




Cite this: *Environ. Sci.: Nano*, 2018, 5, 1117

Different effects of nano-scale and micro-scale zero-valent iron particles on planktonic microorganisms from natural reservoir water†

Nhung H. A. Nguyen,^{*ab} Roman Špánek,^{ab} Vojtěch Kasalický,^c David Ribas,^d Denisa Vlková,^a Hana Řeháková,^a Pavel Kejzlar^a and Alena Ševců ^{*ab}

While nano-scale and micro-scale zero-valent iron (nZVI and mZVI) particles show high potential for remediation of polluted soil aquifers and elimination of cyanobacterial blooms, this has required their release into the environment. This study compares the impact of 100 mg L⁻¹ of nZVI and mZVI on natural planktonic microorganisms from a reservoir, incubated in 1.5 L batches over 21 days. In addition to counting cyanobacterial and algal cell numbers, bacterial community structure was assessed using Ion Torrent sequencing and the number of cultivable bacteria determined using standard cultivation methods. Surprisingly, while mZVI had no significant effect on algal cell number, cyanobacteria numbers increased slightly after 14 days ($P < 0.05$). Algae were only marginally affected by nZVI after seven days ($P < 0.05$), while cyanobacteria numbers remained unaffected after 21 days. Total species richness and less common bacteria increased significantly when treated with mZVI (compared to nZVI). The abundance of *Limnohabitans* (*Betaproteobacteria*), *Roseiflexus* (*Chloroflexi*), hgcL-clade (*Actinobacteria*) and Comamonadaceae_unclassified (*Betaproteobacteria*) increased under nZVI treatment, while mZVI enhanced *Opiritutae_vadinHA64* (*Verrucomicrobia*) and the OPB35_soil_group (*Verrucomicrobia*). Interestingly, the number of cultivable bacteria increased significantly after three days in water with nZVI, and further still after seven days. nZVI shaped bacterial community both directly, through release of Fe(II)/Fe(III), and indirectly, through rapid oxygen consumption and establishment of reductive conditions. The strong physico-chemical changes caused by nZVI proved temporary; hence, it can be assumed that, under natural conditions in resilient reservoirs or lakes, microbial plankton would recover within days or weeks.

Received 24th November 2017,
Accepted 16th March 2018

DOI: 10.1039/c7en01120b

rscl.li/es-nano

Environmental significance

Nanoscale zerovalent iron (nZVI) has been the subject of great interest over the last decade as it is a highly reactive material, capable of rapidly assimilating toxic pollutants (e.g. chlorinated ethenes) in soil and underground waters. To date, however, most field studies have been performed in contaminated underground environments. As nZVI have been postulated as a possible treatment for harmful cyanobacterial blooms, this study focuses on the effect of nZVI on natural pelagic microplankton in a freshwater reservoir. Moreover, we aim to distinguish between the effect of nano- and micro-ZVI particles on the natural microbial community.

1. Introduction

Zero-valent iron (ZVI) has a long tradition of environmental application due to its high capacity to clean-up a wide range

of pollutants, including chlorinated organic compounds and toxic metal cations.^{1–3} Initially applied as granules (>50 μm), and more recently as a micro-scale ZVI (mZVI) material,⁴ ZVI is now more often applied as nano-scale ZVI particles (nZVI).^{5–9} Due to their high specific surface area, micro- and nano-scale materials are more reactive than granular materials, while their small particle size means they can be used for more diverse applications.¹⁰ In addition to the above-mentioned applications, a number of laboratory studies have suggested that nZVI could also prove useful for prevention of cyanobacterial blooms.^{11,12}

Unfortunately, the high capability of ZVI for cleaning-up pollutants in groundwater and soils^{7,13,14} may also have an

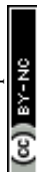
^a Institute for Nanomaterials, Advanced Technologies and Innovation, Technical University of Liberec (TUL), Studentská 2, 46117 Liberec, Czech Republic.
E-mail: nhung.nguyen@tul.cz, alena.sevcu@tul.cz

^b Faculty of Mechatronics, Informatics and Interdisciplinary Studies, Technical University of Liberec (TUL), Studentská 2, 46117 Liberec, Czech Republic

^c Biology Centre of the Czech Academy of Sciences, Institute of Hydrobiology, České Budějovice, Czech Republic

^d Department of Chemical Engineering, Technical University of Catalonia (UPC), ETSEIB, Av. Diagonal, 647, E-08028, Barcelona, Spain

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c7en01120b



effect on indigenous microorganisms mediating fundamental ecosystem processes as they are the first to be exposed to the particles. Hence, there is an urgent need for the evaluation of potential risks relating to ecosystem function changes when using ZVI materials.

Two main mechanisms of direct effect on microbial cells have been identified for ZVI: generation of reactive oxygen species (ROS) and the potential to physically damage bacterial or algal cells due to its strong affinity to cell surfaces,¹⁵ which can lead to a significant decrease in mobility.^{16,17} It is well known that ZVI particles have the potential to release Fe(II) into the surrounding environment, converting less reactive hydrogen peroxide (H₂O₂) into harmful ROS, e.g. hydroxyl radicals (·OH), via the Fenton reaction.^{18,19} ROS are normally scavenged by antioxidants and enzymes; however, elevated concentrations in microbial cells can result in oxidative stress and damage to lipids, proteins and DNA. In addition, the physical interaction of nZVI and microbial cells could cause damage to outer membranes or physical inactivation of cells.^{16,21} Indirect effects of nZVI on microorganisms include a decrease in dissolved oxygen concentration, along with a subsequent decrease in oxidation reduction potential,^{1,22} which could potentially harm aerobic microorganisms. On the other hand, ZVI oxidation or reduction processes could be beneficial for iron-oxidising bacteria.²³

A careful risk assessment of ZVI particles requires different ecotoxicity studies targeting organisms on different ecosystem levels. Consequently, the number of ecotoxicity studies focusing on microorganisms is increasing; however, these have mainly been limited to a single species grown in adequate nutrient media,^{16,24–30} a situation far from relevant to natural environment conditions. Despite this, such studies are important for better understanding more complex studies on microbial communities under natural conditions.

Generally speaking, all studies on ZVI effects to aquifer and soil bacterial communities describe shifts in community structure.^{9,15,31–34} The first toxicity study on a whole bacterial community from the River Thames showed no negative effect from ZVI particles (100 mg L⁻¹, nZVI majority of Fe(0), with 27% of boron on the surface, 30–90 nm; and mZVI 99.5% of Fe(0), 1.5–6.8 μm), the number of bacteria even increasing after 40 days.¹⁵ Similarly, no deleterious effect of nZVI (100 mg L⁻¹, 24–32% Fe(0) and 33% Fe(0) modified with polyaspartate) was observed on total bacterial abundance in a series of microcosm experiments³³ nor prolonged toxicity on dechlorinating microorganism during a 300 day anaerobic column experiment (Nanofer 25, 20% suspension of Fe(0) in water).³⁵ Recently, Němeček *et al.* (2014) showed that the long-term effect of nZVI (Nanofer 25, 20% suspension of Fe(0) in water) on indigenous prokaryotic microorganisms may be neutral, or even positive, as the reduction in toxic pollutants allows further biodegradation.³⁴ This finding was further supported by repeated *in situ* application of nZVI (first round Nanofer STAR, second round Nanofer 25S, 1 g L⁻¹ and 2 g L⁻¹ of 20% suspension in water, respectively) to combat

contamination by hexavalent chromium Cr(VI) and chlorinated ethenes, which resulted in significant stimulation of iron-reducing, sulphate-reducing and chloro-respiring bacterial growth over the 13-month study.⁹ Moreover, no evidence for any negative effect on soil microbial activity was observed when comparing nZVI (ball-milled Fe(0) in 0.18% polyacrylic acid) with mZVI (>98.3% Fe in 0.18% polyacrylic acid) toxicity.³⁶ On the other hand, high levels of mZVI (25 g L⁻¹, >90% Fe(0)) were seen to change anaerobic sludge bacterial community composition,³⁷ with *Flavobacteriaceae*, *Comamonadaceae* and several unclassified iron-reducing bacteria increasing in abundance, considered positive for degradation of toxic flame retardants.³⁷ Velimirovic *et al.* reported five types of mZVI as positively stimulating a CAH-degrading mixed bacterial culture at concentrations up to 500 mg L⁻¹, while 1–15 g L⁻¹ resulted in inhibition depending on mZVI size.³⁸ mZVI in general has a lower impact on microbes than nZVI, though comprehensive studies on natural freshwater communities have yet to be published.

Studies on impact of nZVI on algae or cyanobacteria are still scarce and are generally limited to single-species cultures, while on mZVI effect have yet to be undertaken. A cyanobacterial bloom, dominated by *Microcystis aeruginosa*, was strongly affected by nZVI (Nanofer 25S, >90% Fe(0)) after 24 h exposure (EC50: 50 mg L⁻¹), with the effect significantly higher than on *Desmodesmus subspicatus*.¹¹ Two different nZVI types, Nanofer 25S and Nanofer STAR (70–90% Fe(0) and 10–30% iron oxides), affected the growth rate of freshwater *Pseudokirchneriella subcapitata* at concentrations >8 mg L⁻¹ and ≥12 mg L⁻¹, respectively.¹⁹ Likewise, ball-milled nZVI (FerMEG12, 15–30% Fe(0) in monoethylene glycol) had a negative effect on *P. subcapitata* and *Chlamydomonas* sp. at concentrations <100 mg L⁻¹,^{30,39} while the monoethylene glycol dispersant alone had no effect.³⁰

The major objective of this study was to examine the effect of nZVI and mZVI particles on planktonic microorganisms. The study was carried out on natural freshwater from a reservoir, sampled in late summer in order to include a fully developed phytoplankton community. ZVI inevitably oxidises into Fe-oxides naturally present in the environment at concentrations ranging from 0.1 to 194.7 mg L⁻¹ in rivers⁴⁰ and from 20 to 40 g kg⁻¹ in soils.⁴¹ Hence, we used a concentration of 100 mg L⁻¹ as this was considered relevant to concentrations in the vicinity of ZVI application, where ZVI suspension quickly dilutes^{42–44} and as this is the limit for hazard labelling in EU chemical regulations. Centrifuged and filtered reservoir water was used when assessing the size distribution of both ZVI materials in order to reveal possible differences due to sample preparation. We describe the effect of nZVI and mZVI particles on prokaryotic (bacteria, cyanobacteria) and eukaryotic (algae) microorganisms by determining the number of cultivable bacteria and the number of cyanobacterial and algal cells. For the first time, we apply 16S rRNA amplicon sequencing in order to describe the effect of nZVI and mZVI on the development of a natural bacterial community.



