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Impact of p-type doping on charge transport in blade-coated small-molecule:polymer blend transistors†

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Blade-coating is a roll-to-roll (R2R) compatible processing technique and has the potential to address the industry's needs for scalable manufacturing of future organic electronics. Here we investigate the applicability of blade-coating for the fabrication of organic thin-film transistors (OTFTs) based on best-in-class organic semiconducting blends comprised of the conjugated small-molecule 2,7-diocetyl[1]benzothieno[3,2-b][1]benzothiophene (C₈-BTBT), and the conjugated polymer poly(indacenodithiophene-co-benzothiadiazole) (C₁₆lDT-BT). We show that the operating characteristics of blade-coated transistors consistently outperform devices prepared via spin-coating, showcasing the compatibility of the technique. Introducing the molecular p-dopant C₆₀F₄₈ into the binary C₈-BTBT:C₁₆lDT-BT blend formulation, in combination with carefully optimized blade-coating conditions, helps to enhance the performance of the ensuing transistors further resulting in a maximum hole mobility of $\approx 14 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, and an all-around improvement of the device operating characteristics. Our results show that p-doped blend OTFTs can be manufactured using industry relevant processing techniques without sacrificing their state-of-the-art performance.

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1. Introduction

The vast majority of modern day electronics rely on inorganic materials that require complex and costly manufacturing processes.^{1,2} Emerging semiconductor technologies such as organic semiconductors (OSCs) promise to provide an alternative option for implementation in various future optoelectronics including flexible, foldable, wearable, stretchable and even disposable devices for application in health, security, information, display, sensing and energy generation, among many others.^{3–8} This has resulted in continuous exploration of OSCs as active layers in printable organic photovoltaics (OPVs),

organic light emitting diodes (OLEDs), thermoelectrics, and organic thin-film transistors (OTFTs).^{3,6,9–12}

Among the various demonstrated technologies, OTFTs are highly promising as the building blocks for large-area electronics and more broadly for the internet of things (IoT) device ecosystem.^{13,14} During the past three decades, continuous progress on materials development, interfaces and device engineering have resulted in field-effect mobility (μ), a key figure of merit, of $> 20 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$.^{5,15–18} Laboratory-based research often relies on the widely available film deposition method of spin-coating. Unfortunately, the latter is wasteful (90–95% material waste) and incompatible with continuous, high-throughput roll-to-roll (R2R) printing techniques relevant to manufacturing of the envisaged electronics of the future. To this end, meniscus guided R2R compatible techniques such as blade, knife, bar and slot-die coating, are more relevant to the industry offering 95–99% utilization of the OSC ink.^{5,9,10,19–23} The drying kinetics during solution-phase deposition is known to strongly influence the morphology, crystallinity and microstructure of the resulting layers, ultimately dictating the charge transport properties of the formed OSC films and their devices.^{5,23–28} Though meniscus guided techniques have been used in the past to deposit OSC films with aim to understand the structure–property–performance relationship for OTFTs,^{27,29–34} the majority of these studies relied on low to medium coating speeds.²⁷

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In recent years, molecular doping of OSCs has emerged as promising method to enhance the performance of solution-processed OTFTs.^{18,35–48} In this regard, we have recently developed p-doped OTFTs with hole mobility of $13 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ based on spin-coated blend semiconducting channels based on the small-molecule 2,7-diethyl[1]benzothieno [3,2-*b*] [1]benzothiophene ($\text{C}_8\text{-BTBT}$), and the polymer poly(indacenodithiophene-*co*-benzothiadiazole) ($\text{C}_{16}\text{IDT-BT}$).¹⁶ Key to our success has been the ink-formulation engineering (solvents, material ratios, *etc.*), in combination with careful choice of drying kinetics (spin-coating and melt-processing conditions). Using this work as our starting point, we attempt to bridge the existing gap between lab-scale and industry relevant manufacturing of high-performance blend OSCs and their use in OTFTs. We first studied the role of coating method (spin *vs.* blade coating) on the morphology and charge transport of the blend OSCs and their OTFTs. We find that blade-coated devices consistently outperform their spin-coated counterparts, an observation attributed to the superior morphology of blade-coated layers due to different film formation conditions. By introducing the molecular p-dopant $\text{C}_{60}\text{F}_{48}$ into the $\text{C}_8\text{-BTBT}:\text{C}_{16}\text{IDT-BT}$ blend and blade-coating it at industry relevant speeds of $>1 \text{ m min}^{-1}$, we are able to improve the mobility of the OTFTs further reaching a maximum value of $14 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$. The work highlights the compatibility of high performance blend OSCs with the industry-relevant blade-coating technique while providing important guideline for the processing of high quality blend OSC and the realization of state-of-the-art OTFTs.

