


 Cite this: *RSC Adv.*, 2018, 8, 14597

Received 22nd March 2018

Accepted 11th April 2018

DOI: 10.1039/c8ra02511h

rsc.li/rsc-advances

Trifluoromethylated proline analogues as efficient tools to enhance the hydrophobicity and to promote passive diffusion transport of the L-prolyl-L-leucyl glycineamide (PLG) tripeptide†

 Martin Oliver,^a Charlene Gadais,^a Júlia García-Pindado,^b Meritxell Teixidó,^b Nathalie Lensen,^{id}*^a Grégory Chaume^{ib}*^a and Thierry Brigaud^{ib}^a

The synthesis of four CF₃-proline analogues of the PLG peptide is reported. Our results show that the incorporation of trifluoromethylated amino acids (Tfm-AA) at the N-terminal position of a peptide significantly increases its hydrophobicity. In addition, depending on the relative configuration and the position of the CF₃ group, Tfm-AA can also promote passive diffusion transport.

Introduction

The use of peptides as therapeutic agents are gaining interest and the number of peptides entering into clinical trials is in steady increase.¹ Nonetheless their development is hampered by the lack of metabolic stability, poor transport properties, low oral bioavailability as well as poor target specificity. Therefore, efforts are still needed to circumvent these drawbacks and the incorporation of constrained unnatural amino acids into the peptide chains is one of the common strategies to improve their biological profile. In this context, fluorinated amino acids are very attractive compounds.² The introduction of fluorine atoms and fluoroalkyl groups into peptides and peptidomimetics presents the following advantages: (i) it can greatly increase the local hydrophobicity,³ it also modulates the peptide conformation and assembly,⁴ (ii) it confers a better resistance to proteolysis⁵ and (iii) fluorine atoms can be used as highly sensitive labels for ¹⁹F-NMR/MRI/MRS spectroscopy, because of ¹⁹F 100% natural abundance and the absence of background signals in natural and biological media.⁶ Unfortunately, the use of fluorinated peptides remains limited, especially for amino acids bearing a trifluoromethyl group adjacent to the amino group such as trifluoromethylated amino acids (Tfm-AA). Indeed, only few synthetic methods are available for their synthesis in enantiopure form resulting in a small panel of fluorinated amino acids. Moreover, their incorporation into peptides is still

challenging due to the stereoelectronic effects imparted by the CF₃ group which strongly decrease the nitrogen atom nucleophilicity⁷ and limits the coupling reactions at their N-termini.

Our group develops for several years efficient methods for the synthesis of Tfm-AA⁸ and their incorporation into peptides.⁹ We have reported that, unlike the non-fluorinated series, the coupling reactions at the C-termini of Tfm-AA can be achieved with unprotected amino group because of its strong deactivation by the electron-withdrawing effect of the neighbouring CF₃ group.^{9a} In contrast, the decrease of the nucleophilicity of the amino group together with the steric hindrance of the CF₃ group significantly impedes the coupling reaction of Tfm-AA at their N-terminus and specific activations such as mixed anhydrides or amino acid chlorides have to be used.^{9b,c,10} We applied our methodologies to the synthesis of several Tfm-AA containing peptides in order to investigate their specific physicochemical and/or biological properties.¹¹

Among the targeted peptides, we are interested in the endogenous brain peptide L-prolyl-L-leucyl glycineamide 1 (PLG). Because of its structural simplicity, and its relevance in a wide range of pharmacological activities in the central nervous system (CNS),¹² several conformationally constrained analogues of PLG have been reported in literature.¹³ From our side, we have previously described the synthesis of the (S,S)-2 of PLG. *In vivo* biological activity studies on rat models revealed that (S,S)-2 displays, after intraperitoneal injection, higher analgesic effects in paw pressure test during acute pain and superior anti-opioid effects in stress-induced analgesia compared to PLG 1 (Fig. 1).^{11a,b} Because PLG 1 is known to cross the blood-brain barrier (BBB),¹⁴ we hypothesized that the superior analgesic effect of (S,S)-2 could be probably due to the presence of the CF₃ group. Indeed, the significant decrease of the proline basicity⁷ and its low protonation ability should lead to an increase in the

^aLaboratoire de Chimie Biologique (LCB), Université de Cergy-Pontoise, 5 mail Gay-Lussac, Neuville-sur-Oise, 95031 Cergy-Pontoise, France. E-mail: nathalie.lensen@u-cergy.fr; gregory.chaume@u-cergy.fr

^bInstitute for Research in Biomedicine (IRB Barcelona), Barcelona Institute of Science and Technology (BIST), C/ Baldri Reixac 10, 08028 Barcelona, Spain

† Electronic supplementary information (ESI) available: Extended experimental details and analytical data; copies of the NMR spectra. See DOI: 10.1039/c8ra02511h



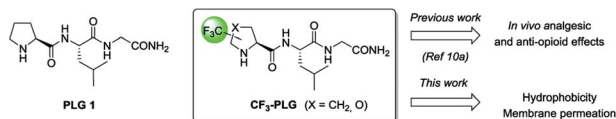


Fig. 1 Chemical structure of PLG and CF₃-PLG analogues.

hydrophobicity parameter enabling the fluorinated PLG analogue (*S,S*)-2 to better cross the BBB and enter the CNS. In order to confirm our hypothesis, we report here the synthesis of four CF₃-proline analogues of PLG. Their hydrophobicity as well as their ability to cross the BBB were assessed and compared to PLG 1.