2. Material and methods

2.1. ZVI particles

A suspension of nZVI in deionised water was obtained from the Polytechnic University of Catalonia, Spain. The material was produced using a novel wet-ball milling process utilising alumina abrasive and monoethylene glycol (MEG; Scharlab S. L.) as the grinding media. The nZVI particles had a mean diameter of 160 nm, a maximum size of 540 nm (determined by scanning electron microscopy [SEM]), a specific surface area of $29.6 \text{ m}^2 \text{ g}^{-1}$ and contained 74.0% Fe(0), 2.7% carbon and 23.3% of iron oxide.⁴⁵ The mZVI reactive powder (no: 72052910) was provided by Hepure Technologies (USA) and had a specific surface area of $0.487 \text{ m}^2 \text{ g}^{-1}$, 95% Fe(0), 1.8% carbon, <1% oxygen, 1% silicon, 0.1% phosphorus and 1% sulphur. A 10 g L^{-1} stock suspension of ZVI in deionised water was freshly prepared before the experiment. Images of the pristine particles are presented in Fig. S1.†

2.2. Experimental design and sampling of reservoir water

Water samples were obtained from Harcov reservoir (50.7702097N, 15.0755733E; Czech Republic) one metre from the reservoir bank and from a point 10 cm below the surface on 22 August 2016. The water was immediately transported in sterile containers (20 L) to the laboratory and the experiment started immediately after arrival. The following variants were prepared in triplicate in sterile 1.5 L glass bottles: fresh reservoir water mixed with stock suspension of either nZVI or mZVI at a final concentration of 100 mg L^{-1} and fresh reservoir water only (control without ZVI). Additionally, sterilised reservoir water and sterilised reservoir water mixed with nZVI were prepared in parallel as reference controls. The bottles, illuminated with two 3 metre light tubes emitting visible light at 670–800 lux (GOSSEN, Germany), were slowly mixed manually for one minute a day and the position of the bottles randomly changed each day. Sub-samples (160 mL) were taken after 0, 1, 3, 7, 14 and 21 days for determination of cultivable bacteria, either as colony forming units (CFU) or number of cyanobacterial and algal cells. Aliquots for DNA extraction and 16S rRNA amplicon sequencing were sampled during days 0, 7 and 21.

2.3. Physico-chemical analysis of reservoir water

The reservoir water (1 L) was sampled separately, in triplicate, for analysis of chemical parameters (Table S1†), determined using iron chromatography (IC; ThermoFisher) and inductively coupled plasma spectrometry (ICP-OES) on an OPTIMA 2100 DV (Perkin Elmer) spectrophotometer, and total organic carbon (TOC) using a multi N/C 2100S analyser (Analytik Jena, Germany).

Concentration of dissolved Fe at 0 and 21 days was measured using ICP-OES. Water samples with nZVI and mZVI (10 mL) were centrifuged for 30 minutes at 4600 rpm (ROTANTA 460R, Hettich, Germany), the supernatant then being passed through Amicon Ultracel 3K ultra-filtration filters with a 3

kDa molecular weight cut-off (Millipore, USA) to separate Fe ions from the particles.

The pH, ORP and dissolved oxygen concentration was measured directly at the reservoir and during the laboratory experiment using a Multi 3430 multi-parameter portable probe (WTW, Germany).

2.4. Characterisation of ZVI particles in reservoir water

The nZVI and mZVI particles (100 mg L^{-1}) were suspended for 24 h in (1) centrifuged reservoir water prepared by spinning for 30 minutes at 6000 rpm, (2) filtered reservoir water prepared by filtration over a $1.4 \mu\text{m}$ membrane filter (Whatman, Life Sciences) and (3) deionised water as a control. The nZVI and mZVI particles were characterised using a Zeiss Ultra Plus field-emission SEM (Zeiss, Germany). The samples were fixed onto aluminium stubs using double-sided carbon tape and cleaned with RF plasma (Evactron) for 10 min before SEM analysis. SEM images were acquired at an accelerating voltage of 5 kV at low probe current (about 15 pA) using an InLens secondary electron detector with SmartSEM software. Subsamples were further analysed for size distribution using differential centrifugal sedimentation (DCS) on a DC24000 centrifuge (CPS Instruments, UK) at a maximum operational speed of 2500 rpm for mZVI and 24 000 rpm for nZVI. In addition, colloidal particles of natural organic compounds were concentrated by centrifugation of filtered ($1.4 \mu\text{m}$) reservoir water for 45 minutes at 6000 rpm, with maximum operational speed set at 10 000 rpm. This protocol was adapted from Barnes *et al.* (2010).¹⁵

2.5. DNA extraction

Reservoir water samples (150 mL) treated with nZVI and mZVI and untreated reservoir water (control) were pre-filtered with $20 \mu\text{m}$ filter papers (Whatman, GE Healthcare Life Science, China) and then filtered through a Hydrophilic Durapore PVDF $0.22 \mu\text{m}$ membrane (Merck Millipore, Germany) for DNA extraction. Briefly, DNA was extracted from the filter using a FastDNA Spin Kit for Soil (MP Biomedicals, CA, USA) according to the manufacturer's protocol, with the Bead Blaster 24 homogenisation unit (Benchmark Scientific, NJ, USA) employed for cell lysis. Extracted DNA concentration was quantified using a Qubit 2.0 fluorometer (Life Technologies, USA) (further details in Dolinová *et al.*, 2016).⁴⁶

2.6. PCR amplification and next-generation sequencing (NGS)

All isolated DNA samples for bacterial and cyanobacterial community analysis were sequenced in duplicate, with two consecutive polymerase chain reactions (PCR) performed per sample to amplify DNA from the V4 region (normal and barcode fusion primers used). *In silico* analysis of primers was performed in order to cover as much diversity as possible while keeping the amplicon size below 400 bp. Amplification of the V4 region of the eubacterial 16S rRNA gene was performed with primers 515F ($5' \text{-TGCCAGCMGCNGCGG-3}'$)⁴⁷ and barcode 802R ($5' \text{-TACNVGGTATCTAATCC-3}'$).⁴⁸ A



MOCK community (collection of 6 bacterial genomes prepared in-house) was subsequently sequenced in order to verify data evaluation (Fig. S2†). PCR cycle conditions were as follows: first PCR 95 °C for 3 min; 10 cycles at 98 °C for 20 s, 50 °C for 15 s and 72 °C for 45 s, with a final extension at 72 °C for 1 min; second PCR 95 °C for 3 min; 35 cycles at 98 °C for 20 s, 50 °C for 15 s and 72 °C for 45 s, with a final extension at 72 °C for 1 min. The concentration of purified PCR products was measured with a Qubit 2.0 fluorometer (Life Technologies, USA). Barcode-tagged amplicons from different samples were then mixed in equimolar concentration. Sequencing of bacterial amplicons was performed on the Ion Torrent platform (Life Technologies, USA), the raw Ion Torrent reads being processed with Mothur software.⁴⁹ Low quality reads were removed and sequences were assigned to each sample. Chimeric sequences were identified using UCHIME⁵⁰ and subsequently removed. Sequences exceeding 400 bases were trimmed and sequences shorter than 180 bases removed. Sequences were classified against Silva database v. 123 with a bootstrap value set at 80%. A cut-off value of 97% was used for clustering sequences into operational taxonomic units (OTUs). Sequence data were normalised to group with the least sequences by randomly selecting a selected number of sequences from each sample. Cluster analysis by Bray–Curtis dissimilarity was undertaken using Vegan in the R statistical package.⁵¹

2.7. Abundance of ferroxidase enzyme

Ferroxidase enzyme (EC1.16.3.1) abundance in prokaryotes was conducted using metabolic inference (preferably, but not exclusively, using NGS) on 16S rRNA gene libraries using Paprica software,⁵² based on the phylogenetic placement approach. Paprica uses pathways/enzymes shared between the members of all clades on a reference tree to determine which pathways/enzymes are likely to be associated with a phylogenetically placed read. Paprica accepts reads pre-processed by Mothur software, allowing removal of low quality reads, short reads and contaminants (e.g. mitochondria, chloroplasts).

2.8. Determination of cultivable bacteria

The number of cultivable bacteria was determined in nZVI and mZVI treated reservoir water by counting CFUs in 1 mL of undiluted and 10-fold diluted samples spread on plate count agar (BIO-RAD, France) in sterile Petri dishes, with CFUs counted after seven days incubation at 20 ± 2 °C. Cultivable bacterial abundance is expressed as CFU per mL.

2.9. Microscopy analysis of phytoplankton

Determination of cell number and filament measurement was based on Czech standard methodologies (CSN 75 77 17). Disintegration of the algal and cyanobacterial colonies into individual cells was undertaken by addition of potassium hydroxide (KOH; 0.1 M), then sucking and releasing a 10 mL sample twenty times using a syringe. The cells were then counted in a Cyrus I chamber (0.01 mL) using an Olympus

BX43 fluorescence microscope (Olympus, Japan) with a blue excitation filter for algae and a green excitation filter for cyanobacteria.