2. Result and discussion

The molecular structures and corresponding energy levels of $\text{C}_8\text{-BTBT}$, $\text{C}_{16}\text{IDT-BT}$ and the p-dopant ($\text{C}_{60}\text{F}_{48}$) are shown in

Fig. 1a and b, respectively, while Fig. 1c depicts the blade-coating process. Further details on the blade-coating process and the relevant parameters are given in Fig. S1 in the ESI.† Blade-coating involves two main coating regimes: (1) the evaporation regime, and (2) the Landau–Levich (L–L) regime, which depend primarily on the coating speed.^{25,49,50} For the purpose of this work we optimized and set the coating speed at 50 mm s^{-1} (3 m min^{-1}) while maintaining the stage temperature at 70°C . The latter falls within the L–L coating regime ($>20 \text{ mm s}^{-1}$) and it is representative of industrial roll-to-roll solution printing requirements.²³ Unlike spin-coating, blade-coating is a meniscus guided coating technique which can help to control the nucleation, crystal growth and orientation of the small molecule (SM) films over a large coating area.⁵¹ Blade-coating and spin-coating of all blend layers investigated was performed entirely under inert nitrogen atmosphere in order to eliminate extrinsic effects such as unintentional doping, variable humidity conditions *etc.*

We employed polarized optical microscopy (POM), atomic force microscopy (AFM), cross sectional transmission electron microscopy (TEM), and grazing incidence wide angle scattering (GIWAXS) techniques to assess the morphology, vertical phase separation and microstructure of the thin films coated by two coating methods. POM micrographs of blade and spin-cast pristine blend films (Fig. 2a and b) reveal that blade-cast films are characterized by small nucleation density and larger polycrystalline domains. The differences are attributed to the different crystallization/phase-transformation mechanisms involved in the two coating techniques.²³

AFM analysis of the layers' surfaces reveal the presence of molecular terraces of $\text{C}_8\text{-BTBT}$ on both spin and blade cast blend films, in agreement with previous reports.^{16,17} The height



Fig. 1 (a) Chemical structures of $\text{C}_8\text{-BTBT}$, $\text{C}_{16}\text{IDT-BT}$ and $\text{C}_{60}\text{F}_{48}$ used in this study. (b) Highest occupied molecular orbital (HOMO) energy levels of $\text{C}_8\text{-BTBT}$ and $\text{C}_{16}\text{IDT-BT}$, and lowest unoccupied molecular orbital (LUMO) energy level of $\text{C}_{60}\text{F}_{48}$. (c) Schematic depicting the blade-coating process and the various parameters used.





Fig. 2 (a and b) Polarized optical microscopy and atomic force microscopy images of spin-coated and blade-coated C₈-BTBT:C₁₆IDT-BT blend layers. (c) Surface height distribution profiles extracted from the AFM images of the spin-coated and blade-coated blend films. (d) X-TEM analysis of the C₈-BTBT:C₁₆IDT-BT blend film deposited on glass. The image on the left is the HAADF-STEM image while the one on the right shows an RGB Composite of S, C, and O maps generated by processing the STEM-EELS data-cubes.

histograms (Fig. 2c) extracted from $20 \times 20 \mu\text{m}^2$ AFM scans, reveal that the majority of the molecular plateaus in blade-coated films have a height in the range 4–5.40 nm (peak at ≈ 4.7 nm) and a smaller fraction of plateaus with heights in the range 1.40–2.70 nm (peak at ≈ 1.9 nm). Similarly, the height distribution for the spin-coated films also exhibit two main peaks, one between 2.40–3.64 nm (peak at ≈ 3.1 nm) and a second one between 4.8–6.3 nm (peak at ≈ 5.75 nm). These differences manifest in a lower root mean square (RMS) surface roughness of 1.46 nm for the blade-coated layers as compared to 1.74 nm for the spin-coated films. The blade-cast films exhibit larger polycrystalline domains, in line with POM images, suggesting a higher film quality than the spin-cast layers. The characteristic height distributions seen in both films is most likely the result of the presence of domains (crystallites) composed of different number of monomolecular layers of the C₈-BTBT lamella, while the broadening of these peaks is most likely the result of the non-uniform nature of the vertical phase-separated system. To this end, a monomolecular terrace step height of ≈ 2.8 nm should be expected for the *c*-axis of C₈-BTBT, which should dominate in films with an out-of-plane molecular packing motif. The measured surface heights are those in broad agreement with the presence of large monomolecular domains.

