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Recent advancement in efficient metal oxide-based flexible perovskite solar cells: a short review

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Solar beams contain solar energy, which releases a great deal of energy. Over the last few years, the interest in and progress toward environmentally sound technologies has increased tremendously. Clean energy is being implemented, and solar cell technology is winning the race. Perovskite solar cells (PSCs) are a possible alternative to conventional Si-based solar cells, which are heavy, expensive, and fragile. As a result, PSC development has advanced very well in a short period of time. For the continuous and efficient conversion of solar energy, a variety of metal oxides have been used to develop different devices for terrestrial and space photovoltaics. These metal oxide materials have an encouraging perspective due to their vast accessibility, non-destructiveness, and chemical stability and can provide an extensive diversity of choices for designing and fabricating the device under ambient conditions. The efficiency and stability of PSC devices are increased thanks to the physical, chemical, and electrical features of metal oxides. An original, reliable, and comprehensive method for developing efficient metal oxide-based flexible PSCs is to optimize the preparative parameters for metal oxides. This short review enthusiastically discusses about metal oxides and how they affect flexible perovskite solar cells, which have been integrated with energy-level compatibility. The recent advancement of metal oxide-based flexible perovskite solar cells has also been discussed and we expect the development of low-priced practicable power stock to advance ecofriendly feasibility in the coming years using flexible perovskite solar cells.

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1. Introduction

The requirement of energy is rising at an astonishing proportion to meet the stride of methodical developments and increasing world population. Currently, the three primary fossil



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fuels (oil, natural gas, and coal) accomplish more than 75–80% of global energy loads.^{1,2} The amount of energy consumed globally each year is currently 15 TW, but it is predicted to increase to 30 TW by 2050. Of this, around 80% of the demand will be met by these fossil fuels, while the rest come from nuclear power, hydropower, and renewable energy sources. On the one side is this increased power consumption, while the other is the depletion of natural resources and raw materials, increasing manufacturing prices, and serious ecological concerns such as the climate catastrophe brought on by the unjustified use of non-renewable energy sources.

Because fossil fuels contribute to global warming and produce greenhouse gases when they are burned, the threat to the earth's ability to support life is increased. These conventional energy sources are insufficient and based on current consumption rates, it is predicted that oil, natural gas, and coal reserves will run out in the next 50, 60, and 120–150 years, respectively. Although it may seem that fossil fuels will continue to dominate the energy sector, one must consider the high risks that these fuels pose to our planet's atmosphere. For example, a 35% increase in CO₂ levels will result in an increase in the global temperature of 1 °C, which would induce a sea level to rise.^{3–5}

The risk to the survival of civilization has increased due to the continuously rising energy consumption, which has a detrimental effect due to the greenhouse emissions and the climate catastrophe, it also indirectly sparks struggle over the availability of non-renewable energy sources. Although the situation has been significantly prevented by improvements in the renewable energy sector, many tasks still need to be completed. Less than 20% of the world's energy demand is met by renewable energy sources.³ Investments, government policies, industrial expansion and demand, public awareness, and other factors, all play a role in the rapid development of the renewable energy sector. Many nations, both developed and developing, are attempting to increase the production of renewable energy, but the demand for clean, sustainable resources has decreased due to a number of issues, including affordability, mobility, and usability. To meet these challenges, considerable financial and human resources are devoted to research and development in renewable energy. Due to the abundance of resources, the direct conversion of energy into electricity without any intermediate states, the stability and long life of solar cells, and photovoltaic energy is one of the primary concentrations among renewable energies. In a decade or two, electricity-based systems will replace hydrocarbon fuels due to the rapid development of electric vehicles by top manufacturers such as Tesla and Mahindra. It will be futile to generate power for cars utilizing fossil fuels. This issue can only be resolved through photovoltaic electricity production. The amorphous Si thin-film solar cell was developed in the late 1970s, and it is comparable to conventional crystalline Si-based solar cells in terms of flexibility and light weight. The structural rigidity and up to this point flexible thin-film modules provide incredibly original energy source possibilities. The mechanically rigid and thus far flexible modules based on thin films offer tremendously unique energy source solutions.⁶ In recent times, new material schemes (*viz.*, organic semiconductors and metal halide-based perovskites) have made a swift advancement. Consequently, in the coming years,

Table 1 Power conversion efficiency (PCE) of different single-junction solar cell devices at 25 °C with a global AM1.5 spectrum (1000 W m⁻²). Adapted and modified with permission from ref. 7, Copyright (2021), Wiley

Classifications	V _{OC} (V)	J _{SC} (mA cm ⁻²)	FF (%)	PCE (%)
Crystalline cell	0.738	42.65	84.9	26.7
Large-crystalline cell	0.747	40.60	83.8	25.5
GaAs (thin film cell)	1.127	29.78	86.7	29.1
GaAs (multi-crystalline)	0.994	23.2	79.7	18.4
InP (crystalline cell)	0.939	31.15	82.6	24.2
GaInP	1.493	16.31	87.7	21.4
Thin film solar cells				
CIGS	0.992	35.70	77.6	22.9
CIGSS (Cd-free)	0.685	39.91	76.4	20.9
CdTe	0.887	31.69	78.5	22.1
CZTSS	0.513	35.21	69.8	12.6
CZTS	0.730	21.74	65.1	11.1
Amorphous/microcrystalline				
Amorphous silicon cell	0.896	16.36	69.8	10.2
Microcrystalline silicon cell	0.550	29.72	75.0	11.9
Other PV technology				
Dye-sensitized cell	0.744	22.47	71.2	11.9
Organic cell	0.896	25.72	78.9	18.2
Perovskite cell	1.179	25.80	84.6	25.7

flexible solar cells are anticipated to be marketed along with several other commercial goods. The efficiency of the current commercial solar cell modules based on crystalline and amorphous Si is 25.1%, but they are too expensive. On the other hand, the biggest obstacles in expanding the commercialization of this solar cell technology are the price of manufacture and the complexity of manufacturing devices. Thin film-based solar cells using cadmium telluride (CdTe/CdZnTe) and copper–indium–gallium–selenide/sulfide (CIGS/CIGSSe) have thus been extensively explored and commercially produced. These cells have demonstrated photovoltaic cell efficiencies and are less expensive and material-intensive.^{7–9} The efficiencies of various single junction-based solar cell technologies are compiled in Table 1.