2.10. Calculation and statistical analysis

For improved comparison, all data were converted into a fold change, calculated by dividing the numbers of cells at day 1, 3, 7, 14 and 21 (D_n) by the number of cells at day 0 (D_0). The differences between treatments and control were compared using ANOVA and Dunnett's test (GraphPad PRISM, USA). Significance levels were set at $*P < 0.05$, $**P < 0.01$ and $***P < 0.001$.

Three diversity metrics were used to compare the impact of ZVI particles on bacterial community composition, as described in Newton and McLellan (2015).⁵³ Species were described using richness, the most abundant taxa/groups within samples were described using the inverse Simpson index and the tail statistic was used to describe the diversity of rare community members. The Mann–Whitney U test was used to examine whether composition of bacterial community exposed to mZVI differed significantly from those exposed to nZVI.

3. Results

3.1. Characterisation of ZVI particles in reservoir water

nZVI and mZVI particles were exposed to two differently prepared reservoir water (centrifugation and filtration) samples with the aim of removing microorganisms and debris that disable characterisation of the particles. Particles exposed to deionised water were considered as a control. Both nZVI and mZVI tended to aggregate in centrifuged reservoir water (Fig. 1A and 2A) and filtered reservoir water (Fig. 1B and 2B), but not in deionised water (Fig. 1C and 2C). mZVI particles had a normal (Gaussian) distribution from 1 to 5 μm , with the maximum ranging from 3.8 to 4 μm in all samples of reservoir and deionised water, as determined by DCS. In contrast, nZVI particles were split into two 'populations', with a dominant peak around 110 nm in centrifuged reservoir water and 120 nm in filtered reservoir water. The smaller population always appeared around 600 nm in both variants. In deionised water, the larger population was detected at around 100 nm and the smaller around 380 nm (Fig. 1D; see SEM images of pristine particles in Fig. S1† for comparison). Compared to centrifuged reservoir water, colloidal particles of natural organic compounds in filtered reservoir water were completely missing at fractions < 60 nm, while relatively few particles were observed at > 60 nm (Fig. S3†).

3.2. Physical–chemical parameters of reservoir water

Temperature remained more or less constant (22.5 ± 0.2 °C) over the 21 day experiment, with the exception of day 0 (sampling in reservoir; 18 °C; Fig. 3A). While the initial pH was 8.0 in untreated reservoir water (control), addition of ZVI particles caused an immediate significant increase in reservoir



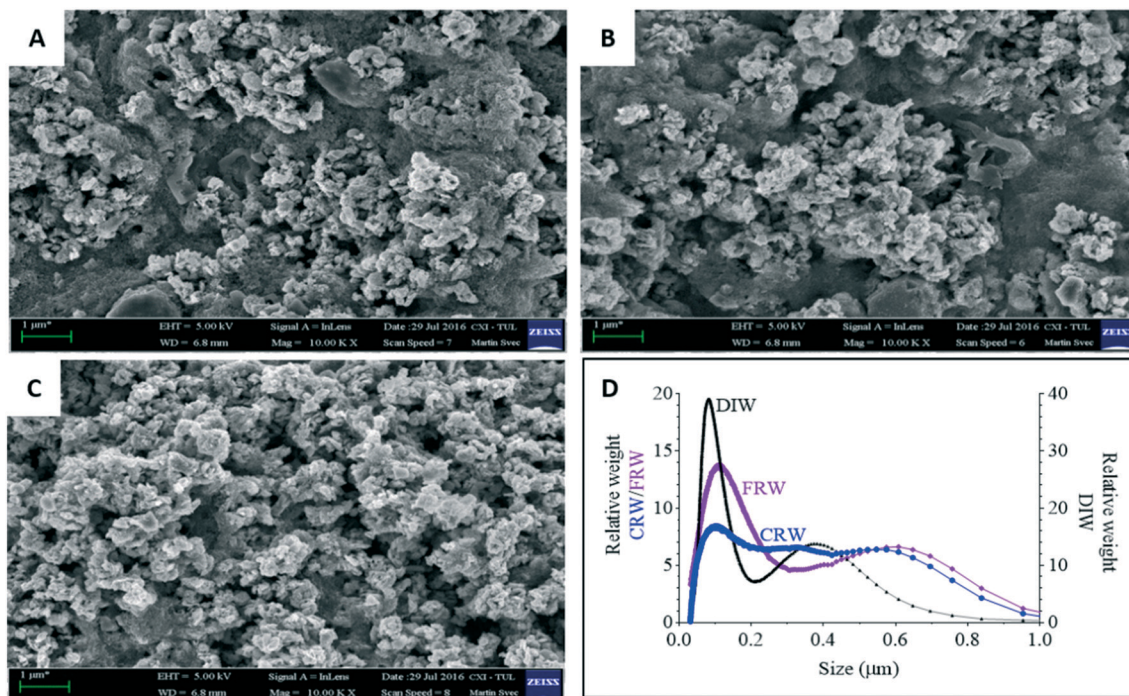


Fig. 1 Scanning electron microscopy images of nZVI exposed for 24 h in reservoir water: (A) nZVI in centrifuged reservoir water (CRW); (B) nZVI in filtered reservoir water (FRW); (C) nZVI in deionised water (DIW) used as a control; scale bar = 1 μm; (D) size distribution of nZVI in CRW, FRW and DIW determined by differential centrifugal sedimentation.

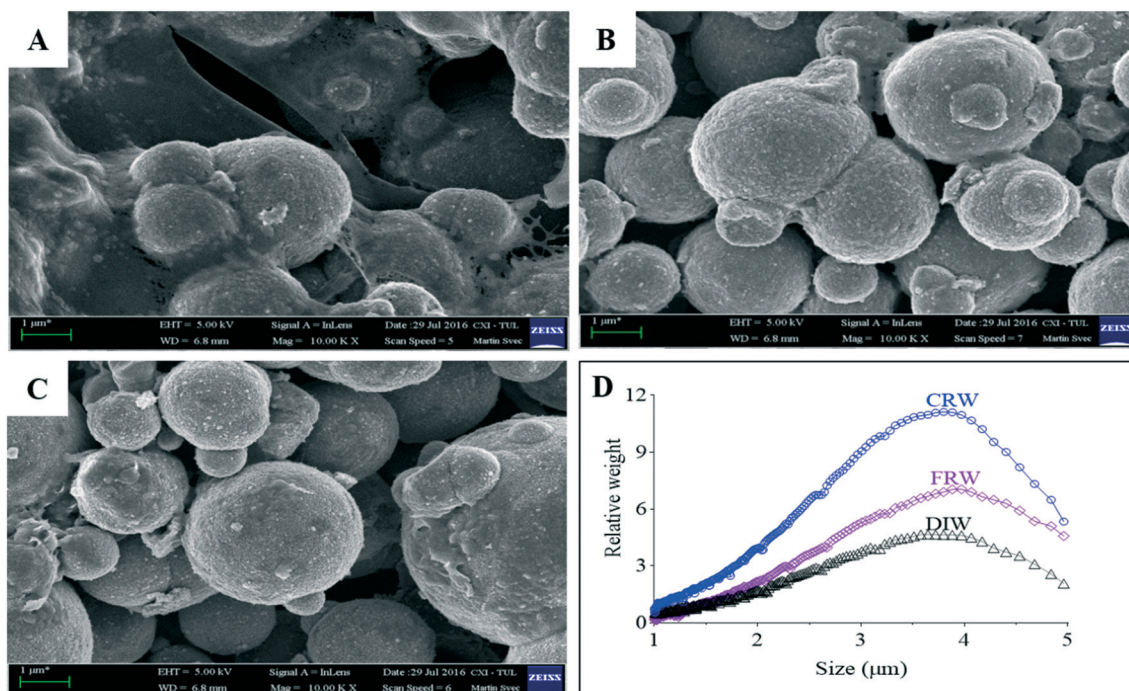


Fig. 2 Scanning electron microscopy images of mZVI exposed for 24 h in reservoir water: (A) mZVI in centrifuged reservoir water (CRW); (B) mZVI in filtered reservoir water (FRW); (C) mZVI in deionised water (DIW) used as a control; scale bar = 1 μm; (D) size distribution of mZVI in CRW, FRW and DIW determined by differential centrifugal sedimentation.

water treated with mZVI (pH 8.8; $P < 0.001$) and nZVI (8.5; $P < 0.01$). The pH then decreased slightly to 7.6 (reservoir water), 8.2 (nZVI) and 8.3 (mZVI), but remained significantly

higher than the control after one day and thereupon remained between 7.2 and 7.6 (with or without ZVI; not significant) until the end of the experiment (Fig. 3B). The ORP



