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## Advances in the synthesis of nitrogen-containing heterocyclic compounds by *in situ* benzyne cycloaddition

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Nitrogen-containing heterocyclic compounds are prevalent in various natural products, medicines, agrochemicals, and organic functional materials. Among strategies to prepare nitrogen-containing heterocyclic compounds, pathways involving benzyne intermediates are attractive given that they can readily assemble highly diverse heterocyclic compounds in a step-economical manner under transition-metal-free conditions. The synthesis of nitrogen-containing heterocyclic compounds from benzyne intermediates offers an alternative strategy to the conventional metal-catalyzed activation approaches. In the past years, chemists have witnessed the revival of benzyne chemistry, mainly attributed to the wide application of various novel benzyne precursors. The cycloaddition of benzenes is a powerful tool for the synthesis of nitrogen-containing heterocyclic compounds, which can be constructed by  $[n + 2]$  cyclization of benzyne intermediates *in situ* generated from benzyne precursors under mild reaction conditions. This review focuses on the application of cycloaddition reactions involving *in situ* benzenes in the construction of various nitrogen-containing heterocyclic compounds.

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### 1 Introduction

Nitrogen-containing compounds have been fundamental for the construction diversity of molecules in the chemical sciences and pharmaceuticals. Among them, nitrogen-containing heterocyclic molecules have a wide range of applications in medicinal chemistry and exhibit several biological functions, such as antibacterial, anticancer, and antimicrobial activities (Scheme 1a).<sup>1–4</sup> Among the many methods for preparing nitrogen-containing heterocyclic compounds, utilizing benzyne intermediates is one of the most attractive strategies, which can effectively construct complex and diverse benzazepine derivatives under mild conditions efficiently and economically. In particular, the research on cyclic isomerization of tethered triynes to generate benzyne species under heating conditions has made benzyne chemistry a field that has flourished in recent years.<sup>5–7</sup>

As one of the classic intermediates, benzyne is widely used in the synthesis of functionalized aromatic compounds.<sup>8</sup> Due to the unique structure of the benzyne intermediate, it is easily captured by other reagents. This capture can establish many new reaction types, and results in the research of benzyne chemistry have always attracted the attention of the field of

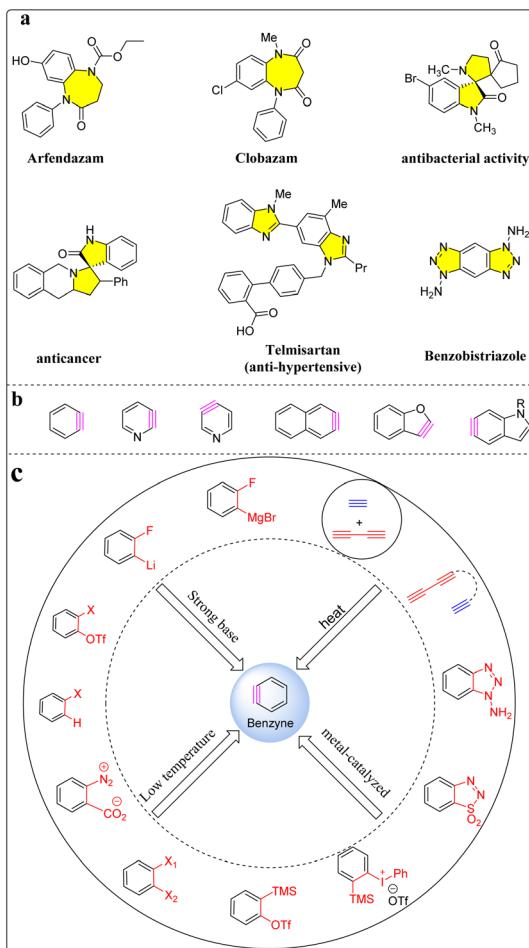
organic synthesis. Benzyne species are often referred to as 1,2-dehydrobenzene or other aromatic heterocyclic analogs (naphthalene, benzofuran, pyridine, and indoleyne, *etc.*), and their varieties are diverse (Scheme 1b).<sup>9,10</sup>

In 1902, Stoermer and Kahlert *et al.* reported the existence of the benzene species by treating 3-bromobenzofuran with a strong base in ethanol solution,<sup>11</sup> the usual strategy for the preparation of benzyne intermediates is to remove two adjacent atoms or groups from the benzene ring. Due to the instability, benzyne intermediate must be generated *in situ* from appropriate precursors under suitable reaction conditions, including thermo-synthesis, oxidation, deprotonation, halogen–metal exchange, C–Si bond cleavage by fluoride source or decarboxylation by transition metal-catalyzed (Scheme 1c).<sup>12–16</sup> Type benzyne precursors include aryl-halides,<sup>17,18</sup> aromatics-diazo-2-carboxylates,<sup>19,20</sup> 1,2,3-benzothiadiazole 1,1-dioxide,<sup>21</sup> 1-aminobenzotriazol,<sup>22,23</sup> and *o*-silyl-nonyl trifluoroester,<sup>24</sup> all of which have a significant feature that a good leaving group is present on the aromatic ring. Although this method can effectively prepare benzyne intermediates, it usually requires strong bases or additives and harsh reaction conditions. These unfavorable factors limit the application range of substrates, and strong bases are prone to nucleophile reactions with benzyne intermediates, reducing the reaction efficiency of benzyne intermediates. With the development of the cycloaddition reaction, the benzyne species can also be obtained from the thermal activation of tethered triynes by hexadehydro-Diels–Alder cycloaddition (HDDA), which can

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Scheme 1 (a) Some examples of functional aromatics, (b) general benzene species, (c) preparation strategy of benzene species.

build functionalized aromatic products with diverse structures.<sup>6,25-29</sup>

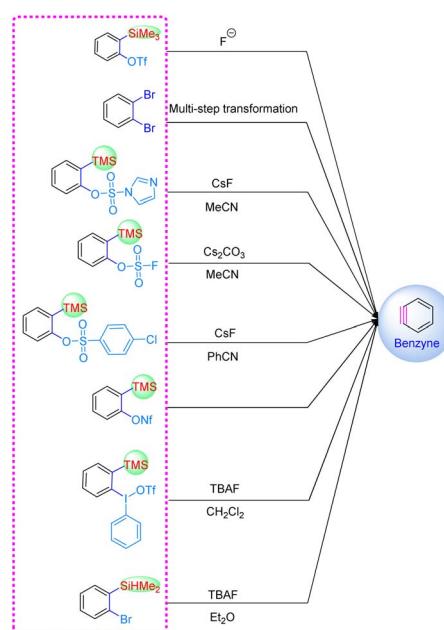
Selecting suitable precursors, reaction conditions, and benzene trapping agents was crucial for effectively utilizing benzene intermediates. The cycloaddition reaction of benzene can achieve 100% atomic economy, which provides an ideal way to efficiently utilize benzene precursors to synthesize various cyclization products. Cyclization requires good LUMO-HUMO orbital matching, and the benzene intermediate has a low LUMO, which is easy for various dienophiles to undergo [4 + 2], [2 + 2] cycloaddition, and [3 + 2] dipolar cycloaddition.<sup>30,31</sup> In the past decade, various benzyl precursor cyclization methods have been developed to synthesize nitrogen-containing heterocyclic compounds with potential applications providing a new route.

In the present minireview, we focus on benzene reactions from various precursors that can react with various dienes under mild conditions to form nitrogen-containing heterocyclic compounds. The application of HDDA in the synthesis of nitrogen-containing heterocycles is also reviewed.

## 2 Construction of nitrogen-containing heterocyclic compounds from *in situ* benzene species by cyclization

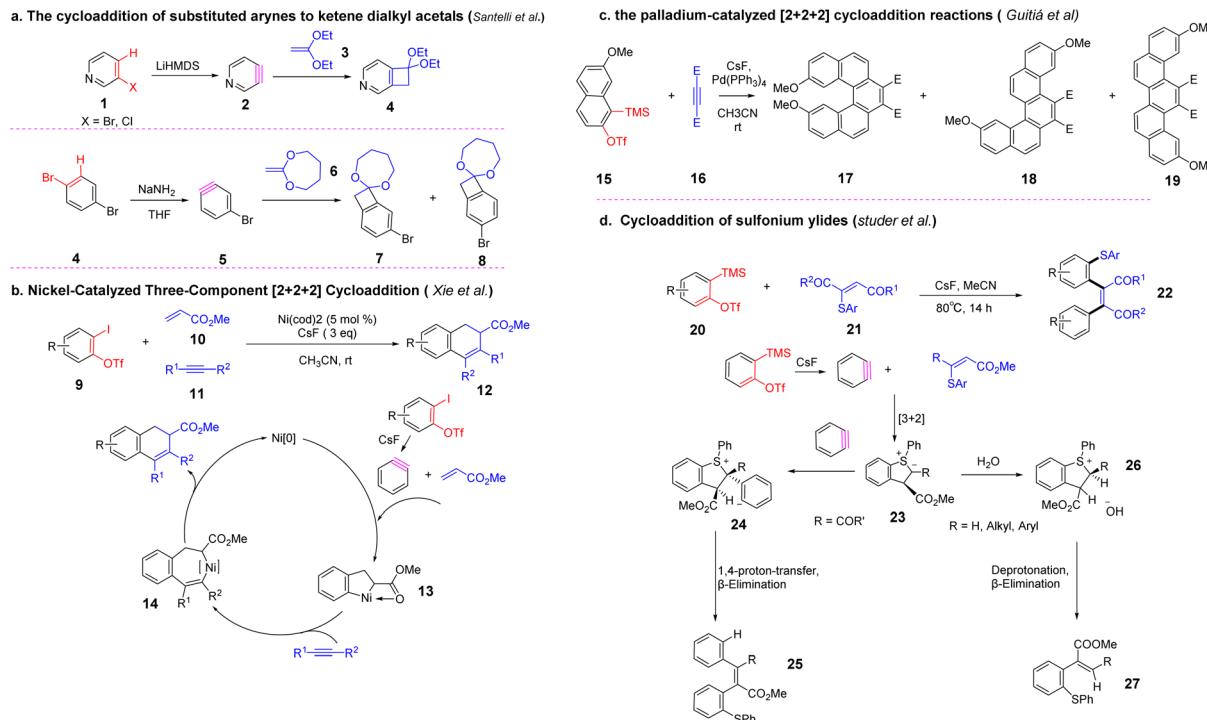
### 2.1 General strategy for the preparation of benzene intermediates

The high reactivity of the benzene intermediate is due to its unique carbon–carbon triple bond, that is, the existence of an  $sp^2$  hybrid orbital in the carbon–carbon triple bond, resulting in very easy functionalization. Up to now, the structure of benzene species has been confirmed from a microscopic point of view using infrared spectroscopy, nuclear magnetic resonance detection, low-temperature scanning tunneling microscopy, and atomic force microscopy.<sup>32-35</sup> In 1983, Kobayashi *et al.* utilized *o*-trimethylsilyl aryl trifluoromethanesulfonate to generate benzene intermediates induced by fluoride ions.<sup>24</sup> In 2000, Fu *et al.* reported a novel phenylyne precursor [2-(hydroxydimethylsilyl) phenyliodoium trifluoromethanesulfonate], which can produce benzene intermediates under mild conditions, which avoided carcinogen hexamethylphosphoramide (HMPA) in the preparation.<sup>36</sup> 2-(Trimethylsilyl)phenyl trifluoromethanesulfonate has unique advantages in organic synthesis as a benzene precursor, but in its synthesis process, toxic, expensive, and highly corrosive trifluoromethanesulfonic anhydride will be used. Therefore, the development of relatively green trimethylsilylphenyl precursors is of great significance for the practical application of benzene species.<sup>37-44</sup> For example: 2-(trimethylsilyl)phenyliodoium trifluoromethanesulfonate, 2-(trimethylsilyl)phenylimidazolyl sulfonate, 2-(trimethylsilyl)phenylnonafluorobutane sulfonate, 2-(trimethylsilyl)phenyl thiofluoride, 2-(trimethylsilyl)phenyl bromide, 2-(trimethylsilyl)phenyl 4-chlorobenzenesulfonate, *etc.*



Scheme 2 Some examples of the preparation of benzyl intermediate.





