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Enhancing the electronic and optical properties of the metal/semiconductor NbS₂/BSe nanoheterostructure towards advanced electronics

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Metal–semiconductor (M–S) contacts play a vital role in advanced applications, serving as crucial components in ultracompact devices and exerting a significant impact on overall device performance. Here, in this work, we design a M–S nanoheterostructure between a metallic NbS₂ monolayer and a semiconducting BSe monolayer using first-principles prediction. The stability of such an M–S nanoheterostructure is verified and its electronic and optical properties are also considered. Our results indicate that the NbS₂/BSe nanoheterostructure is structurally, mechanically and thermally stable. The formation of the NbS₂/BSe heterostructure leads to the generation of a Schottky contact with the Schottky barrier ranging from 0.36 to 0.51 eV, depending on the stacking configurations. In addition, the optical absorption coefficient of the NbS₂/BSe heterostructure can reach up to $5 \times 10^5 \text{ cm}^{-1}$ at a photon energy of about 5 eV, which is still greater than that in the constituent NbS₂ and BSe monolayers. This finding suggests that the formation of the M–S NbS₂/BSe heterostructure gives rise to an enhancement in the optical absorption of both NbS₂ and BSe monolayers. Notably, the tunneling probability and the contact tunneling-specific resistivity at the interface of the NbS₂/BSe heterostructure are low, indicating its applicability in emerging nanoelectronic devices, such as Schottky diodes and field-effect transistors. Our findings offer valuable insights for the practical utilization of electronic devices based on the NbS₂/BSe heterostructure.

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1 Introduction

Two-dimensional (2D) materials, including graphene,¹ phosphorene² and transition metal dichalcogenides,³ have received much attention from the research community owing to their exceptional properties and diverse potential applications. As the first successfully fabricated 2D material, graphene emerges as a promising candidate in various next-generation applications, including electronic,⁴ optoelectronic⁵ and spintronic⁶ applications. However, the absence of a desirable band gap in graphene is crucial, especially in the design of high-frequency applications, where it plays a significant role. Hence, many different strategies, such as doping,⁷ functionalization⁸ and strain⁹ have been proposed to open an intrinsic band gap in graphene. It's worth noting that these strategies may potentially result in

a reduction in the carrier mobility of graphene.¹⁰ Therefore, seeking 2D materials with an appropriated band gap for high-speed applications remains a challenging endeavor.

Recently, the construction of a van der Waals (vdW) heterostructure by stacking two or more 2D materials has emerged as a common strategy to improve the properties and extend the application possibilities of 2D materials.^{11,12} vdW heterostructures can be synthesized in experiments by the chemical vapor deposition (CVD) method,^{13,14} one-step growth¹⁵ or two-step growth.¹⁶ Additionally, these heterostructures also can be predicted through first-principles calculations.^{17,18} Following the success of both experimental and theoretical endeavors, many different vdW heterostructures have been predicted and synthesized, such as graphene-based heterostructures,^{19–23} phosphorene-based heterostructures^{24–26} and MA₂Z₄-based heterostructures.^{27–29} The 2D vdW heterostructures can be categorized into metal/semiconductor (M–S) and semiconductor/semiconductor heterostructures (S–S), which depend on the characteristics of 2D materials. It's worth noting that the M–S heterostructure is one of the most crucial components in electronic devices. Hence, the characteristics of the M–S heterostructure significantly impact the performance of electronic devices. Therefore, there is growing attention on the search for M–S heterostructures towards next-generation electronic devices aiming for high performance.

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Recently, many different M–S heterostructures have been proposed and investigated, involving the combination of both traditional 3D metals or novel 2D metals with 2D semiconductors, such as 3D and 2D metals/MSi₂N₄ (M = Mo, W)³⁰ and 3D metals/MoS₂.³¹ Notably, the traditional 3D metals/semiconductor often exhibit unmodulated Schottky barriers, leading to high contact resistance and lower carrier mobility, thus diminishing device performance. In contrast, the 2D metals/semiconductor heterostructures may feature lower Schottky barriers and they are also tunable under various conditions, including strain and electric fields.^{32–36} Hence, there is a growing focus on the discovery of novel 2D metals and semiconductors, which can be combined to form an M–S contact with enhanced performance.

Currently, a 2D metal NbS₂ monolayer has received tremendous consideration as it can serve as a promising metallic electrode for novel M–S heterostructures.^{37–40} 2D metallic NbS₂ was synthesized in recent experiments by different methods, such as chemical vapor deposition^{41,42} and mechanical/chemical exfoliation.⁴³ The electronic and optical properties of the NbS₂ monolayer have also been investigated through first-principles calculations⁴⁴ along with its tunable properties *via* doping⁴⁵ and intercalations.⁴⁶ All these findings suggest that a metallic NbS₂ monolayer can be considered as a promising material for future applications, including gas sensors^{47,48} and energy storage.⁴⁹ Furthermore, a promising BSe semiconducting monolayer has been predicted to be mechanically and thermally stable at room temperature.⁵⁰ It is evident that the BSe monolayer exhibits a semiconducting feature with an indirect band gap. Additionally, the BSe monolayer is considered as a promising semiconductor for forming contacts with various other 2D materials such as graphene,⁵¹ phosphorene,⁵² MoS₂ (ref. 17) and so forth.^{53,54} However, to date, a combination between a BSe semiconductor and other 2D metals has not yet been extensively predicted and investigated. Therefore, in this work, we perform first-principles calculations to design a M–S heterostructure by combining 2D metal NbS₂ and a BSe semiconductor. Our findings offer valuable insights for the practical utilization of electronic devices based on the NbS₂/BSe heterostructure.

2 Computational model and methods

In this work, we performed first-principles calculations based on the density functional theory (DFT). We use the Quantum Espresso simulation package⁵⁵ in the framework of the generalized gradient approximation (GGA).⁵⁶ In addition, we employed the Perdew–Burke–Ernzerhof (PBE) functional⁵⁷ to calculate the exchange and correlation energy with the projected augmented wave (PAW) pseudopotential.⁵⁸ The vdW interactions that may exist in layered nanostructures can be described by using the Grimme correction DFT-D3 method.⁵⁹ The Heyd–Scuseria–Ernzerhof (HSE) functional⁶⁰ is also used to achieve more accurate band gaps of 2D semiconductors. A cut-off energy of 510 eV and a (9 × 9 × 1) *k*-point mesh are used in all our calculations. A vacuum thickness of 30 Å is included to prevent unnecessary interactions caused by the periodic

boundary conditions. Dipole correction is also applied in all calculations.

