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Multicomponent encapsulation into fully degradable protein nanocarriers *via* interfacial azide–alkyne click reaction in miniemulsion allows the co-delivery of immunotherapeutics†

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Encapsulation of multiple adjuvants along with antigens into nanocarriers allows a co-delivery to antigen-presenting cells for the synergistic induction of robust immune responses. However, loading cargoes of different molar masses, polarities, and solubilities in high efficiencies remains a challenge. Therefore, we developed a strategy to encapsulate a triple combination of the so-called adjuvants, *i.e.* with Resiquimod (R848), muramyl dipeptide (MDP) and polyinosinic-polycytidylic acid (Poly(I:C)) into human serum albumin (HSA) nanocarriers. The loading is conducted *in situ* while the nanocarrier is formed by an orthogonal and metal-free click reaction at the interface of an inverse miniemulsion. By this unique approach, high encapsulation efficiency without harming the cargo during the nanocarrier formation process and regardless of their physical properties is achieved, thus keeping their bioactivity. Furthermore, we demonstrated high control over the encapsulation efficiency and varying the amount of each cargo did not influence the efficiency of multicomponent encapsulation. Azide-modified HSA was crosslinked with hexanediol dipropiolate (HDDP) at the interface of a water-in-oil miniemulsion. Varying the crosslinker amount allowed us to tailor the density and degradation rates of the protein shell. Additional installation of disulfide bonds into the crosslinker created redox-responsive nanocarriers, which degraded both by protease and under reducing conditions with dithiothreitol. The prepared HSA nanocarriers were efficiently taken up by dendritic cells and exhibited an additive cell activation and maturation, exceeding the nanocarriers loaded with only a single drug. This general protocol allows the orthogonal and metal-free encapsulation of various drugs or adjuvants at defined concentrations into the protein nanocarriers.

New concepts

In this work, we demonstrated an efficient *in situ* encapsulation of up to four cargo molecules with different molar mass and solubility into protein nanocarriers, allowing co-delivery of multiple therapeutics for a synergistic therapeutic effect. The novel combination of a reductive-responsive crosslinker with enzymatically degradable proteins as the nanocarrier material created a dual-responsive nanocarrier with complete degradability and release rates in cells. A metal-free azide-alkyne crosslinking chemistry for the formation of the protein nanocarriers ensures a mild and biorthogonal encapsulation of sensible cargoes where their bioactivity is preserved. The developed inverse miniemulsion process enables multicomponent encapsulation of a broad range of cargoes from small to large, hydrophilic to hydrophobic and inorganic, in numerous possible combinations to achieve multifunctional nanocarriers for a broad range of applications, from biomedicine to material science.

Introduction

Biochemical processes in the body rely on a complex interplay of multiple components. If medical therapy aims to mimic or enhance those processes, multiple therapeutic components have to be integrated into one system. For example, in immunotherapy, the biggest challenge is the exhaustion of immune cells due to the immunosuppressive tumor microenvironment.^{1–3} Hence, effective tumor treatment relies on the induction of a strong and durable immune response. Monotherapy proved to be insufficient to overcome these challenges and efforts have to be made to realize a combinatorial treatment with multiple components.^{4,5} True to the motto “The more the merrier”, a combination of multiple

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components enables additive effects for enhanced treatment efficacy. In vaccination approaches, a combination of multiple adjuvants yields a synergistic effect in dendritic cell-directed T cell stimulation, increasing the vaccination effect.^{6–9}

The key to success is an efficient process for multicomponent encapsulation of cargoes with different physicochemical properties such as molar mass, polarity, and solubility.

Common methods to prepare protein nanocarriers include desolvation, self-assembly or gelation. For example, Abraxane[®] is a nanoparticle based on albumin-bound paclitaxel, which is commercially used in cancer therapy. Paclitaxel is entrapped into the albumin nanoparticle during the albumin aggregation by a desolvation process. Although efficient entrapment of drugs can be achieved with those methods, there is a lack of control when multiple drugs should be encapsulated simultaneously into one nanocarrier. Especially when the cargo molecules have different physicochemical properties such as solubility, controlled and efficient encapsulation of all components is challenging with methods relying on random entrapment.