Lastly, we investigated the vertical phase separation in the blend layers using transmission electron microscopy (TEM) in conjunction electron energy loss spectroscopy (EELS) mode. The vertical phase separation in small-molecule:polymer blend

systems is believed to proceed *via* the nucleation and growth of the small-molecule component.^{16,17,52} In particular, the as-deposited wet layer first undergoes a liquid-liquid phase separation (driven by the surface energies of the different blend components) before the small-molecule (*i.e.* C₈-BTBT here) starts to crystallize. This crystallization even becomes particularly pronounced at the surface/air interface of the supersaturated solution, leading to the formation of the small-molecule dominated layer surface. The remaining part of the layer eventually dries up to form the polymer-dominated phase at the bottom interface (substrate/film) of the composite film. Indeed, the Sulphur map of the blend films shown in Fig. 2d, confirms the AFM observations that the surface of the blend layer is dominated by the presence of C₈-BTBT molecules that form a stratified layer approximately ≈ 7 nm in thickness.

To understand the impact of morphology on charge transport properties, we have fabricated top-gate bottom-contact (TG-BC) OTFTs based on spin-coated blade-coated channel layers of undoped C₈-BTBT:C₁₆IDT-BT blends using identical ink-formulation protocols to those reported previously.^{16,17,53} The ensuing devices were electrically characterized under nitrogen atmosphere at room temperature in order to avoid extrinsic effects such as unintentional doping or degradation due to the presence of oxygen, or other atmospheric oxidants *etc.* The representative transfer and output curves, measured in linear and saturation biasing regimes, are shown in Fig. 3a and Fig. S2 (ESI[†]). The average saturation mobility (μ_{SAT}) for spin-cast



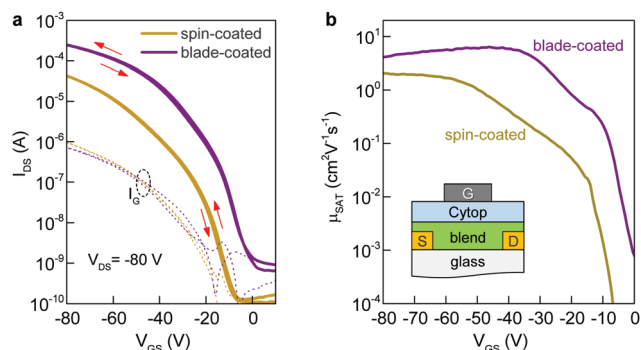


Fig. 3 (a) Transfer characteristics of spin-coated and blade-coated C₈-BTBT:C₁₆IDT-BT OTFTs measured at $V_{DS} = -80$ V. Dotted lines represent the corresponding gate leakage (I_G) currents. The channel width and length of the devices were 1000 μm and 80 μm , respectively. (b) Hole mobility measured in saturation (μ_{SAT}) versus applied gate bias (V_{GS}) for the C₈-BTBT:C₁₆IDT-BT-based OTFTs shown in (a).

devices was found to be $2.4 (\pm 0.7) \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, in excellent agreement to our earlier published reports.^{16,17} Details of the method used to calculate μ_{SAT} are provided in the Experimental section of the manuscript. On the other hand, TGBC devices based on blade-cast films yielded a significantly higher μ_{SAT} value of $6 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$. Fig. S3 (ESI†) shows the statistical comparison of the device performance of spin-coated and blade-coated OTFTs. The measured current-voltage (I - V) characteristics are hysteresis free which is indicative of a high quality semiconductor/dielectric interface.