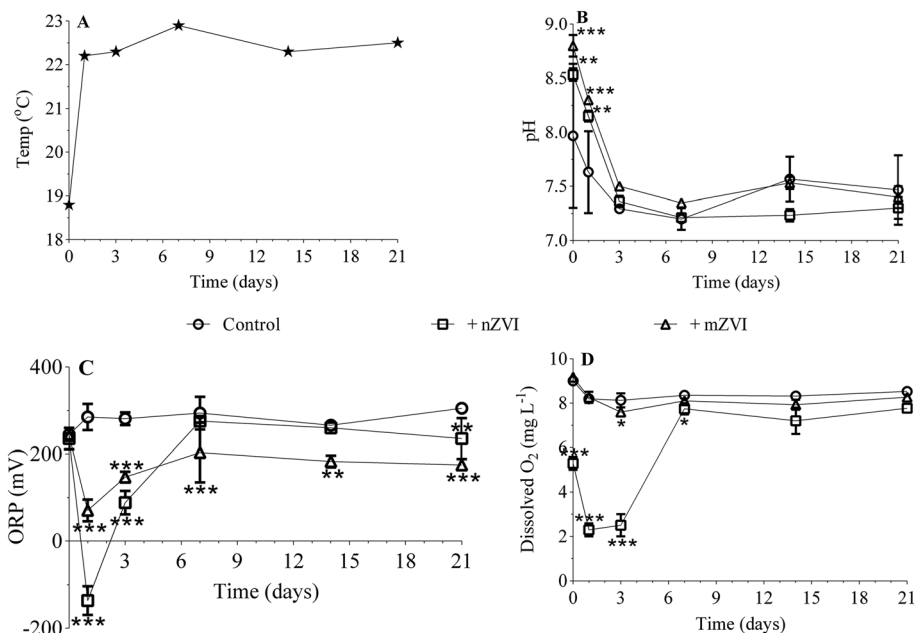


Fig. 3 Physio-chemical parameters monitored over the 21 day experiment with mZVI and nZVI in reservoir water. (A) Temperature, (B) pH, (C) oxidative reductive potential and (D) dissolved oxygen. Error bar $n = 3$. Statistically significant differences (* $P < 0.05$, ** $P < 0.01$ and *** $P < 0.001$) compare the treatment to the control.

of reservoir water was 245 mV at the beginning of the experiment and fluctuated around 300 mV until the end (Fig. 3C). The ORP of reservoir water with nZVI decreased to -136 mV ($P < 0.001$) and to 70 mV ($P < 0.001$) with mZVI after one

day. ORP gradually increased to 88 mV with nZVI ($P < 0.001$) and 148 mV with mZVI reservoir water ($P < 0.001$) after three days, after which there was no significant difference from the control with nZVI while mZVI ORP remained lower (175–200

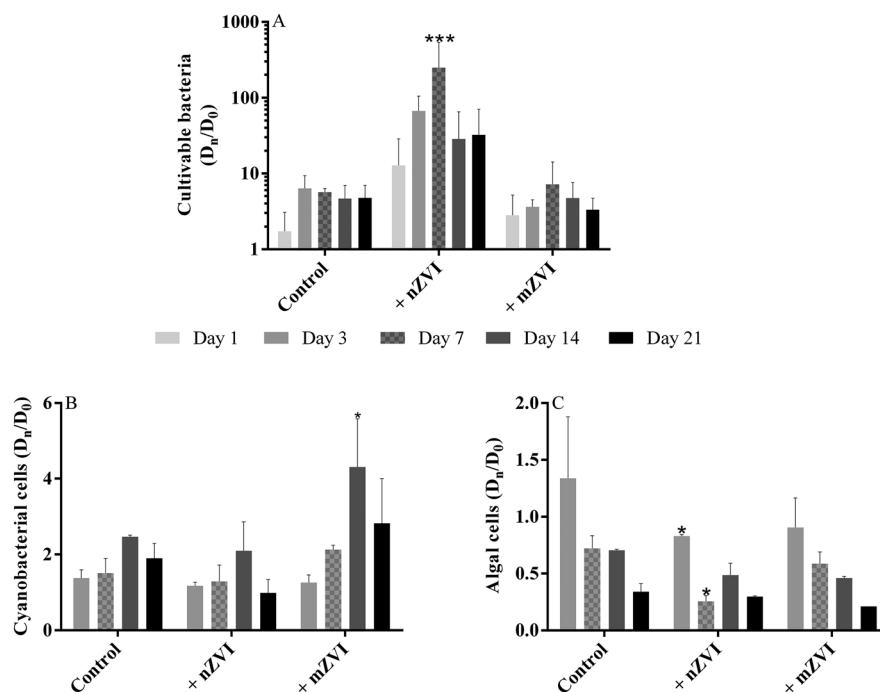


Fig. 4 Abundance of planktonic microorganisms (fold change of D_n/D_0) over the 21 day experiment using reservoir water with nZVI (+ nZVI), reservoir water with mZVI (+ mZVI), and reservoir water only as a control); (A) cultivable bacteria, (B) cyanobacterial number and (C) green algae number. Note: different scales for the y axis. Error bars $n = 3$. Statistically significant differences (* $P < 0.05$, ** $P < 0.01$ and *** $P < 0.001$) compare the treatment to the control. The actual number of cells is provided in the ESI† (Table S2).



mV, $P < 0.01$) over the next 21 days (Fig. 3C). Dissolved oxygen in the control and mZVI sample was around 9 mg L^{-1} at the beginning of the experiment and decreased slightly to $\sim 8 \text{ mg L}^{-1}$ over 21 days. In comparison, nZVI caused a decline to 5.3 mg L^{-1} at day 0, declined further to 2.3 and 2.5 mg L^{-1} after one day and after three days ($P < 0.001$), respectively, then increased to 7.8 mg L^{-1} ($P < 0.05$) after seven days; this situation lasting until the end of the experiment (Fig. 3D). A reddish-brown colour appeared in bottles containing nZVI from day 2, demonstrating presence of oxidised Fe. The reddish colour was observed in reservoir water with both nZVI and mZVI at the end the experiment (Fig. S4†).

3.3. Effect of mZVI and nZVI on planktonic microorganisms

Compared to the untreated control samples, the number of cultivable bacteria in the nZVI treatment increased geometrically, with a maximum at day 7 yielding $171 \times 10^3 \text{ CFU mL}^{-1}$ ($P < 0.001$), but then decreased slowly up to day 21, though still remaining about 3–4 times higher than the control (Fig. 4A, Table S2A†). Both the control and the mZVI treatments displayed a similar number of CFU with no significant difference.

Changes in the number of cyanobacterial cells were recorded in the presence of ZVI particles (Fig. 4B; Table S2B†), with numbers increasing over time with mZVI (significantly higher than the control at day 14; $P < 0.05$) but remaining comparable to the control with nZVI. The dominant cyanobacterial species at the beginning of the experiment was the colony-forming *Microcystis aeruginosa*, followed by *M. flos-aquae* and *Woronichinia naegeliana* at lower abundance, and the genus *Anabaena* sp. rarely recorded. *M. aeruginosa* was still dominant after 7 days, while *M. flos-aquae* and *W. naegeliana* became dominant after 14 days, especially in reservoir water with mZVI.

Early in the experiment, small green algae (*Chlorophyta*) dominated the phytoplankton, especially colony-forming *Planktosphaeria* sp. followed by *Staurastrum* sp., *Scenedesmus* sp., *Tetraedron* sp., *Monoraphidium* sp., *Oocystis* sp. and *Botryococcus braunii*. *Bacillariophyta*, *Dinophyta* and *Cryptophyta* taxa were only rarely recorded, as were ciliates and flagellates. The number of small green algae in nZVI was reduced compared with the control after three days, with this effect remaining significant for up to seven days ($P < 0.05$), while abundance of algae in mZVI remained similar to the control throughout the study (Fig. 4C, Table S2C†).

3.4. Bacterial community composition

NGS was used to characterise bacterial communities in the control and in samples with nZVI and mZVI at day 0, 7 and 21. Diversity tests revealed samples with mZVI as having higher species richness than nZVI ($P < 0.05$), particularly as regards rare community members (tail) ($P < 0.05$, Table 1). Moreover, significant differences were observed at both the family and genus levels when comparing mZVI and nZVI, being more important towards the end of the experiment (day

Table 1 Comparison of diversity indices for reservoir water (control) and reservoir water with nZVI and mZVI. The numbers in brackets represent the standard deviation of duplicate samples

	Richness	Inverse Simpson	Tail
Control	161 (28)	10 (1)	20 (4)
nZVI	147 (18)	11 (3)	18 (2)
mZVI	184 (34)	14 (3)	24 (6)
Mann-Whitney <i>U</i>	Richness	Inverse Simpson	Tail
mZVI-nZVI	31 ^a	29 n.s.	32 ^a

^a Indicates significance at $P < 0.05$; n.s. not significant.

21; family: $P < 0.05$, genus: $P < 0.001$). Statistically significant changes were also detected between the control and nZVI/mZVI samples at genus level by day 21 ($P < 0.05$, Table 2).

At the beginning of the experiment, untreated epilimnion water displayed a typical freshwater bacterial community, being represented in all treatments by the phyla *Proteobacteria* (*Alpha*-, *Beta*- and *Deltaproteobacteria*; together 38%), *Actinobacteria* (23%), *Bacteroidetes* (19%), *Verrucomicrobia* (10%), *Cyanobacteria* (4%) and *Chloroflexi* (2%) (Fig. 5). At day 7, the relative abundance of *Actinobacteria* and *Alphaproteobacteria* had decreased in all treatments, being compensated for by an increase in *Verrucomicrobia* and *Deltaproteobacteria* (control and mZVI) or *Betaproteobacteria* (nZVI). After an additional two weeks (day 21), the abundance of *Actinobacteria* and *Deltaproteobacteria* had recovered while *Verrucomicrobia* stayed enriched and *Alphaproteobacteria* reduced in the control and mZVI treatments. In nZVI, *Betaproteobacteria* (still dominating the bacterioplankton) were reduced by an increase in *Chloroflexi* and *Deltaproteobacteria* at day 21. *Bacteroidetes* and *Cyanobacteria* were not, or only weakly, affected by the treatment over time.