Scheme 3 Some examples of [2 + 2] cycloadditions.

(Scheme 2). These simple and environmentally friendly methods allowed obtaining the cycloaddition products of benzene intermediates.

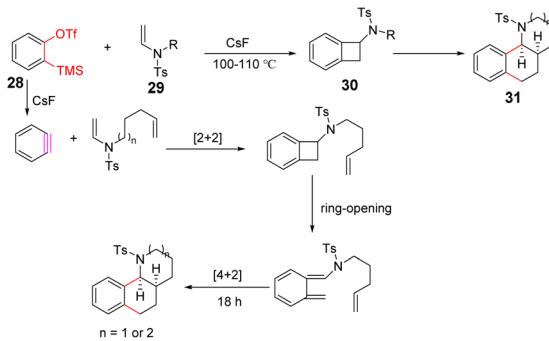
## 2.2 [2 + 2] Cycloaddition reaction

The [2 + 2] cycloaddition reaction is a valuable synthesis toolbox. Various novel compounds can be constructed by combining different mono-enes or mono-acetylenes with benzenes, such as benzocyclobutenes, benzene fused rings, pentahelicene, and bifunctional products (Scheme 3)<sup>45-49</sup>. Subsequently, the use of benzyne intermediates by [2 + 2] cyclization for the construction of nitrogen-containing compounds was developed. In 2009, Hsung *et al.* reported an enamide-benzyne-[2 + 2] cycloaddition, using potassium fluoride as the fluorine source, a benzyne species was formed *in situ* in 1,4-dioxane, and then reacted with enamide to obtain the [2 +

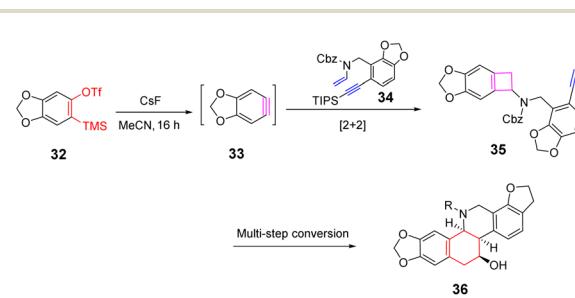
2] product<sup>50</sup> (Scheme 4). The [2 + 2] cycloaddition products might further undergo ring-opening to obtain intramolecular [4 + 2] cycloadducts with high stereoselectivity, which enriches the construction methods of nitrogen-containing nitrogen heterocycles.

In 2012, the *in situ* formation of benzyne intermediate from enamines as starting materials and the total synthesis of *p*-( $\pm$ )-chelidone and ( $\pm$ )-norchelidone *via* cyclization reported by Hsung *et al.*<sup>51</sup> The synthesis method highlights the power of enamines as a building block and the importance of benzyne chemistry, especially the intermolecular [2 + 2] cyclization of benzyne intermediates with enamines as a critical step in initiating this synthesis pathway (Scheme 5).

He *et al.* developed a novel strategy for the preparation of spiroacridines *via* a [2 + 2] cycloaddition/ring-opening sequence of benzenes with *N*-arylimines (Scheme 6). Initially, the benzyne undergoes a [2 + 2] cycloaddition with imine to generate azetidine intermediate **b**. This four-membered ring undergoes rapid

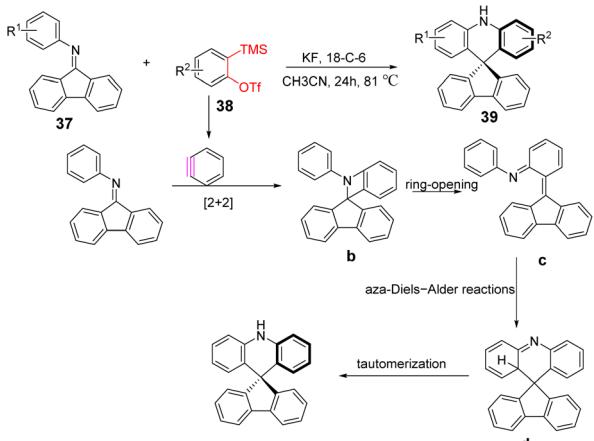


Scheme 4 Tandem benzyne-enamide - [2 + 2]-[4 + 2].

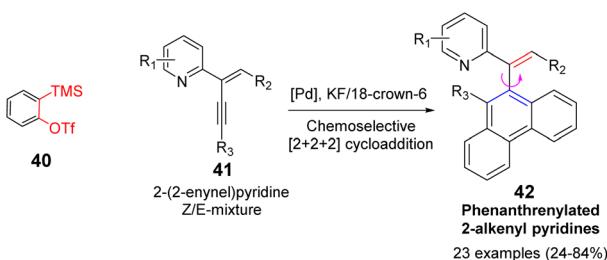


Scheme 5 Tandem enamide-benzyne - [2 + 2]-[4 + 2].





Scheme 6 Benzynes [2 + 2] cycloaddition/ring-opening sequence of benzynes with *N*-arylimines.



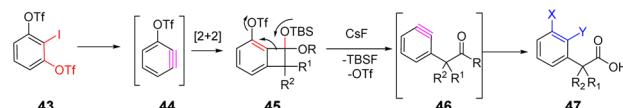
Scheme 7 Palladium-catalyzed selective [2 + 2 + 2] annulation.

intramolecular ring opening by nucleophilic attack on the *N*-*ortho* position of the *N*-aryl ring to form quinomethide imine intermediate. The subsequent intramolecular aza-Diels–Alder reaction **c** leads to the formation of **d**, followed by fluorine anion-assisted tautomerization to access the final product. The reaction provides a powerful tool for the synthesis of valuable nitrogen-containing polyaromatic skeletons.

Wang *et al.* established a palladium-catalyzed chemical and stereoselective [2 + 2 + 2] cyclization reaction of 2-(2-alkenyl) pyridine with alkynes in 2021 (Scheme 7).<sup>52</sup> Palladium catalyst is crucial for the chemical selectivity control and product selectivity of *Z/E* mixture of 2-(2-enynyl)pyridines. A wide range of phenanthrenylated 2-alkenylpyridines can be achieved *via* this strategy, which contains a chiral axis between the olefin and the phenanthrene ring, with good yield (84% yield) and excellent *E*-selectivity.

Li *et al.* developed a novel method for the synthesis of 3-trifluoromethoxyaryne intermediates. Starting from the [2 + 2] cycloaddition, a chemically selective ring-opening method was carefully designed to allow the formation of corresponding 2,3-aryne intermediates.<sup>53</sup> This strategy is universal, efficient, transition metal-free, and can be used for various combinations of aromatic reagents and drug synthesis. The regioselectivity of the reaction is affected by steric hindrance, which preferentially supports pronuclear transformation (Scheme 8).

At present, there are few examples of [2 + 2] cyclizations involving benzenes to construct nitrogen-containing heterocycles

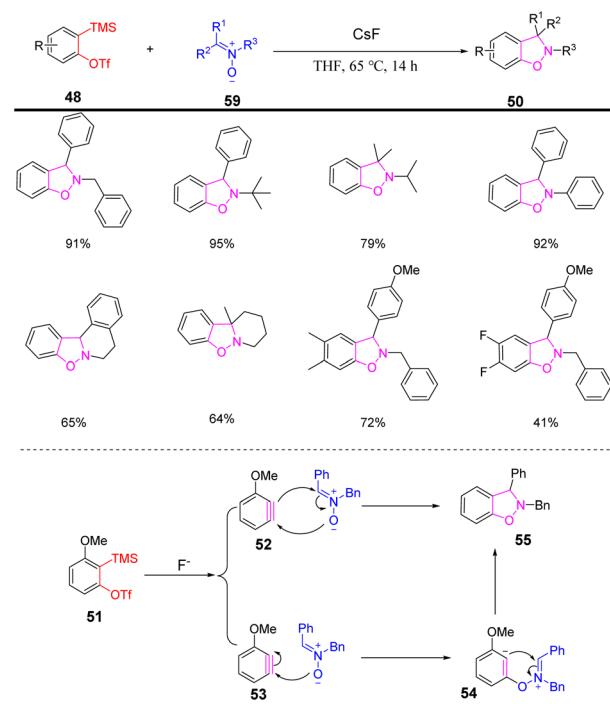


Scheme 8 Multi-functionalization of benzyne precursors by cyclization/ring-opening.

have been reported, which are often accompanied by other cyclization reactions that can be used as an initiated step to construct more structurally complex compounds by 2 + 2 cyclization.

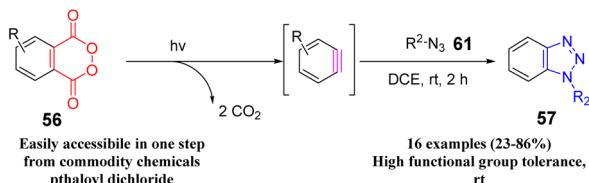
**2.2.1 [3 + 2] Dipole cycloaddition reaction.** The 1,3-dipolar compounds have a three-atom four-electron ( $4\pi$ ) conjugated system, with a charge distribution form of one end showing a negative charge and a positive charge. Therefore, the [3 + 2] dipolar cycloaddition of benzyne intermediates with similar molecular orbitals to allyl anions can produce a variety of benzoheterocycle derivatives. In 2012, Larock *et al.* reported that benzoisoxazoline derivatives were obtained by the [3 + 2] cycloaddition reaction of typical benzyne intermediates with 1,3-dipoles. The reaction has a wide range of substrates, mild conditions, and good functional group compatibility, which provides an effective way to prepare benzoheterocycle compounds (Scheme 9).<sup>54</sup>

In 2015, Shi *et al.* reported the *in situ* generation of benzyne from phthaloyl peroxides by photolysis in the presence of 1,2-dichloroethane to construct benzo[d][1,2,3]triazole by [3 + 2] cycloaddition with azide.<sup>55</sup> (Scheme 10.) The method could be well applied to various organic azides and phthaloyl peroxides,



Scheme 9 Cycloadditions of benzyne with nitrones.





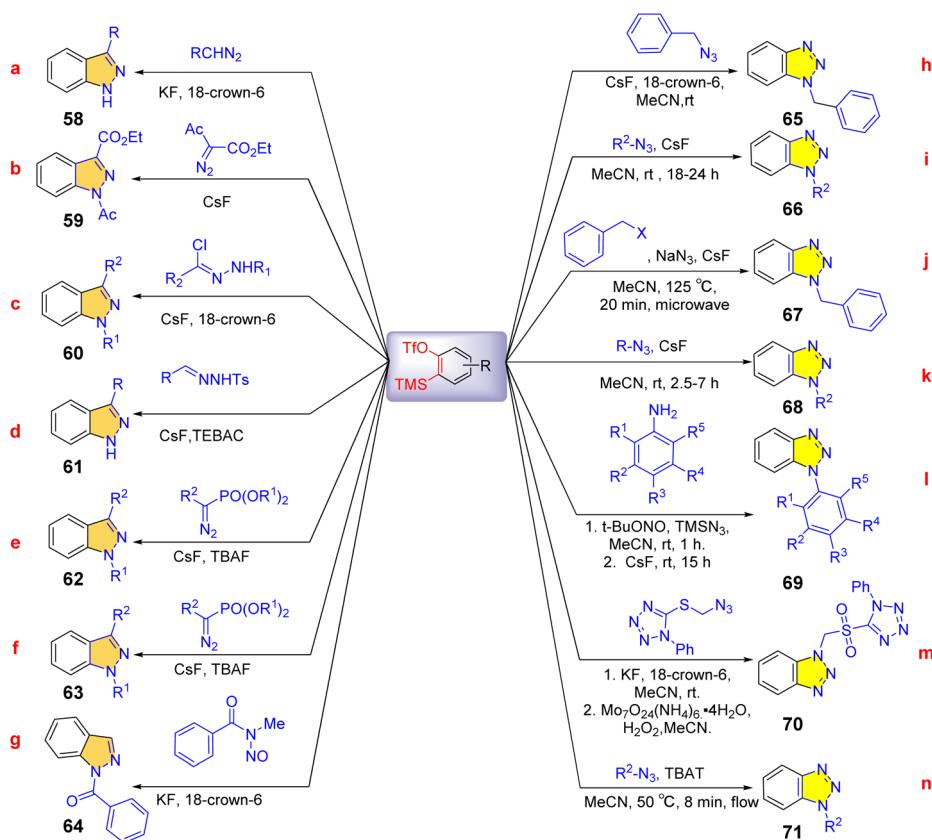
Scheme 10 Photoinduced benzene [3 + 2] cycloaddition.

and the corresponding benzotriazole derivatives can be obtained at room temperature, and the yield is moderate to good.

