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Gold-catalyzed formation of indole derivatives from 2-alkynyl arylazides and oxygen-containing heterocycles[†]

Received 15th April 2016 Accepted 2nd June 2016 Xiaoxiang Zhang,*^a Xiaoping Sun,^a Hui Fan,^b Chang Lyu,^a Ping Li,^a Haifei Zhang^a and Weidong Rao*^a

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A novel method for the formation of indole derivatives *via* goldcatalyzed tandem reactions of 2-alkynyl arylazides and oxygencontaining heterocycles has been developed. A variety of indole derivatives were prepared under mild reaction conditions.

Indoles are very important core structures in organic chemistry, which can be found in many natural products and bioactive compounds.¹ They are also serve as versatile building blocks for functional materials.² For this reason, the development of a novel synthetic methodology to this class of N-containing heterocycles still remains highly active.^{3,4} Recently, an increasing number of studies have demonstrated the use of 2-



Scheme 1 Our design for indole synthesis.

"Jiangsu Key Lab of Biomass-Based Green Fuels and Chemicals, College of Chemical Engineering, Nanjing Forestry University, Nanjing 210037, China. E-mail: zhangxiaoxiang@njfu.edu.cn; weidong@njfu.edu.cn

^bSchool of Petrochemical Engineering, Changzhou University, Changzhou 213164, China

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alkynyl arylazides as efficient substrates in gold-catalyzed indole formation strategies.⁵ In our continuing studies on gold-catalyzed tandem reactions,⁶ we recently reported an efficient and convenient synthetic method to 1*H*-indol-3-yl esters *via* gold-catalyzed tandem reactions of 2-alkynyl arylazides with carboxylic acids (Scheme 1a).⁷ Based on these earlier studies, we reasoned 1-azido-2-(phenylethynyl)benzene **1a** might also undergo a similar intermolecular trapping of putative α -imino





^{*a*} Unless stated otherwise, all reactions were performed in 0.5 mL 1,4dioxane with a catalyst/**1a**/MsOH ratio of 0.02 : 1 : 3. ^{*b*} Isolated yield. ^{*c*} Mixture of unknown side products afforded based on ¹H NMR analysis of the crude mixture. ^{*d*} No reaction was observed. Only starting material left. ^{*c*} Reaction conducted without catalyst. ^{*f*} Reaction conducted without MsOH. ^{*g*} *p*-TsOH·H₂O used as acid additive. ^{*h*} TFA used as acid additive. ^{*i*} TfOH used as acid additive.

Table 2 Gold-catalyzed ring-opening reactions of 2-alkynyl arylazides 1 with $2^{a,b}$



^{*a*} Unless stated otherwise, all reactions were performed at 60 °C in 0.5 mL 1,4-dioxane with a catalyst/1/MsOH ratio of 0.02 : 1 : 3. ^{*b*} Values noted in parentheses denote isolated product yields and reaction times. ^{*c*} Unknown compounds based on ¹H NMR spectroscopic analysis of the crude mixture.

gold carbene species^{5,8,9} generarated *in situ* by methanesulfonic acid. We anticipated the desired 2-phenyl-1*H*-indol-3-yl methanesulfonate **4a** would be afforded under the similar gold-

catalyzed reaction conditions, which could be a potential aryl mesylate cross-coupling reagent.¹⁰ In this reaction, 1,4-dioxane was used as solvent to dilute the methanesulfonic acid due to its

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highly activity. To our surprising, we did not detect the desired product **4a** by analyzing the TLC and crud ¹H NMR spectrum. However, only the unexpected ring-opening product 2-(2-((2-phenyl-1*H*-indol-3-yl)oxy)ethoxy)ethyl methanesulfonate **3a** was obtained in 95% yield (Scheme 1b).¹¹ Herein, we disclose the details of this efficient gold-catalyzed ring-opening reactions of 2-alkynyl arylazides with oxygen-containing heterocycles in the presence of methanesulfonic acid for the synthesis of novel indole derivatives.

