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Sizable bandgaps of graphene in 3d transition metal intercalated defective graphene/WSe₂ heterostructures†

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Controlling the electronic and magnetic properties of G/TMD (graphene on transition metal dichalcogenide) heterostructures is essential to develop electronic devices. Despite extensive studies in perfecting G/TMDs, most products have various defects due to the limitations of the fabrication techniques, and research investigating the performances of defective G/TMDs is scarce. Here, we conduct a comprehensive study of the effects of 3d transition metal (TM = Sc–Ni) atom-intercalated G/WSe₂ heterostructures, as well as their defective configurations having single vacancies on graphene or WSe₂ sublayers. Interestingly, Ni-intercalated G/WSe₂ exhibits a small band gap of 0.06 eV, a typical characteristic of nonmagnetic semiconductors. With the presence of one single vacancy in graphene, nonmagnetic (or ferromagnetic) semiconductors with sizable band gaps, 0.10–0.51 eV, can be achieved by intercalating Ti, Cr, Fe and Ni atoms into the heterostructures. Moreover, V and Mn doped non-defective and Sc, V, Co doped defective G/WSe₂ can lead to sizable half metallic band gaps of 0.1–0.58 eV. Further analysis indicates that the significant electron transfer from TM atoms to graphene accounts for the opening of a large band gap. Our results provide theoretical guidance to future applications of G/TMD based heterostructures in (spin) electronic devices.

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1. Introduction

Heterogeneous systems like van der Waals heterostructures are of considerable interest due to their distinct physical and chemical properties,^{1–3} which opens an avenue for developing new types of applications for field-effect tunneling transistors,^{4–7} catalysis,⁸ solar cells,⁹ and so on. Among different heterogeneous systems, the heterogeneous bilayer structures constructed by locating the graphene (G) layer on proper two-dimensional substrates have been the focus of extensive research because of their potential in high-performance devices. In 2011, Chang and Chen found that G/MoS₂ has excellent rate capability and cycling stability, and has the potential to act as an anode material for lithium ion batteries.¹⁰ In 2014, Tan *et al.* pointed out that the high electronic mobility

of graphene could be maintained in the G/h-BN, G/GaSe and G/WS₂ heterostructures.¹¹ In 2015, Padilha *et al.*¹² and Hu *et al.*¹³ found that the G/phosphorene (G/P) heterostructures can be utilized to control the Schottky barrier by applying a perpendicular electronic field. Using density functional theory calculations, Guo *et al.* found that the G/P heterostructures possess high capacity and good conductivity, excellent Li mobility and ultrahigh stiffness.¹⁴

Importantly, locating graphene on hetero-substrates could preserve a nearly linear band dispersion relation and open the band gap in graphene, which provides a strategy to overcome the biggest obstacle for its logical device application.^{15–21} As a few examples, theoretical studies have shown that the band gap of the G/h-BN heterostructure is sensitive to the interlayer spacing and stacking patterns.^{15,16} By applying a uniaxial electric field, the band structures of G/P heterostructures can be changed from semiconducting to metallic.¹⁷ In addition, a small band gap (2 meV) was opened for graphene when coupling it on MoS₂ or MoSe₂ substrate.^{18,19} Remarkably, sizable band gaps (3.9–28.2 meV) can be achieved for G/WS₂ or G/MoS₂ by regulating the stacking patterns and the interlayer spacing.^{20,21}

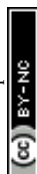
Despite the progress, further improvement of the performances of the graphene-based heterostructures is still challenging. For example, although the weak interlayer van der Waals interaction can introduce small dispersion to the electronic structures of graphene, the opened band gap is too small

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to meet the requirement for nanodevice applications. With the aim of tuning the electronic and magnetic properties, intercalating atoms or molecules between bilayer 2D materials arises as an effective method.^{22–26} For example, by introducing Ti and Mn atoms between bilayer graphene, it is possible to transform them into semiconductors with sizable band gaps of 0.14 eV and 0.82 eV.²³ In addition, due to the limitation of fabrication techniques, it is still challenging to synthesize large-area, high-quality 2D nanomaterials. Experimentally, a variety of defects: vacancies, interstitials, dopants and grain boundaries, and so on, will emerge unavoidably during the growth process of various 2D materials, which will greatly influence their performances.^{27–33} Unfortunately, few reports have been presented on the effects of such defects, and consequently, unveiling the influence of the defects on the performances of the heterostructures is of great importance. Different to MoS₂, monolayer WSe₂ is a p-type semiconductor and shows great potential for further electronic and spintronic applications. On the other hand, the fabrication of WSe₂ appears to be somewhat more difficult, and therefore, the previously mentioned defects must appear in the samples.^{34–36} In this work, based on first-principles calculations, we have systematically studied the structural, electronic and magnetic properties of 3d transition metal atom (TM = Sc–Ni)-intercalated G/WSe₂ heterostructures, as well as the systems involving single vacancies in graphene (G_sv) or WSe₂ (WSe_svSe) sublayers. Based on the different features of the two-sublayers and the choice of the TM elements, the studied systems demonstrate different electronic and magnetic properties.