At the family level, bacterial community structure was also typical for a freshwater reservoir, consisting of *Microbacteriaceae* and *Sporichthyaceae* (both *Actinobacteria*), *Burkholderiaceae* and *Comamonadaceae* (both *Betaproteobacteria*), *Acetobacteraceae*, *Caulobacteraceae*, *Pelagibacteriaceae*, *Rhodospirillaceae* (all *Alphaproteobacteria*), *Opitutae* (*Verrucomicrobia*), *Flavobacteriaceae* and *Saprospiraceae* (both *Bacteroidetes*), and *Roseiflexaceae* (*Chloroflexi*) (Fig. 6). Cluster analysis supported the existence of three sample groups: i) starting samples, ii) mZVI and control samples and iii) nZVI samples. Starting samples were dominated by members of the *Pelagibacteriaceae* (LD12 cluster; 23%), *Sporichthyaceae* (hgcl_clade; 20%), *Flavobacteriaceae* (genus *Flavobacterium*; 7%), *Comamonadaceae* (genus *Limnohabitans*; 3%), *Microbacteriaceae* ('*Candidatus Limnoluna*'; 2%) and *Roseiflexaceae* (genus *Roseiflexus*; 1.4%). While trends differed in the presence of nZVI compared to the control, there was no shift in structure in mZVI samples. For example, while the *Comamonadaceae* and *Flavobacteriaceae* families were gradually reduced in mZVI and the control, the family-like group



Table 2 Differences between control, nZVI and mZVI treatments at the genus and family levels, evaluated using the standard Mann–Whitney *U* test

Genus level	Day 0	Day 7	Day 21
Control/mZVI	n.s.	n.s.	*
Control/nZVI	n.s.	n.s.	*
mZVI/nZVI	n.s.	**	***
Family level			
Control/mZVI	n.s.	n.s.	n.s.
Control/nZVI	n.s.	n.s.	n.s.
mZVI/nZVI	n.s.	*	*

Significance levels: * $P < 0.05$, ** $P < 0.01$ and *** $P < 0.001$; n.s. not significant.

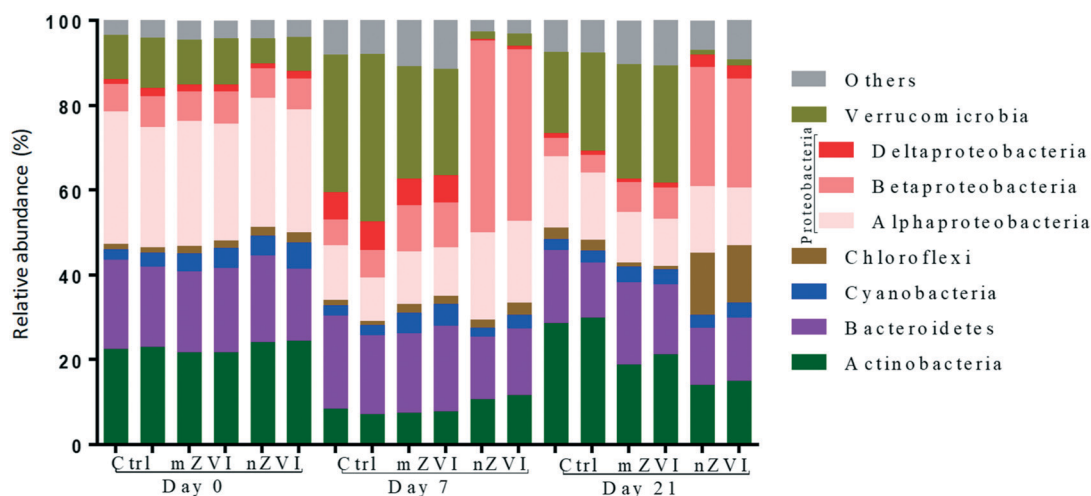


Fig. 5 Phylum and class composition of bacterial populations in the control, mZVI and nZVI treatments. The phylum *Proteobacteria* is represented by *Alfa*-, *Beta*- and *Delta*-classes. Phyla with less than 2% abundance are shown as unclassified. Samples without ZVI particles are used as a control (Ctrl). Note that each sample was sequenced in duplicate.

Opitutae_vadinHA64 increased continuously up to day 21. In the presence of nZVI, however, *Comamonadaceae* increased remarkably after seven days, then decreased after 21 days. *Opitutae_vadinHA64* decreased continuously while *Sporichthyaceae* declined and remained at the same low level from day 7 to the end of the experiment (Fig. 6).

At the genus level, a comparison of relative abundance in the nZVI and mZVI treatments with the control revealed seven groups with >5% change in total bacterial population (Fig. 7). In general, the addition of nZVI particles resulted in more dynamic population shifts (up to 29%) than the addition of mZVI (<11%). A decline in the LD12 cluster and an increase in *Limnohabitans* and in the *Comamonadaceae_unclassified* read tags was observed as a common effect of both materials. *Roseiflexus* showed a continuous increase over days 0, 7, but not day 21, following mZVI treatment, while the opposite trend was observed in the tribe *Opitutae_vadinHA64*. The *hgcI*-clade abundance remained steady until day 7, but was reduced by day 21. The tribe *OPB35_soil_group* was enriched from its initial abundance in nZVI and mZVI treatments, including the control; however, population growth in nZVI treatment was 5% lower

than in control after 7 days. Other genus-like groups also showing 3–5% differences from the control were *Azospirillum*, *Brevundimonas*, *Desulfomonile*, *Flavobacterium*, the NS11-12_marine_group, *Polynucleobacter*, *Roseomonas* and the *Bacteria_unclassified* group. *Flavobacterium*, *Polynucleobacter* and *Roseomonas* all profited from nZVI treatment over both sampling days, while the NS11-12_marine_group was always depleted. A notable decrease in population size was observed for *Desulfomonile* and the *Bacteria_unclassified* group in nZVI at day 7. *Desulfomonile* was also reduced in mZVI at day 7.

3.5. Ferroxidase enzyme

Abundance of enzymes related to iron metabolism in bacteria was calculated using the Paprica⁵² pipeline, which conducts a metabolic inference from 16S rRNA gene sequence libraries. Specifically, we extracted pathway and time changes in abundance of ferroxidase enzyme (EC1.16.3.1).

Compared to the control, the abundance of ferroxidase changed in the presence of nZVI and mZVI over the 21 day experiment (Fig. S5[†]), with the enzyme gradually increasing in *Roseiflexus* sp. and the species *Roseiflexus castenholzii* but



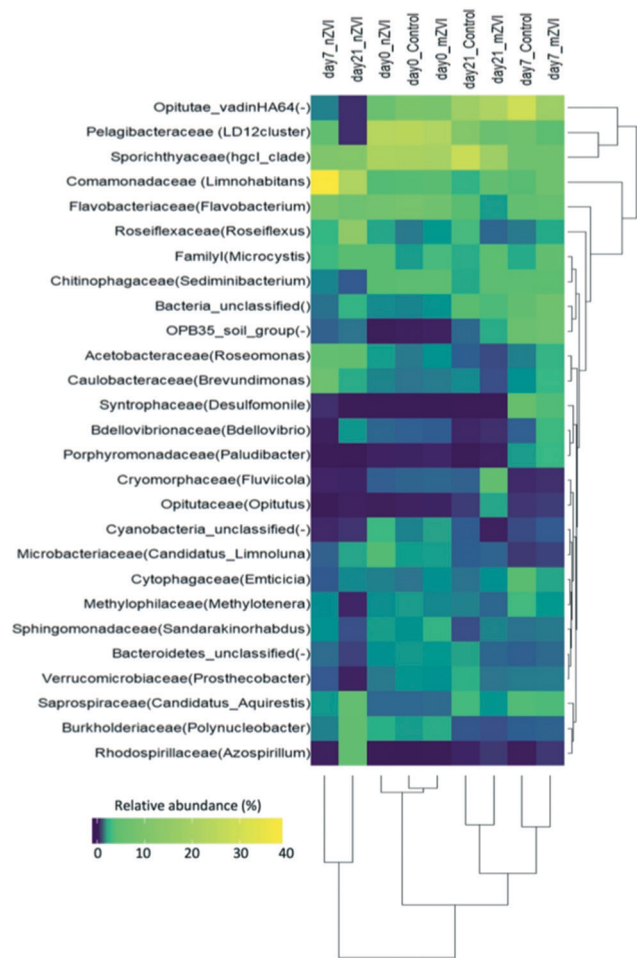


Fig. 6 Heatmap indicating the relative abundance of the major bacterial families for days 0, 7 and 21. The relationship of the bacterial composition in each sample is depicted with a cluster dendrogram based on Bray–Curtis dissimilarity among samples and among families. The dominant genus rank is provided in parentheses after the family name (dash '-', when unclassified at genus level). Only those families with a relative abundance >2% are shown. Data are means from duplicate samples.

decreasing in *Fluviicola taffensis* (Fig. S6[†]). Based on our data, there was no indication of ferroxidase enzyme in the genus *Sediminibacterium*.

4. Discussion

The use of commercially produced ZVI particles for remediation of polluted aquatic habitats has resulted in questions regarding their impact on naturally occurring microbial communities.^{15,54,55} Microbial communities are important in aquatic habitats as they are responsible for dissolved carbon transfer to higher trophic levels (the microbial food web concept). They have also been described as key players in the self-purification of lakes and reservoirs,⁵⁶ and as such possess high ecological potential. In contrast to previous studies,³³ we opened a black box and identified freshwater heterotrophic bacteria that were affected by nZVI and mZVI particles.

4.1. Effect of ZVI on phytoplankton and associated bacteria

We observed different effects of nZVI and mZVI on freshwater reservoir microorganisms (Fig. 4 and 5). Presence of nZVI, for example, significantly stimulated growth of cultivable bacteria but reduced the number of algae on day 3 and day 7, compared to the control (Fig. 4A and C). On the other hand, addition of mZVI resulted in a slight increase in cyanobacteria, which was rather surprising (Fig. 4B and 5). While the use of nZVI has been suggested for prevention of harmful cyanobacterial blooms,^{11,12} our study revealed no significant effect of nZVI on cyanobacteria.