Results and discussion

Synthesis of the CF₃-proline analogues of PLG

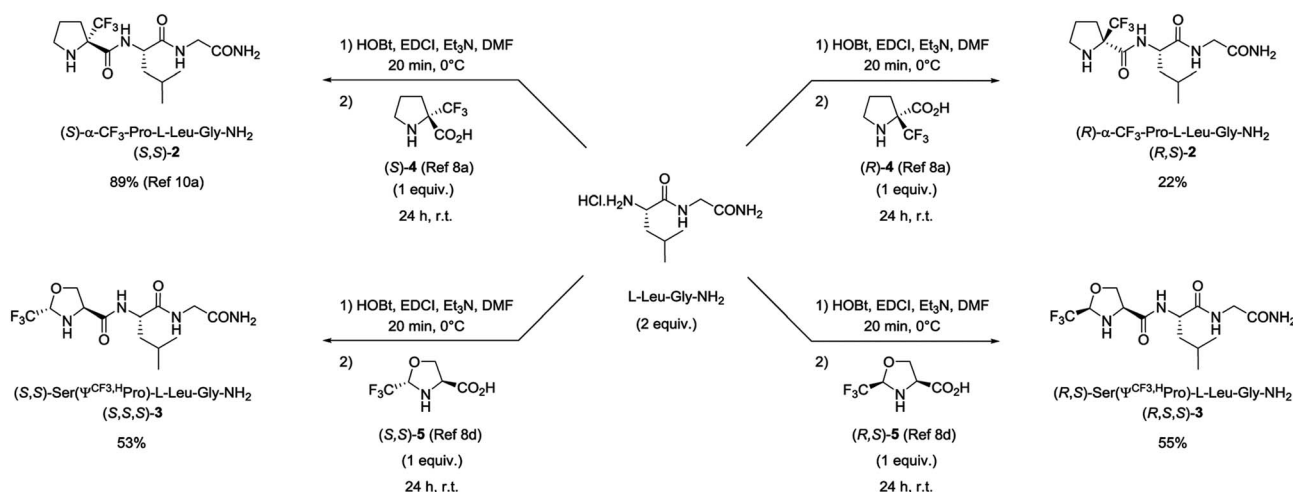
Four fluorinated analogues of PLG, corresponding to two pairs of diastereomers, have been considered to evaluate the effect of the CF₃ group in α position of the N-terminal amino group on hydrophobicity and membrane permeation. The collected data should also provide valuable information related to the influence of the configuration of the chiral center bearing the CF₃ group and its position along the 5-membered ring. In addition to the (*S*)- α -CF₃-proline (*S*)-4 used for the synthesis of peptide (*S,S*)-2, three other fluorinated amino acids, namely the (*R*)- α -CF₃-proline (*R*)-4, the (*S,S*)-CF₃-pseudoproline (*S,S*)-5 and the (*R,S*)-CF₃-pseudoproline (*R,S*)-5, were selected to substitute the proline residue into the PLG sequence. The preparation of these trifluoromethylated amino acids in enantiopure form has been previously reported by our group.^{8a,d}

The synthesis of the corresponding peptides [(*R,S*)-2, (*S,S*)-3 and (*R,S,S*)-3] has been performed following the protocol described for the synthesis of (*S,S*)-2.^{11a} As mentioned before, the incorporation of the fluorinated proline surrogates into the peptide chain does not require preliminary protection of the amino group. Nevertheless, the reverse addition of the unprotected CF₃-amino acids to the preformed hydrochloride salt of

(*S*)-Leu-Gly-NH₂ dipeptide using the classical 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDCI) and 1-hydroxybenzotriazole (HOBt) coupling reagents is required to prevent the formation of diketopiperazine side products. While the synthesis of the (*S*)- α -CF₃-proline containing peptide (*S,S*)-2 occurred in very good yield (89%), the coupling reaction starting from (*R*)- α -CF₃-proline (*R*)-4 gave the corresponding peptide (*R,S*)-2 in low yield (22%) (Scheme 1). This result may be explained by a mismatched double stereo-differentiation between the (*R*)- α -CF₃-proline residue and the (*S*)-Leu-Gly-NH₂ dipeptide. The peptide 3 diastereomers [(*S,S,S*)-3 and (*R,S,S*)-3] were obtained by coupling (*S,S*)- and (*R,S*)-CF₃-pseudoproline 5 with the (*S*)-Leu-Gly-NH₂ dipeptide (53% and 55% respectively).