2. Models and method

All the calculations were performed within the framework of spin polarized DFT as implemented in the Vienna Ab initio Simulation Package (VASP).^{37,38} The exchange-correlation potentials were treated by the generalized gradient approximation (GGA) parameterized by Perdew, Burke and Ernzerhof (PBE).³⁹ For the G/WSe₂ heterostructures, the interaction between valence electrons and ion cores was described by the projector augmented wave (PAW) method⁴⁰ and the DFT-D2 method⁴¹ was taken into account van der Waals (vdW) interaction. It is well accepted that the Coulomb interaction effect (Hubbard U) can help to describe well the magnetism of TM atoms. Comparing with the GGA calculations, the previously tested GGA + U (U = 3.1 eV) calculation in prior literature was proven to give nearly identical results regarding the magnetic features of these graphene/TMDCs systems.^{24,42–44} Therefore, we only explored the electronic and magnetic properties with the GGA method. The TM@(G/WSe₂)s were modelled as a periodic slab by setting the lattice parameters as $a = b = 9.88$ Å (see Fig. 1a–c), in which a 4×4 supercell for the graphene and a 3×3 supercell for the WSe₂ monolayer with the lattice mismatch of 1.1% were employed. The Monkhorst-Pack grid of $9 \times 9 \times 1$ was used for the geometry optimization of G/WSe₂ and a much more dense k point grid of $19 \times 19 \times 1$ was set for electronic properties exploration. To ensure negligible interaction of the molecules from neighbouring unit cells, the lattice constant

along the z axis was set as large as 16 Å. A 400 eV kinetic energy cutoff was chosen for the plane-wave basis set and conjugate gradient (CG) atomic optimization was performed with a criterion of convergence of 0.01 eV Å^{−1}. For the test, we optimized the geometries of monolayer graphene and WSe₂, and the lattice parameters were 2.47 Å and 3.33 Å (see Fig. 1a and b), respectively. On coupling the two sublayers, the interlayer distance of the G/WSe₂ heterostructure is fully relaxed and optimized to be 3.37 Å and the band gap is made as small as approximately 1.0 meV. All the tested results are consistent with previous results,^{24,43} indicating that the computational methods we selected are reasonable.

3. Results and discussion

Four types of TM intercalated G/WSe₂ configurations were considered: (i) TM@(G/WSe₂)s, in which the TM atoms are located between the hollow site of graphene and the top site of a W atom in the WSe₂ layer (see Fig. 1e and f); (ii) and (iii) TM@(G_sv/WSe₂)s and TM@(G_sv₂/WSe₂)s, wherein the graphene layer has a single vacancy and the TM atom is just located between the vacancy of graphene and the top site of a W atom (ii) (see Fig. 1g and h) or the hollow site of the WSe₂ layer (iii) (see Fig. 1i and j); (iv) TM@(G/WSe_svSe)s, where the WSe₂ sublayer has a single Se vacancy and the TM atom is put directly between the vacancy and the hollow site of graphene (see Fig. 1k and l). All the optimized structures are presented in Fig. S1–S4 in the ESI,[†] and most systems intercalated with TM atoms almost maintain their initial configurations. Exceptions are found for the Co and Ni intercalated G/WSe₂, where the graphene layer slides slightly with respect to the WSe₂ sublayer and the TM atoms sit away from the hollow site of graphene. In TM@(G/WSe₂)s, the distances from TM atoms to the graphene layer are around 1.74–2.25 Å, which are much larger than those between TMs and WSe₂ layers, ~ 1.11 – 1.68 Å (see Fig. S5 in the ESI[†]). In addition, introducing one single vacancy to graphene largely shortens the TM–G distances, ~ 0.94 – 1.22 Å, while in the case of TM@(G/WSe_svSe)s, the TM atoms almost fill the Se vacancy.

In general, the charge transfer to graphene and WSe₂ layers decreases gradually from Sc- to Ni-doped systems according to the Bader charge analysis (see Fig. 2a–c). In the case of defective systems, charge transfer to the defective graphene or WSe₂ layers is largely enhanced, which is caused by the shortened distances of TM atoms to the graphene or WSe₂ layer with one single vacancy (see Fig. S5 in the ESI[†]). Fig. 2d–i depict the charge density differences (CDDs) of six Mn/Fe intercalated G/WSe₂ systems, defined as $\Delta\rho = \rho[\text{sys}] - \rho[\text{TM}] - \rho[\text{G}] - \rho[\text{WSe}_2]$, where $\rho[\bullet]$ is the charge density of the whole system, the TM atom, graphene, and WSe₂, respectively. The charges are transferred from TM atoms to the graphene and WSe₂ sublayers, and the TM–G (TM–WSe₂) interaction displays obvious covalent bonding character. The binding energies of these systems are calculated by the equation, $E_b = E_{\text{TM}} + E_{\text{G}} + E_{\text{WSe}_2} - E_{\text{sys}}$, where the E_{TM} , E_{G} , E_{WSe_2} and E_{sys} represent the total energy of the TM atoms, graphene, WSe₂, and the whole system, respectively. All the structures were found to be