The association of distinct bacterial consortia with specific algae is known as the “phycosphere”. In our study, we found similar bacterial clades as described for the *M. aeruginosa* phycosphere in plateau lake,⁵⁷ i.e. *Brevundimonas*, the OM27 clade and *Roseomonas*. The genus *Brevundimonas* (*Caulobacteraceae*) has been described as a keystone species in particle-associated bacterial communities during *M. aeruginosa* blooms⁵⁷ and, while information on clade OM27 is limited, *Roseomonas* sequences have been detected in freshwater bacterial communities associated with cyanobacterial blooms in four Swedish lakes.⁵⁸

Additionally, we also observed that cyanobacterial colonies of *M. aeruginosa* (cells 4–5 μm) dominant up to day 7 in mZVI, were then outcompeted by *M. flos-aquae* and *W. naegeliiana* (cells 2–3 μm). Small-cell colonies were most probably more capable of competition for nutrients, because similar trends were observed in nZVI and control as well. Increasing proportion of disintegrated and partly decayed cyanobacterial cells towards the end of the experiment could be caused by depletion of nutrients, and, moreover algicidal bacteria (e.g. *Brevundimonas*, Fig. 6) might work against cyanobacteria. Algicidal bacteria that can associate with cyanobacteria have the capability to control their proliferation in fresh water and even to lyse them.⁵⁹

The reduction of small green algae after seven days in nZVI was most probably due to unfavourable conditions due to decline in ORP and depletion of bioavailable phosphorus,^{11,12} while the decrease of algal number over the 21 days of the experiment in nZVI, mZVI as well as in control strongly suggests the nutrient limitation. Consequently, bacteria decomposing algal cell walls were favoured by the availability of polysaccharides. We speculate that the increase in relative abundance in the OPB35_soil_group (*Verrucomicrobia*, Fig. 7) could be linked to the decay of algae due to its capability for active polysaccharide degradation (laminarin and xylose) of the closely-related genus *Limnispheera*.⁶⁰

4.2. Oxygen and ORP drive microbial community composition

Compared with the control and mZVI samples, nZVI reductive reactivity resulted in a significant drop in dissolved oxygen and ORP at the start of the experiment (Fig. 3). Consequently, the strictly aerobic genus *Sediminibacterium* (*Chitinophagaceae*)⁶¹ population decreased with nZVI, but not



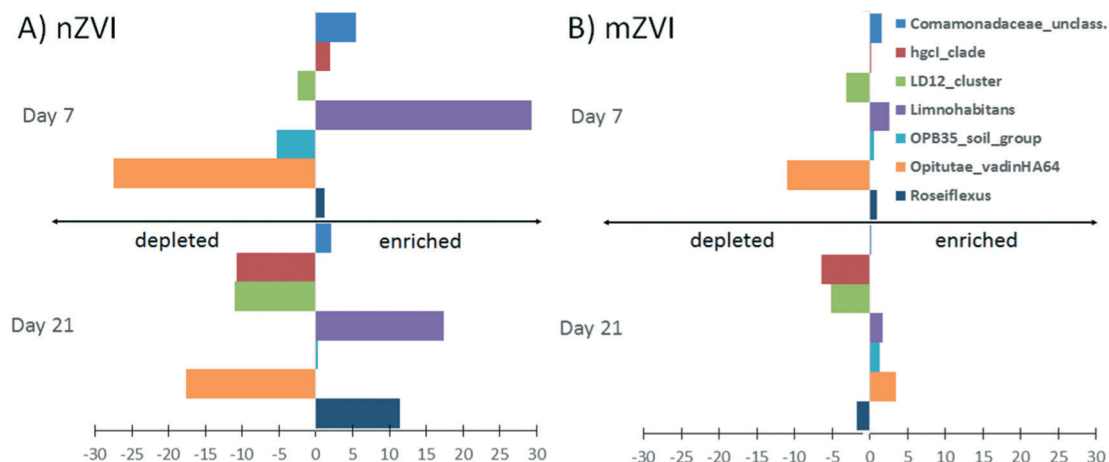


Fig. 7 Impact of ZVI particles on the development of genus-like groups of the most abundant bacteria from the Harcov reservoir. Values were calculated as differences in relative abundance in A) nZVI and the control, or B) mZVI and the control, after 7 and 21 days. The values are means from duplicate measurements. Only those genera or tribes with a relative difference >5% are shown.

in the control or with mZVI (Fig. 6). Moreover, the low dissolved oxygen concentration most probably affected phagotrophic protists, such as heterotrophic flagellates and ciliates. Both these protist groups are efficient grazers on bacteria and, as such, exhibit “top-down” control.⁶² As grazing on bacteria was found to be negligible under anoxic conditions,⁶³ absence of oxygen probably released bacteria from grazing pressure and shifted the community from small-size grazing-resistant prokaryotes to fast-growing opportunists.⁶⁴ In support of our hypothesis, the addition of nZVI strongly favoured growth of *Limnohabitans* (*Betaproteobacteria*), facultative anaerobes^{65,66} and fast growing bacteria that often dominate in bacterivore-free (<0.8 μm) treatments.⁶⁷ In contrast, the freshwater LD12 cluster (*Alphaproteobacteria*) and *Verrucomicrobia*, with their slow-growth life strategy, were depleted in the same treatments. The lesser decrease in ORP (though persistent over the 21 days) caused by mZVI resulted in *Opirituae_vadinHA64* and *OPB35_soil* (*Verrucomicrobia*) being slightly enriched. *Opirituae_vadinHA64* have recently been recorded in the epilimnion of freshwater lakes with higher dissolved oxygen concentrations,⁶⁸ similar to that in the mZVI treatment, and thus we can assume that this cluster prefers aerobic conditions. Unfortunately, *Verrucomicrobia* are still poorly understood, many publications have noted that different members of this group have different oxygen requirements, some being strict aerobes while others are facultative anaerobes or strict anaerobes.^{69–72}

4.3. Response of rare bacterial taxa

Rare bacteria have been proposed as drivers of key functions in aquatic systems as they outnumber the dominant group diversity and each represents a specific metabolic combination.⁷³ Their appearance in samples after experimental manipulation suggests that conditions differed from *in situ* reservoir conditions, where they were almost absent. Our experiment promoted growth of the biofilm-forming taxa

Azospirillum (*Rhodospirillaceae*, *Alphaproteobacteria*) and *Paludibacter* (*Porphyromonadaceae*, *Bacteroidetes*), both having a negligible relative abundance at day 0 but increasing dramatically over the next three weeks of the experiment (Fig. 6). *Azospirillum* bacteria are diazotrophs with the capacity for sulphur-chemolithotrophy.⁷⁴ The strictly anaerobic *Paludibacter* has previously been reported as the most abundant genus on biofilms forming on particles of 0.02–0.2 mm under static conditions.⁷⁵ Similarly, the genus *Desulfomonile*, which bloomed after seven days but then vanished up until day 21 in both treatments (Fig. 6), has previously been reported as a rare taxa from acidic peatlands responding to sulphate incubation.⁷⁶ Populations of the filamentous bacteria ‘*Candidatus Aquirestis*’ (LD2/SOL cluster, *Bacteroidetes*) benefitted from the experimental setup and established strong populations in nZVI and reservoir water after 21 days. These bacteria primarily inhabit the pelagic zone of non-acidic stagnant inland waters and show a strong spring peak, with minor peaks in summer and early autumn.⁷⁷ To conclude, the emergence of rare bacterial taxa indicates changes in water chemistry and the availability of nutrients. In our study, this appears to have been due to increasing nitrogen limitation, a higher availability of sulphur and iron, and the decay of algae with polysaccharide cell walls.

4.4. Oligotrophic bacteria depletion in nZVI

Manipulation experiments may lead to nutrient depletion during transfer from a dissolved to a particulate (biomass) fraction. The freshwater clusters hgcl (*Actinobacteria*) and LD12 (*Alphaproteobacteria*) are small slow-growing bacteria with a passive life style favoured by oligotrophic conditions.⁶⁸ We hypothesise that their relatively high abundance in the reservoir water control and mZVI treatments could serve as indicators of nutrient depletion, while their lower numbers in the nZVI treatment suggest nutrient remineralisation.



In soils, *Verrucomicrobia* generally displays an oligotrophic life strategy.⁷⁸ Currently, members of the phylum *Verrucomicrobia* are increasingly being recognised as abundant species in aquatic habitats,⁶⁸ with prevalence of cluster CL0-14 being negatively correlated with high pH and positively correlated with hydraulic retention time and temperature, as are the oligotrophic clusters LD12 (freshwater SAR11 cluster) and LD28 (*Methylopusillus* spp.).⁷⁹ Our manipulations demonstrated a striking decrease in the abundance of the *Verrucomicrobia* cluster Opitutae_vadinHA64 after nZVI addition, similar to that for the hgcI and LD12 clusters (Fig. 6). The Opitutae_vadinHA64 cluster represents around 80% of all *Verrucomicrobia* in the Harcov reservoir; hence, it has great potential as an indicator clade of nZVI. However, more information is still needed on its ecophysiology, distribution, phylogeny and diversity.

4.5. ZVI particles enhance growth of bacteria with ferroxidase

Typically for natural freshwaters, dissolved Fe(II) was found at very low concentrations (0.1 to 0.01 mg L⁻¹; Table S3†) in this study. On the other hand, oxidation of Fe(0) releasing soluble Fe(II), which was further converted into insoluble Fe(III), was ongoing throughout the experiment.⁸⁰ Iron oxidising bacteria may have caused the increase in cultivable bacteria, also supported by the simultaneous increase in ferroxidase enzyme, a key enzyme in Fe(II) oxidation. The most probable source in this case was the nZVI, which can supply extra Fe(II) for bacterial metabolism.