We also assessed the reliability of the calculated μ_{SAT} values by examining the mobility dependence on V_{GS} (Fig. 3b) at $V_{DS} = -80$ V for both spin- and blade-coated devices. OTFTs prepared via spin-coating exhibit a plateau in μ_{SAT} at high V_{GS} range while for blade-coated transistors the mobility peaks ($6.4 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$)

at lower V_{GS} followed by a reduction ($4.1 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$) at higher voltages. This is a known phenomenon in OTFTs and has been previously associated with the presence of contact resistance that is also affected by the applied gate-field.⁵⁴ To avoid over-estimation, we thus use the mobility values calculated at $V_{GS} = -80$ V as the representative parameter for both spin-cast and blade-coated transistors. The improved μ_{SAT} for blade-coated devices manifests as larger channel on-currents (a factor of six) and lowered V_{TH} as compared to spin-coated transistors (Table S1, ESI†). The noticeable differences in the charge transport between spin-coated and blade-coated OTFTs are almost certainly attributed to morphology and microstructure differences of the channel layer.^{55–57} Importantly, both types of OTFTs show good shelf lifetime with their initial operating parameters being retained for up to six months of storage in a nitrogen atmosphere.

2.1. Effect of molecular p-type doping

We have previously shown that addition of C₆₀F₄₈ in spin-coated C₈-BTBT:C₁₆IDT-BT blend channels increases the hole mobility of the OTFTs to $13 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$.¹⁶ Motivated by this early work, we deposited pristine and C₆₀F₄₈-doped C₈-BTBT:C₁₆IDT-BT layers via blade-coating. Kelvin probe (KP) measurements performed on both types of blend layers (Fig. S4, ESI†) reveal a significant shift in work function (WF) from 4.8 eV, for the pristine blend, to 5.3 eV upon addition of 4.5 mol% C₆₀F₄₈. These results provide direct evidence of successful p-doping of the C₈-BTBT:C₁₆IDT-BT blend by C₆₀F₄₈.

Next we studied the morphology and microstructure of the blade-coated layers using POM, AFM and GIWAXS. The POM and AFM images shown in Fig. 4a and b reveal the presence of large crystalline domains and molecular terraces due to the vertical phase separation of C₈-BTBT at the surface of the blend

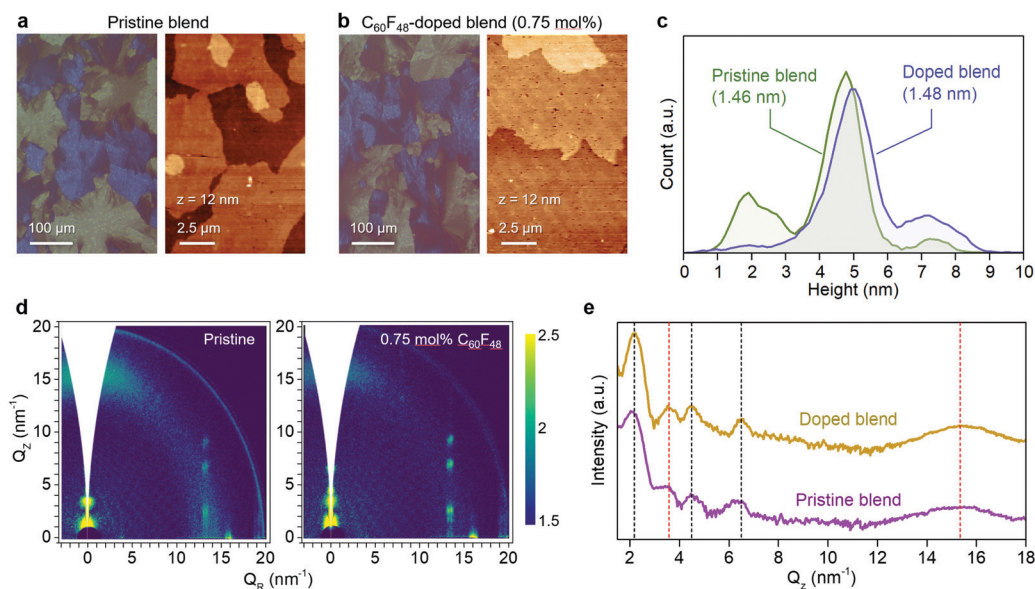


Fig. 4 (a and b) Polarized optical microscopy and atomic force microscopy images of pristine and p-doped C₈-BTBT:C₁₆IDT-BT layers processed via blade-coating. (c) Surface height histograms of the pristine and p-doped C₈-BTBT:C₁₆IDT-BT layers extracted from the AFM images. (d) 2D GIWAXS images of the pristine and p-doped blend layers. (e) Plots of the integrated intensity versus Q (nm^{-1}) for out-of-plane Bragg sheets taken from (d).