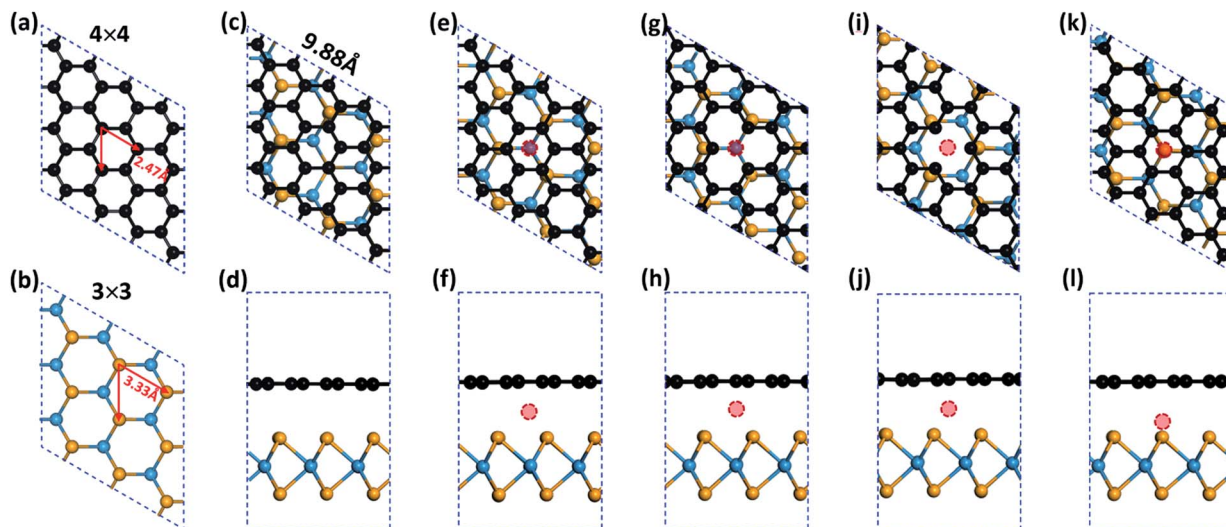


Fig. 1 Schematic illustrations of (a) monolayer graphene, (b) monolayer WSe₂, (c and d) G/WSe₂, (e and f) TM@G/WSe₂, (g and h) TM@G_sv/WSe₂, (i and j) TM@G_sv2/WSe₂, (k and l) TM@G/WSe_svSe heterostructures. Black, yellow, cyan and red balls represent C, Se, W and 3d TM atoms, respectively.

thermodynamically stable for having large binding energies (3.52–11.73 eV) (see Table 1), much larger than those of TM adsorbed graphene.⁴⁵ Of these, the binding energies of the systems with a defective graphene or WSe₂ layer (~5.16–11.73 eV) are much larger than those of TM@G/WSe₂s (~3.52–5.89 eV) and other TM@G/TMD)s.²⁴

Interestingly, versatile electronic and magnetic behaviors are identified for these TM intercalated G/WSe₂ structures (see

Table 1). To accurately explore the real magnetic ground states of such TM@G/WSe₂s, we adopted the $2 \times 1 \times 1$ supercell including two TM atoms to compare the stabilities of their FM and AFM coupling. The TM@G/WSe₂s at TM = Sc–Co demonstrate robust ferromagnetism (FM) with magnetic moments of 1.80 μ_B , 2.46 μ_B , 3.81 μ_B , 4.30 μ_B , 3.05 μ_B , 2.02 μ_B and 1.00 μ_B per unit cell, respectively. In contrast, Ni@G/WSe₂) has a nonmagnetic (NM) ground state. Such magnetic moments

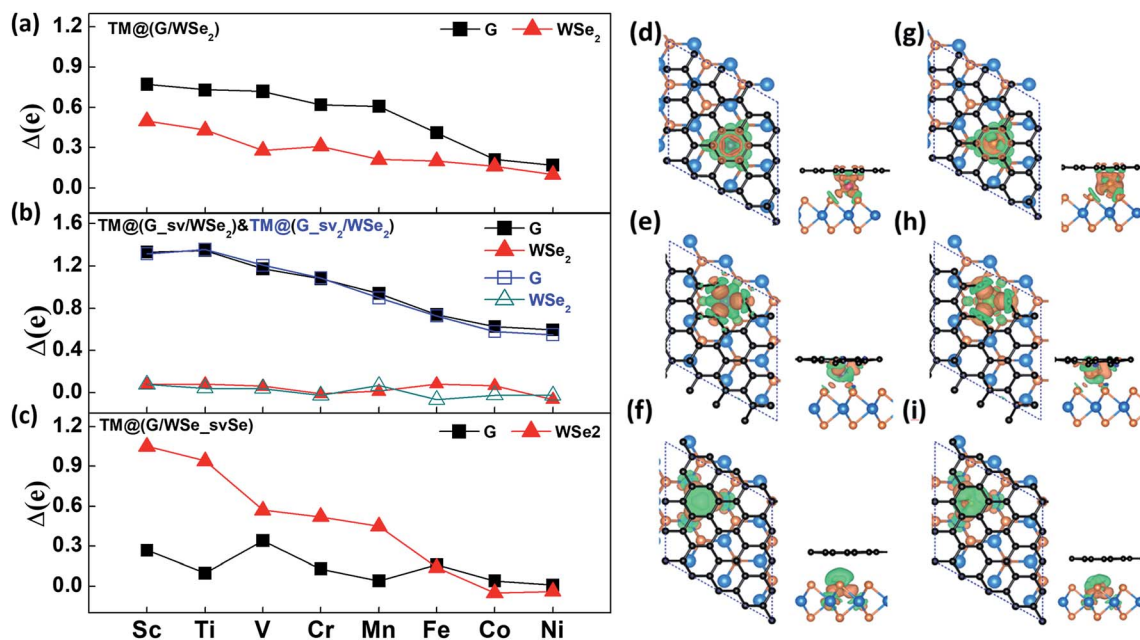


Fig. 2 (a–c) The charges transferred to the graphene layer and the WSe₂ layer in TM@G/WSe₂s, TM@G_sv/WSe₂s, TM@G_sv2/WSe₂s and TM@G/WSe_svSe)s, respectively. (d–i) Top and side views of the CDDs plots for TM@G/WSe₂s, TM@G_sv/WSe₂s and TM@G/WSe_svSe)s (TM = Mn, Fe), respectively. Orange and green isosurfaces denote the accumulation and depletion of electrons, respectively. The isosurface value is $\pm 0.03 \text{ e } \text{\AA}^{-3}$.