Measurement of the isocratic hydrophobicity index ϕ_0

The hydrophobicity of the fluorinated analogues of PLG has then been assessed and compared to the native PLG 1 in order to investigate the specific effect imparted by the CF₃ group. For this purpose, we applied a chromatographic method, first developed by Valko *et al.*,¹⁵ which proved to be very convenient for evaluating the hydrophobicity index of fluorinated peptides.¹⁶ It is based on the measurement of the isocratic hydrophobicity index (ϕ_0) which is defined as the percentage (by volume) of organic solvent required to achieve an equal distribution of compound between the polar mobile and apolar stationary phases. Thus, higher is the ϕ_0 value, more hydrophobic is the compound. The method involves the measurement of the logarithm of retention factor k ($\log k$) values by reversed-phase high-performance liquid chromatography (RP-HPLC) using various organic-phase concentrations in the mobile phase (in our case, the acetonitrile is used). By plotting the $\log k$ values as a function of the acetonitrile concentration, the ϕ_0 value can be obtained from the slope and the intercept of the linear regression ($\phi_0 = -\text{intercept/slope}$). A good correlation between the ϕ_0 and the $\log P$ values was shown.¹⁵ The hydrophobicity index of our peptides was first measured at pH 7. The fluorinated PLG analogues 2 and 3 (entries 2–5, Table 1) display



Scheme 1 Synthesis of CF₃-proline and CF₃-pseudoproline containing PLG analogues.



Table 1 φ_0 values for PLG 1 and its fluorinated analogues 2 and 3

Entry	Compound	φ_0^a	
		pH 2	pH 7
1	1	9,2	12,3
2	(S,S)-2	19,9	31,3
3	(R,S)-2	17,9	31,8
4	(S,S,S)-3	25,3	27,5
5	(R,S,S)-3	30,4	30,8

^a Parameter definition and its calculation is provided in Experimental section.

similar φ_0 values (*ca.* 30) which are significantly higher (by a factor ~ 2.5) compared to the native PLG 1 (entry 1, Table 1). These results show that the incorporation of a CF_3 group in α position of the N-terminal amino group enhances the hydrophobic character of the corresponding peptide. However, the configuration of the stereocenter bearing the CF_3 group as well as its position along the pyrrolidine ring does not seem to play a significant role (entries 2–5, Table 1). The φ_0 values were then measured at pH 2 to investigate the impact of the possible ionic form of the PLG peptides 1–3 on the hydrophobicity parameter. The lower φ_0 value observed for the native PLG 1 is consistent with its expected polar ionic state (entry 1, Table 1). Surprisingly, the fluorinated PLG analogues 2 and 3 display a different behaviour at pH 2. The PLG analogues 2 bearing an α - CF_3 -proline exhibit significant decrease of their hydrophobicity index (entries 2 and 3, Table 1), the (R,S)-2 derivative being more sensitive than its diastereomer (S,S)-2. These results suggest that the protonation of the deactivated N-terminal amino group occurs at pH 2, increasing the polarity of the corresponding ionized peptides. In contrast, the pseudoproline containing derivatives 3 do not seem to be protonated at pH 2. Only a slight decrease in the φ_0 value is observed for the (S,S,S)-3 (entry 4, Table 1), that of the (R,S,S)-3 diastereomer remaining almost unchanged (entry 5, Table 1). Nevertheless, the incorporation of a Tfm-AA residue at the N-terminal position of the PLG allows to significantly enhance the hydrophobic character.

Evaluation of the passive diffusion transport by parallel artificial membrane permeability assay (PAMPA)

We then decided to evaluate the ability of our fluorinated PLG analogues 2 and 3 to cross the BBB through passive diffusion.