film. C₆₀F₄₈-doped film exhibit larger C₈-BTBT crystalline domains/terraces than the pristine layer. The height distribution profile shown in Fig. 4c supports this conclusion since the surface exhibits a single dominant peak at ≈ 5.5 nm, which equates to approximately two monomolecular layers of C₈-BTBT. The pristine layer on the other hand shows a bimodal distribution of terrace heights, one at 1.9 nm (single monomolecular terrace) and a stronger one at 4.8 nm (double monomolecular terrace), supporting the existence of smaller area terraces/crystals with different heights. As a result of these features the RMS surface roughness of pristine and C₆₀F₄₈-doped blend layers are similar and equal to 1.46 and 1.48 nm, respectively. Fig. S5a (ESI†) shows the AFM images of the blend films doped with several different C₆₀F₄₈ concentrations (*i.e.* 0.1%, 2% and 4.25%), while Fig. S5b–d (ESI†) show the evolution of the surface height distribution and surface roughness RMS for blend layers with different dopant concentrations, respectively. AFM images with $20 \times 20 \mu\text{m}^2$ scan area were used to extract the height distribution and RMS roughness data. Evidently, the presence of C₆₀F₄₈, at optimal dopant concentration, enables the formation of blend films with more uniform molecular traces (more compact films) and lower RMS surface roughness.

Further insights into the microstructure of the blend layers and the role of C₆₀F₄₈ were obtained *via* two-dimensional GIWAXS (2D GIWAXS) measurements. Fig. 4d shows the 2D GIWAXS images taken for a pristine and a C₆₀F₄₈-doped (0.75 mol%) blade-coated C₈-BTBT:C₁₆IDT-BT blend films. The crystal lattice of the doped layer appears similar to that of the pristine, in agreement with previous reports on the effect of doping on spin-coated films.⁵⁸ In particular, the 2D GIWAXS patterns show a diffraction in the out-of-plane direction (Fig. 4e) at $Q_z = 3.6 \text{ nm}^{-1}$ where the broad feature at 15 nm^{-1} is associated to C₁₆IDT-BT (red dotted line), while the other highly oriented and sharp diffractions are related to C₈-BTBT (black dotted line). The integrated intensity *vs.* Q_z (nm^{-1}) plots for out-of-plane Bragg reflections revealed only minor differences between the doped and pristine blend films, and are most likely attributed to small differences in the films thicknesses. Overall, the small molecule exhibits a high degree of lamellar ordering, as evident by the well resolved (00 l) diffractions, with higher-order peaks ($L = 1, 2, 3$) in the out-of-plane direction. A similar high intensity of C₈-BTBT related diffractions is also observed along the (11 L) diffractions centered at around 13.4 nm^{-1} (Fig. S6a, ESI†). Addition of C₆₀F₄₈ at two different concentrations of 0.1 and 2 mol% leads to similar observations *i.e.* a well resolved progression of higher-order diffraction peaks corresponding to the (00 l) crystal plane (Fig. S6b, ESI†). Furthermore, the FWHM of the diffraction peak for the doped film appears to be smaller than the pristine film (Fig. S6c, ESI†). The latter feature indicates improved ordering and the existence of larger C₈-BTBT crystallites upon optimal doping, in excellent agreement with the AFM analysis (Fig. 4a and b). Overall, these results suggest that addition of small concentrations of C₆₀F₄₈ improves the degree of microstructural order in blade coated layers, even though the small molecule phase (*i.e.* C₈-BTBT) remains highly crystalline in all samples.

Next, we fabricate OTFTs *via* blade-coating using pristine and p-doped (0 to 4.25 mol% C₆₀F₄₈) C₈-BTBT:C₁₆IDT-BT blends (Fig. S7, ESI†). Processing and device characterization was performed in nitrogen atmosphere to minimize unintentional side effects. Fig. 5a and b shows representative sets of the transfer and output curves, respectively, for an optimised p-doped C₈-BTBT:C₁₆IDT-BT:C₆₀F₄₈ blend OTFT. The device exhibits significantly improved performance manifested in higher on-currents and a maximum μ_{SAT} value of $14 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ as compared to just over $6 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ for the undoped OTFTs. Fig. 5c shows the statistical distribution of μ_{SAT} measured from 80 pristine and 80 optimally p-doped (0.75 mol% C₆₀F₄₈) OTFTs prepared by blade coating. Evidently, the average mobility for p-doped devices remains significantly higher ($9.5 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$) than the values extracted for the pristine devices ($\approx 4.5 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$).