3 Results and discussion

We first examine the atomic and electronic structures of NbS₂ and BSe monolayers, as illustrated in Fig. 1. Both the NbS₂ and BSe monolayers show hexagonal atomic crystals. In the NbS₂ monolayer, one Nb atom is sandwiched between two S atoms on both sides, whereas in the BSe monolayer, each B atom is bonded with one Se atom on one side. The calculated lattice parameters of the NbS₂ and BSe monolayers are 3.31 and 3.23 Å, respectively. These results are in good agreement with previous reports.^{50,61} The NbS₂ monolayer is characterized by its metallic properties, featuring a band that crosses the Fermi level. In contrast, the BSe monolayer exhibits a semiconducting behavior with an indirect band gap of 2.62/3.46 eV, obtained by using the PBE/HSE functional. These values are in good agreement with previous measurements.⁵⁰ The valence band maximum (VBM) and conduction band minimum (CBM) of the BSe monolayer are located at the Γ point and Γ –M path, respectively. Interestingly, we find that both the PBE and HSE functionals predict the similar behavior of NbS₂ and BSe monolayers. The main difference between PBE and HSE methods is in a shift of the VBM of the BSe monolayer. The VBM of the BSe monolayer in the PBE functional is positioned closer to the Fermi level than that in the HSE functional.

We now combine the NbS₂ and BSe monolayers to generate the metal/semiconductor NbS₂/BSe heterostructure by stacking NbS₂ on top of the BSe monolayer. The possible stacking configurations of the NbS₂/BSe heterostructure are depicted in Fig. 2. Due to a small difference in the lattice parameters between NbS₂ and BSe monolayers, the NbS₂/BSe heterostructure has a small lattice mismatch of only 1.2% for all stacking configurations. After the geometric optimization process, the interlayer spacing *d* between NbS₂ and BSe layers in

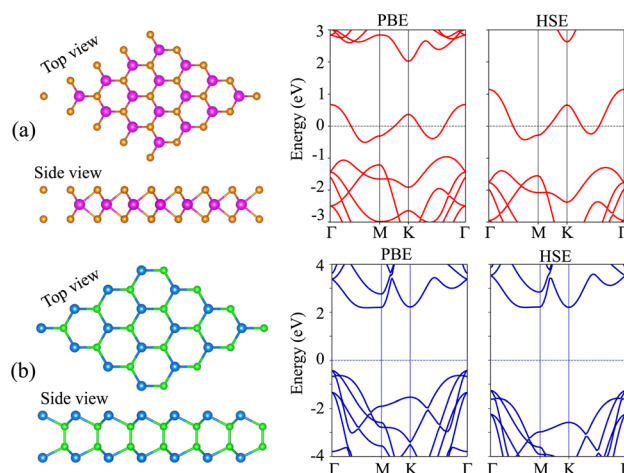


Fig. 1 (From left to right) Atomic structure, PBE and HSE band structures of (a) NbS₂ and (b) BSe monolayers. Orange and purple balls represent the S and Nb atoms, respectively. Dark green and green balls stand for the Se and B atoms, respectively.



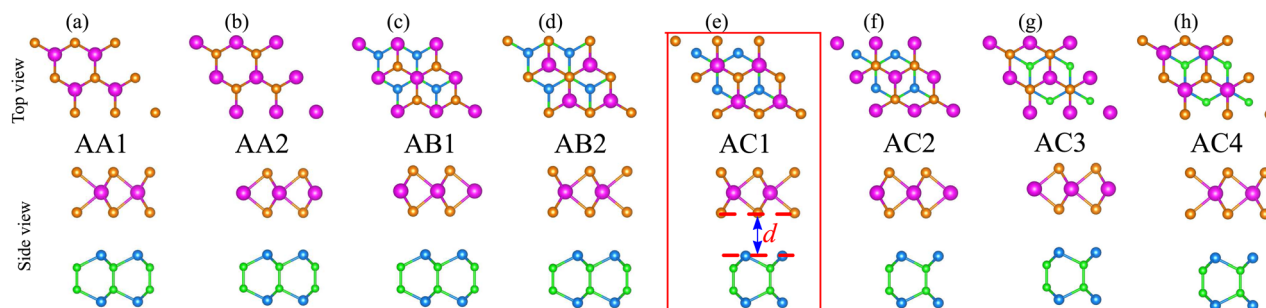


Fig. 2 Atomic structures of the NbS₂/BSe heterostructure for possible different stacking configurations of (a) AA1, (b) AA2, (c) AB1, (d) AB2, (e) AC1, (f) AC2, (g) AC3 and (h) AC4.

their heterostructure can be obtained, as listed in Table 1. It is evident that the interlayer spacing in the AC1 stacking configuration is $d = 2.95 \text{ \AA}$, which is the shortest among the eight stacking configurations. In addition, it should be noted that the interlayer spacing d in all stacking configurations is larger than the sum of covalent radii between S (0.99 \AA) and Se (1.14 \AA) atoms, confirming that there is no covalent bond between the two constituent NbS₂ and BSe layers. Furthermore, to examine the interfacial stability of the NbS₂/BSe heterostructure for all stacks, we calculate the binding energy as follows:

$$E_b = \frac{E_{\text{NbS}_2/\text{BSe}} - E_{\text{NbS}_2} - E_{\text{BSe}}}{S} \quad (1)$$