To date, all known oxygen-dependent, neutrophilic and lithotrophic iron oxidisers have been recorded within the *Proteobacteria*^{23,81} and for *Comamonadaceae* in groundwaters at neutral pH.⁸² In this study, however, we confirmed *Roseiflexus* sp. and *Roseiflexus castenholzii* as being heavily involved in the increase of ferroxidase, a fact not reported elsewhere (Fig. S5 and S6†). Similar trends were observed in a microcosm study on the River Thames, the authors presuming that the increase in total bacteria was attributable to the Fe oxidation process itself.¹⁵ In a previous study, *Sediminibacterium* was identified as an iron-oxidising bacterium actively promoting the corrosion of cast iron pipes alongside sulphur-oxidising and iron-reducing bacteria.⁸³ In our study, the comparable levels recorded in the control and mZVI treatment suggest that *Sediminibacterium* were not promoted by the addition of ZVI particles.

4.6. ZVI particle fate in the aquatic environment

The behaviour of nZVI and mZVI is strongly dependent on environmental conditions. In our case, the detection and measurement of ZVI particles was only possible after centrifugation or filtration, following the removal of large debris particles or microorganisms. As with mZVI, nZVI tended to show higher aggregation/agglomeration in centrifuged rather than filtered reservoir water (Fig. 1 and 2), most probably due to a higher content of colloids that remained after centrifugation. Hence, we believe that centrifuged reservoir water is more

comparable to natural reservoir water than filtered water. Stronger particle aggregation is often connected with higher pH values^{16,19} and, indeed, water from the Harcov reservoir was somewhat alkaline (pH 8.0). The pH increased still further up to 8.8 with ZVI as an electron donor (Fig. 3), though the pH decreased to ~7.4 over the next 21 days in both ZVI treatments. Under the oxic conditions in our study, released Fe(II) quickly oxidised to Fe(III) and most probably formed stable colloids with natural organic matter⁸⁴ that were not harmful to microorganisms. Moreover, the aggregated ZVI particles rapidly formed a sediment, which could reduce their effect on freely-swimming microorganisms in the water column, allowing them to escape direct contact.^{20,30} Oxidation of nZVI led to a decline in ORP and dissolved oxygen concentration after one day (Fig. 3), which had a strong influence on shifts in bacterial community, as described above. However, these values increased again as nZVI oxidised and reductive capacity was spent, resulting in values comparable with the control, the process being driven further by the growth and activity of the microorganisms described above.

5. Conclusions

In general, nZVI showed a stronger effect on planktonic microorganisms than mZVI, most probably due to its higher reactivity. nZVI significantly stimulated the growth of cultivable bacteria and iron-oxidising bacteria but slightly reduced the growth of algae. Phagotrophic flagellates and ciliates were most probably affected by low dissolved oxygen concentration, such conditions releasing bacteria from grazing pressure and shifting the community from small grazing-resistant prokaryotes to fast-growing opportunists. In support of this hypothesis, nZVI strongly favoured growth of the facultative anaerobe *Limnohabitans* and other opportunist bacteria that can come to dominate in bacterivore-free treatments, whereas the LD12 cluster and *Verrucomicrobia*, with slow-growing life strategies, were depleted. Unexpectedly, mZVI resulted in a slight increase in cyanobacterial number, which could be explained by increase in small-cell colonies of *M. flos-aquae*, and *W. naegeliana*. Further *in situ* studies that more realistically mimic reservoir conditions could well reveal more information on the influence of ZVI on natural microbial communities, including parts of the microbial food web not covered here such as heterotrophic flagellates and viruses.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The research reported in this paper was financially supported by the Technology Agency of the Czech Republic within “Nanobiowat” under Project no. TE01020218, and by the Ministry of Education, Youth and Sports under the framework of targeted support for the “National Programme for



Sustainability I', the OPR & DI project (LO 1201) and Nano-Envi, Project no. LM2015073. The work of N. Nguyen was partly supported by MŠMT under SGS Project no. 21176/115 of the Technical University of Liberec. V. Kasalický was supported by Czech Science Foundation Project no. 15-12197S.

References

- P. Bardos, B. Bone, M. Černík, D. W. Elliott, S. Jones and C. Merly, *Remediation*, 2015, **26**, 101–108.
- Y. S. El-Temseh and E. J. Joner, *Chemosphere*, 2012, **89**, 76–82.
- A. Ševců, Y. S. El-Temseh, J. Filip, E. J. Joner, K. Bobčíková and M. Černík, *Environ. Sci. Pollut. Res.*, 2017, **24**, 21191–21202.
- K. J. Cantrell, D. I. Kaplan and T. W. Wietsma, *J. Hazard. Mater.*, 1995, **42**, 201–212.
- R. A. Crane and T. B. Scott, *J. Hazard. Mater.*, 2012, **211**–212, 112–125.
- Wei-xian Zhang and W. X. Zhang, *J. Nanopart. Res.*, 2003, **5**, 323–332.
- N. C. Mueller, J. Braun, J. Bruns, M. Černík, P. Rissing, D. Rickerby and B. Nowack, *Environ. Sci. Pollut. Res.*, 2012, **19**, 550–558.
- S. Comba, A. Di Molfetta and R. Sethi, *Water, Air, Soil Pollut.*, 2011, **215**, 595–607.
- J. Němeček, P. Pokorný, O. Lhotský, V. Knytl, P. Najmanová, J. Steinová, M. Černík, A. Filipová, J. Filip and T. Cajthaml, *Sci. Total Environ.*, 2015, **563**–564, 822–834.
- F. Fu, D. D. Dionysiou and H. Liu, *J. Hazard. Mater.*, 2014, **267**, 194–205.
- B. Marsalek, D. Jancula, E. Marsalkova, M. Mashlan, K. Safarova, J. Tucek and R. Zboril, *Environ. Sci. Technol.*, 2012, **46**, 2316–2323.
- V. K. Sharma, L. Chen, B. Marsalek, R. Zboril, K. E. O'Shea and D. D. Dionysiou, *Water Sci. Technol.: Water Supply*, 2016, **17**, 107–114.
- R. Köber, H. Hollert, G. Hornbruch, M. Jekel, A. Kamptner, N. Klaas, H. Maes, K.-M. Mangold, E. Martac, A. Matheis, H. Paar, A. Schäffer, H. Schell, A. Schiwy, K. R. Schmidt, T. J. Strutz, S. Thümmler, A. Tiehm and J. Braun, *Environ. Earth Sci.*, 2014, **72**, 3339–3352.
- P. Bardos, B. Bone, P. Daly, D. Elliott, S. Jones, G. Lowry and C. Merly, *A Risk/Benefit Appraisal for the Application of Nano-Scale Zero Valent Iron (nZVI) for the Remediation of Contaminated Sites*, WP9 NanoRem, 2014.
- R. J. Barnes, C. J. van der Gast, O. Riba, L. E. Lehtovirta, J. I. Prosser, P. J. Dobson and I. P. Thompson, *J. Hazard. Mater.*, 2010, **184**, 73–80.
- M. Auffan, W. Achouak, J. Rose, M. A. Roncato, C. Chanéac, D. T. Waite, A. Masion, J. C. Woicik, M. R. Wiesner and J. Y. Bottero, *Environ. Sci. Technol.*, 2008, **42**, 6730–6735.
- C. Lei, L. Zhang, K. Yang, L. Zhu and D. Lin, *Environ. Pollut.*, 2016, **218**, 505–512.
- A. Ševců, Y. S. El-Temseh, E. J. Joner and M. Černík, *Microbes Environ.*, 2011, **26**, 271–281.
- A. Keller, K. Garner, R. J. Miller and H. S. Lenihan, *PLoS One*, 2012, **7**, 1–10.
- M. Diao and M. Yao, *Water Res.*, 2009, **43**, 5243–5251.
- M. L. Saccà, C. Fajardo, G. Costa, C. Lobo, M. Nande and M. Martin, *Chemosphere*, 2014, **104**, 184–189.
- W. Zhang, C.-B. Wang and H.-L. Lien, *Catal. Today*, 1998, **40**, 387–395.
- S. Hedrich, M. Schlömann and D. B. Johnson, *Microbiology*, 2011, **157**, 1551–1564.
- M. L. Saccà, C. Fajardo, M. Nande and M. Martín, *Microb. Ecol.*, 2013, **66**, 806–812.
- J. Y. Kim, H. J. Park, C. Lee, K. L. Nelson, D. L. Sedlak and J. Yoon, *Appl. Environ. Microbiol.*, 2010, **76**, 7668–7670.
- C. Fajardo, M. L. Saccà, M. Martinez-Gomariz, G. Costa, M. Nande and M. Martin, *Chemosphere*, 2013, **93**, 1077–1083.
- C. Jiang, X. Xu, M. Megharaj, R. Naidu and Z. Chen, *Sci. Total Environ.*, 2015, **530**–531, 241–246.
- Y. S. El-Temseh, A. Sevcu, K. Bobcikova, M. Cernik and E. J. Joner, *Chemosphere*, 2016, **144**, 2221–2228.
- J. Semerád and T. Cajthaml, *Appl. Microbiol. Biotechnol.*, 2016, **100**, 9809–9819.
- R. Hjorth, C. Coutris, N. H. A. Nguyen, A. Sevcu, A. Baun and E. Joner, *Chemosphere*, 2017, **182**, 525–531.
- C. Fajardo, L. T. Ortíz, M. L. Rodríguez-Membibre, M. Nande, M. C. Lobo and M. Martin, *Chemosphere*, 2012, **86**, 802–808.
- E. L. Tilston, C. D. Collins, G. R. Mitchell, J. Princivalle and L. J. Shaw, *Environ. Pollut.*, 2013, **173**, 38–46.
- T. L. Kirschling, K. B. Gregory, E. G. Minkley, G. V. Lowry and R. D. Tilton, *Environ. Sci. Technol.*, 2010, **44**, 3474–3480.
- J. Němeček and J. Lhotský, *Sci. Total Environ.*, 2014, **485**–486, 739–747.
- A. Schiwy, H. M. Maes, D. Koske, M. Flecken, K. R. Schmidt, H. Schell, A. Tiehm, A. Kamptner, S. Thümmler, H. Stanjek, M. Heggen, R. E. Dunin-Borkowski, J. Braun, A. Schäffer and H. Hollert, *Environ. Pollut.*, 2016, **216**, 1–9.
- L. G. Cullen, E. L. Tilston, G. R. Mitchell, C. D. Collins and L. J. Shaw, *Chemosphere*, 2011, **82**, 1675–1682.
- Y.-h. Shih, H. L. Chou, Y. H. Peng and C.-y. Chang, *Bioresour. Technol.*, 2012, **108**, 14–20.
- M. Velimirovic, Q. Simons and L. Bastiaens, *Chemosphere*, 2015, **134**, 338–345.
- N. H. A. Nguyen, N. R. von Moos, V. I. Slaveykova, K. Mackenzie, R. U. Meckenstock, S. Thümmler, J. Bosch and A. Ševců, *Ecotoxicol. Environ. Saf.*, 2018, **154**, 36–44.
- G. V. Lowry, E. M. Hotze, E. S. Bernhardt, D. D. Dionysiou, J. A. Pedersen, M. R. Wiesner and B. Xing, *J. Environ. Qual.*, 2010, **39**, 1867–1874.
- C. Colombo, G. Palumbo, J. Z. He, R. Pinton and S. Cesco, *J. Soils Sediments*, 2014, **14**, 538–548.
- J. Soukupova, R. Zboril, I. Medrik, J. Filip, K. Safarova, R. Ledl, M. Mashlan, J. Nosek and M. Cernik, *Chem. Eng. J.*, 2015, **262**, 813–822.
- Z. Shi, D. Fan, R. L. Johnson, P. G. Tratnyek, J. T. Nurmi, Y. Wu and K. H. Williams, *J. Contam. Hydrol.*, 2015, **181**, 17–35.