Passive diffusion is a nonsaturable and spontaneous transport process that depends on physicochemical properties such as hydrophobicity, hydrogen bonding, pK_a or molecular weight. Therefore, such evaluation should allow us to assess the relevance of the CF_3 group in the passive diffusion transport and thus, demonstrate the benefit in incorporating Tfm-AAs at the N-terminus of peptides. For this purpose, we used PAMPA method, first introduced by Kansy *et al.*,¹⁷ and selected a mixture of porcine brain polar lipid extract as *in vitro* model of the BBB.¹⁸ It consists in measuring the ratio of peptide that has been transferred from a donor well into an acceptor well through the artificial membrane (see Experimental part). The rate of permeation can be determined by the effective permeability (P_e) as well as the rate of transport values. The membrane retention value can also be obtained and reflects the strength of the interaction between the lipids that mimic the membrane and the peptide. While PLG 1 is known to cross the BBB,¹³ its permeability was found to be very negligible and no peptide was retained in the membrane (entry 1, Table 2). Thus, PLG 1 seems to not display excellent physicochemical properties to transport through passive diffusion. This result correlates with the fact that PLG 1 has been reported to enter brain through a partially saturable transport system.¹⁴ PLG analogues 2, bearing the α - CF_3 -proline residue, show an ability to cross the membrane (entries 2 and 3, Table 2). According to the empirical BBB permeation prediction,¹⁹ the peptide (R,S)-2 displays a poor effective permeability ($P_e < 2.0 \times 10^6 \text{ cm s}^{-1}$) while its diastereomer (S,S)-2 exhibits a moderate transport properties ($2.0 \times 10^6 < P_e < 4.0 \times 10^6 \text{ cm s}^{-1}$). Both peptides (S,S)-2 and (R,S)-2 were also found to have a strong interaction with the membrane. It is worth noting the relevant difference in terms of permeability due to the change of only one stereocenter in the molecules. On the contrary, none of the CF_3 -pseudoproline containing PLG analogues 3 reveal an ability to cross the membrane by passive diffusion nor a membrane affinity (entries 4 and 5, Table 2). The PAMPA assays have been performed at pH 7.4, a pH for which fluorinated PLG analogues 2 and 3 display similar hydrophobicity index values (see Table 1). Therefore, the significant difference between the peptides 2 and 3 suggests that the nature of the 5-membered ring (pyrrolidine or oxazolidine) and the position along the ring as well as the absolute configuration of the carbon bearing the CF_3 group play also a key role in the passive diffusion transport.

Table 2 Effective permeability (P_e), percentage of transport and membrane retention after 4 h in the PAMPA of PLG 1 and its fluorinated analogues 2 and 3^a

Entry	Compound	$P_e (\times 10^6)^b \text{ cm s}^{-1}$	Transport (%) (4 h) ^b	Membrane retention ^b
1	1	nd ^c	nd ^c	nd ^c
2	(S,S)-2	2.68 ± 0.04	5.34 ± 0.08	50%
3	(R,S)-2	0.31 ± 0.03	0.65 ± 0.06	30%
4	(S,S,S)-3	nd ^c	nd ^c	nd ^c
5	(R,S,S)-3	nd ^c	nd ^c	nd ^c

^a Data are expressed as the mean $\pm$ SD. ^b Parameters definitions and their calculations are provided in Experimental section. ^c Not determined.



Conclusions

We demonstrate that the incorporation of a Tfm-AA residue at the N-terminal position of a peptide can be very useful to enhance its hydrophobic character. Moreover, depending on its position along the 5-membered pyrrolidine ring, the CF₃ group may promote passive diffusion transport. Thus, Tfm-AA residues could serve as potent BBB-shuttle to transport into the CNS compounds of interests that cannot cross the BBB. The use of peptide as BBB-shuttles is one of the most promising approaches to deliver safely drugs to the brain.²⁰

Experimental section

Synthesis of the CF₃-Proline analogues of PLG

Representative procedure for the peptide coupling reaction.

To a solution of L-Leu-Gly-NH₂ hydrochloride salt (2 equiv.) in DMF were successively added at 0 °C NEt₃ (4.1 equiv.), HOBT (1.5 equiv.) and EDCI (1.5 equiv.). The reaction mixture was stirred for 20 min at 0 °C and the CF₃-proline surrogate (*R*)-4, (*R,S*)-5 or (*S,S*)-5 (1 equiv.) was added. After 20 min at 0 °C, the resulting solution was stirred overnight at room temperature and then diluted with CH₂Cl₂ and water. The aqueous layer was extracted with CH₂Cl₂ (3×) and the combined chlorinated extracts were washed with water, dried with MgSO₄, filtered and concentrated under reduced pressure. Purification by flash chromatography gave fluorinated peptides (*R,S*)-2, (*S,S,S*)-3 and (*R,S,S*)-3 in 22–53% yield.