The large difference in hole mobility values between the pristine and p-doped blend OTFTs in Fig. 5c most likely indicates different transport mechanisms. Numerous high mobility organic semiconductors were shown to exhibit a characteristic increase in charge carrier mobility with reducing temperature (band-like behaviour) and was described by invoking a dynamic disorder model.^{59–61} Nevertheless, for the vast majority of these molecular systems, this transport regime occurs at relatively high temperatures and above a critical temperature (T_C). Below this characteristic temperature, thermal activation sets-in. The cross-over between the two transport regimes is typically understood in terms of charge carrier localization due to the presence of shallow traps, which tend to dominate charge transport at low temperatures. The origin of these trap states can differ for different materials and may relate to structural defects, grain-boundaries and on the presence of chemical impurities.⁶² If the extent of the disorder is reduced (*i.e.* reducing the trap density) the cross-over between band-like and thermally activated transport is shifted to lower temperatures and the activation energy (E_a) reduces.⁶³ In the case of the polycrystalline layers studied here, however, the grain boundaries present between crystalline domains are known to be the source of the structural disorder⁶⁴ and it remains to be seen whether p-type doping plays a role. For single-crystalline C₈-BTBT-based OTFTs where such grain boundaries are largely absent, band-like transport has indeed been observed.⁶⁵

Fig. 5d and e displays the $I_{\text{DS}}^{1/2}$ *vs.* V_{GS} plots measured at different temperatures for pristine and p-doped blend OTFTs processed *via* blade-coating. Interestingly, both types of devices exhibit band-like transport for temperatures above T_C . At $T < T_C$, the hole mobility becomes thermally activated in both pristine and p-doped transistors, but with the two exhibiting different E_a values. Similar trends have been reported for various organic semiconductors and associated with the presence of structural disorder in the active layers.^{62,63,65} A noticeable difference between the two samples is seen both in terms of T_C and E_a values. In particular, pristine OTFTs show a T_C of 210 K while for the p-doped devices the T_C drops to ≈ 140 K, with a simultaneous decrease of the E_a measured below T_C . Specifically, pristine OTFTs exhibit an activation energy of ≈ 52.6 meV, while