To design the optimal nanocarrier, several requirements are necessary: (1) simultaneous encapsulation of multiple cargoes, (2) selective reaction for the shell formation without harming the cargo, (3) dense carriers for the transport in the body, and (4) degradation of the carriers at the target site. Combining all requirements in one process proved to be challenging. Our developed process combines all four requirements and enables controlled and efficient multicomponent encapsulation into fully degradable protein nanocarriers (PNCs) *via* azide–alkyne click reaction in inverse miniemulsion (Fig. 1).

For the first requirement, an inverse miniemulsion allows the simultaneous encapsulation of different water-soluble cargo molecules into nanocarriers with a defined concentration.¹⁰ In an inverse miniemulsion, the shell material

is crosslinked at the interface of the water-droplets *via* an oil-soluble crosslinker, forming a nanocarrier with a liquid core. The water-soluble cargo molecules are encapsulated inside the nanocarriers during the interfacial shell formation. For the second requirement, click reactions enable selective linkage between shell material and crosslinker without involving the cargo in the shell formation reaction.^{11,12} Previously, protein nanocarriers were prepared by UV-initiated photoclick tetrazole-ene reaction in inverse miniemulsion.¹³ However, UV-light or other harsh initiators, such as metals,¹⁴ can destroy sensitive cargoes leading to loss of bioactivity. A crosslinker such as hexanediol dipropiolate (HDDP) with an activated alkyne reacts without a catalyst in an azide–alkyne click reaction.^{15,16} The amount of crosslinker influences the number of links between the shell material and might control the density of the nanocarrier shell for the third requirement. The last requirement is a degradable shell material ensuring the release of the cargo from nanocarriers. Proteins have been widely used as nanocarrier materials, because of their natural biocompatibility and degradability.^{13,17–19}

Results and discussion

For the azide–alkyne click reaction, human serum albumin (HSA) was functionalized at the lysine residue by a metal-free transfer reaction into azide groups using 1-imidazol azide hydrochloride.²⁰ After functionalization, the fluorescamine assay showed that about 19–24 amines of the 59 lysine residues per HSA (30–35 accessible lysines) were transformed into azides. The degree of functionalization changed with pH value as lower pH values led to lower nucleophilicity of the amine and less transfer reactions, whereas pH 9.5 gave an optimal balance between yielding high number of transfers and maintaining the protein structure (Table S1, ESI[†]). A comparison of the IR



Fig. 1 Multicomponent encapsulation into protein nanocarriers through interfacial azide–alkyne crosslinking in inverse miniemulsion.





Fig. 2 Reaction scheme for the formation of protein nanocarriers through azide–alkyne click reaction (top). (A) IR spectra of native (···), azide-functionalized (---) and human serum albumin nanocarriers (—). (B) SEM images of HSA-NCs using a HDDP (□) and HDDP-SS (□) ratio (Scale bar: 0.5 μm). (C) Analytic data of HSA NCs using HDDP and HDDP-SS.

spectra of natural and azide-modified HSA showed successful modification as demonstrated by the presence of the characteristic azide signal at 2100 cm^{-1} (Fig. 2A). Furthermore, no significant changes in the secondary structure of the protein were observed by CD spectroscopy (Fig. S6, ESI†). Therefore, the modified protein remained highly water-soluble ($>100\text{ mg mL}^{-1}$) further highlighting the mild conditions during the transfer reaction. When the “azidation” was conducted with the horseradish peroxidase (HRP) Table S2 and Fig. S17, ESI† the enzymatic activity was reduced by 40%, probably due to the high basic pH of 11 during the transfer reaction (Fig. S19, ESI†).

For the formation of protein nanocarriers by azide–alkyne click reaction, we chose the dialkyne hexanediol dipropiolate (HDDP) as a crosslinker of the azide-modified proteins. The carbonyl group located next to the alkyne moiety activates the alkyne by an inductive effect, allowing a click reaction without using a metal catalyst.¹⁵ Moreover, inserting a disulfide bond into the chemical structure of HDDP (*i.e.* to HDDP-SS, *cf.* Fig. 2) created a crosslinker prone to degradation by a reducing agent.¹⁶