The highest theoretical efficiency of 33% (for single junction solar cell) predicted by Shockley–Queisser (known as the S–Q limit) is significantly higher than the reported efficiencies of single-junction Si and thin film solar cells, which range between 20 and 25%. Thus, there is a pressing need for simple, inexpensive technologies to produce solar cells that can attain high efficiencies. With the rapid advancement in recent times, some new material schemes, *viz.*, organic semiconductors and metal halide-based perovskites^{10–15} and some emerging photovoltaics concepts such as water-based and integrated solar energy devices have to make a swift advancement,^{16,17} consequently, in the coming years, flexible solar cells are anticipated to be marketed along with several other commercial goods.

2. Metal oxide nanostructures

Metal oxides have such a wide range of physicochemical and optoelectronic features, and they have continued to be one of



the most alluring materials. In comparison to conventional semiconducting materials, metal oxides are advantageous because they have more degrees of freedom, which allows for the better modification of the optoelectronic and other properties. For instance, altering the oxygen-octahedral inclines in perovskite oxides can have a significant impact on the oxides' characteristics. The vast uses of transition-metal oxides (TMOs) in sensors, solar cells, fuel cells, and other devices have drawn broad attention among the various classes of oxides.¹⁸ TMOs are composed of oxygen atoms and transition metals and can exist in the forms of monoxide, dioxide, trioxide, *etc.*, due to the multivalent nature of transition metals. Fig. 1 demonstrates the molecular orbital (MO) diagram of 5d-orbitals, which play an essential part in several applications of TMOs.

One of the widely contested topics in the world of oxide electronics is the use of metal oxide nanostructures for various specific applications. Over the past three decades, scientists have used various chemical or physical evolution techniques with a wide selection of morphologies to produce metal oxide nanostructures (MONs). These nanostructures can be utilized in photovoltaics as a scaffold layer for hybrid perovskite solar cells and dye-sensitized solar cells (DSSCs) (HPSCs). In DSSCs and organic solar cells, these nanostructures can also be employed as the charge transport layers (electrons and holes). The transport layers are used to deport one type of charge carrier block to another type, while the function of the framework/scaffold layer is to separate and transfer the charges. Therefore, the changes to their photophysical characteristics are necessary to develop flexible solar cells with high efficiency.²⁰ Such properties highlight different metal oxides such as titanium dioxide (TiO₂), zinc oxide (ZnO), tungsten trioxide (WO₃), and copper oxides (Cu₂O, CuO) for their use in

flexible solar cells. These materials are the most prevalent and are also the most reliable in terms of profitability, stability, and efficacy. This article also highlights the photophysical and physicochemical characteristics of such metal oxides, as well as new developments, experiments, and modifications made to such metal oxides to overcome their limitations and optimize their performance in photovoltaic applications. Table 2 lists the uses of some interesting materials based on such metal oxides prepared in various nanostructures. Several MONs have been reported, such as nanoparticles, nanowires, nanotubes, nanosheets, nanowhiskers, and nanospheres. These materials in nano form might be used to different applications, *viz.*, catalysis, microelectronic circuits, transistors, energy storage and conversion devices, bio-medicines, and sensors.¹⁸ Titanium dioxide (TiO₂) nanostructures can be developed using a variety of processing methods such as sol-gel, hydrothermal, and co-precipitation methods,²¹ as shown in Fig. 2.

The cross-sectional view of the reasonably homogeneous TiO₂ nanostructure, with a thickness of 149 nm, is shown in the inset of the FESEM image (Fig. 3), which shows the spherical surface morphology of the SILAR (successive ionic layer adsorption and reaction)-processed TiO₂ nanostructure. The elemental surface composition analysis (EDX) verified that the amounts of Ti and O in TiO₂ were stoichiometric. The (101) crystal face of TiO₂ was more stable than the other facets due to its lower surface energy and HR-TEM images that also showed the strongest ring pattern for the (101) plane than others in the SAED spectrum. The diffraction bright and sharp ring pattern approves the nanocrystalline nature of the TiO₂ nanostructure. Based on these properties, a perovskite solar cell was fabricated using SILAR-grown TiO₂ nanostructures as the electron transport layer and efficiency of 9.7% was attained.²³ Besides the