Benzotriazole, indazole scaffolds are antibacterial, antifungal, and antiviral drugs with significant potency.<sup>56-60</sup> Therefore, the simple synthesis of rapidly and scalable widely functionalized benzotriazoles, indazole has important practical significance. Yamamoto *et al.* established a simple, efficient, and versatile method for the synthesis of *N*-unsubstituted indazoles and 1-aryl indazoles by 1,3-dipolar cycloaddition of benzene with diazo-methane derivatives. Under very mild conditions, various indazoles can be obtained, and the yield are from good to high (Scheme 11a).<sup>61</sup> Subsequently, polysubstituted indazoles were obtained by the [3 + 2] cycloaddition of various diazo compounds with *o*-(trimethylsilyl)aryl triflates at room temperature in the presence of CsF and provided a straightforward and effective method for obtaining indazole skeleton. These reactions involve a dipolar cycloaddition of *in situ* generated diazo compounds and

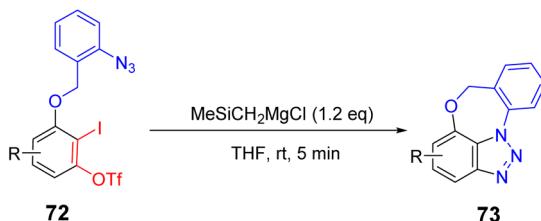
arynes (Scheme 11b-d).<sup>62-64</sup> 3-Alkyl/aryl 1 h-indazoles and 3-alkyl/aryl 3*H*-indazole-3-phosphonates can be synthesized by 1,3-dipolar cycloaddition of  $\alpha$ -substituted  $\alpha$ -diazo-methylphosphonates with arynes under simple reaction conditions. The product distribution was controlled by the phosphoryl group (Scheme 11e and f).<sup>65</sup> A novel cascade cyclization reaction of *N*-alkyl-*N*-nitrosamides with benzene has been developed under transition metal-free conditions at room temperature *via* benzene  $\delta$ -insertion and C(sp<sup>3</sup>)-H bond functionalization strategies. It provided functionalized indazoles with high yield and regioselectivity. The protocol also demonstrates a wide range of substrates (Scheme 11g).<sup>66</sup>

In 2008, Feringa *et al.* reported that fluorine-promoted elimination of *o*-(trimethylsilyl)aryl trifluoroesters to form benzene species that can undergo [3 + 2] cycloaddition with various azides to generate substituted benzotriazoles.<sup>67</sup> The advantages of mild reaction conditions and easy availability of raw materials make it a classic 'click' reaction of azides and benzynes (Scheme 11h and i).<sup>68</sup> Subsequently, in the presence of CsF, various simple and efficient preparation methods for benzotriazole have been developed, including microwave promotion and multi-step reactions (Scheme 11j-m).<sup>69-72</sup> A metal-free flow synthesis of benzotriazole was reported by Heretsch *et al.* (Scheme 11n)<sup>73</sup> Benzotriazoles can be formed by [3 + 2] cycloaddition in a few minutes using azides with *in situ* generated benzene compounds. The thermal strain of dangerous azides and the accumulation of reaction



Scheme 11 Benzotriazoles via [3 + 2] cycloaddition of 1,3-dipolar compounds and benzynes.



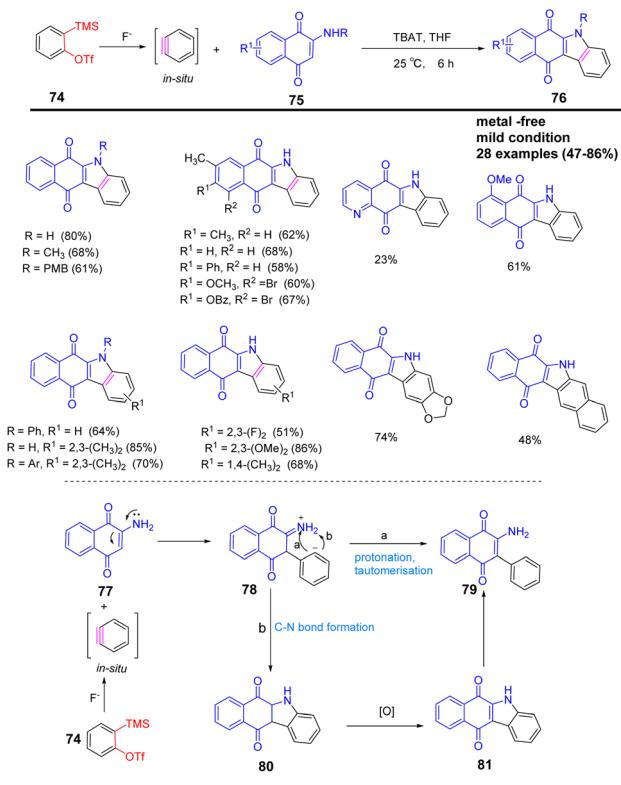


Scheme 12 Intramolecular cycloadditions of aryne precursors bearing an azido.

intermediates are minimized by a short reaction time in the flow, improving process safety.

Hosoya *et al.* established a method for synthesizing novel benzotriazole derivatives based on the intramolecular cycloaddition of containing azide groups.<sup>74</sup> Intramolecular azide aromatic cycloaddition from common raw materials obtained a unique cyclo-fused benzotriazole. This method can be applied to the construction of a unique benzotriazole library. Mechanistic studies have shown that the intermolecular [3 + 2] cycloaddition between 3-alkoxy and azides results in a high stereoselectivity due to the inductive electron-withdrawing effect of alkoxy groups (Scheme 12).

In 2016, the [3 + 2] cyclic addition of aminoquinone derivatives and benzyne intermediates to obtain carbazolquinone compounds with important biological and pharmaceutical significance developed by He *et al.* (Scheme 13).<sup>75</sup> The tandem reaction metal-free is achieved by the continuous formation of

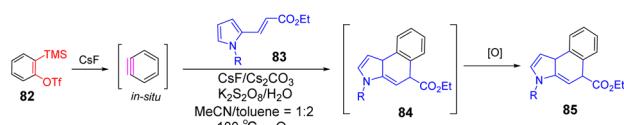


Scheme 13 [3 + 2] Cycloaddition of arynes and 2-aminoquinones.

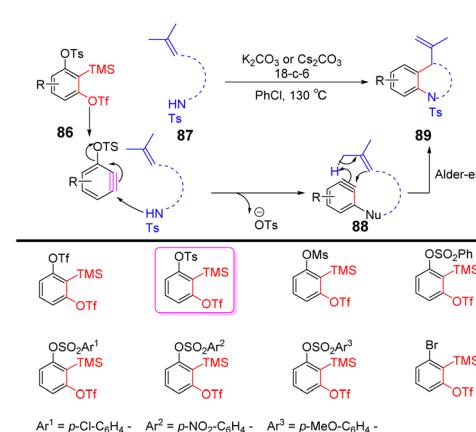
C–C and C–N bonds, and the nucleophilic attack of this reaction is carbon arylation of aminoquinones rather than nitrogen arylation. Control experiments and mechanistic studies showed that initially, the carbon of the enamine moiety undergoes nucleophilic addition to the *in situ* formation of benzyne, resulting in the formation of C-arylated zwitterionic intermediates 83. Then the nucleophilic aryl anions could be added to imine nitrogen and oxidized to obtain the cycle product 84 (path b). The first step, nucleophilic addition, for asymmetric alkynes, determines regional selectivity. The synthesis method can obtain Morson quinolone A and Koniganone B. These studies indicate that this method has potential scalability and flexibility in forming molecular diversity.

Meanwhile, Sha *et al.* reported that benzo[e]indole compounds were synthesized by intermolecular Diels–Alder reaction of benzyne intermediate with 2-vinylpyrrole at 45–86% yield (Scheme 14).<sup>76</sup> Utilizing diphenylmethyl ( $\text{CH}_2\text{Ph}_2$ ) as a protective group, the benzyne intermediate selectively reacts only with the exposed alkenyl group. Due to the wide range of substrates and the avoidance of metal catalysts in this method, it shows potential application value for synthesizing heterocyclic compounds.

Li *et al.* successfully developed an efficient, transition metal-free, broad-spectrum domino arylene annulation approach by nucleophilic–ene cascade domino pathway, which can construct a variety of benzo-fused N-heterocycles and generate substituted arenes.<sup>77</sup> An essential factor for the success of this domino transformation depends on the ability of the second leaving groups to dissociate on the 1,2-benzdiyne equivalents. –OTs were employed as the second leaving group to obtain 77% cyclization yield, while other leaving groups (LGs, *i.e.*, OTf,



Scheme 14 Synthesis of benzo[e]indole from 2-vinylpyrroles and aryne precursors.



Scheme 15 Domino aryne annulation via a nucleophilic–ene.



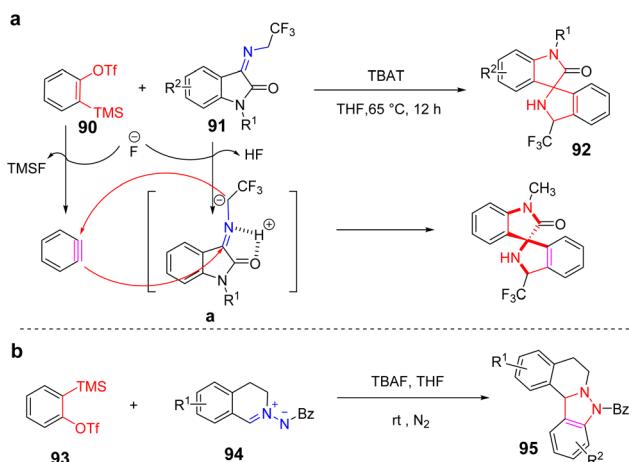
sulfonyloxy groups, bromide) were ineffective in the cyclization process. The strategy provides new avenues for benzyne cyclization reactions (Scheme 15).

Same year, Ko *et al.* red an efficient synthesis of spiro[oxindole-3,2'-pyrrolidine] derivatives containing quaternary carbon centers by [3 + 2] cycloaddition under mild reaction conditions

using TBAT as fluorine sources<sup>78</sup> (Scheme 16a). In the conversion process, imine **91** is converted to azomethine ylide **a** by abstraction of the indicated hydrogen, which is acidic upon treatment with a weak base such as fluoride. The azomethine ylide intermediate **a** undergoes a [3 + 2] cyclization reaction with the *in situ* generated benzyne to give the final product **92**. In 2019, Wu *et al.* established a [3 + 2] cyclization strategy for the synthesis of *N*-substituted indazolo[3,2-*a*]isoquinolines from benzenes and C, *N*-cyclic azomethine imines. This method can yield indazoline[3,2-*a*]isoquinolines with good anti-cancer cell proliferation activity in a one-step reaction under mild conditions (Scheme 16b).<sup>79</sup>

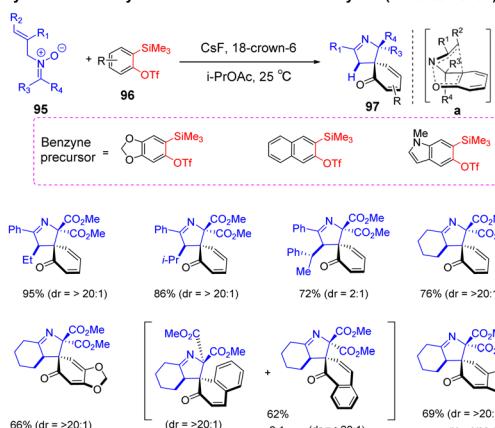
In recent years, the synthesis method of various nitrogen-containing heterocyclic compounds by [3 + 2] cycloaddition of benzenes was established. Anderson *et al.* developed a cycloaddition rearrangement cascade reaction to access 1-pyrrolines from nitrones and arynes in 2020 (Scheme 17a).<sup>80</sup> The reaction provides an altered approach for producing spiropyrrolidines that cannot be obtained through the nitrile pyrrole cycloaddition mechanism and contains unsaturated functional groups that can be further derived by stereoselective functionalization. This cascade reaction is a new pathway in the synthesis of nitrogen-containing heterocyclic compounds *via* [3,3']-sigmatropic rearrangement. In 2022, Mo *et al.* reported similar reactions. Various spirooxindole-benzo[d]oxazoles were

**Scheme 16** 1,3-Dipolar cycloaddition reactions with benzenes. (a) Synthesis of spiro[oxindole-3,2'-pyrrolidine] derivatives, (b) synthesis of *N*-substituted indazolo[3,2-*a*]isoquinolines skeleton.