- 44 T. J. Strutz, G. Hornbruch, A. Dahmke and R. Köber, *J. Contam. Hydrol.*, 2016, **191**, 54–65.
- 45 D. Ribas, M. Cernik, V. Martí and J. A. Benito, *J. Nanopart. Res.*, 2016, **18**, 181.
- 46 I. Dolinová, M. Czinnerová, L. Dvořák, V. Stejskal, A. Ševců and M. Černík, *Chemosphere*, 2016, **157**, 276–285.
- 47 S. E. Dowd, T. R. Callaway, R. D. Wolcott, Y. Sun, T. McKeehan, R. G. Hagevoort and T. S. Edrington, *BMC Microbiol.*, 2008, **8**, 125.
- 48 M. J. Claesson, Q. Wang, O. O'Sullivan, R. Greene-Diniz, J. R. Cole, R. P. Ross and P. W. O'Toole, *Nucleic Acids Res.*, 2010, **38**, 1–13.
- 49 P. D. Schloss, S. L. Westcott, T. Ryabin, J. R. Hall, M. Hartmann, E. B. Hollister, R. A. Lesniewski, B. B. Oakley, D. H. Parks, C. J. Robinson, J. W. Sahl, B. Stres, G. G. Thallinger, D. J. Van Horn and C. F. Weber, *Appl. Environ. Microbiol.*, 2009, **75**, 7537–7541.
- 50 R. C. Edgar, B. J. Haas, J. C. Clemente, C. Quince and R. Knight, *Bioinformatics*, 2011, **27**, 2194–2200.
- 51 A. J. Oksanen, F. G. Blanchet, M. Friendly, R. Kindt, P. Legendre, D. Mcglinn, P. R. Minchin, R. B. O. Hara, G. L. Simpson, P. Solymos, M. H. H. Stevens, E. Szoecs and H. Wagner, *The Comprehensive R Archive Network*, 2016, 2.4-1, pp. 1–291, <https://github.com/vegandevs/vegan>.
- 52 J. S. Bowman and H. W. Ducklow, *PLoS One*, 2015, **10**, 1–18.
- 53 R. J. Newton and S. L. McLellan, *Front. Microbiol.*, 2015, **6**, 1–13.
- 54 A. M. I. Niyas, *J. Crit. Rev.*, 2014, **1**, 26–39.
- 55 E. Lefevre, N. Bossa, M. R. Wiesner and C. K. Gunsch, *Sci. Total Environ.*, 2015, **565**, 889–901.
- 56 M. A. Cooperstein and H. E. Canavan, *Biointerphases*, 2013, **8**, 1–12.
- 57 C. Yang, Q. Wang, P. N. Simon, J. Liu, L. Liu, X. Dai, X. Zhang, J. Kuang, Y. Igarashi, X. Pan and F. Luo, *Front. Microbiol.*, 2017, **8**, 1202.
- 58 A. Eiler and S. Bertilsson, *Environ. Microbiol.*, 2004, **6**, 1228–1243.
- 59 N. Meyer, A. Bigalke, A. Kaulfuß and G. Pohnert, *FEMS Microbiol. Rev.*, 2017, **41**, 880–899.
- 60 M. Martinez-Garcia, D. M. Brazel, B. K. Swan, C. Arnosti, P. S. G. G. Chain, K. G. Reitenga, G. Xie, N. J. Poulton, M. L. Gomez, D. E. D. D. Masland, B. Thompson, W. K. Bellows, K. Ziervogel, C.-C. C. Lo, S. Ahmed, C. D. Gleasner, C. J. Detter and R. Stepanauskas, *PLoS One*, 2012, **7**, 1–11.
- 61 Y. Kim, B. Kim, K. Kang and T.-Y. Y. Ahn, *Int. J. Syst. Evol. Microbiol.*, 2016, **66**, 4501–4505.
- 62 K. Simek, J. Pernthaler, M. G. Weinbauer, J. R. Dolan, J. Nedoma, M. Masin and K. Hornak, *Appl. Environ. Microbiol.*, 2001, **67**, 2723–2733.
- 63 A. Oikonomou, M. Pachiadaki and T. Stoeck, *FEMS Microbiol. Ecol.*, 2014, **87**, 691–703.
- 64 J. Pernthaler, *Nat. Rev. Microbiol.*, 2005, **3**, 537–546.
- 65 M. W. Hahn, V. Kasalický, J. Jezbera, U. Brandt and K. Šimek, *Int. J. Syst. Evol. Microbiol.*, 2010, **60**, 2946–2950.
- 66 V. Kasalický, J. Jezbera, K. Šimek and M. W. Hahn, *Int. J. Syst. Evol. Microbiol.*, 2010, **60**, 2710–2714.
- 67 K. Šimek, K. Hornák, J. Jezbera, J. Nedoma, P. Znachor, J. Hejzlar and J. Sed'a, *Aquat. Microb. Ecol.*, 2008, **51**, 249–262.
- 68 R. J. Newton, S. E. Jones, A. Eiler, K. D. McMahon and S. Bertilsson, *Microbiol. Mol. Biol. Rev.*, 2011, **75**, 14–49.
- 69 P. J. Cabello-Yeves, R. Ghai, M. Mehrshad, A. Picazo, A. Camacho and F. Rodriguez-Valera, *Front. Microbiol.*, 2017, **8**, 1–17.
- 70 A. De Wever, K. Van Der Gucht, K. Muylaert, S. Cousin and W. Vyverman, *Aquat. Microb. Ecol.*, 2008, **50**, 113–122.
- 71 J. Zhang, X. Zhang, Y. Liu, S. Xie and Y. Liu, *Ann. Microbiol.*, 2014, **64**, 1405–1411.
- 72 J. F. Humbert, U. Dorigo, P. Cecchi, B. Le Berre, D. Debross and M. Bouvy, *Environ. Microbiol.*, 2009, **11**, 2339–2350.
- 73 A. Jousset, C. Bienhold, A. Chatzinotas, L. Gallien, A. Gobet, V. Kurm, K. Küsel, M. C. Rillig, D. W. Rivett, J. F. Salles, M. G. A. van der Heijden, N. H. Youssef, X. Zhang, Z. Wei and W. H. G. Hol, *ISME J.*, 2017, **11**, 853–862.
- 74 Y. Kwak and J. H. Shin, *Mar. Genomics*, 2016, **25**, 21–24.
- 75 H. Fang, Y. Chen, L. Huang and G. He, *Sci. Rep.*, 2017, **7**, 1–14.
- 76 B. Hausmann, K.-H. Knorr, K. Schreck, S. G. Tringe, T. Glavina Del Rio, A. Loy and M. Pester, *ISME J.*, 2016, **10**, 2365–2375.
- 77 M. Schauer, C. Kamenik and M. W. Hahn, *Appl. Environ. Microbiol.*, 2005, **71**, 5900–5907.
- 78 G. T. Bergmann, S. T. Bates, K. G. Eilers, C. L. Lauber, J. G. Caporaso, W. A. Walters, R. Knight and N. Fierer, *Soil Biol. Biochem.*, 2011, **43**, 1450–1455.
- 79 M. M. Salcher, S. M. Neuenschwander, T. Posch and J. Pernthaler, *ISME J.*, 2015, **9**, 2442–2453.
- 80 A. J. Feitz, S. H. Joo, J. Guan, Q. Sun, D. L. Sedlak and T. D. Waite, *Colloids Surf., A*, 2005, **265**, 88–94.
- 81 D. Emerson, E. J. Fleming and J. M. McBeth, *Annu. Rev. Microbiol.*, 2010, **64**, 561–583.
- 82 M. Blöthe and E. E. Roden, *Appl. Environ. Microbiol.*, 2009, **75**, 468–473.
- 83 H. Wang, C. Hu, X. Hu, M. Yang and J. Qu, *Water Res.*, 2012, **46**, 1070–1078.
- 84 P. Liao, W. Li, Y. Jiang, J. Wu, S. Yuan, J. D. Fortner and D. E. Giammar, *Environ. Sci. Technol.*, 2017, **51**, 12235–12245.