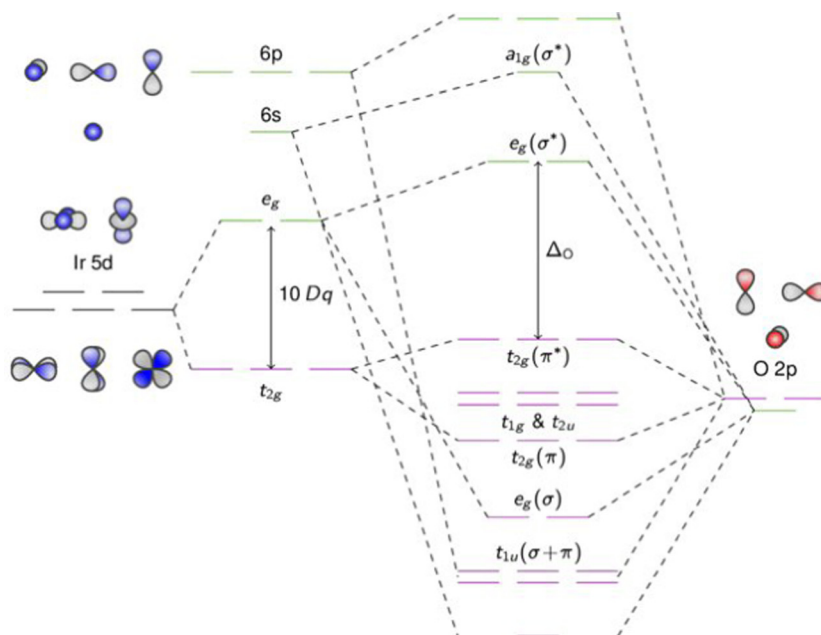


Fig. 1 Molecular orbital (MO) diagram of 5d-orbitals transition-metal oxides (TMOs) cation octahedrally coordinated to six oxygen ligands. The MOs are hybridized and have character arising both from the transition metal and ligand atomic orbitals. Adapted from ref. 19.



Table 2 Uses of some illustrative metal oxides-based materials with various structures. Adapted with permission from ref. 22, Copyright (2020), MDPI

Metal oxide-based nanostructures	Structural design	Preparation method	Usages
ZnO Nanosheets	3D hierarchical flower-like architectures	Solvothermal	Adsorption of triphenylmethane dyes
Fe ₃ O ₄ @UiO-66 composite	Cube-shaped nanoparticles	Sonication	Adsorption
Fe ₃ O ₄ @MIL-100(Fe) core-shell bio-nanocomposites	Core-shell structure having Fe ₃ O ₄ as core	Sonication followed by attaching the nanoparticles on bacteria	Adsorption
ZnO + TiO ₂ /clay	TiO ₂ and ZnO nanoparticles mounted on the clay surface	Sol-gel process	Degradation of MG
Cu/ZnO/Al ₂ O ₃	Cu and ZnO-impregnated γ -Al ₂ O ₃	Penetration method	Removal of CO from renewed coal
Co ²⁺ /Ni ²⁺ -Doped Fe ₃ O ₄ nanoparticles	Cubic lattice	Co-precipitation process	Photodegradation of Carbol Fuchsi
Ce/Fe bimetallic oxides (CFBO)	Flower-like 3D hierarchical architecture	No-pattern hydrothermal process	As ⁵⁺ and Cr ⁴⁺ correctness
Perovskites titanate (ATiO ₃ , A = Sr/Ca/Pb)	Leaf-engineered 3D hierarchical structure	Amalgamation of bio-synthesis from Cherry Blossom, heating, grinding, and photodeposition	Artificial photosynthetic systems for the photoreduction of CO ₂
TiO ₂ polypyrrole	Core-shell nanowires	Seed-assisted hydrothermal process	Elastic supercapacitors
Fe ₃ O ₄ /WO ₃	Hierarchical core-shell structures	Solvothermal growth + oxidation route	Photodegradation of organic-dye materials

**Fig. 2** A schematic representation of the chemical synthesis of MONs employing different processing methods, *i.e.*, (a) sol-gel, (b) hydrothermal, and (c) co-precipitation. Adapted from ref. 21. Copyright (2017), Royal Society of Chemistry.

morphology of metal oxide nanomaterials, doping of metal oxides, which favors the low-temperature process of carrier transport layers for flexible metal oxide-based PCs, can improve the device performances.^{24,25}

The rutile TiO₂ nanorods/films prepared by the hydrothermal method was also discussed, where these nanorods/films were integrated into the flexible PCs.²⁶ It was found that the efficiency was dependent on the nanorod length, while the current density and open-circuited voltage decreased with increasing nanorod length. The preparation of MONs with well-defined size and morphology is an imperative test in photovoltaic applications. The dimension of the MONs may be large, which could be useful for mesostructured-based flexible PCs. Though, for

planar flexible PCs, which have been very widely used in both rigid and flexible photovoltaic, the large nanostructures have big concerns in roughness-related issues. The metal oxide nanomaterials prefer the planar structure-based flexible PCs, while the process temperature needs to be low and the metal oxide layer needs to be high quality (well-pack, smooth, and pin-hole free) for such types of flexible PCs.^{27,28}

3. Metal oxide-based flexible solar cells

One of the most significant sources of renewable energy for sustainable development in the coming years are thin-film