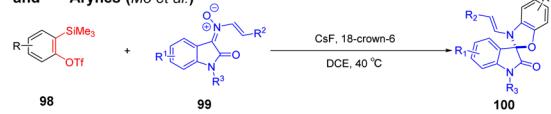


**Scheme 16** 1,3-Dipolar cycloaddition reactions with benzenes. (a) Synthesis of spiro[oxindole-3,2'-pyrrolidine] derivatives, (b) synthesis of *N*-substituted indazolo[3,2-*a*]isoquinolines skeleton.

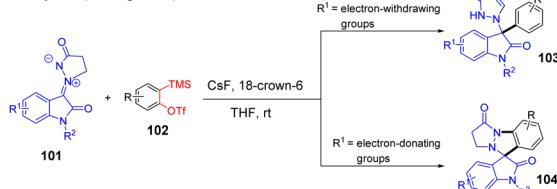
#### a. Synthesis of 1-Pyrrolines from Nitrones and Arynes (Anderson et al.)



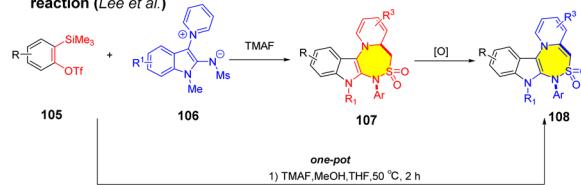
b. [3+2] Cycloaddition and Rearrangement Reaction of N-Vinyl Nitrone and Arynes (Mo et al.)



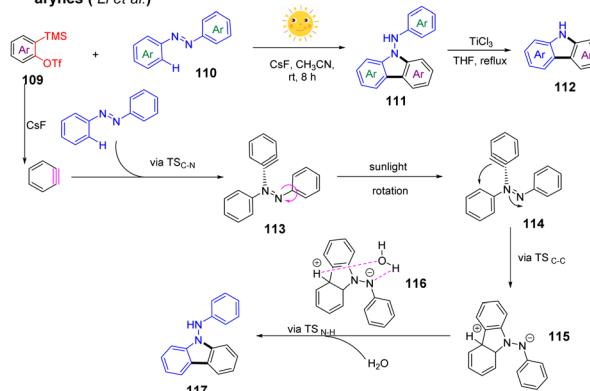
c. [3 + 2] cycloaddition isatin N,N'-cyclic azomethine imine 1,3-dipole with arynes ( *Zhang et al.* )



**d. Sequential N-Arylation and intramolecular imine addition**



e. Sunlight-mediated [3 + 2] cycloaddition of azobenzenes with arynes (*Li et al.*)



**Scheme 17** Synthesis of nitrogen-containing heterocycles by [3 + 2] cycloaddition

prepared by [3 + 2] cycloaddition and selective rearrangement of *N*-vinyl oxindole nitro compounds and benzyne compounds under transition-metal-free conditions in good to excellent yields (Scheme 17b).<sup>81</sup> Zhang *et al.* developed a [3 + 2] cyclization/arylation of isatin *N,N'*-cyclic azomethine imine 1,3-dipole with *in situ* generated benzyne to access 3,3-disubstituted oxindole scaffolds in moderate yields. The method could form various biologically active disubstituted oxindoles with simple and mild reaction advantages, practical and one-pot (Scheme 17c).<sup>82</sup> Lee *et al.* reported that 1,4-thiadiazepine was obtained by sequential *N*-arylation and intramolecular imine addition of indoloazomethine with 2-(trimethylsilyl)aryl triflates in the presence of TMAF and MeOH.<sup>83</sup> Substrates with simple and easy-to-access structures could successfully convert into indolizine compounds in quantitative yield (Scheme 17d). In 2021, Li *et al.* reported an efficient strategy for the preparation of carbazole *via* [3 + 2] cycloaddition of azobenzene and arynes at room temperature in the presence of sunlight. The reaction was carried out catalyst-free and showed good functional group compatibility. DFT calculations show that trace H<sub>2</sub>O plays a crucial role in the formation of the product, the O–H bond cloud promotes aromatization by losing a proton, and the nitrogen anion takes H<sup>+</sup> from H<sub>2</sub>O to realize the protonation reaction forming the final product (Scheme 17e).<sup>84</sup>

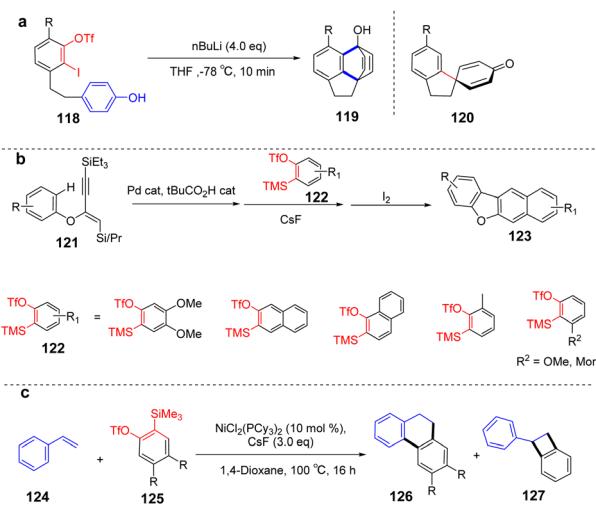
These reactions demonstrate that under relatively green reaction conditions, structurally diverse nitrogen-containing heterocyclic compounds can be constructed by combining suitable benzyne precursors with structurally simple 1,3-dipole dienes. These methods are atomically economical, without involving transition metals, and have the potential for industrial applications.

**2.2.2 [4 + 2] Cycloaddition reaction.** The [4 + 2] cycloaddition of benzyne intermediates with diene is important synthesis method for building nitrogen-containing heterocyclic

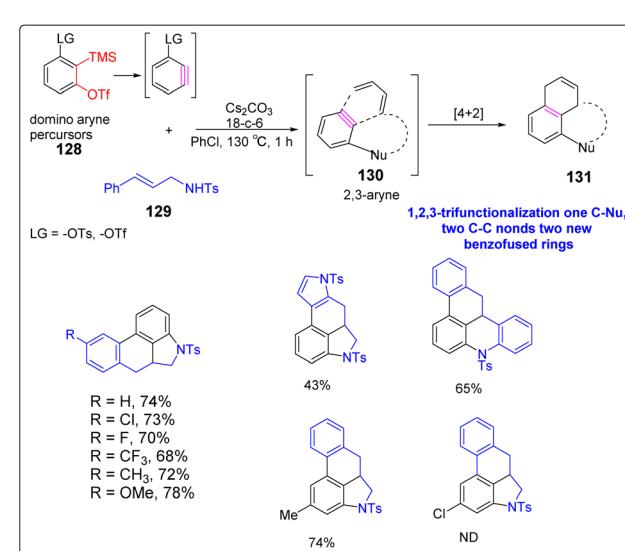
compounds. Suzuki *et al.* reported an intramolecular [4 + 2] cyclization reaction involving a benzyne intermediate in 2020.<sup>85</sup> Employing the iodine group and the sulfonic acid group as benzyne precursors, halogen atom–metal exchange was carried out by *n*-BuLi to make phenol chain groups rapidly undergo molecular [4 + 2] cycloaddition to access a series of benzo barrelenes compounds at 33–99% yield (Scheme 18a). In the same year, Hiyama *et al.* established a new reaction for rapidly constructing naphthyl[2,3-*b*]benzofuran compounds from benzyne intermediates with 1,4-disilyl-2-aryloxy-1,3-enynes *via* sequence catalytic intramolecular hydroarylation, [4 + 2] cycloaddition with a dienophile and oxidative desilylaromatization in one-pot (Scheme 18b).<sup>86</sup> The [4 + 2] cycloaddition reaction of styrene and benzyne was realized by 1 : 1 cross-coupling with nickel as a catalyst. This method is suitable for the *in situ* formation of various benzenes from *o*-(trimethylsilyl) aryl triflates to obtain 9,10-dihydrophenanthrenes containing substituted aromatics. It is easy to synthesize stilbene natural products by this method.<sup>87</sup> (Scheme 18c).

In 2019, with further study of this type of reaction, a new strategy has been successively reported for preparing multi-functional azacyclic compounds from simple and available feedstock.<sup>88</sup> The reaction designed a structurally unique domino benzyne intermediate, which efficiently prepared a more complex benzoheterocycle after a continuous nucleophilic reaction and an intramolecular Diels–Alder reaction with styrene as a structural unit. The approach constructed one C–N bond and two C–C bonds in a metal-free catalyst and had excellent chemical selectivity. (Scheme 19).

Subsequently, a rapid preparation of naphthalene derivatives reaction from linear 1,3-diyne with benzyne intermediates developed by Hoye *et al.*<sup>89</sup> Benzyne intermediate **a** can be generated *in situ* by using K<sub>2</sub>CO<sub>3</sub> as a base, and density functional theory (DFT) shows that the bond angle of carbon–carbon triple bond on intermediate **a** is distorted. It can be inferred that the sulfonamide group on the linear diyne will preferentially

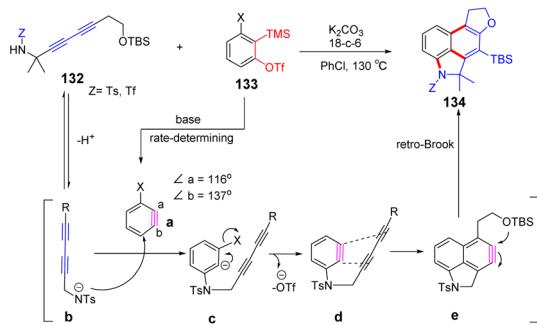


**Scheme 18** Construction of polycyclic compounds *via* [4 + 2] cycloadditions of benzyne with diene. (a) Synthesis of benzo barrelenes compounds, (b) synthesis of naphthyl[2,3-*b*]benzofuran skeleton, (c) synthesis of 9,10-dihydrophenanthrenes containing substituted aromatics.



**Scheme 19** Arene tri-functionalization with highly fused ring systems.





Scheme 20 Arene trifunctionalization with highly fused ring systems.

attack **a** to form intermediate **c**. Because trifluoromethanesulfonate has a good leaving ability, intermediate **d** will be rapidly converted to new benzyne intermediate **e**. Subsequently, the newly formed benzyne intermediate is tripped by linear 1,3-diyne, and an intramolecular [4 + 2] cyclization reaction occurs to obtain the new benzyne intermediate. Finally, the -OTBS is more easily trapped by **e**, which can produce the final product **134** by continuous nucleophilic addition and Brook rearrangement (Scheme 20).