Here, $E_{\text{NbS}_2/\text{BSe}}$, E_{NbS_2} and E_{BSe} are the total energies of the NbS₂/BSe heterostructure, and isolated NbS₂ and BSe monolayers, respectively. $S = 10.69 \text{ \AA}^2$ is the surface area of the heterostructure. The calculated E_b for all stacks of the NbS₂/BSe heterostructure is listed in Table 1. The binding energy of the NbS₂/BSe heterostructure is approximately averaged at about -20 meV \AA^{-2} . The negative binding energy suggests that the NbS₂/BSe heterostructure is stable. Interestingly, this binding energy is comparable to that in graphite⁶² and other typical 2D van der Waals (vdW) heterostructures.^{63,64} This finding indicates that the NbS₂/BSe heterostructure can be synthesized in future experiments by different strategies, including mechanical exfoliation¹² and CVD.⁶⁵ For instance, using the CVD method, Fu *et al.*³⁷ successfully synthesized a metal/semiconductor NbS₂/MoS₂ heterostructure with high quality and a clean interface. Among these stacks, the AC2 stack has the lowest binding

energy of $-23.89 \text{ meV \AA}^{-2}$. The shortest interlayer spacing and the lowest binding energy in the AC1 stacking configuration of the NbS₂/BSe heterostructure indicate that the AC1 stack is the most energetically feasible stacking configuration.

Furthermore, to check the mechanical stability of the NbS₂/BSe heterostructure, we consider its independent elastic constants (C_{ij}). The C_{ij} of the perfect NbS₂ and BSe monolayers are also calculated for comparison, as illustrated in Fig. 3. The calculated elastic constants C_{11} , C_{12} and C_{66} of monolayers NbS₂ and BSe are 183.31, 28.40, 77.45 and 120.10, 34.91 and 42.59 N m^{-1} , respectively. According to the Born–Huang stability criteria,⁶⁶ *i.e.* $C_{11} > C_{12}$, $C_{66} > 0$ and $C_{11}^2 - C_{12}^2 > 0$, both the NbS₂ and BSe monolayers are mechanically stable. Interestingly, the formation of the NbS₂/BSe heterostructure leads to an increase in the elastic constants C_{ij} , as shown in Fig. 3(a). The calculated C_{11} , C_{12} and C_{66} of the NbS₂/BSe heterostructure are about 300, 65 and 130 N m^{-1} , respectively. These values of the elastic constants are still larger than those of the NbS₂ and BSe monolayers. In addition, the elastic constants of the NbS₂/BSe heterostructure meet the Born–Huang stability criteria. This finding suggests that the NbS₂/BSe heterostructure exhibits outstanding mechanical stability. Furthermore, the angle-dependent Young's modulus of NbS₂ and BSe monolayers and their combined NbS₂/BSe heterostructure for the most energetically favorable stacking configuration are depicted in Fig. 3(b). We find that the Young's modulus of the NbS₂/BSe heterostructure is calculated to be 289.74 N m^{-1} , which is greater than that of the NbS₂ (109.95 N m^{-1}) and BSe (178.90 N m^{-1}) monolayers. The phonon dispersion curves of the NbS₂/BSe heterostructure are depicted in Fig. 3(c). We observe that there are no imaginary frequencies in the phonon spectrum of the NbS₂/BSe heterostructure, indicating that such a heterostructure is dynamically stable. All these findings indicate that the NbS₂/BSe heterostructure can be considered as a promising candidate for the design of electronic devices.

The projected band structures of the NbS₂/BSe heterostructure for all stacks are depicted in Fig. 4, in which red and blue bubbles show the contributions of the NbS₂ and BSe layers, respectively. It is obvious that the band structures of the NbS₂/BSe heterostructure appear to be a combination of the band structures of the constituent monolayers. The reason for such a combination is weak interactions between the NbS₂ and BSe layers. These weak interactions play a pivotal role in stabilizing

Table 1 Calculated interlayer spacing (d , \AA), binding energy (E_b , meV \AA^{-2}), and Schottky barrier heights (Φ_n and Φ_p , eV) of the NbS₂/BSe heterostructure

Stacks	D	E_b	Φ_n	Φ_p	Contact types
AA1	3.11	-22.16	2.12	0.36	p-type ShC
AA2	3.58	-14.50	2.05	0.51	p-type ShC
AB1	2.95	-23.88	1.97	0.42	p-type ShC
AB2	3.01	-23.47	2.0	0.43	p-type ShC
AC1	2.95	-23.89	1.97	0.42	p-type ShC
AC2	3.01	-23.47	2.01	0.42	p-type ShC
AC3	3.58	-14.67	2.05	0.51	p-type ShC
AC4	3.04	-23.01	2.10	0.34	p-type ShC



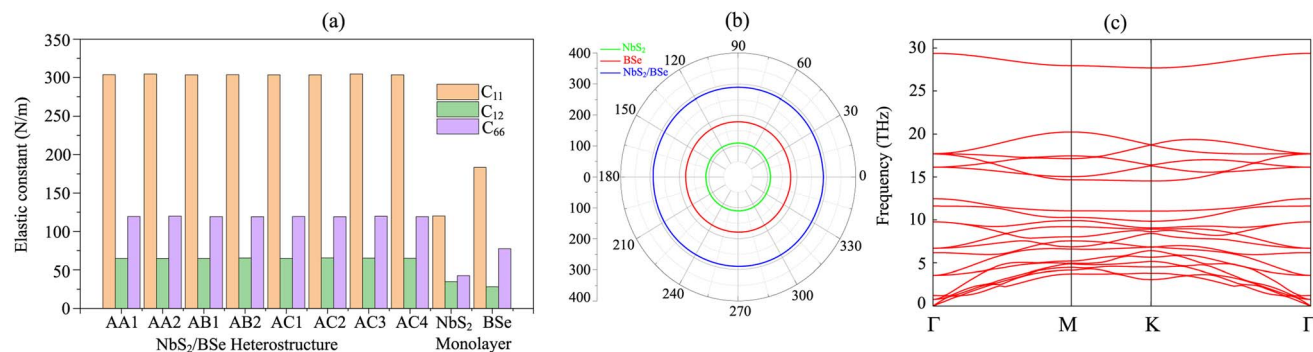


Fig. 3 (a) Calculated independent elastic constants (C_{ij}) of the NbS₂/BSe heterostructure for different stacking configurations and (b) angle-dependent Young's modulus and (c) phonon dispersion curves of the NbS₂/BSe heterostructure for the most energetically favorable stacking configuration.