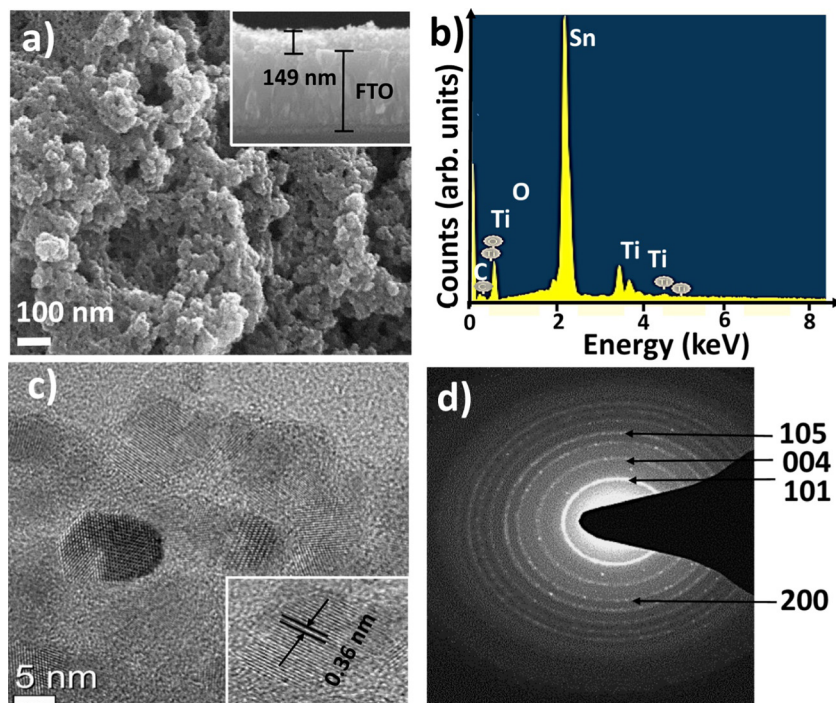


Fig. 3 (a) FESEM plan-view image, (b) EDX pattern, (c) HR-TEM image, and (d) SAED pattern of SILAR-grown TiO_2 film. Adapted with permission from ref. 23. Copyright (2018), Springer Nature Research.

solar cells, quantum-dot solar cells, dye-sensitized solar cells, organic solar cells, and perovskite solar cells. Because they are inexpensive, simple to prepare, and have adjustable optical characteristics, these solar cell devices have received widespread attention.^{6,29} In the last decade or so, the efficiency of perovskite solar cells has grown significantly, rising from 3.8% to 25.7%, and is equivalent to conventional silicon solar cells.⁷ The metal-oxide-based electron transport layers (*viz.*, TiO_2 , ZnO , SnO_2 , NiO , Al_2O_3 , V_2O_5 , ZrO_2 , CeO_2 , and Nb_2O_5) are used to achieve the high PCE. These devices have received widespread approval due to their affordable, high efficacy, and simple manufacturing processes.³⁰ Scientists have also been attracted by the metal-oxide heterojunction devices because of their high stability and low price. The metal oxides are intrinsic semiconducting materials and they have both types of conductivities, *i.e.*, n-type (*viz.* TiO_2 , Fe_2O_3 , and ZnO) and p-type (*viz.* CuO , Cu_2O , and Co_3O_4).^{31,32}

Metal-oxide semiconducting materials with small bandgaps can be employed as an absorber layer, while metal-oxide semiconducting materials with broad bandgaps are typically used as window layers. To prevent any short-circuits, the metal oxide-based transport layers in perovskite solar cells can function as compact layers.³³ Since TiO_2 and ZnO have a broad bandgap more than 3 eV and are transparent to visible light, it makes them ideal alternatives for transparent window layers. ZnO is a direct bandgap material that is frequently utilized in metal oxides-based flexible solar cells.³⁴ TiO_2 is often used to fabricate DSSCs and is primarily found as indirect bandgap semiconductors.³⁵ Among various p-type metal oxides, the copper oxides are the utmost widespread materials for solar cells.

Stable copper oxides (Cu_2O and CuO) are p-type semiconducting materials with a direct energy band gap of ~ 2.1 eV and 1.2 eV, respectively. These semiconducting materials are ideal candidates as an absorber layer for solar cells owing to their high absorption (in the visible area) and decent minority charge carriers diffusion length.³⁶ Cobalt oxide is another potential p-type semiconducting material having good visible-light absorption. Co_3O_4 has a spinel structure and mixed valence states, while CoO is a more stable state of cobalt oxide with a very high bandgap (~ 5 eV); therefore, Co_3O_4 is the phase most frequently used in photovoltaic applications. Numerous metal oxide-based flexible solar cells with different combinations of these metal oxides have been reported in the literature. The most common combination is $\text{ZnO-Cu}_2\text{O}$ -based heterojunction solar cells, but despite having a theoretical efficiency of 18%, these cells exhibit very low efficiencies. Irrespective of the methods used by the investigators, the increase in the efficiency is still quite small. Among many of these reported works, a few outstanding efforts are examined here.

The development of three different forms of $\text{ZnO-Cu}_2\text{O}$ -based flexible solar cells have been done by Wei and co-workers,³⁷ where they used ZnO as thin film, nanowire, and nanotube with a thin film of copper oxides. ZnO wires and tubes demonstrate somewhat greater efficiency as compared to thin films, of about 0.12%. ZnO thin films had relatively low efficiencies of 0.02%. The increased performance of the solar cells made of nanotubes and nanowires may be attributed to an expansion of the heterojunction region between Cu_2O and ZnO . Similar solar cells have been discussed utilizing $\text{ZnO-Cu}_2\text{O}$ heterojunction, where the nanorod and nanotube of ZnO with Cu_2O films have been used.³⁸ ZnO nanorods demonstrated a



photovoltaic efficiency of 0.4%, whereas ZnO nanotubes demonstrated a photovoltaic efficiency of 0.8%. Due to the greater surface area covered by ZnO nanotubes, the efficiency of the nanotubes was double that of nanorods.