Two flexible cascade reactions to prepare multiple functionalized heterocyclic compounds were developed. The approach relies on an easily accessible electron-rich-aza-Diels-

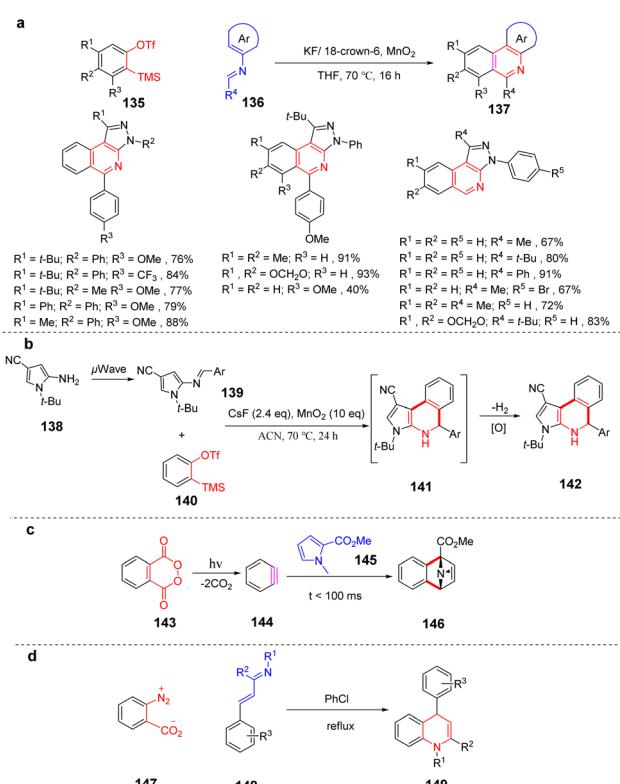
Alder cycloaddition reaction and a subsequent aromatization step. The strategy has a broad substrate and allows for the rapid total synthesis of the natural product benzo[c]phenanthridine alkaloid nornitidine (Scheme 21a).<sup>90</sup>

In 2019, Portilla *et al.* developed a general synthetic strategy for various *E*-2-arylideneaminopyrroles *via* benzyne-promoted azo-Diels–Alder cycloaddition, which is key intermediates for nature product pyrrolo[2,3-*c*]isoquinolines scaffold. There is a strong correlation between the viscosity and fluorescence intensity of these compounds in different solvents, a property that may be useful in fluorescent molecular rotor (FMR) applications (Scheme 21b).<sup>91</sup> Other non-commonly benzyne precursors have been reported in cyclization reaction applications. The photolysis of phtaloyl peroxide by laser flash at 266 nm can rapidly remove two CO<sub>2</sub> molecules to form benzyne, and *N*-containing compounds generated *via* the Diels–Alder reaction of the benzyne intermediate with 1-methylpyrrole-2-carboxylate (Scheme 21c)<sup>92</sup> The [4 + 2] cycloaddition reaction of the benzyne precursor (benzenediazonium-2-carboxylate) with azadiene is an effective strategy for the synthesis of aryl-substituted 1,4-dihydroquinolines. These products could be further oxidized to yield 4-arylquinolin-2-ones with biologically active and converted into aryl substituted quinoline compounds (Scheme 21d).<sup>93</sup>

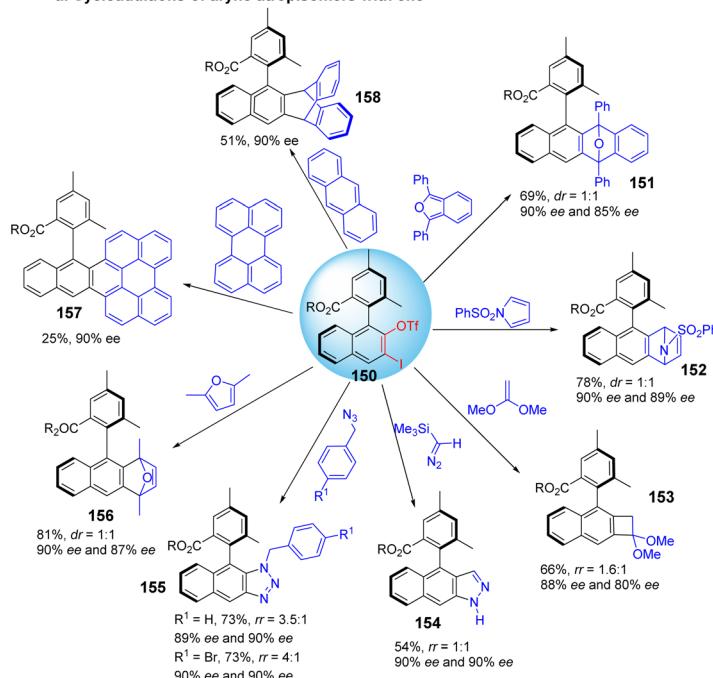
Arynes are widely used as synthetic tools for complex aromatic molecules, such as natural products or large polycyclic aromatic hydrocarbons.<sup>94–96</sup> In 2020, Coquerel *et al.* reported that the active triple bond of enantiomerically enriched aryls could build biaryl stereo-axi compounds.<sup>97</sup> The reaction can easily generate benzyne *in situ* from the corresponding *ortho*-iodoaryl trifluoride and trimethylsilylmethylmagnesium chloride intermediates and perform normal benzyne cycloaddition reactions in an enantiospecific manner. Cycloaddition of benzyne species with various diene occurs in an enantiospecific, opening up new options for the stereo-controlled synthesis of non-racemic anti-compressive polycyclic compounds in a broad sense (Scheme 22a). In 2022, Raminelli *et al.* developed potassium fluoride and 18-crown-6 to promote the reaction of various substituted oxazoles with *o*-(trimethylsilyl)aryl trifluoroesters.<sup>98</sup> At room temperature, deserve functionalized isoquinoline compounds were obtained through the [4 + 2] cycloaddition/ring-opening reaction in moderate to good yields. In addition, at a higher temperature (60 °C), bridge oxygen ether compounds can be obtained by the reaction sequence of [4 + 2] cycloaddition, *retro-DA*, and [4 + 2] cycloaddition. These methods use mild reaction conditions and avoid dangerous aryne precursors and highly toxic reaction conditions (Scheme 22b).

He *et al.* established a simple and transition metal-free method for the synthesis of 4-amino isoquinolin-1(2*H*)-ones by multi-step sequence reaction. The strategy involves the tandem Diels–Alder reaction/dehydrogenation, aromatization/tautomerization reaction of benzyne with 4,5-disubstituted oxazoles to access the final product in moderate to excellent yields. The reaction is easy to scale up, and the product can effectively convert into isoquinoline derivatives (Scheme 22c).<sup>99</sup>

**2.2.3 Others cycloaddition.** Biju *et al.* reported the enantioselective synthesis of bridged azabicyclo [4.3.1] dodecene

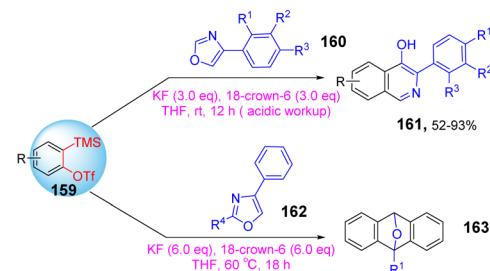
Scheme 21 DA reaction of benzyne with nitrogen-containing dienes. (a) Synthesis of isoquinolines *via* the oxidative aza-DA reaction, (b) synthesis of pyrrolo[2,3-*c*]isoquinolines scaffold *via* aza-DA reactions, (c) photolysis of a benzyne precursor, (d) synthesis of aryl-substituted 1,4-dihydroquinolines by [4 + 2] cycloaddition.

## a. Cycloadditions of aryne atropisomers with ene

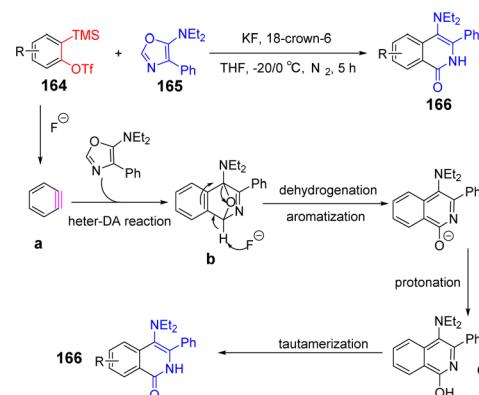


All reactions were performed with 3–5 equiv of the arynophile and 10 equiv of  $\text{Me}_3\text{SiCH}_2\text{MgCl}$  in  $\text{Et}_2\text{O}$  and/or toluene at  $0^\circ\text{C}$ .

## b. Synthesis of isoquinolines via the [4+2] cycloaddition

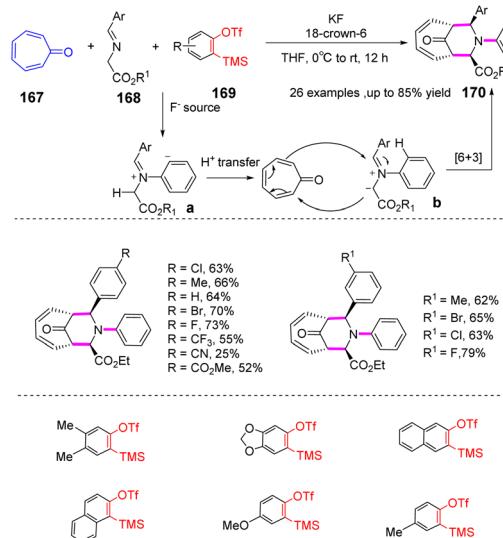


## c. Transition-metal-free synthesis of 4-amino isoquinolin-1(2H)-ones



Scheme 22 Synthesis of *N*-containing compounds via [4 + 2] the cycloaddition reaction.

derivatives *via* a [6 + 3] cyclization reaction of cyclohepta-2,4,6-trien-1-one, iminoester, and benzyne.<sup>100</sup> The critical azaphyllin intermediate in the reaction is generated by imino esters and benzyne. Mechanistic studies have shown that the nucleophilic addition of imino esters to benzynes can generate nitrogen ylide **a** through zwitterionic **b**, which can undergo a synergistic [6 + 3] cycloaddition reaction with tropinone to generate **170**. Various substituted groups can be used as benzyne precursors in this reaction (Scheme 23).

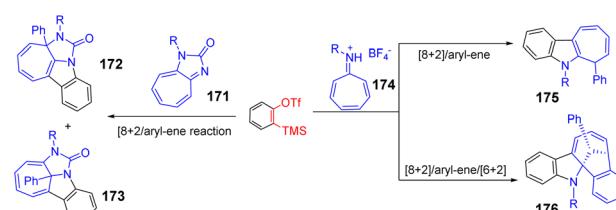


Scheme 23 The three-component diastereoselective [6 + 3] annulation reaction.

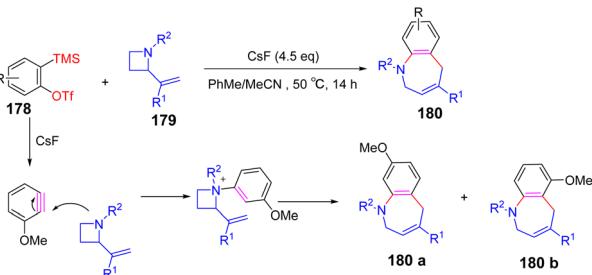
He *et al.* reported the tandem reaction of [8 + 2]/aryl-olefins between *N*-heptafluoroethylene and arylacetylene to synthesize cyclohepta[*b*]indole with biological significance.<sup>101</sup> The [6 + 2] cycloaddition was also applied to alkynes species, which further derivatized oxacyclohepta[*b*]indoles to polycyclic oxhepta[*b*]indoles. The tandem reaction provides a new channel for the synthesis of cyclohepta[*b*]indoles and oxacyclohepta[*b*]indoles (Scheme 24).

Saito *et al.* established a novel method to access benzazocine derivatives by the [6 + 2] cycloaddition reaction of 2-vinyl-azetidines with benzyne without a catalyst<sup>102</sup> (Scheme 25). The regioselectivity of aromatic (**180a**, **180b**) can be explained in terms of electronic effects and steric effects of the substituents. The reaction is initiated by the nucleophilic attack of the nitrogen atom of vinylazetidine to benzyne. The zwitterionic species could be protonated to produce an azetidinium salt. The azetidine salt will be rearranged under reaction conditions to provide ring expansion products.