(*R*)-α-CF₃-Pro-L-Leu-Gly-NH₂ (*R,S*)-2. The tripeptide (*R,S*)-2 was prepared according to the representative procedure, with HCl.L-Leu-Gly-NH₂ (610 mg, 2.73 mmol, 2 equiv.), NEt₃ (844 μL, 5.60 mmol, 4.1 equiv.), HOBT (276 mg, 2.05 mmol, 1.5 equiv.), EDCI (391 mg, 2.05 mmol, 1.5 equiv.), and (*R*)-α-CF₃-proline (*R*)-4 (250 mg, 1.37 mmol, 1 equiv.) in DMF (6 mL). Purification on silica gel (CH₂Cl₂/MeOH, 90 : 10) gave pure (*R,S*)-2 (110 mg, 22%) as colorless oil. *R*_f = 0.33 (CH₂Cl₂/MeOH, 90 : 10); [α]_D +20.1 (c 1.4 in MeOH). IR (neat): 3272, 2955, 1651, 1651, 1519, 1386, 1250 cm⁻¹; ¹H NMR (400 MHz, CD₃OD) δ 0.93 (d, *J* = 6.4 Hz, 3H, H_δ Leu-H), 0.96 (d, *J* = 6.4 Hz, 3H, H_δ Leu-H), 1.58–1.72 (m, 3H, H_β Leu-H, H_γ Leu-H), 1.73–1.90 (m, 2H, H_γ Pro-H), 2.12–2.28 (m, 2H, H_β Pro), 3.02–3.14 (m, 2H, H_δ Pro), 3.78 (d, *J* = 17.0 Hz, 1H, H_α Gly-Ha), 3.91 (d, *J* = 17.0 Hz, 1H, H_α Gly-Hb), 4.41 (m, 1H, H_α Leu-H); ¹³C NMR (100.5 MHz, CD₃OD) δ 22.0 (CH₃, C_δ Leu), 23.4 (CH₃, C_δ Leu), 25.9 (CH, C_γ Leu), 26.1 (CH₂, C_γ Pro), 33.2 (CH₂, C_β Pro), 41.8 (CH₂, C_β Leu), 43.0 (CH₂, C_α Gly), 48.2 (CH₂, C_δ Pro), 53.5 (CH, C_α Leu), 72.2 (q, *J* = 25.9 Hz, C, C_α Pro), 127.4 (q, *J* = 283.7 Hz, C, CF₃), 172.1 (C, C=O), 173.9 (C, C=O), 174.6 (C, C=O); ¹⁹F NMR (376.2 MHz, CD₃OD): δ –78.9 (s, CF₃); HRMS (ESI-TOF) calcd. for C₁₄H₂₄N₄O₃F₃ [M + H]⁺ 353.1801, found 353.1801.

(2*R,4S*)-Ser(Ψ^{CF₃,H}Pro)-L-Leu-Gly-NH₂ (*R,S,S*)-3. The tripeptide (*R,S,S*)-3 was prepared according to the representative procedure, with HCl.L-Leu-Gly-NH₂ (364 mg, 1.63 mmol, 1.5 equiv.), NEt₃ (690 μL, 4.34 mmol, 4 equiv.), HOBT (220 mg, 1.63 mmol, 1.5 equiv.), EDCI (311 mg, 1.63 mmol, 1.5 equiv.) and (2*R,4S*)-Ser(Ψ^{CF₃,H}Pro) (*R,S*)-5 (200 mg, 1.08 mmol, 1 equiv.) in DMF (6 mL). Purification on silica gel (CH₂Cl₂/MeOH, 90 : 10)

gave pure (*R,S,S*)-3 (212 mg, 55%) as hygroscopic white solid. *R*_f = 0.34 (CH₂Cl₂/MeOH, 90 : 10); [α]_D –12.2 (c 1.0 in MeOH); IR (neat): 3326, 2961, 2470, 1638, 1524, 1454, 1288 cm⁻¹; ¹H NMR (400 MHz, CD₃OD) δ 0.93 (d, *J* = 6.4 Hz, 3H, H_δ Leu-H), 0.96 (d, *J* = 6.0 Hz, 3H, H_δ Leu-H), 1.58–1.70 (m, 3H, H_β Leu-H and H_γ Leu-H), 3.77 (d, *J* = 16.9 Hz, 1H, H_α Gly-Ha), 3.90 (d, *J* = 16.9 Hz, 1H, H_α Gly-Hb), 4.11–4.21 (m, 3H, H_α Ψpro-H and H_β Ψpro-Ha), 4.37 (t, *J* = 6.5 Hz, 1H, H_α Leu-H), 5.10 (d, *J* = 6.0 Hz, 1H, H_δ Ψpro-H); ¹³C NMR (100.5 MHz, CD₃OD) δ 21.7 (CH₃, C_δ Leu), 23.5 (CH₃, C_δ Leu), 25.5 (CH, C_γ Leu), 42.0 (CH₂, C_β Leu), 43.1 (CH₂, C_α Gly), 53.1 (CH, C_α Leu), 60.8 (CH, C_α Ψpro), 70.6 (CH₂, C_β Ψpro), 89.1 (q, *J* = 33.6 Hz, CH, C_δ Ψpro), 125.1 (q, *J* = 282.8 Hz, C, CF₃), 174.2 (C, C=O), 174.6 (C, C=O), 175.0 (C, C=O); ¹⁹F NMR (376.2 MHz, CD₃OD) δ –85.3 (bs, CF₃); HRMS (ESI-TOF) calcd. for C₁₃H₂₂N₄O₄F₃ [M + H]⁺ 355.1593, found 355.1601.