Through this method, Zhang's team³⁹ developed flexible solar cells made of ZnO and Cu₂O thin films that were arranged in 3D square-patterned nanorod arrays. These cells showed a substantial improvement over cells without patterns, showing an efficiency of 1.52% and a current density of 9.89 mA cm⁻². By experimenting with different deposition techniques, several other researchers have attempted to improve the performance of Cu₂O-based heterojunction solar cell devices. Wee and colleagues⁴⁰ used pulsed-layer deposition to deposit Cu₂O thin films on the rough metallic substrate. They developed a flexible photovoltaic device, as shown in the schematic in Fig. 4a, with an epitaxial Cu₂O absorber layer, a thick SrRuO₃ conductive oxide layer (about 100 nm) between Cu₂O and SrTiO₃ layers, and then n-type transparent conductive oxide layers made of ZnO and Al-doped ZnO layers on top of the Cu₂O layer. This pure epitaxial phase of Cu₂O produced an efficiency of 1.65%. The flexible ZnO/Cu₂O solar cells exploiting the electrodeposition method was designed by Pławewski and team,⁴¹ and they reported the influence of Cu₂O thickness on the power conversion efficiency, which was found to be 2.7%.

By applying SILAR approach, Chatterjee *et al.*⁴² coated an additional NiO layer above the Cu₂O layer as the hole-transport and electron-block layer and SnO₂ on the ZnO side, as demonstrated in Fig. 5. The prepared heterojunction solar cells with the design of NiO/Cu₂O/ZnO/SnO₂ made it easier for electrons and holes to flow through them efficiently (owing to staircase-like energy levels), resulting in a photon conversion efficiency of 1.12%, an FF of 50%, and a short-circuited current density of 3.50 mA cm⁻². The high purity Cu₂O layers formed using thermal oxidation produced a flexible solar cell with a high efficiency of 3.83%, where high purity copper sheets were being oxidized to Cu₂O by heating at 1010 °C in a carefully controlled environment, whereas the ZnO layers were being doped with aluminium.⁴³ In their study, Kelly's group⁴⁴ used a bending test

to examine various electrodes on the PET substrates and discovered that metal oxide electrodes had evident cracks and a rapid increase in the sheet resistance, while the PEDOT:PSS electrodes had barely changed.

The above discussion make it clear that the purity of copper oxide plays a crucial role in increasing the effectiveness of solar cell devices, yet producing such high purity Cu₂O makes the production of solar cells difficult and expensive. When compared to Cu₂O, there are very few publications on the photophysical characteristics of CuO and the efficiency achieved was less than 2% due to a small carrier concentration, but it has a great bandgap for developing flexible solar cells. CuO-based flexible solar cells can also be made using n-type silicon, which showed an efficiency of 0.36%,⁴⁵ but electroplated TiO₂-CuO-based solar cells had an efficiency of about 1.6%.⁴⁶ Cu₂O and CuO are p-type materials with appropriate work functions and low-lying valence bands that can complement the perovskite layer well and reduce energy loss. Thus, researchers have spent a lot of time recently studying the flexible solar cells, which are crucial to the creation of third-generation solar cells because metal oxides are one of the most critical constituents of third-generation solar cells. Metal oxide electron collector layers are often in charge of transporting electrons in third-generation solar cells. The consistency of components, such as sensitizers in dye-sensitized solar cells, organic dye in organic solar cells, organic-inorganic perovskite in PSCs, quantum dots in quantum dot-based solar cells, and electrodes in CIGS solar cells, stimulated by the Sun and flexible transparent substrates, is another key role of metal oxide layers.

4. Metal oxides-derived flexible perovskite solar cells

Because metal oxide nanostructures are thought to be correlated with promising functional materials, there is a lot of current research being done on them. Metal oxide

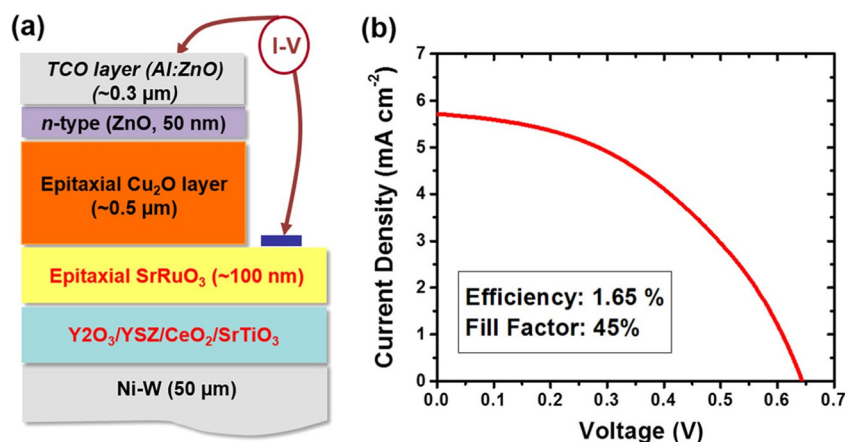


Fig. 4 (a) Device design and (b) current–voltage characteristics of a flexible solar cell, which has an epitaxial Cu₂O absorber-layer on the textured metallic substrate. Adapted from ref. 41, Copyright (2015), Springer Nature Research.