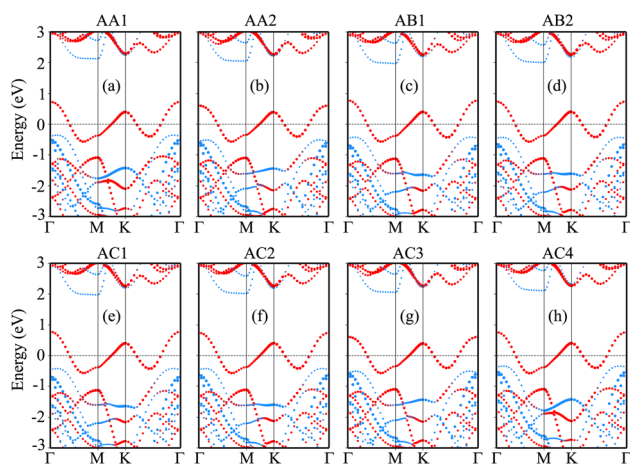


Fig. 4 Projected band structures of the NbS₂/BSe heterostructure for different stacks of (a) AA1, (b) AA2, (c) AB1, (d) AB2, (e) AC1, (f) AC2, (g) AC3 and (h) AC4. Red and blue bubbles represent the contributions of the NbS₂ and BSe layers, respectively.

the heterostructure, making it readily obtainable in experiments. Furthermore, the combination between the metallic NbS₂ monolayer and semiconducting BSe monolayer might give rise to the formation of either a Schottky or ohmic contact, depending on the position of the band edges of the BSe semiconductor in relation to the Fermi level of the metallic NbS₂ layer. As depicted in Fig. 4, the Fermi level of the metallic NbS₂ monolayer lies between the band edges of the semiconductor BSe monolayer. This arrangement leads to the formation of a Schottky contact. In a Schottky contact, Schottky barriers exist, and these barriers can be determined as follows:

$$\Phi_n = E_{\text{CBM}} - E_F \quad (2)$$

and

$$\Phi_p = E_F - E_{\text{VBM}} \quad (3)$$

Here, E_{CBM} and E_{VBM} are the band edge energies of the CBM and VBM of the BSe semiconductor, respectively. E_F is the Fermi level of the heterostructure. The obtained Schottky barriers Φ_n

and Φ_p of the heterostructure are listed in Table 1 and Fig. 5. It is evident that the Schottky barrier Φ_p of the heterostructure is always narrower than the Schottky barrier Φ_n , indicating that the NbS₂/BSe heterostructure tends to exhibit p-type ShC. The p-type Schottky barrier in the NbS₂/BSe heterostructure ranges from 0.34 to 0.51 eV. The AC4 stacking pattern exhibits the narrowest Schottky barrier Φ_p of 0.34 eV, while the AA2 and AC3 stacking patterns exhibit the highest Schottky barrier Φ_p of 0.51 eV. As previously discussed, the AC1 stacking pattern of the NbS₂/BSe heterostructure is the most favorable among the stacking configurations. Hence, we will focus on this stacking configuration in our subsequent calculations.

The projected band structures of the NbS₂/BSe heterostructure for the most energetically favorable stacking patterns using PBE and HSE methods are illustrated in Fig. 6(a). The calculated Schottky barriers Φ_n and Φ_p of the NbS₂/BSe heterostructure are 1.97/2.88 and 0.42/0.43 eV, obtained using the PBE/HSE functional. It is clear that both the PBE and HSE functionals predict the similar behavior of the NbS₂/BSe heterostructure. The CBM calculated using the HSE functional of the NbS₂/BSe heterostructure is positioned further away from the

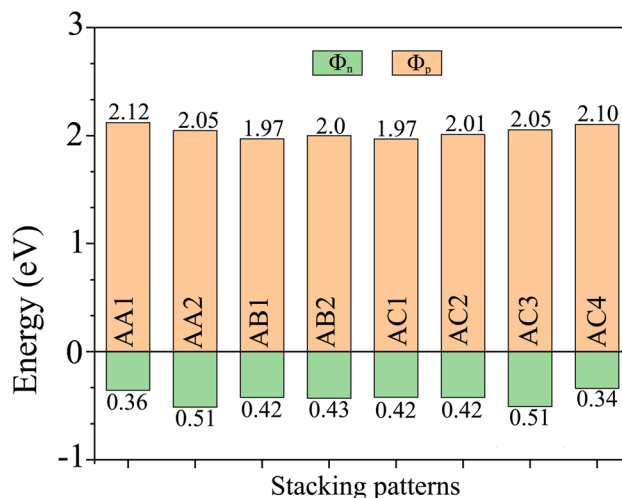


Fig. 5 The Schottky barriers of the NbS₂/BSe heterostructure for different stacking patterns.