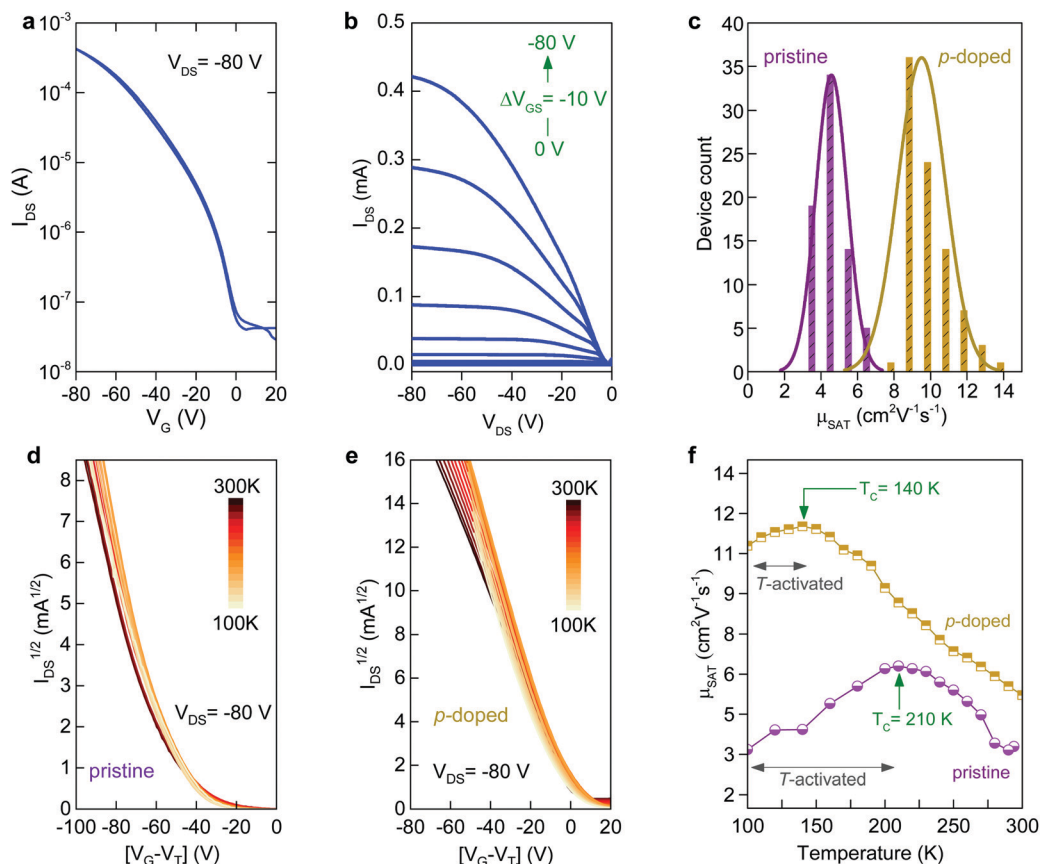


Fig. 5 (a) Transfer and (b) output characteristics of a representative p-doped C₈-BTBT:C₁₆IDT-BT-based OTFT processed *via* blade-coating. (c) Histogram of the hole mobility measured in saturation (μ_{SAT}) for pristine and p-doped blend OTFTs. (d and e) Plots of the square root of the channel current ($I_{\text{DS}}^{0.5}$) versus V_{GS} measured at different temperatures in the range 100–300 K for pristine and p-doped OTFTs prepared by blade-coating. (f) Dependence of μ_{SAT} on temperature for pristine and p-doped blend OTFTs calculated from the data in (d) and (e). The channel width and length for all devices studied were 1000 μm and 80 μm , respectively.

C₆₀F₄₈-doped transistors show a significantly reduced E_{A} of ≈ 29.4 meV. This observation, again, suggests that inclusion of C₆₀F₄₈ at optimal concentrations in the blend leads to reduced energetic disorder and improved charge transport. These observations are in line with the microstructural analysis which shows that the combination of p-doping and blade-coating improves the structural quality of the resulting blend layers, ultimately resulting in the greatly improved hole-transport seen in Fig. 5f.

3. Conclusions

We have studied the influence of molecular p-doping on hole-transport in small-molecule:polymer blend OTFTs processed *via* conventional spin-coating and the industry-relevant technique of blade-coating. OTFTs prepared by blade-coating are found to consistently outperformed those prepared *via* spin-coating yielding a maximum hole mobility of $14 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$. Irrespective of the deposition method, the blend layers exhibit a vertically phase-separated microstructure where the C₈-BTBT segregates atop a C₁₆IDT-BT-dominant phase. Introducing the molecular dopant C₆₀F₄₈ in the semiconducting blend was found to p-dope the formed layers, but also affects the

crystallization kinetics of C₈-BTBT phase leading to enhanced in-plane molecular packing of the C₈-BTBT component, as corroborated by 2D GIWAXS and AFM measurements, ultimately resulting to a higher hole mobility. The work highlights the compatibility of high mobility organic semiconducting blends with high throughput printing techniques relevant to numerous emerging technologies.

4. Experimental

4.1. Material preparation and device fabrication

C₈-BTBT was purchased from 1-Material and used as received. Molecular dopant C₆₀F₄₈ was synthesized from C₆₀ and molecular fluorine, F₂, using a modified method described elsewhere.⁶⁶ Its purity of 98 mol% was confirmed by ¹⁹F NMR, mass spectrometry and high performance liquid chromatography. The solution formulations were prepared using previously reported protocols.¹⁶ In brief, we used a weight ratio of C₈-BTBT to C₁₆IDT-BT of 1:3 with a total solid content of 10 mg per ml⁻¹ of solvent. The solution was prepared in a chlorobenzene and tetralin (1:1) mix solvent. The blend was doped with four different concentration of C₆₀F₄₈ dopant