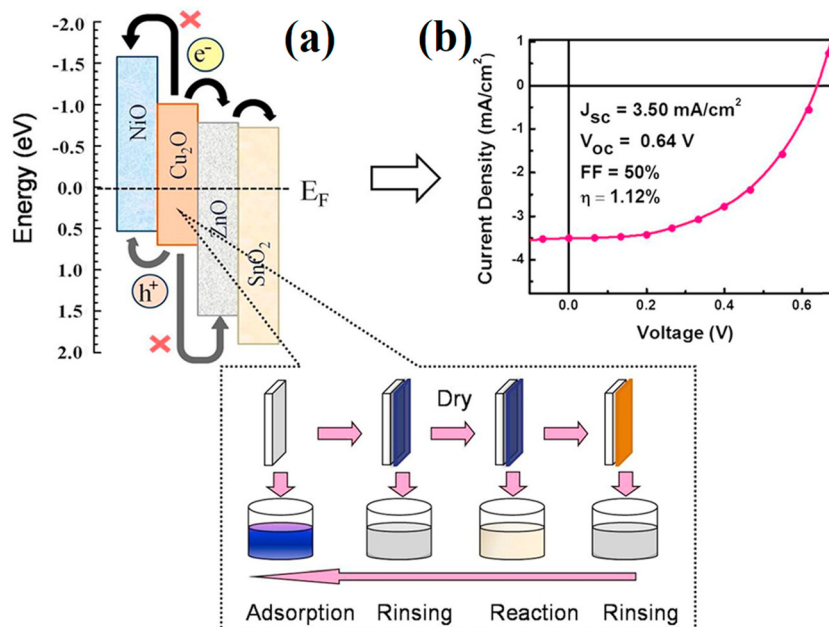


Fig. 5 The schematic energy level diagram and current–voltage characteristics of the NiO/Cu₂O/ZnO/SnO₂ heterojunction flexible solar cell. Adapted with permission from ref. 42, Copyright (2016), Elsevier.

nanostuctures (MONs) are good contenders as flexible transparent electrodes for perovskite solar cells because of their higher transmittance, lower sheet resistance, and superior bending property equated to those of the conventional transparent electrodes. The porous scaffolding layer underlying the perovskite active layer may assist in crystallizing the technology or expand the surface-active-area, while the metal oxide-based nanostructured layers serve as compact layers in PSCs to prevent any short-circuit problems.³³ The sandwiched construction of typical PSCs normally consists of two electrodes and a photoactive layer, as shown in Fig. 6. In recent times, metal oxides have been widely studied as electron transport layers for high-efficiency PSCs owing to their excellent electronic properties, superb versatility, and great stability. The three common structures of PSCs are: (a) semiconductor mesoporous-

designed PSCs, in which a semiconducting material is somewhat solution-administered as a porous scaffolding layer to carry a mesosuper-structured solar cell, (b) thin-film based PSCs, for which no pores is required, while a perovskite active layer is compressed between p- and n-type charge extracting electrodes, and (c) PSCs based on p-n heterojunction, where the perovskite active layer is porous.⁴⁷

In 1958, the Danish scientist Christian Miller⁴⁸ revealed the first semiconducting properties of perovskites along with the first crystallographic studies of cesium–lead–halides ((CsPbX₃, where X = halides, *i.e.*, Cl, Br, or I)).⁴⁸ After 20 years, Dieter Weber⁴⁹ developed novel organic–inorganic hybrid three-dimensional perovskites in 1978, replacing cesium with methyl-ammonium cations. However, Zhai and colleagues were the first to show that organometallic halide perovskites have

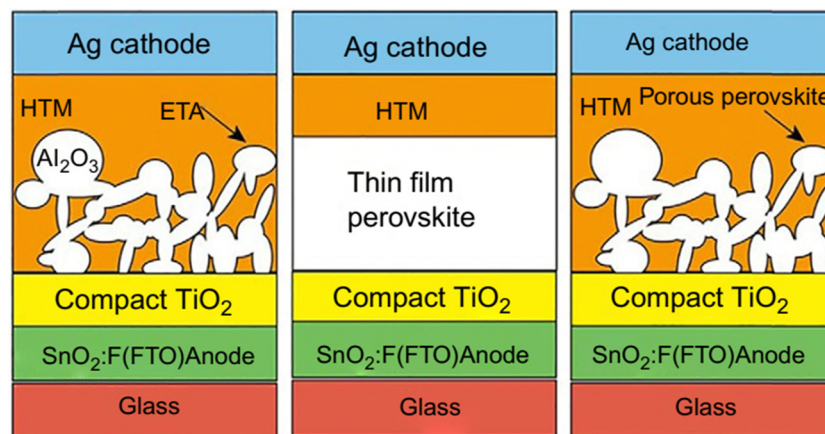


Fig. 6 Typical configuration structures of a PSC. Adapted with permission from ref. 47, Copyright (2018), Elsevier.



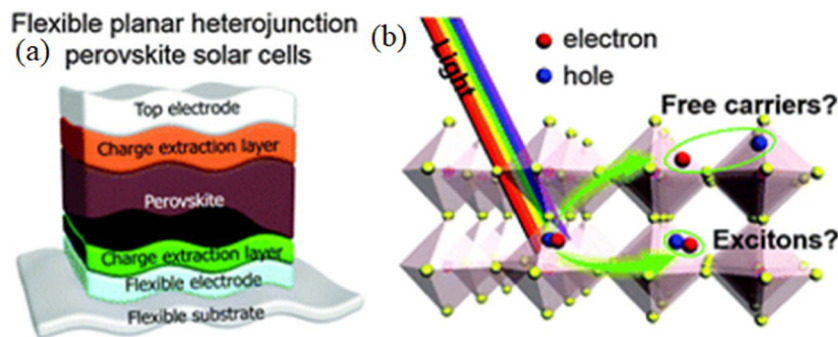


Fig. 7 (a) Planar heterostructure of flexible perovskite solar cell and (b) charge carrier extraction after the photoexcitation of the perovskite. Modified and adapted from ref. 57, Copyright (2016), Royal Society of Chemistry (RSC).