Table 1 The binding energies (E_b), magnetic moments (MMs) and ground states (GS) of TM@(G/WSe₂)s, TM@(G_{sv}/WSe₂)s, TM@(G_{sv2}/WSe₂)s and TM@(G/WSe_{sv}Se)s intercalated with different TM atoms. Here "FM" stands for ferromagnetic state, "NM" for nonmagnetic state, "AFM" for antiferromagnetic state, "M" for metal, "SC" for semiconductor and "HM" for half metal

Sys	TM@(G/WSe ₂)			TM@(G _{sv} /WSe ₂)			TM@(G _{sv2} /WSe ₂)			TM@(G/WSe _{sv} Se)		
	E_b (eV)	MM (μ_B)	GS	E_b (eV)	MM (μ_B)	GS	E_b (eV)	MM (μ_B)	GS	E_b (eV)	MM (μ_B)	GS
Sc	5.55	1.80	FM M	9.30	0.72	FM HM	9.16	0.27	FM M	7.54	0	AFM M
Ti	5.89	2.46	FM M	10.87	0	NM SC	10.71	0	NM SC	7.64	0	NM M
V	5.37	3.81	FM HM	10.68	1.00	FM HM	10.38	1.00	FM HM	6.31	0	AFM M
Cr	3.52	4.30	FM M	9.46	2.00	FM SC	9.2	2.00	FM SC	5.16	0	AFM M
Mn	3.61	3.05	FM HM	9.29	3.00	FM M	9.24	3.00	FM M	5.76	3.05	FM M
Fe	5.85	2.02	FM HM	11.73	0	NM SC	11.530	0	NM SC	7.37	0	AFM M
Co	5.23	1.00	FM HM	10.68	0.96	FM HM	10.64	0.88	FM SC	6.35	1.05	FM M
Ni	5.80	0	NM SC	9.79	0	NM SC	9.77	0	NM SC	6.88	0	NM M

are similar to those of TM adsorbed graphene^{45,46} or other TM@(G/TMDC)s.²⁴ The magnetic properties of these TM-intercalated G/WSe₂ heterostructures can be well understood from the partial density of states (PDOS) (see Fig. 3, S6 and S7 in the ESI†). In the crystal field of graphene and WSe₂, the five d orbitals of TM atoms are split into two degenerate (d_{xy} , $d_{x^2-y^2}$) and (d_{xz} , d_{yz}) orbitals and one singly occupied d_{z^2} orbital. Clearly, the TM-G or TM-WSe₂ interaction upshifts the Fermi level of graphene or WSe₂ simultaneously, and the impurity states are formed near the Fermi level. The magnetic moments of these systems are mainly contributed to by the 3d electrons of TM atoms, meaning fewer come from their nearby atoms in the graphene and WSe₂ layers. Moreover, tunable electronic

properties are identified for these TM intercalated G/WSe₂ heterostructures. Similar to those TM@(G/MoS₂)s,²⁴ the TM@(G/WSe₂) at TM = Sc, Ti and Cr are found to be metals, while those with TM = V, Mn, Fe, and Co are half metals (HM) (electrons of one spin orientation act as metals, while those of the opposite orientation act as semiconductors or insulators) having HM gaps of 0.10 eV, 0.13 eV, 0.05 eV and 0.06 eV, respectively (see Fig. 3 and S8a–g in the ESI†). In contrast, Ni@(G/WSe₂) is a perfect semiconductor (SC) with a direct band gap of 0.06 eV (see Fig. S8h in the ESI†), which is much larger than those in pristine G/TMDs heterostructures.^{10,29,30}

Furthermore, introducing one single vacancy in graphene changes the stability, electronic and magnetic properties of

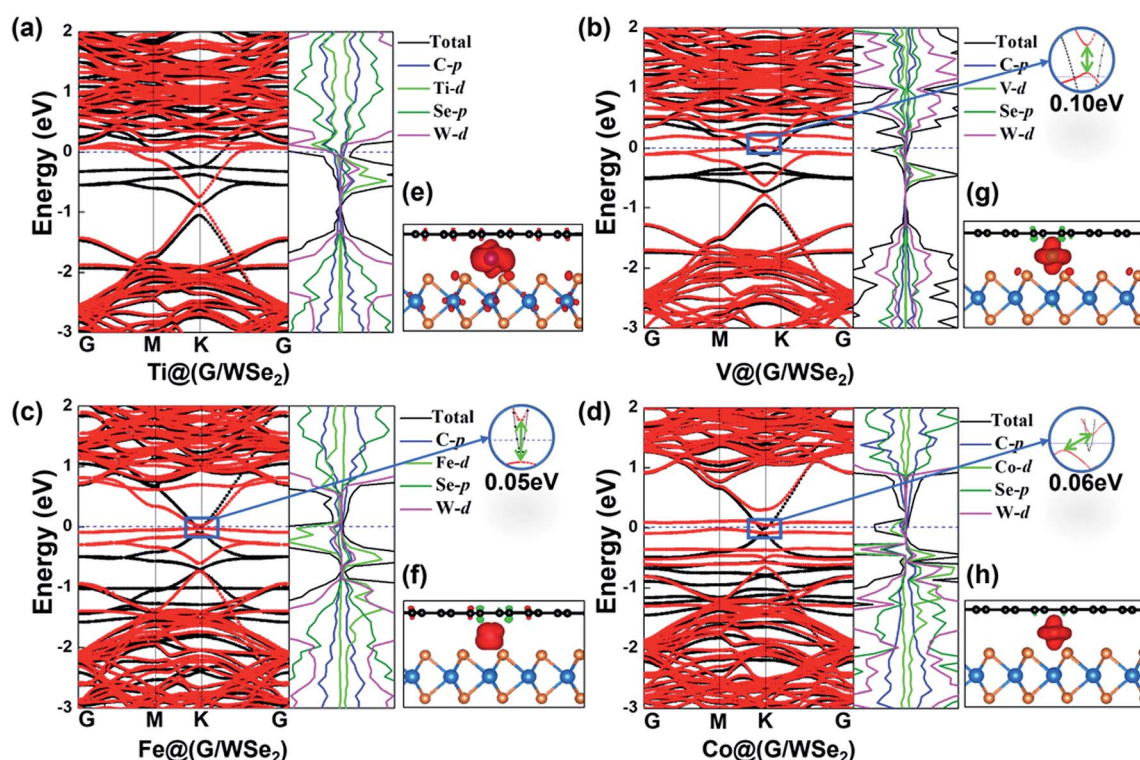
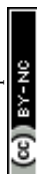


Fig. 3 (a–d) Plots of band structures and partial density of states (PDOS) and (e–h) spin densities of TM@(G/WSe₂)s, TM = Ti, V, Fe, and Co, respectively. Red and black lines represent the bands of spin up and spin down, respectively.