The PNCs were prepared in an inverse water-in-oil miniemulsion with cyclohexane as the continuous and an aqueous buffer as the dispersed phase (Scheme S2, ESI†). The aqueous nanodroplets were prepared with high shear forces using

ultrasonication and were stabilized by the surfactant P((E/B)-*b*-EO) (KLE).²¹ The protein shell formed through an interfacial crosslinking by click reaction at the water droplet interface after addition of HDDP or HDDP-SS to the inverse miniemulsion. In the IR spectra of the PNCs, the azide signal at 2100 cm^{-1} decreased due to the formation of the triazol during the click reaction (Fig. 2A). The hydrodynamic diameters of the PNCs (d_{h}) in cyclohexane were determined to be approx. 300 nm by dynamic light scattering (DLS). After redispersion in water using sodium dodecyl sulfate (SDS, 0.02 mM) as the surfactant followed by washing, the diameters decreased to approx. 250 nm with a zeta potential of approx. -30 mV (Fig. 2C and Table S4, ESI†). Scanning electron micrographs and transmission electron micrographs revealed a core-shell morphology of the PNCs (Fig. 2B and Fig. S8, ESI†). In addition, the PNCs based on HRP exhibited similar results in size and morphology (Fig. S17 and S18, ESI†). This demonstrated the excellent reproducibility and transferability for forming PNCs by the developed protocol. Moreover, the HRP nanocarriers were still enzymatically active, showing approx. 60% of the native enzyme activity (Fig. S19, ESI†).

By varying the amount of HDDP crosslinker, shell thickness and size of PNCs could be controlled (Fig. 3). When the





Fig. 3 Influence of alkyne:azide ratio on the shell thickness (■) and diameter (Δ) of protein nanocarriers. Scanning electron micrographs of HSA nanocarriers using a 2:1 (left) and 20:1 (right) ratio (Scale bar: 0.5 μm).

interfacial crosslinking of azide-modified protein with HDDP was performed with an alkyne:azide ratio of 20:1, PNCs with a $d_h = 250$ nm and approx. 20 nm shell thickness were obtained. With a lower alkyne:azide ratio, the shell thickness decreased, while the diameters increased, probably due to an increased swelling of the softer protein shell (Fig. 3). The amount of crosslinker used additionally influenced the encapsulation efficiency after water transfer. The encapsulation efficiency of dextran-sulforhodamine B ($M = 10$ kDa) decreased from 62% to 39% when 20 equiv. HDDP or only 2 equiv. HDDP were used, respectively, to prepare the PNCs. By encapsulating the low molecular weight dye sulforhodamine-101 ($M = 606.71$ g mol⁻¹), a lower crosslinking density was detected, leading to a faster diffusion through the nanocarrier shell, probably attributed to the thinner and looser shell walls when low amounts of HDDP were used (Fig. S12, ESI[†]).

The degradation of the PNCs was investigated by monitoring the release of dextran-rhodamine B ($M = 10$ kDa, $\lambda_{em} = 570$ nm) under different conditions. Crosslinking with different

amounts of HDDP or HDDP-SS enabled investigating the influence of shell density on the degradation kinetics. The HSA-NCs degraded upon the addition of proteinase K (5 U mL⁻¹) and the amount of released dye was detected in the aqueous supernatant. Depending on the time of the degradation experiment, the amount of released dye increased and reached a plateau of approx. 90% dye after 96 h in all cases (Fig. 4A). However, the crosslinking density, *i.e.* the amount of HDDP used during the PNC formation, influenced the degradation rate of the protein shell and thus the release of the dye. Time and crosslinker dependence of dye-release was also observed for the enzymatic degradation of PNCs crosslinked with HDDP-SS (Fig. 4B). The HSA-HDDP-SS NCs could be cleaved by proteinase K as well as under reducing conditions, *e.g.* by adding dithiothreitol (DTT) as a reducing agent. The PNCs crosslinked with different amounts of HDDP-SS were treated with proteinase K (2 U mL⁻¹) and DTT (3 mM), respectively and in combination, and the amount of released dextran-rhodamine B was measured over time, indicating the



Fig. 4 Release kinetics of dextran-rhodamine B (10 kDa) from HSA NCs prepared with different crosslinking densities as indicated by the ratio of HDDP or HDDP-SS/equiv.: (A) PNCs crosslinked with HDDP and degraded with proteinase K (5 U mL⁻¹), (B) PNCs crosslinked with HDDP-SS and degraded with proteinase K (5 U mL⁻¹), (C) PNCs crosslinked with HDDP-SS and degraded with proteinase K (blue, 5 U mL⁻¹) and dithiothreitol (yellow, 3 mM).



release kinetics were influenced by the crosslinking degree. The resulting PNCs crosslinked with HDDP-SS obtained the property of dual-degradation by DTT and proteinase K.