good performance.⁵⁰ In the organic–inorganic hybrid PSCs, the morphological properties are strongly influenced by the various crystalline phases such as tetragonal, orthorhombic, and cubic, which vary with the operating solvent and temperature. However, the modification in the morphology can affect the electrical characteristics.^{51–53} Acting as a photoactive layer in flexible solar cells, the perovskite demonstrates numerous extraordinary aids as its tunable direct band gap can be altered between 1.24 eV and 3.55 eV by exploiting mixed ions for single-junction flexible solar cell devices. The perovskites having a slender bandgap (<1.65 eV) are suitable for single junction-based solar cell devices, while perovskites having a wide bandgap (1.70–1.80 eV) have been demonstrated to be appropriate for multi-function devices as a top sub-cell.⁵⁴ At present, all efficient PSCs embrace an active absorber layer of intrinsic perovskites material (i), which is between the positive and negative charge-extracting layers (as known as p- and n-type interfaces); these layers are also labeled as ETL and HTL, respectively. The halide perovskite solar cells are typically categorized into three device architectures: mesoporous n–i–p, planar n–i–p, and planar p–i–n solar cell devices.⁵⁵ The foremost effort was made in 2013 to apply perovskites as an absorber layer in the flexible PSCs, where the primary efficiency was about 2.6%.⁵⁶ However, the poor long-term stability and relatively low performance are imperative aspects preventing the development of flexible PSCs. Compared to rigid PSC fabricated on a glass substrate, each layer that makes up the complete flexible PSC device has some special requirements, including the flexible substrate, transparent flexible electrodes, electron transport layer (ETL), perovskite layer (PL), hole transport layer (HTL), and top electrode (E), as demonstrated in Fig. 7.⁵⁷ A sharp upsurge in the efficiency of PSCs is attained in the last few years, which is discussed in the next section.

5. Recent development in metal oxides-based flexible perovskite solar cells

In recent times, metal oxides have been widely studied as electron transport layers for high-efficiency PSCs owing to their excellent electronic properties, superb versatility, and great stability.

Photovoltaic devices based on perovskites are undoubtedly the most promising competitors for next-generation high-efficiency solar cell technologies that are compatible with inexpensive, flexible substrates, and large-area construction developments. High-efficient PSCs are made of perovskites that have a 3D crystalline structure (in the form of ABX_3) and are generally formed by an organic/inorganic mono-valent cation, ($A = \text{CH}_3\text{NH}_3^+$, $\text{CH}(\text{NH}_2)_2^+$, Cs^+ , Rb^+), a bivalent metal-cation, ($B = \text{Pb}^{2+}$; Sn^{2+}), and halide anion motif ($X_3 = \text{I}^-/\text{Br}^-/\text{Cl}^-$). The scientists have become interested in perovskites owing to their easy fabrication methods based on solution precipitation or evaporation, such as a one-step deposition, two step successive deposition, and vapor-induced solution processing. The quality of the perovskite films is determined by the process of growth/deposition. The organic-inorganic hybrid perovskites (OIHPs), which have shown photoconversion efficiencies (PCEs) comparable to those of Si, CdTe, and CIGS solar cells, have advanced significantly over the past decade or so, replacing the methyl-ammonium-lead-halide (MAPbX_3)-based perovskites as light harvesting materials.^{7,53} The most frequently utilized ETL in the systematic development of a PSC is TiO_2 , which is typically produced by sintering a precursor solution at a temperature between 450 °C and 500 °C. The aforementioned practice not merely raises the engineering budgets but also limits the applications of flexible plastic substrates as well. As a result, several alternatives as inorganic ETLs have been developed, including ZnO , SnO_2 , Fe_2O_3 , In_2S_3 , and WO_x . Few triplex metal oxides, *viz.*, Zn_2SnO_4 , BaSnO_3 , BaTiO_3 , SrTiO_3 , and Nb:SnO_2 , have also been investigated as ETLs in perovskite-based solar cell systems.⁵⁸ There are two different types of stabilities that are linked to flexible PSCs. One is the mechanical stability, which is frequently constrained by the electrodes' mechanical robustness. The interfacial layers and perovskite absorbers, which govern this, also contribute to long-term environmental stability. According to Miyasaka and colleagues,⁵⁹ who developed the novel photovoltaic cell devices on a DSSCs outline with the thin perovskite layers of $\text{CH}_3\text{NH}_3\text{PbI}_3$ (MAPbI_3) and $\text{CH}_3\text{NH}_3\text{PbBr}_3$ (MAPbBr_3) on the mesoporous TiO_2 , the organic–inorganic perovskites were first used as a sensitizer in liquid-state DSSCs in 2009, where the devices demonstrated efficiencies within the range of 3.1–3.81%. However, because of the liquid electrolyte's scarring tendencies, the consistency of such devices remained below par. A similar perovskite-sensitized



photovoltaic cell with an enhanced PCE of 6.5% was developed in 2011 by Park and his colleagues⁶⁰ using perovskite quantum dots as sensitizers. Later, in 2012, his team designed solid-state mesoscopic PSCs employing solid-state spiro-MeOTAD as a hole transport material in order to improve the cell stability.⁶¹ This was done by looking into the corrosion processes connected to liquid electrolytes. They used a one-stage/step process to make a perovskite solar cell with a solid-state electrolyte using a solution-based process, and they were able to achieve an efficiency of 9.7%. In addition, they found that the hole-transporting material layer could dissolve in organic solvents (chlorobenzene and toluene), whereas the hybrid perovskite cannot, and that hybrid perovskite was easily deposited on top of this layer. The mesoporous alumina scaffold layer (Al_2O_3) was used by Snaith's group⁶² to develop the meso-superstructure solar cells and achieved a power conversion efficiency of 10.9% for the first time. The large bandgap of alumina prevents it from assisting in electron extraction, which suggests that perovskites can transport electrons effectively without the use of a nanoporous layer (the purpose of the nanoporous layer is to collect electrons that cannot be elevated to thin films by creating additional interfaces between metal oxides and perovskites).⁶³ The PSCs produced using sequential deposition techniques showed a PCE of 15% with excellent repeatability.⁶⁴ Later, Snaith's team developed planar heterojunction PSC for the further development in PCE up to 15.4%⁶⁵ using the vapor deposition method. Subsequently, Seok's team⁶⁶ removed the combined planar and scaffold layer of mixed halide $\text{CH}_3\text{NH}_3\text{Pb}(\text{I}_{1-x}\text{Br}_x)_3$, which raised the efficiency upto 17.9%. In this case, the chemical compositions and layer thickness were both adjusted.