(2*S,4S*)-Ser(Ψ^{CF₃,H}Pro)-L-Leu-Gly-NH₂ (*S,S,S*)-3. The tripeptide (*S,S,S*)-3 was prepared according to the representative procedure, with HCl.L-Leu-Gly-NH₂ (1.11 g, 4.63 mmol, 2 equiv.), NEt₃ (1.5 mL, 9.50 mmol, 4.1 equiv.), HOBT (469 mg, 3.47 mmol, 1.5 equiv.), EDCI (664 mg, 3.47 mmol, 1.5 equiv.), and (2*S,4S*)-Ser(Ψ^{CF₃,H}Pro) (*S,S*)-5 (426 mg, 2.32 mmol, 1 equiv.) in DMF (10 mL). Purification on silica gel (CH₂Cl₂/MeOH, 90 : 10) gave pure (*S,S,S*)-3 (434 mg, 53%) as white solid. *R*_f = 0.34 (CH₂Cl₂/MeOH, 90 : 10); mp 150 °C; [α]_D –41.8 (c 0.9 in MeOH); IR (neat): 3301, 2419, 1782, 1558, 1435, 1289 cm⁻¹; ¹H NMR (400 MHz, CD₃OD) δ 0.95 (d, *J* = 6.4 Hz, 3H, H_δ Leu-H), 0.99 (d, *J* = 6.4 Hz, 3H, H_δ Leu-H), 1.58–1.64 (m, 2H, H_β Leu-H), 1.68 (m, 1H, H_γ Leu-H), 3.64 (t, *J* = 7.3 Hz, 1H, H_β Ψpro-Ha), 3.74 (d, *J* = 17.4 Hz, 1H, H_α Gly-Ha), 3.92 (d, *J* = 17.4 Hz, 1H, H_α Gly-Hb), 3.98 (t, *J* = 7.3 Hz, 1H, H_α Ψpro-H), 4.21 (t, *J* = 7.3 Hz, 1H, H_β Ψpro-Hb), 4.35 (t, *J* = 6.9 Hz, 1H, H_α Leu-H), 5.14 (q, *J* = 5.5 Hz, 1H, H_δ Ψpro-H); ¹³C NMR (100.5 MHz, CD₃OD) δ 22.0 (CH₃, C_δ Leu), 23.3 (CH₃, C_δ Leu), 25.9 (CH, C_γ Leu), 41.4 (CH₂, C_β Leu), 43.1 (CH₂, C_α Gly), 54.0 (CH, C_α Leu), 60.2 (CH, C_α Ψpro), 71.4 (CH₂, C_β Ψpro), 89.4 (q, *J* = 33.6 Hz, CH, C_δ Ψpro), 125.0 (q, *J* = 282.8 Hz, C, CF₃), 172.3 (C, C=O), 174.2 (C, C=O), 175.0 (C, C=O); ¹⁹F NMR (376.2 MHz, CD₃OD) δ –86.1 (d, *J* = 5.5 Hz, CF₃); HRMS (ESI-TOF) calcd. for C₁₃H₂₂N₄O₄F₃ [M + H]⁺ 355.1593, found 355.1587.

Isocratic hydrophobicity index φ₀

An Agilent 1200 series high-performance liquid chromatograph coupled to an ELSD detector was used. The reversed-phase HPLC measurements were carried out on a Zorbax® RX-C18 analytical column (4.6 × 250 mm, 5 μm). The mobile phase was 50 mM ammonium acetate and HPLC grade acetonitrile as organic modifier (pH 7.0) or TFA 0.1% buffer aqueous solution and 0.1% TFA HPLC grade acetonitrile (pH 2). The mobile phase flow rate was ranging from 0.1–1.0 mL min⁻¹. The dead time (*t*₀) was measured by injecting sodium nitrate together with the sample at a concentration of 10 mM. The samples were dissolved at 1 mg/1 mL in water. Few drops of acetonitrile were added when the dissolution of the peptide was incomplete. The logarithm of retention factor *k* (log *k*) values were obtained according to the eqn (1):



$$\log k = \log \left(\frac{t_R - t_0}{t_0} \right) \quad (1)$$

where t_R is the retention time of the compound and t_0 the retention time of the unretained sodium nitrate solute.

The $\log k$ values were plotted as a function of the acetonitrile concentration. The slope (S) and the intercept ($\log k_w$) values were calculated from at least three concentrations of acetonitrile. The correlation coefficients of the linear fit were higher than 0.99. The isocratic hydrophobicity index φ_0 was calculated according to the eqn (2):

$$\varphi_0 = -\log k_w/S \quad (2)$$

where $\log k_w$ is the $\log k$ value extrapolated to the 0% acetonitrile concentration.