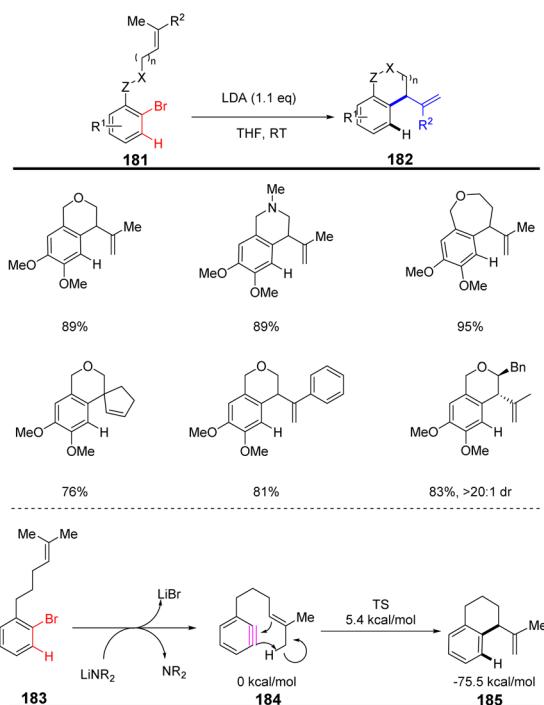
In 2011, Lauten *et al.* developed a highly stereoscopic and regionally selective intramolecular alder-ene reaction by



Scheme 24 The tandem reaction of [8 + 2]/aryl-olefins.



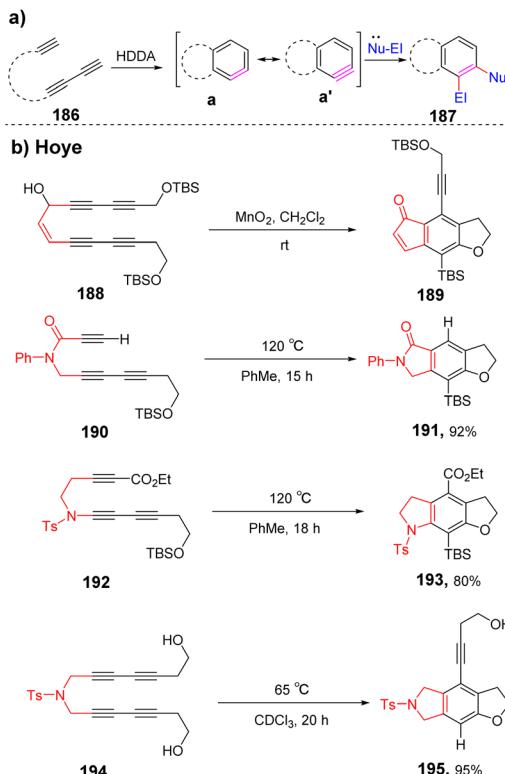
Scheme 25 2-Vinylazetidines with benzene of the [6 + 2] cycloaddition reaction.



Scheme 26 Intramolecular aryne–ene reaction.

a benzene intermediate to obtain a series of aromatic heterocyclic compounds with good yields (Scheme 26).<sup>103</sup> Alder-ene reactions are usually synergistic mechanisms, but a reaction pathway involving free radical intermediates has also been proposed. DFT calculations suggest that the reaction may undergo synergistic or step-by-step mechanisms. Control experiments on *cis*-allyl hydrogen transfer were shown to be carried out in the transition state of *trans*-allyl hydrogen transfer. This process might be caused by the electron-rich alkene group attacking the electron-poor benzene intermediate preferentially, resulting in the formation of C–C bonds taking precedence over C–H bonds, which enables regionally selective hydrogen migration and ensuring stereoselectivity and regional selectivity of the reaction.

**2.2.4 Hexadehydro-Diels–Alder (HDDA) reaction.** The hexadehydro-Diels–Alder reaction was first reported by Johnson and Ueda in 1997.<sup>104,105</sup> As one of the powerful tools of constructing C–C bonds in organic chemistry, it is favored by many researchers.

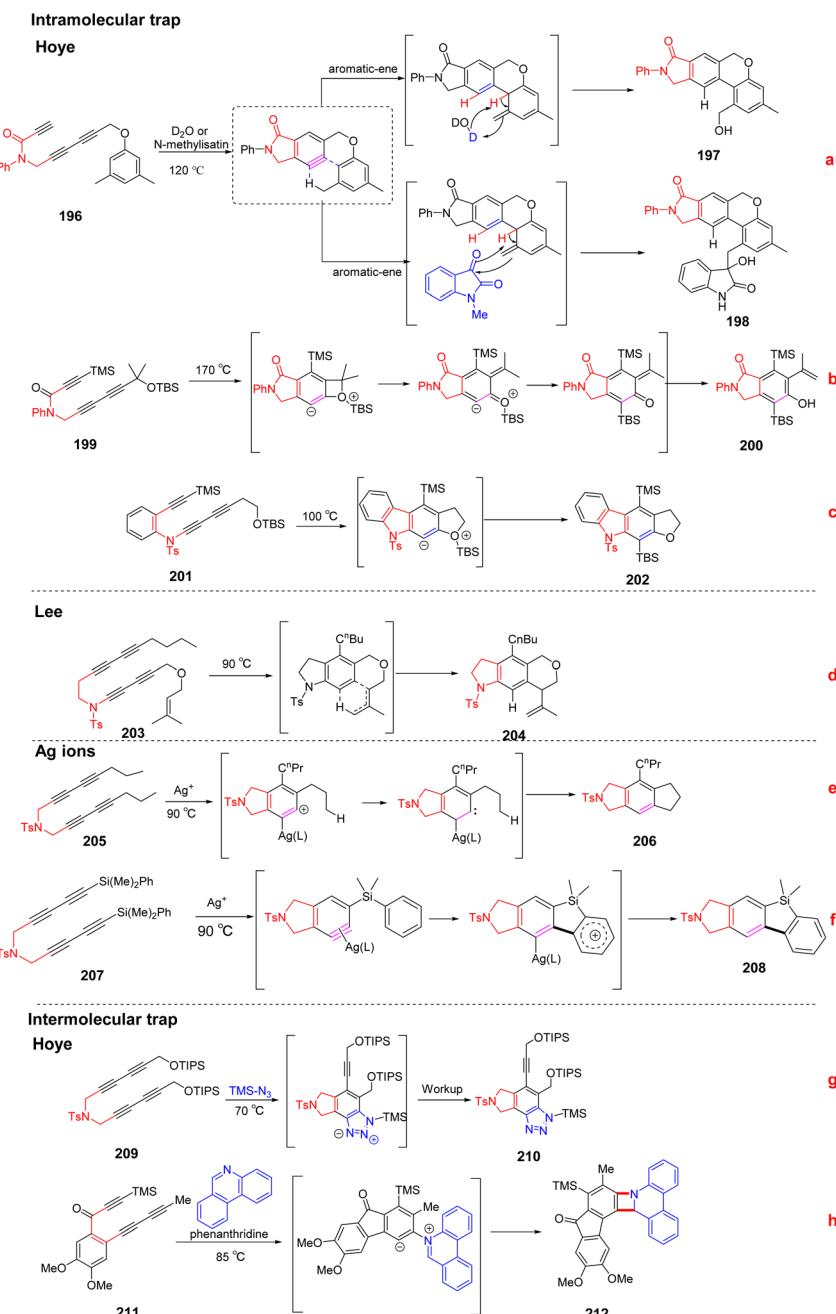


Scheme 27 HDDA cycloaromatization reactions, TBS = *tert*-butyldimethylsilyl, Ts = *para*-toluenesulfonyl. (a) The general mode of HDDA reaction, (b) intramolecular trapping reactions of HDDA.

In 2012, Hoye *et al.* named this type of reaction the six dehydrogenation-Diels–Alder (HDDA) reactions (Scheme 27a)<sup>27</sup>. The first stage of this cascade involves the generation of a highly reactive *ortho*-benzene intermediate by a net [4 + 2] cycloisomerization of a triyne substrate. The benzene can be rapidly ‘trapped’ intramolecularly or intermolecularly with nucleophilic. With their in-depth study of this type of reaction, the method could construct various heterocyclic compounds, especially nitrogen-containing ones.<sup>106–109</sup> The formation of two new carbon–carbon bonds involving the reaction of an intramolecular trap is one of the common HDDA transitions. This transformation is also one of the powerful tools for constructing *N*-containing heterocycles, and some representative reactions are shown in Scheme 27b.

In 2014, Hoye *et al.* reported that HDDA benzenes were utilized to access the first high-yielding aromatic ene reactions.<sup>107</sup> (Scheme 28a) HDDA benzenes intermediates are usually unstable and can react *in situ* with nucleophiles to obtain *N*-containing compounds with complex structures. Subsequently, they have conducted in-depth research on HDDA reaction and enriched the methods for obtaining various nitrogen-containing heterocyclic compounds by changing the HDDA benzenes precursors. In 2019, Hoye *et al.* developed that certain poly-yne substrates could efficiently occur a tandem thermal isomerization by an *o*-QM intermediate (quinone methides) to construct isoindolinone skeleton<sup>110</sup> (Scheme 28b). Experiments and DFT calculations verify that this conversion process may be achieved through multiple reaction





Scheme 28 Examples of the intramolecularly or intermolecularly trap HDDA reaction.

intermediates: thermally generated benzyne, strained benzoxetenonium ion, and *o*-quinone methide base compound. The overall process can be very efficient in obtaining a variety of heterocyclic compounds. A benzenoid ring in fused polycyclic heteroaromatic carbazole (*i.e.*, [2,3]-benzooindole) skeletons can be synthesized in a single step *via* the HDDA reaction of appropriate benzyne precursors. This strategy allows the generation of highly substituted aromatics<sup>111</sup> (Scheme 28c).

Lee *et al.* developed an efficient intramolecular ene reaction of benzene intermediate that directly generate the corresponding benzoheterocycle. Alkene donors are tethered to the

incipient aryne moiety is one of the most favorable reactions for the ene reaction. In addition, the size of the linker linking the alkene donor to the 1,3-diyne moiety and the substituent pattern on the alkene significantly affect the efficiency of the alkene ring closure process, notable, the linking of two 1,3-diyynes or alkenes and heteroatoms in the linker of 1,3-diyne appear to have the most significant impact on the reactivity of various substrates.<sup>112</sup> (Scheme 28d).

It has also been reported that transition metal additives can change the reaction route of benzynes, and one of the earliest examples of a metal addition effect was observed with silver

ions ( $\text{Ag}^+$ ). The current finding is that benzyne is significantly more reactive and selective towards *p*-nucleophiles in the presence of  $\text{Ag}^+$  than the corresponding reaction in the absence of  $\text{Ag}^+$ .  $\text{Ag}^+$  can interact with benzenes to form reactive intermediates that behave like a silver-bound aryl cation or 1,2-carbene-silver carbene (Scheme 28e and f).<sup>16,113–115</sup>

The construction of benzoheterocyclic compounds by HDDA reaction intermolecular trap has also been reported. Hoye *et al.* reported that the 1,3-dipolar (trimethylsilyl azide) and heterocyclic dienes (furan or pyrrole) could be used as an intermolecular trap for benzenes produced by the HDDA reaction of triynes to construct nitrogen-containing heterocycles with relatively high structural complexity fast. This strategy has the potential to synthesize diverse scaffolds and can be a candidate method for the discovery and synthesis of complex drugs<sup>116</sup> (Scheme 28g). The thermal HDDA cycloisomerization of different benzyne precursors (triyne or tetrayne) yields intermediates that can achieve different modes of intermolecular trap reactions<sup>117</sup> (Scheme 28h). The initially formed 1,3-zwitterionic species (i) can collapse intramolecularly, giving a new 1:1 adduct of heterocycle and benzyne; (ii) can react with an added electrophilic third component, giving functionalized heterocyclic products; or (iii) can react with external protic nucleophiles to generate various three-component assemblies after the collapse of ion pairs generated by protonation of zwitterions.