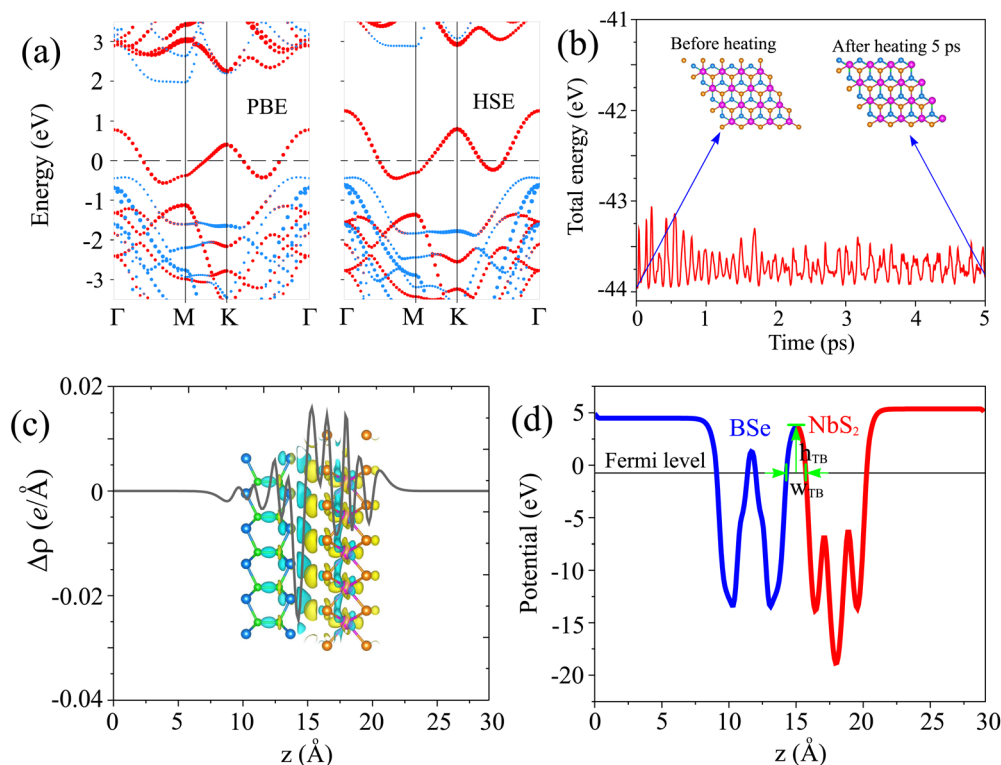


Fig. 6 (a) Projected band structures, (b) AIMD simulation, (c) charge density difference and (d) electrostatic potential of the AC1 stacking pattern of the NbS₂/BSe heterostructure. Red and blue lines represent the contribution of metallic NbS₂ and semiconducting BSe monolayers, respectively. The inset of (c) shows the atomic structure of the heterostructure before and after heating for 5 ps. Yellow and cyan regions in (c) indicate the charge accumulation and depletion, respectively.

Fermi level than that calculated by using the PBE functional. Interestingly, it's worth noting that both the PBE and HSE functional predictions indicate a p-type Schottky contact with the same Schottky barrier Φ_p . This suggests that the PBE functional can be effectively used in the subsequent calculations without compromising the accuracy of the results. Furthermore, in order to verify the thermal stability of the NbS₂/BSe heterostructure, we perform *ab initio* molecular dynamics (AIMD) simulation. The AIMD simulation of the NbS₂/BSe heterostructure is displayed in Fig. 6(b). It is evident that the fluctuation in the total energy of the NbS₂/BSe heterostructure as a function of time steps is small. In addition, the atomic structure of the NbS₂/BSe heterostructure after heating for 5 ps is preserved without any distortion. All these findings confirm that the NbS₂/BSe heterostructure is thermally stable. The stability indicates that the NbS₂/BSe heterostructure could be synthesized and used in recent nanoelectronic and optoelectronic devices, such as Schottky diodes, field-effect transistors and photodetectors.

Furthermore, the charge transfers between the NbS₂ and BSe layers in their heterostructure are also explored by calculating the charge density difference (CDD) as follows:

$$\Delta\rho = \rho_{\text{NbS}_2/\text{BSe}} - \rho_{\text{NbS}_2} - \rho_{\text{BSe}} \quad (4)$$

Here, the charge densities of the NbS₂/BSe heterostructure and isolated NbS₂ and BSe monolayers are denoted as $\rho_{\text{NbS}_2/\text{BSe}}$, ρ_{NbS_2}

and ρ_{BSe} , respectively. The CDD of the NbS₂/BSe heterostructure is depicted in Fig. 6(c). One can find that the charges are accumulated mainly around the NbS₂ layer and depleted mainly around the BSe layer. This finding implies that the electrons are transferred from the NbS₂ to the BSe layer in their corresponding NbS₂/BSe heterostructure. Based on Bader charge analysis, there is a small amount of charge transfer of only 0.02 electrons that flow from the NbS₂ to the BSe layer.

The electrostatic potential of the NbS₂/BSe heterostructure is depicted in Fig. 6(d). One can observe that the NbS₂ layer has a deeper potential than the BSe layer in their heterostructure. The charge transfer between metallic NbS₂ and BSe layers at the interface of the NbS₂/BSe heterostructure gives rise to the formation of an interfacial dipole, which can be obtained using $\Delta V = W_{\text{NbS}_2} - W_{\text{NbS}_2/\text{BSe}}$, where the work functions of the metallic NbS₂ layer and the NbS₂/BSe heterostructure are denoted as the W_{NbS_2} and $W_{\text{NbS}_2/\text{BSe}}$, respectively. The interfacial dipole at the interface of the NbS₂/BSe heterostructure is obtained to be 0.02 eV. The formation of the interfacial dipole at the interface can change the Schottky barriers. However, our results show that both the amount of charge transfer and interface dipole are small; hence, the change in the Schottky barriers at the interface of the NbS₂/BSe heterostructure can be considered negligible. Furthermore, both the charge transfer and interfacial dipole in the NbS₂/BSe heterostructure may give rise to the formation of a built-in electric field at the interface. The built-in electric field can be calculated as follows:⁶⁷



$$E_{\text{in}} = \frac{P}{\epsilon S d} \quad (5)$$

where P stands for the interface dipole. ϵ , S and d are the dielectric constant, surface area and interlayer distance of the NbS₂/BSe heterostructure, respectively. We can observe that the built-in electric field is directly proportional to the interface dipole. As we have discussed above, the amount of charge transfer and interface dipole of the NbS₂/BSe heterostructure are small; hence, the built-in electric field across the interface of the heterostructure is also small and can be considered negligible.