TM@G/WSe₂s significantly (see Table 1, Fig. 4, S9 and S10 in the ESI†). Compared with pristine TM@G/WSe₂s, the binding energies of these graphene-defective systems become much larger, approximately 9.16–11.73 eV, and somewhat larger than those of the TM adsorbed single-vacancy graphene.⁴⁵ Moreover, an FM to NM transition occurs from TM@G/WSe₂s (TM = Ti, Fe) to their defective configurations of TM@G_{sv}/WSe₂s or TM@G_{sv2}/WSe₂s (see Fig. 4a, c, e and g). Apart from this, the TM@G_{sv}/WSe₂s and TM@G_{sv2}/WSe₂s (TM = Sc, V, Cr, Mn, Co) systems are FM. Their total magnetic moments are 0.72 μ_B, 1.00 μ_B, 2.00 μ_B, 3.00 μ_B, 0.96 μ_B and 0.27 μ_B, 1.00 μ_B, 2.00 μ_B, 3.00 μ_B, 0.88 μ_B per unit cell, respectively (see Table 1). In addition, a metal to SC transition occurs for TM@G_{sv}/WSe₂s and TM@G_{sv2}/WSe₂s at TM = Ti, Cr (see Fig. 4a, b, e and f), while a HM to SC transition is found for those doped with Fe atoms (see Fig. 4c and g). Apart from this, the TM@G_{sv}/WSe₂s and TM@G_{sv2}/WSe₂s at TM = V, Co and Ni retain the same electronic properties (see Fig. 4d, h, S9c, g, h, S10c, g and h in the ESI†). The HM gaps for TM@G_{sv}/WSe₂s or TM@G_{sv2}/WSe₂s (TM = V, Co) are 0.36 eV/0.58 eV and 0.29 eV/0.52 eV, respectively. The semiconducting properties of TM@G_{sv}/WSe₂s or TM@G_{sv2}/WSe₂s at TM = Ti, Cr and Fe atoms are similar to their G_{sv}TMG isomers.⁴⁶ Excepting Ni@G_{sv2}/WSe₂, which is a direct band gap SC, all the other SCs have indirect band gaps. It should be mentioned that all the SCs have much larger band gaps (0.10–0.51 eV) than those of G/TMDs and non-defective TM@G/WSe₂s.^{18–21,42} A metal to HM transition occurs for Sc@G_{sv}/WSe₂, while the metallic state is maintained for Sc@G_{sv2}/WSe₂ (see Fig. S9a and S10a in the

ESI†). For both Mn@G_{sv}/WSe₂ and Mn@G_{sv2}/WSe₂, a HM to metal transition is observed (see Fig. S9e and Fig. S10e in the ESI†).

The tunable electronic properties of TM@G/WSe₂, TM@G_{sv}/WSe₂ and TM@G_{sv2}/WSe₂ can be well understood from the electronic configurations near the Fermi level. As shown in Fig. 2b and c, introducing a single vacancy in graphene redistributes the valence electrons of the systems. The impurity states around the Fermi level are mainly from the hybridized d(TM)–π(G) states (Fig. S6 and S7†). For TM@G_{sv}/WSe₂ and TM@G_{sv2}/WSe₂ with TM = Ti, Fe, two hybridized peaks on the majority/minority manifolds of their pristine configurations disappear, meaning that approximately two d electrons from the TM atom are transferred to the C vacant site (see Fig. S6a–c and S7a–c in the ESI†). In addition, the electrons on two spin manifolds distribute evenly, and these systems are thus changed to become NM (see Fig. S6b, c, S7b and c in the ESI†). For the V and Co doped TM@G_{sv}/WSe₂ or TM@G_{sv2}/WSe₂, approximately two flat bands from the minority folds disappear and become empty, leaving the majority bands to stay around the downshifted Fermi level. As a result, they are changed to FM HMs (see Fig. S6e, f, S7e and f in the ESI†). Similar to TM@G_{sv}/WSe₂s or TM@G_{sv2}/WSe₂s, the Ti- and Ni-intercalated G/WSe_{sv}Se are NM and those that are Mn- and Co-intercalated have FM ground states (see Table 1). In contrast, the introduction of an Se vacancy transforms four TM@G/WSe_{sv}Se (TM = Sc, V, Cr, Fe) to AFM (see Table 1). Furthermore, all the studied TM@G/WSe_{sv}Se are found to be metals (see Fig. S11 in the ESI†), wherein the Dirac cones of graphene in most systems are preserved in the Se vacant systems.

