiga toxin type II detection. Reproduced with permission from ref ¹³⁰ (B) Illustration of dual-signal electrochemical and colorimetric detection of *Staphylococcus aureus* enterotoxin B ¹³¹. Copyright Elsevier Science, 2024.

5. Theranostic applications of nanozymes for biotoxins

Nanozymes catalytic theranostics introduces an innovative and novel strategy for theranostic platforms, integrating detection and treatment in a single system ¹³³. The importance of these platforms is highlighted in preventing the spread of biotoxins in contamination outbreaks by intervention to neutralize biotoxins and monitoring them ¹³⁴. In favor of the detection of biotoxins, catalytic color changing reaction conducts for achieving sensitive colorimetric sensors. In terms of naturalization and degradation of biotoxins, the oxidation and hydrolysis of biotoxins break these toxins by-products chemical down into non-toxic components by using nanozymes is an efficient strategy for rendering them harmless ^{135, 136}. Numerous techniques were exploited for biotoxins degradation such as UV method for marine biotoxins, demonstrating resistance to photodegradation, On the other hand, high performance degradation with UV/S2O82- and UV/H₂O₂ was achieved ^{137, 138}. In addition, biodegrading ATX-a into a nontoxic byproduct by Bacillus strains such as Bacillus flexus SSZ01 and Bacillus strain AMRI-03 can supposed high performance method for water treatment like rapid degradation of saxitoxins, which is directly and indirectly associated with food safety ¹³⁹. Furthermore, MOF-derived nanozymes was used as a high potential materials in the neutralization process ¹⁴⁰. Future research should be focused on the theranostic application of nanozymes for biotoxins due to a lack of research studies in this field. In other words, this field can advance biosensors and detoxification systems, managing biotoxin threats in different environments and food matrices.

Table 1. Comparison reported of colorimetric sensors with nanozymes for detection of different biotoxins.

Nanozymes	Enzyme activity	Substrate	Target	Linear range	LOD	Ref.
Au@HgNPs	OXD	TMB	AFB_1	0.125 to 87.5 μg/L	0.08 μg/L	91
CHNPs	OXD	TMB	AFB_1	1 pg/mL to 20 ng/mL	0.73 pg/mL	92
Au-Pt	OXD	TMB	AFB_1	0.1 to 500 ng/mL	5 pg/mL	93
MoS ₂ /Au	POD	TMB	AFB_1	1 to 100 ng/mL	0.25 ng/mL	141
Pt-CN	POD	TMB	AFB_1	1.0 pg/mL to 10 ng/mL	0.22 pg/mL	95
Ag@Au IP6	POD	TMB	AFB_1	2 pg/L to 200 pg/L	0.58 pg/L	98
MnO ₂ NSs	OXD	TMB	AFB_1	0.01 to 150 ng/mL	0.015 ng/mL	99
CuCo@PDA	POD	TMB	AFB_1	0.01 to 50 ng/mL	2.2 pg/mL	101
FeNiNPs	POD	TMB	AFB_1	0.12 to 2 μg/mL	36.57 ng/mL	102
Pt@PCN-222	OXD	ABTS	AFB_1	0.1 to 10 ng/mL	0.074 μg/L	142
NanoPCN-223	POD	TMB	AFB_1	0.05 to 10 ng/mL	0.003 ng/mL	106
MnCo ₂ O ₄	OXD	TMB	OTA	0.1 to 10 ng/mL	0.08 ng/mL	111
Fe ₃ O ₄	POD	TMB	OTA	5 to 100 ng/mL	0.078 ng/mL	113
MnO ₂ NSs	OXD	TMB	OTA	1.25 to 250 nM	0.069 nM	115
Cu ₂ O nanocubes	OXD	TMB	OTA	0.5 ng/L to 5 mg/L	0.26 ng/L	116
Au@Pt	POD	TMB	Okadaic acid	0.8 to 6.8 ng/L	0.5 ng/L	119
AuNPs	POD	TMB	Saxitoxin	78.13 to 2500 pM	42.46 pM	120
AuNPs@Fe ²⁺	POD	TMB	Okadaic acid, dinophysistoxin-1, and dinophysistoxin-2	4688 to 7.5 nM	86.28 pM	123
Fe ₃ O ₄ @Au-Pt	OXD	TMB	Saxitoxin	0.01 to 100 μM	3.1 nM	126
AuNP/Co ₃ O ₄ @Mg/A l cLDH	POD	TMB	Saxitoxin	0 to 1000 ng/mL	3.17 ng/mL	127
Mn/Fe- MIL(53)@AuNSs	POD	TMB	Shiga toxin type II	0.05 to 500 ng/mL	26 pg/mL	130
AuPt@Fe-N-C	POD	TMB	Staphylococcus aureus enterotoxin B	0.0002 to 10.000 ng/mL	0667 pg/mL	131
Rh	POD	TMB	Staphylococcal enterotoxin B	0 to 2 ng/mL	1.2 pg/mL	132