Furthermore, in order to evaluate the efficiency of electron injection through the contact of the NbS₂/MoSSe heterostructure, we add the calculations on the tunneling probability (T) and the tunneling-specific resistivity (ρ_t) as follows:

$$T = \exp\left(\frac{-2w_{\text{TB}}\sqrt{2m_e h_{\text{TB}}}}{\hbar}\right) \quad (6)$$

and

$$\rho_t = \frac{4\pi^2 \hbar w_{\text{TB}}^2}{e^2} \times \frac{\exp\left[\frac{2(2m_e)^{1/2}}{\hbar} w_{\text{TB}} h_{\text{TB}}^{1/2}\right]}{\frac{(2m_e)^{1/2}}{\hbar} w_{\text{TB}} h_{\text{TB}}^{1/2} - 1} \quad (7)$$

$$= \frac{8\pi^2 \hbar^2 w_{\text{TB}}}{3e^2 \sqrt{2m_e h_{\text{TB}}}} \times T^{-1}$$

Here, \hbar is the reduced Planck's constant. e and m_e are the electron magnitude and mass of a free electron, respectively. h_{TB} and w_{TB} are the tunneling barrier height and width, respectively, which can be obtained directly from the electrostatic potential of the NbS₂/BSe heterostructure. The calculated h_{TB} and w_{TB} of the NbS₂/BSe heterostructure for the most energetically favorable stacking configuration are 3.68 eV and 1.35 Å, respectively. Hence, the tunneling probability and contact tunneling-specific resistivity at the interface of the NbS₂/BSe heterostructure are about 7% and $2.09 \times 10^{-10} \Omega \text{ cm}^2$, respectively. We also observe that the value of the tunneling-specific resistivity at the interface of the NbS₂/MoSSe heterostructure exhibits a magnitude similar to that in other metal/semiconductor contacts, such as Bi/MoS₂,⁶⁸ semimetals/TMDs,⁶⁹ metal/MSi₂N₄ ($M = \text{Mo}, \text{W}$)²⁹ and 2D (3D) metals/GeSe.⁷⁰ This finding also suggests that the NbS₂/BSe heterostructure could serve as an efficient contact for electronic devices.

Furthermore, we calculate the optical absorption of the NbS₂/BSe heterostructure as well as that of the constituent NbS₂ and BSe monolayers for comparison. The optical absorption can be obtained as follows:

$$\alpha(\omega) = \sqrt{2}\omega \left[\sqrt{\epsilon_1(\omega)^2 + \epsilon_2(\omega)^2} - \epsilon_1(\omega) \right]^{\frac{1}{2}} \quad (8)$$

Here, the real and imaginary parts of dielectric functions are denoted as ϵ_1 and ϵ_2 , respectively. The optical absorption of the NbS₂/BSe heterostructure and the constituent NbS₂ and BSe monolayers are depicted in Fig. 7. We can find that the optical

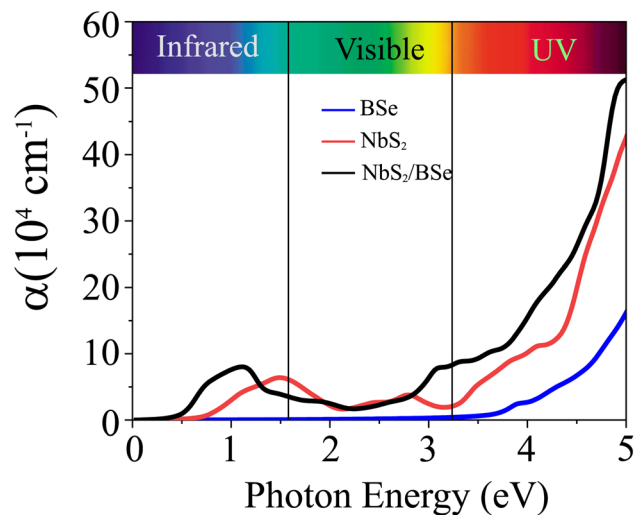


Fig. 7 Calculated optical absorption as a function of the photon energy of the NbS₂/BSe heterostructure and the constituent NbS₂ and BSe monolayers.

absorption coefficient of the NbS₂/BSe heterostructure can reach up to $5 \times 10^5 \text{ cm}^{-1}$ at a photon energy of about 5 eV, which is still greater than that in the constituent NbS₂ and BSe monolayers. Hence, the formation of the M-S NbS₂/BSe heterostructure gives rise to an enhancement in the optical absorption of both NbS₂ and BSe monolayers. The enhancement in the optical absorption of the M-S NbS₂/BSe heterostructure indicates that it can be considered as a promising candidate for the design of optoelectronic devices.

4 Conclusions

In summary, we have designed a metal/semiconductor heterostructure between two different materials 2D NbS₂ metal and 2D BSe semiconductor using first-principles prediction. The combination between NbS₂ and BSe monolayers gives rise to the formation of a metal/semiconductor NbS₂/BSe heterostructure with different stacking configurations. All these stacking configurations of the NbS₂/BSe heterostructure are considered to be structurally, mechanically and thermally stable, suggesting their potential as components in electronic devices. The formation of the NbS₂/BSe heterostructure leads to the generation of a Schottky contact with the Schottky barrier ranging from 0.36 to 0.51 eV, depending on the stacking configurations. In addition, the optical absorption coefficient of the NbS₂/BSe heterostructure can reach up to $5 \times 10^5 \text{ cm}^{-1}$ at a photon energy of about 5 eV, which is still greater than that in the constituent NbS₂ and BSe monolayers. This finding suggests that the formation of the M-S NbS₂/BSe heterostructure gives rise to an enhancement in the optical absorption of both NbS₂ and BSe monolayers. Notably, the tunneling probability and the contact tunneling-specific resistivity at the interface of the NbS₂/BSe heterostructure are low, indicating its applicability in emerging nanoelectronic and optoelectronic devices, such as Schottky diodes, field-effect transistors and photodetectors. Our findings