Parallel artificial membrane permeability assay (PAMPA)

The PAMPA was used to measure the capacity of the peptides 1–3 to cross the BBB by passive diffusion. The effective permeability (P_e) of the compounds was determined at an initial concentration of 5 mM. The buffer solution was prepared from a commercially concentrated one, supplied by PION. Following the manufacturer's instructions, system solution was adjusted to pH 7.4 using a 0.5 M NaOH solution. The compounds to be analyzed were dissolved in a mixture of buffer solution with 20% of 1-propanol used as co-solvent to ensure their solubility. The PAMPA sandwich was separated, and the donor wells were filled with 195 μ L of each compound. A magnetic stirrer was placed in each well. The acceptor plate was put into the donor plate, ensuring that the underside of the membrane was in contact with the buffer. Then, 4 μ L of a mixture of phospholipids from a porcine polar brain extract (20 mg mL⁻¹) in dodecane (composition: 12.6% phosphatidylcholine (PC), 33.1% phosphatidylethanolamine (PE), 18.5% phosphatidylserine (PS), 4.1% phosphatidylinositol (PI), 0.8% phosphatidic acid and 30.9% of other compounds) was added to the filter of each well, followed by 200 μ L of buffer solution. The plate was then covered and incubated at room temperature in a saturated humidity atmosphere for 4 h under orbital agitation at 25 μ m of unstirred water layer (UWL). After this period, the solution from the donor and acceptor plates was transferred to the UPLC vials. 100 μ L of each acceptor and donor samples and 100 μ L of t_0 samples were injected into the UPLC apparatus. Acceptor samples were analyzed by MALDI-TOF in order to study the presence of not degraded peptides and confirm their crossing. The effective permeability (P_e) after 4 h was calculated using eqn (3) and the percentage of transport with eqn (4):

$$P_e = \frac{-218.3}{t} \log \left[1 - \frac{C_A(t)}{C_D(t_0)} \right] 10^{-6} \text{ cm s}^{-1} \quad (3)$$

$$\text{Transport (\%)} = \frac{C_A(t)}{C_D(t_0)} \times 100 \quad (4)$$

where t is time (h), $C_A(t)$ is the peptide concentration in the acceptor well at time t , and $C_D(t_0)$ is the peptide concentration in the donor well at 0 h.

Membrane retention was also calculated as the difference of product at t_0 and after 4 h in the donor and acceptor compartments.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

French partner thanks the French Fluorine Network.

References

- (a) S. Ghosh, *Chem. Today*, 2016, **34**, 5; (b) K. Fosgerau and T. Hoffmann, *Drug Discovery Today*, 2015, **20**, 122; (c) T. Uhlig, T. Kyprianou, F. G. Martinelli, C. A. Oppici, D. Heiligers, D. Hills, X. R. Calvo and P. Verhaert, *EuPa Open Proteomics*, 2014, **4**, 1.
- (a) R. Smits, C. D. Cadicamo, K. Burger and B. Kokscho, *Chem. Soc. Rev.*, 2008, **37**, 1727; (b) X.-L. Qiu and F.-L. Qing, *Eur. J. Org. Chem.*, 2011, 3261; (c) J. L. Aceña, A. E. Sorochinsky and V. A. Soloshonok, *Synthesis*, 2012, **44**, 1591.
- (a) M. Salwiczek, E. K. Nyakatura, U. I. M. Gerling, S. Ye and B. Kokscho, *Chem. Soc. Rev.*, 2012, **41**, 2135; (b) V. H. Dalvi and P. J. Rossky, *Proc. Natl. Acad. Sci. U. S. A.*, 2010, **107**, 13603; (c) Q. A. Huchet, B. Kuhn, B. Wagner, H. Fischer, M. Kansy, D. Zimmerli, E. M. Carreira and K. Müller, *J. Fluorine Chem.*, 2013, **152**, 119; (d) J. R. Robalo, S. Huhmann, B. Kokscho and A. V. Verde, *Chem*, 2017, **3**, 881–897.
- (a) U. I. M. Gerling, M. Salwiczek, C. D. Cadicamo, H. Erdbrink, C. Czekelius, S. L. Grage, P. Wadhwani, A. S. Ulrich, M. Behrends, G. Haufe and B. Kokscho, *Chem. Sci.*, 2014, **5**, 819; (b) S. Huhmann, E. K. Nyakatura, H. Erdbrink, U. I. M. Gerling, C. Czekelius and B. Kokscho, *J. Fluorine Chem.*, 2015, **175**, 32.
- (a) B. Kokscho, N. Sewald, H.-J. Hofmann, K. Burger and H.-D. Jakubke, *J. Pept. Sci.*, 1997, **3**, 157; (b) V. Asante, J. Mortier, G. Wolber and B. Kokscho, *Amino Acids*, 2014, **46**, 2733.
- For review on the use of ¹⁹F NMR, see: (a) E. N. G. Marsh and Y. Suzuki, *ACS Chem. Biol.*, 2014, **9**, 1242; (b) A. Vulpetti and C. Dalvit, *Drug Discovery Today*, 2012, **17**, 890; (c) H. Chen, S. Viel, F. Ziarelli and L. Peng, *Chem. Soc. Rev.*, 2013, **42**, 7971.
- M. Schlosser, *Angew. Chem., Int. Ed.*, 1998, **37**, 1496.
- (a) G. Chaume, M.-C. Van Severen, S. Marinkovic and T. Brigaud, *Org. Lett.*, 2006, **8**, 6123; (b) G. Chaume, M.-C. Van Severen, L. Ricard and T. Brigaud, *J. Fluorine Chem.*, 2008, **129**, 1104; (c) C. Caupène, G. Chaume, L. Ricard and T. Brigaud, *Org. Lett.*, 2009, **11**, 209; (d) G. Chaume, O. Barbeau, P. Lesot and T. Brigaud, *J. Org. Chem.*, 2010, **75**, 4135; (e) J. Simon, T. T. Nguyen, E. Chelain, N. Lensen, J. Pytkowicz, G. Chaume and T. Brigaud, *Tetrahedron: Asymmetry*, 2011, **22**, 309; (f) N. Lensen, J. Marais and T. Brigaud, *Org. Lett.*, 2015, **17**, 342; (g) H. Lubin, J. Pytkowicz, G. Chaume, G. Sizun-Thome and T. Brigaud, *J. Org. Chem.*, 2015, **80**, 2700.