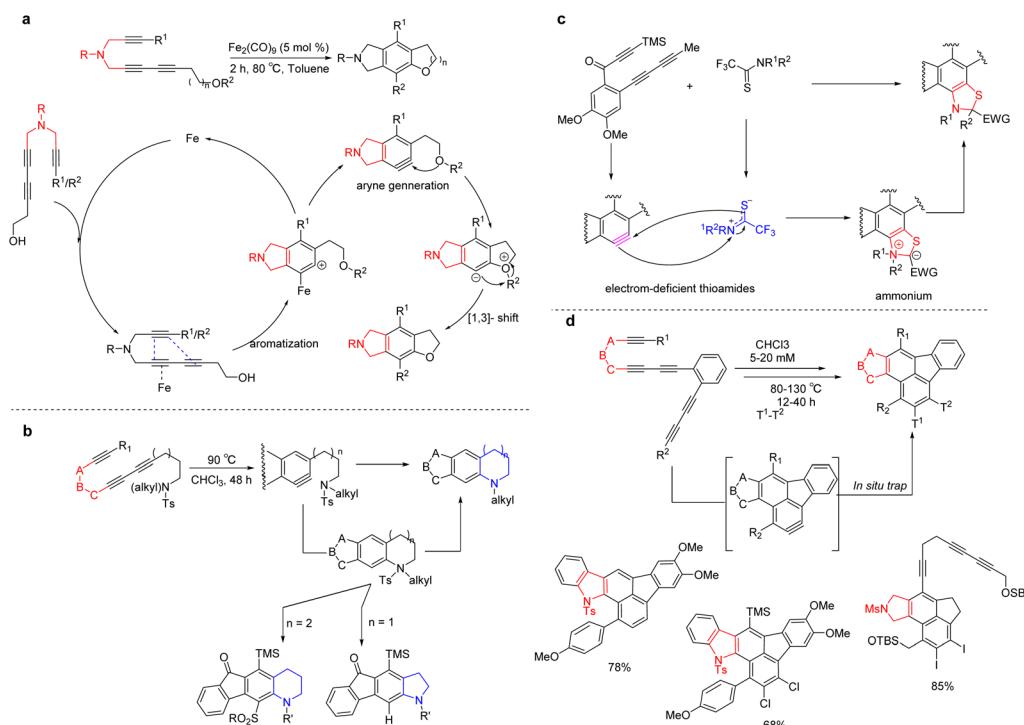
Recently, it has become possible to construct novel nitrogen-containing compounds involving HDDA benzyne intermediates. A facile method for the synthesis of fused

isoindolines *via* the  $\text{Fe}_2(\text{CO})_9$ -catalyzed cascade HDDA strategy was developed by Wang *et al.*<sup>118</sup> The main features of this reaction are simple operation, atomic efficiency, time economy, high yield, and wide substrate range. The reaction process mainly undergoes [4 + 2] cycloaddition to generate the benzyne, intramolecular –OH nucleophile trap, [1, 3]-H shift to give the product (Scheme 29a).

Saturated heterocycles [tetrahydroquinolines ( $n = 2$ ) and indolines ( $n = 1$ )] can be synthesized by using the zwitterionic intermediate produced *via* the reaction of the nitrogen in the tertiary sulfonamide group in the molecule with the benzyne produced by HDDA<sup>119</sup> (Scheme 29b). The different nitrogen heterocycles and the nature of the substituents present on the zwitterionic nitrogen atoms (alkyl *versus* aryl) lead to different reaction pathways. Each reaction result can access a new, saturated, benzo-fused piperidine (*i.e.*, a tetrahydroquinoline) or pyrrolidine (*i.e.*, indoline) ring.

The dihydrobenzothiazole could be obtained by reacting thioamides with electron-withdrawing groups on the carbon atoms of thiocarbonyls with benzenes (produced by the hexahydro-Diels–Alder cycloisomerization reaction)<sup>120</sup> (Scheme 29c). The reaction resulted in the formation of stable ammoniums *via* a rare [3 + 2] cycloaddition reaction. The cycloaddition reactions of different electron-withdrawing groups (including  $-\text{CF}_3$ ,  $-\text{CN}$ , and  $-\text{CO}_2\text{Me}$ ) in thioamides and various benzyne precursors can obtain the target products smoothly.

Construction of polyacene nitrogen-containing fused aromatics by domino HDDA reaction reported by Hoye *et al.* in 2018<sup>121</sup> (Scheme 29d). A series of staged cycloisomerization



**Scheme 29** Cyclization reactions involving different HDDA benzyne precursors. (a) Synthesis of fused isoindolines, (b) construction of saturated heterocycles from triyne, (c) synthesis of dihydrobenzothiazole from thioamides with benzenes, (d) construction of polyacene nitrogen-containing fused aromatics.



reactions can proceed within substrates containing varying numbers of 1,3-diyne units, which each unit separated by a two-atom spacer. The cascade reaction is initiated by a diynophilic alkyne uniquely attached by a three-atom tether. This reagent-free thermal reaction is robust, broad in substrate scope, and yields considerable structural complexity. These methods provide a new channel for the construction of structurally novel and broadly applicable nitrogen-containing compounds.

### 3 Conclusions

In summary, nitrogen-containing heterocyclic compounds are widely found in natural products, drug molecules, and organic functional materials. Among the many methods for preparing benzoheterocycles, the benzyne species cyclization strategy is an attractive method for constructing benzoheterocyclic molecular skeletons because it can efficiently and economically obtain complex and diverse heterocyclic compounds under mild conditions. Examples of new benzyne precursors and new trapping reagents or reactions will continue to emerge as increasing numbers of investigators contemplate the opportunities this approach provides to *in situ* forming benzyne. Studies of benzyne cycloaddition reactions have also provided new mechanistic understanding and insights about benzyne reactivity, and many additional such results are embedded in the primary publications cited here. Given the substantial body of evidence represented by the numerous successes described in this review and the enduring relevance of benzyne to access benzoheterocycle compounds, the field is likely to expand as chemists address challenges in sustainability, atom economy, and new patterns of reactivity for benzazepines synthesis. Overall, the recent developments in benzyne chemistry have been primarily driven by the availability of classical precursors. Further progress in this field is expected to open new avenues for the synthesis of nitrogen-containing heterocyclic compounds under mild, environmentally friendly, low-cost, and operationally simple conditions.

### Conflicts of interest

There are no conflicts to declare.

### Notes and references

- 1 J. A. Garcia-Lopez and M. F. Greaney, *Chem. Soc. Rev.*, 2016, **45**, 6766–6798.
- 2 I. Takahashi, T. Fujita, N. Shoji and J. Ichikawa, *Chem. Commun.*, 2019, **55**, 9267–9270.
- 3 D. Hamprecht, F. Micheli, G. Tedesco, A. Checchia, D. Donati, M. Petrone, S. Terreni and M. Wood, *Bioorg. Med. Chem. Lett.*, 2007, **17**, 428–433.
- 4 W.-T. Jiaang, Y.-S. Chen, T. Hsu, S.-H. Wu, C.-H. Chien, C.-N. Chang, S.-P. Chang, S.-J. Lee and X. Chen, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 687–691.
- 5 J. He, D. Qiu and Y. Li, *Acc. Chem. Res.*, 2020, **53**, 508–519.
- 6 L. L. Fluegel and T. R. Hoye, *Chem. Rev.*, 2021, **121**, 2413–2444.
- 7 Y. Y. Shinichi Saito, *Chem. Rev.*, 2000, **100**, 2901–2915.
- 8 H. H. Wenk, M. Winkler and W. Sander, *Angew. Chem., Int. Ed.*, 2003, **42**, 502–528.
- 9 A. Yoshimura, A. Saito and V. V. Zhdankin, *Chem.*, 2018, **24**, 15156–15166.
- 10 J. Maier and T. B. Marder, *Chem.*, 2021, **27**, 7978–7991.
- 11 R. v. Stoermer and B. Kahlert, *Ber. Dtsch. Chem. Ges.*, 1902, **35**, 1633–1640.
- 12 W. Bachmann and H. Clarke, *J. Am. Chem. Soc.*, 1927, **49**, 2089–2098.
- 13 J. Tan and J. Chen, in *Comprehensive Aryne Synthetic Chemistry*, Elsevier, 2022, pp. 125–221.
- 14 J. Mortier, *Arene chemistry: reaction mechanisms and methods for aromatic compounds*, John Wiley & Sons, 2015.
- 15 P. M. Tadross and B. M. Stoltz, *Chem. Rev.*, 2012, **112**, 3550–3577.
- 16 R. Karmakar and D. Lee, *Chem. Soc. Rev.*, 2016, **45**, 4459–4470.
- 17 M. Uchiyama, T. Miyoshi, Y. Kajihara, T. Sakamoto, Y. Otani, T. Ohwada and Y. Kondo, *J. Am. Chem. Soc.*, 2002, **124**, 8514–8515.
- 18 K. C. Caster, C. G. Keck and R. D. Walls, *J. Org. Chem.*, 2001, **66**, 2932–2936.
- 19 M. Stiles, U. Burckhardt and G. Freund, *J. Org. Chem.*, 1967, **32**, 3718–3719.
- 20 R. S. Berry, J. Clardy and M. E. Schafer, *J. Am. Chem. Soc.*, 1964, **86**, 2738–2739.
- 21 G. Wittig and R. Hoffmann, *Org. Synth.*, 2003, **47**, 4.
- 22 S. Nakazawa, T. Kiyosawa, K. Hirakawa and H. Kato, *J. Chem. Soc., Chem. Commun.*, 1974, 621a.
- 23 S. E. Whitney, M. Winters and B. Rickborn, *J. Org. Chem.*, 1990, **55**, 929–935.
- 24 Y. Himeshima, T. Sonoda and H. Kobayashi, *Chem. Lett.*, 1983, **12**, 1211–1214.
- 25 Y. Liang, X. Hong, P. Yu and K. Houk, *Org. Lett.*, 2014, **16**, 5702–5705.
- 26 T. Wang, D. Niu and T. R. Hoye, *J. Am. Chem. Soc.*, 2016, **138**, 7832–7835.
- 27 T. R. Hoye, B. Baire, D. Niu, P. H. Willoughby and B. P. Woods, *Nature*, 2012, **490**, 208–212.
- 28 R. Karmakar and D. Lee, *Org. Lett.*, 2016, **18**, 6105–6107.
- 29 P. Wessig and G. Müller, *Chem. Rev.*, 2008, **108**, 2051–2063.
- 30 F. García, D. Peña, D. Pérez and E. Gutián, *Modern Aryne Chemistry*, 2021, pp. 27–68.
- 31 K. A. Spence, J. V. Chari, M. Di Niro, R. B. Susick, N. Ukwitegetse, P. I. Djurovich, M. E. Thompson and N. K. Garg, *Chem. Sci.*, 2022, **13**, 5884–5892.
- 32 A. M. Orendt, J. C. Facelli, J. G. Radziszewski, W. J. Horton, D. M. Grant and J. Michl, *J. Am. Chem. Soc.*, 1996, **118**, 846–852.
- 33 R. Warmuth, *Chem. Commun.*, 1998, 59–60.
- 34 J. G. Radziszewski, B. A. Hess Jr and R. Zahradník, *J. Am. Chem. Soc.*, 1992, **114**, 52–57.
- 35 N. Pavliček, B. Schuler, S. Collazos, N. Moll, D. Pérez, E. Gutián, G. Meyer, D. Peña and L. Gross, *Nat. Chem.*, 2015, **7**, 623–628.



36 T. Kitamura, Z. Meng and Y. Fujiwara, *Tetrahedron Lett.*, 2000, **41**, 6611–6614.

37 T. Kitamura, M. Yamane, K. Inoue, M. Todaka, N. Fukatsu, Z. Meng and Y. Fujiwara, *J. Am. Chem. Soc.*, 1999, **121**, 11674–11679.

38 S. Kovacs, A. I. Csincsi, T. Z. Nagy, S. Boros, G. Timári and Z. Novak, *Org. Lett.*, 2012, **14**, 2022–2025.

39 K. Devaraj, F. J. Ingner, C. Sollert, P. J. Gates, A. Orthaber and L. T. Pilarski, *J. Org. Chem.*, 2019, **84**, 5863–5871.