offer valuable insights for the practical utilization of electronic and optoelectronic devices based on the NbS₂/BSE heterostructure.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 K. S. Novoselov, A. K. Geim, S. V. Morozov, D.-e. Jiang, Y. Zhang, S. V. Dubonos, I. V. Grigorieva and A. A. Firsov, *Science*, 2004, **306**, 666–669.
- 2 A. Carvalho, M. Wang, X. Zhu, A. S. Rodin, H. Su and A. H. Castro Neto, *Nat. Rev. Mater.*, 2016, **1**, 1–16.
- 3 S. Manzeli, D. Ovchinnikov, D. Pasquier, O. V. Yazyev and A. Kis, *Nat. Rev. Mater.*, 2017, **2**, 1–15.
- 4 Y. Wu, K. A. Jenkins, A. Valdes-Garcia, D. B. Farmer, Y. Zhu, A. A. Bol, C. Dimitrakopoulos, W. Zhu, F. Xia, P. Avouris, et al., *Nano Lett.*, 2012, **12**, 3062–3067.
- 5 H. Chang and H. Wu, *Adv. Funct. Mater.*, 2013, **23**, 1984–1997.
- 6 W. Han, R. K. Kawakami, M. Gmitra and J. Fabian, *Nat. Nanotechnol.*, 2014, **9**, 794–807.
- 7 T. Wehling, K. Novoselov, S. Morozov, E. Vdovin, M. Katsnelson, A. Geim and A. Lichtenstein, *Nano Lett.*, 2008, **8**, 173–177.
- 8 T. Kuila, S. Bose, A. K. Mishra, P. Khanra, N. H. Kim and J. H. Lee, *Prog. Mater. Sci.*, 2012, **57**, 1061–1105.
- 9 G. Gui, J. Li and J. Zhong, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2008, **78**, 075435.
- 10 M. Y. Han, B. Özyilmaz, Y. Zhang and P. Kim, *Phys. Rev. Lett.*, 2007, **98**, 206805.
- 11 Y. Liu, N. O. Weiss, X. Duan, H.-C. Cheng, Y. Huang and X. Duan, *Nat. Rev. Mater.*, 2016, **1**, 1–17.
- 12 A. Castellanos-Gomez, X. Duan, Z. Fei, H. R. Gutierrez, Y. Huang, X. Huang, J. Queda, Q. Qian, E. Sutter and P. Sutter, *Nat. Rev. Methods Primers*, 2022, **2**, 58.
- 13 L. Samad, S. M. Bladow, Q. Ding, J. Zhuo, R. M. Jacobberger, M. S. Arnold and S. Jin, *ACS Nano*, 2016, **10**, 7039–7046.
- 14 K. Novoselov, A. Mishchenko, A. Carvalho and A. Castro Neto, *Science*, 2016, **353**, aac9439.
- 15 Z. Guo, L. Wang, M. Han, E. Zhao, L. Zhu, W. Guo, J. Tan, B. Liu, X.-Q. Chen and J. Lin, *ACS Nano*, 2022, **16**, 11268–11277.
- 16 Y. Gong, S. Lei, G. Ye, B. Li, Y. He, K. Keyshar, X. Zhang, Q. Wang, J. Lou, Z. Liu, et al., *Nano Lett.*, 2015, **15**, 6135–6141.
- 17 Y. Luo, K. Ren, S. Wang, J.-P. Chou, J. Yu, Z. Sun and M. Sun, *J. Phys. Chem. C*, 2019, **123**, 22742–22751.
- 18 Y. Wang, R. Chen, X. Luo, Q. Liang, Y. Wang and Q. Xie, *ACS Appl. Nano Mater.*, 2022, **5**, 8371–8381.
- 19 Z. Ben Aziza, H. Henck, D. Pierucci, M. G. Silly, E. Lhuillier, G. Patriarche, F. Sirotti, M. Eddrief and A. Ouerghi, *ACS Nano*, 2016, **10**, 9679–9686.
- 20 C.-J. Shih, Q. H. Wang, Y. Son, Z. Jin, D. Blankschtein and M. S. Strano, *ACS Nano*, 2014, **8**, 5790–5798.
- 21 B. Marfoua and J. Hong, *Adv. Mater. Interfaces*, 2023, **10**, 2202387.
- 22 B. Zhao, R. Ngaloy, S. Ghosh, S. Ershadrad, R. Gupta, K. Ali, A. M. Hoque, B. Karpiak, D. Khokhriakov, C. Polley, et al., *Adv. Mater.*, 2023, **35**, 2209113.
- 23 C. K. Borah, P. K. Tyagi and S. Kumar, *Nanoscale Adv.*, 2020, **2**, 3231–3243.
- 24 W. Yu, Z. Zhu, S. Zhang, X. Cai, X. Wang, C.-Y. Niu and W.-B. Zhang, *Appl. Phys. Lett.*, 2016, **109**, 103104.
- 25 Y. Zheng, Y. Chen, B. Gao, B. Lin and X. Wang, *Engineering*, 2021, **7**, 991–1001.
- 26 J. E. Padilha, A. Fazzio and A. J. da Silva, *Phys. Rev. Lett.*, 2015, **114**, 066803.
- 27 C. C. Tho, S.-D. Guo, S.-J. Liang, W. L. Ong, C. S. Lau, L. Cao, G. Wang and Y. S. Ang, *Appl. Phys. Rev.*, 2023, **10**, 041307.
- 28 Y. Yin, Q. Gong, M. Yi and W. Guo, *Adv. Funct. Mater.*, 2023, **33**, 2214050.
- 29 C. C. Tho, C. Yu, Q. Tang, Q. Wang, T. Su, Z. Feng, Q. Wu, C. Nguyen, W.-L. Ong, S.-J. Liang, et al., *Adv. Mater. Interfaces*, 2023, **10**, 2201856.
- 30 Q. Wang, L. Cao, S.-J. Liang, W. Wu, G. Wang, C. H. Lee, W. L. Ong, H. Y. Yang, L. K. Ang, S. A. Yang, et al., *npj 2D Mater. Appl.*, 2021, **5**, 71.
- 31 J. Su, L. Feng, Y. Zhang and Z. Liu, *Phys. Chem. Chem. Phys.*, 2016, **18**, 16882–16889.
- 32 W. Xiong, C. Xia, X. Zhao, T. Wang and Y. Jia, *Carbon*, 2016, **109**, 737–746.
- 33 D. Zhang, Y. Hu, H. Zhong, S. Yuan and C. Liu, *Nanoscale*, 2019, **11**, 13800–13806.
- 34 Q. Luo, S. Yin, X. Sun, G. Guo and X. Dai, *Diamond Relat. Mater.*, 2022, **128**, 109223.
- 35 S. Wang, J.-P. Chou, C. Ren, H. Tian, J. Yu, C. Sun, Y. Xu and M. Sun, *Sci. Rep.*, 2019, **9**, 5208.
- 36 M. Sun, J.-P. Chou, Q. Ren, Y. Zhao, J. Yu and W. Tang, *Appl. Phys. Lett.*, 2017, **110**, 173105.
- 37 Q. Fu, X. Wang, J. Zhou, J. Xia, Q. Zeng, D. Lv, C. Zhu, X. Wang, Y. Shen, X. Li, et al., *Chem. Mater.*, 2018, **30**, 4001–4007.
- 38 X. Ding, S. Zhang, M. Zhao, Y. Xiang, K. H. Zhang, X. Zu, S. Li and L. Qiao, *Phys. Rev. Appl.*, 2019, **12**, 064061.
- 39 B. Wang, H. Luo, X. Wang, E. Wang, Y. Sun, Y.-C. Tsai, H. Zhu, P. Liu, K. Jiang and K. Liu, *ACS Nano*, 2019, **14**, 175–184.
- 40 Z. Golsanamlou, P. Kumari, L. Sementa, T. Cusati, G. Iannaccone and A. Fortunelli, *Adv. Electron. Mater.*, 2022, **8**, 2200020.
- 41 S. Zhao, T. Hotta, T. Koretsune, K. Watanabe, T. Taniguchi, K. Sugawara, T. Takahashi, H. Shinohara and R. Kitaura, *2D Mater.*, 2016, **3**, 025027.