6. Conclusions and future perspectives

Numerous attempts have been undertaken to control biotoxins; however, their contamination remains largely inevitable. Therefore, there is an urgent need for research to explore the biosensors for the rapid, sensitive, and convenient detection of biotoxins, given the detrimental impact of biotoxins on human health and food safety. So, as described in this literature update, owing to their favorable catalytic activity, excellent stability, and low cost, nanozymes and nanozyme-based biosensors have been extensively exploited to identify different biotoxins in food and environmental samples. In particular, the combination of colorimetric biosensors and nanozymes to construct an innovative biosensing scaffold provides a promising outlook in convenient, sensitive, and rapid quantification of various biotoxins. Herein, we discuss the production, characteristics, and application of nanozymes in colorimetric biosensors, focusing on their diverse catalytic activities. Furthermore, a comprehensive overview of the research conducted on the utilization of nanozyme-based colorimetric biosensors for the identification of biotoxins is provided. Through deliberating the sensing approaches employed by nanozyme-based colorimetric biosensors, it becomes evident that these biosensors serve crucial roles in the rapid diagnosis of biotoxins. Besides, we highlight the recent advancements in portable technologies, including hydrogels and paper-based platforms, that can be integrated with smartphones to enable on-site detection. Although nanozyme-based colorimetric biosensors possess the features of fast response, simple operation, high sensitivity, and low cost, they still have some challenges in several aspects. i) Due to the inferior catalytic efficiency of nanozymes in comparison with natural enzymes, the application of nanozymes with multiple catalytic and high activity can be considered one of the efficient strategies. Indeed, multiple catalytic including peroxidase-like, oxidase-like, and catalase-like activities can improve the performance of nanozymes. Furthermore, the concept of

 high activate nanozymes is achieved by integration of nanozymes with different nanomaterials. As well, the lower selectivity of nanozymes towards targets restricts specific recognition in the detection process. Consequently, the development of novel nanozymes with enhanced specificity could involve the integration of biomimetic recognition elements, such as MIPs, and exploring additional technical approaches holds promise for future advancements in this area.

ii) Currently, an increasing number of studies are focusing on nanozymes exhibiting multiple enzyme activities. However, the utilization of nanozymes with multiple enzyme performances in colorimetric biosensors predominantly depends on their POD and OXD-like activities. Regulating the predominant activity of nanozymes with multiple enzyme-like functions simultaneously poses a significant challenge. This task necessitates a more precise comprehension of the catalytic mechanisms underlying various enzyme activities, along with a thorough understanding of the primary factors influencing these activities.

iii) The catalytic reactions occur in the specific regions on the nanozyme's surface which are

considered as active places. Generally, these active places are similarly operated based on the active sites in natural enzymes. However, their function and structure can differ owing to the basic differences between biological macromolecules and nanomaterials ¹⁴³. Elaborately, the presence of a small number of amino acids can able natural enzymes to directly interact with the substrate. Hydrophobic interactions and hydrogen bonding are the most important of these interactions, providing the high efficiency and specificity substrate. On the other hand, specific atoms or clusters of atoms on the surface of nanozymes act as active places. In detail, specific functional groups or metal ions can mimic the catalytic functions of natural enzymes. In addition, the chemical composition, size, and shape of nanozymes can impact on their catalytic activity ¹⁴⁴.

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59 60 To sum up, the investigation of nanozymes and nanozyme-based colorimetric represents merely the initial phase of a vast field of study. Nevertheless, it is evident that these biosensors exhibit significant promise in applications and merit further investigation. As research on progresses and nanozyme-based colorimetric biosensors continue to evolve, it is anticipated that more cutting-edge technologies and portable devices will be developed and extensively employed to safeguard food and environmental integrity.

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