Choy and co-workers have reported several low-temperature processed metal oxides as hole^{67,68} and electron^{69,70} transport layers for high efficiency and stable perovskite solar cells. They demonstrated a controllable approach to form surface-nanostructured and flawless NiO_x film as hole transport layers, while TiO_2 and In_2O_3 -based films as the electron transport layer through a room-temperature solution-processing technique for high-performance flexible perovskite solar cells. There are some reports available in the literature related to the device design and architecture flexibility of perovskites and flexible perovskite solar cells,^{71–79} but any report on the recent advancement in metal oxide-based flexible perovskite solar cells is not available yet; consequently, herein, we summarized the work based on efficient metal oxides flexible perovskite solar cells.

The foremost effort was made in 2013 to apply perovskites as an absorber layer in the flexible PSCs, where the primary efficiency was about 2.6%.⁵⁶ However, the poor long-term stability and relatively low performance are imperative aspects preventing the development of flexible PSCs. As compared to the rigid perovskite solar cell developed on a glass substrate, each layer that makes up the complete flexible PSC device has some special requirements, as stated earlier in Section 4. As stated previously, a rapid increase in the efficiency for flexible perovskite solar cells has been attained in the last decade or so owing to outstanding photophysical and opto-electronic properties of the perovskites and to experiences obtained from OSCs and DSSCs. The flexible perovskite solar cell having a

triple-cation configuration demonstrated an efficiency of 18.6%, when stable metal oxides were combined within an inverted device structure, while the efficiency was extremely stabilized at 17.7%.⁸⁰ The flexible perovskite solar cell achieved a validated efficiency of 17.3% over the highly crystalline oxide-based transporting layer, whereas the flexible all-perovskites tandem photovoltaic systems showed an efficiency of 21.3%.^{81,82} The flexible PSCs fabricated on the PFN substrates showed a thousand times bending sturdiness and greater than 86% of the alleviated power output is reserved after aging of more than 1000 hours.⁸³ The device made on a polyethylene terephthalate-based substrate demonstrated an excellent sturdiness and retained about 95% of its primary/initial efficacy after 5000 times of cyclic bending.⁸⁴

The efficiencies of metal oxide-based PSCs has significantly increased to 25.7% (Fig. 8) at the laboratory scale in 2022 due to the several developments in this field, including altering perovskite characteristics through interface engineering, chemical composition, developing efficient growth methods, and hole transport materials optimization. The promising photophysical and optoelectrical characteristics of perovskite materials such as their high absorption coefficient, increased electron-hole mobility, great life span, and long diffusion-span of the transporting carriers largely contribute to their high efficiency.⁸⁵ In addition, the organic-inorganic halide perovskite materials can produce weakly bonded excitons with low bonding energies, encouraging free-moving carriers with weak electromotive forces within the perovskite material.⁸⁶ Therefore, by combining the designs and advancements of the perovskite materials, the rapid augmentation can actually be close to the potential efficacy limit.^{87–90}

The durability of PSCs can be controlled by a number of factors, including the unpredictability of halide-perovskites, interfaces, and charge-transporting layers (ETLs and HTLs). White LEDs or solar simulators with UV light filters are always used to test the stability of PSCs. The development of amalgamated or double-layered ETLs is considered as a key technique



Fig. 8 The efficiency progress with time for perovskite solar cells based on different metal oxide nanostructures.



improving the quality of the metal-contact layers, and designing a tandem solar cell device structure combining the PSC and the Si/CIGS/CZTSSe. These novel strategies can further increase the efficacy to 29.6%.

The mass manufacture of highly effective solar cells at a somewhat low temperature would be made possible by flexible perovskite-based technologies, which would significantly lower the cost of production. Due to the demands of the moderate growth processes, this photovoltaic technology may now be used to build high-throughput devices. It is anticipated that these new-generation PSCs will find wide applications and eventually result in photovoltaic products that compete with conventional Si-based solar cells. Thus, more effort is necessary to develop flexible electrodes with good electrical conductivity and light transmission for effective flexible PSCs and to bridge the gap between flexible and rigid/standard devices. Also, almost all reported flexible PSCs have small active areas of the concerned devices. It is generally known that the efficiency decreases when the device area grows to a large scale because the films inevitably lose their uniformity. Therefore, the performance of the large-area flexible PSCs is directly dependent on the deposition procedure for the thin films at a large scale. As a result, procedures for fabricating large surface areas are required for all of the layers in flexible PSCs. In order to further reduce the cost of manufacturing, the alternating vacuum deposition method should be included into the roll-to-roll system, which should presumably encourage the development of practical applications. In flexible perovskite solar cells, good encapsulation is more feasible and can give decent efficiency while reducing lead leakage in addition to the pursuit of lead-free devices. Further advancement and encouraging research on flexible PSCs are currently underway,¹⁰² and it is supposed that highly effective metal oxides will make flexible perovskite solar cells more commercially sustainable.

Conflicts of interest

There are no conflicts to declare.

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