- 42 Z. Wang, C.-Y. Cheon, M. Tripathi, G. M. Marega, Y. Zhao, H. G. Ji, M. Macha, A. Radenovic and A. Kis, *ACS Nano*, 2021, **15**, 18403–18410.
- 43 Z. Zhang, P. Yang, M. Hong, S. Jiang, G. Zhao, J. Shi, Q. Xie and Y. Zhang, *Nanotechnology*, 2019, **30**, 182002.
- 44 J. J. Ni, L. Yang and H. X. Yang, *Mater. Today Commun.*, 2022, **33**, 104720.
- 45 V. Kumar, S. Khandelwal, A. Khurajam and N. Halder, *Mater. Today: Proc.*, 2022, **62**, 1580–1584.
- 46 S. Mankovsky, S. Polesya, H. Ebert and W. Bensch, *Phys. Rev. B*, 2016, **94**, 184430.
- 47 Y. Kim, K. C. Kwon, S. Kang, C. Kim, T. H. Kim, S.-P. Hong, S. Y. Park, J. M. Suh, M.-J. Choi, S. Han, et al., *ACS Sens.*, 2019, **4**, 2395–2402.
- 48 T. Li, Q. Miao, Y. Wang and H. Yang, *Surf. Interfaces*, 2023, **42**, 103336.
- 49 J. Zhou, Y. Shen, F. Lv, W. Zhang, F. Lin, W. Zhang, K. Wang, H. Luo, Q. Wang, H. Yang, et al., *Adv. Funct. Mater.*, 2022, **32**, 2204495.
- 50 S. Demirci, N. Avazlı, E. Durgun and S. Cahangirov, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2017, **95**, 115409.
- 51 S.-T. Nguyen, C. Q. Nguyen, Y. S. Ang, N. Van Hoang, N. M. Hung and C. V. Nguyen, *Langmuir*, 2023, **39**, 6637–6645.
- 52 B.-J. Wang, X.-H. Li, R. Zhao, X.-L. Cai, W.-Y. Yu, W.-B. Li, Z.-S. Liu, L.-W. Zhang and S.-H. Ke, *J. Mater. Chem. A*, 2018, **6**, 8923–8929.
- 53 D. Zhang, Q. Gao, Y. Chen, Y. Xia, H. Wang, H. Wang and Y. Ni, *Adv. Theory Simul.*, 2020, **3**, 2000144.
- 54 K. Ren, J. Yu and W. Tang, *J. Alloys Compd.*, 2020, **812**, 152049.
- 55 P. Giannozzi, S. Baroni, N. Bonini, M. Calandra, R. Car, C. Cavazzoni, D. Ceresoli, G. L. Chiarotti, M. Cococcioni, I. Dabo, et al., *J. Phys.: Condens. Matter*, 2009, **21**, 395502.
- 56 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 57 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1998, **80**, 891.
- 58 G. Kresse and D. Joubert, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **59**, 1758.
- 59 S. Grimme, *J. Comput. Chem.*, 2006, **27**, 1787–1799.
- 60 J. Heyd and G. E. Scuseria, *J. Chem. Phys.*, 2004, **121**, 1187–1192.
- 61 K. Ren, Y. Luo, S. Wang, J.-P. Chou, J. Yu, W. Tang and M. Sun, *ACS Omega*, 2019, **4**, 21689–21697.
- 62 S. D. Chakarova-Käck, E. Schröder, B. I. Lundqvist and D. C. Langreth, *Phys. Rev. Lett.*, 2006, **96**, 146107.
- 63 H. Saini, M. Jyothirmai, U. V. Waghmare and R. Thapa, *Phys. Chem. Chem. Phys.*, 2020, **22**, 2775–2782.
- 64 T. Björkman, A. Gulans, A. V. Krasheninnikov and R. M. Nieminen, *Phys. Rev. Lett.*, 2012, **108**, 235502.
- 65 X. Zhou, X. Hu, J. Yu, S. Liu, Z. Shu, Q. Zhang, H. Li, Y. Ma, H. Xu and T. Zhai, *Adv. Funct. Mater.*, 2018, **28**, 1706587.
- 66 F. Mouhat and F.-X. Coudert, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2014, **90**, 224104.
- 67 X. Li, Z. Li and J. Yang, *Phys. Rev. Lett.*, 2014, **112**, 018301.
- 68 P.-C. Shen, C. Su, Y. Lin, A.-S. Chou, C.-C. Cheng, J.-H. Park, M.-H. Chiu, A.-Y. Lu, H.-L. Tang, M. M. Tavakoli, et al., *Nature*, 2021, **593**, 211–217.
- 69 T. Su, Y. Li, Q. Wang, W. Zhao, L. Cao and Y. S. Ang, *J. Phys. D: Appl. Phys.*, 2023, **56**, 234001.
- 70 L. Cao, X. Deng, Z. Tang, G. Zhou and Y. S. Ang, *Appl. Phys. Lett.*, 2022, **121**, 113104.