- 9 (a) G. Chaume, N. Lensen, C. Caupène and T. Brigaud, *Eur. J. Org. Chem.*, 2009, 5717; (b) G. Chaume, J. Simon, C. Caupène, N. Lensen, E. Miclet and T. Brigaud, *J. Org. Chem.*, 2013, **78**, 10144; (c) E. Devillers, J. Pytkowicz, E. Chelain and T. Brigaud, *Amino Acids*, 2016, **48**, 1457.
- 10 B. Kokschi, P. J. L. M. Quaedflieg, T. Michel, K. Burger, Q. B. Broxterman and H. E. Schoemaker, *Tetrahedron: Asymmetry*, 2004, **15**, 1401.
- 11 (a) I. Jlalía, N. Lensen, G. Chaume, E. Dzhambazova, L. Astasidi, R. Hadjiolova, A. Bocheva and T. Brigaud, *Eur. J. Med. Chem.*, 2013, **62**, 122; (b) A. Bocheva, H. Nocheva, I. Jlalía, N. Lensen, G. Chaume and T. Brigaud, *Med. Chem.*, 2013, **3**, 206; (c) A. Botz, V. Gasparik, E. Devillers, A. R. F. Hoffmann, L. Caillon, E. Chelain, O. Lequin, T. Brigaud and L. Khemtémourian, *Biopolymers*, 2015, **104**, 601; (d) J. Simon, J. Pytkowicz, N. Lensen, G. Chaume and T. Brigaud, *J. Org. Chem.*, 2016, **81**, 5381; (e) S. L. Grage, S. Kara, A. Bordessa, V. Doan, F. Rizzolo, M. Putzu, T. Kubař, A. M. Papini, G. Chaume, T. Brigaud, S. Afonin and A. S. Ulrich, *Chem. - Eur. J.*, 2018, **24**, 4328.
- 12 (a) R. K. Mishra, S. Chiu, P. Chiu and C. P. Mishra, *Methods Find. Exp. Clin. Pharmacol.*, 1983, **5**, 203; (b) G. W. Reed, G. A. Olson and R. D. Olson, *Neurosci. Biobehav. Rev.*, 1994, **18**, 519; (c) W. Pan and A. J. Kastin, *Peptides*, 2007, **28**, 2411.
- 13 (a) S. Bhagwanth, R. K. Mishra and R. L. Johnson, *Beilstein J. Org. Chem.*, 2013, **9**, 204 and references cited therein; (b) J. Ferreira da Costa, O. Caamaño, F. Fernández, X. García-Mera, I. E. Sampaio-Dias, J. M. Brea and M. I. Cadavid, *Eur. J. Med. Chem.*, 2013, **69**, 146; (c) I. E. Sampaio-Dias, C. A. D. Sousa, X. García-Mera, J. Ferreira da Costa, O. Caamaño and J. E. Rodríguez-Borges, *Org. Biomol. Chem.*, 2016, **14**, 11065.
- 14 L. Di, E. H. Kerns, K. Fan, O. J. McConnell and G. T. Carter, *Eur. J. Med. Chem.*, 2003, **38**, 223.
- 15 K. Valko and P. Slegel, *J. Chromatogr.*, 1993, **631**, 49.
- 16 C. Gadais, E. Devillers, V. Gasparik, E. Chelain, J. Pytkowicz and T. Brigaud, *ChemBioChem*, 2018, DOI: 10.1002/cbic.201800088.
- 17 M. Kansy, F. Senner and K. Gubernator, *J. Med. Chem.*, 1998, **41**, 1007.
- 18 W. A. Banks and A. J. Kastin, *Peptides*, 1994, **15**, 23.
- 19 P. Crivori, G. Cruciani, P. Carrupt and B. Testa, *J. Med. Chem.*, 2000, **43**, 2204.
- 20 B. Oller-Salvia, M. Sánchez-Navarro, E. Giralt and M. Teixidó, *Chem. Soc. Rev.*, 2016, **45**, 4690.