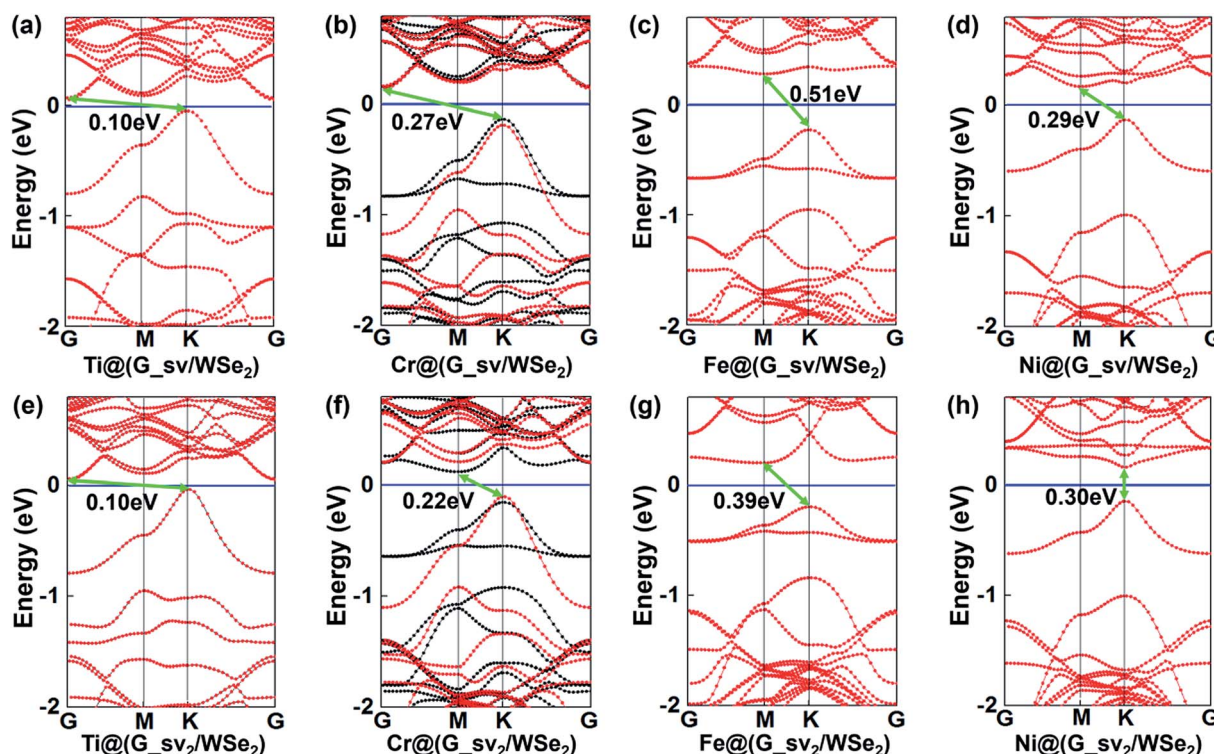
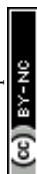


Fig. 4 Band structures of TM@G_{sv}/WSe₂s (a–d) and TM@G_{sv2}/WSe₂s (e–h), TM = Ti, Cr, Fe, Ni, respectively. Red and black lines represent the bands of spin up and spin down, respectively.



Particularly, semimetal is identified for Ni@(G/WSe₂_svSe) (see Fig. S11h in the ESI†).

Finally, let's come to the wide band gaps of half metallic or semiconducting TM@(G_{sv}/WSe₂)s and TM@(G_{sv2}/WSe₂)s. The band structures of free-standing graphene, G/WSe₂ and their defective counterparts are plotted in Fig. 5a–d. Clearly, a very small band gap (~1 meV) is opened for graphene by adding the WSe₂ substrate (see Fig. 5b). Remarkably, making a single vacancy on graphene is an effective way to introduce a sizable band gap. For free standing graphene with a single vacancy, a band gap of 0.28 eV (see Fig. 5c) can be easily obtained. The band gap can be further enlarged to 0.32 eV by adding the WSe₂ substrate (see Fig. 5d). However, the Fermi level of both systems is downshifted, rendering a metal system. Interestingly, if more TM atoms are intercalated, SC or HM with a sizable band gap (both spin or one spin) can be formed (see Fig. 5e and f). Such large gaps for graphene-defective TM@(G_{sv}/WSe₂)s and TM@(G_{sv2}/WSe₂)s systems can be simply understood in terms of the quasi-particle scattering theory.²⁴ As discussed above, there is obvious charge transfer between TM atoms and the graphene or WSe₂ layer (see Fig. 2a–c). The local density of states of graphene in momentum space representation is given by the imaginary part of the Green's function:

$$\rho(\omega) = -\frac{1}{\pi} \text{tr} \sum_{k,k'} \text{Im} [G(k, k', \omega)] \quad (1)$$

The Green's function of graphene in the presence of a single impurity

$$G(k, k', \omega) = \delta_{k,k'} G_0(k) + \sum_{m,m'} [G_0(k) V^m(k) G_{fmm'}(\omega) V^{m'}(k') G_0(k')] \quad (2)$$

where

$$G_0(k, \omega) = [i\omega - H_g(k)]^{-1} \quad (3)$$

$$G_{fmm'}(\omega) = \left[\omega - \varepsilon_0 \delta_{mm'} - \sum_{fmm'} (\omega) + i0^+ \right]^{-1} \quad (4)$$

with

$$\sum_{fmm'} (\omega) = \sum_k V^m(k) G_{fmm'}(\omega) V^{m'}(k) \quad (5)$$

In graphene,

$$H_g(k) = \begin{pmatrix} 0 & t\varnothing(k) \\ t\varnothing(k)^* & 0 \end{pmatrix} \quad (6)$$

where $\varnothing(k) = \sum_{R_i} e^{ikR_i}$, $V^m(k) = \sum_{R_i} V^m e^{ikR_i}$.

Using the local density of states of graphene, we find that the gaps of graphene are moderately opened by the impurity atoms. As shown in Fig. 1, our systems are studied based on the 4×4 supercell lattice, and accordingly, the scattering potential of the impurity atom in the lattice can be viewed as a periodic potential. As a result, the gap in the local density of states is a natural result. The new result of our findings is that the opened gap depends on the electron transfer between the impurity and graphene. For Ni@(G/WSe₂), with small electron