The degradation of the PNCs was also visible by the naked eye, as the turbid PNCs dispersion completely cleared up upon DTT addition (Fig. S15, ESI†). DLS was applied to further quantify the degradation rates of PNCs: Following treatment with proteinase K and DTT, respectively, the size of the NCs decreased significantly from 254 nm to 71 nm with a broadening of the polydispersity index to >0.4 (Fig. S13 and S14, ESI†). Additionally, the derived count rate (DCR) decreased significantly during the degradation experiment, indicating the formation of smaller fragments with lower scattering intensity. The derived count rate was monitored upon DTT treatment and continuously decreased to the lower limit of the DLS apparatus indicating an almost complete degradation of the PNCs (Fig. S15, ESI†).

Since anti-tumor vaccination still lacks the effective response of immune cells against tumors in most cases of vaccinated patients, the role of nanocarriers for co-delivery of vaccine components needs to step into focus. In the vaccination process, the vaccine, consisting of antigen and adjuvants, is taken up by dendritic cells and the antigens are presented on major histocompatibility complex (MHC) molecules of the DCs to the T cells.^{22,23} Subsequently, antigen-specific T cells are activated and proliferate to attack cells bearing the tumor-specific antigen eventually leading to tumor cell killing.²⁴ Crucial for the success of T cell priming is a strong activation of antigen-presenting cells (*e.g.* DCs) by adjuvants inducing expression of costimulatory molecules and the release of

activating pro-inflammatory cytokines. Choosing the correct type and application route of adjuvants is of great importance for the induction of robust immune responses as each adjuvant bind to distinct receptors triggering different signaling pathways in antigen-presenting cells.^{25,26} Therefore, a combination of several different adjuvants simultaneously stimulates different receptors, leading to an additive DC activation.²⁷ To achieve a high local concentration of adjuvants in one DC, all components must be delivered simultaneously.¹⁸

Resiquimod (R848) is a small molecule ($M = 350.8 \text{ g mol}^{-1}$) with low water-solubility ($>1 \text{ mg mL}^{-1}$) and acts as an agonist for the toll-like receptors (TLR) 7 and 8.²⁸ The water-soluble peptidoglycan muramyl dipeptide (MDP; $M = 492.5 \text{ g mol}^{-1}$) has been shown to be recognized by nucleotide-binding oligomerization domain-containing protein (NOD) 2.²⁹ The double-stranded RNA mimic polyinosinic:cytidylic acid (Poly(I:C); 0.2–1 kb) acts as a TLR 3 ligand.³⁰ A major challenge is to find a compatible protocol for the multicomponent encapsulation of cargoes with such different properties in terms of solubility and molecular weight. In this study, we combined R848, MDP, and Poly(I:C) and added Cy5-Oligo (5 kDa) as an additional cargo acting as a fluorescent dye (Fig. 5A), allowing us to investigate the cellular uptake of PNCs by DCs. All four components were successfully encapsulated into PNCs through the azide-alkyne click reaction in inverse miniemulsion demonstrating the excellent feasibility of developed nanocarrier formation for the encapsulation of multiple components. Even though dimethyl sulfoxide (DMSO) was used as a solvent for R848 (14 vol% in the disperse phase) stable droplets could be formed with no influence on the interfacial protein shell



Fig. 5 (A) Chemical structures of Cy5-Oligo dye (●) and adjuvants R848 (▲), MDP (★) and Poly(I:C) (■) encapsulated into HSA NCs. (B) Encapsulation efficiency of cargoes in different combinations into HSA NCs crosslinked with HDDP in percent. (C) Confocal laser scanning microscopy images of bonemarrow-derived dendritic cells (green) and ingested HSA NCs (red). (D) Upregulation of DC maturation markers CD80 and CD86 after stimulation with adjuvant-loaded HSA NCs with either single loading, triple loading or a mixture of single-loaded NCs. BMDCs (2×10^5 cells per mL) were incubated with differently loaded HSA NC formulations ($1\text{--}100 \mu\text{g mL}^{-1}$) or lipopolysaccharid (LPS) (100 ng mL^{-1}) as a positive control for 24 h. Surface expression of CD80 and CD86 of PNC-treated BMDCs was measured by flow cytometry.



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