40 B. Michel and M. F. Greaney, *Org. Lett.*, 2014, **16**, 2684–2687.

41 T. Ikawa, T. Nishiyama, T. Nosaki, A. Takagi and S. Akai, *Org. Lett.*, 2011, **13**, 1730–1733.

42 Q. Chen, H. Yu, Z. Xu, L. Lin, X. Jiang and R. Wang, *J. Org. Chem.*, 2015, **80**, 6890–6896.

43 M. Mesgar and O. Daugulis, *Org. Lett.*, 2016, **18**, 3910–3913.

44 T. Ikawa, S. Masuda, H. Nakajima and S. Akai, *J. Org. Chem.*, 2017, **82**, 4242–4253.

45 Y. Li, C. Mück-Lichtenfeld and A. Studer, *Angew. Chem., Int. Ed.*, 2016, **55**, 14435–14438.

46 N. Mariet, M. Ibrahim-Ouali and M. Santelli, *Tetrahedron Lett.*, 2002, **43**, 5789–5791.

47 P. Maurin, M. Ibrahim-Ouali, J.-L. Parrain and M. Santelli, *J. Mol. Struct.*, 2003, **637**, 91–100.

48 Z. Qiu and Z. Xie, *Angew. Chem.*, 2009, **121**, 5839–5842.

49 J. Caeiro, D. Pena, A. Cobas, D. Perez and E. Guitián, *Adv. Synth. Catal.*, 2006, **348**, 2466–2474.

50 J. B. Feltenberger, R. Hayashi, Y. Tang, E. S. Babiash and R. P. Hsung, *Org. Lett.*, 2009, **11**, 3666–3669.

51 Z.-X. Ma, J. B. Feltenberger and R. P. Hsung, *Org. Lett.*, 2012, **14**, 2742–2745.

52 Z. Xiong, J. Duan, X. Li, X. Wang, Y. Addepalli, M. Lu, W. Yao, L. He and Z. Wang, *J. Org. Chem.*, 2021, **86**, 12403–12411.

53 J. Shi, H. Xu, D. Qiu, J. He and Y. Li, *J. Am. Chem. Soc.*, 2017, **139**, 623–626.

54 C. Lu, A. V. Dubrovskiy and R. C. Larock, *J. Org. Chem.*, 2012, **77**, 2279–2284.

55 D. Chang, D. Zhu and L. Shi, *J. Org. Chem.*, 2015, **80**, 5928–5933.

56 I. Briguglio, S. Piras, P. Corona, E. Gavini, M. Nieddu, G. Boatto and A. Carta, *Eur. J. Med. Chem.*, 2015, **97**, 612–648.

57 Y. Ren, L. Zhang, C.-H. Zhou and R.-X. Geng, *Med. Chem.*, 2014, **4**, 640–662.

58 X.-M. Peng, G.-X. Cai and C.-H. Zhou, *Curr. Top. Med. Chem.*, 2013, **13**, 1963–2010.

59 R. R. Kale, V. Prasad, P. P. Mohapatra and V. K. Tiwari, *Monatsh. Chem.*, 2010, **141**, 1159–1182.

60 E. Loukopoulos and G. E. Kostakis, *Coord. Chem. Rev.*, 2019, **395**, 193–229.

61 T. Jin and Y. Yamamoto, *Angew. Chem., Int. Ed.*, 2007, **46**, 3323–3325.

62 C. Spiteri, S. Keeling and J. E. Moses, *Org. Lett.*, 2010, **12**, 3368–3371.

63 Z. Liu, F. Shi, P. D. Martinez, C. Raminelli and R. C. Larock, *J. Org. Chem.*, 2008, **73**, 219–226.

64 P. Li, J. Zhao, C. Wu, R. C. Larock and F. Shi, *Org. Lett.*, 2011, **13**, 3340–3343.

65 G. Chen, M. Hu and Y. Peng, *J. Org. Chem.*, 2018, **83**, 1591–1597.

66 P. Sureshbabu, V. Bhajammanavar, V. S. K. Choutipalli, V. Subramanian and M. Baidya, *Chem. Commun.*, 2022, **58**, 1187–1190.

67 L. Campbell-Verduyn, P. H. Elsinga, L. Mirfeizi, R. A. Dierckx and B. L. Feringa, *Org. Biomol. Chem.*, 2008, **6**, 3461–3463.

68 G. Singh, R. Kumar, J. Swett and B. Zajc, *Org. Lett.*, 2013, **15**, 4086–4089.

69 F. Shi, J. P. Waldo, Y. Chen and R. C. Larock, *Org. Lett.*, 2008, **10**, 2409–2412.

70 H. Ankati and E. Biehl, *Tetrahedron Lett.*, 2009, **50**, 4677–4682.

71 S. Chandrasekhar, M. Seenaiah, C. L. Rao and C. R. Reddy, *Tetrahedron*, 2008, **64**, 11325–11327.

72 F. Zhang and J. E. Moses, *Org. Lett.*, 2009, **11**, 1587–1590.

73 M. Kleoff, L. Boeser, L. Baranyi and P. Heretsch, *Eur. J. Org. Chem.*, 2021, **2021**, 979–982.

74 S. Yoshida, T. Morita and T. Hosoya, *Chem. Lett.*, 2016, **45**, 726–728.

75 J. Guo, I. C. Kiran, R. S. Reddy, J. Gao, M. Tang, Y. Liu and Y. He, *Org. Lett.*, 2016, **18**, 2499–2502.

76 J. Yuan, X. Y. Wu and F. Sha, *Eur. J. Org. Chem.*, 2016, **2016**, 2929–2932.

77 H. Xu, J. He, J. Shi, L. Tan, D. Qiu, X. Luo and Y. Li, *J. Am. Chem. Soc.*, 2018, **140**, 3555–3559.

78 H. Ryu, J. Seo and H. M. Ko, *J. Org. Chem.*, 2018, **83**, 14102–14109.

79 Y.-K. Li, M.-X. Cui, F. Sha, Q. Li and X.-Y. Wu, *Org. Biomol. Chem.*, 2019, **17**, 8963–8968.

80 A. S. Alshreimi, G. Zhang, T. W. Reidl, R. L. Peña, N. G. Koto, S. M. Islam, D. J. Wink and L. L. Anderson, *Angew. Chem., Int. Ed.*, 2020, **59**, 15244–15248.

81 H. Yuan, D. L. Lu, C. Liang and D. L. Mo, *Adv. Synth. Catal.*, 2022, **364**, 1409–1414.

82 Q. Jin, D. Zhang and J. Zhang, *RSC Adv.*, 2020, **10**, 30620–30623.

83 D. K. Kim, J. Y. Son, I. Jung, N. Heo, S. H. Han, D. Kim and P. H. Lee, *Adv. Synth. Catal.*, 2021, **363**, 1358–1367.

84 W. Zhang, J. Bu, L. Wang, P. Li and H. Li, *Org. Chem. Front.*, 2021, **8**, 5045–5051.

85 H. Takikawa, A. Nishii, H. Takiguchi, H. Yagishita, M. Tanaka, K. Hirano, M. Uchiyama, K. Ohmori and K. Suzuki, *Angew. Chem., Int. Ed.*, 2020, **59**, 12440–12444.

86 Y. Minami, Y. Furuya and T. Hiyama, *Chem. - Eur. J.*, 2020, **26**, 9471–9474.

87 T. Kubo, T. Fujita and J. Ichikawa, *Chem. Lett.*, 2020, **49**, 264–266.

88 J. He, Z. Jia, H. Tan, X. Luo, D. Qiu, J. Shi, H. Xu and Y. Li, *Angew. Chem., Int. Ed.*, 2019, **58**, 18513–18518.

89 X. Xiao and T. R. Hoye, *J. Am. Chem. Soc.*, 2019, **141**, 9813–9818.

90 J.-C. Castillo, J. Quiroga, R. Abonia, J. Rodriguez and Y. Coquerel, *Org. Lett.*, 2015, **17**, 3374–3377.



91 J.-C. Castillo, A. Tigreros, Y. Coquerel, J. Rodríguez, M. A. Macías and J. Portilla, *ACS Omega*, 2019, **4**, 17326–17339.

92 J. Torres-Alacan, *J. Org. Chem.*, 2016, **81**, 1151–1156.

93 S. Stokes, M. Bekkam, M. Rupp and K. T. Mead, *Synlett*, 2012, **2012**, 389–392.

94 J.-A. García-López and M. F. Greeney, *Chem. Soc. Rev.*, 2016, **45**, 6766–6798.

95 H. Takikawa, A. Nishii, T. Sakai and K. Suzuki, *Chem. Soc. Rev.*, 2018, **47**, 8030–8056.

96 I. Pozo, E. Guitián, D. Pérez and D. Peña, *Acc. Chem. Res.*, 2019, **52**, 2472–2481.

97 Y.-L. Wei, G. Dauvergne, J. Rodriguez and Y. Coquerel, *J. Am. Chem. Soc.*, 2020, **142**, 16921–16925.

98 T. R. Silva, V. V. Souza and C. Raminelli, *ChemistrySelect*, 2022, **7**, e202203039.

99 S. Liu, P. Xie, L. Wu, J. Zhao, Z. Cai and L. He, *Org. Chem. Front.*, 2022, **9**, 1550–1555.

100 A. Guin, R. N. Gaykar, S. Deswal and A. T. Biju, *Org. Lett.*, 2021, **23**, 7456–7461.

101 Z. Wang, Y. Addepalli and Y. He, *Org. Lett.*, 2018, **20**, 644–647.

102 T. Aoki, S. Koya, R. Yamasaki and S. Saito, *Org. Lett.*, 2012, **14**, 4506–4509.

103 D. A. Candito, J. Panteleev and M. Lautens, *J. Am. Chem. Soc.*, 2011, **133**, 14200–14203.

104 A. Z. Bradley and R. P. Johnson, *J. Am. Chem. Soc.*, 1997, **119**, 9917–9918.

105 K. Miyawaki, R. Suzuki, T. Kawano and I. Ueda, *Tetrahedron Lett.*, 1997, **38**, 3943–3946.

106 B. Baire, D. Niu, P. H. Willoughby, B. P. Woods and T. R. Hoye, *Nat. Protoc.*, 2013, **8**, 501–508.

107 D. Niu and T. R. Hoye, *Nat. Chem.*, 2014, **6**, 34–40.

108 D. Niu, P. H. Willoughby, B. P. Woods, B. Baire and T. R. Hoye, *Nature*, 2013, **501**, 531–534.

109 L. L. Fluegel and T. R. Hoye, *Chem. Rev.*, 2021, **121**, 2413–2444.

110 H. Shen, X. Xiao and T. R. Hoye, *Org. Lett.*, 2019, **21**, 1672–1675.

111 Y. Wang, L. Zheng and T. R. Hoye, *Org. Lett.*, 2018, **20**, 7145–7148.

112 R. Karmakar, P. Mamidipalli, S. Y. Yun and D. Lee, *Org. Lett.*, 2013, **15**, 1938–1941.

113 D. Lee and S. Ghorai, *Silver Catalysis in Organic Synthesis*, 2019, 33–83.

114 K.-P. Wang, S. Y. Yun, P. Mamidipalli and D. Lee, *Chem. Sci.*, 2013, **4**, 3205–3211.

115 N.-K. Lee, S. Y. Yun, P. Mamidipalli, R. M. Salzman, D. Lee, T. Zhou and Y. Xia, *J. Am. Chem. Soc.*, 2014, **136**, 4363–4368.

116 J. Chen, B. Baire and T. R. Hoye, *Heterocycles*, 2014, **88**, 1191.

117 S. Arora, J. Zhang, V. Pogula and T. R. Hoye, *Chem. Sci.*, 2019, **10**, 9069–9076.

118 J. K. Vandavasi, W.-P. Hu, C.-T. Hsiao, G. C. Senadi and J.-J. Wang, *RSC Adv.*, 2014, **4**, 57547–57552.

119 T. Wang and T. R. Hoye, *J. Am. Chem. Soc.*, 2016, **138**, 13870.

120 J. Zhang, A. C. Page, V. Palani, J. Chen and T. R. Hoye, *Org. Lett.*, 2018, **20**, 5550–5553.

121 X. Xiao and T. R. Hoye, *Nat. Chem.*, 2018, **10**, 838–844.