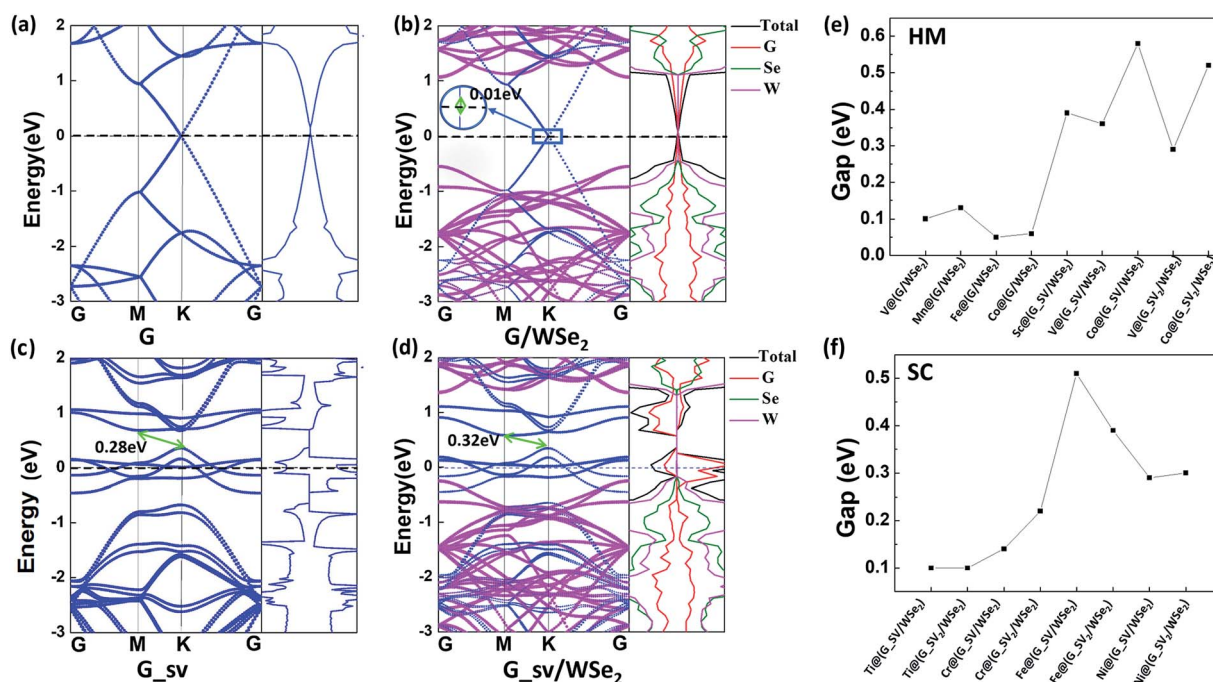


Fig. 5 (a–d) Plots of band structures and DOS of pristine graphene (a), G/WSe₂ (b), graphene with a single vacancy (c) and G_{sv}/WSe₂ (d). (e and f) Band gaps of HM and SC systems.



transfer between the Ni atom and graphene, the band gap is relatively small. However, in the case of TM@(G_{sv}/WSe_2)s and TM@(G_{sv_2}/WSe_2)s systems (TM = Ti, Cr, Fe, Ni), significant electron transfer between the TM atom and graphene is obtained, accounting for large band gap opening (see Fig. 2).

4. Conclusions

Using the density functional theory method, we have systematically studied the structural, electronic and magnetic properties of 3d transition metal atom-intercalated pristine and defective G/WSe_2 heterostructures. Obvious covalent bonding characters are identified for the TM-graphene and TM- WSe_2 interactions. Different from perfect TM@(G/WSe_2)s, introducing a single vacancy in the graphene layer or the WSe_2 layer stabilizes the chemical bonding between the TM atoms and graphene layers. Interestingly, versatile electronic and magnetic properties are found. For the G/WSe_2 s having defective graphene layers, the systems doped with Ti, Fe and Ni atoms can lead to nonmagnetic semiconductors, while those with Cr atoms result in ferromagnetic semiconductors. All these proposed configurations show sizable band gaps ranging from 0.1 eV to 0.51 eV. In addition, V and Mn doped non-defective G/WSe_2 and Sc, V, Co doped defective G/WSe_2 can lead to sizable half metallic band gaps ranging from 0.1 eV to 0.58 eV. Our findings propose an effective way to manipulate the electronic and magnetic properties of graphene/TMD heterostructures.

Conflicts of interest

There are no conflicts of interest to declare.