Initially, we chose to focus our attention on ring-opening reaction of 1-azido-2-(phenylethynyl)benzene 1a and 1,4dioxane in the presence of 3 equivalents of methanesulfonic acid by a variety of gold catalysts to establish the reaction conditions (Table 1). This revealed treating a solution of reaction containing 1a (1 equiv.) and 2a (0.5 mL) with 2 mol% tBuXPhosAuNTf₂ in the presence of 3 equivalents of MsOH at 60 °C for 1 h gave the best result (entry 1). Under these conditions, 2-(2-((2-phenyl-1H-indol-3-yl)oxy)ethoxy)ethyl methanesulfonate 3a was obtained in 95% yield. On the other hand, a slightly lower product yields were observed when the gold catalyst was changed from tBuXPhosAuNTf₂ to PPh₃AuNTf₂, XPhosAuNTf₂ and IPrAuNTf₂ (entries 2-4). In contrast, the use of AuCl₃ as the catalyst led to a mixture of side products that could not be identified by ¹H NMR analysis (entry 5). Notably, only the starting material was recovered when JohnPhosAuCl or AgNTf₂ used as catalyst (entries 6-7). Similar result was obtained when this reaction was repeated in the absence of any gold catalyst (entry 8). The reaction temperature was also checked. High or low temperature led to a markedly lower yields of 59-71% (entries 9-10). Additional control experiment without MsOH led to a mixture of unknown products that could not be identified by TLC and ¹H NMR analysis (entry 11). In contrast, other acid additives were also examined. The corresponding product 2-(2-((2-phenyl-1H-indol-3-yl)oxy)ethoxy)ethyl 4-methylbenzenesulfonate 5 was obtained in 61% yield where the MsOH was replaced with p-TsOH·H₂O as the acid additive (entry 12). However, either no reaction or a mixture of unknown compounds were observed when TFA or TfOH was used as acid additive (entries 13-14).

To define the scope of the present procedure, we next turned our attentions to the reactions of a variety of 2-alkynyl arylazides 1 with oxygen-containing heterocycles 2 (Table 2). These experiments showed that with tBuXPhosAuNTf₂, 2-alkynyl arylazides bearing para-substituted electron-donating or electronwithdrawing substituents on the phenyl ring or thienyl group 1b-f efficiently underwent the tandem ring-opening process and gave the corresponding products 3b-f in good to excellent yields. We also tested the effect of an aliphatic substituent on the alkyne carbon and found that, for substrates where $R^2 =$ nC₅H₁₁ or nC₄H₉, lower yields 63% and 60% were obtained for 3g and 3i, respectively. Under the standard reaction conditions, the desired ring-opening indole 3 h was also provided with the sterically hindered substrate 1 h However, the substrate with cyclopropyl group 1j or terminal alkyne moiety 1k only gave a mixture of unknown side products that could not be identified by TLC and ¹H NMR analysis. Next, arylazides with Cl or Me group on the phenyl ring 3l-n also can provide the desired



Scheme 2 Proposed mechanism for the formation of 3a.

indole derivatives in moderate to excellent yields. Notably, the 5,7-dichloro-2-phenyl-1*H*-indol-3-yl methanesulfonate **4b** was obtained in 29% yield as a side product when the substrate with disubstituted group on the aromatic ring was employed. Next, other oxygen-containing heterocycles were also examined under the optimal reaction conditions. When dioxane was changed to tetrahydro-2*H*-pyran, we found that the corresponding ring-opening indole product **3o** was afforded in 73% yield. Very interesting, the reaction of **1a** with oxetane **2c**, tetrahydrofuran **2d**, or 2-phenyloxirane **2e** only gave a mixture of unknown side products that could not be identified by TLC and ¹H NMR analysis.

We tentatively propose the present $tBuXPhosAuNTf_2$ catalyzed ring-opening indole-forming reaction to proceed by the mechanism outlined in Scheme 2, although it is highly speculative. This could involve activation of 1-azido-2-(phenylethynyl) benzene **1a** through coordination of the gold catalyst with the alkyne, which would be attacked by the azide to deliver intermediate **A**. Subsequently, the α -imino gold carbene **B** was formed after the expulsion of N₂ with the assist of gold catalyst, which was further attacked by 1,4-dioxane to release the intermediate **C**. Ring opening of **C** with methanesulfonic acid would provide the complex **D**, which undergo a final protodemetalation step to deliver the 2-(2-((2-phenyl-1*H*-indol-3-yl) oxy)ethoxy)ethyl methanesulfonate **3a**.

In summary, an unexpected gold-catalyzed synthetic method to indole derivatives based on the ring-opening reaction of 2alkynyl arylazides with oxygen-containing heterocycles in the presence of methanesulfonic acid has been reported. These results show that the reaction tolerates a structurally diverse set of 2-alkynyl arylazides substrates that can be prepared easily in two steps from commercially available materials. Moreover, the gold-catalyzed tandem reactions were also shown to be practical since inert and moisture-free conditions were not required. Further studies on the applications of this synthetic method are currently underway in our laboratory.

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