(0.1, 0.75, 2 and 4.25 mol%). OTFTs were fabricated with a top-gate, bottom-contact (TG-BC) architecture onto $2.54 \times 2.54 \text{ cm}^2$ large glass substrates. The Al/Au (5/35 nm) source and drain electrodes were vapor deposited using shadow mask on the glass substrates and were further treated with the contact modifier Pentafluorothiophenol (PFBT) monolayer. The blend semiconducting layers were deposited using spin-coating and blade-coating. Previously reported experimental conditions were used for the spin-coated films.¹⁷ For blade-coating, the blend solutions were coated on the substrates at 70 °C with the blade speed of 50 mm s^{-1} . The films were then subjected to an additional annealing step at 120 °C for 1 min. A $\approx 900 \text{ nm}$ -thick CYTOP film was deposited on top of the blend semiconductor by spin-coating at 2000 rpm followed by annealing at 50 °C for 1 h in nitrogen atmosphere. The geometrical capacitance of the CYTOP layers was measured to be $\approx 2.1 \text{ nF cm}^{-2}$. Finally, a 70 nm-thick Al layer was vapor deposited atop serving as the gate electrode. The digital photographs of spin-coated and blade-coated devices are shown in Fig. S8 (ESI[†]) along with the device dimensions and source–drain (S–D) electrode distribution. Electrical transistor characterisation was performed using a Keysight B2912A Precision Source/Measure Unit.

4.2. Field-effect charge carrier mobility calculations

The field-effect charge carrier mobility was calculated from the transfer characteristics of the transistors in the saturation regime using.

$$\mu_{\text{SAT}} = \frac{L}{WC_i} \frac{\partial^2 I_{\text{Dsat}}}{\partial V_G^2} \quad (1)$$

In eqn (1), L and W are the transistor channel length and width respectively, V_G is the applied gate voltage, C_i the geometrical capacitance of the gate dielectric employed, and I_{Dsat} the channel current measured in saturation.

4.3. Polarized optical microscopy

Nikon LV-100 microscope was used to perform high speed POM and the images were recorded using a Photron SA-3 CMOS camera.

4.4. Atomic force microscopy

An Agilent 5500 instrument was used to perform all AFM measurements. The scanning speed was set at 0.2 lines s^{-1} . The image processing and the statistical analysis of the images were done using Gwyddion software.

4.5. Transmission electron microscopy

A transmission electron microscope of model Titan G² 80-300 ST from Thermo-Fisher Scientific was used to perform the cross-sectional TEM (X-TEM) analysis of the samples. The TEM instrument is equipped with a spherical aberration corrector for the condenser lens and an energy-filter GIF-Quantum 966 from Gatan Inc. for aberration corrected scanning TEM (AC-STEM) and electron energy loss spectroscopy (EELS), respectively. Further, the microscope was operated at the accelerating voltage of 300 kV and in STEM mode during the

analysis of the samples. The AC-STEM images were generated by collecting the transmitted electron beam with a retractable analogue high angular annular dark-field (HAADF) detector installed above the projection chamber of the microscope. The morphology and thickness of each layer in the stacks was investigated by acquiring the HAADF-STEM images at various magnifications. Whereas, the elemental maps of the main constituents namely, sulphur (S), carbon (C), and oxygen (O) were generated by acquiring the EELS datasets from the samples in a synchronized fashion during the STEM imaging. Furthermore, the elemental maps of S, C, and O were generated by acquiring their S-L_{2,3}, C-K, and O-K energy-loss edges at the energy of 165 eV, 283 eV, and 532 eV, respectively. In this way, the STEM-EELS data-cubes were acquired and processed to generate individual as well as RGB Composite elemental maps.

4.6. Grazing-incidence wide-angle X-ray scattering

Wide-angle X-ray scattering (GIWAXS) measurements were performed at the non-crystalline diffraction beamline (BL11-NCD-Sweet) at ALBA Synchrotron Radiation Facility in Barcelona (Spain). A Rayonix, WAXS LX255-HS detector was used to collect the scattering signals, resolution: 1920×5760 pixels (pixel size = $40 \times 40 \mu\text{m}^2$). Data are expressed as a function of the scattering vector (q) and the angle of incidence was 0.1° . GIWAXS patterns were corrected as a function of the components of the scattering vector with a Matlab script designed for this purpose.⁶⁷ Integrations have been done with Fit2D software.

Conflicts of interest

There are no conflicts to declare.

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