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References

- 1 J. M. Hamm and O. Hess, *Science*, 2013, **340**, 1298–1299.
- 2 A. K. Geim and I. V. Grigorieva, *Nature*, 2013, **499**, 419.
- 3 G. Gao, W. Gao, E. Cannuccia, J. Taha-Tijerina, L. Balicas and A. Mathkar, *Nano Lett.*, 2012, **12**, 3518.
- 4 L. Britnell, R. V. Gorbachev, R. Jalil, B. D. Belle, F. Schedin, A. Mishchenko and T. Georgiou, *Science*, 2012, **335**, 947.
- 5 C. J. Shih, Q. H. Wang, Y. Son, Z. Jin, D. Blankschtein and M. S. Strano, *ACS Nano*, 2014, **8**, 5790.
- 6 M. S. Choi, G. H. Lee, Y. J. Yu, D. Y. Lee, S. H. Lee, P. Kim, W. J. Hone and J. Yoo, *Nat. Commun.*, 2013, **4**, 1624.
- 7 H. Wang, L. Yu, Y. H. Lee, Y. Shi, A. Hsu, M. L. Chin, L. J. Li, M. Dubey, J. Kong and T. Palacios, *Nano Lett.*, 2012, **12**, 4674.
- 8 Y. X. Deng, Z. Luo, N. J. Conrad, H. Liu, Y. J. Gong, S. Najmaei, P. M. Ajayan, J. Lou, X. F. Xu and P. D. Ye, *ACS Nano*, 2014, **8**, 8292.
- 9 H. J. Ling, C. Zhang, Y. P. Feng and L. Shen, *Appl. Phys. Lett.*, 2016, **108**, 122105.
- 10 K. Chang and W. X. Chen, *Chem. Commun.*, 2011, **47**, 4252.
- 11 J. Y. Tan, A. Avsar, J. Balakrishnan, G. K. W. Koon and T. Taychatanapat, *Appl. Phys. Lett.*, 2014, **104**, 183504.
- 12 J. E. Padilha, A. Fazzio and J. R. Antônio da Silva, *Phys. Rev. Lett.*, 2015, **114**, 066803.
- 13 W. Hu, T. Wang and J. L. Yang, *J. Mater. Chem. C*, 2015, **3**, 4756.
- 14 G. C. Guo, D. Wang, X. L. Wei, Q. Zhang, H. Liu, W. M. Lau and L. M. Liu, *J. Phys. Chem. Lett.*, 2015, **6**, 5002.
- 15 G. Giovannetti, P. A. Khomyakov, G. Brocks, P. J. Kelly and J. van den Brink, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **76**, 073103.
- 16 Y. C. Fan, M. W. Zhao, Z. H. Wang, X. J. Zhang and H. Y. Zhang, *Appl. Phys. Lett.*, 2011, **98**, 083103.
- 17 A. Hashmi, U. Farooq and J. Hong, *Curr. Appl. Phys.*, 2016, **16**, 318.
- 18 Y. D. Ma, Y. Dai, W. Wei, C. W. Niu, L. Yu and B. B. Huang, *J. Phys. Chem. C*, 2011, **115**, 20237.
- 19 C. J. Jin, F. A. Rasmussen and K. S. Thygesen, *J. Phys. Chem. C*, 2015, **119**, 19928.
- 20 S. S. Li and C. W. Zhang, *J. Appl. Phys.*, 2013, **114**, 183709.
- 21 Z. L. Wang, Q. Chen and J. L. Wang, *J. Phys. Chem. C*, 2015, **119**, 4752.
- 22 H. K. Jeon, M. Maruyama, K. J. Kawahara, Y. Terao, D. Ding, R. Matsumoto, Y. Matsuo, S. Okada and H. Ago, *Adv. Mater.*, 2017, **29**, 1702141.
- 23 X. Y. Zhang, X. L. Zhao and Y. J. Liu, *J. Phys. Chem. C*, 2016, **120**, 22710.
- 24 X. Y. Zhang, Z. J. Bao, X. S. Ye, W. X. Xu and Y. J. Liu, *Nanoscale*, 2017, **9**, 11231.
- 25 M. Rafique, Y. Shuai, I. Ahmed, R. Shaikh, M. AliTunio and H. P. Tan, *RSC Adv.*, 2018, **8**, 23688.
- 26 X. S. Zhou, L. J. Wan and Y. G. Guo, *Chem. Commun.*, 2013, **4**, 1838.
- 27 X. Y. Zhang, Z. W. Xv, Q. H. Yuan, X. John and F. Ding, *Nanoscale*, 2015, **7**, 20082.
- 28 Y. Shi, W. Zhou, A. Y. Lu, W. Fang, Y. H. Lee, A. L. Hsu, S. M. Kim, K. K. Kim, H. Y. Yang, L. J. Li, J. C. Idrobo and J. Kong, *Nano Lett.*, 2012, **12**, 2784.
- 29 L. L. Yu, Y. H. Lee, X. Ling, J. G. Santos Elton, Y. C. Shin, Y. X. Lin, M. Dubey, E. Kaxiras, J. Kong, H. Wang and T. Palacios, *Nano Lett.*, 2014, **14**, 3055.
- 30 T. Georgiou, R. Jalil, B. D. Belle, L. Britnell, R. V. Gorbachev, S. V. Morozov, Y. J. Kim, A. Gholinia, S. J. Haigh and O. Makarovskiy, *Nat. Nanotechnol.*, 2013, **8**, 100.
- 31 Y. C. Lin, N. Lu, L. J. Nestor Perea-Lopez, Z. Lin, X. Peng, C. H. Lee, C. Sun, L. Calderin, P. N. Browning, M. S. Bresnehan, M. J. Kim, T. S. Mayer, M. Terrones and J. A. Robinson, *ACS Nano*, 2014, **8**, 3715.
- 32 F. Banhart, J. Kotakoski and A. V. Krashenninnikov, *ACS Nano*, 2011, **5**, 26.
- 33 O. V. Yazyev and L. Helm, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **75**, 125408.
- 34 B. Liu, M. Fathi, L. Chen, A. Abbas, Y. Ma and C. Zhou, *ACS Nano*, 2015, **9**, 6119.



- 35 J. Y. Chen, B. Liu, Y. P. Liu, W. Tang, C. T. Nai, L. J. Li, J. Zheng, L. B. Gao, Y. Zheng, H. S. Shin, H. Y. Jeong and K. P. Loh, *Adv. Mater.*, 2015, **27**, 6722.
- 36 Q. L. Feng, M. J. Zhu, Y. h. Zhao, H. Y. Liu, M. Li, J. B. Zheng, H. Xu and Y. M. Jiang, *Nanotechnology*, 2019, **30**, 034001.
- 37 G. Kresse and J. Hafner, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1993, **48**, 13115.
- 38 G. Kresse and J. Furthmüller, *Comput. Mater. Sci.*, 1996, **6**, 15.
- 39 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 40 P. E. Blochl, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **50**, 17953.
- 41 S. J. Grimme, *Comput. Chem.*, 2006, **27**, 1787.
- 42 T. O. Wehling, A. I. Lichtenstein and M. I. Katsnelson, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **84**, 235110.
- 43 S. Li, Z. M. Ao, J. J. Zhu, J. C. Ren, J. Yi, G. Wang and W. Liu, *J. Phys. Chem. Lett.*, 2017, **8**, 1484.
- 44 G. L. Zhang, R. L. Zhou, Y. Gao and X. C. Zeng, *J. Phys. Chem. Lett.*, 2012, **3**, 151–156.
- 45 X. J. Liu, C. Z. Wang, Y. X. Yao, W. C. Lu, M. Hupalo, A. C. Tringides and K. M. Ho, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **83**, 235411.
- 46 H. Johll and H. C. Kang, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2009, **79**, 245416.

